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Interaction of Anesthetics with the Rho GTPase Regulator Rho GDP Dissociation Inhibitor†

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Abstract

The physiological effects of anesthetics have been ascribed to their interaction with hydrophobic sites within functionally relevant CNS proteins. Studies have shown that volatile anesthetics compete for luciferin binding to the hydrophobic substrate binding site within firefly luciferase and inhibit its activity (Franks, N. P., and Lieb, W. R. (1984) Nature 310, 599-601). To assess whether anesthetics also compete for ligand binding to a mammalian signal transduction protein, we investigated the interaction of the volatile anesthetic, halothane, with the Rho GDP dissociation inhibitor (RhoGDIa), which binds the geranylgeranyl moiety of GDP-bound Rho GTPases. Consistent with the existence of a discrete halothane binding site, the intrinsic tryptophan fluorescence of RhoGDIa was quenched by halothane (2-bromo-2-chloro-1,1,1trifluoroethane) in a saturable, concentration-dependent manner. Bromine quenching of tryptophan fluorescence is short range and W192 and W194 of the RhoGDIa are located within the geranylgeranyl binding pocket, suggesting that halothane binds within this region. Supporting this, N-acetyl-geranylgeranyl cysteine reversed tryptophan quenching by halothane. Short chain nalcohols (n<6) also reversed tryptophan quenching, suggesting that RhoGDI α may also bind nalkanols. Consistent with this, E193 was photo-labeled by 3-azibutanol. This residue is located in the vicinity of, but outside, the geranylgeranyl chain binding pocket, suggesting that the alcohol binding site is distinct from that occupied by halothane. Supporting this, N-acetyl-geranylgeranyl cysteine enhanced E193 photo-labeling by 3-azibutanol. Overall, the results suggest that halothane binds to a site within the geranylgeranyl chain binding pocket of RhoGDIa, whereas alcohols bind to a distal site that interacts allosterically with this pocket.

Despite the routine use of volatile anesthetics in surgical procedures for over 150 years, the sites and mechanism of action remain largely unresolved. It has been proposed that these agents interact with hydrophobic sites within functionally relevant CNS proteins (1, 2), and

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examples of such sites within ion channels and receptors have been noted (3). Two mechanisms have been proposed for anesthetic action on protein structure/function (3). In the first mechanism, anesthetics bind selectively to a subset of the conformations that are available to a protein, and thereby act as classical allosteric agents. Such a mechanism has been shown to apply to the interaction of anesthetics with the nicotinic acetylcholine receptor (4, 5), and also with protein kinase C (PKC), where general anesthetics and alcohols appear to allosterically modulate phorbol ester binding to the C1 domains (6–8). In the second mechanism, anesthetics interact directly with binding sites that are normally utilized by endogenous ligands. Experimental evidence supporting this second mechanism comes exclusively from non-mammalian systems where kinetic studies suggest that alcohols and anesthetics compete for the substrate binding sites for luciferin on the firefly luciferase (9) and for aldehyde on the bacterial luciferase (10, 11).

To test whether this second mechanism might also apply to the interaction of anesthetics with a mammalian protein, we chose Rho GDP dissociation inhibitor α (RhoGDI α),¹ which is a pivotal regulator of the biological activities the Rho family of low molecular weight GTPases (Rho GTPases) and shares with the Rho GTPases a central role in the regulation of the cytoskeleton (12–15). The choice of RhoGDI α was also based on the fact that in common with other protein targets for anesthetics, it contains a hydrophobic pocket. This pocket is contained within the C-terminal immunoglobulin-like fold of RhoGDI α and is lined by a number of hydrophobic residues, including W192 and W194 (16, 17). It is formed by two β -sheets that undergo a rigid-body re-orientation upon binding of the C-terminal geranylgeranyl (GG) chain of the Rho GTPases (16), for which it displays a high degree of structural specificity (18). This interaction contributes to the stabilization of the RhoGDI α – Rho GTPase complex, which maintains the Rho GTPase in an inactive, cytosolic GDPbound state (19–22) It also mediates the extraction of the GDP-bound Rho GTPases from the membrane by sequestering the GG chain, which results in the cycling of the Rho GTPases back into the cytosol as well as to other membrane compartments (23, 24).

In the present study, we investigated whether alcohols and anesthetics interact with the hydrophobic binding pocket within the C-terminal domain of RhoGDI α and assessed ensuing effects on GG chain binding to this pocket. Two strategies were used to locate anesthetic sites. First, we took advantage of the presence of two tryptophans within the GG chain binding pocket (W192 and W194) to monitor the interaction of the halogenated anesthetic, halothane, based on quenching of the emission fluorescence of tryptophan, a method utilized previously in studies of anesthetic binding to serum albumins (25, 26), and α -helical bundle proteins (27, 28). Second, we employed photo-labile alcohols to locate the site of alcohol action (6, 29–32).

¹Abbreviations: AGGC, N-acetyl-geranylgeranyl cysteine; FC, flow cell; FRET, fluorescence resonance energy transfer; GMPPNP, guanosine 5'-[β , γ -imido] triphosphate trisodium salt; HAF, 5-hexadecanoylaminofluorescein; Halothane, 2-bromo-2-chloro-1,1,1-trifluoroethane; LUV, large unilamellar vesicles; MANT-GDP, 3'-O-(N-methylanthraniloyl)-guanosine 5'-diphosphate trisodium salt; MANT-GMPPNP, 3'-O-(N-methylanthraniloyl)-guanosine 5'-diphosphate trisodium salt; MANT-GMPPNP, 3'-O-(N-methylanthraniloyl)-guanosine 5'-triphosphate trisodium salt; POPC, *I*-palmitoyl-2-oleoyl-*sn*-glycero-3-phosphocholine; POPS, *I*-palmitoyl-2-oleoyl-*sn*-glycero-3-phosphoserine; RhoGDI α , α -isoform of Rho GDP-dissociation inhibitor; SPR, surface plasmon resonance.

Materials

Grace's insect media, phosphate buffered saline (PBS), fetal bovine serum (FBS), 5hexadecanoylaminofluorescein (HAF), 3'-O-(N-methylanthraniloyl)-β:γ-imidoguanosine 5'triphosphate trisodium salt (MANT-GMPPNP), 2'-(or-3')-O-(Nmethylanthraniloyl)guanosine 5'-diphosphate, disodium salt (MANT-GDP) and Spodoptera frugiperda (Sf9) cells, and primers were each obtained from Invitrogen Technologies (Carlsbad, CA). The L1 Sensor Chip was purchased for use in a BiacoreTM 2000 surace plasmon resonance (SPR) system (Biacore Inc, Piscataway, NJ). Protease Inhibitor Cocktail, guanosine 5'- $[\beta, \gamma$ -imido] triphosphate trisodium salt (GMPPNP), halothane (2-bromo-2chloro-1,1,1-trifluoroethane), n-alkanols, and all other research grade chemicals were from Sigma-Aldrich (St. Louis, MO). The n-alkanol concentration ranges used were normalized on the basis of alcohol hydrophobicity, estimated from octanol/water partition coefficients, as previously described (7). N-acetyl geranylgeranyl cysteine (AGGC) was from Calbiochem (La Jolla, CA); 1-palmitoyl-2-oleoyl-sn-glycero-3-phosphocholine (POPC) and 1-palmitoyl-2-oleoyl-sn-glycero-3-phosphoserine (POPS) were from Avanti Polar Lipids Inc (Alabaster, AL). The photo-labile alcohol, 3-azibutanol was synthesized according to the method of Church et al (33), and 3-azioctanol was synthesized as described previously (34). Baculovirus encoding Human Cdc42 N-terminal tagged with 6xHis was a kind gift from Dr. R. A. Cerione (Department of Molecular Medicine, Veterinary Medical Center, Cornell University, NY); baculovirus encoding 6xHis tagged RhoA was prepared as described previously (35). Baculovirus containing RhoGDIa N-terminal tagged with GST was custom manufactured by Orbigen (San Diego, CA). Molecular graphics images were produced using the UCSF Chimera package (36) from the Resource for Biocomputing, Visualization, and Informatics at the University of California, San Francisco (supported by NIH P41 RR-01081).

Purification of RhoGDIa

Log phase Sf9 cells in 1 L of Grace's insect media and 10% v/v FBS were infected with baculovirus encoding RhoGDIa-GST at a multiplicity of infection of 5 and incubated for 4 days. The cells were then washed once with PBS, pelleted, and lysed using a Dounce homogenizer in Buffer A (10 mM HEPES, pH 8.0, 150 mM NaCl, 5 mM MgCl₂) containing protease inhibitor cocktail. The cell lysate was homogenized and centrifuged at 3500g for 15 min at 4°C in a Sorvall SS-34 rotor to remove whole cell debris and nuclei. The resultant supernatant was centrifuged at 100,000g for 60 min in a Beckman Ti70 rotor to separate cytosolic from membrane and cytoskeleton fractions. The cytosolic fraction (supernatant) was then incubated at 4°C for 60 min with gentle rocking with 1 mL of glutathione agarose resin that had been pre-equilibrated in Buffer A. The resin was then packed into a 5 mL column attached to an ÄKTApurifier[™] FPLC system (GE Healthcare, Piscataway, NJ) and washed with Buffer A until the absorbance of the flow-through at 280 nm was <0.01 absorbance units. RhoGDIa was liberated from the GST tag by thrombin cleavage while the fusion protein was still attached to the glutathione agarose resin. Cleavage was carried out by passing 0.1 units/mL α -thrombin in Buffer A though the column at 25°C for 2 hrs. The fractions containing protein were pooled and applied to a size exclusion column in order to

remove pro-thrombin and α -thrombin. The pooled eluted protein was dialyzed against Buffer A for 36 hours with buffer changes every 12 hours. Protein concentrations were determined by Bradford protein assay (BioRad, Faraday, CA) and protein were stored at -80°C in Buffer A containing 20% glycerol. The concentrated RhoGDI α stock was diluted for experiments with either Buffer A or Buffer B (20 mM Tris/HCl, pH 7.5, 5 mM MgCl₂, 50 mM NaCl).

Purification of RhoGDIa in complex with Rho GTPases

In order to circumvent problems arising from the low aqueous solubility of the Rho GTPases that results from the hydrophobic nature of the C-terminal GG chain, Cdc42 and RhoA were expressed and purified as a 1:1 stoichiometric complex with RhoGDIa. In this complex the geranylgeranyl chain of each Rho GTPase is shielded from the aqueous environment by insertion into the hydrophobic binding pocket on RhoGDIa (16, 17). The complex was purified as described previously (35), except that the Rho GTPase–RhoGDIa complexes were liberated from the GST tag by thrombin cleavage while the fusion protein was still attached to glutathione–agarose. Cleavage was carried out by passing 0.1 units/mL a-thrombin in Buffer A though the column at 25°C for 2 hrs. The fractions containing protein were pooled and applied to a size exclusion column in order to remove pro-thrombin and a-thrombin.

Halothane binding assay

The interaction of halothane with RhoGDIa was determined from measurements of the corresponding quenching of the intrinsic emission tryptophan fluorescence, as described previously (25, 26). Briefly, RhoGDIa (500 nM) was placed into a screw-top sealed 3 mL cuvette with 2.8 mL of Buffer B. The calculated amount of halothane (0-800 µM) was then added to the cuvette in a volume calculated to minimize headspace, which was then promptly sealed. After incubation for 60 min at room temperature, the steady-state emission spectrum of tryptophan was collected using an ISS upgraded SLM 48000 (ISS, Champaign, IL). The excitation wavelength (295 nm) was chosen so as to minimize the contribution of energy transfer from tyrosine residues to the observed fluorescence signal (37). The effects of AGGC on halothane binding was determined by incubating the required concentrations of AGGC (0–300 μ M) with 500 nM RhoGDIa in the presence or absence of 1 mM halothane in 0.5 mL Buffer B for 30 min in gas-tight Hamilton syringes. The mixture was then diluted into 2 mL of Buffer B containing 500 nM RhoGDIa in a 3 mL cuvette, which was promptly sealed prior to recording the tryptophan emission spectra. For determinations of the effects of n-alkanols on halothane binding, the required concentration of halothane was initially added to RhoGDIa (500 nM) in a screw-top sealed 3 mL cuvette with 2.8 mL Buffer B, and the fluorescence intensity at 330 nm upon excitation at 295 nm was monitored as a function of time. After allowing equilibration, n-alkanols were titrated into the cuvette in 10 µL aliquots from concentrated stock solutions in Buffer B. In a separate experiment the operation and timing of each alcohol addition was mimicked by identical additions of Buffer B, in order to control for the effects of halothane evaporation on the fluorescence signal. The increase in baseline fluorescence, due to evaporation of halothane and loss of tryptophan quenching, was found to be <5% of the initial quenched signal. The contribution of inner filter effects to the observed signal was judged to be negligible, based on the observation

that halothane had a molar extinction coefficient of $<100 \text{ M}^{-1} \text{ cm}^{-1}$ at 295 nm (37). The same batch of RhoGDIa was used for each individual halothane quenching experiment, because some batch-to-batch variability was apparent in the extent of tryptophan quenching.

Preparation of Vesicles

Large unilamellar vesicles (LUV) were prepared from lipids as described previously (38). Briefly, chloroform solutions of POPC and POPS (4:1, molar) at a total lipid concentration of 500 μ M were mixed in a test tube and the solvent was evaporated under a stream of nitrogen to form a homogenous thin film. The required volume of buffer (10 mM HEPES, pH 7.4 and 150 mM NaCl) was then added and the lipids were allowed to hydrate for 15 min at 25°C. Following this, multilamellar vesicles were formed by vortexing for 1 min. Large unilamellar vesicles of 100 nm diameter (LUV) were prepared from multilamellar vesicles by the extrusion technique, using an Avestin Liposofast Extruder (MM Developments, Ottawa, Canada), also as previously described (39).

Surface Plasmon Resonance determinations

The dissociation of Cdc42 or RhoA from the complex with RhoGDIa was quantified based on the time dependence of the association of the Rho GTPases with membranes immobilized on the hydrophobic surface of an L1 sensor chip in a Biacore[™] 2000 (Biacore, Inc., Piscataway, NJ), as described previously (35). Briefly, the surface of the L1 sensor chip was initially cleaned by 2 injections of 10 mM CHAPS and POPC/POPS LUV were then captured on this surface using a flow rate of 5 μ L/min and a contact time of 15 min. All measurements were performed at 25°C using a running buffer consisting of 10 mM HEPES, pH 7.4 and 150 mM NaCl. Previous studies have shown that the L1 chip surface is completely covered under the conditions used and that the resultant lipid surface resembles a rough bilayer structure (40, 41). The POPC/POPS surface was then conditioned with sequential injections of 10 mM glycine (pH 1.5) and 10 mM NaOH to remove loosely associated vesicles and to minimize baseline drift. The RhoA-RhoGDIa or RhoGDIa-Cdc42 complex was then diluted to the required final concentration, which was typically 0.5 µM unless otherwise stated, in Buffer A containing 50 µM GMPPNP and the required concentration of n-alkanol. Following this, the free Mg²⁺ concentration was reduced from 5 mM to <0.5 µM by chelation with 14.3 mM EDTA in order to trigger nucleotide exchange and initiate complex dissociation. The resultant mixture was then immediately injected over the POPC/POPS surface at a flow rate of 10 μ L/min, and the increase in the SPR signal (response) due to capture of geranylgeranylated Rho GTPases was monitored as a function of time. The L1 chip surface was regenerated with three 10 µL injections of CHAPS at a flow rate of 5 µL/min.

Analysis of fluorescence quenching data

The extent of tryptophan quenching by halothane (*Q*) was calculated from the fractional decrease in fluorescence intensity, F_Q / F_0 , where F_Q and F_0 are the tryptophan fluorescence intensities measured with and without halothane, respectively. These values were determined by averaging >10 data points at the corresponding emission wavelength maximum. The affinity constant (K_D) and Hill coefficient (*h*) for halothane binding were

obtained by plotting values of F_Q / F_0 against halothane concentration and fitting this data to the following modified Hill equation using non-linear least-squares regression analysis:

$$F_{\rm Q}/F_0 = (F_{\rm max}/F_0).([{\rm halothane}]^h/\{K_{\rm D}{}^h + [{\rm halothane}]^h\})$$
 (Eq. 1)

where F_{max} is the fluorescence intensity at maximal quenching. The effects of n-alkanols and AGGC on halothane binding to RhoGDIa were determined from plots of the recovery in tryptophan fluorescence as a function of ligand concentration. Fluorescence recovery was equal to $(F_{\rm L} - F_0) / (F_0 - F_{\rm H})$ where, F_0 is the tryptophan fluorescence intensity without halothane or ligand (n-alkanol or AGGC), $F_{\rm H}$ is the intensity with halothane, and $F_{\rm L}$ is the intensity with halothane and ligand. The concentration of ligand required for a 50% recovery in tryptophan fluorescence (EC_{50}) and the maximal extent of fluorescence recovery ($F_{\rm max}$) was calculated from fits to the following expression using non-linear least-squares regression analysis:

Fluorescence recovery=
$$F_{\text{max}}$$
.([ligand]^h/{ EC_{50}^{h} +[ligand]^h}) (Eq. 2)

The apparent inhibitory constants for each ligand (K_{I}) were determined from the concentration dependent effects of the ligands on the K_{D} for halothane binding, according to the following expression (42):

$$K_I = EC_{50}/(1 + [\text{halothane}]/K_D)$$
 (Eq. 3)

The estimated incremental changes in the apparent free energy of n-alkanol binding upon addition of a methylene unit were calculated from values of K_{I} obtained for each alcohol using Eq. 3, according to:

$$\Delta\Delta G = -RT \ln(K_{I,n}/K_{I,n+1}) \quad \text{(Eq. 4)}$$

Where R is the universal gas constant, T is the temperature (Kelvin), and *n* is the n-alkanol chain length. Goodness of fit was assessed from values of average squared residual (χ^2).

Analysis of SPR data

Whereas the BiacoreTM system allows affinity constants to be determined from the ratio of association and dissociation rate constants, the rate of dissociation of Cdc42 from the POPC/POPS membranes was found to be too slow for accurate determination of a dissociation rate constant. The binding constant for the interaction of Cdc42 with membranes was therefore determined from measurements of the extent of Cdc42 binding at equilibrium ($R_{eq,Cdc42}$) as a function of the concentration of RhoGDIa–Cdc42 complex injected (35). Values of $R_{eq,Cdc42}$ were obtained by fitting the response (R_t) verses time curves for each analyte concentration to an integrated first-order rate equation using non-linear regression analysis:

$$(R_{\rm t} - R_{\rm eq.Cdc42}) = (R_0 - R_{\rm eq.Cdc42}) \cdot [1 - \exp(-k_{\rm obs} \cdot t)]$$
 (Eq. 5)

Where R_0 is the initial response and k_{obs} is the observed first order rate constant. Values of $R_{eq,Cdc42}$ as a function of n-alkanol concentration were then fitted to a modified Hill equation:

$$R_{\rm eq} = R_0 + R_{\rm max} \cdot \left(\left[n - \text{alkanol} \right]^h / \left\{ K_{\rm app}^h + \left[n - \text{alkanol} \right]^h \right\} \right) \quad \text{(Eq. 6)}$$

Where R_0 is the initial response in the absence of analyte, R_{max} is the response corresponding to maximal occupation of membrane binding sites, K_{app} is the apparent association constant, and *h* is the Hill coefficient. Goodness of fit was assessed from values of average squared residual (χ^2).

Extraction of Cdc42 from membranes by RhoGDIa.

The extraction of Cdc42 from membranes by RhoGDIa was determined from the accompanying decrease in fluorescence energy transfer (FRET) between the MANTfluorophore of MANT-GMPPNP bound to Cdc42 and the membrane probe, HAF, based on a previously described method (23, 43). For these experiments, Cdc42 was expressed in Sf9 cells and purified in the absence of RhoGDIa using procedures that were identical to that described above for the RhoGDIa–Cdc42 complex, except that the purification by glutathione affinity chromatography was omitted. Cdc42 was initially loaded with MANT-GMPPNP as described previously (43). Briefly, 14.3 mM EDTA was added to 60 nM Cdc42 in Buffer B containing 2 µM MANT-GMPPNP, which triggers nucleotide exchange by reducing the free Mg²⁺ concentration from 5 mM to $<0.5 \mu$ M. To this were added LUV composed of POPC/POPS (4:1, molar) containing 2 µM HAF (300 µM total lipid concentration) and 50 mM n-butanol as required, and the resultant solution was allowed to equilibrate for 30 min. The emission fluorescence intensity of MANT-GMPPNP was then measured at 25°C as a function of time at 430 nm upon excitation at 355 nm using an ISS modified LSM 48000 multi-frequency phase and modulation fluorimeter. Upon attaining a stable baseline, the extraction of membrane bound Cdc42 was initiated by the addition of 60 nM RhoGDIa in Buffer B, and monitored by measuring the recovery in the emission fluorescence intensity due to the de-quenching of MANT-GMPPNP as a function of time.

Fluorescence-based nucleotide exchange assay

The effects of n-alkanols on the intrinsic nucleotide exchange rates were determined based on a previously described method (23). Briefly, Cdc42 (100 nM) was incubated with 300 nM MANT-GDP in the presence and absence of the required concentration of n-alkanol for 30 min in 1.5 ml of Buffer B, and the MANT fluorescence intensity at 430 nm upon excitation at 360 nm was measured as a function of time. Nucleotide exchange was initiated by adding 2 μ M GDP in the presence of 16 mM EDTA and the resultant fluorescence intensity decrease was measured as a function of time.

Photo-affinity labeling

RhoGDIa (2.8 nM) in water (150 μ L) was incubated for 30 min without or with AGGC (550 nM) in the presence of 3-azibutanol or 3-azioctanol, which was added from an aqueous solution to yield final alcohol concentrations of 0.01, 0.1 and 1.0 mM in a total reaction volume of 210 μ L. This solution was then irradiated on ice at 365 nm using a hand held

Black-Ray UV lamp model UVL-56 at a distance of ~ 1 cm for 45 min in a 96-well glass plate. The photoreaction was then quenched by the addition of SDS-PAGE sample buffer.

In-gel tryptic digestion

Approximately 10 µg of photolabeled RhoGDIa was separated by SDS-polyacrylamide gel electrophoresis using a 10% polyacrylamide gel followed by Coomassie blue staining. The band corresponding to the molecular weight of RhoGDIa was cut into ~2 mm³ pieces and incubated overnight at room temperature with 200 µL of 50 mM ammonium bicarbonate solution, and then washed with acetonitrile for 10 min. The solvent was then removed under vacuum and the gel pieces were rehydrated with aqueous 50 mM ammonium bicarbonate solution containing 12.5 ng/µL modified sequencing-grade trypsin (Promega, Madison, WI) at 4°C. After 45 min, the excess trypsin solution was removed and replaced with a volume of 50 mM ammonium bicarbonate solution that just covered the gel pieces, which were then incubated at 37°C overnight. Peptides were then extracted from the gel pieces by first removing the ammonium bicarbonate solution and then performing two consecutive 20 min incubations in a solution composed of 50% v/v acetonitrile and 5% v/v formic acid. The extracts were combined, dried under vacuum for ~1 hr and stored at 4°C until analysis.

Online HPLC and Mass Spectrometry

LC-MSMS experiments were performed on an LTQ-FT mass spectrometer (ThermoFinnigan, San Jose, CA). This is a fully integrated hybrid mass spectrometer consisting of a linear ion trap mass spectrometer combined with a Fourier transform ion cyclotron resonance mass spectrometer. Peptides obtained from tryptic digestion were initially reconstituted in 5 μ L of an aqueous solution composed of 2.5% v/v acetonitrile, 0.1% v/v formic acid. A nano-scale reverse-phase HPLC capillary column was created by packing 5 μ m C18 spherical silica (pore size 200 Å) into a 12 cm fused silica capillary tube of 75 μ m inner diameter with a flame-drawn tip fabricated in-house (44). The column was attached to the HPLC system and peptides were eluted by a linear concentration gradient (acetonitrile with 0.1% formic acid increasing from 10 to 38% over 30 min followed by a step increase to 100%. Each eluted peptide was subjected to electrospray ionization. The m/z of each isolated peptide was determined to high accuracy by FTMS. The remainder of the sample was fragmented and its tandem mass spectrum (MSMS) was analyzed to determine its sequence. The position of photo-incorporation was analyzed using the Bioworks browser 3.1 module of the program Xcalibur (ThermoFinnigan, San Jose, CA).

RESULTS

Direct interaction of halothane with RhoGDIa.

The fluorescence emission spectra of RhoGDIa, obtained upon excitation at 295 nm in the absence of halothane, consisted of an asymmetric peak centered at 327 nm (Figure 1). Consistent with a direct interaction of halothane with RhoGDIa, incubation of the protein with increasing levels of the anesthetic resulted in a quenching of fluorescence emission intensity. The anesthetic also induced a slight dose dependent blue shift in the emission maxima (~2 nm), indicating a small decrease in the dielectric local to the tryptophan residues of RhoGDIa (Figure 1). The tryptophan fluorescence intensity decreased as a

function of halothane concentration to a plateau value corresponding to ~30 % of the original signal (Figure 1, *inset*, •). Fitting this hyperbolic curve to a modified Hill equation (Eq. 1), yielded values of $K_D = 147\pm7 \mu$ M, and $h = 0.91\pm0.2$, suggesting the presence of one or more non-interacting halothane binding sites on RhoGDIa. The quenching effect of halothane on free L-tryptophan fluorescence intensity was reduced compared to that on RhoGDIa tryptophan fluorescence (Figure 1, *inset*, •), and was a linear function of anesthetic concentration, contrasting with the saturating effect of halothane. Fitting this data to the Stern-Volmer equation: $F_Q/F_0 = 1 + K_{SV}$ [halothane], using linear regression, yielded a quenching constant, $K_{SV} = 15.7\pm0.2$ M⁻¹, which is close to a previous value (26). The possibility that the observed decrease in tryptophan resonance energy transfer (37), was ruled out by the finding that the halothane concentration-response curve was unaffected upon increasing the excitation wavelength to 305 nm (Figure 1, *inset*, \bigcirc).

AGGC inhibits halothane binding to RhoGDIa.

In order to address the question whether the interaction of halothane with RhoGDIa competes with GG chain binding, the quenching of RhoGDIa tryptophan fluorescence by the anesthetic was determined in the presence of increasing levels of the ligand, AGGC. Although lacking the methoxy group that is attached during post-translational modification, this compound mimics the geranylgeranylated C-terminal cysteine of Rho GTPases and has been shown to bind specifically to RhoGDIa (18). Consistent with the results shown in Figure 1, RhoGDIa tryptophan fluorescence was quenched by 1 mM halothane (Figure 2). The titration of AGGC elicited a saturable decrease in the extent of tryptophan quenching (Figure 2). Plotting the fluorescence intensities at each emission maxima against AGGC concentration yielded a hyperbolic concentration-response curve (Figure 2, inset). Fitting this data to Eq. 2 yielded values of $EC_{50} = 65 \pm 15 \mu M$, $h = 1.2 \pm 0.5$ and $F_{max} = 0.5 \pm 0.1$. The value of $K_{\rm I}$ for the displacement of halothane by AGGC, obtained by substituting EC_{50} into Eq. 3, was 10 μ M, which is close to the apparent dissociation constant for this compound reported in a previous study (18). The possibility that the apparent de-quenching effect of AGGC might have resulted from a change in the local environment of the tryptophans upon binding of the ligand was ruled out in separate control experiments showing that tryptophan fluorescence and emission maxima wavelength were unaffected by AGGC in the absence of halothane (results not shown).

Effects of n-alkanols on halothane binding

The homologous series of n-alkanols of increasing chain length (*n*) are anesthetics and have been used to probe the geometry and structure of anesthetic binding sites within a variety of proteins (9, 45–49). The addition of increasing levels of n-alkanols of n=2 to 5 to RhoGDIa in the presence of halothane (0.2 mM) in each case resulted in a saturable and concentration dependent decrease in the extent of tryptophan fluorescence quenching (Figure 3A–3D). By contrast, control experiments indicated that the tryptophan emission fluorescence of RhoGDIa measured without halothane was unaffected by these n-alkanols within the concentration ranges used (results not shown). Strikingly, the effect of n-hexanol did not reach saturation within the concentration range used and n-hepthanol had negligible effects on halothane-induced tryptophan quenching (Figure 3E– 3F). In keeping with this, values of

 EC_{50} , calculated from fits of the corresponding dose-response curves to Eq. 2 (Table 1), decreased as a non-linear function of n (Figure 4), as revealed by a plot of EC_{50}/C_{sat} against *n* (Figure 4, *inset*). By contrast, values of F_{max} were independent of *n* (Table 1). The apparent non-linear relationship between EC_{50} and n was also evident in the values of K_{I} for the displacement of halothane by each n-alkanol, obtained from Eq. 3 (Table 1), and also by the estimated incremental binding free energy change (G) upon addition of a methylene G was -3 kJ mol⁻¹ for n=2 to 3, which unit, calculated using Eq. 4. Thus, the value of compares favorably with the value of -2 kJ mol^{-1} predicted for the partitioning of a methylene group to a hydrophobic site within a protein (50). By contrast, the values of G were -5 kJ mol^{-1} for n=3 to 4, and $+4 \text{ kJ mol}^{-1}$ for n=4 to 5, indicating a respective increase and decrease in binding energy relative to those predicted based on hydrophobic interactions alone. The Hill coefficients obtained from fits of the corresponding dose-response data (Figure 3), were found to be \sim 1 and independent of *n*, indicating that each of the n-alkanols interact with at least one equivalent binding site.

In order to further investigate the mechanism underlying the displacement of halothane binding by n-alkanols, the concentration-response curves for n-butanol induced tryptophan fluorescence recovery were determined in the presence of increasing levels of halothane (Figure 5). It was found that the values of EC_{50} , obtained from fits of each dose-response curve to Eq. 2, increased as a function of halothane concentration, whereas values of F_{max} were independent of halothane concentration and corresponded to ~50% recovery in halothane induced quenching (Table 2). The calculated values of h were found to be ~1 and also independent of halothane concentration.

Azialcohols photo-affinity label RhoGDla

In order to provide further evidence for an alcohol and anesthetic binding site on RhoGDIa and to identify residues that may contribute to its structure, photo-affinity labeling was performed utilizing the diazirinyl n-alkanols, 3-azibutanol and 3-azioctanol (6, 29–32). RhoGDIa was photolabeled with and without AGGC in the presence of 10 and 100 μ M azialcohol in two independent experiments, digested with trypsin, and subjected to mass spectrometry. Two photolabeled peptides were generated in the presence but not in the absence of AGGC, at 100 μ M 3-azibutanol, which was the lowest concentration at which photolabeling was detected. This indicates that AGGC enhanced the photoincorporation of 3-azibutanol. The characterization of these peptides is shown in Table 3 (Peptides 1 and 2). Photolabeling of RhoGDIa with 3-azioctanol yielded 1 peptide, which was generated in the presence and absence of AGGC, and at 3-azioctanol concentrations as low as 10 μ M (Table 3; Peptide 2). The percentage of RhoGDIa that was sequenced was 79.4% (Table 4), so there is a small chance that other photolabeled peptides were missed.

Photolabeling of E193

The peptides identified above were fragmented by collision with an inert gas to yield peptide fragments with an intact N-terminus, called b-ions, or an intact C-terminus, called y-ions. The MSMS pattern for Peptide 1 obtained from photolabeling with 100 μ M 3-azibutanol in the presence of AGGC is shown in Figure 6. The site of photoincorporation was identified as E193 based on the observed run of strong y-ions from y10 to y5 with loss of modification

between y7 and y6. This assignment was confirmed by a run of doubly charged b-ions from $b17^{++}$ to $b9^{++}$ with loss of photoincorporation between $b13^{++}$ and $b12^{++}$. This residue was not photolabeled by 3-azioctanol at levels up to 1 mM.

Photolabeling of E163

The MSMS pattern for Peptide 2 obtained from photolabeling of RhoGDIa with 10 μ M 3azioctanol is shown in Figure 7. The doubly charged peptide of m/z = 1733.81 yielded a sequence of y-ions from y12 to y3 with loss of modification between y5 and y4, thus identifying E163 as the site of photoincorporation of 3-azioctanol. This was confirmed by the presence of series of strong b-ions from b13 to b4 (except b9), where modification was lost between b11 and b10. This residue was identified in the presence and absence of AGGC. The corresponding residue photolabeled by 3-azibutanol was only labeled in the presence of AGGC (Figure 8).

n-Alkanols enhance Rho GTPase–RhoGDla complex dissociation

Rho GTPase activation is mediated by the dissociation of the Rho GTPase - RhoGDIa complex, which facilitates the association of the RhoGTPases with membranes (Figure 9A). Since this process requires the transfer of the GG chain from the C-terminal binding pocket within RhoGDIa to the membrane interior (19), we investigated the effects of n-alkanols on the kinetics of the dissociation of Cdc42 from its complex with RhoGDIa. The dissociation of the RhoGDIa–Cdc42 complex was triggered by GDP-GMPPNP exchange on Cdc42 induced by chelating the Mg^{2+} ion that serves to stabilize GDP binding (23, 51). Consistent with the results of our previous study (35), the injection of Cdc42 in complex with RhoGDIa in the presence of GMPPNP and excess EDTA resulted in a time dependent increase in binding to the captured POPC/POPS membranes (Figure 9B).

The addition of increasing levels of n-butanol prior to injection of the RhoGDIa-Cdc42 complex over the membrane surface, resulted in a concentration dependent increase in the value of R_{eq} for Cdc42 binding (Figure 9B). Plotting values of $R_{eq,Cdc42}$ as a function of ethanol (Figure 9C) or n-butanol concentration (Figure 9D, •), in each case yielded a hyperbolic curve. Fitting these data to Eq. 6 yielded values of $K_{app} = 50\pm21$ mM and 7.5 ± 1.3 mM for ethanol and n-butanol, respectively, which are close to the corresponding values of $K_{\rm I}$ obtained for the displacement of halothane binding by each n-alkanol. Notably, n-butanol had negligible effects on the association of Cdc42 with membranes in the absence of EDTA and GMPPNP (Figure 9D, O). Consistent with the non-linear chain length dependent effects of the n-alkanols on halothane binding (Figure 3), the association of Cdc42 with membranes was unaffected by n-hexanol (Figure 9E) and n-heptanol (Figure 9F). The concentration dependent effects of 3-azibutanol on $R_{eq,Cdc42}$, which were found to be similar to that obtained for n-butanol (Figure 9D, □), thus validating the use of this alcohol in photolabeling experiments (see above). The observed non-linear chain length dependent effects of the n-alkanols on the membrane association of Cdc42, argues against the possibility that the observed effects resulted from a non-specific perturbation of the membrane structure, since such effects have been shown to be a *linear* function of chain length within the range used in the present study (52).

n-Alkanols inhibit the extraction of Cdc42 from membranes by RhoGDla

It was hypothesized that the extraction of Cdc42 from membranes by RhoGDIa would be inhibited by n-alkanols, since this process involves transfer of the GG chain from the membrane interior to the GG chain binding pocket. In order to address this, the extraction of membrane-bound Cdc42 by RhoGDIa was determined from measurements of FRET between the MANT-donor fluorophore of MANT-GMPPNP loaded onto Cdc42 and the membrane-associated acceptor fluorophore of HAF (35, 43). Consistent with the results of our previous study (35), the addition of POPC/POPS (80:20, molar) labeled with 2 µM HAF quenched the emission fluorescence intensity of MANT-GMPPNP loaded on Cdc42 by ~50% (results not shown). The addition of RhoGDIa (60 nM) resulted in a rapid increase in MANT emission fluorescence due to a decrease in FRET (Figure 10), as reported previously (35, 43). Importantly, the presence of n-butanol (50 mM) attenuated the de-quenching effect of RhoGDIa on MANT fluorescence, suggesting a corresponding reduction in the membrane extraction of Cdc42 in the presence of the n-alkanol (Figure 10). The possibility that the potentiated level of Cdc42 membrane association induced by ethanol and n-butanol may have resulted from modified nucleotide exchange kinetics was ruled out by the finding that neither the rate nor the extent of exchange of MANT-GDP for GDP were affected by 50 mM butanol (results not shown).

DISCUSSION

In this study, we addressed the hypothesis that anesthetics compete with endogenous ligand binding to hydrophobic sites within mammalian proteins. Evidence supporting this was provided by the finding that halothane and n-alkanols bound to sites contained within RhoGDIa and that interact with GG chain binding. However, the results also suggest that halothane and n-alkanols bind to distinct sites within RhoGDIa and that differing mechanisms underlie the effects on GG chain binding. In the case of halothane, the simplest explanation for our data is that halothane bound in the GG binding pocket because its effect was inhibited by the GG chain analog, AGGC. However, this appears not to be the case for the short chain alcohols because AGGC enhanced their association with RhoGDIa. Below, we first critically assess the experimental evidence for these assertions and then discuss the wider implications of the results of this study for the molecular mechanisms of anesthetic action.

Evidence supporting the existence of a discrete halothane binding site within RhoGDIa was provided by the observation that halothane quenched the intrinsic tryptophan fluorescence of RhoGDIa in a concentration dependent and saturable manner. The quenching of tryptophan fluorescence by the heavy atom bromine in halothane (CF₃CHClBr) is a short-range effect (~5 Å) (37). Structural studies have shown that W192 and W194 are contained within the Cterminal Rho GTPase GG chain binding pocket, and the indole side-chain of W194 makes contact (~ 3.5 Å) with carbons C15 and C17 of the GG chain (16, 17, 53). Although the indole ring system of W194 is in van der Waals contact, W192 and W204 are more distant from the GG chain (8–10 Å from the terminal methyl C19/20 atoms), and are shielded from the GG chain by L88 and L86 respectively. The partial observation quenching of tryptophan fluorescence by halothane observed in the present study has also been reported previously

for the beta-barrel protein porcine odorant binding protein (54). The partial quenching may have resulted from an interaction of halothane with a subset of the 3 tryptophan residues or a rapid and transient interaction with all 3 residues. Fluorescence lifetime measurements would be required to address this, which is beyond the scope of this manuscript. Taken together, these observations suggest that the halothane binding site resides within a region of the molecule that is occupied by the GG chain, perhaps close to W194. Consistent with this hypothesis, AGGC inhibits the ability of halothane to quench the intrinsic tryptophan fluorescence of RhoGDIa. However, it is not possible to rule out a negative heterotrophic interaction of halothane with a site that is outside the GG chain binding pocket but close to one of the tryptophans. Furthermore, the finding that AGGC only partially reversed halothane–induced fluorescence quenching (Figure 2) indicates either that halothane binds at more than one site, or that AGGC exerts an allosteric action on the halothane site.

The finding that n-alkanols inhibited the quenching of intrinsic tryptophan fluorescence by halothane provides evidence that these agents also interact with RhoGDIa. The alkanol chain length dependence of this inhibitory effect indicates that the interaction of n-alkanols with RhoGDIa is highly specific, with a chain length cut-off at n=6, and an optimal chain length for binding at n=4. The finding that the apparent EC_{50} for this action increased with halothane concentration (Figure 5), suggests that the displacement of halothane is mediated either by a competitive, or a negative allosteric mechanism. Distinguishing between these two possibilities kinetically would require the EC_{50} for the displacement of halothane by nalkanols to be determined as a function of halothane concentrations that span several orders of magnitude, which for technical reasons is not experimentally feasible. However, the observation that n-alkanols only reversed a fraction of the halothane induced fluorescence quenching and this fraction varied with chain length (Figure 3), strongly suggests an allosteric interaction between the halothane and alcohol sites. Furthermore, it is unlikely that the n-alkanols bind to a site within the GG-binding pocket, because none of the photolabeled residues residued within this pocket, and AGGC enhanced photolabeling by 3azibutanol.

The observation that 3-azibutanol photolabeled both E163 and E193 at the lowest concentration used, and that this occurred only in the presence of AGGC, suggests that the short chain n-alkanols bind to allosteric sites in the vicinity of both of these residues. In order to determine which residue contributes to the allosteric action on halothane–induced fluorescence quenching, we utilized 3-azioctanol, which possesses a chain length that is too long to reverse the quenching effect of halothane. The finding that E193 was *not* photolabeled by 3-azioctanol even at the highest concentration used, indicates that the short chain n-alkanol binding site that allosterically reverses fluorescence quenching is in the vicinity to E193 and not E163. Whereas 3-azioctanol photolabeled E163 at the lowest concentration examined (10 μ M), this photoincorporation was unaffected by AGGC, unlike 3-azibutanol. This prompts the intriguing prediction that the extra four carbon atoms stabilize binding in a conformationally–insensitive fashion.

The solution state structure of RhoGDIa has been determined by NMR in both Rho GTPase–bound and –free forms (17, 53), and the structures of the free C-terminal domain of RhoGDIa (55, 56) and the full length protein bound to Rho GTPases have been solved by

X-ray crystallography at a resolution of 2.6 Å (16, 57, 58). W192 and W194 are on the β 9strand facing the GG pocket and E193 is on the surface of the protein, facing in the opposite direction (see Figure 11). In a previous study, the effect of the insertion of the GG chain into the GG chain binding pocket was assessed by comparing the structure of the free C-terminal globular fold of RhoGDIa with that of the Cdc42-bound full length protein (16). Insertion of the GG chain induces a structural rearrangement that moves the indole ring nitrogen (N ϵ) of W194 1.8 Å, and in doing so moves the backbone in this region 1.0 Å outwards. This structural shift also results in a large displacement of E193, the site of photolabeling by 3azibutanol, and radically changes the surface topology at this residue. Although speculative, this change in the orientation of E193, and in its local topology, might well account for the observed enhancement in the degree of photoincorporation of 3-azibutanol in the presence of AGGC. Structural determinations of RhoGDIa in complex with AGGC are required to resolve this issue fully.

The *initiating* event in Rho GTPase mediated signaling corresponds to the release of the Rho GTPases from the inactive cytosolic complex, which is driven by the transfer of the GG chain from the binding pocket within RhoGDI α into the membrane (51). Conversely, the extraction of Rho GTPases from membranes by RhoGDIa terminates Rho GTPase signaling, and results from the transfer of the GG chain from the membrane interior into the binding pocket within RhoGDIa to form the inactive cytosolic complex (12, 16). The observation that ethanol, n-butanol and 3-azibutanol each potentiated the extent of Cdc42 binding to membranes at equilibrium, and that n-butanol inhibited the extraction of Cdc42 from membranes (Figures 9 and 10), suggests that the interaction of these agents with RhoGDIa promotes the dissociation of the RhoGDIa-Cdc42 complex and thus acts on both the initiating and terminating steps in Rho GTPase signaling. Furthermore, the observed lack of an effect of n-hexanol and n-heptanol on the membrane-association of Cdc42 is reminiscent of the non-linear chain length dependent effects of the n-alkanols on halothane binding. Although this suggests the alcohol site that acts on complex stability may be similar to that which mediates the displacement of halothane binding, our photolabeling studies were based on the RhoGDIa and its complex with AGGC, not on the full RhoGDIa-Cdc42 complex. It would be premature to draw conclusions about the nature and location of the alcohol site on the RhoGDIa-Cdc42 complex. However, it is interesting to note the 3azibutanol site at E163 is in contact with R186 of Cdc42 (16).

Along with the insertion of the GG chain into its binding pocket, the RhoGDIa–Cdc42 complex is also stabilized by an interaction between the N-terminal regulatory arm of RhoGDIa and the switch 1 and 2 regions of the Rho GTPases (16, 53). The conformation of the switch regions and thus their interaction with RhoGDIa are sensitive to GDP-GTP exchange on the Rho GTPases (59). Thus, the finding that nucleotide exchange was an absolute requirement for the potentiation of membrane association of Cdc42 by n-butanol indicates that the corresponding destabilizing effect on the RhoGDIa–Cdc42 complex requires the release of the interaction between the switch regions and the N-terminal regulatory arm of RhoGDIa, and is therefore coupled to GDP–GTP nucleotide exchange activation cycle of the Rho GTPase.

In conclusion, the results provide evidence for the existence of a halothane site within RhoGDIa and support a mechanism in which the anesthetic interacts directly with binding sites that are normally utilized by the GG chain. Moreover, this interaction modified the functional activation of the Rho GTPases, Although caution should be engendered in concluding that an *identical* interaction between anesthetics and RhoGDIa occurs in the cellular environment, it can be speculated that if anesthetics and alcohols destabilize the RhoGDIa–Cdc42 complex in the cellular environment, this might act to sustain signaling to downstream effectors by enhancing the membrane association of Cdc42. It is also notable that the affinity of halothane binding to RhoGDIa ($K_{\rm D} = 147 \pm 7 \,\mu\text{M}$) is within pharmacologically relevant ranges (60). Because RhoGDIa belongs to an extended family of proteins that also possess hydrophobic clefts or pockets that bind isoprenyl chains, it is also intriguing to speculate that members of this diverse class of proteins might also contain functional anesthetic binding sites. Some potential candidates might include, RhoGDI γ , Ly/ D4GDI, RabGDI isoforms (61), as well as other isoprenyl chain binding proteins, such as geranylgeranyl and farnesyl transferases (62). Finally, RhoGDIa constitutes a novel and powerful model for studies of anesthetic-protein interactions by virtue of its solubility, abundance, and the availability of high resolution structural information (16, 17, 53, 55-58).

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Figure 1.

Concentration dependent effects of halothane on the steady-state tryptophan emission fluorescence of RhoGDIa. Tryptophan emission fluorescence spectra of RhoGDIa (500 nM) were acquired upon excitation at 295 nm in the presence of increasing levels of halothane (0–800 μ M). *Inset*, the fluorescence intensities at maximal wavelengths obtained using an excitation wavelength of 295 nm (\odot) or 305 nm (\bigcirc) were plotted as a function of halothane concentration. The solid curve corresponds to a least squares non-linear regression fit of the data obtained using an excitation wavelength of 295 nm to Eq. 1. Also shown is the

concentration dependent effect of halothane on the fluorescence intensities at maximal wavelength obtained for free L-tryptophan upon excitation at 295 nm (•). The solid line through the data was obtained using linear regression. The data are representative of triplicate experiments. The fits of each individual replicate experiment yielded similar parameter values and errors. Other details are given under "Experimental Procedures."

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Figure 2.

Concentration dependent effects of AGGC on the halothane induced quenching of RhoGDIa tryptophan emission fluorescence. The steady-state tryptophan emission fluorescence spectra of RhoGDIa (500 nM) were obtained with (bold) or without (dashed) 1 mM halothane, in the presence of increasing concentrations of AGGC (0–300 μ M). *Inset*, the recovery in emission fluorescence from quenching by 1 mM halothane in the presence of AGGC was plotted as a function of AGGC concentration. The solid curve corresponds to a fit of the data obtained to Eq. 2 using least squares non-linear regression analysis. Data are

representative of at least three independent experiments. The fits of each individual replicate experiment yielded similar parameter values and errors. Other details are described in "Experimental Procedures."



Figure 3.

Chain length dependent effects of a homologous series of n-alkanols on the halothane induced quenching of RhoGDIa tryptophan emission fluorescence. A-F, the steady-state emission fluorescence intensity of RhoGDIa (500 nM) at 330 nm upon excitation at 295 nm was measured as a function of the concentration of n-alkanols with chain lengths, n=2 to n=7. The effects of the n-alkanols are expressed as the recovery in emission fluorescence from quenching by 0.2 mM halothane in the presence of each n-alkanol concentration. The solid curves represent fits of the data to Eq. 2 using least squares non-linear regression

analysis. Calculated values of EC_{50} , h and F_{max} are listed in Table 1. Data are representative of three independent experiments. The fits of each individual replicate experiment yielded similar parameter values and errors. Other details are described in "Experimental Procedures."

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Figure 4.

Chain length dependent effects of n-alkanols on the quenching of RhoGDIa tryptophan fluorescence by halothane (EC_{50}). The values of EC_{50} were calculated from least squares non-linear regression fits of the n-alkanol dose response curves shown in Figure 3 to Eq. 2. A plot of values of C_{sat} for each n-alkanol (63) is shown for comparison of EC_{50} values with aqueous solubilities (dashed line). *Inset*, the non-linear dependence of EC_{50} on *n* as illustrated by a plot of EC_{50} / C_{sat} against *n*. Data are representative of three independent experiments. Other details are described in "Experimental Procedures.

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Figure 5.

Concentration dependent effects of n-butanol on the quenching of RhoGDIa tryptophan emission fluorescence by varying levels of halothane. Tryptophan emission fluorescence recovery at 330 nm upon excitation at 290 nm was measured as a function of n-butanol concentration in the presence of 0.1 mM (\odot), 0.2 mM (\bullet), 0.5 mM (\blacktriangle) and 1 mM (∇) halothane. The solid curves represent fits of the data to Eq. 2 using least squares non-linear regression analysis. Values of EC_{50} , h and F_{max} are listed in Table 2. Data are representative of three independent experiments. The fits of each individual replicate experiment yielded

similar parameter values and errors. Other details are described in "Experimental Procedures."

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Figure 6.

Identification of E193 as a site of photoincorporation of 3-azibutanol (100 μ M) on RhoGDIa (2.8 nM) in the presence of AGGC (550 nM) using LC–MSMS. After photolabeling, RhoGDIa was digested with trypsin and subject to HPLC–MSMS as described in the text. *Inset*, a tryptic peptide bearing a single 72 Da modification and starting at F181 of RhoGDIa. The predicted charge/mass ratios of ions with an intact N-terminus (b-ions) or C-terminus (y-ions) are shown above and below the sequence, respectively, with the indicated charge. The site of photoincorporation for 3-azibutanol was inferred from the MSMS spectrum. The experimentally observed values are colored and in bold and their position indicated on the spectrum. The photolabeled residue is boxed and the statistics are given in Table 3.

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Figure 7.

Identification by LC-MSMS of E163 as a photoincorporation site for 3-azioctanol (10 μ M) on RhoGDIa (2.8 nM) in the absence of AGGC. After photolabeling, RhoGDIa was digested with trypsin and subject to HPLC–MSMS as described in the text. A tryptic peptide bearing a single 128 Dalton modification and starting at A153 of RhoGDIa had the sequence shown in the inset. Statistics are given in Table 3. The site of photoincorporation of 3-azibutanol was inferred from this MSMS spectrum. In the inset, the predicted charge/ mass ratios of ions with an intact N-terminus (b-ions) or C-terminus (y-ions) are shown above and below the sequence, respectively, with the indicated charge. The photolabeled residue is boxed. The experimentally observed values are colored and in bold and their position indicated on the spectrum.

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Figure 8.

Identification by LC-MSMS of E163 as a photoincorporation site for 3-azibutanol (100 μ M) on RhoGDIa (2.8 nM) in the absence of AGGC. After photolabeling, RhoGDIa was digested with trypsin and subject to HPLC–MSMS as described in the text. A tryptic peptide bearing a single 72 Dalton modification and starting at A153 of RhoGDIa had the sequence shown in the inset. Statistics are given in Table 3. The site of photoincorporation for 3-azibutanol was inferred from this MSMS spectrum. The identified b- and y-ions are indicated above the MS/MS spectrum using the same conventions as in Figure 7.



Figure 9.

Enhanced membrane-association of Cdc42 by n-alkanols resulting from the destabilization of the RhoGDI α -Cdc42 complex. *A*, the association of Cdc42 with POPC/POPS (4:1, molar) membranes immobilized on a Biacore L1 chip, was measured from the increase in response (binding) that occurred upon injection of the RhoGDI α -Cdc42 complex in the presence of EDTA and GMPPNP. *B*, representative time courses for the concentration dependent effects of n-butanol on Cdc42 membrane-association. *C*-*F*, values of *R*_{eq,Cdc42}, calculated from fits of the time courses obtained for each concentration of n-alkanols with

chain lengths 2, 4, 6, and 7 to Eq. 5, were plotted as function of n-alkanol concentration. Also shown is the effect of n-butanol on $R_{eq,Cdc42}$ values obtained in the absence of EDTA and GMPPNP (Panel D, \bigcirc), and the concentration dependent effects of 3-azibutanol on $R_{eq,Cdc42}$ (Panel C, \Box). The solid curves represent fits of the data to Eq. 6 using least squares non-linear regression analysis. Values of EC_{50} and h are listed in Table 1. Data are representative of three independent experiments. The fits of each individual replicate experiment yielded similar parameter values and errors. Other details are described in "Experimental Procedures."



Figure 10.

n-Butanol inhibits the extraction of Cdc42 from membranes by RhoGDIa. *A*, The emission fluorescence intensity of MANT-GMPPNP (2 μ M) loaded on Cdc42 (60 nM), at 430 nm upon excitation at 355 nm, was measured in the presence of LUV composed of POPC/POPS (4:1, molar) containing 2 μ M HAF. The recovery of fluorescence from quenching due to MANT–HAF FRET upon addition of 60 nM RhoGDIa was monitored as a function of time in the presence (bold) and absence of 50 mM n-butanol. Results are representative of

experiments carried out at least three times. Other details are described in "Experimental Procedures."



Figure 11.

Location of tryptophans and residues photolabeled by 3-azibutanol within the structure of RhoGDIa. The crystal structure of RhoGDIa in complex with Cdc42 (not shown) was from Hoffman et al (16). Tryptophans, W192, W194 and W202 are shown in green, residues photolabeled by 3-azibutanol are shown in red and the GG chain of Cdc42 is shown in yellow. The C-terminal backbone of Cdc42 (K183-C188) is shown in red. The Protein Data

Bank file number of the structure used is 1DOA and the molecular graphics images were produced using the UCSF Chimera package (36)

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Table 1

Chain Length Dependent Effects of n-Alkanols on Halothane Binding and RhoGDIG- Cdc42 Complex Dissociation.

$n EC_{50}, mM h F_m$			
	K_{Γ}^{c}, mM	$K_{\rm app}$ (mM)	Ч
$2 100 \pm 58 1.4 \pm 0.4 44$	± 5 42 ± 24	70 ± 21	1.2 ± 0.2
3 48 ± 4.5 1.2 ± 0.2 $40 \pm$	$\pm 9 20 \pm 2$	$^{\mathrm{pq}}$	p^{pu}
4 10 ± 3 1.1 ± 0.1 51	$\pm 4 4.2 \pm 1.2$	7.5 ± 1.3	1.5 ± 0.3
5 38 ± 19 1.1 ± 0.2 41	± 8 16 ± 8	p^{pu}	p^{pu}

of the dose-response curves shown in Figure 3 to Eq. 2.

response curves shown in Figure 6 using Eq. 6.

^cValues of KI were calculated using Eq. 3.

 \boldsymbol{d} not determined. See "Experimental Procedures" section for other details.

Table 2

Dependence of the EC_{50} , h, and F_{max} for the Displacement of Halothane by n-Butanol on Halothane Concentration.

[halothane], mM	EC_{50}^{a} , mM	h ^a	F_{\max}^{a}
0.1	3 ± 1	1.0 ± 0.4	0.55 ± 0.06
0.2	11 ± 3	1.1 ± 0.2	0.59 ± 0.04
0.5	22 ± 5	1.0 ± 0.1	0.59 ± 0.05
1.0	52 ± 8	1.2 ± 0.2	0.65 ± 0.08

^{*a*}Values of *EC*₅₀, *h* and F_{max} were obtained from fits of the concentration-response curves shown in Figure 5 to Eq. 2. See "Experimental Procedures" section for other details.

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Residues in RhoGDI-a Photolabeled by 3-Azioctanol and 3-Azibutanol Identified in Peptide Fragments Following Trypsinolysis and LTQ-FT Mass Shectrometry.

Agent	[Alcohol] μM ^a	AGGC	Residue Photolabeled	Peptid <i>e^b</i>	Charge	R _t , min	X _{corr}	$_{\mathcal{O}^+}\mathrm{HW}$	Accuracy ^d ppm
1	100	+	E193	[1]	+3	23.45	5.29	2437.60	2.46
-azibutanol	100	I		Ы	hotolabeled	peptide no	t detecteo	T	
[100	+	E163	[2]	+2	22.52	5.06	1824.96	3.28
-azıbutanoı	100	I		Ρ	hotolabeled	peptide no	t detecte	Ŧ	
lo noto o teo	10	+	E163	[2]	+2	18.82	2.95	1880.90	0.00
- 42100141101	10	I		[2]	+2	18.84	4.30	1880.90	0.00
		I	E193	[1]	+3	16.15	4.77	2365.54	0.00
Unlabeled p	eptide /	I	E163	[2]	+2	21.02	4.75	1752.89	0.00

YEFLTPVEEAPK. Photolabeled residues are indicated in bold.

ton). 2 ŝ

 e Accuracy is the difference between the calculated and observed mass expressed in ppm.

f Unlabeled peptides are those derived from that fraction of RhoGDIa which was not photolabeled during the experiment with 100 µM azibutanol in the presence of AGGC. See "EXPERIMENTAL PROCEDURES" section for other details

Table 4

Sequence Coverage of RhoGDIa Following In-Gel Digestion with Trypsin.^a

001 MAEQEPTAEQ LAQIAAENEE DEHSVNYKPP AQKSIQEIQE LDKDDESLRK YKEALLGRVA VSADPNVPNV 071 VVTGLTLVCS SAPGPLELDL TGDLESFKKQ SFVLKEGVEY RIKISFRVNR EIVSGMKYIQ HTYRKGVKID 141 KTDYMVGSYG PRAEEYEFLT PVEEAPKGML ARGSYSIKSR FTDDDKTDHL SWEMNLTIKK DWKD

			B	Z/	Accuracy ^D
sition	Sequence	Charge	Observed	Calculated	(mdd)
4-49	SIQEIQELDKDDESLR	2	959.4602	959.4736	14.0
4–50	SIQEIQELDKDDESLRK	2	1023.5096	1023.5211	11.2
1–58	YKEALLGR	1	949.5387	949.5465	8.2
3–58	EALLGR	1	658.3844	658.3883	5.9
66-6	VAVSADPNVPNVVVTGLTLVCSSAPGPLELDLTGDLESFKK	4	1039.5316	1039.5526	20.2
0-105	KQSFVLK	1	849.5158	849.5193	4.1
0-105	QSFVLK	1	721.4184	721.4243	8.2
6-111	EGVEYR	1	752.3508	752.3573	8.6
4-117	ISFR	1	522.3012	522.3034	4.2
8-127	VNREIVSGMK	1	1132.617	1132.6143	2.4
1-127	EIVSGMK	1	763.394	763.4019	10.3
8–134	YIQHTYR	1	980.4887	980.4948	6.2
8-135	YIQHTYRK	1	1108.5764	1108.5898	12.1
9–152	IDKTDYMVGSYGPR	2	801.3766	801.385	10.5
2-152	TDYMVGSYGPR	2	623.2784	623.2821	5.9
3-167	AEEYEFLTPVEEAPK	2	876.4141	876.4223	9.4
8-172	GMLAR	1	547.299	547.3021	5.7
3-178	GSYSIK	1	654.3412	654.3457	6.9
9-199	SRFTDDDKTDHLSWEWNLTIK	3	870.0851	870.0884	3.8
1–199	FTDDDKTDHLSWEWNLTIK	3	789.0358	789.0441	10.5
1-200	FTDDDKTDHLSWEWNLTIKK	3	831.7341	831.7424	10.0
7–199	TDHLSWEWNLTIK	ю	548.2857	548.279	12.2

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^aThe sequence of the RhoGDIa is shown in single letter code. The proteolytic fragments sequenced by MSMS are shown in bold. The sequence coverage by amino acid count is 79.4%;

b Accuracy is the difference between the calculated and observed m/z expressed in ppm of the calculated value.