

# Crystal Structure and Substrate Recognition of Cellobionic Acid Phosphorylase, Which Plays a Key Role in Oxidative Cellulose Degradation by Microbes\*

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**Background:** Cellobionic acid phosphorylase (CBAP) catalyzes the reversible phosphorolysis of cellobionic acid into glucose 1-phosphate and gluconic acid.

**Results:** Crystal structures of CBAP complexed with various ligands were determined.

**Conclusion:** CBAP has a unique substrate recognition site for aldonic acids that contains positively charged residues.

**Significance:** This study provided the first insight into the mechanism of sugar catabolism after oxidative cellulose degradation.

The microbial oxidative cellulose degradation system is attracting significant research attention after the recent discovery of lytic polysaccharide mono-oxygenases. A primary product of the oxidative and hydrolytic cellulose degradation system is cellobionic acid (CbA), the aldonic acid form of cellobiose. We previously demonstrated that the intracellular enzyme belonging to glycoside hydrolase family 94 from cellulolytic fungus and bacterium is cellobionic acid phosphorylase (CBAP), which catalyzes reversible phosphorolysis of CbA into glucose 1-phosphate and gluconic acid (GlcA). In this report, we describe the biochemical characterization and the three-dimensional structure of CBAP from the marine cellulolytic bacterium *Saccharophagus degradans*. Structures of ligand-free and complex forms with CbA, GlcA, and a synthetic disaccharide product from glucuronic acid were determined at resolutions of up to 1.6 Å. The active site is located near the dimer interface. At subsite +1, the carboxylate group of GlcA and CbA is recognized by Arg-609 and Lys-613. Additionally, one residue from the neighboring protomer (Gln-190) is involved in the carboxylate recognition of GlcA. A mutational analysis indicated that these residues are critical for the binding and catalysis of the aldonic and uronic acid acceptors GlcA and glucuronic acid. Structural and sequence comparisons with other glycoside hydrolase family 94 phosphorylases revealed that CBAPs have a unique subsite +1

with a distinct amino acid residue conservation pattern at this site. This study provides molecular insight into the energetically efficient metabolic pathway of oxidized sugars that links the oxidative cellulolytic pathway to the glycolytic and pentose phosphate pathways in cellulolytic microbes.

The establishment of cost-efficient degradation systems of cellulosic biomass is a challenging task as cellulose is the most abundant renewable biopolymer on Earth; however, it is intrinsically recalcitrant (1, 2). Hydrolytic cellulase systems involving cellobiohydrolases, endoglucanases, and  $\beta$ -glucosidases have been extensively studied. The recent discovery of lytic polysaccharide mono-oxygenases (3–5), which cleave glycosidic bonds by oxidation at C1 and/or C4 positions to form aldono-lactone or 4-ketoaldose, shifted the paradigm of research on cellulases and chitinases toward oxidative degradation mechanisms (6–9). Cellobiose dehydrogenase has also been demonstrated to catalyze the oxidation of cellobiose at the C1 position to produce cellobiono-1,5-lactone (CbL)<sup>3</sup> (10–12). Extracellular  $\beta$ -glucosidases do not efficiently hydrolyze CbL due to their narrow substrate preference and strong product inhibition (13, 14). Therefore, CbL is a major end product by a combination of the extracellular oxidative (lytic polysaccharide mono-oxygenase and cellobiose dehydrogenase) and hydrolytic (e.g. cellobiohydrolase and endoglucanase) enzymes (15). In our previous study, a novel enzyme, cellobionic acid phosphorylase (CBAP; EC 2.4.1.321), was discovered from two cellulolytic microbes, the plant pathogenic bacterium *Xanthomonas campestris*

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The atomic coordinates and structure factors (codes 4ZLE, 4ZLF, 4ZLG, and 4ZLI) have been deposited in the Protein Data Bank (<http://www.pdb.org/>).

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<sup>3</sup> The abbreviations used are: CbL, cellobiono-1,5-lactone; AA, auxiliary activity; CbA, cellobionic acid; CBAP, cellobionic acid phosphorylase; CBP, cellobiose phosphorylase; CgCBP, *C. gilvus* CBP; ChBP, *N,N'*-diacetylchitobiose phosphorylase; G1P,  $\alpha$ -D-glucose 1-phosphate; GcL, D-glucono-1,5-lactone; GH, glycoside hydrolase; GlcA, D-gluconic acid; GlcUA, D-glucuronic acid; Glc- $\beta$ 1,3-GlcUA, 3-O- $\beta$ -D-glucopyranosyl-D-glucuronic acid; NcCBAP, *N. crassa* CBAP or NCU09425; SdCBAP, *S. degradans* CBAP or Sde\_0906; VpChBP, *V. proteolyticus* ChBP; XcCBAP, *X. campestris* CBAP or XCC4077; BisTris, 2-[bis(2-hydroxyethyl)amino]-2-(hydroxymethyl)propane-1,3-diol.

## Structure of Cellobionic Acid Phosphorylase

(XCC4077; *XcCBAP*) and the red bread mold *Neurospora crassa* (NCU09425; *NcCBAP*) of the phylum Ascomycota (16). CbL is spontaneously hydrolyzed to form cellobionic acid (Glc- $\beta$ 1,4-gluconic acid; CbA), and CBAP catalyzes phosphorylation of this compound to produce  $\alpha$ -D-glucose 1-phosphate (G1P) and D-gluconic acid (GlcA). The function of CBAP was also confirmed *in vivo*. Deletion of the *NcCBAP* gene (*ndvB*) in *N. crassa* resulted in the accumulation of CbA during cultivation on cellulose (17), and the *NcCBAP* gene complemented the *ascB* gene in *Escherichia coli* that is responsible for growth on CbA (18). The reaction of CBAP is reversible, and the synthetic reaction using G1P and GlcA as the donor and acceptor substrates produces CbA and inorganic phosphate ( $P_i$ ). In the synthetic reaction, *XcCBAP* can also utilize D-glucuronic acid (GlcUA) as the acceptor. The reaction with GlcUA exclusively produced a  $\beta$ 1,3-linked disaccharide 3-O- $\beta$ -D-glucopyranosyl-D-glucuronic acid (Glc- $\beta$ 1,3-GlcUA) (16).

In the Carbohydrate-Active enZYme (CAZy) database (19), CBAP belongs to glycoside hydrolase (GH) family 94, which mainly consists of inverting phosphorylases acting on  $\beta$ -linked di- or oligosaccharides such as cellobiose phosphorylase (CBP; EC 2.4.1.20), cellodextrin phosphorylase (EC 2.4.1.49), laminaribiose phosphorylase (EC 2.4.1.31), and *N,N'*-diacetylchitobiose phosphorylase (ChBP; EC 2.4.1.280). Within this family, crystal structures of CBPs from *Cellvibrio gilvus* (*CgCBP*) (20), *Cellulomonas uda* (21), and *Clostridium thermocellum* (22) and ChBP from *Vibrio proteolyticus* (*VpChBP*) (23) have been determined. However, a phylogenetic analysis of GH94 phosphorylases indicated that CABP is distantly related to members that are characterized and have known structures with very low amino acid sequence identity (21–24%) (16). In this study, we characterized another CBAP enzyme (*Sde\_0906*; *SdCBAP*) from the genome of the marine bacterium *Saccharophagus degradans* that can degrade various polysaccharides, including cellulose (24). Biochemical and structural characterization of *SdCBAP* revealed the detailed molecular mechanism of the key metabolic enzyme that connects the oxidative cellulose degradation and downstream pathways.

### Experimental Procedures

**Chemicals**—G1P (disodium salt hydrate), GlcA (sodium salt), and GlcUA (sodium salt) were purchased from Sigma-Aldrich, Nacalai Tesque (Kyoto, Japan), and Wako Pure Chemicals (Osaka, Japan), respectively. CbA and Glc- $\beta$ 1,3-GlcUA were produced by the synthetic reaction of *XcCBAP* and purified as described previously (16).

**Cloning, Expression, and Purification**—A gene encoding *Sde\_0906* (GenBank<sup>TM</sup> accession number AAM43298.1) was amplified by PCR from genomic DNA of *S. degradans* 2-40 using KOD-plus DNA polymerase (Toyobo, Osaka, Japan) with the following oligonucleotides based on the genome sequence (GenBank accession number CP000282) (25): 5'-aaaccatgggct-taaaagccattaacaac-3' and 5'-tttctcgaggtgtgtggcaggtaatag-3' (restriction enzyme sites underlined). The amplified gene was purified using a FastGene Gel/PCR Extraction kit (Nippon Genetics Co., Tokyo, Japan), digested by *Nco*I and *Xho*I (New England Biolabs, Beverly, MA), and inserted into pET28a(+) (Novagen, Madison, WI) to encode a His<sub>6</sub> tag fusion at the C

terminus of the recombinant protein. The expression plasmid was introduced into *E. coli* DH5 $\alpha$  (Toyobo) and verified by sequencing (Operon Biotechnologies, Tokyo, Japan). For native and selenomethionine (SeMet)-labeled protein expression, the plasmid was introduced into *E. coli* BL21 CodonPlus (DE3)-RIL (Stratagene, La Jolla, CA) and *E. coli* B834 (DE3) strains, respectively. The transformants were cultured in Luria-Bertani medium (native protein) or LeMaster medium (SeMet-labeled protein) containing 100 mg/liter kanamycin at 37 °C until the absorbance reached 0.6 at 600 nm. The protein expression was induced by 0.1 mM isopropyl  $\beta$ -D-thiogalactopyranoside and continued at 25 °C for 18 h. The cell-free extract was first purified by nickel affinity chromatography (HisTrap FF crude, GE Healthcare), and the enzyme was eluted with a stepwise increase in the imidazole concentration (20 and 250 mM) in 50 mM HEPES-NaOH (pH 7.0). Next, the enzyme was further purified on a Mono Q column (GE Healthcare) with an increasing linear gradient of 0–100% saturation of NaCl using ÄKTA (GE Healthcare). The protein concentration was determined spectrophotometrically at 280 nm using a theoretical extinction coefficient of  $\epsilon = 146,110 \text{ M}^{-1} \text{ cm}^{-1}$  based on the amino acid sequence. The molecular mass of purified *SdCBAP* was estimated by SDS-PAGE and by gel filtration column chromatography using HiLoad 16/60 Superdex 200 prep grade (GE Healthcare) equilibrated with 50 mM HEPES-NaOH (pH 7.0) and 150 mM NaCl at a flow rate of 1.0 ml/min. A Gel Filtration Markers kit for molecular weights (Sigma-Aldrich) was used for standards.

**Measurement of Enzymatic Activity**—The catalytic activities of *SdCBAP* were measured as described previously (16). The phosphorylase activity was determined by quantifying the G1P released during the reaction in 40 mM MES-NaOH (pH 6.5) containing 10 mM CbA and 10 mM  $P_i$  at 30 °C using the phosphoglucomutase-glucose 6-phosphate dehydrogenase method. The synthetic activity was determined by measuring the increase in  $P_i$  using a reaction mixture containing 10 mM G1P and 10 mM GlcA in 40 mM MES-NaOH (pH 6.5) at 30 °C.

**Structural Determination of Reaction Products**—Structures of reaction products were determined by NMR spectra as described previously (16). Reaction products were generated in 1 ml of reaction mixture (pH 6.5) containing 450 mM G1P and 450 mM GlcA or GlcUA with 22  $\mu\text{M}$  *SdCBAP* at 30 °C for 24 h. After treatment with  $\alpha$ -D-glucose-1-phosphatase from *E. coli* at 30 °C for 24 h, the reaction products were separated on a Toyoparl HW-40S column and then lyophilized. The amount of product obtained was 43 and 42 mg from GlcA or GlcUA, respectively. The structures of the products were identified by comparing their <sup>1</sup>H and <sup>13</sup>C NMR spectra acquired in D<sub>2</sub>O with those of authentic data (16) using a Bruker Avance 500 or Bruker Avance 800 spectrometer (Bruker Biospin, Rheinstetten, Germany).

**Kinetic Analysis**—The initial velocities of the phosphorylase reactions were determined under standard conditions with *SdCBAP* (7 nM–17  $\mu\text{M}$  for the wild-type and mutant enzymes) and a combination of initial concentrations of cellobionic acid (0.5–3.0 mM) and  $P_i$  (0.1–2.0 mM). The kinetic parameters were calculated by curve fitting the experimental data to the theoretical Equation 1 for a sequential bi-bi mechanism using GraFit version 7.0.2 (Erithacus Software Ltd., London, UK).

$$v = V_{\max}[A][B]/(K_{iA}K_{mB} + K_{mB}[A] + K_{mA}[B] + [A][B])$$

• (A = CbA, B = P<sub>i</sub>) (Eq. 1)

Kinetic analysis of the synthetic reactions was performed under the standard conditions with *SdCBAP* (9.0 nM–17 μM for the wild-type and mutant enzymes) and various concentrations of GlcA or GlcUA (0.5–20 mM) as the acceptor or G1P (0.3–5.0 mM) as the donor with 10 mM for each opposite substrate. The kinetic parameters were calculated by curve fitting the experimental data to the Michaelis-Menten equation  $v = k_{\text{cat}} [E]_0 [S]/(K_m + [S])$  using GraFit version 7.0.2.

**Temperature and pH Profiles**—The effects of pH on the phosphorolytic and synthetic activities using 20 nM *SdCBAP* were measured under the standard conditions described above using the following 40 mM buffers: sodium citrate (pH 3.0–5.5), BisTris-HCl (pH 5.5–7.0), HEPES-NaOH (pH 7.0–8.5), and glycine-NaOH (pH 8.5–10.5). The thermal and pH stabilities were evaluated by measuring the residual synthetic activity under the standard conditions after incubation of 9.9 μM *SdCBAP* at a temperature range of 30 to 90 °C for 15 min and at various pH values at 4 °C for 24 h, respectively.

**Site-directed Mutagenesis**—Q190A, R609A, and K613A mutants were constructed using a PrimeSTAR Mutagenesis basal kit (Takara) in accordance with the manufacturer's protocol. The following primers and their complementary strands were used (mutation sites are underlined): 5'-cgccttatgctaaagt-cgccgactacttt-3' (Q190A), 5'-gacgtaggcgcagttaccctcaaaattccca-3' (R609A), and 5'-accaagcattcccaggctctgcagaa-3' (K613A). The mutations were confirmed by DNA sequencing. The mutant enzymes were expressed and purified using the same procedure as described above.

**Crystallography**—All crystals were obtained at 25 °C using the sitting drop vapor diffusion method by mixing 1 μl of protein solution containing 15–30 mg/ml protein with an equal volume of reservoir solution containing 0.1 M sodium citrate (pH 5.6–6.0), 0.1 M Li<sub>2</sub>SO<sub>4</sub>, 0.6 M (NH<sub>3</sub>)<sub>2</sub>SO<sub>4</sub>, and 5% (v/v) glycerol. Crystals completely grew in 3 weeks. Crystals were cryoprotected in the reservoir solutions supplemented with 20% (w/v) glycerol and flash cooled at 100 K in a stream of nitrogen gas. Each complex crystal was obtained by soaking in the cryoprotectant solution containing a ligand (50 mM GlcA, 50 mM CbA, or 20 mM Glc-β1,3-GlcUA) for 2 min. Diffraction data were collected using a charge-coupled device camera on beamlines BL17A (Photon Factory) and NW12A (Photon Factory-Advanced Ring) at the High Energy Accelerator Research Organization (KEK, Tsukuba, Japan) and processed using HKL2000 (26). The initial phase calculation, phase improvement, and automated model building were performed using PHENIX (27). Manual model rebuilding and refinement were achieved using Coot (28) and Refmac5 (29). Initial coordinates and parameters of CbA and Glc-β1,3-GlcUA molecules were built with the JLigand program (30). The molecular interfaces were analyzed using the PDBe PISA server (31). Molecular graphic images were prepared using PyMOL (Schrödinger, LLC, New York, NY).

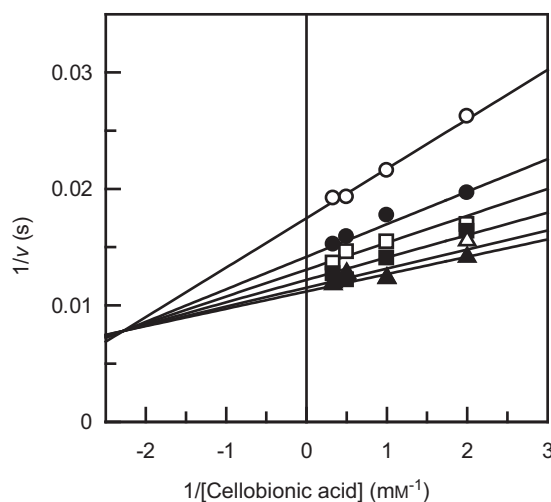


FIGURE 1. Double reciprocal plots of the phosphorolysis of cellobionic acid catalyzed by *SdCBAP*. The initial velocities of the phosphorolytic reaction were determined with a combination of initial concentrations of CbA (0.5, 1.0, 2.0, and 3.0 mM) and P<sub>i</sub> (open circles, 0.1 mM; filled circles, 0.2 mM; open squares, 0.3 mM; filled squares, 0.5 mM; open triangles, 1.0 mM; filled triangles, 2.0 mM).

## Results and Discussion

**Biochemical Characterization of *Sde\_0906* (*SdCBAP*)**—Before our identification of CBAPs from *X. campestris* and *N. crassa*, the *sde\_0906* (*cep94B*) gene of *S. degradans* was predicted as cytoplasmic cellobiose/cellobiose phosphorylase (25). A previous study on recombinant *Sde\_0906* protein (*Cep94B*) showed that the gene product did not exhibit phosphorylase activity on cellobiose or cellobiose (32). Because we noticed that *Sde\_0906* is closely related to *XcCBAP* and *NcCBAP* with high amino acid sequence identities (60.4 and 61.9%, respectively), we prepared its purified recombinant protein and measured the CBAP activities. We subsequently designated the enzyme as *SdCBAP*. The molecular masses of *SdCBAP* as deduced from the amino acid sequence and estimated by SDS-PAGE and calibrated gel filtration chromatography were 88.7, 87, and 147 kDa, respectively, suggesting that it is dimeric in solution. The phosphorolytic reaction followed a sequential bi-bi mechanism, and the kinetic parameters were  $k_{\text{cat}} = 92 \pm 2 \text{ s}^{-1}$ ,  $K_{mA} = 0.12 \pm 0.03 \text{ mM}$ ,  $K_{iA} = 0.44 \pm 0.2 \text{ mM}$ , and  $K_{mB} = 0.061 \pm 0.009 \text{ mM}$  (Fig. 1). The kinetic parameters of the synthetic reactions using GlcA and GlcUA as the acceptor were also determined (Table 1). *SdCBAP* exhibited significant activity to GlcUA, although it was 9-fold lower than that of GlcA due to decreased  $k_{\text{cat}}$  and increased  $K_m$  values. Reaction products synthesized from GlcA and GlcUA were determined by NMR spectra measurements, confirming that they were CbA and Glc-β1,3-GlcUA, respectively, and no other products were detected (data not shown). The  $K_m$  value for G1P was  $0.41 \pm 0.1 \text{ mM}$  when GlcA was used as the acceptor. *SdCBAP* was stable up to 35 °C during 15 min of incubation and in the pH range of 5.5–10.5. The optimum pH for the synthetic reaction was pH 6.5–7.0 (data not shown). The characteristics of *SdCBAP* were similar to those of *XcCBAP* and *NcCBAP* (16).

**Crystal Structure of *SdCBAP***—The crystal structure of *SdCBAP* was solved by the single wavelength anomalous dispersion method using a SeMet derivative. A ligand-free structure and complex structures with GlcA, CbA, and Glc-β1,3-



TABLE 1

Specific activities and kinetic parameters for the synthetic and phosphorolytic reactions of wild-type *SdCBAP* and its mutants

Enzyme	Synthesis acceptor								Phosphorolysis substrate CbA specific activity <sup>a</sup>
	GlcA				GlcUA				
	Specific activity <sup>a</sup> units/mg	$k_{\text{cat}}$ $s^{-1}$	$K_m$ mM	$k_{\text{cat}}/K_m$ $s^{-1}mM^{-1}$	Specific activity <sup>a</sup> units/mg	$k_{\text{cat}}$ $s^{-1}$	$K_m$ mM	$k_{\text{cat}}/K_m$ $s^{-1}mM^{-1}$	
WT	93.2	130 ± 6	1.7 ± 0.2	76	10	34.6 ± 0.7	13.4 ± 0.5	2.58	62
Q190A	3.3	24 ± 3	41 ± 7	0.58	1.1	ND <sup>b</sup>	ND	0.16	9.04
R609A	<0.005	ND	ND	ND	<0.001	ND	ND	ND	<0.001
K613A	<0.01	ND	ND	ND	<0.01	ND	ND	ND	<0.002

<sup>a</sup> The specific activities were measured at 10 mM substrates.<sup>b</sup> ND, not determined.

GlcUA were determined (Table 2). The crystals contained one molecule in the asymmetric unit. The monomer structure consists of four distinct parts (Fig. 2A): an N-terminal  $\beta$  sandwich domain (residues 1–277; *blue*), a helical linker region (278–308; *green*), a catalytic ( $\alpha/\alpha$ )<sub>6</sub> barrel domain (319–709; *yellow*), and a  $\beta$  sheet domain (*red*). The  $\beta$  sheet domain consists of a middle segment (residues 309–318) and the C-terminal segment (710–785). The protomers in the asymmetric units are related by 2-fold crystallographic symmetry to form a dimer (Fig. 2B), which corresponds to the state in solution as confirmed by the biochemical analysis described above. The dimer is tightly connected with an interface area of 2,700 Å<sup>2</sup>. There are 44 hydrogen bonds and 18 salt bridges, and the estimated  $\Delta^iG$  value is –13.2 kcal/mol. The major contact area at the dimer interface comprises the N-terminal and catalytic barrel domains. The overall monomer and dimer structure of *SdCBAP* was basically similar to those of GH94 CgCBP and VpChBP (20, 23) with some variations at peripheral loop regions. The most prominent difference from other GH94 structures is the presence of a long loop region with a short  $\alpha$  helix (186–203; Fig. 2, *magenta*) in the N-terminal domain. This region partly forms the active site in the neighboring protomer (described below).

**Active Site Structure**—Because the crystals grew in high concentrations of ammonium sulfate, all of the crystal structures contained a sulfate ion (SO<sub>4</sub><sup>2-</sup>) at the P<sub>i</sub> binding site formed by Arg-341, Tyr-624, His-626, Thr-694, and Gly-695 as shown in the ligand-free structure (Fig. 3A). These residues at the P<sub>i</sub> binding site, except for Tyr-624, are basically conserved in GH94 phosphorylases (Fig. 4). In the ligand-free structure, a glycerol and an additional sulfate are bound at subsites –1 (the donor glucoside site) and +1 (the acceptor site), respectively. The sulfate at subsite +1 is held by Arg-609, Lys-613, and Gln-190' (the prime mark indicate that it is a residue from the neighboring molecule). Gln-190' is located in the prominent 186–203 loop of the N-terminal domain.

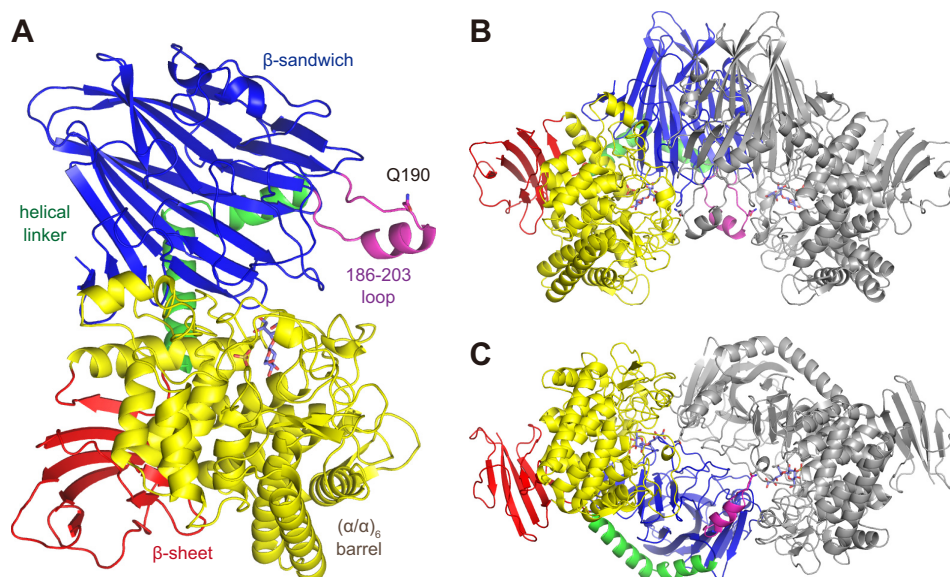
In the CbA complex, a clear electron density of the bound CbA was observed (Fig. 3B). The glucose moiety at subsite –1 is in a standard <sup>4</sup>C<sub>1</sub> conformation with its C6 hydroxyl group in a *gauche-gauche* conformation. At this site, the side chains of Gln-347, Arg-349, Asn-350, and Glu-619 and the main chain of Trp-470 form hydrogen bonds to the hydroxyl groups of the glucopyranoside sugar. The Asp-472 residue of *SdCBAP* corresponds to the general acid residue Asp-490 of CgCBP and Asp-492 of VpChBP (20, 23). Asp-472 is located near the glycosidic bond oxygen atom (3.6 Å) so that it can interact with the glyco-

sidic oxygen by a slight movement of the side chain rotamer. These residues at subsite –1 are also highly conserved in GH94 phosphorylases (Fig. 4). The sulfate ion at the P<sub>i</sub> binding site is located near the anomeric C1 atom of the glucose moiety (3.2 Å). Thus, the P<sub>i</sub> binding site is suitable for the direct nucleophilic attack of the inverting phosphorylase reaction mechanism (23). The gluconate moiety of CbA is bound at subsite +1, and its carboxylate group forms salt bridges with Arg-609 and Lys-613. Gln-190' is located at a relatively far position (4.5 Å) in this structure. Hydroxyl groups of the gluconate moiety are recognized by hydrogen bonds from the *SdCBAP* protein. The C2 and C3 hydroxyl groups form hydrogen bonds with Lys-613, Glu-619, and Asp-472, and the C5 hydroxyl group forms hydrogen bonds with Tyr-624 and Asn-672.

In the GlcA complex, the electron density of GlcA (linear form) was observed at subsite +1, whereas a pyranose sugar molecule was found at subsite –1 (Fig. 3C). Because bonds around the C1 anomeric carbon atom appeared flat, we could confidently place a model of D-glucono-1,5-lactone (GcL) in the electron density map without any skewing of the atoms. It is unclear why a GcL molecule was observed in the complex structure with GlcA. Because subsite –1 appears to be highly suitable for binding of a glucopyranose-like molecule, we presume that the high concentration of GlcA in the soaking solution (50 mM) may have caused occupation of its dehydrated compound (GcL) in this site even though the lactone is not dominant in aqueous solution after the equilibrium (33). At subsite +1, GlcA is bound at a similar but slightly different position from the gluconate moiety of CbA. The carboxylate group was recognized by Arg-609, Lys-613, and Gln-190'. The C2, C3, and C5 hydroxyls form hydrogen bonds to Lys-613, Asp-472, and Gln-347, respectively. Fig. 5A shows a superimposition of the CbA (*yellow*) and GlcA (*cyan*) complex structures. The glucopyranose at subsite –1 almost completely overlaps. At subsite +1, the carboxylate groups of CbA and GlcA are located in a position near Arg-609, Lys-613, and Gln-190', indicating that these residues are critical for recognition of the aldonic acid substrates. The remaining part of the GlcA (C2–C6) is located in a significantly displaced position from CbA. Because the C4 hydroxyl group of GlcA is located far from subsite –1, this binding mode does not perfectly represent the acceptor molecule in the synthetic reaction. This is probably because of the steric hindrance with the C1 ketone group of GcL occupying subsite –1. In the synthetic reaction with G1P, the acceptor GlcA molecule is assumed to bind as in the gluconate moiety in the CbA complex structure.

**TABLE 2**  
Data collection and refinement statistics

	Peak	Ligand-free	CbA	GlcA	Glc- $\beta$ 1,3-GlcUA
<b>Data collection</b>					
Protein Data Bank code		4ZLE	4ZLF	4ZLG	4ZLI
Beamline	BL17A	BL17A	NW12A	BL17A	BL17A
Wavelength (Å)	0.9790	0.9988	1.0000	0.9702	0.9732
Space group	$P3_121$	$P3_121$	$P3_121$	$P3_121$	$P3_121$
Unit cell (Å)	$a = b = 107.2,$ $c = 186.6$	$a = b = 106.9,$ $c = 185.3$	$a = b = 107.2,$ $c = 186.8$	$a = b = 107.0,$ $c = 186.5$	$a = b = 107.1,$ $c = 185.8$
Resolution (Å) <sup>a</sup>	50.00–2.40 (2.44–2.40)	50.00–2.10 (2.14–2.10)	50.00–1.60 (1.63–1.60)	50.00–1.75 (1.78–1.75)	50.00–1.80 (1.83–1.80)
Total reflections	2,184,371	522,451	1,291,267	642,592	830,557
Unique reflections	49,159	72,225	163,884	124,297	114,626
Completeness (%) <sup>a</sup>	98.6 (98.7)	99.5 (100.0)	100.0 (100.0)	98.2 (100.0)	99.5 (100.0)
Redundancy <sup>a</sup>	23.3 (23.3)	7.3 (7.2)	7.9 (7.6)	5.3 (5.4)	7.3 (7.4)
Mean $I/\sigma(I)$ <sup>a</sup>	38.2 (6.8)	17.6 (2.6)	37.3 (2.9)	20.8 (2.1)	24.6 (4.1)
$R_{\text{merge}}$ (%) <sup>a</sup>	10.9 (50.7)	11.2 (73.0)	4.3 (41.1)	6.4 (46.8)	9.2 (54.9)
<b>Refinement</b>					
Resolution (Å)		41.44–2.10	31.14–1.60	41.66–1.75	40.50–1.80
No. of reflections		63,592	155,047	115,743	108,205
$R/R_{\text{free}}$ (%)		17.0/21.9	15.6/17.7	16.9/20.1	15.5/18.8
No. of atoms		6,750	7,118	6,964	6,932
No. of solvents		477 (water), 3 (glycerol), 6 (SO <sub>4</sub> ), 1 (Cl <sup>-</sup> )	809 (water), 1 (CbA), 8 (glycerol), 2 (SO <sub>4</sub> ), 1 (Cl <sup>-</sup> )	688 (water), 1 (GlcA), 1 (GcL), 2 (glycerol), 2 (SO <sub>4</sub> ), 1 (Cl <sup>-</sup> )	629 (water), 1 (Glc- $\beta$ 1,3-GlcUA), 3 (glycerol), 3 (SO <sub>4</sub> ), 1 (Cl <sup>-</sup> )
r.m.s.d. <sup>b</sup> from ideal values					
Bond lengths (Å)		0.020	0.031	0.025	0.027
Bond angles (°)		1.92	2.56	2.09	2.23
Ramachandran plot (%)					
Favored		97.4	98.2	98.2	98.3
Allowed		2.3	1.7	1.5	1.4
Outlier		0.3	0.1	0.3	0.3

<sup>a</sup> Values in parentheses correspond to the highest resolution shell.<sup>b</sup> Root mean square deviation.

**FIGURE 2. Overall structure of SdCBAP shown as monomer (A) and dimer (B and C).** The CbA complex structure is shown and colored by domains: N-terminal  $\beta$  sandwich domain (residues 1–277; blue), helical linker region (278–308; green), a catalytic  $(\alpha/\alpha)_6$  barrel domain (319–709; yellow), and a C-terminal  $\beta$  sheet domain (309–318 and 710–785; red). A prominent loop (186–203) in the N-terminal domain is shown in magenta. CbA (slate blue) and sulfate (yellow) are shown as sticks. In B and C, one monomer is shown in gray, and views from two different orientations (90° rotation around a horizontal axis) are shown.

We also determined the complex structure with Glc- $\beta$ 1,3-GlcUA, which is a product of the synthetic reaction using GlcUA as the acceptor (Fig. 3D). In the electron density map, the glucuronate moiety was observed in a pyranose (ring) form. A superimposition of complex structures with CbA (yellow) and Glc- $\beta$ 1,3-GlcUA (magenta) (Fig. 5B) indicates that the non-reducing end glucose moiety is bound in a similar manner. In the Glc- $\beta$ 1,3-GlcUA complex, however, an additional sulfate is bound at the carboxylate binding pocket in subsite +1, and the glucuronate moiety is significantly displaced from the canonical subsite +1. The carboxylate group forms a water-

mediated hydrogen bond with a glycerol molecule, which is derived from the cryoprotectant. Arg-341, Asn-672, and Asn-693 are involved in the recognition of the GlcUA moiety. Therefore, from this complex structure, Arg-609, Lys-613, and Gln-190' appear not to recognize the carboxylate group of Glc- $\beta$ 1,3-GlcUA.

**Mutational Analysis**—We constructed site-directed mutants of the three residues at the carboxylate binding site in subsite +1 (Q190A, R609A, and K613A) and measured their synthetic (using either GlcA or GlcUA as the acceptor) and phosphorolytic activities. As shown in Table 1, the activities of the mutants

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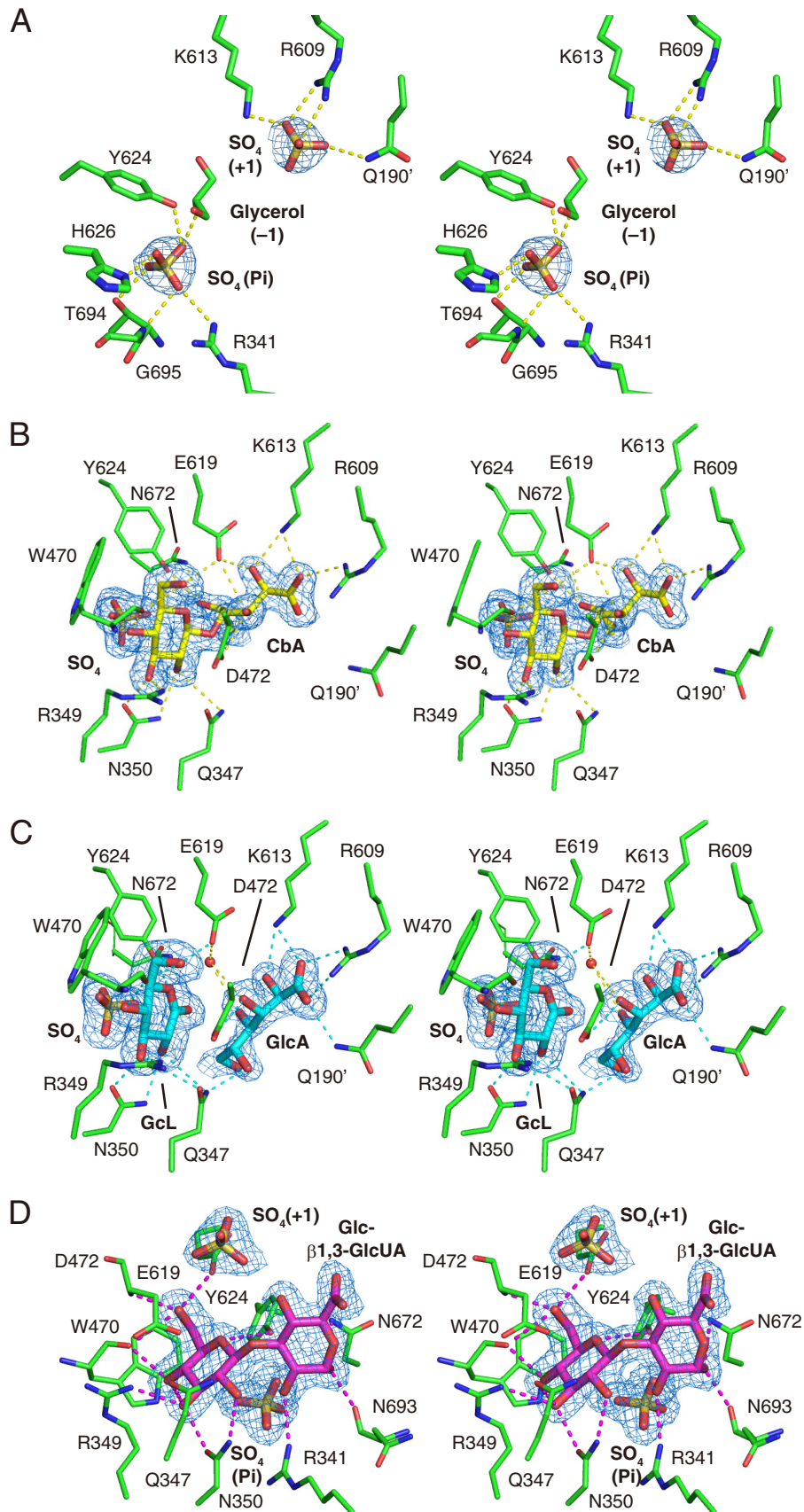


FIGURE 3. Stereoviews of the active site of SdCBAP. A, ligand-free. B, CbA complex (yellow). C, GlcA-GcL complex (cyan). D, Glc-β1,3-GlcUA complex (magenta). The  $|F_o| - |F_c|$  omit electron density maps (4.0σ) of ligands are shown as blue mesh.



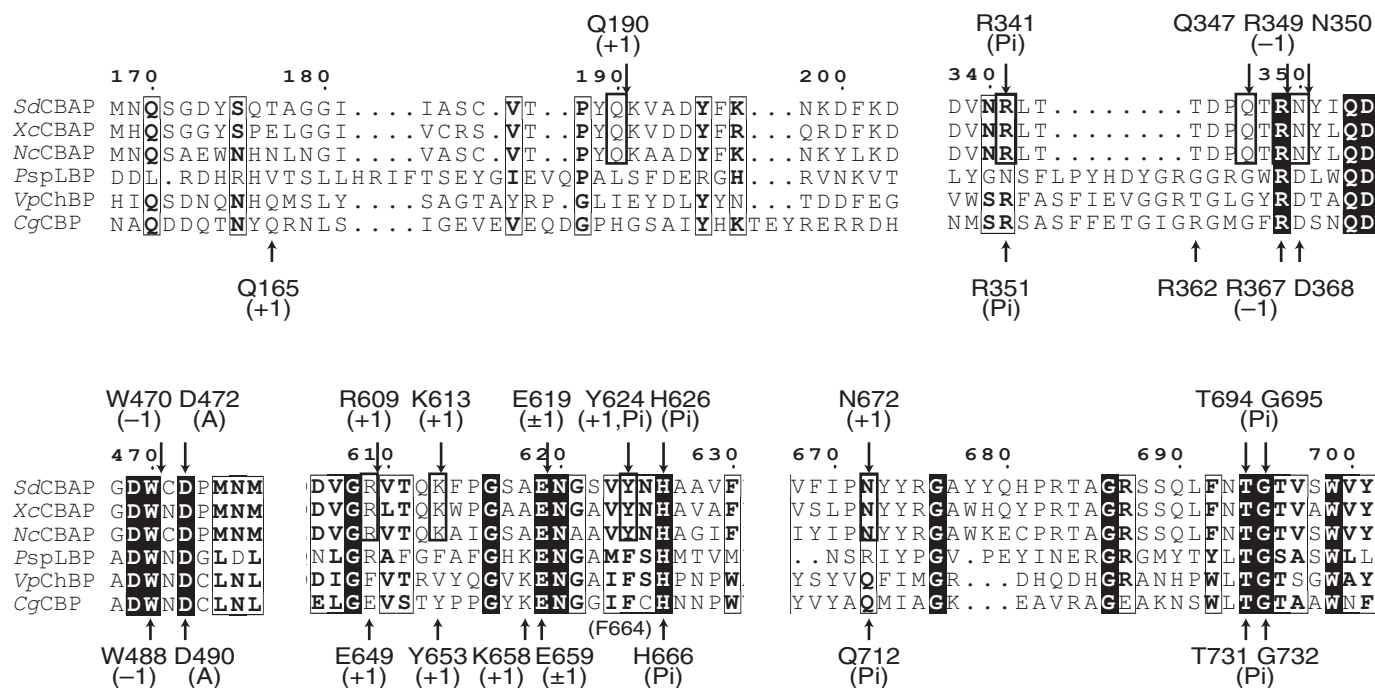


FIGURE 4. **Partial amino acid sequence alignment of GH94 phosphorylases.** Residues at the P<sub>i</sub> binding site and subsites -1 and +1 are indicated above (*SdCBAP*) and below (*CgCBP*) the sequences. Residues involved in the unique subsite +1 of the three CBAPs are shown in the *thick frame*. Residues that have similarity across the sequences are shown in the *thin frame*, and *inverted characters* indicate residues that have complete conservation. *PspLBP*, *Paenibacillus* sp. YM-1 laminaribiose phosphorylase.

at positively charged residues (R609A and K613A) were either not detectable or very low. For the Q190A mutant, the activities were also impaired. The decrease of synthetic activity of Q190A mutant was due to the significant increase and decrease of the  $K_m$  and  $k_{cat}$  values, respectively. We could not determine the  $K_m$  value of the synthetic activity of Q190A mutant using GlcUA as the acceptor because we did not observe any activity saturation up to 20 mM GlcUA. The results indicated that all three of these residues were important for substrate recognition and activity against both GlcA and GlcUA acceptors contrary to the observation in the Glc- $\beta$ 1,3-GlcUA complex structure.

**Implication for the Glc- $\beta$ 1,3-GlcUA Complex Structure**—When a free GlcUA molecule was modeled in subsite +1 of *SdCBAP* by superimposing the C2–C6 atoms to CbA, the C6 carboxylate and the C3 hydroxyl could be placed in appropriate positions for the carboxylate binding pocket and the glycosidic bond oxygen without any steric clash (Fig. 5C). Therefore, the modeled GlcUA appears to be in a position suitable for the synthetic reaction that forms a  $\beta$ 1,3-bond. According to the results of mutational analysis, which emphasize the importance of the three key residues for the reaction using GlcUA, we concluded that the conformation of the Glc- $\beta$ 1,3-GlcUA complex structure was an artifact. Because the concentration of ammonium sulfate in the buffer (600 mM) was higher than that of Glc- $\beta$ 1,3-GlcUA (20 mM), the sulfate anion may have expelled the GlcUA moiety from the canonical position surrounded by Arg-609, Lys-613, and Gln-190'.

**Reaction Mechanism**—Fig. 6 shows a proposed reaction and substrate recognition mechanism of *SdCBAP* from both directions. In the phosphorylase reaction, an oxygen atom of P<sub>i</sub> attacks the anomeric C1 atom of the subsite +1 glucoside, and Asp-472 functions as a general acid by donating a proton to the

glycosidic oxygen atom. Arg-609 and Lys-613 play a crucial role in the recognition of the C1 carboxylate group of the aldonic acid. In the CbA complex structure, Gln-190 from the neighboring protomer was observed in a relatively distant position. However, this residue contributed to the phosphorylase reaction because mutation at this site (Q190A) significantly impaired the activity. Glu-619, Asn-672, and Tyr-624 also contributed to the recognition of the gluconate group by forming hydrogen bonds to C2, C3, and C5 hydroxyls. In the synthetic reaction, Asp-472 works as a general base by accepting a proton from the C4 hydroxyl of GlcA to facilitate the nucleophilic attack to the anomeric C1 atom of G1P. In the GlcA-GcL complex structure (Fig. 5A), the C4 hydroxyl oxygen of GlcA is located far from the C1 atom of GcL (4.7 Å) by the significant displacement of the C4–C5–C6 group compared with CbA. This is probably due to the steric hindrance with the GcL molecule that occupied subsite +1, but G1P would not hinder the correct positioning of GlcA. Residues recognizing the hydroxyls at subsite +1 (Glu-619, Asn-672, and Tyr-624) did not move at all (Fig. 5A). Therefore, we assume that GlcA in the synthetic reaction binds to the protein similarly to the gluconate moiety of CbA. As confirmed by the mutational analysis, Arg-609, Lys-613, and Gln-190' critically contributed to the GlcA recognition in the synthetic reaction.

**Comparison with GH94 Phosphorylases**—Partial amino acid sequence alignment of GH94 phosphorylases is shown in Fig. 4. Because CABP, CBP, ChBP, and laminaribiose phosphorylase act on CbA (Glc- $\beta$ 1,4-gluconic acid), cellobiose (Glc- $\beta$ 1,4-Glc), *N,N'*-diacetylchitobiose (GlcNAc- $\beta$ 1,4-GlcNAc), and laminaribiose (Glc- $\beta$ 1,3-Glc), respectively, these enzymes are expected to have a discrete subsite +1. For comparison, structures of *SdCBAP* (CbA complex) and *CgCBP* were superimposed at the

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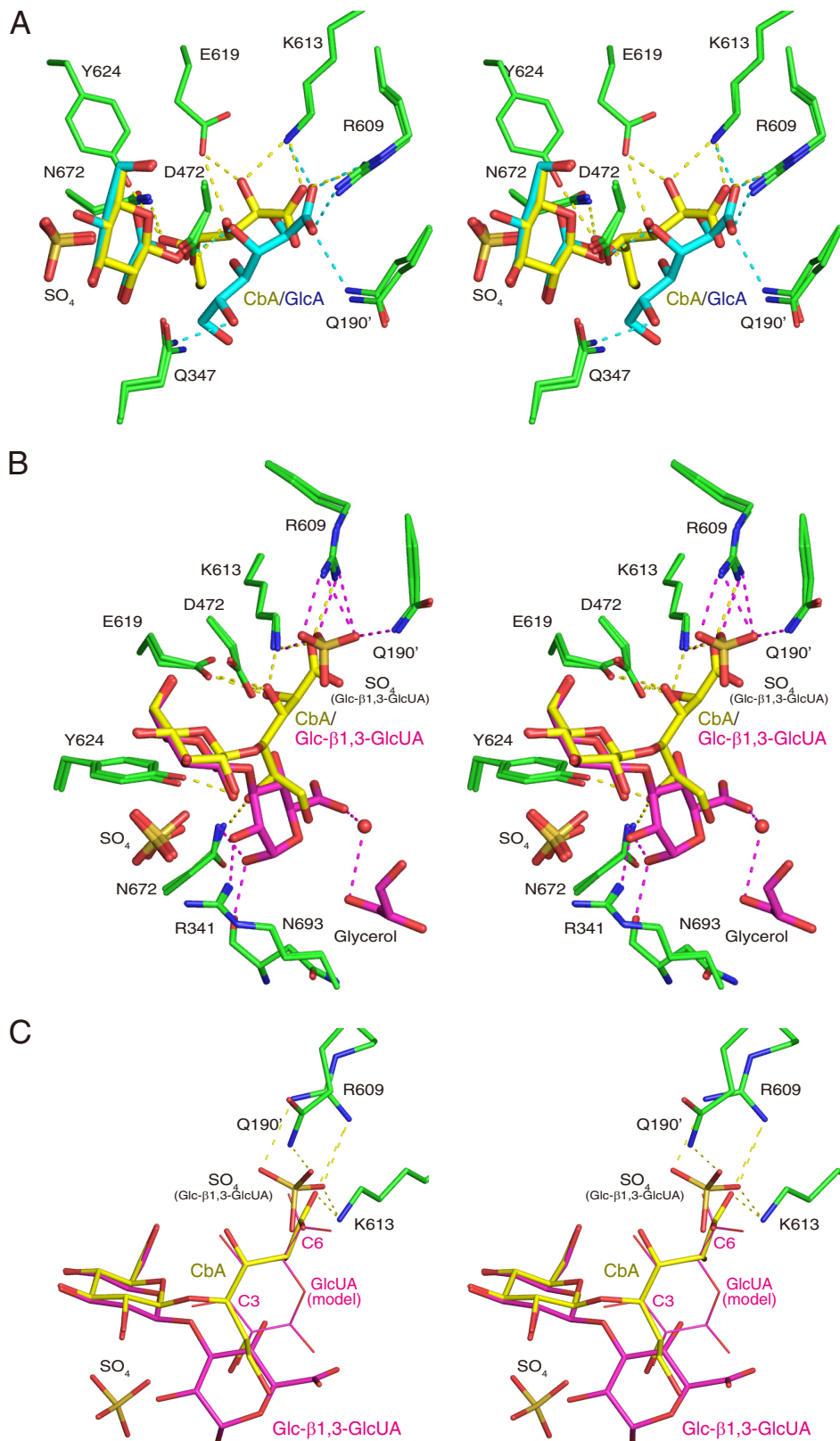


FIGURE 5. **Stereoviews of superimpositions of *SdCBAP* complex structures.** *A*, CbA (yellow) and GlcA-Glc (cyan). *B*, CbA (yellow) and Glc- $\beta$ 1,3-GlcUA (magenta). *C*, CbA (yellow), Glc- $\beta$ 1,3-GlcUA (magenta thick sticks), and a GlcUA molecule modeled at subsite +1 by superimposing the C2–C6 atoms to CbA (magenta thin sticks). Hydrogen bonds present in the CbA, GlcA-Glc, and Glc- $\beta$ 1,3-GlcUA complex structures are shown as yellow, cyan, and magenta dashed lines, respectively.

active site (Fig. 7). The *CgCBP* structure complexed with  $P_i$ , glycerol (subsite -1), and  $\beta$ -D-glucose (subsite +1) was used for comparison (Protein Data Bank code 2CQT) (20). As

expected, residues at the  $P_i$  binding site (Fig. 7A) and subsite -1 (Fig. 7B) are basically conserved except for several residues (Asn-672, Gln-347, and Asn350 in *SdCBAP*). In *CgCBP*, Gln-



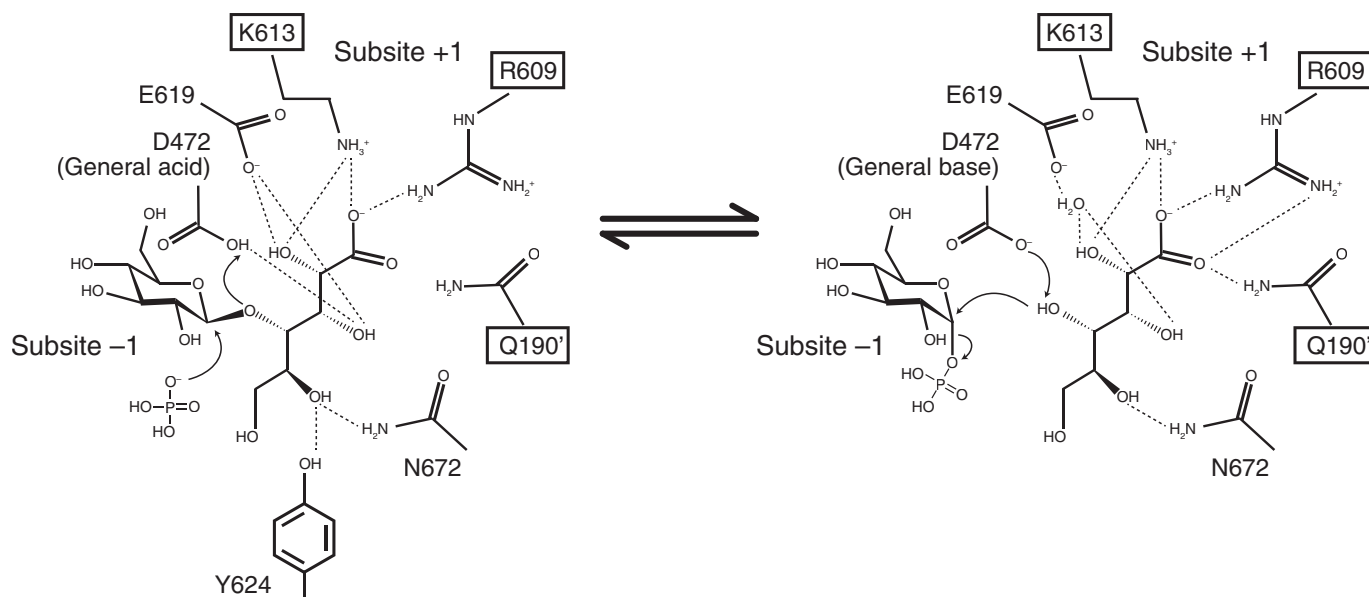


FIGURE 6. Proposed reaction and substrate recognition mechanism of *SdCBAP* from both the phosphorylytic and synthetic directions.

712 forms a hydrogen bond with Pi, whereas the corresponding residue in *SdCBAP* (Asn-672) is involved in subsite +1 recognition (Fig. 7, A and C). Gln-347 in *SdCBAP* forms a hydrogen bond with the C2 hydroxyl of the glucose moiety at subsite -1, but the corresponding residue is Gly in CBP and other enzymes (Figs. 7B and 4). Asn-350 in *SdCBAP* forms hydrogen bonds with C2 and C3 hydroxyls at subsite -1 but is replaced with Asp in other GH94 phosphorylases (Figs. 7B and 4).

Interestingly, the residues involved in subsite +1 recognition of *SdCBAP* are completely conserved in CBAPs (*XcCBAP* and *NcCBAP*), but most of them are not conserved in CBP, ChBP, and laminaribiose phosphorylase (Fig. 4). The region around Gln-190 (186–203 loop in *SdCBAP*) shows low sequence conservation among GH94 phosphorylases. In *CgCBP*, Gln-165 in this region comes from the neighboring protomer and participated in the formation of subsite +1 (Fig. 7C). However, Gln-165 in *CgCBP* and Gln-190 in *SdCBAP* come from different sides. The other two key residues for the carboxylate recognition (Arg-609 and Lys-613) are replaced by Glu-649 and Tyr-653 in *CgCBP*, respectively. Moreover, Tyr-624 and Asn-672 are also not conserved. At subsite +1, Glu-619 is solely conserved because this residue is also involved in subsite -1 recognition (Fig. 7B). In summary, CBAPs have a unique subsite +1 compared with other GH94 phosphorylases.

Sulfate ion and GcL were weak competitive inhibitors of *CgCBP* against phosphate and cellobiose with  $K_i$  values of 79 and 1.1 mM, respectively (34). Those reagents also showed weak inhibition to *SdCBAP* with apparent  $K_i$  values of >30 and ~1 mM, respectively (data not shown).

**A Possible Role of CBAP in Fungal Cellulase Systems**—In this study, we demonstrated that CBAPs from cellulolytic bacteria and fungi shared a conserved active site specific for the binding of aldonic acids. *N. crassa* has a complete oxidative cellulose degradation system, including endoglucanase, cellobiohydrolase, cellobiose dehydrogenase, and C1-oxidizing auxiliary activity (AA) family 9 lytic polysaccharide mono-oxygenases that accept electrons from cellobiose dehydrogenase (35, 36). It

has been shown that cellobiose dehydrogenase enhances cellobiohydrolase activity by relieving product inhibition (37), indicating that the enzymes in the oxidative cellulose degradation system synergistically work to produce CbL. CbL is rapidly and spontaneously hydrolyzed into CbA with a time constant of 61 min at 26 °C (11) and likely transported into the cytoplasm thereafter. The intracellular enzyme CBAP can efficiently degrade CbA, and the products (G1P and GlcA) are further catabolized by the glycolysis and pentose phosphate pathways. The involvement of the glycoside phosphorylase has an energetic advantage in microbial catabolism because CBAP produces the phosphorylated sugar (G1P) without consuming ATP. In addition to *N. crassa*, various fungi of the phylum Ascomycota possess CBAP homologs in their genomes (16), suggesting that the phosphorylytic pathway is also present in these organisms.

**Bacterial Enzymes in Oxidative Cellulose Degradation**—Bacteria do not possess cellobiose dehydrogenase (AA3\_1 subfamily in the CAZy database) at all and must have different systems from fungi. Interestingly, the bacterium *Fibrobacter succinogenes* S85, which does not have cellobiose dehydrogenase, produced CbA and CbL in the culture fluid when grown on cellulosic substrates (38). It has been suggested that pyrroloquinoline quinone-dependent soluble aldose sugar dehydrogenase is responsible for bacterial CbL formation. Aldose sugar dehydrogenase from *E. coli* has wide substrate specificity and can efficiently catalyze the C1 oxidation of various mono-, di-, and trisaccharides, including cellobiose (39). For lytic polysaccharide mono-oxygenases, cellulose-oxidizing AA9 enzymes were exclusively found from eukaryotes. However, several AA10 (CBM33) enzymes from bacteria (*Streptomyces coelicolor* and *Thermobifida fusca*) also oxidize and cleave cellulose (40, 41). Close homologs (amino acid identities >60%) of *SdCBAP* and *XcCBAP* are found in the genomes of cellulose-degrading and plant-pathogenic bacteria, including *Cellvibrio*, *Bacillus*, *Paenibacillus*, and *Xanthomonas* species, which are generally aerobic and possess aldose sugar dehydrogenase and/or AA10 lytic polysaccharide mono-oxygenase genes. This result

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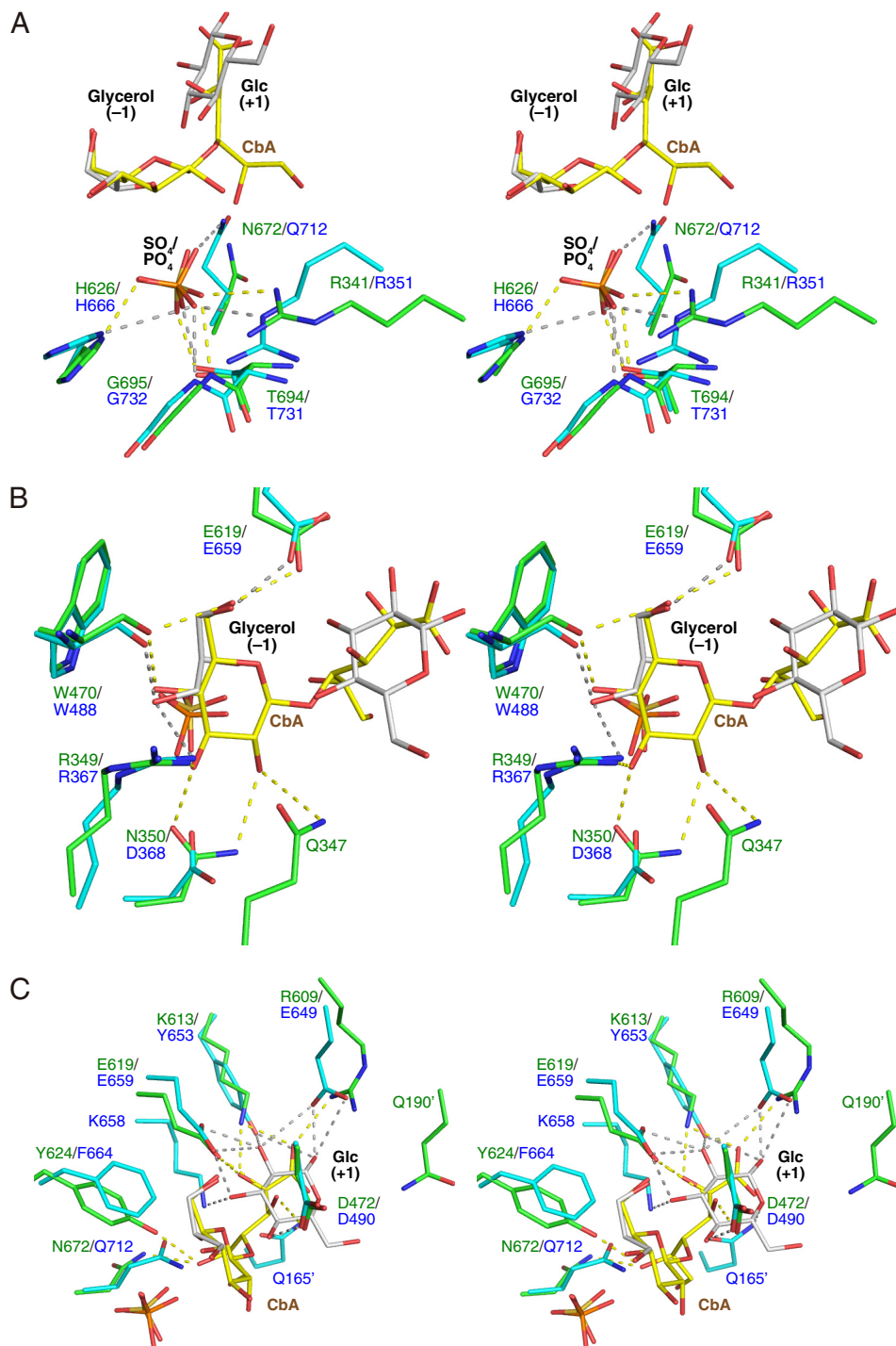


FIGURE 7. Stereoviews of a comparison with CgCBP at the P<sub>1</sub> binding site (A), subsite -1 (B), and subsite +1 (C). A superimposition of SdCBAP (green) complexed with CbA (yellow) and sulfate (orange) and CgCBP (cyan) complexed with P<sub>1</sub> (yellow), glycerol (subsites -1; gray), and β-D-glucose (subsites +1; gray) are shown. Hydrogen bonds present in the SdCBAP and CgCBP structures are shown as yellow and gray dashed lines, respectively.

suggests that bacterial CBAPs also play key roles in oxidative cellulose degradation and infection by decomposing the plant cell wall.

**Role of SdCBAP in the Oxidative Cellulose Degradation System of *S. degradans***—The marine bacterium *S. degradans* strain 2-40 can decompose at least 10 distinct complex polysaccharides from algae, plants, and invertebrates such as cellulose, hemicellulose, and agarose (24). This organism has a novel and complete multienzyme system for cellulose degradation

as its enzymes exhibit unusual architecture via combinations of catalytic and substrate-binding modules (25, 42). Although *S. degradans* apparently lacks a cellobiohydrolase in its genome, three GH5 endoglucanases were shown to be processive and mainly release cellobiose (43). Interestingly, in the genome of *S. degradans*, there are two oxidative enzymes that possibly convert cellobiose into CbL. A putative aldose sugar dehydrogenase gene (*sde\_1897*) shows a certain homology to the cellobiose-active aldose sugar dehy-

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- drogenase from *E. coli* (32% amino acid identity). In addition, there is a putative AA10 gene (*sde\_0633*) attached to the carbohydrate-binding module family 2 domain at its C terminus. The AA10 domain of *sde\_0633* gene exhibits a high homology (54% identity) to cellulose-active and C1-specific lytic polysaccharide mono-oxygenase from *Cellulvibrio japonicus* (44). These facts indicate that the cellulolytic system of *S. degradans* also produces CbL and CbA. A study on possible cellobiose-active enzymes revealed that genes of an extracellular GH3  $\beta$ -glucosidase (*bgl3C*), an intracellular GH1  $\beta$ -glucosidase (*bgl1A*), an intracellular GH94 CBP (*cep94A*), and *SdCBAP* (*cep94B*) were induced by Avicel (32), suggesting the presence of multiple metabolic pathways for cellobiose and CbA by hydrolysis and phosphorylation. The gene architecture of *S. degradans* near the *SdCBAP* (*sde\_0906*) gene locus is interesting as there are genes for a putative major facilitator superfamily for sugar (*sde\_0907*) and a putative gluconate kinase (*sde\_0904*). Therefore, these genes possibly function as an operon to import and metabolize CbA and subsequently send the products into the pentose phosphate and glycolysis pathways. Because the unique and novel cellulolytic system of *S. degradans* has the potential for application to biomass processing (45), more detailed studies are required for a better understanding of its molecular system.
- Concluding Remarks**—This report has provided the first structural basis of CBAP, revealing the key residues for aldonic acid recognition. The discovery of CBAP and its structural basis have expanded our knowledge on microbial cellulose degradation systems. The conservation pattern of the key residues for the CbA recognition is a good indicator to find putative GH94 CBAP genes from genomic and metagenomic information. Consistent with previous studies, this study provided concrete evidence that CBAP participates in the oxidative cellulose degradation systems of cellulolytic bacteria and fungi. In addition to understanding the microbial cellulosic biomass degradation mechanisms, the three-dimensional structure of CBAP will contribute to the future design and engineering of glycoside phosphorylases, which have the potential for application in large scale production of functional oligosaccharides (46).
- Author Contributions**—S. F., H. N., and M. K. conceived and designed the study. S. F., T. A., H. N., and M. K. coordinated the study. S. F., Y.-W. N., and T. N. wrote the paper. Y.-W. N. purified and crystallized the protein and determined the crystal structure. Y.-W. N., T. A., and S. F. collected the x-ray data. Y. S. and T. N. designed and constructed expression vectors. Y.-W. N. constructed mutant expression vectors and purified mutant proteins. T. N. and Y.-W. N. purified and performed enzymatic assay of the wild-type and mutant enzymes. T. N., H. N., and M. K. analyzed kinetic data. All authors reviewed the results and approved the final version of the manuscript.
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