

## Heavy Water Slows the *Gonyaulax* Clock: A Test of the Hypothesis That D<sub>2</sub>O Affects Circadian Oscillations by Diminishing the Apparent Temperature

(rhythms/bioluminescence/dinoflagellates)

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**ABSTRACT** In order to test the hypothesis that deuterium oxide acts in circadian systems by simulating a lower environmental temperature, we have examined the effects of D<sub>2</sub>O on *Gonyaulax polyedra*, where the period of the circadian rhythm of bioluminescent glow in constant environmental conditions is shorter at 16° than at 22°, and the phase of the glow peak relative to a light-dark cycle is advanced at 16° relative to 22°. Addition of low concentrations of D<sub>2</sub>O to *Gonyaulax* does not produce the same effects on period and phase as lowering the temperature; the free running period is lengthened and the phase is delayed. These results show that in the *Gonyaulax* rhythm, the effect of added D<sub>2</sub>O is not equivalent to lowering the temperature.

Although there are many potential causes for the effects of heavy water in biological systems, its action, phenomenologically speaking, is frequently comparable to the effect of lowering the temperature (1-3). In fact, in experiments with the rate of replication of polio virus, Lwoff and Lwoff (4, 5) concluded that D<sub>2</sub>O in that system "diminishes the apparent temperature."

This "low temperature equivalence" hypothesis for D<sub>2</sub>O action is of particular interest in circadian systems, for it is here that one finds a rather unusual effect of temperature (6): the frequency (or period) of the daily oscillations is about the same at different temperatures, in otherwise constant conditions. This has been postulated (7, 8) to be the result of a biochemical compensation mechanism, the functional importance of which is hypothesized to be related to the timekeeping or "biological clock" role of the system (9).

An important test of the "low temperature equivalence" hypothesis for D<sub>2</sub>O is to see whether or not circadian oscillating systems that exhibit temperature compensation in period also exhibit a similar compensation with added D<sub>2</sub>O. Pittendrigh *et al.* (3) recently posed and examined this question, prompted in part by data from Enright (10) which showed that the periods of circadian oscillators were less affected by 20% D<sub>2</sub>O than were those of noncircadian biological oscillating systems. Their results, in experiments with the *Drosophila* eclosion rhythm, supported the low temperature equivalence hypothesis: two different features of the rhythm were affected similarly by D<sub>2</sub>O and by temperature. The period of the rhythm in constant conditions is almost temperature inde-

pendent, and is similarly unaffected by D<sub>2</sub>O. The phase angle of the rhythm, however, is markedly delayed at lower temperatures, which effect can be duplicated by D<sub>2</sub>O.

For several reasons, including their own concern that circadian systems might exhibit general homeostasis (11) not specifically related to the mechanism of D<sub>2</sub>O action, Pittendrigh *et al.* (3) were still wary of the conclusion, and anxious to further test this hypothesis. In particular, they suggested that a circadian system that exhibits "over-compensation" be examined, i.e., one in which the frequency is higher at lower temperature. We have carried out such experiments and find that their caution was justified: D<sub>2</sub>O and temperature do not appear to be equivalent in the *Gonyaulax* system.

### MATERIALS AND METHODS

Cultures of the bioluminescent marine dinoflagellate *Gonyaulax polyedra* were grown at either 22° or 16° on cycles of 12 hr of light and 12 hr of darkness (LD:12,12) on a supplemented sea water medium ("31" medium) as previously described (12). For deuteration, a salt medium containing different amounts of D<sub>2</sub>O, as desired, was added to make up 24% of the final volume. For this salt medium, to 100 ml of H<sub>2</sub>O or D<sub>2</sub>O were added 2.54 g of NaCl, 100 μl of nitrate stock (75 mg of NaNO<sub>3</sub> per ml of deionized distilled water), 10 μl of "31" vitamin mix (12) and 100 μl of "TW" trace elements. One liter of "TW" trace elements contained 800 mg of Na<sub>2</sub>MoO<sub>4</sub>·2H<sub>2</sub>O, 23 mg of CoCl<sub>2</sub>·2H<sub>2</sub>O, 11 mg of CuCl<sub>2</sub>·2H<sub>2</sub>O, and 6 mg of KI. From two stocks of this salt medium, one made with deionized distilled water and the other with 99.8% D<sub>2</sub>O (Bio-Rad), appropriate quantities of each were added to give the desired percentage of D<sub>2</sub>O. Other details are given in the thesis of McDaniel (13).

The bioluminescent "glow" (a continuous, low-intensity, spontaneous light emission) was recorded from 10-ml cultures of cells in vials, by means of a modified automatic scintillation counter (14, 15). To determine the growth rate, we removed aliquots of cells from a culture and determined the cell concentration with the aid of an electronic cell counter.

### RESULTS AND DISCUSSION

The "anomalous" effect of temperature on the period length (i.e., shorter periods at lower temperatures) of the circadian rhythm of bioluminescence (7) was confirmed and can be seen from the two experiments at 16° and 22° in Fig. 1 in which there was no D<sub>2</sub>O added. The Q<sub>10</sub> (ratio of period lengths at a 10° temperature interval) calculated from these two points

Abbreviation: LD:12,12, 12 hr of light and 12 hr of darkness.

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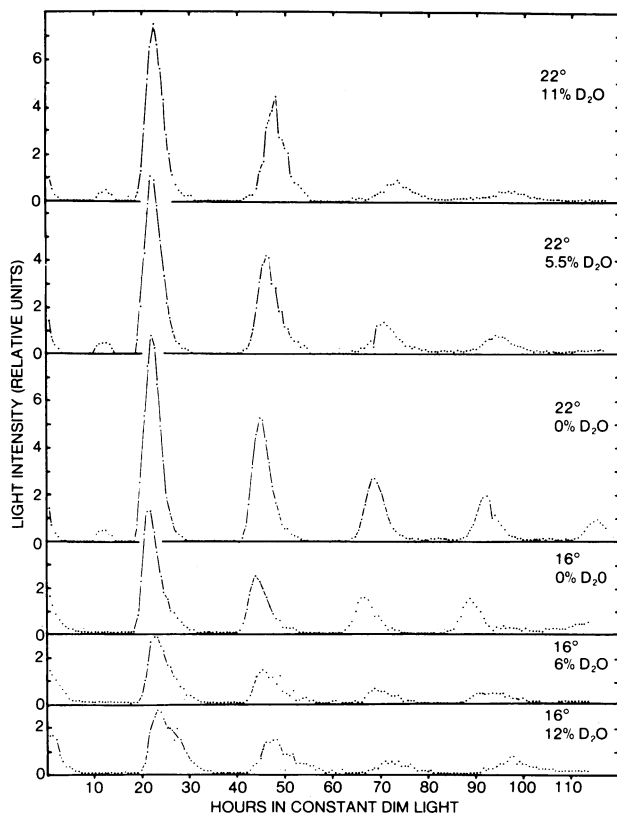


FIG. 1. Free-running period of the *Gonyaulax* glow rhythm. The top three curves illustrate the glow rhythm for cells grown in LD:12,12 at 22° and assayed at 22°. For the three bottom curves the cells were grown and assayed at 16°. D<sub>2</sub>O was added at the end of the dark period, at which time the cells were transferred to constant dim light [110 foot-candles (1180 lux)]. Each of these curves represents the average of the glow of four separate vials. The percent deuteration for each culture is given above each curve on the figure.

was 0.93. At both temperatures the effect of added D<sub>2</sub>O (at all concentrations tried) was to increase the length of the period. The cultures with D<sub>2</sub>O added appeared somewhat less healthy, and there appeared to be more scatter in the data, making the measurements less precise than we would have wished. At 16° (bottom three curves) the measured periods were 22.3, 23.2, and 24.6 ( $\pm 0.7$ , standard error of the mean) hr for 0, 6, and 12% D<sub>2</sub>O, respectively. At 22° (top three curves) the corresponding values were 23.3, 24.2, and 25.3 ( $\pm 0.7$ ) hr for 0, 5.5, and 11% D<sub>2</sub>O, respectively. With 16.5% D<sub>2</sub>O at 22° (not illustrated) the period was 25.5 ( $\pm 0.7$ ) hr. These experiments have been repeated several times with similar results.

Pittendrigh *et al.* (3) also compared the effect of temperature with that of D<sub>2</sub>O on the phase angle of the *Drosophila* rhythm with respect to the light-dark cycle. They found that there was a delay (i.e., the peak occurs later, relative to the onset of light) both at lower temperature and with D<sub>2</sub>O. With the *Gonyaulax* glow rhythm under a light-dark cycle, there is instead an advance in the phase cycle at a lower temperature (Fig. 2). D<sub>2</sub>O does not cause such an advance; at 22° both 5.5 and 11% D<sub>2</sub>O result in a phase delay. D<sub>2</sub>O may actually cause a larger phase delay on the *Gonyaulax* clock, but this is difficult to measure with the glow rhythm, since *Gonyaulax* bioluminescence is photoinhibitable and at 22°, with no D<sub>2</sub>O added, the glow peak occurs just before the lights come on. Regardless of

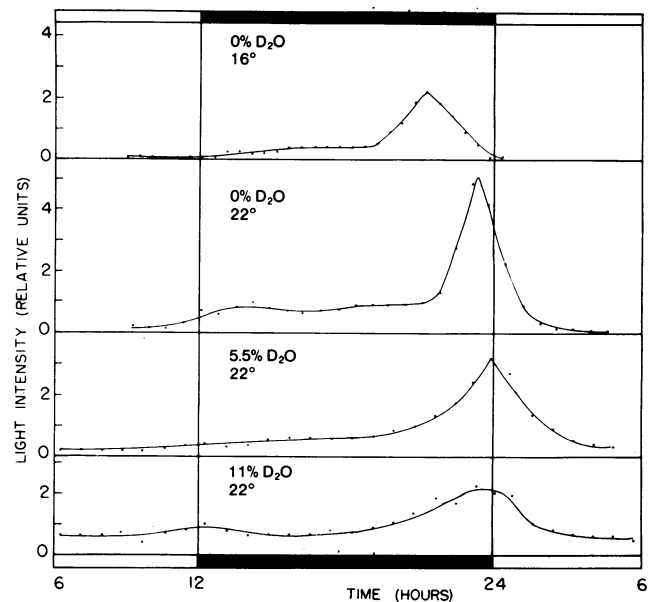


FIG. 2. The phase angle of the *Gonyaulax* glow rhythm with respect to an LD:12,12 cycle. The top curve illustrates the average glow of three separate vials of cells taken from a culture grown at 16° in "31" medium and assayed at 16° in LD:12,12. The bottom three curves illustrate the glow (average of five separate vials) of cells taken from a culture grown at 22° in medium of 0, 5.5, and 11% D<sub>2</sub>O, from top to bottom, respectively, and assayed at 22° in LD:12,12. The shaded areas at the top and bottom of the figure represent the dark part of the light-dark cycle.

the absolute magnitude of the phase delay, the effect of D<sub>2</sub>O is not similar to that of low temperature.

In summary, we have tested the hypothesis that D<sub>2</sub>O affects circadian oscillations by diminishing the apparent temperature, and obtained evidence to reject it. At 22° added D<sub>2</sub>O causes an increase in the period of the oscillation and a delay in the phase angle of the rhythm, whereas lowering the temperature to 16° results in changes in the opposite direction.

With regard to its mode of action, Pittendrigh *et al.* (3) have recently noted that D<sub>2</sub>O stands out as the only chemical agent whose action is uniformly similar and widely (if not universally) effective in circadian systems. Its mechanism of action may be too complex to be of help in elucidating the biochemical basis of circadian rhythms. On the other hand, D<sub>2</sub>O may have a common and possibly even a relatively simple mode of action, such as one involving solvent effects [see Pittendrigh and Cosbey (16) for a discussion and interpretation of rapid effects of D<sub>2</sub>O]. Analysis of the effects of D<sub>2</sub>O may be of help in evaluating models for the circadian clock, such as the one involving membranes proposed recently by Njus *et al.* (8).

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1. Thomson J. F. (1963) *Biological Effects of Deuterium* (Pergamon Press, New York).
2. Katz, J. J. & Crespi, H. L. (1970) "Isotope effects in biological systems," in *Isotopic Effects in Reaction Rates*, eds. Collins, C. J. & Bowman, N. S. (Van Nostrand Reinhold Book Co., New York), chap. V, pp. 286-367.

3. Pittendrigh, C. S., Caldarola, P. C. & Cosbey, E. S. (1973) *Proc. Nat. Acad. Sci. USA* 70, 2037-2041.
4. Lwoff, A. & Lwoff, M. (1960) *C.R. H. Acad. Sci.* 251, 3131-3132.
5. Lwoff, A. & Lwoff, M. (1961) *C.R. H. Acad. Sci.* 252, 223-225.
6. Sweeney, B. M. & Hastings, J. W. (1960) *Cold Spring Harbor Symp. Quant. Biol.* 25, 87-104.
7. Hastings, J. W. & Sweeney, B. M. (1957) *Proc. Nat. Acad. Sci. USA* 43, 804-811.
8. Njus, D., Sulzman, F. M. & Hastings, J. W. (1974) *Nature* 248, 116-120.
9. Pittendrigh, C. S. (1954) *Proc. Nat. Acad. Sci. USA* 40, 1018-1029.
10. Enright, J. T. (1971) *Z. Vergl. Physiol.* 72, 1-16.
11. Pittendrigh, C. S. & Caldarola, P. C. (1973) *Proc. Nat. Acad. Sci. USA* 70, 2697-2701.
12. Fogel, M. & Hastings, J. W. (1971) *Arch. Biochem. Biophys.* 142, 310-321.
13. McDaniel, M. (1974) B.A. Thesis, Biology Dept., Harvard University.
14. Hastings, J. W. (1960) *Cold Spring Harbor Symp. Quant. Biol.* 25, 131-143.
15. Hastings, J. W. & Bode, V. (1962) *Ann. N.Y. Acad. Sci.* 98, 876-889.
16. Pittendrigh, C. S. & Cosbey, E. S. (1974) *Proc. Nat. Acad. Sci. USA* 71, 540-543.