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Phenazine redox cycling enhances anaerobic survival in *Pseudomonas aeruginosa* by facilitating generation of ATP and a proton-motive force

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Summary

While many studies have explored the growth of *Pseudomonas aeruginosa*, comparatively few have focused on its survival. Previously, we reported that endogenous phenazines support the anaerobic survival of *P. aeruginosa*, yet the physiological mechanism underpinning survival was unknown. Here, we demonstrate that phenazine redox cycling enables *P. aeruginosa* to oxidize glucose and pyruvate into acetate, which promotes survival by coupling acetate and ATP synthesis through the activity of acetate kinase. By measuring intracellular NAD(H) and ATP concentrations, we show that survival is correlated with ATP synthesis, which is tightly coupled to redox homeostasis during pyruvate fermentation but not during arginine fermentation. We also show that ATP hydrolysis is required to generate a proton-motive force using the ATP synthase complex during fermentation. Together, our results suggest that phenazines enable maintenance of the proton-motive force by promoting redox homeostasis and ATP synthesis. This work demonstrates the more general principle that extracellular redox-active molecules, such as phenazines, can broaden the metabolic versatility of microorganisms by facilitating energy generation.

Keywords

Pseudomonas aeruginosa; survival; phenazines; anaerobic metabolism; pyruvate fermentation

Introduction

Pseudomonas aeruginosa is an opportunistic pathogen that forms both acute and chronic infections. It is well known to form biofilms in diverse environments, and microbial stasis within biofilms is thought to complicate the treatment of many persistent infections. For example, cells at the base of *P. aeruginosa* biofilms enter a dormant state with decreased levels of transcription and translation, and they undergo physiological adaptations to hypoxic conditions (Alvarez-Ortega and Harwood, 2007; Williamson *et al.*, 2012). These

cells are substantially more resistant to antibiotics than actively growing cells (Williamson *et al.*, 2012). Moreover, nutrient deprivation triggers active responses for *P. aeruginosa* that increase antibiotic resistance in growth-arrested cells (Nguyen *et al.*, 2011). Even under active growth, many bacterial species, including *P. aeruginosa*, maintain a small population of phenotypically resistant cells called persisters (Lewis, 2010). The persister state is thought to be controlled by stochastic changes in cell physiology and metabolism (Keren *et al.*, 2004; Lewis, 2010; Allison *et al.*, 2011). Together, these subpopulations of resistant non-replicating cells can serve as reservoirs for recurrent infections (Lewis, 2010). Our incomplete understanding of the metabolic changes underpinning microbial quiescence thus limits our ability to treat chronic infections.

In addition to forming biofilms and persister cells during infections, pseudomonads synthesize a class of redox-active pigments collectively termed phenazines (Price-Whelan *et al.*, 2006; Mavrodi *et al.*, 2006). Phenazines are important virulence factors (Lau *et al.*, 2005) that serve as antibiotics towards microbial competitors (Baron and Rowe, 1981) and damage mammalian cells (Britigan *et al.*, 1992). Phenazines can benefit *P. aeruginosa* by serving as signaling molecules (Dietrich *et al.*, 2006), regulating persister cell formation (Möker *et al.*, 2010), influencing colony morphology (Dietrich *et al.*, 2008; Dietrich *et al.*, 2013), and promoting iron acquisition and biofilm development (Wang *et al.*, 2011).

We previously reported that micromolar concentrations of phenazines can support anaerobic survival by transferring electrons to an extracellular oxidant (Wang *et al.*, 2010). Although the physiological basis for this survival mechanism was unknown, phenazine redox cycling was correlated with a more oxidized intracellular environment for planktonic cultures (Price-Whelan *et al.*, 2007; Sullivan *et al.*, 2011) and colony biofilms (Dietrich *et al.*, 2013), suggesting that phenazines alter metabolism. *P. aeruginosa* is thought to form biofilms within the lungs of cystic fibrosis patients (Singh *et al.*, 2000; Worlitzsch *et al.*, 2002), and in many patients, the lung sputum contains phenazines at concentrations that can support anaerobic survival (Wilson *et al.*, 1988; Hunter *et al.*, 2012; Wang *et al.*, 2010). Phenazines are positively correlated with the progression of *P. aeruginosa* lung infections (Hunter *et al.*, 2012), and so they may directly alter the physiology of *P. aeruginosa* infections *in situ*.

Here, we sought to elucidate the physiological basis for phenazine-promoted anaerobic survival in *P. aeruginosa*. We discovered that phenazines facilitate the conversion of glucose and pyruvate into acetate. Our data support a model where phenazines dissipate excess reducing equivalents, allowing for the coupling of acetate and ATP synthesis through the activity of acetate kinase. ATP is required for survival, in part to maintain the proton-motive force through the ATP synthase complex. Our results highlight the interplay between redox homeostasis and ATP synthesis in the context of long-term survival, and illuminate the important role extracellular redox active metabolites can play in facilitating core metabolism.

Results

Phenazine redox cycling requires the activity of acetate kinase to promote survival

Phenazine redox cycling occurs when a phenazine molecule undergoes alternating reduction and oxidation reactions, resulting in the transfer of electrons from the reductant (*e.g.* NAD(P)H) to the oxidant (*e.g.* oxygen). Phenazines can cross cell membranes, enabling the transfer of reducing equivalents from inside the cell to the extracellular environment. We have previously shown that this transfer of reducing equivalents can promote anaerobic survival (Wang *et al.*, 2010). We found that a carbon source, specifically glucose, is required for anaerobic survival in the presence of oxidized phenazines (Wang *et al.*, 2010). To understand how phenazines promote anaerobic survival, we began by exploring the role of glucose.

To metabolize glucose, *P. aeruginosa* uses the Entner-Doudoroff pathway to convert one glucose molecule into two pyruvate molecules (Figure 1a). This process is coupled to the synthesis of one net ATP molecule and two excess reducing equivalents (NAD(P)H) (Figure 1a). Pyruvate can be further converted into acetate to produce additional ATP and reducing equivalents (Figure 1b). To sustain anaerobic glycolysis, *P. aeruginosa* must regenerate the oxidants (*i.e.* NAD⁺ and NADP⁺) that were consumed during pyruvate and acetate synthesis. We therefore hypothesized that phenazines, by dissipating excess reducing equivalents, might promote the oxidation of glucose into acetate during anaerobic glycolysis. In this model, acetate synthesis is coupled to ATP synthesis through the activity of acetate kinase (Figure 1b), and it is the ATP synthesis that we expected to promote survival.

To address this hypothesis, we tested mutants with markerless deletions of pyruvate metabolism genes in a phenazine redox cycling survival assay. In this experiment, early stationary phase cells were resuspended in anoxic MOPS-buffered minimal medium containing glucose as the sole carbon and energy source. To control the concentration of phenazines, we used strains with deletions in both phenazine biosynthetic operons (*phzA1-G1 phzA2-G2*, abbreviated *phz1/2*) (Dietrich *et al.*, 2006) and added phenazine-1-carboxylic acid (PCA) back to the culture at a physiologically relevant concentration (75 μ M) (Hunter *et al.*, 2012). We chose PCA because it is the biosynthetic precursor of all other phenazines in *P. aeruginosa* and other pseudomonads (Mavrodi *et al.*, 2001; Mavrodi *et al.*, 2013), and also because it is synthesized whether oxygen is present or not. While oxygen and ferric iron are likely oxidants of phenazines in infection environments (Wang and Newman, 2008), we instead supplied an electrode poised at a potential sufficient to oxidize phenazines in our culture vessels ($E^0 = +207$ mV).

Consistent with the fact that *P. aeruginosa* is not known to ferment glucose, cultures declined to about 0.0001% viability after seven days when provided with glucose alone (Figure 2a). As previously reported (Wang *et al.*, 2010), the addition of phenazines and an extracellular oxidizing potential improved anaerobic survival to 10% after seven days (Figure 2a). A strain lacking lactate dehydrogenase, *ldhA*, was not impaired in survival relative to the *phz1/2* parent strain (Figure 2a). However, strains lacking acetate kinase that are unable to generate ATP by converting pyruvate into acetate, *ackA* and *ackA-pta*,

failed to survive relative to the parent strain, with only about 0.001% of cells remaining viable after seven days (Figure 2a).

We detected current through the extracellular electrode only when phenazines were present (Supplemental Figure 1), consistent with the model that phenazines mediate electron transfer from inside the cells to the electrode. After three days of survival, before we observed a loss in viability (Figure 2a), the *phz1/2* strain transferred a total charge of 11 C to the electrode. Using the relationship $Q = 2F(acv)$, where Q is the total charge transferred (11 C), F is Faraday's constant (96,485 C mole⁻¹), c is the concentration of PCA (7.5×10^{-5} M), v is the reaction volume (0.1 L), and a is the average number of redox cycles, we calculated an average of 8 redox cycles per molecule of PCA, a value similar to previous measurements (Wang *et al.*, 2010).

To test if acetate is produced during phenazine redox cycling, we used high performance liquid chromatography (HPLC) to directly measure the secreted metabolic products in our experiment. We detected the production of acetate, succinate, and two additional unknown products whose identification is a work in progress. Consistent with our hypothesis, cultures with oxidized phenazines produced more acetate than did cultures without phenazines (Figure 2b). After three days, in duplicate measurement, the *phz1/2* strain produced 350 μ M (5.9×10^{-8} nanomoles CFU⁻¹) and 330 μ M (5.0×10^{-8} nanomoles CFU⁻¹) of acetate with phenazine redox cycling but only 78 μ M (1.3×10^{-8} nanomoles CFU⁻¹) and 130 μ M (2.1×10^{-8} nanomoles CFU⁻¹) of acetate without phenazine redox cycling. Succinate and one unknown compound were produced in approximately equal amounts irrespective of phenazines (Figure 2a), with the *phz1/2* strain producing 2.4×10^{-8} nanomoles CFU⁻¹ of succinate after three days with oxidized phenazines. The other unknown product was produced in greater amounts in cultures with oxidized phenazines than in cultures without phenazines (Figure 2a).

We expected that acetate would be synthesized by acetate kinase (Figure 1b), which couples acetate synthesis to ATP synthesis. To our surprise, the *ackA* and *ackA-pta* mutants produced similar amounts of acetate as the parent strain during phenazine redox cycling with glucose (Figure 2b), suggesting that acetate kinase is not the only active acetate-producing enzyme in our experiment. The *P. aeruginosa* PA14 genome contains the gene *poxB*, which codes for a putative pyruvate oxidase that converts pyruvate to acetate and CO₂ without directly generating ATP or NADH (Chang and Cronan, 1983). Using quantitative reverse-transcription PCR, we detected *poxB* transcripts during exponential growth, early stationary phase, and after resuspension in anaerobic MOPS-buffered minimal medium with glucose and 75 μ M oxidized PCA. We also measured an increase in the relative abundance of pyruvate oxidase transcripts to control gene transcripts (pyruvate dehydrogenase and the housekeeping genes *clpX* and *recA*) after resuspension in minimal medium when compared to exponential growth (Supplemental Figure 2). Pyruvate oxidase, encoded by *poxB*, may therefore account for the observed acetate synthesis in the *ackA* and *ackA-pta* mutants during phenazine redox cycling. Pyruvate oxidase may also account for some acetate synthesis in the WT strain.

Together, the survival defect of the *ackA* and *ackA-pta* mutants (Figure 2a) demonstrates that the activity of acetate kinase is essential for long-term survival in our phenazine redox cycling experiment. As the *ackA* and *ackA-pta* mutants still produce acetate (Figure 2b), it is unlikely that acetate itself is sufficient for survival. Instead, our data are consistent with a model where ATP synthesis by acetate kinase is required for anaerobic survival.

Phenazine redox cycling promotes ATP synthesis during pyruvate fermentation

In keeping with the fact that the *phz1/2 ldhA* mutant survived as well as the parent strain *phz1/2* (Figure 2a), we did not observe lactate synthesis during phenazine redox cycling with glucose (Figure 2b). This suggests that lactate dehydrogenase is not used to maintain redox homeostasis during anaerobic glycolysis. In contrast, lactate dehydrogenase is required for survival during pyruvate fermentation (Eschbach *et al.*, 2004). During pyruvate fermentation, *P. aeruginosa* converts pyruvate into succinate, lactate, and acetate (Figure 1b). This process supports anaerobic survival but not growth (Eschbach *et al.*, 2004) and uses acetate kinase to synthesize ATP (Figure 1b). If phenazine redox cycling is a general mechanism for maintaining redox homeostasis, we predicted that phenazine redox cycling should support redox homeostasis and ATP synthesis during pyruvate fermentation.

To test this prediction, we measured the anaerobic survival of the *phz1/2 ldhA* strain using pyruvate as the sole carbon and energy source. As expected, this strain was unable to survive by fermenting pyruvate (Figure 3a). Consistent with our prediction, however, the addition of PCA and an extracellular oxidizing potential improved the survival of the *phz1/2 ldhA* strain during pyruvate fermentation (Figure 3a).

In our model of survival, redox homeostasis allows for additional acetate and ATP synthesis through the oxidative branch of the pyruvate fermentation pathway (Figure 1b). To further confirm this model, we used HPLC to measure acetate synthesis during pyruvate fermentation. The *phz1/2 ldhA* strain produced more acetate when provided with phenazines and an extracellular oxidizing potential (Figure 3b). We observed a difference in acetate production before the *phz1/2 ldhA* cultures began to lose viability (on day 3), suggesting that the difference in acetate synthesis was not simply due to a difference in survival.

To confirm that phenazine redox cycling provides an energetic benefit to *P. aeruginosa*, we also measured the ATP concentrations of cultures with and without phenazine redox cycling. Consistent with a coupling between acetate and ATP synthesis, *phz1/2 ldhA* cultures with oxidized phenazines maintained higher ATP concentrations than those without phenazines (Figure 3c). This difference was again observed before the cultures began to lose viability, suggesting that the difference in ATP concentration was not simply a consequence of cell death.

ATP synthesis is limited by redox homeostasis during pyruvate fermentation

Our model suggests that, by helping to maintain redox homeostasis, phenazines promote the conversion of pyruvate to acetate through the activity of acetate kinase. This model predicts that redox homeostasis is the limiting factor for ATP synthesis during pyruvate

fermentation. From the stoichiometry of pyruvate fermentation, acetate and lactate synthesis are coupled through the common cofactor NAD(H) (Figure 1b). Sustained pyruvate oxidation to acetate, and thus ATP synthesis, requires the regeneration of NADH to NAD⁺, which can be accomplished by the activity of lactate dehydrogenase. To determine whether this is indeed the case, we compared the concentration of relevant metabolites during pyruvate fermentation for wild type (WT) *P. aeruginosa* PA14 and different mutant strains.

To initiate pyruvate fermentation, we cultured PA14 using pyruvate as the sole carbon source aerobically, shifted the cells to an anoxic environment, and washed and resuspended the cells in fresh anoxic medium. As expected, anaerobic WT cultures with pyruvate survived for at least seven days, while cultures without pyruvate quickly declined in viability (Figure 4a). In contrast to previous studies with PAO1 (Eschbach *et al.*, 2004; Schreiber *et al.*, 2006), which demonstrated nearly complete survival, in our study we observed only 50% survival after 7 days of anaerobic incubation with pyruvate.

As expected, mutants deficient in the conversion of pyruvate to acetate, *ackA* and *ackA-pta*, displayed a survival defect during pyruvate fermentation (Figure 4a). A mutant deficient in the conversion of pyruvate to lactate, *ldhA*, had a survival defect similar to that of the *ackA* mutant (Figure 4a). To our surprise, a mutant unable to convert acetyl-CoA to acetyl-phosphate, *ackA-pta*, was more defective in survival than the *ackA* mutant (Figure 4a). The survival difference between the *ackA* and *ackA-pta* mutants might result from a difference in intracellular acetyl-phosphate concentrations, as acetyl-phosphate can serve as a phosphate donor in signaling pathways (Klein *et al.*, 2007). Alternatively, the *ackA-pta* mutant might accumulate acetyl-CoA, starving other essential pathways that require coenzyme A. Similar results have been observed in *E. coli*, where the expression of a pathway to consume acetyl-CoA enhances the survival of a *pta* mutant in stationary phase (Chang *et al.*, 1999).

We used ion chromatography to confirm the biosynthesis of acetate, lactate, and succinate from pyruvate as previously reported (Eschbach *et al.*, 2004) (Supplemental Figure 3). As expected, both the *ackA-pta* and *ldhA* mutants produced significantly less acetate than the WT strain (Supplemental Figure 3), suggesting that the *ldhA* mutant is limited in its ability to generate acetate and ATP using acetate kinase.

To confirm that the *ldhA* strain is impaired in redox homeostasis and ATP synthesis, we measured the NAD⁺, NADH, and ATP concentrations in cultures fermenting pyruvate. After 24 hours of anaerobic incubation with pyruvate as the sole carbon and energy source, both the WT strain and *ackA* mutant maintained an [NADH]/[NAD⁺] ratio of approximately 1 (Figure 4b). In contrast, the *ldhA* mutant contained a more reduced intracellular environment with an [NADH]/[NAD⁺] ratio of approximately 2 (Figure 4b). The *ldhA* mutant also contained significantly less total NAD(H) than the WT strain and *ackA* mutant (Supplemental Figure 4). In support of our model, both the *ldhA* and *ackA* mutants contained significantly less ATP than the WT strain (Figure 4c).

The reduced acetate synthesis and depleted ATP concentration in the *ldhA* mutant suggest that the *ldhA* mutant is unable to synthesize ATP through acetate kinase. The elevated

[NADH]/[NAD⁺] ratio suggests that acetate and ATP synthesis are limited by redox homeostasis, corroborating the fact that phenazine redox cycling improves survival in the *phz1/2 ldhA* mutant during pyruvate fermentation (Figure 3a).

Arginine supports anaerobic ATP synthesis and survival independently of redox state

The survival defects of the *ackA* and *ldhA* mutants during pyruvate fermentation suggest that ATP synthesis is an essential component of survival in *P. aeruginosa*. To test this hypothesis using an independent anaerobic metabolism that produces ATP, we investigated survival using arginine as the sole carbon and energy source. Arginine is used as an ATP source during anaerobic growth through the arginine deiminase pathway (Vander Wauven *et al.*, 1984) (Figure 1c). Previous work has shown that mutants in the arginine deiminase pathway have reduced anaerobic survival in undefined complex media (Schreiber *et al.*, 2006), but to our knowledge there have been no reports of arginine alone serving as a sufficient energy source for survival.

To determine whether arginine is sufficient for anaerobic survival, we cultured *P. aeruginosa* PA14 aerobically using arginine as the sole carbon source, shifted the cells to an anoxic environment, and washed and resuspended the cells in fresh anoxic medium with arginine. As with pyruvate, arginine alone was sufficient as an energy source to promote anaerobic survival (Figure 5a). WT PA14 dropped to 30% viability after one day of anaerobic incubation and slowly decreased in viability to 15% after seven days. In contrast, cells incubated without arginine declined to 0.1% viability after seven days (Figure 5a). A transposon insertion mutant in the arginine deiminase pathway, *arcC::MAR2×T7* (Liberati *et al.*, 2006), rapidly declined in viability to 0.02% after seven days despite the presence of arginine (Figure 5a). Similar results were obtained with a second independent transposon insertion mutant, *arcA::MAR2×T7*, disrupting another gene in the same pathway (Figure 1c) (data not shown). This demonstrates that arginine is sufficient to promote anaerobic survival through the arginine deiminase pathway.

To confirm that disrupting the arginine deiminase pathway also disrupted ATP synthesis, we measured NAD⁺, NADH, and ATP concentrations in cultures incubated with arginine. After 4 hours of anaerobic incubation with arginine as the sole carbon and energy source, both the WT strain and *arcC::MAR2×T7* mutant maintained similarly high [NADH]/[NAD⁺] ratios of approximately 3 (Figure 5b). However, the *arcC::MAR2×T7* mutant contained significantly less ATP than the WT strain (Figure 5c), consistent with the disruption of the ATP-generating step in the arginine deiminase pathway. The ability of the WT strain to anaerobically generate ATP from arginine, despite an elevated [NADH]/[NAD⁺] ratio, is consistent with the fact that the arginine deiminase pathway does not involve redox reactions (Figure 1c).

Together, these measurements are consistent with the hypothesis that ATP synthesis, rather than the [NADH]/[NAD⁺] ratio itself, is essential for long-term anaerobic survival in *P. aeruginosa*. Although redox homeostasis is essential when it is coupled to ATP synthesis, as in glycolysis and pyruvate fermentation, survival correlates most directly with ATP synthesis itself.

ATP synthesis is required to maintain the proton-motive force using the F_1F_0 -ATPase complex during fermentation

The requirement for ATP suggests that *P. aeruginosa* consumes ATP during long-term survival. During fermentation in many bacteria, the F_1F_0 -ATPase complex, also known as ATP synthase, hydrolyses ATP to extrude protons from the cytoplasm to the periplasm, thereby generating a proton-motive force (PMF) (Krulwich *et al.*, 2011). Previous studies have highlighted numerous reasons why a PMF might be required for survival; for example, the PMF is required for the exchange of metabolites and other ions (Krulwich *et al.*, 2011), the localization of cell division and cytoskeletal proteins (Strahl and Hamoen, 2010), protonation of the peptidoglycan layer (Calamita *et al.*, 2001), and the regulation of autolysis (Jolliffe *et al.*, 1981). We therefore hypothesized that *P. aeruginosa* requires ATP to maintain its PMF during fermentation.

To test this hypothesis, we incubated pyruvate-fermenting cells anaerobically with *N,N'*-dicyclohexylcarbodiimide (DCCD), an inhibitor of the proton-translocation channel in the F_0 subunit of ATP synthase (Sebald *et al.*, 1980). DCCD (dissolved in ethanol) was efficient at killing anaerobic *P. aeruginosa*, reducing viability to $0.16\% \pm 0.07\%$ after two days (Figure 6a). Ethanol from the addition of DCCD did not significantly affect survival, as a control culture with 0.2% ethanol maintained $88 \pm 17\%$ viability (Figure 6a). This suggests that the activity of ATP synthase is required for survival during pyruvate fermentation in *P. aeruginosa*.

To confirm that the PMF is required for survival, we incubated cells with carbonyl cyanide 3-chlorophenylhydrazone (CCCP), an ionophore that equilibrates the proton concentration across cell membranes and dissipates the PMF (Hopfer *et al.*, 1968). CCCP (dissolved in ethanol) reduced the viability of anaerobic cultures to below 0.01% after two days (Figure 6a), confirming that the PMF is required for survival.

If the PMF is required for survival, then restoring the PMF should recover survival even in the presence of DCCD. Although DCCD will inhibit oxidative phosphorylation, the cells can still synthesize ATP through pyruvate fermentation. To recover the PMF after DCCD treatment, we incubated cells anaerobically with nitrate, which serves as an alternate terminal electron acceptor for the electron transport chain in *P. aeruginosa* (Carlson and Ingraham, 1983; Van Alst *et al.*, 2009). Cultures incubated with pyruvate, DCCD, and nitrate sustained $46\% \pm 7\%$ viability after two days (Figure 6a), demonstrating that respiration mitigates the killing effects of DCCD. Nitrate also improved the survival of cultures that were depolarized with CCCP to $2\% \pm 1\%$ (Figure 6a). We did not observe significant growth after two days in cultures incubated with pyruvate, 0.2% ethanol, and nitrate, although we did observe growth after three days (data not shown), suggesting a long lag phase for growth by anaerobic nitrate respiration.

Because both pyruvate fermentation and the arginine deiminase pathway support anaerobic ATP synthesis, we also tested the effects of DCCD and CCCP on cultures surviving anaerobically with arginine. Cultures treated with 0.2% ethanol maintained $11\% \pm 1\%$ viability after two days (Figure 6b), consistent with the incomplete long-term survival we observed with arginine (Figure 5a). As with survival on pyruvate, DCCD inhibited the

survival of cultures incubated with arginine, reducing viability to $2.8\% \pm 0.2\%$ (Figure 6b), and 100 μM CCCP reduced viability to below 0.001% (Figure 6b). This suggests that maintenance of the PMF by the F_1F_0 -ATPase complex is a general survival strategy used by *P. aeruginosa* during fermentation.

To validate that DCCD caused death by preventing the generation of a PMF, rather than by some non-specific mechanism, we used flow cytometry with the dye DiOC₂(3) to assess the membrane potential of DCCD-treated cultures. DiOC₂(3) is a green-excitable dye that accumulates intracellularly in polarized cells, where it undergoes a red fluorescence shift that is indicative of a membrane potential (Novo *et al.*, 1999). Since this assay is not quantitative in Gram-negative bacteria (Shapiro and Nebe-von-Caron, 2004), we assessed the membrane potential qualitatively by comparing DCCD-treated samples to controls depolarized with CCCP (for an example analysis, see Supplemental Figure 5). As expected, pyruvate-fermenting cells maintained a detectable membrane potential. We observed a time-dependent depolarization of the cells after treatment with DCCD (Figure 6c), while an ethanol-treated control remained almost fully polarized (Figure 6c), demonstrating that DCCD prevents maintenance of the PMF and induces depolarization of the cell membrane.

These results indicate that *P. aeruginosa* uses the ATP generated during fermentation to maintain a PMF using the F_1F_0 -ATPase complex, and that the PMF is required for long-term anaerobic survival. To integrate our findings, we measured the membrane potential of cells surviving anaerobically on glucose in our phenazine redox cycling experiment. After three days of survival, $38\% \pm 6\%$ of cells maintained a detectable membrane polarization in the presence of oxidized phenazines, while less than 1% of cells maintained a detectable membrane potential in the absence of phenazines. Together, our results suggest that the ATP generated during phenazine redox cycling is used to maintain the PMF.

Discussion

In this study, we sought an explanation for how phenazines promote anaerobic survival in *P. aeruginosa* be it in planktonic (Wang *et al.*, 2010) or biofilm (Dietrich *et al.*, 2013) cultures. We found that phenazines help *P. aeruginosa* convert glucose and pyruvate into acetate (Figure 2b, 3b). This was an interesting result because *P. aeruginosa* is not known to ferment glucose or other sugars (Barnishan and Ayers, 1979; Eschbach *et al.*, 2004). Previous studies have shown that small molecule electron carriers can couple fermentation to extracellular oxidants, leading to the synthesis of more oxidized fermentation products (Emde *et al.*, 1989; Emde and Schink, 1990; Benz *et al.*, 1998; Beck and Schink, 1995). Our results suggest that phenazines perform a similar role, enabling *P. aeruginosa* to generate ATP by oxidizing glucose into acetate. This anaerobic glycolysis promotes survival but not growth (Figure 2a), underscoring the importance of studying survival phenotypes independently from growth phenotypes.

In Figure 7, we integrate the known survival pathways and processes in *P. aeruginosa* with a focus on redox homeostasis and ATP synthesis. By comparing phenazine redox cycling, pyruvate fermentation, and the arginine deiminase pathway, our combined results suggest that ATP synthesis is a key feature of long-term survival. The ATP synthesized during

fermentation and phenazine redox cycling is required, in part, to maintain the PMF using the ATP synthase complex.

During pyruvate fermentation, pyruvate dehydrogenase converts NAD^+ to NADH in a pathway that allows acetate kinase to synthesize ATP (Figure 1b). NAD^+ is regenerated by reducing pyruvate to lactate or succinate (Figure 1b). *P. aeruginosa* produces significantly more lactate than succinate during pyruvate fermentation (Supplemental Figure 3) (Eschbach *et al.*, 2004), and so lactate synthesis by lactate dehydrogenase appears to be the primary pathway for NAD^+ regeneration. This effectively couples the activities of acetate kinase and lactate dehydrogenase through the NAD(H) pool. As a result, the *ldhA* mutant has an elevated $[\text{NADH}]/[\text{NAD}^+]$ ratio during pyruvate fermentation (Figure 4b) and is also unable to maintain ATP levels as high as the WT strain (Figure 4c). Phenazine redox cycling can serve as an alternate pathway for redox homeostasis in the *ldhA* mutant and restore ATP synthesis (Figure 3c) and survival (Figure 3a).

In contrast to pyruvate fermentation, ATP synthesis by the arginine deiminase pathway does not involve redox reactions (Figure 1c). During anaerobic incubation with arginine, both the WT strain and the *arcC::MAR2*×T7 mutant developed an $[\text{NADH}]/[\text{NAD}^+]$ ratio even higher than the *ldhA* mutant did during pyruvate fermentation (Figure 5b). Nonetheless, the WT strain maintained ATP levels considerably higher than the *arcC::MAR2*×T7 mutant (Figure 5c), demonstrating that the arginine deiminase pathway can proceed even with large intracellular redox imbalances. Despite maintaining a high $[\text{NADH}]/[\text{NAD}^+]$ ratio, cultures incubated with arginine were able to survive through the arginine deiminase pathway (Figure 5a). Pyruvate and arginine may therefore offer alternative sources of ATP during long-term anaerobic survival.

The requirement for ATP suggests that *P. aeruginosa* is not dormant during long-term survival. The ATP synthase inhibitor DCCD killed anaerobic PA14 cultures that were fermenting pyruvate or arginine (Figure 6a,b) and also depolarized cultures that were fermenting pyruvate (Figure 6c), demonstrating that *P. aeruginosa* actively maintains its PMF using the ATP synthase complex during fermentation. Previous studies have shown that survival also requires transcription (Hu and Coates, 1999), translation (Reeve *et al.*, 1984), and proteolysis (Weichart *et al.*, 2003), all of which consume ATP. By promoting ATP synthesis, phenazine redox cycling can therefore facilitate a variety of cellular processes that are required for long-term survival.

We do not yet know how native phenazines such as PCA are reduced *in vivo*. Phenazines oxidize NAD(P)H *in vitro* (Cox, 1986), and so PCA might directly oxidize NAD(P)H within the cell. Redox cycling compounds such as phenazines are oxidants of flavins and iron-sulfur clusters (Gu and Imlay, 2011), and so metabolic enzymes such as NADH dehydrogenase, pyruvate dehydrogenase, or pyruvate oxidase might also reduce phenazines. Synthetic electron shuttles such as anthraquinone-2,6-disulfonate and paraquat do not support anaerobic survival by redox cycling because they are reduced slowly by *P. aeruginosa* (Wang *et al.*, 2010), suggesting that cells specifically reduce native phenazines. Our preliminary studies indicate PCA reduction is enzymatically catalyzed, and we are attempting to identify the proteins responsible for this activity.

Consistent with our finding that phenazines can alter metabolism, WT *P. aeruginosa* cultures grown aerobically with glucose secrete pyruvate in the oxygen-limited late stationary phase of growth, while phenazine-null mutant cultures do not (Price-Whelan *et al.*, 2007). It was suggested that superoxide or pyocyanin radicals may inhibit pyruvate dehydrogenase, leading to pyruvate accumulation in the culture supernatant (Price-Whelan *et al.*, 2007). Our results suggest an additional interpretation where glycolysis is limited by the availability of intracellular NAD⁺. By promoting redox homeostasis, phenazines might accelerate the conversion of glucose to pyruvate faster than pyruvate can be consumed. After the consumption of glucose, the secreted pyruvate can be fermented or oxidized in the tricarboxylic acid cycle, depending on the metabolic needs of the cell.

Together, this work underscores the importance of environmental metabolites that are not always accounted for in metabolic pathways. Extracellular electron shuttles, such as phenazines, may permit cells to utilize unexpected energy sources by alleviating redox constraints on a pathway. An understanding of environmental parameters that interface with metabolism is therefore essential to assess the true metabolic potential of some organisms. Furthermore, a distinction must be drawn between growth and survival when considering the environment. Numerous studies have demonstrated the metabolic versatility of *P. aeruginosa* (for example, (Behrends *et al.*, 2009; Stover *et al.*, 2000)), and here we have shown how several anaerobic pathways might enable survival under clinically relevant conditions. Characterizations of cystic fibrosis sputum indicate the presence of up to 100 μM phenazines (Hunter *et al.*, 2012) and significant quantities of nitrate, arginine, and glucose (Palmer *et al.*, 2007). Nitrate concentrations may be as high as 350 μM (Palmer *et al.*, 2005) and oxygen levels vary widely (Worlitzsch *et al.*, 2002; Kolpen *et al.*, 2014), potentially resulting in extended periods of slow growth or quiescence. Given the metabolic heterogeneity of biofilms (Williamson *et al.*, 2012) and the likely heterogeneity of infections, *P. aeruginosa* cells throughout infection environments might exhibit a range of growth rates, including survival without replication. As a result, understanding the multitude of survival mechanisms used by *P. aeruginosa* may be key to identifying new drug targets.

Materials and Methods

Bacterial strains and growth conditions

The strains and plasmids used in this study are shown in Supplemental Table 1. For routine growth, *P. aeruginosa* and *E. coli* were cultured at 37 °C in lysogeny broth (LB) containing 10 g/L tryptone, 5 g/L yeast extract, 10 g/L NaCl, and optionally 15 g/L agar for solid medium. *S. cerevisiae* was cultured at 30 °C in YPD medium containing 10 g/L yeast extract, 20 g/L peptone, 20 g/L d-glucose, and optionally 20 g/L agar for solid medium. Specific experimental growth conditions are described where appropriate. For aerobic growth, liquid cultures were shaken at 250 rpm.

Construction and verification of *P. aeruginosa* mutants

Unmarked deletions in *P. aeruginosa* PA14 were constructed using a modification of previously described methods (Shanks *et al.*, 2006). Briefly, PCR was used to amplify approximately 1-kilobase fragments flanking the target gene. Homologous recombination in

yeast was used to assemble these fragments into the *sacB*-counterselectable suicide shuttle vector pMQ30. The knockout vector was transformed into *E. coli* DH5 α , and triparental mating with DH5 α and *E. coli* HB101/pRK2013 was used to conjugate the constructed knockout vector into PA14; merodiploids containing the chromosomally integrated vector were selected on cetrимide agar (HiMedia) containing 100 $\mu\text{g mL}^{-1}$ gentamicin sulfate. Merodiploids were cultured overnight in nonselective LB, and resolved merodiploids were then selected on LB agar plates with 10% sucrose. Potential deletion mutants were screened using colony PCR with primers flanking the target gene, and clean deletions were confirmed by DNA sequencing of the PCR product (Retrogen). Transposon insertions were verified using colony PCR with primers flanking the annotated transposon position. Primer sequences for constructing and confirming the genetic mutants used in this study are listed in Supplemental Table 2.

Phenazine redox cycling survival

The assay for anaerobic survival enabled by phenazine redox cycling was performed as previously described (Wang *et al.*, 2010). Briefly, strains of PA14 deleted in the core phenazine biosynthesis operons (*phz1/2*) (Dietrich *et al.*, 2006) were grown overnight in LB. The overnight cultures were used to inoculate 250 mL of LB in 1-L flasks to an OD₅₀₀ of 0.05. At early stationary phase (OD₅₀₀ of 2.8) the cells were pelleted, washed twice aerobically in MOPS-buffered medium (100 mM 3-(N-morpholino)propanesulfonic acid (MOPS), 3.5 μM FeSO₄, 43 mM NaCl, 3.7 mM KH₂PO₄, 9.3 mM NH₄Cl at pH 7.2), and resuspended to a concentrated OD₅₀₀ of 70 before being transferred to a nitrogen-only atmosphere (MBraun Unilab glove box). There, 1 mL of concentrated cells was added to sealed glass vessels containing 100 mL of anoxic MOPS-buffered medium amended with 75 μM phenazine-1-carboxylic acid and 20 mM d-glucose or 40 mM sodium pyruvate. To oxidize PCA, a graphite rod (Alfa Aesar #14738) working electrode was poised at a potential of +207 mV vs. NHE against a homemade platinum mesh counter electrode, which was placed in buffer within an attached small chamber separated from the bulk solution by a glass frit. Electrochemical parameters were maintained by a potentiostat (Gamry). An Ag/AgCl electrode (BaSi #RE5B) was used as the reference electrode. Anaerobic cultures of cells were stirred vigorously and maintained at 31 °C for the duration of the experiment, with daily sampling for colony forming units. Phenazine turnover was confirmed by the accumulated charge over the duration of the experiment as previously described (Wang *et al.*, 2010).

Pyruvate and arginine survival

For measuring anaerobic survival using pyruvate or arginine as a sole carbon source, cultures were grown overnight at 37 °C in a minimal medium containing 14.15 mM KH₂PO₄, 35.85 mM K₂HPO₄, 42.8 mM NaCl, 9.3 mM NH₄Cl, 1 mM MgSO₄, 7.2 μM FeCl₂ and trace elements (Widdel *et al.*, 1983), and 40 mM sodium pyruvate or 40 mM arginine (pH adjusted to 7.2 with NaOH or HCl). Overnight cultures were diluted at least 100-fold with fresh medium to an OD₅₀₀ of 0.01 and incubated at 37 °C in shaking flasks to an OD₅₀₀ of 0.4. After centrifuging the culture for 15 min at 5000 \times g, the supernatant was removed and the pellet was transferred to a glove chamber (Coy) containing an atmosphere of 15% CO₂, 80% N₂, and 5% H₂. The pellet was washed twice with fresh anoxic medium

and then resuspended to a final OD₅₀₀ of approximately 0.4 in rubber-stoppered tubes. The sealed tubes were incubated at 37 °C in the anaerobic chamber without shaking.

Viability counting

For measurements of colony forming units (CFU), 20 µL of each culture was sampled anaerobically and then serially diluted in aerobic phosphate-buffered saline. Relevant dilutions were plated in drips of 10 µL on LB agar plates and incubated aerobically for 24 hours at 37°C. The average of at least 4 pipetting replicates was taken as the CFU count for a given CFU measurement.

Measurement of NAD⁺, NADH, and ATP

The method for measuring NAD⁺ and NADH in this study was based on established protocols (San *et al.*, 2002; Price-Whelan *et al.*, 2007). For each condition, two 1-mL samples in 1.7 ml tubes were centrifuged for 1 min at maximum speed (16,000×g). The supernatant was removed and 200 µL of either 0.2 M HCl (for NAD⁺ extraction) or 0.2 M NaOH (for NADH extraction) was added and mixed by vortexing. The samples were then incubated for 10 minutes at 55 °C, cooled on ice, and partially neutralized drop-wise with 200 µL of 0.1 M NaOH (for NAD⁺) or 0.1 M HCl (for NADH). Cell debris was pelleted by centrifugation (5 minutes at maximum speed) and 150 µL of the supernatant was transferred to a fresh tube for immediate analysis. To assay the amount of NAD⁺ or NADH in each sample, 5 µL of sample was added to 90 µL of the reagent mix in a 96-well microtiter plate. The reagent mix contained 2× bicine buffer (1.0 M, pH 8.0), 1× water, 1× 40 mM EDTA, 1× 100% ethanol, 1× 4.2 mM thiazolyl blue, and 2× 16.6mM phenazine ethosulfate. The 96-well plate was then heated to 30 °C. To initiate the colorimetric assay, 5 µL of a 1 mg mL⁻¹ solution of alcohol dehydrogenase II (Sigma A3263) in 0.1 M bicine pH 8.0 was added to each well. A BioTek Synergy 4 plate reader was used to incubate the plate at 30 °C and record the absorbance of reduced thiazolyl blue at 570 nm every minute for 30 minutes. The rate of increase in absorbance at 570 nm over time was used as the measure of NAD(H) concentration. Unknown concentrations of NAD⁺ and NADH were determined from the slope and intercept of a standard curve derived from known concentrations of NADH.

For ATP measurement, 20 µL of culture was added to 180 µL of dimethyl sulfoxide (DMSO) to quench and dissolve the cells. The sample was then diluted with 800 µL of 0.1 M HEPES (pH 7.5) and stored at -80 °C for up to 7 days. ATP was quantified by mixing 25 µL of sample with 25 µL of Promega BacTiter-Glo reagent in a 96-well opaque white microtiter plate. The BacTiter-Glo reagent uses luciferase to produce luminescence in an ATP-dependent manner. Total luminescence was measured at 30 °C in a BioTek Synergy 4 plate reader. In control experiments using pure ATP, DMSO had no effect on luminescence at a final concentration of 9% (data not shown). Unknown concentrations of ATP were determined from the slope and intercept of a standard curve derived from known concentrations of ATP.

Assessment of membrane potential

The membrane potential was measured qualitatively using the dyes 3,3'-diethyloxycarbocyanine iodide (DiOC₂(3)) and TO-PRO-3. Cultures were diluted at least

100-fold into a permeabilization buffer containing 100 mM Tris, 1 mM EDTA, and 80 mM NaCl (pH adjusted to 7.4 using HCl). DiOC₂(3) and TO-PRO-3 (both dissolved in DMSO) were then added to a final concentration of 30 μM and 100 nM, respectively. For depolarized controls, CCCP (dissolved in DMSO) was added to a final concentration of 15 μM. The final DMSO concentration did not exceed 2%. The samples were incubated for 2-5 minutes at room temperature and then analyzed on an Accuri C6 flow cytometer. DiOC₂(3) fluorescence was measured using excitation at 488 nm and emission at 530 ± 15 nm (FL1) and 610 ± 10 nm (FL3). TO-PRO-3 fluorescence was measured using excitation at 640 nm and emission at 675 ± 12.5 nm (FL4). Intact cells were gated on a log-scale scatter plot of FL4 vs. FL2 (585 ± 20 nm) for further analysis. Cells with a membrane potential were distinguished by an increased red/green (FL3/FL1) fluorescence ratio relative to depolarized controls. For an example analysis, see Supplemental Figure 5.

High performance liquid chromatography

Metabolite samples were collected by centrifuging 550 μL of culture and passing the supernatant through a 0.2 μm nylon centrifugal filter. Samples were stored at -80 °C until analysis. Pyruvate, acetate, and succinate were quantified using an Agilent 1100 Series HPLC. A G1312A binary pump was used to draw a continuous flow of 8 mM H₂SO₄ at 0.6 mL/min through a G1322A degasser. Samples were injected using a G1313A autosampler, and separation was effected using an Aminex HPX-87H column (Bio-Rad). Each sample was analyzed using a G1315A diode array detector to measure the UV absorbance at 206 nm referenced to 260 nm (bandwidth 16 nm) with a total run time of 25 min. Retention times for analytes were validated with single species standards. For each analyte, standards ranging from 0 to 40 mM were used to calibrate peak area against a known concentration. Data were analyzed using the ChemStation software (Agilent).

Supplementary Material

Refer to Web version on PubMed Central for supplementary material.

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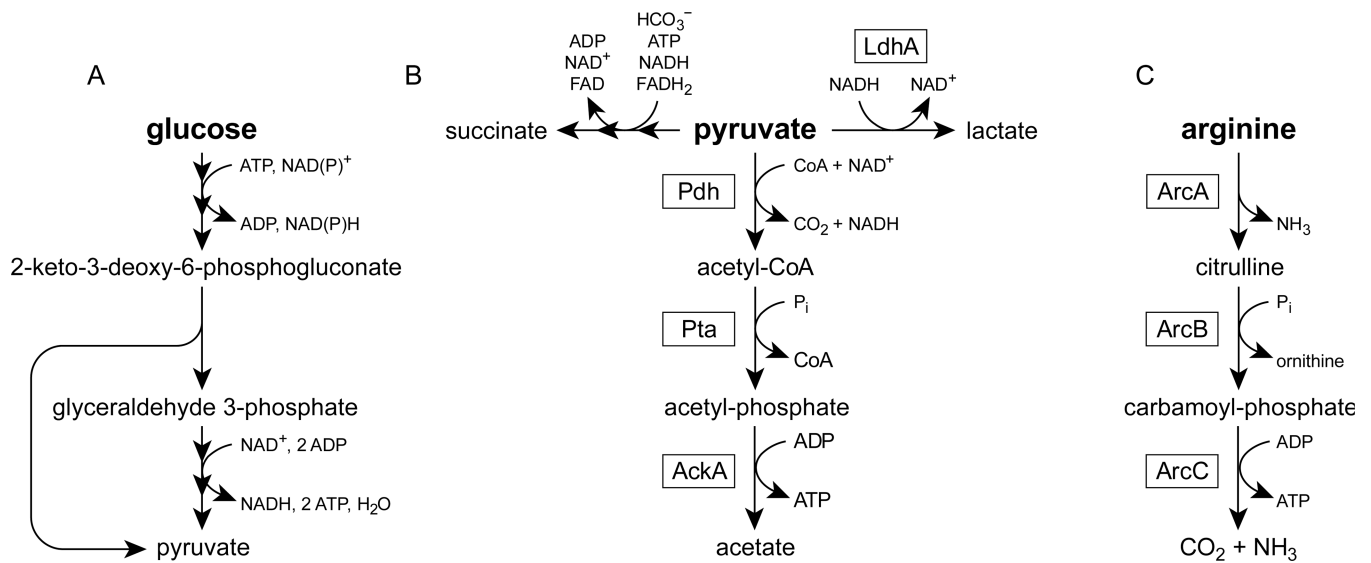


Figure 1.

Metabolic pathways in *P. aeruginosa*. Multiple arrowheads indicate that several reaction steps have been condensed for clarity. Boxes indicate the enzyme that catalyzes each reaction. (A) Glucose oxidation via the Entner-Doudoroff pathway. For clarity, the pathway is abbreviated to the key intermediates of glycolysis. (B) Pyruvate fermentation (Eschbach *et al.*, 2004). LdhA, lactate dehydrogenase; Pdh, pyruvate dehydrogenase; Pta, phosphate acetyltransferase; AckA, acetate kinase. (C) The arginine deiminase pathway (Vander Wauven *et al.*, 1984). ArcA, arginine deiminase; ArcB, ornithine carbamoyltransferase; ArcC, carbamate kinase.

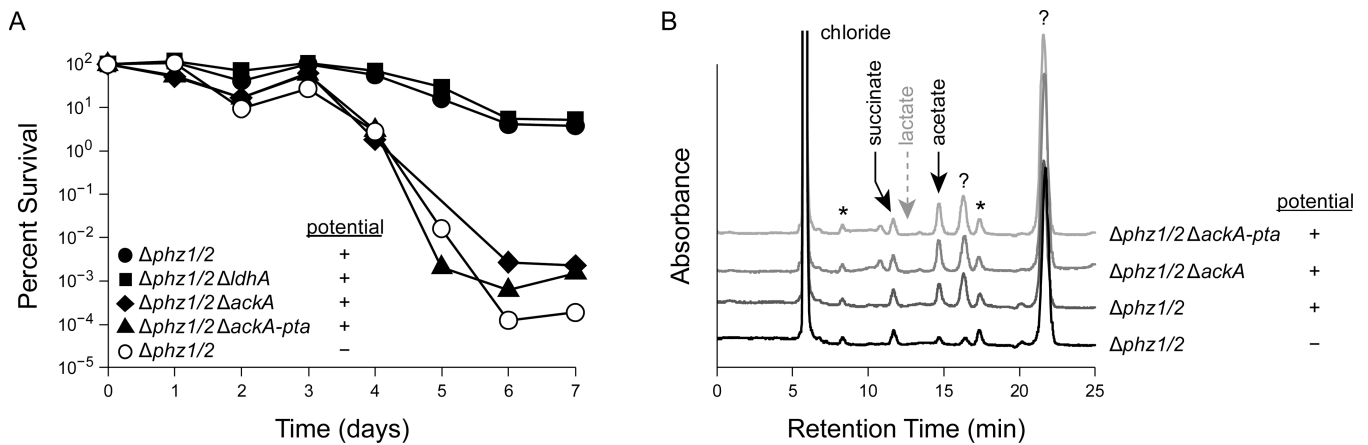
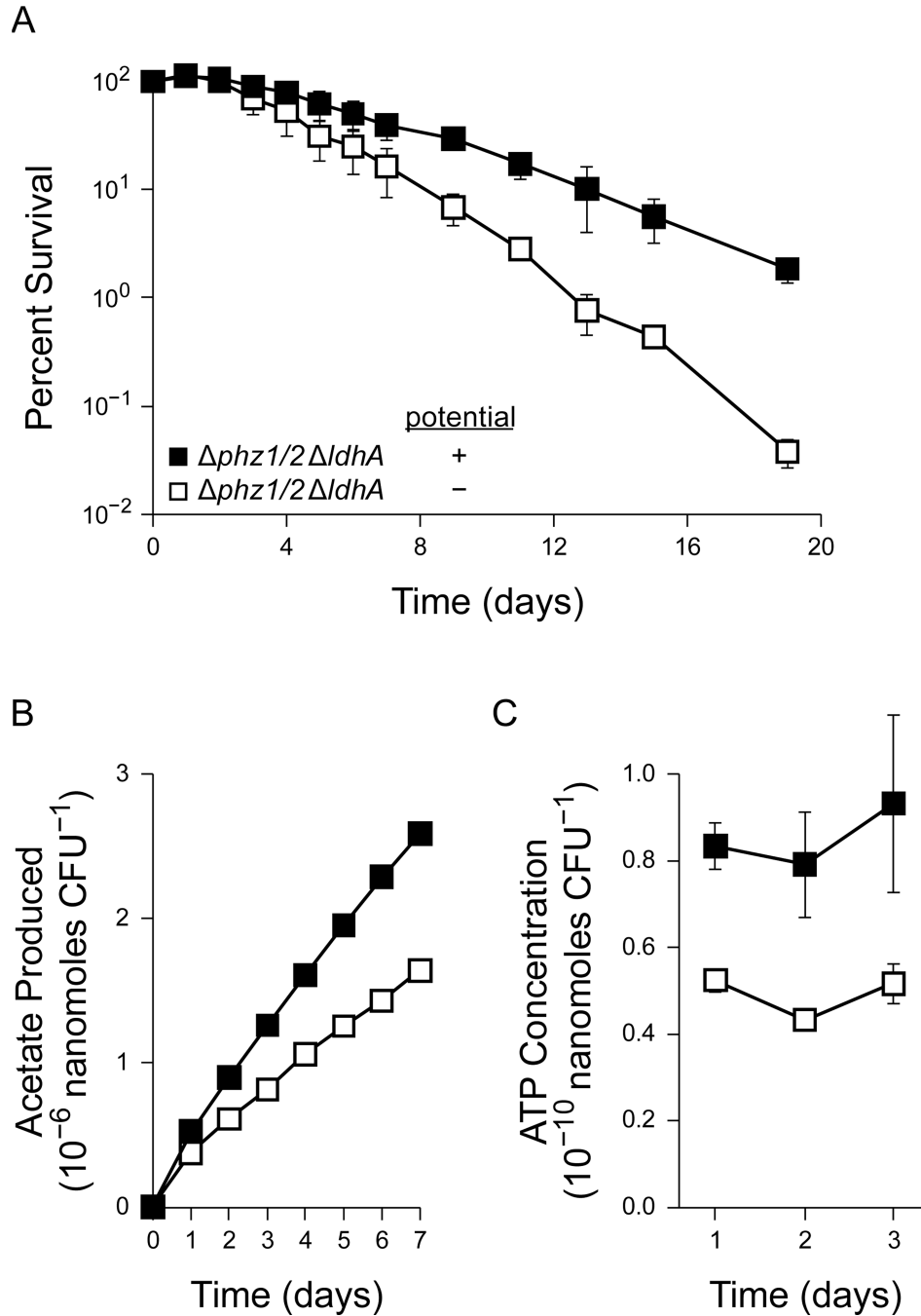


Figure 2.

Phenazine redox cycling in *P. aeruginosa* PA14 *phz1/2*. Cells were grown aerobically in LB to an OD_{500} of 2.8 and then pelleted, washed, and resuspended in anoxic MOPS-buffered minimal medium with 20 mM glucose. Where noted in the ‘potential’ column of the legend, cultures also included 75 μ M PCA and an electrode poised at a potential to oxidize any reduced PCA. Results are typical of at least three independent experiments. (A) Survival of strains with deletions of pyruvate metabolism genes. 100% survival represents approximately 7×10^8 CFU/mL. (B) HPLC traces from culture supernatants after three days of phenazine cycling as in panel A. Peaks were identified by comparing retention times to known standards. The dashed gray arrow indicates the retention time of lactate, which was not observed in this experiment. Question marks (?) indicate unidentified products. Asterisks (*) indicate trace contaminants that were present from the start of the experiment.

**Figure 3.**

Pyruvate fermentation in the PA14 *phz1/2 ldhA* strain with 40 mM pyruvate. Cultures were incubated without phenazines or an oxidizing potential (open squares) or with 75 μ M PCA and an oxidizing potential (closed squares). Error bars indicate the standard error of three independent experiments. (A) Survival of the *phz1/2 ldhA* strain during pyruvate fermentation. 100% survival represents approximately 1.4×10^9 CFU/mL. (B) Acetate produced during pyruvate fermentation in the *phz1/2 ldhA* strain. The amount of acetate produced in each experiment was normalized to the CFU measurement at day 0. Error bars

are present but smaller than the data points. (C) ATP concentrations during pyruvate fermentation in the *phz1/2 ldhA* strain.

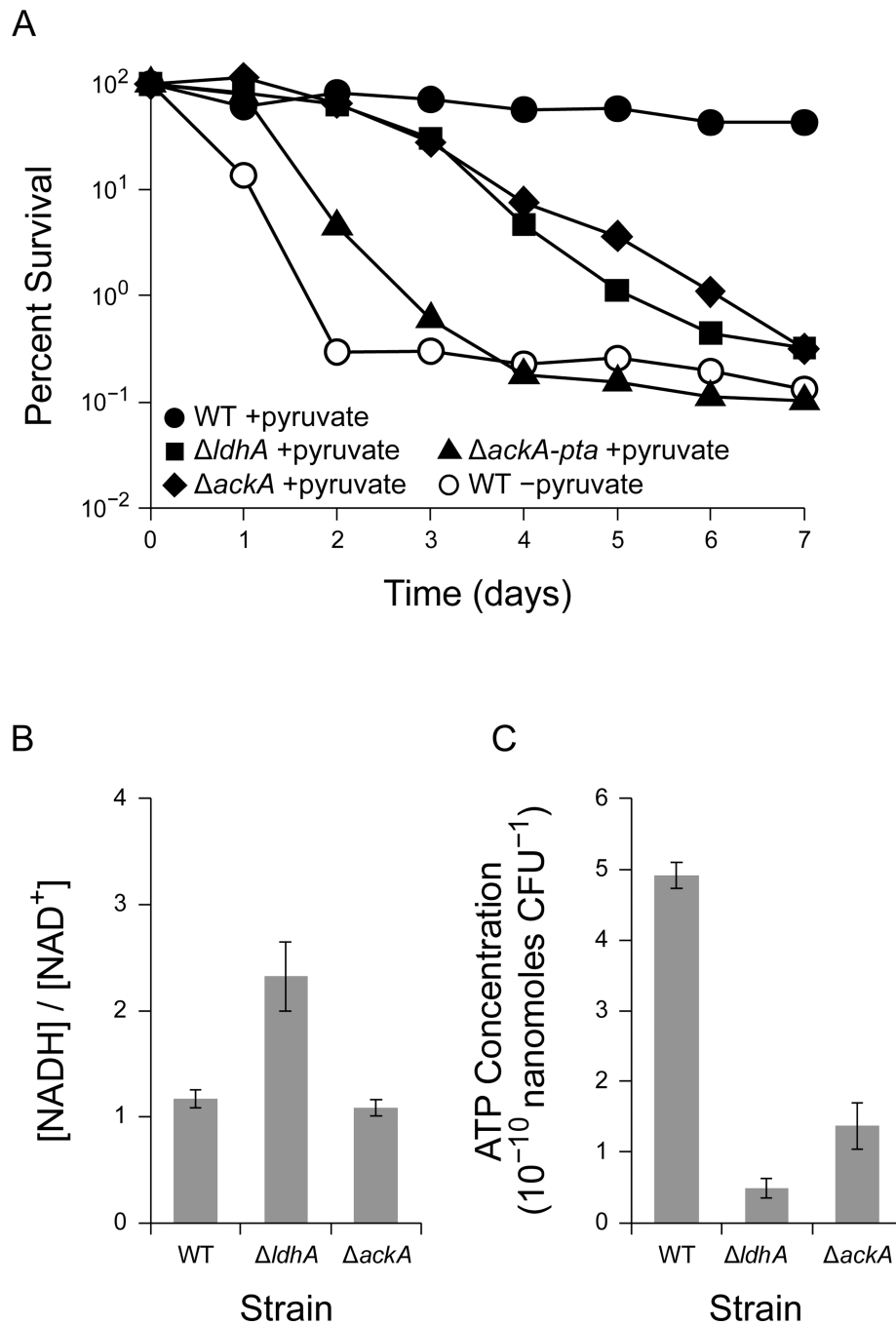


Figure 4. Survival with pyruvate fermentation in *P. aeruginosa*. Where shown, error bars indicate the standard error of three independent experiments. (A) Anaerobic survival for wild type PA14 (WT), PA14 *ldhA* (*ldhA*), PA14 *ackA* (*ackA*), and PA14 *ackA-pta* (*ackA-pta*) in the presence of 40 mM pyruvate (+pyruvate, closed points). WT PA14 without pyruvate is shown for comparison (-pyruvate, open points). 100% survival represents approximately 4×10^8 CFU/mL. Results are typical of least three independent experiments. (B) The

[NADH]/[NAD⁺] ratio after 24 hours of pyruvate fermentation in PA14 WT, *ldhA*, and *ackA*. (C) ATP concentrations from the same experiments shown in panel B.

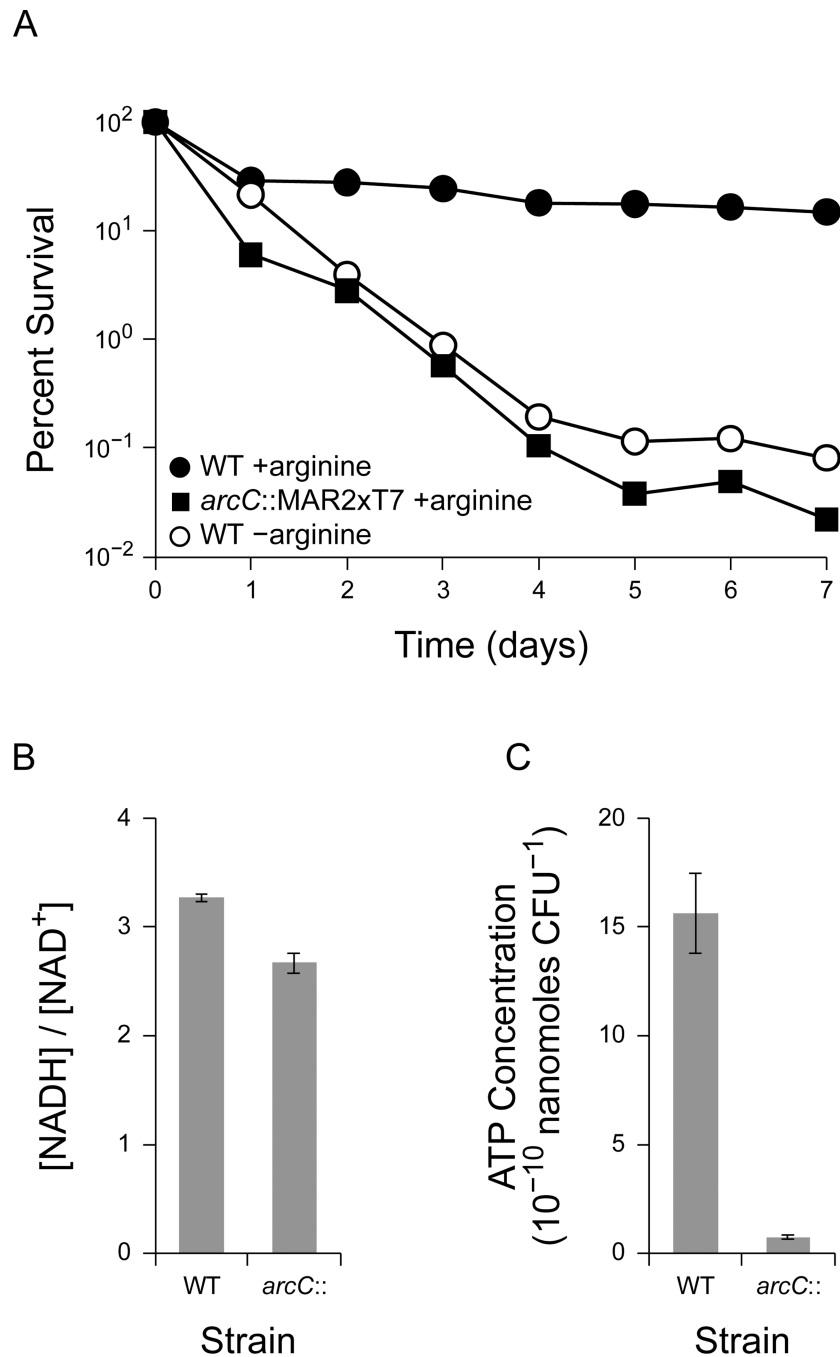
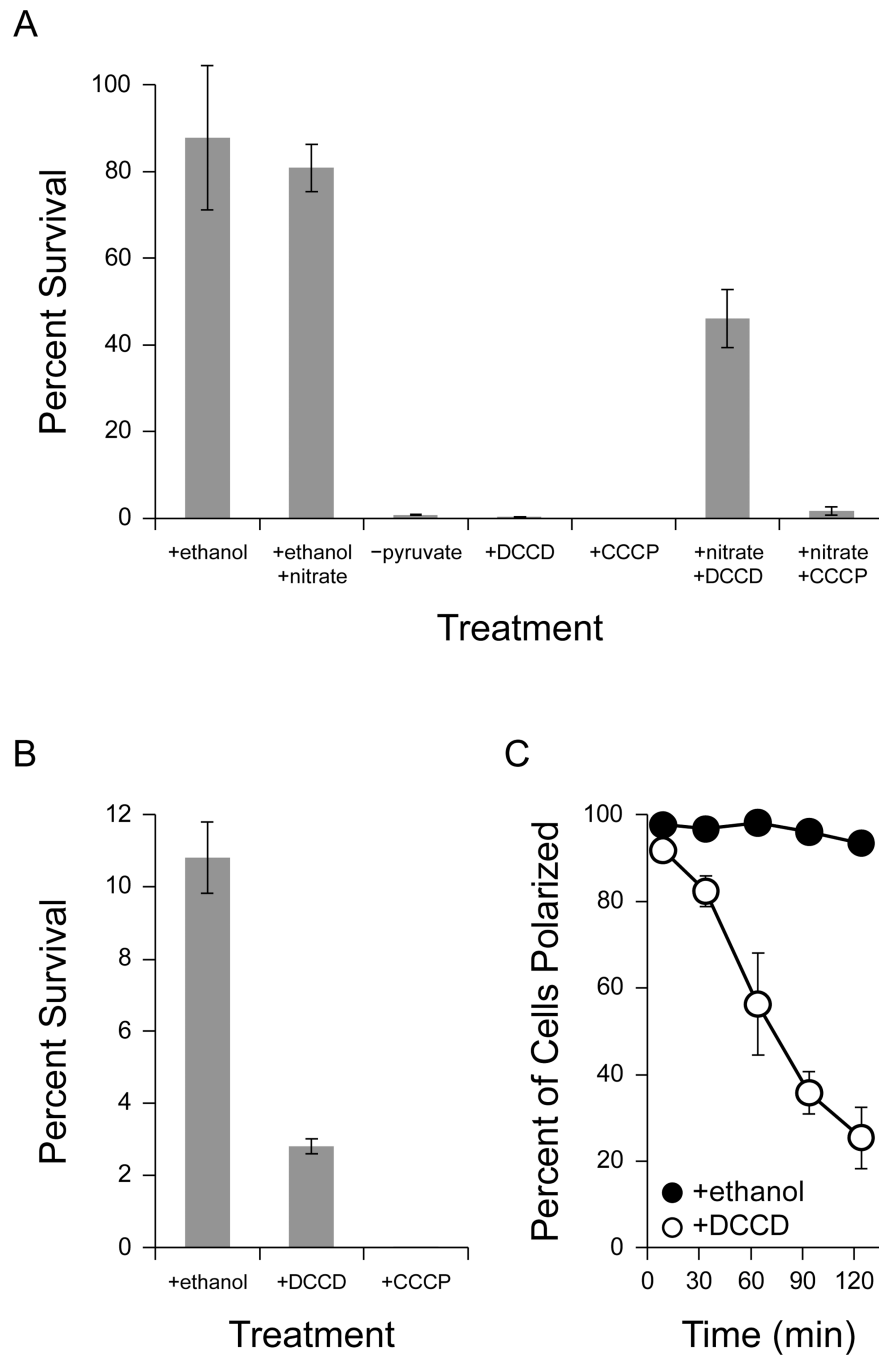


Figure 5. Survival with the arginine deiminase pathway in *P. aeruginosa*. Where shown, error bars indicate the standard error of three independent experiments (A) Anaerobic survival for wild type PA14 (WT) and PA14 *arcC::MAR2xT7* (*arcC::*) in the presence of 40 mM arginine. WT PA14 without arginine is shown for comparison (–arginine, open points). 100% survival represents approximately 4×10^8 CFU/mL. Results are typical of at least three independent experiments. (B) The [NADH]/[NAD⁺] ratio after 24 hours of anaerobic incubation with

arginine in PA14 WT and *arcC::MAR2*×T7. (C) ATP concentrations from the same experiments shown in panel B.

**Figure 6.**

Response of *P. aeruginosa* PA14 to drugs that disrupt the proton-motive force. Error bars represent the standard error of at least three independent experiments. 100% survival represents approximately 4×10^8 CFU/mL. (A) Anaerobic survival of *P. aeruginosa* PA14 after two days of anaerobic incubation with 40 mM pyruvate. At the start of anaerobiosis, potassium nitrate (50 mM), DCCD (2 mM), and/or CCCP (100 μ M) were added as indicated. DCCD and CCCP were dissolved in ethanol, and so as a control ethanol was added to a final concentration of 0.2% where indicated. A culture incubated without

pyruvate is shown for comparison (–pyruvate). (B) Anaerobic survival after two days of anaerobic incubation with 40 mM arginine. At the start of anaerobiosis, ethanol (0.2%), DCCD (2 mM), or CCCP (100 μ M) were added as indicated. (C) Effect of DCCD on membrane polarization. After 24 hours of anaerobiosis, 2 mM DCCD was added to cultures surviving anaerobically on 40 mM pyruvate (+DCCD, open circles). Since DCCD was dissolved in ethanol, an equivalent concentration of ethanol (0.2%) was added to control cultures (+ethanol, closed circles). Membrane polarization was monitored over time as described in Methods (for an example analysis, see Supplemental Figure 5). Time 0 indicates the addition of DCCD and ethanol to the cultures.

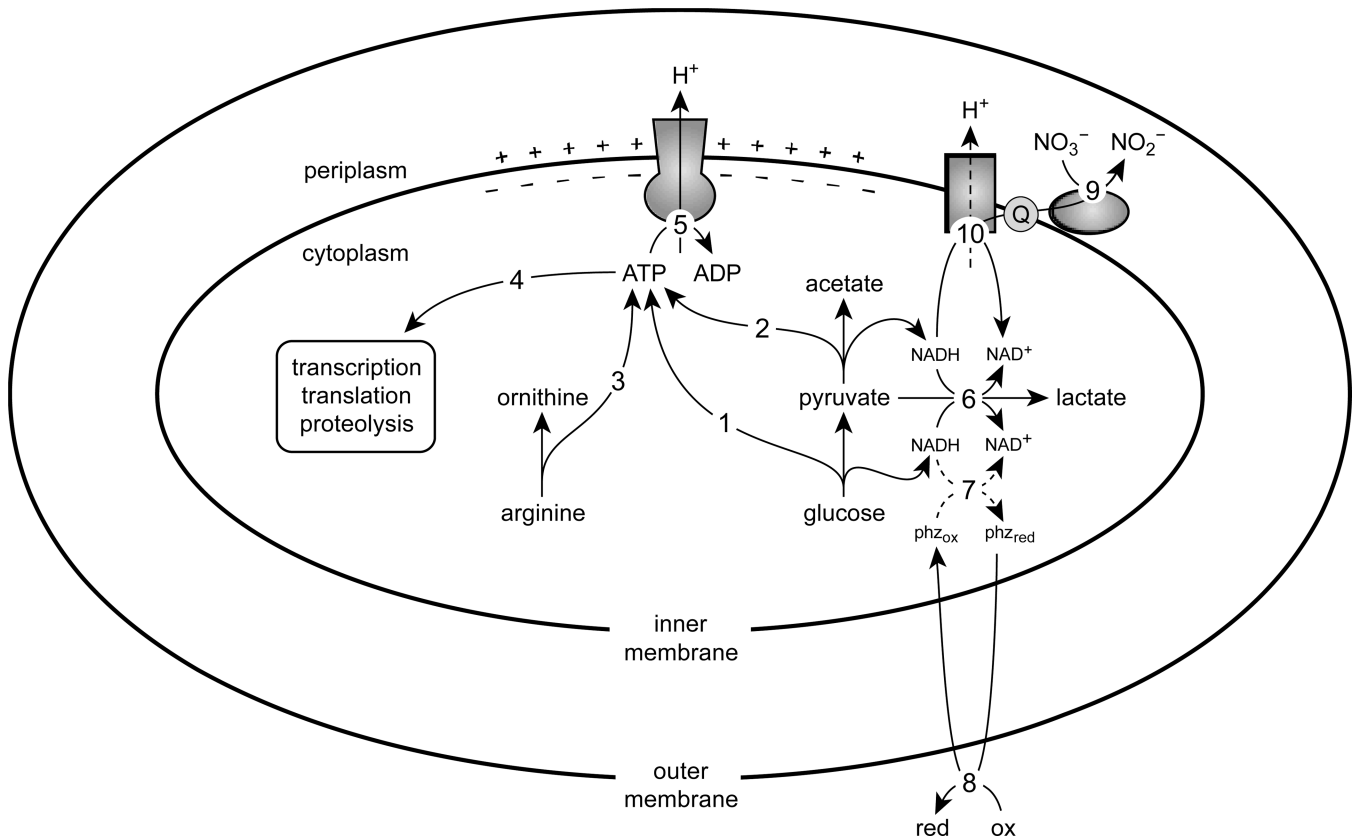


Figure 7.

A metabolic model of survival in *P. aeruginosa*. The ATP synthesized from glycolysis (1), pyruvate fermentation (2), and the arginine deiminase pathway (3) is used for essential processes including transcription, translation, and proteolysis (4). The F_1F_0 -ATPase complex hydrolyses ATP to translocate protons across the inner membrane (5), thereby generating a proton-motive force. Excess reducing equivalents from pyruvate oxidation can be dispensed by converting pyruvate to lactate (6). Phenazines can also regenerate oxidants for metabolism (7); although NAD(P)H is a possible electron donor for phenazines *in vivo* (Price-Whelan *et al.*, 2007), the dashed lines indicate that the nature of this interaction has not been elucidated. Phenazines are then oxidized extracellularly (8) and can be reused. Nitrate reductase can reduce nitrate to nitrite using electrons from the quinone pool (9), permitting oxidation of NADH via NADH dehydrogenase (10), which can result in proton translocation depending on which dehydrogenase complex is used.