Concomitant Determination of Folar Nitrogen Loss, Net Carbon Dioxide Uptake, and Transpiration¹

Received for publication June 29, 1979 and in revised form September 17, 1979

R. TIMOTHY WEILAND AND CHARLES A. STUTTE

Department of Agronomy, Altheimer Laboratory, University of Arkansas, Fayetteville, Arkansas 72701

ABSTRACT

A closed system was designed for concomitant determination of net $CO₂$ uptake, transpiration, and foliar nitrogen (N) loss in soybeans (Glycine max $[L]$ Merr.). The $CO₂$ uptake was monitored by infrared analysis of system air removed periodically in aliquots. Leaf vapors were trapped in calibrated tubes at Dry Ice temperature, and transpiration rate was determined from the quantity of condensate. Subsequent pyrochemiluminescent analysis of this condensate revealed reduced N forms, although a s -all percentage (4 to 15%) of oxidized forms was found.

Maximum CO₂ uptake (22.3 milligrams per square decimeter per hour), transpiration (2.5 grams H_2O per square decimeter per hour), and the total (4.9 micrograms per square decimeter per hour), reduced (4.1 micrograms per square decimeter per hour), and oxidized (0.7 microgram per square decimeter per hour) N loss rates were measured between 6.5 and 10.5 hours (30 C) of a simulated 13-hour photoperiod in leaves of V4 to V5 (three to four trifoliolate stage) soybeans. During a temperature study (20, 25, 30, and 35 C) with these plants and several leaf positions of older vegetative soybeans (V9 to Vll), total and reduced N loss were maximum at 30 C (V4 to V5: 9.8 and 9.3 micrograms per square decimeter per hour, V9 to VII: 3.3 and 2.9 micrograms per square decimeter per hour); transpiration was maximal at 30 C in V4 to V5 plants (2.6 grams H_2O per square decimeter per hour) and at 35 C in V9 to V11 plants (2.4 grams H_2O per square decimeter per hour); and CO_2 uptake was maximal at 25 and 30 C during both sampling dates (V4 to V5: 22.7 to 23.5 milligrams CO2 per square decimeter per hour: V9 to VII: 14.1 to 14.5 milligrams $CO₂$ per square decimeter per hour). At 30 and 35 C these parameters were highest in the youngest tissue of V9 to V11 plants.

In V9 to VII plants at all four temperatures, reduced N loss was correlated with total N loss ($r = 0.99$ at 20, 25, and 30 C and 0.97 at 35 C), and $CO₂$ uptake with transpiration ($r = 0.47$ at 20 C, 0.75 at 25 C, 0.85 at 30 C, and 0.81 at 35 C). Transpiration was correlated with both total and reduced N loss at 30 C (total N: $r = 0.69$; reduced N: $r = 0.70$) and 35 C (total N: $r = 0.58$; reduced N: $r = 0.54$). In addition, CO₂ uptake was directly related to total ($r = 0.69$ at 30 C and 0.56 at 35 C) and reduced $(r = 0.67$ at 30 C and 0.56 at 35 C) N loss at these two temperatures.

Plant foliage has been found to be a site for significant losses of N (6, 8, 21, 22, 24). This loss was greater in senescing or stressed tissue (8, 24) and in younger vegetative tissue than in the older vegetative tissue or the foliage of plants in reproductive stages (22). On ^a unit area basis, weed species generally evolved more N than crop species (21). Nitrogen dissipation also seemed to be correlated directly with temperature and transpiration (21, 22), and most of the N evolved as a reduced form $(8, 24)$, although oxidized forms could be detected (24).

Data on the relationship of this N loss and $CO₂$ utilization are limited. Farquhar et al. (8) have found significant ammonia losses (about $3 \mu g/dm^2 \cdot h$) from senescing, yet photosynthetically active, leaves of maize (Zea mays L.) cultured in low N (3.4 meq/liter). Younger leaves, as well as older leaves of plants cultured in high N (35.0 meq/liter), had smaller foliar N losses but greater photosynthetic assimilation of $CO₂$. No loss could be detected with their technique from the younger leaves of maize grown with high N.

In order to characterize the relationships of these separate processes, an apparatus for concomitant measurement of net $CO₂$ uptake, transpiration, and foliar N loss from field, growth chamber, or greenhouse plants has been prepared. With this apparatus, correlated N loss, $CO₂$ uptake, and transpiration in soybean (Glycine max [L.] Merr.) leaves at several vegetative maturities, over a simulated solar day, and at several temperatures have been investigated. By analysis of chemical and physical properties of several N compounds, we can deduce possible forms in which this N evolves.

MATERIALS AND METHODS

Soybean (Glycine max [L.] Merr. cv. Davis) seed from a single plant (20) were germinated in horticultural Vermiculite moistened daily with tap water in a controlled environment. Fluorescent and incandescent lighting gave an intensity of 3.0×10^4 lux that included 22 w/m² of blue, 12 w/m² of red, and 6 w/m² of far red light at the plant apex as measured with an International Light, Inc.2 plant growth photometer. To simulate a solar day, a noninductive 13-h photoperiod started at 0630 with 1.5 h of one-third maximum irradiance, progressed through 1.5 h of two-thirds maximum to maximum, and then ended with 1.25 h of two-thirds and 1.0 ^h of one-third maximum irradiance. Day/night RH was 50/70%, and maximum/minimum air temperatures were 30 C during the day and ¹⁸ C during the night.

At the VI (unifoliate) stage of development (9) plants of similar appearance were transferred to a nitrate hydroponic culture of mineral nutrient solution as described by Stutte et al. (20). Plants were grown in separate containers with aeration $(20-30 \text{ ml/min} \cdot$ liter) in 3.5 liters of the solution. This volume was maintained throughout the experiment by daily additions of one-eighth- to one-fourth-strength nutrient solution.

Leaf vapors were collected in a calibrated tube at Dry Ice temperatures (about -50 C) in a closed system impermeable to $CO₂$ (Fig. 1). A leaflet was encased in a Saranex bag (Automated Packaging Systems, Inc., Twinsburg, Ohio) secured with a twisttie. Directed holes through the inlet valve into the bag mixed the

^{&#}x27;Contribution by permission of the Director of the Arkansas Agric. Exp. Stn. This research was supported in part by the Ben. J. Altheimer Foundation and the Arkansas Soybean Promotion Board.

² Trade names are included for the benefit of the reader and do not imply endorsement by the Arkansas Agric. Exp. Stn.

FIG. 1. Diagram of closed system used to determine net $CO₂$ uptake, transpiration, and foliar N loss.

incoming air throughout this flexible leaf chamber. The system, with an internal volume of 2.80 liters, also included a diaphragm pump (5.0 liters/min), an additional 1.8-liter mixing chamber, Millipore filter (3- μ m pores), 0.48-cm (i.d.) Teflon tubing, and 0.64-cm (i.d.) aluminum pipe. Five such units were operated simultaneously.

Two experiments consisted of sampling the center leaflet of the fourth node leaf (numbered acropetally) from five plants (V4-V5) over ^I simulated solar day at 75-min intervals and at four temperatures (20, 25, 30, and 35 C). In addition, the center leaflets of the five uppermost expanded leaves were sampled from five plants (V9-VlI1) at the same four temperatures. For plant temperature acclimation, ² days were allowed from the 30, 25, and 20 C runs and ³ days between the 20 and ³⁵ C run. Except for the simulated solar day study, sampling was conducted between 1300 and 1530 h. The upper three leaves sampled during the leaf position study received direct chamber light. The lower two leaves were shaded $(3-6 \text{ w/m}^2 \text{ of blue}, 3-6 \text{ w/m}^2 \text{ of red}, \text{ and } 2-3 \text{ w/m}^2 \text{ of far red})$ light). A wire support to each sampling bag was used to maintain leaf position in the plant canopy. Temperature of the nutrient culture was approximately that of the chamber air during each sampling period. The internal air temperature of any of the closed sampling systems did not fluctuate from the chamber temperature during the course of an experiment.

The rate of net $CO₂$ uptake was determined by monitoring at intervals the disappearance of $CO₂$ in each system during an experiment. Chamber $CO₂$ concentration levels ranged between 450 and 650 μ l/liter during this time. After each leaflet was sealed in the bag, the air was circulated for 2 min to purge the system of any atmospheric vapors and of the initial leaf vapors. Then all pumps were switched off (time 0), and clean, calibrated gas trap tubes were exchanged for those removing initial vapors. Circulation was then resumed and supplemental $CO₂$ (about 3.5 ml) added immediately via a syringe through a septum. Five-ml gaseous aliquots were removed with syringes for $CO₂$ analysis from each system at 3, 5, 7, 9, and 11 min. The $CO₂$ in each aliquot was determined with a Beckman model 215A IR analyzer. Concentrations of CO_2 at 3 min were usually 700 to 900 μ l/l and decreased to 400 to 600 μ 1/I at 11 min. This sample removal from each system reduced the internal volume 1.0-1.5% which was considered negligible. The decrease of $CO₂$ over the 8-min sampling period was used for net $CO₂$ uptake determinations; the $CO₂$ readings at 5, 7, and 9 min were used to ensure linear removal of this gas. After 12 min, the gas trap tubes were removed and stoppered.

When the tubes had equilibrated to air temperature, the transpiration rate was determined from the volume of condensate collected during the 12-min period. The condensate was then analyzed for soluble N with ^a pyrochemiluminescent technique which was sensitive to 10 µg N/liter. The technique allowed determination of total N (21, 22) and oxidized N (24) concentrations, and reduced N was estimated by difference.

Nitrogen loss, $CO₂$ uptake, and transpiration rates were based

on leaf area measured at the initiation of each experiment with a LI-COR model LI 3000 area meter.

RESULTS AND DISCUSSION

Rates of $CO₂$ uptake and transpiration, similar to those derived by other methods, were determined using this technique (Tables I and II). Net CO_2 assimilation of 12-24 (4), 17-19 (23), and 20-30 mg/dm² \cdot h (15) has been measured for soybeans with closed systems. Similar soybean transpiration rates of 1.2-2.3 (3), 1.4-1.8 (14), and 1.5-2.0 g/dm^2 -h (16) employing gravimetric methods and $2.7-3.8$ g/dm² \cdot h (7) using psychrometers have also been reported.

Net $CO₂$ uptake, transpiration, and N loss were all maximal from 1315 to 1700 of the simulated solar day (Fig. 2). Although $CO₂$ uptake and transpiration were also high at 1200 , when the chamber air temperature had just increased to ³⁰ C, total N loss was near minimum during this time; a similar trend had been noted during an earlier field study with soybeans (22). This decrease in N loss with corresponding increases in $CO₂$ uptake and transpiration indicates that N loss is ^a separate process. Regression analysis of all data present in Figure 2 ($r = 0.38, 0.01$) $> P > 0.001$, for total N loss and net CO₂ uptake; $r = 0.37, 0.01$ $> P > 0.001$, for total N loss and transpiration) suggests that the processes are statistically related. Both reduced and oxidized N losses were correlated to total N loss $(r = 0.99$ and 0.76, respectively) as was reduced N loss to oxidized loss ($r = 0.65$, $P < 0.001$). Additionally, net $CO₂$ uptake and transpiration were highly correlated over the solar day's samplings $(r = 0.77, P < 0.001)$. The diurnal changes in $CO₂$ uptake and transpiration corresponded to those found by Pallas (16) with soybeans.

Highest rates of $CO₂$ uptake and $H₂O$ loss by V4-V5 plants occurred at ²⁵ and ³⁰ C (Table I). Total, reduced, and oxidized N losses were maximal at 30 C. In the older plants (V9-V11) the greatest rates (although not as great as found in V4-V5 plants) of $CO₂$ uptake, H₂O transpired, and total and reduced N loss occurred in the younger leaves (Table II) at the higher temperatures (30 and 35 C). Farquhar et al. (8) found the highest N losses in the oldest leaves of N-stressed maize grown hydroponically. However, Weiland and Stutte (24) reported the least amount of N loss in the middle foliage of flowering field-grown soybeans (R2). Oxidized N loss appeared minimal in the middle leaves sampled at 30 C; it was maximal in these leaves at ³⁵ C (Table II). Lowest $CO₂$ assimilation rates were determined from shaded leaves in the plant canopy at all four temperatures.

Linear regression analysis (Table III) of the data in Table II suggested ^a direct relationship between total and reduced N loss, as well as between $CO₂$ uptake and transpiration, at all four temperatures. These relationships were also found in the solar day study (Table I). At 30 and 35 C, both $CO₂$ uptake and transpiration

Table I. Net $CO₂$ Uptake, Transpiration, and N Loss Rates Determined in V4- V5 Soybean Foliage at Four Temperatures

Values are means of five samples.

Table II. Net $CO₂$ Uptake, Transpiration, and N Loss Rates Determined from Five Soybean Leaf Positions of V9- VII Plants at Four Temperatures Values are means of five samples.

 $LSD_{0.06}$ within any temperature = 0.08; LSD_{0.05} over temperatures = 0.04

^a Expanded trifoliolate soybean leaves numbered basipetally from the plant apex.

were correlated to total and reduced N losses. Field studies by Stutte et al. (22) have also shown a direct correlation between transpiration and N loss as well as between temperature and N loss, and temperature and transpiration. The similarities in rates of total N loss throughout the plant canopy at ²⁰ and ²⁵ C contribute to ^a lack of correlation of N loss and temperature with $CO₂$ uptake and transpiration rates. The reduced $CO₂$ uptake and N loss rates observed at ³⁵ C compared to ³⁰ C could be the result of high root temperature and/or partial closing of stomates. Preliminary measurements with an adapted sphygmomanometer using clamping scissors (2) indicated that stomatal opening was less at 35 than at 30 C.

A direct correlation between total N and oxidized N loss was found at 25 and 30 C, but not at 20 and 35 C. As in V4-V5 plants highest rates of oxidized N loss were at the two highest temperatures. Weiland and Stutte (24) also observed that under stress conditions (i.e. H_2O or temperature) a higher rate of oxidized N loss occurs. Similarly, Klepper (13) has shown herbicide-treated soybeans to evolve significant quantities of oxidized N forms.

Several inorganic N compounds were introduced into vapor entrapment systems to test possible entrapment of forms or chemical products of these compounds. Five-ml aliquots of nitrogen dioxide, nitrous oxide, nitric oxide, and ammonia were injected into separate closed systems containing 0.2 ml of reagent water and ambient air. Only nitrous oxide was not detected in the water after 10 min and thus would not be removed under conditions of the experimental procedures.

Because nitrous oxide (b.p. -88.5 C), as well as nitric oxide $(b.p. -151.8 C)$, has a condensation point lower than the temperature of the Dry Ice, this organic form would not be entrapped. Also, nitrous oxide is stable in air. Nitric oxide is rapidly oxidized

FIG. 2. Net CO₂ uptake, transpiration, and total, reduced, and oxidized N loss rates over ^I simulated solar day. The 13-h photoperiod started at 0630. Values are the mean of five samples. $LSD_{0.05}$ for $CO₂$ uptake = 2.2 mg/dm²·h, for transpiration = 0.3 g H₂O/dm²·h, for total N loss = 0.7 μ g/dm²·h, for reduced N loss = 0.7 μ g/dm²·h, and for oxidized N loss = $0.2 \mu g/dm^2 \cdot h$.

Table III. Linear Regression Coefficients of Parameters Determined from Five Leaf Positions of Five Plants at Four Temperatures

Regression	Temperature (C)			
	20	25	30	35
PS [*] vs TR	$0.47*$	$0.75***$	$0.85***$	$0.81***$
PS vs N	-0.14	-0.30	$0.69***$	$0.56***$
PS vs RN	-0.17	-0.28	$0.67***$	$0.56***$
PS vs ON	0.34	-0.24	0.37	-0.11
TR vs N	0.18	0.14	$0.69***$	$0.58**$
TR vs RN	0.15	0.15	$0.70***$	$0.54***$
TR vs ON	0.24	0.18	0.20	-0.03
N vs RN	$0.99***$	$0.99***$	$0.99***$	$0.97***$
N vs ON	0.29	$0.56***$	$0.45*$	0.13
RN vs ON	0.20	$0.51***$	0.33	0.22

* PS = net $CO₂$ uptake; TR = transpiration; N = total N loss; RN = reduced N loss; $ON = oxidized N$ loss.

 b *, **, *** Significant at the 0.05, 0.01, and 0.001 level of probability, respectively.

in air to nitrogen dioxide (b.p. 21.2 C), or its dimer nitrogen tetroxide (N_2O_4) , both of which can be condensed at this temperature and are soluble in water (19). Both nitrogen dioxide and tetroxide, as well as nitrogen pentoxide (b.p. 47.0 C), dissolve in water to form nitric acid or nitric oxide. Nitric oxide with some nitrogen dioxide has been reported to evolve from herbicidetreated soybeans (13).

Hydroxylamine is very soluble in water but does undergo slow decomposition in solution to ammonia and even possibly to nitrous oxide or nitrogen; N_2 is not detected by pyrochemiluminescence. More probable reduced N forms which could be volatilizing from the tissue are ammonia (b.p. -33.4 C) and hydrogen cyanide (b.p. 26.0 C). Hydrogen cyanide has been found to be lost from leaves of some cultivars of sorghum (Sorghum bicolor [L.] Moench) (10), and ammonia has been detected from plant tissue by Farquhar et al. (8) and Hooker et al. (11).

Stutte and Weiland (21) and Stutte et al. (22) estimated a minimal foliar loss from a soybean field of 45 kg/ha throughout the ¹⁹⁷⁷ growing season. On a per plant basis this loss would amount to at least 5% of the plant's N. A loss of ⁴⁵ kg of N/ha was also estimated by Daigger et al. (5) from the foliage of wheat (Triticum aestivum L.) during the growth stages between anthesis and maturity. The significance of this volatilization from plant foliage could explain in part Allison's enigma in his soil N balance studies (1), as well as N imbalance in several other recent experiments (e.g. 12, 17, 18).

Although it cannot specifically define N forms, the technique described does allow estimations of total foliar N loss and the amounts in either reduced or oxidized forms. At the same time, it is a quick, repeatable method to determine net $CO₂$ uptake and transpiration from plants in the field, greenhouse, or environment chamber. The technique is currently being used in a study on nitrate reduction and foliar N loss at several temperatures from plants at different growth stages.

Acknowledgments-We acknowledge Drs. M. Davis, J. L. Wickliff, and J. B. Murphy for helpful comments and discussion, and S. Rader for preparation of figures.

LITERATURE CITED

- 1. ALLISON FE ¹⁹⁵⁵ The enigma of soil nitrogen balance sheets. Adv Agron 7: 213- 250
- 2. ALVIM PDET 1966 Stomatal opening as a practical indicator of moisture stress in cotton. Physiol Plant 19: 308-312
- 3. BEARDSELL MF, KJ MITCHELL, RG THOMAS ¹⁹⁷³ Transpiration and photosynthesis in soybean-effects of temperature and vapour pressure deficit. ^J Exp Bot 24: 587-595
- 4. CuRTis, PE, WL OGREN, RH HAGEMAN ¹⁹⁶⁹ Varietal effects in soybean photosynthesis and photorespiration. Crop Sci 9: 323-327
- 5. DAIGGER LA, DH SANDER, GA PETERSON ¹⁹⁷⁶ Nitrogen content of winter wheat
- during growth and maturation. Agron J 68: 815-818 6. DA SILVA PRF, CA STUTTE ¹⁹⁷⁹ Loss of gaseous N from rice leaves with transpiration. Ark Farm Res 28(4): ³
- 7. DORNHOFF GM, RM SHIBLES ¹⁹⁷⁰ Varietal differences in net photosynthesis of soybean leaves. Crop Sci 10: 42-45
- 8. FARQUHAR GD, R WETSELAAR, PM FIRTH ¹⁹⁷⁹ Ammonia volatilization from senescing leaves of maize. Science 203: 1257-1258
- 9. FEHR WR, CE CAVINESS, DT BURMOOD, JS PENNINGTON 1971 Stage of development descriptions for soybeans, Glycine max (L.) Merrill. Crop Sci 11: 929-931
- 10. FRANZKE CJ, AM HUME ¹⁹⁴⁵ Liberation of HCN in sorghum. ^J Am Soc Agron 37: 848-851
- 11. HOOKER ML, GA PETERSON, DH SANDER, LA DAIGGER ¹⁹⁷⁸ Gaseous nitrogen losses from winter wheat. Agron Abstr p 155
- 12. JONES MB, CC DELWICHE, WA WILLIAMS 1977 Uptake and losses of ¹⁵N applied to annual grass and clover in lysimeters. Agron J 69: 1019-1023
- 13. KLEPPER LA ¹⁹⁷⁸ Nitric oxide (NO) evolution from herbicide-treated soybean plants. Plant Physiol 61: S-65
- 14. NEWMANN HH, GW THURTELL, KR STEVENSON 1974 In situ measurements of leaf water potential and resistance to water flow in corn, soybean, and sunflower at several transpiration rates. Can ^J Plant Sci 54: 175-184
- 15. OHKI K ¹⁹⁷⁸ Zinc concentration in soybean as related to growth, photosynthesis, and carbonic anhydrase activity. Crop Sci 18: 79-82
- 16. PALLAS JE JR 1973 Diurnal changes in transpiration and daily photosynthetic rate of several crop plants. Crop Sci 13: 82-84
- 17. PATRICK WH JR, KR REDDY ¹⁹⁷⁶ Fate of fertilizer nitrogen in ^a flooded rice soil. Soil Sci Soc Am ^J 40: 678-681
- 18. RUMBURG CB, FA SNEVA ¹⁹⁷⁰ Accumulation and loss of nitrogen during growth and maturation of cereal rye (Secale cereale). Agron J 62: 311-313
- 19. SIENKO MJ, RA PLANE, EDS ¹⁹⁶⁶ Chemistry, Ed ³ McGraw-Hill, New York, pp 513-524
- 20. STUTTE CA, AR BLEM, RT WEILAND 1979 Single plant seed sources for uniform soybean physiology tests. Crop Sci 19: 324-326
- 21. STUTTE CA, RT WEILAND 1978 Gaseous nitrogen loss and transpiration of several crop and weed species. Crop Sci 18: 887-889
- 22. STUTrE CA, RT WEILAND, AR BLEM ¹⁹⁷⁹ Gaseous nitrogen loss from soybean foliage. Agron J 71: 95-97
- 23. WEILAND RT, RD NOBLE, RE CRANG ¹⁹⁷⁵ Photosynthetic and chloroplast ultrastructural consequences of manganese deficiency in soybean. Am ^J Bot 62: 501-508
- 24. WEILAND RT, CA STUTTE 1979 Pyro-chemiluminescent differentiation of oxidized and reduced nitrogen forms evolved from plant foliage. Crop Sci 19: 545-547