

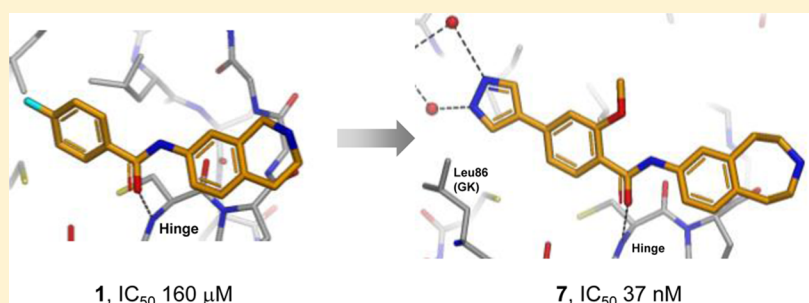
Fragment-Based Discovery of Type I Inhibitors of Maternal Embryonic Leucine Zipper Kinase

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Supporting Information



ABSTRACT: Fragment-based drug design was successfully applied to maternal embryonic leucine zipper kinase (MELK). A low affinity (160 μM) fragment hit was identified, which bound to the hinge region with an atypical binding mode, and this was optimized using structure-based design into a low-nanomolar and cell-penetrant inhibitor, with a good selectivity profile, suitable for use as a chemical probe for elucidation of MELK biology.

KEYWORDS: Maternal embryonic leucine zipper kinase, fragment-based drug design, structure-based optimization

Maternal embryonic leucine zipper kinase (MELK) is a serine/threonine kinase belonging to the CAMK family of kinases.¹ MELK is structurally related to AMPK, SNARK, and NUAKE, which are implicated in metabolism² but, in contrast to these, is not regulated by LKB1.³ Early studies suggested that MELK might be a promising target for the treatment of several cancers.⁴ Increased MELK expression has been reported in brain, breast, prostate, melanoma, and colorectal carcinomas, and is correlated with poor prognosis.^{5–10} Normal adult cells have low expression of MELK, offering promise for selective treatment, rendering the assessment of the role of MELK in carcinogenesis, tumor growth, and metastasis even more urgent. In addition, it has been shown recently that MELK plays an important role in the proliferation of glioma stem-like cells in glioblastoma multiforme¹¹ through phosphorylation of the transcription factor FOXM1, which can be disrupted by the antibiotic Siomycin A. Similarly, depletion of MELK by shRNA decreases tumor growth and malignancy and increases survival in a GSC-derived tumor mouse model.¹² MELK has also been shown to be upregulated in mammary tumor-initiating cells, and its function is required for mammary tumorigenesis *in vivo*.¹³ Consequently, inhibition of MELK was shown to prevent mammosphere formation in breast cancer cells.¹⁴

Elucidation of the role of MELK in cancer^{15,16} is hampered by the lack of well-characterized, selective inhibitors of the kinase activity. At the time of the execution of this project, only nonselective inhibitors of MELK were described.¹⁷ Most potent was dorsomorphin, developed as a KDR (VEGFR) inhibitor¹⁸ and subsequently described as an AMPK inhibitor.^{19,20} More recently, a number of publications on MELK inhibitors have appeared,^{21–25} for example, OTSSP167,¹⁴ but none discloses both ligand structure and data showing a broad selectivity profile.

This letter describes the discovery of a dual-selective MELK inhibitor using fragment-based drug design (FBDD). FBDD has been successfully applied across a number of target classes, and fragment-derived kinase inhibitors have progressed to clinical trials.^{26,27} An important factor in the success of FBDD campaigns is access to high quality structural information in the form of multiple protein–ligand X-ray crystal structures. For this work, soakable crystal forms of MELK were developed that were suitable for high throughput structure determination of

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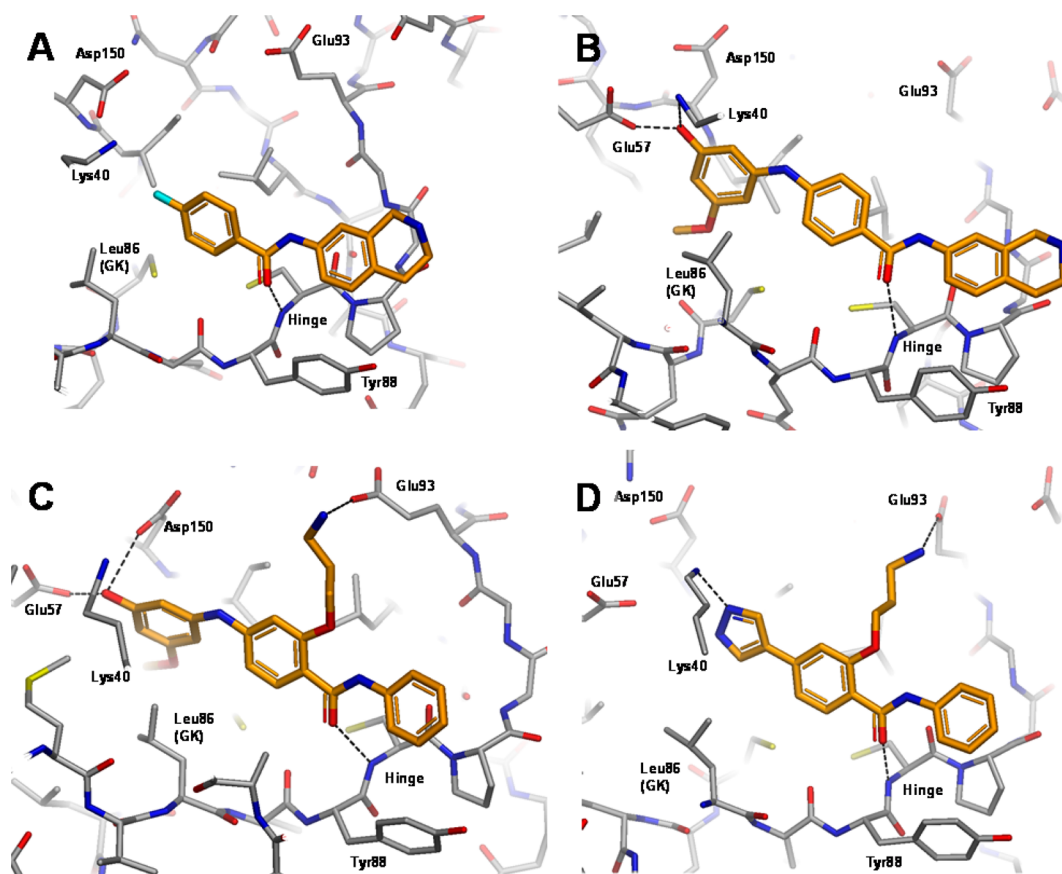


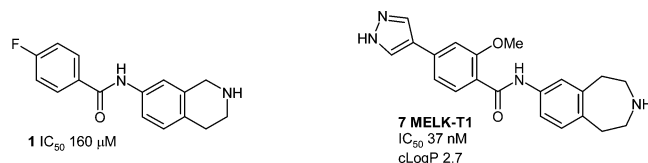
Figure 1. X-ray crystal structures of *trans*-amide ligands bound to MELK: (A) compound 1; (B) compound 2; (C) compound 3; (D) compound 4 (see Table 1); with hydrogen bond and salt bridge interactions shown as dotted lines.

protein–ligand complexes (see Supporting Information). Notably, crystal structures for MELK have recently been published.^{25,28}

Approximately 1500 compounds constituting an Astex fragment library were screened using ligand observed nuclear magnetic resonance (NMR) spectroscopy and protein thermal shift (T_m). Hits from both biophysical assays were progressed to X-ray crystallography, from which a large number of structurally validated hinge-binding compounds were obtained. Multiple factors are taken into account when selecting fragment hits for progression to hit-to-lead chemistry, for example, high ligand efficiency (LE)²⁹ and the presence of suitable growth vectors. However, in this case, it was also important to obtain high selectivity over other kinases in order to obtain a good quality chemical probe. Furthermore, selectivity versus closely related AMPK (60% identity in kinase domain) was highly desirable since off-target inhibition of AMPK has been reported as the cause of cardiotoxicity induced by sunitinib.³⁰ We hypothesized that a hinge-binding fragment with an atypical binding motif might provide an inherent bias toward selective inhibition during the optimization process compared with a typical “frequent-hitter” kinase hinge binding motif. An atypical hinge binder was also thought to be desirable in terms of exploring highly novel chemical space for kinase inhibition.

Here we describe the efficient optimization of a secondary amide hinge binder **1** leading to a potent and selective MELK inhibitor **7**, designated MELK-T1. The binding mode of **1** from the X-ray crystal structure is shown in Figure 1A. Notably, the amide binds in the *E*- or *trans*-form (a rarely precedented binding mode), and this key feature distinguishes it from cyclic

amide- and pyridone-containing hinge binders previously reported. One exception is a series of p38 inhibitors,³¹ though these molecules are structurally distinct from the series derived from **1**.



Another key feature of the binding mode is the movement of the side chain position of Tyr88 in the hinge region, which was not seen in structures of more typical hinge binding fragments identified in the MELK fragment screen, or in the published crystal structures.^{25,28} Examination of Protein Data Bank kinase structures showed that this residue movement was rare and hence might be an important contributor to obtaining a beneficial selectivity profile. The structure shows a DFG-in conformation of the kinase activation loop; hence, **1** binds as a Type I inhibitor. The benzo ring of the tetrahydroisoquinoline (THIQ) moiety forms an edge-to-face interaction with the displaced Tyr88, while the unsaturated ring is oriented toward the solvent exposed region of the ATP binding cleft, and the potentially charged cyclic amine therein provides a means of conferring favorable aqueous solubility for the chemotype. Although **1** (IC₅₀ 160 μM, LE 0.26) was by no means the most ligand efficient hit obtained from the fragment screen, the novel features described above made it an attractive starting point for testing our selectivity hypothesis.

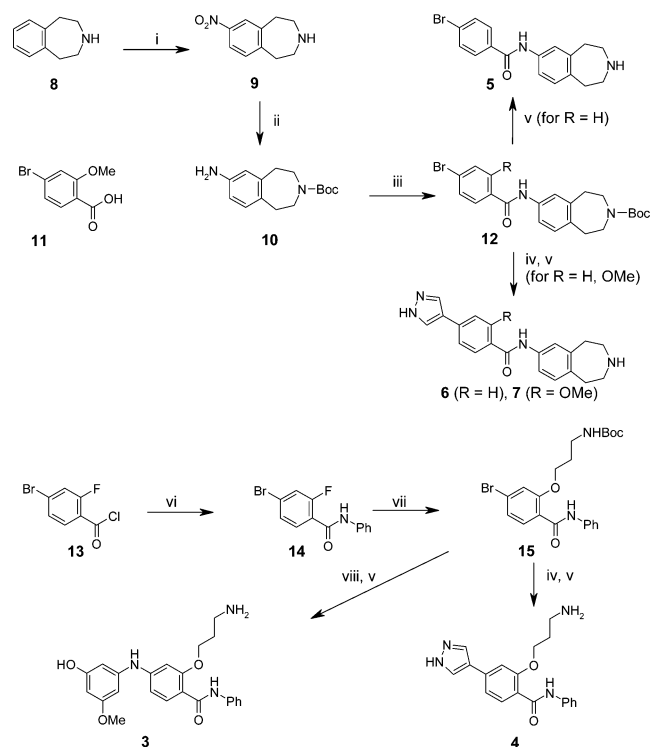
Table 1. Structure-Based Optimization of 1

Cpd	Structure	MELK IC ₅₀ μM ^a	LE
1		160	0.26
2		2.1	0.27
3		4.2	0.24
4		1.0	0.33
5		31	0.29
6		0.41	0.35
7		0.037	0.38

^aAssay details described in Supporting Information.

Compounds 3–7 (Table 1) were prepared as illustrated in Scheme 1 (for experimental details and compound characterization, see Supporting Information). Nitration of benzazepine **8** gave **9**, which was Boc-protected and then reduced by catalytic hydrogenation to give the protected bicyclic aniline **10**. Acylation gave amides **12**, which were then subjected to palladium-mediated cross-coupling with an optionally protected pyrazole boronate ester, followed by acid mediated deprotection to give final compounds **6** and **7**. Alternatively **12** (R = H) could be deprotected directly to give bromobenzamide **5**. Synthesis of aminopropoxy derivatives **3** and **4** started from 2-fluoro-4-bromobenzoyl chloride **13**. Amide coupling gave anilide **14**, then nucleophilic substitution of the 2-fluoro substituent with the appropriately protected amino-alcohol under basic conditions gave the required aryl alkyl ether **15**. Introduction of the pyrazole and final deprotection proceeded as previously described to afford compound **4**. The *N*-linked phenol derivative **3** was also synthesized from **15** using a palladium catalyzed coupling with 3-amino-5-methoxyphenol and subsequent deprotection. An analogous two step sequence starting from *N*-1,2,3,4-tetrahydroisoquinolin-7-yl-4-bromobenzamide afforded compound **2**. Compounds were assessed for inhibition of MELK kinase activity using a radioactive filter binding assay (see Supporting Information).

Compound **1** provided good growth vectors toward the back pocket via conserved catalytic Lys40 and the DFG motif, a region that has been targeted successfully in kinase

Scheme 1. Preparation of Compounds 3–7^a

^aReagents and conditions: (i) 1. H₂SO₄ (aq), 1,4-dioxane, RT. 2. HNO₃, H₂SO₄, 0 °C; (ii) 1. (Boc)₂O, CH₂Cl₂, RT. 2. H₂, Pd/C, MeOH, RT; (iii) **11**, EDC, HOBT, Et₃N, CH₂Cl₂, RT; or 4-bromobenzoyl chloride, NEt₃, CH₂Cl₂, RT; (iv) 4-(4,4,5,5-tetramethyl-[1,3,2]dioxaborolan-2-yl)-pyrazole or 4-(4,4,5,5-tetramethyl-[1,3,2]dioxaborolan-2-yl)-pyrazole-1-carboxylic acid *tert*-butyl ester, Pd(PPh₃)₄, Na₂CO₃, H₂O, DME, reflux; (v) HCl, 1,4-dioxane, EtOAc, RT; (vi) PhNH₂, NEt₃, CH₂Cl₂, RT; (vii) *N*-Boc-3-hydroxy-1-propylamine, Cs₂CO₃, DMF, 60 °C; (viii) 3-amino-5-methoxyphenol, Pd₂dba₃, XPhos, K₃PO₄, DME, 140 °C. For yields, see experimental procedures in Supporting Information.

optimizations. Using C-4 of the benzamide moiety as a growth point, a range of groups bearing H-bond donors and/or acceptors was designed, with a view to forming interactions with polar residues and backbone amides in this area. Encouragingly, substituted phenol **2** (IC₅₀ 2.1 μM, LE 0.27; Figure 1B) gave an 80-fold improvement in potency compared to **1** while maintaining ligand efficiency. The phenol OH functions as both donor and acceptor, engaging both the catalytic lysine and Glu57 situated on the α-C helix. The methoxy group fills the lipophilic selectivity pocket adjacent to the gate-keeper residue, while the NH spacer is important in allowing the bulky disubstituted phenyl group to bypass the relatively large Leu86 gate-keeper.

An alternative growth strategy was explored with the goal of forming a salt bridge interaction with Glu93. A targeted set of basic side chains was designed, attached via an oxygen atom to the ortho position of the benzamide phenyl ring. The oxygen linker was deemed important in ensuring a coplanar arrangement of the aryl benzamide core as seen in the crystal structures of **1** and **2**, by virtue of forming an intramolecular H-bond to the adjacent amide NH, an arrangement that would not be possible in the case of an all-carbon linker. Aminopropoxy derivative **3** (IC₅₀ 4.2 μM, LE 0.24; Figure 1C) duly formed the desired salt bridge interaction, but this change resulted in a less

ligand efficient molecule than **2**. However, replacement of the substituted phenol group with 4-pyrazolyl **4** significantly improved potency and ligand efficiency (IC_{50} 1.0 μ M, LE 0.33). Pyrazole containing fragments have been reported as kinase hinge binders,³² so there was a perceived risk that the molecule would switch orientation and bind with the pyrazole adjacent to the hinge. However, the crystal structure (Figure 1D) clearly showed that such a reorientation did not occur; conversely, the binding mode of original hit **1** was recapitulated.

Further optimization was carried out in the solvent exposed region. Examination of the surface contacts around the THIQ moiety in the co-crystal structure of **1** showed that the saturated ring did not optimally fill the solvent exposed pocket. Docking studies suggested that expansion of the saturated ring to seven atoms would be advantageous, and accordingly, symmetrical benzazepine **5** was prepared. Gratifyingly this change was well tolerated (IC_{50} 31 μ M, LE 0.29), and the seven-membered ring showed excellent shape complementarity with the protein surface (Figure 2).

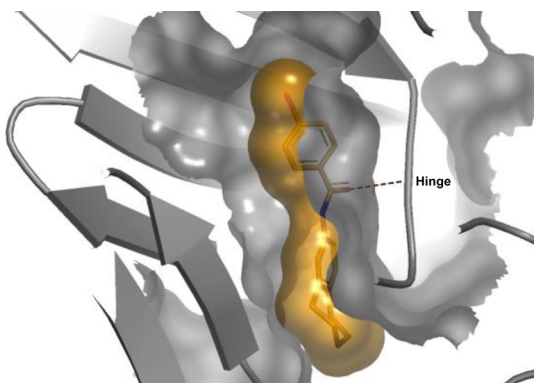


Figure 2. X-ray crystal structure of compound **5** bound to MELK viewed from the solvent exposed area, showing Connolly surfaces of ligand (orange) and protein (gray). The hydrogen bond between the amide carbonyl and the hinge is shown as a dotted line.

Replacement of the bromo substituent in **5** with 4-pyrazolyl then gave a 75-fold increase in potency, with **6** achieving a submicromolar IC_{50} value. A common feature observed in all crystal structures is the near-coplanar arrangement of the phenyl benzamide core. It is known from small molecule X-ray crystal structures that benzamides preferentially adopt a conformation where the amide is twisted by 30–40° out of the plane of the aryl ring.³³ It was reasoned that introduction of a methoxy substituent ortho to the amide carbonyl in **6** would stabilize the desired planar conformation, resulting in synthesis of compound **7**. This change resulted in a further 10-fold increase in affinity (IC_{50} 37 nM) and a further improvement in LE. Figure 3 shows the binding mode of **7**, wherein the pyrazole now forms a network of through-water H-bond interactions with Lys40 and Glu57.

Compound **7** (MELK-T1) satisfied our potency criteria for a chemical probe, but it was necessary to establish whether the selectivity profile was appropriate and also provide evidence that the molecule was able to enter cells and thereby access the target. Across a panel of 235 kinases, compound **7** showed >50% inhibition at 1 μ M against only six enzymes; of these, by far the most potent interactions were with Flt3 (IC_{50} 18 nM) and MELK (IC_{50} 23 nM) (Table 2). Importantly, no inhibitory

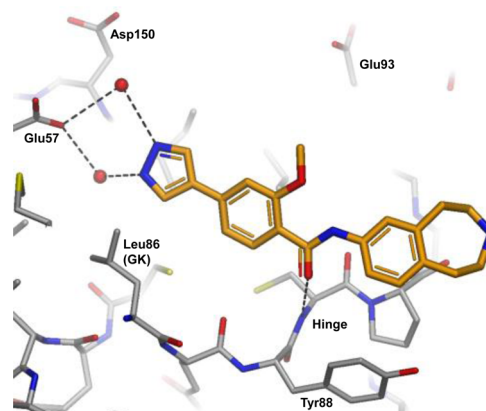


Figure 3. X-ray crystal structure of compound **7** bound to MELK. Hydrogen bond interactions are shown as dotted lines and water molecules as red spheres.

Table 2. Kinase Selectivity of MELK-T1 (7)

kinase ^a	% inh @ 1 μ M (IC_{50} nM)
MELK	97 (23)
Flt3	90 (18)
CAMKII δ	60 (810)
Mnk2	59 (760)
CAMKII γ	55 (1000)
MLCK	54 (1000)
AMPK α 1	5
AMPK α 2	-12

^aMillipore kinase panel; for details, see Supporting Information

activity was detected for the key antitarget AMPK, despite the close structural relationship with MELK.

The potent inhibition of Flt3 provided a means of testing the cell permeability of **7** using engineered Flt3-driven Ba/F3 cell lines.³⁴ In the Ba/F3-Flt3 assay, **7** showed inhibition of proliferation with IC_{50} 1.5 μ M in the absence of IL-3, while no inhibitory activity was observed in the presence of IL-3. In Ba/F3 cell lines transfected with either FGFR1, FGFR3, or KDR, compound **7** did not inhibit proliferation, either in the presence or absence of IL-3, thus confirming both intracellular target engagement and specificity for Flt3. Given such a demonstration of intracellular activity, compound **7** should therefore be a suitable chemical probe for elucidating cellular phenotypic effects arising from inhibition of MELK.

In conclusion, we have demonstrated that the choice of an atypical hinge binding motif as a starting point for FBDD can be a successful strategy for obtaining a kinase inhibitor with a good selectivity profile. In the case of MELK, fragment screening generated many high LE hits that fell into the category of “frequent hitter” hinge binding elements. Such frequent hitters are by definition predisposed to interact with multiple kinases, so here one may face the challenge of designing out off-target activity at each stage of the fragment growing process. The benefit of starting from a nonfrequent hitter is that selectivity is potentially built in at the outset, though there may be a trade-off in terms of lower on-target potency and/or LE, as in the MELK example here. A consequence of choosing a less ligand-efficient hinge binding fragment, such as **1**, is the need to identify several high energy interactions at multiple sites in the binding pocket. This was successfully achieved for MELK, hence a notable feature of the

optimization from **1** to **7** is the significant increase in LE with molecular size. The availability of multiple protein–ligand crystal structures enabled a highly efficient optimization process (~35 compounds prepared in total), the end result being the discovery of dual-selective and cell penetrant chemical probe MELK-T1 **7**, suitable for the further elucidation of MELK biology and as a starting point for further optimization.

■ ASSOCIATED CONTENT

■ Supporting Information

Assay details; structures of dorsomorphin and OTSSP167; fragment screen details; crystallographic details and PDB accession codes; experimental procedures for the preparation of novel compounds and intermediates. This material is available free of charge via the Internet at <http://pubs.acs.org>.

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■ Notes

The authors declare no competing financial interest.

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■ ABBREVIATIONS USED

DME, 1,2-dimethoxyethane; XPhos, 2-dicyclohexylphosphino-2',4',6'-triisopropylbiphenyl; Pd₂dba₃, tris(dibenzylideneacetone)dipalladium(0)

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