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The Cysteine Proteome

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Abstract

The cysteine (Cys) proteome is a major component of the adaptive interface between the genome and the exposome. The thiol moiety of Cys undergoes a range of biologic modifications enabling biological switching of structure and reactivity. These biological modifications include sulfenylation and disulfide formation, formation of higher oxidation states, S-nitrosylation, persulfidation, metallation, and other modifications. Extensive knowledge about these systems and their compartmentalization now provides a foundation to develop advanced integrative models of Cys proteome regulation. In particular, detailed understanding of redox signaling pathways and sensing networks is becoming available to discriminate network structures. This research focuses attention on the need for atlases of Cys modifications to develop systems biology models. Such atlases will be especially useful for integrative studies linking the Cys proteome to imaging and other omics platforms, providing a basis for improved redox-based therapeutics. Thus, a framework is emerging to place the Cys proteome as a complement to the quantitative proteome in the omics continuum connecting the genome to the exposome.

Introduction

The sulfur atom of cysteine (Cys) provides a considerable range of chemical reactivity and structural flexibility in the proteome. The presence of conserved Cys in motifs of proteins found in essentially all life forms indicates that these chemical characteristics were harnessed in early evolution to support enzyme catalysis, transcriptional regulation, protein folding and 3-dimensional structure. A remarkable range of biologic functions is supported by Cys because sulfur is stable in multiple coordinate covalent bonds with the major atoms of living organisms (C, H, O, N, P), forms stable coordinate complexes with transition metal ions (Zn, Fe, Cu) and is stable at a range of oxidation states (-SH, -SS-, -SO₂⁻¹, -SO₃⁻¹). Additionally, Cys thiols differentially undergo reversible ionization to the negatively charged thiolate form over the physiologic range of pH to flexibly optimize the functions of specific peptidyl Cys.

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The human genome encodes about 214,000 Cys. Cys is widely distributed among proteins, with most expressing at least one and many having multiple coordinated Cys and Cys-rich domains. The reactivity and diverse functions of Cys are mirrored by a spectrum of susceptibilities and dysfunctions of their respective proteins, resulting in central roles of the Cys proteome in development, signal transduction, biologic defenses, aging and disease. Considerable knowledge has accumulated for the symptoms and etiology of major human diseases, such as cardiovascular disease, Alzheimer's disease, lung disease, metabolic syndrome, eye disease and diseases of other organ systems. These often share related oxidative phenotypes and mechanisms, and progress in redox systems biology is beginning to provide an understanding of the integration of the underlying redox systems.

Early photosynthetic organisms transformed the earth by fixing the carbon atom of CO₂ into hydrocarbons and releasing bound atomic oxygen as gaseous molecular O₂. This changed the atmosphere from <1% to about 21% O₂ and had a profound effect on life forms by creating a diversity of external environments with different oxidative conditions. Contemporary redox proteomic systems have evolved as a result of the selective pressures of an oxygen environment and aerobic respiration, with extant genomes having selectively advantageous redox proteomic structures that protect the genome and ensure its replication despite environmental, organismal and developmental variation in oxidative and reductive challenges. Although oxidative pressure influences multiple critical loci of the redox proteome such as methionine (Met) and selenocysteine (Sec), the Cys proteome is central within this spectrum of biological plasticity and is the focus of this review.

The Cys proteome (Fig. 1) serves as a central adaptive interface between the functional genome and the external environment of an organism [1]. The external environment is a critical determinant of individual survival; in human health, this is now discussed in terms of an individual's "exposome" [2]. The exposome is defined as the cumulative measure of environmental influences and associated biological responses throughout the life span, including exposures from the environment, diet, behavior, and endogenous processes [3]. In addition to utility in contemporary environmental health research [3, 4], this provides a useful way to think about the evolution of complex multicellular organisms as the content of O₂ increased in the atmosphere. Specifically, the redox sensing and signaling mechanisms supporting responses to challenge also provided sensing and signaling systems for temporal and spatial execution of differentiation and development in multicellular organisms.

In the present review of the Cys proteome, we start with a summary of lessons learned from study of the redox proteome, that portion of reversible and irreversible covalent modifications that link redox metabolism to biologic structure and function [1]. We follow this with a more specific consideration of the chemistry of sulfur and the spectrum of oxidative reactions that defines Cys proteomic functions. In a third section, we address the redox regulation systems, especially focusing on major reducing systems that balance the oxidative reactions and support a rapidly evolving understanding of the spatiotemporal nature of redox regulation. In this, we provide specific examples illustrating the organizational structure of stable, kinetically controlled functional redox networks and summarize other modifications such as S-nitrosylation and persulfidation. Finally, we

discuss the Cys proteome within the context of integrated -omics approaches, underscoring the ongoing need for development of quantitative redox biology models.

Lessons from the redox proteome

Central dogma

Cys is incorporated into protein as the thiol (RSH) form, with apparently no exception. Thiols are oxidized to sulfenic acids (RSO^-) intermediate to formation of disulfides (RSSR) and higher oxidation states (e.g., RSO_2^-). Thiols and disulfides also undergo exchange reactions in which the thiol reacts to form a new disulfide and liberates a different thiol ($\text{RSH} + \text{R}_1\text{SSR}_1 \leftrightarrow \text{RSSR}_1 + \text{R}_1\text{SH}$) [5]. Disulfide is a common post-translational modification for 3-dimensional structure, as a component of multi-step vectorial processing and transport, as a switching mechanism in regulation or signaling and as a consequence of oxidative stress. Research with redox western and mass spectrometry-based redox proteomics methods show that partial oxidation is common for Cys residues throughout the proteome of mammalian systems [6–10]. This contradicts earlier interpretations that Cys thiol oxidation only represents an artifact of extraction [11] and supports the hypothesis that significant speciation of different peptidyl Cys in basal oxidation, organization and function occurs within the Cys proteome.

Discussions of procedures and pitfalls in measurements of thiols and thiol/disulfide have been reviewed [12–14]. Measurements involving extraction and chemical modification reflect the efficiency of alkylation relative to oxidation in trapping the thiol form [See Hansen and Winther [13] for detailed discussion]. Importantly, processing must be rapid under conditions to minimize both oxidative *and* reductive artifacts. Artifacts are more likely in experimental studies with added oxidants, metal ions or electrophiles, especially if these are not removed prior to processing biologic materials for thiol analysis. Extraction at 0° slows reaction rates about 10-fold relative to 37°, and higher O₂ in air-equilibrated solution increases rates about 5-fold that of tissues *in vivo* in mammals. Thus, in the absence of added oxidants and metals, oxidation of Cys during assay occurs at relatively slow rates, similar to the ongoing autoxidation rates *in vivo*. High concentrations of thiol-reactive chemicals can be readily achieved under most experimental conditions to minimize significant artifact [13]. An alternate reductive artifact can also occur with addition of thiol-trapping reagents under conditions where NADPH-dependent reduction continues.

Discussion of the redox activities of the Cys proteome requires care in consideration of assumptions upon which interpretations are based. For instance, redox potential refers to equilibrium conditions, which can be approximated experimentally in dilute solution with sufficient equilibration time. However, total protein concentration in biological systems is very high, many proteins are continuously undergoing reversible interaction with other proteins and many processes are kinetically limited and displaced from equilibrium. Local pH and metal ion concentrations may not be uniform, and dynamic processes of phosphorylation/dephosphorylation, acetylation/deacetylation, methylation/demethylation, reversible metal ion binding, etc., may impact *in vivo* properties. Most of the time, data and analytic methods are insufficient to fully evaluate these possibilities. Furthermore, site-directed mutagenesis studies to test function of a Cys can be misleading if experimental

conditions do not replicate conditions under which the Cys impacts function. Such issues represent challenges for the future to improve integrated redox biology models.

Protein thiols exist in a dynamic steady state

Targeted redox analyses of thioredoxin-1 (Trx-1), thioredoxin reductase-1, peroxiredoxin-1, mitochondrial Trx-2, thioredoxin reductase-2, peroxiredoxin-3, NF- κ B (p50), redox factor-1, protein disulfide isomerase and other proteins [for summary, see [15, 16]] show partial oxidation in cell culture under control conditions. Studies of glutathionylation also reveal modified residues under basal conditions, including Cys³⁷⁴ of actin in A431 cells [17] (more discussion of Cys modification below). However, these targeted studies mostly involve proteins that are regulatory or signaling in nature, leaving open the possibility that steady-state oxidation is an uncommon event. An important breakthrough during the past decade is derived from mass spectrometry-based proteomic methods, which provide much greater coverage of oxidation of the Cys proteome.

Sethuraman et al. [18] showed that Isotope-Coded Affinity Tag (ICAT) reagents [19] could be used to measure oxidation of individual Cys-containing peptides, specifically comparing differences between control and H₂O₂-treated samples. The design in this study did not permit evaluation of steady-state oxidation without oxidant. Related methods used by Le Moan et al [20], Leichert et al [21] and Go et al [22, 23] allow determination of the percentage reduction of Cys under control as well as experimentally treated conditions. For instance, proteins can be rapidly extracted, treated with a heavy stable isotopic ICAT reagent, treated with reductant and finally treated with a light stable isotopic ICAT reagent. With liquid chromatography-mass spectrometry/mass spectrometry (LC-MS/MS), this allows measurement of the percent oxidation of specific residues in peptide sequences. Importantly, Le Moan et al [20] showed that in yeast, Cys in many proteins are partially oxidized under control conditions. Leichert et al [21] obtained similar results in studies of *Escherichia coli*, and Go et al found steady-state oxidation of proteins in mouse aortic endothelial cells [22], HT29 colon carcinoma cells [6], lung fibroblasts [8], mouse liver [24] and liver mitochondria [24].

More sensitive mass spectrometers and newer peptide tagging methods are becoming available to further enhance sensitivity and coverage of the Cys proteome [6] and efforts to identify oxidatively sensitive Cys, the collective “sulfenylome” of protein Cys that forms sulfenic acids (RSO⁻) in response to multiple stimuli, have yielded increasingly promising results [25, 26]. While there remains a need for caution until more independent analyses and confirmations are available, the mass spectrometry-based methods are enhancing ability to characterize both stable and transient, partially oxidized Cys residues within the mammalian Cys proteome.

Knowledge of the functions of the redox-sensitive Cys residues is of critical importance. The peptides detected are from relatively abundant proteins and ionize well with available instrumentation. This leaves open the possibility that the Cys residues measured by LC-MS/MS are non-functional or function in a non-specific way to protect critical sites from thiol-reactive chemicals [27]. Three types of evidence suggest that many have more specific functions. Comparison among species of the measured protein Cys to other amino acid

residues within a sampling of mitochondrial proteins showed that the Cys are more highly conserved (Fig. 2A; [6]). Examination of the positions of the measured protein Cys within the 3-dimensional crystal structures for homologous proteins shows that the Cys are often in positions expected to be functionally important (Fig. 2B). These include positions near catalytic sites, protein-protein interaction sites, and RNA-binding or DNA-interaction sites. Finally, the measured Cys peptides map to functional pathways according to percentage oxidation (Fig. 3) [6]. Some specific examples are discussed in more detail below. The results are sufficient to conclude that this is a general phenomenon, although not ubiquitous. Functional networks identified include the cytoskeleton, translation, nuclear import, RNA processing, energy metabolism and oxidative stress signaling [6, 8, 24].

Selective pressure against Cys in protein

Cys codons represent 3.28% of all codons, yet the percentage of Cys in protein is typically only 2.2% or less [28], indicating that there is evolutionary selection against utilization of Cys. At the same time, the percentage of Cys in the proteome is substantially greater in complex eukaryotes (2.2%) than in archaeobacteria and eubacteria (0.5%; [28]). The Cys residues are relatively conserved in metazoan evolution, leading to the interpretation that much of the Cys proteome is functional [16]. If such function were restricted to catalytic or other “active sites”, then one or two Cys within each of the 20,000 gene products would account for only 10–20% of the 214,000 Cys in the proteome. Instead, the data suggest that up to 80% could be functional, although most without essential “life/death” function. Alternative functions for Cys include stabilization of long-lived proteins by formation of Zn clusters [28], presentation of decoy thiols to protect critical thiols from oxidants and reactive electrophiles [27], and function as redox sensing elements for coordinate regulation of functional pathways [16]. These are linked together by the concept that the Cys proteome functions as an interface between the genome and the exposome (Fig. 1; [1]). Specifically, a Cys proteome network provides a defensive barrier and sensing system to enhance fitness to tolerate exogenous oxidants, reactive electrophiles, and toxic trace metal ions, while at the same time being used to optimize food utilization.

Early forms of life evolved in significantly more reducing conditions than in the current age of highly efficient photosynthesis. Prior to the evolution of an O₂-rich atmosphere between 600 and 900 million years ago, living systems were already utilizing inorganic sulfur and Cys. Iron-sulfur clusters, for instance, are ancient prosthetic groups which may have functioned to abiotically catalyze formation of C-C bonds [29]. Along with the evolving Met proteome and associated redox reactions to regulate physiology [30, 31], Cys became increasingly utilized with organizational complexity [28]. Notably, selective pressure against the use of Cys due to vulnerability to oxidation is a result of the same chemical features that confer selective advantage when used as a redox sensor and switch [28]. The Cys proteome thus enabled evolution of systems to regulate O₂ availability in tissues and subcellular compartments as well as support developmental systems to tolerate and adapt to complex oxidative environments.

Redox compartmentalization

Oxidation of protein Cys occurs within the secretory pathway as part of a vectorial processing system for proteins targeted to specific membrane regions and for protein export [32]. Measurements of steady-state oxidation of proteins show that during processing, protein disulfide isomerase (PDI) is partially oxidized [33, 34]. Recent studies show similar redox processing of proteins occurs with mitochondrial protein import [35, 36]. Dynamic redox processes are also involved in nuclear protein import and export [37–39]. These processes depend upon different control systems, discussed in more detail below, with many utilizing thioredoxins (Trx) and thioredoxin-like proteins.

Targeted studies of steady-state oxidation of Trx in subcellular compartments [40] show that oxidation of Trx differs among major compartments of mammalian cells, with oxidation in cytoplasm > nuclei > mitochondria [15, 41–44]. Dynamic control within organelles is illustrated by NF- κ B reporter activity being decreased by generation of H₂O₂ within nuclei [9] and increased by targeting peroxiredoxin-1 to nuclei [45]. Increased abundance of nuclear Trx-1 in a transgenic mouse model showed increased NF- κ B activity and a proinflammatory state potentiating cardiovascular disease [46] and increased sensitivity to influenza infection [46]. Thus, control of the Cys proteome within subcellular compartments impacts Cys reactivity and biologic function.

Functional networks within the Cys proteome

Differences in steady-state oxidation of Trx and GSH systems among compartments suggest an organizational structure in which the central thiol redox systems control subcellular functions. Studies using mass spectrometry-based redox proteomics revealed, however, that steady-state percent oxidation of Cys in proteins within cytosol, mitochondria and nuclei mapped to functional networks rather than to subcellular compartments [6]. These results support a role for the Cys proteome in metabolic and structural organization of cells, not just as a series of isolated elements in insulated pathways within subcellular compartments but rather as a redox network connected throughout the cell.

Emerging data indicate that redox networks coordinate large numbers of redox elements in function and resistance to environmental challenges. Barabasi and coworkers described metabolic organization in scale-free hierarchical structures that are inherently stable [47, 48]. A bilateral structure for the Cys proteome can accommodate different oxidant and reductant systems and has advantages over parallel branching pathways in which protein Cys residues are physically or kinetically isolated from each other [6, 7]. Furthermore, redox proteomics studies with inhibition of thioredoxin reductase (TrxR) show that many protein Cys are simultaneously oxidized [49]. Along with the above evidence for compartmentalization, and recent data showing that redox pathways can switch due to abundance of reactive targets [50], these observations indicate that interactions within the Cys proteome are best described as a network with spatial determinants [6, 7]. Redox signaling pathways are present within this network, characterized by being localized [51] and having rapid flux compared to the relatively slow and distributed electron flow of the overall Cys proteome. We discuss this in more detail below under “Protein Cys redox regulation”, section on “Trx and GSH systems”.

Cys proteome chemistry and functions

Chemical properties of cysteinyl sulfur

The six valence electrons of sulfur in Cys are distributed among four sp^3 orbitals, which consist of two lone pairs and two sigma bonds with C and H, respectively. Compared to the more abundant element oxygen, sulfur is less electronegative and has a valence shell radius nearly twice as large. In the absence of local effects in protein structure, the thiol of Cys is 5–6 orders of magnitude more acidic than the hydroxyl group of serine. At physiologic pH, local environments in proteins can result in predominantly the thiol (most common) or thiolate (most reactive) form. Sulfur also has greater nucleophilicity, with reactivity being substantially enhanced by ionization and local protein structure. These properties favor the use of cysteinyl sulfur in biological switching, as it has increased tendency both to donate electrons as well as to accept them [52].

Reactions of protein Cys thiol

Cys thiolate donates an electron pair to thermodynamically favorable targets resulting in the oxidation of the thiol group. Sulfenic acids (RSOH), present mostly as sulfenates (RSO^-), are a common product of Cys oxidation and rapidly resolve by reaction with a thiol, eliminating water to form a disulfide (RSSR). The reaction of RSO^- with RSH to form RSSR is $>10^5 M^{-1} s^{-1}$ [52]. RSO^- are central to the oxidative modification of large numbers of Cys targets [25]. Dimeric or internal disulfides often describe a biologically relevant half-cell with the reduced Cys species to form a redox couple maintained in non-equilibrium steady-state based on interacting redox couples, compartmentation, cell cycle progression and other parameters [40]. In addition to internal disulfides, protein Cys react with GSH or free Cys to form S-glutathionyl or S-cysteinyl derivatives, with protein Cys varying in selectivity for these modifications versus formation of internal disulfides [53].

The sulfur of Cys can be sequentially oxidized by strong oxidants, progressing from sulfenic to sulfinic and sulfonic acids (present as sulfinate, RSO_2^- and sulfonate, RSO_3^- respectively). These higher oxidation states of Cys are significantly more difficult to reduce and may be considered irreversible, although reduction of certain Cys- SO_2^- [e.g., peroxiredoxin (Prx)-2] is catalyzed by sulfiredoxin [54]. Higher oxidation states are detected in mass spectral analysis of proteins but the extent of natural occurrence and possible functions remain unclear.

Cysteinyl sulfur undergoes a variety of other reversible and irreversible covalent modifications including nitrosylation, persulfidation, alkylation and arylation that can have unique effects on Cys redox regulation, function and signaling (Fig 4). At least 18 biologically occurring non-radical modifications of Cys thiol have been described, including the above reactions. These have widely varying stability and multiple reviews of their biological impacts are available [55–57]. A remarkable selectivity in reaction with different targets underscores the versatility of protein Cys in use. For instance, the redox sensor Keap1 has multiple Cys residues that react independently with alkenals, metals and nitric oxide to provide a selective detection system [58]. This system also discriminates acrylamide and H_2O_2 by reaction with different Cys residues [59, 60]. RS^- can also be

converted to thiyl radicals during ionizing radiation and other conditions. These radicals rapidly react with RS^- and O_2 in a series of reactions that convert more reactive radicals to superoxide [57, 61]. A Cys thiyl radical intermediate is used in the catalytic mechanism of all classes of ribonucleotide reductase, abstracting a hydrogen from the 3' carbon of ribonucleotide ribose to permit deoxygenation at the 2' position for deoxyribonucleotide synthesis [62, 63].

In addition to Cys, the genome includes special instructions for the translation of Sec, the so-called 21st amino acid [64]. Sec is identical to Cys except for the replacement of sulfur with selenium (Se). Cys and Sec undergo many comparable reactions, but Se is much more acidic and nucleophilic than sulfur [65]. Sec is incorporated into important thiol regulatory and antioxidant proteins, including TrxR, GPx and Met-R-sulfoxide reductase B1, formerly known as selenoprotein R [64, 66, 67]. These interactions warrant inclusion of the selenoproteome along with the Cys proteome and Met proteome in broader considerations of redox systems biology.

Factors affecting protein thiol reactivity

Common chemical features for thiol reactivity exist for reactions involving metal chelation, substrate reduction and/or conjugation or thiol-disulfide exchange. Specific Cys are, however, often limited to reaction with a fraction of potential thiol-reactive chemicals, e.g., individual thiols within the same monomeric protein are selective for thiol-disulfide exchange or *S*-nitrosylation [68]. Cys reactivity is mediated in part by the overall availability of the thiolate form because Cys thiol shares its donor electron pair in a sigma bond with hydrogen. As a consequence, pH and pK_a interdependently regulate cysteinyl sulfur reactivity. For two hypothetical thiolates in solvent pH 7.4, sharing a reaction mechanism for H_2O_2 removal and having pK_a 7.4 (50% RS^-) and 8.4 (9.1% RS^-) respectively, in absence of other influences, the more acidic thiol reacts approximately 5-fold more rapidly with the target species. Therefore, pK_a exerts influence on the relationship of a thiol to solvent pH, which differs between subcellular compartments in eukaryotes. For instance, the mitochondrial matrix is alkaline at pH 7.8 compared to the intermembrane space, which is maintained approximately 1 unit lower [69]. This affects thiol redox potential (E_h) with specificity to each thiol. For a thiol-disulfide half-cell, the effect on the standard redox potential (E_o) is -5.9 mV/0.1 pH unit. In the case of vicinal dithiols, the effect is more limited (-2.95 mV/0.1 pH) [70]. In vicinal dithiols that collaborate in a single catalytic function, one is typically more acidic and the overall behavior depends upon the presence and characteristics of other reactants. Stable RSO^- offer another means of differentiating pH regulation of redox couples, as they tend to be very strong acids [71]. However, the pK_a - and pH-determined fraction of Cys available as thiolate varies over only a 10^2 -fold range, and thus cannot solely explain the 10^6 -fold range of reactivity of protein thiols with H_2O_2 and similar ranges for other oxidants [72, 73].

The nucleophilicity index (determined by highest occupied molecular orbital and lowest unoccupied molecular orbital) of a thiol also greatly contributes to its reactivity. Specifically, a strongly acidic thiol with poor nucleophilicity may be similar in reactivity to a weakly acidic thiol with high nucleophilicity [74]. The reactive thiolate species is a “soft”

(polarizable) nucleophile and targets of Cys therefore tend to be “soft” (polarizable) electrophiles, e.g. 4-hydroxynonenal [74].

Protein Cys are also highly effective in coordinating metal ion complexes in enzyme active sites and are widely used to regulate protein structure, maintain metal ion homeostasis or sequester a metal toxicant [75, 76]. Cys cooperates with inorganic sulfur in >100 iron-sulfur (Fe-S) proteins, such as ferredoxins and Rieske proteins [77]. Proximal Cys interaction with heme iron is a conserved feature of cytochromes P450 (CYP) which enables monooxygenase activity [78]. Without the coordination of Cys234, CYP 2B4 is converted to a form with NADPH oxidase activity [79].

Cysteinylyl sulfur (alone or in combination with histidine nitrogen) coordinates Zn^{2+} in 3% of human gene products [80]. Cu^+ generally has higher affinity for Cys than other metals, and Cys-rich metal ion chaperones (e.g., metallothionein and ferritin) maintain metal ion homeostasis to prevent so-called “mis-metallation” of maturing enzymes [81]. Cys involved in metal ion coordination also participate in thiol-disulfide exchange during changes in the steady state potential of cellular antioxidants [82]. Subcellular compartmentalization of metal ions (particularly in the endomembrane system) is dependent on expression and oxidation of cysteinylyl redox-active transporters such as sarcoplasmic reticulum Ca^{2+} -ATPase (SERCA) [81, 83].

The role of Cys in catalytic sites of enzymes

The flexible reactivity of Cys allows it to confer a range of functions to enzyme catalytic mechanisms. A common role of Cys in enzyme catalysis is as a soft nucleophile, acting as the electron donor for a wide range of substrates. In addition, many proteins utilize Cys to facilitate protein-protein (e.g., intermolecular disulfide formation), protein-DNA (e.g., promoter binding) and protein-lipid interactions (e.g., S-palmitoylation) [84–86]. Furthermore, a number of proteins use Cys to permit the inclusion or function of enzyme prosthetic groups such as heme, Fe and Zn [78, 87, 88], discussed elsewhere in this review. Cys has direct catalytic involvement in at least five of the six Enzyme Commission classes of enzymes as briefly described here and summarized in Table 1.

Major subtypes of oxidoreductases involving Cys in enzyme catalysis include disulfide exchange, peroxidation, thiyl radical formation, thioesterification and persulfidation (Fig 4). Each of these reactions begins with reduced Cys thiol(ate) and involves the formation of covalent bonds between Cys thiol and substrates containing carbon, oxygen, sulfur or phosphorous but not nitrogen, which only bonds with catalytic Cys as a means of enzyme regulation. Many of these reactions transfer electrons to or from NAD(P)H via Cys. Note that while Cys thiol is oxidized in the initial step of all of these reactions, the substrates and catalytic cycle determine whether energy (often in the form of NADPH) is generated or consumed.

Protein Cys thiyl radicals are only known to occur in ribonucleotide reductases (RNR), which are essential for DNA synthesis in all kingdoms of life [63]. All RNR utilize a Cys thiyl radical to initiate the catalytic cycle but differ in how it is generated: class I RNR utilizes a stable tyrosyl radical, class II utilizes adenosylcobalamin and class III utilizes a

glycyl radical generated by Fe-S clusters and S-adenosylmethionine [63]. The Cys thiyl radical initiates reaction by oxidizing C3 of ribonucleotide ribose. The carbon-centered radical shifts to C2, dehydrates and undergoes 1-electron reduction (again forming a radical) by a pair of additional Cys residues. Finally, the radical oxidizes the initiating Cys, regenerating the original thiyl radical and releasing deoxyribonucleotide [62].

Disulfide- and peroxide-reducing functions of catalytic Cys have been extensively reviewed. These reactions transfer electrons to substrate sulfur or oxygen in an oxidation state of 0 (RSOH sulfur) or -1 (RSSR and peroxides). The result is formation of disulfide or RSOH on the nucleophilic Cys and the liberation of RSH, H₂O or ROH from the substrate. The initiating Cys, typically oxidized to RSSR, must then be reduced to regenerate the enzyme. These reactions make up the bulk of redox regulation by members of the TrxR (thioredoxin reductase) and glutathione reductase (GR) antioxidant nodes. These reactions are also found in important chaperone proteins, such as protein-disulfide isomerase, which are classified by the Enzyme Commission as isomerases because they rearrange protein disulfides rather than cause net oxidation or reduction [89].

Sulfurtransferases are proteins that catalyze the formation, interconversion and reactions of compounds containing sulfane sulfur atoms [90]. Although characterization of many is incomplete, Cys persulfide (Cys-SSH) is an important catalytic intermediate for some [91]. The catalytic Cys is persulfidated by thiosulfate, hydrogen sulfide or mercaptopyruvate, creating a highly reactive sulfur donor, which produces low molecular weight sulfur metabolites such as thiocyanate (SCN) and methanethiol [92]. H₂S can be liberated from the Cys persulfide as well as the original persulfide donor [92]. As persulfidation is also a putative peptidyl Cys regulatory mechanism, it is covered elsewhere in this review.

Catalytic Cys-formed thioesters fall into two classes: thioacyl esters (S-CR bond) and thiophosphate esters (S-PO₃⁻ bond). Thioacyl esters are intermediates of aldehyde dehydrogenases (ALDH) and cysteine protease catalytic cycles. In the case of glyceraldehyde-3-phosphate dehydrogenase (GAPDH) and ALDH, Cys thioacyl formation is preceded by a thiohemiacetal intermediate which donates a hydride to NADP⁺ [93]. Similarly, Cys proteases attack the amide carbon peptide backbones at select residues (e.g., aspartate), causing cleavage of the amide bond [94]. Cys proteases are unlike Asp proteases in that they do not participate in classical acid-base hydrolysis. Rather, the Cys thioacyl ester provides a good leaving group to facilitate hydrolysis and protease regeneration. Similarly, the Cys-SPO₃⁻ thiophosphate ester formed following nucleophilic attack on phosphotyrosine in the protein tyrosine phosphate (PTP) catalytic cycle is an excellent leaving group [95]. However, as in acid-base catalysis, these reactions are supported by negatively charged side chains such as Asp and Glu, which coordinate the protonation of water, substrates and Cys itself.

Protein Cys redox regulation

Trx and GSH systems

Extensive research in the last decades shows that redox responses of the Cys proteome occurring at the extracellular and cellular levels are critical in initiation and progression of

disease. Two major cellular redox regulation systems, dependent upon thioredoxin (Trx) and glutathione (GSH), are central to regulation of the Cys proteome. These systems maintain a reduced redox environment in cells by catalyzing electron flux from NADPH through Trx reductase (TrxR) and glutathione disulfide reductase (GR) to Trx and GSH, respectively. In addition, mixed disulfides of GSH with protein Cys (S-glutathionylation) can have broad impact on protein function and disease [88]. S-Glutathionyl groups are removed by glutaredoxin (Grx) plus GSH. In some cases Grx may transfer electrons from the GR-GSH node into the TrxR-Trx node by reducing Trx disulfide [96]. Numerous studies of Cys modification and redox regulation provide evidence that redox-sensitive regulatory Cys residue(s) (see Fig 5 legend) play critical roles in cell signaling and control as part of a redox network structure. These systems are involved in multiple signaling pathways through direct or indirect interactions with a variety of target proteins controlled by the GSH and Trx redox systems (Fig. 5). Proteins and associated protein networks are often selectively regulated by Trx or GSH systems but also commonly regulated by both GSH and Trx systems, e.g., S-glutathionylation in C³⁷⁴ and S-nitrosylation in C²⁸⁵ of actin (Fig. 5).

As indicated above, mass spectrometry-based redox proteomics together with bioinformatics enables identification of redox-sensitive Cys residues, proteins and networks of Cys proteome in association with cellular conditions [6, 8, 49]. Studies in yeast showed that the redox proteome was preserved in cells deficient in the GSH system but not in cells with compromised Trx systems. These results show that the Trx system is most central in regulation of the yeast Cys proteome [97]. In mammalian cells, responses of the Cys proteome were tested by selectively disrupting the Trx- and GSH-dependent systems with inhibitors, auranofin for thioredoxin reductase and buthionine sulfoximine (BSO) for γ -glutamylcysteine synthetase, the first enzyme in the GSH synthesis pathway. Results showed that the Trx and GSH systems support distinct functions in cellular redox control by controlling different subsets of protein Cys. Consistent with the yeast findings, the Trx system regulated a much broader range of proteins than the GSH system [49]. However, GSH and Grx can act as a backup to the TrxR-Trx system to promote the reduction of Trx during aurothioglucose exposure [96]; the extent to which this alters redox proteomic effects of TrxR inhibition is unclear. Examples from studies with selective inhibition of TrxR and GSH systems are available [49, 98].

Regulation of actin cytoskeleton proteins

Cytoskeleton remodeling through actin dynamics regulates vital biological processes, e.g., cell morphogenesis, migration and growth, membrane trafficking, exocytosis and endocytosis [99–101]. Actin is susceptible to multiple post-translational modifications in Cys residues including reversible thiol oxidation (Cys^{217, 257, 272, 285, 374}) [102–104], glutathionylation (Cys^{10, 217, 257, 374}) [105–107] and nitrosylation (Cys^{217, 257, 285, 374}) [108–110]. Oxidation disrupts actin dynamics, e.g. filamentous (assembled) and monomeric (disassembled) actin formation [111, 112]. Glutathionylation/deglutathionylation of actin, regulated by the concentration of GSH and Grx activity, is also associated with actin dynamics [113–115].

In addition to actin itself, a cohort of actin-associated cytoskeletal proteins control actin dynamics. Oxidation of Cys¹³⁹ and Cys¹⁴⁷ in actin-binding protein cofilin stimulates apoptotic signaling as a consequence of inhibiting interaction with actin as well as increasing mitochondrial translocation [116], and cofilin oxidation impairs cytoskeletal function in T cells [117]. Furthermore, destrin, a cofilin family member, is an actin depolymerizing factor regulated by Cys redox status. Destrin nuclear translocation was associated with redox disruption of actin cytoskeleton proteome [8]. Actin cytoskeleton proteins were also susceptible to oxidation during inflammatory signaling associated with mitochondrial oxidation [6, 22]. Redox-dependent control of actin dynamics was observed from interaction of actin with the Trx system, Trx1 and TrxR1 [110, 118, 119]. Release of Trx1 from actin resulted in degradation of Trx1, which then stimulated cell death [119]. Denitrosylation of actin by TrxR via focal adhesion kinase restored cellular function [110].

Extensive knowledge of the redox modulation of the cytoskeleton by GSH and Trx systems show the central importance of redox control in cell function. The data demonstrate that GSH and Trx do not indiscriminately “equilibrate” the peptidyl Cys environment to a consistent reduction potential but sensitively and selectively maintain discrete non-equilibrium dynamics for individual proteins and protein networks. The following examples further illustrate the use of kinetically controlled systems in nuclear protein trafficking and in maintenance of the integrity of the genome. Recent reviews are also available on the redox processing in the secretory pathway [33, 34, 120] and in mitochondrial import [35, 36]. Relatively less information is available on vectorial redox processing during endocytosis and proteolysis in lysosomes, but evidence is available suggesting that analogous systems are present [121, 122].

Cys proteome in nuclear protein trafficking

Redox-dependent transcription has been known for many years, with divergent mechanisms involved for AP-1, Nrf2, NF- κ B, glucocorticoid receptor and other systems [38, 123–126]. Both Trx1 and its inhibitory binding partner Txnip are translocated into nuclei to regulate the Cys proteome. For instance, nuclear accumulation of Txnip was detected when cells were treated with the anti-cancer histone deacetylase inhibitor, suberoylanilide hydroxamic acid [127]. Rch1 (importin- α) is required for Txnip nuclear translocation and growth control, and this interaction is controlled by a Trx-dependent redox system [128]. Interestingly, Trx1 is also translocated into nuclei, and import required karyopherin- α (importin- α) and nitric oxide (NO). Site-directed mutagenesis of Lys^{81,82} of Trx1 showed that these residues are important for the translocation of Trx1 into nuclei [129], suggesting that the protein translocation mechanism is complex, requiring localized signal motifs in addition to redox-dependent regulation.

In addition to enhanced Trx1 translocation to nuclei, transport and translocation of a wide range of proteins are regulated in a redox-dependent manner. For instance, oxidative stress stimulates nuclear translocation of SBP 2 [SECIS (selenocysteine insertions sequence) binding protein 2] via oxidation of Cys within a redox-sensitive Cys-rich domain. Reversal by Trx1 or Grx suggests that redox state of SBP 2 is critical to compartmental translocation and selenoprotein synthesis [130]. In bronchopneumonia, protein *S*-glutathionylation

significantly co-localized with Fas ligand, and Grx1 knockout mice had increased Fas S-glutathionylation and improved murine lung infection survival [131]. Mass spectrometry-based redox proteomics also showed functional redox networks of nuclear import proteins (Fig 3; [6]). Additionally, targeted expression of nuclear Trx1, stimulation of nuclear H₂O₂ production with nuclear-targeted D-amino acid oxidase, and enhanced nuclear peroxidase activity by expressing nuclear targeted Prx1 further show distinct redox control in nuclei [9, 45]. Thus, multiple approaches show the nuclear Cys proteome is controlled in a semiautonomous manner by transport and compartment-specific regulation.

Nuclear functions of the Cys proteome

The dynamic nature and functional importance of compartment-specific control of the Cys proteome are further illustrated by Trx-dependent protection against cellular stresses associated with growth inhibition and cell death induced by H₂O₂, TNF α and chemotherapeutic agents [132–134]. Trx-interacting protein [Txnip/Trx binding protein (TBP)/vitamin D3-upregulated protein] directly binds to Trx1 by forming a disulfide bond between Cys²⁴⁷ of Txnip and catalytically active Cys residues of Trx (Cys³² and Cys³⁵), inhibiting redox activity of the Trx [135, 136]. The complex of Txnip-bound Trx contributes to pathogenesis of multiple diseases including diabetes, autoimmune disease and cancer [137–139]. Upregulation of Txnip under sustained high glucose exposure caused increased ROS generation and caspase-3 activation [140]. The pathological role of Txnip was revealed with release of Txnip from oxidized Trx1 to stimulate inflammation [141] and cell death [140]. Similarly, the reduced form of Trx1 binds to apoptosis signal-regulating kinase-1 (ASK-1) and inhibits its activity to prevent stress- and cytokine-induced cell death [142]. Activation of ASK-1 is stimulated under stress signaling after dissociation with Trx1 [142] and plays a key role in apoptosis with increased DNA damage [143, 144]. Direct interaction of Trx with Txnip (Cys²⁴⁷) and ASK-1 (Cys²⁵⁰) via reduced Cys³² (Cys³⁰ of Trx2) supports a key role of Trx thiol-disulfide redox regulatory function against oxidative stress-mediated DNA damage and cell death signaling [141].

Trx1 also plays a key role in DNA repair mechanisms by regulating the base excision repair pathway via the p53 downstream gene, growth arrest, and DNA-damage-inducible protein 45- α and its interaction with apurinic/apyrimidinic endonuclease 1 [APE1, also called redox factor (Ref)-1] [145]. This mechanism is regulated by the interaction between Trx1, Ref-1 and redox sensitive p53 transcription factor [146, 147]. TrxR-Trx1 also interacts with Ref-1, and this interaction controls p53 activity supporting the Trx1 redox system in modulation of redox-sensitive transcription factors [148]. Moreover, Trx1 protection against cell death was shown to occur in growth regulation in B-cell chronic lymphocytic leukemia (CLL) by control of the interaction of protein disulfide isomerase with tumor necrosis factor receptor (TNFR) [149]. This finding additionally supports the beneficial role of Trx1 in survival, perhaps via activation of NF- κ B transcription factor [147].

Oxidation of the Cys proteome by H₂O₂ and derivative species

Measurements of bulk protein Cys in mouse liver and HT29 cell culture show that 92% and 88%, respectively, are in the thiol form [6]. With procedures to minimize artifacts in mass spectrometry measurements, similar mean (84.3%) and median (87.7%) values were

obtained for HT29 cells in culture. Specific protein Cys had a distribution of steady-state oxidation ranging from less than 8% to greater than 99% reduced. These observations raise key questions about the oxidants involved and the underlying enzymatic and non-enzymatic mechanisms contributing to steady-state oxidation.

The balance of oxidation and reduction of the Cys redox proteome is governed by the thermodynamics, selectivity and kinetics of different peptidyl Cys. As indicated above, details of the reduction systems are becoming clear. In contrast, the fine-tuned oxidative signaling has been ascribed extensively to H₂O₂, a species that is relatively unreactive with most Cys thiols. At least three factors impact specificity of oxidation, i.e., localized high activity generation of H₂O₂, catalytic mediators of oxidation, and generation of alternative low-molecular weight chemical oxidants. Enzymes with high reactivity with H₂O₂ [e.g., members of the Prx and GSH S-transferase(GST)] may act as special carriers of the oxidative signal to interacting proteins [150]. Alternatively, H₂O₂ is used to generate other oxidants with altered kinetics (e.g., hypohalous acids) [151], which in turn can form further derivative oxidizing species (e.g., haloamines). These latter processes support speciation of an H₂O₂ oxidative signal into various paths, which regulate not only the fate of H₂O₂ but also the effect of its oxidative signal.

H₂O₂

Although O₂ is present at five orders of magnitude higher concentration than H₂O₂ in aerobic organisms, H₂O₂ is usually considered the most central oxidant in maintenance of steady-state oxidation of the Cys proteome. H₂O₂ is a product of at least 31 human enzymes (Table 2) and has been measured in mammalian tissues at concentrations in the range of 1 to 50 nM [152–155]. Measurements of H₂O₂ based upon steady-state kinetics of endogenous catalase in perfused rat liver developed by Sies [156] may be the most reliable tissue measurements available because there are no possible artifacts due to transfection or calibration of imaging reagents. Catalase concentration is too low to support this type of measurement in most cells and tissues, however, and more sensitive fluorescent probes have been widely used. Some, such as dichlorofluorescein diacetate (DCFDA), are non-specific and require caution when evaluating cell and tissue H₂O₂ production. Other probes can also be non-specific, insensitive to rapid flux changes, or misleading due to redistribution of probes within cells [157]. In contrast to high reactivity of H₂O₂ with peroxiredoxins, GSH peroxidases and catalase, H₂O₂ reacts quite slowly with the majority of thiols. For low molecular weight thiolates, the rate constant of reaction with H₂O₂ is about 20 M⁻¹ s⁻¹ but the effective rate constant for total thiol (thiol+thiolate) is <5 M⁻¹ s⁻¹ [158]. Using 5 M⁻¹s⁻¹, total protein Cys at 20 mM and H₂O₂ at 10 nM, non-catalytic oxidation of protein Cys by H₂O₂ in cells occurs at 0.0003 % per minute. This rate is too slow to maintain the steady-state oxidation as indicated above.

Multiple alternatives exist which could account for higher steady-state oxidation, e.g., a fraction of protein Cys with higher rate constants, more reactive oxidants derived from H₂O₂, or catalytic oxidation of protein Cys by peroxiredoxins, glutathione S-transferases, or other enzymes in the presence of H₂O₂. Studies of different protein Cys show that rate constants for reaction with H₂O₂ vary over more than six orders of magnitude [57]. A

network model of H₂O₂ clearance with 24 elements and 28 kinetic parameters, including pseudo-enzymatic oxidative turnover of protein thiols, the enzymatic actions of catalase, glutathione peroxidase, peroxiredoxin, and glutaredoxin, and the redox reactions of Trx and GSH, correctly reproduced experimentally measured oxidation profiles for the GSH and Trx redox couples [159]. The model suggested that approximately 10% of intracellular protein thiols react with H₂O₂ at substantial rates and predicted an unequal distribution of the intracellular anti-oxidative burden between Trx-dependent and GSH-dependent antioxidant pathways, with the former contributing the majority of the cellular antioxidant defense due to peroxiredoxins (Prx) and protein disulfides. Importantly, if enzymes with peroxidase activity, such as the Prx and GST families, use protein Cys as direct reductant, supported by Trx or GSH systems to reduce the protein Cys back to thiol form, then a complete enzymatic system exists to maintain the dynamic Cys proteome [7]. The overall organizational structure is one in which the Cys proteome is not collectively equilibrated with NADP⁺/NADPH, O₂/H₂O or at any single redox potential. Instead, a continuum of kinetically limited non-equilibrium steady states exist for protein Cys, varying in abundance and spatial and temporal distributions [40].

The system is dynamic and responds to changes in H₂O₂ during inflammation and signaling for proliferation, cell migration or host defense [151, 160–162]. Research shows substantial circadian variation [163], changes during development [164] and spatiotemporal temporal changes during wound healing [165]. The oxidative second messenger [166] with localized generation [51] is amplified by a “flood-gate” mechanism [167] involving excessive oxidation of Prx and Trx1 (see below) [168]. This mechanism could alter redox signaling via diverting H₂O₂ flux to oxidize other proteins or, perhaps more likely, support enzymatic oxidation or other redox-activated signaling mechanism [166].

Earlier analytic methods only detected protein disulfides; development of reagents for trapping RSO⁻, however, showed that RSO⁻ is the predominant modification of the Cys proteome [25] and central to oxidative Cys regulation (Fig. 4). For instance, RSO⁻ in multiple PTPs is associated with inhibition of CD8⁺ T cell activation [169], and formation of RSO⁻ in IQGAP1 plays a direct role in endothelial cell migration and angiogenesis [170]. Recent bioinformatic analysis using a dimedone-tagged click chemistry probe has revealed multiple peptidyl Cys with basal RSO⁻ modification, including Cys¹⁸ of SIRT6 and multiple Cys of pyruvate kinase M2 [26]. RSO⁻ were detected in solvent-accessible peptidyl Cys from proteins associated with most subcellular compartments and included enzymes regulating phosphorylation, acetylation and ubiquitylation [26]. Peptidyl Cys with detectible RSO⁻ modification were often associated with Glu and Lys residues and less often with other Cys residues [26].

Oxidation of RSO⁻ to RSO₂⁻ (sulfinate) in peroxiredoxin has a special role in signaling because this oxidation inhibits peroxidase activity, redirects reducing equivalents from the Trx system away from Prx2 and changes protein Cys redox regulation [171]. This process is termed “hyperoxidation”, and detailed studies with Cys⁵⁰ of Prx2 [172], including thermodynamics and contributions of proximal amino acids [60, 73, 173, 174] are available to describe the underlying mechanism. Ultimately, hyperoxidation of Prx2 diverts H₂O₂ to other antioxidant systems. Such diversion alters redox signaling by changing elements of the

Cys proteome that react with oxidant [166]. Furthermore, hyperoxidation of Prx2 is biochemically regulated, as sulfiredoxin reduces Prx2 Cys sulfinate back to the sulfenate form, which may be reduced by Trx [54].

The Cys proteome includes sulfinic and sulfonic acids, as well as series of sulfenamides, sulfnamides and sulfonamides. With the exception of sulfinic acid and possible precursor RSONR₂ [175] in peroxiredoxins, there is little evidence that these represent functional variations. Similarly, cyclic sulfonamide (isothiazolidinone) structures are formed from the catalytic Cys of certain phosphatases, e.g. Cys²¹⁵ of PTP1B [175, 176], but the functional and physiologic relevance of these is unclear. Cysteinylyl sulfur and amine interaction also provides means to cross-link molecules by strong oxidants, observed in model cysteinylyl peptides and GSH and in vivo [177–179]. Thiosulfinates and thiosulfonates are documented in thiol chemistry [52], and thiosulfinate has been found as an intermediate of sulfiredoxin (Srx) Cys⁹⁹ reduction of Prx2 sulfinic phosphoryl ester [180]. Thus, a diverse range of oxidized sulfur and sulfamine modifications of the Cys proteome can occur, but only a very small number have been characterized in terms of biologic functions.

Oxidant species derived from H₂O₂

The Cys proteome is subject to oxidation by peroxide-derived oxidants, some of which have similar thermodynamic potential but lesser kinetic limitations than H₂O₂. Hypohalous acids (hypochlorous acid, HOCl; hypobromous acid, HOBr) are strong oxidants generated by phagocytic myeloperoxidase (MPO) during inflammation and widely recognized as dual agents of host defense and host injury [151, 181]. HOCl and HOBr rapidly react with thiolates by transfer of Cl⁺ or Br⁺ [52]. The resulting sulfenyl halide is unstable and rapidly hydrolyzes to release Cl⁻ or Br⁻ and RSO⁻ [182]. Similarly, eosinophil peroxidase is released in allergic events and generates HOBr, but little or no HOCl at physiologic pH [183, 184].

Low-level generation of HOCl and HOBr can also occur in tissues without inflammation via the activity of peroxidasins (also known as vascular peroxidases), which participate in collagen cross-linkage and are essential for tissue development [185–188]. However, the impact of constitutive HOCl and HOBr generation on Cys proteomic regulation has not been extensively characterized. In principle, these oxidants could oxidize selected subsets of the Cys proteome as suggested by the finding that HOCl independently activates Nrf2 and MAPK signaling in airway epithelia [189]. A potential limitation for specific Cys oxidation is that HOCl and HOBr have high and indiscriminate reactivity with most Cys [190] and also react with non-Cys residues, e.g., Met and primary amines [191, 192]. On the other hand, conversion of hypohalous acids into less reactive haloamines (e.g., RNHCl) may confer specificity. For instance, taurine chloramine oxidizes Cys²⁸² of creatine kinase with 3-fold increased selectivity compared to HOCl [193].

Hypothiocyanous acid (HOSCN) is a related oxidant that is constantly produced from endogenous thiocyanate (SCN), lactoperoxidase (LPO) and H₂O₂ in human secretory fluids (e.g., saliva, airway lining fluid, tears and milk) [194–196]. HOSCN has been studied extensively in saliva and is maintained at concentrations >5-fold more abundant than the precursor H₂O₂ [197]. In addition to enzymatic production by LPO, MPO and other

peroxidases (REF=Van Dalen et al Biochem J 1997)[186], HOSCN is produced by the direct reduction of HOCl or HOBr by SCN^- , a reaction which competes effectively with physiologic concentrations of GSH [52, 198]. HOSCN reacts with protein Cys to form RSO^- and SCN^- [52, 198].

The most critical aspect of HOSCN is that it is significantly less reactive than H_2O_2 , HOCl or HOBr [173, 199–201], which suggests greater specificity for Cys proteomic regulation. Support for this possibility is limited, but HOSCN, and not HOCl or HOBr, induces cell signaling by reaction with serum factor(s) [202, 203]. HOSCN selectively targets salivary GSTP, but not GSTA or GSTM [204]. Altered redox signaling of HOSCN may be reflected in altered cellular responses to this oxidant instead of others, e.g., removal of H_2O_2 and HOCl by SCN^- protects against mammalian cell death and decreases inflammation [179, 205]. Furthermore, mammalian innate immunity exploits specificity of HOSCN reactivity to place selective oxidative pressure on microbes [206]. The biochemical and physiological effects of SCN^- and HOSCN have been recently reviewed [207].

S-Nitrosylation

Nitric oxide (NO) is a precursor for S-nitrosylation of the Cys proteome, governed by multiple reaction pathways, e.g., metal catalysis of NO^+ transfer, with the ultimate action being formation of protein S-nitrosocysteine (RSNO). S-Nitrosylation is regulated by GSH and Trx systems, which reduce RSNO [208, 209]. Trans-S-nitrosylation is important in NO signaling and as a negative feedback mechanism for NO synthases and effector proteins [209]. SERCA is nitrosylated at Cys³⁴⁴ and Cys³⁴⁹ following ONOO⁻ exposure, which enhances subsequent protein glutathionylation [210]. The oxidation of critical SERCA Cys residues, e.g. S-nitrosylation and S-glutathionylation, may be a guanylate cyclase-independent means to alter vascular tone [211]. Skeletal muscle ryanodine receptor/ Ca^{2+} -release channel (RyR1) Cys³⁶³⁵ is regulated by S-nitrosylation while other Cys mediate redox sensitivity through SNO-independent means [209, 212, 213]. Several mitochondrial metabolic enzymes are targets of S-nitrosylation [214]. Thus, S-nitrosylation represents a series of modifications of the Cys proteome, which can be considered parallel to the oxidative systems in providing a network of regulatory elements impacting diverse aspects of cell function.

Persulfidation

Persulfidation, also termed S-sulphydration [215, 216], is another mechanism for modification of the Cys proteome that can operate in parallel with the oxidative and S-nitrosylation systems. The protein persulfide (RSS^-) is derived from H_2S generated from Cys via intestinal microbes, mercaptopyruvate sulfotransferase, cystathionine β -synthase (CBS) and cystathionine γ -lyase (CGL) [217, 218]. At physiologic pH, H_2S is principally ionized (HS^-) and reacts with a disulfide to produce persulfide (RSS^-) and release a thiol. For instance, GSH persulfide (GSS^-) is formed from reaction of GSSG with H_2S [219]. The sulfur added to the thiol in the persulfide moiety is present at the oxidation state of elemental sulfur (sulfane), which commonly exists as octasulfane (S_8). Cysteinyll polysulfides have been documented in crystal structures of bacterial sulfide quinone oxidoreductases [215], while free polysulfides (H_2S_n) in brain tissue can persulfidate multiple protein Cys targets

[220]. Sulfane sulfur can be transferred between protein thiols and low molecular weight thiols (trans-persulfidation or trans-sulfanation). The extent of persulfide modification of proteins is currently unknown [221, 222].

Persulfide formation generally interferes with normal protein Cys reactivity in much the same way as *S*-nitrosylation or sulfenylation [223] but differs in that the highly nucleophilic auxiliary sulfane is more reactive toward electrophiles [224]. Inhibition of H₂S generation leads to mitochondrial dysfunction and increased O₂⁻ flux, suggesting a role for H₂S in regulation of mitochondrial function [225]. Persulfidation of 39 proteins, including GAPDH Cys¹⁵⁰ was detected by mass spectrometry [226]. Persulfidation of GAPDH Cys¹⁵⁰ by treatment with NaSH increased enzyme activity, with an EC₅₀ of 15 μM; GAPDH is also a *S*-nitrosylation target, highlighting a potential for H₂S/NO cross-talk [226]. After TNF-α stimulation of CGL, increased persulfidation of p65 Cys³⁸ regulated its DNA binding and anti-apoptotic function [227]. The C⁵⁹S mutant of P66Shc is resistant to anti-phosphorylation effects of H₂S or CBS overexpression, implicating regulation by persulfidation [228]. H₂S also regulates eNOS activity by persulfidation of Cys⁴⁴³, which promotes activity of the NOS in opposition to NO binding [216].

Alkylation

Many modifications of the Cys proteome are irreversible under normal circumstances in vivo. Some of these, such as modification by endogenously derived reactive lipids, function to provide a response to stress [229]. Reactive aldehydes make up a highly significant and well-recognized fraction of permanent Cys adduction [230]. Many α/β-unsaturated aldehydes such as 4-hydroxynonenal (4-HNE) are “soft” electrophiles that react readily with soft nucleophiles and are chiefly regulated in vivo by Cys thiolate [74, 174]. 4-HNE reacts with Cys residues of many important proteins. The thioredoxin system, including TrxR (Cys⁴⁹⁶ along with Sec⁴⁹⁷ [65]) and Trx (Cys³², Cys³⁵), is very sensitive to inhibition by 4-HNE [231]. Similarly, the catalytic and modifying subunits of Glu-Cys ligase (GCLC/M) are targets of 4-HNE (GCLC Cys⁵⁵³, GCLM Cys³⁵) [232]. 4-HNE also induces antioxidant genes (e.g., TrxR1; heme oxygenase-1, HO-1) through induction of the Nrf2 promoter by modifying Cys²⁸⁸ of Keap1 [58, 233]. Conversely, 4-HNE inhibits the release of NF-κB by modifying Cys¹⁷⁹ of IKK-β [234]. Together, these examples emphasize that irreversible modification of Cys in signaling proteins is functionally important to respond to electrophile exposures [229].

Selectivity of protein Cys to electrophilic modification has been demonstrated by comparisons of proteins modified by iodoacetyl- and maleimido- reagents [235]. For instance, of 1255 distinct Cys-containing peptides from isolated mitochondria, the electrophiles only overlapped 35% in Cys adduction and had distinctly different effects on cytotoxicity and apoptosis. Similarly in nuclear-cytosolic fractions, there was only 14% peptidyl Cys overlap between the electrophiles [236]. The targeted residues represented proteins involved in a variety of central cellular pathways including Trx-2, heat shock proteins, permeability transition pore complex, nuclear/cytosolic cytoskeleton, DNA replication/repair, protein synthesis and RNA processing [236].

As described above, the Cys proteome is part of an adaptive interface of an organism with its environment, and covalent modifications of the Cys proteome are useful to characterize environmental exposures to pollutants, dietary metabolites, drug metabolites and endogenous toxicants [237]. In human serum albumin, 66 adducts of Cys³⁴ have been detected [238]. Similarly in drug metabolism, quinone structures from metabolites of acetaminophen (APAP), diclofenac and related drugs have distinct 1,4-Michael addition reaction with thiols to form a covalently modified Cys³⁴ of albumin. The APAP metabolite N-acetyl-p-benzoquinone imine (NAPQI) targets peptidyl Cys of essential metabolic and regulatory proteins, including TrxR1 which is covalently modified by NAPQI at residues Cys⁵⁹ (GR-homologous active residue), Cys⁴⁹⁷ and Sec⁴⁹⁸ (vicinal thiol-selenol pair), associates these modifications with loss of TrxR activity in APAP overdose mouse liver [239]. Other modifications also occur, such as, GAPDH Cys¹⁴⁹ modification by NAPQI [240].

Integrated function of the Cys proteome

The proteome is the embodiment of the genome, serving to translate the genetic and epigenetic systems into function and interaction with the environment. The series of post-translational modifications of the translated proteome, collectively termed the epiproteome [241], dramatically expands the structural variation and range of reactivity beyond that found in the translated proteome. Perhaps most importantly, bi- and multiphasic switches of the Cys proteome provide organizational hubs to broadly control and integrate complex biologic functions.

While not fully elucidated, recent studies have begun to define multiphasic switching by the Cys proteome through detection of proteins with *S*-nitrosylation, *S*-glutathionylation, persulfidation and sequential oxygenation [242]. Some Cys residues can undergo multiple modifications, potentially providing pathway switches to change biologic function. For instance, Cys³⁷⁴ of actin can be nitrosylated as well as glutathionylated [107, 109]. However, comparison of sulfenylated Cys residues in 193 proteins with previously identified targets of glutathionylation, disulfide formation and *S*-nitrosylation shows minimal overlap [25]. These results suggest that different Cys modifications are selectively targeted toward different subsets of the Cys proteome rather than serving as multidirectional switches. Expansion of this pioneering research is needed to provide an understanding of the integrated function of the Cys proteome.

At the same time, multiple omics platforms are being used to translate basic science into medical applications [243]. Comparison to genomics and transcriptomics illustrates current limitations at the level of the Cys proteome. Unlike the genome and epigenome, where systematic methods provide nearly comprehensive coverage, only a few percent of the Cys in the proteome are captured in the best mass-spectrometry proteomics analyses. In studies of gene-expression-redox proteomic interaction [244] common pathway responses were observed, but the data were too sparse to allow direct correlation. On the other hand, more targeted analysis of redox proteomic associations with metabolomics in isolated mitochondria provided direct correlations [24]. The data indicate that with improved coverage of the proteome, integrated omics research will enable mechanistic understanding

at the level of systems biology, i.e., directly linking changes in the Cys proteome with altered biologic function.

Perspective and future directions

A conceptual framework is emerging for elucidation of the Cys proteome as a central omics layer within the continuum of interactive systems linking the genome and exposome (Fig. 1). Major gaps exist, however, in the hierarchy of interactions of Cys proteome sub-networks, i.e., how do the different oxidation, nitrosylation, persulfidation, metallation and alkylation sub-networks synergize or antagonize each other? Sufficient detailed understanding of individual Cys proteomics systems is available to initiate construction of more complex systems models using a “bottom-up” approach. However, a limitation of a bottom-up approach is that an atlas of Cys modifications is required to place individual signaling elements within the context of others. This highlights a need for systematic development of resources for systems biology of the Cys proteome. With advances in mass spectrometry-based methods, increased coverage of large segments of the Cys proteome is possible. But consortium efforts will be needed to approach the coverage currently available for the genome and transcriptome. Such resources could allow development of quantitative models to test function of the different Cys proteome sub-networks with existing physiologic, pharmacologic, toxicologic and genetic tools.

Such quantitative descriptions of Cys proteome pathways and networks are essential to understand sites of vulnerability in disease mechanisms and provide a foundation for a new generation of network-based redox therapeutics. Many manipulations of the Cys proteome have been used in human research and some are effective in clinical practice. For instance, global regulation of Trx-1 and GSH systems occurs through activation of the Nrf2 system, a system targeted to prevent neurodegeneration [245], in cancer prevention [246], cardiovascular disease [247] and other disease processes [248].

While optimism is appropriate concerning such applications, there is also need for recognition of the failures of free radical scavenging antioxidant trials [249, 250], as manipulations of the Cys proteome are pursued for health outcomes. The research discussed in the present review shows that the Cys proteome is a complex network, apparently with an evolved function to support adaption to diet and environment. Therapeutic manipulations are, in effect, environmental perturbations. Hence, outcomes from treatments to modify the Cys proteome may suffer the same fate as the antioxidant trials if network structures are not considered in study design.

Fortunately, the range of modifications for biological switching of structure and reactivity is now well established. Methods are available to quantify abundance and reactivity, to incorporate reactions with electrophilic chemicals and binding to metal ions, to elucidate the details of the switching systems and to define their compartmentalized functions. With these capabilities, advanced integrative models, with links to imaging and other omics platforms, will provide capabilities to recognize the therapeutic potential available through targeting the cysteine proteome.

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Highlights

- The cysteine proteome provides an interface for an organism to sense and adapt to its environment, including resources such as food and O₂, and threats, such as infection and toxicants.
- Functional switching of protein structure and reactivity occurs through purposeful sulfenylation, disulfide formation, oxidation to higher oxidation states, S-nitrosylation, persulfidation, metallation, and other modifications
- Systematic mapping of cysteine proteome reactivities and modifications will support integrated systems biology models connecting cell signaling and other complex functions; this mapping will enable new therapeutic approaches targeting redox networks.

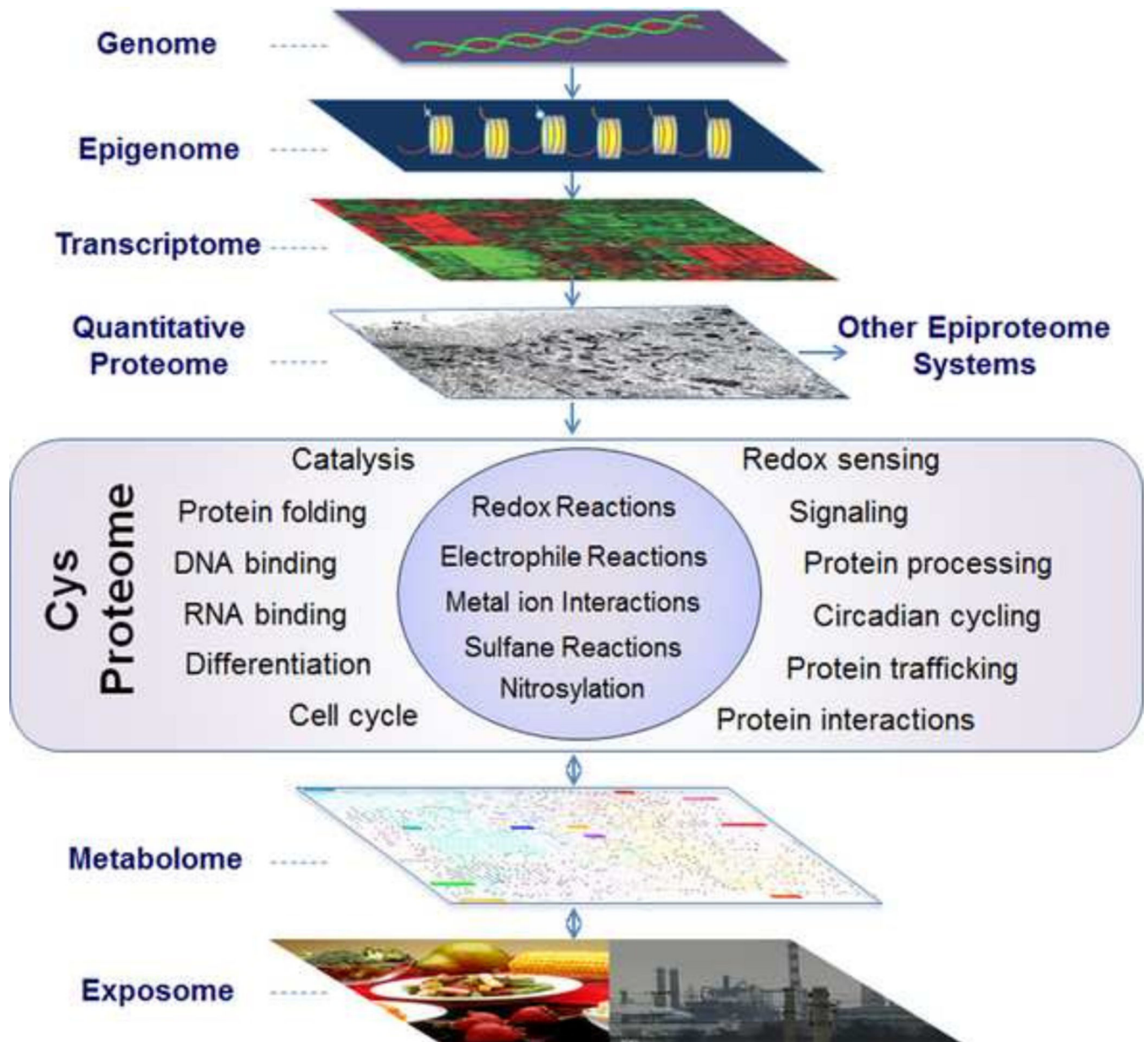


Fig. 1. The Cys proteome exists as an adaptive interface between the exposome and the functional genome. Research on integrated omics has emphasized the role of the metabolome and the proteome as intermediates between environmental exposures and the genome, epigenome and transcriptome. The Cys proteome has a more direct role to support sensing and signaling of nutritional and environmental conditions and also to provide an array of reactive nucleophiles to neutralize toxic electrophilic chemicals. Thus, the Cys proteome is an important functional component of the translated proteome, directly interacting with the metabolome to utilize environmental resources and defend against environmental threats.

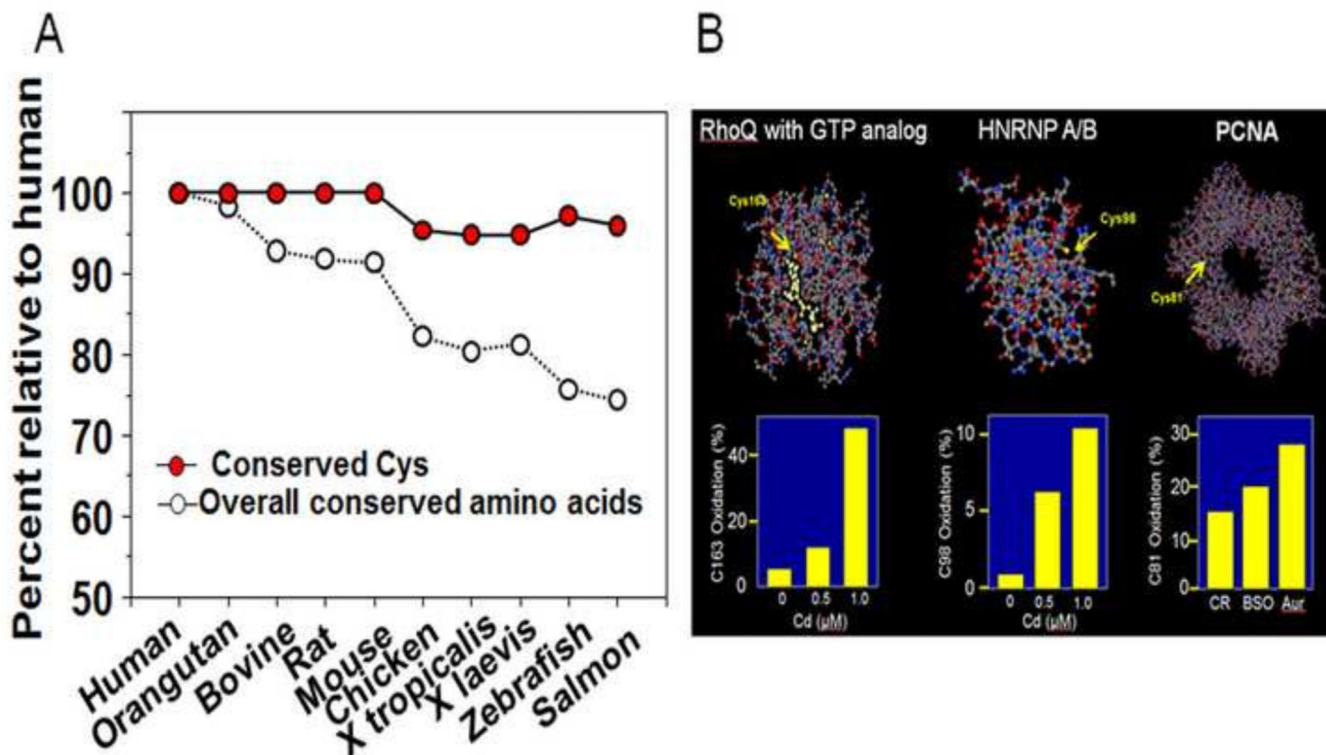


Fig. 2. Multiple lines of evidence support function of redox-sensitive Cys measured by mass spectrometry. **A.** Targeted searches show that Cys detected in redox proteomic analyses are highly conserved across metazoans compared to other amino acids in the same proteins. **B.** Examination of x-ray crystal structures shows that redox-sensitive Cys are commonly present in functional motifs of proteins. Examples shown include specific Cys adjacent to GTP-binding site in RhoQ, Cys in RNA binding site of heteronuclear ribonucleoprotein (HNRNP) A/B and DNA interaction site of PCNA. Lower panels show oxidation in response to challenge with cadmium or inhibition of thioredoxin with auranofin (Aur) or GSH depletion with buthionine sulfoximine (BSO).

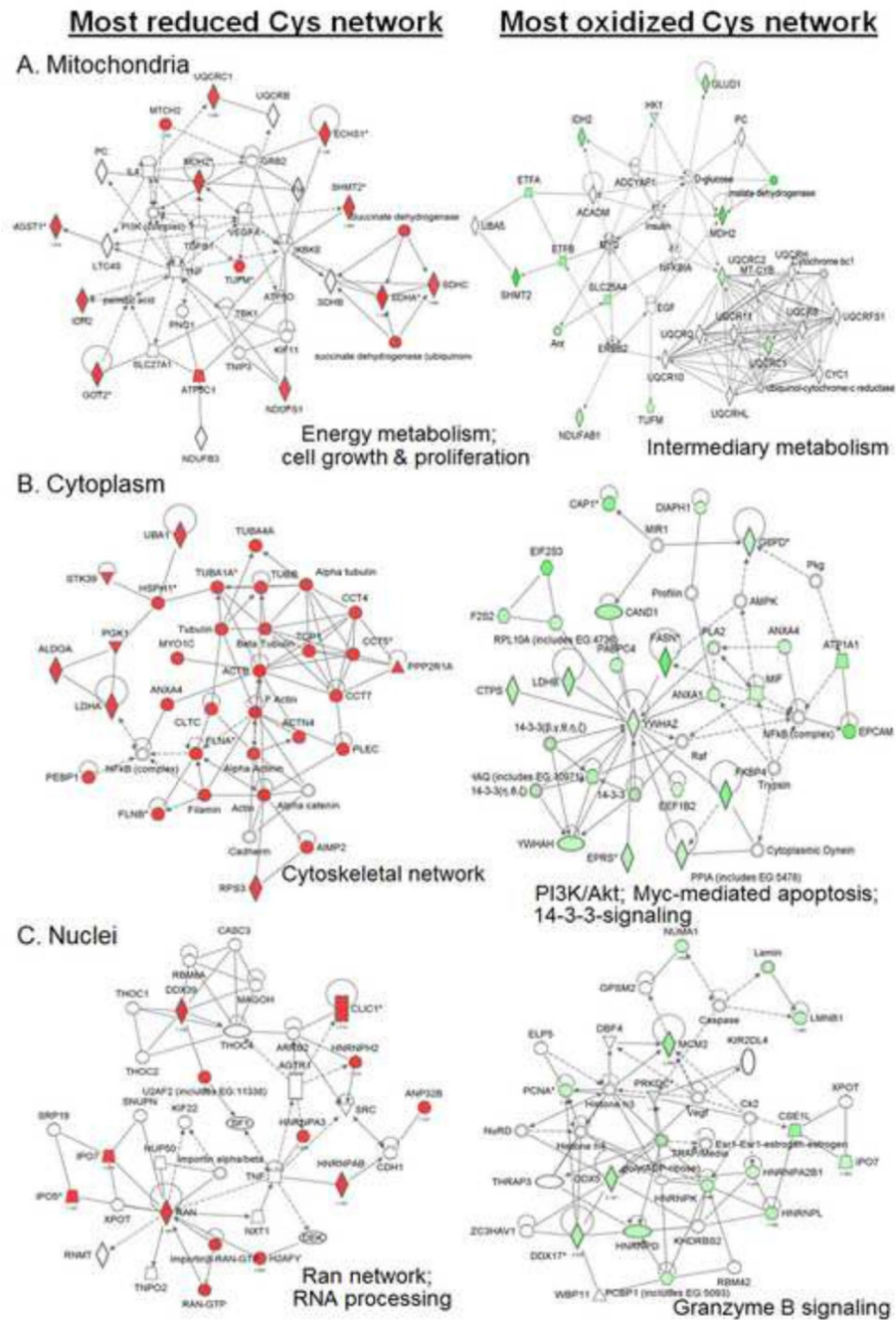


Fig. 3. Protein Cys map to functional pathways based upon percent oxidation. Ingenuity Pathway Analysis of most reduced protein Cys in colon carcinoma HT29 cells under steady-state non-challenged conditions showed associations with mitochondrial pathways for energy metabolism, cell growth and proliferation; cytoplasmic cytoskeletal pathways and nuclear ran-import mechanisms and RNA processing. Most oxidized protein Cys were associated with mitochondrial intermediary metabolism, cytoplasmic cell survival, Myc-mediated apoptosis and 14-3-3 signaling, and nuclear granzyme B signaling. Data from [6, 8, 24, 49].

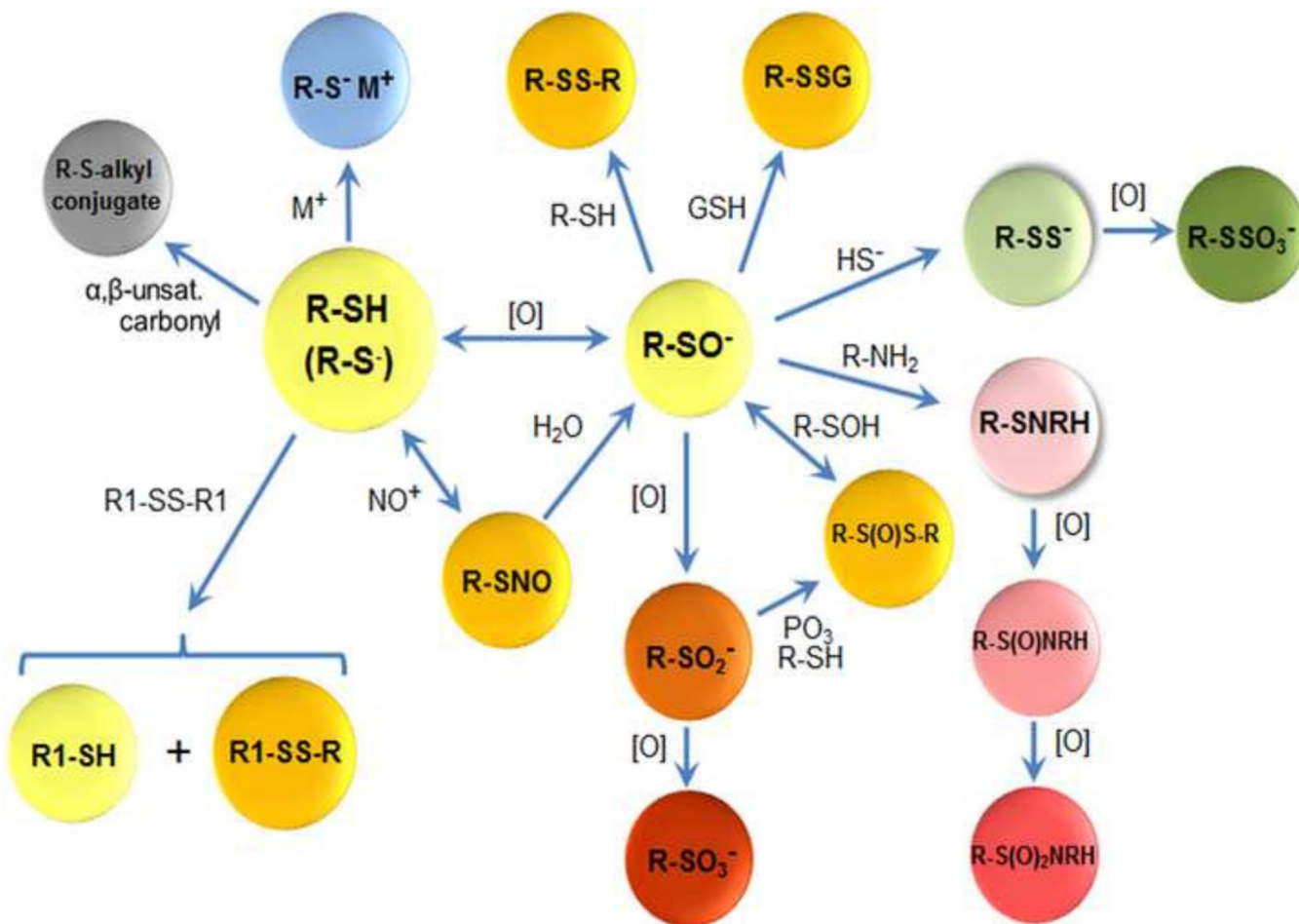


Fig. 4.

Functional modifications of the Cys proteome. Protein Cys (RSH) undergoes a range of enzymatic and non-enzymatic modifications, often mediated by pH-dependent ionization to Cys thiolate (RS^- ; large yellow sphere). Interaction of RS^- with metal ions (blue) is common as a structural element, as a binding motif for nucleic acids and as a component of catalytic sites. RS^- is subject to oxidation ([O]) to Cys sulfenate (RSO^- , yellow) by H_2O_2 and other 2-electron oxidants. Thiol (RSH) also undergoes exchange with disulfide (R_1S-SR_1) to generate different thiol (R_1SH) and disulfide ($RS-SR_1$) in a process termed thiol-disulfide exchange. Non-equilibrium steady-state oxidation of specific RSH occurs due to the presence of low nanomolar concentrations of H_2O_2 in cells. RSO^- reacts with RSH or GSH to produce the respective disulfides (RSSR, RSSG; dark yellow, top). Many RSO^- can undergo hyperoxidation to sulfinic acid (RSO_2^- ; orange, bottom) and sulfonic acid (RSO_3^- ; red) in the presence of excess oxidant. RS^- also reacts with hypothiocyanous acid (HOSCN) to produce sulfenyl thiocyanate (RSSCN; dark yellow, upper left), which rapidly hydrolyzes, and with nitric oxide through multiple means of nitronium ion transfer (NO^+) to produce a corresponding nitrosothiol (RSNO; orange, bottom left). RSO^- can also react with hydrogen sulfide (HS^-) to form a cysteinyl persulfide (RSS^- , green), which can be oxidized to a cysteinyl thiosulfate ($RSSO_3^-$). Alternatively, RSO^- can react with primary amines of

neighboring amino acids or separate biomolecules to form sulfenamide (RSNHR) which can be further oxidized to sulfinamide (RS(O)NHR) and sulfonamide (RS(O)₂NHR), frequently resulting in cyclic intramolecular structures.

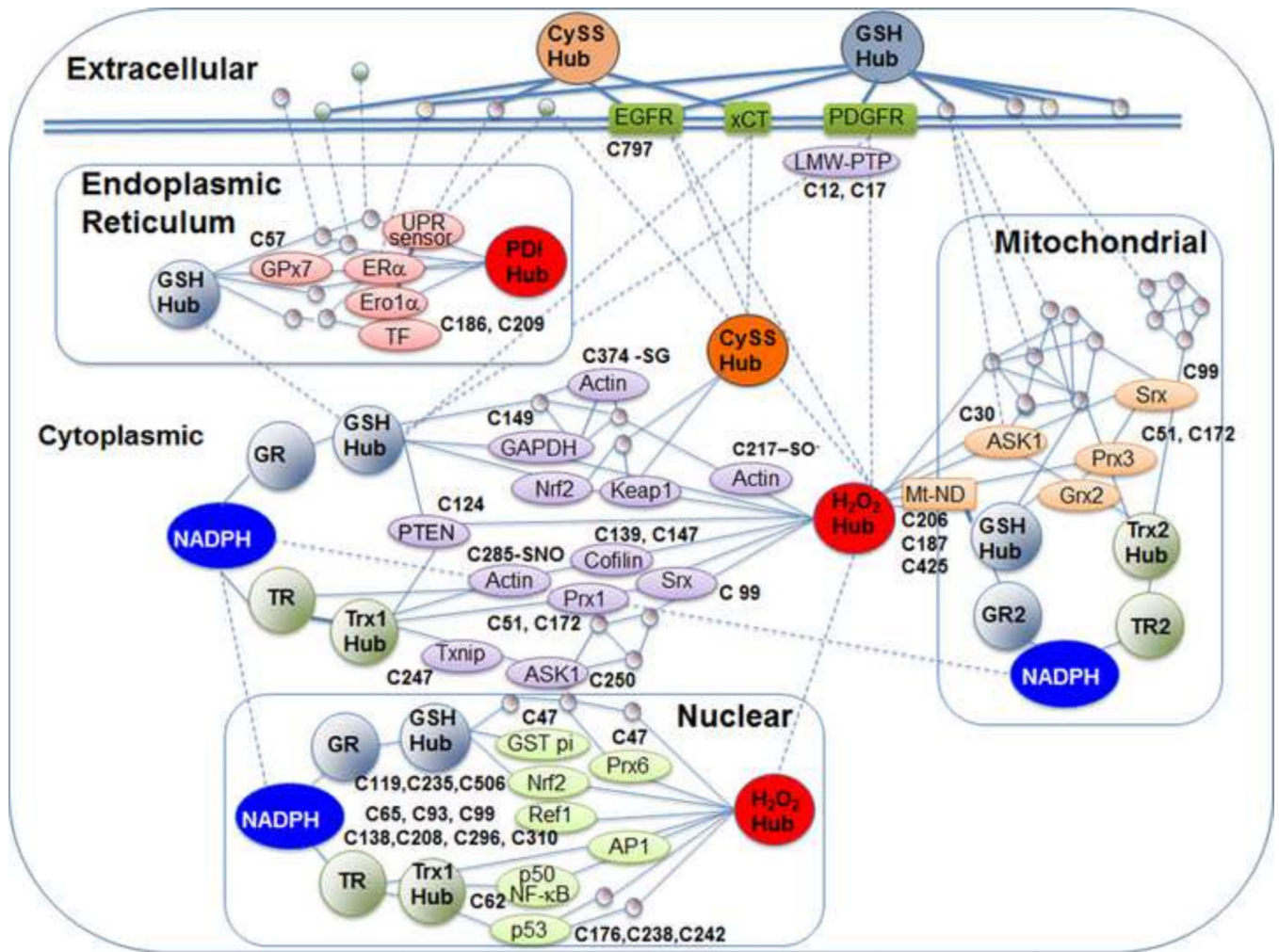


Fig. 5.

Cys proteome networks regulated by Trx and GSH systems. Accumulating evidence shows that redox networks exist in different subcellular compartments supported by NADPH as reductant functioning in opposition to oxidation by H_2O_2 and other oxidants. Trx is the most common intermediary reductant, with a smaller number of proteins supported by GSH. Cysteine and cystine constitute the most abundant extracellular low molecular weight thiol/disulfide couple, apparently functioning in regulation of protein Cys on the extracellular surface. References provide detailed information on interactions of Cys proteins and functional Cys residues illustrated in this figure. TrxR-actin [110]; Trx-dependent regulation of proteins [139]; Ref1-AP1 [258]; TrxR/AP1 [259]; PDGFR [260]; xCT [261]; PTEN-Trx [262, 263]; PTEN-GSH [264]; ASK1-Trx2 [264]; GSTp1-Prx6 [265]; GAPDH-GSH [266, 267]; ND-GSH [268]; p53-Trx [269]; Trx2-Prx3 [270]; Srx-Prx1 and Srx-Prx3 [271, 272]; Grx2-Prx3 [273]; Grx2-Prx3-Trx2 [274]; PDI-UPR sensor [275]; Era-PDI [276]; PDI-TF [277]; Ero- α -PDI and GSH [278]; GPx7-ER [279]; Nrf2 regulatory Cys residues [280]; Srx functional Cys residues [281].

Table 1

Cys in catalytic sites of enzymes. The sulfur of Cys functions in the active site of at least five of the six classes of enzymes classified by Enzyme Commission. These include different types of oxidoreductases linked to the NADP or NAD systems and also the radical-mediated ribonucleotide reduction. Cys functions in transfer of sulfane in some transferases. Cys is present in active sites of hydrolases, such as cysteine proteases and phosphatases. Protein disulfide isomerases (PDI) catalyze rearrangement of thiol and disulfide bonds without any net oxidation or reduction. Some ligases have active site Cys functioning in covalent bond formation.

Function	Enzyme	Substrate(s)	Reference
Oxidoreductase	GR	GSSG	[251]
	Grx	GSS-Cys, Trx-SS	[96, 252]
	GPx	Lipid, inorganic peroxides	[64]
	GST	Electrophiles, lipid peroxides	[253]
	TrxR	Trx-SS	[64]
	Prx	Peroxides	[60]
	GAPDH	G3P	[59]
	ADH	Aldehydes	[93]
	RNR (radical-mediated reduction)	Ribonucleotides	[63, 254]
Transferase	Thiosulfate sulfurtransferase, mercaptopyruvate sulfurtransferase	H ₂ S, S ₂ O ₃ , mercaptopyruvate	[91]
	Thymidylate synthase	dUMP	[255]
Hydrolase	γ-Glu hydrolase	γ-Glu-peptides	[256]
	Caspases, cathepsins, calpains	Polypeptides	[94]
	PTP, Cdc25	pTyr	[176]
Isomerase	PDI	RSSR-x-x-x-RSH	[89]
Ligase	Parkin E3 ubiquitin ligase	Cys-His-Glu	[257]

Hydrogen peroxide-generating human enzymes. Enzymes producing superoxide anion (O_2^-) are included because O_2^- is commonly dismutated to produce H_2O_2 .

Table 2

Name	Protein	Gene	Accession	Location	Product
Aldehyde oxidase	ADO	AOX1	Q06278	C	H_2O_2
Amine oxidase (flavin-containing) A	AOFA	MAOA	P21397	M	H_2O_2
Amine oxidase (flavin-containing) B	AOFB	MAOB	P27338	M	H_2O_2
D-Amino acid oxidase	OXDA	DAO	P14920	Px	H_2O_2
L-Amino acid oxidase	OXLA	IL4II	Q96RQ9	L	H_2O_2
D-Aspartate oxidase	OXDD	DDO	Q99489	Px	H_2O_2
Amiloride-sensitive amino oxidase (copper containing)	ABP1	ABP1	P19801	S	H_2O_2
FAD-linked sulphydryl oxidase	ALR	GFER	P55789	C, M, S	H_2O_2
Hydroxyacid oxidase 1	HAOX1	HAO1	Q9UJIM8	Px	H_2O_2
Hydroxyacid oxidase 2	HAOX2	HAO2	Q9NYQ3	Px	H_2O_2
Membrane primary amine oxidase	AOC3	AOC3	Q16853	PM	H_2O_2
Peroxisomal N(1)-acetyl-spermine/spermidine oxidase	PAOX	PAOX	Q6QHE9	Px, C	H_2O_2
Peroxisomal acyl-coenzyme A oxidase 1	ACOX1	ACOX1	Q15067	Px	H_2O_2
Peroxisomal acyl-coenzyme A oxidase 3	ACOX3	ACOX3	Q15254	Px	H_2O_2
Peroxisomal sarcosine oxidase	SOX	PIPOX	Q9P0Z9	Px	H_2O_2
Prenylcysteine oxidase 1	PCYOX	PCYOX1	Q9UHG3	L	H_2O_2
Protein-lysine 6-oxidase	LYOX	LOX	P28300	S	H_2O_2
Pyridoxine-5-phosphate oxidase	PNPO	PNPO	Q9NVS9	C	H_2O_2
Retina-specific copper amine oxidase	AOC2	AOC2	O75106	PM, C	H_2O_2
Spermine oxidase	SMOX	SMOX	Q9NWM0	C, N	H_2O_2
Sulphydryl oxidase 1	QSOX1	QSOX1	O00391	G	H_2O_2
Sulphydryl oxidase 2	QSOX2	QSOX2	Q6ZRP7	N, PM, S	H_2O_2
Sulfite oxidase, mitochondrial	SUOX	SUOX	P51687	M	H_2O_2

Name	Protein	Gene	Accession	Location	Product
Xanthine dehydrogenase/oxidase	XDH	XDH	P47989	C, Px, S	H ₂ O ₂
NADPH oxidase 1	NOX1	NOX1	Q9Y5S8	PM	O ₂ ⁻
Cytochrome b-245 heavy chain (gp91-phox)	CY24B (NOX2)	CYBB	P04839	PM	O ₂ ⁻
NADPH oxidase 3	NOX3	NOX3	Q9HBY0	PM	O ₂ ⁻
NADPH oxidase 4	NOX4	NOX4	Q9NPH5	E, PM, N	H ₂ O ₂
NADPH oxidase 5	NOX5	NOX5	Q96PH1	E	O ₂ ⁻
Dual oxidase 1	DUOX1	DUOX1	Q9NRD9	PM	H ₂ O ₂
Dual oxidase 2	DUOX2	DUOX2	Q9NRD8	PM	H ₂ O ₂

Abbreviation used for subcellular localization: C, cytoplasm; E, endoplasmic reticulum; G, golgi; L, lysosome; M, mitochondria; N, nucleus; PM, plasma membrane; Px, peroxisome; S, secreted. Other proteins are likely to produce O₂⁻ or H₂O₂ but not included due to limited characterization. Enzymes generating lipid peroxides are not included.