Kinetics of Cellobiohydrolase (Cel7A) Variants with Lowered Substrate Affinity

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Jeppe Kari^{‡1}, **Johan Olsen**^{‡1}, **Kim Borch**[§], **Nicolaj Cruys-Bagger**[‡], **Kenneth Jensen**[§], **and Peter Westh**^{‡2} From [‡]Department of Science, Systems and Models, Research Unit for Functional Biomaterials, Roskilde University, 1 Universitetsvej, DK-4000 Roskilde, Denmark and [§]Novozymes A/S, Krogshøjvej 36, Bagsværd DK-2880, Denmark

Background: To elucidate the rate-determining steps of cellobiohydrolase Cel7A from *T. reesei*, variants with lower sub-strate affinity were designed.

Results: Mutant (W38A) had reduced substrate affinity but a 2-fold increase in the maximum quasi-steady-state rate.

Conclusion: Dissociation of stalled *Tr*Cel7A is the rate-limiting step in the initial phase of hydrolysis. **Significance:** This work offers a new perspective for the design of faster cellulases.

Cellobiohydrolases are exo-active glycosyl hydrolases that processively convert cellulose to soluble sugars, typically cellobiose. They effectively break down crystalline cellulose and make up a major component in industrial enzyme mixtures used for deconstruction of lignocellulosic biomass. Identification of the rate-limiting step for cellobiohydrolases remains controversial, and recent reports have alternately suggested either association (on-rate) or dissociation (off-rate) as the overall bottleneck. Obviously, this uncertainty hampers both fundamental mechanistic understanding and rational design of enzymes with improved industrial applicability. To elucidate the role of on- and off-rates, respectively, on the overall kinetics, we have expressed a variant in which a tryptophan residue (Trp-38) in the middle of the active tunnel has been replaced with an alanine. This mutation weakens complex formation, and the population of substrate-bound W38A was only about half of the wild type. Nevertheless, the maximal, steady-state rate was twice as high for the variant enzyme. It is argued that these opposite effects on binding and activity can be reconciled if the rate-limiting step is after the catalysis (i.e. in the dissociation process).

Enzymatic breakdown of crystalline cellulose has proven challenging to describe in molecular detail. This is probably the result of a complex enzyme architecture with different domains and moieties that interacts with the heterogeneous surface of the insoluble substrate (1–4). Moreover, the linear β -1,4-glucan chains in the highly condensed substrate are connected by strong intermolecular forces, which compete with enzymesubstrate interactions during the hydrolysis and hence add to the complexity of the process (5, 6). The most extensively studied cellulase is the cellobiohydrolase Cel7A from *Trichoderma reesei* (an anamorph of the fungus *Hypocrea jecorina*). *Tr*Cel7A is a multidomain enzyme comprising a catalytic domain (core), which is connected by a highly glycosylated linker peptide to a family 1 carbohydrate-binding module (CBM1)³ (7). The threedimensional crystal structure of the catalytic domain in complex with different cello-oligosaccharides has been solved, revealing a 50-Å-long active site tunnel with 10 glucosyl-binding sites, numbered from -7 at the entrance of the tunnel to +3at the exit (8). The enzyme hydrolyzes cellulose processively from the reducing end with the scissile bond between subsite -1 and +1, hence leaving cellobiose in the +1, +2 expulsion site. Analysis of processive hydrolysis requires consideration of a number of steps, including adsorption, surface diffusion, recognition, "threading," complex formation, catalysis, product expulsion, processive sliding, and dissociation (4, 9). Many of these steps remain poorly understood partly because their isolation has proven difficult in experiments (4) and partly as a result of the heterogeneous environment in which these enzyme work (2, 10).

It is of particular importance to identify the slowest and hence rate-determining step in the catalytic cycle. Numerous earlier studies have addressed this question, and interestingly, it appears as if the conclusions fall mainly into two mutually conflicting groups. Thus, a number of works have used either direct experimental evidence or re-examination of broader, previously published material to conclude that the overall rate is governed by the association of enzyme and substrate, particularly the placement of a cellulose strand in the long active tunnel (11-18). In contrast to this, other studies, including work by this group on the pre-steady state kinetics of wild type Cel7A, have suggested that the formation of the activated complex is quite rapid and that dissociation of enzyme, which is in a position where further processive movement is hindered, is the rate-limiting step (19-25). To elucidate this question, we have designed, expressed, and purified variants of TrCel7A with lower affinity for the substrate. The idea was that the kinetic behavior of an enzyme with a less stable substrate complex could reveal whether the bottleneck was located before or after the activated complex (i.e. whether rate limitation was one of



¹ Both authors contributed equally to this work.

² To whom correspondence should be addressed: Research Unit for Functional Biomaterials, NSM, Roskilde University, 1 Universitetsvej, 4000 Roskilde, Denmark. Tel.: 45-4674-2879; Fax: 45-4674-3011; E-mail: pwesth@ ruc.dk.

³ The abbreviations used are: CBM, carbohydrate-binding module; BisTris, 2-[bis(2-hydroxyethyl)amino]-2-(hydroxymethyl)propane-1,3-diol.

Kinetics of Off-rate Variants: W38A



FIGURE 1. Structure of TrCeI7A. Shown is a surface representation of TrCeI7A in complex with a cellulose chain and highlighted tryptophan residues in the catalytic tunnel. The image was made in PyMOL (Schrödinger, LLC, New York) using the crystal structure of the catalytic domain (Protein Data Bank code 8CEL) and carbon binding module (Protein Data Bank code 1CBH).

the associating or dissociating steps). Specifically, we replaced Trp-38 located in subsite -4 (Fig. 1) in the active tunnel with Ala. Previous work has suggested that this tryptophan residue interacts with the substrate (8, 26), and its replacement with a non-aromatic residue was expected to weaken the complex. Here, we report a comparative analysis of steady-state kinetics, adsorption, and processivity for wild type *Tr*Cel7A and the Trp-38 \rightarrow Ala variant, henceforth termed WT and W38A, respectively. To assess possible effects of the CBM, we have made this comparison both for the intact enzyme and the catalytic domain alone.

EXPERIMENTAL PROCEDURES

All experiments were conducted in a standard 50 mm acetate buffer, 2 mm CaCl $_2$, pH 5.0.

Mutagenesis, Expression, and Purification—All enzymes were expressed heterologously in *Aspergillus oryzae* and purified as described elsewhere (27). Core variants (*i.e.* the catalytic domain alone) were made by introducing a BamHI restrictionsite in the *Tr*Cel7A gene at the end of the core domain, making it possible to splice out the core domain from the *Tr*Cel7A plasmid or W38A plasmid. All purified enzyme stocks showed a single band in SDS NuPAGE 4–12% BisTris gel (GE Healthcare), and the absence of specific contamination by β -glucosidase was confirmed by the lack of activity against *p*-nitrophenyl β -Dglucopyranoside. Henceforth, we will use subscript "intact" for enzymes with and "core" for enzymes without the CBM. Enzyme concentrations were determined by absorbance at 280 nm using molar extinction coefficients of 84,810 M⁻¹ cm⁻¹, 79,120 M⁻¹ cm⁻¹, 79,210 M⁻¹ cm⁻¹, and 73,520 M⁻¹ cm⁻¹ for WT_{intact}, W38A_{intact}, WT_{core}, and W38A_{core}, respectively.

Kinetic Measurements—The cellobiose production was monitored by an amperometric enzyme biosensor based on cellobiose dehydrogenase from *Phanerochaete chrysosporium* adsorbed onto the surface of a benzoquinone-modified carbon paste electrode as described in detail previously (28). Measurements were carried out at 25 °C in a 5-ml water-jacketed glass cell connected to a water bath. Avicel PH 101 (Sigma-Aldrich) suspensions were stirred at 600 rpm, and enzyme was injected to a final concentration of 100 nM from a Chemyx Fusion 100 syringe pump with an injection time of 1.0 s. All measurements were made in duplicate with calibration before and after each run, using the average sensitivity as the calibration factor (28).

Adsorption—The adsorption of enzyme on Avicel was measured in 96-well microtiter plates by adding substrate and enzyme to a final volume of 300 μ l in each well. The substrate load was 10 g/liter for enzymes with CBM and 30 g/liter for the core variants, and the enzyme concentration was varied from 0 to 5 μ M. The plate was sealed and incubated for 1 h at 25 °C and 1200 rpm orbital mixing. After incubation, the plate was centrifuged at 3500 rpm for 5 min, and 100 μ l of supernatant was transferred to black 96-well microtiter plates. Intrinsic protein fluorescence was determined by measuring the emission at 340 nm at an excitation wavelength of 280 nm. In all experiments, an enzyme standard curve was made using the enzyme in question as reference.

Product Profile—The production of glucose, cellobiose, and cellotriose was measured at different time points by taking 500- μ l subsets of a 50-ml reaction mixture, containing 10 g/liter Avicel and 0.50 μ M enzyme at 25.0 °C. Subsets were quenched by mixing with the same volume 0.1 M NaOH, and following centrifugation, the concentrations of glucose, cellobiose, and cellotriose in the supernatants were measured on a Dionex ICS-5000 ion chromatograph (Thermo Fisher Scientific, Waltham, MA) as described elsewhere (27, 29).

RESULTS

Kinetics—Fig. 2 shows real-time data obtained by the *P. chrysosporium* cellobiose dehydrogenase biosensor. The *two top panels* illustrate the cellobiose production for the enzymes



FIGURE 2. **Real-time progress curves.** Shown is the initial hydrolysis (0–300 s) of the four investigated *Tr*CeI7A variants, WT_{intact}, W38A_{intact}, WT_{core}, and W38A_{core}. The enzyme concentration was 100 nm, and the Avicel load ranged from 0.5 to 60 g/liter.

with CBM, WT_{intact}, and W38A_{intact}. Analogously, the *bottom* panels present data for the two core variants, WT_{core} and W38A_{core}. The enzyme concentration was 100 nM in all trials, whereas the load of Avicel was varied over 2 orders of magnitude (0.5–60 g/liter). The results show that enzymes with the W38A mutation produce more cellobiose than the corresponding wild type except at very low Avicel loads.

We have previously suggested (23, 30) that experimental data for processive enzymes may be analyzed by a reaction scheme (Scheme 1), according to which the free enzyme, *E*, combines with a cellulose strand, C_m , to form an activated complex, EC_m . The enzyme subsequently moves along the strand, which is sequentially shortened by one cellobiose, moiety *C*, for each step (so the strand is converted to C_{m-1} , C_{m-2} , etc.). In this process, the activated complexes (EC_{m-1} , EC_{m-2} ...) are all allowed two reaction pathways. They may either go through another catalytic cycle to produce a cellobiose and a correspondingly shortened cellulose strand, or they may dissociate. This reaction mechanism can be kinetically described by four parameters. These are the rate constants for association (k_{con}), catalysis (k_{cat}), and dissociation (k_{off}) and the processivity num-



SCHEME 1. **Simplified reaction scheme and its structural interpretation.** The scheme defines the three rate constants (k_{onv} , k_{catv} and k_{off}) used and the processivity number (*n*).

ber, *n* (the average number of sequential steps made following the attack of one cellulose strand).

At quasi-steady state, the rate of cellobiose production according to the processive reaction in Scheme 1, $_{p}V_{SS}$, attains the usual hyperbolic form (23),

$${}_{p}V_{ss} = \frac{{}_{p}V_{max}S}{{}_{p}K_{m}+S}$$
(Eq. 1)

where *S* is the load of substrate in g/liter, and the processive kinetic parameters, ${}_{p}K_{m}$ and ${}_{p}V_{max}$, are defined by the processivity, *n*, and the rate constants from Scheme 1.





FIGURE 3. **Steady-state kinetics.** The steady-state hydrolytic rate, $_pV_{ss}$, for the four investigated enzymes is plotted as a function of the substrate load. *Symbols* represent experimental values derived from the slopes of linear fits between 100 and 150 s in Fig. 2. The *lines* are best fits of Equation 1. The kinetic parameters $_pK_m$ and $_pV_{max}$ (Equation 1) are given in Table 1.

TABLE 1

Steady-state kinetic and adsorption parameters for the four investigated TrCeI7A variants, WT_{intact}, W38A_{intact}, WT_{core}, and W38A_{core}

	Kinetic parameters				Adsorption parameters		
Enzyme	$_{p}V_{\max}$	$_{p}K_{m}$	$_{p}V_{\max}/_{p}K_{m}$	Γ_{max}	K_d	$\Gamma_{\rm max}/K_d$	
	nm/s	g/liter	nmol/s/g	nmol/g	μ_M	liter/g	
WT _{intact}	19.5 ±0.6	1.54 ± 0.23	12.7 ± 1.93	195 ± 7	0.145 ± 0.019	1.345 ± 0.183	
W38A _{intact}	38.4 ± 0.7	5.95 ± 0.34	6.5 ± 0.39	114 ± 4	0.544 ± 0.061	0.210 ± 0.025	
WT _{core}	34.4 ± 1.8	12.6 ± 1.82	2.7 ± 0.42	30 ± 2	1.43 ± 0.195	0.021 ± 0.003	
W38A _{core}	53.2 ± 2.9	24.3 ± 2.87	2.2 ± 0.28	22 ± 3	2.97 ± 0.620	0.008 ± 0.002	

$$_{D}K_{m} = \frac{k_{\text{off}}}{k_{\text{on}}}$$
 (Eq. 2)

and

$${}_{p}V_{\max} = \beta E_0 k_{\text{cat}} \tag{Eq. 3}$$

Equations 1–3 rely on the assumption that the substrate is in excess (*i.e.* that the concentration of attack sites for enzyme on the substrate surface is much larger than the enzyme concentration). This has previously been shown to be a reasonable assumption for conditions similar to those used here (22, 23). We note that the main difference between the parameters for processive enzyme activity given in Equations 2 and 3 and the analogous parameters in the conventional Michaelis-Menten equation is the constant, β . This so-called kinetic processivity coefficient is defined, $\beta = 1 - (k_{cat}/(k_{cat} + k_{off}))^n$ and may be interpreted as the probability that the enzyme will dissociate from the cellulose strand before completing the average number (*n*) of processive steps (23).

The experimental data in Fig. 2 are typical for cellulases in the sense that a truly constant reaction rate (*i.e.* a fully linear progress curve) is never observed. As argued previously (23), this requires some compromise in the definition of the steady-state rate, and it was recommended to assign $_{\nu}V_{SS}$ as the slope of the

concentration trace after the initial burst, which represents a non-steady-state condition (23, 31) but before the rate has been significantly reduced by other (as of yet partially unknown) factors. In accordance with this, we fitted a line to the data from 100 to 150 s in Fig. 2 and used the slope as a measure of $_{n}V_{ss}$. Results are plotted as a function of the Avicel load in Fig. 3. Comparisons of data for the two enzymes with CBM (WT_{intact} and W38A_{intact}) and the two catalytic domains (WT_{core} and $W38A_{core}$), respectively, again show that the W38A mutation slightly decreases activity at very low substrate load but significantly promotes pV_{SS} when the Avicel load approaches saturating levels. Fig. 3 also shows that Equation 1 accounts well for the experimental data for all four enzymes, and the kinetic parameters derived from the regression analysis based on Equation 1 are listed in Table 1. These parameters reveal that the two W38A variants have 1.6-2 times higher $_{p}V_{max}$ compared with the corresponding enzymes without this mutation. Conversely, the affinity for the substrate is lowered by the W38A mutation $({}_{p}K_{m}$ is increased 4 times for the intact enzyme and 2 times for the core variant).

Adsorption—The effect of the W38A mutation on substrate affinity was further investigated by adsorption measurements. Results in Fig. 4 show the characteristic adsorption saturation and a strong effect of the CBM, which have been discussed in



FIGURE 4. Adsorption isotherms for the four investigated enzymes. The coverage of enzyme on Avicel, Γ , is plotted as a function of the free enzyme concentration, *E. Points* represent the experimental data, and the *lines* are best fits of the Langmuir isotherm.

detail earlier both for TrCel7A and numerous related cellulases (32-34). In the current context, it is more important to note that the W38A mutation significantly reduces binding both in the intact enzyme and in the core variant. To quantify this, we analyzed the data with respect to a standard Langmuir isotherm, Γ , where the variables *E* and Γ represent free concentration and adsorbed amount of enzyme, respectively, and Γ_{\max} and K_d are the usual Langmuir parameters (saturation coverage and dissociation constant). The Langmuir isotherm was found to account well for the experimental data, and the parameters are listed in Table 1. It appears that the mutation weakens the interaction significantly because K_d increased 3.7 times in the intact enzyme and 2.1 times in the core variant. The saturation coverage, Γ_{max} , was also reduced (30-40%) by the mutation, indicating a reduced substrate accessibility for the mutant.

Product Profile-Progress curves for glucose, cellobiose, and cellotriose produced by WT_{intact} and W38A_{intact} are shown in Fig. 5. These measurements were made to assess the influence of the W38A mutation on the processivity. It has previously been suggested that processivity can be estimated from the cellobiose/cellotriose ratio of either concentrations or steadystate production rates (31, 35, 36). Applying these principles, we found processivities of 23-24 for WT_{intact} and 15-16 for W38A_{intact}. Product profiles were only measured for one enzyme concentration, and the following discussion of processivity hence neglects a possible dependence of *n* on the enzyme/ substrate ratio. We note, however, that the enzyme/substrate ratio in the product profile measurements (Fig. 5) was comparable with the ratio in the kinetic measurements (Fig. 2) for substrate loads around ${}_{\nu}K_m$ and hence that the two types of experiments appear compatible. The processivity of the wild type TrCel7A on Avicel has been measured before (15, 22, 31),

and the current result is in line with these earlier reports. We note that relationships between product profiles and processivity have been extensively discussed (4, 35, 37), and somewhat deviant principles for the calculation of *n* have been put forward (22, 31, 36, 38). A discussion of the validity and weaknesses of these strategies is beyond the current scope, but using different approaches, we consistently found that the processivity of W38A was about 25% lower than the wild type. Hence, we conclude that the product profile suggests a moderately decreased processivity in the W38A variant compared with the wild type enzyme.

DISCUSSION

Tryptophan residues in the active site of glycoside hydrolases are important for processive hydrolysis (8, 39-42). Many mutational studies in which Trp has been replaced with nonaromatic residues have shown negative effects on activity and processivity of, for example, cellulases (16, 36, 38, 43-46) and chitinases (47–49). One suggested role of Trp is that "stacking" interactions of π -electrons in its aromatic ring and the α -face of the carbohydrate (50) enable the necessary sliding of the bound chain (4, 51), possibly through stabilization of the transition state between adjacent positions in the processive movement (52, 53). Tryptophan residues near the entrance of the active tunnel are also important for the initial "threading" and recognition of the cellulose strand, particularly on crystalline substrates (16, 26, 38, 44, 54). In the current work, we used the W38A mutation in T. reesei Cel7A to obtain an enzyme with weakened substrate affinity. The purpose was to study the kinetics of a low affinity variant over a broad range of substrate loads and use this information to elucidate whether the ratelimiting step was before or after the activated complex, EC_{m-1} in Scheme 1. Detailed molecular analysis of Trp-carbohydrate





FIGURE 5. Product profile. Shown are concentrations of glucose, cellobiose, and cellotriose produced by 0.5 µM WT_{intact} and W38A_{intact}, respectively, during hydrolysis of Avicel at 25 °C and an initial load of 10 g/liter.

interactions, on the other hand, was beyond the current scope. In the simplified view of Scheme 1, the question we are addressing is whether k_{on} or k_{off} limits the overall reaction. As outlined in the Introduction, this question has recently attracted wide-spread interest, and its elucidation appears to be central for the fundamental understanding of processive cellulases. We chose Trp-38 as target for a low affinity mutant because this residue has previously been shown to interact with the cellulose chain (8, 26). Moreover, it is located in the middle of the tunnel quite far from the scissile bond unlike, for example, Trp-367 and Trp-376, which are directly involved in the twist of the cellulose strand that is important for the actual catalysis (55).

Substrate Affinity-Results in Table 1 confirmed that the affinity for Avicel was reduced by the W38A mutation. This was seen both for the enzymes with CBM (WT_{intact} versus W38A_{intact}) and when comparing the two core variants, WT_{core} and W38A_{core}. Moreover, lower affinity was reflected in both kinetic and adsorption measurements (both $_{p}K_{m}$ and K_{d} were increased by the W38A mutation). The absolute values of ${}_{\nu}K_{m}$ and K_d cannot be directly compared because the insoluble nature of the substrate prevents conversion of ${}_{p}K_{m}$ to molar units. However, the contribution of Trp-38 to the affinity may be assessed from the ratio of parameters, K_i (W38A)/ K_i (WT), where K_i represents either ${}_{p}K_m$ or K_d . For the intact enzyme, this ratio was 3.8 for K_d and 3.9 for ${}_{\nu}K_m$ (Table 1). Thus, affinity changes derived independently by either kinetic or adsorption measurements were in perfect accord, and we may estimate the contribution to the standard free energy of binding brought about by Trp-38 as follows.

$$\Delta \Delta G_{W38} = -RT \ln \left(K_i (W 38 \text{ A}) / K_i (WT) \right)$$
(Eq. 4)

Insertion of the above values for the intact enzyme yields $\Delta\Delta G_{W38}^0 = -3.3$ kJ/mol. For the core variants (WT_{core} and W38A_{core}), the ratios were 2.1 (K_d) and 1.9 ($_pK_m$), and this translates into $\Delta\Delta G_{W38}^0 = 1.7$ kJ/mol. Provided that alanine in position 38 interacts negligibly with the cellulose strand (50) and that the overall conformation of the mutant is mainly unchanged, this suggests that Trp-38 contributes -2 to -3kJ/mol to the standard free energy of substrate binding. This value perfectly matches general assessments of the interaction free energy for carbohydrates and aromatic side chains (50, 56), but it is lower than the contribution from Trp-38 to ligand binding (-15.9 kJ/mol) derived from molecular dynamics simulations (57). We note that a similar analysis of the contribution of the CBM (again using the appropriate ratios of parameters from Table 1) suggested $\Delta\Delta G^0_{
m CBM}$ values of -5.4 kJ/mol for the wild type (WT $_{\rm intact}$ versus WT $_{\rm core}$), and -3.9 kJ/mol for the W38A mutant (W38A_{intact} versus W38A_{core}). This result for the wild type is in accord with values calculated from the adsorption isotherms reported by Stahlberg et al. (32). Comparing this and the results for the Trp mutant suggests that the CBM contributes about twice as much to the net affinity compared with Trp-38. Because the CBM has three subsites, a 1:3 ratio might have been expected, but the somewhat weaker subsite contribution in the CBM may reflect differences between interactions of a planar interface (CBM) and a tunnel (core). Moreover, the CBM has Tyr in all three subsites, and this residue has previously been shown to bind less tightly than Trp (58).

We note that contributions from either Trp-38 or the CBM were not simply additive. Thus, the W38A mutation was more

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detrimental to binding of the intact enzyme compared with the core, and the favorable effect of the CBM was larger in the wild type than in the W38A mutant. Origins of this non-additivity cannot be assessed on the basis of the current data, but we note that a related behavior has previously been observed by Palonen *et al.* (59). These workers found that the affinity of Cel7A core was much lower than the intact enzyme, thus suggesting a major role of the CBM for adsorption. However, when purified CBM was tested in separate experiments, it did not show high affinity, and this clearly suggests a synergy (non-additivity) in the adsorption process.

Kinetics-The remainder of this discussion will address the respective influence of association and dissociation on the overall rate. As outlined in the Introduction, this focus was motivated partly by recent discussions of rate limitation, which have highlighted these two steps, and partly by the deduction that the hydrolytic step (governed by k_{cat} in Scheme 1) is an unlikely candidate for rate determination. The latter relies on numerous measurements of k_{cat} for TrCel7A, which have shown that at room temperature this parameter is in the range of 2-10 s⁻¹ (19, 21, 22, 24, 43, 55, 60, 61), and this is much faster than typically measured specific rates. Because the overall rate is obviously dictated by the slowest step, this discrepancy strongly suggests that catalysis is not the bottleneck and that rate determination must be some much slower process. To illustrate this for the current system, we note that the maximal specific rate at steady-state, $_{p}V_{max}/E_{0}$, for wild type TrCel7A on Avicel was about 0.2 s⁻¹ (Table 1; E_0 was 100 nM). This is over an order of magnitude less than k_{cat} for the same system (4.8 s⁻¹) (24), and we will therefore not consider this parameter further in the discussion of rate limitation. We note, however, that the above arguments pertain to hydrolysis by monocomponent TrCel7A, and different conclusions regarding k_{cat} may be reached for mixtures of synergistically acting cellulases (21, 55). Hence, Jalak et al. (21) found that hydrolysis of bacterial cellulose by an optimally composed mixture of Cel7A and the endo-active enzyme Cel5A could reach a specific hydrolysis rate at steady state (about 2 s^{-1}), which was close to the rate constant for the inner catalytic cycle (including hydrolysis, product expulsion, and processive movement). This implies that, under these conditions, k_{cat} was indeed important for rate limitation. More recently, Knott et al. (55) have used molecular dynamics simulation and free energy calculation to suggest that the rate-limiting step in the hydrolytic chain of reactions is glycosylation of the nucleophile (Glu-212), which had a rate constant of about 11 s^{-1} . This theoretical result is in remarkable accordance with $k_{\rm cat}$ values derived from either high speed atomic force microscopy (60) or pre-steady-state kinetic measurements (24). Knott et al. (55) also used computational methods to analyze the kinetics of other steps, including processive motion, deglycosylation, catalytic activation, and "threading" the cellulose strand into the active tunnel, but these latter steps were found to be faster than glycosylation, which was therefore suggested to be potentially rate-limiting. The off-rate was not assessed by Knott et al. A combined interpretation of these results suggests that glycosylation is likely to be rate-limiting for the conditions studied by Jalak et al. However, specific rates reported for both monocomponent cellobiohydrolases (15, 16, 19, 30, 62) and

enzyme mixtures (63, 64) are well below 1 s^{-1} and hence suggest a rate-limiting step outside the catalytic cycle.

The results we present here show that the W38A mutation in TrCel7A decreases substrate affinity and processivity but increases the specific hydrolytic rate for both enzymes with and without CBM. Although this decoupling appears puzzling at first, it may form the basis for a qualitative interpretation of the results. This is illustrated, at least to a first approximation, in the simplified energy diagram in Fig. 6. Thus, a variant with lowered substrate affinity (green line) compared with the wild type (blue line) will obviously show less binding (the EC_m complex for the variant has higher free energy). Moreover, the variant also has a lower activation barrier, $\Delta\Delta G_D^{\neq}$, for dissociation (*c.f.* Fig. 6), and if $\Delta\Delta G_D^{\neq} > \Delta\Delta G_A^{\neq}$ so that dissociation is ratelimiting, the low affinity variant will show a faster overall reaction. It follows that decoupling of affinity and activity, as seen in Table 1, may be rationalized along the lines of Fig. 6, provided that dissociation is rate-limiting. One corollary of a dissociation-controlled rate is that not all adsorbed enzyme can be catalytically active at the same time (20, 23), and this is congruent with suggestions of a bound but inactive population of processive cellulases. This may rely on irregularities in the cellulose structure, which obstruct processive movement (20, 30, 65) or "traffic jams" in which enzymes collide and stall (60). In any case, slow release will be associated with an adsorbed but catalytically inactive population of enzyme. It has been argued that the kinetic processivity factor, β (Equation 3), provides a measure of the catalytically active population and that it may be as low as 3-8% for TrCel7A (23). If so, a dissociation-controlled mechanism can be understood as a situation where dissociation of predominantly inactive EC_{m-1} complexes (and hence recruitment for attacks on new cellulose strands) is rate-limiting.

None of the rate constants in Scheme 1 can be calculated directly from the parameters in Table 1, but some of them may be assessed on the basis of the relationship,

$$_{p}V_{\max} \approx nk_{\text{off}}E_{0}$$
 (Eq. 5)

provided that $k_{\text{cat}} \gg k_{\text{off}}$ (23). This latter assumption has previously proven valid for TrCel7A hydrolyzing both Avicel (24) and other substrates (19, 22). If we insert the processivities for WT_{intact} (*n* = 23) and W38A_{intact} (*n* = 16) found in Fig. 5 and $_{p}V_{max}$ values from Table 1 into Equation 5, we find off-rate constants of 0.0085 s⁻¹ for WT_{intact} and 0.024 s⁻¹ for the $W38A_{intact}$ mutant. Expressed as half-times for the dissociation of enzyme-substrate complexes ($t_{\frac{1}{2}} = \frac{\ln 2}{k_{\text{off}}}$), this translates into 82 and 29 s, respectively. When off-rate constants and $_{n}K_{m}$ values are known, the (second order) on-rate constants can be readily calculated from Equation 2. We found k_{on} values of $0.0055 \text{ (g/liter)}^{-1} \text{ s}^{-1} \text{ and } 0.0040 \text{ (g/liter)}^{-1} \text{ s}^{-1} \text{ for WT}_{\text{intact}}$ and W38A_{intact}, respectively. In conclusion, this analysis suggests that the lowered net affinity of the W38A mutant reflected a moderate (\sim 27%) reduction in $k_{\rm on}$ and significant (\sim 2.8-fold) increase in k_{off} . Because k_{on} and k_{off} are of different kinetic orders, their values cannot be directly compared. However, an apparent on-rate constant at a fixed substrate load, $k_{on}S$ (in units of s⁻¹) can be compared with k_{off} to assess the importance of association and dissociation, respectively. If, for example, we





FIGURE 6. **Simplified energy diagram for TrCeI7A catalysis with emphasis on association and dissociation.** The *blue curve* represents a possible course for the wild type, and the *green curve* illustrates differences for a variant with lower substrate affinity (such as W38A). The reaction has been condensed into the same steps as in Scheme 1, and this is illustrated by the *drawings* at the *top*. The plot emphasizes how lower substrate affinity will tend to decrease the energy barrier for dissociation (ΔG_{Φ}^{\pm}) rather than the barrier for association (ΔG_{A}^{\pm}).

consider a very dilute substrate, such as S = 1 g/liter, together with the rate constants given above, we find that $k_{\text{off}} > k_{\text{on}}S$. This implies an association controlled reaction. However, the apparent on-rate constant grows proportionally with S, whereas k_{off} is independent of the substrate load, and this means that at, for example, 10 g/liter, the situation is reversed, and $k_{on}S > k_{off}$. To put this in more general terms, we note that the processive Michaelis constant was defined (Equation 5) as $_{p}K_{m} = k_{\text{off}}/k_{\text{on}}$. It follows that the apparent on-rate constant, $k_{on}S$, is equal to k_{off} when $S = {}_{p}K_{m}$ and that $k_{on}S > k_{off}$ (*i.e.* off-rate controlled hydrolysis) when $S > {}_{p}K_{m}$. This analysis allows a more detailed interpretation of the rate limitation. Thus, off-rate limitation, as suggested above, only occurs for $S > K_m$, because at lower loads, the scarcity of attack sites inevitably delays association to the extent where it becomes rate-limiting. This distinction is obviously important in discussions of rate limitation, and conclusions based on experiments at a single load will only provide an incomplete picture of the limiting step. As seen from Table 1, the changeover from on- to off-rate limitation occurs at low loads for the wild type enzyme $({}_{n}K_{m} = 1.5 \text{ g/liter})$, but it increases significantly for the low affinity variants and reaches 24 g/liter for W38A_{core}. This shows that the potential of fast hydrolysis of this latter enzyme only materializes at quite high substrate loads. This may be interesting from an applied perspective because plans for industrial conversion of biomass usually include very high solid loads of 100-400 g/liter (66, 67). Although typically only around half of this dry mass is cellulose, it appears most likely that these conditions are in the regime, where the initial rate is governed by k_{off} .

Some studies on processive cellulases have found that replacement of Trp residues near the entrance of the catalytic tunnel leads to enhanced activity against soluble and/or amorphous substrates, whereas the hydrolysis of highly crystalline substrate is reduced in comparison with the wild type (16, 38, 44). In these cases, the activity was only tested at one (low) substrate load (0.7–2.5 g/liter). Because the activity curves in Fig. 3 showed a crossover at intermediate substrate loads, it appears that a broader range of loads may be necessary to reach a conclusion on the effect of the mutation on activity. We emphasize, however, that the W38A mutants studied here for their activity against Avicel (with an intermediate crystallinity) could behave differently on other substrates and hence that further experimental work would be required to elucidate possible kinetic differences between Trp mutations near the opening (16, 38, 44) and (as Trp-38) in the middle of the catalytic tunnel, respectively.

Conclusion-The W38A mutation in Cel7A moderately reduced the strength of binding to cellulose. Specifically, the standard free energy of association was about 3 kJ/mol less negative compared with the wild type enzyme. Kinetic analysis suggested that the reduced net affinity can be broken into a small reduction of k_{on} and a larger increase of k_{off} . At substrate loads approaching saturating levels, these changes led to a 2-fold reduction in adsorption but a 2-fold increase in activity. Conversely, at very low substrate loads, both activity and adsorption were lessened by the mutation. These observations strongly suggest that except in very dilute Avicel suspensions, where infrequent enzyme-substrate collisions predictably become limiting, the rate-determining step must be in the dissociation process. The results cannot be reconciled with an associationlimited interpretation because the on-rate constant for the (more active) mutant was lower than the on-rate constant for the (less active) wild type. Scheme 1 represents a simplification, where several potential steps (see the Introduction) have been pooled into one association and one dissociation reaction. It

follows that k_{on} and k_{off} are composite rate constants reflecting the kinetics of passing through a sequence of steps. The dissociation process, which we have found to be the bottleneck, may, for example, be partitioned into the separate release of core and CBM domains. We cannot unambiguously separate these steps on the basis of the current data, but we note that the W38A mutation brought about similar kinetic changes in both the intact enzyme and the separate core domain. In light of this, we speculate that the release of substrate from the active tunnel is rate-limiting under the conditions studied here. As discussed in the Introduction, other studies have reached quite different conclusions regarding the rate-limiting step for cellobiohydrolases, and this raises the question of whether rate determination may be different for different conditions and particularly whether it may shift as the substrate is gradually modified during hydrolysis. This cannot be addressed from the current measurements of initial rates, but it would be relevant for future work.

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