

A Matter of Timing: Contrasting Effects of Hydrogen Sulfide on Oxidative Stress Response in *Shewanella oneidensis*

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ABSTRACT

Hydrogen sulfide (H_2S), well known for its toxic properties, has recently become a research focus in bacteria, in part because it has been found to prevent oxidative stress caused by treatment with some antibiotics. H_2S has the ability to scavenge reactive oxygen species (ROS), thus preventing oxidative stress, but it is also toxic, leading to conflicting reports of its effects in different organisms. Here, with *Shewanella oneidensis* as a model, we report that the effects of H_2S on the response to oxidative stress are time dependent. When added simultaneously with H_2O_2 , H_2S promoted H_2O_2 toxicity by inactivating catalase, KatB, a hemecontaining enzyme involved in H_2O_2 degradation. Such an inhibitory effect may apply to other heme-containing proteins, such as cytochrome *cbb*₃ oxidase. When H_2O_2 was supplied 20 min or later after the addition of H_2S , the oxidative-stress-responding regulator OxyR was activated, resulting in increased resistance to H_2O_2 . The activation of OxyR was likely triggered by the influx of iron, a response to lowered intracellular iron due to the iron-sequestering property of H_2S . Given that *Shewanella* bacteria thrive in redox-stratified environments that have abundant sulfur and iron species, our results imply that H_2S is more important for bacterial survival in such environmental niches than previously believed.

IMPORTANCE

Previous studies have demonstrated that H_2S is either detrimental or beneficial to bacterial cells. While it can act as a growthinhibiting molecule by damaging DNA and denaturing proteins, it helps cells to combat oxidative stress. Here we report that H_2S indeed has these contrasting biological functions and that its effects are time dependent. Immediately after H_2S treatment, there is growth inhibition due to damage of heme-containing proteins, at least to catalase and cytochrome *c* oxidase. In contrast, when added a certain time later, H_2S confers an enhanced ability to combat oxidative stress by activating the H_2O_2 -responding regulator OxyR. Our data reconcile conflicting observations about the functions of H_2S .

s an important signaling molecule, along with nitric oxide (NO) and carbon monoxide (CO), hydrogen sulfide (H₂S) has been a focus of research in mammals in recent years and has been found to have many beneficial functions (1). The most important one is its cytoprotective effect against oxidative stress (2). Although studies of the physiological roles of H₂S in bacteria have lagged behind significantly, it was recently reported that H₂S is important in the abilities of certain species, such as Escherichia coli, Bacillus anthracis, Pseudomonas aeruginosa, and Staphylococcus aureus, to survival and grow well in various niches (3). By stimulating protection against reactive oxygen species (ROS), H₂S provides bacteria a general defense against antibiotics (3). However, H₂S has also long been known for its toxic properties, especially as a growth inhibitor by damaging DNA, by denaturing proteins through disruption of disulfide cross-links, by inactivating the redox centers of metalloenzymes, and by enhancing oxidative stress (4-7). H₂S significantly inhibits the growth of several bacteria, including E. coli, Bacillus subtilis, Salmonella enterica serovar Typhimurium, and S. aureus, as well as some marine bacteria (7, 8). In addition, H₂S was also found to act as an antifungal agent for several pathogens, including Aspergillus niger and Penicillium ita*licum*, by decreasing the activities and expression of superoxide dismutase (SOD) and catalase (CAT) (8). While the contrasting effects of H₂S reported in these studies may be attributed to differences in the concentrations used, it is possible that H₂S has distinct impacts on different species of bacteria.

Shewanella bacteria are facultative Gram-negative gammaproteobacteria with a remarkable respiratory versatility and potential

for bioremediation of metals, as well as for its use in microbial fuel cells (9, 10). Shewanella species are also gradually emerging as human and animal pathogens, as reports of Shewanella infections have been increasingly reported (11). Owing to these properties, some species, the model species Shewanella oneidensis in particular, have been intensively studied. There is endogenous H₂S generation in S. oneidensis (12), but not until recently was its enzymatic foundation determined (13-15). Through anaerobic respiration of inorganic sulfur compounds in the periplasm, including thiosulfate $(S_2O_3^{2-})$, sulfite (SO_3^{2-}) , tetrathionate $(S_4O_6^{2-})$, and elemental sulfur (S⁰), S. oneidensis generates H₂S with sulfite reductase SirACD, as well as thiosulfate and polysulfide reductase PsrABC (13, 14, 16). Cysteine degradation via methionine γ -lyase MdeA is the predominant source of endogenous H₂S production, with other mechanisms via SO_1095 and SseA being of lesser importance (15). While MdeA and SO_1095 are homologous to P.

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TABLE 1 Strains and plasmids used in this study

| Strain or plasmid | Description | Reference or source |
|-------------------------------|---|----------------------------------|
| E. coli | | |
| DH5a | Host strain for plasmids and general use | Lab stock |
| WM3064 | Donor strain for conjugation; $\Delta dapA$ | W. W. Metcalf, UIUC ^a |
| S. oneidensis | | |
| MR-1 | Wild type | ATCC 700550 |
| HG1070 | $\Delta katB$ mutant derived from MR-1 | 19 |
| HG1095 | $\Delta SO1095$ mutant derived from MR-1 | 15 |
| HG1261 | Δ sseA mutant derived from MR-1 | 15 |
| HG1328 | $\Delta oxyR$ mutant derived from MR-1 | 19 |
| HG1812 | $\Delta m deA$ mutant derived from MR-1 | 15 |
| HG2364 | $\Delta ccoN$ mutant derived from MR-1 | 40 |
| HG2903 | $\Delta cysK$ mutant derived from MR-1 | 15 |
| HG4056 | $\Delta met B$ mutant derived from MR-1 | 15 |
| $\Delta cysK^{C}$ | $\Delta cysK$ mutant with copy of <i>cysK</i> integrated into chromosome | 15 |
| $\Delta triple$ | $\Delta m deA \Delta SO1095 \Delta sseA$ mutant derived from MR-1 | This study |
| $\Delta penta$ | $\Delta m deA \Delta SO1095 \Delta sseA \Delta psrA \Delta sirA$ mutant derived from MR-1 | This study |
| Plasmids | | |
| pHGM01 | Ap ^r Gm ^r Cm ^r suicide vector | 23 |
| pHG101 | Promoterless broad-host-range Km ^r vector | 24 |
| pHG102 | pHG101 containing S. oneidensis arcA promoter | 24 |
| pHGEI01 | Integrative <i>lacZ</i> reporter vector | 32 |
| pBBR-Cre | Sp ^r helper plasmid for antibiotic cassette removal | 39 |
| pHGE-P _{katB} -lacZ | Reporter vector carrying P _{katB} -lacZ | This study |
| pHGE-P _{dps} -lacZ | Reporter vector carrying P_{dps} -lacZ | This study |
| pHGE-P _{ahpC} -lacZ | Reporter vector carrying P_{ahpC} -lacZ | This study |
| pHGE-P _{katG1} -lacZ | Reporter vector carrying P_{katG1} -lacZ | This study |

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aeruginosa cystathionine γ -lyase (CSE), SseA is a homologue of 3-mercaptopyruvate sulfurtransferase (3MST) (3, 15, 17).

Endogenous H₂S is particularly important for Shewanella species because it is critical to iron reduction (18). In alkaline environments containing multiple electron acceptors, including sulfur and iron species, S. oneidensis first generates H₂S (HS⁻), which in turn reduces iron compounds abiotically. In parallel, in the redox-stratified environments where Shewanella bacteria thrive, ROS are likely to occur. Hence, for these bacteria, H₂S may not only function as a reductive chemical for abiotic iron reduction but also play a role in the cellular response to oxidative stress. Like E. coli and many other bacteria, S. oneidensis utilizes OxyR, a LysR family transcriptional regulator, as the predominant regulator mediating the cellular response to H_2O_2 (19–21). OxyR controls a large number of genes by acting as both an activator and a repressor. Two such genes under OxyR repression, katB and dps, are particularly important in protecting cells from H_2O_2 damage (19). While KatB is a CAT dominating H₂O₂ degradation, Dps, as an iron-sequestering protein, plays a role in the control of cellular iron homeostasis, especially when cells are challenged by H_2O_2 .

In this study, we have attempted to understand the effects of H_2S on the response of *S. oneidensis* to H_2O_2 . We found that H_2S either aggravates or protects from H_2O_2 lethality, depending on the time of H_2O_2 addition. Enhanced killing was due primarily to H_2S inactivation of KatB, and such a mechanism appeared to be applicable to some other heme-containing proteins. In contrast, protection against H_2O_2 induced by pretreatment with H_2S depended on the activation of OxyR, the master regulator mediating the cellular response to H_2O_2 (19–21). This appears to be due to

iron influx triggered by lowered intracellular free-iron levels resulting from the iron-sequestering activity of H₂S.

MATERIALS AND METHODS

Bacterial strains, plasmids, and culture conditions. The bacterial strains and plasmids used in this study are described in Table 1. All chemicals were acquired from Sigma Co. (Shanghai, China) unless otherwise noted. Information about the primers used in this study is available upon request. For genetic manipulation, *E. coli* and *S. oneidensis* strains were grown in lysogeny broth (LB; Difco, Detroit, MI) under aerobic conditions at 37 and 30°C, respectively. When needed, the growth medium was supplemented with 2,6-diaminopimelic acid at 0.3 mM, ampicillin at 50 μ g/ml, kanamycin at 50 μ g/ml, and gentamicin at 15 μ g/ml.

H₂**O**₂ **sensitivity.** The response of *S. oneidensis* to H₂O₂ was assessed as previously described (19). For growth analysis, overnight cultures from a single colony on LB plates were inoculated into fresh LB to an optical density at 600 nm (OD₆₀₀) of ~0.01. The cultures were supplemented with chemicals as indicated in Results and the figure legends and monitored for growth by recording OD₆₀₀ values. To assess susceptibility to H₂O₂, properly diluted mid-log-phase cultures (OD₆₀₀ of ~0.2) were spread onto fresh LB plates (200 µl of culture, approximately 10⁶ CFU). After a bacterial lawn had become visible, paper discs 6 mm in diameter containing 10 µl of 5 M H₂O₂ were placed on plates, which were incubated for 16 h at 30°C before photography. To determine the effects of chemicals on the survival of *S. oneidensis*, mid-log-phase cultures were challenged and at various time intervals, samples were serially diluted and plated onto LB. Colonies were counted after overnight incubation at 30°C. All experiments were repeated at least three times.

 H_2S detection. To keep levels of H_2S from both endogenous and exogenous sources in growing cultures relatively stable, cells for H_2S -related physiological assays were grown under aerobic conditions on a shaker

at 50 rpm at 30°C. H_2S production in *S. oneidensis* was monitored by a lead acetate detection method and the methylene blue formation assay (3, 15, 22).

Mutagenesis and genetic complementation. *S. oneidensis* in-frame deletion strains were constructed by the *att*-based fusion PCR method (23). In brief, two fragments flanking the target gene were generated by PCR with primers containing *attB* and gene-specific sequences and joined by a second round of PCR. The fused fragments were introduced into plasmid pHGM01 by site-specific recombination with BP Clonase (Invitrogen) according to the manufacturer's instructions. The resulting vectors were transformed into *E. coli* WM3064 and then transferred into relevant *S. oneidensis* strains via conjugation. Mutagenesis constructs integrated into the chromosome were selected by resistance to gentamicin and confirmed by PCR. These transconjugants were grown in LB broth in the absence of NaCl and plated on LB supplemented with 10% sucrose. Gentamicin-sensitive and sucrose-resistant colonies were screened by PCR for deletion of the target gene. All mutations were verified by sequencing the mutated regions.

All mutants from previous studies were successfully complemented by a copy of the corresponding gene on plasmids pHG101 and pHG102 (Table 1) (24, 25). In this study, these complementation vectors were used with similar results.

DNA damage assays. Measurement of DNA damage was initially performed by quantitative PCR (qPCR) (26). Total genomic DNA was isolated from 10 ml of mid-log-phase cultures with a DNeasy Tissue kit (Qiagen) and quantified with NanoVue (GE Health Care). A fragment of \sim 10 kb close to the *ccm* (encodes the cytochrome *c* maturation system) region was selected for analysis. qPCR was performed with an ABI 7300 96-well qPCR system (Applied Biosystems) as described previously (27), and DNA damage was calculated by an established method (26). The amount of amplified product from the treated samples was normalized to that from an untreated control. DNA damage was also assessed by measurement of the rate of mutagenesis with the thyA (encodes thymidylate synthase) forward mutagenesis assay scoring resistance to trimethoprim (TMP) (28). TMP is a dihydrofolate reductase inhibitor that suppresses the growth of wild-type cells but not that of mutants lacking *thyA* as long as thymine is present in the medium. Mid-log-phase cultures were treated with H₂O₂, H₂S, or both and incubated for 20 min. After serial dilution, 250 µl was plated on LB containing 0.2 mg/ml thymine and 0.1 mg/ml TMP and incubated at 30°C.

Enzyme assays. Mid-log-phase cultures were centrifuged, and cell pellets were washed with 0.2 M phosphate-buffered saline (PBS, pH 7.2), resuspended in PBS to an OD₆₀₀ of ~0.5, and lysed by sonication. After centrifugation for 10 min at 10,000 rpm, the supernatants were used as crude enzyme extracts. H₂O₂ concentrations in aliquots of crude extracts were determined at various time points, as indicated in the relevant figures, by a method described previously (29). The same procedure was used to measure H₂O₂-degrading activity by extracts and by a commercial heme CAT from bovine liver (Sigma, St. Louis, MO). SOD and peroxidase (POD) activities were measured on the basis of pyrogallol autoxidation and guaiacol as the electron donor, respectively (29, 30). Quantitative analysis of cytochrome *c* oxidase (Cco) was carried out as described previously (31).

Expression analyses. β-Galactosidase assays were performed to determine gene expression with the integrative *lacZ* reporter plasmid pHGEI01 (32). In brief, the sequence of ~400 bp upstream of a gene of interest was amplified and placed immediately upstream of the full-length *E. coli lacZ* gene. The resulting vector was maintained in *E. coli* WM3064 and transferred to *S. oneidensis* after verification by sequencing. Log-phase cells (OD₆₀₀, ~0.2) were harvested by centrifugation, washed with PBS, suspended in lysis buffer (0.25 M Tris HCl, 0.5% Triton X-100, pH 7.5) for 30 min, and assayed with *ortho*-nitrophenyl-β-D-galactopyranoside as described previously (24). Activity was determined by monitoring color development at 420 nm with a Synergy 2 Multi-Detection microplate reader (M200 Pro; Tecan) and reported in Miller units. In addition, gene

expression was determined by qRT-PCR, which was performed with an ABI 7300 96-well qRT-PCR system (Applied Biosystems) essentially as described above.

Quantification of intracellular total and free iron species. The total cellular iron content of 1-liter cultures was assayed by inductively coupled plasma mass spectrometry (ICP-MS) as described elsewhere (33). For sample preparation, log-phase cells (OD_{600} of ~ 0.2) grown under the conditions indicated in Results or the figure legends were harvested by centrifugation, washed with PBS, and lysed by sonication. The supernatants were analyzed by ICP-MS on an iCAP (Thermo Scientific). The iron content was normalized to the total protein in the lysates. All data were evaluated with at least three independent biological experiments. Measurements of free iron species were performed with a calcein assay as described previously (21, 34).

Other analyses. Experimental values were subjected to statistical analyses and presented as the mean \pm standard deviation (SD). Student's *t* test was performed for pairwise comparisons of groups.

RESULTS

Endogenous H₂S has a negligible role in protecting cells from H_2O_2 . S. oneidensis is able to produce H_2S through both cysteine degradation and respiration of sulfur species (13-15). Under aerobic conditions, H₂S produced through respiration is negligible, especially when the relevant sulfur species is not added (15). Thus, endogenous H₂S results from the activity of the CSE homologues MdeA and SO1095, as well as from the 3MST homologue SseA, with MdeA accounting for >70% (15) (Fig. 1A). In addition, loss of CysK resulted in H₂S hyperproduction. To test whether endogenous H₂S plays a cytoprotective role against ROS in S. oneidensis, as reported in other bacteria (3), an H₂O₂ susceptibility assay was done (Fig. 1B). Surprisingly, the wild type and all of the mutant strains with an impaired ability to generate H₂S exhibited similar levels of susceptibility to H_2O_2 . In contrast, the $\Delta cysK$ mutant, which overproduces H₂S, displayed substantially increased sensitivity. The lower sensitivity was restored by cysK expression in trans, indicating that high levels of intracellular H₂S sensitize S. oneidensis to H₂O₂. To further confirm that the loss of H₂S generation does not affect resistance to H₂O₂, we constructed a strain lacking all three of the genes contributing to cysteine degradation (Δ triple mutant). This mutant, which produced very little H₂S under aerobic conditions (Fig. 1A), had a level of H₂O₂ resistance comparable to that of the wild type (Fig. 1B). Moreover, additional removal of the *psrA* and *sirA* genes ($\Delta penta$ mutant), which completely eliminated the ability to produce H₂S (Fig. 1A), did not compromise H_2O_2 resistance (Fig. 1B).

We also determined the abilities of the $\Delta cysK$, $\Delta triple$, and $\Delta penta$ mutant strains to degrade H₂O₂ and their viability after exposure to H₂O₂. Cells in the early stationary phase (4 h after the end of exponential growth), when endogenous H₂S levels were relatively stable (see Fig. S1 in the supplemental material), were collected for the two assays. Loss of endogenous H₂S production did not significantly compromise the ability to scavenge H_2O_2 (Fig. 2C). However, the $\Delta cysK$ mutant strain, consistent with its hypersensitivity to H₂O₂, degraded H₂O₂ at a significantly lower rate. With respect to survival, the $\Delta triple$ and $\Delta penta$ mutant strains resembled the wild type whereas the $\Delta cysK$ mutant strain was substantially impaired (Fig. 1D). Overall, these data demonstrate that endogenously generated H₂S in S. oneidensis provides little protection from exogenous H2O2 damage; rather, H2S at high concentrations negatively regulates the ability to cope with the stress.



FIG 1 Endogenous H_2S does not protect bacteria against H_2O_2 . (A) H_2S production in various *S. oneidensis* strains. Lead acetate-soaked paper strips show a brown or black PbS stain as a result of reaction with H_2S under aerobic conditions. The relative H_2S levels shown were obtained by normalization to the average level of the wild type (WT), which was set to 100%. The $\Delta triple$ ($\Delta mdeA \Delta SO1095 \Delta sseA$) mutant strain lacks the ability to produce H_2S via cysteine metabolism, while the $\Delta penta$ mutant strain carries additional deletions in the *psrA* and *sirA* genes, thus removing the ability to produce H_2S through respiration. (B) H_2O_2 sensitivity assay. Paper disks of 6 mm loaded with 10 µl of 5 M H_2O_2 were placed on a bacterial lawn and photographed after 16 h at 30°C. The relative H_2O_2 susceptibilities shown were obtained by normalization to the average level of the wild type, which was set to 100%. The $\Delta cysK^C$ mutant is a complemented mutant carrying a copy of the *cysK* gene integrated into the chromosome and produces H_2S like the wild type. (C) H_2O_2 consumption assay. H_2O_2 at 0.5 mM was added to mid-log-phase cultures (OD_{600} of ~0.2, the same afterward), and the H_2O_2 remaining at the time points indicated was measured. (D) Survival assay. One millimolar H_2O_2 was added to mid-log-phase cultures. After 5 and 30 min, samples were diluted and plated on LB. Colony counting was done after 24 h. For panels A and B, experiments were performed five times and representative results are shown. In panels B and D, data are reported as the mean \pm SD (n = 4).

The impact of H₂S on H₂O₂-treated cells is temporally dependent. The unexpected findings obtained with H₂S-deficient and -overproducing mutants raise questions about the roles that H₂S plays in bacteria, since it is apparently not simply acting as a protective molecule. Given that the $\Delta triple$ and $\Delta penta$ mutant strains responded the same way, the $\Delta penta$ mutant strain was used as the H₂S-deficient strain. Moreover, comparable results were obtained in all subsequent experiments with the wild-type and $\Delta penta$ mutant strains, and H₂S concentrations in mutant cultures can be accurately controlled with NaHS as the source of exogenous H₂S. For clarity, we present data from the H₂S-deficient strain rather than the wild type unless otherwise noted. The impact of H₂O₂ on the growth and viability of *S. oneidensis* has been extensively studied recently, showing that it is able to inhibit

growth even at concentrations as low as 100 μ M, although the MIC is ~1.25 mM (19). At physiological pH, most H₂S exists as HS⁻ (bisulfide ion) and there are only small amounts of H₂S and S²⁻. It should be noted that we mention only H₂S for simplicity, but all of the ionic forms are included. Unlike H₂O₂, H₂S did not noticeably inhibit the growth of *S. oneidensis* at concentrations of up to 1 mM (Fig. 2A).

We then assessed the impact of H_2S on growth inhibition by H_2O_2 . H_2S and H_2O_2 were added to mid-log-phase cultures (OD_{600} of ~ 0.2) at the same and different times, and the growth of the cultures was monitored. Strikingly, in the presence of 0.1 mM H_2S , the timing of H_2O_2 addition was found to be important for eliciting a detectable effect on growth (Fig. 2B). When both compounds were added simultaneously (T_0), H_2S significantly in-



FIG 2 Effects of H₂S, H₂O₂, and both together on the growth of the *S. oneidensis* $\Delta penta$ mutant. Mid-log-phase cultures were inoculated into LB medium containing the chemicals indicated and incubated statically at 30°C in 24-well plates. (A) Effect of H₂S (from NaHS) on growth. (B) Effects of 0.1 mM H₂S, 1 mM H₂O₂, and both on the growth of the $\Delta penta$ mutant strain. T_0 , T_{10} , T_{20} , and T_{30} represent the addition of H₂O₂ at the same time as H₂S and 10, 20, and 30 min later, respectively. In both panels A and B, -- represents the LB control. Data are reported as the mean \pm SD (n = 4).



FIG 3 H_2S promotes H_2O_2 killing when added promptly after H_2O_2 . (A) Synergistic inhibition of *S. oneidensis* growth by H_2S and H_2O_2 when added at the same time. Cells at an OD₆₀₀ of ~0.01 were inoculated into LB in 24-well plates containing H_2S , H_2O_2 , or both at the concentrations indicated and incubated statically at 30°C. The plates were photographed 24 h after inoculation. Experiments were performed five times, and similar results were obtained. (B) Mid-log-phase cells (OD₆₀₀ of ~0.2) were treated for 20 min with H_2S , H_2O_2 , or both at the concentrations indicated. The treated cultures were diluted, plated on LB, and incubated at 30°C. Survival was calculated as the ratio of the number of colonies in the treated cultures to that in the untreated control. Only plates containing 100 to 300 colonies were counted. (C) H_2S - H_2O_2 treatment generates extensive DNA damage, as determined by qPCR. Mid-log-phase cells were treated with 0.25 mM H_2S , 1.0 mM H_2O_2 , or both. Total genomic DNA was extracted, and qPCR was performed with equivalent amounts of template DNA. The relative fluorescence was determined by thyA mutation analysis. Mid-log-phase cells grown in LB were diluted 100-fold with fresh LB (--) or with LB containing 0.1 mM H_2S , 0.25 mM H_2O_2 , or both. At each time point, CAT was added to degrade H_2O_2 , and both total viability and the frequency of thyA mutants were determined. Data are reported as the mean \pm SD (n = 4).

creased the growth-inhibiting effects of 1 mM H₂O₂. In contrast, when H₂O₂ was added 30 min after H₂S (T_{30}), its effect on growth was protective. The balancing point was about 20 min after the addition of H₂S, at which time growth was comparable to that of cultures containing only H₂O₂. Before 20 min, the general trend was that the later the addition of H₂O₂, the lower the growth inhibition. These data indicate two distinct time-dependent effects of H₂S on the action of H₂O₂ in *S. oneidensis*.

H₂S promotes H₂O₂ killing when added promptly after the addition of H₂O₂. To assess whether H₂S aggravates growth inhibition by H2O2 when added promptly, we measured growth in the presence of various concentrations of H₂S or H₂O₂ (Fig. 3A). Similar levels of inhibition could be observed at different combinations; the higher the level of H₂S, the lower the level of H₂O₂. For instance, in the presence of 0.025 mM H₂S and 1 mM H₂O₂, growth was completely prevented and a similar level of inhibition was achieved with 0.25 mM H₂S and 0.25 mM H₂O₂. To determine whether growth inhibition by the simultaneous addition of H₂S and H₂O₂ was due to the bactericidal effect of H₂O₂, we examined the survival of treated cells. Plating assays showed that cells were readily killed by H_2O_2 in the presence of H_2S (Fig. 3B). After treatment for 20 min, a combination of 0.5 mM H₂S and 0.25 mM H_2O_2 reduced the proportion of viable cells to only \sim 4%, whereas killing by either 0.25 mM H₂O₂ alone or by 1 mM H₂S alone was insignificant. These results show that H₂S per se is not bactericidal but that it facilitates killing by H_2O_2 .

In general, DNA damage is the most lethal impact of oxidative

stress on viability (35). To test this, chromosomal DNA damage was measured by qPCR, which detects any lesions or strand breaks that block the progression of the PCR DNA polymerase and result in fewer PCR products than with intact DNA (36). A fragment of \sim 10 kb near the *ccm* gene (encodes the cytochrome *c* maturation system) was used for analysis because this PCR product can be reliably obtained (37). We found that the yield of full-length PCR fragments from the treated samples decreased by approximately 3-fold in comparison to that from the untreated control (Fig. 3C). We also measured H₂O₂-induced mutagenesis in cells challenged with a combination of 0.1 mM H₂S and 0.25 mM H₂O₂, a condition where a majority of the cells are not killed but mutations could accrue. The number of thyA mutants increased 7-fold in H₂S-H₂O₂-treated cells, whereas neither molecule alone was mutagenic compared to the control (Fig. 3D). On the basis of these results, we conclude that H₂S-H₂O₂-treated cells suffer gross DNA damage.

Reduced CAT activity underlies H_2S -induced enhancement of H_2O_2 killing. Since H_2O_2 does not oxidize DNA directly, the observed DNA damage most likely results from the product of the Fenton reaction (35), in which H_2O_2 reacts with unincorporated iron to produce hydroxyl radicals, extremely strong oxidizing agents that can react with virtually all organic molecules (35). As H_2S can bind ferric or ferrous iron to form insoluble FeS or Fe₂S₃, it is conceivable that the levels of unincorporated iron would not increase. It is possible, therefore, that H_2S treatment blocks the degradation of H_2O_2 and/or other ROS agents, as indicated by the



FIG 4 Effects of H_2S , H_2O_2 , and both on selected heme-containing enzymes. (A) Mid-log-phase cells were harvested, washed, and sonicated for the preparation of crude enzyme extracts. H_2S at 0.25 mM, H_2O_2 at 0.5 mM, or the two together were added to crude enzyme extracts at the times indicated, and SOD, POD, and CAT activities were measured. Subscript numbers are the times (in minutes) when H_2O_2 was added after the addition of H_2S . The activities of the treated samples were normalized to the activity of the untreated control and are reported as relative activities (RA). Importantly, the activities of all three enzymes from untreated samples were found to be stable for 30 min. (B) Crude enzyme extracts prepared as described for panel A were treated with the reagents indicated and assayed for H_2O_2 degradation. Concentrations: H_2S , 0.25 mM; NaN₃, 0.1 mM; dipyridyl (Dip), 0.5 mM. WT, wild type. (C) Two thousand five hundred units of bovine liver CAT was treated with the chemicals indicated at various concentrations for 5 min and then assayed for the ability to degrade 1 mM H_2O_2 at 37°C compared to the control (--). After 5 min, the remaining H_2O_2 concentrations were measured. (D) Cco assay. Crude enzyme extracts prepared as described for panel A were streated as a negative control, and the $\Delta katB$ mutant strain was also included for reference. All chemicals were added at the same time. Concentrations: H_2S , 0.25 mM; H_2O_2 , 0.5 mM. In all panels, data are reported as the mean \pm SD (n = 4).

compromised H_2O_2 -scavenging capacity of the *cysK* mutant (Fig. 1B). *S. oneidensis* has a full set of ROS-scavenging enzymes, including CAT, POD, and SOD (19, 20). To determine whether H_2S affects the activities of these enzymes, exponentially growing cells (OD_{600} of ~0.2) were harvested to prepare crude enzyme extracts as described in Materials and Methods. The extracts were treated with H_2S and/or H_2O_2 and assayed 5 min later (Fig. 4A). Consistent with previous observations (19), the SOD activities were relatively constant, implying a negligible role for SOD in the enhanced killing seen. In contrast, the activities of POD and CAT were inhibited by H_2S , H_2O_2 , or both, but there were substantial differences. H_2S alone had an insignificant effect on POD activity, whereas H_2O_2 in the presence or absence of H_2S caused a reduction to about 60% of the wild-type level. These data indicate that neither SOD nor POD was the target of H_2S .

In the case of CAT, the inhibitory effect of H_2O_2 was rather minor but the addition of H_2S resulted in a drastic reduction of activity (Fig. 4A). Given that the CAT KatB is the predominant enzyme for H_2O_2 degradation in *S. oneidensis* (19), we hypothesized that the observed enhanced killing by the combination of H_2S and H_2O_2 may be due to the reduced activity of KatB, thus preventing removal of H_2O_2 . This hypothesis was supported by the finding that a *katB* mutant was unable to respond to H_2S addition with respect to its ability to degrade H_2O_2 (Fig. 4B). To confirm this, we repeated the experiment with the heme-binding agent sodium azide (NaN₃), which specifically inhibits heme-containing CAT (38). As shown in Fig. 4B, the presence of NaN₃ resulted in extensive reduction of H₂O₂ degradation, supporting evidence that KatB is directly inhibited by H₂S. H₂S is an ironsequestering agent; thus, it is possible that this property plays a role in the inactivation of KatB. To test this, 0.5 mM dipyridyl, a specific iron-chelating agent, was added to the extracts but H₂O₂ degradation was hardly affected, suggesting that it is unlikely that H₂S inactivates KatB through its interaction with unincorporated iron species. As a further confirmation, we examined the effects of H₂S, NaN₃, and dipyridyl on commercially available heme CAT from bovine liver. Not surprisingly, we found that the activity was drastically inhibited by either H₂S or NaN₃ but not by dipyridyl (Fig. 4C). NaN₃ appears to be a stronger inhibitor than H_2S . Moreover, the addition of dipyridyl and H2O2 did not elicit enhanced inhibition of growth (see Fig. S2 in the supplemental material), supporting the in vivo evidence that H₂S negatively regulates the activity of heme-containing CAT but this activity is independent of its iron-sequestering ability.

To test whether H_2S inhibits other heme-containing proteins, we examined its effect on the *cbb*₃ oxidase. This oxidase is composed of at least three subunits, CcoN, CcoP, and CcoO, and is the primary system for respiration of oxygen and the only enzyme that has Cco activity (39, 40). As shown in Fig. 4D, the activity of the cbb_3 oxidase was inhibited by 0.25 mM H₂S to approximately 55% of the untreated level after a 5-min treatment. Although H₂O₂ was also able to inhibit the cbb_3 oxidase, the effect was rather modest, with approximately 65% of the activity remaining in the presence of H₂O₂ at the growth-inhibitory concentration of 1.25 mM (19). Besides, unlike the activity of the enzyme in the H₂S-treated samples, that in H₂O₂-treated samples was restored to the untreated level 20 min after the addition, most likely because of the removal of H₂O₂. When the experiment was repeated with the *katB* mutant strain, the activity of the cbb_3 oxidase was no longer recoverable. Furthermore, the combination of H₂O₂ (1.25 mM) and H₂S (0.25 mM) failed to further reduce this activity. These data show that H₂S is effective in negatively regulating the activity of the two heme-containing proteins KatB and cbb_3 oxidase.

H₂S induces the OxyR-mediated stress response in S. oneidensis. In contrast to facilitating killing by H₂O₂, H₂S improves the growth of S. oneidensis under H₂O₂ stress conditions if pretreatment for 20 min or longer is applied (Fig. 2B), indicating a protective role, as observed in four other bacterial species (3). It is proposed that H₂S, as for NO, confers resistance to H₂O₂-induced oxidative stress by (i) suppressing the damaging Fenton reaction and (ii) improving the antioxidant capacity of bacterial cells involving the major ROS scavengers CAT, POD, and SOD (3, 41). These two mechanisms are in accord with the OxyR-mediated stress response, suggesting that the cytoprotective role of these molecules relies on the OxyR system. To test whether this is true for S. oneidensis, we examined the amounts of ROS-scavenging enzymes after a 30-min treatment with H₂S by in-gel staining analysis of proteins separated by native PAGE. As previously reported (19), KatB, whose expression is directly repressed by OxyR, was the only CAT visible in the staining analysis (Fig. 5A). As expected, production of KatB was substantially increased in samples treated with H_2O_2 or in an *oxyR* mutant. H_2S_2 , although less effective, was also able to induce KatB production. Noticeable induction was observed with H₂S at 0.25 mM, and at higher concentrations (0.5 to 2 mM), induction was more robust.

The observation that H₂S alone induces enhanced production of KatB, similar to that from treatment with H₂O₂, supports the idea that the OxyR system mediates the cytoprotection of H₂S. To confirm this possibility, we examined the expression of three other genes (*ahpC*, *dps*, and *katG1*) of the OxyR regulon, as well as the katB gene, in samples treated with H₂S by using an integrative lacZ reporter system. While the *ahpC* and *katG1* genes (encode alkylhydroperoxide reductase and CAT KatG1, respectively) are positively regulated by OxyR, the dps gene, which encodes an iron storage protein, is repressed by OxyR, as is the katB gene (19). As shown in Fig. 5B, in the presence of H_2S or H_2O_2 , the β -galactosidase activities driven by promoters of the test genes were upregulated in the $oxyR^+$ background but no longer responsive to either compound in the absence of OxyR. A similar trend was observed by qRT-PCR (see Fig. S3 in the supplemental material). In summary, these data indicate that H₂S, when added after 10 min, enhanced resistance to H₂O₂ by activating OxyR.

Iron sequestration likely triggers activation of OxyR. Because iron is both essential and potentially harmful (42), bacteria have evolved various mechanisms to ensure adequate supplies and at the same time tight restriction of free-iron levels to protect against iron-induced toxicity. As H_2S is an iron-sequestering agent, we first tested whether this feature is associated with the



FIG 5 H₂S induces the OxyR-mediated stress response. (A) CAT staining analysis. Cells were harvested just prior to (-) and 30 min after the addition of 0.25 mM H₂S or 0.1 mM H₂O₂ (top). Protein samples of \sim 10 µg from the cell lysates indicated were separated by native PAGE and stained for CAT activity. The Δ*katB* and Δ*oxyR* mutant strains (constitutive high-level expression) were used as negative and positive controls. In the analysis shown at the bottom, various concentrations of H₂S were examined for the ability to induce expression of the *katB* gene. (B) Impact of H₂S on the expression of four members of the OxyR regulon. β-Galactosidase assays were carried out with *lacZ* reporter vectors. Cells grown to mid-log phase were treated with the chemicals indicated for 30 min and then harvested for the assays. Concentrations: H₂S, 0.25 mM; H₂O₂, 0.2 mM; dipyridyl (Dip), 0.25 mM. β-Galactosidase activities are reported as the mean ± SD (n = 4). Similar results were obtained by qRT-PCR assay (see Fig. S3 in the supplemental material).

activation of OxyR. The intracellular levels of total and unincorporated iron after the addition of H₂S were measured (Fig. 6A). Surprisingly, the addition of H₂S had a minor effect on the concentrations of unincorporated iron but substantially increased the amounts of total iron. This result implies that H₂S may cause an iron influx, presumably triggered by the decreased intracellular levels of free iron immediately after its addition. If this were the case, then the addition of dipyridyl should produce a similar result. Indeed, total intracellular iron levels were elevated in the presence of 0.5 mM dipyridyl, although unincorporated iron was diminished to some extent, implying that dipyridyl may be able to activate OxyR, although it differs from H₂S in its inability to promote H_2O_2 killing when added simultaneously with H_2O_2 , as shown in Fig. 4. To test this, we measured KatB activity by in-gel staining and found that it was increased after the addition of dipyridyl (Fig. 6B). Furthermore, the expression pattern of four OxyR-regulated genes in cells treated with dipyridyl resembled that found after H₂S treatment (Fig. 6B). These data, collectively,



FIG 6 H₂S induces the OxyR-mediated stress response. (A) Intracellular iron levels induced by H₂S. One-liter cultures grown to mid-log phase (OD₆₀₀ of \sim 0.2) were harvested just before (0 min) and 5, 10, and 30 min after the addition of 0.25 mM H₂S or 0.2 mM dipyridyl, and the unincorporated (U) and total (T) intracellular iron concentrations were measured. The experiments were performed at least three times. Error bars show standard deviations. (B) CAT staining analysis. Cells were harvested 30 min after the addition of 0.25 mM H₂S or 0.2 mM dipyridyl along with the untreated control (--). Ten-microgram samples of protein from the cell lysates indicated were separated by native PAGE and stained for CAT activity. The $\Delta katB$ and $\Delta oxyR$ mutant strains (constitutive high-level expression) were used as negative and positive controls. In the right panel, two higher concentrations of dipyridyl were tive of three independent experiments.

suggest that iron sequestration is the likely mechanism by which H_2S activates the OxyR-mediated cellular response to oxidative stress.

DISCUSSION

Previous studies have demonstrated that H_2S may be either detrimental or beneficial to bacterial cells (3, 7, 8). Growth inhibition is attributed to the abilities of H_2S to damage DNA and denature proteins. In contrast, a beneficial role played by H_2S is its ability to combat oxidative stress. However, there are reports that, in some organisms, H_2S stimulates rather than inhibits ROS production (4, 5, 43). The basis of these differences has not been resolved. One possible explanation is that species inhabiting various niches may evolve different mechanisms for responding to H_2S , given the unparalleled diversity of all physiological aspects of living organisms. But the presence of a general strategy is equally possible. Here, we present evidence that H_2S is indeed able to perform these contrasting functions, depending upon the time of addition.

In this study, endogenous H_2S was found to be dispensable for protecting *S. oneidensis* cells from oxidative stress, in contrast to findings recently reported (3). Conceivably, the production of H_2S from metabolic processes is continuous and steady but unable to reach levels sufficiently high to trigger a cellular response. This possibility is supported by the finding that mutation of *cysK*,

resulting in substantially enhanced production of H₂S, sensitizes cells to H₂O₂. Using exogenous H₂S, we demonstrated that it promotes H₂O₂ killing by specifically damaging heme-containing proteins, especially the CAT KatB, even though it is able to induce the protective OxyR regulon, as described in other bacterial species (3). Such an inhibitory effect on heme-containing proteins such as Cco (mitochondrial aa3 type), myoglobin, and hemoglobin by H_2S is well documented in eukaryotes (44–46). In general, the reaction between these proteins and H₂S induced modification of the heme component, reversibly inhibiting activity (46, 47). Sulfheme species, resulting from the binding of H_2S , are reduced because of this interaction (48). Intriguingly, although the inhibitory effect of H₂S on mitochondrial Cco was first reported more than 80 years ago, the cellular targets of H₂S are still largely restricted to proteins implicated in oxygen transport and consumption (46). Results presented here clearly show that CATs, at least those from S. oneidensis and bovine liver, and cytochrome *cbb*₃ oxidase are inactivated by H₂S, expanding the repertoire of H₂S targets. It is very likely that there are more H₂S targets, based on the very large pool of heme-containing proteins. Nevertheless, with respect to H₂O₂ killing, CAT is the most crucial target of H₂S because its loss largely disables the cell's ability to degrade H_2O_2 . Furthermore, it should be noted that the effect of H₂S on different heme-containing proteins may vary considerably, given that amino acid residues near the heme play an essential role in sulfheme formation (46).

In contrast to the inhibitory effect, activation of OxyR by H₂S appears to rely on its iron-sequestering capability, as a similar result was obtained with the iron-chelating agent dipyridyl. Ironand heme-containing proteins, mononuclear iron proteins in particular, are primary targets of H_2O_2 (49, 50). Some bacteria, such as E. coli, are able to replace the iron atom of these enzymes with manganese to maintain activity in the presence of H_2O_2 (51, 52). However, in S. oneidensis, this protective mechanism is likely to be heavily compromised because of the extremely low Mn/Fe ratio (19, 53). More importantly, S. oneidensis lacks an H₂O₂-inducible manganese transporter, preventing the accelerated import of Mn for protection during oxidative stress. As a result, free iron is abundant for the Fenton reaction when challenged by exogenous H₂O₂. To effectively prevent the formation of lethal hydroxyl radicals, S. oneidensis activates OxyR, resulting in the rapid removal of H2O2 and reduction of intracellular unincorporated iron concentrations. For the former, the strategy is to produce multiple scavenging enzymes, with KatB dominating. For the latter, the situation is rather complicated, however, because there are a large number of iron- and heme-containing proteins in S. oneidensis (such as up to 42 c-type cytochromes, while E. coli has 5 to 7), thus requiring relatively high levels of iron for its normal physiological activities (54, 55).

To control intracellular free-iron levels under stress conditions, the iron-sequestering ferritin Dps is extensively induced upon the activation of OxyR (19). In addition, *S. oneidensis* limits its iron uptake rate to prevent a sudden increase, as evidenced by its accumulation of 20 times less iron than by *E. coli* in a 24-h period (53). It is therefore conceivable that although addition of H_2S , as well as dipyridyl, reduces levels of free iron to some extent and thus transiently impedes growth, the process overall triggers an iron influx, which likely serves as a signal for activation of OxyR. This hypothesis gains support from the finding that the total iron concentration is significantly elevated but the amounts of free iron are rather constant. Presumably, most of the imported free iron is bound to Dps under oxidative stress conditions. This protective effect of H_2S as an iron-sequestering agent seems to coincide with that of superoxide (56). While superoxide can lead to cellular damage and death, its accumulation induces the protective SoxRS and MarRAB regulons (57). Thus, pretreatment with low-to-moderate concentrations of superoxide provides cells a greater ability to cope with subsequent more severe stress and, more importantly, prevents cells from entering programmed cell death (56).

Shewanella bacteria thrive in redox-stratified environments where ROS are likely to form (9) and microbes with high resistance to ROS are presumably favored. However, S. oneidensis is very susceptible to ROS, in comparison to E. coli (19, 53, 58). We do not yet know the reason for this difference, but it may be associated with novel properties of this species. We have previously shown that S. oneidensis prefers insoluble to soluble electron acceptors for respiration, such as Fe (III) oxide versus oxygen, thus preventing endogenous generation of ROS from autoxidation of components of the respiratory chain (35, 59). In addition, unlike other organisms containing cytochrome caa3-type and cbb3-type oxidases, where the low-affinity (caa3-type) oxidase plays a dominant role under high-oxygen conditions and the cbb_3 oxidase is induced only at low O₂ concentrations, S. oneidensis utilizes the *cbb*₃-type oxidase under both high- and low-oxygen conditions and does not express the caa_3 oxidase (39, 40). As respiration of oxygen by the cbb₃ oxidase is relatively slow, less ROS would be generated. Moreover, H₂S that is produced endogenously may play a beneficial role. The elevated sensitivity to H₂O₂ is observed only in the H₂S-hyperproducing ($\Delta cysK$ mutant) strain, implying that H₂S generated in the wild type is not sufficient to elicit a detrimental response. Hence, H₂S may simply act as an agent to help maintain a relatively reduced environment for cells, either inside or outside in proximity.

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