



HHS Public Access

Author manuscript

Crit Rev Toxicol. Author manuscript; available in PMC 2015 April 02.

Published in final edited form as:

Crit Rev Toxicol. 2015 March ; 45(3): 245–272. doi:10.3109/10408444.2014.999365.

Metabolism and metabolites of polychlorinated biphenyls (PCBs)

FA Grimm,

Interdisciplinary Graduate Program in Human Toxicology, Department of Pharmaceutical Sciences & Experimental Therapeutics, University of Iowa

D Hu,

Department of Civil and Environmental Engineering, University of Iowa

I Kania-Korwel,

Department of Occupational & Environmental Health, University of Iowa

HJ Lehmler,

Interdisciplinary Graduate Program in Human Toxicology, Department of Occupational & Environmental Health, University of Iowa

G Ludewig,

Interdisciplinary Graduate Program in Human Toxicology, Department of Occupational & Environmental Health, University of Iowa

KC Hornbuckle,

Interdisciplinary Graduate Program in Human Toxicology, Department of Civil and Environmental Engineering, University of Iowa

MW Duffel,

Interdisciplinary Graduate Program in Human Toxicology, Department of Pharmaceutical Sciences & Experimental Therapeutics, University of Iowa

A Bergman, and

Swedish Toxicology Sciences Research Center (SWETOX), Forskargatan 20, SE-151 36 Södertälje, SWEDEN

LW Robertson*

Interdisciplinary Graduate Program in Human Toxicology, Department of Occupational & Environmental Health, University of Iowa

Abstract

The metabolism of polychlorinated biphenyls (PCBs) is complex and has an impact on toxicity and thereby assessment of PCB risks. A large number of reactive and stable metabolites are formed in the processes of biotransformation in biota in general and in humans in particular. The aim of this document is to provide an overview of PCB metabolism and to identify metabolites of

* **Corresponding author and complete contact info** Larry W. Robertson, Ph.D., M.P.H., ATS, Professor, Department of Occupational and Environmental Health, The University of Iowa, College of Public Health, 100 Oakdale Campus #219 IREH, Iowa City, Iowa 52242-5000, USA, Phone: 319-335-4346, Fax: 319-335-4290, larry-robertson@uiowa.edu.

concern and their occurrence. Emphasis is given to mammalian metabolism of PCBs and their hydroxyl, methylsulfonyl, and sulfated metabolites, especially those that persist in human blood. Potential intracellular targets and health risks are also discussed.

Keywords

Hydroxylated PCBs; polychlorobiphenyls; PCB methyl sulfones; PCB sulfates; human exposure; reactive intermediates

Introduction

Polychlorinated biphenyls (PCBs) are a class of industrial chemicals that were mass-produced globally from the late 1920s until their commercial production was banned, initially by the Toxic Substances Control Act (TSCA) in the United States in 1979, as a reaction to increasing numbers of reports of PCBs in humans and concern for adverse human health effects.

The commercial production of PCBs involved the batch chlorination of biphenyl with chlorine gas in the presence of a catalyst, resulting in the formation of complex mixtures containing a range of PCBs rather than just individual congeners (Erickson, 1997, Hansen, 1999, Erickson and Kaley, 2011). Typically, these PCB mixtures were sold under various trade names (such as Aroclor in the U.S., Kanechlor in Japan, or Clophen in Germany), often with numerical designations indicating percent chlorine by weight (e.g., Aroclor 1254 is 54% and Clophen A60 is 60%) (Silberhorn et al., 1990).

PCBs were used in a wide variety of applications that resulted in release to the environment. Open-ended applications included their uses as plasticizers in rubber and resins, in carbonless copy paper, in adhesives, in wax extenders, in dedusting agents, in paints, and in inks, while nominally closed system uses were in hydraulic fluids, heat-transfer fluids, and in lubricants. Finally, closed system uses of PCBs were primarily in capacitors and in transformers (Kimbrough et al., 1989). The estimated world production of PCBs has been variously estimated from about 1.2 million tons to 2 million tons, of which 0.2 to 0.4 million tons have become “environmentally available” (Tanabe, 1988, ATSDR, 2000). PCBs can now be detected all over the planet, from highly populated areas to the arctic region (Christensen et al., 2010, Gutleb et al., 2010, Macdonal et al., 2000, Ockenden et al., 2001, Wethington and Hornbuckle, 2005, Hu et al., 2010a). Although different for each congener, their resistance towards chemical and biological degradation explains their environmental persistence and omnipresence more than thirty years after their withdrawal from commercial mass production. From a public health perspective, this widespread distribution of PCBs indicates the need for a thorough understanding of the potential adverse effects associated with the parent congeners and their metabolites. This is of particular importance for populations living near, or eating fish from, PCB reservoirs such as the Great Lakes area, the Hudson River in New York or the area around Anniston, Alabama, a former Monsanto production site of PCBs (Custer et al., 2010, Fitzgerald et al., 2008, Goncharov et al., 2010, Goncharov et al., 2011, Silverstone et al., 2012).

PCBs that are the substrates for metabolic attack in exposed individuals arise from two sources. The major source of PCBs for the general population is the food supply (Feinberg et al., 2011, Schechter et al., 2010). A second and often overlooked source of exposure to PCBs is city air and the air of buildings that were constructed using PCBs in sealants, caulking, and other building materials (Ludewig et al., 2008, Persoon et al., 2010). Among these airborne PCBs are also most of the recently discovered nonlegacy PCBs, PCBs that are unintentionally formed as by-products during the manufacture of paints and dyes that are currently sold (Hu and Hornbuckle, 2010, Rodenburg et al., 2010a, Grossman, 2013).

PCBs in the food supply tend to be more highly chlorinated and are therefore poorer substrates for metabolic attack, while airborne PCBs tend to be more volatile, and possess fewer chlorine atoms (McFarland and Clarke, 1989, Robertson and Ludewig, 2011). The former group is longer lived/ more persistent, while the latter one, often referred to as transient or episodic, is composed of PCBs with comparatively short half-lives that are relatively quickly metabolized (Hansen, 2001, Zhao et al., 2010, Robertson and Ludewig, 2011). However, to date the metabolism of PCBs and the physiologic fate of the individual metabolites remain poorly understood.

Many biologic effects of PCBs are receptor-mediated (Luthe et al., 2008), including the well-described characteristics of PCBs as inducers of xenobiotic metabolism (Parkinson et al., 1983, Safe et al., 1985). On the other hand, PCB metabolism is generally regarded as a detoxication process due to the fact that a large proportion of all hydroxylated PCB metabolites (OH-PCBs) being formed are excreted from the body as such or after conjugation (Birnbaum, 1983, Ohta et al., 2015). However, it has become apparent that a variety of PCB toxication processes involves or depends on the metabolism of parent PCBs or their metabolic progeny. Electrophilic metabolic intermediates such as arene oxides may cause harm through their reactions with protein, DNA, or lipids (Pereg et al., 2001, Morck et al., 2002). In the event that OH-PCBs are further oxidized to (semi)quinones, these highly reactive species may also form covalent adducts with proteins, DNA and other endogenous compounds (Lin et al., 2000, Pereg et al., 2001, Robertson and Gupta, 2000, Song et al., 2009, Srinivasan et al., 2002, Amaro et al., 1996). Additionally, metabolites like OH-PCBs, PCB sulfates, and PCB methyl sulfones (MeSO₂-PCBs) might be equally persistent as parent congeners and elicit their own toxicities.

In this review, we will attempt to emphasize the often overlooked issues of metabolism of PCBs and the role metabolism and metabolites play in toxication processes.

Environmental sources of PCBs

Traditional manufactured PCBs

Based on differences in their number of chlorine substituents, PCB congeners can be subdivided into semivolatile and relatively nonvolatile, with the higher chlorinated PCBs (HC-PCBs) typically being less volatile. These differences are also key determinants for their environmental availability and their routes of exposure. The majority of airborne PCBs found in major cities in the United States are lower-chlorinated ones (LC-PCBs) containing four or less chlorine substituents (Hu et al., 2010a, Persoon et al., 2010, Wethington and

Hornbuckle, 2005). Both LC-PCBs and HC-PCBs were contained in traditionally manufactured commercial PCB products, however, HC-PCBs have a higher potential for bioaccumulation and biomagnification along the food chain (Barron et al., 1994, Troisi et al., 2001). As a result, human populations are typically exposed to less volatile, HC-PCBs via contaminated food, particularly fish (Domingo and Bocio, 2007, Weintraub and Birnbaum, 2008). Epidemiological studies have revealed a correlation between the consumption of contaminated fish and increased serum concentrations of such PCBs (Weintraub and Birnbaum, 2008).

Airborne PCBs

Although all PCBs are semivolatile, the LC-PCBs are most commonly reported. The highest concentrations of airborne PCBs are found in indoor air and in industrial and densely populated urban areas, for example in the cities of Chicago, Milwaukee, Toronto, Philadelphia and New York (Breivik et al., 2007, Hu et al., 2010a, Ockenden et al., 2001, Wethington and Hornbuckle, 2005, Sun et al., 2006, Du et al., 2009, Melymuk et al., 2012). The twenty most abundant of these airborne PCBs are summarized in Table 1. Volatilization of airborne PCBs is temperature dependent and can result in their release from environmental or industrial reservoirs, such as rivers, lakes, landfills or contaminated building materials (Persoon et al., 2010, Desborough and Harrad, 2011, Zhang et al., 2011, Simcik et al., 1999, Achman et al., 1993, Hsu et al., 2003, Rudel and Perovich, 2009).

Human populations are more likely being exposed to lower-chlorinated, airborne PCBs by inhalation, rather than by ingestion (Harrad et al., 2009, Robertson and Ludewig, 2011). Indoor inhalation exposure to PCBs is of concern in schools and other buildings that were built in the 1950s and 1960s, as demonstrated by a number of studies investigating indoor exposure to PCBs in the United States and Europe (Herrick, 2010, Herrick et al., 2004, MacIntosh et al., 2012, Gabrio et al., 2000, Jamshidi et al., 2007, Harrad et al., 2010, Zhang et al., 2011). During this time, the caulking and other building materials used in construction contained high levels of PCBs and affected buildings still represent a major source for chronic inhalation exposures. A German study reporting measurements of several indicator PCBs in indoor air of contaminated schools found high concentrations of the lower-chlorinated PCB congeners 28 and 52 (Gabrio et al., 2000). Moreover, there was a significant correlation between PCB exposure and increased blood concentrations of PCBs in teachers who had worked in these contaminated school buildings. Another study reported a correlation between indoor PCB levels and serum PCB concentrations in residents in the Hudson River area (Fitzgerald et al., 2011). The public health relevance of indoor inhalation exposure is further highlighted by several studies indicating that indoor air concentrations of PCBs significantly exceeded those determined in their respective outdoor environments (Jamshidi et al., 2007, Menichini et al., 2007).

Nonlegacy PCBs

The common perception that PCBs are long-banned industrial contaminants that are no longer produced is unfortunately misleading. In recent years, it has been conclusively shown that non-Aroclor, or nonlegacy PCBs contaminate the environment of homes and cities and accumulate in the bodies of exposed populations. In 2008, the Iowa Superfund Research

Program (ISRP) published the results of a large-scale air toxics monitoring program, demonstrating for the first time that PCB 11 (3,3'-di-chlorobiphenyl), a non-Aroclor PCB that has been previously detected in New York/ New Jersey harbor surface and waste water (Litten et al., 2002), was ubiquitous in Chicago (Hu et al., 2008) and in Cleveland (Persoon et al., 2010) air. Since that initial discovery, researchers have reported the presence of nonlegacy PCBs in air samples around the world and it has been revealed that the most likely source of these contaminants is volatilization from common household paint (Basu et al., 2009, Choi et al., 2008). In 2010, a study of PCBs in pigments manufactured and sold as colorants in household paint revealed the presence of more than 50 nonlegacy PCBs (Hu and Hornbuckle, 2010). PCB 11 was the most commonly detected congener but many other PCBs, including several dioxin-like PCBs, were also highly abundant. Yellow, green, blue, and red pigments also contained PCB congeners, while white, black, and brown did not. The distinctive pattern of PCB congeners in the different pigments was an important clue to the chemical manufacturing processes that inadvertently produced these unwanted byproducts. Since then additional studies have shown the widespread presence of non-Aroclor PCBs in the environment and in consumer products. Their presence is most likely due to use of paint and pigments containing PCBs.

Even though they are only being discovered now, non-Aroclor PCBs have been present in the environment for at least 80 years, as evidenced by the presence of paint pigment derived PCBs in sediment core samples from the Great Lakes (Hu et al., 2011). As opposed to the Aroclor PCBs, however, PCB congeners associated with pigments are not exhibiting strong declines in the environment. And unlike Aroclors, these PCB congeners are still legally produced and distributed in the public. At this point, it has become clear that pigments are an important source of PCBs into the environment, especially into air but there are probably other sources. Theoretically, PCBs can be unintentionally produced from any chemical process that involves carbon, chlorine, and elevated temperatures or catalysts (Rudel and Perovich, 2009, Erickson, 2001). Consequently, many consumer products, including building materials that have been manufactured involving these chemical processes, might be tainted with PCBs which are eventually released into the environment (Rodenburg et al., 2010b, Shang et al., 2014).

Despite this knowledge, human uptake, metabolism and toxic response to PCB 11 and other nonlegacy PCBs remains poorly understood. Initial laboratory studies with animals and plants have demonstrated the potential of PCB 11 to become bioavailable to humans through inhalation and to be accessible for biotransformation (Hu et al., 2013, Zhu et al., 2013, Hu et al., 2012, Hu et al., 2010b, Hu et al., 2014). In 2013, PCB 11 was reportedly detected in human serum for the first time, thereby further emphasizing the need for a better understanding of the exposure, metabolism and toxicities of nonlegacy PCBs (Marek et al., 2013b).

PCB metabolism and relevant classes of PCB metabolites

Nomenclature of PCB metabolites

PCB metabolite nomenclature is based on PCB structure as originally numbered by Ballschmiter and coworkers (Ballschmiter et al., 1993). The Ballschmiter system assigns

congeners in ascending numerical order according to their chlorination status beginning with monochlorinated PCBs and ending with the only decachlorinated congener, PCB 209. Accordingly, it is not possible to derive the nomenclature from the correct IUPAC name of the metabolite. A short stepwise guidance for abbreviations of hydroxyl, methylsulfonyl, sulfate, and glucuronic acid metabolites of PCBs can be derived from the nomenclature system originally proposed for OH-PCBs and MeSO₂-PCBs (Maervoet et al., 2004). **Step 1:** Identify the Ballschmiter number of the PCB as if no substituents are attached to the molecule and note which one of the two phenyl rings is carrying the primed chlorine positions. **Step 2:** A functional group (e.g. hydroxyl (-OH), methylsulfonyl (-MeSO₂) or sulfate (-OSO₃⁻) is given a non-primed or a primed number depending on the position where it is attached to the biphenyl moiety. Examples of full metabolite nomenclature are given in Tables 2 and 3. Eight-hundred-thirty-seven mono-hydroxylated metabolites are theoretically possible from the 209 PCB congeners, and this would also be true of the analogous PCB sulfates and PCB glucuronides derived from OH-PCBs. These OH-PCB congeners are listed in Table 2. In Table 3, those OH-PCBs identified in human plasma are listed together with their suggested or experimentally determined parent PCB congeners (Sjödén et al., 1998).

PCB metabolism

The rate and extent of PCB metabolism depends on the number and positions of chlorines in the molecule (Kato et al., 1980, Matthews and Anderson, 1975, Mills et al., 1985, Schnellmann et al., 1985). Overall, the fewer number of chlorine atoms on biphenyl, the faster the metabolism. Also, availability of vicinal non-chlorine substituted positions, especially in *meta*- and *para*-positions of the biphenyl core, increases the chances of cytochrome P450 (CYP) mediated transformation (Mills et al., 1985). Consequently, the fate of individual PCBs within the human body depends on their structural properties. Conventional HC-PCBs are fairly resistant towards biotransformation reactions and due to their high lipophilicity, they tend to be retained in adipose tissues or in plasma where they can be frequently detected at high concentrations of more than 10 µg/g lipid weight (Fangstrom et al., 2002, Kutz et al., 1991). LC-PCBs, by contrast, are often transiently detected in serum and their rapid disappearance is assumed to be related to their higher susceptibility for metabolic conversion (Robertson and Ludewig, 2011, Hansen, 1999)

The main pathways for PCB metabolism are shown in Figure 1 (Safe, 2001, Letcher et al., 2000), with enzymes involved suggested. The initial step includes the oxidation of PCBs to OH-PCBs by hepatic cytochromes P-450 (CYP) enzymes (Bergman et al., 1994b, Matthews and Kato, 1979). Structurally different PCB congeners may be metabolized by different enzymes of the CYP superfamily. Non-ortho substituted (so called co-planar or dioxin-like) PCB congeners are metabolized predominately by CYP1A enzymes, while multiple ortho-substituted PCBs are substrates for CYP2B enzymes (Kaminsky et al., 1981, Lu et al., 2013, Lu and Wong, 2011, McGraw and Waller, 2006, Waller et al., 1999, Warner et al., 2009). These observations are also confirmed by more efficient binding of structurally related PCBs to appropriate cytochrome P-450 isoforms (Hrycay and Bandiera, 2003, Kania-Korwel et al., 2008a).

Depending on the PCB congener, the initial CYP-dependent monooxygenation can result from direct electrophilic addition of oxygen or may involve the formation of a transient reactive arene oxide (Guengerich, 2001, Jerina and Daly, 1974, Preston et al., 1983). In fact, OH-PCBs are most easily formed from PCBs with 2,3-, 2,5- and 2,6-dichlorination or 2,3,6-trichlorination patterns or PCB epoxides where the phenyl ring has none or a lower number of chlorine atom substituents in any position except the 4-position. PCB congeners with 4-, 3,4-, 3,5-, 2,4,5-, 2,3,4,6- or 2,3,5,6- chlorine substitutions are less efficiently metabolized and they tend to form epoxides between a chlorine substituted carbon and an unsubstituted carbon. Each of the epoxides will yield two isomeric OH-PCBs. While OH-PCBs with at least one unsubstituted carbon next to the phenol group appear to be rapidly eliminated from organisms, including humans, the chlorine may undergo a 1,2-shift (Guroff et al., 1967) yielding rearranged polychlorobiphenyls. OH-PCBs with neighboring chlorine substituted carbons to the phenol group are retained in the blood (Letcher et al., 2000).

Transient PCB epoxides may also form adducts with biomacromolecules, isomerize to mono-hydroxy PCBs or hydrolyze to form PCB dihydrodiols (McLean et al., 1996b, McLean et al., 1996a, Kaminsky et al., 1981). Alternatively, PCB epoxides may react with glutathione (GSH) to form a dihydro-glutathione-hydroxyl-substituted PCB metabolite that can form a fully aromatic PCB glutathione conjugate through the loss of water (Figure 1) (Letcher et al., 2000, Bakke et al., 1983, Bakke et al., 1982). Each PCB undergoing this type of metabolism will yield two PCB glutathione conjugates. These GS-PCBs are degraded in a stepwise fashion to the corresponding cysteine conjugates that may be *N*-acetylated to form mercapturic acids, a route of degradation known as the mercapturic acid pathway (Bakke et al., 1982, Letcher et al., 2000). The corresponding PCB thiols are formed via cysteine *S*-conjugate β -lyase catalyzed cleavage of the C-S bond in the cysteine conjugate. The PCB thiols are methylated and oxidized in two steps to sulfoxides and finally to the corresponding MeSO₂-PCBs, highly hydrophobic Lewis bases (Kallenborn and Huhnerfuss, 2001).

OH-PCBs have also been shown to undergo multiple oxidation reactions leading to more than a single hydroxyl substituent on the biphenyl structure (Bergman et al., 1994b, James, 2001). Alternatively, the formation of dihydroxylated PCB metabolites can be directly catalyzed from parent PCBs by CYP2B enzymes (Lu et al., 2013, McLean et al., 1996a, Waller et al., 1999). Dihydroxylation can result in the formation of catechols and other hydroquinones (McLean et al., 1996a), that possess toxicological relevance in that they promote oxidative stress (Song et al., 2008b). OH-PCBs also represent substrates for conjugation reactions catalyzed by sulfotransferases (SULTs) or UDP-glucuronosyl transferases (UGTs) to yield their respective sulfate or glucuronic acid conjugates (Daidoji et al., 2005, Dhakal et al., 2012, Matthews and Kato, 1979).

Chirality of PCB metabolites

Nineteen PCBs and their metabolites are optically active (or chiral) because they exist as stable rotational isomers, called atropisomers, that are non-superimposable mirror images of each other (Lehmler et al., 2010, Mannschreck et al., 1985, Püttmann et al., 1989, Püttmann et al., 1986). Only non-coplanar PCBs with three or four *ortho*-chlorine substituents possess

sufficient torsional strain to facilitate the formation of stable atropisomer pairs that can be separated chromatographically (Kaiser, 1974, Norstrom et al., 2006, Haglund, 1996). PCBs 84, 91, 95, 132, 136, 149, 174 and/ or their metabolites have been shown to accumulate enantioselectively in mammalian tissues (Lehmler et al., 2010), in birds (Jörundsdottir et al., 2006), in plants (Zhai et al., 2011) and in humans (Ellerichmann et al., 1998, Hovander et al., 2004).

Theoretically, 456 of the 837 possible mono-hydroxylated PCB metabolites are chiral (Nezel et al., 1997). These atropisomers may or may not be generated in a 1:1 ratio, but the metabolites, like PCB atropisomers, seem to have different half-lives in animal tissues and in biota (Püttmann et al., 1989, Larsson et al., 2004, Larsson et al., 2002, Chu et al., 2003a). The atropselective formation of chiral OH-PCBs has been shown both *in vivo* (Kania-Korwel et al., 2008b, Kania-Korwel et al., 2012) and *in vitro* (Kania-Korwel et al., 2011, Wu et al., 2011, Lu et al., 2013, Wu et al., 2013a, Wu et al., 2014, Wu et al., 2013b, Zhai et al., 2013b). The recently identified OH-PCB metabolites of five chiral PCBs, PCB 91, 95, 132, 136, and 149, are chiral themselves, but were not previously identified in environmental samples, including human blood, due to the lack of authentic standards. The atropselective formation of these OH-PCBs results in changes of enantiomeric fractions of the parent compound (Warner et al., 2009). It was demonstrated using pure atropisomers, that biotransformation of (–)-PCB 136 leads to the formation of single enantiomer of 5-OH-PCB 136, while the biotransformation of (+)-PCB 136 results in the formation of the other enantiomer of that major metabolite (Wu et al., 2011). Considering that pure PCB atropisomers can elicit different toxicological responses (Pessah et al., 2009, Lehmler et al., 2005, Yang et al., 2014), these findings may have implications for risk assessment associated with those metabolites. Optically active MeSO₂-PCBs identified in humans and laboratory animals to date are atropisomers of 5'-MeSO₂-PCB 132 and 3-MeSO₂-PCB 149 (Ellerichmann et al., 1998, Norstrom et al., 2006).

Reactive (epoxide and (semi)quinone) PCB intermediates

Hepatic microsomes are capable of metabolizing lower chlorinated biphenyls, mono-, di-, and trichlorobiphenyls to catechols and hydroquinones (Robertson and Gupta, 2000, McLean et al., 1996a, Oakley et al., 1996). Likewise, the potential for microsomal formation of PCB catechols derived from penta- and hexachlorinated PCBs (e.g. PCB 136) was demonstrated (Lu et al., 2013, Wu et al., 2013a, Wu et al., 2014). One-electron oxidation of a PCB hydroquinone or catechol, or single-electron reduction of a PCB quinone, results in a semi-quinone radical with subsequent formation of reactive oxygen species (e.g. superoxide anion radical, hydrogen peroxide, and hydroxyl radical) and the PCB quinone (Song et al., 2008a, Song et al., 2008b). In addition to the potential for generation of toxic oxygen species, the metabolic pathways of PCBs may include the formation of electrophilic PCB arene oxides and quinones that may bind to nucleophilic sites on cellular macromolecules (Robertson and Gupta, 2000, Lin et al., 2000, Qin et al., 2013, Wangpradit et al., 2009, James, 2001). In fact, a large number of *in vitro* studies have demonstrated adduct formation of PCBs and their metabolites, in particular PCB quinones, to proteins, RNA, DNA or lipids (Robertson and Gupta, 2000, Morck et al., 2002, Ludewig, 2001, Klasson Wehler et al., 1989, Klasson Wehler et al., 1993, Zhao et al., 2004).

Even though most evidence of PCB adduct formation points towards a primary involvement of LC-PCBs, there is limited evidence available supporting the potential of HC-PCBs for adduct formation with DNA/RNA and/or protein. An *in vivo* study in mice demonstrated covalent binding of 2,2',3,3',6,6'-hexachlorobiphenyl (PCB 136) to RNA, proteins, and DNA in liver, muscle, and kidneys and of 2,2',4,4',5,5'-hexachlorobiphenyl (PCB 153) to RNA and proteins in liver (Morales and Matthews, 1979). Another study showed binding of PCB 153 to nuclear proteins and DNA in livers of treated rats (Daubeze and Narbonne, 1984). Further evidence of the presence of reactive intermediates of PCBs forming reaction products with biomolecules is the observation that non-extractable residues are present after exposure to radiolabeled PCBs (Pereg et al., 2001, Klasson Wehler et al., 1989, Klasson Wehler et al., 1993, Morck et al., 2002, Tampal et al., 2003). However, the identity of these adducts have so far only been poorly characterized, although two classes of PCB electrophiles, arene oxides and (semi)quinones, appear to be involved. The binding to lipids appears to involve phospholipids (Morck et al., 2002).

The ability of eight mono- to hexachlorinated biphenyls to form DNA adducts following bioactivation with hepatic microsomes from different species (rat, mouse, and human) was investigated (Pereg et al., 2002). Interestingly, only the lower chlorinated congeners with up to three chlorine atoms were capable of DNA adduction. Based on structural identification of PCB adducts to DNA, the suggested formation of DNA adducts involves PCB quinone metabolites (Zhao et al., 2004).

Binding indices have shown 15- to 30-fold greater binding of PCBs to peptides than to DNA (Pereg et al., 2001). Protein adduction is initiated through a reaction of the PCB metabolites with nitrogen and sulfur nucleophiles (Amaro et al., 1996). Sulfur nucleophiles are much more reactive with PCB-derived electrophiles than nitrogen nucleophiles. Therefore one would predict that protein sulfhydryls would be a preferred target (both on the basis of reactivity and abundance). In addition, using glutathione (GSH) as a target peptide, two distinctive, structure-dependent mechanisms by which PCB quinones are capable of forming protein adducts were revealed (Song et al., 2009). While PCB quinones without chlorine substituents in the quinone ring typically undergo a Michael addition to form a GSH adduct, GSH adduction of PCB quinones with chlorine substituents in the quinone ring tends to involve nonenzymatic displacement of such chlorines. Lin et al. reported an implied involvement of reactive quinones in the liver and brain of rats exposed to 2,2',5,5'-tetrachlorobiphenyl (PCB 52) (Lin et al., 2000). The adducts seemed to be unstable *in vivo* since the estimated half-life of the adducts was 2.5-fold shorter than the turnover rate of liver cytosolic protein.

Altogether, it is evident that PCBs are precursors of biomacromolecule adducts. This has most clearly been shown for the lower chlorinated biphenyls, but a number of studies on higher chlorinated biphenyls confirm the possibility of adduct formation. This is mediated by the metabolic formation of reactive intermediates, arene oxides, and (semi)quinones.

Hydroxylated PCB metabolites

Some OH-PCBs are stable (and extractable) metabolites that are retained in the body. Approximately 40 different OH-PCBs have been identified in human blood as first reported

by Bergman (Bergman et al., 1994a) and thereafter in a series of other studies (Marek et al., 2013b, Quinete et al., 2014, Park et al., 2008, Park et al., 2007, Park et al., 2009), as well as in blood from wildlife (Fängström et al., 2005a, Letcher et al., 2000, Gutleb et al., 2010). There are 837 possible OH-PCB congeners (Table 2); the chemical structures of several that have been commonly detected in human blood (Table 3) are shown in Figure 2. All of these have 5 or more chlorines. Human plasma or serum concentrations of these most abundant OH-PCB congeners, as determined in human populations from Canada, the Faroe Islands, Latvia, the Netherlands, Nicaragua, Russia, Slovakia, and Sweden, are given in Table 4. All studies, except for the study of humans from the Russian Arctic, provided comparisons with PCB 153, the most abundant PCB congener. The OH-PCBs are present in concentrations similar in range as many PCB congeners, except for those PCBs that are the most prevalent/persistent. Since OH-PCBs in the blood are reversibly bound to proteins, their accumulation is not lipid dependent (Gutleb et al., 2010, Purkey et al., 2004). Therefore it is more correct to report concentrations in serum or plasma on the basis of fresh weights. However, for comparative reasons the concentrations are often provided based on lipid weight. It was shown that the OH-PCB congener concentrations may reach median concentrations of 2–300 ng/g fat, with individual levels up to 940 ng/g lipid weight (Table 4). Comparisons with plasma levels of PCB 153 indicate the high prevalence of various OH-PCBs. In fact, plasma concentrations of the most abundant OH-PCB congeners reach about 30% of those determined for PCB 153, with a variation among studies of 11–82%. The practical implication is that OH-PCBs present at the highest concentrations always exceed the concentrations of a large number of individual PCB congeners.

OH-PCBs are efficiently transferred from the maternal to the fetal blood via the placenta (Gruvenius et al., 2003, Meerts et al., 2002, Morse et al., 1996, Soechitram et al., 2004). In fact, the transplacental transfer seems to be more efficient for the OH-PCBs (Lucier et al., 1978) than for the PCBs (Gallenberg et al., 1990, Ring et al., 1988) themselves. This is most likely due to a lower relative lipid content in the blood of the fetus. It is also necessary to consider this difference in maternal-fetal transfer from a risk perspective. On the other hand, basically no, or a very limited, transfer of OH-PCBs occurs from mothers to their nursing children via the milk, which was shown both in laboratory animals (Meerts et al., 2002, Morse et al., 1996) and in humans (Malmberg et al., 2004, Fangstrom et al., 2005, Gruvenius et al., 2002). The concentrations of PCB 153 in mother's milk have been shown to be almost three orders of magnitude higher than those determined for 4-OH-PCB 187, the most common OH-PCB. This partitioning behavior is presumably due to the more polar and less lipophilic characteristics of the OH-PCBs.

To date, the *in vivo* formation of OH-PCBs from LC-PCBs has not been extensively characterized and deserves further scientific attention. Initial studies using rats exposed to PCB 11 by inhalation exposure, however, clearly indicate their rapid hydroxylation and subsequent elimination (Hu et al., 2014, Hu et al., 2013). While many OH-PCB metabolites are good substrates for conjugation reactions (James, 2001), the ease of formation of glucuronic acid and sulfate conjugates is highly structure-dependent, an observation that is supported by the persistence of certain OH-PCBs in serum (Tampal et al., 2002, Liu et al., 2009, Bergman et al., 1994b, Gutleb et al., 2010). In addition, both conjugation reactions,

glucuronidation and sulfation, may be inhibited by OH-PCBs (Ekuase et al., 2011, James, 2001, Kester et al., 2000, Liu et al., 2006, Liu et al., 2011, Liu et al., 2009, Schuur et al., 1998b, van den Hurk et al., 2002, Wang et al., 2006).

PCB methyl sulfones

About 60 MeSO₂-PCBs have been detected in environmental samples (Letcher et al., 2000). The precursor PCB congeners are well known for all MeSO₂-PCBs that so far have been identified in humans and in wildlife. All the 22 MeSO₂-PCB metabolites that have been structurally identified, and their parent PCB congeners, are listed in Table 5. In addition, recent data show the occurrence of up to fifty or more MeSO₂-PCB congeners in human serum, but so far the majority of these are still not structurally identified (Hovander et al., 2006). Studies on the formation and retention of MeSO₂-PCBs in humans and wildlife published through 1999 were reviewed in detail by Letcher and co-workers (Letcher et al., 2000).

Given their neutral lipophilic character, MeSO₂-PCBs are present primarily in body lipids. In humans, MeSO₂-PCBs that have been identified are generally present only at low concentrations, 1% or less compared to the PCB concentrations. They accumulate with high selectivity in certain tissues, such as liver and lung (Letcher et al., 2000, Bergman et al., 1979, Larsson and Bergman, 1998). The MeSO₂-PCB concentrations are notably high in the liver of mammals (Larsson and Bergman, 1998, Bergman et al., 1994b) and likewise, human livers have been shown to accumulate MeSO₂-PCBs (Weistrand and Noren, 1997, Ellerichmann et al., 1998). The majority of these metabolites have the MeSO₂ substituent in the *meta*-position of their respective biphenyl ring. The binding characteristics of MeSO₂-PCBs in liver tissue have not been elucidated, but 4-MeSO₂-PCBs that accumulate in lung bronchial epithelium of rodents were specifically located in non-ciliated cells, the Clara cells, which contain uteroglobin, alternatively known as PCB binding protein (Lund et al., 1985, Lund et al., 1987). This accumulation in the lung may be seen as an excretion pathway for the MeSO₂-PCBs. However, if transported up the airways they may be swallowed and possibly reabsorbed from the gut. Data on human concentrations of MeSO₂-PCBs are summarized in Table 6.

PCB sulfates

A number of studies have been published that identified OH-PCBs as substrates and inhibitors for recombinant human sulfotransferases SULT1A1, SULT1E1, and SULT2A1 (Ekuase et al., 2011, Kester et al., 2000, Liu et al., 2006, Wang et al., 2006). Moreover, the susceptibility of several OH-PCBs to serve as substrates or inhibitors for rat SULT1A1 and rat SULT2A3, the presumptive rat homolog for human SULT2A1, was assessed (Liu et al., 2009). The majority of the OH-PCBs examined were found to be excellent substrates for certain SULT isoforms while being inhibitors for others. A general conclusion of these studies on structure-activity relationships was that 4'-hydroxylated PCBs without chlorine substituents in the adjacent 3' or 5' positions were typically substrates for SULT1A1 and inhibitors for SULT2A1, whereas OH-PCBs possessing the 3',5'-dichloro-4'-hydroxy substitution pattern were usually identified as substrates for SULT2A1 and inhibitors of SULT1A1. The latter group also constitutes the most potent inhibitors of SULT1E1 (Kester

et al., 2000). Based on the small number of OH-PCBs with only a single chlorine atom in a position *ortho* to the phenol that have been examined to date, they appear to have more subtle and complex interactions with family 1 and 2 SULTs that affect substrate/inhibitor specificities of the enzymes.

However, evidence for the formation of PCB sulfates *in vivo* was lacking until sulfate ester metabolites derived from PCB 3 were recently detected in serum and urine samples collected from male Sprague-Dawley rats exposed to PCB 3 by intraperitoneal injection or inhalation (Dhakal et al., 2013, Dhakal et al., 2012, Dhakal et al., 2014). Interestingly, PCB sulfates appeared to be major metabolites in these studies, far outweighing glucuronidation. Moreover, the most abundant PCB sulfate in serum was identified as 4'-OSO₃⁻-PCB 3 and its serum concentrations greatly exceeded those determined for 3'-OSO₃⁻-PCB 3 and 2'-OSO₃⁻-PCB 3, thereby indicating the 4' position on the unsubstituted aryl ring as a primary target for oxidation and subsequent sulfation reactions. In addition, serum concentrations of 4'-OSO₃⁻-PCB 3 were approximately 60-fold higher than those of its hydroxylated precursor, 4'-OH-PCB 3. Thus, this indicated that sulfation is a major metabolic pathway for this lower-chlorinated OH-PCB in the rat. The presence of sulfate metabolites derived from PCB 3 has also been confirmed in poplar plants (Zhai et al., 2013a).

PCB metabolite associated toxicities

Despite the fact that xenobiotic metabolism is primarily regarded as a detoxication process, there is increasing evidence that various classes of PCB metabolites exhibit their own toxicities, including carcinogenic, neurologic and endocrine effects (Brouwer et al., 1999, Knerr and Schrenk, 2006, Silberhorn et al., 1990, Tilson and Kodavanti, 1998). These findings raise important questions such as whether many of the toxic effects previously attributed to parent PCBs are in fact caused by their metabolites. In this section, we attempt to summarize the scientific evidence for the involvement of PCB metabolites in PCB-associated toxicities with an emphasis on carcinogenesis and thyroid disruption.

Reactive (epoxide and (semi)quinone) PCB intermediates

PCBs have recently been classified as “carcinogenic to humans” by the International Agency for Research on Cancer (IARC) (Lauby-Secretan et al., 2013). While this classification was based on positive correlation between PCB exposure and incidences of melanoma and limited evidence of an involvement in the development of breast cancer and non-Hodgkin lymphoma, exposure to PCBs has also been correlated with an increased incidence of other malignancies including hepatocellular carcinoma and lung cancer (Onozuka et al., 2009, Todaka et al., 2009). In addition, exposure to several Aroclor PCB mixtures resulted in the formation of thyroid neoplasms in male Sprague-Dawley rats, a finding that is in agreement with similar *in vivo* studies (Mayes et al., 1998, Vansell et al., 2004). Although for a long time PCBs were thought to be strictly promoting carcinogens, a lower chlorinated congener (PCB3) was shown to induced point mutations in rat livers *in vivo* (Lehmann et al., 2007), and, in agreement with this, recent evidence suggests that particularly the lower-chlorinated congeners can be oxidized to genotoxic metabolites, such as arene oxides and quinone species (Espandiarri et al., 2003, Espandiarri et al., 2004, Robertson and Ludewig, 2011). In fact, the quinone metabolites of PCB3 increase gene

This finding was in agreement with previous observations *in vivo* indicating that LC-PCBs were more estrogenic than higher-chlorinated congeners, which could be a result of their increased potential for metabolic conversion (Bitman and Cecil, 1970, Hansen, 1998). Also, exposure of ovarian cells to ng/ml amounts of PCB3 or its hydroxylated metabolites increased secretion of estradiol, an effect that was caused by an increase in aromatase activity and which was most pronounced with the dihydroxy PCB3 (Ptak et al., 2005, Ptak et al., 2006). In addition to exerting direct estrogenic and antiestrogenic effects through receptor binding or increased estrogen production, OH-PCBs have also been shown to induce estrogenicity indirectly through inhibition of the estrogen sulfotransferase (i.e., SULT1E1) (Kester et al., 2000, Kester et al., 2002). Inhibition of this SULT isoform slows down the rate of inactivation of estrogens by sulfation and thus results in increased levels of the active hormones.

OH-PCBs have also been identified as one class of PCB metabolites that act as disruptors of thyroid homeostasis (Meerts et al., 2002, Morse et al., 1996, Meerts et al., 2004) and a clear relationship between elevated OH-PCB levels and decreased thyroid hormone levels has been observed in animal (Dallaire et al., 2009, Otake et al., 2007, van den Berg et al., 1988, Morse et al., 1993) and human (Kato et al., 2004) studies. Serum hypothyroxinemia is among the most frequently reported adverse health effects in human populations exposed to PCBs (Kodavanti and Curras-Collazo, 2010, Patrick, 2009). Alongside observed direct thyroid effects in wildlife, these observations led to the classification of PCBs as thyroid-disrupting chemicals (TDCs) (Knerr and Schrenk, 2006, Mayes et al., 1998, Pearce and Braverman, 2009). Due to the role of thyroid hormones as stimulants and regulators for cellular proliferation and differentiation, thyroid disruption is particularly critical during human fetal development (Patrick, 2009, Giera et al., 2011). Serum concentrations of both the pro-hormone L-thyroxine (T_4) and the active hormone triiodothyronine (T_3) are extremely low (3.5–6.5 pM free T_3 , 0.9–2.8 nM total T_3 , 10–23 pM free T_4 , 58–161 nM total T_4) and even small changes in these concentrations can result in developmental toxicities (Patrick, 2009). Neurodevelopmental effects of alterations in thyroid hormones have been observed in infants and laboratory animals as a result of subclinical maternal hypothyroidism during the first trimester of pregnancy, and include decreased cognitive and motor function, mental retardation, and low IQ scores (Vermiglio et al., 2004, Zoeller and Rovet, 2004). Epidemiological studies report visual recognition deficits, impaired executive functioning and speech problems as adverse neurodevelopmental effects in PCB-exposed populations (Jugan et al., 2010, Pearce and Braverman, 2009). In adults, subclinical hypothyroidism primarily affects lipid metabolism and is associated with increased serum lipid concentrations and a higher incidence of obesity (Asvold et al., 2007, Patrick, 2009). These effects may also contribute to the increased risk of cardiovascular effects such as atherosclerosis and myocardial infarction that was observed in PCB exposed populations (Razvi et al., 2008).

Interestingly, while most studies report negative correlations between PCB and thyroid hormone levels, fewer studies indicate positive correlations, particularly between serum PCB levels and free thyroid hormone concentrations (Bloom et al., 2009, Langer et al., 2008). In addition, a few studies report no correlation between elevated plasma PCB and

thyroid hormone levels (Jugan et al., 2010). These findings indicate that thyroid effects may not just be related to the extent of the exposure but are in fact congener- or metabolite-dependent. *In vivo* animal models clearly indicate a negative correlation between PCB concentrations and T₄ levels (Jugan et al., 2010, Meerts et al., 2002, Meerts et al., 2004). PCBs have been suggested to affect the thyroid homeostatic system at various stages in a congener-dependent manner and there is increasing evidence for distinct roles of various classes of PCB metabolites.

A suggested major contributing mechanism in environmental contaminant induced hypothyroxinemia is the displacement of T₄ from its binding sites on the thyroid hormone transport protein transthyretin (TTR) (Kodavanti and Curras-Collazo, 2010). As opposed to parent PCBs, all OH-PCBs retained in human and wildlife blood seem to be competitors for T₄ binding sites on TTR (Brouwer et al., 1998, Malmberg et al., 2004, Darnerud et al., 1996, Gutleb et al., 2010). In addition, the binding potencies of OH-PCBs have been shown to be up to an order of magnitude greater than the binding of T₄ (Rickenbacher et al., 1986, Chauhan et al., 2000). Moreover, TTR has been suggested to facilitate the transport of bound ligands across the blood-brain barrier and the placenta and thus, binding to TTR may also play a role in the distribution of OH-PCBs to the placenta and the brain (Brouwer et al., 1998, Lans et al., 1993, Mortimer et al., 2012). In fact, exposure of pregnant mice to PCBs resulted in elevated serum and brain levels of OH-PCBs in the developing fetus (Meerts et al., 2002, Meerts et al., 2004, Morse et al., 1995, Morse et al., 1996). After birth, the OH-PCBs in brain were no longer detected, suggesting an increased susceptibility exists in this highly sensitive moment of brain development (Jacobson et al., 1990, Boucher et al., 2009, Darras, 2008).

Another potential mechanism by which OH-PCBs interfere with thyroid homeostasis is represented by the inhibition of SULT-catalyzed sulfation of thyroid hormones (Schuur et al., 1998a), since OH-PCBs are a well-known class of substrates and inhibitors of a variety of sulfotransferase enzymes (Liu et al., 2009, Ekuase et al., 2011, Kester et al., 2000, Kester et al., 2002, Wang et al., 2006).

OH-PCBs also possess neurotoxic potential (Pessah et al., 2006, Londono et al., 2010, Sharma and Kodavanti, 2002) in that they have been shown to interact more potently with the RyR receptors than parent compounds (Pessah et al., 2006) and they have been shown to affect the Ca₂₊ homeostasis in neuronal cells (Londono et al., 2010). More recently it was revealed that OH-PCB induced muscle dysfunction actually depends on their interactions with RyR receptors (Niknam et al., 2013).

PCB methyl sulfones

The selective tissue retention of MeSO₂-PCBs in lung tissue has been hypothesized to be at least partially accountable for respiratory problems in Yusho patients (Kato et al., 1995). In liver, several MeSO₂-PCBs have been found to induce microsomal drug metabolizing enzymes, and in some cases metabolites were even more potent than the respective parent compounds (Kato et al., 1995, Kato et al., 1999, Kato et al., 1997). In particular, certain MeSO₂-PCBs were capable of inducing CYP2B1, CYP2B2, CYP3A2, and CYP2C6 (Kato et al., 1995). In addition, they were strong inhibitors of CYP11B1, an enzyme that is

required in corticosterone biosynthesis and antagonists for the glucocorticoid receptor, which further indicated their potential as endocrine disruptors (Johansson et al., 1998). The MeSO₂-PCBs may also induce UDP-glucuronosyltransferase (Kato et al., 2000), which in turn was hypothesized to affect the thyroid hormone homeostasis (Kato et al., 2000). Finally, MeSO₂-PCBs are also capable of reducing thyroid hormone levels in laboratory animals (Kato et al., 1998, Kato et al., 1999, Kato et al., 2000). These effects, were observed both in mink dams and also in their offspring, where metabolites were transferred from mothers (Lund et al., 1999).

PCB sulfates

While OH-PCBs have long been known as a class of competitive ligands for TTR, we have recently demonstrated that PCB sulfates represent another group of high-affinity ligands for the T₄-binding sites on the protein (Grimm et al., 2013). Interestingly, the sulfates bound with similar or higher affinity than their corresponding OH-PCBs, which for the first time indicated a potential toxicological significance of PCB sulfates in thyroid hormone disruption. This increased affinity as compared to OH-PCBs may be the result of the presence of the anionic sulfate group, which resembles the alanyl moiety on T₄ and is assumed to facilitate hydrogen bonding interactions with lysine residues in the T₄ binding site. Authentic PCB sulfate standards were not available until relatively recently (Li et al., 2010), and this may explain the lack of previous mechanistic studies on their biological and/or toxicological potential.

Toxicological relevance of PCB metabolites and research needs

So why should we care about the formation and physiologic fate of PCB metabolites, particularly those derived from the more readily metabolized LC-PCBs?

First, due to their environmental persistence and widespread occurrence, traditionally manufactured PCBs remain a public health hazard. Even though human exposure levels appear to be overall declining (Jugan et al., 2010), the measured PCB concentrations can vary greatly between exposed individuals, from almost undetectable to peak concentrations of more than 100 µg/g lipid weight (Langer et al., 2008). Assuming an average plasma lipid concentration of 7.35 g/l (El Majidi et al., 2014) and an average molecular weight of 300 g/mol for PCBs, these 31 measurements translate into plasma PCB concentrations from the picomolar to low micromolar range (2.45 µM). OHPCB and MeSO₂-metabolite concentrations are frequently correlated to parent PCB levels and are typically one to two orders of magnitude below PCB concentrations, a finding that is reflective of the relative resistance toward bioconversion exhibited by mostly HC-PCBs (Letcher et al., 2000, Marek et al., 2013b).

Second, recent evidence for the omnipresence of lower-chlorinated and/or nonlegacy PCBs in our environment and the increasing evidence for their associated toxicities, clearly indicate their role as chemicals of interest from a public health perspective. A target group of particular concern are school children daily exposed to lower-chlorinated, airborne PCBs in older buildings built primarily in the 1950s and 1960s (Herrick, 2010, Herrick et al., 2004, MacIntosh et al., 2012). As opposed to HC-PCBs, these LC-PCBs are readily

metabolized which may be the primary reason, why the parent congeners are typically present only at very low to undetectable concentrations (Hansen, 2001, Robertson and Ludewig, 2011). Low blood levels are often misinterpreted as an indication for low exposure and therefore low relevance when in fact they should be taken as an indication of continuous, steady-state, and therefore potentially significant overall exposure to compounds that have a high propensity to be bioactivated to potentially harmful metabolites. To date research has revealed four groups of potentially toxic metabolites of primarily these LC-PCBs: *i*: reactive metabolic intermediates from PCBs, particularly the quinones; *ii*: hydroxylated PCB metabolites, *iii*: PCB methyl sulfones and *iv*: PCB sulfates.

While there is substantial evidence for the formation and toxic potential of these metabolites derived from both *in vitro* and *in vivo* studies, the extent of their formation and exposure levels in human populations have not yet been explored. However, considering air PCB concentrations of up to of 5.5 ng/m³, for example in Chicago (Hu et al., 2010a, Persoon et al., 2010), and several hundred to thousand ng/m³ in indoor air of schools (Herrick et al., 2004, Liebl et al., 2004) and evidence for the presence of certain lower-chlorinated, airborne PCBs, such as PCB 11, in plasma of exposed populations (Marek et al., 2013b), quantitative assessment of LC-PCB and metabolite exposure levels is highly desirable.

It is not yet possible to assess human exposure to reactive metabolic intermediates of PCBs (arene oxides, semiquinones, quinones or their bound residues) even though the neutral, lipophilic MeSO₂-PCB congeners may be taken as a measure of arene oxide formation. Moreover, it is relevant to assess human levels of OH-PCBs and their conjugated PCB metabolites, such as MeSO₂-PCBs, PCB sulfates, and PCB glucuronides, since the parent PCB congeners are in many cases present only at trace concentrations or remain undetectable. The major OH-PCBs are present at concentrations in the same range as many of the PCB congeners and hence relevant for exposure assessments to PCBs. MeSO₂-PCBs have strong tissue and cell specific retention leading to higher local (cellular) concentrations than general tissue levels. The physiological fates and biological activities of PCB sulfates are still relatively poorly understood and additional studies will be required to assess their significance in PCB metabolism and toxicities.

Research needs for reactive (epoxide and (semi)quinone) PCB intermediates

PCB adducts with DNA and proteins in humans may be impossible to identify. Risks of adduct formation to DNA and other biomacromolecules can therefore only fully be assessed indirectly if we know the complete PCB congener profile and concentration pattern of the individual exposure to PCBs. This requirement is primarily based on the need for concentration data for lower chlorinated biphenyls and other congeners that are rapidly metabolized and therefore disappearing in humans despite often continuous, ongoing exposure from the 33 environment (Grossman, 2013). Exposures to lower chlorinated biphenyls would be primarily through inhalation of contaminated air in cities, buildings or locations near PCB-contaminated waste sites (Herrick, 2010, Hu and Hornbuckle, 2010, Hu et al., 2010a, MacIntosh et al., 2012, Marek et al., 2013b). The majority of exposures to PCBs in the general population, however, is thought to arise through dietary sources (Domingo and Bocio, 2007). Food as source for LC-PCBs is often overlooked, but may be

high for certain food items such as plant oils, possibly due to airborne exposure onto crops or from contaminated sewage sludge used as fertilizer (Ludewig et al., 2007). However, since most HC-PCBs biomagnify in the food chain and they are therefore the major contributors to dietary PCB exposure. HC-PCBs also bioaccumulate in humans resulting in the observed age-dependent increase in body burden over a lifetime. For these congeners it would seem possible to model the reactive intermediate concentrations based on internal concentrations of PCB congeners. The exposure to LC-PCBs with short half-lives may have a small impact on the total PCB body burden at any given time, but may well be the major source of PCBs that form reactive intermediates (Robertson and Gupta, 2000, Robertson and Ludewig, 2011). The formation of reactive intermediates, arene oxides, semi-quinones, and quinones, is therefore related to PCB concentration patterns in individuals. Other important questions relate to organ and species-specificity of activation pathways and therefore likelihood of adduct formation. As described above, the chlorination pattern of a PCB congener determines the substrate specificity for CYP forms which in turn influence organ and cell-specific formation of oxidized metabolites. In addition, cells with specific oxidizing enzymes, like myeloperoxidase in bone marrow cells, may be most likely to produce reactive quinone metabolites and therefore more likely to form DNA and protein adduct (Xie et al., 2010). Understanding of these specific enzyme requirements may elucidate the mechanism of the observed organ-specific carcinogenicity of 34 PCBs. Finally, an understanding of the activation pathway and mechanisms of toxicity will enable the development of chemoprotective regimens. Potential toxicities of the reactive intermediates are thus 1. directly related to the internal PCB congener concentrations including those from short-lived lower chlorinated congeners that can only be estimated by assessing the external exposures, 2. affected by species, organ and cell-specific bioactivation capacities, and 3. determined by the cellular consequences of the adduction to a specific macromolecule. Data for all three parameters are currently incomplete, thereby preventing true data-driven risk assessment.

Research needs for OH-PCBs

Human populations can be exposed to OH-PCBs by two different sources. While metabolic conversion of PCBs was for a long time considered the only source of OH-PCBs, it has also been demonstrated that OH-PCBs were originally present in Aroclors and they are still present in detectable amounts in the environment (Bergman et al., 1994b, Marek et al., 2013a, Matthews and Kato, 1979). Moreover, little or no data exist about OH-PCBs in our foodstuffs. OH-PCBs are either excreted as such or as their conjugates, or they may be further metabolized to dihydroxylated PCBs. However, some of the OH-PCBs are strongly retained in blood as protein bound phenolic metabolites with reasonably long half-lives (Bergman et al., 1994b, Malmberg et al., 2004, Oberg et al., 2002). Considering that OH-PCBs formed from internal PCB congeners lead to a continuous exposure to these metabolites, it should be possible to model some OH-PCB concentrations based on known PCB levels in a healthy normal human. However, since OH-PCB congeners can be present in high concentrations (10% – 30% of the PCB level) which are often higher than many of the individual PCB congeners (Marek et al., 2013b, Hisada et al., 2013, Nomiya et al., 2010), it is relevant to determine the concentrations of these metabolites and 35 potential sources separately. The goal is primarily to promote dose-response linkages between effects

of the OH-PCBs and their concentrations, a research field that needs further development. Assessing PCB induced adverse health effects for many toxic endpoints can be improved by including data on human concentrations and effects of OH-PCBs.

Research needs for PCB methyl sulfones

In contrast to the retained OH-PCB metabolites, MeSO₂-PCBs are neutral metabolites that are present at much lower concentrations in humans than OH-PCBs (<4% of the dominant OH-PCB congener) (Bergman et al., 1979, Letcher et al., 2000). MeSO₂-PCBs have specific and strong tissue or even cell specificity, particularly in the liver and lung (Bergman et al., 1979, Larsson and Bergman, 1998, Letcher et al., 2000). Whether or not this translates into organ and cell specific toxic effects has not been fully investigated. Another difference between methylsulfone and OH PCBs is that the MeSO₂-PCBs are the final product of the most rapidly metabolized PCB congeners which consequently are present in humans only at trace or even non-detectable concentrations. Since these metabolites derive from exposure to some of the most rapidly metabolized PCB congeners, they may serve as markers for exposure to these congeners and of the non-quantifiable arene oxide and quinone metabolites of these PCBs.

Research needs for PCB sulfates

Although the toxicological relevance of PCB sulfation is not yet known, a variety of *in vitro* and *in vivo* studies suggest that the formation of PCB sulfates is a potentially significant metabolic pathway for LC-PCBs in humans (Dhakal et al., 2012, Dhakal et al., 2014, Ekuase et al., 2011, Grimm et al., 2013, Liu et al., 2006, Liu et al., 2011, Liu et al., 2009, Zhai et al., 2013a). The fact that, in the past, human serum has been almost exclusively analyzed for parent PCBs and only recently for their hydroxylated metabolites indicates that overall PCB exposure levels in exposed populations may have been underestimated, at least with respect to the LC-PCBs (Hovander et al., 2000, Marek et al., 2013b, Dirtu and Covaci, 2010). Thus development of sulfate metabolites in urine or blood as biomarker of exposure to LC-PCBs is a promising avenue to obtain realistic exposure data for humans in their natural environments.

Acknowledgements

This review was in part the work product of the European Food Safety Authority (EFSA) working group on non-dioxin like PCBs, 2005. Where the authors own work was mentioned, those studies were supported by grants from the NIH (ES07380 and ES013661), DOD, EPA and the EU R&D projects PCBRIK, COMPARE, ANEMONE and RENCO. LWR would also like to recognize the Alexander von Humboldt Foundation for financial support.

References

- Alvares AP, Fischbein A, Anderson KE, Kappas A. Alterations in drug metabolism in workers exposed to polychlorinated biphenyls. *Clin Pharmacol Ther.* 1977; 22:140–146. [PubMed: 407043]
- Amaro AR, Oakley GG, Bauer U, Spielmann HP, Robertson LW. Metabolic activation of PCBs to quinones: reactivity toward nitrogen and sulfur nucleophiles and influence of superoxide dismutase. *Chem Res Toxicol.* 1996; 9:623–629. [PubMed: 8728508]
- Asvold BO, Vatten LJ, Nilsen TI, Bjoro T. The association between TSH within the reference range and serum lipid concentrations in a population-based study. *The HUNT Study. Eur J Endocrinol.* 2007; 156:181–186. [PubMed: 17287407]

- ATSDR. Toxicological profile for polychlorinated biphenyls (PCBs). U.S. Dept. Health Services, Public Health Service; 2000.
- Bakke JE, Bergman AL, Larsen GL. Metabolism of 2,4',5-trichlorobiphenyl by the mercapturic acid pathway. *Science*. 1982; 217:645–647. [PubMed: 6806905]
- Bakke JE, Feil VJ, Bergman A. Metabolites of 2,4',5-trichlorobiphenyl in rats. *Xenobiotica*. 1983; 13:555–564. [PubMed: 6419479]
- Ballschmitter K, Mennel A, Buyten J. Long chain alkyl-polysiloxanes as non-polar stationary phases in capillary gas chromatography. *Fresenius' Journal of Analytical Chemistry*. 1993; 346:396–402.
- Ballschmitter K, Zell M. Analysis of polychlorinated-biphenyls (PCB) by glass-capillary gas-chromatography - composition of technical Aroclor-PCB and Clophen-PCB mixtures. *Fresenius J Anal Chem*. 1980; 302:20–31.
- Barron MG, Yurk JJ, Crothers DB. Assessment of Potential Cancer Risk from Consumption of PCBs Bioaccumulated in Fish and Shellfish. *Environ Health Perspect*. 1994; 102:562–567. [PubMed: 9679116]
- Basu I, Arnold KA, Venier M, Hites RA. Partial pressures of PCB-11 in air from several Great Lakes sites. *Environmental Science & Technology*. 2009; 43:6488–6492. [PubMed: 19764206]
- Bender RP, Lehmler HJ, Robertson LW, Ludewig G, Osheroff N. Polychlorinated biphenyl quinone metabolites poison human topoisomerase IIalpha: altering enzyme function by blocking the N-terminal protein gate. *Biochemistry*. 2006; 45:10140–10152. [PubMed: 16906772]
- Bender RP, Osheroff N. Mutation of cysteine residue 455 to alanine in human topoisomerase IIalpha confers hypersensitivity to quinones: enhancing DNA scission by closing the N-terminal protein gate. *Chem Res Toxicol*. 2007; 20:975–981. [PubMed: 17516663]
- Bergman A, Brandt I, Jansson B. Accumulation of methylsulfonyl derivatives of some bronchial-seeking polychlorinated biphenyls in the respiratory tract of mice. *Toxicol Appl Pharmacol*. 1979; 48:213–220. [PubMed: 112715]
- Bergman A, Klasson-Wehler E, Kuroki H. Selective retention of hydroxylated PCB metabolites in blood. *Environ Health Perspect*. 1994a; 102:464–469. [PubMed: 8593850]
- Bergman A, Norstrom RJ, Haraguchi K, Kuroki H, Beland P. PCB and DDE methyl sulfones in mammals from Canada and Sweden. *Environmental Toxicology and Chemistry*. 1994b; 13:121–128.
- Birnbaum LS. Distribution and excretion of 2,3,6,2',3',6'- and 2,4,5,2',4',5'-hexachlorobiphenyl in senescent rats. *Toxicol Appl Pharmacol*. 1983; 70:262–272. [PubMed: 6414105]
- Bitman J, Cecil HC. Estrogenic activity of DDT analogs and polychlorinated biphenyls. *J Agric Food Chem*. 1970; 18:1108–1112. [PubMed: 5483049]
- Bloom MS, Vena JE, Olson JR, Kostyniak PJ. Assessment of polychlorinated biphenyl congeners, thyroid stimulating hormone, and free thyroxine among New York state anglers. *Int J Hyg Environ Health*. 2009; 212:599–611. [PubMed: 19493696]
- Boucher O, Muckle G, Bastien CH. Prenatal exposure to polychlorinated biphenyls: a neuropsychologic analysis. *Environ Health Perspect*. 2009; 117:7–16. [PubMed: 19165381]
- Bradbury S, Mekenyan O, Ankley G. Quantitative structure-activity relationships for polychlorinated hydroxybiphenyl estrogen receptor binding affinity—an assessment of conformer flexibility. *Environ Toxicol Chem*. 1996; 15:1945–1954.
- Breivik K, Sweetman A, Pacyna JM, Jones KC. Towards a global historical emission inventory for selected PCB congeners--a mass balance approach 3. An update. *Sci Total Environ*. 2007; 377:296–307. [PubMed: 17395248]
- Brouwer A, Longnecker MP, Birnbaum LS, Cogliano J, Kostyniak P, Moore J, Schantz S, Winneke G. Characterization of potential endocrine-related health effects at low-dose levels of exposure to PCBs. *Environ Health Perspect*. 1999; 107(Suppl 4):639–649. [PubMed: 10421775]
- Brouwer A, Morse DC, Lans MC, Schuur AG, Murk AJ, Klasson-Wehler E, Bergman A, Visser TJ. Interactions of persistent environmental organohalogenes with the thyroid hormone system: mechanisms and possible consequences for animal and human health. *Toxicol Ind Health*. 1998; 14:59–84. [PubMed: 9460170]

- Chauhan KR, Kodavanti PR, Mckinney JD. Assessing the role of ortho-substitution on polychlorinated biphenyl binding to transthyretin, a thyroxine transport protein. *Toxicol Appl Pharmacol.* 2000; 162:10–21. [PubMed: 10631123]
- Choi SD, Baek SY, Chang YS, Wania F, Ikonomou MG, Yoon YJ, Park BK, Hong S. Passive air sampling of polychlorinated biphenyls and organochlorine pesticides at the Korean Arctic and Antarctic research stations: implications for long-range transport and local pollution. *Environmental Science & Technology.* 2008; 42:7125–7131. [PubMed: 18939536]
- Christensen H, Heggberget TM, Gutleb AC. Polychlorinated biphenyls and reproductive performance in otters from the Norwegian coast. *Arch Environ Contam Toxicol.* 2010; 59:652–660. [PubMed: 20383701]
- Chu S, Covaci A, Haraguchi K, Voorspoels S, van de Vijver K, Das K, Bouquegneau JM, De Coen W, Blust R, Schepens P. Levels and enantiomeric signatures of methyl sulfonyl PCB and DDE metabolites in livers of harbor porpoises (*Phocoena phocoena*) from the Southern North Sea. *Environ Sci Technol.* 2003a; 37:4573–4578. [PubMed: 14594363]
- Chu S, Covaci A, Jacobs W, Haraguchi K, Schepens P. Distribution of methyl sulfone metabolites of polychlorinated biphenyls and p,p'-DDE in human tissues. *Environ Health Perspect.* 2003b; 111:1222–1227. [PubMed: 12842777]
- Connor K, Ramamoorthy K, Moore M, Mustain M, Chen I, Safe S, Zacharewski T, Gillesby B, Joyeux A, Balaguer P. Hydroxylated polychlorinated biphenyls (PCBs) as estrogens and antiestrogens: structure-activity relationships. *Toxicol Appl Pharmacol.* 1997; 145:111–123. [PubMed: 9221830]
- Cuadra SN, Linderholm L, Athanasiadou M, Jakobsson K. Persistent organic pollutants in children working in a waste disposal site and in female high fish consumers in Managua, Nicaragua. *Ambio.* 2006; 35:109–116. [PubMed: 16846198]
- Custer TW, Custer CM, Gray BR. Polychlorinated biphenyls, dioxins, furans, and organochlorine pesticides in belted kingfisher eggs from the upper Hudson River basin, New York, USA. *Environ Toxicol Chem.* 2010; 29:99–110. [PubMed: 20821424]
- Daidoji T, Gozu K, Iwano H, Inoue H, Yokota H. UDP-glucuronosyltransferase isoforms catalyzing glucuronidation of hydroxy-polychlorinated biphenyls in rat. *Drug Metab Dispos.* 2005; 33:1466–1476. [PubMed: 16006569]
- Dallaire R, Muckle G, Dewailly E, Jacobson SW, Jacobson JL, Sandanger TM, Sandau CD, Ayotte P. Thyroid hormone levels of pregnant inuit women and their infants exposed to environmental contaminants. *Environ Health Perspect.* 2009; 117:1014–1020. [PubMed: 19590699]
- Darnerud PO, Morse D, Klasson-Wehler E, Brouwer A. Binding of a 3,3',4,4'-tetrachlorobiphenyl (CB-77) metabolite to fetal transthyretin and effects on fetal thyroid hormone levels in mice. *Toxicology.* 1996; 106:105–114. [PubMed: 8571380]
- Darras VM. Endocrine disrupting polyhalogenated organic pollutants interfere with thyroid hormone signalling in the developing brain. *Cerebellum.* 2008; 7:26–37. [PubMed: 18418666]
- Daubeze M, Narbonne JF. Incorporation of labeled 2,4,5,2',4',5' - hexachlorobiphenyl into the nuclear fraction of rat hepatocytes in vivo. *Toxicology.* 1984; 31:315–318. [PubMed: 6204423]
- Decastro BR, Korrick SA, Spengler JD, Soto AM. Estrogenic activity of polychlorinated biphenyls present in human tissue and the environment. *Environ Sci Technol.* 2006; 40:2819–2825. [PubMed: 16683629]
- Dhakal K, Adamcakova-Dodd A, Lehmler HJ, Thorne PS, Robertson LW. Sulfate conjugates are urinary markers of inhalation exposure to 4-chlorobiphenyl (PCB3). *Chem Res Toxicol.* 2013; 26:853–855. [PubMed: 23713983]
- Dhakal K, He X, Lehmler HJ, Teesch LM, Duffel MW, Robertson LW. Identification of sulfated metabolites of 4-chlorobiphenyl (PCB3) in the serum and urine of male rats. *Chem Res Toxicol.* 2012; 25:2796–2804. [PubMed: 23137097]
- Dhakal K, Uwimana E, Adamcakova-Dodd A, Thorne PS, Lehmler HJ, Robertson LW. Disposition of Phenolic and Sulfated Metabolites after Inhalation Exposure to 4-Chlorobiphenyl (PCB3) in Female Rats. *Chem Res Toxicol.* 2014
- Dirtu AC, Covaci A. Estimation of daily intake of organohalogenated contaminants from food consumption and indoor dust ingestion in Romania. *Environ Sci Technol.* 2010; 44:6297–6304. [PubMed: 20704229]

- Domingo JL, Bocio A. Levels of PCDD/PCDFs and PCBs in edible marine species and human intake: a literature review. *Environ Int.* 2007; 33:397–405. [PubMed: 17270272]
- Du S, Wall SI, Cacia D, Rodenburg LA. Passive air sampling for polychlorinated biphenyls in the Philadelphia metropolitan area. *Environ Sci Technol.* 2009; 43:1287–1292. [PubMed: 19350892]
- Ekuase EJ, Liu Y, Lehmler HJ, Robertson LW, Duffel MW. Structure-activity relationships for hydroxylated polychlorinated biphenyls as inhibitors of the sulfation of dehydroepiandrosterone catalyzed by human hydroxysteroid sulfotransferase SULT2A1. *Chem Res Toxicol.* 2011; 24:1720–1728. [PubMed: 21913674]
- El Majidi N, Bouchard M, Carrier G. Systematic analysis of the relationship between standardized biological levels of polychlorinated biphenyls and thyroid function in pregnant women and newborns. *Chemosphere.* 2014; 98:1–17. [PubMed: 24200047]
- Ellerichmann T, Bergmann A, Franke S, Huehnerfuss H, Jakobsson E, Koenig WA, Larsson C. Gas chromatographic enantiomer separations of chiral PCB methyl sulfons and identification of selectively retained enantiomers in human liver. *Fresenius Environmental Bulletin.* 1998; 7:244–257.
- Erickson, MD. Analytical Chemistry of PCBs. Boca Raton: CRC Lewis Publishers; 1997.
- Erickson, MD. Introduction: PCB properties, uses, occurrence, and regulatory history. In: Robertson, LW.; Hansen, LG., editors. PCBs: recent advances in environmental toxicology and health effects. Lexington, Kentucky: The University Press of Kentucky; 2001.
- Erickson MD, Kaley RG 2nd. Applications of polychlorinated biphenyls. *Environ Sci Pollut Res Int.* 2011; 18:135–151. [PubMed: 20848233]
- Espandiari P, Glauert HP, Lehmler HJ, Lee EY, Srinivasan C, Robertson LW. Polychlorinated biphenyls as initiators in liver carcinogenesis: resistant hepatocyte model. *Toxicol Appl Pharmacol.* 2003; 186:55–62. [PubMed: 12583993]
- Espandiari P, Glauert HP, Lehmler HJ, Lee EY, Srinivasan C, Robertson LW. Initiating activity of 4-chlorobiphenyl metabolites in the resistant hepatocyte model. *Toxicol Sci.* 2004; 79:41–46. [PubMed: 14976334]
- Fängström B, Athanasiadou M, Athanassiadis I, Weihe P, Bergman A. Hydroxylated PCB metabolites in nonhatched fulmar eggs from the Faroe Islands. *Ambio.* 2005a; 34:184–187. [PubMed: 16042274]
- Fangstrom B, Athanasiadou M, Grandjean P, Weihe P, Bergman A. Hydroxylated PCB metabolites and PCBs in serum from pregnant Faroese women. *Environ Health Perspect.* 2002; 110:895–899. [PubMed: 12204824]
- Fängström B, Hovander L, Bignert A, Athanassiadis I, Linderholm L, Grandjean P, Weihe P, Bergman A. Concentrations of PBDE, PCB and OH-PCBs in serum from pregnant Faroese women and their children seven years later. *Environmental Science & Technology.* 2005b; 39:9457–9463. [PubMed: 16475322]
- Fangstrom B, Strid A, Grandjean P, Weihe P, Bergman A. A retrospective study of PBDEs and PCBs in human milk from the Faroe Islands. *Environ Health.* 2005; 4:12. [PubMed: 16014177]
- Feinberg M, Soler L, Contenot S, Verger P. Assessment of seasonality in exposure to dioxins, furans and dioxin-like PCBs by using long-term food-consumption data. *Food Addit Contam Part A Chem Anal Control Expo Risk Assess.* 2011; 28:502–512. [PubMed: 21416416]
- Fitzgerald EF, Belanger EE, Gomez MI, Cayo M, Mccaffrey RJ, Seegal RF, Jansing RL, Hwang SA, Hicks HE. Polychlorinated biphenyl exposure and neuropsychological status among older residents of upper Hudson River communities. *Environ Health Perspect.* 2008; 116:209–215. [PubMed: 18288320]
- Fitzgerald EF, Shrestha S, Palmer PM, Wilson LR, Belanger EE, Gomez MI, Cayo MR, Hwang SA. Polychlorinated biphenyls (PCBs) in indoor air and in serum among older residents of upper Hudson River communities. *Chemosphere.* 2011; 85:225–231. [PubMed: 21724230]
- Flor S, Ludewig G. Polyploidy-induction by dihydroxylated monochlorobiphenyls: structure-activity-relationships. *Environ Int.* 2010; 36:962–969. [PubMed: 20471090]
- Gabrio T, Piechotowski I, Wallenhorst T, Klett M, Cott L, Friebe P, Link B, Schwenk M. PCB-blood levels in teachers, working in PCB-contaminated schools. *Chemosphere.* 2000; 40:1055–1062. [PubMed: 10739046]

- Gallenberg LA, Ring BJ, Vodcnik MJ. The influence of time of maternal exposure to 2,4,5,2',4',5'-hexachlorobiphenyl on its accumulation in their nursing offspring. *Toxicol Appl Pharmacol.* 1990; 104:1–8. [PubMed: 2113719]
- Giera S, Bansal R, Ortiz-Toro TM, Taub DG, Zoeller RT. Individual polychlorinated biphenyl (PCB) congeners produce tissue- and gene-specific effects on thyroid hormone signaling during development. *Endocrinology.* 2011; 152:2909–2919. [PubMed: 21540284]
- Goncharov A, Bloom M, Pavuk M, Birman I, Carpenter DO. Blood pressure and hypertension in relation to levels of serum polychlorinated biphenyls in residents of Anniston, Alabama. *J Hypertens.* 2010; 28:2053–2060. [PubMed: 20644494]
- Goncharov A, Pavuk M, Foushee HR, Carpenter DO. Blood pressure in relation to concentrations of PCB congeners and chlorinated pesticides. *Environ Health Perspect.* 2011; 119:319–325. [PubMed: 21362590]
- Grimm FA, Lehmler HJ, He X, Robertson LW, Duffel MW. Sulfated metabolites of polychlorinated biphenyls are high-affinity ligands for the thyroid hormone transport protein transthyretin. *Environ Health Perspect.* 2013; 121:657–662. [PubMed: 23584369]
- Grossman E. Nonlegacy PCBs: pigment manufacturing by-products get a second look. *Environ Health Perspect.* 2013; 121:A86–A93. [PubMed: 23454657]
- Guengerich FP. Common and uncommon cytochrome P450 reactions related to metabolism and chemical toxicity. *Chem Res Toxicol.* 2001; 14:611–650. [PubMed: 11409933]
- Guroff G, Daly JW, Jerina DM, Renson J, Witkop B, Udenfriend S. Hydroxylation-induced migration: the NIH shift. Recent experiments reveal an unexpected and general result of enzymatic hydroxylation of aromatic compounds. *Science.* 1967; 157:1524–1530. [PubMed: 6038165]
- Gutleb AC, Cenijn P, Velzen M, Lie E, Ropstad E, Skaare JU, Malmberg T, Bergman A, Gabrielsen GW, Legler J. In vitro assay shows that PCB metabolites completely saturate thyroid hormone transport capacity in blood of wild polar bears (*Ursus maritimus*). *Environ Sci Technol.* 2010; 44:3149–3154. [PubMed: 20345174]
- Guvenius DM, Aronsson A, Ekman-Ordeberg G, Bergman A, Noren K. Human prenatal and postnatal exposure to polybrominated diphenyl ethers, polychlorinated biphenyls, polychlorobiphenyls, and pentachlorophenol. *Environ Health Perspect.* 2003; 111:1235–1241. [PubMed: 12842779]
- Guvenius DM, Hassanzadeh P, Bergman A, Noren K. Metabolites of polychlorinated biphenyls in human liver and adipose tissue. *Environ Toxicol Chem.* 2002; 21:2264–2269. [PubMed: 12389902]
- Haglund P. Isolation and characterization of polychlorinated biphenyl (PCB) atropisomers. *Chemosphere.* 1996; 32:8.
- Hagmar L, Rylander L, Dyremark E, Klasson-Wehler E, Erfurth EM. Plasma concentrations of persistent organochlorines in relation to thyrotropin and thyroid hormone levels in women. *Int Arch Occup Environ Health.* 2001; 74:184–188. [PubMed: 11355292]
- Hansen LG. Stepping backward to improve assessment of PCB congener toxicities. *Environ Health Perspect.* 1998; 106(Suppl 1):171–189. [PubMed: 9539012]
- Hansen, LG. The ortho side of PCBs: Occurrence and disposition. Boston: Kluwer Academic Publishers Boston; 1999.
- Hansen, LG. Identification of steady state and episodic PCB congeners from multiple pathway exposures. In: Robertson, LW.; Hansen, LG., editors. *PCBs: Recent Advances in Environmental Toxicology and Health Effects.* Lexington, Kentucky, US: The University Press of Kentucky; 2001.
- Harrad S, Goosey E, Desborough J, Abdallah MA, Roosens L, Covaci A. Dust from U.K. primary school classrooms and daycare centers: the significance of dust as a pathway of exposure of young U.K. children to brominated flame retardants and polychlorinated biphenyls. *Environ Sci Technol.* 2010; 44:4198–4202. [PubMed: 20441148]
- Harrad S, Ibarra C, Robson M, Melymuk L, Zhang X, Diamond M, Douwes J. Polychlorinated biphenyls in domestic dust from Canada, New Zealand, United Kingdom and United States: implications for human exposure. *Chemosphere.* 2009; 76:232–238. [PubMed: 19356786]
- Herrick RF. PCBs in school-persistent chemicals, persistent problems. *New Solut.* 2010; 20:115–126. [PubMed: 20359995]

- Herrick RF, McClean MD, Meeker JD, Baxter LK, Weymouth GA. An unrecognized source of PCB contamination in schools and other buildings. *Environ Health Perspect.* 2004; 112:1051–1053. [PubMed: 15238275]
- Hisada A, Shimodaira K, Okai T, Watanabe K, Takemori H, Takasuga T, Noda Y, Shirakawa M, Kato N, Yoshinaga J. Serum levels of hydroxylated PCBs, PCBs and thyroid hormone measures of Japanese pregnant women. *Environ Health Prev Med.* 2013; 18:205–214. [PubMed: 23054994]
- Hofvander, L. Ph.D. Thesis. Department of Environmental Chemistry, University of Stockholm; 2006. Polychlorinated biphenyls and their metabolites in human blood: Method development, identification and quantification.
- Hovander L, Athanasiadou M, Asplund L, Jensen S, Wehler EK. Extraction and cleanup methods for analysis of phenolic and neutral organohalogenes in plasma. *J Anal Toxicol.* 2000; 24:696–703. [PubMed: 11110024]
- Hovander L, Linderholm L, Athanasiadou M, Athanassiadis I, Bignert A, Fängström B, Kocan A, Petrik J, Trnovec T, Bergman A. Levels of PCBs and their metabolites in the serum of residents of a highly contaminated area in eastern Slovakia. *Environmental Science & Technology.* 2006; 40:3696–3703. [PubMed: 16830529]
- Hovander L, Linderholm L, Athanasiadou M, Athanassiadis I, Trnovec T, Kocan A, Petrik J, Bergman A. Analysis of PCB and PCB metabolites in humans from eastern Slovakia. *Organohalogen Compounds.* 2004; 66:3525–3531.
- Hrycay EG, Bandiera SM. Spectral interactions of tetrachlorobiphenyls with hepatic microsomal cytochrome P450 enzymes. *Chem. Biol. Interact.* 2003; 146:285–296. [PubMed: 14642740]
- Hu D, Hornbuckle KC. Inadvertent polychlorinated biphenyls in commercial paint pigments. *Environ Sci Technol.* 2010; 44:2822–2827. [PubMed: 19957996]
- Hu D, Lehmler HJ, Martinez A, Wang K, Hornbuckle KC. Atmospheric PCB congeners across Chicago. *Atmos Environ* (1994). 2010a; 44:1550–1557. [PubMed: 21918637]
- Hu D, Martinez A, Hornbuckle KC. Discovery of non-aroclor PCB (3,3'-dichlorobiphenyl) in Chicago air. *Environ Sci Technol.* 2008; 42:7873–7877. [PubMed: 19031874]
- Hu D, Martinez A, Hornbuckle KC. Sedimentary Records of Non-Aroclor and Aroclor PCB mixtures in the Great Lakes. *Journal of Great Lakes Research.* 2011; 37:359–364. [PubMed: 23538476]
- Hu X, Adamcakova-Dodd A, Lehmler HJ, Hu D, Hornbuckle K, Thorne PS. Subchronic inhalation exposure study of an airborne polychlorinated biphenyl mixture resembling the Chicago ambient air congener profile. *Environ Sci Technol.* 2012; 46:9653–9662. [PubMed: 22846166]
- Hu X, Adamcakova-Dodd A, Lehmler HJ, Hu D, Kania-Korwel I, Hornbuckle KC, Thorne PS. Time course of congener uptake and elimination in rats after short-term inhalation exposure to an airborne polychlorinated biphenyl (PCB) mixture. *Environ Sci Technol.* 2010b; 44:6893–6900. [PubMed: 20698547]
- Hu X, Adamcakova-Dodd A, Thorne PS. The fate of inhaled (14)C-labeled PCB11 and its metabolites in vivo. *Environ Int.* 2014; 63:92–100. [PubMed: 24275706]
- Hu X, Lehmler HJ, Adamcakova-Dodd A, Thorne PS. Elimination of inhaled 3,3'-dichlorobiphenyl and the formation of the 4-hydroxylated metabolite. *Environ Sci Technol.* 2013; 47:4743–4751. [PubMed: 23582014]
- Jacobson JL, Jacobson SW, Humphrey HE. Effects of in utero exposure to polychlorinated biphenyls and related contaminants on cognitive functioning in young children. *J Pediatr.* 1990; 116:38–45. [PubMed: 2104928]
- Jacobus JA, Flor S, Klingelutz A, Robertson LW, Ludewig G. 2-(4'-Chlorophenyl)-1,4-Benzoquinone Increases the Frequency of Micronuclei and Shortens Telomeres. *Environ Toxicol Pharmacol.* 2008; 25:267–272. [PubMed: 18438462]
- James, MO. Polychlorinated Biphenyls: Metabolism and Metabolites. In: Robertson, L.; Hansen, LG., editors. *PCBs, Recent Advances in Environmental Toxicology and Health Effects.* Lexington, KY: The University Press of Kentucky; 2001.
- Jamshidi A, Hunter S, Hazrati S, Harrad S. Concentrations and chiral signatures of polychlorinated biphenyls in outdoor and indoor air and soil in a major U.K. conurbation. *Environ Sci Technol.* 2007; 41:2153–2158. [PubMed: 17438756]

- Jerina DM, Daly JW. Arene oxides: a new aspect of drug metabolism. *Science*. 1974; 185:573–582. [PubMed: 4841570]
- Johansson M, Larsson C, Bergman A, Lund BO. Structure-activity relationship for inhibition of CYP11B1-dependent glucocorticoid synthesis in Y1 cells by aryl methyl sulfones. *Pharmacol Toxicol*. 1998; 83:225–230. [PubMed: 9834972]
- Jörundsdóttir H, Norstrom K, Olsson M, Pham-tuan H, Hühnerfuss H, Bignert A, Bergman A. Temporal trend of bis(4-chlorophenyl) sulfone, methylsulfonyl-DDE and -PCBs in Baltic guillemot (*Uria aalge*) egg 1971–2001 - A comparison to 4,4'-DDE and PCB trends. *Environ. Pollut*. 2006; 141:226–237. [PubMed: 16225974]
- Jugan ML, Levi Y, Blondeau JP. Endocrine disruptors and thyroid hormone physiology. *Biochem Pharmacol*. 2010; 79:939–947. [PubMed: 19913515]
- Kaiser K. On the optical activity of polychlorinated biphenyls. *Environ. Poll*. 1974; 7:93–101.
- Kallenborn, R.; Hühnerfuss, H., editors. *Chiral Environmental Pollutants - Trace Analysis and Ecotoxicology*. Berlin, Heidelberg: Springer Verlag; 2001.
- Kaminsky LS, Kennedy MW, Adams SM, Guengerich FP. Metabolism of dichlorobiphenyls by highly purified isozymes of rat liver cytochrome P-450. *Biochemistry*. 1981; 20:7379–7384. [PubMed: 6798990]
- Kania-korwel I, Barnhart CD, Stamou M, Truong KM, El-komy MH, Lein PJ, Veng-pedersen P, Lehmler HJ. 2,2',3,5',6-Pentachlorobiphenyl (PCB 95) and its hydroxylated metabolites are enantiomerically enriched in female mice. *Environ Sci Technol*. 2012; 46:11393–11401. [PubMed: 22974126]
- Kania-korwel I, Duffel MW, Lehmler HJ. Gas chromatographic analysis with chiral cyclodextrin phases reveals the enantioselective formation of hydroxylated polychlorinated biphenyls by rat liver microsomes. *Environ. Sci. Technol*. 2011; 45:9590–9596. [PubMed: 21966948]
- Kania-korwel I, Hrycay EG, Bandiera S, Lehmler H-J. 2,2',3,3',6,6'-hexachlorobiphenyl (PCB 136) atropisomers interact enantioselectively with hepatic microsomal cytochrome P450 enzymes. *Chem. Res. Toxicol*. 2008a; 21:1295–1303. [PubMed: 18494506]
- Kania-korwel I, Vyas S, Song Y, Lehmler HJ. Gas chromatographic separation of methoxylated polychlorinated biphenyl atropisomer. *J. Chromatogr. A*. 2008b; 1207:146–154. [PubMed: 18760792]
- Kato S, McKinney JD, Matthews HB. Metabolism of symmetrical hexachlorobiphenyl isomers in the rat. *Toxicol. Appl. Pharma*. 1980; 53:389–398.
- Kato Y, Haraguchi K, Kawashima M, Yamada S, Masuda Y, Kimura R. Induction of hepatic microsomal drug-metabolizing enzymes by methylsulphonyl metabolites of polychlorinated biphenyl congeners in rats. *Chem Biol Interact*. 1995; 95:257–268. [PubMed: 7728896]
- Kato Y, Haraguchi K, Shibahara T, Masuda Y, Kimura R. Reduction of thyroid hormone levels by methylsulfonyl metabolites of polychlorinated biphenyl congeners in rats. *Arch Toxicol*. 1998; 72:541–544. [PubMed: 9765071]
- Kato Y, Haraguchi K, Shibahara T, Shimura Y, Masuda Y, Kimura R. The induction of hepatic microsomal UDP-glucuronosyltransferase by the methylsulfonyl metabolites of polychlorinated biphenyl congeners in rats. *Chem Biol Interact*. 2000; 125:107–115. [PubMed: 10699571]
- Kato Y, Haraguchi K, Shibahara T, Yumoto S, Masuda Y, Kimura R. Reduction of thyroid hormone levels by methylsulfonyl metabolites of tetra- and pentachlorinated biphenyls in male Sprague-Dawley rats. *Toxicol Sci*. 1999; 48:51–54. [PubMed: 10330683]
- Kato Y, Haraguchi K, Tomiyasu K, Hiroyuki S, Isogai M, Masuda Y, Kimura R. Structure-dependent induction of CYP2B1/2 by 3-methylsulfonyl metabolites of polychlorinated biphenyl congeners in rats. *Environ Toxicol Pharmacol*. 1997; 3:137–144. [PubMed: 21781771]
- Kato Y, Ikushiro S, Haraguchi K, Yamazaki T, Ito Y, Suzuki H, Kimura R, Yamada S, Inoue T, Degawa M. A possible mechanism for decrease in serum thyroxine level by polychlorinated biphenyls in Wistar and Gunn rats. *Toxicol Sci*. 2004; 81:309–315. [PubMed: 15254343]
- Kester MH, Bulduk S, Tibboel D, Meinel W, Glatt H, Falany CN, Coughtrie MW, Bergman A, Safe SH, Kuiper GG, Schuur AG, Brouwer A, Visser TJ. Potent inhibition of estrogen sulfotransferase by hydroxylated PCB metabolites: a novel pathway explaining the estrogenic activity of PCBs. *Endocrinology*. 2000; 141:1897–1900. [PubMed: 10803601]

- Kester MH, Bulduk S, Van toor H, Tibboel D, Meinel W, Glatt H, Falany CN, Coughtrie MW, Schuur AG, Brouwer A, Visser TJ. Potent inhibition of estrogen sulfotransferase by hydroxylated metabolites of polyhalogenated aromatic hydrocarbons reveals alternative mechanism for estrogenic activity of endocrine disrupters. *J Clin Endocrinol Metab.* 2002; 87:1142–1150. [PubMed: 11889178]
- Kimbrough, RD.; Jensen, AA., editors. Halogenated Biphenyls, Terphenyls, Naphthalenes, Dibenzodioxins and Related Products. 2nd Ed.. 1989.
- Klasson wehler E, Jonsson J, Bergman A, Brandt I, Darnerud PO. 3,3',4,4'-Tetrachlorobiphenyl and 3,3',4,4',5-pentachlorobiphenyl: tissue localization and metabolic fate in the mouse. *Chemosphere.* 1989; 19:809–812.
- Klasson Wehler E, Lindberg L, Joensson CJ, Bergman A. Tissue retention and metabolism of 2,3,4,3',4'-pentachlorobiphenyl in mink and mouse. *Chemosphere.* 1993; 27:2397–2412.
- Knerr S, Schrenk D. Carcinogenicity of "non-dioxinlike" polychlorinated biphenyls. *Crit Rev Toxicol.* 2006; 36:663–694. [PubMed: 17050081]
- Kodavanti PR, Curras-collazo MC. Neuroendocrine actions of organohalogen: thyroid hormones, arginine vasopressin, and neuroplasticity. *Front Neuroendocrinol.* 2010; 31:479–496. [PubMed: 20609372]
- Krishnan V, Safe S. Polychlorinated biphenyls (PCBs), dibenzo-p-dioxins (PCDDs), and dibenzofurans (PCDFs) as antiestrogens in MCF-7 human breast cancer cells: quantitative structure-activity relationships. *Toxicol Appl Pharmacol.* 1993; 120:55–61. [PubMed: 7685553]
- Kutz FW, Wood PH, Bottimore DP. Organochlorine pesticides and polychlorinated biphenyls in human adipose tissue. *Rev Environ Contam Toxicol.* 1991; 120:1–82. [PubMed: 1899728]
- Langer P, Kocan A, Tajtakova M, Koska J, Radikova Z, Ksinantova L, Imrich R, Huckova M, Drobna B, Gasperikova D, Sebkova E, Klimes I. Increased thyroid volume, prevalence of thyroid antibodies and impaired fasting glucose in young adults from organochlorine cocktail polluted area: outcome of transgenerational transmission? *Chemosphere.* 2008; 73:1145–1150. [PubMed: 18790515]
- Lans MC, Klasson-Wehler E, Willemsen M, Meussen E, Safe S, Brouwer A. Structure-dependent, competitive interaction of hydroxy-polychlorobiphenyls, -dibenzo-p-dioxins and -dibenzofurans with human transthyretin. *Chem Biol Interact.* 1993; 88:7–21. [PubMed: 8330325]
- Larsdotter M, Darnerud P, Aune M, Glynn A, Bjerselius R. Serum concentrations of pentachlorophenol (PCP), polychlorinated biphenyls (PCBs), and hydroxylated metabolites of PCB during pregnancy and lactation (in Swedish). Swedish EPA Report on Contract 2190104, 2005. 2005 /<http://www.naturvardsverket.se/dokument/mo/modok/export/klorfenoler.pdf/>.
- Larsson C, Bergman A. Synthesis of radiolabelled methylsulphonyl CBs with specific retention in the rat liver. *Organohalogen Compounds.* 1998; 35:127–130.
- Larsson C, Ellerichmann T, Huhnerfuss H, Bergman A. Chiral PCB methyl sulfones in rat tissues after exposure to technical PCBs. *Environ Sci Technol.* 2002; 36:2833–2838. [PubMed: 12144255]
- Larsson C, Norstrom K, Athanasiadis I, Bignert A, Konig WA, Bergman A. Enantiomeric specificity of methylsulfonyl-PCBs and distribution of bis(4-chlorophenyl) sulfone, PCB, and DDE methyl sulfones in grey seal tissues. *Environ Sci Technol.* 2004; 38:4950–4955. [PubMed: 15506185]
- Lauby-secretan B, Loomis D, Grosse Y, El ghissassi F, Bouvard V, Benbrahim-tallaa L, Guha N, Baan R, Mattock H, Straif K. Carcinogenicity of polychlorinated biphenyls and polybrominated biphenyls. *Lancet Oncol.* 2013; 14:287–288. [PubMed: 23499544]
- Lehmann L, H LE, P AK, L WR, Ludewig G. 4-monochlorobiphenyl (PCB3) induces mutations in the livers of transgenic Fisher 344 rats. *Carcinogenesis.* 2007; 28:471–478. [PubMed: 16950798]
- Lehmler HJ, Harrad SJ, Huhnerfuss H, Kania-korwel I, Lee CM, Lu Z, Wong CS. Chiral polychlorinated biphenyl transport, metabolism, and distribution: a review. *Environ Sci Technol.* 2010; 44:2757–2766. [PubMed: 20384371]
- Lehmler HJ, Robertson LW, Garrison AW, Kodavanti PR. Effects of PCB 84 enantiomers on [3H]-phorbol ester binding in rat cerebellar granule cells and 45Ca2+-uptake in rat cerebellum. *Toxicol Lett.* 2005; 156:391–400. [PubMed: 15763638]

- Letcher, RJ.; Klasson-wehler, E.; Bergman, A. Methyl sulfone and hydroxylated metabolites of polychlorinated biphenyls. In: Paasivirta, J., editor. *New Types of Persistent Halogenated Compounds*. Berlin: Springer-Verlag; 2000.
- Li X, Parkin S, Duffel MW, Robertson LW, Lehmler HJ. An efficient approach to sulfate metabolites of polychlorinated biphenyls. *Environ Int*. 2010; 36:843–848. [PubMed: 19345419]
- Liebl B, Schettgen T, Kerschler G, Broding HC, Otto A, Angerer J, Drexler H. Evidence for increased internal exposure to lower chlorinated polychlorinated biphenyls (PCB) in pupils attending a contaminated school. *Int J Hyg Environ Health*. 2004; 207:315–324. [PubMed: 15471095]
- Lin PH, Sangaiah R, Ranasinghe A, Upton PB, La DK, Gold A, Swenberg JA. Formation of quinonoid-derived protein adducts in the liver and brain of Sprague-Dawley rats treated with 2,2',5,5'-tetrachlorobiphenyl. *Chem Res Toxicol*. 2000; 13:710–718. [PubMed: 10956058]
- Litten S, Fowler B, Luszniak D. Identification of a novel PCB source through analysis of 209 PCB congeners by US EPA modified method 1668. *Chemosphere*. 2002; 46:1457–1459. [PubMed: 12002476]
- Liu Y, Apak TI, Lehmler HJ, Robertson LW, Duffel MW. Hydroxylated polychlorinated biphenyls are substrates and inhibitors of human hydroxysteroid sulfotransferase SULT2A1. *Chem Res Toxicol*. 2006; 19:1420–1425. [PubMed: 17112228]
- Liu Y, Lehmler HJ, Robertson LW, Duffel MW. Physicochemical properties of hydroxylated polychlorinated biphenyls aid in predicting their interactions with rat sulfotransferase 1A1 (rSULT1A1). *Chem Biol Interact*. 2011; 189:153–160. [PubMed: 21130751]
- Liu Y, Smart JT, Song Y, Lehmler HJ, Robertson LW, Duffel MW. Structure-activity relationships for hydroxylated polychlorinated biphenyls as substrates and inhibitors of rat sulfotransferases and modification of these relationships by changes in thiol status. *Drug Metab Dispos*. 2009; 37:1065–1072. [PubMed: 19196841]
- Londono M, Shimokawa N, Miyazaki W, Iwasaki T, Koibuchi N. Hydroxylated PCB induces Ca²⁺ oscillations and alterations of membrane potential in cultured cortical cells. *J Appl Toxicol*. 2010; 30:334–342. [PubMed: 19924679]
- Lu Z, Kania-korwel I, Lehmler HJ, Wong CS. Stereoselective formation of mono- and dihydroxylated polychlorinated biphenyls by rat cytochrome P450 2B1. *Environ Sci Technol*. 2013; 47:12184–12192. [PubMed: 24060104]
- Lu Z, Wong CS. Factors affecting phase I stereoselective biotransformation of chiral polychlorinated biphenyls by rat cytochrome P-450 2B1 isozyme. *Environ Sci Technol*. 2011; 45:8298–8305. [PubMed: 21863805]
- Lucier GW, McDaniel OS, Schiller CM, Matthews HB. Structural requirements for the accumulation of chlorinated biphenyl metabolites in the fetal rat intestine. *Drug Metab Dispos*. 1978; 6:584–590. [PubMed: 30609]
- Ludewig, G. Cancer initiation by PCBs. In: Robertson, LW.; Hansen, LG., editors. *PCBs, Recent Advances in Environmental Toxicology and Health Effects*. Lexington: The University Press of Kentucky; 2001.
- Ludewig, G.; Esch, H.; Robertson, L. Polyhalogenierte Bi- und Terphenyle. In: Dunkelberg, H.; Gebel, T.; Hartwig, A., editors. *Handbuch der Lebensmitteltoxikologie: Belastungen, Wirkungen, Lebensmittelsicherheit, Hygiene*. Weinheim: Wiley-VCH; 2007.
- Ludewig G, Lehmann L, Esch H, Robertson LW. Metabolic Activation of PCBs to Carcinogens in Vivo - A Review. *Environ Toxicol Pharmacol*. 2008; 25:241–246. [PubMed: 18452002]
- Lund B-O, Örborg J, Bergman Å, Larsson C, Bergman A, Bäcklin B-M, Håkansson H, Madej A, Brouwer A, Brunström B. Chronic and reproductive toxicity of a mixture of 15 methylsulfonyl-polychlorinated biphenyls and 3-methylsulfonyl-2,2-bis-(4-chlorophenyl)-1,1-dichloroethene in mink (*Mustela vison*). *Environmental Toxicology and Chemistry*. 1999; 18:292–298.
- Lund J, Brandt I, Poellinger L, Bergman A, Klasson-wehler E, Gustafsson JA. Target cells for the polychlorinated biphenyl metabolite 4,4'-bis(methylsulfonyl)-2,2',5,5'-tetrachlorobiphenyl. Characterization of high affinity binding in rat and mouse lung cytosol. *Mol Pharmacol*. 1985; 27:314–323. [PubMed: 3918256]
- Lund J, Nordlund L, Devereux T, Glaumann H, Gustafsson JA. Physicochemical and immunological characterization of binding protein for PCB methyl sulfones. *Chemosphere*. 1987; 16:1677–1680.

- Luthe G, Jacobus JA, Robertson LW. Receptor interactions by polybrominated diphenyl ethers versus polychlorinated biphenyls: a theoretical Structure-activity assessment. *Environ Toxicol Pharmacol.* 2008; 25:202–210. [PubMed: 19768137]
- Ma R, Sassoon DA. PCBs exert an estrogenic effect through repression of the Wnt7a signaling pathway in the female reproductive tract. *Environ Health Perspect.* 2006; 114:898–904. [PubMed: 16759992]
- Macdonal RW, Barrie LA, Bidleman TF, Diamond ML, Gregor DJ, Semkin RG, Strachan WM, Li YF, Wania F, Alaee M, Alexeeva LB, Backus SM, Bailey R, Bewers JM, Gobeil C, Halsall CJ, Harner T, Hoff JT, Jantunen LM, Lockhart WL, Mackay D, Muir DC, Pudykiewicz J, Reimer KJ, Smith JN, Stern GA. Contaminants in the Canadian Arctic: 5 years of progress in understanding sources, occurrence and pathways. *Sci Total Environ.* 2000; 254:93–234. [PubMed: 10885446]
- Machala M, Blaha L, Lehmler HJ, Pliskova M, Majkova Z, Kapplova P, Sovadinova I, Vondracek J, Malmberg T, Robertson LW. Toxicity of hydroxylated and quinoid PCB metabolites: inhibition of gap junctional intercellular communication and activation of aryl hydrocarbon and estrogen receptors in hepatic and mammary cells. *Chem Res Toxicol.* 2004; 17:340–347. [PubMed: 15025504]
- Macintosh DL, Minegishi T, Fragala MA, Allen JG, Coghlan KM, Stewart JH, Mccarthy JF. Mitigation of building-related polychlorinated biphenyls in indoor air of a school. *Environ Health.* 2012; 11:24. [PubMed: 22490055]
- Maervoet J, Covaci A, Schepens P, Sandau CD, Letcher RJ. A reassessment of the nomenclature of polychlorinated biphenyl (PCB) metabolites. *Environ Health Perspect.* 2004; 112:291–294. [PubMed: 14998742]
- Malmberg T, Hoogstraate J, Bergman A, Klasson wehler E. Pharmacokinetics of two major hydroxylated polychlorinated biphenyl metabolites with specific retention in rat blood. *Xenobiotica.* 2004; 34:581–589. [PubMed: 15277017]
- Mannschreck A, Pustet N, Robertson L, Oesch F, Püttmann M. Enantiomers of polychlorinated biphenyls: Semi-preparative enrichment by liquid chromatography. *Liebigs Ann. Chem.* 1985:2101–2103.
- Marek RF, Martinez A, Hornbuckle KC. Discovery of hydroxylated polychlorinated biphenyls (OH-PCBs) in sediment from a lake Michigan waterway and original commercial aroclors. *Environ Sci Technol.* 2013a; 47:8204–8210. [PubMed: 23862721]
- Marek RF, Thorne PS, Wang K, Dewall J, Hornbuckle KC. PCBs and OH-PCBs in serum from children and mothers in urban and rural U.S. communities. *Environ Sci Technol.* 2013b; 47:3353–3361. [PubMed: 23452180]
- Matthews HB, Anderson MW. Effect of chlorination on the distribution and excretion of polychlorinated biphenyls. *Drug Metab. Disp.* 1975; 3:371–380.
- Matthews HB, Kato S. The metabolism and disposition of halogenated aromatics. *Ann N Y Acad Sci.* 1979; 320:131–137. [PubMed: 110188]
- Mayes BA, Mcconnell EE, Neal BH, Brunner MJ, Hamilton SB, Sullivan TM, Peters AC, Ryan MJ, Toft JD, Singer AW, Brown JF Jr, Menton RG, Moore JA. Comparative carcinogenicity in Sprague-Dawley rats of the polychlorinated biphenyl mixtures Aroclors 1016, 1242, 1254, and 1260. *Toxicol Sci.* 1998; 41:62–76. [PubMed: 9520342]
- Mcfarland VA, Clarke JU. Environmental occurrence, abundance, and potential toxicity of polychlorinated biphenyl congeners: considerations for a congener-specific analysis. *Environ Health Perspect.* 1989; 81:225–239. [PubMed: 2503374]
- Mcgraw JESR, Waller DP. Specific human CYP 450 isoform metabolism of a pentachlorobiphenyl (PCB-IUPAC# 101). *Biochem Biophys Res Commun.* 2006; 344:129–133. [PubMed: 16616008]
- Mclean MR, Bauer U, Amaro AR, Robertson LW. Identification of catechol and hydroquinone metabolites of 4-monochlorobiphenyl. *Chem Res Toxicol.* 1996a; 9:158–164. [PubMed: 8924585]
- Mclean MR, Robertson LW, Gupta RC. Detection of PCB adducts by the 32P-postlabeling technique. *Chem Res Toxicol.* 1996b; 9:165–171. [PubMed: 8924587]

- Meerts IA, Assink Y, Cenijn PH, Van den berg JH, Weijers BM, Bergman A, Koeman JH, Brouwer A. Placental transfer of a hydroxylated polychlorinated biphenyl and effects on fetal and maternal thyroid hormone homeostasis in the rat. *Toxicol Sci.* 2002; 68:361–371. [PubMed: 12151632]
- Meerts IA, Hoving S, Van den berg JH, Weijers BM, Swarts HJ, Van der beek EM, Bergman A, Koeman JH, Brouwer A. Effects of in utero exposure to 4-hydroxy-2,3,3',4',5-pentachlorobiphenyl (4-OH-CB107) on developmental landmarks, steroid hormone levels, and female estrous cyclicity in rats. *Toxicol Sci.* 2004; 82:259–267. [PubMed: 15310862]
- Meijer L, Weiss J, Van velzen M, Brouwer A, Bergman A, Sauer P. Flame retardants, polychlorinated biphenyls and insecticides in pregnant women in the northern part of The Netherlands. *Organohalogen Compounds.* 2004; 66:3552–3556.
- Melymuk L, Robson M, Helm PA, Diamond ML. PCBs, PBDEs, and PAHs in Toronto air: spatial and seasonal trends and implications for contaminant transport. *Sci Total Environ.* 2012; 429:272–280. [PubMed: 22578845]
- Menichini E, Iacovella N, Monfredini F, Turrio-baldassarri L. Atmospheric pollution by PAHs, PCDD/Fs and PCBs simultaneously collected at a regional background site in central Italy and at an urban site in Rome. *Chemosphere.* 2007; 69:422–434. [PubMed: 17604079]
- Mills RA, Millis CD, Dannan GA, Guengerich FP, Aust SD. Studies on the structure-activity relationships for the metabolism of polybrominated biphenyls by rat liver microsomes. *Toxicol Appl Pharmacol.* 1985; 78:96–104. [PubMed: 2994255]
- Mills SA 3rd, Thal DI, Barney J. A summary of the 209 PCB congener nomenclature. *Chemosphere.* 2007; 68:1603–1612. [PubMed: 17499337]
- Morales NM, Matthews HB. In vivo binding of 2,3,6,2',3',6' - hexachlorobiphenyl and 2,4,5,2',4',5' - hexachlorobiphenyl to mouse liver macromolecules. *Chem Biol Interact.* 1979; 27:99–110. [PubMed: 113111]
- Morck A, Larsen G, Wehler EK. Covalent binding of PCB metabolites to lipids: route of formation and characterization. *Xenobiotica.* 2002; 32:625–640. [PubMed: 12162858]
- Morse DC, Groen D, Veerman M, Van amerongen CJ, Koeter HB, Smits van prooije AE, Visser TJ, Koeman JH, Brouwer A. Interference of polychlorinated biphenyls in hepatic and brain thyroid hormone metabolism in fetal and neonatal rats. *Toxicol Appl Pharmacol.* 1993; 122:27–33. [PubMed: 8378931]
- Morse DC, Wehler EK, Van DE, Pas M, De bie AT, Van bladeren PJ, Brouwer A. Metabolism and biochemical effects of 3,3',4,4'-tetrachlorobiphenyl in pregnant and fetal rats. *Chem Biol Interact.* 1995; 95:41–56. [PubMed: 7697753]
- Morse DC, Wehler EK, Wesseling W, Koeman JH, Brouwer A. Alterations in rat brain thyroid hormone status following pre- and postnatal exposure to polychlorinated biphenyls (Aroclor 1254). *Toxicol Appl Pharmacol.* 1996; 136:269–279. [PubMed: 8619235]
- Mortimer RH, Landers KA, Balakrishnan B, Li H, Mitchell MD, Patel J, Richard K. Secretion and transfer of the thyroid hormone binding protein transthyretin by human placenta. *Placenta.* 2012; 33:252–256. [PubMed: 22264585]
- Newsome WH, Davies D. Determination of PCB metabolites in Canadian human milk. *Chemosphere.* 1996; 33:559–565. [PubMed: 8680832]
- Nezel T, Müller-plathe F, Müller MD, Buser H-R. Theoretical considerations about chiral PCBs and their methylthio and methylsulfonyl metabolites being possibly present as stable enantiomers. *Chemosphere.* 1997; 35:1895–1906.
- Niknam Y, Feng W, Cherednichenko G, Dong Y, Joshi SN, Vyas SM, Lehmler HJ, Pessah IN. Structure-activity relationship of selected meta- and para-hydroxylated non-dioxin like polychlorinated biphenyls: from single RyR1 channels to muscle dysfunction. *Toxicol Sci.* 2013; 136:500–513. [PubMed: 24014653]
- Nomiyama K, Murata S, Kunisue T, Yamada TK, Mizukawa H, Takahashi S, Tanabe S. Polychlorinated biphenyls and their hydroxylated metabolites (OH-PCBs) in the blood of toothed and baleen whales stranded along Japanese coastal waters. *Environ Sci Technol.* 2010; 44:3732–3738. [PubMed: 20426459]

- Noren K, Lunden A, Pettersson E, Bergman A. Methylsulfonyl metabolites of PCBs and DDE in human milk in Sweden, 1972–1992. *Environ Health Perspect.* 1996; 104:766–772. [PubMed: 8841763]
- Norstrom K, Eriksson J, Haglund J, Silvani V, Bergman A. Enantioselective formation of methyl sulfone metabolites of 2,2',3,3',4,6'-hexachlorobiphenyl in rat. *Environ Sci Technol.* 2006; 40:7649–7655. [PubMed: 17256508]
- Oakley GG, Robertson LW, Gupta RC. Analysis of polychlorinated biphenyl-DNA adducts by 32P-postlabeling. *Carcinogenesis.* 1996; 17:109–114. [PubMed: 8565118]
- Oberg M, Sjodin A, Casabona H, Nordgren I, Klasson-wehler E, Hakansson H. Tissue distribution and half-lives of individual polychlorinated biphenyls and serum levels of 4-hydroxy-2,3,3',4',5-pentachlorobiphenyl in the rat. *Toxicol Sci.* 2002; 70:171–182. [PubMed: 12441362]
- Ockenden WA, Lohmann R, Shears JR, Jones KC. The significance of PCBs in the atmosphere of the southern hemisphere. *Environ Sci Pollut Res Int.* 2001; 8:189–194. [PubMed: 11505903]
- Ohta S, Haraguchi K, Kato Y, Endo T, Kimura O, Koga N. Distribution and excretion of 2,2',3,4',5,5',6-heptachlorobiphenyl (CB187) and its metabolites in rats and guinea pigs. *Chemosphere.* 2015; 118:5–11. [PubMed: 25433397]
- Onozuka D, Yoshimura T, Kaneko S, Furue M. Mortality after exposure to polychlorinated biphenyls and polychlorinated dibenzofurans: a 40-year follow-up study of Yusho patients. *Am J Epidemiol.* 2009; 169:86–95. [PubMed: 18974082]
- Otake T, Yoshinaga J, Enomoto T, Matsuda M, Wakimoto T, Ikegami M, Suzuki E, Naruse H, Yamanaka T, Shibuya N, Yasumizu T, Kato N. Thyroid hormone status of newborns in relation to in utero exposure to PCBs and hydroxylated PCB metabolites. *Environ Res.* 2007; 105:240–246. [PubMed: 17490634]
- Park JS, Bergman A, Linderholm L, Athanasiadou M, Kocan A, Petrik J, Drobna B, Trnovec T, Charles MJ, Hertz-picciotto I. Placental transfer of polychlorinated biphenyls, their hydroxylated metabolites and pentachlorophenol in pregnant women from eastern Slovakia. *Chemosphere.* 2008; 70:1676–1684. [PubMed: 17764717]
- Park JS, Linderholm L, Charles MJ, Athanasiadou M, Petrik J, Kocan A, Drobna B, Trnovec T, Bergman A, Hertz-picciotto I. Polychlorinated biphenyls and their hydroxylated metabolites (OH-PCBS) in pregnant women from eastern Slovakia. *Environ Health Perspect.* 2007; 115:20–27. [PubMed: 17366814]
- Park JS, Petreas M, Cohn BA, Cirillo PM, Factor-litvak P. Hydroxylated PCB metabolites (OH-PCBs) in archived serum from 1950–60s California mothers: a pilot study. *Environ Int.* 2009; 35:937–942. [PubMed: 19439357]
- Parkinson A, Safe SH, Robertson LW, Thomas PE, Ryan DE, Reik LM, Levin W. Immunochemical quantitation of cytochrome P-450 isozymes and epoxide hydrolase in liver microsomes from polychlorinated or polybrominated biphenyl-treated rats. A study of structure-activity relationships. *J Biol Chem.* 1983; 258:5967–5976. [PubMed: 6304102]
- Patrick L. Thyroid disruption: mechanism and clinical implications in human health. *Altern Med Rev.* 2009; 14:326–346. [PubMed: 20030460]
- Pearce EN, Braverman LE. Environmental pollutants and the thyroid. *Best Pract Res Clin Endocrinol Metab.* 2009; 23:801–813. [PubMed: 19942155]
- Pereg D, Robertson LW, Gupta RC. DNA adduction by polychlorinated biphenyls: adducts derived from hepatic microsomal activation and from synthetic metabolites. *Chem Biol Interact.* 2002; 139:129–144. [PubMed: 11823002]
- Pereg D, Tampal N, Espandiari P, Robertson LW. Distribution and macromolecular binding of benzo[a]pyrene and two polychlorinated biphenyl congeners in female mice. *Chem Biol Interact.* 2001; 137:243–258. [PubMed: 11566292]
- Persoon C, Peters TM, Kumar N, Hornbuckle KC. Spatial distribution of airborne polychlorinated biphenyls in Cleveland, Ohio and Chicago, Illinois. *Environ Sci Technol.* 2010; 44:2797–2802. [PubMed: 20384374]
- Pessah IN, Hansen LG, Albertson TE, Garner CE, Ta TA, Do Z, Kim KH, Wong PW. Structure-activity relationship for noncoplanar polychlorinated biphenyl congeners toward the ryanodine

- receptor-Ca²⁺ channel complex type 1 (RyR1). *Chem Res Toxicol*. 2006; 19:92–101. [PubMed: 16411661]
- Pessah IN, Lehmler HJ, Robertson LW, Perez CF, Cabrales E, Bose DD, Feng W. Enantiomeric specificity of (–)-2,2',3,3',6,6'-hexachlorobiphenyl toward ryanodine receptor types 1 and 2. *Chem Res Toxicol*. 2009; 22:201–207. [PubMed: 18954145]
- Pliskova M, Vondracek J, Canton RF, Nera J, Kocan A, Petrik J, Trnovec T, Sanderson T, Van den berg M, Machala M. Impact of polychlorinated biphenyls contamination on estrogenic activity in human male serum. *Environ Health Perspect*. 2005; 113:1277–1284. [PubMed: 16203234]
- Preston BD, Miller JA, Miller EC. Non-arene oxide aromatic ring hydroxylation of 2,2',5,5'-tetrachlorobiphenyl as the major metabolic pathway catalyzed by phenobarbital-induced rat liver microsomes. *J Biol Chem*. 1983; 258:8304–8311. [PubMed: 6408087]
- Ptak A, Ludewig G, Lehmler HJ, Wojtowicz AK, Robertson LW, Gregoraszczyk EL. Comparison of the actions of 4-chlorobiphenyl and its hydroxylated metabolites on estradiol secretion by ovarian follicles in primary cells in culture. *Reprod Toxicol*. 2005; 20:57–64. [PubMed: 15808786]
- Ptak A, Ludewig G, Robertson L, Lehmler HJ, Gregoraszczyk EL. In vitro exposure of porcine prepubertal follicles to 4-chlorobiphenyl (PCB3) and its hydroxylated metabolites: effects on sex hormone levels and aromatase activity. *Toxicol Lett*. 2006; 164:113–122. [PubMed: 16412591]
- Purkey HE, Palaninathan SK, Kent KC, Smith C, Safe SH, Sacchettini JC, Kelly JW. Hydroxylated polychlorinated biphenyls selectively bind transthyretin in blood and inhibit amyloidogenesis: rationalizing rodent PCB toxicity. *Chem Biol*. 2004; 11:1719–1728. [PubMed: 15610856]
- Püttmann M, Mannschreck A, Oesch F, Robertson L. Chiral effects in the induction of drug-metabolizing enzymes using synthetic atropisomers of polychlorinated biphenyls (PCBs). *Biochem Pharmacol*. 1989; 38:1345–1352. [PubMed: 2495802]
- Püttmann M, Oesch F, Robertson L, Mannschreck A. Characteristics of polychlorinated biphenyl (PCB) atropisomers. *Chemosphere*. 1986; 15:2061–2064.
- Qin X, Lehmler HJ, Teesch LM, Robertson LW, Duffel MW. Chlorinated Biphenyl Quinones and Phenyl-2,5-benzoquinone Differentially Modify the Catalytic Activity of Human Hydroxysteroid Sulfotransferase hSULT2A1. *Chem Res Toxicol*. 2013
- Quinete N, Schettgen T, Bertram J, Kraus T. Occurrence and distribution of PCB metabolites in blood and their potential health effects in humans: a review. *Environ Sci Pollut Res Int*. 2014
- Rayne S, Forest K. pK(a) values of the monohydroxylated polychlorinated biphenyls (OH-PCBs), polybrominated biphenyls (OH-PBBs), polychlorinated diphenyl ethers (OH-PCDEs), and polybrominated diphenyl ethers (OH-PBDEs). *J Environ Sci Health A Tox Hazard Subst Environ Eng*. 2010; 45:1322–1346. [PubMed: 20658412]
- Razvi S, Shakoor A, Vanderpump M, Weaver JU, Pearce SH. The influence of age on the relationship between subclinical hypothyroidism and ischemic heart disease: a metaanalysis. *J Clin Endocrinol Metab*. 2008; 93:2998–3007. [PubMed: 18505765]
- Rickenbacher U, McKinney JD, Oatley SJ, Blake CC. Structurally specific binding of halogenated biphenyls to thyroxine transport protein. *J Med Chem*. 1986; 29:641–648. [PubMed: 3009810]
- Ring BJ, Seitz KR, Vodicknik MJ. Transfer of 2,4,5,2',4',5'-hexachlorobiphenyl across the in situ perfused guinea pig placenta. *Toxicol Appl Pharmacol*. 1988; 96:7–13. [PubMed: 3142100]
- Robertson, LW.; Gupta, R. Metabolism of polychlorinated biphenyls (PCBs) generates electrophiles and reactive oxygen species that damage DNA. In: Williams, GM.; Aruoma, OI., editors. *Molecular Drug Metabolism and Toxicology*. OICA International; 2000.
- Robertson LW, Ludewig G. Polychlorinated Biphenyl (PCB) carcinogenicity with special emphasis on airborne PCBs. *Gefahrst Reinhalt Luft*. 2011; 71:25–32. [PubMed: 21686028]
- Rodenburg LA, Du S, Fennell DE, Cavallo GJ. Evidence for widespread dechlorination of polychlorinated biphenyls in groundwater, landfills, and wastewater collection systems. *Environ Sci Technol*. 2010a; 44:7534–7540. [PubMed: 20828204]
- Rodenburg LA, Guo J, Du S, Cavallo GJ. Evidence for unique and ubiquitous environmental sources of 3,3'-dichlorobiphenyl (PCB 11). *Environmental Science & Technology*. 2010b; 44:2816–2821. [PubMed: 20384375]
- Rudel RA, Perovich LJ. Endocrine disrupting chemicals in indoor and outdoor air. *Atmos Environ* (1994). 2009; 43:170–181. [PubMed: 20047015]

- Safe S. Hydroxylated polychlorinated biphenyls (PCBs) and organochlorine pesticides as potential endocrine disruptors. *Handbook of Environmental Chemistry*. 2001; 3:155–167.
- Safe S, Bandiera S, Sawyer T, Robertson L, Safe L, Parkinson A, Thomas PE, Ryan DE, Reik LM, Levin W, et al. PCBs: structure-function relationships and mechanism of action. *Environ Health Perspect*. 1985; 60:47–56. [PubMed: 2992927]
- Sandanger TM, Dumas P, Berger U, Burkow IC. Analysis of HO-PCBs and PCP in blood plasma from individuals with high PCB exposure living on the Chukotka Peninsula in the Russian Arctic. *J Environ Monit*. 2004; 6:758–765. [PubMed: 15346180]
- Sandau CD, Ayotte P, Dewailly E, Duffe J, Norstrom RJ. Pentachlorophenol and hydroxylated polychlorinated biphenyl metabolites in umbilical cord plasma of neonates from coastal populations in Quebec. *Environ Health Perspect*. 2002; 110:411–417. [PubMed: 11940460]
- Schechter A, Colacino J, Haffner D, Patel K, Opel M, Papke O, Birnbaum L. Perfluorinated Compounds, Polychlorinated Biphenyl, and Organochlorine Pesticide Contamination in Composite Food Samples from Dallas, Texas. *Environ. Health Perspect*. 2010; 118:796–802. [PubMed: 20146964]
- Schnellmann, RG.; Vickers, AEM.; Sipes, IG. Metabolism and disposition of polychlorinated biphenyls. In: Hodgson, EBJR.; Philpot, RM., editors. *Reviews in Biochemical Toxicology*. New York, Asterdam, Oxford: Elsevier; 1985.
- Schuur AG, Brouwer A, Bergman A, Coughtrie MW, Visser TJ. Inhibition of thyroid hormone sulfation by hydroxylated metabolites of polychlorinated biphenyls. *Chem Biol Interact*. 1998a; 109:293–297. [PubMed: 9566753]
- Schuur AG, Legger FF, Van meeteren ME, Moonen MJ, Van leeuwen-bol I, Bergman A, Visser TJ, Brouwer A. In vitro inhibition of thyroid hormone sulfation by hydroxylated metabolites of halogenated aromatic hydrocarbons. *Chem Res Toxicol*. 1998b; 11:1075–1081. [PubMed: 9760282]
- Senthikumar PK, Klingelutz AJ, Jacobus JA, Lehmler H, Robertson LW, Ludewig G. Airborne polychlorinated biphenyls (PCBs) reduce telomerase activity and shorten telomere length in immortal human skin keratinocytes (HaCat). *Toxicol Lett*. 2011; 204:64–70. [PubMed: 21530622]
- Shang H, Li Y, Wang T, Wang P, Zhang H, Zhang Q, Jiang G. The presence of polychlorinated biphenyls in yellow pigment products in China with emphasis on 3,3'-dichlorobiphenyl (PCB 11). *Chemosphere*. 2014; 98:44–50. [PubMed: 24231041]
- Sharma R, Kodavanti PR. In vitro effects of polychlorinated biphenyls and hydroxy metabolites on nitric oxide synthases in rat brain. *Toxicol Appl Pharmacol*. 2002; 178:127–136. [PubMed: 11858728]
- Silberhorn EM, Glauert HP, Robertson LW. Carcinogenicity of polyhalogenated biphenyls: PCBs and PBBs. *Crit Rev Toxicol*. 1990; 20:440–496. [PubMed: 2165409]
- Silverstone AE, Rosenbaum PF, Weinstock RS, Bartell SM, Foushee HR, Shelton C, Pavuk M. Polychlorinated biphenyl (PCB) exposure and diabetes: results from the Anniston Community Health Survey. *Environ Health Perspect*. 2012; 120:727–732. [PubMed: 22334129]
- Sjodin A, Hagmar L, Klasson-wehler E, Bjork J, Bergman A. Influence of the consumption of fatty Baltic Sea fish on plasma levels of halogenated environmental contaminants in Latvian and Swedish men. *Environ Health Perspect*. 2000; 108:1035–1041. [PubMed: 11102293]
- Sjodin A, Tullsten AK, Klasson-wehler E. Identification of the parent compounds to selectively retained hydroxylated PCB metabolites in rat blood plasma. *Organohalogen Compounds*. 1998; 37:365–368.
- Sleight S. Effects of PCBs and related compounds on hepatocarcinogenesis in rats and mice. *Environ Health Perspect*. 1985; 60:35–39. [PubMed: 2992924]
- Soechitram SD, Athanasiadou M, Hovander L, Bergman A, Sauer PJ. Fetal exposure to PCBs and their hydroxylated metabolites in a Dutch cohort. *Environ Health Perspect*. 2004; 112:1208–1212. [PubMed: 15289169]
- Song Y, Buettner GR, Parkin S, Wagner BA, Robertson LW, Lehmler HJ. Chlorination increases the persistence of semiquinone free radicals derived from polychlorinated biphenyl hydroquinones and quinones. *J Org Chem*. 2008a; 73:8296–8304. [PubMed: 18839991]

- Song Y, Wagner BA, Lehmler HJ, Buettner GR. Semiquinone radicals from oxygenated polychlorinated biphenyls: electron paramagnetic resonance studies. *Chem Res Toxicol.* 2008b; 21:1359–1367. [PubMed: 18549251]
- Song Y, Wagner BA, Witmer JR, Lehmler HJ, Buettner GR. Nonenzymatic displacement of chlorine and formation of free radicals upon the reaction of glutathione with PCB quinones. *Proc Natl Acad Sci U S A.* 2009; 106:9725–9730. [PubMed: 19497881]
- Srinivasan A, Lehmler HJ, Robertson LW, Ludewig G. Production of DNA strand breaks in vitro and reactive oxygen species in vitro and in HL-60 cells by PCB metabolites. *Toxicol Sci.* 2001; 60:92–102. [PubMed: 11222876]
- Srinivasan A, Robertson LW, Ludewig G. Sulfhydryl binding and topoisomerase inhibition by PCB metabolites. *Chem Res Toxicol.* 2002; 15:497–505. [PubMed: 11952335]
- Sun P, Basu I, Hites RA. Temporal trends of polychlorinated biphenyls in precipitation and air at Chicago. *Environ Sci Technol.* 2006; 40:1178–1183. [PubMed: 16572772]
- Tampal N, Lehmler HJ, Espandiari P, Malmberg T, Robertson LW. Glucuronidation of hydroxylated polychlorinated biphenyls (PCBs). *Chem Res Toxicol.* 2002; 15:1259–1266. [PubMed: 12387623]
- Tampal N, Myers S, Robertson LW. Binding of polychlorinated biphenyls/metabolites to hemoglobin. *Toxicol Lett.* 2003; 142:53–60. [PubMed: 12765239]
- Tanabe S. PCB problems in the future: foresight from current knowledge. *Environ Pollut.* 1988; 50:5–28. [PubMed: 15092651]
- Tilson HA, Kodavanti PR. The neurotoxicity of polychlorinated biphenyls. *Neurotoxicology.* 1998; 19:517–525. [PubMed: 9745906]
- Todaka T, Hori T, Hirakawa H, Kajiwara J, Yasutake D, Onozuka D, Iida T, Furue M. Concentrations of polychlorinated biphenyls in blood of Yusho patients over 35 years after the incident. *Chemosphere.* 2009; 74:902–909. [PubMed: 19070886]
- Troisi GM, Haraguchi K, Kaydoo DS, Nyman M, Aguilar A, Borrell A, Siebert U, Mason CF. Bioaccumulation of polychlorinated biphenyls (PCBs) and dichlorodiphenylethane (DDE) methyl sulfones in tissues of seal and dolphin morbillivirus epizootic victims. *J Toxicol Environ Health A.* 2001; 62:1–8. [PubMed: 11205532]
- Van den berg KJ, Zurcher C, Brouwer A. Effects of 3,4,3',4'-tetrachlorobiphenyl on thyroid function and histology in marmoset monkeys. *Toxicol Lett.* 1988; 41:77–86. [PubMed: 3128898]
- Van den hurk P, Kubiczak GA, Lehmler HJ, James MO. Hydroxylated polychlorinated biphenyls as inhibitors of the sulfation and glucuronidation of 3-hydroxy-benzo[a]pyrene. *Environ Health Perspect.* 2002; 110:343–348. [PubMed: 11940451]
- Vansell NR, Muppidi JR, Habeebu SM, Klaassen CD. Promotion of thyroid tumors in rats by pregnenolone-16alpha-carbonitrile (PCN) and polychlorinated biphenyl (PCB). *Toxicol Sci.* 2004; 81:50–59. [PubMed: 15201439]
- Vermiglio F, Lo presti VP, Moleti M, Sidoti M, Tortorella G, Scaffidi G, Castagna MG, Mattina F, Violi MA, Crisa A, Artemisia A, Trimarchi F. Attention deficit and hyperactivity disorders in the offspring of mothers exposed to mild-moderate iodine deficiency: a possible novel iodine deficiency disorder in developed countries. *J Clin Endocrinol Metab.* 2004; 89:6054–6060. [PubMed: 15579758]
- Waller SC, He YA, Harlow GR, He YQ, Mash EA, Halpert JR. 2,2',3,3',6,6'-hexachlorobiphenyl hydroxylation by active site mutants of cytochrome P450 2B1 and 2B11. *Chem Res Toxicol.* 1999; 12:690–699. [PubMed: 10458702]
- Wang LQ, Lehmler HJ, Robertson LW, James MO. Polychlorobiphenyls are selective inhibitors of human phenol sulfotransferase 1A1 with 4-nitrophenol as a substrate. *Chem Biol Interact.* 2006; 159:235–246. [PubMed: 16413005]
- Wangpradit O, Mariappan SV, Teesch LM, Duffel MW, Norstrom K, Robertson LW, Luthe G. Oxidation of 4-chlorobiphenyl metabolites to electrophilic species by prostaglandin H synthase. *Chem Res Toxicol.* 2009; 22:64–71. [PubMed: 19105592]
- Warner NA, Martin JW, Wong CS. Chiral polychlorinated biphenyls are biotransformed enantioselectively by mammalian cytochrome P-450 isozymes to form hydroxylated metabolites. *Environ Sci Technol.* 2009; 43:114–121. [PubMed: 19209593]

- Weintraub M, Birnbaum LS. Catfish consumption as a contributor to elevated PCB levels in a non-Hispanic black subpopulation. *Environ Res.* 2008; 107:412–417. [PubMed: 18407261]
- Weistrand C, Noren K. Methylsulfonyl metabolites of PCBs and DDE in human tissues. *Environ Health Perspect.* 1997; 105:644–649. [PubMed: 9288499]
- Weistrand C, Noren K, Nilsson A. Occupational exposure. Organochlorine compounds in blood plasma from potentially exposed workers. PCB, PCN, PCDD/F, HCB and methylsulfonyl metabolites of PCB. *Environmental Science and Pollution Research International.* 1997; 4:2–9. [PubMed: 19002410]
- Wethington DM 3RD, Hornbuckle KC. Milwaukee, WI, as a source of atmospheric PCBs to Lake Michigan. *Environ Sci Technol.* 2005; 39:57–63. [PubMed: 15667075]
- Wu X, Duffel M, Lehmler HJ. Oxidation of polychlorinated biphenyls by liver tissue slices from phenobarbital-pretreated mice is congener-specific and atropselective. *Chem Res Toxicol.* 2013a; 26:1642–1651. [PubMed: 24107130]
- Wu X, Kammerer A, Lehmler HJ. Microsomal oxidation of 2,2',3,3',6,6'-hexachlorobiphenyl (PCB 136) results in species-dependent chiral signatures of the hydroxylated metabolites. *Environ Sci Technol.* 2014; 48:2436–2444. [PubMed: 24467194]
- Wu X, Kania-korwel I, Chen H, Stamou M, Dammanahalli KJ, Duffel M, Lein PJ, Lehmler HJ. Metabolism of 2,2',3,3',6,6'-hexachlorobiphenyl (PCB 136) atropisomers in tissue slices from phenobarbital or dexamethasone-induced rats is sex-dependent. *Xenobiotica.* 2013b; 43:933–947. [PubMed: 23581876]
- Wu X, Pramanik A, Duffel MW, Hrycay EG, Bandiera SM, Lehmler HJ, Kania-korwel I. 2,2',3,3',6,6'-Hexachlorobiphenyl (PCB 136) is enantioselectively oxidized to hydroxylated metabolites by rat liver microsomes. *Chem. Res. Toxicol.* 2011; 24:2249–2257. [PubMed: 22026639]
- Xie W, Wang K, Robertson LW, Ludewig G. Investigation of mechanism(s) of DNA damage induced by 4-monochlorobiphenyl (PCB3) metabolites. *Environ Int.* 2010; 36:950–961. [PubMed: 20129669]
- Yang D, Kania-korwel I, Ghogha A, Chen H, Stamou M, Bose DD, Pessah IN, Lehmler HJ, Lein PJ. PCB 136 atropselectively alters morphometric and functional parameters of neuronal connectivity in cultured rat hippocampal neurons via ryanodine receptor-dependent mechanisms. *Toxicol Sci.* 2014; 138:379–392. [PubMed: 24385416]
- Zettner MA, Flor S, Ludewig G, Wagner J, Robertson LW, Lehmann L. Quinoid metabolites of 4-monochlorobiphenyl induce gene mutations in cultured Chinese hamster v79 cells. *Toxicol Sci.* 2007; 100:88–98. [PubMed: 17686921]
- Zhai G, Hu D, Lehmler HJ, Schnoor JL. Enantioselective biotransformation of chiral PCBs in whole poplar plants. *Environ Sci Technol.* 2011; 45:2308–2316. [PubMed: 21329345]
- Zhai G, Lehmler HJ, Schnoor JL. Sulfate metabolites of 4-monochlorobiphenyl in whole poplar plants. *Environ Sci Technol.* 2013a; 47:557–562. [PubMed: 23215248]
- Zhai G, Wu X, Lehmler HJ, Schnoor JL. Atropisomeric determination of chiral hydroxylated metabolites of polychlorinated biphenyls using HPLC-MS. *Chem Cent J.* 2013b; 7:183. [PubMed: 24360245]
- Zhang X, Diamond ML, Robson M, Harrad S. Sources, emissions, and fate of polybrominated diphenyl ethers and polychlorinated biphenyls indoors in Toronto, Canada. *Environ Sci Technol.* 2011; 45:3268–3274. [PubMed: 21413794]
- Zhao HX, Adamcakova-dodd A, Hu D, Hornbuckle KC, Just CL, Robertson LW, Thorne PS, Lehmler HJ. Development of a synthetic PCB mixture resembling the average polychlorinated biphenyl profile in Chicago air. *Environ Int.* 2010; 36:819–827. [PubMed: 19375801]
- Zhao S, Narang A, Ding X, Eadon G. Characterization and quantitative analysis of DNA adducts formed from lower chlorinated PCB-derived quinones. *Chem Res Toxicol.* 2004; 17:502–511. [PubMed: 15089092]
- Zhu Y, Mapuskar KA, Marek RF, Xu W, Lehmler HJ, Robertson LW, Hornbuckle KC, Spitz DR, Aykin-burns N. A new player in environmentally induced oxidative stress: polychlorinated biphenyl congener, 3,3'-dichlorobiphenyl (PCB11). *Toxicol Sci.* 2013; 136:39–50. [PubMed: 23997111]

Zoeller RT, Rovet J. Timing of thyroid hormone action in the developing brain: clinical observations and experimental findings. *J Neuroendocrinol.* 2004; 16:809–818. [PubMed: 15500540]

Author Manuscript

Author Manuscript

Author Manuscript

Author Manuscript

Summary of PCB metabolism and associated toxicities

- Although many biologic effects of parent PCBs are receptor-mediated, including the well-described characteristics of PCBs as inducers of xenobiotic metabolism, other PCB toxication processes involve the metabolism of PCBs themselves or their metabolic progeny.
- PCBs are in general metabolized via initial oxidation to arene oxides, but may also undergo direct insertion of hydroxyl groups. These reactions, catalyzed by CYPs, are regio- and stereo-selective. Retention of PCB atropisomers in tissues, for example, may be highly enantioselective.
- Reactive electrophilic PCB metabolites, arene oxides, semi-quinones and quinones, may form adducts to biomacromolecules, i.e. proteins, DNA, RNA and lipids.
- The major stable PCB metabolites are polychlorobiphenylols (OH-PCBs) that are, depending on their structure, either rapidly metabolized, excreted or retained in certain compartments in the body, primarily the blood.
- Major OH-PCB congeners derived from HC-PCBs that are found in blood are present in concentrations similar to the most persistent individual PCB congeners. They are also more easily transferred than the parent compounds via the placenta to the fetus.
- PCB congeners with non-chlorinated meta-/para-positions and chlorinated neighboring ortho-/meta-positions and a slowly reacting second phenyl ring are rapidly metabolized; these PCB congeners form OH-PCBs and MeSO₂-PCBs.
- Several MeSO₂-PCBs are accumulated in a highly tissue-specific manner, especially in liver and lung.
- Recent evidence supports sulfation as a major metabolic pathway for LC-PCBs *in vitro* and *in vivo*, and provides initial evidence for biological activity of the resulting sulfate ester metabolites.
- Conjugated PCB metabolites such as PCB sulfates and glucuronides may have been overlooked classes of PCB metabolites in the past and total PCB exposure levels, particularly to LC-PCBs in exposed populations, may have been underestimated. Novel procedures to reliably quantify PCB sulfates and potentially other conjugated metabolites, such as glucuronic acid derivatives, in human serum could fill this significant gap in the literature and could help to more accurately estimate human exposure levels.

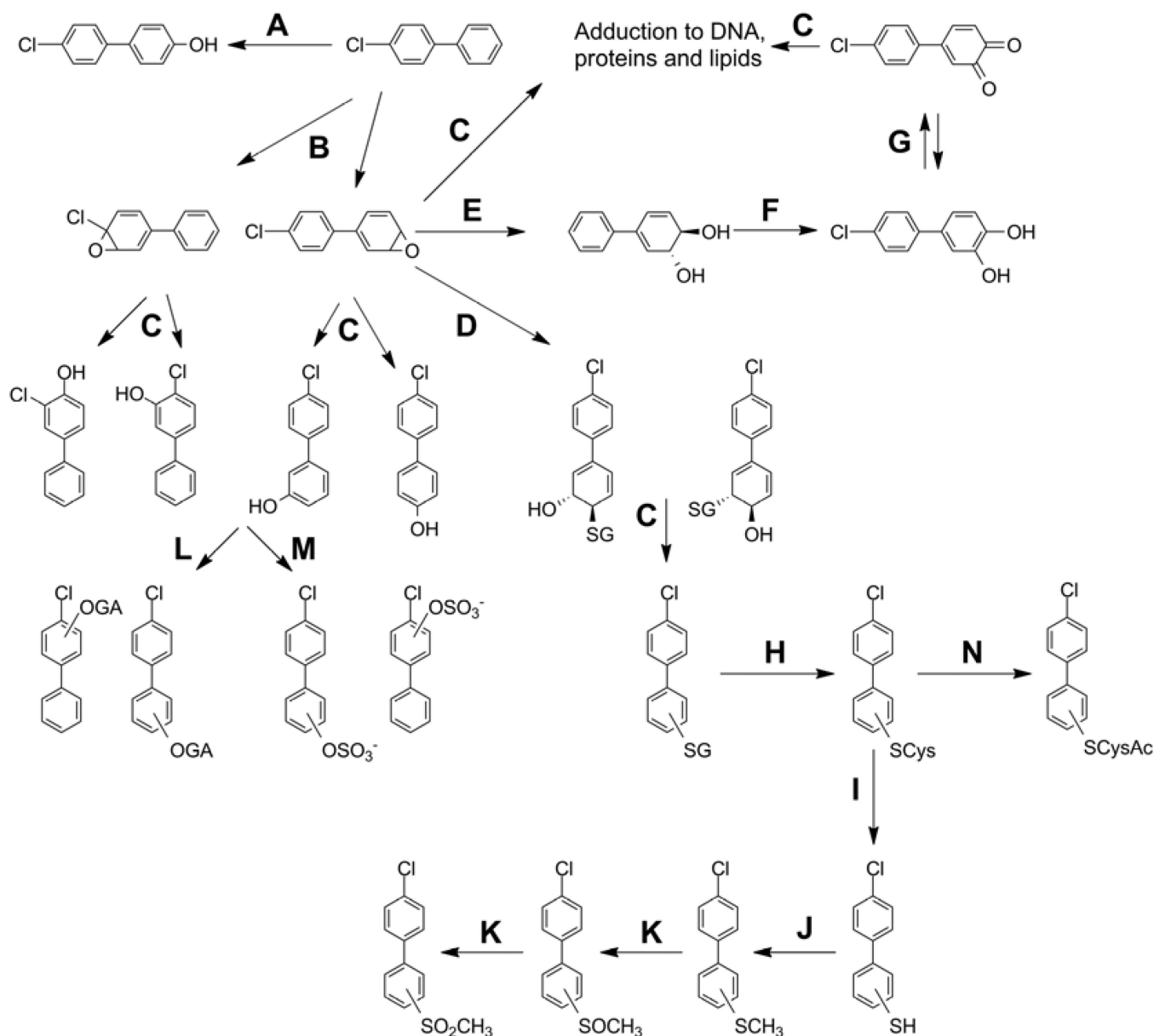


Figure 1.

General metabolic scheme for a representative lower-chlorinated PCB congener, PCB 3. Enzymes involved in the metabolism are indicated by the letters A, B and D-N with the letter C indicating non-enzymatic transformations. Enzymes suggested for these transformations are listed as follows: A, Cytochrome P-450 (CYP) enzyme system, Direct insertion in meta position; CYP2B (rodents); B, Cytochrome P-450 enzyme system; CYB2B1 (rodents); CYP3A4 (humans); Non-coplanar PCBs: CYB2B, 2C, 3A ; C, Non-enzymatic reaction; D, Glutathione *S*-transferase; E, Epoxide hydrolase; F, Dihydrodiol dehydrogenase (AKR1C); G, Autooxidation and/or Peroxidases; H, γ -Glutamyl transpeptidase, then cysteinylglycine dipeptidase; I, Cysteine *S*-conjugate β -lyase; J, Thiol *S*-methyltransferase; K, CYP and/or FAD-containing monooxygenases (FMO); L, UDP-

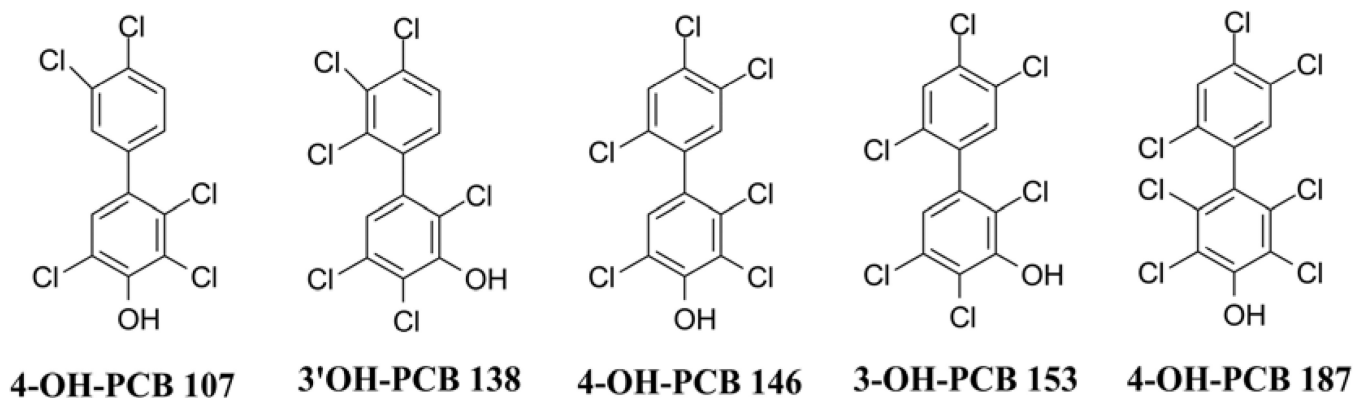
glucuronosyl transferase (UGT); M, Sulfotransferase (SULT); N, Cysteine *S*-conjugate *N*-acetyltransferase

Author Manuscript

Author Manuscript

Author Manuscript

Author Manuscript

**Figure 2.**

Chemical structures of the most common OH-PCB congeners present in human blood (c.f. Table 2). The full names of the five OH-PCBs are: 2,3,3',4',5-pentachloro-4-biphenylol (4-OH-PCB 107); 2,2',3',4,4',5- hexachloro-3-biphenylol (3'-OH-PCB 138); 2,2',3,4',5,5'- hexachloro-4-biphenylol (4-OH-PCB 146); 2,2',4,4',5,5'- hexachloro-3-biphenylol (3-OH-PCB 153) and 2,2',3,4',5,5',6 - heptachloro-4-biphenylol (4-OH-PCB 187).

Table 1

The twenty most frequently detected PCB congeners in air and their potential sources

	Major Sources	Detection in	
		Paint pigments	Human serum
52	Aroclors 1016, 1242, 1248, 1254	Y	Y
20/28	Aroclors 1016, 1242, 1248	Y	Y
11	Pigment and dye production	Y	Y
95	Aroclor 1254	Y	Y
31	Aroclors 1016, 1242, 1248	Y	Y
18/30	Aroclors 1016, 1242	Y	Y
8	Aroclors 1221, 1016, 1242	Y	Y
61/70/74/76	Aroclors 1242, 1248, 1254	Y	Y
3	Aroclor 1221	Y	Y
4	Aroclor 1221, 1016, 1242	Y	Y
90/101/113	Aroclor 1254	Y	Y
21/33	Aroclors 1016, 1242	Y	Y
16	Aroclors 1016, 1242 (>3%)	Y	Y
15	Aroclors 1016, 1242	Y	Y
110	Aroclor 1254	Y	Y
17	Aroclors 1016, 1242 (>3%)	ND	Y
49/69	Aroclors 1016, 1248 (>3)	ND	Y
83/99	Aroclor 1254	ND	Y
118	Aroclor 1254	Y	Y

1) Congeners were ranked by frequency of occurrence in analyzed air samples.

2) For the co-eluting peaks, the bolded congeners are believed to be primary congeners.

3) Y: detected; ND: non-detectable

Table 2

List of all possible mono-OH-PCB congeners, showing the calculated pKa values and octanol:water coefficients of the un-ionized state

ID ^I	Compound ^I	pKa ^I	log p ^I
	2-Chlorobiphenyl ⁵		
1	3-OH-PCB 1	8.36	4.07
2	4-OH-PCB 1	8.77	4.07 4.04 ⁴
3	5-OH-PCB 1	9.29	4.04
4	6-OH-PCB 1	9.79	4.05
5	2'-OH-PCB 1	10.63	4.03
6	3'-OH-PCB 1	9.81	4.03
7	4'-OH-PCB 1	9.55	4.05
	3-Chlorobiphenyl		
8	2-OH-PCB 2	9.1	4.04
9	4-OH-PCB2	7.89	4.07
10	5-OH-PCB 2	9.05	4.03
11	6-OH-PCB 2	9.99	4
12	2'-OH-PCB 2	10.49	3.99
13	3'-OH-PCB 2	9.85	4
14	4'-OH-PCB 2	9.41	4.01
	4-Chlorobiphenyl		
15	2-OH-PCB 3	9.74	4
16	3-OH-PCB 3	8.33	4.04
17	2'-OH-PCB 3	10.5 9.48 ²	3.98 3.51 ²
18	3'-OH-PCB 3	9.87 9.62 ²	3.98 3.80 ²
19	4'-OH-PCB 3	9.42 9.71 ² 9.61 ³	3.97 3.77 ² 3.77 ³ 4.04 ⁴
	2,2'-Dichlorobiphenyl		
20	3-OH-PCB 4	8.26	4.82
21	4-OH-PCB4	8.69	4.81
22	5-OH-PCB 4	9.19	4.79
23	6-OH-PCB4	9.86	4.78
	2,3-Dichlorobiphenyl		
24	4-OH-PCB 5	7.2	4.83
25	5-OH-PCB 5	8.43	4.82
26	6-OH-PCB 5	9.17	4.78

ID ^I	Compound ^I	pKa ^I	log p ^I
27	2'-OH-PCB 5	10.52	4.78
28	3'-OH-PCB 5	9.75	4.8
29	4'-OH-PCB 5	9.5	4.79
	2,3'-Dichlorobiphenyl		
30	3-OH-PCB 6	8.3	4.8
31	4-OH-PCB 6	8.71	4.77
32	5-OH-PCB 6	9.23	4.75
33	6-OH-PCB 6	9.67	4.76
34	2'-OH-PCB 6	9.14	4.77
35	4'-OH-PCB 6	7.98 8.17 ³	4.81 4.21 ³
36	5'-OH-PCB 6	8.95	4.77
37	6'-OH-PCB 6	10.01	4.76
	2,4-Dichlorobiphenyl		
38	3-OH-PCB 7	6.81	4.78
39	5-OH-PCB 7	7.71	4.8
40	6-OH-PCB 7	8.92	4.77
41	2'-OH-PCB 7	10.53	4.76
42	3'-OH-PCB 7	9.76	4.75
43	4'-OH-PCB 7	9.51	4.77
	2,4'-Dichlorobiphenyl		
44	3-OH-PCB 8	8.31	4.77
45	4-OH-PCB 8	8.73 8.70 ³	4.77 4.57 ³
46	5-OH-PCB 8	9.24	4.75
47	6-OH-PCB 8	9.68	4.75
48	2'-OH-PCB 8	9.77	4.76
49	3'-OH-PCB 8	8.23	4.79
	2,5-Dichlorobiphenyl		
50	3-OH-PCB 9	7.49	4.8
51	4-OH-PCB 9	7.2	4.83
52	6-OH-PCB 9	8.28	4.75
53	2'-OH-PCB 9	10.52	4.77
54	3'-OH-PCB 9	9.75	4.75
55	4'-OH-PCB 9	9.5 9.25 ³	4.78 4.17 ³ 4.63 ⁴
	2,6-Dichlorobiphenyl		
56	3-OH-PCB 10	7.73	4.82
57	4-OH-PCB 10	7.91	4.84

ID ^I	Compound ^I	pKa ^I	log p ^I
58	2'-OH-PCB 10	10.66	4.78
59	3'-OH-PCB 10	9.71	4.8
60	4'-OH-PCB 10	9.47	4.8 4.63 ⁴
	3,3'-Dichlorobiphenyl		
61	2-OH-PCB 11	8.99	4.75
62	4-OH-PCB 11	7.84 8.23 ³	4.77 4.29 ³
63	5-OH-PCB 11	8.99	4.73
64	6-OH-PCB 11	9.87	4.71
	3,4-Dichlorobiphenyl		
65	2-OH-PCB 12	8.25	4.75
66	5-OH-PCB 12	7.47	4.79
67	6-OH-PCB 12	9.12	4.76
68	2'-OH-PCB 12	10.39	4.75
69	3'-OH-PCB 12	9.81	4.75
70	4'-OH-PCB 12	9.37 9.26 ³	4.77 4.21 ³
	3,4'-Dichlorobiphenyl		
71	2-OH-PCB 13	9	4.73
72	4-OH-PCB 13	7.85	4.75
73	5-OH-PCB 13	9.01	4.72
74	6-OH-PCB 13	9.88	4.7
75	2'-OH-PCB 13	9.63	4.71
76	3'-OH-PCB 13	8.27	4.74
	3,5-Dichlorobiphenyl		
77	2-OH-PCB 14	8.48	4.75
78	4-OH-PCB 14	6.35 6.94 ³	4.8 4.27 ³
79	2'-OH-PCB 14	10.38	4.74
80	3'-OH-PCB 14	9.79	4.75
81	4'-OH-PCB 14	9.35 6.94 ²	4.76 4.27 ²
	4,4'-Dichlorobiphenyl		
82	2-OH-PCB 15	9.64	4.68
83	3-OH-PCB 15	8.29	4.73
	2,2',3-Trichlorobiphenyl		
84	4-OH-PCB 16	7.13	5.24
85	5-OH-PCB 16	8.32	5.15
86	6-OH-PCB 16	9.24	5.22

ID ^I	Compound ^I	pKa ^I	log p ^I
87	3'-OH-PCB 16	8.2	5.19
88	4'-OH-PCB 16	8.64	5.2
89	5'-OH-PCB 16	9.13	5.16
90	6'-OH-PCB 16	9.74	5.2
	2,2',4-Trichlorobiphenyl		
91	3-OH-PCB 17	6.71	5.25
92	5-OH-PCB 17	7.61	5.14
93	6-OH-PCB 17	8.99	5.17
94	3'-OH-PCB 17	8.21	5.15
95	4'-OH-PCB 17	8.65	5.26
96	5'-OH-PCB 17	9.14	5.22
97	6'-OH-PCB 17	9.75	5.1
	2,2',5-Trichlorobiphenyl		
98	3-OH-PCB 18	7.4	5.19
99	4-OH-PCB 18	7.12	5.17
100	6-OH-PCB 18	8.36	5.23
101	3'-OH-PCB 18	8.2	5.15
102	4'-OH-PCB 18	8.64	5.26
103	5'-OH-PCB 18	9.13	5.21
104	6'-OH-PCB 18	9.74	5.11
	2,2',6-Trichlorobiphenyl		
105	3-OH-PCB 19	7.64	5.23
106	4-OH-PCB 19	7.87	5.18
107	3'-OH-PCB 19	8.16	5.21
108	4'-OH-PCB 19	8.65	5.19
109	5'-OH-PCB 19	9.08	5.15
110	6'-OH-PCB 19	9.89	5.22
	2,3,3'-Trichlorobiphenyl		
111	4-OH-PCB 20	7.15	5.19
112	5-OH-PCB 20	8.37	5.16
113	6-OH-PCB 20	9.05	5.12
114	2'-OH-PCB 20	9.02	5.2
115	4'-OH-PCB 20	7.93	5.18
116	5'-OH-PCB 20	8.89	5.16
117	6'-OH-PCB 20	9.9	5.1
	2,3,4-Trichlorobiphenyl		
118	5-OH-PCB 21	6.85	5.3
119	6-OH-PCB 21	8.3	5.27
120	2'-OH-PCB 21	10.41	5.2

ID ^I	Compound ^I	pKa ^I	log p ^I
121	3'-OH-PCB 21	9.7	5.16
122	4'-OH-PCB 21	9.46	5.23
	2,3,4'-Trichlorobiphenyl		
123	4-OH-PCB 22	7.16	5.18
124	5-OH-PCB 22	8.38	5.17
125	6-OH-PCB 22	9.06	5.11
126	2'-OH-PCB 22	9.65	5.1
127	3'-OH-PCB 22	8.17	5.15
	2,3,5-Trichlorobiphenyl		
128	4-OH-PCB 23	5.66	5.32
129	6-OH-PCB 23	7.66	5.26
130	2'-OH-PCB 23	10.4	5.12
131	3'-OH-PCB 23	9.69	5.17
132	4'-OH-PCB 23	9.45	5.26
	2,3,6-Trichlorobiphenyl		
133	4-OH-PCB 24	6.35	5.32
134	5-OH-PCB 24	6.87	5.3
135	2'-OH-PCB 24	10.54	5.21
136	3'-OH-PCB 24	9.65	5.16
137	4'-OH-PCB 24	9.42	5.22
	2,3',4-Trichlorobiphenyl		
138	3-OH-PCB 25	6.75	5.18
139	5-OH-PCB 25	7.65	5.15
140	6-OH-PCB 25	8.81	5.1
141	2'-OH-PCB 25	9.03	5.11
142	4'-OH-PCB 25	7.94 7.77 ³	5.23 4.79 ³
143	5'-OH-PCB 25	8.9	5.23
144	6'-OH-PCB 25	9.91	5.12
	2,3',5-Trichlorobiphenyl		
145	3-OH-PCB 26	7.43	5.14
146	4-OH-PCB 26	7.14	5.18
147	6-OH-PCB 26	8.17	5.17
148	2'-OH-PCB 26	9.02	5.12
149	4'-OH-PCB 26	7.93	5.22
150	5'-OH-PCB 26	8.89	5.21
151	6'-OH-PCB 26	9.9	5.12
	2,3',6-Trichlorobiphenyl		
152	3-OH-PCB 27	7.67	5.15

ID ^I	Compound ^I	pKa ^I	log p ^I
153	4-OH-PCB 27	7.86	5.18
154	2'-OH-PCB 27	9.17	5.21
155	4'-OH-PCB 27	7.9	5.18
156	5'-OH-PCB 27	8.84	5.15
157	6'-OH-PCB 27	10.04	5.11
	2,4,4'-Trichlorobiphenyl		
158	3-OH-PCB 28	6.76	5.14
159	5-OH-PCB 28	7.66	5.17
160	6-OH-PCB 28	8.82	5.1
161	2'-OH-PCB 28	9.66	5.12
162	3'-OH-PCB 28	8.18	5.17
	2,4,5-Trichlorobiphenyl		
163	3-OH-PCB 29	5.95	5.29
164	6-OH-PCB 29	7.43	5.26
165	2'-OH-PCB 29	10.41	5.12
166	3'-OH-PCB 29	9.7	5.19
167	4'-OH-PCB 29	9.46	5.27
	2,4,6-Trichlorobiphenyl		
168	3-OH-PCB 30	6.18	5.29
169	2'-OH-PCB 30	10.55	5.13
170	3'-OH-PCB 30	9.66	5.17
171	4'-OH-PCB 30	9.44	5.27 5.22 ⁴
	2,4',5-Trichlorobiphenyl		
172	3-OH-PCB 31	7.45	5.15
173	4-OH-PCB 31	7.16	5.2
174	6-OH-PCB 31	8.18	5.11
175	2'-OH-PCB 31	9.65	5.12
176	3'-OH-PCB 31	8.17	5.16
	2,4',6-Trichlorobiphenyl		
177	3-OH-PCB 32	7.69	5.15
178	4-OH-PCB 32	7.87	5.19
179	2'-OH-PCB 32	9.79	5.1
180	3'-OH-PCB 32	8.13	5.15
	2,3',4'-Trichlorobiphenyl		
181	3-OH-PCB 33	8.25	5.15
182	4-OH-PCB 33	8.68 8.33 ³	5.27 5.01 ³
183	5-OH-PCB 33	9.18	5.23

ID ^I	Compound ^I	pKa ^I	log p ^I
184	6-OH-PCB 33	9.57	5.1
185	2'-OH-PCB 33	8.28	5.21
186	5'-OH-PCB 33	7.37	5.18
187	6'-OH-PCB 33	9.15	5.12
	2,3',5'-Trichlorobiphenyl		
188	3-OH-PCB 34	8.24	5.15
189	4-OH-PCB 34	8.66	5.27
190	5-OH-PCB 34	9.17	5.24
191	6-OH-PCB 34	9.55	5.1
192	2'-OH-PCB 34	8.52	5.19
193	4'-OH-PCB 34	6.43 6.57 ² 6.71 ³	5.2 4.74 ² 4.74 ³
	3,3',4-Trichlorobiphenyl		
194	2-OH-PCB 35	8.13	5.17
195	5-OH-PCB 35	7.41	5.15
196	6-OH-PCB 35	9.01	5.11
197	2'-OH-PCB 35	8.88	5.11
198	4'-OH-PCB 35	7.8 7.82 ² 7.82 ³	5.23 4.74 ² 4.74 ³
199	5'-OH-PCB 35	8.95	5.23
200	6'-OH-PCB 35	9.77 8.56 ³	5.14 4.86 ³
	3,3',5-Trichlorobiphenyl		
201	2-OH-PCB 36	8.36	5.11
202	4-OH-PCB 36	6.29 6.72 ² 6.78 ³	5.18 4.83 ² 4.83 ³
203	2'-OH-PCB 36	8.87	5.1
204	4'-OH-PCB 36	7.78 7.86 ² 7.86 ³	5.24 4.87 ² 4.87 ³
205	5'-OH-PCB 36	8.93	5.23
206	6'-OH-PCB 36	9.76	5.15
	3,4,4'-Trichlorobiphenyl		
207	2-OH-PCB 37	8.14	5.12
208	5-OH-PCB 37	7.43	5.15
209	6-OH-PCB 37	9.02	5.13
210	2'-OH-PCB 37	9.52	5.13
211	3'-OH-PCB 37	8.23	5.19

ID ^I	Compound ^I	pKa ^I	log p ^I
	3,4,5-Trichlorobiphenyl		
212	2-OH-PCB 38	7.62	5.26
213	2'-OH-PCB 38	10.27	5.13
214	3'-OH-PCB 38	9.75	5.2
215	4'-OH-PCB 38	9.32	5.26
	3,4',5'-Trichlorobiphenyl		
216	2-OH-PCB 39	8.38	5.11
217	4-OH-PCB 39	6.31 6.81 ²	5.18 4.84 ² 5.22 ⁴
218	2'-OH-PCB 39	9.51	5.15
219	3'-OH-PCB 39	8.21	5.2
	2,2',3,3'-Tetrachlorobiphenyl		
220	4-OH-PCB 40	7.08	5.89
221	5-OH-PCB 40	8.26	5.89
222	6-OH-PCB 40	9.12	5.86
	2,2',3,4'-Tetrachlorobiphenyl		
223	5-OH-PCB 41	6.75	5.87
224	6-OH-PCB 41	8.37	5.85
225	3'-OH-PCB 41	8.15	5.88
226	4'-OH-PCB 41	8.6	5.93
227	5'-OH-PCB 41	9.08	5.92
228	6'-OH-PCB 41	9.64	5.85
	2,2',3,4'-Tetrachlorobiphenyl		
229	4-OH-PCB 42	7.09	5.9
230	5-OH-PCB 42	8.28	5.91
231	6-OH-PCB 42	9.13	5.86
232	3'-OH-PCB 42	6.65	5.86
233	5'-OH-PCB 42	7.55	5.88
234	6'-OH-PCB 42	8.88	5.84
	2,2',3,5'-Tetrachlorobiphenyl		
235	4-OH-PCB 43	5.58	5.91
236	6-OH-PCB 43	7.74	5.85
237	3'-OH-PCB 43	8.14	5.89
238	4'-OH-PCB 43	8.59	5.94
239	5'-OH-PCB 43	9.07	5.91
240	6'-OH-PCB 43	9.63	5.86
	2,2',3,5'-Tetrachlorobiphenyl		
241	4-OH-PCB 44	7.08	5.9

ID ^I	Compound ^I	pKa ^I	log p ^I
242	5-OH-PCB 44	8.26	5.9
243	6-OH-PCB 44	9.12	5.86
244	3'-OH-PCB 44	7.34	5.87
245	4'-OH-PCB 44	7.07	5.9
246	6'-OH-PCB 44	8.25	5.85
	2,2',3,6-Tetrachlorobiphenyl		
247	4-OH-PCB 45	6.31	5.9
248	5-OH-PCB 45	6.77	5.88
249	3'-OH-PCB 45	8.1	5.88
250	4'-OH-PCB 45	8.6	5.93
251	5'-OH-PCB 45	9.02	5.92
252	6'-OH-PCB 45	9.78	5.87
	2,2',3,6'-Tetrachlorobiphenyl		
253	4-OH-PCB 46	7.09	5.89
254	5-OH-PCB 46	8.22	5.89
255	6-OH-PCB 46	9.27	5.87
256	3'-OH-PCB 46	7.58	5.88
257	4'-OH-PCB 46	7.82	5.92
	2,2',4,4'-Tetrachlorobiphenyl		
258	3-OH-PCB 47	6.66	5.87
259	5-OH-PCB 47	7.56	5.9
260	6-OH-PCB 47	8.89	5.87
	2,2',4,5-Tetrachlorobiphenyl		
261	3-OH-PCB 48	5.85	5.88
262	6-OH-PCB 48	7.51	5.85
263	3'-OH-PCB 48	8.15	5.89
264	4'-OH-PCB 48	8.6	5.93
265	5'-OH-PCB 48	9.08	5.91
266	6'-OH-PCB 48	9.64	5.86
	2,2',4,5'-Tetrachlorobiphenyl		
267	3-OH-PCB 49	6.65	5.87
268	5-OH-PCB 49	7.55	5.9
269	6-OH-PCB 49	8.88	5.87
270	3'-OH-PCB 49	7.35	5.89
271	4'-OH-PCB 49	7.08	5.91
272	6'-OH-PCB 49	8.26	5.84
	2,2',4,6-Tetrachlorobiphenyl		
273	3-OH-PCB 50	6.09	5.87
274	3'-OH-PCB 50	8.11	5.88

ID ^I	Compound ^I	pKa ^I	log p ^I
275	4'-OH-PCB 50	8.61	5.93
276	5'-OH-PCB 50	9.03	5.91
277	6'-OH-PCB 50	9.79	5.86
	2,2',4,6'-Tetrachlorobipheny		
278	3-OH-PCB 51	6.61	5.87
279	5-OH-PCB 51	7.5	5.87
280	6-OH-PCB 51	9.03	5.85
281	3'-OH-PCB 51	7.59	5.89
282	4'-OH-PCB 51	7.83	5.93
	2,2',5,5'-Tetrachlorobipheny		
283	3-OH-PCB 52	7.34	5.89
284	4-OH-PCB 52	7.07	5.91
285	6-OH-PCB 52	8.25	5.84
	2,2',5,6'-Tetrachlorobipheny		
286	3-OH-PCB 53	7.3	5.87
287	4-OH-PCB 53	7.08	5.9
288	6-OH-PCB 53	8.41	5.85
289	3'-OH-PCB 53	7.58	5.88
290	4'-OH-PCB 53	7.82	5.93
	2,2',6,6'-Tetrachlorobipheny		
291	3-OH-PCB 54	7.54	5.89
292	4-OH-PCB 54	7.83	5.92
	2,3,3',4-Tetrachlorobipheny		
293	5-OH-PCB 55	6.79	5.87
294	6-OH-PCB 55	8.19	5.85
295	2'-OH-PCB 55	8.92	5.85
296	4'-OH-PCB 55	7.89	5.92
297	5'-OH-PCB 55	8.84	5.9
298	6'-OH-PCB 55	9.79	5.87
	2,3,3',4'-Tetrachlorobipheny		
299	4-OH-PCB 56	7.11	5.91
300	5-OH-PCB 56	8.32	5.9
301	6-OH-PCB 56	8.95	5.87
302	2'-OH-PCB 56	8.17	5.84
303	5'-OH-PCB 56	7.31	5.88
304	6'-OH-PCB 56	9.03	5.84
	2,3,3',5-Tetrachlorobipheny		
305	4-OH-PCB 57	5.61	5.88
306	6-OH-PCB 57	7.55	5.85

ID ^I	Compound ^I	pKa ^I	log p ^I
307	2'-OH-PCB 57	8.91	5.83
308	4'-OH-PCB 57	7.87	5.92
309	5'-OH-PCB 57	8.83	5.9
310	6'-OH-PCB 57	9.78	5.88
	2,3,3',5'-Tetrachlorobiphenyl		
311	4-OH-PCB 58	7.1	5.91
312	5-OH-PCB 58	8.31	5.9
313	6-OH-PCB 58	8.93	5.88
314	2'-OH-PCB 58	8.4	5.83
315	4'-OH-PCB 58	6.38	5.9
	2,3,3',6'-Tetrachlorobiphenyl		
316	4-OH-PCB 59	6.3	5.88
317	5-OH-PCB 59	6.81	5.86
318	2'-OH-PCB 59	9.06	5.86
319	4'-OH-PCB 59	7.85	5.92
320	5'-OH-PCB 59	8.78	5.91
321	6'-OH-PCB 59	9.92	5.87
	2,3,4,4'-Tetrachlorobiphenyl		
322	5-OH-PCB 60	6.81	5.86
323	6-OH-PCB 60	8.2	5.84
324	2'-OH-PCB 60	9.55	5.85
325	3'-OH-PCB 60	8.12	5.89
	2,3,4,5-Tetrachlorobiphenyl		
326	6-OH-PCB 61	6.81	5.92
327	2'-OH-PCB 61	10.3	5.87
328	3'-OH-PCB 61	9.64	5.9
329	4'-OH-PCB 61	9.41	5.93 5.81 ⁴
	2,3,4,6-Tetrachlorobiphenyl		
330	5-OH-PCB 62	5.33	5.94
331	2'-OH-PCB 62	10.44	5.87
332	3'-OH-PCB 62	9.6	5.9
333	4'-OH-PCB 62	9.39	5.94
	2,3,4',5-Tetrachlorobiphenyl		
334	4-OH-PCB 63	5.62	5.88
335	6-OH-PCB 63	7.56	5.83
336	2'-OH-PCB 63	9.54	5.87
337	3'-OH-PCB 63	8.11	5.9
	2,3,4',6-Tetrachlorobiphenyl		

ID ^I	Compound ^I	pKa ^I	log p ^I
338	4-OH-PCB 64	6.31	5.89
339	5-OH-PCB 64	6.82	5.87
340	2'-OH-PCB 64	9.68	5.86
341	3'-OH-PCB 64	8.07	5.9
	2,3,5,6-Tetrachlorobiphenyl		
342	4-OH-PCB 65	4.81	5.95
343	2'-OH-PCB 65	10.43	5.87
344	3'-OH-PCB 65	9.59	5.89
345	4'-OH-PCB 65	9.37	5.94
	2,3',4,4'-Tetrachlorobiphenyl		
346	3-OH-PCB 66	6.7	5.88
347	5-OH-PCB 66	7.6	5.9
348	6-OH-PCB 66	8.7	5.87
349	2'-OH-PCB 66	8.18	5.83
350	5'-OH-PCB 66	7.33	5.89
351	6'-OH-PCB 66	9.04	5.87
	2,3',4,5-Tetrachlorobiphenyl		
352	3-OH-PCB 67	5.89	5.86
353	6-OH-PCB 67	7.31	5.84
354	2'-OH-PCB 67	8.92	5.84
355	4'-OH-PCB 67	7.89	5.91
356	5'-OH-PCB 67	8.84	5.89
357	6'-OH-PCB 67	9.79	5.89
	2,3',4,5'-Tetrachlorobiphenyl		
358	3-OH-PCB 68	6.69	5.88
359	5-OH-PCB 68	7.59	5.9
360	6-OH-PCB 68	8.69	5.87
361	2'-OH-PCB 68	8.41	5.85
362	4'-OH-PCB 68	6.39 6.30 ² 6.30 ³	5.91 5.33 ² 5.33 ³
	2,3',4,6-Tetrachlorobiphenyl		
363	3-OH-PCB 69	6.12	5.86
364	2'-OH-PCB 69	9.07	5.84
365	4'-OH-PCB 69	7.87	5.92
366	5'-OH-PCB 69	8.8	5.9
367	6'-OH-PCB 69	9.93	5.88
	2,3',4',5-Tetrachlorobiphenyl		
368	3-OH-PCB 70	7.39	5.89

ID ^I	Compound ^I	pKa ^I	log p ^I
369	4-OH-PCB 70	7.11	5.92
370	6-OH-PCB 70	8.06	5.84
371	2'-OH-PCB 70	8.17	5.83
372	5'-OH-PCB 70	7.31	5.89
373	6'-OH-PCB 70	9.03	5.87
	2,3',4',6-Tetrachlorobiphenyl		
374	3-OH-PCB 71	7.63	5.9
375	4-OH-PCB 71	7.82	5.92
376	2'-OH-PCB 71	8.32	5.84
377	5'-OH-PCB 71	7.27	5.88
378	6'-OH-PCB 71	9.17	5.84
	2,3',5,5'-Tetrachlorobiphenyl		
379	3-OH-PCB 72	7.37	5.9
380	4-OH-PCB 72	7.09	5.91
381	6-OH-PCB 72	8.05	5.84
382	2'-OH-PCB 72	8.4	5.85
383	4'-OH-PCB 72	6.38	5.91
	2,3',5',6-Tetrachlorobiphenyl		
384	3-OH-PCB 73	7.62	5.9
385	4-OH-PCB 73	7.81	5.93
386	2'-OH-PCB 73	8.55	5.84
387	4'-OH-PCB 73	6.36	5.89
	2,4,4',5-Tetrachlorobiphenyl		
388	3-OH-PCB 74	5.9	5.85
389	6-OH-PCB 74	7.32	5.83
390	2'-OH-PCB 74	9.55	5.88
391	3'-OH-PCB 74	8.12	5.9
	2,4,4',6-Tetrachlorobiphenyl		
392	3-OH-PCB 75	6.14	5.86
393	2'-OH-PCB 75	9.69	5.87
394	3'-OH-PCB 75	8.08	5.9
	2,3',4',5'-Tetrachlorobiphenyl		
395	3-OH-PCB 76	8.19	5.89
396	4-OH-PCB 76	8.62	5.93
397	5-OH-PCB 76	9.12	5.91
398	6-OH-PCB 76	9.45	5.85
399	2'-OH-PCB 76	7.66	5.85
	3,3',4,4'-Tetrachlorobiphenyl		
400	2-OH-PCB 77	8.03	5.84

ID ^I	Compound ^I	pKa ^I	log p ^I
401	5-OH-PCB 77	7.37	5.9
402	6-OH-PCB 77	8.9	5.87
	3,3',4,5-Tetrachlorobiphenyl		
403	2-OH-PCB 78	7.51	5.83
404	2'-OH-PCB 78	8.77	5.85
405	4'-OH-PCB 78	7.75 6.36 ²	5.92 5.28 ²
406	5'-OH-PCB 78	8.89	5.9
407	6'-OH-PCB 78	9.65	5.88
	3,3',4,5'-Tetrachlorobiphenyl		
408	2-OH-PCB 79	8.01	5.85
409	5-OH-PCB 79	7.35	5.89
410	6-OH-PCB 79	8.89	5.87
411	2'-OH-PCB 79	8.26	5.86
412	4'-OH-PCB 79	6.26 6.36 ³	5.91 5.28 ³
	3,3',5,5'-Tetrachlorobiphenyl		
413	2-OH-PCB 80	8.25	5.86
414	4-OH-PCB 80	6.24	5.91
	3,4,4',5-Tetrachlorobiphenyl		
415	2-OH-PCB 81	7.52	5.84
416	2'-OH-PCB 81	9.41	5.87
417	3'-OH-PCB 81	8.17	5.89
	2,2',3,3',4-Pentachlorobiphenyl		
418	5-OH-PCB 82	6.69	6.55
419	6-OH-PCB 82	8.26	6.51
420	4'-OH-PCB 82	7.04	6.57
421	5'-OH-PCB 82	8.22	6.57
422	6'-OH-PCB 82	9.02	6.53
	2,2',3,3',5-Pentachlorobiphenyl		
423	4-OH-PCB 83	5.53	6.56
424	6-OH-PCB 83	7.63	6.51
425	4'-OH-PCB 83	7.03	6.57
426	5'-OH-PCB 83	8.2	6.58
427	6'-OH-PCB 83	9.01	6.53
	2,2',3,3',6-Pentachlorobiphenyl		
428	4-OH-PCB 84	6.26	6.59
429	5-OH-PCB 84	6.72	6.55
430	4'-OH-PCB 84	7.04	6.59

ID ^I	Compound ^I	pKa ^I	log p ^I
431	5'-OH-PCB 84	8.16	6.57
432	6'-OH-PCB 84	9.16	6.54
	2,2',3,4,4'-Pentachlorobiphenyl		
433	5-OH-PCB 85	6.7	6.54
434	6-OH-PCB 85	8.27	6.52
435	3'-OH-PCB 85	6.6	6.52
436	5'-OH-PCB 85	7.5	6.54
437	6'-OH-PCB 85	8.77	6.52
	2,2',3,4,5-Pentachlorobiphenyl		
438	6-OH-PCB 86	6.89	6.5
439	3'-OH-PCB 86	8.09	6.56
440	4'-OH-PCB 86	8.55	6.61
441	5'-OH-PCB 86	9.02	6.58
442	6'-OH-PCB 86	9.52	6.53
	2,2',3,4,5'-Pentachlorobiphenyl		
443	5-OH-PCB 87	6.69	6.54
444	6-OH-PCB 87	8.26	6.52
445	3'-OH-PCB 87	7.29	6.55
446	4'-OH-PCB 87	7.03	6.56
447	6'-OH-PCB 87	8.15	6.51
	2,2',3,4,6-Pentachlorobiphenyl		
448	5-OH-PCB 88	5.23	6.53
449	3'-OH-PCB 88	8.05	6.57
450	4'-OH-PCB 88	8.56	6.62
451	5'-OH-PCB 88	8.97	6.58
452	6'-OH-PCB 88	9.67	6.52
	2,2',3,4,6'-Pentachlorobiphenyl		
453	5-OH-PCB 89	6.65	6.55
454	6-OH-PCB 89	8.41	6.51
455	3'-OH-PCB 89	7.53	6.57
456	4'-OH-PCB 89	7.78	6.6
	2,2',3,4',5-Pentachlorobiphenyl		
457	4-OH-PCB 90	5.55	6.54
458	6-OH-PCB 90	7.64	6.52
459	3'-OH-PCB 90	6.59	6.54
460	5'-OH-PCB 90	7.49	6.55
461	6'-OH-PCB 90	8.76	6.53
	2,2',3,4',6-Pentachlorobiphenyl		
462	4-OH-PCB 91	6.27	6.57

ID ^I	Compound ^I	pKa ^I	log p ^I
463	5-OH-PCB 91	6.73	6.57
464	3'-OH-PCB 91	6.55	6.54
465	5'-OH-PCB 91	7.45	6.56
466	6'-OH-PCB 91	8.91	6.53
	2,2',3,5,5'-Pentachlorobiphenyl		
467	4-OH-PCB 92	5.53	6.54
468	6-OH-PCB 92	7.63	6.52
469	3'-OH-PCB 92	7.28	6.55
470	4'-OH-PCB 92	7.02	6.58
471	6'-OH-PCB 92	8.13	6.52
	2,2',3,5,6-Pentachlorobiphenyl		
472	4-OH-PCB 93	4.77	6.55
473	3'-OH-PCB 93	8.04	6.58
474	4'-OH-PCB 93	8.55	6.62
475	5'-OH-PCB 93	8.96	6.58
476	6'-OH-PCB 93	9.66	6.53
	2,2',3,5,6'-Pentachlorobiphenyl		
477	4-OH-PCB 94	5.55	6.57
478	6-OH-PCB 94	7.79	6.5
479	3'-OH-PCB 94	7.52	6.57
480	4'-OH-PCB 94	7.77	6.61
	2,2',3,5',6-Pentachlorobiphenyl		
481	4-OH-PCB 95	6.26	6.58
482	5-OH-PCB 95	6.72	6.56
483	3'-OH-PCB 95	7.24	6.58
484	4'-OH-PCB 95	7.03	6.59
485	6'-OH-PCB 95	8.3	6.51
	2,2',3,6,6'-Pentachlorobiphenyl		
486	4-OH-PCB 96	6.27	6.57
487	5-OH-PCB 96	6.68	6.56
488	3'-OH-PCB 96	7.48	6.58
489	4'-OH-PCB 96	7.78	6.6
	2,2',3,4',5'-Pentachlorobiphenyl		
490	4-OH-PCB 97	7.04	6.56
491	5-OH-PCB 97	8.22	6.57
492	6-OH-PCB 97	9.02	6.53
493	3'-OH-PCB 97	5.79	6.53
494	6'-OH-PCB 97	7.4	6.51
	2,2',3,4',6'-Pentachlorobiphenyl		

ID ^I	Compound ^I	pKa ^I	log p ^I
495	4-OH-PCB 98	7.05	6.59
496	5-OH-PCB 98	8.17	6.57
497	6-OH-PCB 98	9.17	6.53
498	3'-OH-PCB 98	6.03	6.54
	2,2',4,4',5-Pentachlorobiphenyl		
499	3-OH-PCB 99	5.81	6.51
500	6-OH-PCB 99	7.41	6.5
501	3'-OH-PCB 99	6.6	6.54
502	5'-OH-PCB 99	7.5	6.55
503	6'-OH-PCB 99	8.77	6.52
	2,2',4,4',6-Pentachlorobiphenyl		
504	3-OH-PCB 100	6.04	6.54
505	3'-OH-PCB 100	6.57	6.55
506	5'-OH-PCB 100	7.46	6.56
507	6'-OH-PCB 100	8.92	6.53
	2,2',4,5,5'-Pentachlorobiphenyl		
508	3-OH-PCB 101	5.79	6.52
509	6-OH-PCB 101	7.39	6.51
510	3'-OH-PCB 101	7.29	6.54
511	4'-OH-PCB 101	7.03	6.59
512	6'-OH-PCB 101	8.14	6.52
	2,2',4,5,6'-Pentachlorobiphenyl		
513	3-OH-PCB 102	5.76	6.54
514	6-OH-PCB 102	7.56	6.49
515	3'-OH-PCB 102	7.53	6.56
516	4'-OH-PCB 102	7.78	6.61
	2,2',4,5',6-Pentachlorobiphenyl		
517	3-OH-PCB 103	6.03	6.54
518	3'-OH-PCB 103	7.25	6.56
519	4'-OH-PCB 103	7.04	6.58
520	6'-OH-PCB 103	8.31	6.52
	2,2',4,6,6'-Pentachlorobiphenyl		
521	3-OH-PCB 104	5.99	6.55
522	3'-OH-PCB 104	7.49	6.56
523	4'-OH-PCB 104	7.79	6.62
	2,3,3',4,4'-Pentachlorobiphenyl		
524	5-OH-PCB 105	6.75	6.53
525	6-OH-PCB 105	8.08	6.51
526	2'-OH-PCB 105	8.06	6.49

ID ^I	Compound ^I	pKa ^I	log p ^I
527	5'-OH-PCB 105	7.27	6.54
528	6'-OH-PCB 105	8.93	6.51
	2,3,3',4,5-Pentachlorobiphenyl		
529	6-OH-PCB 106	6.69	6.46
530	2'-OH-PCB 106	8.8	6.51
531	4'-OH-PCB 106	7.84	6.57
532	5'-OH-PCB 106	8.78	6.55
533	6'-OH-PCB 106	9.68	6.53
	2,3,3',4',5-Pentachlorobiphenyl		
534	4-OH-PCB 107	5.57	6.54
535	6-OH-PCB 107	7.44	6.51
536	2'-OH-PCB 107	8.05	6.5
537	5'-OH-PCB 107	7.25	6.52
538	6'-OH-PCB 107	8.92	6.51
	2,3,3',4,5'-Pentachlorobiphenyl		
539	5-OH-PCB 108	6.73	6.54
540	6-OH-PCB 108	8.07	6.51
541	2'-OH-PCB 108	8.3	6.51
542	4'-OH-PCB 108	6.34	6.55
	2,3,3',4,6-Pentachlorobiphenyl		
543	5-OH-PCB 109	5.27	6.5
544	2'-OH-PCB 109	8.96	6.51
545	4'-OH-PCB 109	7.81	6.58
546	5'-OH-PCB 109	8.74	6.57
547	6'-OH-PCB 109	9.82	6.53
	2,3,3',4',6-Pentachlorobiphenyl		
548	4-OH-PCB 110	6.26	6.57
549	5-OH-PCB 110	6.76	6.55
550	2'-OH-PCB 110	8.21	6.51
551	5'-OH-PCB 110	7.21	6.54
552	6'-OH-PCB 110	9.06	6.54
	2,3,3',5,5'-Pentachlorobiphenyl		
553	4-OH-PCB 111	5.55	6.54
554	6-OH-PCB 111	7.43	6.51
555	2'-OH-PCB 111	8.29	6.52
556	4'-OH-PCB 111	6.33	6.55
	2,3,3',5,6-Pentachlorobiphenyl		
557	4-OH-PCB 112	4.76	6.53
558	2'-OH-PCB 112	8.95	6.51

ID ^I	Compound ^I	pKa ^I	log p ^I
559	4'-OH-PCB 112	7.8	6.59
560	5'-OH-PCB 112	8.72	6.58
561	6'-OH-PCB 112	9.81	6.54
	2,3,3',5',6-Pentachlorobiphenyl		
562	4-OH-PCB 113	6.24	6.57
563	5-OH-PCB 113	6.75	6.56
564	2'-OH-PCB 113	8.44	6.52
565	4'-OH-PCB 113	6.31	6.57
	2,3,4,4',5-Pentachlorobiphenyl		
566	6-OH-PCB 114	6.7	6.48
567	2'-OH-PCB 114	9.43	6.52
568	3'-OH-PCB 114	8.06	6.54
	2,3,4,4',6-Pentachlorobiphenyl		
569	5-OH-PCB 115	5.28	6.52
570	2'-OH-PCB 115	9.57	6.53
571	3'-OH-PCB 115	8.02	6.55
	2,3,4,5,6-Pentachlorobiphenyl		
572	2'-OH-PCB 116	10.32	6.5
573	3'-OH-PCB 116	9.54	6.59
574	4'-OH-PCB 116	9.33	6.61
	2,3,4',5,6-Pentachlorobiphenyl		
575	4-OH-PCB 117	4.77	6.54
576	2'-OH-PCB 117	9.56	6.53
577	3'-OH-PCB 117	8.01	6.55
	2,3',4,4',5-Pentachlorobiphenyl		
578	3-OH-PCB 118	5.84	6.51
579	6-OH-PCB 118	7.21	6.5
580	2'-OH-PCB 118	8.06	6.5
581	5'-OH-PCB 118	7.27	6.53
582	6'-OH-PCB 118	8.93	6.52
	2,3',4,4',6-Pentachlorobiphenyl		
583	3-OH-PCB 119	6.08	6.54
584	2'-OH-PCB 119	8.22	6.51
585	5'-OH-PCB 119	7.22	6.53
586	6'-OH-PCB 119	9.07	6.51
	2,3',4,5,5'-Pentachlorobiphenyl		
587	3-OH-PCB 120	5.83	6.51
588	6-OH-PCB 120	7.2	6.49
589	2'-OH-PCB 120	8.3	6.5

ID ^I	Compound ^I	pKa ^I	log p ^I
590	4'-OH-PCB 120	6.34	6.56
	2,3',4,5',6-Pentachlorobiphenyl		
591	3-OH-PCB 121	6.06	6.54
592	2'-OH-PCB 121	8.45	6.52
593	4'-OH-PCB 121	6.32	6.55
	2,3,3',4',5'-Pentachlorobiphenyl		
594	4-OH-PCB 122	7.06	6.56
595	5-OH-PCB 122	8.26	6.56
596	6-OH-PCB 122	8.83	6.53
597	2'-OH-PCB 122	7.55	6.5
	2,3',4,4',5'-Pentachlorobiphenyl		
598	3-OH-PCB 123	6.64	6.52
599	5-OH-PCB 123	7.54	6.55
600	6-OH-PCB 123	8.59	6.51
601	2'-OH-PCB 123	7.56	6.5
	2,3',4',5,5'-Pentachlorobiphenyl		
602	3-OH-PCB 124	7.33	6.53
603	4-OH-PCB 124	7.05	6.57
604	6-OH-PCB 124	7.95	6.51
605	2'-OH-PCB 124	7.55	6.49
	2,3',4',5',6-Pentachlorobiphenyl		
606	3-OH-PCB 125	7.57	6.56
607	4-OH-PCB 125	7.77	6.6
608	2'-OH-PCB 125	7.7	6.51
	3,3',4,4',5-Pentachlorobiphenyl		
609	2-OH-PCB 126	7.41	6.49
610	2'-OH-PCB 126	7.91	6.5
611	5'-OH-PCB 126	7.31	6.53
612	6'-OH-PCB 126	8.79	6.51
	3,3',4,5,5'-Pentachlorobiphenyl		
613	2-OH-PCB 127	7.39	6.5
614	2'-OH-PCB 127	8.15	6.5
615	4'-OH-PCB 127	6.21	6.53
	2,2',3,3',4,4'-Hexachlorobiphenyl		
616	5-OH-PCB 128	6.64	7.13
617	6-OH-PCB 128	8.15	7.11
	2,2',3,3',4,5-Hexachlorobiphenyl		
618	6-OH-PCB 129	6.78	7.04
619	4'-OH-PCB 129	6.99	7.18

ID ^I	Compound ^I	pKa ^I	log p ^I
620	5'-OH-PCB 129	8.16	7.2
621	6'-OH-PCB 129	8.9	7.12
	2,2',3,3',4,5'-Hexachlorobiphenyl		
622	5-OH-PCB 130	6.63	7.15
623	6-OH-PCB 130	8.14	7.12
624	4'-OH-PCB 130	5.49 5.04 ²	7.13 6.17 ²
625	6'-OH-PCB 130	7.53	7.11
	2,2',3,3',4,6'-Hexachlorobiphenyl		
626	5-OH-PCB 131	5.17	7.11
627	4'-OH-PCB 131	7	7.17
628	5'-OH-PCB 131	8.11	7.2
629	6'-OH-PCB 131	9.05	7.11
	2,2',3,3',4,6'-Hexachlorobiphenyl		
630	5-OH-PCB 132	6.59	7.14
631	6-OH-PCB 132	8.29	7.1
632	4'-OH-PCB 132	6.22	7.16
633	5'-OH-PCB 132	6.67	7.14
	2,2',3,3',5,5'-Hexachlorobiphenyl		
634	4-OH-PCB 133	5.48	7.13
635	6-OH-PCB 133	7.52	7.1
	2,2',3,3',5,6'-Hexachlorobiphenyl		
636	4-OH-PCB 134	4.7	7.17
637	4'-OH-PCB 134	6.99	7.17
638	5'-OH-PCB 134	8.1	7.2
639	6'-OH-PCB 134	9.05	7.12
	2,2',3,3',5,6'-Hexachlorobiphenyl		
640	4-OH-PCB 135	5.49	7.14
641	6-OH-PCB 135	7.68	7.1
642	4'-OH-PCB 135	6.21	7.17
643	5'-OH-PCB 135	6.66	7.15
	2,2',3,3',6,6'-Hexachlorobiphenyl		
644	4-OH-PCB 136	6.22	7.16
645	5-OH-PCB 136	6.62	7.15
	2,2',3,4,4',5'-Hexachlorobiphenyl		
646	6-OH-PCB 137	6.79	7.07
647	3'-OH-PCB 137	6.54	7.12
648	5'-OH-PCB 137	7.44	7.2
649	6'-OH-PCB 137	8.66	7.12

ID ^I	Compound ^I	pKa ^I	log p ^I
	2,2',3,4,4',5'-Hexachlorobiphenyl		
650	5-OH-PCB 138	6.64	7.18
651	6-OH-PCB 138	8.15	7.13
652	3'-OH-PCB 138	5.75	7.11
653	6'-OH-PCB 138	7.29	7.08
	2,2',3,4,4',6'-Hexachlorobiphenyl		
654	5-OH-PCB 139	5.18	7.11
655	3'-OH-PCB 139	6.51	7.13
656	5'-OH-PCB 139	7.4	7.2
657	6'-OH-PCB 139	8.81	7.12
	2,2',3,4,4',6'-Hexachlorobiphenyl		
658	5-OH-PCB 140	6.6	7.15
659	6-OH-PCB 140	8.3	7.11
660	3'-OH-PCB 140	5.98	7.12
	2,2',3,4,5,5'-Hexachlorobiphenyl		
661	6-OH-PCB 141	6.78	7.05
662	3'-OH-PCB 141	7.23	7.17
663	4'-OH-PCB 141	6.98	7.22
664	6'-OH-PCB 141	8.03	7.1
	2,2',3,4,5,6'-Hexachlorobiphenyl		
665	3'-OH-PCB 142	8	7.25
666	4'-OH-PCB 142	8.51	7.21
667	5'-OH-PCB 142	8.92	7.17
668	6'-OH-PCB 142	9.56	7.12
	2,2',3,4,5,6'-Hexachlorobiphenyl		
669	6-OH-PCB 143	6.94	7.05
670	3'-OH-PCB 143	7.47	7.15
671	4'-OH-PCB 143	7.73	7.23
	2,2',3,4,5',6'-Hexachlorobiphenyl		
672	5-OH-PCB 144	5.17	7.11
673	3'-OH-PCB 144	7.19	7.16
674	4'-OH-PCB 144	6.99	7.22
675	6'-OH-PCB 144	8.19	7.1
	2,2',3,4,6,6'-Hexachlorobiphenyl		
676	5-OH-PCB 145	5.14	7.08
677	3'-OH-PCB 145	7.43	7.16
678	4'-OH-PCB 145	7.74	7.23
	2,2',3,4',5,5'-Hexachlorobiphenyl		

ID ^I	Compound ^I	pKa ^I	log p ^I
679	4-OH-PCB 146	5.49 5.04 ²	7.19 6.23 ²
680	6-OH-PCB 146	7.53	7.1
681	3'-OH-PCB 146	5.73	7.11
682	6'-OH-PCB 146	7.28	7.08
	2,2',3,4',5,6'-Hexachlorobiphenyl		
683	4-OH-PCB 147	4.73	7.12
684	3'-OH-PCB 147	6.49	7.14
685	5'-OH-PCB 147	7.39	7.19
686	6'-OH-PCB 147	8.8	7.11
	2,2',3,4',5,6'-Hexachlorobiphenyl		
687	4-OH-PCB 148	5.51	7.14
688	6-OH-PCB 148	7.69	7.1
689	3'-OH-PCB 148	5.97	7.14
	2,2',3,4',5,6'-Hexachlorobiphenyl		
690	4-OH-PCB 149	6.22	7.2
691	5-OH-PCB 149	6.67	7.17
692	3'-OH-PCB 149	5.7	7.13
693	6'-OH-PCB 149	7.45	7.1
	2,2',3,4',6,6'-Hexachlorobiphenyl		
694	4-OH-PCB 150	6.23	7.18
695	5-OH-PCB 150	6.63	7.15
696	3'-OH-PCB 150	5.93	7.12
	2,2',3,5,5',6'-Hexachlorobiphenyl		
697	4-OH-PCB 151	4.72	7.23
698	3'-OH-PCB 151	7.18	7.15
699	4'-OH-PCB 151	6.98	7.21
700	6'-OH-PCB 151	8.19	7.11
	2,2',3,5,6,6'-Hexachlorobiphenyl		
701	4-OH-PCB 152	4.73	7.11
702	3'-OH-PCB 152	7.42	7.16
703	4'-OH-PCB 152	7.73	7.23
	2,2',4,4',5,5'-Hexachlorobiphenyl		
704	3-OH-PCB 153	5.75	7.15
705	6-OH-PCB 153	7.29	7.09
	2,2',4,4',5,6'-Hexachlorobiphenyl		
706	3-OH-PCB 154	5.71	7.12
707	6-OH-PCB 154	7.46	7.1
708	3'-OH-PCB 154	5.98	7.15

ID ^I	Compound ^I	pKa ^I	log p ^I
	2,2',4,4',6,6'-Hexachlorobiphenyl		
709	3-OH-PCB 155	5.94	7.15
	2,3,3',4,4',5-Hexachlorobiphenyl		
710	6-OH-PCB 156	6.59	7.1
711	2'-OH-PCB 156	7.95	7.09
712	5'-OH-PCB 156	7.21	7.18
713	6'-OH-PCB 156	8.81	7.16
	2,3,3',4,4',5'-Hexachlorobiphenyl		
714	5-OH-PCB 157	6.69	7.18
715	6-OH-PCB 157	7.97	7.13
716	2'-OH-PCB 157	7.44	7.09
	2,3,3',4,4',6-Hexachlorobiphenyl		
717	5-OH-PCB 158	5.22	7.12
718	2'-OH-PCB 158	8.1	7.1
719	5'-OH-PCB 158	7.16	7.19
720	6'-OH-PCB 158	8.95	7.14
	2,3,3',4,5,5'-Hexachlorobiphenyl		
721	6-OH-PCB 159	6.58	7.09
722	2'-OH-PCB 159	8.18	7.11
723	4'-OH-PCB 159	6.29	7.2
	2,3,3',4,5,6-Hexachlorobiphenyl		
724	2'-OH-PCB 160	8.84	7.11
725	4'-OH-PCB 160	7.76	7.17
726	5'-OH-PCB 160	8.68	7.17
727	6'-OH-PCB 160	9.7	7.11
	2,3,3',4,5',6-Hexachlorobiphenyl		
728	5-OH-PCB 161	5.21	7.11
729	2'-OH-PCB 161	8.33	7.1
730	4'-OH-PCB 161	6.27	7.2
	2,3,3',4',5,5'-Hexachlorobiphenyl		
731	4-OH-PCB 162	5.52	7.18
732	6-OH-PCB 162	7.33	7.1
733	2'-OH-PCB 162	7.43	7.09
	2,3,3',4',5,6-Hexachlorobiphenyl		
734	4-OH-PCB 163	4.72	7.12
735	2'-OH-PCB 163	8.09	7.11
736	5'-OH-PCB 163	7.15	7.17
737	6'-OH-PCB 163	8.94	7.14
	2,3,3',4',5',6-Hexachlorobiphenyl		

ID ^I	Compound ^I	pKa ^I	log p ^I
738	4-OH-PCB 164	6.21	7.21
739	5-OH-PCB 164	6.7	7.16
740	2'-OH-PCB 164	7.59	7.1
	2,3,3',5,5',6-Hexachlorobiphenyl		
741	4-OH-PCB 165	4.71	7.13
742	2'-OH-PCB 165	8.32	7.11
743	4'-OH-PCB 165	6.26	7.19
	2,3,4,4',5,6-Hexachlorobiphenyl		
744	2'-OH-PCB 166	8.16	6.94
745	3'-OH-PCB 166	7.57	6.95
	2,3',4,4',5,5'-Hexachlorobiphenyl		
746	3-OH-PCB 167	5.78	7.13
747	6-OH-PCB 167	7.09	7.08
748	2'-OH-PCB 167	7.44	7.12
	2,3',4,4',5',6-Hexachlorobiphenyl		
749	3-OH-PCB 168	6.02	7.16
750	2'-OH-PCB 168	7.6	7.09
	3,3',4,4',5,5'-Hexachlorobiphenyl		
751	2-OH-PCB 169	7.29	7.11
	2,2',3,3',4,4',5-Heptachlorobiphenyl		
752	6-OH-PCB 170	6.67	7.55
753	5'-OH-PCB 170	6.58	7.65
754	6'-OH-PCB 170	8.04	7.62
	2,2',3,3',4,4',6-Heptachlorobiphenyl		
755	5-OH-PCB 171	5.12	7.6
756	5'-OH-PCB 171	6.54	7.65
757	6'-OH-PCB 171	8.19	7.63
	2,2',3,3',4,5,5'-Heptachlorobiphenyl		
758	6-OH-PCB 172	6.66	7.57
759	4'-OH-PCB 172	5.44 4.73 ²	7.65 6.55 ²
760	6'-OH-PCB 172	7.41	7.62
	2,2',3,3',4,5,6-Heptachlorobiphenyl		
761	4'-OH-PCB 173	6.95	7.68
762	5'-OH-PCB 173	8.05	7.67
763	6'-OH-PCB 173	8.94	7.6
	2,2',3,3',4,5,6'-Heptachlorobiphenyl		
764	6-OH-PCB 174	6.83	7.49
765	4'-OH-PCB 174	6.17	7.68

ID ^I	Compound ^I	pKa ^I	log p ^I
766	5'-OH-PCB 174	6.61	7.65
2,2	3,3',4,5',6-Heptachlorobiphenyl		
767	5-OH-PCB 175	5.11	7.6
768	4'-OH-PCB 175	5.45	7.66
769	6'-OH-PCB 175	7.58	7.62
2,2	3,3',4,6,6'-Heptachlorobiphenyl		
770	5-OH-PCB 176	5.08	7.56
771	4'-OH-PCB 176	6.18	7.67
772	5'-OH-PCB 176	6.57	7.66
2,2	3,3',4,5',6'-Heptachlorobiphenyl		
773	5-OH-PCB 177	6.53	7.65
774	6-OH-PCB 177	8.18	7.63
775	4'-OH-PCB 177	4.68	7.62
2,2	3,3',5,5',6-Heptachlorobiphenyl		
776	4-OH-PCB 178	4.67	7.64
777	4'-OH-PCB 178	5.44	7.66
778	6'-OH-PCB 178	7.57	7.61
2,2	3,3',5,6,6'-Heptachlorobiphenyl		
779	4-OH-PCB 179	4.68	7.6
780	4'-OH-PCB 179	6.17	7.67
781	5'-OH-PCB 179	6.56	7.65
2,2	3,4,4',5,5'-Heptachlorobiphenyl		
782	6-OH-PCB 180	6.67	7.61
783	3'-OH-PCB 180	5.69	7.64
784	6'-OH-PCB 180	7.18	7.61
2,2	3,4,4',5,6-Heptachlorobiphenyl		
785	3'-OH-PCB 181	6.45	7.64
786	5'-OH-PCB 181	7.34	7.65
787	6'-OH-PCB 181	8.7	7.63
2,2	3,4,4',5,6'-Heptachlorobiphenyl		
788	6-OH-PCB 182	6.84	7.56
789	3'-OH-PCB 182	5.92	7.65
2,2	3,4,4',5',6-Heptachlorobiphenyl		
790	5-OH-PCB 183	5.12	7.64
791	3'-OH-PCB 183	5.65	7.64
792	6'-OH-PCB 183	7.34	7.61
2,2	3,4,4',6,6'-Heptachlorobiphenyl		
793	5-OH-PCB 184	5.09	7.6
794	3'-OH-PCB 184	5.88	7.65

ID ^I	Compound ^I	pKa ^I	log p ^I
2,2',3,4,5,5',6-Heptachlorobiphenyl			
795	3'-OH-PCB 185	7.13	7.66
796	4'-OH-PCB 185	6.94	7.69
797	6'-OH-PCB 185	8.08	7.49
2,2',3,4,5,6,6'-Heptachlorobiphenyl			
798	3'-OH-PCB 186	7.37	7.66
799	4'-OH-PCB 186	7.69	7.69
2,2',3,4',5,5',6-Heptachlorobiphenyl			
800	4-OH-PCB 187	4.68 4.08 ²	7.65 6.77 ²
801	3'-OH-PCB 187	5.64	7.64
802	6'-OH-PCB 187	7.33	7.61
2,2',3,4',5,6,6'-Heptachlorobiphenyl			
803	4-OH-PCB 188	4.69	7.64
804	3'-OH-PCB 188	5.87	7.65
2,3,3',4,4',5,5'-Heptachlorobiphenyl			
805	6-OH-PCB 189	6.47	7.6
806	2'-OH-PCB 189	7.33	7.6
2,3,3',4,4',5,6-Heptachlorobiphenyl			
807	2'-OH-PCB 190	7.99	7.59
808	5'-OH-PCB 190	7.1	7.65
809	6'-OH-PCB 190	8.84	7.63
2,3,3',4,4',5',6-Heptachlorobiphenyl			
810	5-OH-PCB 191	5.16	7.62
811	2'-OH-PCB 191	7.48	7.61
2,3,3',4,5,5',6-Heptachlorobiphenyl			
812	2'-OH-PCB 192	8.22	7.61
813	4'-OH-PCB 192	6.22	7.66
2,3,3',4',5,5',6-Heptachlorobiphenyl			
814	4-OH-PCB 193	4.67	7.64
815	2'-OH-PCB 193	7.47	7.61
2,2',3,3',4,4',5,5'-Octachlorobiphenyl			
816	6-OH-PCB 194	6.56	8.04
2,2',3,3',4,4',5,6-Octachlorobiphenyl			
817	5'-OH CB 195	6.48	8.14
818	6'-OH CB 195	8.08	8.09
2,2',3,3',4,4',5,6'-Octachlorobiphenyl			
819	6-OH-PCB 196	6.72	7.97
820	5'-OH-PCB 196	5.06	8.14

ID ¹	Compound ¹	pKa ¹	log p ¹
	2,2',3,3',4,4',6,6'-Octachlorobiphenyl		
821	5-OH-PCB 197	5.03	8.1
	2,2',3,3',4,5,5',6-Octachlorobiphenyl		
822	4'-OH-PCB 198	5.4	8.14
823	6'-OH-PCB 198	7.46	7.98
	2,2',3,3',4,5,5',6'-Octachlorobiphenyl		
824	6-OH-PCB 199	6.71	7.97
825	4'-OH-PCB 199	4.63	8.15
	2,2',3,3',4,5,6,6'-Octachlorobiphenyl		
826	4'-OH-PCB 200	6.13	8.14
827	5'-OH-PCB 200	6.51	8.16
	2,2',3,3',4,5',6,6'-Octachlorobiphenyl		
828	5-OH-PCB 201	5.02	8.02
829	4'-OH-PCB 201	4.64	8.15
	2,2',3,3',5,5',6,6'-Octachlorobiphenyl		
830	4-OH-PCB 202	4.63	8.12
	2,2',3,4,4',5,5',6-Octachlorobiphenyl		
831	3'-OH-PCB 203	5.59	8.14
832	6'-OH-PCB 203	7.23	7.98
	2,2',3,4,4',5,6,6'-Octachlorobiphenyl		
833	3'-OH-PCB 204	5.82	8.14
	2,3,3',4,4',5,5',6-Octachlorobiphenyl		
834	2'-OH-PCB 205	7.37	8.06
	2,2',3,3',4,4',5,5',6-Nonachlorobiphenyl		
835	6'-OH-PCB 206	6.61	8.19
	2,2',3,3',4,4',5,6,6'-Nonachlorobiphenyl		
836	5'-OH-PCB 207	4.97	8.23
	2,2',3,3',4,5,5',6,6'-Nonachlorobiphenyl		
837	4'-OH-PCB 208	4.59	8.25

¹ Taken from Rayne and Forest (2010) unless otherwise denoted.

² Taken from (Tampal et al., 2002).

³ Calculated from Advanced Chemistry Development (ACD) I-Lab Web Service using ACD/pKa 8.03 or ACD/LogP 8.02 as appropriate.

⁴ Calculated LogP values from Bradbury et al. (1996).

⁵ Parent congeneric PCB nomenclature consistent with Ballschmiter (1980) and EPA on-line list. See Mills et al. (2007) for a comparison of congener nomenclatures.

Table 3

Polychlorinated biphenyls (OH-PCBs) identified in human plasma and their suggested parent compound. Parent compounds in bold as determined by Sjödin and coworkers (Soechitram et al., 2004).

Abbreviation	Structure	Parentcompound	
		via epoxide or direct insertion	via 1,2-shift
5'-OH-PCB 66	3-OH-2',4,4',5-tetraCB	PCB 66	PCB 68
4'-OH-PCB 79	4-OH-3,3',4',5-tetraCB	PCB 79	PCB 77
3'-OH-PCB 85	3-OH-2,2',3',4,4'-pentaCB	PCB 85	PCB 105 PCB 82
4'-OH-PCB 120	4-OH-2',3,4',5,5'-pentaCB	PCB 120	PCB 118
4'-OH-PCB 101	4-OH-2,2',4',5,5'-pentaCB	PCB 101	PCB 99
3-OH-PCB 118	3-OH-2,3',4,4',5-pentaCB	PCB 118	PCB 107 PCB 126
4-OH-PCB 107	4-OH-2,3,3',4',5-pentaCB	PCB 107	PCB 105, PCB 118
4'-OH-PCB 108	4-OH-2',3,3',4',5-pentaCB	PCB 108	PCB 105
4-OH-PCB 134	4-OH-2,2',3,3',5,6-hexaCB	PCB 134	PCB 131
4'-OH-PCB 97	4-OH-2,2',3,4',5'-pentaCB	PCB 97	PCB 99
3'-OH-PCB 184	3-OH-2,2',3',4,4',6,6'-heptaCB	PCB 184	PCB 176
3-OH-PCB 153	3-OH-2,2',4,4',5,5'-hexaCB	PCB 153	PCB 146 PCB 167
4-OH-PCB 146	4-OH-2,2',3,4',5,5'-hexaCB	PCB 146	PCB 138, PCB 153
3',4-diOHCB 90	3',4-diOH-2,2',3,4',5-pentaCB		
2',4-diOH-PCB 107	2',4-diOH-2,3,3',4',5-pentaCB		PCB 105 PCB 118
4'-OH-PCB 127	4-OH-3,3',4',5,5'-pentaCB	PCB 127	PCB 126
3'-OH-PCB 138	3-OH-2,2',3',4,4',5-hexaCB	PCB 138	PCB 130 PCB 157
4'-OH-PCB 130	4-OH-2,2',3,3',4',5-hexaCB	PCB 130	PCB 138 PCB 128
4-OH-PCB 163	4-OH-2,3,3',4',5,6-hexaCB	PCB 163	PCB 158
4-OH-PCB 178	4-OH-2,2',3,3',5,5',6-heptaCB	PCB 178	PCB 175
3'-OH-PCB 182	3-OH-2,2',3',4,4',5',6-heptaCB	PCB 182	PCB 174 PCB 180
3'-OH-PCB 183	3-OH-2,2',3',4,4',5,6'-heptaCB	PCB 183	PCB175 PCB 191
4'-OH-PCB 175	4-OH-2,2',3,3',4',5,6'-heptaCB	PCB 175	PCB 183 PCB 171
4-OH-PCB 187	4-OH-2,2',3,4',5,5',6-heptaCB	PCB 187	PCB 183
4'-OH-PCB 159	4-OH-2',3,3',4',5,5'-hexaCB	PCB 159	PCB 156
4-OH-PCB 162	4-OH-2,3,3',4',5,5'-hexaCB	PCB 162	PCB 157
4-OH-PCB 177	4-OH-2,2',3,3',4',5,6-heptaCB	PCB 177	PCB 171
4-OH-PCB 202	4-OH-2,2',3,3',5,5',6,6'-octaCB	PCB 202	PCB 201
3'-OH-PCB 180	3-OH-2,2',3',4,4',5,5'-heptaCB	PCB 180	PCB 172 PCB 189
4'-OH-PCB 172	4-OH-2,2',3,3',4',5,5'-heptaCB	PCB 172	PCB 170 PCB 180
4-OH-PCB 193	4-OH-2,3,3',4',5,5',6-heptaCB	PCB 193	PCB 191
4,3'-diOH-PCB 187	4,3'-diOH-2,2',3,4',5,5',6-heptaCB	PCB 187	PCB 183
4,4'-diOH-PCB 178	4,4'-diOH-2,2',3,3',5,5',6-heptaCB		PCB 183
3'-OH-PCB 203	3-OH-2,2',3',4,4',5,5',6'-octaCB	PCB 203	PCB 198 PCB 205

Abbreviation	Structure	Parentcompound	
		via epoxide or direct insertion	via 1,2-shift
4'-OH-PCB 198	4-OH-2,2',3,3',4',5,5',6'-octaCB	PCB 198	PCB 195 PCB 203
4'-OH-PCB 199	4-OH-2,2',3,3',4',5,5',6'-octaCB	PCB 199	PCB 196
4,4'-diOH-PCB 202	4,4'-diOH-2,2',3,3',5,5',6'-octaCB		
4'-OH-PCB 208	4-OH-2,2',3,3',4',5,5',6,6'-nonaCB	PCB 208	PCB 207

Author Manuscript

Author Manuscript

Author Manuscript

Author Manuscript

Table 4

Median and range (min-max) concentrations (ng/g lipid weight) of OH-PCBs and CB-153 in plasma from humans at different geographical areas.

Country	Year ^d	N	4-OH-PCB 107	3'-OH-PCB 138	4-OH-PCB 146	3-OH-PCB 153	4-OH-PCB 187	PCB-153
Slovakia <i>R1</i>								
Stropkov/Svidnik	2001	175	26 (0.95–84)	26 (6.5–92)	58 (15–260)	240 (51–1200)		
Micahlovec		122	62 (3.7–710)	73 (6.7–940)	130 (11–1500)	570 (100–7200)		
The Netherlands <i>R2</i>								
	1998/00	51	10 (0.8–38)	7 (1.3–26)	10 (3–27)	5 (1.4–13)	20 (7–49)	100 (43–290)
Faroe Islands <i>R3</i>								
	1994/95	51	14 (4–350)	18 (6–50)	23 (8–58)	14 (3–38)	38 (17–69)	115 (25–250)
	1994/95	57	54 (5.7–300)	33 (7.5–260)	88 (22–360)	31 (5.8–230)	150 (39–600)	430 (99–1500)
	2000/01	42	130 (8.5–580)	31 (2.0–130)	68 (4.9–480)	23 (1.9–100)	99 (12–470)	310 (25–1300)
Sweden <i>R4</i>								
	1995	16	36 (0–59)		41 (4–72)	50 (13–73)		
		16	81 (28–480)		93 (43–270)	290 (150–460)		
Sweden <i>R5</i>								
	1991	20	36 (15–110)	20 (7.5–45)	39 (12–140)	15 (5.1–31)	74 (54–110)	220 (120–390)
		12	58 (27–290)	28 (14–68)	66 (43–290)	20 (11–51)	68 (57–280)	450 (280–1000)
Latvia <i>R5</i>								
	1993	19	82 (31–150)	18 (9.1–33)	31 (17–62)	12 (7.3–28)	34 (23–61)	160 (100–230)
		26	290 (87–770)	74 (29–230)	160 (57–540)	57 (16–280)	120 (66–430)	920 (320–1700)
The Netherlands <i>R6</i>								
	2001/02	90	3.5 (n.d.–20)	14 (5.7–110)	11 (4.7–74)	63 (19–230)		
		9	7.7 (n.d.–23)	41 (13–180)	30 (13–87)	62 (26–110)		
Sweden <i>R7</i>								
		6	52 (27–130)	26 (9.9–59)	40 (30–84)	95 (41–160)		
		8	14 (9.3–31)	14 (5.3–22)	26 (14–39)	64 (32–75)		
Nicaragua <i>R8</i>								
	2002	4			2 (1–5)	14 (6–61)		
		4			16 (9–27)	100 (49–220)		
Canada <i>e, f, R9</i>								
Nuнавик								
	1993/96	10	6 (1.5–22)	5 (1.5–18)	18 (2–67)	9.5 (2–32)	24 (6.5–78)	131 (24–670)
Lower N. Shore								
		10	24 (3–84)	11 (4.5–46)	40 (8–250)	12 (5–37)	48 (27–130)	215 (54–680)
Southern Quebec								
		10	5.5 (1.5–22)	2.5 (1.5–8)	6 (2–29)	3 (1.5–7)	14 (5–48)	52 (15–100)
Sweden ^{e, R10}								
	2000/01	15	5 (2–15)	4.5 (1–27)	14 (6–61)	3.5 (0.5–18)	24 (12–49)	56 (27–200)
		15	2.5 (<0.05–5.5)	4.5 (1–28)	10 (4–27)	2.5 (0.5–16)	12 (6.5–22)	44 (20–110)

Country	Year ^a	N	4-OH-PCB	3-OH-L-PCB	4-OH-PCB	3-OH-L-PCB	4-OH-PCB	3-OH-L-PCB	4-OH-PCB	PCB-153
Russian Arctic ^g , <i>R11</i>	2003	15	280 (200–920)	170 (92–340)	190 (87–350)	190 (100–490)	146	138	187	140 (74–460)

n.d. = not detected;

^a) Sampling year

^b) The groups are divided according to PCB153 concentrations, low and high, respectively.

^c) Divided in groups according to consumption of fatty fish: low = 0–1 meals/month and high > 12 fish meals/month, 10th–90th percentile range

^d) Pregnancy time, early: 9–13 weeks and late: 31–36 weeks

^e) Data has been normalized for lipid content at 0.2% according to (Soechitram et al., 2004) and (Gruvenius et al., 2003)

^f) Geometric mean

^g) Samples chosen with high PCB value.

References :

R1 : (Fångström et al., 2005b);

R2 : (Hagmar et al., 2001);

R3 : (Sjodin et al., 2000);

R4 : (Meijer et al., 2004);

R5 : (Larsdotter et al., 2005);

R6 : (Cuadra et al., 2006);

R7 : (Sandau et al., 2002);

R8 : (Gruvenius et al., 2003);

R9 : (Sandanger et al., 2004);

R10 : (Hovander et al., 2006);

R11 : (Hofvander, 2006).

Table 5

PCB methyl sulfone (MeSO₂-PCBs) identified in humans and their parent compounds are indicated. Chiral MeSO₂-PCBs are marked with an x in the table.

Structure	Abbreviation	Chiral MeSO ₂ -CBs	Parent compound
3-MeSO ₂ -2,2',4',5-tetraCB	3'-MeSO ₂ -PCB 49		PCB 49
4-MeSO ₂ -2,2',4',5-tetraCB	4'-MeSO ₂ -PCB 49		PCB 49
3-MeSO ₂ -2,2',5,5'-tetraCB	3-MeSO ₂ -PCB 52		PCB 52
4-MeSO ₂ -2,2',5,5'-tetraCB	4-MeSO ₂ -PCB 52		PCB 52
3-MeSO ₂ -2,4',5,6-tetraCB	5-MeSO ₂ -PCB 64		PCB 64
4-MeSO ₂ -2,3,4',6-tetraCB	4-MeSO ₂ -PCB 64		PCB 64
3-MeSO ₂ -2,3',4',5-tetraCB	3-MeSO ₂ -PCB 70		PCB 70
4-MeSO ₂ -2,3',4',5-tetraCB	4-MeSO ₂ -PCB 70		PCB 70
3-MeSO ₂ -2,2',3',4',5-pentaCB	3'-MeSO ₂ -PCB 87		PCB 87
4-MeSO ₂ -2,2',3',4',5-pentaCB	4'-MeSO ₂ -PCB 87		PCB 87
3-MeSO ₂ -2,2',4',5,6-pentaCB	5-MeSO ₂ -PCB 91	x	PCB 91
4-MeSO ₂ -2,2',3,4',6-pentaCB	4-MeSO ₂ -PCB 91	x	PCB 91
3-MeSO ₂ -2,2',4',5,5'-pentaCB	3'-MeSO ₂ -PCB 101		PCB 101
4-MeSO ₂ -2,2',4',5,5'-pentaCB	4'-MeSO ₂ -PCB 101		PCB 101
3-MeSO ₂ -2,2',3',4',5,6-hexaCB	5'-MeSO ₂ -PCB 132	x	PCB 132
4-MeSO ₂ -2,2',3,3',4',6-hexaCB	4'-MeSO ₂ -PCB 132	x	PCB 132
3-MeSO ₂ -2,2',3',4',5,5'-hexaCB	3'-MeSO ₂ -PCB 141		PCB 141
4-MeSO ₂ -2,2',3',4',5,5'-hexaCB	4'-MeSO ₂ -PCB 141		PCB 141
3-MeSO ₂ -2,2',4',5,5',6-hexaCB	3-MeSO ₂ -PCB 149	x	PCB 149
4-MeSO ₂ -2,2',3,4',5',6-hexaCB	4-MeSO ₂ -PCB 149	x	PCB 149
3-MeSO ₂ -2,2',3',4',5,5',6-heptaCB	5'-MeSO ₂ -PCB 174	x	PCB 174
4-MeSO ₂ -2,2',3,3',4',5',6-heptaCB	4'-MeSO ₂ -PCB 174	x	PCB 174

Table 6

Median and range (min-max) concentrations (ng/g lipid weight) of methylsulfonyl-PCBs in humans from four European countries and from Canada. References to the scientific reports are given in the table.

Country	year ^d	n	4'-MeSO ₂ -PCB 87	4'-MeSO ₂ -PCB 101	4-MeSO ₂ -PCB 149	ΣMeSO ₂ -PCB	PCB-153	No. cong. ^b
Plasma								
Slovakia ^{R1}	2001							
Stropkov/Svidnik		175	0.24 (LOQ-1.4)	0.22 (LOQ-1.4)	1.0 ^c (0.19-9.2)	1.5 (0.34-11)	240 (51-1200)	3
Micahlovec		122	0.85 (LOQ-22)	0.75 (LOQ-9.9)	2.7 ^c (0.25-72)	4.2 (0.41-100)	570 (100-7200)	3
Faroe Islands ^{R12}								
maternal	1994/95	10	5.1 (1.9-15)	4.5 (1.6-11)	7.5 ^c (2.4-25)	73 (24-189)	730 (490-1500)	11
children	2000/01	10	4.0 (1.7-6.5)	3.6 (1.3-6.6)	8.6 ^c (3.4-17)	70 (23-103)	820 (620-1300)	11
Sweden ^{R12}	2002	9	0.88 (0.14-2.4)	0.50 (0.14-1.5)	1.2 ^c (0.28-4.6)	5.6 (0.90-17)	290 (140-800)	11
Sweden ^{R13}	1997 ^d	11				2.0 (0.81-5.6)	210 (110-380)	17
Milk								
Sweden ^{R14}	1972	p ^e /75	2.1	0.78	2.0	9.2	210	23
Sweden ^{R14}	1984/85	p ^e /10 ²	0.70	0.32	0.64	3.1	100	23
Sweden ^{R14}	1992	p ^e /20	0.33	0.13	0.35	1.6	96	23
Canada ^{R15}	1992	50	0.33 ^f	0.087 ^f		0.81 ^f		12
Adipose tissue								
Sweden ^{R16}	1997 ^d	7	0.82 (0.48-3.6)	0.40 (0.21-2.6)	0.71 (0.41-5.1)	2.9 (2.0-9.0)	280 (140-590)	24
Sweden ^{R17}	1994	5	1 (0.2-2)	0.3 (0.2-1)	1 (0.4-2)	6 (2)	240 (140-740)	24
Belgium ^{R18}	2002	11	0.33 ^g (0.13-0.98)	0.27 ^g (0.12-0.93)	0.08 ^g (n.d.-0.17)	1.57 ^g (0.33-4.3)	110 (40-250)	26
Liver								
Sweden ^{R16}	1997 ^d	7	1.2 (0.61-2.61)	0.54 (0.18-1.5)	1.7 (0.69-11)	28 (12-358)	230 (110-610)	24
Sweden ^{R17}	1994	5	1 (0.4-2)	0.3 (0.2-2)	1 (0.1-11)	34 (12-358)	200 (120-620)	24
Belgium ^{R18}	2002	11	0.35 ^g (n.d.-0.87)	0.23 ^g (n.d.-1.0)	0.10 ^g (n.d.-0.42)	9.3 ^g (1.7-27)	78 (11-240)	26

Country	year ^d	n	4'-MeSO ₂ -PCB 87	4'-MeSO ₂ -PCB 101	4-MeSO ₂ -PCB 149	ΣMeSO ₂ -PCB	PCB-153	No. cong. ^b
Belgium <i>R18</i>	2002	11	0.558 ^g (n.d.-1.1)	0.468 ^g (n.d.-1.2)	0.088 ^g (n.d.-0.89)	2.78 ^g (n.d.-12)	1208 ^g (19-800)	26

n.d. = not detected; LOQ = limit of quantification;

- a)* Sampling year;
- b)* Number of MeSO₂-PCB congeners in the ΣMeSO₂-PCB;
- c)* include concentration of the MeSO₂-hexaCB (unknown);
- d)* The publication year;
- e)* pooled sample/number of subjects in the pool;
- f)* Data has been recalculated for a lipid content of 4%;
- g)* Mean values.
- References:
- R1* : (Weistrand et al., 1997);
- R12* : (Noren et al., 1996);
- R13* : (Newsome and Davies, 1996);
- R14* : (Weistrand and Noren, 1997);
- R15* : (Weistrand and Noren, 1997);
- R16* : (Guvinius et al., 2002);
- R17* : (Chu et al., 2003b);
- R18* : (Chu et al., 2003b).