

The HIV-1 Nucleocapsid Protein Recruits Negatively Charged Lipids To Ensure Its Optimal Binding to Lipid Membranes

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ABSTRACT

The HIV-1 Gag polyprotein precursor composed of the matrix (MA), capsid (CA), nucleocapsid (NC), and p6 domains orchestrates virus assembly via interactions between MA and the cell plasma membrane (PM) on one hand and NC and the genomic RNA on the other hand. As the Gag precursor can adopt a bent conformation, a potential interaction of the NC domain with the PM cannot be excluded during Gag assembly at the PM. To investigate the possible interaction of NC with lipid membranes in the absence of any interference from the other domains of Gag, we quantitatively characterized by fluorescence spectroscopy the binding of the mature NC protein to large unilamellar vesicles (LUVs) used as membrane models. We found that NC, either in its free form or bound to an oligonucleotide, was binding with high affinity ($\sim 10^7 \text{ M}^{-1}$) to negatively charged LUVs. The number of NC binding sites, but not the binding constant, was observed to decrease with the percentage of negatively charged lipids in the LUV composition, suggesting that NC and NC/oligonucleotide complexes were able to recruit negatively charged lipids to ensure optimal binding. However, in contrast to MA, NC did not exhibit a preference for phosphatidylinositol-(4,5)-biphosphate. These results lead us to propose a modified Gag assembly model where the NC domain contributes to the initial binding of the bent form of Gag to the PM.

IMPORTANCE

The NC protein is a highly conserved nucleic acid binding protein that plays numerous key roles in HIV-1 replication. While accumulating evidence shows that NC either as a mature protein or as a domain of the Gag precursor also interacts with host proteins, only a few data are available on the possible interaction of NC with lipid membranes. Interestingly, during HIV-1 assembly, the Gag precursor is thought to adopt a bent conformation where the NC domain may interact with the plasma membrane. In this context, we quantitatively characterized the binding of NC, as a free protein or as a complex with nucleic acids, to lipid membranes and showed that the latter constitute a binding platform for NC. Taken together, our data suggest that the NC domain may play a role in the initial binding events of Gag to the plasma membrane during HIV-1 assembly.

The nucleocapsid protein (NC) of human immunodeficiency virus type 1 (HIV-1) is a small basic nucleic acid binding protein with two strictly conserved CCHC motifs that strongly bind zinc (1). Mature NC results from the protease-mediated cleavage of the Gag polyprotein precursor and plays key roles in HIV-1 replication (2–5). During the early steps, NC acts as a cofactor of the reverse transcriptase (RT), to promote the initiation of reverse transcription (6–8), as well as the two obligatory strand transfers (9, 10). Moreover, NC also reduces RT pauses at the initiation step (11–13), increases the overall RT processivity (14, 15), promotes synthesis through pause sites (16–19), helps to remove the RNA fragments resulting from the RT RNase H activity (20), and may contribute to the protection of the viral DNA (vDNA) during its nuclear import (21). Therefore, NC is thought to play a central role in the production of full-length vDNA as well as in its integration into the host cell genome (22–25).

The late steps of the viral life cycle are orchestrated by the Gag polyprotein precursor, formed of four major structural domains, namely, the matrix (MA), capsid (CA), nucleocapsid (NC), and p6. While the MA domain (GagMA) (26) targets Gag to the inner leaflet of the plasma membrane (PM) through a myristoyl group at its N terminus and a highly basic region (HBR) (27–31), the NC domain (GagNC) specifically interacts with the genomic RNA (gRNA) among the large excess of cellular RNAs. This GagNC/gRNA recognition constitutes a prerequisite for gRNA packaging (32–40) and dimerization (41–43). Interaction of GagNC with

gRNA is also important for the CA-mediated multimerization of Gag (44–50), albeit cellular mRNAs can replace gRNA in the formation of virus-like particles (51).

Deleting GagNC or replacing all its basic residues with neutral ones was reported to partially impair the anchoring and the multimerization of Gag at the PM (47, 52). This phenotype was observed when the mutated Gag was expressed either alone or coexpressed with wild-type Gag in the same cell. Though the role of GagNC in Gag oligomerization could be a consequence of its binding to RNAs, which act as a scaffold to direct Gag-Gag interactions, a nonexclusive hypothesis is that GagNC possibly contributes to the binding of Gag to the PM. Along this line, recombinant Gag molecules were found to adopt a bent globular

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conformation (53–55) and were proposed to interact with the gRNA through both their MA and NC domains (56–60). Moreover, in the absence of RNAs, Gag proteins in their bent conformation were shown to bind to negatively charged model membranes, suggesting that in addition to the MA domain, the NC domain may bind to membranes. Interestingly, RNAs have been shown to decrease the affinity of MA for lipid membranes (61, 62). Thus, RNAs and notably gRNA may act as negative regulators of nonspecific membrane binding, likely by reducing the electrostatic interactions of the HBR with acidic lipids and preventing myristate exposure (62, 63). Phosphatidylinositol-(4,5)-bisphosphate [PI(4,5)P₂] and probably other phosphoinositides (64, 65), which are essentially found on the cytoplasmic leaflet of the PM, can outcompete gRNA (61, 66) and promote the anchoring of the myristoylated MA domain to the PM (62, 67). The PI(4,5)P₂-induced release of GagMA from the gRNA is also thought to promote Gag trimerization and a structural switch of Gag to its rod-shaped conformation that is observed in virus particles (68–70).

In this context, as the gRNA decreases the affinity of GagMA for the membrane, it can be hypothesized that GagNC participates at least transiently, together with GagMA, to the initial binding of the Gag/gRNA complex to the PM. In order to further characterize the possible interaction of GagNC with lipid membranes, we used the mature NC protein, to avoid any interfering effects brought by the other domains of Gag, and large unilamellar vesicles (LUVs) as lipid membrane models. By using a combination of fluorescence spectroscopy techniques, we found a strong interaction of NC, in its free form and bound to nucleic acids, with negatively charged lipid membranes, but with no preferential binding to PI(4,5)P₂. Taken together, our data support a possible role of NC in the initial binding of Gag to the inner leaflet of the PM.

MATERIALS AND METHODS

Materials. All chemicals and solvents for spectroscopic measurements were from Sigma-Aldrich. Dioleoylphosphatidylcholine (DOPC), dioleoylphosphatidylethanolamine (DOPE), dioleoylphosphatidylserine (DOPS), dioleoylphosphatidylglycerol (DOPG), and phosphatidylinositol-4,5-bisphosphate [PI(4,5)P₂] were from Sigma-Aldrich. Rhodamine-DOPE (Rh-DOPE) was purchased from Avanti Polar Lipids (Birmingham, AL). The concentration of phospholipid stock solutions in chloroform was determined by dry weight. The HIV-1 unlabeled dTAR sequence and dTAR labeled with a fluorescein at its 3' end were synthesized and purified by high-pressure liquid chromatography (HPLC) by IBA GmbH (Germany). The concentration of dTAR was calculated from its absorbance measured with a Cary 400 spectrophotometer (Varian), using a molar extinction coefficient at 260 nm of 515,070 M⁻¹ · cm⁻¹ and at 488 nm of 80,000 M⁻¹ · cm⁻¹ for dTAR and dTAR-6-carboxyfluorescein (dTAR-FAM), respectively.

Preparation of peptides. NC peptides were prepared by solid-phase peptide synthesis on a 433A synthesizer (ABI, Foster City, CA), HPLC purified, and characterized by ion spray mass spectrometry, as previously described (71, 72). The MFL probe, a functionalized derivative of 4'-(dimethylamino)-3-hydroxyflavone, was prepared and coupled to the N terminus of the NC peptides, as previously described (73). The lissamine rhodamine dye (LRh) was synthesized as described previously (74) and coupled to the N terminus of the NC peptides in the same way as the MFL probe. To get the zinc-bound form of NC peptides, 2.2 molar equivalents of ZnSO₄ was added to the peptide and pH was raised to 7.4. Noticeably, in the case of MFL-labeled peptides, a large excess of Zn²⁺ ions should be avoided since this ion could affect its fluorescence. Peptide concentration was determined using an extinction coefficient of 5,700 M⁻¹ · cm⁻¹ at 280 nm for nonlabeled peptides, 33,000 M⁻¹ · cm⁻¹ at 400 nm for the MFL-

labeled peptides, and 80,000 M⁻¹ · cm⁻¹ at 555 nm for LRh-labeled NC. Note that the nucleic acid chaperone properties (75, 76) of the labeled NC peptide were tested as described previously (72) and found to be close to those of the unlabeled NC, indicating that the MFL label marginally perturbs the peptide structure and activity (data not shown).

Preparation of LUVs. Large unilamellar vesicles (LUVs) as well as LUVs labeled with 1% Rh-DOPE (LUVs-Rh), were prepared in 20 mM phosphate buffer, 150 mM NaCl, pH 7.4, by the classical extrusion method (77). Suspensions of multilamellar vesicles were extruded with a Lipex Biomembranes extruder (Vancouver, Canada), using first filters of 0.2 μm (7 passages) and thereafter 0.1 μm (10 passages). This protocol leads to LUVs with a mean diameter of 0.11 μm, as measured by dynamic light scattering (DLS) with a Malvern Zetamaster 300 (Malvern, United Kingdom). To calculate the vesicle concentration, the extended radius of the vesicles (*R*) was considered to be 535 Å, as determined by DLS measurements. The thickness of the lipid bilayer (*t*) and the average lipid density (*d*) were assumed to be 40 Å and 70 Å²/lipid, respectively (78). The number of lipids per vesicle was determined as $n = 4\pi[R^2 + (R - t)^2]/d = 9.54 \times 10^4$ lipids/vesicle, so that the vesicle concentration corresponds to $C(\text{vesicles}) = C(\text{lipids})/n$.

Fluorescence spectroscopy. Fluorescence spectra were recorded at 20°C on a FluoroMax or Fluorolog (Jobin Yvon Horiba) spectrofluorometer equipped with a thermostated cuvette holder. All spectra were corrected for the emission of the corresponding blank solution (neat solvent, lipid vesicles). Quantum yields (QY) were calculated, using 4'-(dialkylamino)-3-hydroxyflavone in ethanol (QY = 0.52 [79]), as a reference. To calculate the insertion depth of the MFL label in the lipid bilayer when the MFL-labeled NC peptides were bound to LUVs, the parallax quenching method using nitroxide lipids was used, as described previously (80–82). To determine the NC/LUV binding parameters, 0.1 μM MFL-NC was titrated with increasing concentrations of LUVs at 20°C. To determine the binding parameters of the MFL-NC/dTAR (3:1) complexes to LUVs, the complexes were prepared first by mixing 0.1 μM MFL-NC with 33 nM dTAR and then titrated with increasing concentrations of LUVs at 20°C.

The average number *ν* of moles of MFL-NC bound per mole of LUVs is calculated from the fluorescence intensities using $\nu = (I_0 - I)/(I_0 - I_t)P_t/L_t$, where *P_t* and *L_t* designate the total concentration of MFL-NC and LUVs (calculated as indicated above), respectively, and *I_t* designates the fluorescence at the plateau when all the peptide is bound, whereas *I₀* and *I* correspond to the fluorescence intensities of the peptide in the absence and in the presence of a given concentration of LUVs, respectively. Furthermore, the free peptide concentration *P* is deduced by $P = P_t - \nu L_t$. The same calculations are used for the binding of MFL-NC/dTAR complexes to LUVs.

To further demonstrate the interaction between the binding partners in the ternary complex NC/dTAR/LUVs, fluorescence resonance energy transfer (FRET) experiments were performed. In the FRET approach, an excited fluorophore, acting as a FRET donor, transfers nonradiatively its energy to an acceptor through dipole-dipole coupling. The efficiency of this energy transfer is inversely proportional to the sixth power of the distance between the donor and acceptor, making FRET extremely sensitive to small changes in distance. Due to the short distances where FRET applies (<8 nm), this method is used to evidence the interaction between two molecular species. The interactions between our partners were monitored two at a time, so that three combinations were used, namely, MFL-NC with LUVs-Rh, dTAR-FAM with LRh-NC, and dTAR-FAM with LUVs-Rh. The NC/dTAR (3:1) complexes (0.3 μM NC added to 0.1 μM dTAR) were prepared first and then added to an equivalent volume of a LUV solution (100 μM in lipid concentration). Excitation wavelengths were 400 nm and 480 nm for MFL-NC and dTAR-FAM, respectively. Emission spectra were recorded between 420 nm and 700 nm for the MFL/Rh couple and between 500 and 700 nm for the FAM/Rh and FAM/LRh couples. All experiments were performed in 20 mM phosphate buffer in the presence of 150 mM NaCl, pH 7.4, to limit nonspecific electrostatic interactions.

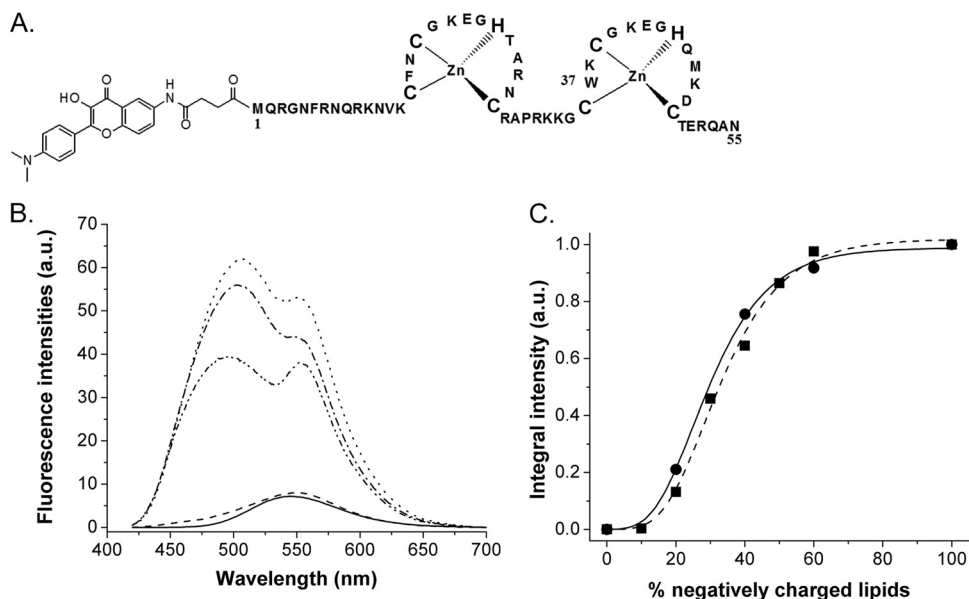


FIG 1 Binding of NC-MFL to LUVs, taken as a membrane model. (A) Structure of the NC peptide N terminally labeled with the MFL probe. (B) Fluorescence spectra of MFL-NC in the absence (solid line) and in the presence of LUVs composed of either neutral DOPC (dashed line), negatively charged DOPS (line with alternating dashes and dots) or DOPG (dotted line), or LUV_{mix1}, a lipid mixture composed of DOPC-DOPE-DOPS-SM-PI(4,5)P₂ at a ratio of 16:46:25:8:5 (mol% of lipids) (line with dashes and double dots). Concentrations of NC and lipids were 0.25 μ M and 100 μ M, respectively. (C) Dependence of MFL-NC/LUV binding on the percentage of negatively charged DOPS (■) or DOPG (●) lipids in mixture with DOPC, as monitored by the integral fluorescence intensity of the label. The data points were fitted by a sigmoidal function (solid and dashed lines) to guide the eye. The concentration of peptide was 0.3 μ M. The concentrations of DOPC/DOPS and DOPC/DOPG lipids were 200 and 100 μ M, respectively. Experiments were performed in 20 mM phosphate buffer, pH 7.4, in the presence of 150 mM NaCl to limit nonspecific electrostatic interactions. Excitation wavelength was 400 nm.

RESULTS

Fluorescence assay to monitor the binding of NC to model membranes. To show and characterize the binding of NC to membranes, we used a synthetic NC protein covalently labeled at its N terminus by MFL (73, 83), an environment-sensitive probe of the 3-hydroxyflavone family (Fig. 1A), and large unilamellar vesicles (LUVs) of 100 nm in diameter as membrane models. Due to an excited-state intramolecular proton transfer reaction, the MFL probe is characterized by two excited-state species, namely, the normal and tautomeric forms (N* and T*), showing two well-separated emission bands, differently sensitive to the polarity and hydration of the probe environment (73, 79, 84).

While the labeled NC showed a single emission band (Fig. 1B) similar to that of the free MFL label in water (73), negatively charged LUVs, composed of either DOPS or DOPG phospholipids added at a molar ratio of 400 lipids per MFL-NC peptide, caused a dual emission (Fig. 1B) with a strong increase (\sim 8.5-fold) in quantum yield. These important changes in the MFL emission can be ascribed to the transfer of the MFL probe from the polar environment of the buffer to the aprotic environment of the hydrophobic membrane. These data clearly show that the MFL-NC peptide binds to negatively charged LUVs, irrespective of the nature of the lipid polar head group. In sharp contrast, neutral DOPC LUVs did not significantly affect the MFL-NC fluorescence spectrum (Fig. 1B), suggesting that NC does not bind to neutral lipid vesicles. To further correlate NC binding to LUVs with the presence of negatively charged lipids, we increased the molar fraction of DOPS to DOPC and monitored the binding of these LUVs to MFL-NC through the changes in the probe fluorescence intensity (Fig. 1C). While no change in MFL intensity was

observed with a molar fraction of 10%, a sharp increase was observed when the molar ratio was increased from 20 to 60 mol%, where a plateau was reached. The dependence of NC binding on the molar fraction of PS (Fig. 1C) was similar to the one reported for the MA protein (85) or the Gag protein (86). This indicates that, similarly to MA or Gag (85, 86), electrostatic interactions play an important role in the binding of the positively charged NC protein to negatively charged lipids. This conclusion is further substantiated by the strong analogy of MFL-NC with MFL-labeled poly-L-lysine that was reported to bind through electrostatic interactions to negatively charged LUVs and to induce similar changes in MFL fluorescence parameters (73). Interestingly, LUVs composed of a mixture of lipids, DOPC-DOPE-DOPS-SM-PI(4,5)P₂, in a ratio of 16:46:25:8:5 (mol% of lipids), which mimics the content in negatively charged phospholipids of the inner leaflet of the PM (87–90), led to changes in the MFL-NC spectrum, close to those observed upon addition of negatively charged LUVs (Fig. 1B). The only difference between these LUVs (called LUV_{mix1}) and pure DOPS LUVs is their somewhat lower quantum yield, probably due to their lower content (30%) in negatively charged lipids. Thus, it appears that NC can probably bind to the inner leaflet of the PM through electrostatic interactions with negatively charged lipids.

Determination of the NC binding parameters to lipid membranes. In order to determine the binding parameters of NC to LUVs, we titrated MFL-NC using increasing concentrations of LUVs with different lipid compositions (Fig. 2A). To fit the binding curves, we first used an equation that considers the binding of NC to lipid bilayers to be governed by a molar partition coefficient K_m :

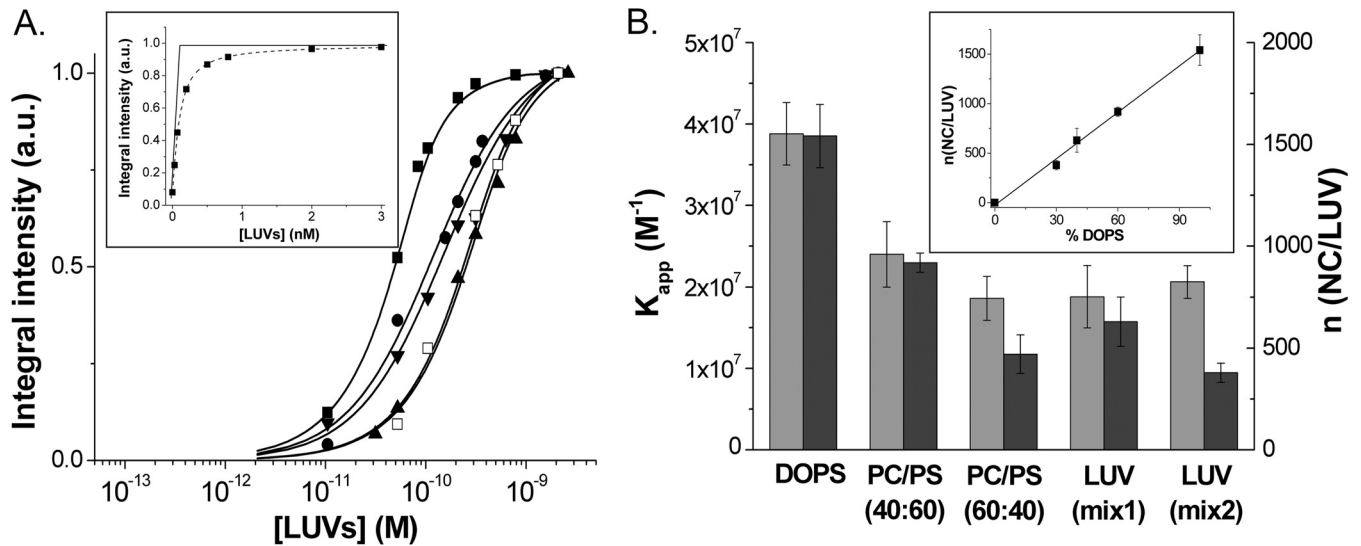


FIG 2 Determination of the binding parameters of NC protein to model lipid membranes. (A) Titration of MFL-NC (0.1 μM) with LUVs of different lipid compositions: DOPS (■), DOPC-DOPS (40:60) (●), DOPC-DOPS (60:40) (▲), DOPC-DOPE-DOPS-SM-PI(4,5)P₂ (16:46:25:8:5) (LUV_{mix1}, ▼), and DOPC-DOPE-DOPS-SM (16:46:30:8) (LUV_{mix2}, □). Solid lines correspond to the fit of the data points with equation 2, using the binding constant and the number of binding sites per LUVs indicated in panel B. (Inset) Determination of the number of binding sites from the intercept of the tangent to the first points of the titration with the tangent to the fluorescence plateau. (B) Binding affinity (light gray) and number of binding sites per LUV (dark gray) as a function of the LUV composition. The values of K_{app} and n were determined from the binding curves in panel A. (Inset) Linear dependence of the number of binding sites as a function of the percentage of negatively charged lipids in the LUV composition. All experiments were in 20 mM phosphate buffer, 150 mM NaCl, pH 7.4. Excitation wavelength was 400 nm.

$$\frac{P_m}{P_t} = \frac{(I - I_0)}{(I_f - I_0)} = \frac{K_m[L']}{1 + K_m[L']} \quad (1)$$

where P_m and P_t are the concentrations of NC protein bound to the membrane and total protein, respectively. I_0 , I , and I_f are the fluorescence intensities in the absence of lipid, in the presence of a given lipid concentration, and in the presence of a saturating lipid concentration, respectively. Assuming that NC can access only lipids at the external surface of the vesicles and that the concentration of bound NC is negligible compared to that of accessible lipids, the concentration of free lipids can be approximated by the equation $[L'] = [L']_t/2$, where $[L']_t$ is the total concentration of lipids. Using this approximation, equation 1 was used to fit the binding curves of MA to lipid bilayers (85, 91). Using the same equation, we found that the K_m value of NC for DOPS LUVs was $(4.5 \pm 0.7) \times 10^5 \text{ M}^{-1}$ (Table 1) and, thus, was close to the affinity

$[(3.5 \pm 1) \times 10^5 \text{ M}^{-1}]$ reported for the binding of MA to the same kind of LUVs (92). Due to this rather high affinity, it is likely that the concentration of bound peptide is probably not negligible in comparison to $[L']_t$, especially for the first points of the titration, where the concentration of peptide is higher than the concentration of LUVs, so that the approximation $[L'] = [L']_t/2$ is not fully valid. Moreover, equation 1 is formally equivalent to the equation describing a classical protein-ligand interaction with a one-to-one stoichiometry (93, 94). Due to its multiple charges and its large area compared to that of lipid head groups, it is unlikely that NC binds to only one lipid in a LUV. In fact, the real entities that bind NC are the lipid vesicles. Assuming that each vesicle can bind a number n of NC molecules with an apparent binding constant K_{app} , we can use the equation that we derived for the binding of NC to oligonucleotides, which is based on the same binding model (95, 96):

$$I = I_0 - \frac{(I_0 - I_f)}{P_t} \times \frac{[1 + (P_t + nL_t)K_{\text{app}}] - \sqrt{[1 + (P_t + nL_t)K_{\text{app}}]^2 - 4P_t nL_t K_{\text{app}}^2}}{2K_{\text{app}}} \quad (2)$$

where P_t is the total concentration of peptide and L_t is the total concentration of LUVs, calculated as indicated in Materials and Methods. This equation does not make any approximation on the concentration of the free species and allows a direct fitting of the fluorescence intensity I of the labeled protein. The number of binding sites n was determined from the intercept of the tangent to the initial points of the curve with the tangent to the plateau value (Fig. 2A, inset). For DOPS LUVs, we obtained a binding stoichiometry of $1,500 (\pm 150)$ NC protein molecules per LUV, which corresponds to about $29 (\pm 3)$ lipids per NC molecule. By taking

TABLE 1 Comparison of K_m and K_{app} values obtained for the binding of NC derivatives to LUVs of different compositions^a

Peptide	LUV composition	K_m (10^{-5} M^{-1})	K_{app} (10^{-5} M^{-1})
NC	DOPS	4.5 (± 0.7)	390 (± 20)
	DOPC/DOPS (40:60)	2.0 (± 0.1)	220 (± 70)
	DOPC/DOPS (60:40)	0.9 (± 0.1)	180 (± 10)
	LUV _{mix1}	1.4 (± 0.2)	170 (± 30)
	LUV _{mix2}	0.7 (± 0.1)	200 (± 20)
NC _{H23C}	DOPC/DOPS (40:60)	2.2 (± 0.2)	220 (± 20)
NC _{SSH5}	DOPC/DOPS (40:60)	2.6 (± 0.3)	160 (± 20)
NC(11–55)	DOPC/DOPS (40:60)	0.30 (± 0.02)	20 (± 1)

^a K_m and K_{app} values were obtained by fitting the titration curves of MFL-labeled NC derivatives by the LUVs of indicated compositions, using equations 1 and 2, respectively. The compositions of LUV_{mix1} and LUV_{mix2} were DOPC-DOPE-DOPS-SM-PI(4,5)P₂ (16:46:25:8:5) and DOPC-DOPE-DOPS-SM (16:46:30:8), respectively.

70 Å² as the surface of a lipid, each peptide thus occupies about 20 nm² on the vesicle surface. As this occupied surface is substantially larger than the ~4-nm² area calculated from the largest section (3.2 × 1.5 nm) of the three-dimensional (3D) structure of NC (PDB 1ESK), this suggests that NC molecules do not fully coat the LUV surface, probably as a consequence of the repulsive electrostatic forces between the nonneutralized charges of the NC molecules bound to the lipid surface. Using the stoichiometry value obtained from the inset of Fig. 2A, we determined from the fit of the binding curves with equation 2 a K_{app} value of $(3.9 \pm 0.9) \times 10^7 \text{ M}^{-1}$ (Fig. 2B; Table 1) for each NC binding site on the DOPS vesicles. As this K_{app} value was obtained through the same formalism as that applied for the binding of NC to nucleic acids (95, 96), a direct comparison is thus possible and shows that the affinity of NC for negatively charged lipids is of the same order as the affinity of NC to nucleic acids (97).

Interestingly, addition of an increasing proportion of DOPC to DOPS LUVs was found to sharply decrease the number of NC binding sites per LUV to 920 (± 50), 470 (± 90), and 630 (± 100) for DOPC-DOPS (40:60), DOPC-DOPS (60:40), and LUV_{mix1}, respectively (Fig. 2B). In contrast, the binding affinities were only slightly modified, since we observed only a 2-fold decrease between pure DOPS LUVs and LUV_{mix1}, which corresponds to a negligible change of the binding energy ($< 0.5 \text{ kcal} \cdot \text{mol}^{-1}$). Since GagMA binds to the PM via an electrostatic interaction between its HBR domain and negatively charged lipids and also through specific recognition of PI(4,5)P₂, we investigated whether NC could also specifically interact with this lipid. To this aim, we replaced 5% PI(4,5)P₂ with 5% DOPS in the composition of LUV_{mix1}, so that the total percentage of DOPS becomes 30%. Thus, by using LUV_{mix2}, composed of DOPC-DOPE-DOPS-SM (16:46:30:8), the number of binding sites decreased to 380 (± 50) NC/LUV with no significant change in the binding affinity (Fig. 2B). As PI(4,5)P₂ molecules are characterized by 3 negative charges versus only one for DOPS, one PI(4,5)P₂ molecule may be equivalent to 3 DOPS molecules in the binding process. Therefore, the real percentage of negative charges in LUV_{mix1} that contains 5% PI(4,5)P₂ and 25% DOPS is 40%, explaining the similarities with the DOPC-DOPS (60:40) composition. Moreover, the negligible changes in the K_{app} values upon the replacement of PI(4,5)P₂ by DOPS further suggested that in contrast to myristylated MA (91), NC does not show a preferential binding for PI(4,5)P₂. Together with the limited changes in the K_{app} values, a linear correlation was observed between the number of binding sites and the percentage of negatively charged lipids in the LUVs (Fig. 2B, inset), strongly suggesting that NC can recruit negatively charged lipids to ensure a tight binding to LUVs.

Taken together, our data indicate that NC binds with high affinity to lipid membranes, supporting the notion that GagNC could participate, at least transiently, in the initial binding of Gag to the PM.

Determinants of NC binding to lipid membranes. In order to determine which regions of NC are instrumental for membrane binding, we used three NC mutants (Fig. 3A). The role of the basic unfolded N-terminal (amino acids [aa] 1 to 10) domain was evaluated with the NC(11–55) derivative. The role of the central folded zinc finger domain, which is critical for a number of NC functions (5, 95, 98–100), was evaluated with the NC_{SSHS} mutant, where all cysteine residues were replaced by serines to prevent the coordination of zinc ions and therefore the folding of the fingers

(10). Finally, to examine the contribution of the hydrophobic plateau at the top of the central finger motif, we used the NC_{H23C} mutant, where the His23 residue is replaced by a Cys residue. This substitution induces a misfolding of the proximal finger motif that prevents formation of the hydrophobic plateau (101).

To monitor their interaction with LUVs, the NC mutants were N terminally labeled with the MFL probe. As for the labeled native NC, addition of a large excess of negatively charged LUVs (200 μM in lipids) to all labeled mutants induced a two-band emission and strongly increased their quantum yield, indicating that all peptides were able to bind to these LUVs (Fig. 3B). Interestingly, the N^*/T^* ratio values were significantly lower with the NC_{H23C} ($I_{N^*}/I_{T^*} = 1.07$) and NC_{SSHS} ($I_{N^*}/I_{T^*} = 1.14$) mutants than with the native protein ($I_{N^*}/I_{T^*} = 1.27$), indicating that the MFL probe linked to the amino acid at position 1 experienced a less polar environment in these mutants than in native NC, upon binding to LUVs. This suggests that the position of the NC N-terminal (aa 1 to 10) domain on LUVs depends on the proper folding of the fingers. In contrast, the fluorescence parameters of the labeled NC(11–55) peptide ($I_{N^*}/I_{T^*} = 1.24$) were indistinguishable from those of MFL-NC, indicating that the amino acids at position 11 in NC(11–55) and at position 1 in NC(1–55) experience a similar environment at the membrane.

To further determine the localization of the NC probe in the membrane, parallax measurements were performed to determine the depth of the label from the center of the bilayer (73). This distance is recovered through the amount of quenching of the MFL emission observed with lipids carrying fluorescence quenchers at different depths in the bilayer. Using DOPC-DOPS (40:60) vesicles, we obtained average depths of 16.2 to 17.7 Å from the center of the bilayer for both MFL-NC and the three NC mutants (data not shown). These values pointed to a shallow location of the probe close to the lipid head groups for all four NC peptides. However, the accuracy of this technique is not sufficient to show binding depth differences that could explain the differences in the N^*/T^* ratio values observed for the NC_{H23C} and NC_{SSHS} mutants compared to the native protein.

Next, we compared the binding of the four labeled NC peptides to LUVs containing an increasing fraction of negatively charged lipids (Fig. 3C). The binding of the NC mutants to LUVs was found to strongly rely on the percentage of negatively charged lipids. In fact, as for the native NC, an optimal binding for NC_{H23C} and NC_{SSHS} peptides was reached with LUVs containing 60% negatively charged lipids. In contrast, NC(11–55) needs a higher percentage of negatively charged lipids, suggesting that it probably binds to LUVs with a lower affinity. By investigating the binding parameters of the NC mutants to DOPC-DOPS (40:60) LUVs, we found that the MFL-labeled NC, NC_{H23C}, and NC_{SSHS} showed nearly indistinguishable binding parameters. This indicates that the proper folding of the zinc fingers, as well as the hydrophobic plateau at the top of the zinc fingers, is not essential for the interaction of NC with the lipids. Nevertheless, the differences in polarity of the MFL probe observed in Fig. 3B suggest that the properly folded fingers may control the position of the N-terminal domain of NC in the membrane. Moreover, as the binding is thought to be mainly electrostatic, it can be speculated that the numerous positively charged amino acids distributed along the NC sequence indifferently intervene in the binding to the lipids, so that their number is more important than their localization in the sequence. In agreement with this conclusion, NC(11–55), which

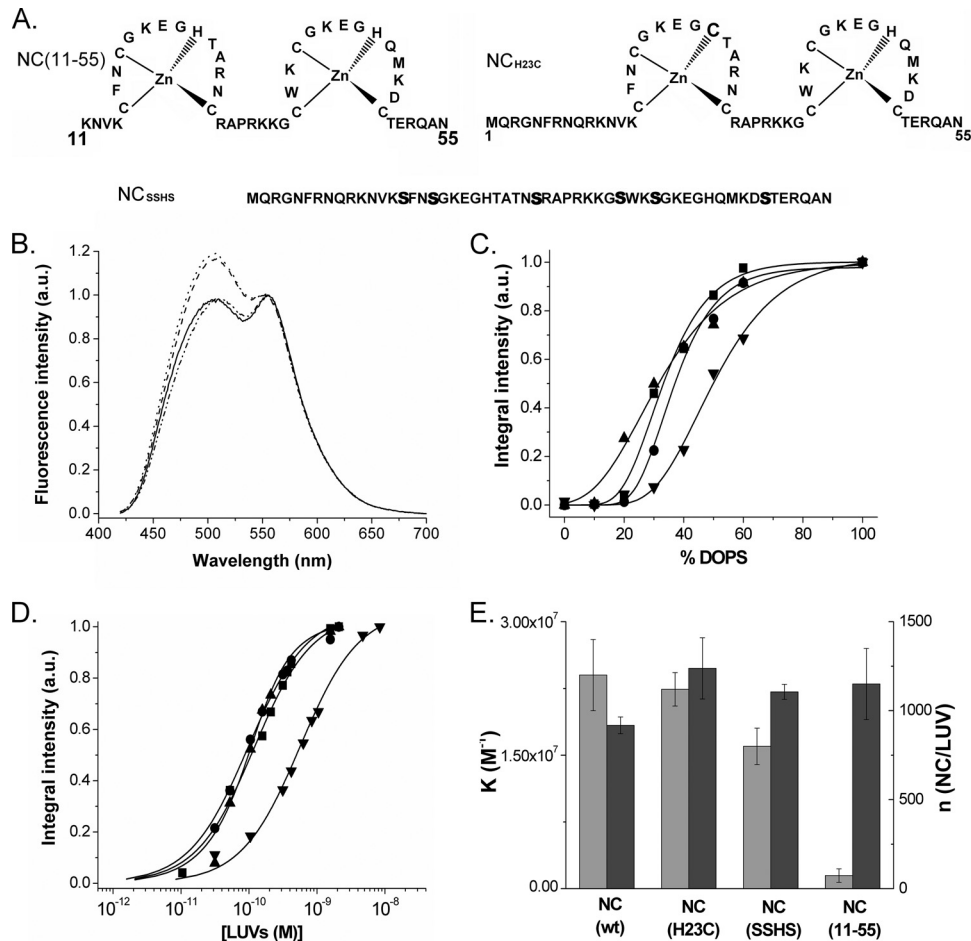


FIG 3 Interaction of NC mutants with LUVs. (A) Sequences of the NC mutants. Mutations are shown in bold. (B) Fluorescence spectra of MFL-labeled NC (dashed line), NC_{H23C} (line with dashes and double dots), NC_{SSHs} (solid line), and NC(11-55) (dotted line) peptides in the presence of DOPS LUVs. Concentrations of peptides and lipids were 0.25 and 200 μ M, respectively. (C) Dependence of the binding of MFL-labeled NC (■), NC_{H23C} (●), NC_{SSHs} (▲), and NC(11-55) (▼) peptides on the percentage of negatively charged DOPS lipids in DOPC-DOPS LUVs. The data points were fitted by a sigmoidal function (solid lines) to guide the eye. Concentrations of peptides and LUVs were 0.25 and 200 μ M, respectively. (D) Titrations of MFL-labeled NC, NC_{H23C}, NC_{SSHs}, and NC(11-55) peptides (0.1 μ M) with DOPC-DOPS (40:60) LUVs. Symbols are as in panel C. The number of binding sites per LUV and the binding affinities were obtained as described for Fig. 2A. Solid lines correspond to the fit of the data points with equation 2, using the parameters given in panel E. (E) Binding parameters of NC mutants to DOPC-DOPS (40:60) LUVs (K_{app} in light gray, n in dark gray). Buffer and excitation wavelength were as in Fig. 2.

contains three basic residues fewer than NC(1-55), shows a >10-fold decrease in its binding constant (Fig. 3D and E).

Binding properties of NC/DNA complexes to lipid membranes. Since the functions of NC and GagNC in virus replication are mediated by an interplay with viral nucleic acids (97), we studied the possible binding of NC/nucleic acid complexes to lipid membranes. As a nucleic acid model sequence, we used the dTAR 55-nucleotide (nt) sequence corresponding to the DNA version of the HIV-1 RNA transactivation response element. To show the binding of the NC/dTAR (3:1) complex to lipid membranes, we first performed a series of FRET experiments by monitoring the interactions between two partners at a time. As we have three partners, the FRET efficiencies of the three combinations of two labeled partners were determined in the presence or the absence of the third nonlabeled partner. As both the interactions of NC with LUVs (this work) and with nucleic acids (95, 102, 103) are dependent on electrostatic interactions, we systematically investigated the FRET efficiencies at two salt concentrations (30 mM and 150

mM NaCl) (Table 2). For the interactions between MFL-NC and DOPS LUVs labeled by 1% Rh-DOPE (LUV-Rh), we observed a strong FRET efficiency (83 to 90%) that was slightly affected by the addition of nonlabeled dTAR or by the salt concentration. These data indicate that NC is bound to LUVs both in the absence and in the presence of dTAR, being close enough to the Rh-DOPE molecules to give a high FRET efficiency. It can also be concluded that dTAR does not significantly outcompete NC from the LUVs to form binary dTAR/NC complexes.

Next, we monitored the FRET efficiency of dTAR labeled with fluorescein at its 3' end (dTAR-FAM) to LUV-Rh in the absence and the presence of nonlabeled NC. As expected, due to the inability of the negatively charged oligonucleotide to bind to negatively charged lipid vesicles (104), we did not observe any FRET in the absence of NC. In contrast, a significant FRET efficiency (9% at 150 mM NaCl and 27% at 30 mM NaCl) was observed when NC was added, revealing the formation of a ternary NC/dTAR/LUV complex. Finally, we observed a strong FRET efficiency (>70%)

TABLE 2 FRET efficiencies of binary and ternary complexes between NC, dTAR, and LUVs

Monitored interaction ^a	Nonlabeled partner	FRET efficiency for NaCl concn:	
		150 mM	30 mM
MFL-NC/LUV-Rh	–	0.84	0.90
	dTAR	0.83	0.85
dTAR-FAM/LUV-Rh	–	0	0
	NC	0.09	0.27
dTAR-FAM/LRh-NC	–	0.72	0.80
	LUVs	0.37	0.62

^a FRET experiments were performed between two labeled partners in the absence (–) and in the presence of the third nonlabeled partner in 20 mM phosphate buffer with 150 mM or 30 mM NaCl (pH 7.4).

between dTAR-FAM and NC labeled at its N terminus by lissamine rhodamine (LRh-NC) in the binary NC/dTAR complex. Addition of nonlabeled LUVs to Rh-NC/dTAR-FAM was found to preserve a good FRET efficiency (37% at 150 mM NaCl and 62% at 30 mM NaCl). Taken together, our data showing a high FRET efficiency between MFL-NC and LUV-Rh in the presence of dTAR, as well as a high FRET efficiency between dTAR-FAM and LRh-NC in the presence of LUV, unambiguously confirmed the formation of NC/dTAR/LUV ternary complexes. Noticeably, the lower FRET efficiencies observed for dTAR-FAM/LUV-Rh and dTAR-FAM/LRh-NC in the ternary mixtures at 150 mM NaCl than at 30 mM NaCl suggest that fewer dTAR molecules may be bound to the LUVs, probably as a consequence of a limited coating of dTAR by NC molecules at the highest salt concentration. However, it cannot be excluded that the two salt concentrations induce different conformations of the NC/dTAR complexes bound to the LUVs.

Next, we determined the binding parameters of the MFL-NC/dTAR complexes to LUVs of various compositions (DOPS, DOPC-DOPS [40:60], and LUV_{mix1}) by titrating the complexes with increasing concentrations of LUVs in 20 mM phosphate buffer, 150 mM NaCl, pH 7.4 (Fig. 4A). For all three compositions of LUVs, the binding constants of the NC/dTAR complexes were found to be similar to those of NC but with a lower number of binding sites, as expected from the increased size of the NC/dTAR complexes compared to free NC (Fig. 4B). The reduction in the number of binding sites was limited to 20% for DOPS LUVs but up to 45% for LUV_{mix1}. Since, by analogy with free NC, only the number of binding sites, but not the affinity of the NC/dTAR complexes for the LUVs, was found to vary with the percentage of negatively charged lipids, it can be concluded that NC/dTAR complexes are also able to recruit negatively charged lipids in LUVs in order to optimize their binding.

Taken together, our data indicate that the NC/DNA complexes bind with high affinity to negatively charged lipid membranes, supporting the notion that the NC domain of Gag may interact with the inner leaflet of the PM and in so doing might participate in Gag assembly at the level of the PM.

DISCUSSION

The aim of the present work was to investigate the possible interactions of HIV-1 NC protein with lipid membranes. The possible binding of GagNC to membranes has already been evoked (58, 59, 105) or deduced from the cellular localization of Gag lacking the

NC domain (52), but the evidence of this interaction together with its characterization was still missing. To that end, we investigated the binding of fluorescently labeled synthetic NC, free or bound to nucleic acids, to LUVs with different lipid compositions, taken as model membranes. This was carried out with the mature NC in order to avoid any influence from the other domains of Gag.

Our results show that NC, free or bound to an oligonucleotide, binds with high affinity to negatively charged LUVs but not to neutral ones. The NC-membrane interaction is thought to be mostly electrostatic, which was confirmed by the analysis of the binding of NC mutants to LUVs. Though it cannot be excluded that the MFL probe contributes to the binding of the labeled peptides to LUVs, its contribution is likely limited as it is not sufficient to promote the binding of the labeled peptides to neutral LUVs. Importantly, while the number of binding sites was found to decrease with the percentage of negatively charged lipids in the LUV composition, the binding constant value was nearly independent of this percentage, suggesting that NC and NC-oligonucleotide

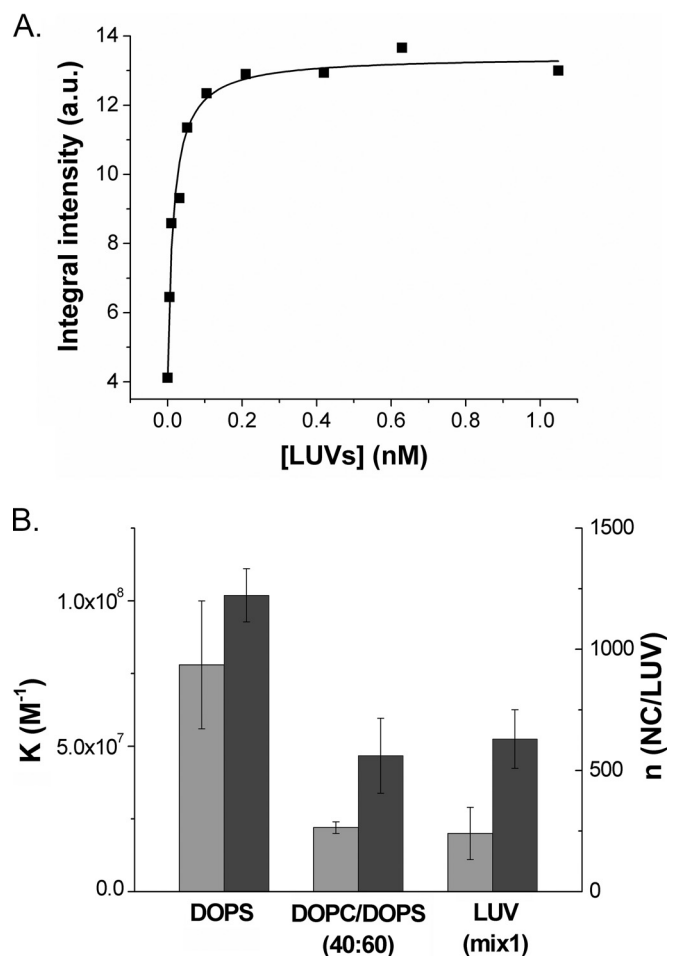


FIG 4 Determination of the binding parameters of the NC/dTAR complexes to model lipid membranes. (A) Titration curve of MFL-NC/dTAR (3:1) complexes with DOPS LUVs. The number of binding sites per LUV and the binding affinities were obtained as described for Fig. 2A. The solid line corresponds to the fit of the data points with equation 2. Concentrations of NC and dTAR were 100 nM and 33 nM, respectively. (B) Binding parameters of NC/dTAR complexes to DOPS LUV, DOPC/DOPS (40:60) LUVs, and LUV_{mix1} (K_{app} in light gray, n in dark gray). Buffer and excitation wavelength were as in Fig. 2.

complexes can recruit negatively charged lipids to ensure optimal binding. Such a recruitment of negatively charged lipids by NC is in line with the recent hypothesis that Gag multimerization at the PM could induce the formation of acidic lipid-enriched microdomains (ALEM) (106, 107).

Interestingly, the binding constant of NC for negatively charged lipids is similar to that of the interaction of unmyristoylated MA and Gag proteins with DOPS-containing LUVs (91, 92), which is mainly governed by electrostatic interactions through the HBR sequence of MA (59, 60, 85, 106). However, in contrast to the myristoylated GagMA domain (67, 91, 108), which binds avidly and specifically to PI(4,5)P₂ through a hydrophobic interaction with the exposed myristyl chain, no specific binding was found for the NC-PI(4,5)P₂ interaction. Therefore, it is likely that the affinity of NC for negatively charged lipids is substantially lower than the final affinity of the myristoylated MA domain for PI(4,5)P₂, following the myristyl switch. In agreement with this conclusion, mutating the N-terminal glycine of GagMA, to which the myristate is attached, impairs Gag-membrane association (27, 28, 109).

Another major difference between the MA and NC domains is their relative affinities for RNA and lipids. Indeed, due to its high affinity for the myristoylated Gag, PI(4,5)P₂ can outcompete RNA bound to the myristoylated GagMA domain (58, 61, 63, 66). In contrast, results in Fig. 4B indicate that PI(4,5)P₂ cannot dissociate the NC/nucleic acid complexes, suggesting that GagNC might be able to simultaneously bind RNA and the PM. This hypothesis is supported by the multivalent nature of NC, which allows it to bind at the same time two nucleic acid molecules (110). These differences between the relative stabilities of the different complexes, together with the ability of NC and NC/nucleic acid complexes to interact with negatively charged lipid membranes, lead us to revisit the HIV-1 assembly model, previously proposed by several groups (55, 61, 70, 111) (for reviews, see references 59, 60, and 63). In this revised model, we propose that the NC domain of Gag participates in a transient manner in the binding of Gag to the PM, allowing the recruitment of negatively charged lipids (Fig. 5).

This model stipulates that at the start of Gag assembly, the NC and MA domains are involved in the selection and recruitment of the gRNA (Fig. 5, step 1), which is facilitated by the globular or bent conformation of Gag (36, 55, 59, 61, 62, 69, 70, 112). At this stage, the binding of the gRNA to GagMA is thought to be mediated by the HBR sequence of Gag (61, 64), while the myristyl group is still buried in the GagMA core. Next, as NC bound to a nucleic acid can interact with negatively charged membranes (Fig. 4), the Gag-gRNA complex may bind to the cytoplasmic leaflet of the PM through its MA and NC domains. To strengthen its interaction with the PM, the gRNA-bound NC domain is thought to recruit negatively charged phospholipids, notably PI(4,5)P₂ and PS, in order to form an ALEM (Fig. 5, step 2). This NC-induced recruitment of ALEM is consistent with the reported ability of GagNC to accelerate and stabilize the binding of Gag to cellular membranes (113). Once recruited, PI(4,5)P₂ should induce the myristyl switch which stabilizes the anchoring of the GagMA into the PM. After the dissociation of the gRNA from the MA domain, the binding energy of the NC-gRNA complex for the PM is probably not sufficient to maintain the globular/bent conformation of Gag, so that the GagNC-gRNA complex is released from the membrane, allowing Gag to adopt an extended rod-shape conformation (Fig. 5, step 3) found in immature viral particles (114, 115).

This model proposes that the initial steps of HIV-1 assembly

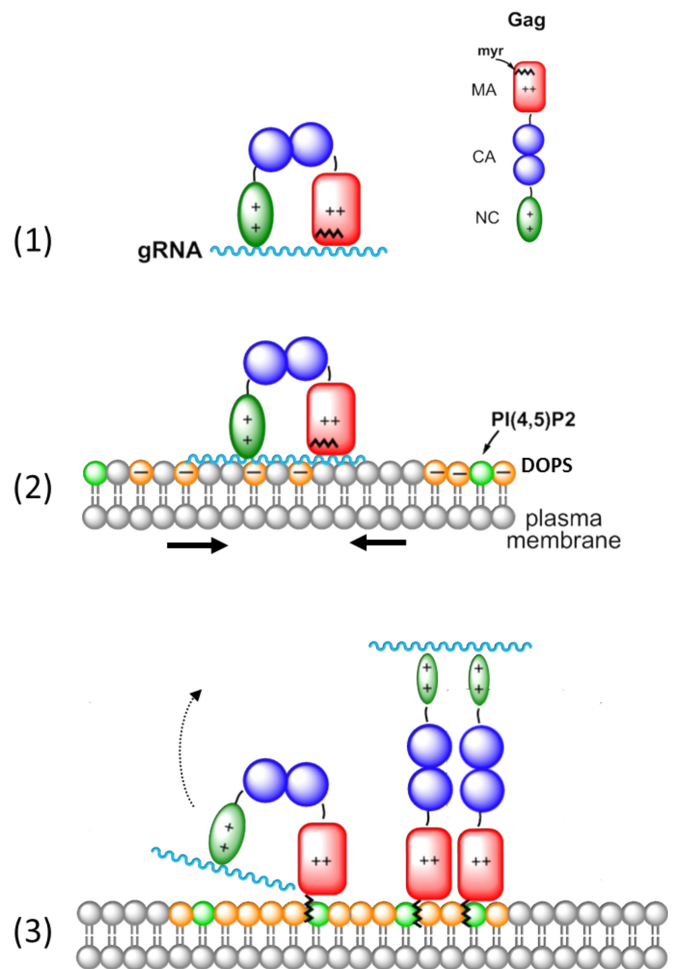


FIG 5 Proposed role of the GagNC domain in the binding of Gag to the plasma membrane. (1) Newly made Gag polyprotein binds in its bent conformation to the gRNA via its NC and MA domains. (2) The Gag-gRNA complex interacts with the plasma membrane via its MA and NC domains. (3) The NC domain recruits negatively charged lipids, including PI(4,5)P₂, that form an acidic lipid-enriched microdomain. This in turn leads to the binding of GagMA to PI(4,5)P₂ through its previously sequestered myristyl group. This may finally cause the dissociation of GagMA from the gRNA and promote the rod-shape conformation of Gag.

might result from binding reactions of the GagMA and GagNC domains with both the gRNA and the PM. These binding events associated with conformational changes could thus control the Gag assembly process at the PM. Such a dual role of the NC domain in the binding of Gag to the PM and the gRNA could explain, at least in part, the decrease in the accumulation of Gag at the PM, upon deleting the NC domain (52). This view may differ from previous reports showing that GagZip chimeric proteins (50, 116–118), where GagNC is replaced by a heterologous transcription factor leucine zipper, can undergo assembly with the release of immature virus-like particles resembling those formed by wild-type Gag (50, 116–119). However, these immature particles contain very little genomic or cellular RNA (116), indicating that the GagZip proteins probably do not interact with nucleic acids in the cellular context. As a consequence of the fused leucine zipper motif, the myristyl group of GagMA may be exposed very early during the assembly process, directly anchoring the GagZip proteins to

the PM and bypassing the steps described in Fig. 5. Further validation of the possible role of GagNC, as proposed in Fig. 5, will be elucidated in due course by monitoring the assembly of Gag and Gag mutants in the cellular context.

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The authors declare that they have no conflict of interest.

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