

peptides range from 42.7 to 71.8 °C (14–16). Similarly, the antigenic properties of pHLAs also depend on the bound peptides. We previously reported that HLA-B*35:01-restricted cytotoxic T lymphocytes specific for two HIV-1 Nef peptides, VPLRPMTY (VY8) and RPQVPLRPMTY (RY11), showed substantially different antiviral activities despite their sequence similarity and the comparable binding affinity between each pHLA and cognate TCR (15). HLA-B*35:01 exhibited a higher melting temperature, longer cell surface presentation, and more potent susceptibility to killing by cytotoxic T lymphocytes when in complex with VY8 (HLA/VY8) than when in complex with RY11 (HLA/RY11). This relationship was further demonstrated by HLA-B*35:01 in complex with various variant peptides of VY8 and RY11, where the surface expression levels of pHLAs were controlled to be constant (15). Although the interaction mode between the pHLA and TCR is a key factor in determining the potency of T-cell responses (17, 18), these results suggest that pHLA stability itself has a significant impact on the mounting of efficient T-cell responses. However, the determinant of the peptide-dependent stabilization mechanism remains elusive. Observed differences in thermostability and antigenicity are difficult to explain by static structural differences between the peptide binding domain and the bound peptide itself. Rather, it is thought that the conformational dynamics of the pHLA affect its thermostability and antigenic properties.

In the present study we determined the crystal structure of HLA-B*35:01 in complex with RY11 and also examined conformational fluctuations of three HLA-B*35:01 complexes exhibiting different thermostability and cytotoxic T lymphocyte activity profiles using relaxation dispersion NMR spectroscopy (19, 20). Although the crystal structure was nearly identical to previously determined structures of HLA-B*35:01 in complex with different bound peptides, the NMR analyses identified peptide-dependent fluctuations and a transiently formed minor state at the interface of the peptide binding domain. Our data revealed that the minor state is responsible for peptide-dependent stabilization of HLA-B*35:01.

EXPERIMENTAL PROCEDURES

Sample Preparation—For crystallographic studies, residues 1–274 of the HLA-B*35:01 heavy chain were cloned into the plasmid pET21a (+) (Invitrogen) and expressed in BL21 (DE3) cells. The heavy chain was refolded with the HIV-1 RY11 peptide and β_2 -microglobulin and purified as described elsewhere (21). The collected protein fractions were concentrated to 10 mg/ml.

For NMR measurements, ^2H and ^{15}N or ^2H , ^5N , and ^{13}C uniformly labeled heavy chain and ^2H -labeled β_2 -microglobulin were expressed as inclusion bodies in BL21 (DE3) cells grown in M9 minimal medium. The protein-containing cells were lysed, and the pellets were separately dissolved in 6 M guanidine-HCl buffer. The HLA-binding peptides VY8, VY8(P5A), VY8(L3A), and RY11(P8A) were purchased from GenScript. The HLA complex was refolded in the presence of β_2 -microglobulin and one of the peptides and then purified as described previously (15).

Crystallography Methods—HLA/RY11 was crystallized via the hanging-drop vapor diffusion method at 18 °C. The crystals grew in a reservoir solution of 0.1 M MES monohydrate (pH 6.5) and 12% (w/v) polyethylene glycol 20,000. All crystals were flash-cooled in liquid nitrogen after a brief soaking in reservoir solution with the addition of 17% (v/v) glycerol. The x-ray diffraction data were collected at Shanghai Synchrotron Radiation Facility using beamline 17U. All data were processed with HKL2000 software (22).

The HLA/RY11 structure was solved by molecular replacement using Phaser (23) from the CCP4 program suite (24) with the structure of HLA/VY8 (PDB code 1A1N) serving as a search model. The initial model was refined using REFMAC5 (25), and extensive model building was performed using COOT (26). Further rounds of refinement were carried out using the phenix.refine program implemented in the PHENIX package (27), with energy minimization, group and individual B-factor refinement, bulk solvent modeling, and TLS refinement. The stereochemical quality of the final model was assessed with the PROCHECK program (28). The formula for R-factors is as stated in Brünger *et al.* (29). The structure was deposited in the PDB under accession number 4LNR. The mutated structures HLA/VY8(P5A), HLA/VY8(L3A), and HLA/RY11(P8A) were modeled from the structure of HLA/VY8 or HLA/RY11 using the PyMOL Molecular Graphics System, Version 1.6.0.0 (Schrödinger, LLC.) and YASARA force field (30).

NMR Methods—Three-dimensional transverse relaxation-optimized spectroscopy (TROSY) spectra of HNCO, HN(CA)CO, HNCA, HN(CO)CA, HNCACB, HN(CO)CACB, CC(CO)NH, and ^{15}N -NOESY were measured on an AVANCE DMX750 spectrometer (Bruker BioSpin) for sequential assignments of the backbone ^1H , ^{13}C , and ^{15}N chemical shifts (31) of the HLA complexes at 0.8–0.9 mM in NMR buffer (95% H_2O , 5% D_2O , 20 mM PBS (pH 7.40), 143 mM NaCl, 2 mM NaN_3). NMR data were processed using the program NMRPipe (32), and signals were assigned using the programs KUIJIRA (33) and NMRView (34). ^{15}N R_2 relaxation rates were measured on AVANCE DRX600 and AVANCE DMX750 spectrometers (Bruker BioSpin) using the TROSY-type constant-time Carr-Purcell-Meiboom-Gill pulse sequence (19, 20) at 10, 15, 20, and 25 °C for HLA/VY8(P5A) and at 20 °C for HLA/RY11(P8A) and HLA/VY8(L3A). The spectra were processed, and effective R_2 rates were calculated as described previously (35). Relaxation dispersion curves were fitted to a two-state exchange model (major \leftrightarrow minor) using the program GLOVE (36). The relaxation dispersion curves were initially fitted for each individual residue with fitting parameters of the population-average intrinsic relaxation rate (R_2^0), the exchange rate constant ($k_{\text{ex}} = k_{\text{major} \rightarrow \text{minor}} + k_{\text{minor} \rightarrow \text{major}}$), the chemical shift difference between states ($\Delta\omega$), and the product of the populations of the major and minor states ($p_{\text{major}}p_{\text{minor}}$). The population of the minor state was calculated according to the formula $p_{\text{minor}} = (1 - (1 - 4p_{\text{major}}p_{\text{minor}})^{1/2})/2$. The kinetic rates $k_{\text{major} \rightarrow \text{minor}}$ and $k_{\text{minor} \rightarrow \text{major}}$ were calculated as $k_{\text{major} \rightarrow \text{minor}} = p_{\text{minor}}k_{\text{ex}}$ and $k_{\text{minor} \rightarrow \text{major}} = p_{\text{major}}k_{\text{ex}}$, respectively. Data for which R_2 changed by $<1 \text{ s}^{-1}$ over the entire range of τ_{cp} or for which an F test showed less than a 99% confidence limit were excluded from further analysis. Because the fluctuating residues were

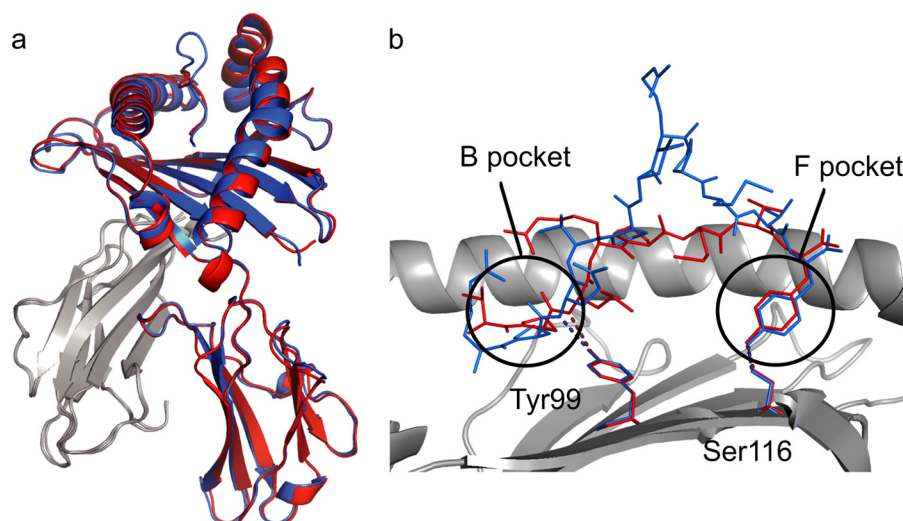


FIGURE 1. **Similarity of the crystal structures of HLA/VY8 and HLA/Ry11.** *a*, the crystal structures of HLA/VY8 (red; PDB code 1A1N) and HLA/Ry11 (blue; PDB code 4LNR) superposed using the HLA heavy chain. The structure of β_2 -microglobulin is shown in gray. *b*, closed view around the peptide-binding domain in the structures of HLA/VY8 (red) and HLA/Ry11 (blue). The regions around the B and F pockets are circled, and hydrogen bonds important for peptide recognition involving Tyr-99 and Ser-116 are represented as dashed lines.

concentrated in the peptide binding domain, all relaxation dispersions, except those for a few linker residues, were fitted globally with global parameters of k_{ex} and p_{major}/p_{minor} .

van't Hoff Fitting of Fluctuation Parameters—The equilibrium constant of the fluctuation between the major and minor states was calculated as $K_{eq} = k_{minor \rightarrow major}/k_{major \rightarrow minor}$, and $\ln K_{eq}$ was plotted as a function of the inverse of the absolute temperature, $1/T \times 10^3$. The data were fitted to the nonlinear van't Hoff equation (26), $RT \ln K_{eq} = \Delta H - T\Delta S + \Delta C_p(T - T_0) - T\Delta C_p \ln(T/T_0)$, where $T_0 = 293.15$, and R is the gas constant. The terms ΔC_p , ΔH , and ΔS represent the changes in heat capacity, enthalpy, and entropy, respectively.

Calculation of ΔC_p from the Buried Surface Area (BSA) Value—The BSA of the peptide and heavy chain interface was calculated using the Protein Interfaces, Surfaces, and Assemblies (PISA) service of the European Bioinformatics Institute (37).

Circular Dichroism Analysis—For circular dichroism measurements, the change in ellipticity (θ) with heating from 4 to 90 °C was monitored at 222 nm using a J-820 spectropolarimeter (JASCO), with a sample-cell volume of ~ 0.4 ml in a quartz cell with an optical path length of 2 mm at the rate of $1^\circ \text{C} \cdot \text{min}^{-1}$. The concentration of the sample was set between 0.1 and 0.2 mg/ml. The experiment was conducted single time for each sample. The T_m value in the circular dichroism analysis was calculated using the software provided by the manufacturer (JASCO).

RESULTS

Comparison of the Crystal Structures of Peptide-HLA-B*35:01 Complexes Exhibiting Different Thermostabilities—To address the correlation between the structure and thermostability of various peptide-HLA-B*35:01 complexes, we determined the crystal structure of HLA/Ry11 at a resolution of 2.0 Å (Fig. 1*a*; Table 1). The structure was compared with previously determined crystal structures of HLA-B*35:01 (Table 2), especially HLA/VY8, as HLA/VY8 is more stable than HLA/Ry11 despite the similar sequences of the bound peptides. The

TABLE 1

Data collection and refinement statistics (molecular replacement)

HLA/Ry11	
Data collection	
Space group	P2 ₁ 2 ₁ 2 ₁
Cell dimensions	
<i>a</i> , <i>b</i> , <i>c</i> (Å)	50.80, 81.42, 108.97
α , β , γ (°)	90.00, 90.00, 90.00
Resolution (Å)	50.00–2.00 (2.00–2.07) ^a
<i>R</i> _{sym} or <i>R</i> _{merge}	0.062 (0.331)
<i>I</i> / σ <i>I</i>	36.3 (8.1)
Completeness (%)	97.3 (96.2)
Redundancy	10.2 (9.8)
Refinement	
Resolution (Å)	24.00–2.00
No. reflections	29915
<i>R</i> _{work} / <i>R</i> _{free}	0.1887/0.2084
No. atoms	
Protein	3178
Ligand/ion	0
Water	443
<i>B</i> -factors	
Protein	29.1
Ligand/ion	0
Water	37.9
r.m.s.d.	
Bond lengths (Å)	0.003
Bond angles (°)	0.734

^a Values in parentheses are for the highest resolution shell.

overall structure of HLA/Ry11 was nearly identical to all of the 21 crystal structures of HLA-B*35:01, including HLA/VY8 (11), with all-atom r.m.s.d. of 0.22–1.09 Å for the heavy chain. However, the structures of the bound VY8 and Ry11 peptides differed, with the middle of Ry11 being bulged, as is the case with long HLA-binding peptides (Fig. 1*b*) (38, 39). Nevertheless, both peptides were bound to HLA-B*35:01 in the typical manner in which the B and F pockets prefer peptides with a proline residue at position 2 and either a tyrosine, phenylalanine, methionine, or leucine residue at the carboxyl terminus (40, 41). The remaining pockets (A, C, D, and E) exhibit little preference with respect to the amino acid sequence of the bound peptide (12, 13). The second proline and the carboxyl-terminal tyrosine in both VY8 and Ry11 form hydrogen bonds with Tyr-99 in the B pocket and Ser-116 in the F pocket, respectively

TABLE 2
Structural comparison of HLA-B*35:01

PDB ID	r.m.s.d. ^a	Peptide sequence	Comment	References
4LNR	0	RPQVPLRPMTY	HLA/RV11	
1A1N	0.597	RPQVPLRAMTY	HLA/RV11 (P8A)-modeled structure	11
	0.358	VPLRPMTY	HLA/VY8	
	0.566	VPARPMTY	HLA/VY8 (L3A)-modeled structure	
	0.524	VPLRAMTY	HLA/VY8 (P5A)-modeled structure	
2CIK	0.36	KPIVVHLGY		52
1ZHK	0.365	LPEPLPQGQLTAY		60
2FYY	0.208	HPVGEADYFEY		55
2H6P	0.384	KPIVVHLGY		61
3LKP	0.299	LPFDKSTIM		53
3LKN	0.37	LPFERATIM		53
3LKS	0.367	LPFEKSTVM		53
3LKQ	0.347	LPFDKTTIM		53
1ZSD	0.167	EPLPQGQLTAY		59
1XH3	0.153	LPAVVGLSPGEQY		38
3LKO	0.243	LPFDRTTIM		53
2AXG	0.186	APQPAPENAY		57
3LKR	0.252	LPFERATVM		53
3MV8	0.552	HPVGEADYFEY	Complex with TCR	54
3MV9	0.856	HPVGEADYFEY	Complex with TCR	54
3MV7	0.82	HPVGEADYFEY	Complex with TCR	54
2NX5	0.946	EPLPQGQLTAY	Complex with TCR	58
1A9B	1.063	LPPLDITPY		56
1A9E	1.029	LPPLDITPY		56
1CG9	1.064	LPPLDITPY		56

^a The r.m.s.d. of the all atom of heavy chain was calculated using PyMOL Molecular Graphics System, Version 1.6.0.0. (Schrödinger, LLC.) against the structure of HLA/RV11.

(Fig. 1b). Curiously, HLA/RV11 is less thermostable, and the RV11 peptide is easy to dissociate from HLA (15), but the BSA of the HLA-peptide contact surface of HLA/RV11 was similar to that of HLA/VY8 (1780 Å² versus 1710 Å²). Although RV11 has a larger B-factor than the other part of HLA/RV11, no significant difference in B-factor was found in other parts of HLA/RV11 and HLA/VY8. It is known that pHLA with bulged peptides tends to have low thermostability. However, it is unclear from the crystal structures alone how the large thermostability difference could arise by the difference in the peptide.

Analysis of the Conformational Dynamics of HLA-B*35:01 in Solution—We attempted to analyze the conformational dynamics of HLA/VY8 and HLA/RV11 in solution by relaxation dispersion NMR spectroscopy. However, disintegration and aggregate formation hampered the relaxation dispersion measurements. The aggregation was most likely due to insufficient stability of the pHLAs under the sample condition and NMR measurements (~1 week) employed. Therefore, we used a VY8 peptide variant containing an Ala substituted for Pro-5 (VY8(P5A)), as this variant is known to improve thermostability while preserving TCR specificity (15). There was no significant interaction observed between Pro5 and the heavy chain, whereas the change in BSA resulting from the mutation was very small (Δ BSA = 1% of the total interaction surface). The HSQC spectrum of HLA/VY8(P5A) was virtually identical to that of HLA/VY8 (Fig. 2). The secondary structure of HLA/VY8(P5A) estimated from the ¹³C_α, ¹³C_β, ¹³CO, and ¹⁵N chemical shifts closely matched that of the crystal structure of HLA/VY8.

The relaxation dispersion experiment identified fluctuating residues in the peptide binding domain of HLA/VY8(P5A) at 20 °C (Fig. 3a). No fluctuation was observed in the Ig-like C1-type domain. We fitted all the relaxation dispersions globally to a two-state exchange model (major ↔ minor), yielding an exchange rate ($k_{ex} = k_{major \rightarrow minor} + k_{minor \rightarrow major}$) of $766 \pm 41 \text{ s}^{-1}$ and a minor population (p_{minor}) of $4.37 \pm 0.09\%$. Anal-

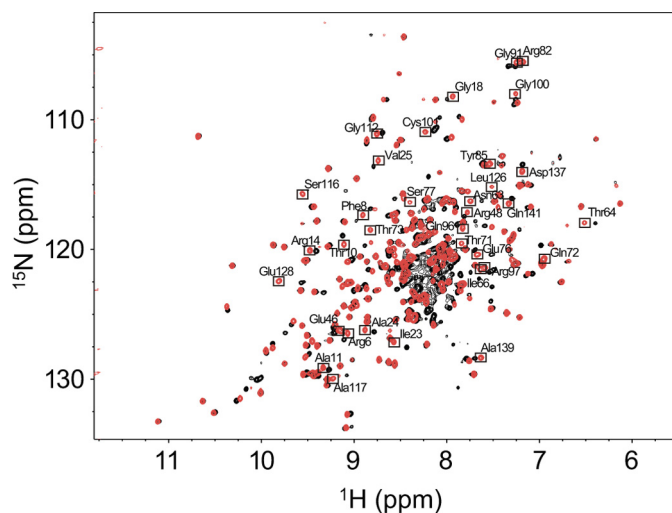


FIGURE 2. HSQC spectra of HLA/VY8 and HLA/VY8(P5A). The ¹H,¹⁵N TROSY HSQC spectra of HLA/VY8 and HLA/VY8(P5A) are shown in black and red, respectively. The resonances of HLA/VY8(P5A) that do not overlap with those of HLA/VY8 are boxed with the assignments.

ysis of the chemical shift differences ($\Delta\omega$, representing the amplitude of the conformational change between the major and minor states) showed large-amplitude conformational changes in the B and F pockets compared with the other fluctuating regions (Fig. 3a). It is of note that the fluctuating residues included Tyr-99 in the B pocket and Ser-116 in the F pocket, both of which are important for peptide recognition by HLA-B*35:01.

To further characterize the minor state, we also measured relaxation dispersions of HLA/VY8(P5A) at 10, 15, and 25 °C. The exchange rates obtained at these temperatures were converted to the equilibrium constants ($K_{eq} = k_{minor \rightarrow major}/k_{major \rightarrow minor}$) and fitted to the van't Hoff equation (Fig. 3b). This analysis provided a change in the heat capacity (ΔC_p) of $-0.88 \pm 0.46 \text{ kcal}\cdot\text{mol}^{-1}\cdot\text{K}^{-1}$, a change in enthalpy (ΔH) of -261 ± 138

Peptide-dependent Conformational Dynamics of HLA

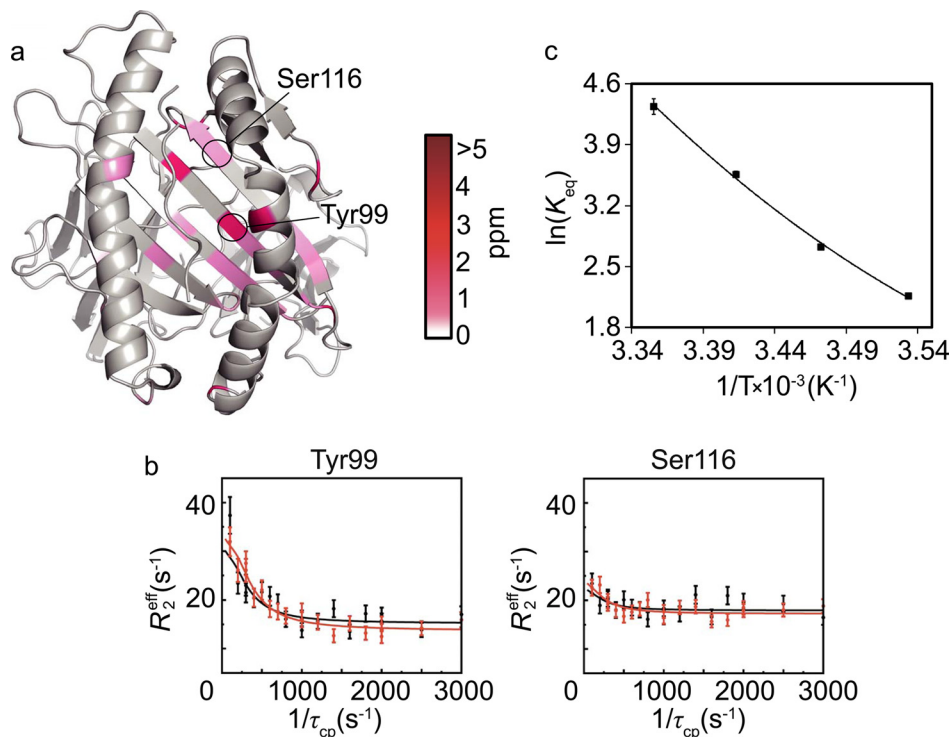


FIGURE 3. **Conformational fluctuations of HLA/VY8(P5A).** *a*, chemical shift differences ($\Delta\omega$) between the major and minor states plotted on the crystal structure of HLA/VY8 as a continuous color scheme from gray to red. The positions around Tyr-99 and Ser-116, which are important for peptide recognition, are circled. *b*, relaxation dispersion profiles for Ser-116 and Tyr-99 recorded at 14.1 tesla (black) and 17.6 tesla (red). *c*, the van't Hoff plot for the fluctuations of pHLA between the major and minor states. The best-fit curve is shown as a solid line. The determinant factor (R^2) is 0.99.

$\text{kcal}\cdot\text{mol}^{-1}$, and a change in entropy (ΔS) of $-6.16 \pm 3.22 \text{ cal}\cdot\text{mol}^{-1}\cdot\text{K}^{-1}$. It is generally appreciated that a negative ΔC_p value for a protein complex is strong evidence that dehydration has occurred due to interaction and/or induced fit (42, 43). In the case of HLA/VY8(P5A), only $-0.25 \text{ kcal}\cdot\text{mol}^{-1}\cdot\text{K}^{-1}$ of the total ΔC_p value was calculated as being due to interaction with the peptide. Therefore, the remainder of the ΔC_p value ($-0.63 \text{ kcal}\cdot\text{mol}^{-1}\cdot\text{K}^{-1}$) can be ascribed to an induced-fit process. Because the secondary structure predicted by the chemical shift matched that of the crystal structure well, the fluctuating regions would already have formed a well folded conformation in the major state. Therefore, the minor state most likely forms a more packed dehydrated conformation. Note that the peptide binding domain is composed of many fluctuating hydrophobic residues (Fig. 4), and dehydration in a hydrophobic area generally contributes to the formation of a strong interaction (43). Furthermore, the negative ΔH and ΔS values are both consistent with a more packed dehydrated conformation because they can be interpreted as indicating the formation of new electrostatic- or hydrogen-bonding interactions and an increasingly ordered state, respectively (44). Therefore, it is likely that the interaction between the peptide binding domain and the peptide becomes stronger in the dehydrated minor state.

Together with the nearly identical crystal structures of the pHLAs, these results led us to hypothesize that the minor state is important for the stabilization of the pHLA to prevent disintegration of the complex. We also hypothesized that the loosely packed major conformation enables binding of a variety of peptides without inducing substantial structural changes in HLA.

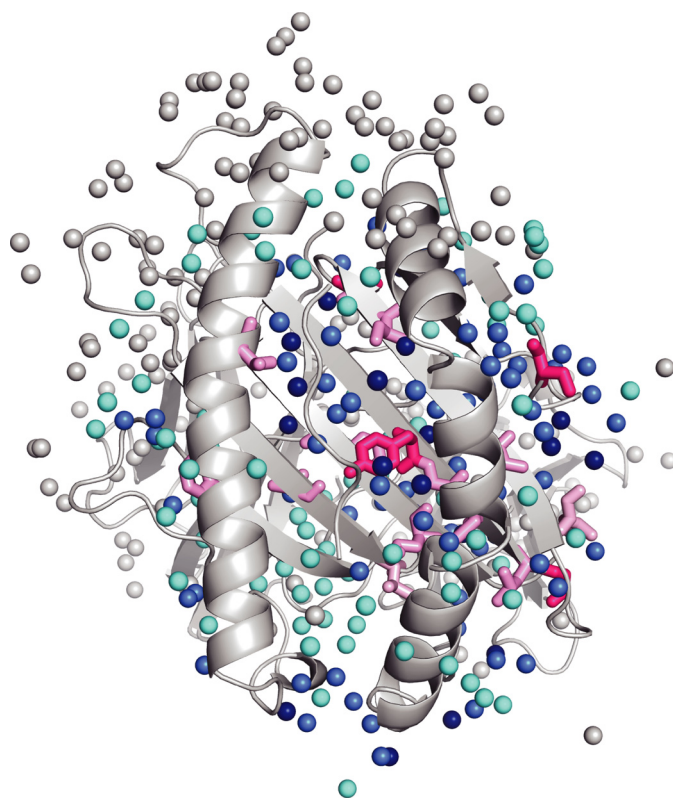


FIGURE 4. **The water molecules around the fluctuating residues.** Water molecules are shown as spheres in the HLA/VY8 structure. The fluctuating hydrophobic residues are shown as sticks in the same color scheme as in Fig. 2. Dark blue, blue, and cyan waters are located within 3.2, 6.4, and 9.6 Å from the fluctuating residues, respectively.

Comparison of the Minor State among Different HLA-B*35:01 Complexes—We postulated that pHLAs of differing thermostability have minor states that differ in abundance and/or nature. To test this hypothesis, we analyzed the conformational dynamics of HLA-B*35:01 in complex with other peptide variants: Ala at p8 of RY11 and Ala at p3 of VY8, designated as RY11(P8A) and VY8(L3A), respectively. The peptide variants differentially increased the stability of the pHLAs in the previous study while preserving the specificity of the cognate TCR (15). We avoided introducing mutations into the fluctuating peptide binding domain of the HLA heavy chain because such mutations may cause unwanted effects, such as changing the peptide sequence preference.

The HSQC spectra of both HLA/RY11(P8A) and HLA/VY8(L3A) were similar to that of HLA/VY8(P5A) (Fig. 5). The resonances of the Ig-like C1-type domain of HLA/RY11(P8A)

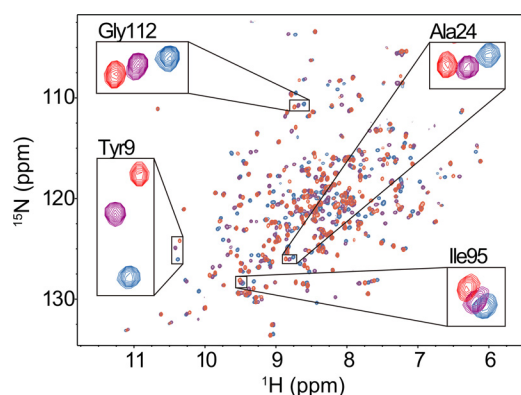


FIGURE 5. Similarity in the solution structures of HLA/VY8(P5A), HLA/VY8(L3A), and HLA/RY11(P8A). ^1H , ^{15}N TROSY HSQC spectra of HLA/VY8(P5A), HLA/VY8(L3A), and HLA/RY11(P8A) are shown in red, purple, and blue, respectively. Residues showing relatively large chemical shift differences in the peptide binding groove are shown in the inset.

and HLA/VY8(L3A) superimposed well on those of HLA/VY8(P5A), whereas many residues in the peptide binding domains of HLA/RY11(P8A) and HLA/VY8(L3A) showed slightly different chemical shifts compared with those of HLA/VY8(P5A) (Figs. 5 and 6). Almost all the fluctuating residues in HLA/VY8(P5A) have different chemical shifts of HLA/RY11(P8A) and HLA/VY8(L3A) (Fig. 7). This suggests that the NMR signals were observed at the population-averaged chemical shifts of the major and minor state. Interestingly, most of the shifted resonances of HLA/VY8(L3A) appeared in between those of HLA/VY8(P5A) and HLA/RY11(P8A), suggesting that the minor populations are different among the pHLAs. The residues showing chemical shift differences between the pHLAs were spread over a large area of the peptide binding domain and included residues that have no direct contacts with the peptide.

The relaxation dispersion experiments with HLA/RY11(P8A) and HLA/VY8(L3A) showed that both pHLAs fluctuate in the peptide binding domain, as was the case for HLA/VY8(P5A) (Fig. 8). However, fewer residues fluctuated in HLA/RY11(P8A) and HLA/VY8(L3A) than in HLA/VY8(P5A). For HLA/VY8(L3A), the fluctuating residues were located more distal from the center of the peptide binding domain than were those in HLA/VY8(P5A). For HLA/RY11(P8A), only a small number of residues near the A and B pockets fluctuated. It is notable that Tyr-99 and Ser-116, which are responsible for peptide binding, fluctuated only in HLA/VY8(P5A) (Fig. 5), suggesting that fluctuations at these residues play an important role in the peptide-dependent minor state. Fitting of relaxation dispersion curves yielded p_{minor} of $1.25 \pm 0.05\%$ and k_{ex} of $742 \pm 48 \text{ s}^{-1}$ for HLA/VY8(L3A) and p_{minor} of $0.58 \pm 0.02\%$ and k_{ex} of $1590 \pm 96 \text{ s}^{-1}$ for HLA/RY11(P8A). The minor populations of both HLA/RY11(P8A) and HLA/VY8(L3A) were smaller than that of HLA/VY8(P5A) (4.37%). The kinetic rate

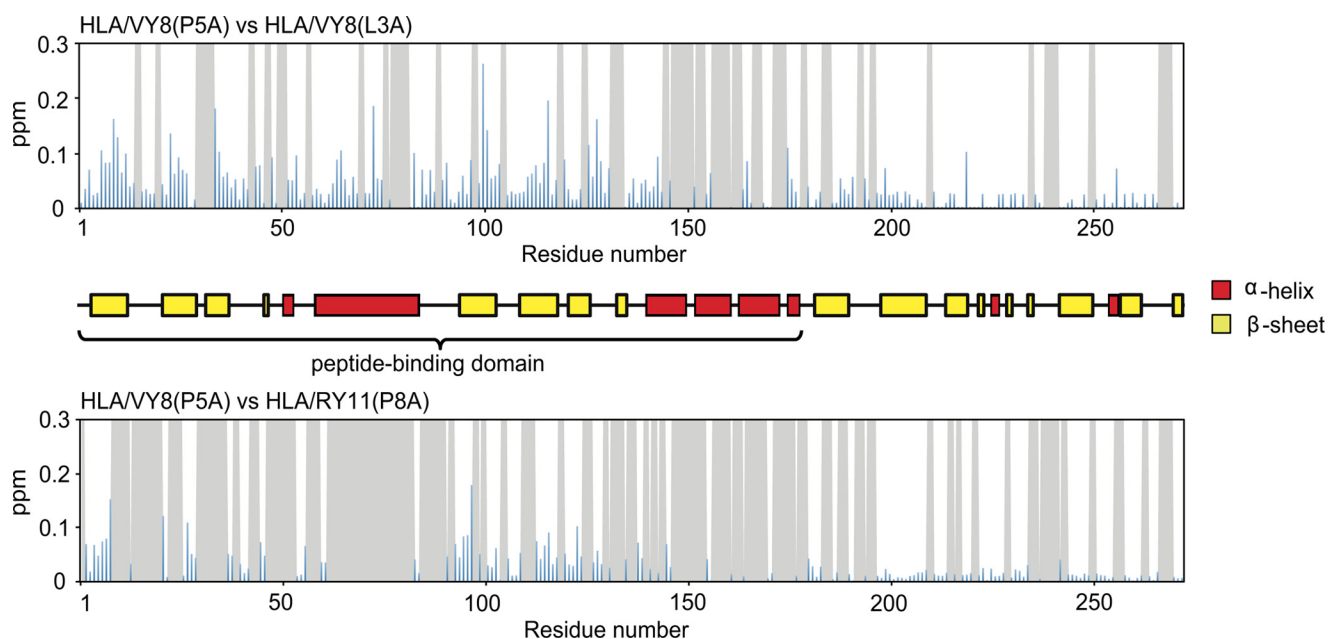


FIGURE 6. The chemical shift differences of HLA/VY8(L3A) and HLA/RY11(P8A) from HLA/VY8(P5A). The chemical shift differences $(\Delta\delta_{\text{H}})^2 + (\Delta\delta_{\text{N}}/5)^2$ between HLA/VY8(P5A) and HLA/RY11(P8A) and between HLA/VY8(P5A) and HLA/VY8(L3A) are shown as blue bars in the upper and lower graphs, respectively. The gray background shows the unassigned residues in HLA/VY8(L3A) and HLA/RY11(P8A). The yellow and red boxes in the middle indicate the secondary structural components observed in the crystal structure of HLA/VY8.

Peptide-dependent Conformational Dynamics of HLA

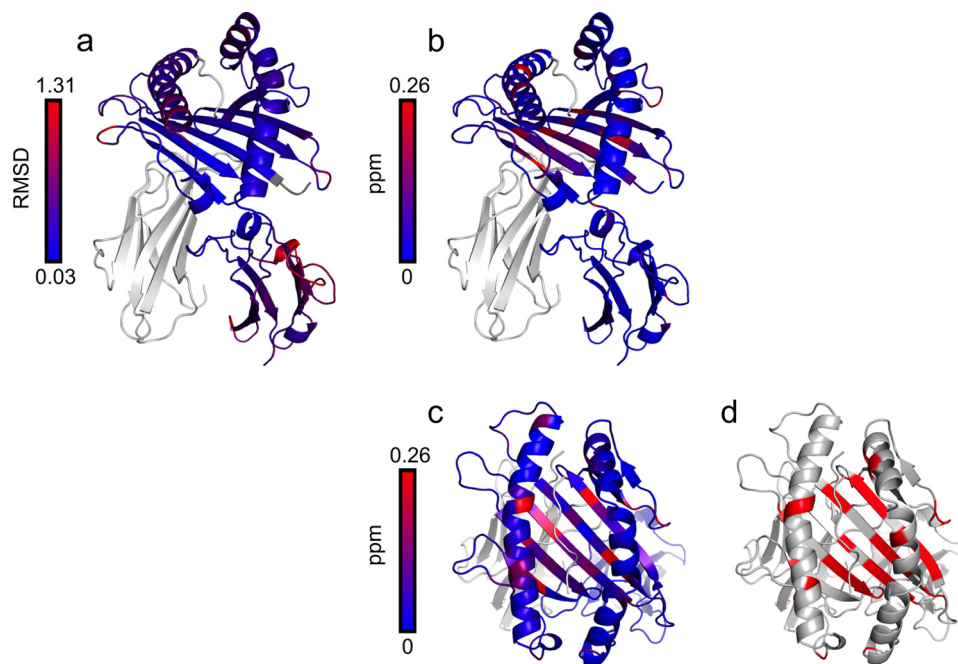


FIGURE 7. **Comparison of the r.m.s.d. differences and chemical shift differences.** *a*, the r.m.s.d. for each residue in the crystal structures of HLA/VY8 and HLA/Ry11 are shown. *b* and *c*, the chemical shift differences between HLA/VY8(P5A) and HLA/VY8(L3A) are shown from different angles. *d*, the residues showing relaxation dispersions in HLA/VY8(P5A) and HLA/VY8(P3A) are shown in red.

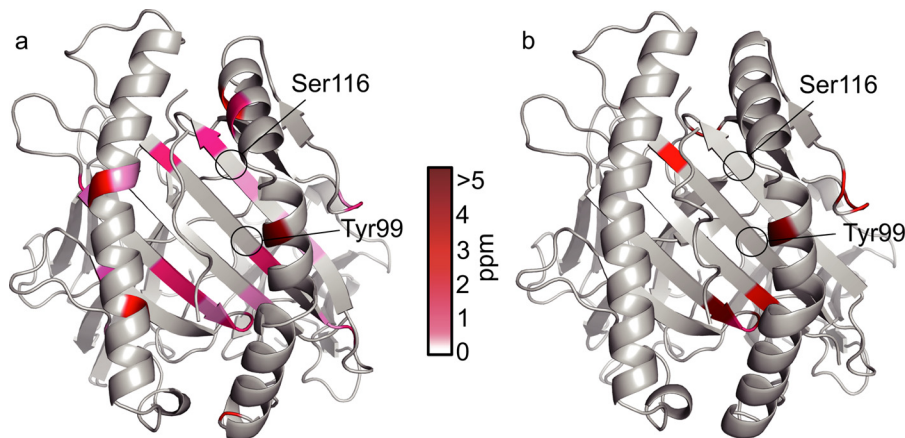


FIGURE 8. **Conformational fluctuations of HLA/VY8(L3A) and HLA/Ry11(P8A).** Chemical shift differences ($\Delta\omega$) are plotted on the crystal structures of HLA/VY8 (*a*) and HLA/Ry11 (*b*) in the same manner as in Fig. 3*a*.

constants $k_{\text{major} \rightarrow \text{minor}}$ and $k_{\text{minor} \rightarrow \text{major}}$ had an interesting correlation among the pHLAs (Fig. 9). HLA/Ry11(P8A) and HLA/VY8(L3A) had similar $k_{\text{major} \rightarrow \text{minor}}$ rates (9.29 ± 0.07 and $9.33 \pm 0.07 \text{ s}^{-1}$, respectively) which were slower than that of HLA/VY8(P5A) ($33.54 \pm 0.06 \text{ s}^{-1}$). In contrast, HLA/VY8(L3A) and HLA/VY8(P5A) had similar $k_{\text{minor} \rightarrow \text{major}}$ rates (733.55 ± 0.06 and $732.56 \pm 0.07 \text{ s}^{-1}$, respectively), which were slower than that of HLA/Ry11(P8A) ($1581.06 \pm 0.07 \text{ s}^{-1}$). Thus, HLA/VY8(L3A) exhibits the characteristics of HLA/VY8(P5A) regarding the minor-to-major transition and the characteristics of HLA/Ry11(P8A) regarding the major-to-minor transition. A comparison of the fluctuating regions among the three pHLAs suggested that the $k_{\text{minor} \rightarrow \text{major}}$ rate is determined by the residues in the E and F pockets fluctuating in HLA/VY8(L3A) and HLA/VY8(P5A) and that the $k_{\text{major} \rightarrow \text{minor}}$ rate is determined by the residues in the D pocket fluctuating only in HLA/VY8(P5A) (Fig. 9). Analysis of the relaxation dis-

persions also revealed chemical shift differences ($\Delta\omega$) between the major and minor states of HLA/Ry11(P8A) and HLA/VY8(L3A). Almost all of the $\Delta\omega$ values of HLA/Ry11(P8A) and HLA/VY8(L3A) were larger than those of HLA/VY8(P5A) (Figs. 3*a* and 8), suggesting that the conformational changes from the major to the minor states are larger for HLA/Ry11(P8A) and HLA/VY8(L3A) than for HLA/VY8(P5A).

Analysis of the Correlation between Conformational Dynamics, Thermostability, and pHLA Decay on the Cell Surface—We conducted thermostability analyses of HLA/VY8(L3A), HLA/VY8(P5A), and HLA/Ry11(P8A) using circular dichroism. The melting temperatures were 50.1°C for HLA/Ry11(P8A), 57.7°C for HLA/VY8(L3A), and 61.4°C for HLA/VY8(P5A). These values are different from our previous determined values (15). The differences were caused most likely because we used a different instrument and newly prepared samples. Although the values were different, we observed the T_m ten-

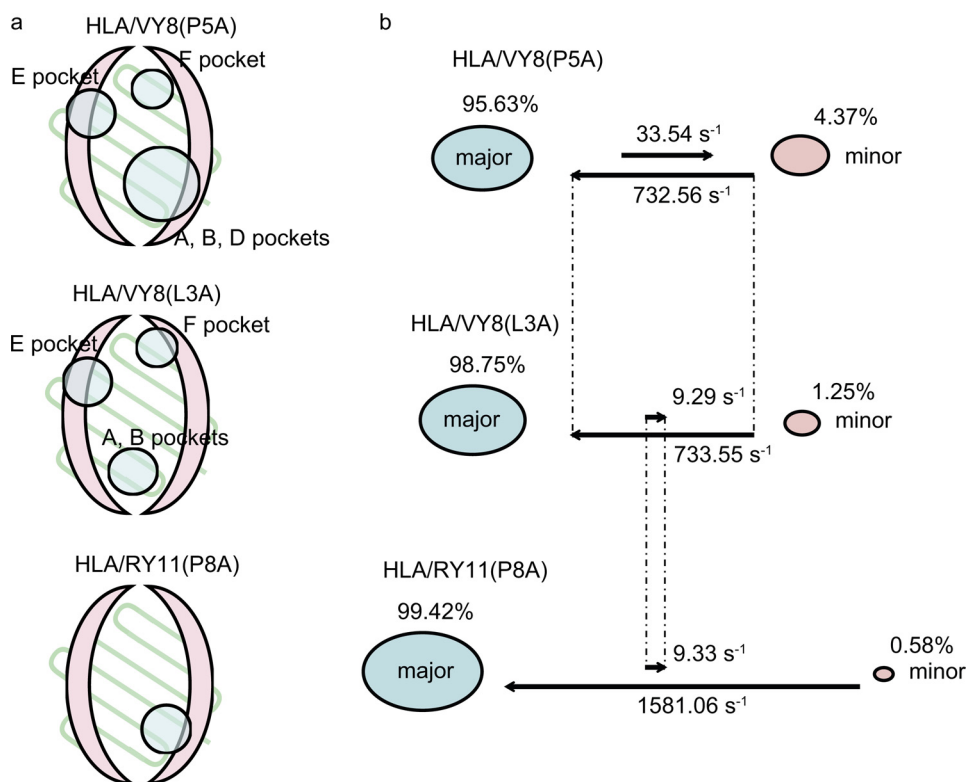


FIGURE 9. **Fluctuating regions in three pHLA complexes.** *a*, fluctuating regions in HLA/VY8(P5A), HLA/VY8(L3A), and HLA/RV11(P8A) are shown as schematic drawings. The α -helices and β -strands in the peptide binding domain are shown in pink and light green, respectively. The fluctuating regions are circled with the name of the pockets. *b*, the kinetic rates and the populations of the major and minor states are visualized for the three pHLA complexes.

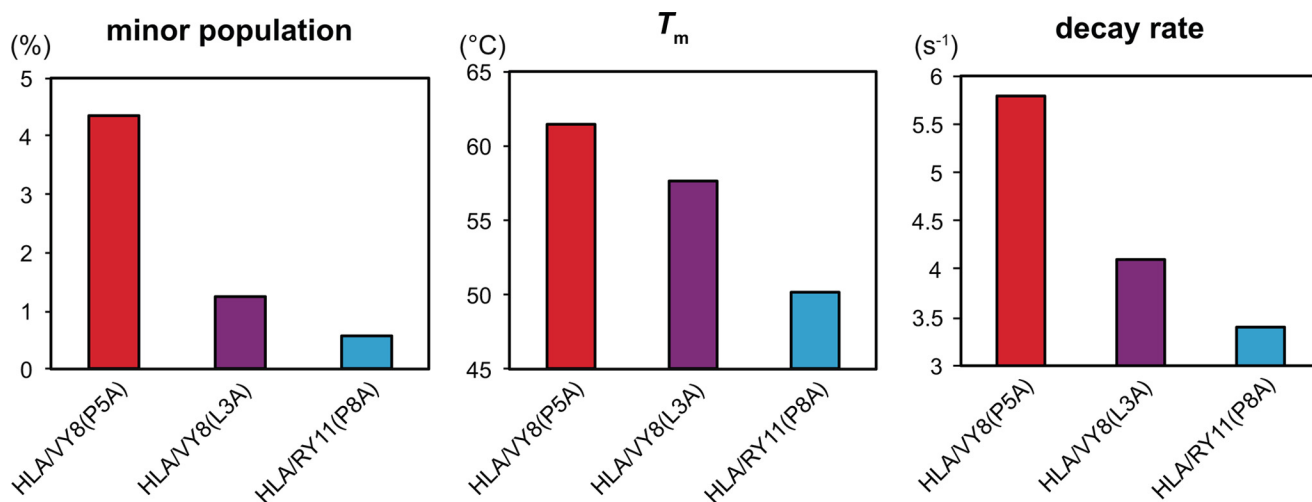


FIGURE 10. **Correlation between minor populations, stability, and T-cell activity decay rate.** The left, middle, and right graphs show the minor populations, melting temperatures, and T-cell activity decay rates of the three pHLAs. The formula $y = a \times \exp(-b \times t)$ was used to calculate the T-cell activity decay rate with previously reported data (15), where t represents the time for activity measurement, and $\log b$ was determined as the decay rate.

dency (HLA/VY8(P5A) > HLA/RV11(P8A)) being preserved. Interestingly, as the melting temperatures of the three pHLAs increased, so did the sizes of the minor populations determined by the relaxation dispersion experiments (Fig. 10). In other words, the pHLAs with larger major populations tended to disintegrate more rapidly. These results strongly support our hypothesis that the minor state is responsible for pHLA thermostability and that the peptides are loosely bound in the major state.

We then calculated the pHLA decay rates on the antigen-presenting cell surface measured by the cognate T-cell

responses determined from previously published data (15). Interestingly, the rates of pHLA decay on the antigen-presenting cell surface also correlated well with the minor populations. That is, the larger the minor populations, the longer the pHLAs were presented on the cell surface (Fig. 10).

DISCUSSION

In this study we report the conformational fluctuations that are responsible for the stabilization of peptide-HLA-B*35:01 complexes. In solution, pHLAs predominantly form a bound peptide-independent conformation that would be similar to the

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crystal structure except that they form a peptide-dependent dehydrated minor conformation at the peptide binding domain. The minor populations of the pHLAs correlated well with the length of time antigenic peptides are presented on the cell surface, which is important for T-cell recognition. It is, therefore, likely that the conformational fluctuations that we observed are part of the molecular basis of pHLA antigenicity.

A number of crystal structures of HLA class I allomorphs in complex with various peptides have been reported, but the structures of each HLA heavy chain are virtually identical (Table 2). In contrast, thermodynamic analyses using differential scanning calorimetry, circular dichroism (14, 15), infrared spectrometry (45), and hydrogen/deuterium exchange mass spectrometry (46) have indicated that pHLAs are flexible in solution. These studies indicated that the overall flexibility of pHLAs differs depending on the bound peptides, suggesting the importance of the fluctuation of HLA on its function (47). However, they did not show how pHLAs are stabilized by the bound peptides.

Here, we characterized for the first time peptide-dependent site-specific pHLA fluctuations. Our results enabled us to explain observed peptide-dependent differences in the stability of various pHLAs. For example, Tyr-99 and Ser-116, which are important for peptide recognition by HLA-B*35:01, fluctuated only in the most stable pHLA analyzed in this study (HLA/VY8(P5A)). This result suggests that peptide recognition by a pHLA and further stabilization of the pHLA are dynamically regulated by conformational fluctuations.

With respect to the process of pHLA disintegration, previous molecular dynamic simulation studies indicated that in the absence of a bound peptide, HLA-B*27, becomes partially unfolded on a nanosecond time scale (48). Winkler *et al.* (49) showed that the disintegration of HLA-B*27 is initiated by dissociation of the peptide or β_2 -microglobulin, with disintegration rates of 10^{-4} – 10^{-7} s $^{-1}$, depending on the peptide. Consistent with these data, we previously reported that peptide-HLA-B*35:01 complexes, including HLA/VY8(L3A), HLA/VY8(P5A), and HLA/RV11(P8A), disintegrate with rate constants of 1×10^{-6} – 4×10^{-6} s $^{-1}$ (15). In contrast, the rate constants of the major-to-minor transition observed in this study (9–34 s $^{-1}$) were much faster than the disintegration rate constants. Taken together, these data suggest that a pHLA rapidly forms the more packed minor conformation (constituting only up to 5% of the population) before it enters the disintegration pathway.

How does the mutation incorporated into the position where the side chain of the residue points out of the binding pocket affect the fluctuation of the residues even at the bottom of the peptide-binding groove? As the mutation changed the chemical shifts of the wide area of the heavy chain including the regions distal to the peptide (Fig. 2), the fluctuation change in the peptide by the mutation may be transmitted to the heavy chain through the intermolecular interactions between the peptide and the heavy chain and through the intramolecular interactions in the heavy chain.

What is the physiological role of the pHLA major state? Antigenic peptides are selected and loaded onto HLA class I proteins at the endoplasmic reticulum (ER) by various ER-resident

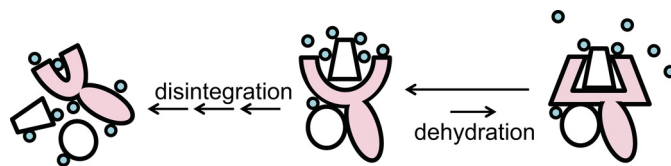


FIGURE 11. **The pHLA transient induced-fit model.** A schematic illustration of pHLA fluctuation is shown. The HLA heavy chain is shown in pink, and the peptide and β_2 -microglobulin are represented as a trapezoid and a circle, respectively. Water molecules are shown as light blue circles.

chaperones in the so-called “peptide-loading complex” (50). As a component of this large complex, HLA would need to assume a similar conformation to accommodate a peptide. Our data suggest that the major state of HLA seems to accommodate an antigenic peptide more loosely compared with the well packed minor conformation, as if it allows the loading of various structurally different peptides in a similar structure. We propose such a two-step “transient induced-fit model” for the loading of a peptide to HLA. In the first step, the major state loads a peptide loosely, without the need for conformational adjustment, and in the second step, conformational tuning to the well packed minor state enables the pHLA to avoid rapid disintegration (Fig. 11). It would be intriguing to examine whether such dynamic conformational transitions of pHLA play a role in the endoplasmic reticulum peptide-loading complex as well as in TCR binding on the cell surface.

The limited use of an HLA class I allomorph in this study may be of concern. It is not sure if a similar correlation between the minor population and the thermostability will generally be observed in other HLAs. The analysis of various HLA class I allomorphs in complex with different peptides of immunological interest using the same NMR approach is, therefore, warranted. Elucidating the stabilization mechanisms of various pHLAs will help facilitate the rational design of more stable therapeutically useful pHLAs (51).

Acknowledgments—We thank Dr. Jose M. M. Caaveiro for critically reading the manuscript and Hao Cheng for protein preparation.

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