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Transferrin-Mediated Iron Sequestration As a Novel Therapy for Bacterial and Fungal Infections

Kevin W. Bruhn¹ and Brad Spellberg²

¹Department of Microbiology & Immunology, Keck School of Medicine at the University of Southern California (USC), Los Angeles, CA

²Department of Medicine, Keck School of Medicine at USC, Los Angeles, CA

Abstract

Pathogenic microbes must acquire essential nutrients, including iron, from the host in order to proliferate and cause infections. Iron sequestration is an ancient host antimicrobial strategy. Thus, enhancing iron sequestration is a promising, novel anti-infective strategy. Unfortunately, small molecule iron chelators have proven difficult to develop as anti-infective treatments, in part due to unacceptable toxicities. Iron sequestration in mammals is predominantly mediated by the transferrin family of iron-binding proteins. In this review, we explore the possibility of administering supraphysiological levels of exogenous transferrin as an iron sequestering therapy for infections, which could overcome some of the problems associated with small molecule chelation. Recent studies suggest that transferrin delivery may represent a promising approach to augment both natural resistance and traditional antibiotic therapy.

Keywords

transferrin; antibiotic resistance; bacteria; fungi; combination therapy

Introduction

In recent years the critical role of iron in microbial growth and pathogenesis has garnered increasing attention. Virtually all microbes must obtain iron in order to survive and propagate [1–5], including disease-causing pathogens that establish infections in mammalian hosts. Many microbes have thus evolved specialized mechanisms to acquire this limited resource. Conversely, an evolutionarily conserved, common host strategy to control microbial growth is to strictly regulate levels of available free iron. The microbial requirement for iron suggests that strategies aimed at blocking iron acquisition by microbes

Corresponding author: Kevin W. Bruhn, PhD, 1441 Eastlake Avenue, NTT 6427, Los Angeles, CA 90033, (323) 865-3737; kbruhn@usc.edu.

Disclosures

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might form the basis for promising, novel anti-infective therapies. Unfortunately, previous approaches using small molecule iron chelators have not proven safe and effective for treating clinical infections. In this review, we examine the potential of transferrin, a mammalian iron-binding protein, to be developed as a novel therapeutic for bacterial and fungal disease.

Limitations of small molecule iron chelators as therapeutics

The simplest approach to blocking iron acquisition by pathogens is the use of small molecule chelating agents that sequester iron and prevent microbial uptake. Numerous investigators over several decades have characterized a variety of iron sequestering agents that inhibit the growth of microbes *in vitro* [6–10]. However, several critical barriers have limited development of small molecule chelators as therapies for infection. First is the production by bacteria of siderophores (e.g., *S. aureus* staphyloferrin A and B, and *A. baumannii* acinetobactin) that are secreted by the microbe to scavenge and acquire iron in environments where bioavailability is low [11,12]. Siderophores are extremely strong binders to ferric (Fe³⁺) iron, and often possess iron affinities that are 10¹⁰- to 10²⁰-fold higher than small molecule or biological iron chelators [13,14]. Similarly, fungal species such as *Candida* also produce high affinity iron siderophores, and both fungi and bacteria (e.g., *Acinetobacter*) can uptake high affinity xeno-siderophores that are produced by other bacteria (e.g., desferoxamine) [15–21]. The extremely high iron affinities of microbial-derived siderophores, which are far higher than affinities for small molecule chelators, has led to the perception that iron acquisition by high affinity siderophores cannot be overcome *in vivo* by chelation-based therapy.

An additional problem is that small molecule chelators alter metabolic disposition of iron in ways that may be injurious to the host. For example, chelators reduce iron availability to myeloid cells, which are normally the predominant recyclers of iron in the host, and increase its excretion into renal tubules where iron is not normally found. Thus, serious toxicity to bone marrow, kidneys, and other organ systems can occur during small molecule iron chelator therapy [22,23].

As a result of these factors, *in vivo* testing of iron chelation strategies has focused on eukaryotic pathogens (e.g., malaria and molds) [24,25]. Unfortunately, the most advanced effort to develop a small molecule chelator into clinical trials for infection failed, as a recent randomized, controlled trial of patients with mucormycosis found that small molecule iron chelation was not safe or effective [26]. Nevertheless, the profound requirement for iron acquisition of microbes continues to spur translational efforts to develop novel therapies.

Transferrin as an innate immune mediator

Given how fundamental iron acquisition is to microbial survival, it is not surprising that in mammals, the concentration of free iron in tissue fluids is less than 10⁻²⁴ M. This exceedingly low concentration is maintained predominantly by the iron-binding protein transferrin [27]. Transferrin is an abundant serum glycoprotein that mediates transport and homeostasis of iron levels in the plasma and extracellular tissue fluid. The protein contains two homologous lobes, each with a single high-affinity iron-binding site. Average

transferrin levels in the serum are between 1 and 4 mg/mL, and under normal physiological conditions, the protein remains approximately 30% iron-saturated [28]. Normal levels of unsaturated transferrin help to maintain the concentration of free iron in tissue fluids at levels that are restrictive for uncontrolled microbial growth. Many studies have identified transferrin as one of the major components necessary for the antimicrobial activity of serum [29,30]. Conversely, increased iron stores have been reported to correlate with increased frequency and severity of many bacterial and fungal infections, as well as sepsis [31–34].

***In vitro* Antimicrobial Effects of Transferrin**

These observations have led investigators to consider a biological-based strategy for iron sequestration using exogenous transferrin. Numerous studies have demonstrated the ability of transferrin to restrict microbial growth *in vitro* due to its iron sequestration capacity [35–37]. Pathogenic organisms whose growth is inhibited by transferrin include both Gram-negative and Gram-positive bacterial pathogens such as *Pseudomonas aeruginosa* [38], *Klebsiella pneumoniae* [39], *Yersinia pseudotuberculosis* [40], *Acinetobacter baumannii* [8], and *Bacillus anthracis* [41], as well as fungal pathogens, such as *Candida* species and *Histoplasma capsulatum* [42,43].

Our group has also assessed the *in vitro* efficacy of transferrin against diverse bacterial and fungal pathogens. We conducted time-kill curves and determined minimum inhibitory concentrations (MICs) of recombinant human transferrin (rhTransferrin) against *S. aureus* (Gram-positive bacterium), *A. baumannii* (Gram-negative bacterium) and *C. albicans* (fungus). Transferrin had an MIC of 6 µg/ml for the virulent strains *S. aureus* LAC and *A. baumannii* HUMC1, and a 60 µg/ml MIC for virulent *C. albicans* SC5314, demonstrating concentration-dependent static activity against all three pathogens [44]. At the 60 µg/ml concentration (10-fold above the MIC), transferrin mediated a >3 log reduction in *S. aureus* CFUs at 24 hours compared to growth control [44]. For *A. baumannii*, both the 6 and 60 µg/ml concentration mediated 10–100-fold reductions in CFUs/ml at 24 hours compared to growth control. For *C. albicans*, the 60 µg/ml dose mediated minor growth inhibition at 6 hours, and 3-fold reductions in CFUs at 24 hours. Higher concentrations (120 and 360 µg/ml) mediated substantial inhibition of growth at all time points [44].

Because transferrin targets host iron, rather than a biochemical target on microbes, we hypothesized that it would exert minimal selective pressure driving resistance. We found that serial passage of each organism in the presence of a sub-inhibitory concentration of rhTransferrin for 20 generations led to no change in susceptibility. Antimicrobial activity was inhibited by the addition of exogenous iron or iron-loaded siderophores, as well as anti-transferrin antibodies. In addition, intracellular iron levels in all three pathogens were markedly reduced following exposure to rhTransferrin in a dose-dependent manner [44]. Thus, transferrin acts as a static, not cidal, agent against a broad spectrum of human pathogens.

Iron is a critical electron acceptor in the oxidative phosphorylation cascade that leads to ATP generation in both prokaryotes and eukaryotes [45]. Treatment of *A. baumannii*, *S. aureus*, and *C. albicans* with rhTransferrin resulted in disrupted membrane potentials in all

three pathogens in a dose-dependent manner, as early as 1 hour following treatment, with increased effect at 6 hours [44]. Although both *C. albicans* and *S. aureus* experienced some degree of membrane potential recovery at 24 hours, this may have been due to the liberation of intracellular iron stores from dying organisms, which became available to saturate transferrin. Disrupted membrane potentials were maintained when the transferrin was separated by a filter from the microbes, and were totally reversed by the addition of exogenous iron. Thus, the effect of transferrin on microbial membrane potentials appeared to be due to iron sequestration.

Challenges to Standardizing *In vitro* Testing of Transferrin

When testing transferrin MICs to a variety of organisms, we noted substantial variability between assays that used serum from different batches, due to variability in the concentration of iron and iron binding proteins in the serum [44]. Thus, reproducible transferrin MIC testing requires conducting the assay in media without serum. Rich growth media that are normally used for susceptibility testing also posed challenges due to the high levels of free iron. Such media do not recapitulate the normal, exceedingly low free iron levels in human blood and tissues. When tested in RPMI in the absence of serum, human apo-transferrin MICs were highly reproducible. However, the apo-transferrin MICs were substantially lower than the physiological concentration of transferrin (a mixture of apo- and holo-transferrin) in human blood [28]. The amount of apo-transferrin that is required to be added into biological matrices to inhibit microbial growth will be difficult to predict given the complex dynamics of free vs. bound iron in such matrices. Thus, while *in vitro* reproducibility of MIC testing is likely to require assays run in the absence of serum, clinical investigation is going to be required to define how breakpoints set by such assays predict *in vivo* efficacy.

In vivo Validation of Transferrin Efficacy Against Infection

In contrast to the numerous reports on the *in vitro* inhibitory effects of transferrin on microbial growth, there have been fewer investigations into the potential for exogenous transferrin to effectively treat infections *in vivo*. One early study demonstrated decreased mortality rates in mice with experimental candidiasis that were preadministered transferrin [46]. More recently, our group infected mice intravenously with various pathogens to test the ability of rhTransferrin to confer survival against distinct, lethal bloodstream infections. Separate groups of mice were administered *S. aureus*, *A. baumannii*, and *C. albicans*, and treated with human transferrin or placebo. Four doses of 90 and 270 mg/kg/d of rhTransferrin substantially improved survival against all three infections compared to placebo-treated mice [44]. Transferrin treatment significantly reduced tissue bacterial/fungal burden for all three pathogens, as well. Protection mediated by rhTransferrin was reversed by administration of exogenous free iron, or by treatment with transferrin saturated with iron before injection into the mice.

We also sought to determine the potential for transferrin to synergize with antimicrobial therapy, as well as help prevent the emergence of resistance to antimicrobial therapy. We selected rifampin as a representative antibiotic to test against *S. aureus*, because emergence

of bacterial resistance occurs rapidly during rifampin monotherapy due to a single step mutation that results in high level resistance [47]. *In vitro*, a sub-MIC concentration of rhTransferrin (3 µg/ml) synergized with antibiotic treatment, decreasing the *S. aureus* MIC of rifampin from 0.15 µg/ml to 0.019 µg/ml (8-fold decrease) [44]. *In vivo*, when rhTransferrin was combined with rifampin treatment to treat *S. aureus* bacteremia, the emergence of rifampin-resistance escape mutants was markedly reduced [44]. Prevention of resistance could be due to synergy with antibiotics or geometric mass action (antibiotic escape mutant rate * transferrin escape mutant rate = much lower frequency of viable, resistant escape mutants).

Evidence of safety profile from clinical trials

One potential advantage of administering a naturally occurring biologic molecule, as opposed to small molecule chelators, is that supraphysiological levels of transferrin would presumably not alter the normal pathways of iron sequestration and trafficking, but would rather enhance these pathways. Transferrin has already been extensively studied in clinical trials for patients with iron overload, providing important insights into its safety profile [36]. In a series of studies over 10 years ago, stem cell transplant patients undergoing myeloablative chemotherapy were administered intravenous transferrin, in order to counter their high levels of non-transferrin-bound iron (NTBI). A single dose of apo-transferrin (100 mg/kg) was well-tolerated by six stem cell transplant patients following myeloablation, and correlated with a reduction in transferrin saturation, as well as a reduction of NTBI to undetectable levels [48]. Serum samples taken directly after apo-transferrin administration inhibited the growth of *S. epidermidis* [49]. Later studies showed that repeated administration of divided daily doses of 1040 mg/kg total of apo-transferrin were well-tolerated by stem cell transplant patients, with no observable toxicities, and marked decreases in transferrin saturation and unbound iron [50]. An ongoing phase II/III dose escalation study is reported on clinicaltrials.gov [NCT01797055]. Thus, although transferrin is not yet FDA-approved, no substantial toxicity signals have yet been revealed in these trials [36]. Therefore, adjunctive transferrin therapy to treat infections or to reduce resistance emergence may therefore be rapidly translatable.

Conclusions and Future Needs

The impending crisis of antibiotic resistance has created a need for alternative therapies to treat bacterial infections, and strategies that treat the host rather than the bacteria are attractive because they are theoretically less likely to induce resistance. We and others have demonstrated that transferrin has broad, cross-kingdom efficacy both *in vitro* and *in vivo*, and likely represents a superior approach to sequestering iron by using small molecule chelating agents. Multiple clinical trials treating patients with iron overload have already demonstrated acceptable safety profiles, which would thus enable rapid clinical translation of transferrin as a viable antimicrobial therapy. Delivery of supraphysiological levels of recombinant transferrin may also inhibit or slow the development of intrinsic antibiotic resistance. For these reasons, transferrin represents a potentially promising clinical alternative or adjunct to traditional antibiotic treatment, and is worthy of continued study.

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Highlights

- Mammalian transferrin maintains iron homeostasis in serum and extracellular tissue fluid.
- Sequestration of iron is an innate host strategy to prevent pathogenic microbial growth.
- Addition of transferrin limits growth of numerous diverse pathogens in the bloodstream.
- Delivery of exogenous transferrin may offer an attractive antimicrobial approach.