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Photocatalytic killing effect of TiO₂ nanoparticles on Ls-174-t human colon carcinoma cells

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Abstract

AIM: To investigate the photocatalytic killing effect of photoexcited $TiO₂$ nanoparticles on human colon carcinoma cell line (Ls-174-t) and to study the mechanism underlying the action of photoexcited $TiO₂$ nanoparticles on malignant cells.

METHODS: Ls-174-t human colon carcinoma cells were cultured in RPMI 1640 medium supplemented with 199 mL/L calf serum in a humidified incubator with an atmosphere of 50 mL/L CO₂ at 37 °C. Viable cells in the samples were measured by using the MTT method. A GGZ-300 W high pressure Hg lamp with a maximum ultraviolet-A (UVA, 320- 400 nm) irradiation peak at 365 nm was used as light source in the photocatalytic killing test.

RESULTS: The photocatalytic killing of Ls-174-t cells was carried out *in vitro* with $TiO₂$ nanoparticles. The killing effect was weak by using UVA irradiation without $TiO₂$ nanoparticles. In our studies, the photocatalytic killing effect was correlated with the concentration of $TiO₂$ and illumination time. Once $TiO₂$ was added, Ls-174-t cells were killed at a much higher rate. In the presence of 1 000 μ g/mL TiO₂, 44% of cells were killed after 10 min of UVA irradiation, and 88% of cells were killed after 30 min of UVA irradiation.

CONCLUSION: When the concentration of $TiO₂$ is below $200 \mu g/mL$, the photocatalytic killing effect on human colon carcinoma cells is almost the same as that of UVA irradiation alone. When the concentration of TiO₂ is above 200 μ g/mL, the remarkable killing effect of photoexcited $TiO₂$ nanoparticles can be found.

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INTRODUCTION

The application of $TiO₂$ photocatalysis has received increasing attention since the first report of microbiocidal effects by Matsunaga *et al.* in 1985^[1]. In recent years, in contrast to many studies using $TiO₂$ powder for photodecomposition of organic pollutants^{$[2-9]$}, few studies have investigated the application of $TiO₂$ in life science, especially in the field of cancer treatment^[10-12]. The incidence of colon cancer is rising in China. Despite that surgical operation is used currently, people have recognized its limitation. The way to treat cancer usually includes radiation therapy and chemical therapy, which may generate severe side effects in human body. Therefore, this study tried to investigate a new therapy for cancer. Ls-174-t cells were used as experiment objects in this study. The photocatalytic killing effect of $TiO₂$ nanoparticles on malignant cells and its killing mechanism were investigated.

MATERIALS AND METHODS

Reagent preparation

 $TiO₂$ colloid solutions were prepared^[13,14] by hydrolysis of titanium isopropoxide, Ti [OCH (CH3) 2] 4 (97%, Aldrich Chemical Co). In brief, 12.5 mL of Ti [OCH $(CH_3)_2$] was added to 2 mL isopropanol, then the mixture was added to 150 mL of distilled deionized water containing 2 mL of 700 mL/L nitric acid and vigorously stirred for 6 h at 75 °C. Approximately 150 mL of $TiO₂$ colloid solution being stable for several months at 4 °C was obtained after the organic layer was removed. The average diameter determined by Zetasizer 3000 HS_A(USA) was 21.2 nm.

The pH value of $TiO₂$ colloid solutions used in the subsequent experiment had to be adjusted from 1.8 to 5.5-6.5 in order not to damage the normal growth of cells. Therefore, 1 mol/L NaOH aqueous solution and 1.5 mL/L polyvinyl alcohol were added to the colloid solutions before the pH adjustment to prevent the $TiO₂$ from precipitation. The final $TiO₂$ colloid solutions were sterilized by autoclaving and then diluted to the required concentration. Other chemical reagents used were all of analytical purity from commercial sources.

Cell culture and treatment

Human colon carcinoma cell lines Ls-174-t were purchased from Shanghai Institute of Cell Biology, Chinese Academy of Sciences, Shanghai, China. The cell lines were cultured in RPMI 1640 (Gibco) medium supplemented with calf serum 100 ml/L, penicillin ($100\times10^3 \mu/L$) and streptomycin (100 mg/L). pH was maintained at 7.2-7.4 by equilibration with 50 mL/L $CO₂$. Temperature was maintained at 37 °C. Cells were sub-cultured with a mixture of ethylenedinitrile tetraacetic acid (EDTA) and trypsin. All experiments were performed using cells during the exponential growth phase. Cell concentration was determined by using a hemocytometer and the cell density was adjusted to the required final concentration.

Ls-174-t cells were treated with $TiO₂$ diluted in RPMI 1640 medium for 2 h at 37 $^{\circ}$ C. Then the solutions were irradiated with a GGZ-300W high pressure Hg lamp (E_{max} = 365 nm) at room temperature. A UV pass filter was used to obtain a light wavelength between 300-400 nm. The light intensity at the liquid surface was measured by a VLX-3W radiometer-photometer (USA). The incident light intensity was 3.7 mW/cm^2 . In our study, three groups of tests were carried out. One group was treated in the absence of TiO₂. Another group was treated in the absence of UVA. The third group was treated with different $TiO₂ concentrations and irradiated by UVA.$

Measurement of the viability of Ls-174-t cells

Viable cells in the samples were measured by using the MTT staining method^[15]. MTT [3- (4, 5-dimethylthiazol-2-yl) -2, 5diphenyl tetrazolium bromide] was dissolved in phosphatebuffered saline (PBS, pH 7.4) at 5 mg/mL and filtered to be sterilized. Twenty microliters of stock MTT solution were added to all wells for an assay, and plates were incubated at 37 \degree C for 4 h. One hundred and fifty microliters of DMSO were added to all wells and mixed thoroughly to dissolve the blue-violet crystals. After a few minutes at room temperature to ensure that all crystals were dissolved, the plates were read on a Bio-Rad Novapathtm microplate reader (Japan), using a test wavelength of 595 nm, taking the solution without MTT as control. Then optical absorptions [*A*] were obtained. Plates were normally read within 1 h after DMSO was added. The survival rate could be calculated according to $[A]/[A]$ _i, where $[A]_i$ is the optical absorption of untreated cells.

RESULTS

Determination of the average diameter of TiO2 particles

Zetasizer 3000HSA (USA) was used to determine the average diameter of $TiO₂$ nanoparticles. The result is shown in Figure 1. The average volume size of $TiO₂$ nanoparticles was 21.2 nm.

Cytotoxicity of TiO2 nanoparticles

The cytotoxicity of $TiO₂$ nanoparticles (without UVA irradiation) was determined by exposing cells to various concentrations of $TiO₂$ in RPMI 1640 (Gibco) medium for 24 h. The surviving fraction of the cells was greater than 90% when the concentration of $TiO₂$ was in the range of 1 000 µg/mL, as shown in Figure 2. The result confirmed that nonirradiated $TiO₂$ nanoparticles were not toxic to the Ls-174-t cells. It was consistent with those in literatures $[16-17]$.

Effect of photoexcited TiO2 nanoparticles on Ls-174-t cells

The fact that the surviving fraction was greater than 90% after 30 min as shown in Figure 3 (A) indicated that $TiO₂$ nanoparticles without UVA irradiation showed little toxicity to living cells. The killing effect of UVA without $TiO₂$ is shown in (B) with the surviving fraction of Ls-174-t cells given as a function of the UVA light irradiation time. About 20% cells were killed after 30 min exposure, whereas after 20 min exposure, more than 90% of the cells survived. Once $TiO₂$ was added, the Ls-174-t cells were killed at a much higher rate as shown in (C). For example, in the presence of $1000 \mu g/mL$ of TiO₂, 44% of the cells were killed after 10 min of UVA irradiation, and after 30 min irradiation, 80 % of the cells were killed. Therefore it was concluded that photoexcited $TiO₂$ nanoparticles had an active killing effect on Ls-174-cells.

Effect of TiO₂ concentration on Ls-174-t cells activity

The effect of the $TiO₂$ concentration ranging from 200 to 1 000 µg/mL on the rate of cell killing by UVA light is shown in Figure 4. The light intensity was 3.7 mW/cm² and kept constant. The experimental results demonstrated that cell viability decreased monotonically as $TiO₂$ concentration increased and cell viability decreased with time.

Although a higher concentration of $TiO₂$ could achieve a higher reaction rate, the difficulties in separation and measurement had to be considered. So the effect of a higher $TiO₂$ concentration $(C>1029 \mu g/mL)$ on Ls-174-t cell activity was not investigated. The concentration of TiO₂ was set at <1 029 μ g/mL with which the dark cytotoxicity was considered to be negligible. The range of TiO₂ concentrations was close to that used previously^[18].

Morphological changes of Ls-174-t cells

Untreated and $TiO₂$ -treated cells were collected by centrifugation and resuspended in RPMI 1640 medium. The samples were pipetted into a 24-well plate, which was directly observed with

an inverted phase-contrast microscope. When treated by photoexcited $TiO₂$, the cellular shape was condensed and nuclei were dispersed in fragments.

Figure 1 Volume size distribution of TiO₂.

Figure 3 Effect of light and $TiO₂$ on viability of Ls-174-t cells. (A) TiO₂ (1 000 μ g/mL) in the dark; (B) no TiO₂ in the light; (C) TiO₂ (1 000 µg/mL) in the light. Initial cell concentration: 5×10^{5} cell/mL, light intensity: 3.7 mW/cm2 .

Figure 4 Influence of TiO₂ concentration on Ls-174-t cells activity. (A) 204 µg/mL,(B) 524 µg/mL,(C) 804 µg/mL,(D) 1 029 µg/mL.

DISCUSSION

In this experiment, the $TiO₂$ nanoparticle system was used, displaying its superiority. TiO₂ nanoparticles were easy to attach to the cellular membranes and accumulate. They were also easy to enter into the cytoplasm via phagocytosis^[19]. It could lead to accumulation of ROS on the surface of cell membranes and in the cytoplasm. Hence under light irradiation, $TiO₂$ nanoparticles had more significant cell killing effect *in vitro*.

Human colon carcinoma cells treated with photoexcited TiO₂ nanoparticles $(C>200 \mu g/mL)$ were effectively damaged, with cells contracted and a lot of cell fragments simultaneously observed by an inverted phase-contrast microscope^[18].According to these characteristics, we assumed that the mechanism of photoexcited $TiO₂$ in killing human colon carcinoma cells might be through a series of oxidized chain reactions and in inducing cell death by reactive oxygen species^[20-25]. Human colon carcinoma cell damages occurred in two stages. The initial oxidative damage took place on the cell membranes, where the $TiO₂$ photocatalytic surface had its first contact with intact cells, the membranes became somewhat permeable. At this stage the cells did not lose their viability. Photocatalytic action made the cell membranes permeable, intracellular components began to leak from the cells and free $TiO₂$ nanoparticles might also diffuse into the damaged cells and directly attack intracellular components, eventually leading to cell death. It is different from the bactericidal effect of $TiO₂$ photocatalytic reaction. Bacteria are simple prokaryotic cells that do not contain the nucleus characteristics of eukaryotic cells. Whereas human colon carcinoma cells are eukaryotic cells and their structure is complex. Based on their structural differences, we assumed that killing cancer cells might be more difficult than killing bacteria by the photocatalytic reaction of $TiO₂$ nanoparticles.

 In the present study, cultured human colon carcinoma cells were effectively killed by photoexcited TiO₂ nanoparticles *in virto*. The concentration of $TiO₂$ affected the photocatalytic killing effect. When the concentration of $TiO₂$ was below 200 μ g/mL, there was only a slight decrease in survival ratio after UVA irradiation for more than 30 min. It was almost the same as that of UVA irradiation alone. It indicated that minor cell membrane leakage might occur and the cell viability was not lost. When the concentration of TiO₂ was above 200 μ g/mL, the survival ratio decreased rapidly with increasing $TiO₂$ concentration. It indicated that major rupture of cell membranes and decomposition of essential intracellular components might take place, thus accelerating cell death. It verified the mechanism of $TiO₂$ nanoparticles in killing human colon carcinoma cells.

The photocatalytic killing effect of $TiO₂$ nanoparticles on human colon carcinoma cells suggested the idea of cancer treatment using $TiO₂$ nanoparticles and light irradiation. Under these conditions, it could be adapted to an anticancer modality by the local or regional treatment of the tumor with $TiO₂$ nanoparticles, followed by light irradiation focusing on the tumor. Although UVA light (320-400 nm) cannot penetrate the human body deeply, it may be possible that the modality will be applied to several human tumors in the future.

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