

Evaluation of the Effect of Four Surface Conditioning Methods on the Shear Bond Strength of Metal Bracket to Porcelain Surface

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Abstract

Objective: This study evaluated the effect of superpulse CO₂ laser irradiation and deglazing of porcelain surfaces on the shear bond strength (SBS) of metal orthodontic brackets, and compared it with two conventional etching techniques. **Methods:** Forty-eight Feldspathic porcelain fused to metal specimens embedded in cylindrical acrylic resin tubes were fabricated, and all the specimens were divided into four groups. In Group 1, the specimens were roughened with a diamond bur and etched with hydrofluoric acid (HFA) gel for 4 min. In Group 2, the specimens were roughened with a bur and irradiated by a CO₂ laser with a 2 W power setting for 20 sec. In Group 3, the specimens were only irradiated by a CO₂ laser. In Group 4, the porcelain surface was sandblasted with 50 μm aluminum oxide. Before bonding, the bracket silane was applied on the porcelain surfaces. SBS was evaluated by a Universal testing machine (Zwickroll, Germany). The remaining adhesive after the bond failure was evaluated using an adhesive remnant index (ARI). Statistical analysis was conducted by analysis of variance (ANOVA), Tukey, and Kruskal–Wallis tests. **Results:** ANOVA revealed significant differences in SBS among the four groups ($p < 0.001$). Group 1 demonstrated significantly higher bond strength (13.13 ± 2.47) when compared with the other groups. Group 2 showed higher bond strength (9.60 ± 1.91) when compared with group 4 (6.40 ± 1.67) ($p = 0.016$). Group 1 displayed the highest ARI scores among the groups. **Conclusions:** Deglazing combined with HFA etching produced the highest bond strength, but CO₂ laser irradiation provided adequate bond strength and allowed for elimination of the HFA step. Deglazing is not recommended as a preliminary step before CO₂ laser conditioning.

Introduction

ONE OF ORTHODONTISTS' PROBLEMS IS the need to bond orthodontic brackets to artificial surfaces such as porcelain, rather than the normal tooth surface.^{1–4} It has been proven that conventional acid etching is not effective in preparing the porcelain surface for the mechanical retention of orthodontic brackets.^{5,6}

During the past few years, a number of methods have been introduced to overcome this problem.^{7–9} One of these techniques is deglazing the porcelain surface by roughening the surface with a diamond bur, or microetching the surface with aluminum oxide particles and after that bonding the bracket with or without a coupling agent.^{5,10} Whereas some laboratory studies have concluded that mechanical or che-

mical abrasion to deglaze the porcelain surface is necessary for enhanced bond strength,^{11–14} others report no advantage to deglazing, and suggest silane bonding.^{15,16} In a number of studies,^{17–19} silanes have been used to increase the bond strength to either glazed or roughened dental porcelain.⁶

Chemical preparation of the previously deglazed porcelain surface has been suggested.⁵ Application of orthophosphoric acid (OPA) as one of these chemical agents failed to present sufficient bond strength for clinical purposes.^{20,21} Hydrofluoric acid (HFA), on the other hand, seemed to be able to produce bond strength of acceptable values.^{4,20,21} It has even been reported that airborne-particle abrasion and acid etching with HFA increase the surface area of ceramic surfaces and create an irregular topography that enhances the potential for micromechanical retention of the adhesive cement.¹

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With the introduction of the laser, the idea of using it as a means of reinforcing bond strength has become popular. In previous studies, the efficacy of different types of lasers on porcelain surface preparation has been evaluated.^{5,22–26} Some studies claimed that Nd:YAG laser with appropriate parameters can be used as an alternative method for porcelain etching.^{22,23} The results on the efficacy of Er:YAG laser in porcelain conditioning is more controversial.^{22,24,25} Two studies evaluated the effect of CO₂ laser irradiation of porcelain surfaces on the bond strength of metal orthodontic brackets.^{5,26} Akova et al.⁵ concluded that superpulse CO₂ laser irradiation provided adequate bonding strength between metal brackets and porcelain surfaces. Ahrari et al.²⁶ evaluated the effect of fractional CO₂ laser conditioning of porcelain surfaces on shear bond strength (SBS) of orthodontic brackets. This study recommended laser treatment as a suitable alternative to HFA for deglazed feldspathic porcelain.

Considering the challenge for orthodontists in preparation of adequate bond strength to the porcelain surfaces and the fact that there are few studies^{5,26} that evaluate the effect of new techniques such as CO₂ laser irradiation in this regard, this study aimed to compare the effect of four different methods of porcelain surface treatment on the SBS of orthodontic metal brackets.

Materials and Methods

For this experimental study, 48 porcelain fused to metal specimens (Ceramco, Dentsply, York, PA; Heraus Kulzer, Hanau, Germany) were fabricated. Each specimen was formed in the form of a disc by the same operator. Feldspathic porcelain was placed on 3 mm nickel-chromium disc to produce 5-mm-thick specimens with a radius of 8 mm. A vacuum dental furnace (model: Centurion VPC, Ney Dental, Inc., Burlington, NJ) was used to fire the porcelain. The porcelain for the alloy had a heat up rate of 55°C/min to a peak firing temperature of 960°C and 0 min hold. All specimens were embedded in cylindrical acrylic resin tubes (Cold cure acrylic, Acropars, Marlic Inc., Iran) with a mounting jig. The mounting jig was used to align the direction of debonding force parallel to the surface of the porcelain during shear strength measurements. Metal standard edgewise maxillary central incisor brackets with 0.022 slot (American Orthodontics, Sheboygan, WI) and a surface area of 12.09 mm² was placed on the porcelain surface. Before bonding the brackets, the surface area border on which the bracket was to be placed later was drawn by waterproof marker on the specimens so that the area that had to be treated could be controlled.

All the specimens were divided into four groups by systematic random sampling. The groups were divided as follows:

Deglazed + HFA group (Group1): The specimens were roughened with a diamond bur (863 Grit; Drendell and Zewilling, Berlin, Germany) to remove the glaze and then etched with 9.6% HFA gel (Porcelain Etch Gel, Pulpdent Corp., Watertown, MA) for 4 min, rinsed with distilled water for 30 sec, and then air dried for 15 sec.

Deglazed + CO₂ group (Group2): The specimens were roughened with a diamond bur to remove the glaze, and then irradiated by superpulse CO₂ laser (Ultradream pulse V, Daeshin enterprise, Hungary) with a 2 W power setting in the superpulse mode (15 msec, 2 Hz)

for 20 sec as suggested by Akova et al.⁵ via contacting the laser probe on the porcelain surface.

CO₂ group (Group 3): The specimens were only irradiated by CO₂ laser as was done with the second group, without removing the glaze.

Sandblasted group (Group 4): The porcelain surface was sandblasted (Micro-etcher, Sandblaster, Masel, 2701 Bartram road, Bristol, USA) with 50 μm aluminum oxide at 80 psi for 5 sec. In order to remove the remnants of aluminum oxide, after the sandblasting procedure, all the specimens were cleaned with compressed air for 20 sec.

In the four groups, a silane coupling agent (Silane Bond Enhancer; Pulpdent Corp.) was applied after porcelain conditioning. Then, the brackets were bonded with light-curing orthodontic adhesive according to the manufacturer's instructions (Transbond XT, 3M Unitek, Monrovia, CA). The excess resin was removed by an explorer and the each surface was cured with light curing device (L.E.D. curing unit, Smartlife IQ2, Dentsply, Milford, USA) for a total of 40 sec from mesial, distal, occlusal, and gingival directions (10 sec each).

The specimens were then placed in an incubator (Vafaei, Iran) in water at 37°C for 24 h, and then SBS was evaluated by Universal testing machine (Zwickroll, Germany) with a crosshead speed of 1 mm/sec. During the debonding procedure, the bonded surface of the porcelain was parallel to the direction of force application, and the blade of the Instron was placed against the wing of the bracket. The shear force on the specimen was continued until debonding occurred. The maximum load applied to the surface (in Newtons) before debonding occurred was then divided by the cross-sectional area of the bracket (12.09 mm²) so that the results were obtained in megapascals (MPa). The process of applying the shear forces against the brackets and porcelain surface treatment was accomplished by operators who were blinded to the study.

After debonding the brackets, the porcelain specimens were examined under a stereomicroscope (SZ 40, Olympus, Tokyo, Japan) of ×20 magnification to assess the mode of failure. Adhesive remnants indexes (ARI) were graded as suggested by Artun and Bergland²⁷:

Grade 0: no adhesive remnant on former bracket site

Grade 1: adhesive remnants covering ≤50% of the former bracket site

Grade 2: adhesive remnants covering >50% of the former bracket site

Grade 3: all adhesive left behind on the former bracket site

Statistical analysis

To compare mean SBS among different groups of the study, statistical analysis was conducted by means of one way analysis of variance (ANOVA). At the next step (post-hoc), in order to compare the groups two by two, Tukey test with a correction in *p* value was used. Kruskal–Wallis and Mann–Whitney tests were used to compare the ARI scores. The statistical significance level was established at 0.05. All the statistical analysis was performed by SPSS software (Statistical Package for the Social Sciences, Version 11.5, SPSS, Chicago, IL).

Results

The descriptive statistics of the shear bond strength in all four groups are summarized in Table 1. ANOVA revealed significant differences in SBS among the four groups ($p < 0.001$). Two by two comparison of the mean SBS in the study groups is outline in Table 2. Multiple comparisons by Tukey test demonstrated that the highest bond strength was in the deglazed+HFA group (Group 1) and that the lowest was in the sandblasted group (Group 4).

The distribution of ARI scores is shown in Table 3. Kruskal–Wallis showed that there were significant differences among the four groups ($p < 0.001$) regarding the ARI scores. In the deglazed+HFA group (Group 1), an increased frequency of ARI score 2 was reported. In the deglazed+CO₂ group (Group 2), a higher frequency of ARI score 1 was present, and none of the failures occurred in ARI scores of 2 or 3. In the sandblasted and CO₂ laser groups (Groups 3 and 4), most of the bond failures happened at ARI scores of 0. No ARI scores of 3 were reported in any of the specimens (Fig. 1).

Discussion

Because of the chemical properties of bonding agents as well as the glazed surfaces of porcelain, bonding orthodontic brackets to porcelain surfaces has always been an issue for orthodontists.⁵ Conventional methods such as acid etching provide an inadequate bond strength for orthodontic brackets and, therefore, are not as effective.⁷ A number of alternative methods have been suggested to overcome this problem, such as deglazing and roughening the porcelain surface with a diamond bur, microetching the surface with aluminum oxide particles, conditioning the porcelain surface with the application of silane, and, most recently, lasers.^{5,10,13,14,18,21–26}

Some of the previous studies^{22–25} have evaluated the effect of Nd:YAG and Er:YAG lasers on porcelain surface treatment for efficient bracket bonding, whereas some other studies^{5,26} have assessed the effectiveness of CO₂ laser for this purpose.

During the process of heat induction of porcelain surfaces with a focused CO₂ laser, conchoidal tears, a typical effect of surface warming, appear. It is said that these tears provide mechanical retention between the composite resin and porcelain.^{5,28} Considering several advantages CO₂ lasers provide, such as time saving and biocompatibility,⁵ their effectiveness on the SBS of orthodontic brackets was compared with that of sandblasting and HFA etching in our study. The deglazed+HFA group (Group 1) demonstrated significantly higher bond strength (13.13 ± 2.47) when compared with other groups; whereas the deglazed+CO₂ group (Group 2) showed higher bond strength (9.60 ± 1.91)

TABLE 1. DESCRIPTIVE STATISTICS OF SHEAR BOND STRENGTH (MPa) IN FOUR GROUPS

Study group	Mean (MPa)	SD	Range
1 (Deglazed + HFA)	13.13	2.47	9.59–17.70
2 (Deglazed + CO ₂)	9.60	1.91	6.10–12.03
3 (CO ₂)	8.38	3.74	4.17–18.53
4 (Sandblast)	6.40	1.67	4.05–9.93

HFA, hydrofluoric acid.

TABLE 2. TWO BY TWO COMPARISON OF THE MEAN SHEAR BOND STRENGTH IN THE STUDY GROUPS

Study groups	Mean difference	Significance
1 (Deglazed + HFA), 2 (Deglazed + CO ₂)	3.53598	0.009*
1 (Deglazed + HFA), 3 (CO ₂)	4.75738	0.000**
1 (Deglazed + HFA), 4 (sandblast)	6.72801	0.000**
2 (Deglazed + CO ₂), 3 (CO ₂)	1.22140	0.655
2 (Deglazed + CO ₂), 4 (sandblast)	3.19203	0.021*
3 (CO ₂), 4 (sandblast)	1.97064	0.256

* $p < 0.05$, ** $p < 0.001$.
HFA, hydrofluoric acid.

when compared with the sandblasted group (Group 4) (6.40 ± 1.67) ($p = 0.016$).

According to Reynolds,²⁹ a minimum bond strength of 6–8 MPa is required for efficient clinical orthodontic bonding, whereas 13 MPa has been considered the maximum authorized strength between porcelain and adhesive to prevent cohesive porcelain fracture.³⁰ In our study, the mean SBS in the deglazed+HFA group (group 1) was 13.13, which is not only higher than the values reported for other groups, but also higher than the maximum authorized strength between porcelain and adhesive.³⁰ The values reported in the rest of the groups in our study were well within acceptable limits.

The effectiveness of different types of lasers on porcelain surface treatment has been evaluated in previous studies.^{5,22–26}

It has been proven that CO₂ wavelength is absorbed almost totally by porcelain, suggesting that it as a good candidate for porcelain surface treatment.^{5,28} Some investigators^{5,26} used the CO₂ laser as the present study, and reported satisfying results in SBS as a result of conditioning the porcelain surface by CO₂ laser. Akova et al.⁵ studied the effect of conventional CO₂ laser with the output energy of 2 W on the bond strength of orthodontic brackets to porcelain surfaces, and reported that the average debonding force was 6.26 ± 0.58 MPa. All the samples in their study were deglazed by grinding. It has been stated that this method has the capability of introducing microcracks along the porcelain and, therefore, is not recommended by some investigators.^{5,7} Ahrari et al.²⁶ used a fractional CO₂ laser with different output energies on glazed and deglazed porcelain surfaces. This study showed that deglazing followed by CO₂ laser treatment at 10 W produced the highest mean bond strength (11.4 MPa), whereas the glazed surfaces treated by CO₂ laser at 20 W yielded the lowest SBS value (4.4 MPa).

TABLE 3. DESCRIPTIVE STATISTICS OF ARI SCORES IN ALL GROUPS

Study Group	Adhesive remnant index (ARI) scores			
	0	1	2	3
1 (Deglazed + HFA)	0 (0%)	2 (16.7%)	10 (83.3%)	0 (0%)
2 (Deglazed + CO ₂)	2 (16.7%)	10 (83.3%)	0 (0%)	0 (0%)
3 (CO ₂)	7 (58.3%)	4 (33.3%)	1 (8.3%)	0 (0%)
4 (Sandblast)	8 (66.7%)	4 (33.3%)	0 (0%)	0 (0%)

HFA, hydrofluoric acid.

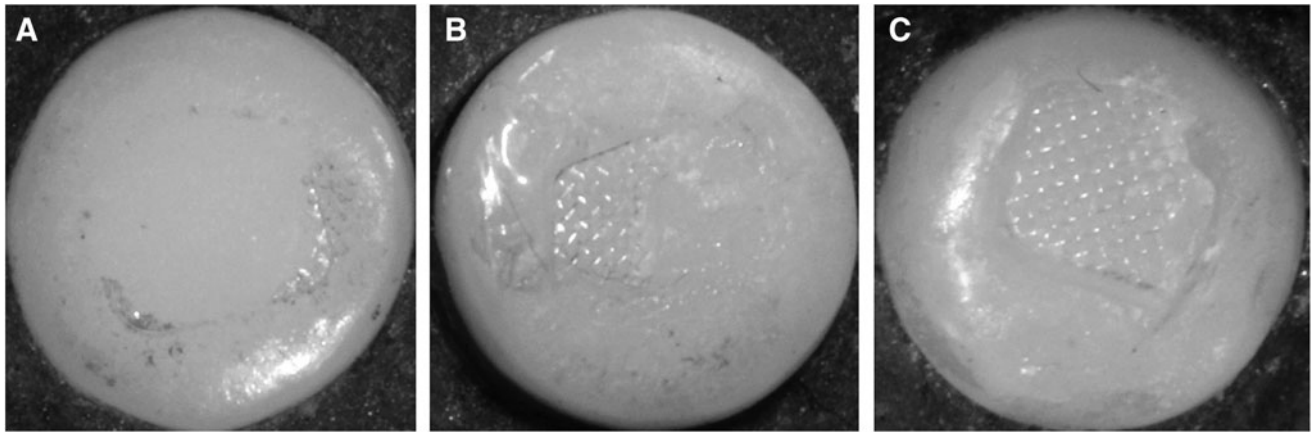


FIG. 1. After debonding the brackets, we found three modes of bond failure in the porcelain specimens under the stereomicroscope. (A) Adhesive remnant index (ARI) 0: no adhesive remnant on the former bracket site. (B) ARI 1: adhesive remnants covering $\leq 50\%$ of the former bracket site. (C) ARI 2: adhesive remnants covering $> 50\%$ of former bracket site.

In our study, laser irradiation with an output energy of 2 W was used for treating the porcelain surface. The power of the laser used in the study of Akova et al.⁵ was also 2 W, which was obtained from a pilot study comparing the effect of 2, 3, 5, 10, and 15 W powers on surface bond strength. That pilot study found that output energies other than 2 W were not suitable for porcelain surface treatment and produced surface cracks, which could reduce the fracture resistance of porcelain. However, in the study of Ahrari et al.,²⁶ porcelain surfaces were irradiated with the different powers of 10, 15, and 20 W, using a fractional CO₂ laser device. Their results indicated that the bond strengths of the aforementioned groups were not significantly different from each other.

In the present study, the deglazed and laser-irradiated group (Group 2) showed higher SBS than the group only irradiated by laser (Group 3), but the difference was not statistically significant. This finding is in agreement with that of Akova et al.⁵ who did not find any significant differences between the SBS of glazed and deglazed surfaces. Ahrari et al.,²⁶ however, reported that the SBS of the deglazed and laser-irradiated group was higher than that of the glazed group, and concluded that CO₂ laser can produce higher bond strength on deglazed porcelain when compared with HFA. The difference between the results of the present study and the study of Ahrari et al. can be attributed to different parameters and duration of laser application. The specimens in the latter study were treated with a fractional CO₂ laser in dynamic mode for 10 sec, with an output energy of 10 mJ, a frequency of 200 Hz, and powers of 10 and 15 W, which were different from the parameters used in our study [2 W power setting in the superpulse mode (15 msec, 2 Hz) for 20 sec]. Moreover, the bond strength of the specimens treated by HFA were lower in the study of Ahrari et al.²⁶ than in our study, which may be related to different etching time (2 vs. 4 min). In the study conducted by Ahrari et al.,²⁶ the type of bracket and adhesive used was the same as our study; moreover, none of the specimens were thermocycled, whereas in the study conducted by Akova et al.,⁵ a different type of bracket and adhesive was used and the specimens underwent thermocycling prior to SBS testing. However, the results of our study were mostly in agreement with the results of Akova et al., rather than with those of Ahrari et al. There-

fore, it can be deduced that surface conditioning techniques as well as laser parameters play a more crucial role in the bond strength of brackets to ceramic surfaces than do the aforementioned factors.

Deglazing may increase the risk of crack propagation and probability of porcelain fracture during debonding,³¹ and the aesthetic characteristics may be irreversibly damaged by the roughening process.³² Considering the comparable bond strength of the glazed and deglazed porcelain surfaces irradiated by laser in the present study, grinding the porcelain surface before bonding in cases in which CO₂ laser is to be applied cannot be justified.

The findings of the present study indicate that porcelain treated with either CO₂ laser irradiation or sandblasting can yield bond strength values within the acceptable clinical range. Considering the need for rinsing and drying the porcelain surface following the application of HFA, it is obvious that a more time-consuming process is required when working with HFA (~ 4 min) than when working with CO₂ laser (20 sec). Moreover, HFA requires extra care to avoid soft tissue damage.

The evaluation of the remaining composite after the shear bond test showed that none of the specimens had an ARI score of 3, which means that in none of the experimental groups did the entire adhesive remain on the porcelain surface. The highest frequency of an ARI score of 2 was found in the deglazed+HFA group (Group 1), which had the highest reported SBS as well. It can be concluded that HFA produces the best wettability of porcelain surface. This finding could also indicate that more time is needed to remove the remaining adhesive from the porcelain surface treated by HFA than with CO₂ laser after bracket debonding.

It has been proven that adhesive failure during debonding is preferred to cohesive failure, because the latter may lead to porcelain fracture.³³ If the bond strength between the porcelain and composite resin is > 13 MPa, the risk of cohesive fracture increases.³⁰ This is in agreement with the results of our study, because in the deglazed+HFA group (Group 1), the reported bond strength was > 13 MPa, and all of the bond failures were cohesive, whereas in the CO₂ and

sandblasted groups (Groups 3 and 4) most of the bond failures were adhesive. A higher frequency of adhesive failure in the latter two groups compared with HFA can be attributed to the lower bond strength.

It is highly recommended that further *in vivo* studies be conducted to obtain further information regarding different surface treatment procedures.

Conclusions

The results of this study indicated that HFA provides a higher-than-recommended SBS, whereas CO₂ laser irradiation with or without porcelain deglazing provides adequate bond strength between metal brackets and porcelain surfaces. Therefore, considering the more time-consuming process required for porcelain etching with HFA, possible soft tissue injuries, and the excessive mean bond strength, CO₂ laser can be recommended as an appropriate substitute for HFA in porcelain surface treatment, with a diminished risk of porcelain fracture. Considering the comparable bond strength in the laser-treated groups with or without porcelain deglazing, and the risks engendered by deglazing on the integrity of porcelain, deglazing is not recommended prior to CO₂ laser irradiation. Moreover, based on the results of the ARI scores, albeit HFA provides the best porcelain surface wetting, complete adhesive removal following a bond failure will be more time-consuming.

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Author Disclosure Statement

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