

Antioxidant Metabolism during Acclimation of *Begonia* × *erythrophylla* to High Light Levels

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This study examined the influence of high light levels on antioxidant metabolism and the photosynthetic properties of *Begonia* × *erythrophylla* leaves. The pigment composition of shaded leaves and those developing in full sunlight was typical of shade- and sun-leaves, respectively. After 28 d in full sunlight, the preformed leaves of shade plants transferred to full sunlight (transferred-leaves) showed photo-bleaching with lower Chl (*a* + *b*) content and Chl *a* : Chl *b* ratios than shade-leaves, with Chl (*a* + *b*) : carotenoid ratios not significantly different. The variable/maximal fluorescence (F_v/F_m) of sun-leaves was not significantly different from that of shade-leaves, but transferred-leaves had reduced $F_v : F_m$ ratios. Light response curves for the electron transport rate (ETR), the oxidation state of photosystem II (qP) and non-photochemical quenching (NPQ) showed significant differences between the three leaf types, with transferred-leaves not able to acclimate completely to full sunlight, having lower ETR, qP and NPQ values at high light levels than sun-leaves. Transfer to full sunlight caused a rapid increase in H₂O₂ and lipid hydroperoxides, and a slight increase in protein oxidation. Ascorbate and glutathione levels decreased rapidly, as did the size of the total glutathione pool and, in addition to the general oxidation of proteins, rapid decreases in both the initial and total activities of chloroplastic fructose-1,6-bisphosphatase and glyceraldehyde-3-phosphate dehydrogenase were observed. These results suggest that a more oxidizing cellular environment is the likely cause of the photo-bleaching observed upon transfer of shade-leaves to full sunlight. Acclimation of transferred-leaves to full sunlight involved gradual increases in the activities of enzymes involved in antioxidant metabolism, including superoxide dismutase, catalase, glutathione reductase, ascorbate peroxidase, dehydroascorbate reductase and monodehydroascorbate reductase, but the levels of these enzymes still remained at levels lower than those found in sun-leaves. © 2003 Annals of Botany Company

Key words: Active oxygen species, antioxidants, *Begonia* × *erythrophylla*, light, oxidative stress, photosynthesis.

INTRODUCTION

Antioxidant metabolism plays an important role in protecting plants from a wide variety of environmental stresses, such as drought, extreme temperatures, pollutants, ultraviolet radiation and high levels of light (Foyer *et al.*, 1994; Smirnoff, 1995). Although light is required for plant growth and development, when exposed to photosynthetically active radiation (PAR) at photosynthetic photon flux densities (PPFD) greater than those required for CO₂ assimilation, electron carriers can be over-reduced and plants can suffer photo-inhibition (Demmig-Adams and Adams, 1992; Foyer *et al.*, 1994).

Plants grown under higher PPFD usually have a decreased capacity to absorb incident radiation and an increased ability to dissipate excess excitation energy. Chlorophyll fluorescence has proved useful for measuring the photochemical and non-radiative dissipation activities in leaves under different PPFD. In two recent studies, Greer and Jeffares (1998) and Laing *et al.* (2000) demonstrated that the quantum yield of photosystem II (PSII) and the oxidation state of PSII (qP) decline as PPFD increases, while electron transport rate (ETR) and non-photochemical quenching (NPQ) increase.

Under conditions of excess photon energy the NPQ capacity of the photosynthetic apparatus can be exceeded and the photosynthetic electron transport system becomes a source of active oxygen species (AOS) and can generate singlet oxygen, ¹O₂, and superoxide, O₂⁻ (Asada, 1994). The half-life of O₂⁻ in plant cells is relatively short. It can disproportionate spontaneously to form hydrogen peroxide (H₂O₂) and molecular oxygen (O₂), or this reaction can be catalysed by the enzyme superoxide dismutase (SOD) (Noctor and Foyer, 1998). The H₂O₂ produced by these reactions is not as reactive as O₂⁻ and, by comparison, is a relatively long-lived molecule. However, in cells under oxidative stress, O₂⁻ can reduce transition metals which can in turn reduce H₂O₂, generating highly reactive hydroxyl radicals (OH). These can trigger the autocatalytic process of lipid peroxidation (Halliwell and Gutteridge, 1989).

The ability to withstand the oxidative stress imposed by AOS depends on the antioxidative capacity of the cell. Plant cells contain both enzymic and non-enzymic antioxidants. SOD catalyses the disproportionation of H₂O₂ to O₂, and the resultant H₂O₂ is decomposed by catalase (CAT) or ascorbate peroxidase (APOX), the latter enzyme requiring reduced ascorbate, provide by the ascorbate–glutathione pathway, for the reaction to occur. In the ascorbate–glutathione pathway, reduced glutathione (GSH) is needed for the reduction of dehydroascorbate (DHA), which is

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formed via monodehydroascorbate (MDHA) produced by the action of APOX or by non-enzymatic reactions of ascorbate (ASA) with oxidants. In addition to playing roles in enzymic antioxidant metabolism, both ASA and GSH can directly scavenge AOS (Asada, 1994). GSH is especially important as it helps protect thiol-containing enzymes such as the Calvin-cycle enzymes, NADP⁺-dependent glyceraldehyde-3-phosphate dehydrogenase (G3PDH) and chloroplast fructose-1,6-bisphosphatase (FBPase). The maintenance of a reduced pool of glutathione is mediated by the activity of glutathione reductase (GR) at the expense of NADPH oxidation (Hausladen and Alscher, 1993).

There are many reports in the literature detailing changes in the activities of enzymes involved in antioxidant metabolism in response to high-light stress (Gillham and Dodge, 1987; Foyer *et al.*, 1989; Mishra *et al.*, 1995; Grace and Logan, 1996; Logan *et al.*, 1998). From these and other studies it has become clear that plants grown under low PPFD are less able to protect themselves from AOS, due to a lower capacity to dissipate excess light energy and a lower capacity to scavenge AOS, compared with plants grown in full sun.

Begonias are popular ornamental plants, with many species grown indoors as potted plants. Numerous studies have shown light to be of great importance for the growth and development of begonia, and hence their commercial production (Nowak and Feild, 1993; Rudnicki *et al.*, 1993; Myster, 1999). Many begonias are classified as shade-demanding understory species, and have leaves that are highly adapted to deep shade (Lee *et al.*, 1990). As a result they are often propagated under very low light levels (Myster, 1999) and can suffer damage if transferred to full sunlight. Despite the importance of light for the development of begonias, very little is known about the physiology of how these commercially valuable plants respond to light.

Begonia × erythrophylla J. Neuman, also known as *B. × Feistii* Hort. ex L.H. Bailey, is a hybrid begonia (*B. manicata × B. hydrocotylifolia*) probably of garden origin and is a popular ornamental plant, mostly grown indoors as a pot plant. *B. × erythrophylla* plants have large well-developed leaves to maximize light interception and, when placed in full sunlight, fully expanded leaves exhibit rapid photo-bleaching.

There have been no reports in which the influence of light levels on the production of AOS and antioxidant metabolism in begonia plants has been considered. The objectives of this study were to determine the role that antioxidant metabolism plays in the acclimation of *B. × erythrophylla* plants following transfer of plants, grown in the shade, to full sunlight, and to test the hypotheses that pre-existing foliage suffers oxidative damage, does not acclimate completely to full sunlight and that only new leaves are fully acclimated to full sunlight. Specifically, we monitored chlorophyll fluorescence parameters, the production of AOS, oxidative damage to cellular macromolecules and the development of the antioxidant systems following transfer of *B. × erythrophylla* plants from shade to full sunlight.

MATERIALS AND METHODS

Plant material and sampling protocols

Begonia × erythrophylla stock plants were grown in a glasshouse under shade-cloth, which reduced PAR to approx. 10–15 % of ambient, with a mean midday PAR of 155 $\mu\text{mol m}^{-2} \text{s}^{-1}$ during the experimental period. The photoperiod in the glasshouse was extended to 16 h by artificial illumination (Phillips SonT Agro 400 lamps) and the night temperature was maintained at a minimum of 15 °C by electric heating.

The leaves of plants that were maintained in the shade for the duration of the experiment are referred to as shade-leaves, while those that developed fully in the shade and were then transferred to full sunlight are referred to as transferred-leaves. The leaves of the transferred plants that expanded and developed in full sunlight are referred to as sun-leaves. Sun-leaves were only sampled from plants grown in full sunlight for 21 and 28 d. Leaf samples were harvested and measurements of variable/maximal fluorescence (F_v/F_m) were made between 1100 and 1300 h to minimize diurnal fluctuations. Irradiance curves were generated 28 d after the transfer of plants to full sunlight.

Pigment quantification

Chlorophylls and carotenoids were extracted in dimethylsulfoxide (DMSO). Discs (1 cm²) were weighed and placed in 5 ml DMSO for 48 h, at room temperature. Chlorophyll and carotenoid pigments were measured using a Jasco V-550 dual-beam spectrophotometer and determined, on a fresh weight basis, following the formulae of Chapelle *et al.* (1992).

Chlorophyll fluorescence measurements

Chlorophyll fluorescence was measured using a Pulse Amplitude Modulation (PAM-2000) portable chlorophyll fluorometer (Walz, Effeltrich, Germany). Six plants from each treatment were brought into the laboratory and adapted to the dark, at ambient temperature for 30 min. To quantify the efficiency of photon capture by open PSII reaction centres, leaf surfaces were exposed to a saturating pulse of light for determination of the ratio of $F_v : F_m$ (Butler and Kitajima, 1975).

The light response curves of ETR, qP and NPQ were also determined. Minimal fluorescence F_o and F_m were determined for dark-adapted leaves. The leaves were then acclimated at 100 $\mu\text{mol m}^{-2} \text{s}^{-1}$ for 10 min before being exposed to gradual increases in PPFD provided by an external Halogen lamp (2050-H). At each light intensity the leaves were allowed to acclimate for 6.5 min and then given a saturating pulse of light. Following the saturating pulse, the actinic light was turned off and minimal fluorescence in the light (F_o') measured after a 5.5 s exposure to far-red light. The quantum yield of PSII was derived as per Gentry *et al.* (1989): $\Delta F/F_m' = (F_m' - F_t)/F_m'$, where F_m' is maximal fluorescence in the light, and F_t is fluorescence at time t . ETR was determined from yield \times PAR \times 0.5 \times 0.84, qP from $(F_m' - F_o)/(F_m' - F_o')$ and NPQ from $(F_m - F_m')/F_m'$.

Hydrogen peroxide determination

Frozen leaf tissue was ground in 50 mM ice-cold potassium phosphate buffer (3 ml g⁻¹ f. wt), containing 200 mM perchloric acid and the catalase inhibitor hydroxylamine (1 mM), and centrifuged at 10 000 g for 15 min at 4 °C. The perchloric acid was removed and hydrogen peroxide levels determined using the spectrophotometric method described by Cui *et al.* (1999).

Lipid extraction and peroxide analysis

Leaf tissue was boiled in isopropanol and lipids were extracted, as described by Navari-Izzo *et al.* (1991). Lipid hydroperoxides were determined using the method of Droillard *et al.* (1987). Briefly, 2 ml lipid extract was added to a solution of 5 ml ethanol, 0.2 ml 1 M HCl and 0.1 ml 35 mM ferrous ammonium sulfate. One millilitre of 2.6 M ammonium thiocyanate was added and the absorbance was read at 480 nm. A calibration curve with *t*-butyl hydroperoxide was used for quantification.

Oxidative damage to proteins

Frozen leaf tissue was ground in 50 mM ice-cold potassium phosphate buffer (pH 7.0) (3 ml g⁻¹ f. wt), containing 1 mM EDTA and 2.5 µg ml⁻¹ phenylmethylsulfonyl fluoride and leupeptin, and centrifuged at 10 000 g for 15 min at 4 °C. Contaminating nucleic acids were removed by treatment with streptomycin sulfate, and the oxidative damage to proteins was estimated as the protein carbonyl content, as determined by reaction with 2,4-dinitrophenylhydrazine (Reznick and Packer, 1994).

Glutathione extraction and determination

Tissue was homogenized in ice-cold 5 % (w/v) sulfosalicylic acid (5 ml g⁻¹ f. wt), centrifuged at 10 000 g for 15 min at 4 °C, and the supernatant was used for total and GSSG determinations by the 5,5'-dithio-bis-(2-nitrobenzoic acid)/GSSG reductase recycling procedure (Anderson *et al.*, 1992). GSSG was determined after removal of GSH by 2-vinylpyridine derivatizations, according to Anderson (1985). The contents of total glutathione and GSSG were calculated as per Sgherri *et al.* (1994), and GSH was determined by subtraction of GSSG (as GSH equivalents) from the total glutathione content.

Ascorbate extraction and determination

Tissue was homogenized in 5 % (w/v) metaphosphoric acid (10 ml g⁻¹ f. wt) and the homogenate centrifuged at 10 000 g for 15 min at 4 °C. Total and reduced ASA in the supernatant were determined using the method of Hodges *et al.* (1996). Briefly, total ascorbate was determined in a 1 ml reaction mixture containing 200 µl supernatant, 60 mM K-Na phosphate (pH 7.4), 3 mM EDTA and 1 mM dithiothreitol (DTT). After 10 min at 25 °C 100 µl of 40 mM *N*-ethylmaleimide was added. To determine reduced ascorbate the reaction mixture was as above, except 200 µl

H₂O replaced the DTT and *N*-ethylmaleimide. Colour was developed in both sets of reaction mixtures by the addition of 400 µl of 0.61 M trichloroacetic acid, 400 µl of 0.8 M orthophosphoric acid, 400 µl of 0.26 M α,α'-dipyridyl in 70 % ethanol and 200 µl of 0.19 M FeCl₃, and after 1 h at 40 °C, the absorbance was measured at 525 nm. Oxidized ascorbate (DHA) was calculated by subtracting reduced ascorbate from total levels.

Extraction and assay of antioxidant enzymes

Tissue was ground to a fine powder in liquid nitrogen and total proteins were extracted by homogenizing the powdered tissue in extraction buffer (3 ml g⁻¹ f. wt). The extraction buffer contained 100 mM potassium phosphate (pH 7.0), 40 mM KCl, 10 % (w/v) glycerol, 0.25 % (w/v) Triton X-100 and 2 % (w/v) Polyclar AT (Serva Chemicals Ltd, Heidelberg, Germany). Ascorbate (1 mM) was included in the extraction buffer when extracts were to be used for assaying APOX. Homogenates were centrifuged at 10 000 g for 20 min at 4 °C. The supernatants obtained were divided into aliquots and stored at -80 °C. Protein concentrations were determined according to the method of Pederson (1977), using bovine serum albumin as a standard.

APOX (EC 1.11.1.11) was assayed by following the decrease in A₂₉₀ as ascorbate disappeared (Rao *et al.*, 1996). The reaction mixture (1 ml) contained 100 mM potassium phosphate (pH 7.0), 0.5 mM ascorbate, 0.2 mM H₂O₂ and up to 50 µl extract. Activity was calculated using the extinction coefficient 2.8 mM⁻¹ cm⁻¹.

Dehydroascorbate reductase (DHAR) (EC 1.8.5.1) was assayed by monitoring the change in absorbance at 265 nm, as described by Miyake and Asada (1992), and monodehydroascorbate reductase (MDHAR) (EC 1.6.5.4) by measuring the decrease in absorbance at 340 nm, as described by Foyer *et al.* (1989). For DHAR the reaction mixture (1 ml) contained 50 mM HEPES/KOH (pH 7.0), 2.5 mM GSH 0.2 mM DHA, 0.1 mM EDTA and up to 50 µl of enzyme extract. DHAR activity was calculated using an extinction coefficient of 7.0 mM⁻¹ cm⁻¹. For MDHAR the reaction mixture (1 ml) contained 0.4 units ascorbate oxidase, 100 mM HEPES/KOH (pH 7.6), 2.5 mM ascorbate, 25 µM NADPH and up to 100 µl enzyme extract.

GR (EC 1.6.4.2) activity was determined following the procedure of Sgherri *et al.* (1994), by measuring the decrease in absorbance at 340 nm. The reaction mixture (1 ml) contained 0.2 M potassium phosphate (pH 7.5), 0.2 mM Na₂ EDTA, 1.5 mM MgCl₂, 0.25 mM GSSG, 25 µM NADPH and up to 50 µl of enzyme extract. The reaction was initiated by the addition of NADPH, and corrections for GSSG-independent NADPH oxidation were not necessary. Activity was calculated using the extinction coefficient 6.2 mM⁻¹ cm⁻¹.

Superoxide dismutase (EC 1.15.1.1) was assayed by the inhibition of xanthine oxidase-dependent reduction of nitroblue tetrazolium (McCord and Fridovich, 1969). The reaction mixture (1 ml) contained 50 mM potassium phosphate (pH 7.8), 0.5 mM nitroblue tetrazolium, 0.5 mM EDTA, 0.1 mM xanthine and 0.05 units xanthine oxidase. One unit of SOD is defined as the amount of enzyme that

TABLE 1. Chlorophyll and carotenoid contents of *B. × erythrophylla* leaves that developed in the shade (Shade), developed in the shade and were then transferred to full sunlight (Transferred), or developed in full sunlight (Sun)

	Chl <i>a</i> : Chl <i>b</i> ratio	Chl <i>a</i> + <i>b</i> (mg g ⁻¹ f. wt)	Carotenoids (mg g ⁻¹ f. wt)	Chl (<i>a</i> + <i>b</i>) : carotenoids ratio
Shade	2.41 ± 0.13 ^a	1.55 ± 0.21 ^b	0.41 ± 0.07 ^b	3.78 ± 0.34 ^b
Transferred	2.79 ± 0.21 ^b	0.99 ± 0.19 ^a	0.28 ± 0.03 ^a	3.54 ± 0.29 ^b
Sun	3.11 ± 0.25 ^b	1.16 ± 0.14 ^a	0.43 ± 0.09 ^c	2.70 ± 0.31 ^a

Mean values ± s.e. (*n* = 6).

Letters indicate values that differ significantly at *P* < 0.05.

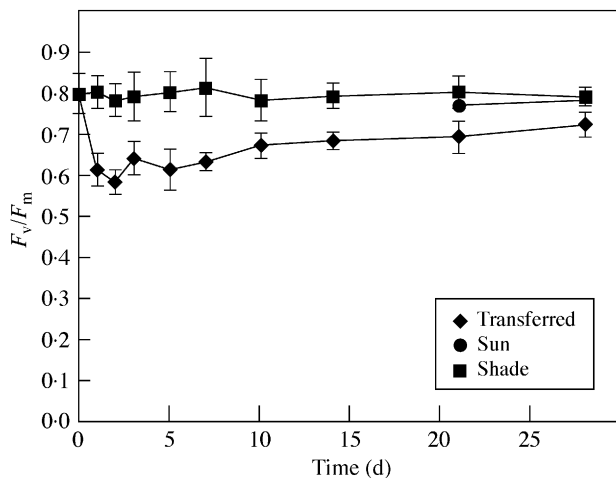


FIG. 1. The $F_v : F_m$ ratios of *B. × erythrophylla* leaves that developed in the shade (Shade), developed in the shade and were then transferred to full sunlight (Transferred), or developed in full sunlight (Sun). Mean values ± s.e. (*n* = 6).

inhibits, by 50 %, the control rate (0.025 units of absorbance at 560 nm min⁻¹).

Catalase (EC 1.11.1.6) was assayed by following the decrease in A_{240} as H₂O₂ was consumed (Rao *et al.*, 1996). The reaction mixture (1 ml) contained 50 mM potassium phosphate (pH 7.0), 37.5 mM H₂O₂ and up to 50 µl enzyme. Activity was calculated using the extinction coefficient 39.4 mM⁻¹ cm⁻¹.

All enzyme assays were conducted at 25 °C using a Jasco 550 spectrophotometer fitted with a MFC-132 temperature control cell. The conditions for all assays were chosen so that the rate of reaction was constant for the entire experimental period and proportional to the amount of enzyme added.

Extraction and determination of G3PDH and chloroplast FBPase

Tissue was ground to a fine powder in liquid nitrogen and total proteins were extracted by homogenizing the powdered tissue in extraction buffer (3 ml g⁻¹ f. wt). For G3PDH (EC 1.2.1.13) the extraction buffer contained 100 mM potassium phosphate (pH 8.0), 0.1 mM

EDTA and 4 % (w/v) Polyclar AT, and for FBPase (EC 1.11.1.9), 100 mM potassium phosphate (pH 8.0), 1.0 mM EDTA, 10 mM MgCl₂ and 4 % (w/v) Polyclar AT. Enzymes were assayed using the procedures of Harten and Eickmeier (1986) for G3PDH and Hurry *et al.* (1995) for FBPase. For G3PDH the reaction mixture (1 ml) contained 100 mM Tris-HCl (pH 8.0), 5 mM MgCl₂, 2 mM ATP, 1 mM 3-phosphoglyceric acid, 0.06 units of 3-phosphoglyceric phosphokinase from bakers' yeast, 0.14 mM NADPH and up to 50 µl enzyme extract. For chloroplast FBPase the reaction mixture (1 ml) contained 100 mM Tris-HCl (pH 8.0), 0.5 mM Na₂EDTA, 10 mM MgCl₂, 0.6 mM fructose-1,6-bisphosphate, 0.6 units glucose-6-phosphate dehydrogenase, 1.2 units glucose phosphate isomerase, both from bakers' yeast, 0.3 mM NADPH and up to 100 µl enzyme extract. For both enzymes the decrease in A_{340} was measured, and the enzyme activity was calculated using the extinction coefficient 6.2 mM⁻¹ cm⁻¹. Total activity was assayed on aliquots of enzyme extract incubated for 20 min with 20 mM DTT.

Statistical analysis

The significance of differences between mean values was determined by Duncan's multiple range test (SPSS 10 for Mac computers).

RESULTS

Effect of full sunlight on chlorophyll pigments

The chlorophyll and carotenoid contents of pre-existing and new leaves of plants transferred from shade to full sunlight were compared with those of shaded control plants, 28 d after transfer. The pigment composition of the leaves of shaded plants (shade-leaves) was typical of a shade-demanding species, with a high Chl (*a* + *b*) content, a low Chl *a* : Chl *b* ratio and a high Chl (*a* + *b*) : carotenoid ratio (Table 1). In contrast, new leaves developing on plants transferred to full sunlight (sun-leaves) had lower Chl (*a* + *b*) contents, a higher Chl *a* : Chl *b* ratio and a lower Chl (*a* + *b*) : carotenoid ratio (Table 1). Transferred-leaves had a significantly lower Chl (*a* + *b*) content and Chl *a* : Chl *b* ratio than shade-leaves, but no significant difference in the Chl (*a* + *b*) : carotenoid ratio (Table 1). No significant difference in pigment composition was found when the pre-existing and

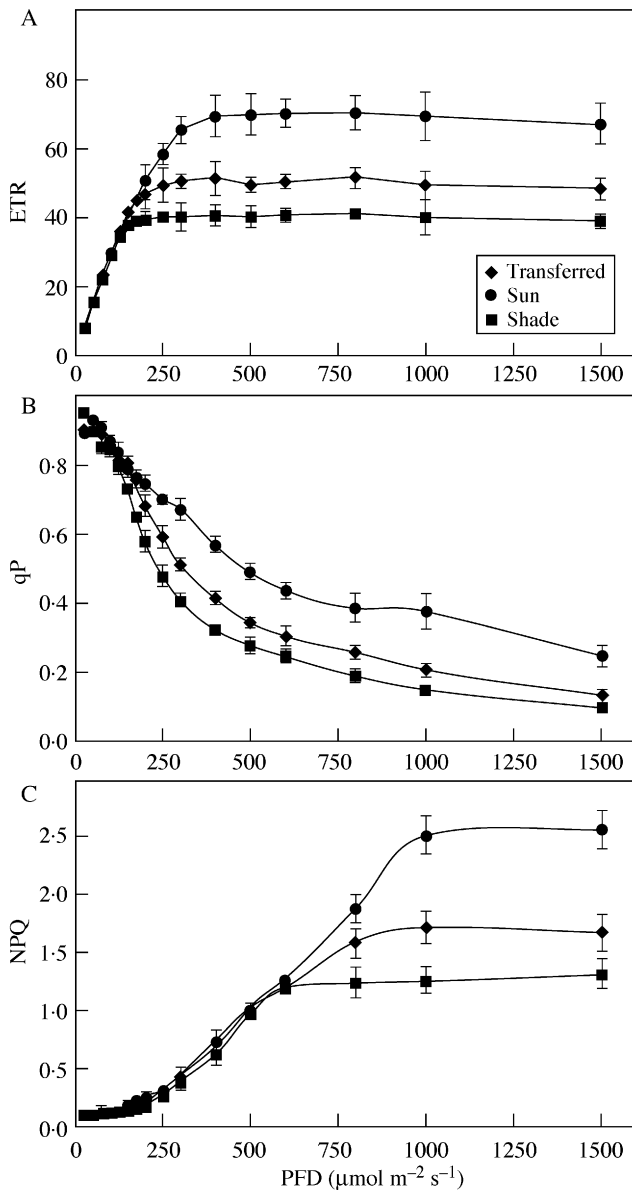


FIG. 2. Light response curves for ETR (A), qP (B) and NPQ (C) of *B. × erythrophylla* leaves that developed in the shade (Shade), developed in the shade and were then transferred to full sunlight (Transferred), or developed in full sunlight (Sun). Mean values \pm s.e. ($n = 6$).

new leaves of shaded plants were compared (data not presented).

Chlorophyll a fluorescence

Transfer to full sunlight rapidly caused a significant decline in F_v/F_m values (Fig. 1). The F_v/F_m values of transferred-leaves failed to recover completely, remaining significantly less than those of shade- and sun-leaves. The F_v/F_m values of sun-leaves were not significantly different to those of shade-leaves (Fig. 1).

The light response curves for ETR, qP and NPQ showed significant differences between the three leaf types (Fig. 2). ETR was irradiance-saturated for shade-leaves at $148 \mu\text{mol}$

$\text{m}^{-2} \text{s}^{-1}$, transferred-leaves at $248 \mu\text{mol m}^{-2} \text{s}^{-1}$ and for sun-leaves at $405 \mu\text{mol m}^{-2} \text{s}^{-1}$ (Fig. 2A). The ETRs of sun-leaves were higher than those of shade-leaves at irradiances greater than $148 \mu\text{mol m}^{-2} \text{s}^{-1}$, increasing to a maximum 75 % higher at irradiances of $405 \mu\text{mol m}^{-2} \text{s}^{-1}$ or greater. The ETRs of transferred-leaves were also higher than those of shade-leaves at irradiances greater than $148 \mu\text{mol m}^{-2} \text{s}^{-1}$, but only reached a maximum 25 % higher than those of shade-leaves at irradiances of $248 \mu\text{mol m}^{-2} \text{s}^{-1}$ or greater.

Typically, qP decreased in shade-, sun- and transferred-leaves as irradiance was increased (Fig. 2B). At low irradiances (up to $100 \mu\text{mol m}^{-2} \text{s}^{-1}$) there was little difference in qP values between the shade-, transferred- and sun-leaves, but above $148 \mu\text{mol m}^{-2} \text{s}^{-1}$ qP values of transferred- and sun-leaves were significantly higher than those of the shade-leaves. At irradiances above $200 \mu\text{mol m}^{-2} \text{s}^{-1}$ qP values of sun-leaves exceeded those of transferred-leaves.

The dependence of NPQ on irradiance is useful for estimating the relative increase in the non-radiative dissipation of absorbed excitation energy. Higher NPQ values were observed for sun- and transferred-leaves than shade-leaves, although the maximum NPQ value of transferred-leaves was still significantly lower than that of sun-leaves (Fig. 2C). Leaves from all sources showed irradiance-saturation of NPQ; for shade-leaves this occurred at $597 \mu\text{mol m}^{-2} \text{s}^{-1}$ and for transferred- and sun-leaves at $1004 \mu\text{mol m}^{-2} \text{s}^{-1}$.

Effect of full sunlight on H_2O_2 levels, lipid hydroperoxides and protein oxidation

Transfer to full sunlight rapidly caused a significant increase in the levels of leaf H_2O_2 (Fig. 3A). Twenty-four hours following transfer, levels of H_2O_2 had increased by nearly 300 %, remaining at high levels for 5 d and then declining to levels twice those found in shade-leaves (Fig. 3A). Levels of H_2O_2 in sun-leaves were not significantly different from those found in shade-leaves (Fig. 3A).

As observed for H_2O_2 , levels of lipid hydroperoxides increased rapidly when plants were transferred to full sunlight (Fig. 3B). Levels remained high for 7 d following transfer and then declined to levels approx. 30 % greater than those of shade-leaves. The levels of lipid hydroperoxides in sun-leaves were not significantly different from those found in shade-leaves (Fig. 3B).

The carbonyl content of leaves increased slightly following transfer to full sunlight, with levels remaining significantly elevated for 14 d (Fig. 3C). The carbonyl content of sun-leaves was not significantly different from that of shade-leaves (Fig. 3C).

Effect of full sunlight on metabolites

No significant differences in the total ascorbate pool were observed when transferred- and shade-leaves were compared (Fig. 4A). However, the total ascorbate pool of sun-leaves was 50 % greater than that of both shade- and transferred-leaves (Fig. 4A). Although no differences in total ascorbate were observed, in transferred-leaves the

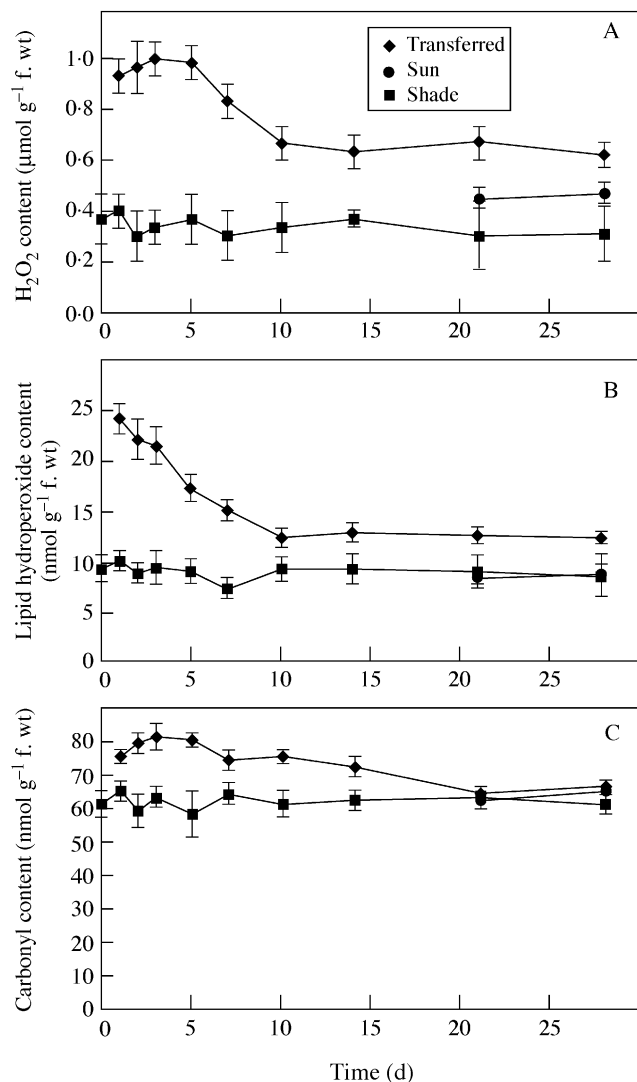


FIG. 3. The influence of growth conditions on the levels of H₂O₂ (A), lipid hydroperoxides (B), and carbonyl content (C) of *B. × erythrophylla* leaves that developed in the shade (Shade), developed in the shade and were then transferred to full sunlight (Transferred), or developed in full sunlight (Sun). Mean values ± s.e. ($n = 5$).

ASA:DHA ratio declined following transfer to full sunlight, and by day 7 was only 60 % that of shade-leaves (Fig. 4B). This decline reflects a decrease in reduced ascorbate (ASA) and an increase in oxidized ascorbate (DHA). Sun-leaves contained both larger pools of ascorbate and higher ASA : DHA ratios than shade-leaves, indicating a greater capacity for AOS scavenging (Fig. 4A and B). Unlike the ascorbate pool, the total glutathione pool in transferred-leaves declined rapidly upon exposure to full sunlight, dropping to almost 50 % that of shade-leaves (Fig. 4C). Glutathione levels then increased, reaching levels about 30 % greater than those found in shade-leaves after 10 d and remained at this level (Fig. 4C). Changes in the GSH : GSSG ratio paralleled the changes observed for total glutathione, initially decreasing and then increasing to a

value greater than that of shade-leaves after 14 d (Fig. 4D). As seen with ascorbate, sun-leaves contained both a larger pool of glutathione and a higher GSH : GSSG ratio than shade leaves (Fig. 4C and D).

Effect of full sunlight on antioxidative enzymes

With the exception of catalase all of the antioxidant enzymes assayed showed increased activity 24 h after transfer to full sunlight (Fig. 5A–F). SOD activity increased rapidly, reaching a peak of activity four times greater than that of shade-leaves within 7 d, and remained at this level (Fig. 5B). SOD activity of sun-leaves was four times greater than that of shade-leaves and was not significantly different from that of transferred leaves at 21 and 28 d (Fig. 5B).

GR, APOX, DHAR and MDHAR activities all showed a more gradual increase than SOD, peaking 5–10 d following transfer to full sunlight (Fig. 5C–F). After 28 d, activities had stabilized at approx. three, three, two and three times those found in shade-leaves, for GR, APOX, DHAR and MDHAR, respectively. In sun-leaves GR, APOX, DHAR and MDHAR activities were higher than those found in transferred-leaves after 28 d in full sunlight (Fig. 5C–F).

CAT activity initially declined upon transfer to full sunlight, dropping by 50 % within 24 h, before increasing to levels almost three times those found in shade-leaves (Fig. 5A). CAT activity in sun-leaves was higher than in transferred-leaves exposed to full sunlight for 28 d (Fig. 5A).

Effect of full sunlight on G3PDH and chloroplast FBPase activities

Both G3PDH and FBPase activities in transferred-leaves showed the same trends, with declines in both total activity and activation state upon transfer to full sunlight (Fig. 6). However, after 28 d in full sunlight the total activities of these enzymes had recovered to similar levels found in the shade, but the activation states of both enzymes remained lower than in shaded plants (Fig. 6). In sun-leaves, while the total activities of G3PDH and FBPase were almost 30 % higher than in shade-leaves, the activations states were similar (Fig. 6).

DISCUSSION

Many studies have shown that photo-inhibition occurs when plants are exposed to a PPFD higher than that required for the rate of CO₂ fixation and higher than that to which they have been acclimated (Asada, 2000). When compared with plants grown in the shade, those grown under higher PPFD usually have a decreased capacity to absorb incident radiation and an increased ability to dissipate excess excitation energy. *Begonia* × *erythrophylla* plants grown in the shade had F_v/F_m values of approx. 0.80, close to the 0.83 value typical for non-photo-inhibited vascular plants (Björkman and Demmig, 1987). However, upon transfer to full sunlight a decrease in F_v/F_m was observed, indicating a strong inhibition of photosynthetic efficiency. F_v/F_m remained depressed for over 1 week and even after 28 d had not fully recovered. This rapid decrease in F_v/F_m

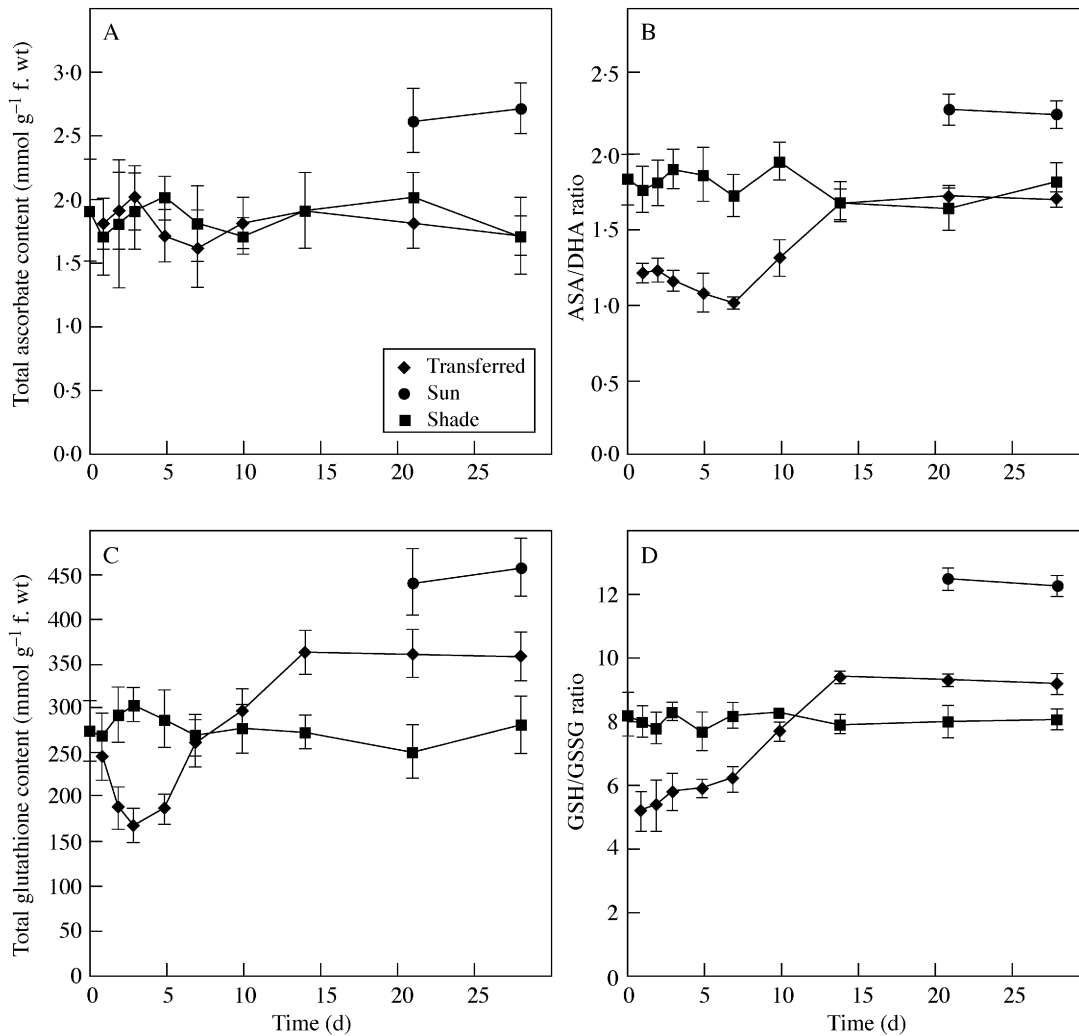


FIG. 4. The influence of growth conditions on total ascorbate (A), the reduced (ASA) : oxidized (DHA) ascorbate ratio (B), total glutathione (C) and the reduced (GSH) : oxidized (GSSG) glutathione ratio (D) of *B. × erythrophylla* leaves that developed in the shade (Shade), developed in the shade and were then transferred to full sunlight (Transferred), or developed in full sunlight (Sun). Mean values \pm s.e. ($n = 5$).

suggests a significant transient loss of PSII function followed by a slow and possibly incomplete recovery.

The limitations of pre-existing leaves transferred to full sunlight were also evident when the irradiance response curves for ETP and qP were compared with those of leaves that developed in full sunlight. At irradiances above $200 \mu\text{mol m}^{-2} \text{s}^{-1}$ the ETR and qP of sun-leaves greatly exceeded those of transferred-leaves, clearly indicating that transferred-leaves, unlike sun-leaves, were not fully adapted to utilize higher PPFD. As acclimation to high light levels involves a complex array of complementary and interdependent structural and functional alterations that enable efficient carbon gain in a new environment (Naidu and Delucia, 1997), it is logical to assume that full acclimation to high light levels can only be achieved in developing begonia leaves that have significant developmental plasticity and not in fully developed leaves.

A major contributor to the protection of the photosynthetic apparatus in plants growing in an environment with

excess excitation energy is an increased ability to dissipate this energy via non-photochemical quenching mechanisms such as the xanthophyll cycle, which is one of a variety of protection and repair mechanisms, including AOS scavenging systems, that help chloroplasts avoid damage due to excess PPFD (Gilles and Vidaver, 1990; Adamska, 1997; Logan *et al.*, 1998). According to Demmig-Adams (1990), the irradiance resulting in an increase in $1 - \text{qP}$ to a value above 0.6 is excess irradiance. Therefore the begonia plants grown in the shade ($155 \mu\text{mol m}^{-2} \text{s}^{-1}$) were grown under a low to moderate excitation pressure ($1 - \text{qP} < 0.3$), while those under excess excitation pressure grown in the sun were $1 - \text{qP} > 0.6$. NPQ is thought to be a good indicator of the concentration of dissipating complexes (Gilmore *et al.*, 1995), and hence the ability of a plant to dissipate light energy in excess of that required for CO_2 assimilation. Compared with leaves that developed in full sunlight, those of shaded plants and the pre-existing leaves of plants transferred to full sunlight had lower NPQ values at higher

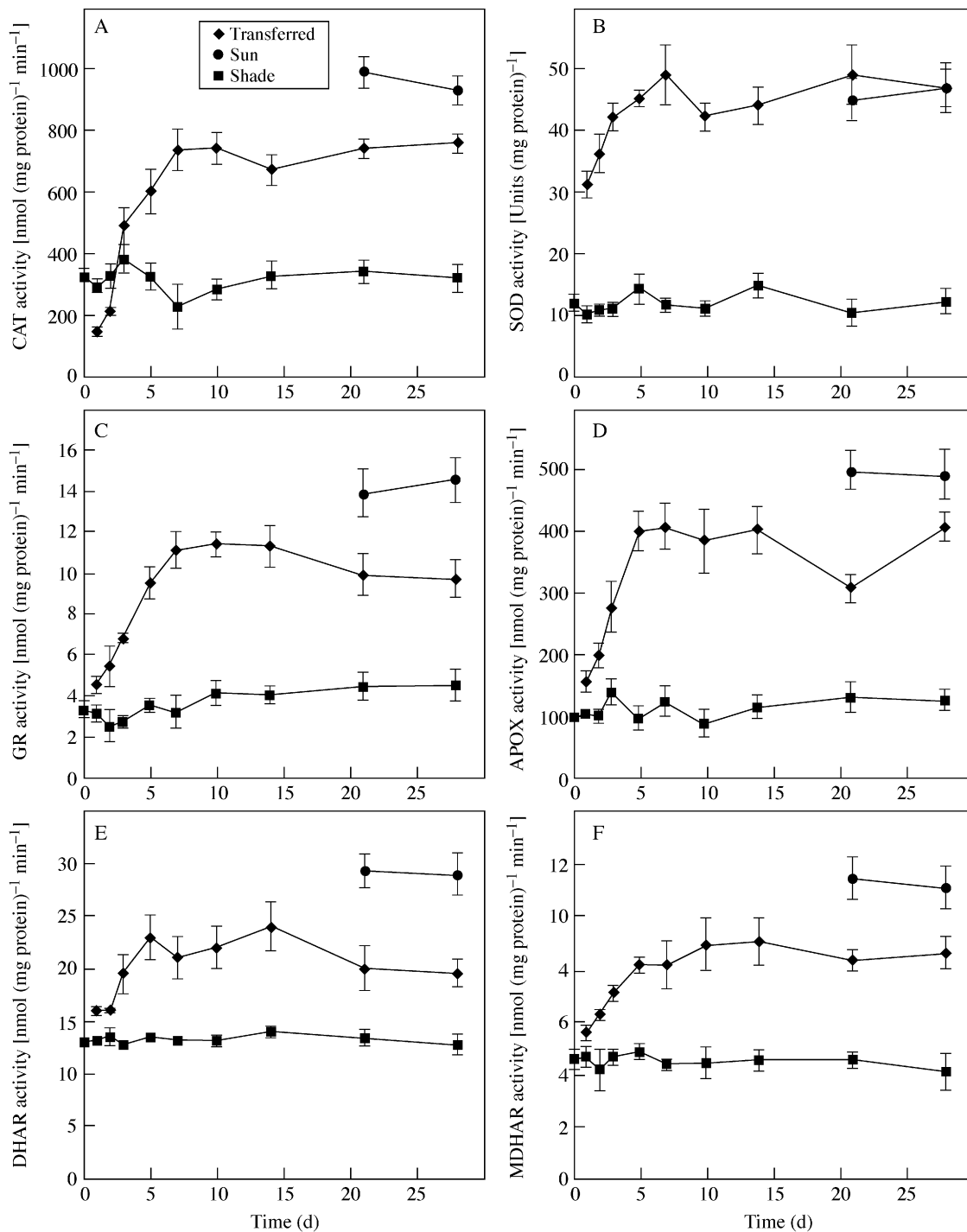


FIG. 5. The influence of growth conditions on the levels of CAT (A), SOD (B), GR (C), APOX (D), DHAR (E) and MDHAR (F), in *B. × erythrophylla* leaves that developed in the shade (Shade), developed in the shade and were then transferred to full sunlight (Transferred), or developed in full sunlight (Sun). Mean values \pm s.e. ($n = 5$).

light levels than sun-leaves. This would probably result in greater damage to the photosynthetic apparatus and the generation of higher concentrations of AOS when shade-leaves are exposed to high PPFD. This could lead rapidly to photo-bleaching and hence the loss of photosynthetic

pigments seen in this study. In the long term, the loss of photosynthetic pigments could be viewed as a protection mechanism as it would decrease the capacity of the leaf to absorb incident radiation and hence reduce the amount of excess excitation energy that has to be dissipated by NPQ.

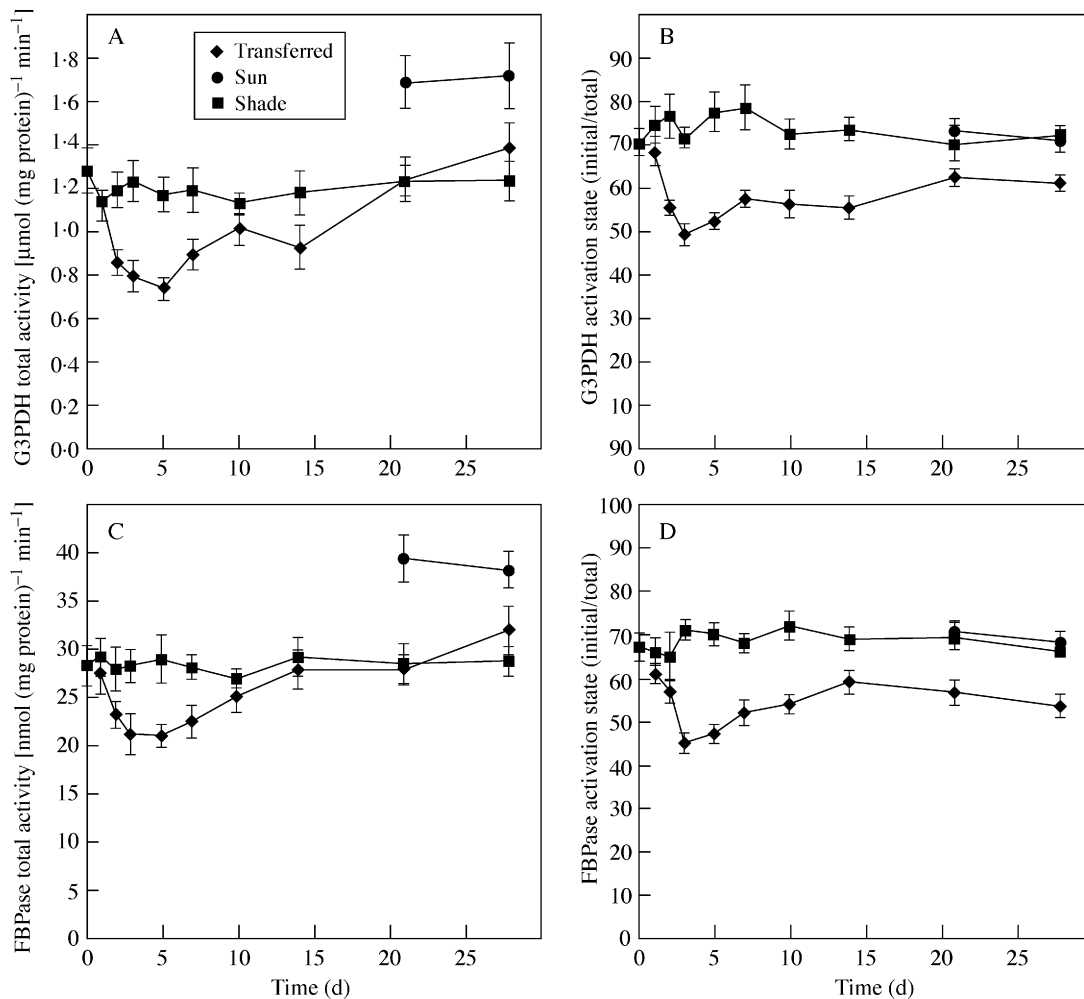


FIG. 6. The influence of growth conditions on the total activity of G3PDH (A), the activation state of G3PDH (B), the total activity of FBPase (C), and the activation state of FBPase (D) of *B. × erythrophylla* leaves that developed in the shade (Shade), developed in the shade and were then transferred to full sunlight (Transferred), or developed in full sunlight (Sun). Mean values \pm s.e. ($n = 5$).

The results presented here show that H_2O_2 is rapidly produced following transfer of *B. × erythrophylla* plants from shade to full sunlight. The amount of H_2O_2 produced was within the range $0.1\text{--}1 \mu\text{mol g}^{-1} \text{f. wt}$ and is similar to levels found previously in plant cells (Veljovic-Jovanovic *et al.*, 2003). Relatively low levels of H_2O_2 have been shown to cause the de-activation of key enzymes required for photosynthetic carbon reduction, such as chloroplastic FBPase and G3PDH (Charles and Halliwell, 1981; Takeda *et al.*, 1995). Therefore, it is important that photosynthetic cells have only low levels of hydrogen peroxide and that it is scavenged efficiently. When *B. × erythrophylla* plants grown in the shade were transferred to full sunlight, ASA and GSH levels decreased rapidly as did the size of the total glutathione pool. In addition to the general oxidation of proteins, rapid decreases in both the initial and total activities of chloroplastic FBPase and G3PDH were observed. GSH is less stable than GSSG under conditions of oxidative stress because AOS can rapidly oxidize GSH to GSSG, but the reduction of GSSG to GSH requires GR and NADPH. High H_2O_2 levels, combined with low GSH : GSSG

and ASA : DHA ratios and a reduction in the size of the glutathione pool would leave the essential sulfhydryl groups of FBPase and G3PDH vulnerable to oxidative damage and could account for the reduced activities observed in this study.

Although the initial activities of FBPase and G3PDH recovered partially as transferred-leaves began to acclimate to higher irradiances, and the glutathione levels and ASA : DHA and GSH : GSSG ratios increased, the activation states of both enzymes remained lower than before transfer to full sunlight and lower than those of sun-leaves. This suggests that even though the activities of antioxidant enzymes increased progressively in response to the oxidative stress that occurred following the transfer of leaves to full sunlight, this increased AOS scavenging capacity was still not sufficient to allow the activation states of FBPase and G3PDH to recover to levels found in shade- and sun-leaves. This reduction in enzyme activity would result in pre-existing leaves having a lower capacity to fix carbon in full sunlight than newly formed leaves or leaves of plants grown in the shade, and hence having reduced

photosynthetic efficiency. Interestingly, in addition to elevated levels of antioxidant enzymes, the pools of both ascorbate and glutathione in sun-leaves were greater than those in shade-leaves and transferred-leaves, even after 28 d in full sunlight. Therefore it appears that under full sunlight, an increase in the size of both the ascorbate and glutathione pools, and not just an increase in the ability to regenerate ASA and GSH, could be important in keeping AOS at low levels and controlling the cellular redox status.

In addition to inhibiting the activity of key enzymes, cellular H_2O_2 can react with O_2^- in the presence of Fe^{2+} , to form highly reactive hydroxyl radicals ($OH\cdot$) that can trigger the autocatalytic process of lipid peroxidation and cause severe membrane damage (Halliwell and Gutteridge, 1989). The increase in lipid hydroperoxides observed upon transfer of plants to full sunlight is a clear indicator that membrane damage occurs in the leaves of *B. × erythrophylla* plants transferred from shade to full sunlight. Using isolated thylakoids, Jakob and Heber (1996) demonstrated that hydroxyl radicals can also inactivate both PSI and PSII, although PSI is considered to be more vulnerable to hydroxyl radical inactivation as the Fe-S centres in the vicinity of P700 provide an environment ideal for hydroxyl radical formation (Sonoike, 1998). Inactivation of PSII and/or PSI would reduce the ability of pre-existing leaves to utilize higher irradiances, and hence result in a further reduction in photosynthetic efficiency upon transfer to full sunlight. The combined effects on photosynthetic efficiency of oxidative damage to the photosystems and key enzymes required for carbon acquisition would result in pre-existing leaves presenting a greater maintenance load on the plant and hence a temporary reduction in growth efficiency while pre-existing leaves are replaced with new, high-light acclimated, leaves.

Previous studies have indicated that increases in both the levels of cellular antioxidants and the activities of enzymes involved in antioxidant metabolism generally accompany exposure to high light levels (Foyer *et al.*, 1997). Logan *et al.* (1998) suggested that in *Cucurbita* and *Vinca* plants exposed to increasing light intensities, photo-oxidative stress is linked to photo-inhibition and that protection against photo-oxidative stress is important for the acclimation of plants to high light levels. The change in the size of the glutathione pool, the redox status of the glutathione and ascorbate pools and the activities of enzymes involved in the scavenging of AOS, following transfer of *B. × erythrophylla* plants to full sunlight, are indicative of oxidative stress. In particular, the changes in the ratios of reduced to oxidized antioxidants represent a change in the redox state of cells indicating a more strongly oxidizing environment for the first 7–10 d following transfer to full sunlight. This shift in redox status is followed by a progressive activation of the antioxidant systems that help protect cells against AOS, and so could be important in regulating antioxidant metabolism (Foyer and Noctor, 2000; Noctor *et al.*, 2000). These findings suggest that in *B. × erythrophylla*, as in *Cucurbita* and *Vinca* plants, oxidative stress and high-light stress are linked.

It has been suggested that enhancements in the activities of activated oxygen scavenging enzymes during high-light

stress occur mostly in the chloroplasts, because in photosynthetic tissues the ascorbate–glutathione cycle is localized principally in the chloroplasts, which are the major sites of AOS production in leaves (Foyer *et al.*, 1997). The rapid increase in SOD activity and the gradual increase in the activities of APOX, GR, DHAR and MDHAR under full sunlight indicate that this is true in *B. × erythrophylla*. However, the activity of antioxidant enzymes in other cell compartments may also be important for scavenging AOS. There is increasing evidence indicating the importance of the cytosolic isoforms of antioxidative enzymes, and those associated with the mitochondria and peroxisomes, in plants grown under various environmental stress conditions (Alscher *et al.*, 1997; Karpinski *et al.*, 1997; Logan *et al.*, 1998).

Of the enzymes assayed in this study only CAT showed a decrease in activity. This decrease was transient and occurred immediately upon transfer to full sunlight. This response is not surprising, as it has been reported that both the synthesis and degradation of CAT are light sensitive (Hertwig *et al.*, 1992). The decline in CAT activity combined with the rapid oxidation of both the ascorbate and glutathione pools, and the rapid increase in SOD activity that would have resulted in the rapid conversion of O_2^- to H_2O_2 , could alone have resulted in the rapid increase in cellular H_2O_2 observed following transfer to full sunlight. However, a reduction in CAT could also lead to reduced breakdown of the H_2O_2 formed as a result of photorespiration and mitochondrial electron transport. Hence H_2O_2 from different sources could contribute to the increased levels of cellular H_2O_2 observed in *B. × erythrophylla*. Karpinski *et al.* (1997) demonstrated that exposure of arabidopsis plants to light stress not only triggers AOS production in the chloroplast, but also in the cytosol. They suggested that induction of AOS scavenging mechanisms in the cytosol provides an important secondary defence against AOS that occurs before the chloroplast AOS scavenging systems are saturated and that cytosolic APOXs play an important role in this process. Such a mechanism could be important in *B. × erythrophylla* as the transient decline in CAT activity following transfer to full sunlight could result in a rapid build up of cytosolic H_2O_2 .

The results of this study show that shade-acclimated leaves of *B. × erythrophylla* have a limited ability for photosynthetic acclimation following transfer to full sunlight, as demonstrated by lower NPQ values than sun-leaves, and an inability to completely recover F_v/F_m and the activation states of FBPase and G3PDH, elevated levels of AOS and oxidative damage compared with sun- and shade-leaves. However, pre-existing leaves do have the ability to rapidly up-regulate their capacity to scavenge AOS by increasing levels of SOD, CAT and the enzymes of the ascorbate–glutathione cycle, following transfer to full sunlight. Increased GR activity would help maintain the pool of glutathione in the reduced state, allowing GSH to be used by DHAR to reduce DHA to ASA (Noctor *et al.*, 1998). In addition, elevated levels of MDHAR would allow more MDHA to be regenerated directly back to ASA providing enough NAD(P)H is available. ASA can then be used as a reductant by APOX to catalyse the reduction of H_2O_2 to

H₂O. Elevated APOX in combination with elevated CAT would lower H₂O₂ levels, which would in turn lead to a reduction in lipid peroxidation and reduced damage to the photosystems under full sunlight.

Of particular interest is what triggers the increased activities of these enzymes. As increased H₂O₂ levels, oxidative damage and a shift in redox to a more oxidized cellular environment occur within 24 h of transfer to full sunlight, redox signalling could be important for the up-regulation of antioxidant enzyme activities. There is evidence that both the ASA : DHA ratio and the glutathione redox state may function as cellular regulatory signals (Foyer *et al.*, 1997; Noctor *et al.*, 2000). How the activity of the scavenging enzymes is up-regulated in response to cellular signals is also unknown and will require further investigation. Mobilization of inactive enzyme pools, adaptive changes in the catalytic properties of the enzymes induced by the cellular environment and/or the transcription of usually silent genes are all possibilities.

In conclusion, AOS metabolism is clearly important for *B. × erythrophylla* plants developing in full sunlight. Although leaves that develop in full sunlight have reduced pigment concentrations and a greater capacity for NPQ than those of plants grown in the shade, they still develop significantly larger ascorbate and glutathione pools, and have greater activities of CAT, SOD, GR, APOX, DHAR and MDHAR. In addition, shade-leaves of *B. × erythrophylla* possess a limited capacity for acclimation to high light levels, with an enhanced capacity for scavenging AOS appearing to play an important role.

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