

Structure of LacY with an α -substituted galactoside: Connecting the binding site to the protonation site

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The X-ray crystal structure of a conformationally constrained mutant of the *Escherichia coli* lactose permease (the LacY double-Trp mutant Gly-46→Trp/Gly-262→Trp) with bound *p*-nitrophenyl- α -D-galactopyranoside (α -NPG), a high-affinity lactose analog, is described. With the exception of Glu-126 (helix IV), side chains Trp-151 (helix V), Glu-269 (helix VIII), Arg-144 (helix V), His-322 (helix X), and Asn-272 (helix VIII) interact directly with the galactopyranosyl ring of α -NPG to provide specificity, as indicated by biochemical studies and shown directly by X-ray crystallography. In contrast, Phe-20, Met-23, and Phe-27 (helix I) are within van der Waals distance of the benzyl moiety of the analog and thereby increase binding affinity nonspecifically. Thus, the specificity of LacY for sugar is determined solely by side-chain interactions with the galactopyranosyl ring, whereas affinity is increased by non-specific hydrophobic interactions with the anomeric substituent.

X-ray structure | membrane protein | transport | major facilitator superfamily | conformational change

The lactose permease of *Escherichia coli* (LacY) binds and catalyzes the coupled stoichiometric transport of D-galactose or β -D-galactopyranosides and H⁺ (galactoside/H⁺ symport) but does not interact with glucopyranosides. Biochemical studies (1–7) indicate that affinity and specificity are distinct properties determined by different interactions with LacY. Specificity is determined entirely by interactions with the galactopyranosyl ring, whereas affinity is better with α - than β -galactopyranosides (anomeric at C1) and can be increased dramatically by hydrophobic anomeric substituents with no effect on specificity.

By using the free energy released from the energetically downhill movement of H⁺ in response to the electrochemical H⁺ gradient ($\Delta\tilde{\mu}_{H^+}$), LacY catalyzes uphill (active) transport of galactosides against a concentration gradient. Because coupling between sugar and H⁺ translocation is obligatory, in the absence of $\Delta\tilde{\mu}_{H^+}$, LacY can also transduce the free energy released from the downhill transport of sugar to drive uphill H⁺ transport with the generation of $\Delta\tilde{\mu}_{H^+}$, the polarity of which depends upon the direction of the sugar gradient (reviewed in refs. 8–10).

Rates of equilibrium exchange and counterflow (exchange of one substrate molecule for another labeled molecule from the other side of the membrane) are unaffected by imposition of $\Delta\tilde{\mu}_{H^+}$. Therefore, it is apparent that alternating accessibility of sugar- and H⁺-binding sites to either side of the membrane is the result of galactoside binding and dissociation and not $\Delta\tilde{\mu}_{H^+}$ (reviewed in refs. 8–10). Moreover, downhill lactose/H⁺ symport from a high to a low lactose concentration in the absence of $\Delta\tilde{\mu}_{H^+}$ exhibits a primary deuterium isotope effect that is not observed for $\Delta\tilde{\mu}_{H^+}$ -driven lactose/H⁺ symport, equilibrium exchange, or counterflow (11, 12). Thus, it is likely that the rate-limiting step for downhill symport is deprotonation (13, 14), whereas in the presence of $\Delta\tilde{\mu}_{H^+}$, opening of a cavity on the other side of the membrane after dissociation of sugar and H⁺ is limiting (15). Based on these and other findings, a detailed mechanism for symport by LacY has been proposed (10).

Initial X-ray structures of LacY without bound sugar exhibit two pseudosymmetrical bundles of mostly irregular transmembrane helices surrounding a large aqueous cavity in the middle of the

molecule; these initial structures were open on the cytoplasmic side and sealed on the periplasmic side (an inward-open conformation) (16–19). However, our recent X-ray crystallography studies (20) of the conformationally trapped double-Trp mutant G46W/G262W cocrystallized with β -D-galactopyranosyl-1-thio- β -D-galactopyranoside (TDG) reveal an almost occluded conformation with a narrowly outward (periplasmic)-open conformation and a tightly sealed cytoplasmic side [Protein Data Bank (PDB) ID code 4OAA]. In addition, a molecule of TDG is bound in a central cavity. The evidence shows that specific galactoside binding is consistent with prior findings from mutagenesis (21–23) and uses induced fit to interact with the surrounding protein (20). The findings also provide a strong indication that the transport mechanism of LacY involves a substrate-bound, occluded, intermediate conformation.

Lactose has only one galactopyranosyl ring. Similarly, one galactopyranosyl ring of TDG lies against Trp-151 (helix V), confirming hydrophobic stacking between the bottom of the galactopyranosyl ring and the aromatic indole ring as suggested (24, 25). Glu-269 (helix VIII) is the acceptor of hydrogen bonds from the C4-OH group of the galactopyranosyl ring (21, 26). The η 1 NH₂ group of Arg-144 (helix V) donates a hydrogen bond to O5 in the ring and is within hydrogen-bond distance of the C6-OH, whereas the η 2 NH₂ group of Arg-144 donates hydrogen bonds to the C2'-OH (the other galactopyranosyl ring) of TDG and to Glu-126

Significance

Substituted D-galactopyranosides, particularly those in the α configuration and/or with hydrophobic constituents at the anomeric position, bind to LacY with higher affinity than the physiological substrate lactose that has a β configuration. The structure of a conformationally restricted LacY mutant with bound *p*-nitrophenyl- α -D-galactopyranoside (α -NPG), a high-affinity lactose analog, is described. Higher affinity, gained by nonspecific hydrophobic interaction of the nitrophenyl group, shows identical interaction at the key galactosyl moiety as in thio-digalactoside and so validates the highly specific, oriented set of hydrogen bonds with the key galactosyl moiety of substrates. Confirmation of galactose-specific binding interactions delineates a directional hydrogen-bonding network that couples the binding site to sites that are sensitive to protonation in the mechanism.

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Data deposition: The atomic coordinates and structure factors have been deposited in the Protein Data Bank, www.pdb.org (PDB ID code 4ZYR).

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Oε2 (27–29). Glu-126 (helix IV) acts as hydrogen-bond acceptor from the C2'-OH of TDG (27–29). His-322 (helix X) acts as a hydrogen-bond donor/acceptor between the εNH of the imidazole ring and the C3-OH of TDG (29–33) and is stabilized by a hydrogen-bond donor/acceptor interaction with the δNH of the imidazole and the OH of Tyr-236 (29, 34, 35). Finally, Asn-272 (helix VIII) donates a hydrogen bond to the C4-OH of TDG (23). These interactions define the specificity of LacY (summarized in ref. 20).

Cys-148 (helix V), well known with respect to substrate protection against alkylation (reviewed in ref. 22), is also close to bound TDG but not sufficiently close to interact directly. Similarly, replacement of Ala-122 (helix IV) with a bulky side chain or alkylation of A122C with bulky thiol reagents causes LacY to become specific for the monosaccharide galactose. Disaccharide binding and transport are blocked sterically (36). Although Ala-122 does not make direct contact with TDG, bulky substituents at position 122 would clearly impact disaccharide binding.

p-Nitrophenyl- α -D-galactopyranoside (α -NPG) is a FRET acceptor from Trp-151 (37) and binds to LacY with \sim eight times higher affinity than the twofold symmetric TDG, \sim two orders of magnitude better than β -NPG, and \sim three orders of magnitude better than the physiological substrate lactose or the monosaccharide galactose (6, 7, 38, 39). As we show here, the side-chain interactions with the galactopyranosyl moiety of α -NPG that provide specificity are almost identical to those described for TDG. In contrast, the increased affinity of α -NPG versus TDG is probably attributable primarily to hydrophobic interactions between the nitrophenyl group of NPG and Phe-20, Met-23, and Phe-27 from helix I. In addition, the nitro group is in close contact with polar groups that can sustain polar interactions of the type that also pertain to the glucose moiety of lactose.

Results

Global Fold. The X-ray crystal structure of LacY G46W/G262W cocrystallized in the presence of 1 mM α -NPG was determined to 3.3-Å resolution (Fig. 1). Data collection and refinement statistics are summarized in Table 1. Crystals of the G46W/G262W

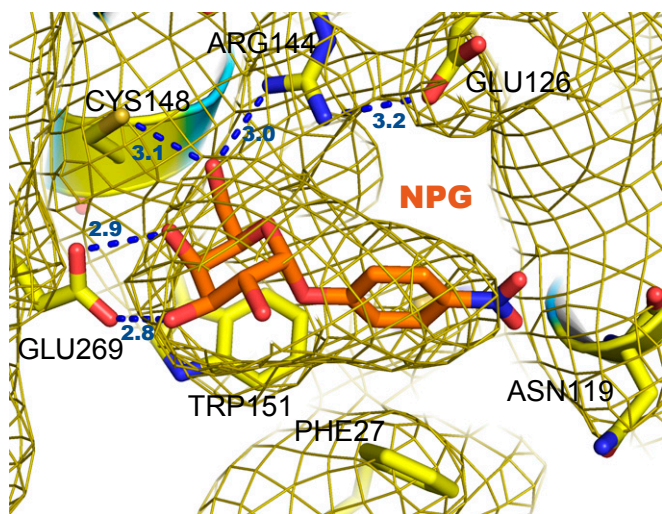


Fig. 1. Electron density map of the α -NPG ligand site in molecule A, LacY G46W/G262W. Density contoured at 1.3σ (golden yellow mesh) is superimposed on the cartoon representation of the structure. Residues surrounding the ligand are labeled. Carbon atoms in α -NPG are shown in orange, nitrogen in blue, and oxygen in red. Side-chain atoms are shown as carbon in yellow, nitrogen in blue, oxygen in red, and sulfur in light yellow. Broken blue lines show hydrogen bonds with lengths labeled in Å. The cytoplasmic side is upward, and the normal to the membrane plane is approximately vertical.

Table 1. Data collection and refinement statistics

Wavelength, Å	1.116
Resolution range, Å	37–3.31 (3.42–3.31)
Space group	C 2 2 2 ₁
Unit cell	103.2 122.0 266.1
Total reflections	114045 (10755)
Unique reflections	25009 (2429)
Multiplicity	4.6 (4.4)
Completeness, %	97.6 (97.2)
Mean $I/\sigma(I)$	7.8 (0.4)
Wilson B-factor	126
R-merge	0.16 (4.06)
R-measured	0.18
CC1/2	0.998 (0.082)
CC*	0.999 (0.390)
R-work	0.25 (0.40)
R-free	0.28 (0.41)
Nonhydrogen atoms	6157
Ligand atoms	84
No. waters	0
Protein residues	772
rms (bonds), Å	± 0.004
rms (angles), °	± 0.85
Ramachandran favored, %	95
Ramachandran outliers, %	0.39
Clash score	14.20
Average B-factor	142
Macromolecules	142
Ligands	141

Parameters for the outermost shell are shown in parentheses. CC1/2, percentage of correlation between intensities from random half datasets. Correlation significant at the 0.1% level (59). CC*, the CC of the full dataset against the true intensities (59).

mutant with α -NPG are in space group C222₁ with the cell dimensions $a = 103.17$ Å, $b = 122.01$ Å, $c = 266.13$ Å. The structure was determined by molecular replacement using the almost occluded outward-open structure of G46W/G262W LacY alone after removing the ligand TDG from PDB ID code 4OAA (20) as the search molecule. The structure of the protein with bound α -NPG is essentially the same as the TDG-bound structure (PDB ID code 4OAA) (rmsd, 0.52 Å), in contrast to the inward-facing WT LacY structure (19) (PDB ID code 2V8N).

There are two independent molecules of LacY in the asymmetric unit that are adjacent to one another but of opposite-facing orientation relative to the putative plane of the membrane (Fig. S1). The protein occupies 28% of the unit cell volume. The Trp replacements G46W/G262W on the periplasmic side induce the transition to an almost occluded, outward-facing conformation and allow substrate binding in an almost identical manner to TDG with respect to the galactopyranosyl moiety (Fig. S2).

The two independent α -NPG-bound LacY molecules in the unit cell (termed A and B) are both in an almost-occluded, outward-open conformation with a single molecule of α -NPG in the central sugar-binding site, suggesting that they are in their most stable solution structure (Fig. 2). The two structures differ slightly from each other only in the orientation of the substrate and the slight rearrangement of the interacting side chains (Table S1). There are also two molecules of nonyl- β -D-glucopyranoside bound in the interface between the LacY molecules. (Fig. S1).

Compared with WT LacY [without substrate bound (PDB ID code 2V8N)], the overall rmsd between 390 C α s of the G46W/G262W is 4.5 Å, reflecting the major conformational change to the almost occluded, outward-facing conformation (leaving out the ligand and immediate surroundings).

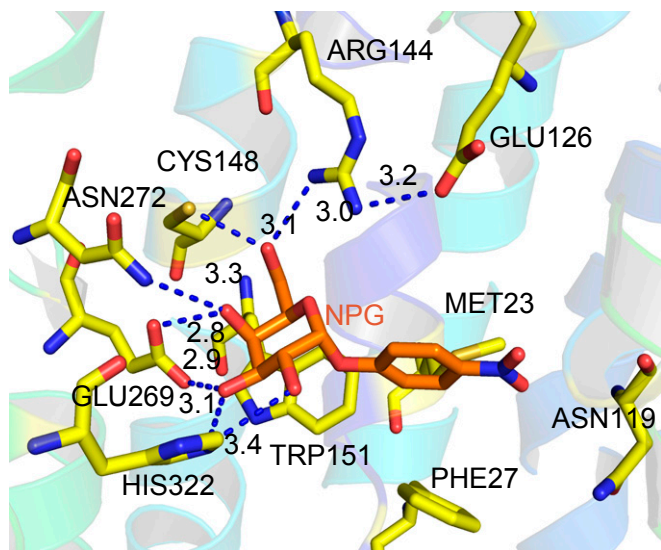


Fig. 2. The galactoside-binding site in LacY G46W/G262W, molecule A. The cytoplasmic side is upward, and the normal to the membrane plane is approximately vertical. Atoms in α -NPG are shown with carbon in orange, nitrogen in blue, and oxygen in red. Side-chain atoms are shown with carbon in yellow, nitrogen in blue, oxygen in red, and sulfur in light yellow. Dotted blue lines identify hydrogen bonds.

The Substrate-Binding Site. The conformation of a higher-affinity α -NPG molecule (K_d , $\sim 2 \mu\text{M}$) is apparent in the almost occluded central cavity (Figs. 1 and 2 and Fig. S1) and allows assignment of probable hydrogen-bond interactions with the protein (Table S1). Although at 3.3-Å resolution interatomic distances are an approximation, their mutual orientation is key to defining hydrogen-bonding interactions. Electron density for the α -NPG is also detected in an unbiased omit map after refinement without ligand clearly showing its orientation and interactions.

Cys-148, which is protected from alkylation by substrate binding, is somewhat closer to the galactopyranosyl ring here (3.1 Å versus 4.6 Å) than observed previously with the TDG-bound structure (PDB ID code 4OAA). Therefore, Cys-148 may be hydrogen bonded to C6-OH of the galactoside in the primary specificity site. (Table S1).

His-322 in both the TDG and α -NPG bound structures provides a hydrogen-bonded bridge between the C2-OH and C3-OH of the galactoside on one side of the imidazole ring and Glu-325 and Tyr-236 on the other side (Fig. S3). In turn, Tyr-236 accepts a hydrogen bond from donor Arg-302. Without substrate bound, this complex internal site could rearrange to redistribute charge, decrease the effective $\text{p}K_a$ of the site around Glu-325, and account for coupled H^+ transport.

α -NPG is tilted toward Met-23, and the *p*-nitrophenyl group is in van der Waals contact with it (Fig. 2); the sulfur atom in Met-23 is ~ 3.5 Å from the C10 (Fig. S4 and Table S1) of α -NPG. Together, Phe-20, Phe-27, and Met-23 form a hydrophobic cavity. A polar contact from the carbonyl oxygen of Phe-118 (helix IV) is rotated toward one nitrophenyl oxygen (O7) of α -NPG at a distance of ~ 3 Å. The side chain of Asn-119 (helix IV) also contributes to the polar environment in the vicinity of the nitro moiety of the nitrophenyl ring at a distance of ~ 3.7 – 4.0 Å.

Discussion

In order for LacY to transport galactopyranosides sufficiently rapidly for the cell to grow at optimal rates, binding affinity cannot be too high. Otherwise, the dissociation rate of the sugar would be limiting for growth. Thus, lactose, the physiological substrate for LacY, binds with millimolar affinity but is transported

at an optimal rate. However, the low affinity for lactose precludes direct determination of binding, and in this regard, protection of Cys-148 against alkylation by substrate is particularly useful for measuring relative affinities of low-affinity substrates (5–7, 38, 40, 41). By using this technique, it was shown that despite having the lowest affinity, the monosaccharide galactose binds as specifically as substituted galactopyranosides that bind orders of magnitude better, but glucose or glucopyranosides have no effect. It has also been shown (5, 6) that although the specificity of LacY is strongly directed toward the C4-OH of the galactopyranosyl ring, the OH groups are important in the following order: C4-OH \gg C6-OH $>$ C3-OH $>$ C2-OH. These interactions seen in the structure broadly explain the relative preferences in terms of the number of hydrogen bonding groups that coordinate the galactopyranosyl hydroxyls.

TDG appears to be unique as a relatively high-affinity ligand, because the anomeric functional group is hydrophilic rather than hydrophobic. However, the hydrophilic moiety is another galactopyranosyl molecule, and because the two moieties are linked one-to-one, TDG is perfectly symmetrical. Therefore, the molecule can associate with the binding site in LacY from either end. This being the case, the symmetrical digalactopyranoside TDG may not be able to dissociate readily from the binding site, which causes TDG to bind almost three orders of magnitude better than lactose but also results in much lower transport activity relative to lactose (26, 42).

Furthermore, galactopyranosides in the α -anomeric (C1) configuration bind with better affinity than β -anomeric (C1) galactopyranosides, and affinity can be increased dramatically by hydrophobic additions to the anomeric position C1 of galactose with no effect on specificity (5). For example, LacY exhibits no demonstrable affinity for *p*-nitrophenyl- α -D-glucopyranoside in which the C4-OH has the alternate chirality (43, 44). These results also emphasize that the interactions seen in the structure determine specificity.

The galactopyranosyl rings of α -NPG and TDG in the binding site are in an essentially identical configuration, although they differ in chirality at C1, α in NPG (an α -galactoside, downward in Fig. 3) versus β in TDG (a β -thio di-galactoside, upward in Fig. 3). This difference in chirality is offset by the nitrophenyl in α -NPG lying in a more hydrophobic cavity; with more polar surroundings of the second galactose ring in TDG, where Glu-126 (helix IV) and the $\eta 2$ NH₁ of Arg-144 are hydrogen-bonded to the C2'-OH. The Arg-144

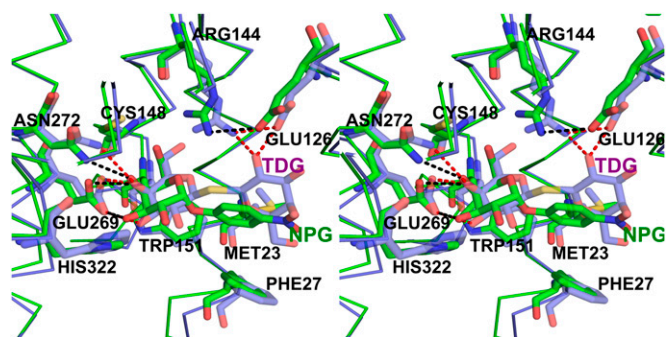


Fig. 3. Comparison of the sugar-binding site in α -NPG-bound with TDG-bound LacY G46W/G262W. The view is crossed-eyed stereo. The TDG structure (PDB ID code 4OAA chain A) [light blue ribbons and side chains as sticks (carbon in blue, oxygen in red, nitrogen in dark blue, and sulfur in yellow) with TDG (shown as carbon in blue, oxygen in red, nitrogen in dark blue, and sulfur in yellow) hydrogen bonds in dashed red] is superimposed on the α -NPG structure [green ribbons with side chains as sticks (carbon in green, oxygen in red, nitrogen in dark blue, and sulfur in yellow) with NPG (shown as carbon in green, oxygen in red, and nitrogen in dark blue) hydrogen bonds in dashed black].

η_2 NH₁ can no longer hydrogen bond to the nitrophenyl group, and instead hydrogen bonds to O5 of the galactopyranosyl ring (Fig. 3). However, Glu-126 O ϵ_2 remains as a hydrogen bond acceptor from the η_2 NH₂ group of Arg-144 (Figs. 1–4). The hydrogen bond of the η_1 NH₂ group of Arg-144 (helix V) to C6-OH is maintained with both ligands. Thus, Arg-144 makes two hydrogen bonds with the galactopyranosyl ring in α -NPG (Fig. 4).

In both α -NPG and TDG, the primary galactopyranosyl ring lies against the aromatic face of Trp-151 (helix V), providing hydrophobic stacking, as found in other sugar-binding sites. Glu-269 (helix VIII) is acceptor of two hydrogen bonds, each with almost ideal geometry, from the C3-OH and C4-OH groups of the galactopyranosyl ring and so provides selectivity-determining interactions. The ϵ N2 of His-322 (helix X) forms a bifurcated hydrogen bond to both C2-OH and C3-OH. If Glu-269 is unprotonated and negatively charged it would imply that C3-OH provides the donor hydrogen to Glu-269. Therefore, the His ϵ N2 must be protonated and a binary donor to two oxygen substituents of the pyranosyl ring. On the other side of the imidazole ring, the δ NH of the imidazole of His-322 forms a critical linkage with groups that are essential for protonation/deprotonation particularly the carboxyl of Glu-325 and also the hydroxyl of Tyr-236 (Fig. 4). If Glu-269 is charged, it will act as acceptor from C4-OH of the pyranose, therefore Asn-272 (helix VIII) most probably donates a hydrogen bond to the oxygen of C4-OH in α -NPG.

Glu-325 (helix X) and Arg-302 (helix IX) are directly involved in coupled H⁺ translocation; neutral replacement of either residue yields mutants that are defective in all transport reactions that involve lactose/H⁺ symport but bind substrate well and catalyze equilibrium exchange and/or counterflow as well or better than WT (45–47). The mutants therefore undergo the conformational change allowing alternate access to either side of the membrane and hence translocation of galactopyranoside across the membrane, a fundamental property of the symporter. However, the mutants are no longer coupled to the pK_a changes in Glu-325 that must occur during coupled H⁺ transport (10).

The phenyl ring of α -NPG derives its affinity primarily from hydrophobic interactions. The side chains of Met-23, Phe-20, and Phe-27 in helix I (~3.6 Å) provide a hydrophobic cavity accounting for higher affinity. α -NPG is tilted toward Met-23, and the *p*-nitrophenyl group is in van der Waals contact; the sulfur atom in Met-23 is ~3.5 Å from the C10 of α -NPG (Fig. 2,

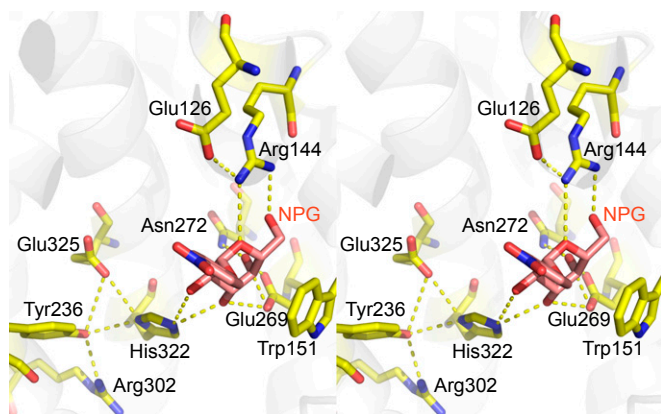


Fig. 4. Connection between the ligand-binding site and residues known to be involved in protonation/deprotonation during the transport cycle are indicated. The figure is in crossed-eyed stereo. The cytoplasmic surface is upward, and the normal to the membrane plane is approximately vertical. The figure shows the hydrophobic contact of the galactoside ring with the indole ring of Trp-151. O2 and O3 of the galactose are hydrogen-bonded to the imidazole of His-322 that in turn forms hydrogen bonds with Glu-325 and Tyr-236. The hydroxyl of Tyr-236 is also acceptor to a hydrogen bond from Arg-302.

Fig. S4, and Table S1). A polar contact from the carbonyl oxygen of Phe-118 (helix IV) is rotated toward one nitrophenyl oxygen (O7) of α -NPG at a distance of ~3 Å. The side chain of Asn-119 (helix IV) also contributes to the polar environment in the vicinity of the nitro moiety of the nitrophenyl ring at a distance of ~3.7–4.0 Å. The large cavity around the nitrophenyl group provides both hydrophobic and polar counterparts in what would be the vicinity of the glucose moiety of lactose. Notably, glucose is often bound using hydrophobic and polar groups in proteins.

α -NPG binds 10 times better than phenyl- α -D-galactopyranoside (lacking the nitro group), and the nitro group provides ~three times higher affinity in the *para* position, as opposed to the *ortho* or *meta* position (7). This effect could thus be entropic because of the higher entropic cost of ordering the *ortho* and *meta* substituents or to the plausible solvent linkage to Asn-119. The methyl group of Ala-122 (helix IV) is ~4.5 Å from O7 of the nitrophenyl ring, and mutational analysis is also consistent with α -NPG being relatively distant from Ala-122.

In addition to the nitrophenyl group, methyl, allyl, cyclohexyl, phenyl, methylphenyl, naphthyl (7), or dansyl (48) increase affinity to varying degrees. However, these hydrophobic interactions not only make no contribution to specificity, but individual mutation of the surrounding side chains in LacY has little or no effect on the ability of LacY to catalyze lactose/H⁺ symport (21, 49). Thus, they are not involved in the symport mechanism. Perhaps surprising is the large volume around the substituents of the primary galactosyl moiety, suggesting a role for water in the coordination of substrate and, particularly, a role for water after the galactoside dissociates, when LacY must return empty in a state that deprotonates before transitioning into the empty occluded state.

In the mechanism proposed recently (10), Glu-325 plays the key role in coupled H⁺ translocation. To bind galactoside, Glu-325 must be protonated, and because this side chain appears to have a pK_a of ~10.5 for binding substrates (29, 50, 51), LacY is protonated over the physiological pH range. Furthermore, replacement of any single side chain that interacts directly with the galactosyl moiety of substrate drastically decreases affinity (29). As shown in the current structure (Fig. 4), Glu-325 is linked to the galactoside-binding site via a hydrogen bond with His-322N δ H. The His-322N δ H is hydrogen donor to 3OH of the galactopyranose, because if Glu-269 is negatively charged, the 3OH in turn acts as hydrogen donor to Glu-269. If the site returns to the occluded state without substrate bound, it must lose an H⁺, implying a drop in the effective pK_a. It does so presumably through interaction of protonated Glu-325 with Arg-302 and water molecules and loses the H⁺ to the side that is closing. This proposal can then account for the vectorial nature of the deprotonation, determined primarily by the water access of the closing side, whichever that may be. Although Glu-325 and Arg-302 are essential for protonation/deprotonation and His-322 is essential for binding, all three may be required for the pK_a change in Glu-325 upon closing to the occluded, substrate-free conformation.

Materials and Methods

Materials. NPG and buffers were from Sigma-Aldrich. Talon superflow resin was from BD Clontech. Dodecyl- β -D-maltopyranoside (DDM) and n-nonyl- β -D-glucoside (NG) were from Affymetrix. All other materials were of reagent grade and obtained from commercial sources.

Growth, Expression, and Purification. Plasmid pT7-5 encoding LacY G46W/G262W/His₆ tag was expressed in *E. coli* C41. A detailed procedure is described in *SI Materials and Methods*. Membranes were prepared and solubilized in 2% (wt/vol) DDM, and LacY was purified on a Co (II) column and eluted in 20 mM Hepes/0.2% NG/200 mM imidazole (pH 6.5). Details are given in *SI Materials and Methods*.

Crystallization, Data Collection, and Structure Determination. NPG was added (1 mM, final concentration) to a protein solution [10 mg/mL in 20 mM Hepes (pH 6.5)/0.2% NG] before crystallization trials. Crystallization

screens were performed using the hanging-drop vapor-diffusion method on a Mosquito Crystal Robot (TTP Labtech) in a 96-well plate with a drop ratio of 100 nL of well solution:300 nL of protein. Crystals appeared in 24 h and grew to $600 \times 200 \times 150 \mu\text{m}$ in a week. Crystals were harvested after 7 d and screened for diffraction. Crystals were reproducible and appeared in wide range of PEG concentrations. They also appeared at different pH values and with different buffers. The best diffracting crystals grew in 1.0 M NaCl/0.05 M Tris (pH 8.0)/26% PEG 600. No cryoprotectant was added before data collection.

Diffraction data were collected at the Lawrence Berkeley National Laboratory Advanced Light Source Beamline 8.3.1, at -170°C at a wavelength of 1.115 Å. The highest-resolution crystals diffracted to 3.3 Å. Data were processed with XDS (52) and HKL2000 (53).

The structure was determined by molecular replacement using LacY_{46WG262W} (PDB ID code 4OAA) as the search model, in the program suite PHENIX (54). The program COOT was used for density fitting (55), and refinement was carried out using PHENIX and Refmac from the CCP4 suite (56). All covalent hydrogen atoms were added in expected positions. Table 1 shows data and refinement statistics. The MolProbity server (57) was used for

structural validation leading to improvement. Restrained refinement was performed with noncrystallographic symmetry (NCS) and stereochemistry restraint for all of the refinement, at later stage crystallographic refinement server developed by Haddadian et al. (58) (godzilla.uchicago.edu) was used for further refinement.

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