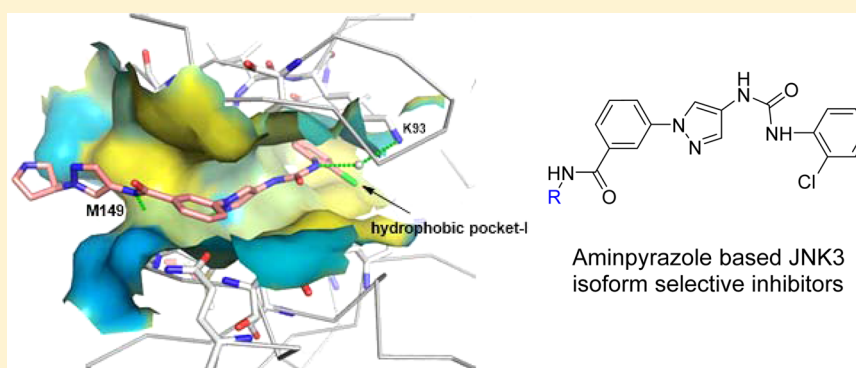


Design and Synthesis of Highly Potent and Isoform Selective JNK3 Inhibitors: SAR Studies on Aminopyrazole Derivatives

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Supporting Information



ABSTRACT: The c-jun N-terminal kinase 3 (JNK3) is expressed primarily in the brain. Numerous reports have shown that inhibition of JNK3 is a promising strategy for treatment of neurodegeneration. The optimization of aminopyrazole-based JNK3 inhibitors with improved potency, isoform selectivity, and pharmacological properties by structure–activity relationship (SAR) studies utilizing biochemical and cell-based assays, and structure-based drug design is reported. These inhibitors had high selectivity over JNK1 and p38 α , minimal cytotoxicity, potent inhibition of 6-OHDA-induced mitochondrial membrane potential dissipation and ROS generation, and good drug metabolism and pharmacokinetic (DMPK) properties for iv dosing. **26n** was profiled against 464 kinases and was found to be highly selective hitting only seven kinases with >80% inhibition at 10 μ M. Moreover, **26n** showed good solubility, good brain penetration, and good DMPK properties. Finally, the crystal structure of **26k** in complex with JNK3 was solved at 1.8 Å to explore the binding mode of aminopyrazole based JNK3 inhibitors.

■ INTRODUCTION

As a member of the mitogen-activated protein kinase (MAPK) family, the c-Jun N-terminal kinases (JNKs) regulate the serine/threonine phosphorylation of several transcription factors¹ when they are activated in response to various stimuli such as oxidative stress, neurotoxins, cytokines, and fatty acids.^{2–8} There are three human JNK isoforms: JNK1, JNK2, and JNK3.^{9–12} JNK1 and JNK2 are ubiquitously expressed in most tissues, while JNK3 is primarily expressed in the brain and, to a lesser extent, in the heart and testes.^{1,10–17} Recent studies have shown that JNK3 plays a central role in the brain to mediate neurodegeneration, such as β -amyloid processing and neuronal apoptosis in Alzheimer's disease,¹⁸ as well as the mediation of neurotoxicity in rodent models of Parkinson's disease.^{3,19–22} The selective expression of JNK3 in the brain, along with findings that JNK3 knockout mice exhibit amelioration of neurodegeneration in animal models of Parkinson and Alzheimer's disease, makes inhibiting this isoform a particularly promising therapeutic target for neurodegenerative diseases.^{18,23,24}

Identifying potent and selective inhibitors of JNK3 may contribute toward neuroprotective therapies with reduced untoward side effect profiles if JNK1 inhibition has such negative effects. To date, however, the development of specific small molecule inhibitors with high isoform selectivity for JNK3, especially against JNK1, is still a relatively untapped area as the three JNK isoforms share more than 90% sequence identity in the ATP pocket. Many published JNK3 inhibitors are also potent for JNK1, JNK2, and some inhibit p38 α as well, because of their high degree of amino acid sequence similarity, which might lead to potential side effect profiles on immune and inflammatory systems.^{25,26} Thus, developing isoform-specific JNK3 inhibitors as therapeutics has gained considerable interest over the past few years despite most reports being centered on pan-JNK inhibitors.^{25,27–45} Our previous effort toward the development of isoform selective JNK3 inhibitors led to the identification of a class of aminopyrazole compounds.

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These aminopyrazole-based JNK3 inhibitors not only had high selectivity against p38 α but also showed noticeable isoform selectivity against JNK1 (compound SR-4326, 18.5-fold, Figure 1).³⁶ Herein, we describe extensive SAR studies in a continuing

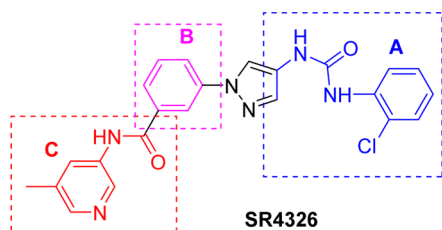


Figure 1. A previously disclosed isoform selective JNK3 inhibitor SR-4326.

effort to develop highly isoform selective, efficacious, and pharmacologically viable JNK3 inhibitors from this novel aminopyrazole scaffold. The focused medicinal chemistry efforts led to the discovery of several highly potent and isoform selective JNK3 inhibitors with an isoform selectivity of >50-fold over JNK1. In addition, these JNK3 inhibitors generally had high selectivity against the closely related protein kinase p38 α , were potent in protecting against ROS generation and mitochondrial dysfunction, and were optimized to have good DMPK properties for topical use and/or in iv dosing.

CHEMISTRY

Several short routes were used to synthesize inhibitors 6–8 which possessed different substituents on the urea moiety (Scheme 1). Ullman coupling of 4-nitro-1H-pyrazole 1 with 3-bromobenzoic acid 2 yielded acid 3. Amide formation to 3 gave intermediate 4, followed by a Pd/C mediated hydrogenation to afford amine 5. This intermediate was directly subjected to

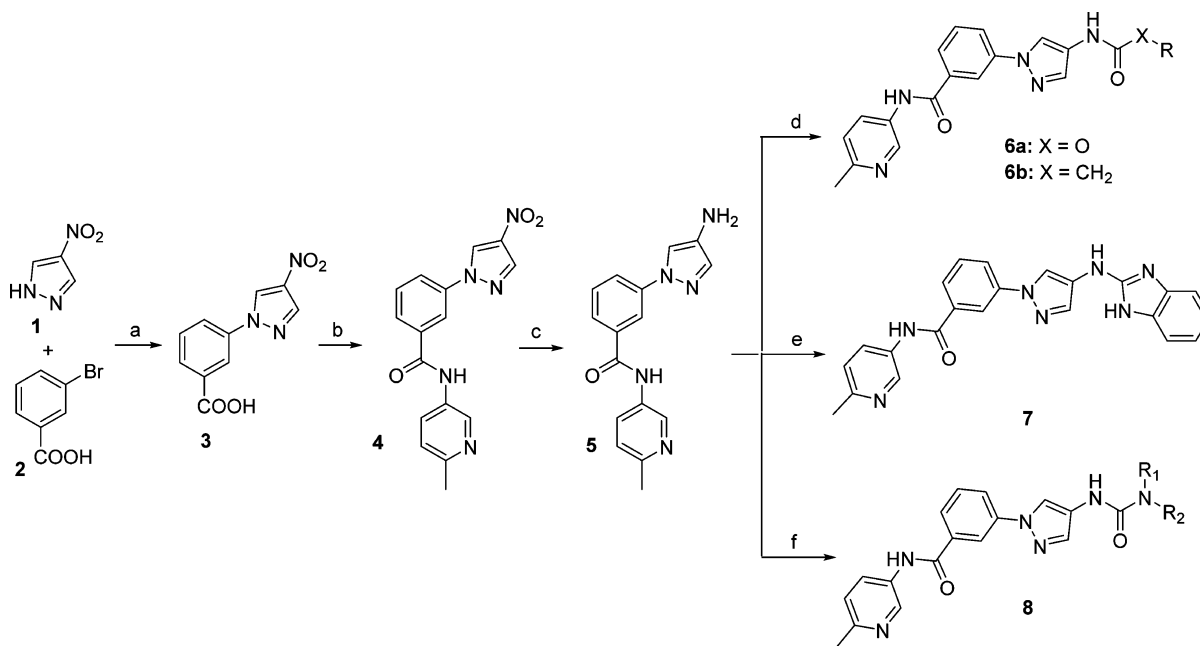
amide formation to furnish inhibitor 6. Inhibitor 7 was formed via a one-step coupling reaction from amine 5 under microwave reaction conditions. Inhibitors 8a–l were synthesized by treating amine 5 with different isocyanates or with triphosgene and a secondary amine, respectively.

The preparation of inhibitor 12 is described in Scheme 2. Treatment of ethyl 1H-pyrazole-4-carboxylate 9 with 3-bromobenzoic acid 2 followed by amide formation gave amide 10. Amide 10 was hydrolyzed under basic conditions to yield acid 11, which was then subjected to an amide formation reaction using EDC as the coupling reagent to furnish the final product 12.

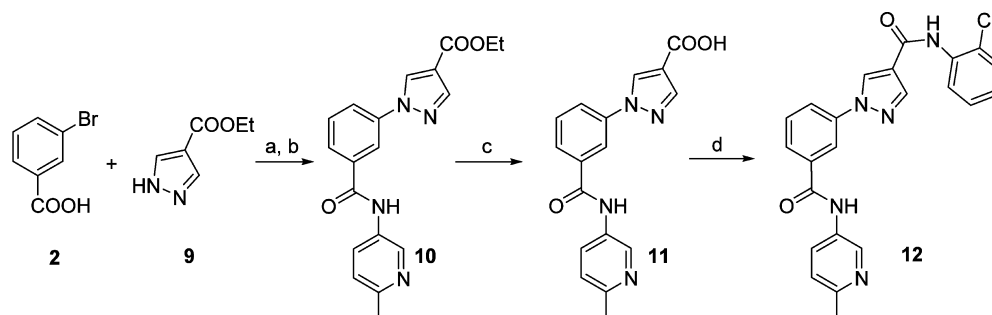
Inhibitors 16a–d, all of which have a fluoro substitution on the middle phenyl ring, were synthesized following procedures shown in Scheme 3. The commercially available 4-nitro-1H-pyrazole 1 was N-arylated with ethyl 3-bromobenzoate 13 using standard Ullman procedures, followed by ester hydrolysis under basic conditions to form acid 14. Amide formation followed by Pd/C mediated reduction of 14 was applied to yield amine 15. Finally, intermediate 15 was subjected to a urea formation to furnish product 16.

The preparation of inhibitors 20–22 is outlined in Scheme 4. Ullman coupling of 4-nitro-1H-pyrazole 1 with 1-bromo-3-iodobenzene (for synthesis of compounds 20) or 3-bromobenzonitrile (for synthesis of compounds 21) or 3-bromobenzoic acid (for synthesis of compounds 22), followed by a Pd/C mediated reduction, afforded the intermediate amine 18. These amines were reacted with 1-chloro-4-isocyanobenzene in DCM to furnish intermediates 19a–c. Target compounds 20 were prepared through a Suzuki coupling reaction between intermediate 19a and a boronic acid pinacol ester catalyzed by Pd(PPh₃)₄. Target compounds 21a and 21b were prepared through treatment of intermediate 19b with hydrazine monohydrate, followed by a cyclization reaction with

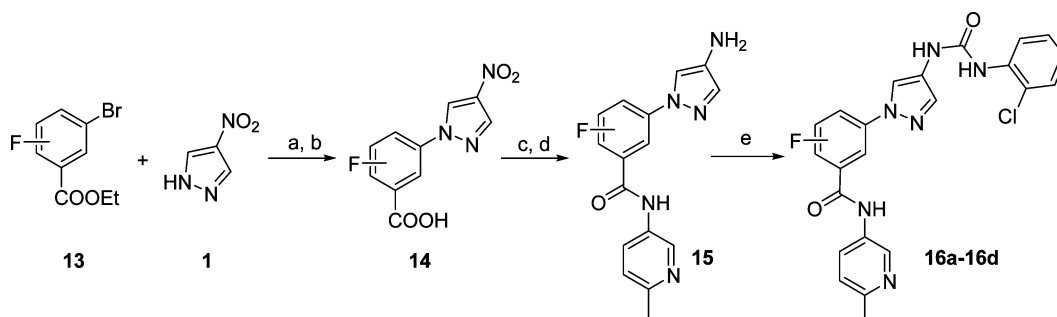
Scheme 1. Synthesis of Inhibitors 6–8^a



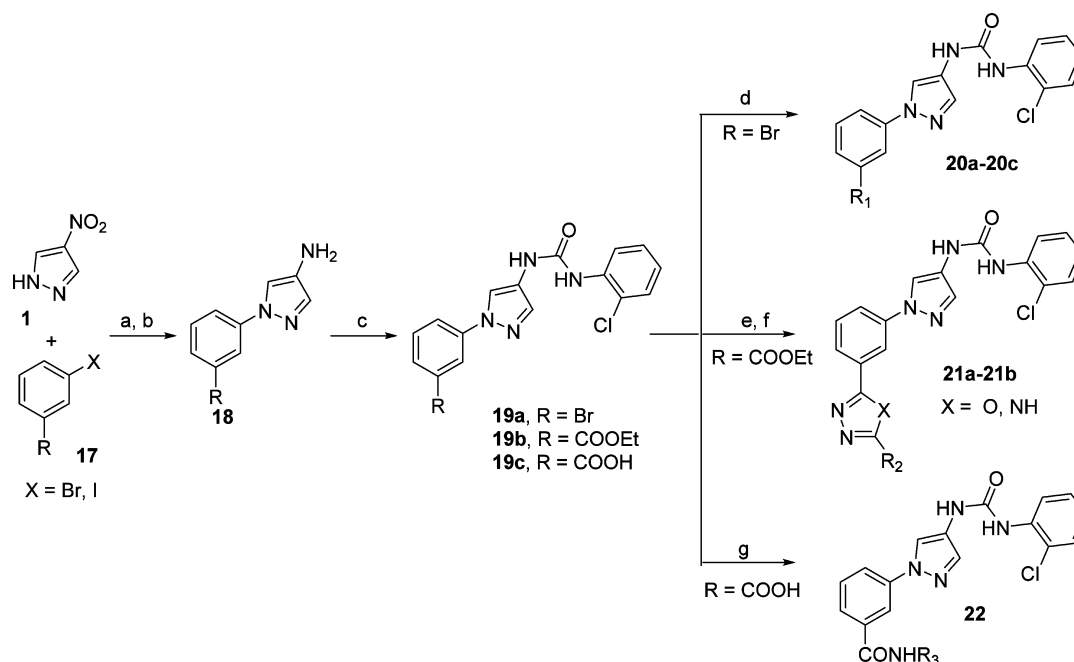
^aReagents and conditions: (a) CuI, *trans*-*N,N*-dimethylcyclohexane-1,2-diamine, Cs₂CO₃, DMF, 90 °C; (b) 6-methylpyridin-3-amine, EDC, HOBT, DIEA, CH₂Cl₂, 25 °C; (c) Pd/C, MeOH, H₂; (d) K₂CO₃, THF or EDC, HOBT, DIEA, CH₂Cl₂, 25 °C; (e) 2-chloro-1H-benzo[d]imidazole, EtOH, HCl, microwave, 140 °C; (f) isocyanate, CH₂Cl₂, 25 °C, or secondary amine; triphosgene, 50 °C, or CDI, THF, microwave, 150 °C.

Scheme 2. Synthesis of Inhibitors 12^a

^aReagents and conditions: (a) CuI, *trans*-*N,N*-dimethylcyclohexane-1,2-diamine, Cs₂CO₃, DMF, 90 °C; (b) 6-methylpyridin-3-amine, EDC, HOBT, DIEA, CH₂Cl₂, 25 °C; (c) LiOH, THF/H₂O (1:1), 100 °C; (d) 2-chloroaniline, EDC, HOBT, DIEA, CH₂Cl₂, 25 °C.

Scheme 3. Synthesis of Inhibitors 16^a

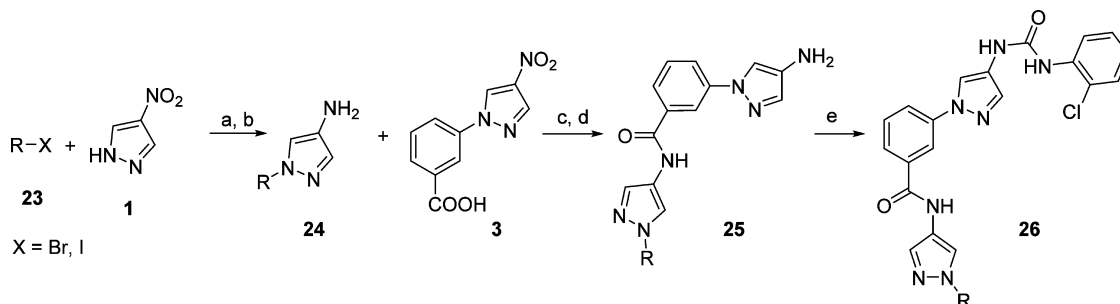
^aReagents and conditions: (a) CuI, *trans*-*N,N*-dimethylcyclohexane-1,2-diamine, Cs₂CO₃, DMF, 90 °C; (b) LiOH, THF/H₂O (1:1), 100 °C; (c) 6-methylpyridin-3-amine, EDC, HOBT, DIEA, CH₂Cl₂, 25 °C; (d) Pd/C, MeOH, H₂; (e) 1-chloro-2-isocyanatobenzene, CH₂Cl₂, 25 °C.

Scheme 4. Synthesis of Inhibitors 20–22^a

^aReagents and conditions: (a) CuI, *trans*-*N,N*-dimethylcyclohexane-1,2-diamine, Cs₂CO₃, DMF, 100 °C; (b) Pd/C, MeOH, H₂; (c) 1-chloro-2-isocyanatobenzene, CH₂Cl₂, 25 °C; (d) R₁B(OH)₂, Ph(PPh₃)₄, K₂CO₃, dioxane/H₂O (3:1), 95 °C; (e) N₂H₄·xH₂O, EtOH, reflux; (f) acetimidamide, EtOH, NaOMe or acetic acid, propylphosphonic anhydride, TEA, DMF, 110 °C; (g) R₃NH₂, EDC, HOBT, DIEA, CH₂Cl₂, 25 °C.

acetimidamide or acetic acid. Target compounds **22** were prepared through amide formation in the presence of HATU in

DMF. All final inhibitors were purified by reverse-phase preparative HPLC.

Scheme 5. Synthesis of Inhibitors 26^a

^aReagents and conditions: (a) K_2CO_3 , DMF, 40 °C; (b) Pd/C, MeOH, H_2 ; (c) EDC, HOBT, DIEA, CH_2Cl_2 , 25 °C; (d) Pd/C, MeOH, H_2 ; (e) 1-chloro-2-isocyanatobenzene, CH_2Cl_2 , 25 °C.

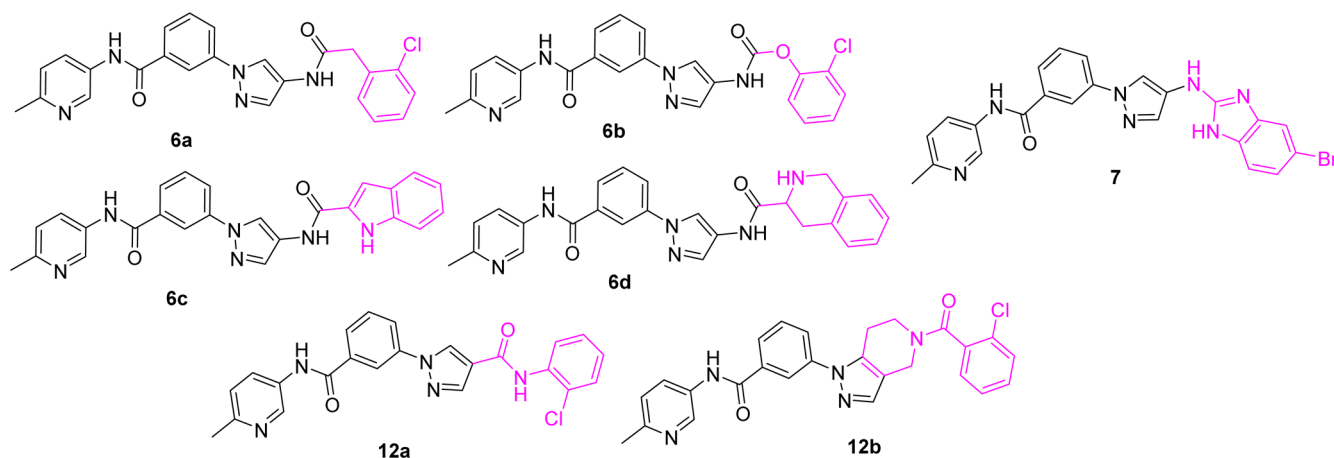


Figure 2. Compounds with low JNK3 inhibition activity with $IC_{50} > 10 \mu M$.

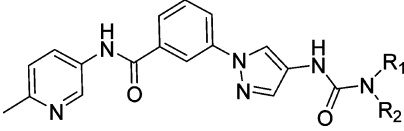
Applications of N-substituted pyrazole amines led to inhibitors 26. As shown in Scheme 5, the N1 alkylation of 3-nitropyrazole with a primary iodo- or bromoalkane using potassium carbonate as the base was first used to furnish N-substituted pyrazoles, followed by reduction of the nitro group to give pyrazole amines 24. Amide formation between 24 and acid 3, followed by a Pd/C mediated reduction furnished amine 25. A one-pot urea formation of 25 with 1-chloro-4-isocyanobenzene in DCM afforded the target inhibitor 26 which was purified by reverse-phase preparative HPLC.




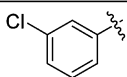
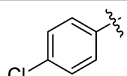

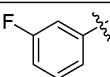
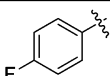
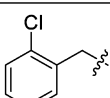


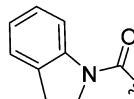
RESULTS AND DISCUSSION

In the crystal structures published in our previous work, the aniline–urea moiety (Figure 1) binds to the hydrophobic selectivity pocket (hydrophobic pocket I) and is responsible for the isoform selectivity of JNK3 over JNK1.³⁶ We therefore started our SAR investigations by replacing the aniline–urea moiety in region A (Figure 1) with selected bioisosters. As shown in Figure 2, very few replacements to the urea moiety were tolerated at region A, as all compounds had JNK3 IC_{50} values of $>10 \mu M$. For example, replacement of the aniline NH group with a carbon (6a) or oxygen (6b) or a direct heterocyclic ring connection (6c,d) all resulted in complete loss of JNK3 inhibition. Similarly, structural changes to the other NH group of urea were not tolerated either. For example, amide compound 12a and pyrazolo[4,3-*c*]pyridine compound 12b were completely inactive. Some other replacements of the urea moiety were also investigated (such as compound 7; for more data see Supporting Information), but no improvement

was observed. The observed results for modifications to the urea moiety suggested that the urea core was essential for achieving a high JNK3 affinity. Therefore, this urea moiety will be retained in all further optimizations.

We next investigated a series of replacements and substitutions on the phenyl group of the urea moiety. As shown in Table 1, replacing the aromatic group with an alkyl group was not tolerated, as indicated by the significant loss of JNK inhibition activity for 8a. Compound 8b, which had no substitutions on the urea phenyl ring, showed a JNK3 inhibition activity similar to the lead compound SR-4326 but with almost no isoform selectivity. In the case of chloro substitutions, both the meta- and the para-Cl substitutions (8d and 8e) led to a decrease in JNK affinity, which could be due to space limitation inside the binding pocket. However, when a Cl substituent was introduced at the ortho-position of the phenyl ring (8c), where there should be enough space to accommodate a Cl group based on the crystal structures,³⁶ higher JNK3 potency was obtained probably because of the extra hydrophobic interactions from the Cl group. More interestingly, some isoform selectivity (JNK3 over JNK1) was achieved as well (4.4-fold). On the other hand, fluoro substitutions on this phenyl ring exhibited a significantly different SAR profile. Compared to the lead compound SR-4326 and compound 8c, a mono-F-substitution at all three positions resulted in reduction of JNK3 inhibition activity, although these substitutions could improve the isoform selectivity (compounds 8f, 8g, 8h). While the meta-F compound 8g and the para-F compound 8h showed a slight

Table 1. Biochemical IC₅₀ Values for JNK3 and JNK1 for SAR Studies of the Urea Moiety^a


compd	R ₁	R ₂	JNK3 IC ₅₀ (nM)	JNK1 IC ₅₀ (nM)	JNK1/JNK3
8a	H		N/I ^b	N/I ^b	-
8b	H		115	138	-
8c	H		38	170	4.4
8d	H		596	N/I ^b	-
8e	H		1829	N/I ^b	-
8f	H		313	1607	5.1
8g	H		139	653	3.3
8h	H		162	400	2.5
8i	H		N/I ^b	N/I ^b	-
8j	Me		3063	N/I ^b	-
8k	HO-CH ₂ -CH ₂ -		N/I ^b	N/I ^b	-
8l			N/I ^b	N/I ^b	-

^aIC₅₀ values are the mean of two or more experiments (with triplicate replicates for each experiment) with errors within 80% of the mean. ^bNo inhibition up to 10 μM.

decrease of JNK3 potency, a substitution at the ortho-position resulted in around 3-fold loss of potency as compared to the unsubstituted analog **8b** (Table 1).

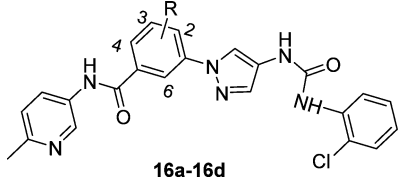
The unusual SAR observed for F-substituted compounds in Table 1 might be due to the F-bonding/interactions.^{46,47} Different types of interactions between fluorine atoms with protein functional groups have been reported (not found in other halogens), and in some protein environments, compounds with a fluorine substitution behave very differently from other halogen-substituted compounds.⁴⁷ Such fluorine inter-

actions include F–H hydrogen bonds, π-interactions with aromatic or guanidine groups, multipolar interactions, and/or F–nitrogen interactions (halogen bonding).⁴⁷ For the aminopyrazole based JNK inhibitors, these special F-interactions could result in a change of inhibitor conformations in the pocket or a twist of the orientation of the phenyl ring inside hydrophobic pocket I. These changes in conformation or orientation could produce unfavorable binding of the inhibitor in the ATP pocket, and therefore cause a decrease in inhibition of JNK activity for compounds **8f–h**.

Significant loss of potency also occurred with replacement of the *o*-Cl-phenyl moiety by an *o*-Cl-benzyl group (**8i**, Table 1). This result indicated that the hydrophobic selectivity pocket might not have enough room to tolerate a larger group or that the phenyl ring of the benzyl group could not assume a conformation/orientation that could give maximal hydrophobic interactions. The effect of substitutions on the phenyl urea NH was also studied. Unfortunately, any substitution to this NH group greatly reduced the JNK3 potency (Table 1). For example, a simple methyl substitution resulted in compound **8j** which exhibited a JNK3 IC_{50} of 3063 nM. Larger substituents yielded even lower JNK3 inhibitory activity (compounds **8k** and **8l**, $IC_{50} > 10 \mu M$). As is demonstrated in the crystal structure (see Figure 4), this urea NH is involved in H-bonding interactions with protein residues, and its alkylation is predicted to reduce the JNK3 inhibitory activity.

The feasibility of applying some minor structural modifications to the central phenyl group attached to the pyrazole N1 (region B, Figure 1) was explored next. Our preliminary SAR showed that a methyl substitution was tolerated here.³⁶ We envisioned that introduction of a fluoro substitution on the benzene ring might introduce additional interactions that could perturb the JNK3 activity and selectivity, as well as improve DMPK properties. Therefore, compounds with a simple fluoro substitution at different positions of this benzene ring were investigated (Table 2). Although improvement in isoform

Table 2. Biochemical IC_{50} Values for JNK3 and JNK1 for SAR Studies for the Middle Phenyl Moiety^a



compd	R	JNK3 IC_{50} (nM)	JNK1 IC_{50} (nM)	JNK1/JNK3
16a	2-F	80	2369	29.6
16b	3-F	4588	N/I ^b	–
16c	4-F	71	180	2.5
16d	6-F	230	3691	16.1

^a IC_{50} values are the mean of two or more experiments (with triplicate replicates for each experiment) with errors within 80% of the mean.
^bNo inhibition up to $10 \mu M$.

selectivity was achieved (compared to SR-4326 and **8c**) for the 2-fluoro substitution (**16a**) and the 6-fluoro substitution (**16d**), loss of JNK3 inhibition potency was observed for all fluoro-substituted compounds **16a–d**, especially for the 3-fluoro-derivative **16b**, which showed an IC_{50} value against JNK3 in the micromolar range (4588 nM). The reduction in JNK inhibitory potency could be due to the F-bondings/interactions^{46,47} previously mentioned and as observed in compounds **8f–h** (Table 1). In addition, the 2-F analog **16a** exhibited potent inhibition of cytochrome P450 isoforms CYP2C9 and CYP2D6 (see Table 5). Because of these detrimental features, F-substitutions on this benzene ring will not be applied in future optimizations.

The crystal structure published in our previous work indicated that these compounds were type I kinase inhibitors and showed that the amide carbonyl and the aromatic ring (in the amide moiety) did not have any significant interactions with the protein (only an H-bond interaction was observed between

the amide NH and the protein backbone).³⁶ Thus, structural modifications in this part will likely not affect the binding mode significantly. To test this observation, a series of replacements for the amide moiety in region C (Figure 1) with some amide bioisosters were investigated. As shown in Figure 3, the

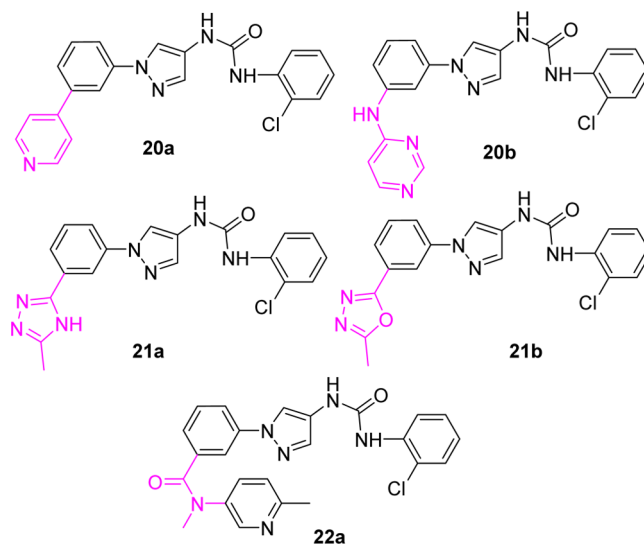
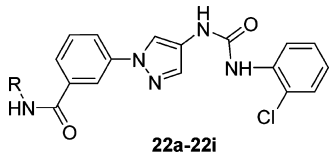












Figure 3. Compounds with low JNK3 inhibition activity with $IC_{50} > 10 \mu M$.

pyridine and pyrimidine based structures (**20a–b**), which have heteroatoms on the aromatic ring to form H-bonds, showed no JNK inhibitory activity. The triazole (**21a**) and the oxodiazole (**21b**) based amide isosters, which mimic the binding of the anilinoamide in region C, also gave no inhibition for JNK3 and JNK1. Moreover, alkylation of the amide NH (**22a**) led to complete loss of JNK3 inhibitory activity. These results indicated that the amide structure was critical for efficient binding and the amide moiety –CONH– was indispensable for this scaffold to achieve a high JNK3 affinity.

SAR studies of the amide moiety are a very important part of our efforts to improve both JNK3 inhibitory potency and isoform selectivity (against JNK1) for this aminopyrazole-based scaffold. The effect of substitution patterns on the pyridine ring for lead compound **8c** was first explored. As shown in Table 3, moving the methyl group from 4'-position to 2'-position reduced JNK3 inhibitory potency by 3.7-fold but it reduced the JNK1 potency as much as 15.8-fold providing a benefit for the overall JNK3/JNK1 isoform selectivity (**22b** vs **8c**). The dimethylpyridine compound **22c** showed a JNK3 potency similar to that of the monosubstituted analogue **22b** but a reduced isoform selectivity. Nevertheless, lead compound **8c** still had the best JNK3 potency (38 nM). While the 2'-pyridine analogs **22d** and **22e** exhibited a significant loss of JNK3 inhibition potency, the 4'-pyridine compound, **22f**, presented just a slightly lower JNK3 potency (98 nM vs 38 nM) but a better isoform selectivity (28-fold vs 4.4-fold, **22f** vs **8c**). The dimethyl-4'-pyridine compound **22g** exhibited an even higher isoform selectivity (35-fold), although with a sacrifice in JNK3 potency. Interestingly, the pyrimidine amide **22h** showed a similar JNK3 potency but had isoform selectivity slightly better than compound **8c**. It is also important to note that the benzyl type amide **22i** gave a weaker JNK3 inhibition but improved isoform selectivity as compared to the corresponding aniline type amide **8c** and SR-4326. In summary, for pyridine and/or

Table 3. Biochemical IC₅₀ Values for JNK3 and JNK1 for SAR Studies for the Amide Moiety^a


compd	R	JNK3 IC ₅₀ (nM)	JNK1 IC ₅₀ (nM)	JNK1/JNK3
SR-4326		117	2169	18.5
8c		38	170	4.4
22b		141	2689	19
22c		141	1028	7
22d		N/I ^b	N/I ^b	-
22e		5184	N/I ^b	-
22f		98	2740	28
22g		130	4544	35
22h		62	537	9
22i		311	7791	25

^aIC₅₀ values are the mean of two or more experiments (with triplicate replicates for each experiment) with errors within 80% of the mean.

^bNo inhibition up to 10 μM.

pyrimidine based amides, 3'-N derivatives gave better JNK3 potency but not satisfactory isoform selectivity (8c and 22h). On the other hand, 4'-N analogs exhibited good isoform selectivity but lower JNK3 inhibition activity (22f and 22g). Nevertheless, no compounds from this class could present both good JNK3 inhibition potency (IC₅₀ < 50 nM) and excellent isoform selectivity (>50-fold).

Although the benzamide moiety (region C, Figure 1) extends away from hydrophobic pocket II and is exposed to solvents, results in Table 3 indicated that structures in region C can still significantly affect both JNK3 inhibitory activity and isoform selectivity. Considering that pyridine/pyrimidine amide compounds might also have cytotoxicity and poor microsomal stabilities (see Table 5 and Table 6), we expanded SAR studies for this part to other amide structures. Thus, amides based on five-membered heterocyclic ring systems were investigated. As shown in Table 4, the *N*-methylpyrazole compounds 26a–d showed a similar potency and isoform selectivity compared to lead compound SR-4326. Application of an *N*-methyl-triazole (26e) resulted in a 5-fold reduction in JNK3 inhibitory activity. Compounds having larger substituents on the NH of pyrazole

groups gave better properties. For example, an *N*-isopropyl substitution on the pyrazole ring yielded a JNK3 inhibitor (26f) having better isoform selectivity compared to the *N*-methyl analog (26a).

Remarkably, those analogs with a substitution containing a dimethylamino, diethylamino, piperidine, or pyrrolidine group (26g–k) exhibited high JNK3 potency (IC₅₀ of 21–34 nM) and good to excellent isoform selectivity (22- to 60-fold over JNK1). One compound, 26k, which had a substituent containing a pyrrolidine moiety, stood out and showed the highest JNK3 potency (IC₅₀ < 1 nM) and isoform selectivity (>500-fold over JNK1 and >210-fold over JNK2; see Table 5). Careful inspection of the crystal structure of 26k indicated the potential for a hydrogen bond between the carbonyl of N89 in JNK3 with the pyrrolidinyl NH of 26k with an H-bond distance of 2.9 Å (Figure 4). This potential for tight interaction with optimal hydrogen bond distance and optimal H-bond angle was likely the reason for the potent inhibition of 26k (the energy minimized state of a computational model showed the *N*-pyrrolidinyl of 26k could rotate ~90° to make a H-bond with N89). In contrast, 26j, which contained the six-membered piperidinyl ring and was not optimized for H-bond distance and angle, had weaker inhibition to JNK3 than 26k. This observation is likely true for all of the other analogs presented in Table 4. Excellent JNK3 isoform selective inhibitors were also obtained from amides containing no aromatic groups (alkylamides, 26l–q). As shown in Table 4, all the inhibitors prepared gave excellent isoform selectivity and fair to excellent JNK3 inhibitory activity. The much better JNK3 potency and the higher isoform selectivity (JNK3 over JNK1) for these compounds containing a side chain amino group could be due to extra H-bond interactions from the amino group, and a combination of these interactions and the hydrophobic interaction inside hydrophobic pocket I might be more favorable for JNK3 than for JNK1. These interactions are indicated in the crystal structure of 26k with JNK3 (Figure 4).

To better understand the observed SAR for these JNK3 isoform selective inhibitors, the crystal structure for 26k with JNK3 was solved at 1.8 Å (Figure 4). This crystal structure showed that 26k (the *R*-enantiomer) interacted with JNK3 through H-bonding with the hinge residue M149 with a bond distance of 2.7–3.2 Å. The hydrogen bond between the benzamide NH of 26k and the main-chain atoms of M149 in JNK3 would explain why substitution on the amide NH (22a) resulted in no JNK3 affinity (shown as yellow dotted lines). An extra H-bond also occurred between the urea NH (on the aniline side) and the side chain of residue K93 bridged by a water molecule. The existence of this interaction explains why *N*-alkylation of the urea NH led to loss of JNK3 inhibitory activity (compounds 8j, 8k, and 8l, Table 1). As mentioned, the *N*-pyrrolidinylpyrazole group made optimal H-bond interactions based on length and angle with residue N89, and the rest of the ring structure of 26k made tight hydrophobic interactions with residues in ATP binding pocket. The center phenyl ring made hydrophobic interactions with side chain of I70 and backbone of A151. The central pyrazole ring also made hydrophobic interactions with the side chains of A91 and V196.

The most significant hydrophobic interactions occurred at the terminal *o*-Cl-phenyl ring (attached to the urea moiety) and were tightly sandwiched between the side chains of K93 and M146. Nearby hydrophobic residues V78, M115, L126, L144, and L206 construct a hydrophobic pocket for the *o*-Cl-phenyl ring. These structural observations are consistent with the

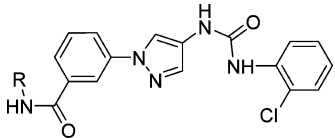
Table 4. Biochemical IC₅₀ Values for JNK3 and JNK1 for SAR Studies for the Amide Moiety^a











compd	R	JNK3 IC ₅₀ (nM)	JNK1 IC ₅₀ (nM)	JNK1/JNK3
26a		133	2057	16
26b		203	2625	28
26c		162	3162	20
26d		116	1284	11
26e		594	5385	9
26f		114	5046	44
26g		34	758	22
26h		21	1249	60
26i		23	1226	54
26j		23	679	30
26k		< 1	528	>500
26l		137	4495	33
26m		100	4728	47
26n		206	N/I ^b	>50
26o		169	N/I ^b	>60
26p		34	1873	55
26q		29	2756	94

^aIC₅₀ values are the mean of two or more experiments (with triplicate replicates for each experiment) with errors within 80% of the mean. ^bNo inhibition at 10 μM.

observed SAR results indicating that there was almost no JNK3 affinity if the urea moiety was replaced or the urea NH was

alkylated. This structure also explains why there was a reduction of JNK3 inhibitory activity when the phenyl ring was

Table 5. Inhibitor Selectivity and in Vitro DMPK Data^a


compd	R	JNK2 IC ₅₀ (nM)	p38α IC ₅₀ (nM)	Microsomal stability T _{1/2} (min) ^b		Cyp-450 % inh. at 10 μM 1A2/2C9/2D6/3A4	Solubility in 1% DMSO/buffer (μM)	
				Human	Mouse		pH 3.5	pH 7.4
22f		2836	N/I ^c	32	26	17 / 60 / 25 / 44	23	0.2
22g		561	N/I ^c	28	6	10 / 19 / -25 / 26	23	0.2
22i		428	N/I ^c	11	17	17 / 48 / 83 / 49	29	0.9
16a ^d		148	N/I ^c	50	86	19 / 76 / 93 / 10	9	0.5
26f		158	3623	39	37	15 / 49 / 54 / 16	2	1.4
26g		66	N/I ^c	35	4	-5 / 12 / 23 / -9	71	72
26j		25	N/I ^c	408	54	3 / 22 / 35 / -16	63	53
26k		210	N/I ^c	113	33	38 / 66 / 43 / 70	n.d ^b	n.d ^b
26l		N/I ^c	N/I ^c	40	6	-12 / -1 / 17 / 15	37	45
26n		311	N/I ^c	305	72	-8 / 1 / 8 / 4	151	79

^aIC₅₀ values are the mean of two or more experiments (with triplicate replicates for each experiment) with errors within 80% of the mean. ^bNot determined. ^cNo inhibition up to 10 μM. ^dWith fluoro-substitution on the middle phenyl ring at Region B.

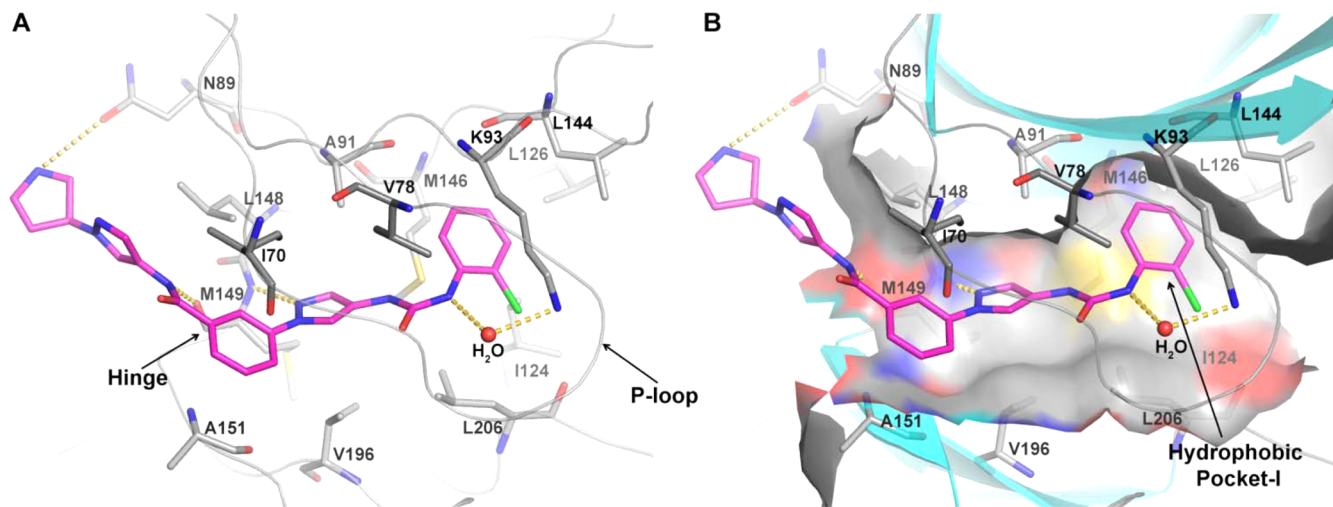


Figure 4. X-ray crystal structure of 26k in JNK3.

substituted with a Cl on its meta- or para-position (8d and 8e), since there was not enough space inside this hydrophobic pocket. The isoform selectivity of JNK3 over JNK1 for aminopyrazole-based JNK3 inhibitors was also mainly due to interactions in this pocket, which has been documented in our previous publications.³⁶

To evaluate the selectivity of these aminopyrazole based JNK3 inhibitors, several selected compounds were subjected to additional counter screening against JNK2 and p38. As shown in Table 5, almost all aminopyrazole compounds had no inhibition on the most closely related kinase p38 at a concentration of 10 μM with the exception of the *N*-alkyl substituted pyrazole amide 26f (3623 nM for p38). Moreover, these compounds had inhibitory activity for JNK2 with potency

similar to that for JNK3 with only a few exceptions (22f, 22g, 22i, 26k, and 26l). To further evaluate the general kinase selectivity of these aminopyrazole based JNK3 inhibitors, compound 26n was subjected to a profiling study at a concentration of 10 μM in the full panel KINOMEScan (Ambit, San Diego, CA),^{48–50} a high throughput method for screening kinase inhibitors against a panel of 464 kinases. The TREEspot maps in Figure 5 revealed that 26n possessed an extremely high selectivity with significant inhibitions (>80% at 10 μM) observed only for JNK1/2/3, Clk2, haspin, Mek3, and Ysk4 (see Supporting Information for detailed profiling data).

Inhibition of four selected cytochrome P450 isoforms (1A2, 2C9, 2D6, and 3A4) was also tested at 10 μM, and results demonstrated that most of these aminopyrazole-based JNK3

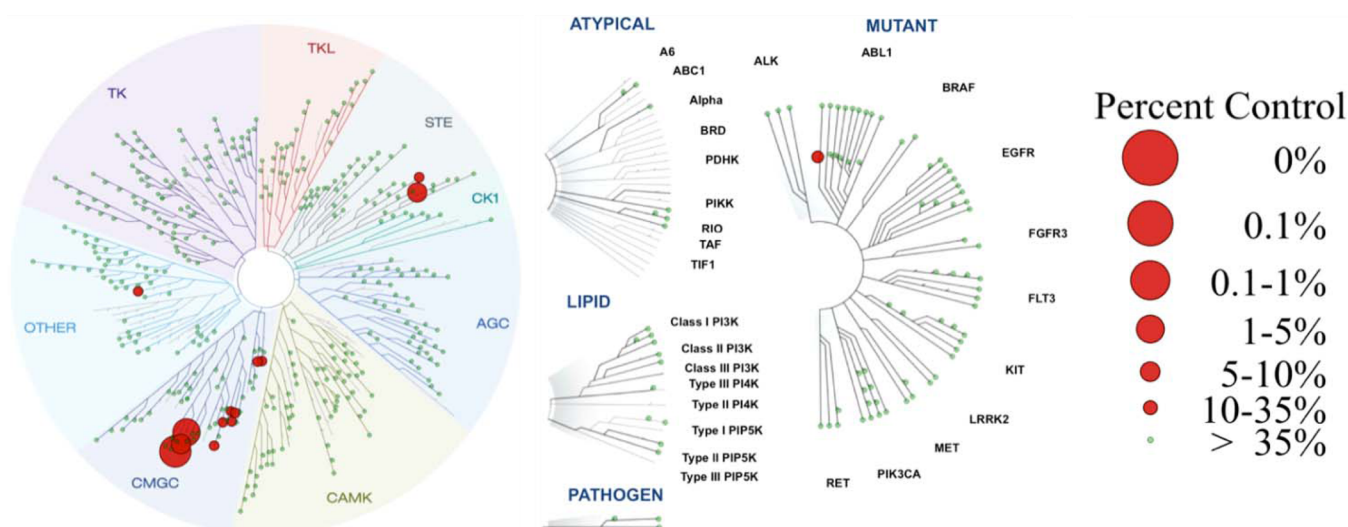


Figure 5. TREEspot interaction maps for **26n** (also coded as SR-11935) in the panel profiling of 464 kinases.

Table 6. Cytotoxicity and Cell Based Potency Data for Selected Compounds

cmpd	cytotoxicity in SHSY5Y cells after 48 h		in-cell Western, ^a SHSY5Y IC ₅₀ (nM)	inhibition of 6-OHDA induced cell death, SHSY5Y IC ₅₀ (nM)	inhibition of 6-OHDA induced mitochondrial membrane depolarization, SHSY5Y IC ₅₀ (nM)
	% cell viability at 10 μM	% cell viability at 30 μM			
22f	7	5	866	2976	40
22g	24	5	2331	nd ^b	nd ^b
22i	80	56	905	13	17
26f	84	16	3250	nd ^b	130
26j	91	98	1436	568	25
26g	105	97	N/I ^c	N/I ^c	N/I ^c
26k	112	97	N/I ^c	N/I ^c	31
26n	92	98	1895	281	4

^aIC₅₀ values are the mean of two or more experiments (with triplicate replicates for each experiment) with errors within 80% of the mean. ^bNot determined. ^cNo inhibition up to 10 μM.

inhibitors had generally low inhibition against all four enzymes. Generally, the pyrazole-amide compounds **26f**, **26j**, **26g**, **26k** presented lower inhibition for P450 isoforms compared to those pyridineamide based compounds (**22e**, **22f**, **22i**, **16a**). The alkylamide based compounds gave even better P450 inhibition profiles. Compounds **26l** and **26n** exhibited almost no inhibition for all four selected isoforms even when tested at 10 μM. It is noteworthy that the F-substituted compound **16a** exhibited potent inhibitions for isoforms CYP2C9 and CYP2D6.

The microsomal stability of these compounds in both human and mouse liver microsomes were evaluated. Briefly, the microsomal stabilities of most inhibitors based on the pyridine amide (**16a**, **22f**, **22g**, **22i**) or the *N*-alkyl substituted pyrazoleamide (**26f**) were low with the exception of the F-substituted compound **16a** (half-life was 50 min in human and 86 min in mouse). The presence of a secondary amine pyrrolidine ring or piperidine ring at the amide part was clearly a favorable factor for metabolic stability. For example, inhibitors **26g** and **26l**, which had no secondary amine substitutions, exhibited a stability of $t_{1/2}$ = 35–40 min in human microsomes and only 4–6 min in mouse microsomes, while the corresponding inhibitors containing a secondary amine moiety (**26j**, **26k**, **26n**) had a much greater microsomal stability ($t_{1/2}$ = 113–408 min in human and $t_{1/2}$ = 33–72 min in mouse). The aqueous solubility was also tested for these compounds at both

pH 3.5 and pH 7.4. As shown in Table 5, while the solubility for compounds containing the pyridineamide (**22f**, **22g**, **22i**) and *N*-alkyl substituted pyrazoleamide **26f** were low, inhibitors with substituents containing a secondary or tertiary amine exhibited excellent solubility at both pH 3.5 and pH 7.4. Indeed, **26n** had excellent solubility at both pH 3.5 and pH 7.4 (Table 5).

The cytotoxicity in SHSY5Y cells was evaluated for some selected lead inhibitors. As shown in Table 6, pyridineamide based inhibitors **22f** and **22g** showed significant cytotoxicity at both 10 and 30 μM, while the benzylamide based inhibitor **22i** and the simple alkyl substituted pyrazoleamide **26f** exhibited significant toxicity only at 30 μM. Remarkably, other pyrazoleamide and alkylamide based compounds were all nontoxic even at a high concentration of 30 μM in SHSY5Y cells, indicating that they were relatively safe JNK3 inhibitors. This was especially true for **26n** which had >90% cell viability at both 10 and 30 μM.

Some compounds were also evaluated in vitro for their ability to inhibit the phosphorylation of c-Jun in neuronal SHSY5Y cells. In-cell Western blot analysis is a direct measure of JNK activity and was employed to investigate the ability of these JNK3 isoform selective inhibitors to inhibit c-Jun phosphorylation.^{20,36,41,51,52} As shown in Table 6, most of the compounds had an IC₅₀ of around 1 μM or higher in this assay, indicating that they were reasonable JNK inhibitors. It is

noteworthy that there was a large right shift between JNK3 biochemical potency and cell-based potency for these isoform selective JNK3 inhibitors (Table 6). The lack of JNK1 inhibition and the cell types might be part of the reasons.^{27,53} Indeed, because c-Jun phosphorylation is dependent on all three isoforms and SHSYSY cells contain all three isoforms, the uninhibited JNK1 activity in these cells was the likely reason for the high cell-based IC₅₀ values. In contrast, these same compounds showed good cell potency in other types of cell assays (Table 6) which were not as dependent on JNK1 nuclear activity for substrate phosphorylation but rather more centered on the role of JNK3 in mitochondrial dysfunction.^{3,4,51,52}

These compounds showed protection from 6-OHDA-induced mitochondrial membrane depolarization and cell death in SHSYSY cells, both of which are regulated and promoted by JNK activation.⁵⁴ Improved cell potency was obtained in these two assays as compared to that of the in-cell Western assay. Interestingly, very good cell potency was seen in the mitochondrial membrane potential assays (IC₅₀ < 50 nM) for most compounds (Table 6). The lower IC₅₀ value in mitochondrial membrane potential assays compared to that in in-cell Western blot assays for these compounds (Table 6) could be due to the fact that JNK2/3 has a greater contribution to mitochondrial dysfunction than JNK1. This interpretation is consistent with the in vivo observation of Zhao and Herdegen where minimal to no JNK1 activity is seen on the mitochondria in neurodegenerative models.⁵⁴ Thus, a small dose of these JNK2/3 isoform selective inhibitors is sufficient to inhibit JNK2/3 activity and to rescue mitochondrial membrane potential.

The high cell potency for isoform selective JNK2/3 inhibitors was further confirmed in our mitochondrial ROS inhibition assays in SHSYSY cells. As shown in Figure 6, compound **26n** significantly reduced the ROS generated by 6-OHDA (35 μM) at a concentration as low as 40 nM (IC₅₀ < 40 nM). Similar cell potency in this specific assay was also observed for compound **26j** (IC₅₀ < 40 nM), showing this

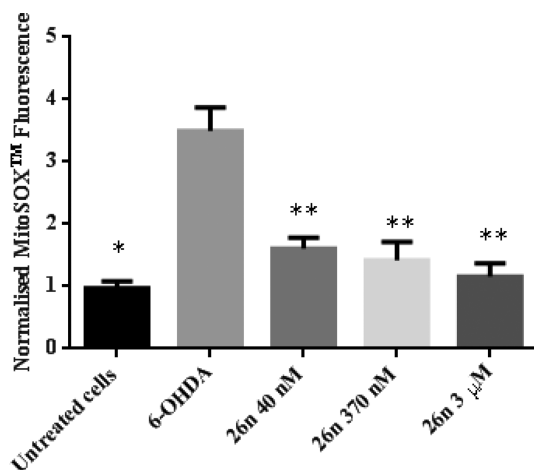


Figure 6. Inhibition of mitochondrial ROS generation by inhibitor **26n** in SHSYSY cells. SHSYSY cells were treated with 35 μM 6-OHDA for 5 h, and mitochondrial ROS was measured by normalized MitoSOX fluorescence in the presence or absence of 40 nM, 370 nM, and 3 μM inhibitor. Statistical significance ($p < 0.05$) between control, untreated group, and 6-OHDA treated group is shown by *. Significance ($p < 0.05$) between 6-OHDA-treated groups and different concentrations of the inhibitor and 6-OHDA-treated groups is shown by **.

phenomenon is shared among the compounds in this class. Indeed, in contrast to biochemical assays, IC₅₀ determinations of functional end points in cell-based assays often have steep slopes indicative of the functional readout that is being measured as opposed to the direct substrate competitive inhibition seen in a biochemical assay, and this could be why the cell-based IC₅₀ is at the biochemical IC₅₀ for **26k** and lower than that for **26n**. We have seen similar behavior in our aminopyrimidine class³⁷ where the inhibition of ROS generation (~1 nM) was lower than that of the biochemical IC₅₀ value (50 nM). Thus, it may be that only a small amount of JNK3 and JNK2 inhibition goes a long way toward shutting down ROS generation and JNK mitochondrial function. Like **26n**, the highlighted compound in the Kamenecka et al. study (**9l**, SR-3562) had exquisite selectivity over a panel of kinases. Thus, while we cannot rule out off-target kinase activity as a possible explanation for the lower IC₅₀ in the cell-based assays, it is unlikely given the high selectivity of these compounds. Finally, we cannot rule out alternative interpretations such as other signaling pathways that do not have a kinase component contributing to ROS or MMP changes in the mitochondria, so this remains a possibility for the greater potency in ROS generation and MMP dissipation for these compounds.

In order to evaluate the feasibility of using these JNK3 isoform selective inhibitors for in vivo applications, in vivo pharmacokinetic (PK) properties for selected compounds were studied in mice. As demonstrated in Table 7, these select lead compounds showed good PK properties in iv dosing. They had good C_{max} and AUC (area under curve) values and fair half-lives (~1–1.5 h). The clearance (Cl) and the volume of distribution (V_d) properties were also good for **22i**, **26j**, and **26k** but were only fair for compound **26n**. On the other hand, these compounds showed poor PK properties in oral dosing. For example, all of them exhibited almost no oral bioavailability (F values are all ≤1%). Therefore, the focus of future optimizations will be to improve oral PK properties if oral administration is preferred. However, these compounds, as JNK2/3 inhibitors for topical applications and/or in iv dosing, may be very useful to protect against retinal degeneration in glaucoma⁵⁵ and could conceivably be used in this indication. Finally, since JNK3 selective inhibitors have the potential to be utilized in CNS disorders, we measured the brain penetration properties of **26n**. After ip dosing (10 mg/kg in mice) brain levels at 2 h after dosing was approximately 55% of that found in the plasma. In addition, we also measured the brain concentration of **26n** at 2 and 6 h after po and iv dosing. Because of the poor oral bioavailability of **26n** (see Table 7) the brain levels at 2 and 6 h were below the lower limit of quantitation (~5 nM) after oral dosing but were found to be 60 and 40 nM at 2 and 6 h, respectively, after iv dosing of 2 mg/kg (Table 8).

CONCLUSION

We have developed a series of aminopyrazoleamide based JNK3 isoform selective inhibitors starting from the initial lead compound SR-4326. SAR investigation and optimization successfully yielded potent JNK3 inhibitors with high selectivity over p38 kinase, as well as high isoform selectivity against JNK1 and in some cases over JNK2. For example, compound **26n** had an isoform selectivity of >50-fold for JNK3 against JNK1 and were potent in all three types of cell-based assays in SHSYSY cells (in-cell Western target modulation, inhibition of 6-OHDA induced cell death, and inhibition of 6-OHDA induced

Table 7. In Vivo PK Data on Mice for Selected Lead JNK3 Inhibitors^a

compd	C _{max} ^a (μM), iv	AUC ^a (μM·h), iv	T _{1/2} ^a (h), iv	Cl ^a ((mL/min)/kg), iv	V _d ^a (L/kg), iv	F ^a (%) po
22i	2.4	1.8	1.0	10	0.6	1.0
26j	3.2	1.5	1.6	12	0.7	0.4
26k	1.9	1.1	1.1	17	0.9	0.6
26n	1.7	1.0	1.5	21	1.2	0.5

^aDosed at 0.5 mg/kg for iv and at 3 mg/kg for po. All data were based on three determinations.

Table 8. Brain Concentration of 26n in Mice^a

dosing	2 h	6 h
iv (2 mg/kg)	60 nM	40 nM
po (5 mg/kg)	<5 nM	<5 nM

^aData were generated from three determinations.

mitochondrial membrane depolarization). Another lead inhibitor **26k** was highly potent and selective with an IC₅₀ value against JNK3 in the subnanomolar range and possessed an isoform selectivity of >500-fold over JNK1 and >20000-fold against p38α. In addition, **26n** and **26k** were noncytotoxic, potently inhibited mitochondrial dysfunction, and possessed good PK properties in iv dosing. In addition, **26n** showed a plasma/brain ratio of ~2:1. The X-ray crystal structure of **26k** in JNK3 revealed a binding mode that can very well support the observed SAR, and this structure was also helpful for designing potent and selective JNK3 inhibitors. Future optimizations for these aminopyrazole-based compounds will be mainly focused on improving the pharmacokinetic properties in order to obtain better JNK3 isoform selective inhibitors for oral dosing. These studies will be reported in due course.

EXPERIMENTAL SECTION

Commercially available reagents and anhydrous solvents were used without further purification unless otherwise specified. Thin layer chromatography (TLC) analyses were performed with precoated silica gel 60 F254. The mass spectra were recorded by LC/MS with Finnigan LCQ Advantage MAX spectrometer of Thermo Electron. Flash chromatography was performed on prepacked columns of silica gel (230–400 mesh, 40–63 μm) by CombiFlash with EtOAc/hexane or MeOH/DCM as eluents. The preparative HPLC was performed on SunFire C₁₈ OBD 10 μm (30 mm × 250 mm) with CH₃CN + 50% MeOH/H₂O + 0.1% TFA as eluents to purify the targeted compounds. Analytic HPLC was performed on Agilent Technologies 1200 series with CH₃CN (solvent B)/H₂O + 0.9% CH₃CN + 0.1% TFA (solvent A) as eluents, and the targeted products were detected by UV in the detection range of 215–310 nm. All compounds were determined to be >95% pure by this method. NMR spectra were recorded with a Bruker 400 MHz spectrometer at ambient temperature with the residual solvent peaks as internal standards. The line positions of multiplets were given in ppm (δ), and the coupling constants (J) were given in hertz. The high-resolution mass spectra (HRMS, electrospray ionization) experiments were performed with Thermo Finnigan Orbitrap mass analyzer. Data were acquired in the positive ion mode at resolving power of 100 000 at *m/z* 400. Calibration was performed with an external calibration mixture immediately prior to analysis.

General Synthetic Procedures. A mixture of 4-nitro-1H-pyrazole **1** (10 mmol), 3-bromobenzoic acid **2** (20 mmol), CuI (2.0 mmol), *trans*-N,N-dimethylcyclohexane-1,2-diamine (4.0 mmol), and Cs₂CO₃ (30 mmol) in DMF (20 mL) was purged with argon and stirred for 12 h at 100 °C in a sealed tube. The reaction mixture was cooled, and filtered through a pad of silica gel, and rinsed with EtOAc. The resulting solution was concentrated in vacuo to yield a crude residue which was purified by chromatography on silica gel (EtOAc/hexane) to provide 3-(4-nitro-1H-pyrazol-1-yl)benzoic acid **3**.

To a solution of 3-(4-nitro-1H-pyrazol-1-yl)benzoic acid **3** (5.0 mmol) in CH₂Cl₂ (20 mL) were added EDC (10 mmol), HOBt (10 mmol), and diisopropylethylamine (15 mmol), and the mixture was stirred for 30 min. Then the 6-methylpyridin-3-amine (5.5 mmol) was added, and the resulting mixture was stirred overnight. Water (50 mL) was added to the reaction mixture and extracted with EtOAc (2 × 100 mL). The resulting solution was concentrated in vacuo to yield a crude product (**4**).

This intermediate was hydrogenated in anhydrous methanol (100 mL) in the presence of 10% Pd/C (1.0 g) under a balloon of hydrogen for 3 h. The mixture was filtered through a Celite pad and evaporated. The residue was purified by chromatography on silica gel (dichloromethane/methanol) to give the 3-(4-amino-1H-pyrazol-1-yl)-N-(6-methylpyridin-3-yl)benzamide **5**.

1-Chloro-2-isocyanatobenzene (0.12 mmol) was added to 3-(4-amino-1H-pyrazol-1-yl)-N-(6-methylpyridin-3-yl)benzamide (0.1 mmol) in CH₂Cl₂ (1.0 mL) at room temperature and stirred for 1 h. The solvent was evaporated and the residue was purified by reverse-phase preparative HPLC to give 3-(4-(3-(2-chlorophenyl)ureido)-1H-pyrazol-1-yl)-N-(6-methylpyridin-3-yl)benzamide **8c** as a white powder.

3-(4-(2-(2-Chlorophenyl)acetamido)-1H-pyrazol-1-yl)-N-(6-methylpyridin-3-yl)benzamide (6a). 66% yield in four steps (TFA salt, white solid); ¹H NMR (400 MHz, DMSO) δ 10.93 (s, 1H), 10.51 (s, 1H), 9.08 (d, *J* = 2.0 Hz, 1H), 8.70 (s, 1H), 8.42 (dd, *J* = 8.4, 2.0 Hz, 1H), 8.38 (s, 1H), 8.10–8.03 (m, 1H), 7.89 (d, *J* = 8.0 Hz, 1H), 7.82 (s, 1H), 7.68 (t, *J* = 8.0 Hz, 2H), 7.42–7.47 (m, 2H), 7.37–7.28 (m, 2H), 3.85 (s, 2H), 2.61 (s, 3H). LC/MS (ESI) *m/z*: [M + H]⁺ calcd for C₂₄H₂₀ClN₅O₂, 446.1; found, 446.4

2-Chlorophenyl (1-(3-(6-Methylpyridin-3-yl)carbamoyl)phenyl)-1H-pyrazol-4-yl)carbamate (6b). 58% yield in four steps (TFA salt, white solid); ¹H NMR (400 MHz, DMSO) δ 10.60–10.54 (m, 2H), 8.82 (d, *J* = 2.4 Hz, 1H), 8.52 (s, 1H), 8.38–8.33 (m, 1H), 8.11 (dd, *J* = 8.4, 2.6 Hz, 1H), 8.07–7.98 (m, 1H), 7.90–7.83 (m, 1H), 7.79 (s, 1H), 7.65 (t, *J* = 8.0 Hz, 1H), 7.60 (d, *J* = 7.6 Hz, 1H), 7.43 (m, 2H), 7.38–7.28 (m, 2H), 2.47 (s, 3H). LC/MS (ESI) *m/z*: [M + H]⁺ calcd for C₂₃H₁₈ClN₅O₃, 448.1; found, 448.4

N-(1-(3-(6-Methylpyridin-3-yl)carbamoyl)phenyl)-1H-pyrazol-4-yl)-1H-indole-2-carboxamide (6c). 62% yield in four steps (TFA salt, white solid); ¹H NMR (400 MHz, DMSO) δ 11.00 (s, 1H), 10.76 (s, 1H), 10.18 (s, 1H), 9.12 (s, 1H), 8.88 (s, 1H), 8.53–8.37 (m, 2H), 8.23 (t, *J* = 4.9 Hz, 2H), 8.12 (dd, *J* = 8.0, 1.2 Hz, 1H), 7.90 (d, *J* = 7.8 Hz, 2H), 7.71 (dd, *J* = 16.4, 8.4 Hz, 2H), 7.53–7.40 (m, 1H), 7.23–7.12 (m, 2H), 2.63 (s, 3H). LC/MS (ESI) *m/z*: [M + H]⁺ calcd for C₂₅H₂₀N₆O₂, 437.2; found, 437.4.

N-(1-(3-(6-Methylpyridin-3-yl)carbamoyl)phenyl)-1H-pyrazol-4-yl)-1,2,3,4-tetrahydroisoquinoline-3-carboxamide (6d). 35% yield in four steps (TFA salt, white solid); ¹H NMR (400 MHz, DMSO) δ 11.10 (s, 1H), 10.89 (s, 1H), 9.80 (s, 1H), 9.69 (s, 1H), 9.04 (s, 1H), 8.83 (s, 1H), 8.48 (s, 1H), 8.37 (s, 1H), 8.17 (dd, *J* = 8.4, 1.6 Hz, 1H), 8.00 (d, *J* = 9.8 Hz, 2H), 7.84–7.71 (m, 1H), 7.59 (s, 1H), 7.37 (m, 3H), 4.49 (s, 2H), 3.56–3.52 (m, 1H), 3.49 (dd, *J* = 9.2, 4.4 Hz, 1H), 3.20 (dd, *J* = 16.8, 12.2 Hz, 1H), 2.62 (s, 3H). LC/MS (ESI) *m/z*: [M + H]⁺ calcd for C₂₆H₂₄N₆O₂, 453.2; found, 453.3.

3-(4-(6-Bromo-1H-benzo[d]imidazol-2-yl)amino)-1H-pyrazol-1-yl)-N-(6-methylpyridin-3-yl)benzamide (7). 48% yield in four steps (TFA salt, white solid); ¹H NMR (400 MHz, DMSO) δ 12.98 (s, 1H), 10.90 (s, 1H), 10.80 (s, 1H), 10.58 (s, 1H), 9.01 (s, 1H), 8.96 (s, 1H), 8.50 (s, 1H), 8.28 (s, 1H), 8.13 (d, *J* = 8.0 Hz, 1H), 8.04 (s, 1H), 8.00 (d, *J* = 8.0 Hz, 1H), 7.76 (t, *J* = 8.0 Hz, 1H), 7.36

(ddd, $J = 56.6, 6.0, 3.2$ Hz, 4H), 2.53 (s, 3H). LC/MS (ESI) m/z : $[M + H]^+$ calcd for $C_{23}H_{18}BrN_7O$, 517.4; found, 517.6.

3-(4-(3-Cyclopentylureido)-1H-pyrazol-1-yl)-N-(6-methylpyridin-3-yl)benzamide (8a). 45% yield in four steps (TFA salt, white solid); 1H NMR (400 MHz, DMSO) δ 10.78 (s, 1H), 8.96 (s, 1H), 8.41 (s, 1H), 8.28 (s, 2H), 8.17 (s, 1H), 7.95 (dd, $J = 8.1, 1.3$ Hz, 1H), 7.79 (d, $J = 8.1$ Hz, 1H), 7.66 (s, 1H), 7.63–7.49 (m, 2H), 6.22 (s, 1H), 3.87 (dd, $J = 13.4, 6.7$ Hz, 1H), 2.51 (s, 3H), 1.76 (dt, $J = 11.6, 5.8$ Hz, 2H), 1.61–1.51 (m, 2H), 1.43–1.48 (m, 2H), 1.36–1.24 (m, 2H). LC/MS (ESI) m/z : $[M + H]^+$ calcd for $C_{22}H_{24}N_6O_2$, 405.2; found, 405.3.

N-(6-Methylpyridin-3-yl)-3-(4-(3-phenylureido)-1H-pyrazol-1-yl)benzamide (8b). 58% yield in four steps (TFA salt, white solid); 1H NMR (400 MHz, DMSO) δ 10.82 (s, 1H), 8.99 (s, 1H), 8.81 (s, 1H), 8.69 (s, 1H), 8.52 (s, 1H), 8.32 (s, 2H), 8.07–7.96 (m, 1H), 7.81 (d, $J = 8.0$ Hz, 1H), 7.78 (d, $J = 14.4$ Hz, 1H), 7.61 (t, $J = 8.0$ Hz, 2H), 7.41 (d, $J = 7.6$ Hz, 2H), 7.26–7.17 (m, 2H), 6.90 (t, $J = 7.2$ Hz, 1H), 2.52 (s, 3H). LC/MS (ESI) m/z : $[M + H]^+$ calcd for $C_{23}H_{20}N_6O_2$, 413.2; found, 413.3.

3-(4-(3-(2-Chlorophenyl)ureido)-1H-pyrazol-1-yl)-N-(6-methylpyridin-3-yl)benzamide (8c). 63% yield in four steps (TFA salt, white solid); 1H NMR (DMSO- d_6 , 400 MHz) δ 10.88 (s, 1H), 9.45 (s, 1H), 9.04 (s, 1H), 8.62 (s, 1H), 8.40 (m, 3H), 8.20 (dd, $J = 8.2, 1.6$ Hz, 1H), 8.09 (d, $J = 8.2$ Hz, 1H), 7.94–7.81 (m, 2H), 7.69 (t, $J = 8.0$ Hz, 1H), 7.63 (s, 1H), 7.47 (dd, $J = 8.0, 1.4$ Hz, 1H), 7.38–7.25 (m, 1H), 7.10–6.96 (m, 1H), 2.59 (s, 3H). LC/MS (ESI) m/z : $[M + H]^+$ calcd for $C_{23}H_{19}FN_6O_2$, 447.1; found, 447.4.

3-(4-(3-(3-Chlorophenyl)ureido)-1H-pyrazol-1-yl)-N-(6-methylpyridin-3-yl)benzamide (8d). 64% yield in four steps (TFA salt, white solid); 1H NMR (400 MHz, DMSO) δ 10.89 (s, 1H), 9.21 (s, 1H), 9.05 (s, 1H), 8.99 (s, 1H), 8.61 (s, 1H), 8.39 (d, $J = 8.6$ Hz, 2H), 8.08 (dd, $J = 8.1, 1.3$ Hz, 1H), 7.88 (d, $J = 8.2$ Hz, 1H), 7.84 (s, 1H), 7.78 (s, 1H), 7.68 (t, $J = 8.0$ Hz, 1H), 7.64 (d, $J = 7.9$ Hz, 1H), 7.34–7.27 (m, 2H), 7.06–6.92 (m, 1H), 2.58 (d, $J = 10.6$ Hz, 3H). LC/MS (ESI) m/z : $[M + H]^+$ calcd for $C_{23}H_{19}FN_6O_2$, 447.1; found, 447.4.

3-(4-(3-(4-Chlorophenyl)ureido)-1H-pyrazol-1-yl)-N-(6-methylpyridin-3-yl)benzamide (8e). 58% yield in four steps (TFA salt, white solid). 1H NMR (400 MHz, DMSO) δ 10.87 (s, 1H), 9.04 (s, 2H), 8.82 (s, 1H), 8.59 (s, 1H), 8.38 (s, 2H), 8.07 (d, $J = 7.8$ Hz, 1H), 7.88 (d, $J = 7.8$ Hz, 1H), 7.83 (s, 1H), 7.67 (t, $J = 8.0$ Hz, 2H), 7.52 (d, $J = 8.8$ Hz, 2H), 7.37–7.22 (m, 2H), 2.58 (s, 3H). LC/MS (ESI) m/z : $[M + H]^+$ calcd for $C_{23}H_{19}FN_6O_2$, 447.1; found, 447.3.

3-(4-(3-(2-Fluorophenyl)ureido)-1H-pyrazol-1-yl)-N-(6-methylpyridin-3-yl)benzamide (8f). 56% yield in four steps (TFA salt, white solid); 1H NMR (400 MHz, DMSO) δ 10.92 (s, 1H), 9.21 (s, 1H), 9.07 (s, 1H), 8.93 (s, 1H), 8.61 (s, 1H), 8.40 (s, 2H), 8.08 (dd, $J = 8.1, 1.4$ Hz, 1H), 7.89 (d, $J = 8.1$ Hz, 1H), 7.84 (s, 1H), 7.67 (dd, $J = 15.0, 7.0$ Hz, 2H), 7.54 (dt, $J = 12.0, 2.4$ Hz, 1H), 7.38–7.22 (m, 1H), 7.15 (dd, $J = 8.2, 1.1$ Hz, 1H), 6.78 (td, $J = 8.4, 2.4$ Hz, 1H), 2.58 (d, $J = 13.2$ Hz, 3H). LC/MS (ESI) m/z : $[M + H]^+$ calcd for $C_{23}H_{19}FN_6O_2$, 431.2; found, 431.3.

3-(4-(3-(3-Fluorophenyl)ureido)-1H-pyrazol-1-yl)-N-(6-methylpyridin-3-yl)benzamide (8g). 55% yield in four steps (TFA salt, white solid); 1H NMR (400 MHz, DMSO) δ 10.98 (s, 1H), 9.14 (d, $J = 7.2$ Hz, 2H), 8.74 (d, $J = 2.3$ Hz, 1H), 8.68 (s, 1H), 8.46 (dd, $J = 5.2, 3.6$ Hz, 2H), 8.22 (td, $J = 8.3, 1.6$ Hz, 1H), 8.19–8.11 (m, 1H), 7.99–7.90 (m, 2H), 7.73 (dd, $J = 16.4, 8.8$ Hz, 2H), 7.30–7.34 (m, 1H), 7.27–7.14 (m, 1H), 7.14–7.01 (m, 1H), 2.66 (s, 3H). LC/MS (ESI) m/z : $[M + H]^+$ calcd for $C_{23}H_{19}FN_6O_2$, 431.2; found, 431.4.

3-(4-(3-(4-Fluorophenyl)ureido)-1H-pyrazol-1-yl)-N-(6-methylpyridin-3-yl)benzamide (8h). 58% yield in four steps (TFA salt, white solid); 1H NMR (400 MHz, DMSO) δ 10.94 (s, 1H), 9.08 (s, 1H), 9.03 (s, 1H), 8.87 (s, 1H), 8.59 (s, 1H), 8.46–8.36 (m, 2H), 8.07 (dd, $J = 8.1, 1.3$ Hz, 1H), 7.89 (d, $J = 8.2$ Hz, 1H), 7.83 (s, 1H), 7.68 (t, $J = 8.0$ Hz, 2H), 7.56–7.45 (m, 2H), 7.19–7.08 (m, 2H), 2.61 (s, 3H). LC/MS (ESI) m/z : $[M + H]^+$ calcd for $C_{23}H_{19}FN_6O_2$, 431.2; found, 431.3.

3-(4-(3-(3-Chlorobenzyl)ureido)-1H-pyrazol-1-yl)-N-(6-methylpyridin-3-yl)benzamide (8i). 32% yield in four steps (TFA salt, white solid); 1H NMR (400 MHz, DMSO) δ 10.97 (s, 1H), 9.45 (s, 1H), 9.12 (d, $J = 2.0$ Hz, 1H), 8.74 (s, 1H), 8.59 (s, 1H), 8.46 (dd, $J =$

8.6, 2.0 Hz, 1H), 8.42 (s, 1H), 8.14–8.07 (m, 1H), 7.93 (d, $J = 8.0$ Hz, 1H), 7.87 (s, 1H), 7.72 (t, $J = 8.0$ Hz, 2H), 7.50 (ddd, $J = 8.0, 4.8, 2.0$ Hz, 2H), 7.42–7.32 (m, 2H), 3.89 (s, 2H), 2.65 (s, 3H). LC/MS (ESI) m/z : $[M + H]^+$ calcd for $C_{24}H_{21}N_6O_2$, 461.1; found, 461.4.

3-(4-(3-(2-Chlorophenyl)-3-methylureido)-1H-pyrazol-1-yl)-N-(6-methylpyridin-3-yl)benzamide (8j). 21% yield in four steps (TFA salt, white solid); 1H NMR (400 MHz, DMSO) δ 10.96 (s, 1H), 9.10 (s, 1H), 8.51 (s, 1H), 8.48–8.37 (m, 2H), 8.34 (t, $J = 1.8$ Hz, 1H), 8.05–7.97 (m, 1H), 7.86 (dd, $J = 6.8, 1.5$ Hz, 1H), 7.75–7.70 (m, 2H), 7.63 (m, 2H), 7.55–7.34 (m, 3H), 3.18 (s, 3H), 2.62 (s, 3H). LC/MS (ESI) m/z : $[M + H]^+$ calcd for $C_{24}H_{21}N_6O_2$, 461.1; found, 461.2.

3-(4-(3-(2-Chlorophenyl)-3-(2-hydroxyethyl)ureido)-1H-pyrazol-1-yl)-N-(6-methylpyridin-3-yl)benzamide (8k). 23% yield in four steps (TFA salt, white solid); 1H NMR (400 MHz, DMSO) δ 10.59 (s, 1H), 8.84 (s, 1H), 8.52 (s, 1H), 8.31 (s, 1H), 8.30–8.23 (m, 1H), 8.16–8.08 (m, 1H), 8.02–7.94 (m, 1H), 7.84 (s, 1H), 7.73 (s, 1H), 7.63 (d, $J = 4.0$ Hz, 2H), 7.54 (s, 1H), 7.46 (s, 2H), 7.37–7.29 (m, 1H), 3.51–3.47 (m, 2H), 3.40–3.44 (m, 2H), 3.17 (s, 1H), 2.48 (s, 3H). LC/MS (ESI) m/z : $[M + H]^+$ calcd for $C_{25}H_{23}N_6O_3$, 491.2; found, 491.4.

N-(1-(3-(6-Methylpyridin-3-yl)carbamoyl)phenyl)-1H-pyrazol-4-yl)indoline-1-carboxamide (8l). 28% yield in four steps (TFA salt, white solid); 1H NMR (400 MHz, DMSO) δ 11.46 (s, 1H), 8.98 (s, 1H), 8.66 (s, 1H), 8.40 (s, 1H), 8.12 (dd, $J = 8.0, 1.6$ Hz, 1H), 8.00–7.91 (m, 3H), 7.91–7.82 (m, 2H), 7.71 (t, $J = 8.0$ Hz, 1H), 7.21 (d, $J = 6.8$ Hz, 1H), 7.13 (t, $J = 7.6$ Hz, 1H), 6.91 (dd, $J = 11.6, 4.2$ Hz, 1H), 6.57 (s, 1H), 4.10 (t, $J = 8.8$ Hz, 2H), 3.21 (t, $J = 8.6$ Hz, 2H), 2.66 (s, 3H). LC/MS (ESI) m/z : $[M + H]^+$ calcd for $C_{25}H_{22}N_6O_2$, 439.1; found, 439.4.

N-(2-Chlorobenzyl)-1-(3-((6-methylpyridin-3-yl)carbamoyl)phenyl)-1H-pyrazole-4-carboxamide (12a). 71% yield in four steps (TFA salt, white solid); 1H NMR (400 MHz, DMSO) δ 10.93 (s, 1H), 9.12 (s, 1H), 9.08 (s, 1H), 8.83 (t, $J = 5.8$ Hz, 1H), 8.48 (t, $J = 1.8$ Hz, 1H), 8.42 (d, $J = 9.2$ Hz, 2H), 8.30 (s, 1H), 8.15 (dd, $J = 8.0, 1.6$ Hz, 1H), 7.98 (d, $J = 6.8$ Hz, 1H), 7.75 (t, $J = 8.0$ Hz, 1H), 7.70 (d, $J = 6.2$ Hz, 1H), 7.48 (dd, $J = 7.6, 1.6$ Hz, 1H), 7.42 (dd, $J = 7.4, 2.0$ Hz, 1H), 7.34 (m, 2H), 4.55 (d, $J = 5.7$ Hz, 2H), 2.61 (s, 3H). LC/MS (ESI) m/z : $[M + H]^+$ calcd for $C_{24}H_{20}ClN_5O_2$, 446.1; found, 446.3.

3-(5-(2-Chlorobenzyl)-4,5,6,7-tetrahydro-1H-pyrazolo[4,3-cl]pyridin-1-yl)-N-(6-methylpyridin-3-yl)benzamide (12b). 45% yield in four steps (TFA salt, white solid); 1H NMR (400 MHz, DMSO) δ 10.88 (s, 1H), 9.02 (s, 1H), 8.54 (s, 1H), 8.40 (t, $J = 1.8$ Hz, 1H), 8.34 (d, $J = 8.4$ Hz, 1H), 8.02 (dd, $J = 8.1, 1.4$ Hz, 1H), 7.94 (d, $J = 8.0$ Hz, 1H), 7.86–7.77 (m, 1H), 7.69 (t, $J = 8.0$ Hz, 1H), 7.64 (dd, $J = 7.6, 1.6$ Hz, 1H), 7.50–7.56 (m, 3H), 4.61 (s, 2H), 4.40–4.45 (m, 2H), 3.10–3.14 (m, 2H), 2.56 (s, 3H). LC/MS (ESI) m/z : $[M + H]^+$ calcd for $C_{26}H_{23}ClN_5O_2$, 472.2; found, 472.5.

3-(4-(3-(2-Chlorophenyl)ureido)-1H-pyrazol-1-yl)-4-fluoro-N-(6-methylpyridin-3-yl)benzamide (16a). 47% yield in five steps (TFA salt, white solid); 1H NMR (400 MHz, DMSO) δ 10.92 (s, 1H), 9.46 (s, 1H), 9.05 (s, 1H), 8.50 (dd, $J = 7.6, 2.4$ Hz, 1H), 8.46–8.31 (m, 3H), 8.18 (dd, $J = 8.4, 1.4$ Hz, 1H), 8.02 (ddd, $J = 8.6, 4.4, 2.4$ Hz, 1H), 7.91 (s, 1H), 7.74–7.69 (m, 1H), 7.69–7.60 (m, 1H), 7.47 (dd, $J = 8.0, 1.4$ Hz, 1H), 7.35–7.26 (m, 1H), 7.11–6.93 (m, 1H), 2.60 (s, 3H). LC/MS (ESI) m/z : $[M + H]^+$ calcd for $C_{23}H_{18}ClFN_6O_2$, 465.1; found, 465.3.

3-(4-(3-(2-Chlorophenyl)ureido)-1H-pyrazol-1-yl)-5-fluoro-N-(6-methylpyridin-3-yl)benzamide (16b). 53% yield in five steps (TFA salt, white solid); 1H NMR (400 MHz, DMSO) δ 10.58 (s, 1H), 9.45 (s, 1H), 8.80 (d, $J = 2.4$ Hz, 1H), 8.65 (s, 1H), 8.39 (s, 1H), 8.27 (s, 1H), 8.21 (dd, $J = 8.4, 1.2$ Hz, 1H), 8.07 (dd, $J = 8.4, 2.4$ Hz, 1H), 7.94 (dt, $J = 10.0, 2.0$ Hz, 1H), 7.90 (s, 1H), 7.66 (d, $J = 9.0$ Hz, 1H), 7.45 (dd, $J = 8.0, 1.2$ Hz, 1H), 7.36–7.20 (m, 2H), 7.00–7.04 (m, 1H), 2.45 (s, 3H). LC/MS (ESI) m/z : $[M + H]^+$ calcd for $C_{23}H_{18}ClFN_6O_2$, 465.1; found, 465.2.

3-(4-(3-(2-Chlorophenyl)ureido)-1H-pyrazol-1-yl)-6-fluoro-N-(6-methylpyridin-3-yl)benzamide (16c). 22% yield in five steps (TFA salt, white solid); 1H NMR (400 MHz, DMSO) δ 10.91 (s, 1H), 9.44 (s, 1H), 8.54 (d, $J = 6.4$ Hz, 1H), 8.48 (s, 1H), 8.39 (s, 1H), 8.24–8.14 (m, 2H), 8.14–8.07 (m, 1H), 8.04 (s, 1H), 7.94 (d, $J = 6.4$

H_z, 1H), 7.85 (s, 1H), 7.57 (t, *J* = 9.2 Hz, 1H), 7.47 (dd, *J* = 8.0, 1.6 Hz, 1H), 7.36–7.23 (m, 1H), 7.09–6.97 (m, 1H), 2.67 (s, 3H). LC/MS (ESI) *m/z*: [M + H]⁺ calcd for C₂₃H₁₈ClFN₆O₂, 465.1; found, 465.4.

3-(4-(3-(2-Chlorophenyl)ureido)-1H-pyrazol-1-yl)-2-fluoro-N-(6-methylpyridin-3-yl)benzamide (16d). 15% yield in five steps (TFA salt, white solid); ¹H NMR (400 MHz, DMSO) δ 10.79 (s, 1H), 9.53 (s, 1H), 8.93 (s, 1H), 8.74 (s, 1H), 8.48 (s, 1H), 8.37 (s, 1H), 8.27 (s, 2H), 8.05 (s, 1H), 7.94–7.99 (m, 1H), 7.56 (d, *J* = 8.0 Hz, 1H), 7.39 (d, *J* = 8.0, 1H), 7.29 (s, 1H), 7.14–7.18 (m, 1H), 7.10–7.13 (m, 1H), 7.00–7.06 (m, 1H), 2.60 (s, 3H). LC/MS (ESI) *m/z*: [M + H]⁺ calcd for C₂₃H₁₈ClFN₆O₂, 465.1; found, 465.4.

1-(2-Chlorophenyl)-3-(1-(3-(pyridin-4-yl)phenyl)-1H-pyrazol-4-yl)urea (20a). 38% yield in four steps (TFA salt, white solid); ¹H NMR (400 MHz, DMSO) δ 9.43 (s, 1H), 8.84 (d, *J* = 4.8 Hz, 2H), 8.69 (s, 1H), 8.39 (s, 1H), 8.28 (s, 1H), 8.24–8.09 (m, 3H), 8.02 (d, *J* = 8.1 Hz, 1H), 7.87–7.76 (m, 2H), 7.68 (t, *J* = 8.0 Hz, 1H), 7.46 (dd, *J* = 8.0, 1.2 Hz, 1H), 7.36–7.24 (m, 1H), 7.03 (td, *J* = 8.0, 1.6 Hz, 1H). LC/MS (ESI) *m/z*: [M + H]⁺ calcd for C₂₁H₁₆ClN₅O, 390.1; found, 390.2.

1-(2-Chlorophenyl)-3-(1-(3-(pyrimidin-4-ylamino)phenyl)-1H-pyrazol-4-yl)urea (20b). 25% yield in four steps (TFA salt, white solid); ¹H NMR (400 MHz, DMSO) δ 10.72 (s, 1H), 9.41 (s, 1H), 8.87 (s, 1H), 8.44 (s, 1H), 8.37 (s, 2H), 8.18 (d, *J* = 6.8 Hz, 2H), 7.81 (s, 1H), 7.58 (d, *J* = 7.6 Hz, 2H), 7.51 (d, *J* = 8.0 Hz, 1H), 7.46 (d, *J* = 7.0 Hz, 1H), 7.30 (s, 1H), 7.03 (s, 1H), 6.98 (d, *J* = 6.6 Hz, 1H). LC/MS (ESI) *m/z*: [M + H]⁺ calcd for C₂₁H₁₆ClN₇O, 406.1; found, 406.3.

1-(2-Chlorophenyl)-3-(1-(3-(5-methyl-4H-1,2,4-triazol-3-yl)phenyl)-1H-pyrazol-4-yl)urea (21a). 28% yield in five steps (TFA salt, white solid); ¹H NMR (400 MHz, DMSO) δ 9.40 (s, 1H), 8.52 (s, 1H), 8.39 (s, 2H), 8.20 (d, *J* = 8.0, 1H), 7.87 (d, *J* = 8.0, 1H), 7.83 (m, 2H), 7.58 (t, *J* = 7.8 Hz, 1H), 7.47 (d, *J* = 8.0, 1H), 7.31 (t, *J* = 7.2 Hz, 1H), 7.16 (s, 1H), 7.04 (t, *J* = 7.6 Hz, 1H), 2.42 (s, 3H). LC/MS (ESI) *m/z*: [M + H]⁺ calcd for C₁₉H₁₆ClN₇O, 394.1; found, 394.3.

1-(2-Chlorophenyl)-3-(1-(3-(5-methyl-1,3,4-oxadiazol-2-yl)phenyl)-1H-pyrazol-4-yl)urea (21b). 37% yield in five steps (TFA salt, white solid); ¹H NMR (400 MHz, DMSO) δ 9.50 (s, 1H), 8.59 (s, 1H), 8.42 (s, 1H), 8.29 (s, 1H), 8.20 (d, *J* = 7.1 Hz, 1H), 8.04 (d, *J* = 8.0 Hz, 1H), 7.83 (s, 1H), 7.78 (d, *J* = 7.4 Hz, 1H), 7.63 (t, *J* = 7.8 Hz, 1H), 7.47 (d, *J* = 8.0 Hz, 1H), 7.31 (t, *J* = 7.8 Hz, 1H), 7.04 (t, *J* = 7.8 Hz, 1H), 2.45 (s, 3H). LC/MS (ESI) *m/z*: [M + H]⁺ calcd for C₁₉H₁₅ClN₆O₂, 395.1; found, 395.2.

3-(4-(3-(2-Chlorophenyl)ureido)-1H-pyrazol-1-yl)-N-methyl-N-(6-methylpyridin-3-yl)benzamide (22a). 48% yield in four steps (TFA salt, white solid); ¹H NMR (DMSO-*d*₆, 400 MHz) δ 9.88 (s, 1H), 9.42 (s, 1H), 8.59 (s, 1H), 8.36–8.39 (m, 1H), 8.32 (s, 1H), 8.21 (dd, *J* = 8.4, 1.6 Hz, 1H), 8.01 (d, *J* = 9.6 Hz, 1H), 7.90 (s, 1H), 7.86–7.78 (m, 2H), 7.62 (t, *J* = 8.0 Hz, 1H), 7.47 (dd, *J* = 8.0, 1.4 Hz, 1H), 7.31 (t, *J* = 7.2 Hz, 1H), 7.03 (dd, *J* = 10.8, 4.4 Hz, 1H), 3.77 (s, 3H), 2.17 (s, 3H). LC/MS (ESI) *m/z*: [M + H]⁺ calcd for C₂₄H₂₁ClN₆O₂, 461.2; found, 461.5.

3-(4-(3-(2-Chlorophenyl)ureido)-1H-pyrazol-1-yl)-N-(2-methylpyridin-3-yl)benzamide (22b). 55% yield in four steps (TFA salt, white solid); ¹H NMR (400 MHz, DMSO) δ 10.80 (s, 1H), 9.63 (s, 1H), 8.77–8.64 (m, 2H), 8.52 (dd, *J* = 4.8, 3.0 Hz, 2H), 8.44 (d, *J* = 8.0 Hz, 1H), 8.26 (dd, *J* = 8.3, 1.5 Hz, 1H), 8.17 (dd, *J* = 8.4, 1.2 Hz, 1H), 7.98 (d, *J* = 8.0 Hz, 1H), 7.93 (s, 1H), 7.88–7.80 (m, 1H), 7.75 (t, *J* = 8.0 Hz, 1H), 7.53 (dd, *J* = 8.0, 1.6 Hz, 1H), 7.42–7.32 (m, 1H), 7.14–7.06 (m, 1H), 2.73 (s, 3H). LC/MS (ESI) *m/z*: [M + H]⁺ calcd for C₂₃H₁₉ClN₆O₂, 447.1; found, 447.4.

3-(4-(3-(2-Chlorophenyl)ureido)-1H-pyrazol-1-yl)-N-(2,6-dimethylpyridin-3-yl)benzamide (22c). 48% yield in four steps (TFA salt, white solid); ¹H NMR (400 MHz, DMSO) δ 10.55 (s, 1H), 9.45 (s, 1H), 8.62 (s, 1H), 8.41 (d, *J* = 7.6 Hz, 2H), 8.19 (d, *J* = 8.3 Hz, 2H), 8.09 (d, *J* = 8.4 Hz, 1H), 7.89 (d, *J* = 7.6 Hz, 1H), 7.85 (s, 1H), 7.68 (t, *J* = 8.0 Hz, 1H), 7.57 (s, 1H), 7.47 (dd, *J* = 8.0, 1.6 Hz, 1H), 7.37–7.26 (m, 1H), 7.08–7.00 (m, 1H), 2.63 (s, 3H), 2.57 (s, 3H). LC/MS (ESI) *m/z*: [M + H]⁺ calcd for C₂₄H₂₁ClN₆O₂, 461.1; found, 461.4.

3-(4-(3-(2-Chlorophenyl)ureido)-1H-pyrazol-1-yl)-N-(6-methylpyridin-2-yl)benzamide (22d). 32% yield in four steps (TFA salt,

white solid); ¹H NMR (400 MHz, DMSO) δ 11.10 (s, 1H), 9.60 (s, 1H), 8.72 (s, 1H), 8.43 (d, *J* = 14.2 Hz, 2H), 8.24–8.18 (m, 1H), 8.11–8.01 (m, 2H), 7.90 (d, *J* = 7.8 Hz, 1H), 7.84–7.74 (m, 2H), 7.62 (t, *J* = 8.0 Hz, 1H), 7.46 (dd, *J* = 8.0, 1.4 Hz, 1H), 7.30 (d, *J* = 7.2 Hz, 1H), 7.14–7.05 (m, 1H), 7.04 (d, *J* = 7.8 Hz, 1H), 2.53 (s, 3H). LC/MS (ESI) *m/z*: [M + H]⁺ calcd for C₂₃H₁₉ClN₆O₂, 447.1; found, 447.4.

3-(4-(3-(2-Chlorophenyl)ureido)-1H-pyrazol-1-yl)-N-(5-methylpyridin-2-yl)benzamide (22e). 43% yield in five steps (TFA salt, white solid); ¹H NMR (400 MHz, DMSO) δ 10.74 (s, 1H), 9.44 (s, 1H), 9.08 (s, 2H), 8.61 (s, 1H), 8.40 (d, *J* = 5.2 Hz, 2H), 8.21 (d, *J* = 8.2 Hz, 1H), 8.08 (d, *J* = 7.6 Hz, 1H), 7.88 (d, *J* = 12.4 Hz, 2H), 7.68 (t, *J* = 8.0 Hz, 1H), 7.47 (d, *J* = 7.8 Hz, 1H), 7.31 (t, *J* = 7.6 Hz, 1H), 7.04 (t, *J* = 7.2 Hz, 1H), 2.62 (s, 3H). LC/MS (ESI) *m/z*: [M + H]⁺ calcd for C₂₃H₁₉ClN₆O₂, 447.1; found, 447.3.

3-(4-(3-(2-Chlorophenyl)ureido)-1H-pyrazol-1-yl)-N-(2-methylpyridin-4-yl)benzamide (22f). 22% yield in four steps (TFA salt, white solid); ¹H NMR (400 MHz, DMSO) δ 11.42 (s, 1H), 9.45 (s, 1H), 8.63 (s, 2H), 8.40 (s, 2H), 8.20 (d, *J* = 6.8 Hz, 1H), 8.12 (s, 2H), 8.06–7.99 (m, 1H), 7.87 (s, 2H), 7.73 (d, *J* = 8.0 Hz, 1H), 7.48 (d, *J* = 9.4 Hz, 1H), 7.31 (s, 1H), 7.06 (d, *J* = 6.0 Hz, 1H), 2.66 (s, 3H). LC/MS (ESI) *m/z*: [M + H]⁺ calcd for C₂₃H₁₉ClN₆O₂, 447.1; found, 447.3.

3-(4-(3-(2-Chlorophenyl)ureido)-1H-pyrazol-1-yl)-N-(2,6-dimethylpyridin-4-yl)benzamide (22g). 20% yield in four steps (TFA salt, white solid); ¹H NMR (400 MHz, DMSO) δ 11.10 (s, 1H), δ 9.58 (s, 1H), 8.72 (s, 1H), 8.47–8.37 (m, 2H), 8.25–8.18 (m, 1H), 8.12–8.02 (m, 2H), 7.96–7.89 (m, 1H), 7.84–7.74 (m, 2H), 7.63 (t, *J* = 7.6 Hz, 1H), 7.51–7.42 (m, 1H), 7.31 (t, *J* = 7.8 Hz, 1H), 7.10 (d, *J* = 8.0 Hz, 1H), 7.05 (t, *J* = 7.6 Hz, 1H), 2.53–2.52 (m, 6H). LC/MS (ESI) *m/z*: [M + H]⁺ calcd for C₂₄H₂₁ClN₆O₂, 461.1; found, 461.4.

3-(4-(3-(2-Chlorophenyl)ureido)-1H-pyrazol-1-yl)-N-(2-methylpyrimidin-5-yl)benzamide (22h). 46% yield in four steps (TFA salt, white solid); ¹H NMR (400 MHz, DMSO) δ 10.74 (s, 1H), 9.44 (s, 1H), 9.08 (s, 2H), 8.61 (s, 1H), 8.40 (d, *J* = 5.2 Hz, 2H), 8.21 (d, *J* = 8.2 Hz, 1H), 8.08 (d, *J* = 7.6 Hz, 1H), 7.88 (d, *J* = 12.4 Hz, 2H), 7.68 (t, *J* = 8.0 Hz, 1H), 7.47 (d, *J* = 7.8 Hz, 1H), 7.31 (t, *J* = 7.6 Hz, 1H), 7.04 (t, *J* = 7.2 Hz, 1H), 2.62 (s, 3H). LC/MS (ESI) *m/z*: [M + H]⁺ calcd for C₂₃H₁₉ClN₆O₂, 448.1; found, 448.4.

3-(4-(3-(2-Chlorophenyl)ureido)-1H-pyrazol-1-yl)-N-(6-methylpyridin-3-yl)methyl)benzamide (22i). 66% yield in four steps (TFA salt, white solid); ¹H NMR (400 MHz, DMSO) δ 9.46 (s, 1H), 9.42 (t, *J* = 5.6 Hz, 1H), 8.76 (s, 1H), 8.58 (s, 1H), 8.41 (s, 1H), 8.34 (d, *J* = 8.2 Hz, 1H), 8.30 (s, 1H), 8.19 (dd, *J* = 8.4, 1.4 Hz, 1H), 8.01 (dd, *J* = 8.1, 1.3 Hz, 1H), 7.85–7.77 (m, 3H), 7.61 (t, *J* = 8.0 Hz, 1H), 7.47 (dd, *J* = 8.0, 1.4 Hz, 1H), 7.35–7.27 (m, 1H), 7.04 (td, *J* = 8.0, 1.6 Hz, 1H), 4.63 (d, *J* = 5.6 Hz, 2H), 2.67 (s, 3H). LC/MS (ESI) *m/z*: [M + H]⁺ calcd for C₂₄H₂₁ClN₆O₂, 461.1; found, 461.2.

3-(4-(3-(2-Chlorophenyl)ureido)-1H-pyrazol-1-yl)-N-(1-methyl-1H-pyrazol-4-yl)benzamide (26a). 72% yield in four steps (TFA salt, white solid); ¹H NMR (400 MHz, DMSO) δ 10.61 (s, 1H), 9.43 (s, 1H), 8.59 (s, 1H), 8.39 (s, 1H), 8.34 (s, 1H), 8.20 (dd, *J* = 8.3, 1.4 Hz, 1H), 8.06 (s, 1H), 8.06–7.98 (m, 1H), 7.84 (t, *J* = 3.6 Hz, 2H), 7.69–7.58 (m, 2H), 7.46 (dd, *J* = 8.0, 1.4 Hz, 1H), 7.36–7.24 (m, 1H), 7.04 (td, *J* = 8.0, 1.6 Hz, 1H), 3.84 (s, 3H). LC/MS (ESI) *m/z*: [M + H]⁺ calcd for C₂₁H₁₈ClN₇O₂, 436.1; found, 436.2.

3-(4-(3-(2-Chlorophenyl)ureido)-1H-pyrazol-1-yl)-N-(1-methyl-1H-pyrazol-3-yl)benzamide (26b). 68% yield in four steps (TFA salt, white solid); ¹H NMR (400 MHz, DMSO) δ 11.07 (s, 1H), 9.42 (s, 1H), 8.66 (s, 1H), 8.38 (s, 2H), 8.23 (dd, *J* = 8.4, 1.6 Hz, 1H), 8.02 (dd, *J* = 8.1, 1.3 Hz, 1H), 7.88 (d, *J* = 8.0 Hz, 1H), 7.82 (s, 1H), 7.67–7.57 (m, 2H), 7.47 (dd, *J* = 8.0, 1.5 Hz, 1H), 7.38–7.26 (m, 1H), 7.08–6.97 (m, 1H), 6.63 (d, *J* = 2.4 Hz, 1H), 3.81 (s, 3H). LC-MS: 436 (M + H). LC/MS (ESI) *m/z*: [M + H]⁺ calcd for C₂₁H₁₈ClN₇O₂, 436.1; found, 436.4.

3-(4-(3-(2-Chlorophenyl)ureido)-1H-pyrazol-1-yl)-N-(1-methyl-1H-pyrazol-5-yl)benzamide (26c). 65% yield in four steps (TFA salt, white solid); ¹H NMR (400 MHz, DMSO) δ 10.58 (s, 1H), 9.50 (s, 1H), 8.68 (s, 1H), 8.45 (s, 2H), 8.26 (dd, *J* = 8.4, 1.6 Hz, 1H), 8.14 (dd, *J* = 8.0, 1.4 Hz, 1H), 7.99–7.88 (m, 2H), 7.72 (t, *J* = 8.0 Hz, 1H), 7.56–7.45 (m, 2H), 7.41–7.32 (m, 1H), 7.14–7.05 (m, 1H), 6.33 (d, *J*

= 1.9 Hz, 1H), 3.79 (s, 3H). LC-MS: 436 (M + H). LC/MS (ESI) *m/z*: [M + H]⁺ calcd for C₂₁H₁₈ClN₇O₂, 436.1; found, 436.2.

3-(4-(3-(2-Chlorophenyl)ureido)-1H-pyrazol-1-yl)-N-(1,3-dimethyl-1H-pyrazol-4-yl)benzamide (26d). 75% yield in four steps (TFA salt, white solid); ¹H NMR (400 MHz, MeOD) δ 8.36 (s, 1H), 8.19 (s, 1H), 8.06–8.00 (m, 1H), 7.86 (d, *J* = 8.0 Hz, 1H), 7.79–7.72 (m, 2H), 7.65 (s, 1H), 7.53 (t, *J* = 8.0 Hz, 1H), 7.31 (dd, *J* = 8.0, 1.2 Hz, 1H), 7.21–7.17 (m, 1H), 6.99–6.92 (m, 1H), 3.74 (s, 3H), 2.16 (s, 3H). LC-MS: 436 (M + H). LC/MS (ESI) *m/z*: [M + H]⁺ calcd for C₂₂H₂₀ClN₇O₂, 450.1; found, 450.4.

3-(4-(3-(2-Chlorophenyl)ureido)-1H-pyrazol-1-yl)-N-(1-methyl-1H-1,2,4-triazol-3-yl)benzamide (26e). 66% yield in four steps (TFA salt, white solid); ¹H NMR (400 MHz, DMSO) δ 10.91 (s, 1H), 9.43 (s, 1H), 8.63 (s, 1H), 8.39 (t, *J* = 13.4 Hz, 3H), 8.21 (dd, *J* = 8.4, 1.6 Hz, 1H), 8.05 (dd, *J* = 8.0, 1.4 Hz, 1H), 7.92–7.81 (m, 2H), 7.63 (t, *J* = 7.9 Hz, 1H), 7.47 (dd, *J* = 8.0, 1.4 Hz, 1H), 7.36–7.26 (m, 1H), 7.08–6.99 (m, 1H), 3.87 (s, 3H). LC-MS: 436 (M + H). LC/MS (ESI) *m/z*: [M + H]⁺ calcd for C₂₀H₁₇ClN₈O₂, 437.1; found, 437.3.

3-(4-(3-(2-Chlorophenyl)ureido)-1H-pyrazol-1-yl)-N-(1-isopropyl-1H-pyrazol-4-yl)benzamide (26f). 44% yield in five steps (TFA salt, white solid); ¹H NMR (400 MHz, DMSO) δ 10.60 (d, *J* = 7.2 Hz, 1H), 9.44 (s, 1H), 8.59 (s, 1H), 8.44–8.32 (m, 2H), 8.21 (dd, *J* = 8.3, 1.5 Hz, 1H), 8.09 (d, *J* = 3.6 Hz, 1H), 8.06–7.97 (m, 1H), 7.84 (d, *J* = 6.8 Hz, 2H), 7.70–7.58 (m, 2H), 7.47 (dd, *J* = 8.0, 1.5 Hz, 1H), 7.36–7.26 (m, 1H), 7.09–7.00 (m, 1H), 4.55–4.47 (m, 1H), 1.42 (dd, *J* = 6.8, 2.0 Hz, 7H). LC-MS: 436 (M + H). LC/MS (ESI) *m/z*: [M + H]⁺ calcd for C₂₃H₂₂ClN₈O₂, 464.2; found, 464.4.

3-(4-(3-(2-Chlorophenyl)ureido)-1H-pyrazol-1-yl)-N-(1-(2-dimethylamino)ethyl)-1H-pyrazol-4-yl)benzamide (26g). 54% yield in four steps (TFA salt, white solid); ¹H NMR (400 MHz, MeOD) δ 8.36 (s, 1H), 8.20 (t, *J* = 1.8 Hz, 1H), 8.13 (s, 1H), 8.02 (dd, *J* = 8.4, 1.6 Hz, 1H), 7.87–7.82 (m, 1H), 7.74 (d, *J* = 8.4 Hz, 1H), 7.68 (s, 1H), 7.64 (s, 1H), 7.52 (t, *J* = 8.0 Hz, 1H), 7.31 (dd, *J* = 8.0, 1.4 Hz, 1H), 7.18 (d, *J* = 1.6 Hz, 1H), 6.95 (dd, *J* = 7.8, 1.6 Hz, 1H), 4.57–4.32 (m, 2H), 3.59 (t, *J* = 5.7 Hz, 2H), 2.87 (s, 6H). LC/MS (ESI) *m/z*: [M + H]⁺ calcd for C₂₄H₂₅ClN₈O₂, 493.2; found, 493.5.

3-(4-(3-(2-Chlorophenyl)ureido)-1H-pyrazol-1-yl)-N-(1-(2-diethylamino)ethyl)-1H-pyrazol-4-yl)benzamide (26h). 54% yield in four steps (TFA salt, white solid); ¹H NMR (400 MHz, MeOD) δ 8.48 (s, 1H), 8.32 (t, *J* = 2.0 Hz, 1H), 8.28 (s, 1H), 8.14 (dd, *J* = 8.4, 1.6 Hz, 1H), 8.01–7.93 (m, 1H), 7.86 (dd, *J* = 5.4, 4.0 Hz, 1H), 7.78 (d, *J* = 14.8 Hz, 2H), 7.65 (t, *J* = 8.0 Hz, 1H), 7.43 (dd, *J* = 8.0, 1.4 Hz, 1H), 7.33–7.24 (m, 1H), 7.10–6.98 (m, 1H), 4.61 (t, *J* = 5.9 Hz, 2H), 3.72 (t, *J* = 6.0 Hz, 2H), 3.32–3.28 (m, 4H), 1.34 (t, *J* = 7.2 Hz, 6H). LC/MS (ESI) *m/z*: [M + H]⁺ calcd for C₂₆H₂₉ClN₈O₂, 521.2; found, 521.3.

3-(4-(3-(2-Chlorophenyl)ureido)-1H-pyrazol-1-yl)-N-(1-(2-pyrrolidin-1-yl)ethyl)-1H-pyrazol-4-yl)benzamide (26i). 55% yield in four steps (TFA salt, white solid); ¹H NMR (400 MHz, DMSO) δ 10.74 (s, 1H), 9.49 (s, 1H), 8.59 (s, 1H), 8.42 (s, 1H), 8.36 (s, 1H), 8.28 (s, 1H), 8.20 (dd, *J* = 8.4, 1.6 Hz, 1H), 8.03 (dd, *J* = 8.0, 1.4 Hz, 1H), 7.86 (d, *J* = 7.0 Hz, 2H), 7.74 (s, 1H), 7.65 (t, *J* = 7.9 Hz, 1H), 7.47 (dd, *J* = 8.0, 1.4 Hz, 1H), 7.36–7.27 (m, 1H), 7.09–7.00 (m, 1H), 4.52 (t, *J* = 6.0 Hz, 2H), 3.64 (s, 2H), 3.49 (s, 2H), 3.01 (s, 2H), 1.90–1.94 (m, 4H). LC/MS (ESI) *m/z*: [M + H]⁺ calcd for C₂₃H₂₂ClN₈O₂, 519.2; found, 519.5.

3-(4-(3-(2-Chlorophenyl)ureido)-1H-pyrazol-1-yl)-N-(1-(piperidin-4-yl)-1H-pyrazol-4-yl)benzamide (26j). 62% yield in four steps (TFA salt, white solid); ¹H NMR (400 MHz, MeOD) δ 8.50 (s, 1H), 8.33 (s, 1H), 8.25 (s, 1H), 8.15 (dd, *J* = 8.4, 1.6 Hz, 1H), 7.99 (d, *J* = 6.0 Hz, 1H), 7.87 (d, *J* = 8.4 Hz, 1H), 7.76 (s, 1H), 7.73 (s, 1H), 7.66 (t, *J* = 7.9 Hz, 1H), 7.46–7.39 (m, 1H), 7.30 (d, *J* = 6.9 Hz, 1H), 7.11–7.01 (m, 1H), 3.59 (d, *J* = 13.6 Hz, 2H), 3.51–3.46 (m, 1H), 3.29–3.19 (m, 2H), 2.40–2.27 (m, 4H). LC/MS (ESI) *m/z*: [M + H]⁺ calcd for C₂₅H₂₅ClN₈O₂, 505.2; found, 505.4.

3-(4-(3-(2-Chlorophenyl)ureido)-1H-pyrazol-1-yl)-N-(1-(pyrrolidin-3-yl)-1H-pyrazol-4-yl)benzamide (26k). 53% yield in four steps (TFA salt, white solid); ¹H NMR (400 MHz, DMSO) δ 10.72 (s, 1H), 9.48 (s, 1H), 9.10 (s, 1H), 8.59 (d, *J* = 7.7 Hz, 1H), 8.42 (s, 1H), 8.36 (s, 1H), 8.28 (s, 1H), 8.20 (dd, *J* = 8.4, 1.6 Hz, 1H), 8.02 (dt, *J* = 11.4, 5.6 Hz, 1H), 7.86 (d, *J* = 5.2 Hz, 1H), 7.75 (s, 1H), 7.62

(dt, *J* = 19.8, 8.0 Hz, 1H), 7.47 (dt, *J* = 8.0, 2.0 Hz, 1H), 7.36–7.22 (m, 1H), 7.04 (ddd, *J* = 11.0, 4.6, 2.3 Hz, 1H), 5.33–5.12 (m, 1H), 3.77–3.27 (m, 5H), 2.42 (m, 1H), 2.26 (m, 1H). LC/MS (ESI) *m/z*: [M + H]⁺ calcd for C₂₄H₂₃ClN₈O₂, 496.2; found, 496.4.

3-(4-(3-(2-Chlorophenyl)ureido)-1H-pyrazol-1-yl)-N-(2-(diethylamino)ethyl)benzamide (26l). 73% yield in four steps (TFA salt, white solid); ¹H NMR (400 MHz, CH₃OD) δ 8.44 (s, 1H), 8.23 (d, *J* = 1.7 Hz, 1H), 8.12 (dd, *J* = 8.2, 1.4 Hz, 1H), 7.97–7.91 (m, 1H), 7.81–7.72 (m, 2H), 7.60 (t, *J* = 7.9 Hz, 1H), 7.41 (dd, *J* = 8.0, 1.4 Hz, 1H), 7.33–7.25 (m, 1H), 7.09–7.02 (m, 1H), 3.53 (t, *J* = 6.4 Hz, 2H), 3.27–3.20 (m, 4H), 2.05 (dt, *J* = 19.2, 5.6 Hz, 2H), 1.33 (t, *J* = 7.2 Hz, 6H). LC/MS (ESI) *m/z*: [M + H]⁺ calcd for C₂₃H₂₇ClN₆O₂, 455.2; found, 455.4.

3-(4-(3-(2-Chlorophenyl)ureido)-1H-pyrazol-1-yl)-N-(3-(pyrrolidin-1-yl)propyl)benzamide (26m). 69% yield in four steps (TFA salt, white solid); ¹H NMR (400 MHz, MeOD) δ 8.45 (s, 1H), 8.23 (t, *J* = 1.8 Hz, 1H), 8.13 (dd, *J* = 8.4, 1.5 Hz, 1H), 8.00–7.89 (m, 1H), 7.84–7.71 (m, 2H), 7.60 (t, *J* = 8.0 Hz, 1H), 7.43 (dd, *J* = 8.0, 1.4 Hz, 1H), 7.30 (ddd, *J* = 8.0, 7.6, 1.5 Hz, 1H), 7.12–6.92 (m, 1H), 3.78–3.62 (m, 2H), 3.55 (t, *J* = 6.5 Hz, 2H), 3.29 (d, *J* = 8.0 Hz, 2H), 3.11 (dd, *J* = 10.8, 8.0 Hz, 2H), 2.23–2.13 (m, 2H), 2.13–1.90 (m, 4H). LC-MS: 467 (M + H). LC/MS (ESI) *m/z*: [M + H]⁺ calcd for C₂₄H₂₇ClN₆O₂, 467.2; found, 467.5.

3-(4-(3-(2-Chlorophenyl)ureido)-1H-pyrazol-1-yl)-N-(piperidin-4-ylmethyl)benzamide (26n). 67% yield in four steps (TFA salt, white solid); ¹H NMR (400 MHz, DMSO) δ 9.45 (s, 1H), 8.79 (t, *J* = 6.0 Hz, 1H), 8.56 (s, 1H), 8.40 (s, 1H), 8.24 (t, *J* = 1.7 Hz, 1H), 8.19 (dd, *J* = 8.3, 1.5 Hz, 1H), 7.98 (dd, *J* = 8.0, 1.2 Hz, 1H), 7.82 (s, 1H), 7.75 (d, *J* = 8.0 Hz, 1H), 7.58 (t, *J* = 8.0 Hz, 1H), 7.47 (dd, *J* = 8.0, 1.6 Hz, 1H), 7.34–7.27 (m, 1H), 7.07–7.00 (m, 1H), 3.43 (d, *J* = 11.3 Hz, 2H), 3.22 (t, *J* = 6.0 Hz, 2H), 2.90 (q, *J* = 10.4 Hz, 2H), 2.75 (s, 1H), 1.89 (d, *J* = 14.2 Hz, 2H), 1.80–1.84 (m, 1H), 1.38 (dd, *J* = 24.8, 10.8 Hz, 2H). LC/MS (ESI) *m/z*: [M + H]⁺ calcd for C₂₃H₂₅ClN₆O₂, 453.2; found, 453.9.

3-(4-(3-(2-Chlorophenyl)ureido)-1H-pyrazol-1-yl)-N-((1-methylpiperidin-4-yl)methyl)benzamide (26o). 72% yield in four steps (TFA salt, white solid); ¹H NMR (400 MHz, DMSO) δ 9.47 (s, 1H), 8.80 (t, *J* = 6.0 Hz, 1H), 8.56 (d, *J* = 6.4 Hz, 2H), 8.41 (s, 1H), 8.24 (d, *J* = 1.6 Hz, 2H), 8.19 (dd, *J* = 8.3, 1.5 Hz, 1H), 7.97 (dd, *J* = 8.0, 1.2 Hz, 1H), 7.81 (s, 1H), 7.75 (d, *J* = 8.0 Hz, 1H), 7.58 (t, *J* = 8.0 Hz, 1H), 7.46 (dd, *J* = 8.0, 1.6 Hz, 1H), 7.33–7.27 (m, 1H), 7.11–6.98 (m, 1H), 3.35 (s, 3H), 3.28 (d, *J* = 12.6 Hz, 2H), 3.23 (t, *J* = 6.1 Hz, 2H), 2.94–2.77 (m, 2H), 1.83 (d, *J* = 12.4 Hz, 2H), 1.35 (d, *J* = 12.4 Hz, 2H), 1.29–1.23 (m, 1H). LC/MS (ESI) *m/z*: [M + H]⁺ calcd for C₂₄H₂₇ClN₆O₂, 467.2; found, 467.6.

N-(Azetidin-3-yl)-3-(4-(3-(2-chlorophenyl)ureido)-1H-pyrazol-1-yl)benzamide (26p). 58% yield in four steps (TFA salt, white solid); ¹H NMR (400 MHz, DMSO) δ 9.51 (s, 1H), 9.32 (d, *J* = 6.4 Hz, 1H), 8.56 (s, 1H), 8.44 (s, 1H), 8.28 (t, *J* = 2.0 Hz, 1H), 8.23–8.15 (m, 1H), 8.05–7.98 (m, 1H), 7.84 (s, 1H), 7.77 (d, *J* = 8.0 Hz, 1H), 7.66–7.59 (m, 1H), 7.47 (dd, *J* = 8.0, 1.4 Hz, 1H), 7.36–7.27 (m, 1H), 7.08–6.99 (m, 1H), 4.85 (m, 1H), 4.18 (m, 4H), 3.26 (m, 1H). LC/MS (ESI) *m/z*: [M + H]⁺ calcd for C₂₀H₁₉ClN₆O₂, 411.1; found, 411.4.

3-(4-(3-(2-Chlorophenyl)ureido)-1H-pyrazol-1-yl)-N-(pyrrolidin-3-yl)benzamide (26q). 65% yield in four steps (TFA salt, white solid); ¹H NMR (400 MHz, DMSO) δ 9.45 (s, 1H), 8.80 (s, 1H), 8.56 (s, 1H), 8.41 (s, 1H), 8.25 (s, 1H), 8.19 (d, *J* = 8.4 Hz, 1H), 8.00 (d, *J* = 8.1 Hz, 1H), 7.84 (s, 1H), 7.76 (d, *J* = 7.6 Hz, 1H), 7.61 (t, *J* = 8.0 Hz, 1H), 7.52–7.40 (m, 1H), 7.30 (dd, *J* = 11.6, 4.4 Hz, 1H), 7.10–6.95 (m, 1H), 4.54 (d, *J* = 6.1 Hz, 1H), 3.47 (d, *J* = 6.3 Hz, 3H), 3.30 (s, 1H), 3.25–3.16 (m, 1H), 2.23 (dd, *J* = 13.2, 7.6 Hz, 1H), 2.12–2.00 (m, 1H). LC/MS (ESI) *m/z*: [M + H]⁺ calcd for C₂₁H₂₁ClN₆O₂, 425.1; found, 425.4.

JNK-1/2/3 and p38 Assays. Homogeneous time-resolved fluorescence assay—enzyme inhibition studies were performed in 384-well polystyrene homogeneous time-resolved fluorescence plates (Grainier) for 1 h at ambient temperature (~22 °C) with 0.5 μM biotinylated FL-ATF-2, 1.25 μM ATP, 0.75 nM activated JNK3α1 or JNK2 or JNK1 (with a control in the absence of kinase for

determining the basal signal) in 10 μ L volumes containing the final concentrations of the following: 50 mM Hepes, pH 7.0, 2.5 mM $MgCl_2$, 0.1 mg/mL bovine serum albumin, 1 mM DL-dithiothreitol, 0.01% Triton X-100 (all from Sigma-Aldrich), and 5% DMSO (with or without compound). A 10-point titration of all compounds was carried out in 3-fold dilutions from 10 μ M to 0.5 nM. After 22 min, the kinase reaction was terminated by addition of 10 μ L of quenching solution (1 \times Lance buffer, detection reagents, streptavidin-xLAPC (200 nM), and europium cryptate-labeled rabbit polyclonal anti-phospho-ATF-2 (1 nM) were from Cis-Bio). The homogeneous time-resolved fluorescence signal was detected using an EnVision plate reader 1 h after quenching. The data from four different experiments were averaged and presented as the mean \pm SD. IC₅₀ values were determined by fitting the data to the equation for a four-parameter logistic.^{3,4,37,51}

Cell Culture. SHSYSY cells (ATCC) were grown at 37 °C and 5% CO₂ in DMEM/F:12 (Invitrogen) supplemented with 10% fetal bovine serum (FBS) and penicillin/streptomycin. To ensure that the cells were actively growing, only cells at ~80% confluency and between passages five and 15 were used in the experiments. The cells were serum starved for 24 h in DMEM/F:12 medium containing 2% FBS before any treatment.

Mitochondrial Membrane Depolarization. Mitochondrial membrane depolarization was monitored by MitoTracker Orange CMTRos (Invitrogen) fluorescence. SHSYSY cells (60 000 cells/well) were seeded in black walled, clear bottomed 96-well plates. Cells were incubated with compounds for 30 min before the addition of 35 μ M 6-hydroxydopamine (6-OHDA) for 4 h. After the incubation, cells were stained with 500 nM MitoTracker Orange CMTRos for 30 min under growth conditions. Cells were washed twice in Hank's buffer salt solution (HBSS), and placed in prewarmed HBSS for fluorescent recordings. Fluorescence was detected at 576 nm (exciting at 554 nm) on a SpectraMax e5 plate reader (Molecular Devices). Mitochondrial membrane depolarization was normalized to cell abundance by staining the cells with Hoechst 33342 (excitation, 350 nm; emission, 450 nm).

Cell Viability. Cell viability of SHSYSY cells was monitored by MTT assay (Cayman Chemical). Cells (60 000 cells/well) were seeded in a 96-well plate (clear bottom) and treated as described in the text. At the culmination of each treatment the cells were treated with the MTT reagent. Absorbance was monitored in a SpectraMax e5 plate reader (Molecular Devices).

Measurement of Compound Cytotoxicity. Cytotoxicity of the compounds was monitored by MTT assay (Cayman Chemical). SHSYSY cells (60 000 cells/well) were seeded in a 96-well plate (clear bottom) and incubated with different concentrations (0–30 μ M) of compounds for 48 h. At the culmination of the treatment the cells were treated with the MTT reagent. Absorbance was monitored in a SpectraMax e5 plate reader (Molecular Devices).

In Cell Western Assay. SHSYSY cells (60 000 cells/well) were plated in a clear-bottomed Packard View black 96-well plate in 100 μ L of 10% FBS DMEM:F12 medium and were allowed to attach overnight. Next day, the cells were treated with the compounds for 1 h prior to induction of the JNK pathway activation. The cells were treated with 35 μ M 6-OHDA for 4–5 h. Cells were then fixed in 4% paraformaldehyde in PBS for 20 min at room temperature with no shaking. They were then washed once with 0.1 M glycine to neutralize paraformaldehyde for 5 min. Cells were permeabilized with 0.2% Triton X-100 in PBS for 20 min at room temperature on orbital shaker after which they were washed once with PBS for 5 min. They were then incubated with Licor blocking buffer in PBS (1:1 dilution in PBS) for 1–1.5 h rocking at room temperature. Cells were incubated with primary antibody p-c-Jun S63 Ab (Cell Signaling no. 9261) 1:100 dilution in Licor blocking buffer overnight at 4 °C. Next day, they were washed twice with PBS–0.1% Tween 20 (PBST) washing solution for 5 min each at room temperature on the orbital shaker, followed by one wash with Licor blocking buffer containing 0.05% Tween-20 for 5 min on the shaker at room temperature. The cells were then incubated with secondary antibody goat anti-rabbit IR800 (1:500 dilution) for 1 h at room temperature in the dark (covered the plate with foil) in

Licor blocking buffer-containing Tween-20. Following this, cells were washed twice with PBST for 5 min each at room temperature and then once with Licor blocking buffer-containing 0.05% Tween-20. The wells were then incubated with ToPro 3 stain (nucleic acid staining), diluted 1:4000 in Licor blocking buffer or Licor blocking buffer with 0.05% Tween-20 for 30 min at room temperature in the dark. Finally the plates were washed twice with PBS and analyzed using the Odyssey LICOR infrared scanner.^{3,4,20,36,37,51,52,56}

Crystallization. Purification of JNK3 39-402 and its crystallization with AMPNP was done following previously published procedure.^{36,37,57} Compound **26k** was soaked into the crystal by adding 2 mM compound into the crystallization drop and incubating for 24 h. The crystal was then transferred to a mounting loop, and excess soaking solution was removed and flash frozen by plunging into liquid nitrogen.

Data Collection and Refinement. A diffraction data set was collected at LS-CAT beamline 21-ID-G (APS) using Marmosiac 300 CCD detector (Mar Research). The data set was processed with autoProc with XDS as the data reduction engine. The PDB code 1JNK was used as the molecular replacement model. Quick molecular replacement using Phaser in Phenix suite properly positioned JNK3, **26k** was identified as positive densities in ATP binding pocket. Restraints and coordinates for the compound was generated using eLBOW in Phenix suite and incorporated into the JNK3 coordinate using the graphics program Coot. The model was then refined using autoBuster with TLS (translation, libration, and screw-motion), water update, and unknown ligand search options turned on. The model was manually inspected and adjusted after each refinement cycle using Coot. The refinement was completed after the free and crystallographic R-factors stabilized. Data processing and refinement statistics are given in Supporting Information. Structural analysis and figure preparations were done using PyMol. The coordinate and the structure factor are deposited to wwPDB with the PDB code 4WH2.

■ ASSOCIATED CONTENT

● Supporting Information

Details for the synthesis of **12**, **16**, **22**, and **26**; LC purity data; protocols for JNKs and p38 enzyme assays; protocols of DMPK studies. This material is available free of charge via the Internet at <http://pubs.acs.org>.

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Notes

The authors declare no competing financial interest.

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■ ABBREVIATIONS USED

JNK, c-jun NH2-terminal kinase; MAPK, mitogen-activated protein kinase; ATF2, activating transcription factor 2; ATP, adenosine triphosphate; 6-OHDA, 6-hydroxydopamine; MMP, mitochondrial membrane potential; AD, Alzheimer's disease; PD, Parkinson's disease; ALS, amyotrophic lateral sclerosis; AUC, pharmacokinetic area under curve; Cl, pharmacokinetic clearance; C_{max} , pharmacokinetic maximum concentration; F , percent oral bioavailability; HLMS, human liver microsomal stability; V_d , volume of distribution; SAR, structure–activity relationship; DMPK, drug metabolism and pharmacokinetics

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