

# NIH Public Access

**Author Manuscript** 

*Chemistry*. Author manuscript; available in PMC 2014 May 22.

# Published in final edited form as:

Chemistry. 2014 April 14; 20(16): 4554-4558. doi:10.1002/chem.201400307.

# One-Pot Protection-Glycosylation Reactions for Synthesis of Lipid II Analogues

# Katsuhiko Mitachi, Priya Mohan, Shajila Siricilla, and Michio Kurosu

Department of Pharmaceutical Sciences, College of Pharmacy, University of Tennessee Health Science Center, 881 Madison Avenue, Memphis, TN 38163-0001 (USA)

Michio Kurosu: mkurosu@uthsc.edu

# Abstract

(2,6-Dichloro-4-methoxyphenyl)(2,4-dichlorophenyl) methyl trichloroacetimidate (**3**) and its polymer-supported reagent **4** can be successfully applied to a one-pot protection-glycosylation reaction to form the disaccharide derivative **7d** for the synthesis of lipid II analogues. The temporary protecting group or linker at the C-6 position and *N*-Troc protecting group of **7d** can be cleaved simoultaneousely *via* a reductive condition. Overall yield of syntheses of lipid II, **1** and neryl-lipid II-*N*<sup>e</sup>-dansylthiourea are significantly improved *via* the described methods.

Lipid II, 1 is a membrane-anchored cell-wall precursor that is essential for the growth and replication of both Gram-positive and Gram-negative bacteria. The effectiveness of targeting the enzymes associated with lipid II or lipid II itself as an antibacterial strategy is highlighted by the fact that it is the target for at least four different classes of antibiotics, including the clinically important glycopeptide antibiotics.<sup>[1]</sup> The reactions necessary for the biogenesis of peptidoglycan (PG) have been known for decades, and the biosynthesis of peptidoglycan of *E. coli* has been discussed extensively in reviews by van Heijenoort.<sup>[2]</sup> Lipid II is an important biochemical tool for studying MurG, flippase that translocates lipid II across the cytoplasmic membrane, penicillin binding proteins (PBPs), and the mode of action of glycopeptide antibiotics.<sup>[3]</sup> Lipid II displays very poor water solubility, and thus, is not trivial to apply the natural form of lipid II to their biological investigations or to assay for screening inhibitor molecules. To date, several lipid II analogs have been synthesized and applied to the biochemical studies.<sup>[4]</sup> We have recently identified that nervl lipid II-N<sup>e</sup>dansyl analogue, 2-N<sup>e</sup>-dansylthiourea, could be recognized by mycobacterial transglycosylase (TGase) to form polymerized products with significant enhancement of visible-light absorption at 400 nm. Because the dansyl group itself does not have absorption in visible-light region, bathochromic shift of the polymerized- $2-N^{e}$ -dansylthiourea is an unusual physicochemical observation. Thus, the polymerized-2-N<sup>e</sup>-dansylthiourea via Mtb TGase can be distinguished from unreacted 2-N<sup>e</sup>-dansylthiourea in the reaction mixture via visible-light without separation (see, supporting information). The observed physicochemical property of polymerized- $2-N^{e}$ -dansylthiourea has been applied as a

Correspondence to: Michio Kurosu, mkurosu@uthsc.edu.

Supporting information for this article is available on the WWW under http://dx.doi.org/10.1002/chem.201xxxxxx.

convenient assay for lead discovery of new TGase inhibitor molecules in our laboratory. In order to perform high-throughput screening against TGase, it is indispensable to establish a convenient synthetic route to access neryl-lipid II, **2**. To date, total chemical and biochemical syntheses of lipid II and its derivatives have been accomplished by a few research groups.<sup>[4]</sup> We have accomplished chemoenzymatic and total chemical synthesis of UDP-MurNAc-pentapeptide (Park's nucleotide) and prenyl-MurNAc-pentapeptide (lipid I).<sup>[5]</sup> However, to the best of our knowlege, total chemical synthesis is more feasible than other methods for generation of enough lipid II analogues for high-throughput screening (HTS). Herein, we report an improved chemical synthesis of lipid II, **1** and neryl-lipid II analogue, **2**-*N*<sup>e</sup>-dansylthiourea *via* one-pot protection and glycosylation reactions.

Total chemical synthesis of lipid II reported previously revealed that limited combinations of the protecting groups of glycosyl acceptors and donors (**i** and **ii** in Figure 2) can be applied to successful synthesis of the lipid II disaccharide (**iii** and **iv**); the glycosyl donors such as *N*-phthaloyl 3,4,6-*O*-triacetyl-2-deoxyl-2-amino-D-glucopyranosyl 1-bromide ( $P_2 =$ phthaloyl, L = Br in **i**), *N*-2,2,2-trichloroethoxycarbonyl 3,4,6-*O*-triacetyl-2-deoxyl-2-amino-D-glucopyranosyl 1-bromide ( $P_2 =$  Troc, L = Br in **i**), or *N*-phthaloyl 2-deoxy-2amino-3,4,6-*O*-triacetate-D-glucopyranosyl 1-(2,2,2-trichloroacetoimidate) ( $P_2 =$  phthaloyl, L = *O*-imidate in **i**) have been utilized in the synthesis of the disaccharide **iv** with the C6protected donor **ii** ( $R_3 =$  Ac or Bn).[4] In general, the glycosyl acceptors **ii**, whose C6position was protected with acyl groups showed slow reaction rate and low conversion.<sup>[4f]</sup>

In our studies, glycosylation of the C6-acetylated MurNAc derivative **6a**<sup>[5a]</sup> with the glycosyl imidate 5a did not yield the desired product 7a (e.g. Entry 1 in Table 1). Because the N-Troc-protected imidate 5a is one of the convenient GlcNAc sources in chemical glycosylations, we have sought an appropriate C6-ether-protected MurNAc to efficiently synthesize the lipid II disaccharide iv (Figure 2).<sup>[6]</sup> The C6-benzyl ether-protected MurNAc 6b was introduced for Königs-Knorr type glycosylation of 5b by Saha et al.<sup>[7]</sup> The AgOTfcatalysed glycosylation of 6c with 5b was performed in our laboratory; however, these reactions resulted in the formation of the desired  $\beta$ -linked disaccharide 7c in moderate yield (Entry 5 in Table 1). The same glycosyl acceptor 6c was applicable to the glycosylation with the imidate 5a; TMSOTf-catalyzed glycosylation of 6c with 5a furnished 7c in a moderate yield of 50% (Entry 2 in Table 1).<sup>[8]</sup> Although synthesis of the C6-benzyl-protected glycosyl acceptor **6b** from the corresponding diol **ii** ( $R_3 = H$  in Figure 2) and selective debenzylation of the *primary* benzyl ether **7b** after the glycosylation were reported, in our hands, each step for these transformations provided the undesired diastereomers(s) that require timeconsuming chromatography, and/or the yield of each step was 45–65%.<sup>[7]</sup> We have previously reported that primary alcohols can be protected selectively with (2,6-dichloro-4methoxyphenyl)(2,4-dichlorophenyl)methyl trichloroacetimidates [3, monomethoxydiphenylmethyl acetimidate (MDPM-imidate)] to afford the corresponding ethers in good to excellent yields at controlled temperatures.<sup>[9]</sup> The MDPM-ether protecting group of **6d** was stable under Schmidt glycosylation conditions for glycosyl trichloroacetimidate; TMSOTf-catalyzed glycosylation of **6d** with **5a** gave rise to the  $\beta$ glycoside 7d as a mixture of two inseparable diastereomers in 65% yield in 3h (Entry 3 in

Table 1). The isolation yield of the same glycosylation reaction of **6d** with **5a** was increased by 20% using  $BF_3$ •OEt<sub>2</sub> (Entry 4 in Table 1).

MDPM-imidate **3** reacted against only the *primary* alcohol of **6a** even at room temperatures with near quantitative yield, and the by-product, 2,2,2-trichloroacetamide is an innocent species in Schmidt glycosylations. Thus, a one-pot protection-glycosylation protocol was envisioned for the synthesis of **7d** directly from the diol **6a** (Figure 2). As expected, the desired lipid II disaccharide derivative **7d** could be synthesized from **6a** in 75–85% yields in a one-pot two steps strategy (Scheme 1). It is worth mentioning that (2,6-dichloro-4-methoxyphenyl)(2,4-dichlorophenyl)methyl ether protecting group possesses a characteristic UV absorption, and isolation of the disaccharide **7d** from the crude reaction mixture *via* chromatography was relatively simple compared to that of **7c** (Entry 2 in Table 1).

In order to further facilitate the synthesis of the key intermediate iv in Figure 2, (2,6dichloro-4-alkoxyphenyl)(2,4-dichlorophenyl)methyl trichloroacetimidate linker resin 4 was applied to a one-pot two steps strategy for the synthesis of **iii** (Figure 2).<sup>[10]</sup> Loading of the diol 6a onto the linker resin 4 was completed with BF3•OEt2 (5 equivalents) at room temperature in 1h; in this step, progress of the reaction was monitored by consumption of **6a** via LC-MS. Once the loading step was completed, the imidate 5a was added into the reaction mixture to afford the desired  $\beta$ -glycoside resin 7e in 65–80% yield which was determined based on the isolated 7a (Table 1) after the cleavage of 7e with 30% TFA in CH<sub>2</sub>Cl<sub>2</sub> for 1h. Accordingly, convenient synthetic procedures for 7d and 7e for the syntheses of lipid II analogues were accomplished. The other convenient feature of (2,6dichloro-4-alkoxyphenyl)(2,4-dichlorophenyl)methyl ether-protecting group and -linker is that they can be deptrotected or cleaved simultaneously when the N-Troc group is removed under a reductive condition. The N-Troc and C6-ether protecting group or -linker of 7d or 7e were deprotected under the condition of Zn in 30% TFA-AcOH to furnish the amino-alcohol 8 (Scheme 2). Acetylations of the free amine and alcohol of 8 afforded 9 in 85–90% yield from 7d or 7e. a-Phosphorylation and diphosphate ester formation of 9 were carried out via the established protocols with minor modifications. Deprotection of the anomeric Bn protecting group was performed by Pd-C catalysed hydrogenation reaction to afford a mixture of  $\alpha/\beta$ -anomers, which were subjected to  $\alpha$ -selective phosphite formation using dibenzyl N,N-diisopropylphosphoramidite and 5-(ethylthio)-1H-tetrazole. The generated aphosphite intermediate was oxidized with  $tBuO_2H$  to afford the  $\alpha$ -phosphate 10 in 70% overall yield for three steps. Deprotection of the 2-(phenylsulfonyl)ethanol protecting group of 10 was achieved by the treatment with DBU to furnish the  $\alpha$ -phosphoryl GlcNAc-MurNAc-monopeptide derivative. The tetrapeptide, HCl•H-L-Ala-yD-Glu(OMe)-L-Lys(COCF<sub>3</sub>)-D-Ala-D-Ala-OMe (11) was synthesized in water media via water-soluble reagents (glyceroacetonide-Oxyma (12), EDCI, and NaHCO<sub>3</sub>) (see supporting information). [11] Coupling of the free carboxylic acid, a-phosphoryl GlcNAc-MurNAc-monopeptide with the tetrapeptide 11 under a mild condition (12, EDCI, and NaHCO<sub>3</sub>) in  $H_2O$  yielded the a-phosphoryl GlcNAc-MurNAc-pentapeptide 13 in over 90% overall yield from 10. Conveniently, all reagents and excess tetrapeptide used in this step could be removed via a basic water work-up. Hydrogenolytic debenzylations of 10 followed by the treatment with Et<sub>3</sub>N resulted in the corresponding monotriethylammonium phosphate 14 in quantitative

Chemistry. Author manuscript; available in PMC 2014 May 22.

yield, whose structure was established by <sup>1</sup>H-NMR analysis. Triethylammonium phosphate 14 was then applied to a carbonyldiimidazole (CDI) promoted diphosphate-formation reaction.<sup>[4e,4f]</sup> Triethylammonium a-phosphoryl GlcNAc-MurNAc-pentapeptide 14 was first activated with CDI and the excess CDI was quenched with MeOH to afford 1Himidazole-1-carboxylic (phosphoric) anhydride and methyl 1*H*-imidazole-1-carboxylate, which was not reactive against the phosphate nucleophiles. All volatiles were extensively removed and the remaining mixture was subjected to the cross-coupling reaction with the ammonium undecaprenyl phosphate (15) or neryl phosphate (16). Progress in the coupling reactions was monitored via reverse-phase HPLC (0.05 M NH<sub>4</sub>HCO<sub>3</sub> : MeOH to MeOH for lipid II). The reaction mixture was lyophilized and the fully-protected product was subjected to global deprotection reactions with aq. LiOH. Lipid II, 1 was synthesized in 45% overall yield from 13 after purification via reverse-phase HPLC. The structure of 1 was confirmed by <sup>1</sup>H-NMR, negative ESI-TOF-MS spectroscopy, and retention time in HPLC analysis.<sup>[4e]</sup> Similary, nervl-lipid II analogue, 2 could be synthesized in 75% overall yield by using excess ammonium nervl phosphate (16). The dansyl group was conjugated to the  $N^{e}$ -lysine moiety of 2 with 4-(dansylamino)phenyl isothiocyanate to furnish neryl-lipid II- $N^{\epsilon}$ dansylthiourea, 2-N<sup>e</sup>-dansylthiourea, in greater than 90% yield.

In conclusion, chemical syntheses of lipid II and neryl-lipid II analogues were accomplished via one-pot protection-glycosylation protocols. (2,6-Dichloro-4-methoxyphenyl)(2,4dichlorophenyl)methyl trichloroacetimidate (3) and its linker-resin 4 were demonstrated to be useful temporary protecting groups for the *primary* alcohol of the diol **6a** that could be compatible with Schmidt glycosylation reactions. In addition, (2,6-dichloro-4-alkoxyphenyl) (2,4-dichlorophenyl)methyl ether linkage could be deprotected simoutaneousely during the deprotection of N-Troc groups. Accodingly, the synthetic intermediate 10 for lipid II was efficiently synthesized from **6a** in 45–54% overall yield with minimum number of chemical steps. Gram-quantity of the tetrapeptide building block 11 can readily be synthesized in water media, and the peptide-forming reagents used for the synthesis of 13 could be removed via simple water work-ups. Detailed experimental procedures for improved syntheses of lipid II and its neryl analogues are summarized in Supporting Information. As mentioned in the introduction, neryl lipid II-N<sup>e</sup>-dansylthiourea, **2**-N<sup>e</sup>-dansylthiourea is a very useful biochemical tool for studying transglycosylase (TGase). Characterization of polymerized 2-N<sup>e</sup>-dansylthiourea by TGase and development of high-throughput screening (HTS) against TGase will be reported elsewhere.

# **Experimental Section**

## One-pot protection and glycosylation reaction of 6a

To a stirred suspension of **6a** (0.20 g, 0.32 mmol), (2,6-dichloro-4-methoxyphenyl)(2,4dichlorophenyl)methyl trichloroacetimidate (**3**, 0.29 g, 0.57 mmol), and MS 3Å (0.5 g) in  $CH_2Cl_2$  (2.5 mL) was added BF<sub>3</sub>•OEt<sub>2</sub> (0.5 mmol) at 0 °C. After 1h, the imidate **5a** (0.34 g, 0.54 mmol) in dichloromethane (1.0 mL) was added. The reaction mixture was stirred for 3 h at 0 °C, and quenched with sat. aq. NaHCO<sub>3</sub> solution. The resulting suspension was filtered through celite. The aqueous layer was extracted with ethyl acetate and the combined organic layer was dried over  $Na_2SO_4$ , filtered, and evaporated. Purification by silica gel column chromatography (hexanes/ethyl acetate = 20/80) afforded **7d** (0.39 g, 85%).

#### (2,6-Dichloro-4-alkoxyphenyl)(2,4-dichlorophenyl)methyl trichloroacetimidate linker resin 4

(2,6-Dichloro-4-alkoxyphenyl)(2,4-dichlorophenyl)methanol linker resin was prepared according to the procedure reported previously. The resin (5g, active surface: ~1.0 mmol/g) was suspended in  $CH_2Cl_2$  (10 mL) and  $CCl_3CN$  (5 mmol) and DBU (1 mmol). The suspension was gently stirred for 3h. The polymer resins were washed with THF-*i*PrOH (3/1), THF, and THF-CH<sub>2</sub>Cl<sub>2</sub> (2/1), and dried under high vacuum to furnish **4** (5.72g, quantitative yield).

#### One-pot loading and glycosylation of 6a

To a stirred suspension of **6a** (0.20 g, 0.32 mmol) and (2,6-dichloro-4-alkoxyphenyl)(2,4dichlorophenyl)methyl trichloroacetimidate linker resin **4** (1.6 g, ~1.0 mmol/g), and in CH<sub>2</sub>Cl<sub>2</sub> (5.0 mL) was added BF<sub>3</sub>•OEt<sub>2</sub> (0.64 mmol) at r.t. After 1h, the imidate **5a** (0.596 g, 0.96 mmol) in dichloromethane (2.0 mL) was added at 0 °C. The reaction mixture was gently stirred for 3h at 0 °C, and the resins were washed THF-*i*PrOH (3/1), THF, and THF-CH<sub>2</sub>Cl<sub>2</sub> (2/1), and dried under high vacuum to furnish **7e** (0.25 g). In order to determine the overall yield, the disaccharide resin **7e** was cleaved with 30% TFA for 1h and the generated **7a** was quantitated *via* LC-MS (65–80% yield).

# Deprotections of *N*-Troc and (2,6-dichloro-4-methoxyphenyl)(2,4-dichlorophenyl)methyl ether groups and acetylations

To a stirred solution of **7d** (0.35 g, 0.25 mmol) in 30% TFA-AcOH (3.5 mL) was added zinc powder (0.16 g, 2.5 mmol). After stirring the solution for 6h at r.t, the reaction mixture was filtered and all volatiles were evaporated to afford the crude **8**. To a solution of 8 in pyridine (2.0 mL) was added Ac<sub>2</sub>O (2.0 mL). After 4h at r.t, all volatiles were removed to afford the crude product. Purified by silica gel column chromatography (ethyl acetate/methanol = 95/5) to afford **9** (0.24 g, 95% overall yield). Similarly, the disaccharide resin **7e** was converted to **9** (85% overall yield).

# **Supplementary Material**

Refer to Web version on PubMed Central for supplementary material.

## Acknowledgments

The National Institutes of Health is greatly acknowledged for financial support of this work (AI084411). We also thank University of Tennessee for generous financial support. NMR data were obtained on instruments supported by the NIH Shared Instrumentation Grant.

# References

 a) Varneyl KM, Bonvin AMJJ, Pazgier M, Malin J, Yu W, Ateh E, Oashi T, Lu W, Huang J, Diepeveen-de Buin M, Bryant J, Breukink E, MacKerell AD Jr, de Leeuw EPH. PLOS. 2013; 9:e1003732.b) Yuan Y, Fuse S, Ostash B, Sliz P, Kahne D, Walker S. ACS Chem Biol. 2008; 3:429–436. [PubMed: 18642800] c) Martin NI, Breukink E. Future Micro. 2007; 2:513–525.d) Breukink E, de Kruijff B. Nature Rev Drug Disc. 2006; 5:321–323.

Chemistry. Author manuscript; available in PMC 2014 May 22.

- Page 6
- a) van Heijenoort J. Mol Biol Rev. 2007; 71:620–635.b) Auger G, van Heijenoort J, Mengin-Lecreulx D, Blanot D. FEMS Microbiol Lett. 2003; 219:115–119. [PubMed: 12594032] c) Bupp K, van Heijenoort J. J Bacteriol. 1993; 175:1841–1843. [PubMed: 8449890]
- a) Zervosen A, Sauvage E, Frère JM, Charlier P, Luxen A. Molecules. 2012; 17:12478–12505. [PubMed: 23095893] b) van Damk V, Sijbrandi R, Kol M, Swiezewska E, de Kruijff B, Breukink E. Mol Micro. 2007; 64:1105–1114.c) Zawadzke LE, Wu P, Cook L, Fan L, Casperson M, Kishnani M, Calambur D, Hofstead SJ, Padmanabha R. Anal Biochem. 2003; 314:243–252. [PubMed: 12654311] d) Hu Y, Chen L, Ha S, Gross B, Falcone B, Walker D, Mokhtarzadeh M, Walker S. Proc Natl Acad Sci USA. 2003; 100:845–849. [PubMed: 12538870]
- 4. a) Lebar MD, Lupoli TJ, Tsukamoto H, May JM, Walker S, Kahne D. J Am Chem Soc. 2013; 135:4632–4635. [PubMed: 23480167] b) Chen KT, Kuan YC, Fu WC, Liang PH, Cheng TJR, Wong CH, Cheng WG. Chem Eur J. 2013; 19:834–838. [PubMed: 23229320] c) Shih HW, Chen KT, Cheng TJR, Wong CH, Cheng WC. Org Lett. 2011; 13:4600–4003. [PubMed: 21797279] d) Yi Zhang Y, Fechter EJ, Wang T-SA, Barrett D, Walker S, Kahne DE. J Am Chem Soc. 2007; 129:3080–3081. [PubMed: 17323951] e) Van Niieuwenhze MS, Mauldin SC, Zia-Ebrahimi M, Winger BE, Hornback WJ, Saha SL, Aikins JA, Blaszack LC. J Am Chem Soc. 2002; 124:3656– 3660. [PubMed: 11929255] f) Schwartz B, Markwalder JA, Wang Y. J Am Chem Soc. 2001; 12:11638–11643. [PubMed: 11716719] g) Ha S, Chang E, Lo MC, Park P, Ge M, Walker S. J Am Chem Soc. 1999; 121:8415–8426.h) Hitchcock SA, Eid CN, Aikins JA, Zia-Ebrahimi M, Blaszczak LC. J Am Chem Soc. 1998; 120:1916–1917.
- a) Kurosu M, Li K. Hetrocycles. 2008; 76:445–469.b) Kurosu M, Mahapatra S, Narayanasamy P, Crick DC. Tetrahedron Lett. 2007; 48:799–803.
- Zhang Z, Ollmann I, Ye XS, Wischnat R, Baasov T, Wong CH. J Am Chem Soc. 1999; 121:734– 753.
- Saha SL, Van Nieuwenhze MS, Hornback WJ, Aikins JA, Blaszczak LC. Org Lett. 2001; 3:3575– 3577. [PubMed: 11678712]
- 8. Termin A, Schmidt RR. Liebigs Ann Chem. 1992:527-533.
- a) Kurosu M, Li K. Synthesis. 2009; 21:3633–3641.b) Kurosu M, Biswas K, Crick DC. Org Lett. 2007; 9:1141–1145. [PubMed: 17311394] c) Ali IAI, El Ashry EH, Schmidt RR. Eur J Org Chem. 2003:4121–4131.
- 10. Wang Y, Kurosu M. Tetrahedron. 2012; 68:4797-4804. [PubMed: 22711936]
- 11. Wang Q, Wang Y, Kurosu M. Org Lett. 2012; 14:3372-3375. [PubMed: 22697488]

NIH-PA Author Manuscript

2

Me

H<sub>3</sub>C

0 -0

H

N−CH<sub>3</sub>

Me

Me





**NIH-PA Author Manuscript** 

**NIH-PA** Author Manuscript





One-pot protection-glycosylation to synthesize GlcNAc-MurNAc-peptide iv for lipid II.



#### Scheme 1.

One-pot protection-glycosylation to synthesize GlcNAc-MurNAc-peptide 7.

Chemistry. Author manuscript; available in PMC 2014 May 22.



**Scheme 2.** Synthesis of lipid II and neryl lipid II analogues.

t NIH-PA

**NIH-PA** Author Manuscript

NIH-PA Author Manuscript NIH-PA Author Manuscript

Table 1

Mitachi et al.

Synthesis of lipid II disaccharide  $7.^{[a]}$ 

5	Ac $\alpha(\mathbf{x}, \beta = 7:1)$ Br $(\alpha: \beta = 1:0)$ $(\alpha: \beta = 1:0)$	$\begin{array}{c} \overbrace{\qquad} \overbrace{\qquad} \overbrace{\qquad} \overbrace{\qquad} \overbrace{\qquad} \overbrace{\qquad} \overbrace{\qquad} \qquad$	$\mathbb{R}_3$ for 6d and 7d $\mathbb{R}_3^{(2)} \xrightarrow{\mathbb{R}_3^{(2)}} \mathbb{R}_3^{(2)}$ $\mathbb{R}_3^{(2)} \xrightarrow{\mathbb{R}_3^{(2)}} \mathbb{R}_3^{(2)}$	7b: R <sub>3</sub> = Ac 7c: R <sub>3</sub> = Ac 7c: R <sub>3</sub> = Bc 7d: R <sub>3</sub> = (2.6-Dich phenyl)(2.4-dichloi phe	Product	Yield (%)
-	6b	TMSOTf	0	12	7b	0
	9C	TMSOTf	0	3	7с	50
	<b>6</b> d	TMSOTF	0	ю	7d	65
	6d	$BF_3 \bullet OEt_2$	0	ŝ	7d	85
	90	AgOTf	0	3	7с	45[b]

Chemistry. Author manuscript; available in PMC 2014 May 22.

[b] The reported isolation yield is 74%.