

Effects of clarification on physicochemical characteristics, antioxidant capacity and quality attributes of açai (*Euterpe oleracea* Mart.) juice

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Abstract The effects of a clarifying process using pectinases and chitosan on the physicochemical characteristics, antioxidant capacity and quality attributes of açai fruit (*Euterpe oleracea* Mart.) juice were evaluated. Clarification of açai pulp resulted ($P \leq 0.05$) in a 50 % loss of total anthocyanin (4.2730 mg/100 mL) and 29 % reduction in antioxidant capacity

(33.60 $\mu\text{M FeSO}_4/\text{g}$). A high association ($P \leq 0.05$) was found between the decrease of antioxidant capacity and total anthocyanin loss. The use of pectinases associated with chitosan as an aid for clarification of açai juice proved to be highly effective and resulted in a clear juice with a brighter purple to red color that was free of lipids, insoluble solids, and others substances that cause hazes. The obtained clarified açai juice is a genuinely high-value anthocyanin-rich product that could be used as colorant and functional ingredient to fruit juices and soft drinks.

Dr. Isabella Brasil is the responsible person for this project

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Introduction

In recent years, the increase in consumption of fruits and vegetables has been associated with the prevention of modern life-style related degenerative disease (Pacheco-Palencia and Talcott 2010). More attention has been given to the presence of dietary polyphenols in fruits and vegetables once they contribute towards maintaining a good health (Netzel et al. 2007). Açai (*Euterpe oleracea* Mart.) is a slender, multi-stemmed, monoecious palm, widely distributed in the Amazon floodplains (Muniz-Miret et al. 1996). Moreover, açai is a highly perishable fruit with short shelf life (Tonon et al. 2008). Beyond being a highly energetic fruit, açai has been recognized for its functional properties for use in food and nutraceutical products, due to its high antioxidant activity, which is related to its high anthocyanin and phenolic content (Coisson et al. 2005; Schauss et al. 2006). Del Pozo-Insfran et al. (2006) reported that açai was found to have higher antioxidant capacity than other anthocyanin-rich fruits, such as highbush blueberries, blackberries, cranberries and others,

and verified that the predominant anthocyanin present in açaí pulp was cyanidin-3-glucoside (1040 mg/L pulp).

Few studies have reported the potential health benefits of açaí (Del Pozo-Insfran et al. 2006; Lichtenthaler et al. 2003) and the stability of the phytochemical compounds upon processing and storage. Due to its highly perishable nature, consumption and commercialization of açaí was mainly restricted to the regional level in Brazil. However, an increased interest from international markets has made açaí pulp and its derived products widely available to the general public. Commercialization of açaí pulp imposes new challenges since various processing steps such as pasteurization, freezing, dilution or dehydration are often employed during manufacture of retail products. Clarification of açaí juice improves aesthetic properties and market acceptability while removing lipids and insoluble solids. More recently, the consumer market has been equally receptive to clear fruit juice (Carvalho et al. 2008). Clarification is an important step in the processing of fruit juice and is most often achieved through microfiltration, enzymatic treatment or by using common clarifying aids like gelatin, bentonite, silica sol, polyvinyl pyrrolidone or a combination of these compounds. Nowadays, a great variety of new clarified fruit juices is offered in international and local markets; however, there is no clarified exotic tropical fruit juices. Therefore, clarified açaí juice with its unique exotic color and novelty flavor may be able to diversify the market, thus providing a totally new experience to consumers. For clarified products, clarity and homogeneity are two important characteristics, which are achieved by the complete removal of all suspended solids (Pacheco-Palencia et al. 2007). The edible pulp of açaí is commonly macerated with water to produce a thick, purple beverage of creamy texture, oily appearance and characteristic flavor which requires the use of enzymes in order to obtain a pulpy juice with lower viscosity due to a much smaller amount of pectin and starch which is advantageous for the filtration process. Chitosan (poly- β (1–4) N-acetyl-glucosamine) has been reported to have a number of potential industrial uses such as an adhesive, a paper-sizing agent, a chelating agent for metal ions and as fruit-juice clarifying aids (Knorr 1984). Chitosan being nontoxic and biodegradable may be used as an alternative agent for refining of fruit juices. Moreover, acid soluble crab shell chitosan and water soluble chitosan salt proved equally effective as fining agent for apple and carrot juices (Imeri and Knorr 1988; Soto-Peralta et al. 1989). The use of chitosan in this respect is hindered due to its solubility in organic acids

The use of one compound only as compared to the silica sol/gelatin/bentonite treatment and the ease of handling acid soluble chitosan should make it an attractive alternative to conventional juice fining procedures (Chatterjee et al. 2004; Rungsardthong et al. 2006). The use of pectolytic enzymes in association with clarifying agents in fruit processing is

essential to get better juice yields, improve filtration rate and produce clear juices of higher quality for the concentration process. Even though there are a vast number of studies describing chemical properties of açaí and açaí products, there is lack of information on the literature regarding processing technologies of clarified açaí juice using enzymes and chitosan as clarifying agents. Therefore, the aim of this work was to evaluate the effects of a clarifying process using pectinases and chitosan on the physicochemical characteristics, antioxidant capacity and quality attributes of açaí juice. Results from this investigation will assist on assessing factors that influence the quality and physicochemical stability of açaí-based food.

Material and methods

Fruit material

Pasteurized, frozen açaí pulp (4°Brix) was obtained from local market (Fortaleza, CE, Brazil). The pulp was divided into suitable small portions and kept frozen (–18 °C) until use.

Chemicals and enzymes

Citrozym-Ultra L from Novozymes Latin America Ltda (Araucária, Brazil) was used for enzymatic treatment for açaí pulp and stored at 4 °C. Citrozym-Ultra L is a commercial enzyme preparation from *Aspergillus aculeatus* and *Aspergillus niger*, used in the food industry for fruit juice processing to reduce viscosity. It contains different food grade (generally recognized as safe – GRAS) pectinolytic and cellulolytic enzymes [endo-polygalacturonase (EC 3.2.1.15; C.A.S. No. 9032-75-1), endopectinylase (EC 4.2.2.10; C.A.S. No. 9033-35-6) and pectin esterase (EC 3.1.1.11; C.A.S. No. 9025-98-3)]. The declared activity of Citrozym-Ultra L is 4500 PECTU units/mL (pectin unit per mL). The optimum enzyme reaction conditions are at pH 3.5–6.0 and temperature range below 50 °C (Kashyap et al. 2001). Commercial chitosan from Polymar Ciência e Nutrição S/A (Fortaleza, Brazil) was used in the clarification of açaí juice. Chitosan powder was dissolved in (1 % w/v) acetic acid solution to yield a 4 % (w/v) solution.

Enzymatic treatment

The enzymatic hydrolysis of the pulp as a pretreatment for clarification was carried out on a laboratory scale. Samples of açaí pulp were gently thawed overnight under refrigeration (5 °C) the day prior to use on the clarifying experiments. The range of the variables for enzymatic treatment conditions were based on preliminary experiments conducted earlier and included incubation time of 15, 30, 45, and 60 min and enzyme concentration of 0.01–0.2 % (v/v). The fresh pulp (FP) was

divided into six equal portions of 100 mL and initially heated in a water-bath at 45 °C followed by the addition of different Citrozym-Ultra L concentrations (0.01, 0.05, 0.10, 0.15 and 0.20 % (v/v)) under constant agitation at 100 rpm (± 5), in which one portion remained unprocessed (control). The pH of the pulp was kept constant at its natural value of 4.0. During the mash treatment 10 mL of each treated pulp were drawn out every 15 min to accomplish the enzyme inactivation by heating the pulp to 90 °C for 5 min in a water bath. After that, in order to evaluate the optimal conditions for pulp hydrolysis, the treated samples were filtered through a Whatman No.1 filter paper and the filtrate was submitted to acidified alcohol test (IAL 2005) and standard iodine test (AOAC 1995) to detect pectin and insoluble starch, respectively.

Clarifying assays

According to preliminary assays based on an enzymatic treatment optimization (data not showed) 0.1 % Citrozym-Ultra L at 45 °C for 60 min was selected to attain optimum enzymatic hydrolysis conditions since it showed negative results for starch and pectin presence. A negative test by iodine indicates that all of the starch has been reduced to a chain length of less than nine to twelve glucose units, a size sufficiently reduced to not produce post-bottling hazes (Abdullah et al. 2007). According to Grassin and Fauquembergue (1996) the acidified alcohol test is positive when pectin reacts with ethanol acidified with chlorhydric acid forming a viscous gel.

After the enzymatic treatment the hydrolyzed pulp (HP) was suspended with distilled water to obtain 3 L of açai cloudy juice (CJ) (30 % (w/v) pulp). The CJ juice was submitted to a filtration through cheesecloth and pasteurized in a water bath (90 °C/5 min) for clarifying experiments. As pretreatment to define the optimum clarification conditions, the CJ was divided into six equal portions of 50 mL and transferred to 250 mL Erlenmeyer flasks. Volumes ranging from 2.5 to 22.5 mL of 4 % chitosan solution in water (100 to 900 mg of soluble chitosan) were added to each flask. The control flask received only 2.5 mL of distilled water instead of chitosan solution. Flasks were then left at room temperature (28 °C) and gently agitated for 120 min and finally the juice was filtered through Whatman No. 2 filter paper in order to obtain a brighter purple to red colored juice free of visual sediments or cloudiness (pulpless juice), 5 mL aliquots of açai juice were taken from each flask at intervals of 30 min to determine the clarity and color. Clarity of the clarified açai juice was determined by measuring the absorbance at 660 nm according to Abdullah et al. (2007) using a UV-Vis spectrophotometer (Model B582, Micronal, Brazil) after filtering through cheesecloth and diluting with water (all treatments have the same dilution (w/v)). The analysis of color was accomplished by transmittance in the Hunter system using a Chroma Meter (Minolta Co., CR10,

Japan) and was expressed with the values of Hunter L^* (lightness), a^* (redness-greenness), b^* (yellowness-blueness). The chroma meter was calibrated using the standard white and black plates. The Commission Internationale de L'Eclairage (CIE) system reference measures the lightness (L^* value) on a numerical scale, where white=100 and black=0 (Saxena et al. 2012). The a^* and b^* have no specific numerical limits (positive a^* (red), negative a^* (green), positive b^* (yellow), negative b^* (blue)). The criteria applied for clarifying optimization were as follows: maximum clarity, a^* and L^* values, since they are important physical indexes for quality attributes of the clarified açai juice.

Juice clarifying process

After analyzing statistically the results from the experimental clarifying assay (data not showed) the optimum conditions were 700 mg of soluble chitosan/50 mL of treated açai juice during 90 min for flocculation. Approximately 10 kg of 0.1 % Citrozym-Ultra L treated pulp were used for further laboratory pilot-scale experiment. The treated pulp was diluted in spring water 1:3 (v/v) to produce 30 L of pulpy juice (PJ) (33 %v/v, 1.26°Brix). The juice was filtered through cheesecloth to reduce fat content and insoluble solids. The filtered moderately cloudy juice obtained was then pasteurized at 90 °C for 5 min before proceeding with the clarifying process. According to previous clarifying assay data for 30 L of filtered juice, was added 3 L of 4 % chitosan solution (700 mg of soluble chitosan/50 mL fruit juice) and held at room temperature under gentle agitation for 90 min until flocculation was complete. Following, the juice was immediately filtered using cheesecloth to obtain a clear juice (CJ). Schematic diagram of the clarifying process is showed in Fig. 1.

Physical and physicochemical analysis

Samples of the products obtained throughout the clarifying process were analyzed (Fig. 1): Fresh pulp (FP), hydrolyzed pulp (HP), pulpy juice (PJ) and clear juice (CL). The physicochemical analyses were: clarity (Abdullah et al. 2007); color (Brasil et al. 2012); pH was measured using a glass electrode potentiometer according to AOAC method 981.12 (1995); titratable acidity was determined by titration with 0.1 N NaOH to a pH 8.1 and expressed as % citric acid using AOAC method 932.14 (AOAC 1995); total soluble solids was measured refractometrically by use of a digital refractometer type RX 5000 (Atago, Tokio, Japan) using AOAC method 932.14 (1995); reducing and total sugar was estimated by DNSA (dinitrosalicylic acid) reagent (Miller 1959) and Anthrone reagent methods (Carvalho et al. 2008); ascorbic acid was measured following AOAC method 985.33 (2,6-dichloroindophenol titrimetric method, (AOAC 1995)); total anthocyanin content was evaluated

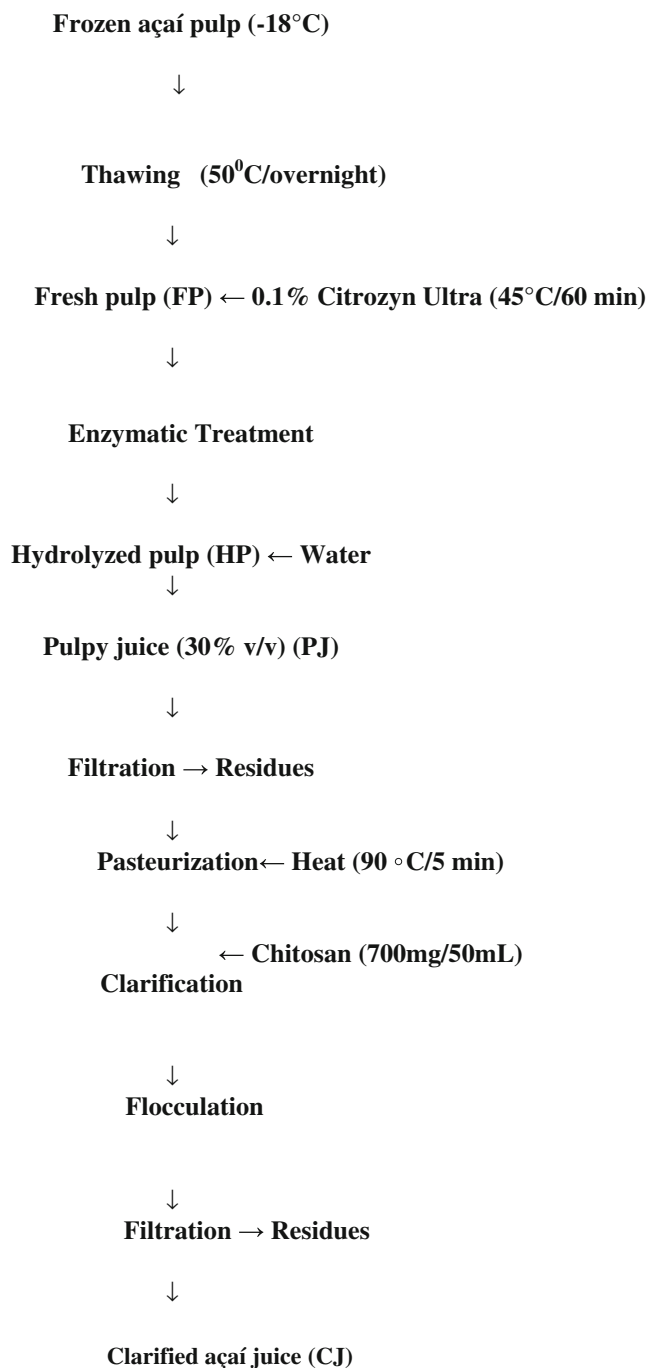


Fig. 1 General processing diagram for obtaining clarified açai juice by pectinase and chitosan

using the method described by Giusti and Wrolstad (2003)); starch was determined using the method described by AOAC method 996.11 (AOAC 1995); and total lipid was analyzed using Soxhlet method (IAL 2005).

Determination of antioxidant capacity (FRAP Method)

Changes in antioxidant capacity during processing (FP, HP, PJ, and CJ) were determined by the ferric reducing

antioxidant power (FRAP) assay method (Benzie and Strain 1996) with minor modifications. This method consists in estimating the antioxidant capacity of the products from their ability to reduce Fe(III)-2,4,6-Tri(2-pyridyl)-s-triazine (TPTZ) complex to Fe(II)-TPTZ, the resulting intense blue color is linearly related to the amount of reductant (antioxidant) present. The absorbance at 593 nm was measured 4 min after 1 mL of a ten-fold dilution of the samples was added to 3 mL of Fe(III)-TPTZ and ferric reducing antioxidant potential (FRAP value) was interpolated from a standard curve prepared from a stock solution (Benzie and Strain 1996). The results are expressed as micromoles of ferrous equivalents per gram of fresh weight.

Statistical analysis and experimental design

The experimental design and statistical analysis were performed using Statistical Analyses System software Version 9.1 (SAS 2006). For all analyses, determinations were made in triplicate as independent experiments. The components of variance for balanced split-plot experiments were determined, considering a complete factorial design with first factor distributed in the whole plots (four incubation times: 15, 30, 45 and 60 min), and a second factor in the subplots (six chitosan solution concentrations: 0, 100, 300, 500, 700 and 900 mg/50 mL of fruit juice). The dependent variables measured were clarity and color values. Analysis of variance (ANOVA) and regression were assessed to define the optimum chitosan concentration and flocculation time. Differences between the physical, physicochemical, and antioxidant activity data for each processing phase (FP, HP, PJ and CL) were tested for significance ($\alpha=5\%$) by one-way ANOVA. Significant different means ($P\leq 0.05$) were separated by Tukey's HSD test.

Results and discussion

Physicochemical characteristics, antioxidant capacity, and quality attributes

Physical and physicochemical characteristics, and antioxidant capacity from samples of each product (FP, HP, PJ, and CJ) obtained during the clarifying process are reported on Table 1. The results indicate a significant effect ($P\leq 0.05$) in all studied parameters among the products obtained during the clarification process. Although the pH and titratable acidity were significantly affected by processing, the typical sour characteristic of açai juice was kept during the clarifying process. This indicates the concentrations of organic acids in the clarified juice remained unaltered as a constant net balance. The total soluble solids (TSS) content decreased ($P\leq 0.05$) during the clarification of açai juice, a typical behaviour for clarifying processes. The TSS content found in the açai fresh pulp is

Table 1 Physical and physicochemical characteristics, and antioxidant capacity of clarified açai juice at different phases of the clarification process

Parameters*	Fresh Pulp (FJ)	Hydrolysed Pulp (HP)	Pulpy Juice (PJ)	Clear Juice (CJ)
pH	4.6908 ^a ± 1.24	4.7233 ^a ± 0.34	4.5733 ^b ± 0.99	3.4233 ^c ± 1.98
Titrateable acidity (% citric acid)	0.0410 ^b ± 1.23	0.0490 ^b ± 0.998	0.0086 ^b ± 1.12	0.1756 ^a ± 0.98
Total Soluble Solids (°Brix)	3.98476 ^a ± 1.02	4.8333 ^b ± 1.98	1.2667 ^c ± 301	1.0333 ^c ± 2.98
Reducing Sugars (% glucose)	1.1667 ^a ± 1.23	2.3646 ^b ± 0.65	0.3157 ^c ± 1.29	0.2996 ^c ± 2.56
Total sugars (%)	1.5587 ^a ± 1.02	1.7852 ^a ± 2.01	0.4212 ^b ± 9.12	0.3595 ^c ± 2.65
Ascorbic Acid (mg/100 mL)	15.3181 ^a ± 0.76	15.3067 ^a ± 0.99	4.7367 ^b ± 11.98	4.3600 ^b ± 0.76
Total Lipids (%)	6.7198 ^a ± 0.76	6.7094 ^a ± 0.70	0.5163 ^b ± 1.231.36	0.0190 ^c ± 2.34
Starch (%)	1.1345 ^a ± 0.99	0.7670 ^b ± 0.96	0.4467 ^b ± 1.13	nd
Total Anthocyanins (mg/100 mL)	53.9789 ^a ± 0.87	53.9000 ^a ± 0.97	8.0730 ^b ± 1.02	4.2730 ^c ± 1.12
Lightness (<i>L</i> *)	21.79 ^a ± 0.87	20.10 ^a ± 1.98	19.76 ^b ± 1.34	34.34 ^c ± 1.98
Color <i>a</i> *	4.01 ^a ± 0.56	2.95 ^a ± 0.52	33.89 ^b ± 0.90	22.51 ^c ± 0.23
Color <i>b</i> *	0.88 ^a ± 1.76	0.66 ^a ± 0.871.12	-0.29 ^b ± 0.99	10.87 ^c ± 0.760
Clarity (ABS. at 660 nm)	nr	nr	0.016 ^a	0.007 ^b
FRAP (µM FeSO ₄ /g fresh weight)	54.02 ^a ± 0.98	53.77 ^a ± 1.28	47.54 ^b ± 0.98	33.60 ^c ± 1.98

FRAP ferric reducing antioxidant power; Each data is a mean ± SD of three replicate experiments (n=3)

*values followed by different letters in the same line are significantly different (Tukey HSD test, $P \leq 0.05$), *nr* not realized, *nd* not detected.

lower ($P \leq 0.05$) than other tropical fruits such as guava (Brasil et al. 1995), acerola (Brasil et al. 2007), mango (Gupta and Jain 2012), and cashew apple (Maia et al. 2004). This is probably related to the presence of high-suspended solids content in the pulp such as starch that can interfere with the measurement of the refractive index (Carvalho et al. 2008). On the contrary, the increase of TSS in treated pulp is associated to the increase in reducing sugar content due to the action of pectinases (polygalacturonases and pectin lyases) on polygalacturonic chains as well as hydrolysis of non reducing sugars by weak acids at higher temperatures (pulp treatment) (Hernandez and Villegas 1987). Brasil et al. (1995) found an increase of 275 % in the reducing sugar content of guava pulp treated with 600 ppm of Clarelx-L at 45 °C for 120 min. Floribeth and Lastreto (1981) detected an increase of 20 % in the reducing sugar content (galactose, arabinose and xylose) using a combination of pectic enzymes and cellulases to clarify apple juice. Carvalho et al. (2008) reported an increase of sucrose, glucose, and fructose in hydrolyzed pineapple juice using commercial pectinase (Ultrasym 100 G). Cheirsilp and Umsakul (2008) found a 15 and 39 % increase in total soluble sugars and reducing sugars in banana pulp after incubating with 0.05 % (w/w) of pectinase at 40 °C for 2 h. The decrease in TSS and reducing and total sugars in pulpy juice and clear juice is related to the pulp dilution to produce pulpy juice (33 %v/v) and the pulpy juice filtration, respectively. According to Youn et al. (2004), polysaccharides, proteins and colloidal materials are present as solid materials in juices forming gels and accumulating on the filter surface forming a secondary membrane as filtration goes on.

Açai fruit is a tropical fruit that contains considerable amounts of starch, as much as 10 % (Bobbio et al. 2002; Coisson et al. 2005; Del Pozo-Insfran et al. 2006; Lichtenthaler et al. 2003). However, the starch content found in non treated pulp (Table 1) is very low compared to the literature. According to Carvalho et al. (2008) some molecules may interfere with the starch determination by increasing or decreasing its value. Starch is estimated as glucose units or reducing sugars and if the starch granules have not been broken down completely, short-chained dextrans are left leading to retrogradation, when the short-chained dextrin recrystallize into a form that is no longer susceptible to hydrolysis, regardless of heating. Moreover, there was a significant decrease of starch content in treated pulp. This is supported by the hydrolysis of pectin present in the pulp at 45 °C. According to Carrin et al. (2004) when an aqueous suspension of starch is heated, the hydrogen bonds weaken, water is absorbed, the granules swell, rupture, and finally gelatinize causing a practically complete breaking down of the starch granule. The quantitative starch assays in CJ demonstrated that Citrozyn Ultra L (0.1 %, 45 °C for 60 min) was highly effective, even in doses lower than those recommended by the manufacturer, resulting in a complete break down of the gelatinized starch. Floribeth and Lastreto (1981) reported that up to 1 % starch may be present in clarified juice after milling and pressing causing post-process cloudiness and hindering filtration. According to Carrin et al. (2004) in the presence of starch, the following problems may occur: (i) slow filtration, (ii) membrane fouling, (iii) gelling after concentration, and (iv) post-concentration haze.

In addition, the chitosan (700 mg/50 mL for 90 min) was a highly effective clarifying agent by removing starch cloudiness, since there were significant differences in clarity and L^* values for clarified açai juice, lowest absorbance and highest values, respectively. Chitosan as a clarifying agent complexes with protein, polyphenols, and others insoluble solids inducing flocculation and sedimentation thus resulting in removal of these potential haze precursors (Soto-Peralta et al. 1989). Indeed, the total lipids of pulpy juice were significantly reduced (90 %). This result could be associated to the polycationic nature of chitosan acting as an effective clarifying agent in the separation of lipid particles from açai pulpy juice, therefore improving aesthetic properties and further market acceptability.

Ascorbic acid content suffered a significant decrease (71.5 %) throughout the clarifying process with values ranging from 15.3 mg/100 mL (fresh pulp) to 4.3 mg/100 mL (clear juice). The vitamin C content in clarified citrus juices has been reported by other studies from no measurable loss up to 33 % compared to fruit pulp, depending on the types of clarifying aids and processing conditions (Youn et al. 2004). The low value obtained in the final product could be related to the processing phases such as pulp dilution, heat treatment, fining and due to oxygen exposure. An important criterion for the final product quality after heat treatment is the ascorbic acid retention. Yuyama et al. (1999) compared vitamin C content of açai fruit to different exotic fruits from Amazon region such as bacuri (1.3 mg/100 g), maracujá-do-mato (6.7 mg/100 g) and murici (8.1 mg/100 g) and detected that açai had the highest vitamin C content; however, when compared to cupuaçu (38.3 mg/100 g) and ituí (62.9 mg/100 g) both from the Amazon region, açai fruit value was lower approaching barely to taperebá (17.2 mg/100 g). Vitamin C is highly bio-available and is consequently one of the most important water-soluble antioxidants in cells, efficient in scavenging reactive oxygen species such as O_2^* , OH^* , peroxy radicals and singlet oxygen (Halliwell 1996). Consequently, when considering the antioxidant activities of fruit juices to disease risk and health, it is important to account the contribution of vitamin C in addition to phenolic compounds with antioxidant activity in chemical systems (Williams 1995). According to Ozkan (2002) foods naturally low in ascorbic acid, such as açai juice are particularly good candidates for fortification.

The clarifying process induced a significant decrease (50 %) in total anthocyanins content between pulpy juice and clear juice. Pacheco-Palencia et al. (2007) reported a 27 % loss in total polyphenolics (197 ± 6.9 mg gallic acid/100 mL) and a 20 % reduction in total anthocyanins (729 ± 3.4 mg/L) during clarification of açai pulp. Anthocyanins are labile compounds, subject to numerous detrimental reactions during processing and storage (Wrolstad et al. 2005), among which the transformation of monomeric forms into oligomeric or polymeric pigments gives rise to important color changes toward brownish-red hues, that are generally more stable (Monagas

et al. 2006). This decrease is probably related to the polycationic chitosan that bound to anthocyanin-based polymers dissolved in pulpy juice. Pacheco-Palencia et al. (2007) reported an initial detrimental effect on anthocyanins, non-anthocyanin polyphenolics, and antioxidant capacity during the clarification process of açai pulp using diatomaceous earth.

The negative effect of chitosan on anthocyanins and non-anthocyanin polyphenolics reduction in contrast to the beneficial adsorption to lipids and others insoluble solids during the clarifying process resulted in an increase ($P \leq 0.05$) in lightness (L^*) and decrease ($P \leq 0.05$) in red color (a^*). Aside from the loss of anthocyanins during clarification, there was a significant increase in red intensity (a^*) after pulp dilution (pulpy juice), and this could be explained by co-pigmentation of anthocyanins. Co-pigmentation reactions involve the formation of weak linkages between anthocyanin glycosides and other non-colored components such as phenolic acids, flavonoids, and flavonol derivatives (Kirca et al. 2006). Furthermore, Boulton (2001) reported that such reactions cause anthocyanins to exhibit far stronger colors than would be expected from their concentration resulting in a hyperchromic response that may result in an over-estimation of total pigments in spectrophotometric assays. The b^* values followed the a^* values trend and pulpy juice samples showed a decrease ($P \leq 0.05$) of the yellow intensity, with negative results. Additionally, for clear juice samples the b^* values showed a significant increase that could be related to interactions between polyphenolics and carbohydrate/ascorbic acid degradation products, such as furfurals and other aldehydes, that will influence the formation of brown pigments during storage of fruit-based foods (Brasil et al. 2012). Aldehydes generally promote anthocyanin polymerization with flavonols, flavan-3-ols and their derivatives resulting in the formation of both colorless and yellow-colored compounds that contribute to browning reactions and decreased color stability of anthocyanins (Maia et al. 2004). Kirca et al. (2006) found that polyphenolic, antioxidant, and color stabilities of açai juice are dependent on interactions among its matrix components and are influenced by processing, storage, temperature and chemical composition. Greater anthocyanin stability in açai pulp and semi-clarified juice may be linked to several factors including the stabilizing effect of non-anthocyanin polyphenolics (Garzon and Wrolstad 2002; Skrede et al. 1992); differences in polymeric anthocyanin concentrations, or the presence of residual lipids or insoluble solids that altered oxidative reaction rates (Pacheco-Palencia et al. 2007). Color deterioration of anthocyanin-containing products during storage not only results from anthocyanin degradation but also from the transformation of monomeric anthocyanins into higher molecular weight polymeric forms (Baranac et al. 1996; Francia-Aricha et al. 1997; Johnston and Morris 1997).

There was a significant difference in antioxidant capacity between pulpy juice and clear juice and this result indicated that clarifying treatment affects the functional properties of clarified açai juice. Clarification of açai juice resulted in a 29 % loss in antioxidant capacity (33.60 $\mu\text{M FeSO}_4/\text{g}$) and a high correlation was found between the decrease of antioxidant capacity and total anthocyanins losses (50 %). This result is comparable to previously reported values for açai juice (Pacheco-Palencia et al. 2007) and açai pulp (Lichtenthaler et al. 2003). Even though, there was a decrease in anthocyanin content and antioxidant capacity of the clarified açai juice due to the clarifying process when compared to other anthocyanin-rich fruits such as highbush blueberries, blackberries, and cranberries, considered novel sources of antioxidants for food (Del Pozo-Insfran et al. 2006), its antioxidant capacity is still significantly higher.

Conclusions

The use of **pectinolytic** enzymes associated to chitosan as a clarifying aid for açai juice proved to be highly effective once it resulted in a clear juice with a brighter purple to red color that was free of lipids, insoluble solids, and others substances that cause hazes. Even though, the clarifying process showed a significant decrease on total anthocyanin content and antioxidant capacity of the clarified açai juice, its antioxidant capacity is still higher when compared to other fruits recognized as rich sources of antioxidants. The obtained clarified açai juice is a genuinely high-value anthocyanin-rich product that could be used as colorant and functional ingredient to fruit juices and soft drinks as an interesting alternative to the growing market of food products associated to health and well-being.

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