

Structure of a partially unfolded form of *Escherichia coli* dihydrofolate reductase provides insight into its folding pathway

Joseph R. Kasper, Pei-Fen Liu, and Chiwook Park*

¹Department of Medicinal Chemistry and Molecular Pharmacology, Bindley Bioscience Center, Purdue University, West Lafayette, Indiana 47907

Received 26 August 2014; Revised 19 September 2014; Accepted 22 September 2014

DOI: 10.1002/pro.2555

Published online 23 September 2014 proteinscience.org

Abstract: Proteins frequently fold via folding intermediates that correspond to local minima on the conformational energy landscape. Probing the structure of the partially unfolded forms in equilibrium under native conditions can provide insight into the properties of folding intermediates. To elucidate the structures of folding intermediates of *Escherichia coli* dihydrofolate reductase (DHFR), we investigated transient partial unfolding of DHFR under native conditions. We probed the structure of a high-energy conformation susceptible to proteolysis (cleavable form) using native-state proteolysis. The free energy for unfolding to the cleavable form is clearly less than that for global unfolding. The dependence of the free energy on urea concentration (*m*-value) also confirmed that the cleavable form is a partially unfolded form. By assessing the effect of mutations on the stability of the partially unfolded form, we found that native contacts in a hydrophobic cluster formed by the F-G and Met-20 loops on one face of the central β -sheet are mostly lost in the partially unfolded form. Also, the folded region of the partially unfolded form is likely to have some degree of structural heterogeneity. The structure of the partially unfolded form is fully consistent with spectroscopic properties of the near-native kinetic intermediate observed in previous folding studies of DHFR. The findings suggest that the last step of the folding of DHFR involves organization in the structure of two large loops, the F-G and Met-20 loops, which is coupled with compaction of the rest of the protein.

Keywords: DHFR; partially unfolded form; folding intermediate; proteolysis; folding pathway

Introduction

Escherichia coli dihydrofolate reductase (DHFR; 5,6,7,8-tetrahydrofolate: NADP⁺ oxidoreductase, EC

Abbreviations: DHFR, C85A/C152S *E. coli* dihydrofolate reductase; EDTA, ethylenediaminetetraacetic acid; PAGE, polyacrylamide gel electrophoresis; SDS, sodium dodecyl sulfate.

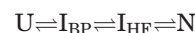
Additional Supporting Information may be found in the online version of this article.

Pei-Fen Liu's current address is Department of Food Science and Biotechnology, National Chung Hsing University, Taichung, Taiwan

Grant sponsor: National Science Foundation; Grant number: 1021652.

*Correspondence to: Chiwook Park, 575 Stadium Mall Drive, Purdue University, West Lafayette, IN 47907.
E-mail: chiwook@purdue.edu

1.5.1.3) is one of the best characterized model systems in protein folding studies as well as in enzymology and structural biology.^{1–4} The protein is a 159 amino acid monomeric enzyme that catalyzes the reduction of 7,8-dihydrofolate to 5,6,7,8-tetrahydrofolate using NADPH as a cofactor. Folding of DHFR has been shown to occur through two groups of distinctive kinetic intermediates, I_{BP} and I_{HF} ^{5–12}:



Each group is believed to include several different intermediates that share similar spectroscopic properties. Within microseconds, unfolded DHFR collapses into I_{BP} , a group of early intermediates that are responsible for the burst phase amplitude observed by circular dichroism (CD). Formation of

I_{BP} is followed by slower formation of I_{HF} , another group of intermediates characterized by high fluorescence (I_{HF}) and finally by formation of a group of native conformers.^{6,12}

Although DHFR folding and structure have been the subject of numerous studies, the structure of the folding intermediates remains elusive. Hydrogen/deuterium exchange pulse labeling during refolding showed that two hydrophobic clusters form in I_{BP} (Fig. 1).⁷ The high fluorescence of I_{HF} has been attributed to formation of a hydrophobic cluster around Trp47 and Trp74 (Fig. 1).¹³ However, the structure of I_{HF} is largely unknown. Pulse labeling could not provide information on I_{HF} because prolonged incubation at acidic pH necessary for pulse quenching causes DHFR to aggregate.⁷ Although several attempts have been made to computationally model the folding of DHFR, these efforts do not converge upon a single structural description of DHFR folding.^{14–16}

The structures of partially unfolded forms at equilibrium with native protein provide information on protein folding intermediates.^{17,18} The current understanding of protein folding holds that proteins fold along the surface of funnel-like conformational energy landscapes, which contain an ensemble of partially unfolded forms in local energy minima.^{19–21} This view of protein folding suggests that, once folded, a natively folded protein makes transient excursions to the partially unfolded forms. Therefore, from the structures of the partially unfolded forms in equilibrium under native conditions, one can obtain valuable information on the sequence of structural acquisition along the folding pathway.^{22,23}

To obtain structural information on the folding intermediates of DHFR, we investigated a partially unfolded form of DHFR by native-state proteolysis. Native-state proteolysis is a method to determine the energetics of partial unfolding under native conditions.²⁴ Combined with site-directed mutagenesis, native-state proteolysis also provides information on the structure of partially unfolded forms.²⁵ The method is based on the observation that proteolysis of folded proteins requires transient unfolding to a cleavable conformation as shown in the following scheme:



where k_{op} and k_{cl} are the forward and the reverse rate constant for unfolding to a cleavable form, and k_{int} is the rate constant for proteolysis of the cleavable form. When k_{int} is significantly smaller than k_{cl} (EX2 condition), the apparent rate constant of the reaction (k_{obs}) is expressed as:

$$k_{obs} = K_{op}k_{int}, \quad (1)$$

where K_{op} is the equilibrium constant for unfolding to the cleavable form. By approximating k_{int} with the rate for proteolysis of a peptide substrate, one

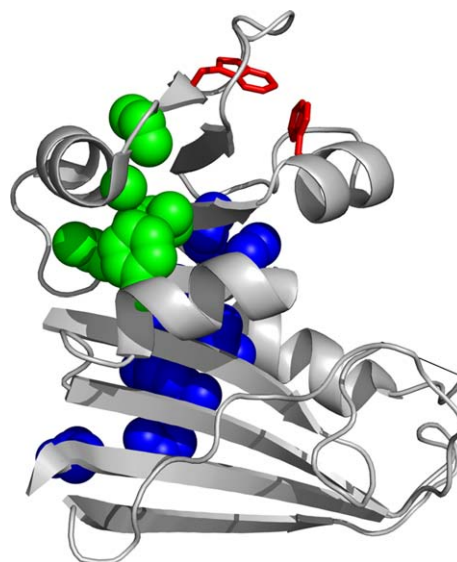


Figure 1. Structure of DHFR. The ribbon representation of the structure of DHFR (PDB CODE: 5DFR) is shown. The side chains of the residues in the two hydrophobic clusters that form in I_{BP} are shown in blue and green spheres. The side chains of Trp47 and Trp74, which are responsible for high fluorescence in I_{HF} , are shown in red sticks. The invisible part of the Met20 loop is shown in a straight line. The image was created with PyMOL.

can calculate K_{op} from k_{obs} and also the free energy required for unfolding (ΔG_{op}°) from K_{op} .

Native-state proteolysis is conceptually analogous to native-state hydrogen/deuterium (H/D) exchange, in which partially unfolded forms are probed by the exchange between amide protons with solvent deuterium.^{18,22,23} Although native-state H/D exchange may report the presence of multiple partially unfolded forms, native-state proteolysis reports only the most accessible partially unfolded forms, because unfolding to the most accessible form dominates proteolysis kinetics. Despite this limitation, native-state proteolysis is a valuable alternative to native-state H/D exchange and a suitable choice for DHFR. Because proteolysis requires unfolding of a segment with at least 8–12 residues,²⁶ native-state proteolysis is relatively insensitive to local fluctuation, which frequently complicates detection of partial unfolding by native-state H/D exchange. Also, while the adjustment of the intrinsic exchange rate (k_{int}) in H/D exchange requires a change in pH, the intrinsic proteolysis rate of unfolded peptides in native-state proteolysis can be controlled simply by a change in protease concentration. This aspect of native-state proteolysis provides flexibility in the choice of pH. Because DHFR tends to aggregate at low pH,⁷ which is frequently necessary to probe partial unfolding by native-state H/D exchange, the use of a neutral pH in native-state proteolysis is especially advantageous.

Here, we report the structure of a partially unfolded form of DHFR we probed by native-state

Table I. *Equilibrium Unfolding Parameters*

DHFR	m -Value (kcal mol ⁻¹ M ⁻¹)	C_m (M)	ΔG_{U-N}° (kcal mol ⁻¹)	$\Delta\Delta G_{U-N}^{\circ a}$ (kcal mol ⁻¹)
Wild type	2.3 ± 0.1	2.87 ± 0.01	6.6 ± 0.3	—
L4A	2.0 ± 0.1	1.47 ± 0.03	2.9 ± 0.1	-3.7 ± 0.3
L8A	2.1 ± 0.1	2.12 ± 0.01	4.5 ± 0.1	-2.1 ± 0.3
I61A	2.8 ± 0.2	1.77 ± 0.01	5.0 ± 0.4	-1.6 ± 0.5
I91A	2.6 ± 0.1	2.29 ± 0.01	6.0 ± 0.2	-0.6 ± 0.4
L110V	2.6 ± 0.1	2.14 ± 0.01	5.6 ± 0.2	-1.0 ± 0.4
L112V	2.2 ± 0.1	2.21 ± 0.01	4.9 ± 0.2	-1.7 ± 0.4
V136A	2.5 ± 0.1	2.26 ± 0.01	5.7 ± 0.1	-0.9 ± 0.3
L156A	2.1 ± 0.1	1.75 ± 0.02	3.7 ± 0.1	-2.9 ± 0.3

The m -values, C_m , and the global stability (ΔG_{U-N}°) of each variant were determined by monitoring unfolding in urea by circular dichroism. Standard errors of m -values and C_m values are from nonlinear curve-fitting of urea unfolding to a two-state model. Standard errors of ΔG_{U-N}° and $\Delta\Delta G_{U-N}^\circ$ are from error propagation.

^a $\Delta\Delta G_{U-N}^\circ = \Delta\Delta G_{U-N}^\circ(\text{mutant}) - \Delta\Delta G_{U-N}^\circ(\text{wild type})$.

proteolysis in combination with site-directed mutagenesis. We compare the identified partially unfolded form with known folding intermediates of DHFR, using the previously reported structural features of the folding intermediates. We also postulate the molecular events at the final stage of DHFR folding based on the structure of the partially unfolded form.

Results

Proteolysis of DHFR occurs through partial unfolding

DHFR used in this study is a cysteine-free version of DHFR (C85A/C152S DHFR), which is more resistant to oxidation than wild-type DHFR. Still, the cysteine-free DHFR has been shown to have activity, stability, and a folding mechanism comparable to wild-type DHFR.²⁷ For simplicity, here we refer to the cysteine-free DHFR as wild type.

We probed unfolding of DHFR under native conditions by native-state proteolysis with a nonspecific protease, thermolysin. We determined the apparent first-order rate constant for proteolysis (k_{obs}) by monitoring the change in the band intensities of remaining intact protein on a SDS-PAGE gel from a proteolysis reaction quenched at various time points. Determination of the equilibrium constant for unfolding to a cleavable form (K_{op}) requires EX2-like conditions ($k_{\text{cl}} \gg k_{\text{int}}$). Because k_{int} is dependent on the protease concentration, EX2-like conditions can be achieved and also confirmed by adjusting the protease concentration. To identify the optimal range of protease concentration for EX2-like conditions, we determined k_{obs} at varying concentrations of thermolysin and confirmed that proteolysis of DHFR satisfies EX2-like conditions up to 70 $\mu\text{g mL}^{-1}$ thermolysin (Supporting Information Fig. S1). Then, we determined K_{op} from k_{obs} by estimating k_{int} with k_{cat}/K_m for cleavage of an unstructured peptide substrate. From the K_{op} value, the free energy for unfolding to the cleavable form ($\Delta G_{\text{op}}^\circ$) was deter-

mined to be 4.9 ± 0.1 kcal mol⁻¹, which is significantly less than the free energy required for global unfolding (6.6 ± 0.3 kcal mol⁻¹) as determined by equilibrium unfolding monitored by circular dichroism (Table I). The observation that unfolding to the cleavable form requires less free energy than global unfolding indicates that proteolysis occurs through a form distinct from the globally unfolded form.

To assess the scale of unfolding to the cleavable form, we measured $\Delta G_{\text{op}}^\circ$ at varying concentrations of urea. The free energy for cooperative unfolding in proteins depends linearly on urea concentration, and the dependence (m -value) is proportional to the amount of buried surface area exposed upon unfolding.²⁸ The plot of $\Delta G_{\text{op}}^\circ$ versus urea reveals two phases with distinct slopes (Fig. 2). Because the slope is proportional to the change in the solvent accessible surface area upon unfolding, the presence of two phases indicates that proteolysis occurs through two distinct cleavable forms depending on

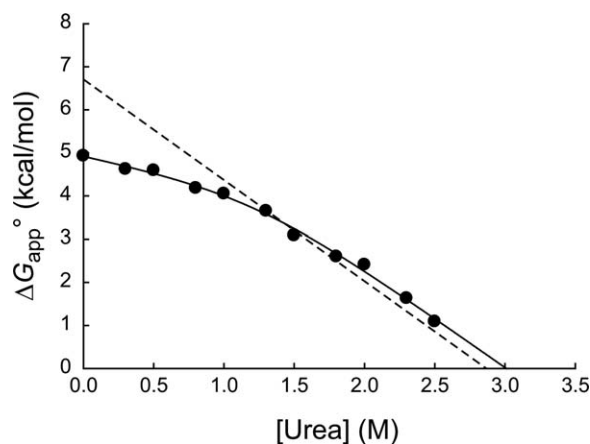


Figure 2. Native-state proteolysis of wild-type DHFR. The dependence of the free energy of unfolding ($\Delta G_{\text{app}}^\circ$) determined by proteolysis (\bullet) and by equilibrium unfolding (---) is plotted against urea concentration. The plot of $\Delta G_{\text{app}}^\circ$ versus urea concentration was fit to Eq. (2) (solid line) to estimate parameters for partial unfolding (m_{op} and $\Delta G_{\text{op,H}_2\text{O}}^\circ$) and global unfolding (m_{unf} and $\Delta G_{\text{unf,H}_2\text{O}}^\circ$).

Table II. Proteolysis Kinetics and φ_c Values

DHFR	k_p^a ($\times 10^{-3} \text{ s}^{-1}$)	$k_p(\text{mut})/k_p(\text{wt})$	$\Delta\Delta G_{C-N}^{\circ b}$ (kcal mol $^{-1}$)	φ_c
Wild type	0.087 ± 0.003	—	—	—
L4A	2.2 ± 0.2	25 ± 2	-1.91 ± 0.05	0.49 ± 0.04
L8A	2.4 ± 0.1	28 ± 2	-1.97 ± 0.04	0.05 ± 0.1
I61A	0.20 ± 0.01	2.3 ± 0.1	-0.49 ± 0.02	0.69 ± 0.09
I91A	0.140 ± 0.008	1.6 ± 0.1	-0.28 ± 0.04	0.5 ± 0.2
L110V	1.3 ± 0.1	14 ± 1	-1.56 ± 0.04	-0.6 ± 0.5
L112V	3.3 ± 0.2	38 ± 2	-2.15 ± 0.03	-0.3 ± 0.3
V136A	0.20 ± 0.02	2.3 ± 0.2	-0.49 ± 0.05	0.5 ± 0.2
L156A	3.3 ± 0.2	38 ± 2	-2.15 ± 0.03	0.24 ± 0.07

^a k_p is the first-order rate constant for the proteolysis of 0.10 mg mL $^{-1}$ DHFR by 20 $\mu\text{g mL}^{-1}$ thermolysin. The standard error of k_p is from nonlinear curve-fitting. The standard errors of $\Delta\Delta G_{C-N}^{\circ}$ and φ_c are from error propagation.

^b $\Delta\Delta G_{C-N}^{\circ} = \Delta G_{C-N}^{\circ}(\text{mutant}) - \Delta G_{C-N}^{\circ}(\text{wild type})$.

the concentration of urea. The shallow phase at lower concentrations of urea suggests proteolysis through partial unfolding, and the steep phase at higher concentrations of urea suggests proteolysis through global unfolding. At lower concentrations of urea, the partially unfolded form is energetically more accessible than the globally unfolded form. As the urea concentration increases, the globally unfolded form becomes more accessible than the partially unfolded form.

We fit the plot of $\Delta G_{\text{op}}^{\circ}$ versus urea using a model in which proteolysis occurs through unfolding to two cleavable forms with different m -values (Fig. 2). The m -values for the shallow phase and the steep phase were determined to be $0.7 \pm 0.3 \text{ kcal mol}^{-1} M^{-1}$ and $2.3 \pm 0.2 \text{ kcal mol}^{-1} M^{-1}$, respectively. The m -value of the steep phase agrees well with the m -value of global unfolding ($2.3 \pm 0.1 \text{ kcal mol}^{-1} M^{-1}$) determined by equilibrium unfolding monitored by circular dichroism (Table I). This consistency in m -value corroborates that proteolysis occurs through global unfolding in the steep phase. The smaller m -value of the shallow phase also confirms that proteolysis occurs through partial unfolding in the shallow phase. Compared with the m -value for global unfolding ($2.3 \pm 0.1 \text{ kcal mol}^{-1} M^{-1}$), the m -value of the shallow phase ($0.7 \pm 0.3 \text{ kcal mol}^{-1} M^{-1}$) suggests that the partial unfolding exposes about 30% of the buried surface that is exposed upon global unfolding. The free energies of the shallow phase and the steep phase were extrapolated to be $4.9 \pm 0.1 \text{ kcal mol}^{-1}$ and $6.9 \pm 0.6 \text{ kcal mol}^{-1}$ in 0M urea, respectively. Proteolysis of the unfolded protein occurs before proline residues reach their *cis-trans* equilibrium. The lack of proline isomerization in the unfolded forms is expected to increase the free energy of global unfolding measured by proteolysis by $0.7 \text{ kcal mol}^{-1}$ in DHFR.²⁹ Even after this correction, the free energy of the steep phase agrees well with the free energy for global unfolding determined by equilibrium unfolding ($6.6 \pm 0.3 \text{ kcal mol}^{-1}$). The uncertainty in the rate constant for intrinsic proteolysis (k_{int}) that we estimate with a peptide substrate may also contribute to the slight discrepancy.

Probing the partially unfolded form by site-directed mutagenesis

To determine the region of DHFR that becomes unfolded in the partially unfolded form, we probed the structure of the partially unfolded form by site-directed mutagenesis. To destabilize the protein with only minimal structural perturbation, we limited our mutations to leucine to valine, leucine, and isoleucine to alanine, or valine to alanine.³⁰ Specifically, we prepared L4A, L8A, I61A, I91A, L110V, L112V, V136A, and L156A DHFR. We confirmed that all variants maintain native-like structures by circular dichroism (Supporting Information Fig. S2).

From the global stability and the free energy of partial unfolding of each variant, we determined the φ_c value, which is the ratio of the change in the stability of the partially unfolded form ($\Delta\Delta G_{U-C}^{\circ}$) to that of the native form ($\Delta\Delta G_{U-N}^{\circ}$) caused by a mutation.²⁵ If the mutated residue lies within the folded region of the partially unfolded form, the partially unfolded form and the native form will be equally destabilized, and the φ_c value of the variant will be close to 1. If a mutated residue lies within the unfolded region of the partially unfolded form, only the native form will be destabilized, and the φ_c value of the variant will be close to 0. Therefore, the distribution of φ_c values in a protein structure reports which part of the protein is unfolded in a partially unfolded form.²⁵ We calculated the destabilization of the native form ($\Delta\Delta G_{U-N}^{\circ}$) by comparing the global stability of each variant of DHFR to that of wild-type DHFR (Table I) after determining the global stability of each by equilibrium unfolding in urea (Supporting Information Fig. S3). We determined the destabilization of the partially unfolded form ($\Delta\Delta G_{U-C}^{\circ}$) indirectly from the change in the free energy for partial unfolding ($\Delta\Delta G_{C-N}^{\circ}$) determined by native-state proteolysis (Table II and Supporting Information Fig. S4).

Mutations show a broad spectrum of φ_c values (Table II). Interestingly, no mutation has a φ_c value close to 1. Ile61 has the highest φ_c value of 0.69. Leu4, Ile91, and Val136 have φ_c values close to 0.5.

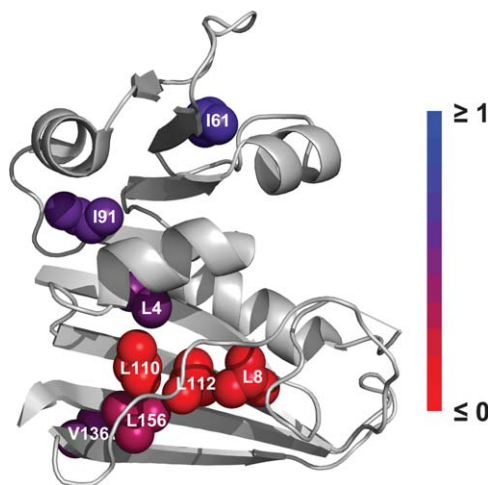


Figure 3. φ_c values of the mutated residues. Residues probed by mutation are shown in spheres in the native structure of DHFR (PDB CODE: 5DFR), colored by φ_c value from 0 (red) to 1 (blue). For clarity, φ_c values less than 0 are shown as 0. The invisible part of the Met20 loop is shown in a straight line. The image was created with PyMOL.

Leu156 has a low φ_c value of 0.24. Finally, Leu8, Leu110, and Leu112 have φ_c values close to 0. When shown on the structure, the distribution of φ_c values clearly suggests the location of unfolding in the partially unfolded form (Fig. 3). Residues with φ_c values close to 0 (Leu8, Leu110, and Leu112) form a cluster on one face of the central β -sheet of the protein. This face of the β -sheet is covered by two extended loops (F-G loop and Met-20 loop). The low φ_c values of Leu8, Leu110, and Leu112 signify that the loops are unfolded and the face of β -sheet is exposed in the partially unfolded form. Moving away from the cluster of Leu8, Leu110, and Leu112, residues show increasing φ_c values. The relatively high φ_c values of Ile61 and Ile91 indicate that the adenosine-binding domain maintains a significant portion of its native contacts in the partially unfolded form. The relatively high φ_c value of Val136 also indicates the back of the central β -sheet does not lose its native contacts completely in the partially unfolded form. The gradual increase in φ_c values from Leu110 to Leu156 and to Val136 also indicates that the loss of the native contacts in the cluster of Leu8, Leu110, and Leu112 is mostly confined to the front face of the central β -sheet (Fig. 3).

Discussion

The structure of the partially unfolded form

The pattern of φ_c values reveals the location of unfolding in the partially unfolded form of DHFR (Fig. 3). Leu8, Leu110, and Leu112 have φ_c values close to 0 and form a continuous cluster, suggesting that the region experiences significant structural disruption in the partially unfolded form. However,

none of the mutated residues has a φ_c value close to 1. The observation that a majority of the mutated residues have fractional φ_c values is distinct from the observation we made with *E. coli* maltose binding protein in which the majority of the mutated residues from an initial survey show φ_c values close to 1.²⁵ We interpret the fractional φ_c values similarly to the way that fractional transition-state φ values have been interpreted.³¹ Fractional φ_c values could mean that some but not all of a residue's native contacts are lost, or an ensemble exists with forms that are close in energy but possess somewhat different structural features. One of the folding intermediates of DHFR, I_{HF} , has been described as a group of four species with similar properties.^{6,32} The fractional φ_c values may have resulted from a different contribution of each species in a group of partially unfolded forms. Although the partially unfolded form we probed may be a group of structurally distinct forms with similar energetic properties, the residues with φ_c values close to 0 allow us to determine the common unfolded region of the partially unfolded forms in the group. Another source of the fractional φ_c values is underestimation of ΔG_{C-N}° due to proteolysis through global unfolding. The free energy for unfolding of the partially unfolded form (ΔG_{U-C}°) is just 1.8 ± 0.3 kcal mol⁻¹. If a mutation destabilizes the partially unfolded form by more than 1.8 kcal mol⁻¹, proteolysis would occur through the globally unfolded form rather than the partially unfolded form even in 0M urea. In this case, ΔG_{C-N}° determined by native-state proteolysis is not the free energy for partial unfolding but the global stability of the mutant. Underestimated ΔG_{C-N}° results in underestimation of $\Delta \Delta G_{U-C}^\circ$ and subsequent underestimation of φ_c . We found one mutant that may have this scenario. The global stability of L4A is only 2.9 kcal mol⁻¹, and its ΔG_{C-N}° determined by proteolysis is 3.0 kcal mol⁻¹. This similarity between the global stability and ΔG_{C-N}° suggests that proteolysis of L4A may occur through global unfolding and the φ_c value of 0.49 may be an underestimated value. Even if L4A has a φ_c value of 1, underestimation of $\Delta \Delta G_{U-C}^\circ$ to 1.8 kcal mol⁻¹ would result in φ_c value of ~ 0.5 .

A significant portion of the native contacts of I61 and I91 seems to be retained in the partially unfolded form (Fig. 3). This result suggests that the adenosine-binding domain mostly remains in the native-like conformation in the partially unfolded form. From m -values for partial and global unfolding, we estimate that partial unfolding exposes only $\sim 30\%$ of the buried surface that is exposed upon global unfolding. The relatively small m -value of partial unfolding is also consistent with intact adenosine-binding domain in the partially unfolded form. Still, the fractional φ_c values of I61 and I91 may arise from structural heterogeneity or partial

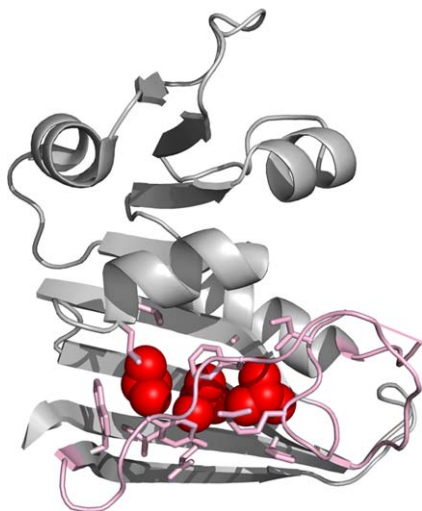


Figure 4. Unfolded region in the partially unfolded form. Leu8, Leu110, and Leu112, which have low φ_c values, are shown in red spheres. Side chains in contact with the three residues are shown in pink sticks. Contacts are defined as residues within 5 Å of any carbon atom lost upon mutation. The F-G loop and the Met-20 loops are also shown in pink. The invisible part of the Met20 loop is shown in a straight line. The image was created with PyMOL.

loss of the native contacts within the adenosine-binding domain in the partially unfolded form.

The native contacts of Leu8, Leu110, and Leu112 seem to be mostly lost in the partially unfolded form (Fig. 3). These residues belong to the discontinuous loop domain, the large domain containing the N- and C-termini, and contact quite a few residues in the central β -sheet, the F-G loop, and the Met-20 loop (Fig. 4). The low φ_c values of Leu8, Leu110, and Leu112, therefore, indicate that the partial unfolding involves a concerted loss of structure in the F-G and Met-20 loops and exposure of the residues on the β -sheet that the two loops cover. The Met-20 loop is known to undergo conformational changes between the “occluded” and “closed” conformations along the catalytic steps of DHFR.^{33,34} Moreover, residues 16–20 of the Met-20 loop are invisible in the structure of the apo form of DHFR determined by X-ray crystallography. This apparent disorder of the Met-20 loop was suggested to result from an interconversion between the occluded and closed conformations.³⁵ The conformations that the two loops experience during the catalytic cycle seem to be distinct from their conformations in the partially unfolded form. The significant free energy (4.9 ± 0.1 kcal mol⁻¹) required for partial unfolding indicates that the partially unfolded form is rarely populated under native conditions (1 out of 4000 molecules), while the conformations accessible during the catalytic cycle are significantly populated under native condition.³⁴ Although the partially unfolded form is energetically

distinct from the conformations accessed during the catalytic cycle, the partial unfolding in the F-G and Met-20 loops seems to be the consequence of the conformational energy landscape of DHFR optimized for its function. To achieve the dynamic nature of the catalytic site, the protein structure may have sacrificed the local stability of the loop regions, which results in transient partial unfolding of the region under native conditions.

The relatively high φ_c values of Leu4 and Val136 could indicate the boundary of the unfolded region in the partially unfolded form (Fig. 3). Again, the fractional φ_c values may arise from structural heterogeneity or a partial loss of the contacts around Leu4 and Val136. Still, the significant difference in the φ_c values of the two residues from those of Leu8, Leu110, and Leu112 implies that the degree of unfolding is much less severe outside of the cluster of Leu8, Leu110, and Leu112. The relatively high φ_c value of Val136 supports that the central β -sheet is largely intact in the partially unfolded form, which is consistent with the observation that many of the amide hydrogens in the central β -sheet are protected quite early in the folding pathway of DHFR.⁷

The partially unfolded form and folding intermediates

The structure and stability of the partially unfolded form are consistent with the folding intermediate I_{HF} . The folding of DHFR has been shown to involve multiple intermediates.^{5,6,10} Matthews and coworkers describe DHFR folding as occurring through an early intermediate (I_{BP}) and a group of four later intermediates (I_{HF}) that lead to four native conformers via parallel folding channels.⁶ The parallel channels may form even earlier,⁷ and the native β -sheet topology begins to form before the formation of I_{BP} .^{11,12}

Pulse labeling H/D exchange and the effect of mutations on the burst-phase CD amplitude suggest that I_{BP} contains native-like secondary structure and limited hydrophobic packing.^{7,9} The burst phase amplitude is affected by mutations at Ile91, Ile94, and Ile155 but is not substantially affected by mutations at Ile2, Ile61, and Leu112, indicating that the later group does not participate in hydrophobic packing in I_{BP} .⁹ The φ_c values of Ile91 (0.5 ± 0.2) and Leu122 (-0.3 ± 0.3) are in line with this result. However, the relatively high φ_c value of Ile61 (0.69 ± 0.09) clearly indicates that I_{BP} is different from the partially unfolded form we observed. I_{BP} buries about 20% of the surface buried in the native form¹⁰ and is 1.5 ± 0.5 kcal mol⁻¹ more stable than globally unfolded DHFR.⁹ Our partially unfolded form has stability similar to I_{BP} (1.8 ± 0.3 kcal mol⁻¹) but buries a significantly larger fraction of surface ($\sim 70\%$) than I_{BP} . Therefore, the partially unfolded form is distinct from I_{BP} .

The later intermediate I_{HF} forms within the range of hundreds of milliseconds and is an essential, on-pathway intermediate.¹⁰ I_{HF} has a higher fluorescence than I_{BP} , native, or unfolded DHFR. The high fluorescence is attributed to the formation of a hydrophobic cluster in the adenosine-binding domain in which Trp47 and Trp74 form their native interaction and become protected from solvent.^{10,13,36} Ile61 forms a hydrophobic cluster with Trp47 and Trp74. The high φ_c value of Ile61 strongly supports that the hydrophobic cluster including Trp47 and Trp74 is intact in the partially unfolded form. Furthermore, I_{HF} was previously found to bury about 65% of the buried surface of the native form,¹⁴ which agrees well with our finding that ~70% of the buried surface of the native form remains buried in the partially unfolded form. The structure and fraction of buried surface suggest that the partially unfolded form and I_{HF} are likely to be the same species.

Frieden and coworkers explain the fluorescence change upon DHFR refolding with a linear model containing three intermediates (I_1 – I_3).^{5,8} I_1 forms with an increase in fluorescence intensity and occurs on the timescale of hundreds of milliseconds.^{5,8} I_1 and I_{HF} are the species with the highest fluorescence in their respective models, and both form on the millisecond timescale. Because the structure of our partially unfolded form is similar to I_{HF} , it is possible that the partially unfolded form corresponds to I_1 . I_2 and I_3 form with a decrease in fluorescence. I_3 has been shown to be only 1–2 kcal mol⁻¹ less stable than the native form.⁸ Because our partially unfolded form is closer in energy to the unfolded form, I_3 is clearly distinct from the partially unfolded form. However, we cannot rule out the possibility that the partially unfolded form corresponds to I_2 .

The partially unfolded form is also similar to an equilibrium intermediate of DHFR, which was discovered in an extensive analysis of the dependence of spectroscopic signal on temperature and urea concentration.³⁷ The equilibrium intermediate has a compact structure and, like I_{HF} , has higher tryptophan fluorescence than native or unfolded DHFR. The m -value for the equilibrium between the native form and the equilibrium intermediate was determined to be 0.7 ± 0.2 kcal mol⁻¹ M⁻¹, which agrees well with the m -value (0.7 ± 0.3 kcal mol⁻¹ M⁻¹) for unfolding to our partially unfolded form. The equilibrium intermediate, the kinetic intermediate I_{HF} , and the partially unfolded form detected by native-state proteolysis all appear to be the same species that is the most accessible non-native form of DHFR.

Comparison with computational studies

Computational approaches to DHFR unfolding have previously suggested the structures of folding intermediates. Sham *et al.* investigated thermal unfold-

ing of DHFR by molecular dynamics simulation.¹⁶ The first step of unfolding was the loss of the contacts between β -strands 1 and 2, which include Ile61 and Trp74. The effect of mutations on the burst-phase CD intensity showed that this region is folded in I_{HF} but not I_{BP} .⁹ Disruption of the interaction between β -strands 1 and 2 captured by the molecular dynamics simulation seems to indicate that the unfolding simulation quickly reaches a conformation that resembles I_{BP} . The unfolding to I_{HF} or a conformation equivalent to our partially unfolded form may be of too small a scale to be detected in this simulation because the analysis only takes into account contacts with >70% occupancy during a control simulation.

Pan *et al.* investigated unfolding of DHFR using the COREX algorithm.¹⁵ This approach estimates the energetic impact of iteratively unfolding small segments of a protein to identify residues that have a high probability of unfolding.²¹ Their study found that residues 60–90 have the highest probability to unfold in DHFR. Interestingly, the region includes Ile61 and Trp74, which are believed to be folded in our partially unfolded form. The high-energy conformation that COREX predicts is clearly different from the partially unfolded form we observed in this study. A possible explanation for this discrepancy is the difference in the choice of the native state. The structure used for the COREX study was the crystallographic structure of DHFR:folate:NADP⁺ ternary complex with the two ligand removed, which is clearly different from the structure of the apo-DHFR under our experimental conditions.

Clementi *et al.* simulated DHFR folding only explicitly accounting for the native topology of the protein.¹⁴ Their simulation predicted that the adenosine binding domain is mostly folded but the discontinuous loop domain is mostly unfolded in I_{HF} . This structure is clearly distinct from the model of the partially unfolded form we describe here in which the central β -sheet of the discontinuous loop domain is mostly intact. The extensive protection of the central β -sheet in the hydrogen/deuterium exchange pulse chase experiment⁷ is rather more in line with our model of I_{HF} than the model proposed from the simulation. It is possible that the intermediate observed in the simulation occurs much earlier in the folding pathway than I_{HF} and is not observable in our native-state proteolysis experiment. Still, their calculation shows that contacts between the Met-20 and F-G loops have some of the lowest probabilities of contacts at the intermediate stage of folding, which is consistent with the concerted unfolding of the loops in the partially unfolded form.

Conclusion

We report here the structure of a partially unfolded form of DHFR that transiently forms under native

conditions. Our comparison with previously observed folding intermediates of DHFR suggests that the identified partially unfolded form is likely the same species as the folding intermediate I_{HF} . The partially unfolded form shows a similar degree of unfolding to that of I_{HF} . Also, the spectroscopic nature of I_{HF} is consistent with the structure of the partially unfolded form. The structure of the partially unfolded form provides much greater details of the structure of I_{HF} ; the adenosine-binding domain is mostly folded, and the hydrophobic core formed by the F-G and Met-20 loops and one face of the central β -sheet is largely unfolded. However, even the folded region seems to have some degree of structural heterogeneity. The final and also the slowest step of DHFR folding is then the docking of the two loops to the central β -sheet as observed in the structure of the native form. The completion of the folding of the two loops may also cause the rest of the protein to achieve more compact packing and reduce the structural heterogeneity. As shown by previous native-state H/D exchange studies,^{22,23} our study clearly demonstrates that investigating partially unfolded forms under native conditions is a valuable means to elucidate structures of folding intermediates. Also, partial unfolding of the F-G and Met-20 loops whose dynamics have functional importance for catalysis by DHFR implies that optimization of the conformational energy landscape of a protein for its function may influence how the protein achieves its native structure along the folding pathway.

Materials and Methods

Preparation of proteins

We expressed DHFR in *E. coli* BL21(DE3)pLysS cells grown to OD₆₀₀ of 0.6 and induced with 0.50M isopropyl- β -D-thiogalactopyranoside (IPTG). Plasmids with mutated DHFR genes were prepared by thermal cycling of the plasmid carrying the DHFR gene sequence and mutagenic oligonucleotide primers according to the protocol for Quikchange Site-Directed Mutagenesis (Agilent Technologies, Santa Clara, CA). All the DHFR proteins we used in this study have C85A/C152S mutations. We purified DHFR by DEAE Sepharose Fast Flow (GE Healthcare Life Sciences; Piscataway, NJ) anion exchange chromatography and Superdex 200 (GE Healthcare Life Sciences) size exclusion chromatography. Thermolysin was prepared by dissolving lyophilized thermolysin (Type X; Sigma-Aldrich; St. Louis, MO) in 2.5M NaCl and 10 mM CaCl₂.³⁸ Concentrations of all proteins were determined by absorbance at 280 nm using extinction coefficients determined according to their amino acid composition.³⁹ We confirmed that less than 10% of purified DHFR was bound to the cofactor NADPH by monitoring change in fluorescence upon NADPH titration.

Native-state proteolysis

For wild-type DHFR, we determined proteolysis kinetics at varying concentrations of thermolysin (0.6–70 $\mu\text{g mL}^{-1}$) in 0–2.5M urea. We initiated proteolysis by adding concentrated thermolysin to DHFR in a buffer to achieve the final condition of 0.10 mg mL⁻¹ DHFR, 20 mM Tris-HCl buffer (pH 8.0), 100 mM NaCl, 10 mM CaCl₂, and 0–2.5M urea. At desired time points, 15- μL aliquots were quenched with 5 μL of 50 mM EDTA. The low protein concentration (0.10 mg mL⁻¹ DHFR) was chosen to minimize the inhibition by cleavage products (Kasper and Park, unpublished result).

We analyzed quenched reaction samples by 15% SDS-PAGE gels. Mark12 Protein Standard (Life Technologies; Grand Island, NY) was used as a molecular weight marker. Gels were stained with SYPRO Red Protein Gel Stain (Life Technologies), and fluorescent images were taken with a Typhoon scanner (GE Healthcare Life Sciences). Intact protein gel bands were quantified from images with ImageJ software. Apparent rates of proteolysis (k_p) were calculated from fitting the change in band intensity over time to a first-order rate equation in OriginPro 8.5.1 (OriginLab; Northampton, MA). We confirmed proteolysis through the EX2-like kinetics by examining the linear dependence of k_p on thermolysin concentration at each urea concentration. Thermolysin maintains its structure under our experimental condition, but the activity decreases as urea concentration is increased.^{24,40} We approximated the k_{int} value at each urea concentration using the k_{cat}/K_m value for proteolysis of a generic peptide substrate for thermolysin, 2-aminobenzoyl-Ala-Gly-Leu-Ala-4-nitrobenzylamide by thermolysin.²⁵ Because proteolysis of DHFR by thermolysin does not generate any cleavage products observable by SDS PAGE, the initial cleavage site is not known, and a generic peptide substrate is used instead of a peptide substrate with the actual sequence of the initial cleavage site. We determined the equilibrium constant for unfolding to the cleavable form (K_{op}) from the slope of the plot of k_p versus thermolysin concentration and the k_{cat}/K_m value for proteolysis of the peptide substrate.²⁴ Then, we calculated the free energy for unfolding to the cleavable form ($\Delta G_{\text{app}}^\circ$) from the K_{op} value at each urea concentration.

We fit the plot of $\Delta G_{\text{app}}^\circ$ versus urea (Fig. 2) to

$$\Delta G_{\text{app}}^\circ = -RT \ln(e^{\Delta G_{\text{op}}^\circ / -RT} + e^{\Delta G_{\text{unf}}^\circ / -RT}), \quad (2)$$

in which the free energies $\Delta G_{\text{op}}^\circ$ and $\Delta G_{\text{unf}}^\circ$ are expressed as functions of urea by the linear-extrapolation method:

$$\Delta G_{\text{op}}^\circ = -m_{\text{op}}[\text{urea}] + \Delta G_{\text{op},\text{H}_2\text{O}}^\circ \quad (3)$$

$$\Delta G_{\text{unf}}^\circ = -m_{\text{unf}}[\text{urea}] + \Delta G_{\text{unf},\text{H}_2\text{O}}^\circ \quad (4)$$

From this fitting, we determined the m -value and free energy of unfolding for both partial

unfolding (m_{op} , $\Delta G_{op,H_2O}^\circ$) and global unfolding (m_{unf} , $\Delta G_{unf,H_2O}^\circ$). The use of the approximate k_{int} value from the generic substrate may introduce a systematic error in K_{op} and ΔG_{op} . However, the m -values are independent of k_{int} and can be determined reliably regardless of the uncertainty in k_{int} .

Equilibrium unfolding

We conducted equilibrium unfolding experiments with 1.0 μM DHFR in 20 mM Tris-HCl buffer (pH 8.0) containing 100 mM NaCl, 10 mM CaCl₂, and varying concentrations of urea. We incubated all samples overnight at 25°C before measurement. We measured ellipticity for each sample at 222 nm by a JASCO J-815 CD spectrophotometer (JASCO, Easton, MD). We determined C_m and m -value by fitting the dependence of ellipticity on urea concentration to a two-state unfolding model using OriginPro 8.5.1 (OriginLab, Northampton, MA).⁴¹ ΔG_{U-N}° was calculated as the product of C_m and m , and $\Delta\Delta G_{U-N}^\circ$ was calculated by subtracting ΔG_{U-N}° of wild type DHFR from that of a mutant.

Determination of φ_c values

We calculated φ_c by

$$\varphi_c = \frac{\Delta\Delta G_{U-C}^\circ}{\Delta\Delta G_{U-N}^\circ} = 1 - \frac{\Delta\Delta G_{C-N}^\circ}{\Delta\Delta G_{U-N}^\circ}, \quad (5)$$

using the relationship of $\Delta\Delta G_{U-C}^\circ = \Delta\Delta G_{U-N}^\circ - \Delta\Delta G_{C-N}^\circ$.²⁵ ΔG_{U-C}° is the free energy for unfolding of the partially unfolded form, ΔG_{U-N}° is the free energy for global unfolding, and ΔG_{C-N}° is the free energy for partial unfolding. We determined $\Delta\Delta G_{C-N}^\circ$ by native-state proteolysis with the assumption that proteolysis occurs with EX2-like kinetics for all DHFR variants under our experimental condition, using

$$\begin{aligned} \Delta\Delta G_{C-N}^\circ &= -RT \ln \frac{K_{op}(\text{mut})}{K_{op}(\text{wt})} = -RT \ln \frac{k_p(\text{mut})/k_{int}}{k_p(\text{wt})/k_{int}} \\ &= -RT \ln \frac{k_p(\text{mut})}{k_p(\text{wt})}, \end{aligned} \quad (6)$$

in which K_{op} is the equilibrium constant for unfolding to the cleavable form, k_p is the apparent rate constant for proteolysis, and k_{int} is the intrinsic rate of proteolysis for the cleavable form.²⁵ The k_p value of each variant was determined at a single concentration of thermolysin (20 μg mL⁻¹).

Acknowledgments

The authors thank Youngil Chang for generation of the cysteine-free version of DHFR and Chen Chen and Nathan Gardner for helpful comments on this manuscript.

References

- Fierke CA, Johnson KA, Benkovic SJ (1987) Construction and evaluation of the kinetic scheme associated with dihydrofolate reductase from *Escherichia coli*. *Biochemistry* 26:4085–4092.
- Bolin JT, Filman DJ, Matthews DA, Hamlin RC, Kraut J (1982) Crystal structures of *Escherichia coli* and *Lactobacillus casei* dihydrofolate reductase refined at 1.7 Å resolution. I. General features and binding of methotrexate. *J Biol Chem* 257:13650–13662.
- Touchette NA, Perry KM, Matthews CR (1986) Folding of dihydrofolate reductase from *Escherichia coli*. *Biochemistry* 25:5445–5452.
- Schnell JR, Dyson HJ, Wright PE (2004) Structure, dynamics, and catalytic function of dihydrofolate reductase. *Ann Rev Biophys Biomol Struct* 33:119–140.
- Frieden C (1990) Refolding of *Escherichia coli* dihydrofolate reductase: sequential formation of substrate binding sites. *Proc Natl Acad Sci USA* 87:4413–4416.
- Jennings PA, Finn BE, Jones BE, Matthews CR (1993) A reexamination of the folding mechanism of dihydrofolate reductase from *Escherichia coli*: verification and refinement of a four-channel model. *Biochemistry* 32:3783–3789.
- Jones BE, Matthews CR (1995) Early intermediates in the folding of dihydrofolate reductase from *Escherichia coli* detected by hydrogen exchange and NMR. *Protein Sci* 4:167–177.
- Clark AC, Frieden C (1999) Native *Escherichia coli* and murine dihydrofolate reductases contain late-folding non-native structures. *J Mol Biol* 285:1765–1776.
- O'Neill JC, Jr, Matthews CR (2000) Localized, stereochemically sensitive hydrophobic packing in an early folding intermediate of dihydrofolate reductase from *Escherichia coli*. *J Mol Biol* 295:737–744.
- Heidary DK, O'Neill JC, Jr, Roy M, Jennings PA (2000) An essential intermediate in the folding of dihydrofolate reductase. *Proc Natl Acad Sci USA* 97:5866–5870.
- Arai M, Kondrashkina E, Kayatekin C, Matthews CR, Iwakura M, Bilsel O (2007) Microsecond hydrophobic collapse in the folding of *Escherichia coli* dihydrofolate reductase, an a/b-type protein. *J Mol Biol* 368:219–229.
- Arai M, Iwakura M, Matthews CR, Bilsel O (2011) Microsecond subdomain folding in dihydrofolate reductase. *J Mol Biol* 410:329–342.
- Kuwajima K, Garvey EP, Finn BE, Matthews CR, Sugai S (1991) Transient intermediates in the folding of dihydrofolate reductase as detected by far-ultraviolet circular dichroism spectroscopy. *Biochemistry* 30:7693–7703.
- Clementi C, Jennings PA, Onuchic JN (2000) How native-state topology affects the folding of dihydrofolate reductase and interleukin-1b. *Proc Natl Acad Sci USA* 97:5871–5876.
- Pan H, Lee JC, Hilser VJ (2000) Binding sites in *Escherichia coli* dihydrofolate reductase communicate by modulating the conformational ensemble. *Proc Natl Acad Sci USA* 97:12020–12025.
- Sham YY, Ma B, Tsai CJ, Nussinov R (2002) Thermal unfolding molecular dynamics simulation of *Escherichia coli* dihydrofolate reductase: thermal stability of protein domains and unfolding pathway. *Proteins* 46:308–320.
- Englander SW (2000) Protein folding intermediates and pathways studied by hydrogen exchange. *Ann Rev Biophys Biomol Struct* 29:213–238.
- Englander SW, Mayne L, Krishna MM (2007) Protein folding and misfolding: mechanism and principles. *Q Rev Biophys* 40:287–326.

19. Dill KA, Chan HS (1997) From Levinthal to pathways to funnels. *Nat Struct Biol* 4:10–19.
20. Street TO, Barrick D (2009) Predicting repeat protein folding kinetics from an experimentally determined folding energy landscape. *Protein Sci* 18:58–68.
21. Hilser VJ, Garcia-Moreno EB, Oas TG, Kapp G, Whitten ST (2006) A statistical thermodynamic model of the protein ensemble. *Chem Rev* 106:1545–1558.
22. Bai Y, Sosnick TR, Mayne L, Englander SW (1995) Protein folding intermediates: native-state hydrogen exchange. *Science* 269:192–197.
23. Chamberlain AK, Handel TM, Marqusee S (1996) Detection of rare partially folded molecules in equilibrium with the native conformation of RNase H. *Nat Struct Biol* 3:782–787.
24. Park C, Marqusee S (2004) Probing the high energy states in proteins by proteolysis. *J Mol Biol* 343:1467–1476.
25. Chang Y, Park C (2009) Mapping transient partial unfolding by protein engineering and native-state proteolysis. *J Mol Biol* 393:543–556.
26. Hubbard SJ, Eisenmenger F, Thornton JM (1994) Modeling studies of the change in conformation required for cleavage of limited proteolytic sites. *Protein Sci* 3:757–768.
27. Iwakura M, Jones BE, Luo J, Matthews CR (1995) A strategy for testing the suitability of cysteine replacements in dihydrofolate reductase from *Escherichia coli*. *J Biochem* 117:480–488.
28. Myers JK, Pace CN, Scholtz JM (1995) Denaturant m values and heat capacity changes: relation to changes in accessible surface areas of protein unfolding. *Protein Sci* 4:2138–2148.
29. Huyghues-Despointes BM, Scholtz JM, Pace CN (1999) Protein conformational stabilities can be determined from hydrogen exchange rates. *Nat Struct Biol* 6:910–912.
30. Fersht A (1999) Structure and mechanism in protein science: a guide to enzyme catalysis and protein folding. New York: W.H. Freeman.
31. Fersht AR, Matouschek A, Serrano L (1992) The folding of an enzyme. I. Theory of protein engineering analysis of stability and pathway of protein folding. *J Mol Biol* 224:771–782.
32. Jones B, Jennings P, Pierre R, Matthews C (1994) Development of nonpolar surfaces in the folding of *Escherichia coli* dihydrofolate reductase detected by 1-anilino-naphthalene-8-sulfonate binding. *Biochemistry* 33:15250–15258.
33. Bystrhoff C, Oatley SJ, Kraut J (1990) Crystal structures of *Escherichia coli* dihydrofolate reductase: the NADP⁺ holoenzyme and the folate-NADP⁺ ternary complex. Substrate binding and a model for the transition state. *Biochemistry* 29:3263–3277.
34. Boehr DD, McElheny D, Dyson HJ, Wright PE (2006) The dynamic energy landscape of dihydrofolate reductase catalysis. *Science* 313:1638–1642.
35. Sawaya MR, Kraut J (1997) Loop and subdomain movements in the mechanism of *Escherichia coli* dihydrofolate reductase: crystallographic evidence. *Biochemistry* 36:586–603.
36. Garvey EP, Swank J, Matthews CR (1989) A hydrophobic cluster forms early in the folding of dihydrofolate reductase. *Protein* 6:259–266.
37. Ionescu RM, Smith VF, O'Neill JC, Jr, Matthews CR (2000) Multistate equilibrium unfolding of *Escherichia coli* dihydrofolate reductase: thermodynamic and spectroscopic description of the native, intermediate, and unfolded ensembles. *Biochemistry* 39:9540–9550.
38. Inouye K, Kuzuya K, Tonomura B (1998) Sodium chloride enhances markedly the thermal stability of thermolysin as well as its catalytic activity. *Biochim Biophys Acta* 1388:209–214.
39. Pace CN, Vajdos F, Fee L, Grimsley G, Gray T (1995) How to measure and predict the molar absorption coefficient of a protein. *Protein Sci* 4:2411–2423.
40. Park C, Marqusee S (2005) Pulse proteolysis: a simple method for quantitative determination of protein stability and ligand binding. *Nat Methods* 2:207–212.
41. Pace CN (1986) Determination and analysis of urea and guanidine hydrochloride denaturation curves. *Methods Enzymol* 131:266–280.