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Bivalent Ligands Targeting Chemokine Receptor Dimerization: Molecular Design and Functional Studies

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Abstract

Increasing evidence has shown that chemokine receptors may form functional dimers with unique pharmacological profiles. A common practice to characterize such G protein-coupled receptor dimerization processes is to apply bivalent ligands as chemical probes which can interact with both receptors simultaneously. Currently, two chemokine receptor dimers have been studied by applying bivalent compounds: the CXCR4-CXCR4 homodimer and the CCR5-MOR heterodimer. These bivalent compounds have revealed how dimerization influences receptor function and may lead to novel therapeutics. Future design of bivalent ligands for chemokine receptor dimers may be aided with the recently available CXCR4 homodimer, and CCR5 monomer crystal structures by more accurately simulating chemokine receptors and their dimers.

Keywords

bivalent ligand; CCR5; chemokine receptor; CXCR4; dimerization; GPCR; MOR

1. Introduction

All chemokine receptors are G protein-coupled receptors (GPCR), which have seven transmembrane helices (TM) and couple to heterotrimeric G proteins. The GPCR superfamily of proteins has approximately 791 genes encoding for the six different receptor subtypes [1]. Currently, all 19 known chemokine receptors belong to the class A (rhodopsin-like) family of GPCRs and are classified into four main subfamilies based upon which chemokines they bind: CC, CXC, XC, and CX₃C receptors [2–4]. Many of the chemokine receptors are promiscuous and bind to several chemokines within their family and allow for tailored chemokine response and redundancy [2,4].

Originally, it was postulated that GPCRs functioned in a monomeric fashion and that there was a general stoichiometry of 1:1 for the receptor ligand interaction [5,6]. However,

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Conflict of Interest

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increasing evidence has begun to support the possibility that they may act in dimeric, or even oligomeric assemblies [7–9]. One of the first observations of dimerization in rhodopsin-like GPCRs was seen in β -adrenergic receptors; it was seen that binding of one ligand decreased the binding of a second one [10]. This type of “cross-talk,” or better known as negative cooperativity, occurs when a dimer bound ligand either inhibits the binding, or signaling of a second ligand to the dimer pair [8,9].

One of the earliest methods for elucidating dimer pairs was to use co-immunoprecipitation (co-IP) techniques. First used for the β_2 -adrenergic receptor, two populations of the receptor were engineered to either have an influenza hemagglutinin (HA) or a myc-epitope tag incorporated into the receptor [11]. These two receptor subtypes were then co-expressed, and using an anti-myc-epitope antibody, immunoprecipitation was performed. If only monomers were present, then only the myc-epitope tagged β_2 -adrenergic receptor should show up on a Western blot analysis due to the selectivity of the anti-myc-epitope antibody. However, it was found that the HA tagged β_2 -adrenergic receptor was present (by co-staining with an anti-HA antibody) along with the myc-epitope tagged β_2 -adrenergic receptor [10]. Therefore, the two receptor populations had to be directly interacting with each other to both be isolated using co-immunoprecipitation, and thus showed that the β_2 -adrenergic receptor was able to homodimerize [10]. This technique has subsequently been used as a preliminary technique to study the homo and heterodimerization of numerous GPCRs [7–9].

Another important technique for GPCR dimerization/oligomerization detection is Fröster Resonance Energy Transfer (FRET). Both bioluminescence and fluorescence (BRET and FRET respectively) have been used in this technique. For FRET detection, the two receptors suspected of dimerizing are tagged with two different fluorescent proteins: i.e., a cyan fluorescent protein (CFP) and a yellow fluorescent protein (YFP). It is essential for the FRET that the excited state of one fluorescent protein (donor chromophore) can transfer energy at a very specific distance to an acceptor chromophore and permit it to emit its unique excitation wavelength. In order for FRET to take place, the two chromophores (and associated proteins) must be in close proximity (10 to 100 Å) [11]. Therefore, excitation of the CFP at ~436 nm would give only one emission wavelength at ~480 nm if no dimerization was present since the two GPCRs are not in close proximity. If the receptors dimerized, exciting the CFP would yield both the emission wavelength at ~480 nm (for CFP) and an additional emission wavelength at ~535 nm, which would correspond to the excitation/emission from the YFP on the other GPCR in close proximity. This technique can also be coupled with a bioluminescent luciferase enzyme instead of the CFP to excite the YFP through BRET [12]. The combination of co-immunoprecipitation and FRET/BRET has led to a network of GPCR homodimers and heterodimers being discovered [7].

In addition to the biochemical techniques, direct observation of GPCR dimers and oligomers has been obtained using both GPCR crystallization and atomic-force microscopy techniques [13–15]. Using atomic-force microscopy, oligomer formations of rhodopsin were able to be observed, giving the first direct visualization of GPCR oligomization [15]. Additionally, both chemokine receptor CXCR4 (CXCR4) and the mu opioid receptor (MOR) were observed to form dimers within their crystal lattice (Fig. **1a** and **1b**). While these observed

dimers might partially be due to an artifact of the crystallization process, it did lend credence to GPCR dimerization [13,14].

Several types of interactions between dimerized GPCRs have been proposed, and two main dimerization models have subsequently been described: a contact dimer model, and a domain-swapped dimer [9]. The domain-swapped model proposes that TM6 and TM7 are exchanged between monomers to form a dimer [16]. The contact dimer model proposes that dimerization occurs through direct contact between the different interfaces of helices of GPCRs: TM5/TM6, TM3/TM4, and TM1/TM2 interfaces have been postulated and observed [13,14,17,18]. Both hypotheses have been supported by mutation and computational studies, but due to observations of GPCR crystal structure, the contact dimer may represent a more realistic model.

An important aspect of GPCR dimerization is its effect on receptor function and signaling. As alluded to earlier, a possible outcome of dimerization is positive and negative cooperativity (Fig. 2) [7,8]. Positive cooperativity occurs when binding of a ligand to one receptor leads to partial, full, or enhanced activation of the second receptor [7,8]. It may also occur when two ligands bind both receptors, and an enhanced action is seen. Negative cooperativity can occur when one bound ligand leads to either inhibition of a second ligand binding to the dimer, or inhibition of signaling from a second bound ligand [7,8].

Functionally, heterodimers may allow for different mechanisms of signal regulation for GPCRs [19]. For example, within the CCR2-CCR5 heterodimer, dimerization led to the receptors being able to couple with $G_{\alpha_{q/11}}$, which, as individual receptors, they normally do not couple with [20]. A similar effect was seen for the MOR-delta opioid receptor (DOR) heterodimer; when the receptors were expressed alone, pertussis toxin inhibited agonist stimulated G_{α} -dependent signaling from both receptors, but when expressed together, pertussis toxin did not inhibit their G_{α} -dependent signaling [21]. These results suggested that the heterodimer could couple to different G proteins than the monomers by themselves. Dimerization of GPCRs may also affect receptor desensitization and internalization [7,8,21]. There are several comprehensive reviews on chemokine receptor homodimerization and heterodimerization, which reveal the extent of their dimerization and the functional consequences [20,22–29]. Similarly, chemokine receptor dimers led to unique pharmacological profiles which can add upon the already intricate receptor-ligand interactions. Targeting these chemokine dimer interactions may lead to unique therapies with marked potential. Currently, the most direct way to monitor these interactions is to target them with chemical probes such as bivalent compounds [5]. A bivalent compound is defined as a compound that contains two distinct pharmacophores which can interact simultaneously with two receptors at once [6].

2. Chemokine Receptor Bivalent Ligands

Bivalent compounds are indispensable for studying the relationship between GPCRs in both homodimers and heterodimers pairs [5]. By targeting dimers of GPCRs, new pharmacological profiles are obtainable because of their unique properties [30]. Using bivalent ligands may lead to ligands that have higher affinity, higher selectivity, and

improved or altered physiological responses. The possible synergistic effects are due to the cooperativity between the receptors and an overall drop in the entropy of interaction by targeting two receptors at once [6]. As such, it is imperative and advantageous to target homodimers or heterodimers of chemokine receptors with bivalent compounds [20,23,25,29,31].

Generally, bivalent compounds can either be classified as homo-bivalent or hetero-bivalent, that is, they either have two of the same pharmacophores, or two different ones. These two pharmacophores are attached to each other with a linker that will not interfere with receptor binding and is the appropriate length to allow the two pharmacophores to interact with both receptors. The average distance between GPCR dimers is thought to be between 27 Å and 32 Å [5]; therefore, the linker length should ideally be in that range. Several different linker types have been reported, and range from aliphatic chains to polyethers [30]. The pharmacophores of choice usually have high affinity and selectivity for the targeted receptor(s) dimer and can tolerate added substitutions onto their structure to facilitate the addition of the linker. Currently, two chemokine receptor dimers have been targeted with bivalent compounds: the CXCR4-CXCR4 homodimer and the CCR5-MOR heterodimer. However, more examples are expected in the future with the expanding field of GPCR dimerization and the examples below are very typical within bivalent ligand studies.

2.1. Bivalent Ligands Targeting the CXCR4-CXCR4 Homodimer

Chemokine receptor CXCR4 is one of the two main co-receptors for the human immunodeficiency virus type 1 (HIV-1) for viral invasion into human cells [32–35]. Homodimerization of CXCR4 has been shown to contribute to “warts, hypogammaglobulinemia, infections and myelokathexis” (WHIM) syndrome and may be involved in other physiological functions [36,37]. Bivalent ligands have been specifically designed to interact with the CXCR4 homodimer in order to study its biological significance [5,30].

The first report of a chemokine receptor CXCR4 bivalent ligand was by Tanaka *et al.* in 2010 [38]. Using an analogue of the cyclic peptide CXCR4 antagonist FC131 [cyclo(-D-Tyr¹-Arg²-Arg³-Nal⁴-Gly⁵-)] as the principle pharmacophore, different length of poly(L-proline) linkers were used to estimate the distances between binding pockets of CXCR4 homodimers. The rigid poly(L-proline) linkers are known to maintain a predetermined, constant distance between two ligands and therefore, could act as a molecular ruler. Fig. 3 shows the two bivalent ligand scaffolds with either a poly(L-proline) linker (**1-8**) or a polyethylene glycolated (PEG) poly(L-proline) linker (**9-14**) and the two monovalent controls (**15, 16**). The first difficulty in designing these bivalent ligands was finding high affinity ligands with linker attachment site(s) that did not affect their affinity. The CXCR4 antagonist FC131 was chosen for the bivalent compound due to its high binding affinity ($K_i = 31.5$ nM) [38–40] and an accessible attachment site. Previous structure-activity relationship (SAR) studies of FC131 have shown that the carbonyl oxygen of Gly⁵, but not the side chain's hydrogen atoms, plays a role in binding [41–44]. Therefore, different amino acids can take the place of Gly⁵ without drastically altering the binding affinity. Replacing

Gly⁵ with a D-Cys only had a 2-fold decrease in its binding affinity to CXCR4 and maintained a nanomolar K_i value of 53.4 nM [38].

Once the pharmacophore and the linker attachment site were chosen the linker was then designed based upon proline oligomers. Proline oligomers can adopt a constant helical structure that maintains a length of 0.9 nm per turn by which three prolines may give a rise of 3.1 Å per residue [38,45]. This structure allowed for two pharmacophores to be separated by specific distances. Previous bivalent ligand studies have shown that the length of the linker between pharmacophores was essential for activity [5,30,46–49]. The poly(L-proline) linker in compounds **1-8** and **9-14** varied in length from 2 to 8 nm. Since the poly(L-proline) linker is rigid, more flexible alkyl chains (**1-8**) or PEG groups (**9-14**) attached the linker to the FC131 units making the bivalent ligand less rigid and allowing them to adopt more conformations.

Fig. 4 shows the effects that linker length had on binding affinity of the two sets of compounds. For both series the optimum linker length was approximately 6 nm with compounds **6** and **12** having a K_i value of 9.9 nM and 13.9 nM respectively [38]. Increasing or decreasing the linker length from 6 nm led to loss of binding affinity, resulting in a maximum 6-fold loss in activity. The 2 to 3-fold increase in binding affinity for the bivalent compounds compared to FC131 may due to the synergistic effect of binding two receptors at once [5,30]. Additionally, the two control compounds, **15** and **16**, had K_i values of 294 nM and 72 nM, which was less than FC131, and thus, showed that the synergistic effects were not due to the poly(L-proline) linkers.

Compounds **1** through **14** revealed several important aspects of designing chemokine receptor dimer bivalent compounds. First, linker composition is pivotal; when comparing the two series of compounds, there was a general trend that the non-PEGylated linker (**1-8**) was up to two times more potent than the PEGylated linker (**9-14**). One explanation of the loss of potency was that the PEGylated linker added more rotatable bonds making the linker less rigid, which had been shown to decrease binding affinity [30,50]. Second, linker length is essential for the binding affinity of CXCR4 bivalent ligands. As shown in figure 4, there was a global minimum for both sets of compounds at a length of 6 nm. Overall, this study demonstrated the importance of linker length and composition in designing chemokine receptor bivalent ligands.

While linker length and composition are important in bivalent ligand development, Choi *et al.* showed that synthetic linkers may not be needed at all for chemokine receptor bivalent ligands [37]. Utilizing a small, all D-amino acid peptide (Fig. 5, DV1, **17**) based upon the N-terminus of viral macrophage protein II (vMIP-II) they were able to synthesize a bivalent ligand which lacked the classical linker seen in traditional bivalent compounds. The DV1 dimer, **18**, consists of two DV1 peptides linked together through a disulfide bond between cysteines already present in the peptides. Competition binding assays utilizing a CXCR4 specific antibody indicated that bivalent peptide (**18**) had 14 times greater affinity for CXCR4 than the monomer (**17**), with IC₅₀ values of 3 nM and 43 nM respectively [37]. A similar trend was seen in antiviral activity against the HIV-1 IIIIB strain; compound **18** had an IC₅₀ value of 4.4 μM, whereas **17** had an IC₅₀ value of 12.1 μM [37]. The synergistic

increase in binding affinity and functional activity may be from the bivalent peptide by simultaneously interacting with both binding pockets within a CXCR4 homodimer. Guided by site directed mutagenesis, molecular modeling verified that **18** could interact with both binding pockets of a CXCR4 homodimer [37].

Further studies utilizing peptides for the basis of bivalent CXCR4 homodimer ligands led to simplifying the DV1 (**17**) peptide into a shortened version named DV3 (**19**) [51]. Using competition binding assays utilizing a CXCR4 specific antibody, DV1 (**17**) and DV3 (**19**) showed IC₅₀ values of 236 nM and 440 nM respectively [51]. The dimer version of the DV3 peptide (**20**) had an IC₅₀ value of 133 nM, which was 3 times higher than the monomer **19**, and was consistent with the increase in binding affinity seen for bivalent compounds [51]. The binding data for the DV3 peptide dimer (**20**) indicated that the residues after the cysteine residue of dimeric DV1 (**18**) were essential for CXCR4 binding. An additional bivalent ligand consisting of DV1 and DV3 linked through C-terminus lysine residues, DV1-K-DV3 (**21**), had an IC₅₀ of 4 nM, which was 33 times higher than the DV3 dimer (**20**) [51]. These results suggested that **21** was more capable of interacting with both binding pockets within the CXCR4 homodimer. Reasonable explanations for the increase in interaction could be the increase in length of the bivalent compound or a different tertiary structure that was only adopted in **21**.

Both examples of peptide-based bivalent ligands for CXCR4 homodimers revealed that bivalent compounds can be synthesized without the traditional linker connecting two pharmacophores. Compounds **18**, **20**, and **21** illustrate a logical progression which suggested that optimization of chemokine receptor bivalent ligands may rely heavily on pharmacophore choice and attachment site.

2.2 Bivalent Ligands Targeting the CCR5-MOR Heterodimer

Due to modern antiretroviral therapies, HIV-1 infected patients have longer lifespans and a better quality of life [52]. However, several neurological complications are now being seen due to HIV-1 associated injury of neurons by infected microglia and astrocytes (neuroAIDS) [53–55]. Furthermore, these effects are further exacerbated with opiate use and abuse [54,56–59]. A possible mechanism for the potentiation effects of opiates is the interaction of the mu opioid receptor (MOR) with chemokine receptor CCR5 (CCR5), a known HIV-1 co-receptor in the CNS [35,54,60–64]. The progression of neuroAIDS has been linked to opiate abuse that may arise from the synergistic interactions between CCR5 and MOR [54,57–59,65,66]. A key example of this was that MOR agonists can up-regulate the expression of CCR5 and promote HIV-1 infection, which can be blocked by MOR antagonists [67]. Opiates can also exacerbate the amount of indirect neuronal injury in neurons and glia through HIV-1 induced CNS inflammation.[68,69]. The specific opioid dependent neuronal injury may be primarily induced by MOR expressing glia in the CNS [70].

Importantly, MOR and CCR5 have been shown to heterodimerize with each other and undergo crosstalk [71,72]. The interaction has been shown to affect immune cell function and may produce the synergistic effects seen in neuroAIDS progression.[67,73]. In order to explore the pharmacological profile of the CCR5-MOR heterodimerization and its relation

with neuroAIDS, Yuan *et al.* designed a bivalent ligand (**22**) containing both MOR and CCR5 antagonist pharmacophores [49].

The premise of the bivalent compound **22** was to use both a MOR and a CCR5 antagonist to try to inhibit both receptors at the same time (Fig. 7). Naltrexone (**28**) and maraviroc (**29**) were chosen for their high binding affinities and well known pharmacological profiles [52,74]. However, both molecules had to be functionalized with an amine group in order to allow for attachment of the linker. 6 β -Naltrexamine has been synthesized before, but 4-aminophenyl-maraviroc had never been reported, so a new synthetic route was devised [49]. The linker connecting the two pharmacophores was chosen based on the work of Daniels *et al.* with MOR-DOR bivalent compounds [48]. They found that a 21-atom spacer made of an aliphatic diamine flanked by two diglycolic groups was optimal for opioid receptor heterodimers [48].

In order to test if the bivalent ligand still recognized both receptors, binding assays were first conducted. Table 1 shows the results of both CCR5 and MOR radiobinding assays for selected compounds. Within the MOR binding assay, all of the compounds showed higher K_i values than naltrexone (**28**) [75]. The bivalent compound **22** had a 70-fold loss in binding affinity to the MOR, whereas the monovalent compound **24** had a 13-fold loss in affinity. The data for the CCR5 binding assay indicated that any substitution on maraviroc's phenyl ring is detrimental; there was a clear trend of decreasing affinity with increasing size of the group at the 4-position [75]. Overall, there was about a 1000-fold loss of affinity seen for **22** compared to maraviroc [75]. However, it did still bind CCR5 at a nanomolar level, meaning that its affinity wasn't completely abolished.

Calcium mobilization assays were used to determine the functional activity of the compounds to both the MOR and the CCR5 (Table 2). Compounds containing a morphinan group (**22**, **24**, **28**) were tested for their MOR antagonism [75]. Overall, substitution on naltrexone (**28**) was much more tolerated than for maraviroc as seen in Table 1 [75]. All of the compounds had similar IC_{50} values which meant that the difference in maraviroc attachment sites and lack of maraviroc did not affect MOR antagonism.

The CCR5 antagonism results from the calcium mobilization assays indicated that modification of maraviroc (**29**) with phenyl substituents was not well tolerated. When an amino group was added to the 4-position (**27**), there was about 7-fold loss in CCR5 inhibition [75]. An even more drastic effect was seen for the bulkier substituents in **25** and **26**, with losses in activity of 3600-fold and 700-fold respectively [75]. Therefore, smaller substituents on the phenyl ring of maraviroc might be better tolerated compared than more sterically bulky groups. However, this observation was not seen to the same extent for the bivalent and monovalent compound. The monovalent compound **23** had a larger substituent than **25** and **26**, but they showed only a 200-fold and 60-fold decrease in activity compared to **29** [75]. These results suggest that the longer monovalent compounds may adopt a different binding mode than **25** and **26** and retain some of their CCR5 antagonism.

A HIV-1 infection assay was conducted using primary human astrocytes; primary human astrocytes were chosen because they are one of the primary sites of infection in NeuroAIDS.

They are localized on the blood brain barrier and are the sites where opioids can synergistically potentiate the pathophysiological effects of HIV-1 infection [54]. Upon infection with R5 HIVSF162 (with and without morphine), there was a significant increase in Tat (transactivator of transcription) expression in astrocytes that coincided with virus invasion (Fig. 8) [76]. When maraviroc was added, virus invasion was decreased, as expected. However, when morphine was added along with maraviroc (**29**), its antiviral effects were completely abolished, which was indicated by a significant 4-fold increase in HIV Tat expression in the astrocytes. Treatment with naltrexone (**28**), or a combination of naltrexone (**28**) and maraviroc (**29**), had no effect on virus invasion with and without morphine's presence. On the other hand, addition of the bivalent compound **22** ("bivalent") had a significant effect compared to maraviroc and maraviroc with morphine stimulation. Overall, there was a 3.3-fold decrease in virus entry compared to maraviroc alone and a 7-fold decrease when compared to maraviroc with morphine [76]. Importantly, morphine stimulation had no effect on the bivalent compound's viral entry inhibition activity. Cytotoxicity assays indicated neither maraviroc nor **22** had any toxicity in the astrocytes [76]. The results showed that in a native system, the bivalent compound could act as a potent virus invasion inhibitor without deleterious effects caused by morphine stimulation.

While astrocytes harbor HIV in a more latent state, microglia and macrophages are the primary location for viral production in the brain [76,77]. Accordingly, bivalent compound **22** was also tested in microglia to elucidate cell-specific interactions. As previously observed in astrocytes, the antiviral effect of maraviroc (**29**) was significantly decreased by morphine administration. However, unlike in astrocytes, the bivalent compound **22** was unable to prevent HIV-1 infection in microglia [76]. Several explanations exist for the differential effects seen in astrocytes and microglia for **22**. One possibility for the difference is that the expression levels of CCR5 and MOR differ greatly in the two cell types. The ratio of CCR5:MOR in astrocytes is roughly 2:1, while in microglia it is roughly 4:1 [76]. This difference may lead to more CCR5-MOR heterodimers forming in astrocytes than microglia. Bivalent compound **22** was designed to selectively target those heterodimers; therefore, it should have a greater affect in astrocytes where there are potentially more CCR5-MOR heterodimers present [49,75,76].

Another plausible reason for the difference in compound **22**'s anti viral potency between the cell types are the different MOR splice variants present in them. The MOR undergoes extensive alternative splicing and these splice variants may lead to cell-specific effects and unique pharmacological profiles [78–80]. Increasing evidence has shown that MOR splice variants could directly affect HIV infection, susceptibility, and progression [78,79]. When comparing the expression rates of three different MOR splice variants (MOR-1, MOR-1A, and MOR-1X), Dever *et al.* found that they were differentially expressed in the individual CNS cell types [79]. MOR-1 is the canonical variant while MOR-1A is the shortest C-terminal splice variant and MOR-1X is the longest C-terminal variant. Importantly, astrocytes were found to express all three subtypes, while microglia only express MOR-1A [79]. The MOR-1 splice variant is known to dimerize with CCR5, while the other splice variants have not been fully studied for CCR5-MOR heterodimerization [72]. Therefore, the MOR-1A variant might not be able to heterodimerize with CCR5 in microglia, which would

lead to the lack of antiviral activity for bivalent compound **22** seen in microglia. Both the differences in expression ratios of CCR5 to MOR and differences in splice variants between astrocytes and microglia helped explain the cell specific effects for **22**.

3. Current Chemokine Receptor Crystal Structures

In 2010, the first chemokine receptor (CXCR4) was crystallized by the Stevens lab at the Scripps Research Institute [14]. The CXCR4 crystal structure was the first peptide GPCR to be solved and represented a major breakthrough in this field. It is important to note that several structural changes were used in order to stabilize the receptor for crystallography [13,14,81–91]. Using the T4 Lysozyme (T4L) strategy, intracellular 3 (IL3) of CXCR4 was replaced with T4L along with truncating its C-terminal and using point mutations for stabilizing [14]. While these techniques have been used successfully to crystallize GPCRs, they may introduce or induce unnatural receptor conformations [18,92]. However, GPCR crystal structures have provided a wealth of knowledge concerning ligand binding that have confirmed or disproved modeling and mutagenesis data [18,93]. More importantly, the chemokine receptor CXCR4 was crystallized in two dimeric forms with resolutions of 2.5 and 3.2 Å and these two dimers had either a TM5 and TM6 interface or a TM3 and TM4 interface respectively [14].

The antagonists utilized in the CXCR4 crystallization process were a small molecule antagonist IT1t and a cyclic peptide antagonist CVX15 [14]. In the presence of either ligand the binding pocket was shown to be much larger than comparable aminergic receptors, which was most likely due to its much larger endogenous peptide agonists [14,93]. The increase in the size of the binding pocket presumably led to both antagonists binding shallowly near extracellular loop 2 (ECL2), which is important in ligand recognition and receptor activation [14,93–95]. While the large binding pocket may make computational modeling and docking difficult, the CXCR4 crystal structure has led to a magnitude of studies using structural based drug design in order to make new CXCR4 specific ligands [51,93,96–108].

In 2013 chemokine receptor CCR5 (CCR5) was crystallized by Tan et al. and revealed both similarities and differences within the chemokine receptor family [91]. Like CXCR4, the binding pocket of CCR5 was large due to its endogenous peptide agonists, but the crystallized antagonist, maraviroc, occupied a deeper domain of the pocket compared to CXCR4. Due to the depth of the maraviroc binding pocket, ECL2 did not play a role in binding, which was in contrast to the CXCR4 structure [91]. The variances could reflect the differences in the mechanisms of antagonists used in the crystallization processes or general structural differences between the receptors. In all, both chemokine receptor crystal structures will allow for the rational design of new ligands and for the homology modeling of other chemokine receptors.

In addition to modeling ligand binding, the CXCR4 crystal structure allows for a unique approach to model chemokine receptor dimers. Previously, the only methods to model GPCR dimers were either sequence-based or docking-based, which are largely pragmatic and easily biased [18]. However, crystallized GPCR homodimers permit a new method of

modeling other GPCR dimers. For example, using the crystal structure of the bovine rhodopsin homodimer, Gorinski *et al.* were able to model the 5-HT_{1A} homodimer by superimposing monomer units over the dimer based upon sequence similarities [109]. When combined with site directed mutagenesis, the work supported a TM4/TM5 interface for 5-HT_{1A} homodimers [109].

Presumably different receptor types may lead to different dimer interfaces; therefore, it is critical to choose a dimer template that is similar to the target dimer being modeled [18]. The CXCR4 homodimer crystal structure thus allows for other chemokine receptor dimers to be more confidently modeled. For example, the CCR5-MOR heterodimer was recently modeled utilizing the CXCR4 homodimer crystal structure as the template (Fig. 9a) [76]. The heterodimer was based on the TM5-TM6 dimer interface that was seen in the IT1t bound CXCR4 crystal structure [14]. Utilizing this method the CCR5-MOR heterodimer model showed favorable electrostatic and hydrophobic interactions between receptors and could represent a possible conformation of the heterodimer. Furthermore, such models can be used to map the interactions between bivalent compounds and their respective receptor dimer. Fig. 9b and 9c show the CCR5-MOR selective ligand (22) bound to both receptors by spanning across the TM5-TM6 interface [110]. By further understanding the interactions between bivalent compounds and dimers, insight can be gained both in receptor function and dimer interaction. For example, the observed synergism in the HIV-1 inhibitory effects of ligand 22 was explained through dynamic simulations of 22 bound to the CCR5-MOR dimerization model [110].

4. Conclusion

Bivalent ligands represent a very promising technique for the study of GPCR dimerization. As indicated in this review, the groundwork for discovering new bivalent ligands has been accomplished within the field of chemokine receptor dimerization. First these studies indicated, a rational process must drive the bivalent ligand design in order to achieve interpretive results. Aspects such as pharmacophore identification, linker attachment site, linker composition, and linker length must be addressed first, as seen in the examples outlined in this review. Secondly, biological techniques such as co-immunoprecipitation, fluorescent lifetime imaging microscopy, FRET, protein fragment complementation assays, and even crystallization studies will be imperative to confirm the receptor(s) of interest may actually dimerize [20,22–29]. Thirdly, in order to fully explore the pharmacological and therapeutic implication of these dimers, bivalent ligands should be exploited in disease relevant cellular and molecular models.

For chemokine receptors, there are currently bivalent ligands targeting the CXCR4-CXCR4 homodimer and the CCR5-MOR heterodimer. For the CXCR4 homodimer three independent studies synthesized bivalent ligands that successfully targeted the homodimer. Linker length and composition proved to be essential when combining CXCR4 antagonist pharmacophores; more rigid linkers having a length of approximately 6 nm were favored [38]. Peptide based bivalent ligands were also capable of interacting with the homodimer. Both the attachment site and peptide lengths were influential in binding affinity [37,51]. For the CCR5-MOR heterodimer a bivalent compound was synthesized containing two

antagonists linked together with an aliphatic linker [49]. The CCR5-MOR bivalent ligand proved to be a potent and cell type specific inhibitor for NeuroAIDS where the known treatment, maraviroc, is less efficacious and fails to inhibit virus entry in the presence of morphine [75,76]. These studies show the effectiveness and potential of targeting chemokine receptor dimers with bivalent ligands and have laid the foundation for future studies.

The exponential increase of GPCR crystal structures within the last decade will undoubtedly aid in the design and implementation of bivalent ligands targeting chemokine receptor dimers. This is especially true since two chemokine receptors have been successfully crystallized and one of them was crystallized as a homodimer. When combined, these discoveries will allow for more reliable homology models of chemokine receptors and the ability to study their dimerized state. By utilizing computational modeling to support bivalent ligand design, a better understanding of receptor(s)-ligand(s) interaction can be gained. Furthermore, dynamics simulations of dimer-bivalent ligand complex models can improve our understanding on the dimerization mechanisms and its relationship to the synergism seen in bivalent ligands. This strategy has become even more advantageous due to the availability of chemokine receptor crystal structures.

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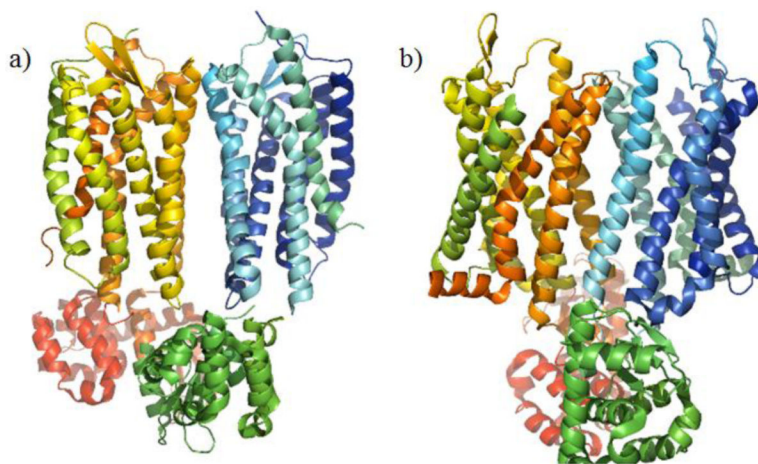


Fig. (1). Observed GPCR dimer construct from crystal structures. a) Crystal structure of the CXCR4 dimer with a TM5/TM6 interface (PDB code 3ODU) [14]. b) Crystal structure of the MOR dimer with a TM5/TM6 interface (PDB code 4DKL) [13].

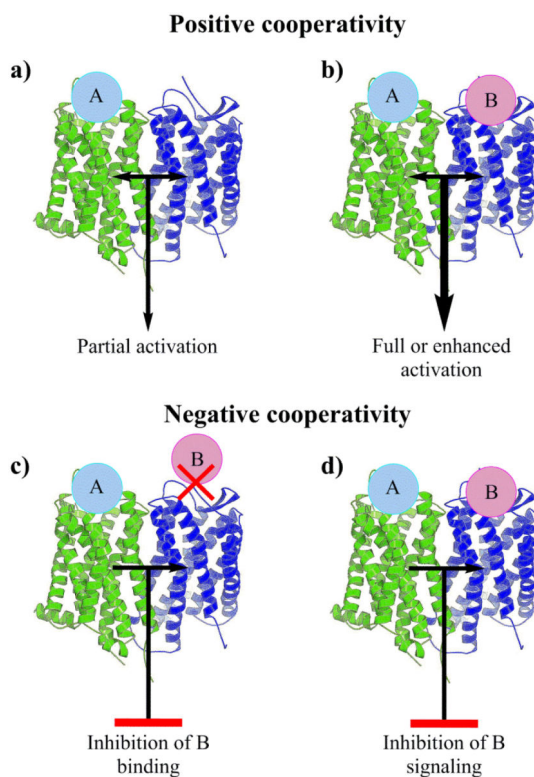


Fig. (2). Positive and negative cooperativity in GPCR dimerization. a) Agonist A binding to one GPCR (in green) results in partial activation of another GPCR (in blue). b) When two agonists, A and B, bind to the GPCRs there will be enhanced activation, synergism. c) In negative cooperativity binding of A to one GPCR (in green) leads to inhibition of the binding of B to another GPCR (in blue), leading to suppression of ligand B related signaling. d) Binding of A leads to inhibition of signaling from the GPCR (in blue) even with B bound to it.

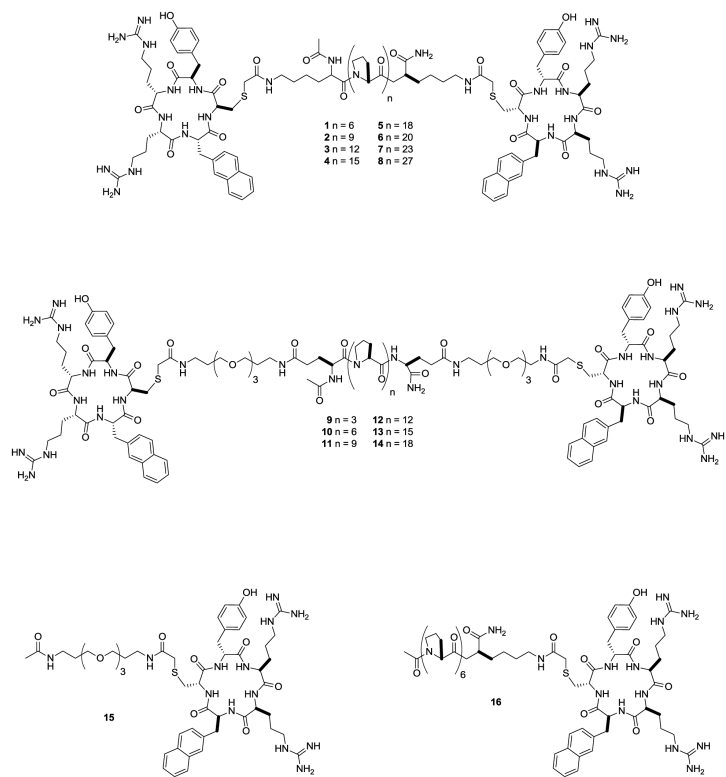


Fig. (3).
 Bivalent ligands for the CXCR4 homodimer with various linker lengths [38].

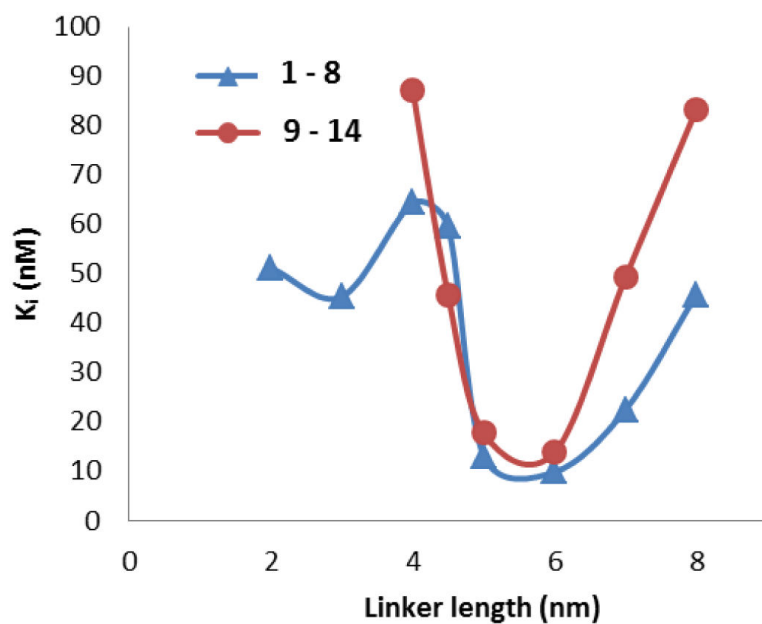


Fig. (4). Relationship of CXCR4 binding affinity (K_i) versus the linker length of compounds **1** through **14** [38].

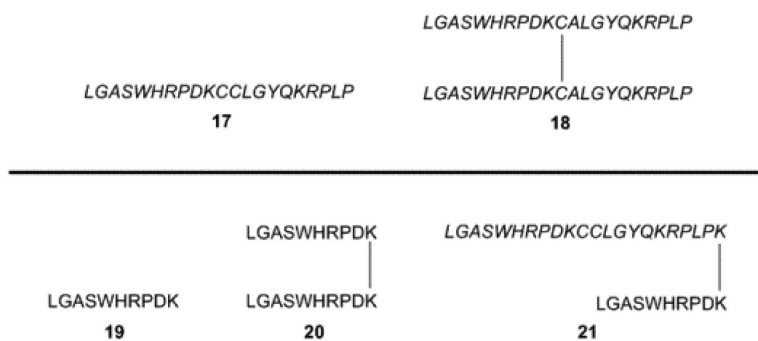


Fig. (5).
 Peptide based bivalent ligands for the CXCR4 homodimer [37,51].

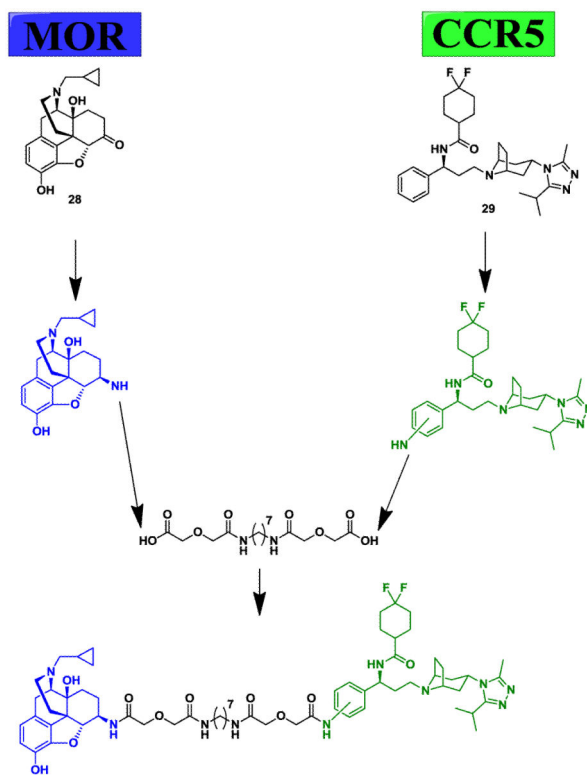


Fig. (7).
Bivalent compound strategy for targeting the CCR5-MOR heterodimer [49].

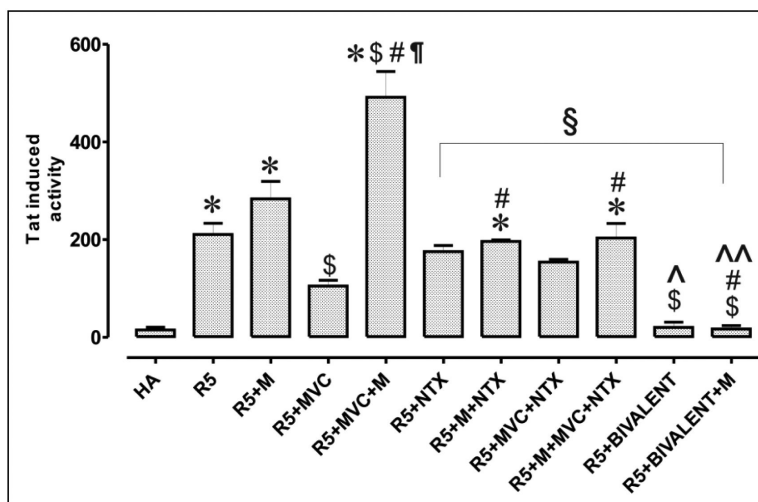


Fig. (8). HIV-1 infection assay. HIV-1_{SF162} infectivity in human glial was determined based on the relative amount of Tat protein expressed by the virus using a luciferase based assay. (HA) human astrocytes, (R5) HIV-1_{SF162}, (M) morphine at 500 nM, (MVC) maraviroc at 100 nM, (bivalent) compound **49** at 100 nM, and (NTX) naltrexone at 1500 nM. Values are absorbance \pm SEM of 3 independent experiments at 18 h post-infection (* p < 0.005 vs. uninfected cells; \$ p < 0.05 vs. R5 HIV-1; # p < 0.05 vs. opioid; ¶ p < 0.05 vs. maraviroc (MVC); § p < 0.05 vs. morphine + MVC; ^ p < 0.05 vs. MVC + NTX; ^^ p < 0.05 vs. morphine + MVC + NTX; Ω p < 0.05 vs. bivalent) [76].

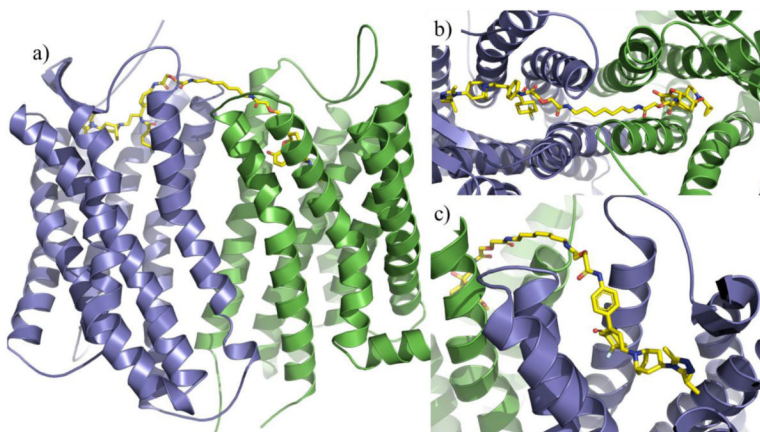


Fig. (9). CCR5-MOR heterodimer model based on CXCR4 dimer crystal structure (PDB code: 3ODU) with bivalent compound **22** bound. The CCR5 is colored in blue whereas the MOR in green. Compound **22** is colored in yellow [110].

Table 1

CCR5 and MOR radioligand binding assay results.

Compound	MOR K _i (nM) ^a	CCR5 K _i (nM) ^b
28	0.7 ± 0.1	-
29	-	0.24 ± 0.06
22	51.8 ± 7.9	239 ± 56
23	-	151 ± 44
24	9.2 ± 3.4	-
27	-	15.3 ± 4.8

^a[³H]naloxone was used in hMOR-CHO membranes.^b[¹²⁵I]MIP-1α was used in CCR5 rhesus macaque membranes. All values are means ± S.E.M. of three independent experiments [75].

Table 2

Antagonism of DAMGO and RANTES stimulated calcium mobilization in hMOR-CHO and MOLT-4 cells respectively.

Compound	MOR IC ₅₀ (nM) ^a	CCR5 IC ₅₀ (nM) ^b
28	8.9 ± 0.9	-
29	-	2.2 ± 0.3
22	40.0 ± 4.8	126 ± 28
23	-	622 ± 36
24	37.8 ± 4.4	-
25	-	7.91 ± 0.76
26	-	1.57 ± 0.18
27	-	14.2 ± 1.9

^ahMOR-CHO cells were stimulated with DAMGO,

^bMOLT-4 cells were stimulated with RANTES, (-) denotes that the compound was not tested [75].