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Oxidation precursor dependence of atomic layer deposited Al_2O_3 films in a-Si:H(i)/ Al_2O_3 surface passivation stacks

Yuren Xiang, Chunlan Zhou*, Endong Jia and Wenjing Wang

Abstract

In order to obtain a good passivation of a silicon surface, more and more stack passivation schemes have been used in high-efficiency silicon solar cell fabrication. In this work, we prepared a-Si:H(i)/ Al_2O_3 stacks on KOH solution-polished n-type solar grade mono-silicon(100) wafers. For the Al_2O_3 film deposition, both thermal atomic layer deposition (T-ALD) and plasma enhanced atomic layer deposition (PE-ALD) were used. Interface trap density spectra were obtained for Si passivation with a-Si films and a-Si:H(i)/ Al_2O_3 stacks by a non-contact corona C-V technique. After the fabrication of a-Si:H(i)/ Al_2O_3 stacks, the minimum interface trap density was reduced from original 3×10^{12} to $1 \times 10^{12} \text{ cm}^{-2} \text{ eV}^{-1}$, the surface total charge density increased by nearly one order of magnitude for PE-ALD samples and about $0.4 \times 10^{12} \text{ cm}^{-2}$ for a T-ALD sample, and the carrier lifetimes increased by a factor of three (from about 10 μs to about 30 μs). Combining these results with an X-ray photoelectron spectroscopy analysis, we discussed the influence of an oxidation precursor for ALD Al_2O_3 deposition on Al_2O_3 single layers and a-Si:H(i)/ Al_2O_3 stack surface passivation from field-effect passivation and chemical passivation perspectives. In addition, the influence of the stack fabrication process on the a-Si film structure was also discussed in this study.

Keywords: Atomic layer deposition; Al_2O_3 ; A-Si:H(i)/ Al_2O_3 stack; Interface trap density

Background

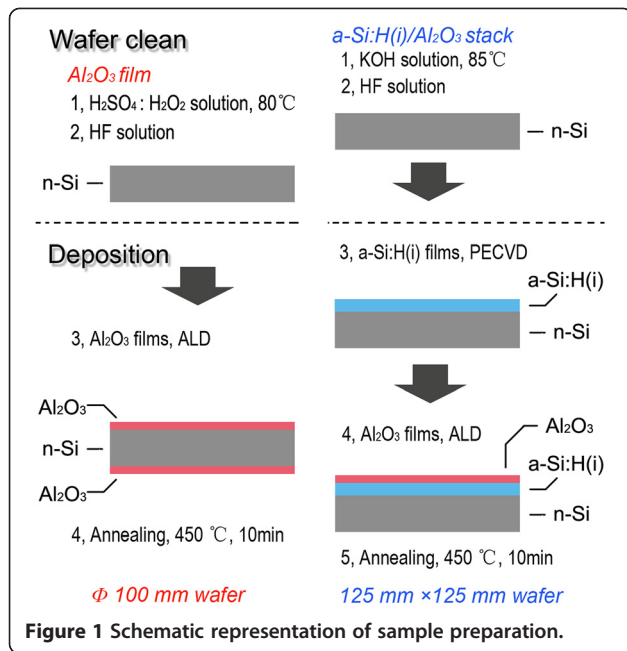
An excellent interface passivation has been considered as the key point for high-efficiency solar cells such as passivated emitter and rear cell (PERC), heterojunction with intrinsic thin layer (HIT) or interdigitated back contact (IBC) device structures. The properties of aluminum oxide (Al_2O_3) films and hydrogenated amorphous silicon (a-Si:H) films have been widely investigated for solar cell fabrication. Both of them have shown excellent performances, such as remarkable passivation behavior on both n- and p-type Si surfaces and the cost-saving deposition using atomic layer deposition (ALD) and plasma-enhanced vapor chemical deposition (PECVD) at low temperatures, respectively [1]. The effective passivation of Al_2O_3 film is related to the field-effect passivation associated with the fixed negative charges (Q_f , about $3\text{--}10 \times 10^{12} \text{ cm}^{-2}$) and chemical passivation associated with low interface trap

density ($D_{it} \leq 10^{11} \text{ eV}^{-1} \text{ cm}^{-2}$) [2]. The passivation effect of a-Si:H films with low D_{it} is generally attributed to the saturation of dangling bonds on Si surface by hydrogen [3].

Various passivation stack schemes, such as $\text{SiO}_2/\text{Al}_2\text{O}_3$, $\text{Al}_2\text{O}_3/\text{a-SiN}_x\text{:H}$, a-Si:H/ SiN_x , etc., have already been investigated to improve the passivation effect, giving consideration to both low temperature deposition process and stability of thermal and ultraviolet (UV) radiation in the photovoltaic field [4-7]. In $\text{SiO}_2/\text{Al}_2\text{O}_3$ stacks, Al_2O_3 films play the role of a capping layer to improve the passivation effect [4]. The low D_{it} values obtained through this passivation scheme were explained by an effective hydrogenation of defects present at the buried Si/ SiO_2 interface under the influence of the Al_2O_3 capping layer [4,8]. In $\text{Al}_2\text{O}_3/\text{a-SiN}_x\text{:H}$ and $\text{Al}_2\text{O}_3/\text{SiN}_x$ stacks, Al_2O_3 films have been applied as an interface layer due to the excellent surface passivation and parasitic shunting free on p-type silicon, e.g., PERC [5,9-11]. A-Si:H/ SiO_x [12], a-Si:H/ SiN_x [13], and a-Si:H/ $\text{SiN}_x\text{:H}$ [7] stacks have been studied as stable alternatives to thermally grown SiO_2 because of the low a-Si:H deposition temperature. Those schemes use a-

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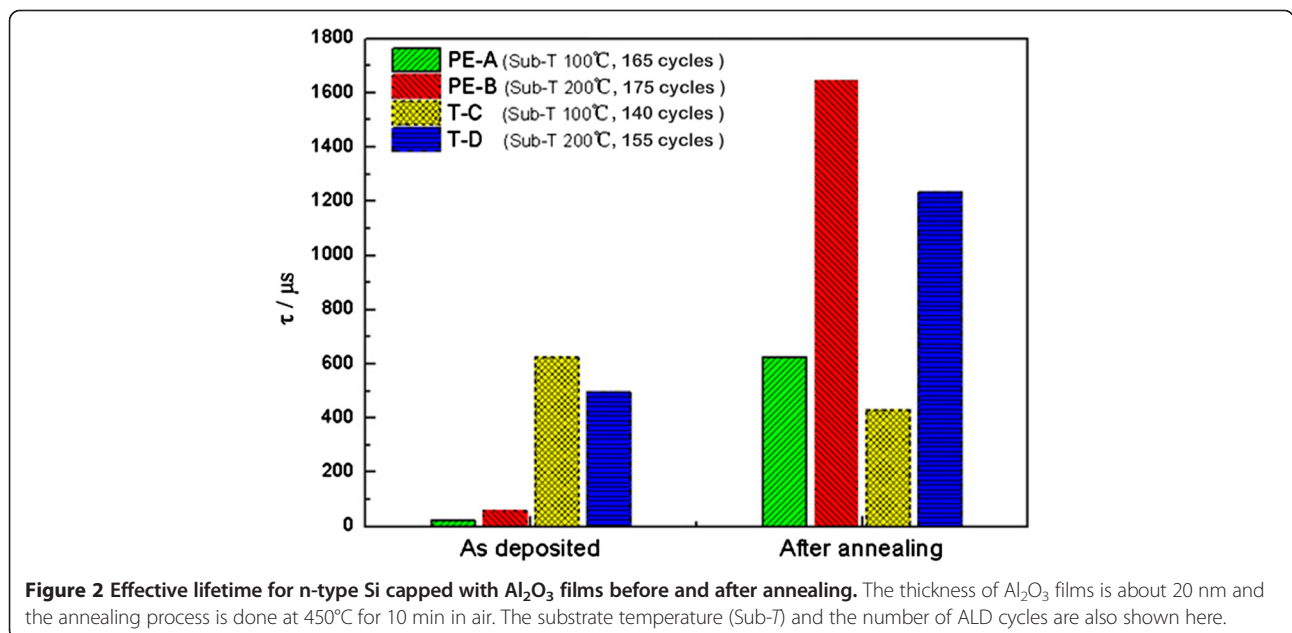
Si:H films as interface layers to ensure the relatively high quality passivation and eliminate the parasitic shunting in case of directly deposited SiN_x or SiN_x:H films on p-type silicon surface with parasitic shunting [7,14].

In this paper, a-Si:H(i)/Al₂O₃ passivation stacks for silicon were chosen as a model system for studying the passivation effect of the ALD Al₂O₃ films. It is surmised that the a-Si:H(i) films act as a hydrogen reservoir to improve the interfacial chemical passivation during the a-Si:H(i)/Al₂O₃ stack annealing. For the stack preparation, a-Si:H(i)

films were deposited on just one side of silicon wafer by PECVD; after annealing of the wafers, ALD Al₂O₃ films were deposited on top of the a-Si:H layers. For reference purposes, the passivation of Al₂O₃ single layers on high-quality n-type Si wafers (without an intermediate a-Si:H layer) was also shown in this study. Combining the passivation testing results with an X-ray photoelectron spectroscopy (XPS) analysis, we discussed the influence of oxidation precursor for ALD Al₂O₃ deposition on Al₂O₃ single layer and a-Si:H(i)/Al₂O₃ stack surface passivation from field passivation and chemical passivation perspectives.

Methods

For the a-Si/Al₂O₃ stack passivation samples, 125 mm × 125 mm 0.9-Ω · cm phosphorus-doped solar grade monocrystalline silicon(100) wafers (Jinglong Industry and Commerce Group Co. Ltd., Xingtai, Hebei Province, China) were used as substrates. The wafers were polished in KOH solution (Arkonic Gases & Chemicals Inc., Wuhu, Anhui Province, China) to a thickness of 160 μm and cleansed by a hydrogen fluoride (HF) solution (Arkonic Gases & Chemicals Inc., Wuhu, Anhui Province, China) to remove the native oxide layer. a-Si:H(i) films with thicknesses of 80 and 170 nm were deposited on just one side of the Si wafers by PECVD at 160°C using hydrogen (H₂) and silane (SiH₄) as precursor gases. Those films were annealed at 250°C for 10 min in air. For the reference Al₂O₃ single layer passivation samples, Φ100-mm polished n-type high quality silicon wafers were used as substrates. All the wafers were cleaned by H₂SO₄:H₂O₂ solution (4:1 vol) at 80°C. Before the Al₂O₃ film deposition, the native oxide layer present on the wafer surfaces was also removed using an HF solution. For the Al₂O₃ film deposition, both



plasma-enhanced atomic layer deposition (PE-ALD) and thermal atomic layer deposition (T-ALD) were used. The ALD substrate temperatures were 200°C for the a-Si/Al₂O₃ stacks, while 100°C and 200°C for Al₂O₃ single layers. Al (CH₃)₃ (trimethylaluminum, TMA; Jiangsu Nata Opto-electronic Material Co., Ltd., Suzhou, Jiangsu Province, China) served as Al precursor, and either remote O plasma (for PE-ALD) or H₂O (for T-ALD) was used as O precursor. After Al₂O₃ layer deposition, all the samples (both the Al₂O₃ single layers and the a-Si/Al₂O₃ stacks) were annealed at 450°C for 10 min in air. In addition, only the Al₂O₃ single layer passivation samples were covered with Al₂O₃ films (Jiangsu Nata Opto-electronic Material Co.,

Ltd., Suzhou, Jiangsu Province, China) on both sides of Si wafers. The sample preparation process was shown in Figure 1. The thickness of films was measured by a step profiler. The composition of Al₂O₃ films was measured by XPS (Beijing Synchrotron Radiation Facility, Beijing, China) before and after annealing. The interface trap density spectra were obtained from a non-contact corona C-V technique for a-Si single films and a-Si:H(i)/Al₂O₃ stacks (Institute of Electrical Engineering, CAS, Beijing, China) coating Si surface.

In this study, the X-ray photoelectron spectroscopy used 700-eV synchrotron-based light as X-ray radiation source and the adventitious C 1 s signal (at 284.6 eV) to

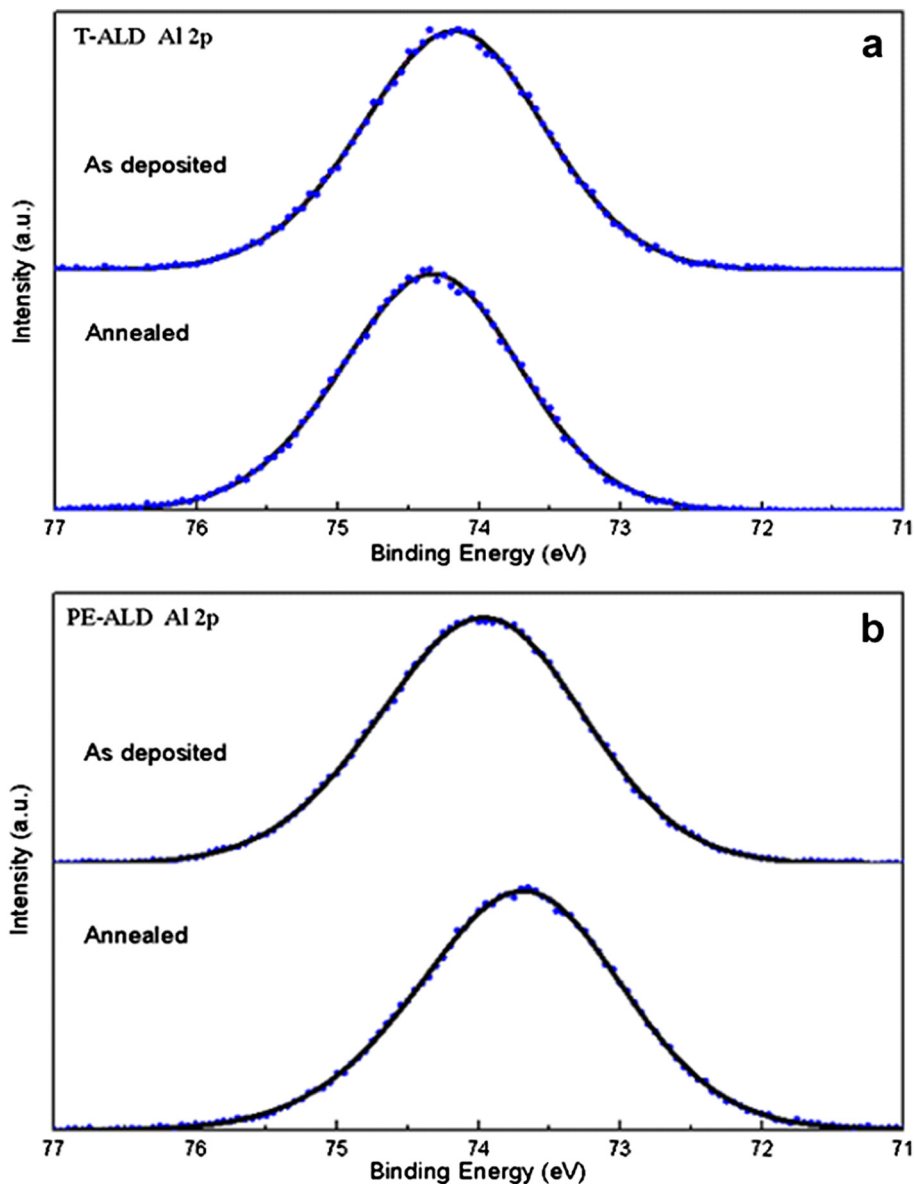


Figure 3 Al 2p XPS spectra of Al₂O₃ films. The films were deposited by T-ALD (a) and by PE-ALD (b). The solid black lines represent the fitted curves.

calibrate the peak positions. Al 2p and O 1s core-level spectra were measured in order to investigate the chemical state of the Al_2O_3 films.

The non-contact corona C-V technology is based on charge biasing of the dielectric with a precise charge dose, ΔQ_C , deposited by a corona discharge in air. The corresponding surface voltage change including the change in surface barrier, V_{SB} , is measured with a vibrating Kelvin probe. This technology monitors the interface trap density (D_{it}) spectra, D_{it} vs. V_{SB} position in the Si energy gap, and the total charge density by means of the multi-metrology Semilab PV-2000 platform (Semilab Co. Ltd., Shanghai, China) [15]. The effective minority carrier lifetime was measured by the Semilab WT2000 set-up (Semilab Co. Ltd., Shanghai, China).

Results and discussion

The difference in the passivation between PE-ALD and T-ALD deposition

For the reference samples ($\Phi 100$ mm) covered by Al_2O_3 films on both sides, the effective minority carrier lifetimes before and after annealing are shown in Figure 2. The T-ALD samples have higher effective minority carrier lifetimes just before deposition, but the advantage is lost after annealing contrasted with PE-ALD samples, especially for the Al_2O_3 deposition at the low substrate temperature of 100°C . The samples with the high substrate temperature of 200°C for both deposition types show higher effective minority carrier lifetimes after annealing ($1,647 \mu\text{s}$ for PE-ALD deposition, $1,232 \mu\text{s}$ for T-ALD deposition). The reasons for the different performances in the effective minority carrier lifetime for the two ALD deposition Al_2O_3 films are considered as follows: First, D_{it} of PE-ALD Al_2O_3 film-passivated Si surfaces is higher than that of T-ALD Al_2O_3 film-passivated Si surfaces due to the presence of vacuum ultraviolet radiation in the O_2 plasma. Thus PE-ALD interfaces need an annealing step to improve performance. Second, the fixed charge densities in T-ALD Al_2O_3 films are lower than those in PE-ALD Al_2O_3 films [2]. The fixed charge density is related to field-effect passivation. The origin of the negative fixed charge in the Al_2O_3 film is considered to be related to the effect of the interfacial SiO_x film which is formed during the ALD deposition [16].

Moreover, the T-ALD Al_2O_3 films with a substrate temperature of 100°C have a passivation degradation after annealing (see Figure 2). Due to the reaction between H_2O and TMA, the Al_2O_3 film quality depends on the reaction activity of H_2O and changes according to the variation of substrate temperature [17]. A lower substrate deposition temperature leads to a lower Al_2O_3 film quality. This degradation may be caused by loosely bound H atoms escaping from the loose interfacial layer by thermal driving.

Chemical composition analysis of Al_2O_3 layer using XPS

The difference in the chemical binding state between PE-ALD and T-ALD deposition (substrate temperature of 200°C) and the effect of annealing process on the chemical structure of the Al_2O_3 films were investigated by XPS measurement to study the passivation mechanism. Figure 3a,b shows the Al 2p core level spectra of Al_2O_3 films deposited by PE-ALD and T-ALD before and after the annealing process, respectively. Peaks de-convolution are not processed for analysis in here because the Al 2p position of AlOOH or $\text{Al}(\text{OH})_3$ is similar with that of Al_2O_3 [18]. Al 2p peak positions of as-deposited films are almost 74.1 eV irrespective of ALD type. After annealing at 450°C for 10 min in air, the positions shift to 74.3 eV (T-ALD) and 73.7 eV (PE-ALD). Compared with the reported Al 2p peak position of ALD Al_2O_3 films [19,20], those position shifts may be attributed to the charge accumulation, especially for the T-ALD high energy direction shift.

Figure 4a,b show the O 1s core level spectra of Al_2O_3 films deposited by PE-ALD and T-ALD before and after annealing. Two components are de-convoluted from O 1s peak - one centered at approximately 531.8 eV corresponding to Al-OH of AlOOH or $\text{Al}(\text{OH})_3$ and the other centered at approximately 530.7 eV corresponding to O-Al-O bonds of Al_2O_3 . The fitted component curves (Al_2O_3 , red

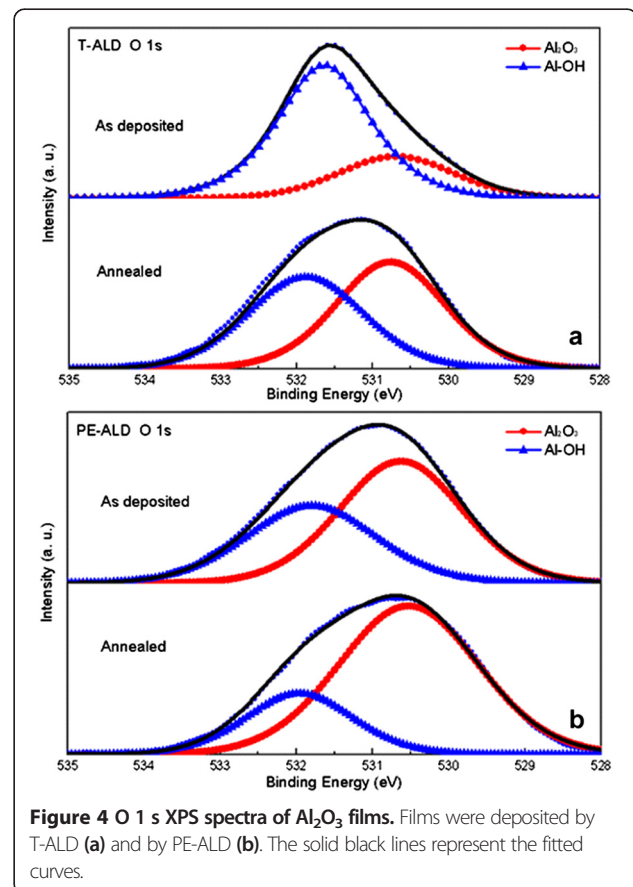


Figure 4 O 1s XPS spectra of Al_2O_3 films. Films were deposited by T-ALD (a) and by PE-ALD (b). The solid black lines represent the fitted curves.

line) in Figure 4b show that O-Al-O bonds have a higher occupation area than Al-OH bonds (Al-OH, blue line) in the PE-ALD-deposited film, and the area ratio ($A_{Al_2O_3}/A_{Al-OH}$) increases from 3:2 to 3:1 after annealing. By contrast

with PE-ALD film, the area ratio of the T-ALD-deposited film is lower. But the variation trends of PE-ALD- and T-ALD-deposited films are similar after annealing, and the area ratio of the T-ALD film also increases from 3:7 to 5:4,

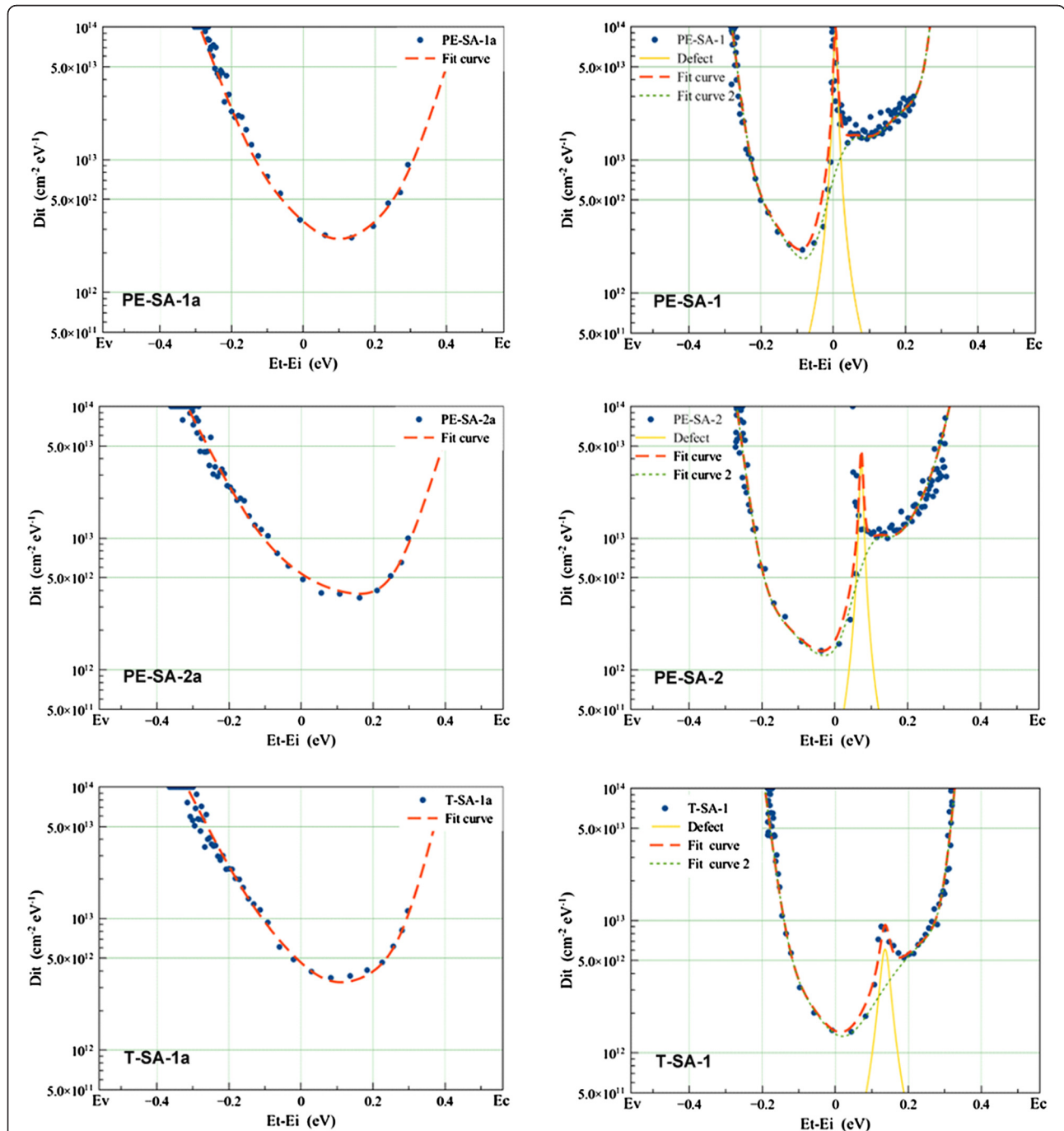


Figure 5 DIT spectra of a-Si:H (left side) and Al₂O₃/a-Si:H stack (right side). The Al₂O₃ films deposited on PE-SA-1 and PE-SA-2 by PE-ALD at 200°C. The Al₂O₃ films deposited on T-SA-1 by T-ALD at 200°C. PE-SA-1a, PE-SA-2a, and T-SA-1a are the samples only covered by the a-Si:H films and PE-SA-1, PE-SA-2, and T-SA-1 refer to the stack structure samples. The a-Si:H(i)/Al₂O₃ stack samples (the structures of samples are shown in Table 1) have been annealed at 450°C for 10 min in air. ‘Dots’ are the measured data by the non-contact corona C-V technology. ‘Fit curve’ are fitted curves for the measured data using the simplified density of an interface state model, which can be de-convoluted to ‘Defect’ and ‘Fit curve 2’.

showed in Figure 4a. The increase of the T-ALD area ratio after annealing is attributed to the residual O-H bonds breaking at high temperature. The bond breaking releases the interstitial H atoms which diffuse into the Al₂O₃ film and the Al₂O₃/Si interface [21], and saturate the dangling bonds of the silicon surface as discussed above. Simultaneously, the residual O-H bond breaking also releases Al atoms to form more O-Al-O bonds of Al₂O₃. By contrast with the PE-ALD film, the lower area ratio of the T-ALD film indicates that the reaction activity of H₂O is lower than that of plasma O and residual OH group absorbed during reaction between TMA and H₂O. In another aspect, the relatively stable area ratios for PE-ALD film before and after annealing are due to the high reaction activity of plasma O, which allows for a higher reaction ratio with TMA.

The passivation effect of a-Si:H(i)/Al₂O₃ stack on mc-Si(n)

Figure 5 shows the D_{it} spectra data of a-Si:H/Si interface with and without Al₂O₃ layer (the structures of samples are showed in Table 1). The directly measured voltage scale has been changed into energy scale referred to the silicon band edges. This conversion from the surface barrier (V_{SB}) to silicon bulk energy level is based on [22] the following: i.) the Fermi level position of the n-type silicon wafer using in this study is $(E_f - E_i) = 0.34$ eV, where E_i refers to midgap ($E_g = 1.12$ eV), and ii.) the corresponding energy scale for the D_{it} spectra is $(E_t - E_i) = ((E_f - E_i) - V_{SB})$. So, it is obvious that $(E_t - E_i) = (E_f - E_i) = 0.34$ eV in the flat-band condition ($V_{SB} = 0$). A simplified interface state model has been used to fit the directly measured data (dots in the Figure 5). This model consists of the valence and conduction band tails originating from weak Si-Si bonds and a dangling-bond distribution with two Gaussian components, which correspond to donor and acceptor state in the a-Si:H(i)/Si interface [23]. 'Fit curve' shown in Figure 5 are fitted results using the simplified interface state density model, which can be de-convoluted to the discrete energy trap ('Defect') and 'Fit curve 2'.

From the D_{it} spectra of a-Si:H(i)/Si interface, the three different samples have the same interface state density distribution and all get their lowest point ($D_{it} < 4.0 \times 10^{12} \text{ cm}^{-2} \cdot \text{eV}^{-1}$) at about 0.1 eV above midgap. The fitted curves show that the spectra have a trend to be asymmetric, and curves change slower with energy level in the side near to valence

band than the side near conduction band. After the Al₂O₃ film deposition and annealing, the curves change faster in both sides near the conduction and valence bands and have lower minimum interface trap density (Min- D_{it}) than the samples without Al₂O₃ film deposition. Those may be attributed to two reasons: first, the thermal process releases the stress of weak Si-Si bonds to decrease the density of the band tail states; second, the H diffusion shifts the weak Si-Si bonds to dangling-bonds with H saturation.

It is worth noting that the D_{it} spectra of samples deposited with Al₂O₃ film seem to have two different regions, one with a low D_{it} bottom (Min- $D_{it} \leq 1 \times 10^{12} \text{ cm}^{-2} \cdot \text{eV}^{-1}$) and another with a high D_{it} bottom (Min- $D_{it} \geq 1 \times 10^{12} \text{ cm}^{-2} \cdot \text{eV}^{-1}$). An asymmetric curve of the bulk density of states calculated by Winer [24] and Powell [25] is similar with the 'Fit curve 2' in Figure 5, and in their model, this curve is referred to as doped amorphous silicon (p type).

The discrete energy trap ('Defect' in Figure 5) has an energy level position at $E_t - E_i = 0.14$ eV for T-SA-1, at the middle of the energy gap $E_t - E_i = 0$ eV for PE-SA-1, and at $E_t - E_i = 0.07$ eV for PE-SA-2. The largest peak density of the trap in PE-ALD samples is higher than that in T-ALD sample, and the PE-ALD defect peak shapes are sharp. It is well known that the defects in the middle of the Si energy gap can be called deep level centers, which will lead to an effective minority carrier lifetime reduction.

The field-effect passivation effect as a function of the surface total charges (Q_{tot} , including the charged on the film top surface, in the film and interface) was detected by the Q-V measurement (see Table 2). In this study, Q_{tot} of each sample are negative. After Al₂O₃ film deposited, Q_{tot} increased by near one order of magnitude for PE-ALD samples and about $0.4 \times 10^{12} \text{ cm}^{-2}$ for T-ALD sample. The negative charges of a-Si:H(i)/Si originate from the electrons occupying the band gap states of donor and acceptor energy level. Those electrons diffuse from n-type silicon wafer into a-Si:H(i) films caused by the energy level mismatch [26]. The negative charge density increase for the a-Si:H(i)/Al₂O₃ stacks may be mostly benefit from the large amount of negative charges in Al₂O₃ films. The PE-ALD Al₂O₃ films show more charges than T-ALD films in Table 2. PE-ALD precursor plasma O reaction with TMA shows a better property for negative charges supply, which

Table 1 The structures of samples

	Before deposition	After deposition
PE-SA-1	80 nm a-Si:H/Si	30 nm Al ₂ O ₃ (210 cycles) + 80 nm a-Si:H/Si
PE-SA-2	80 nm a-Si:H/Si	10 nm Al ₂ O ₃ (85 cycles) + 80 nm a-Si:H/Si
T-SA-1	170 nm a-Si:H/Si	30 nm Al ₂ O ₃ (200 cycles) + 170 nm a-Si:H/Si

The numbers of ALD deposition cycles are shown behind the Al₂O₃ film thickness.

Table 2 Electrical interface parameters

	PE-SA-1	PE-SA-2	T-SA-1
Before deposition ^a Q_{tot} (cm ⁻²)	-9.23×10^{11}	-7.70×10^{11}	-9.93×10^{11}
V_{SB} (V)	-0.177	-0.134	-0.169
After deposition Q_{tot} (cm ⁻²)	-8.93×10^{12}	-5.64×10^{12}	-1.38×10^{12}
V_{SB} (V)	-0.546	-0.499	-0.192

^aThe samples before deposition refer to the samples only covered by a-Si:H films. The samples after deposition were annealed at 450°C for 10 min in air.

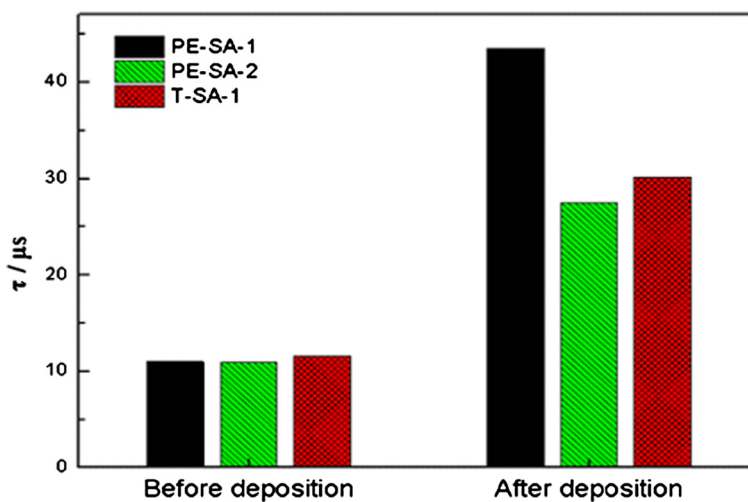


Figure 6 Effective lifetime for single-side a-Si:H capped Si before and after Al_2O_3 deposited. The Al_2O_3 films were deposited on PE-SA-1 and PE-SA-2 using PE-ALD at 200°C. The Al_2O_3 films were deposited on T-SA-1 using T-ALD at 200°C. After Al_2O_3 film deposition, the samples were annealed at 450°C for 10 min in air.

agrees with the Al_2O_3 single layer passivation data and the XPS analysis above.

Figure 6 shows that the effective minority carrier lifetimes of the a-Si:H(i)/ Al_2O_3 stack samples increase by a factor of three (from about 10 μs to about 30 μs) compared with only a-Si film passivation samples. The increases in the effective minority carrier lifetime mean that the chemical passivation or field-effect passivation improved, which are dominated by Extended-Shockley-Read-Hall recombination at the Si surface. In this case after capping with Al_2O_3 films, the reduction of average D_{it} and the increase of negative charges weaken the trap effect of defects and reduce the interface recombination to improve the effective minority carrier lifetime. From the data above, a-Si:H(i)/ Al_2O_3 stack also shows a good thermal stability at 450°C that can extend the a-Si:H(i) film application field in the solar cell fabrication.

Conclusions

In this study, we demonstrated that the a-Si:H(i)/ Al_2O_3 stack passivation layer can provide satisfactory passivation effect and thermal stability after annealing at 450°C for 10 min. The minimum interface defect density of samples with a-Si:H(i)/ Al_2O_3 stack passivation after annealing was reduced from $3 \times 10^{12} \text{ cm}^{-2} \cdot \text{eV}^{-1}$ (only with a-Si:H(i) film passivation) to $1 \times 10^{12} \text{ cm}^{-2} \cdot \text{eV}^{-1}$; Q_{tot} increased by nearly one order of magnitude for PE-ALD samples and about $0.4 \times 10^{12} \text{ cm}^{-2}$ for T-ALD sample, and the effective minority carrier lifetimes increased by a factor of three (from about 10 μs to about 30 μs) compared with only a-Si:H(i) film passivation. Combining those passivation test results with the XPS analysis, we also discussed the oxide precursor influence of direct ALD Al_2O_3 films and a-Si:H(i)/ Al_2O_3

stack on silicon passivation. The T-ALD Al_2O_3 films were low-density and hydrogen-rich films compared with PE-ALD Al_2O_3 films, leading to a low density of interface trap (excellent chemical passivation). The PE-ALD Al_2O_3 films also can provide excellent field effect passivation due to large number of negative charges. The results suggest the ALD Al_2O_3 film using H_2O as O precursor could be helpful for hydrogen transport and interface hydrogenation, while the ALD Al_2O_3 film using O plasma precursor can be beneficial for negative charge formation. Furthermore, negative charge-rich Al_2O_3 films together with optimization of the a-Si:H(i) film deposition had more possibility to improve the a-Si:H(i)/ Al_2O_3 stack passivation effect. The findings in this study can offer advantages for silicon passivation optimization of ALD Al_2O_3 films and the a-Si:H(i)/ Al_2O_3 stack and improve the thermal stability of a-Si:H(i) films during solar cell fabrication process.

Competing interests

The authors declare that they have no competing interests.

Authors' contributions

YX performed the sample preparation and characterization, analyzed the results, and wrote the manuscript. CZ and WW discussed the results and improved the manuscript. EJ carried out some sample preparation. All authors read and approved the final manuscript.

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