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## Dragon fruit-like biocage as an iron trapping nanoplatform for high efficiency targeted cancer multimodality imaging

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## Abstract

Natural biopolymer based multifunctional nanomaterials are perfect candidates for multimodality imaging and therapeutic applications. Conventional methods of building multimodal imaging probe require either cross-linking manners to increase its *in vivo* stability or attach a target module to realize targeted imaging. In this study, the intrinsic photoacoustic signals and the native strong chelating properties with metal ions of melanin nanoparticle (MNP), and transferrin receptor 1 (TfR1) targeting ability of apoferritin (APF) was employed to construct an efficient nanoplatform (AMF) without tedious assembling process. Smart APF shell significantly increased metal ions loading (molar ratio of 1:800, APF/Fe<sup>3+</sup>) and therefore improved magnetic resonance imaging (MRI) sensitivity. Moreover, synergistic use of Fe<sup>3+</sup> and APF contributed to high photoacounstic imaging (PAI) sensitivity. AMF showed excellent bio-stability and presented good *in vivo* multimodality imaging (PET/MRI/PAI) properties (good tumor uptake, high specificity and high tumor contrast) in HT29 tumor because of its targeting property combined with the enhanced permeability and retention (EPR) effect, making it promising in theranostics and translational nanomedicine.

## Keywords

Apoferritin; Melanin; PET; MRI; PAI

#### Appendix A. Supplementary data

Supplementary data related to this article can be found at http://dx.doi.org/10.1016/

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## 1. Introduction

Non-invasive imaging techniques have attracted much attention due to their considerable roles in tumor diagnosis, prognosis and personalized treatment guidance [1-3]. Multimodality imaging provides comprehensive information and accomplishes synergistic advantages over any single modality alone [4-6]. The advance on nanomaterials research has offered numerous opportunity for designing multifunctional nanoprobes with unique chemical and physical properties [7, 8]. Nanomaterials with various building blocks have been explored for multimodal imaging, however, it is usually contradictory to simultaneously satisfy expected demands and avoid intrinsic limitations. For instance, polymeric micelles are with easy synthesis and modification, but they are not suitable for loading of hydrophilic payloads [9-11]. Inorganic nanoparticles possess good biocompatibility, controllable size and shape, but accompanying poor in vivo stability [12– 15]. Particularly, there is a growing need to accurately image biological targets (e.g. receptors over-expressed in tumor), which not only increases the cost of imaging probes, but also makes the preparation process troublesome. To address those issues, biological nanoparticles are perfect candidates because of their biodegradability and homogeneity. But their handling and further modification are very difficult [16–18]. Therefore, facile synthesis and construction of efficient multimodal imaging platforms is very significant.

As indispensable elements for living organisms, metal ions are also appropriate reporters for bioimaging because of their attractive physical characteristics such as magnetic, optical and radioactive properties [19, 20]. However, free metal ions show weak in vivo stability and cause severe toxixicity, therefore metal ions are usually bound with or encapsulated into a carrier. Melanin nanoparticle (MNP) is such a carrier that not only shows intrinsic chelating ability with metal ions (<sup>64</sup>Cu<sup>2+</sup>, Fe<sup>3+</sup>) for positron emission tomography (PET) and magentic resonance imaging (MRI) with a high loading capacity, but also possesses native optical properties for photoacoustic imaging (PAI) [21]. However, MNP complexed with metal ions could precipitate in aqueous solutions, and PEGylation was utilized to solve this problem and increase the water solubility of MNP. Moreover, in order to make naoparticles access to targeted tissues, apart from enhanced permeability and retention (EPR) effect, attaching a targeting moiety such as folic acid or RGD peptide is usually necessary [22–25], which makes the preparation process troublesome and also could result in heteogenous products.

Ferritin is a natural iron storage protein. It shows cage-like structure with small size (around 10 nm), large surface for modification, multi-channel (<1 nm) connecting the exterior with interior, dissociation-reassemble characters upon pH changes, making it a widely used nanoplatform for metal capture, drug delivery, bioimaging techniques and phototheramal therapy [18, 26–28]. More importantly, it possesses targeting ability to transferrin receptor 1 (TfR1) which is overexpressed in numerous type of cancer cells [29]. Ferritin that is not combined with iron is called apoferritin, and it is extensively utilized for cargo loading. These features suggest ferritin a smart platform with intrinsic targeting ability.

Herein, we report the integrated endogenous nanomaterials for enhanced targeted multimodality imaging through ingeniously space matching, i.e., embedding ultrasmall MNPs into the cavity of apoferritin (APF). APF played synergistic roles of guaranteeing

targeting and stability of the nanoplatform. First, we prepared ultra-small and water-soluble MNPs and then encapsulated one MNP together with high concentration of ferric irons into the cavity of APF to prepare nanomaterials AMF with core-shell structures. <sup>64</sup>Cu<sup>2+</sup> and Fe<sup>3+</sup> can be easily loaded into AMF nanoparticles cavities without adding additional chelating agents. More importantly, AMF can achieve TfR1 targeted cancer trimodal (PET/MRI/PAI) imaging (Fig. 1). To the best of our knowledge, AMF represents the first organic nanoplatform integrating both melanin and apoferritin with high and stable metal ions load for targeted cancer molecular imaging.

## 2. Results and discussion

## 2.1 Preparation and characterization of AMF

The inside cavity diameter of APF is ~8 nm, thus, ultrasmall MNPs below 8 nm were synthesized from commercial melanin granules (Supporting Information) according to previous report [21]. The average sizes of MNPs were ~4.0 nm measured by transmission electron microscopy (TEM) (Fig. 2B). Hydrodynamic diameters of MNPs were ~ 5.6 nm determined by dynamic light scattering (DLS, Fig. 2E), and zeta potential of MNP was -31.8 mV (Table S1). The molecular weight of MNP was determined to be around 61 kDa by matrix-assisted laser desorption/ionization time of flight mass spectrometry (MALDI-TOF-MS) (Fig. S2).

APF cage shows reversible pH-responsiveness, it disassembles into discrete subunits in strong acidic environment (pH 2.0 or 3.5), and re-assembles to a cage-like structure under basic conditions (pH 8) [30-32]. This smart architecture renders it convenient to encapsulate MNP and Fe<sup>3+</sup> within the cavity of APF to prepare AMF (Fig. 1). Usually, the loading yields of metal ions into apoferritin are very low; for example, it was reported that about 8-10 Gd-HPDO3A molecules and one molecule of [DFOFe] complex were trapped within an APF molecule [33, 34]. Since MR signal is positively associated with Fe<sup>3+</sup>, higher ferric ion capture is preferred. APF:Melanin:Fe<sup>3+</sup> at different ratios were used to prepare AMF, and the optimal conditions were determined as 1:1:1000 (Supporting Information). The loading vield of Fe<sup>3+</sup> to AMF was 71.36  $\pm$  2.56%, achieving a molar ratio of 1:1:800 APF:MNP:Fe<sup>3+</sup>. This was a very high loading. Previously, we found MNP quickly precipitated after adding Fe<sup>3+</sup>. Even after PEGylation, the maximum ratio of Fe<sup>3+</sup> to MNP-PEG was only around 90:1 [21]. In this study, the solution containing APF remained a clear brown solution at room temperature for at least 1 month (Fig. S3), whereas black precipitate was immediately observed in APF-free condition. On the other hand, in a MNP-free solution, Fe<sup>3+</sup> was stably encapsulated in acid condition, but after transfer to neutral or basic condition, Fe<sup>3+</sup> could leak freely from APF through channels between the exterior and interior cavities, resulting in red-brown precipitates. Therefore, in this circumstance, the final Fe<sup>3+</sup> loading vield within APF alone was only 4%. In contrast, using the APF-MNP dual platform, Fe<sup>3+</sup> ion loading increased more than 10 times and MNP inside APF dramatically enhanced the stability of Fe<sup>3+</sup> loading.

The nanomaterials prepared in aqueous solution were examined by TEM and DLS. APF showed in Fig. 2A was stained with uranyl acetate, which cannot penetrate the protein shell and therefore cannot stain the cavity, revealing APF as lighter round patches, which was

consistent with previous results.[30–32] The diameter was ~12.1 nm measured by TEM, the hydrodynamic diameter and zeta potential were ~13.6 nm and -17.2 mV respectively measured by DLS (Table S1). Fig. 2D showed that the AMF nanoparticle had the similar core-shell structure as APF, the diameter increased to 16.4 nm (Fig. 2E) and zeta potential of AMF was -18.4 mV, indicating both MNP and Fe<sup>3+</sup> were encapsulated inside the protein cavity. Fig. 2C showed a non-stained sample of AMF, discrete electron-dense sphere was clearly observed and the diameter was ~6 nm, which was a little larger than MNP (Fig. 2B), this suggested Fe<sup>3+</sup> was chelated by MNP inside the cavity of AMF. This was consistent with our previous work that the diameter of PEG-MNP increased from 7.0 nm to 8.9 nm after Fe<sup>3+</sup> loading [21]. All these confirmed that one MNP and many Fe<sup>3+</sup> ions were successfully loaded into the cage of APF.

## 2.2 Stability of AMF

Next, we investigated the stability of AMF. The synthesized AMF brown solution was subjected to dialysis in either phosphate-buffered saline (PBS,  $1\times$ , pH 7.4) or acetate buffer (pH=5). At pH 5, precipitation was observed within the dialysis bag (molecular weight cut-offs: 3 k and 100 k were both used in pH 7.4 and pH 5) after 1 h, while no color changed at pH 7. The dialysis process was followed by UV–VIS spectroscopy and ICP-MS analysis. No Fe<sup>3+</sup> or MNP release were detected within the dialysis solution out of bag within 120 h at 37 °C (Fig. S4), which clearly indicated that the Fe-MNP complex remained trapped inside the APF at pH 7 for long periods. The diameter of Fe-MNP complex was about 6 nm, inhibiting its leakage from the APF channels (< 1 nm). At pH 5, the protein cage disassembled, Fe-MNP was released from APF cage, however, the precipitated Fe-MNP complex was not separated, neither free Fe<sup>3+</sup> nor melanin were detected out of the dialysis bag.

## 2.3 Stability of <sup>64</sup>Cu-AMF

After stable and water-soluble AMF was prepared,  ${}^{64}Cu^{2+}$  was selected to label AMF for PET imaging due to its moderate half-life (12.7 h). Traditionally, the labeling of  ${}^{64}Cu^{2+}$  is performed by preconjugating chelators such as 1,4,7,10-tetraazacyclododecane-1,4,7,10-tetraacetic acid (DOTA) with the platform in a multistep process [35–37]. In this study,  ${}^{64}Cu^{2+}$  was loaded directly into AMF at pH 6 through multichannels of APF without any linker or nanocage disassembly. A dose calibrator (Capintec) showed that 80% of the radionuclide could be loaded into AMF after 30 min incubation at room temperature. This loading yield was higher than the previous report (~60%) [18]. To probe the stability of the loaded  ${}^{64}Cu^{2+}$ ,  ${}^{64}Cu^{2+}$ -labeled AMF was then subjected to dialysis in either PBS or fetal bovine serum (FBS), and the radioactivity decay was monitored over time. In both media,  ${}^{64}Cu^{2+}$  was well retained in the nanocages (Fig. S4), with less than 10% of the radioactivity (decay-corrected) released from the nanocages within the observation period of 24 h.

## 2.4 In vitro cell studies

Cell line HT29 has been reported to overexpress transferrin receptor 1 (TfR1), while HepG2 has low TfR1 expression [38, 39]. Therefore, HT29 and HepG2 cells were further used for cell uptake and blocking studies to investigate the targeting specificity of <sup>64</sup>Cu-AMF. As shown in Fig. S8, the uptake of <sup>64</sup>Cu-AMF decreased significantly (P < 0.05) at 1 and 2 h in

HT29 cell lines incubated with the blocking dose of free APF (1  $\mu$ M), and their uptakes were close to those in HepG2 cells. By contrast, the uptake of <sup>64</sup>Cu-AMF in HepG2 cells didn't show obvious alteration with or without APF blocking. These results indicate that APF coated nanoparticles specifically recognize the transferring receptor positive cells. Moreover, competitive binding of <sup>64</sup>Cu-AMF to HT29 cells was determined using APF as competitor (Fig. S9). The apparent eqilibrium dissociation constant was calculated and Kd value for <sup>64</sup>Cu-AMF was 20.2pM. Lastly, MTT assay also indicates that AMF has high biocomatability. No observed toxicity to the cells at even 10  $\mu$ M of AMF (Fig. S11).

#### 2.5 In vivo imaging

After confirming the structural composition and stability of <sup>64</sup>Cu-AMF, in vivo validation experiments were conducted. In brief, the nanoprobe was injected intravenously into tumorbearing mice and then imaged by PET, MRI and PAI. Human colon cancer HT29 [38] overexpressing TfR1 and liver cancer HepG2 with low TfR1 expression [39] were used to set-up animal models for PET/MRI/PAI imaging.

**2.5.1 PET imaging**—For PET studies, mice (n=4) were injected with about 100 µCi of <sup>64</sup>Cu-AMF via tail vein. Tomographic images were acquired on a Siemens Inveon microPET/CT scanner at various time points. The decay-corrected whole body images were shown in Fig. 3. It can be observed that the signal intensities gradually increased in the HT29 tumor area and reached maximum at 4 h post-injection (p.i.), followed by very slow signal intensity decrease (Fig. 3A). Quantification analysis of three-dimensional regions of interest (ROIs) over the images showed that the probe's biodistribution pattern was consistent with that previous reported [18, 21], and <sup>64</sup>Cu-AMF showed high accumulation in liver, which indicated that AMF was cleared mainly through hepatobiliary system. Moreover, <sup>64</sup>Cu-AMF displayed higher tumor uptakes in comparison to RGD-modified melanin, and the tumor uptakes were  $4.82 \pm 0.59$ ,  $6.14 \pm 0.77$ ,  $7.34 \pm 0.93$ ,  $7.26 \pm 1.32$ , and  $6.74 \pm 0.51$  % ID/g at 1, 2, 4, 18 and 26 h p.i., respectively (Fig. S10). Such high tumor accumulations were possibly caused by synergistic apoferritin-TfR1 specific interaction (Fig. S8) and EPR effect. To verify this, HepG2 liver cancer model was used as a negative control (n=4). Significant lower tumor uptakes were observed at all time points, which were  $2.95 \pm 0.40$ ,  $3.55 \pm 0.58$ ,  $4.33 \pm 1.16$ ,  $4.61 \pm 1.58$  and  $4.77 \pm 0.47$  %ID/g at 1, 2, 4, 18 and 26 h p.i., respectively. This observation corroborated our hypothesis that specific targeting was involved in the accumulation and retention of AMF in the TfR1 positive tumor. The uptakes of <sup>64</sup>Cu-AMF in most normal tissues were very low after 24 h, with persistent high uptake only in the liver (the site of transferrin production and metabolism). The liver uptakes of HT29 and HepG2 models were comparable (P > 0.05), reaching a peak of 1–2 h p.i before decreasing. Even after 48 h p.i., around 15 ID%/g liver uptakes were observed, confirming liver as the main metabolism and retaining organ for <sup>64</sup>Cu and nanoparticles [40-42].

**2.5.2 MRI imaging**—In MRI studies, relaxation times of the phantoms were first measured. By determining the variation in relaxation times of different concentrations of AMF, the relaxation rates (reciprocal values of relaxation times) were plotted against the amount of  $Fe^{3+}$  (1/T vs. [Fe<sup>3+</sup>]). Relaxivity values (r1; the slopes of each plot, mM<sup>-1</sup> s<sup>-1</sup>)

were determined to be 2.54 by using FeCl<sub>3</sub> as control and saturation test software in the 1.5T Bruker micro-MR system (Fig. S5). This value is over 2 fold compared to the r1 value of Fe-RGD-PEG-MNP ( $1.2 \text{ mM}^{-1} \text{ s}^{-1}$ ) [21], this high loading efficiency of Fe<sup>3+</sup> in AMF can largely improved MRI sensitivity. Tumor bearing mice (n=4) were then administered 10 nmol AMF through the tail vein. Significantly enhanced signals in HT29 tumor were observed with times ranging from 1 to 4 h. (Fig. 4). Ratios of HT29 tumor uptake after injection vs. pre-scan levels (T/T<sub>0</sub>) were 1.58, 1.73 and 2.02 at 1, 2 and 4 h p.i., respectively. However, HepG2 tumor mice showed lower signals with quick clearance of nanoprobes, the corresponding T/T<sub>0</sub> ratios were 1.21, 1.44 and 1.03 at 1, 2 and 4 h p.i., respectively. These results demonstrated that AMF can be successfully used for targeted MRI of tumor, and they also confirmed that AMF helped the uptake in TfR1 positive tumors.

**2.5.3 PAI imaging**—To study the PAI applications of AMF, we first tested the photoacoustic intensities of a tube phantom. Different concentrations of MNP, Fe-PEG-MNP, AMF, and AMF without Fe (APF+MNP) were loaded into separate tubes (Supporting Information). The AMF signal was about 2 times higher than the other 3 nanoparticles (p < 0.01), reaching maximum emission strength at 500 µg/mL (based on MNP concentration) (Fig. S6). These data suggested that the strong signal intensity of AMF stemmed from the combination of APF and Fe, and AMF has higher PAI sensitivity than MNP, Fe-PEG-MNP, and AMF without Fe. Furthermore, *in vivo* imaging studies were performed in HT29 mice models (n = 4). After injection of 20 nmol AMF, PAI contrast increased with time in HT29 tumor (Fig. 5). The tumor uptake reached 1.03 ± 0.02, 7.31 ± 2.33 and 24.15 ± 2.86 A.U. at 0, 2 and 4 h p.i., respectively. In contrast, no obvious PAI contrast was observed in the negative HepG2 tumor bearing mice (Fig. S12). These data clearly indicated that AMF was an excellent nanoprobe for PAI in addition to its capability for PET and MRI.

## 3. Conclusion

In summary, we report AMF as a nanoplatform for high efficiency targeted cancer multimodality imaging. APF is a focusing heat point because of its pH-dependant disassembly-assembly behavior and TfR1 targeting ability can behave as an efficient platform to simplify the combining of loading different imaging moieties and control the size of nanomaterial. Combining APF and MNP can increase the metal ions loading and improve the stability of the nanoplatform. At last but not least, AMF presented good *in vivo* tumor imaging properties. The nanoplatform is safe with complete utilization of natural materials, which is important for future translation into clinic use. The technique developed in this work can be potentially extended to load other types of functionalities such as imaging agents (Gd, <sup>89</sup>Zr, Mn and etc.), therapeutic radionuclides (<sup>90</sup>Y, <sup>177</sup>Lu and etc.) and targeting motifs (peptides, affibody, antibody and etc.).

## 4. Experimental section

#### 4.1 Chemicals and reagents

Apoferritin (APF) from equine spleen (A3641, 50 mg/mL) and melanin (M8631) were both obtained from Sigma-Aldrich. PD-10 column (17-0851-01), NAP-5 columns were

purchased from GE Health, and Slide-A-Lyzer® Dialysis Cassettes,  $100-500 \mu$ l, 3500 MWCO (66335) from Thermo Fisher Scientific.

#### 4.2 Preparation of small melanin nanoparticles (MNPs)

Ultra small and water soluble melanin nanoparticles were prepared according to the previous report [21]. Melanin was first dissolved in 1M NaOH aqueous solution before adjusting pH to 7 using aqueous 1M HCl solution under strong sonication. The bright black melanin aqueous solution was further purified and lyophilized to obtain black MNPs in solid state. The diameter was measured in water by Dynamic Light Scattering (DLS) using the ZetaSizer Nano-ZS (Malvern), and TEM images were taken using the JEOL JEM-2010 transmission electron microscope. Mass of MNPs was measured using Matrix-Assisted Laser Desorption/Ionization Time of Flight Mass Spectrometry (MALDI-TOF-MS) in Stanford University.

## 4.3 Preparation of melanin-Fe-loaded apoferritin nanoparticles (AMF)

Different ratios of APF, MNP and Fe<sup>3+</sup> were tested to identify the optimal conditions (Fig. S3) before selecting 1:1:1000. APF was first diluted with water (90 nmol in 4 mL water). The solution pH was then adjusted to 2 with 1M HCl while gently shaking at room temperature. MNP (90 nmol in 5 ml water) and FeCl<sub>3</sub> (90 µmol in 135 µL acidic water) were added into the acid solution. The pH value was maintained for about 15 min and then adjusted to pH 8–10 by adding 1 M NaOH. The mixture was continuously stirred for 2 h at room temperature. The resulting solution was purified by PD-10 column pre-equilibrated with PBS buffer and concentrated by centrifugal-filter (Amicon, MWCO = 100 kDa) to get AMF.

## 4.4 Preparation of <sup>64</sup>Cu-AMF

Contrary to the literature [18], <sup>64</sup>Cu radiolabeled AMF without disrupting the cage at pH=2. <sup>64</sup>CuCl<sub>2</sub> (1 mCi) was added into AMF solution directly. The mixture was incubated at temperature for 30 minutes. The products were then purified by PD-10 column to remove free <sup>64</sup>Cu<sup>2+</sup>. The fractions eluted from PD-10 were measured for radioactivity and the most concentrated fractions were used for animal experiments. For a stability test, purified <sup>64</sup>Cu-AMF (300  $\mu$ l, ~100  $\mu$ Ci, respectively) were added to a 500  $\mu$ L mini-dialysis unit. The samples were then dialyzed against 10 mL PBS buffer or FBS. The radioactivity remaining in the dialysis units and those leaked into the outer solution were both measured by a radioisotope calibrator.

#### 4.5 PET imaging

The details of small-animal PET imaging and region-of-interest (ROI) analysis have been reported previously [43]. In brief, each mouse was injected with about 100  $\mu$ Ci of <sup>64</sup>Cu-AMF via tail vein. Mice bearing tumors (n=4) underwent microPET/CT (Seimens Inveon) scans at various time points after the injection. For each microPET/CT scan, 3-dimensional ROIs were drawn over the tumor or other normal organs on decay-corrected whole body coronal images. The %ID/g was then derived from the readings by IAW software.

### 4.6 MR imaging

Phantom Imaging: AMF was prepared in Eppendorf tubes at varying concentrations for imaging. The T<sub>1</sub> relaxation times were measured on a 1.5 T Bruke MRI scanner. FeCl<sub>3</sub> was measured in the same manner as control. The parameters were as follows: TR/TE=50, 100, 200, 500, 750, 1000, 1500, 2000/8.8 ms; SI: 3.0 mm; FOV: 6.0 cm; MTX: 256/128). The signal intensities for each of the ROIs on the T<sub>1</sub> map were measured for each concentration. T<sub>1</sub> saturation method within the machine was selected to calculate T<sub>1</sub> values. Specific relaxivities (r1) of AMF or FeCl<sub>3</sub> were calculated from the plots of T<sub>1</sub><sup>-1</sup> versus concentration of Fe in AMF or FeCl<sub>3</sub>.

Small animal MR Imaging: The animals (n=4) were anesthetized and set into the 1.5 T micro-MR-compatible cradle. During MRI, the animal was anesthetized with inhalation of 2% isoflurane with oxygen-enriched air through a facemask. Temperature was maintained at  $36 \pm 1$  °C using a water bath underneath the cradle and electrocardiogram (ECG) was used to detect the heart rate. The prepared AMF (10 nmol/200 µL per mouse) were intravenously administered. To investigate the time course distributions of AMF in the mouse body, MRI was performed before and 1, 2 and 4 h after the administrations. High-resolution AMF contrast-enhanced multislice MR images were obtained from each tumor by using T<sub>1</sub>-flash MRI sequence: repetition time (TR)/echo time (TE) = 300/6.1 ms; number of experiment (NEX) = 16; Matrix: 256 × 256; slice thickness: 1.25 cm; FOV:3.0 cm; and 15 slices. All images were analyzed using ImageJ software. ROIs were drawn over the tumor and the MR signal was quantified as SItumor, where SItumor is the mean signal intensity of the tumor accumulation. The SI ratio at time interval after AMF injected was calculated as T/T<sub>0</sub>.

#### 4.7 Photoacoustic imaging

PA/Ultrasound (US) coregistered images were acquired with a LAZR commercial instrument (VevoLAZR; VisualSonics) equipped with a LZ-250 linear array transducer and a tunable Nd:YAG laser system (680–970 nm) [44]. Polyethylene (PE) tubes with 1~ 2 cm lengths and 0.7 mm inner diameters were heat-sealed at one end, then filled with water, FeCl<sub>3</sub>, MNP, Fe-PEG-MNP, Apoferritin-MNP-Fe (AMF), AMF without Fe (Apoferritin-MNP), and finally heat-sealed on the opposing side. According to previous report [21], we selected the laser excitation wavelength as 680 nm. Mice bearing HT29 tumors (n= 4) were imaged at 680 nm before and 2 and 4 h after the intravenous injection of 20 nmol AMF. Images were exported as 3-dimensional volume tiff files that were stacked together using ImageJ. Normalization and quantification of the tumor images were also done in ImageJ. The PA signals reported were the average pixel intensities from within the ROI.

## Supplementary Material

Refer to Web version on PubMed Central for supplementary material.

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**Fig. 1.** Schematic illustration of AMF nanocage synthesis.



## Fig. 2.

AMF nanocages characterized in A (negative stained TEM images of APF) and D (negative stained images of AMF). B: Unstained TEM images of MNP. C: Fe-MNP in AMF. E: Diameters of MNP, APF and AMF by DLS.

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## Fig. 3.

Representative coronal small animal PET images of HT29 (top) and HepG2 (bottom) tumorbearing mice (n=4) at 1, 2, 4, 18 and 26 h after administration of <sup>64</sup>Cu-AMF. White circles indicate tumors location.



## Fig. 4.

T1 MR images of HT29 (upper) and HepG2 (lower) tumor model after administration of AMF, red circles indicate the area of tumors.

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Pre

2h

4h

## Fig. 5.

The US (grey, top), PA (red, middle) and overlayed coronal sections (bottom) of HT29 tumor models before and after tail-vein injection of AMF nanocages.