



# Structure of the Varicella Zoster Virus Thymidylate Synthase Establishes Functional and Structural Similarities as the Human Enzyme and Potentiates Itself as a Target of Brivudine

Kelly Hew<sup>1</sup>, Sue-Li Dahlroth<sup>1</sup>, Saranya Veerappan<sup>1</sup>, Lucy Xin Pan<sup>1</sup>, Tobias Cornvik<sup>1</sup>, Pär Nordlund<sup>1,2,3</sup>\*

- 1 Division of Structural Biology and Biochemistry, Nanyang Technological University, School of Biological Sciences, Singapore, Singapore, 2 Division of Biophysics, Department of Medical Biochemistry and Biophysics, Karolinska Institutet, Stockholm, Sweden, 3 Institute of Molecular and Cell Biology, Agency for Science, Technology and Research, Singapore, Singapore
- \* pnordlund@ntu.edu.sg



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# **Abstract**

Varicella zoster virus (VZV) is a highly infectious human herpesvirus that is the causative agent for chicken pox and shingles. VZV encodes a functional thymidylate synthase (TS), which is the sole enzyme that produces dTMP from dUMP de novo. To study substrate binding, the complex structure of TS<sub>VZV</sub> with dUMP was determined to a resolution of 2.9 Å. In the absence of a folate co-substrate, dUMP binds in the conserved TS active site and is coordinated similarly as in the human encoded TS (TSHS) in an open conformation. The interactions between TS<sub>VZV</sub> with dUMP and a cofactor analog, raltitrexed, were also studied using differential scanning fluorimetry (DSF), suggesting that TS<sub>VZV</sub> binds dUMP and raltitrexed in a sequential binding mode like other TS. The DSF also revealed interactions between TS<sub>VZV</sub> and in vitro phosphorylated brivudine (BVDU<sub>P</sub>), a highly potent anti-herpesvirus drug against VZV infections. The binding of BVDUP to TSVZV was further confirmed by the complex structure of  $TS_{VZV}$  and  $BVDU_P$  solved at a resolution of 2.9 Å.  $BVDU_P$  binds similarly as dUMP in the TS<sub>HS</sub> but it induces a closed conformation of the active site. The structure supports that the 5-bromovinyl substituent on BVDU<sub>P</sub> is likely to inhibit TS<sub>VZV</sub> by preventing the transfer of a methylene group from its cofactor and the subsequent formation of dTMP. The interactions between TS<sub>VZV</sub> and BVDU<sub>P</sub> are consistent with that TS<sub>VZV</sub> is indeed a target of brivudine in vivo. The work also provided the structural basis for rational design of more specific TS<sub>VZV</sub> inhibitors.



## Introduction

Varicella zoster virus (VZV) is a highly contagious human herpesvirus that causes varicella, which is commonly known as chicken pox [1]. Like all other human herpesviruses, VZV infection is life-long and the virus establishes its latency in dorsal root ganglia [2, 3]. Reactivation of VZV from latency can cause herpes zoster or shingles, which occurs more frequently in older adults and immunocompromised individuals [1]. The herpes zoster lesions have been associated with acute and persisting pain that may last beyond a month after the initial outbreak [4].

Herpesviral DNA synthesis is pivotal for the production of new infectious virus particles in the host cells. This is governed by activities of DNA synthesis proteins, which have been the focus in the development of many anti-herpesvirus drugs [5, 6]. A large proportion of the clinically approved anti-herpesvirus drugs are nucleoside analogs that have inhibitory actions against the viral DNA polymerase upon their conversions to the nucleotide triphosphosphates [7, 8]. These include the acyclic nucleoside analogs aciclovir, valaciclovir, penciclovir, famiciclovir, ganciclovir and valgancyclovir, the deoxycytidine analog cidofovir, and the 5-substituted pyrimidine nucleoside analogs idoxuridine (IDU) and brivudine (BVDU) [7, 8]. The safety and selectivity of most of these nucleoside analogs are dependent on the first phosphorylation step by the viral thymidine kinase (TK) that is only encoded in a virally infected cell [9, 10].

Thymidylate synthase (TS) is a highly conserved protein found through all kingdoms of life including human, mouse, bacteria, protozoa and some viruses. It catalyzes the transfer of the methylene group from cofactor 5,10-methylenetetrahydrofolate (mTHF) to substrate deoxyuridine monophosphate (dUMP) [11]. In this process, deoxythymidine monophosphate (dTMP) is produced *de novo* and mTHF is converted to dihydrofolate (DHF). Subsequently, dTMP undergoes another two successive phosphorylation steps to produce deoxythymidine triphosphate (dTTP), which is added to the elongating DNA chain by the DNA polymerase [11].

Due to its essential role in the DNA replication, the human TS ( $TS_{HS}$ ) has been extensively studied and is a well-proven drug target for cancer therapy [12]. Inhibition of  $TS_{HS}$  has been shown to lead to a deficiency in dTTP and DNA damage due to insertion of deoxyuridine triphosphate (dUTP) into DNA. This results in a cell death, so called thymineless death [13]. Different  $TS_{HS}$  inhibitors have been derived from modifications of its natural substrate dUMP and cofactor mTHF. These include the nucleobase analog 5-fluorouridine and the folate analogs raltitrexed and pemetrexed, which are currently used in treatments against solid tumors [14–16]. Molecular features of  $TS_{HS}$  and other TS have been elucidated in complex structures with different substrates and/or inhibitors [17–30]. These structures have contributed greatly to rational structure based drug designs that improved the specificity and efficacy of  $TS_{HS}$  inhibitors [31].

For reasons not completely clear, only VZV and Kaposi's sarcoma associated herpesvirus (KSHV) encode a TS (TS<sub>VZV</sub> and TS<sub>KSHV</sub> respectively) [32, 33]. Both TS<sub>VZV</sub> and TS<sub>KSHV</sub> share more than 50% sequence identity with TS<sub>HS</sub> but unlike the well-studied human counterpart, the homologous herpesviral TSs remain poorly characterized and no structural information is available on these proteins. TS<sub>VZV</sub> has also been shown to be non-essential for the viral replication *in vitro* [34]. In an attempt to gain insights into TS<sub>VZV</sub> and to differentiate TS<sub>VZV</sub> from TS<sub>HS</sub>, biophysical analysis and structural analyzes of TS<sub>VZV</sub> with its substrate dUMP and a TS<sub>HS</sub> inhibitor raltitrexed were carried out. The structures of TS<sub>VZV</sub> and TS<sub>VZV</sub> in complex with dUMP were solved to a resolution of 3.1 Å and 2.9 Å respectively. This constitutes the first crystal structure of an eukaryotic virus TS. The biophysical assays and the complex structure of TS<sub>VZV</sub> with dUMP demonstrated similar modes of substrate binding for TS<sub>VZV</sub> and TS<sub>HS</sub>. Structural comparisons of TS<sub>VZV</sub> and TS<sub>HS</sub> also revealed similar substrate binding details but with slight amino acids variations in the TS active site. Biophysical and crystallographic



studies of  $TS_{VZV}$  with an *in vitro* phosphorylated BVDU supports binding of phosphorylated BVDU to  $TS_{VZV}$ . This study also provided further support for that  $TS_{VZV}$  is a potential target enzyme for BVDU after metabolic activation in infected cells.

## Results

# Structure of apo TS<sub>VZV</sub>

The structure of the apo  $TS_{VZV}$  (apo- $TS_{VZV}$ ) was determined and refined to a final resolution of 3.1 Å, with four molecules in each asymmetric unit (<u>Table 1</u> and <u>S1 Fig</u>). Apo- $TS_{VZV}$  was crystallized with a protein construct comprising of amino acids 8 to 295, which lacks the N-terminus seven amino acids and C-terminus six amino acids. This model has electron densities for amino acids 15–295, except for a missing loop containing amino acids 38–39 in one of the four molecules in the asymmetric unit. The four  $TS_{VZV}$  monomers in the asymmetric unit are arranged as two homo-dimers. The monomers are virtually the same and the  $C\alpha$  atoms superimpose with a root mean square deviation (rmsd) of 0.78 Å.

The structures of apo- $TS_{VZV}$  and the human encoded apo TS ( $TS_{HS}$ ) superimpose with a rmsd of 0.9 Å for 558 residues, with the highest similarity at the active sites [22]. A phosphate ion was found at one end of the active site in each of the  $TS_{VZV}$  monomers and is held in place by Arg 203, Ser 204, as well as Arg 163' and Arg 164' that are located on the loop from the dimer partner (Fig 1a).

## Complex structure of TS<sub>VZV</sub> with dUMP

 $TS_{VZV}$  was crystallized in a second crystal form that was used for subsequent determination of the structure of  $TS_{VZV}$  in complex with dUMP ( $TS_{VZV}$ +dUMP). The  $TS_{VZV}$ +dUMP structure was determined and refined to a resolution of 2.9 Å ( $Table\ 1$ ). Each asymmetric unit contains four  $TS_{VZV}$  subunits arranged into two homo-dimers. Each subunit is made up of amino acids 15 to 295. This is with exception to amino acids 98 to 117 in one monomer of the  $TS_{VZV}$  dimer, and amino acids 38 to 39 and 136 to 139 in the other monomer of the  $TS_{VZV}$  dimer. Electron densities in these regions are not continuous and could therefore not be modeled.

A molecule of dUMP binds in the  $TS_{VZV}$  active site, where the phosphate group is coordinated by Arg 38, Arg 203, and Arg 163' of the dimer partner (Fig 1b). O3' of the pentose sugar group makes a hydrogen bond to the hydroxyl group of Tyr 246 and the pyrimidine ring forms hydrogen bonds with Asp 206 and Asn 214 (Fig 1b). O2 interacts with the main chain amide group of Asp 206 while O4 binds N $\delta$ 1 and O $\delta$ 1 of Asn 214.

Comparison of the apo- $TS_{VZV}$  and  $TS_{VZV}$ +dUMP structures reveals no major conformational difference upon the binding of dUMP (Fig 1a and 1b). C $\alpha$  backbones of  $TS_{VZV}$  monomers in both structures overlay well with an average rmsd of 0.69 Å. Only small side chain rearrangements are seen for the amino acids lining the active site. The most noticeable changes are observed for amino acids interacting directly with dUMP (Fig 1a and 1b). In the presence of dUMP, the side chains of Asp 206, Asn 214 and Tyr 246 shifted for direct interactions with the pentose sugar and pyrimidine ring. The loop spanning amino acids 204 to 206 also inclined towards dUMP for direct interactions. The side chain configurations of Arg 163' and Arg 164' from the other monomer are also slightly displaced. Although Tyr 123 does not make direct interactions with dUMP, the tyrosine ring is rotated in the presence of dUMP.

## Structural comparison of the coordination of dUMP in TS<sub>VZV</sub> and TS<sub>HS</sub>

Complex structures of different TS have been described to exist in either an open or a closed conformation [22, 35]. Both conformations are structurally conserved but the thiol group of



Table 1. Data collection and refinement statistics on the apo and complex structures of TS<sub>VZV</sub>.

	Apo TS <sub>VZV</sub> (4XSE)	TS <sub>VZV</sub> +dUMP (4XSD)	TS <sub>VZV</sub> +BVDU <sub>P</sub> (4XSC)
Data Collection			
Beamline	AS MX1	AS MX1	AS MX1
Wavelength (Å)	0.9537	0.9537	0.9537
Resolution Range (Å)	29.7–3.1 (3.1–3.2) <sup>a</sup>	30.0–2.9 (2.9–3.0) <sup>b</sup>	30.0-2.9 (2.9-3.0) <sup>b</sup>
Space group	p3 <sub>2</sub>	p3 <sub>2</sub>	p3 <sub>2</sub>
Unit cell	153.4 Å 153.4 Å 89.2 Å	150.2 Å 150.2 Å 89.2 Å	149.5 Å 149.5 Å 89.0 Å
	90.0° 90.0° 120.0°	90.0° 90.0° 120.0°	90.0° 90.0° 120.0°
Total reflections	155476	1195011	94756
Unique reflections	42243	49612	46238
Ι/Ι(σ)	8.5 (2.4) <sup>a</sup>	19.8 (3.9) <sup>b</sup>	8.1 (1.9) <sup>b</sup>
Multiplicity	3.7 (3.4) <sup>a</sup>	5.5 (5.2) <sup>b</sup>	1.5 (1.5) <sup>b</sup>
Completeness (%)	99.2 (97.5) <sup>a</sup>	100.0 (99.9) <sup>b</sup>	96.2 (95.3) <sup>b</sup>
R <sub>merge</sub> <sup>c</sup> (%)	12.7 (47.5) <sup>a</sup>	6.9 (38.0) <sup>b</sup>	9.7 (40.3) <sup>b</sup>
Refinement			
R <sub>factor</sub> d (%)	23.8	24.0	19.6
R <sub>free</sub> e (%)	26.8	28.0	23.5
Protein residues	1116	1081	1244
Solvent	0	9	30
Ligands	4 phosphate ions	4 dUMP	4 BVDU <sub>P</sub>
			4 1PE
RMSD bonds (Å)	0.5	0.0045	0.002
RMSD angles (°)	1.134	0.882	0.585
Ramachandran favored (%)	95	96	94.4
Ramachandran allowed (%)	5	4	5.1
Ramachandran outliers (%)	0	0	0.5
Rotamer outliers (%)	4.4	6.4	0.8

<sup>&</sup>lt;sup>a</sup> The values in the parentheses are for the highest resolution shell (3.1–3.2 Å).

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the catalytic cysteine in the closed conformation is in an active complex and binds covalently with C6 of the dUMP pyrimidine ring [22, 36]. The catalytic cysteine corresponds to Cys 183 in  $TS_{VZV}$  or Cys 195 in  $TS_{HS}$  (Fig 1b and 1c). The structure of  $TS_{VZV}$ +dUMP superimposes well with the open and closed conformations of  $TS_{HS}$  with an average rmsd of 0.7 Å and 0.9 Å along the C $\alpha$  backbone respectively.

The coordination of dUMP in the active site of  $TS_{VZV}$  was compared with the closed  $TS_{HS}$ , since the latter was reported to represent the conformation of a catalytically active  $TS_{HS}$  [22]. The overall coordination of dUMP in the active sites of both  $TS_{VZV}$  and  $TS_{HS}$  is highly similar with slight differences. The dUMP phosphate group is coordinated by three conserved arginines (Arg 38, Arg 203, and Arg 164') in  $TS_{VZV}$  but four conserved arginines (Arg 50, Arg 215, Arg 176' and Arg 177') in  $TS_{HS}$  (Fig 1b and 1c). O2 of the dUMP pyrimidine ring binds to the

<sup>&</sup>lt;sup>b</sup> The values in the parentheses are for the highest resolution shell (2.9–3.0 Å).

 $<sup>^{\</sup>text{c}}R_{\text{merge}} = \frac{|\Sigma_{\text{Nel}}\Sigma_i|I_i(hkl)-\bar{I}(hkl)|}{|\Sigma_{\text{Nel}}\Sigma_i/(hkl)|} \times 100$ , where  $I_i$  is the ith intensity measurement of reflection hkl,  $\bar{I}(hkl)$  is the mean intensity measurement of the symmetry related or replicated reflections of the unique reflection hkl.

 $<sup>^{\</sup>rm d}$   $R_{\rm factor} = \frac{|\Sigma_{\it hkl}|F_{\it obs}(\it hkl) - F_{\it calc}(\it hkl)|}{|\Sigma_{\it hkl}F_{\it obs}(\it hkl)|} imes 100$ , where  $F_{\it obs}$  and  $F_{\it calc}$  are the observed and calculated structure factors respectively.

e R<sub>free</sub> is equivalent to R<sub>factor</sub> but 5% of the measured reflections have been excluded from refinement and set aside for cross validation.



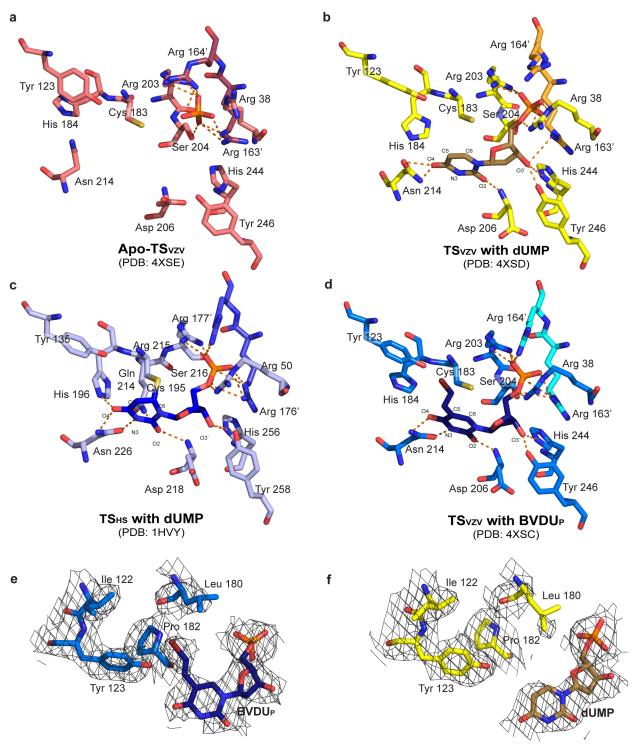


Fig 1. Stereo views of the  $TS_{VZV}$  and  $TS_{HS}$  active sites. Corresponding stereo views of the ligands and amino acids lining the TS active sites in the structures of (a) apo- $TS_{VZV}$  with a phosphate ion, (b)  $TS_{VZV}$  with dUMP, (c)  $TS_{HS}$  with dUMP (PDB ID: 1HVY) [35] and (d)  $TS_{VZV}$  with BVDU<sub>P</sub>. The polar interactions between the amino acids and the different ligands are illustrated by orange dotted lines.  $2F_0$ - $F_C$  electron densities of the binding (e) BVDU<sub>P</sub> and (f) dUMP with the surrounding hydrophobic amino acids are also shown (contoured at  $1\sigma$  with carved = 2.0).

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main chain amide of Asp 206 in  $TS_{VZV}$  and the corresponding Asp 218 in  $TS_{HS}$ . Both Asn 214 in  $TS_{VZV}$  and the corresponding Asn 226 in  $TS_{HS}$  bind to O4 of the dUMP pyrimidine ring.

Differences are also observed in the coordination of dUMP in TS<sub>VZV</sub> and TS<sub>HS</sub> (Fig 1b and 1c). While the catalytic cysteine is conserved in TS<sub>VZV</sub>, it does not appear to be covalently bound to dUMP as in some analog complexes from other TS (Fig 1b) [22, 36]. Distance of the catalytic cysteine thiol group to C6 of the dUMP binding to TS<sub>HS</sub> in the closed TS<sub>HS</sub> structure (PDB ID: 1HVY) was reported to be 2.1 Å [22]. The distance between the thiol group of the corresponding Cys 183 and C6 of dUMP in the TS<sub>VZV</sub>+dUMP structure varies across the four TS<sub>VZV</sub> subunits in the asymmetric unit, ranging from 3.2 Å to 3.7 Å and averages at 3.4 Å. This is more similar to the structure of TS<sub>HS</sub> in complex with only dUMP (PDB ID: 3HB8), where the distance between the catalytic thiol and C6 of the dUMP is also approximately 3.4 Å. This is consistent with the proposal that folate is required for formation of the covalent bond between dUMP and the catalytic cysteine [35]. In addition, O3' of the dUMP pentose sugar group is hydrogen-bonded to His 246 in TS<sub>VZV</sub> and the corresponding His 258 in TS<sub>HS</sub>. However, O3' is also hydrogen-bonded to Tyr 256 in TS<sub>HS</sub> but not the corresponding Tyr 244 in TS<sub>VZV</sub>. The distance between O3' and NE2 on Tyr 244 in TS<sub>VZV</sub> is 3.5 Å, which is too far for Tyr 244 to be a hydrogen bond donor. An additional hydrogen bond also exists between N3 of the pyrimidine ring and Oδ1 of Asn 226 in TS<sub>HS</sub> but not between N3 and the corresponding Asn 214 in TS<sub>VZV</sub>. Another distinctive difference is located at His 184 (His 196 in TS<sub>HS</sub>). In TS<sub>HS</sub>, there exists a direct interaction between the imidazole ring of His 196 and O4 of the pyrimidine ring but in TS<sub>VZV</sub>, the corresponding His 184 is tilted and does not make any direct interactions with dUMP (Fig 1c and 1d).

# Structure of TS<sub>VZV</sub> in complex with BVDU<sub>P</sub>

The structure of  $TS_{VZV}$  in complex  $BVDU_P$  was also determined by soaking  $BVDU_P$  with crystals of the second crystal form. The complex structure was determined and refined to a resolution of 2.9 Å (<u>Table 1</u>). There are also four  $TS_{VZV}$  monomers that are arranged into two homodimers in each asymmetric unit. Each monomer is made up of amino acids 15 to 295, except the disordered amino acids 98 to 117 in one monomer of the  $TS_{VZV}$  dimer.

The active site of each monomer contains a molecule of BVDU<sub>P</sub> (Fig 1d). However, there is no electron density for the bromide atom in two of the four subunits in the asymmetric unit. The missing bromide atom in BVDU<sub>P</sub> is likely attributed to radiation damage during data collection [37, 38]. A molecule of PEG 400, that is likely to be contributed by the crystallization buffer, is also found in the active site. It binds to BVDU<sub>P</sub> and is located in the  $TS_{VZV}$  active site where the folate binds.

The BVDU<sub>P</sub> phosphate group is coordinated by the conserved Arg 38, Arg 203 and Ser 204 from one monomer, and Arg 163' and Arg 164' from the other monomer (Fig 1d). The pentose sugar on BVDU<sub>P</sub> interacts with the surrounding residues through hydrogen bonds involving O3'. O3' is hydrogen-bonded to both the hydroxyl group on Tyr 246 and N $\epsilon$ 2 on the imidazole ring of His 244. BVDU<sub>P</sub> is also held in place through interactions involving the pyrimidine ring. O2 of the BVDU<sub>P</sub> pyrimidine ring is hydrogen-bonded to the main chain amide of Asp 206. O4 and N3 of the BVDU<sub>P</sub> pyrimidine ring are also hydrogen-bonded to N $\delta$ 1 and O $\delta$ 1 of Asn 214 respectively. The 5-bromovinyl substituent on BVDU<sub>P</sub> stretches towards and binds the hydrophobic patch on the active site that is lined by amino acids Tyr 100, Ile 122, Tyr 123, Leu 180 and Pro 182 (Fig 1e).

No major conformational difference is observed between the structures of  $TS_{VZV}+BVDU_P$  and  $TS_{VZV}+dUMP$ . However, the position and coordination of  $BVDU_P$  deviate somewhat from dUMP in the active site of  $TS_{VZV}$ . As a result, the direct interactions of  $BVDU_P$  and



 $TS_{VZV}$  is not conserved with that observed between dUMP and  $TS_{VZV}$  (Fig 1b and 1d). The O3' of the pentose sugar in BVDU<sub>P</sub> is hydrogen-bonded to His 244 and Tyr 246 but that in dUMP only interacts with Tyr 246. An additional interaction also exists between N3 of the BVDU<sub>P</sub> pyrimidine ring and O81 of Asn 214. The distance between the thiol group atom on Cys 183 to C6 of the pyrimidine ring in dUMP and BVDU<sub>P</sub> decreased from 3.4 Å to 3.0 Å respectively. The 5-bromovinyl substituent in BVDU<sub>P</sub> also caused slight side chain perturbations at Tyr 100, Ile 122, Tyr 123, Leu 180 and Pro 182 (Fig 1e and 1f).

In contrast, the coordination of BVDU<sub>P</sub> in  $TS_{VZV}+BVDU_P$  is more similar with the coordination of dUMP in the closed- $TS_{HS}$  structure (PDB ID: 1HVY) [22] (Fig 1c and 1d). The coordination of the phosphate groups, pentose sugars and pyrimidine rings of both BVDU<sub>P</sub> and dUMP is highly conserved in both  $TS_{VZV}$  and  $TS_{HS}$ . This is with exception to His 184. In  $TS_{HS}$ , O4 of the dUMP pyrimidine ring is hydrogen-bonded to N $\epsilon$ 2 of His 196 but in  $TS_{VZV}$ , the corresponding His 184 is not hydrogen-bonded to O4 of the BVDU<sub>P</sub> pyrimidine ring. The imidazole ring of His 184 in three of the four  $TS_{VZV}$  subunits in the asymmetric unit is rotated approximately 50–60° about C $\beta$  from the corresponding His 196 in  $TS_{HS}$  and does not bind to O4 of the BVDU<sub>P</sub> pyrimidine ring directly. Even though the conformation of His 184 in one of the  $TS_{VZV}$  subunits in the asymmetric unit is more similar to the corresponding His 196 in  $TS_{HS}$ , the distance between His 184 and BVDU<sub>P</sub> is still too far for direct interaction.

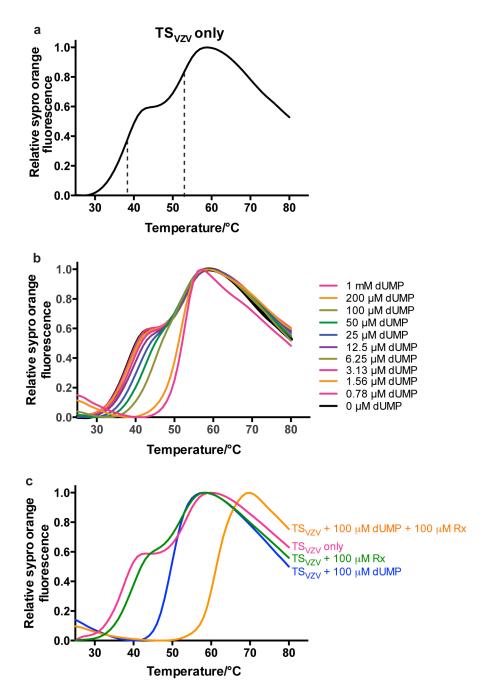
# Binding studies of TS<sub>VZV</sub> using differential scanning fluorimetry (DSF)

To study the binding of ligands to  $TS_{VZV}$ , we explored a thermal stability shift assay based on differential scanning fluorimetry (DSF). In the absence of any ligand, DSF shows that  $TS_{VZV}$  has a dual-transition melting curve that is made up of two sigmoidal curves (Fig 2a). The melting temperature ( $T_m$ ) of the first and second melting transition of  $TS_{VZV}$  in the absence of any ligand were estimated to be 38°C and 53°C respectively (Fig 2a). When  $TS_{VZV}$  was titrated with different concentrations of dUMP, a shift of the first transition of the  $TS_{VZV}$  melting curve was seen, while the second melting transition remained unchanged (Fig 2b). The first transition eventually disappeared with 100  $\mu$ M of dUMP and further increase of the dUMP concentration could not increase the  $T_m$  beyond the second melting transition (Fig 2b).

The  $T_m$  of  $TS_{VZV}$  is 50°C in the presence of 100  $\mu$ M dUMP (Fig 2c). When 100  $\mu$ M raltitrexed and 100  $\mu$ M dUMP was added to  $TS_{VZV}$ , the  $T_m$  of  $TS_{VZV}$  increased from 50°C to 61°C (Fig 2c). In the absence of dUMP, the addition of 100  $\mu$ M raltitrexed produced a dual-transition melting curve that is largely similar to the melting curve of  $TS_{VZV}$  without any ligand (Fig 2c). No significant stabilization effect on the first and second melting transitions was observed with raltitrexed (Fig 2c). This is consistent with the notion where dUMP must be bound in order for the successive binding of raltitrexed in the active site of  $TS_{HS}$  [18].

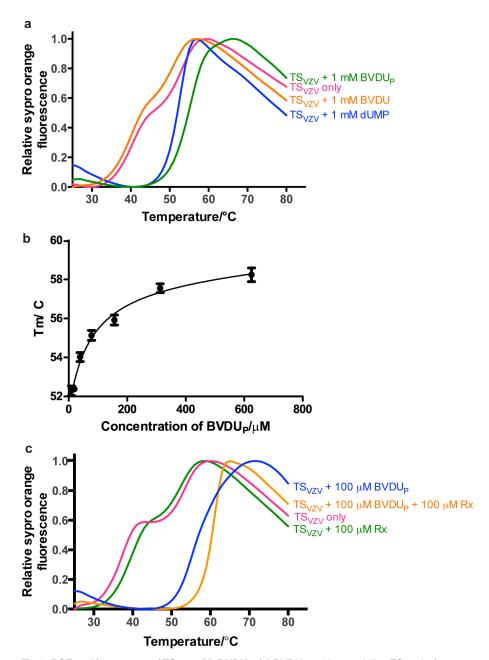
We were also interested in whether BVDU, a nucleoside based VZV drug with some similarities to the substrate dUMP, could also bind  $TS_{VZV}$ . The binding of  $TS_{VZV}$  to BVDU before (BVDU) and after *in vitro* phosphorylation (BVDU<sub>P</sub>) with  $TK_{HS}$  was therefore studied with DSF. The addition of 100  $\mu$ M BVDU to  $TS_{VZV}$  produced a dual-transition melting curve that is similar to the ligand free  $TS_{VZV}$  (Fig 3a). On the contrary, BVDU<sub>P</sub> produced a single melting transition with a  $T_m$  of approximately 57°C in DSF. This observed stabilization of  $TS_{VZV}$  by BVDU<sub>P</sub> is approximately 7°C higher than the stabilization of  $TS_{VZV}$  by 100  $\mu$ M dUMP (Fig 3a). Varying concentrations of BVDU<sub>P</sub> produced a dose responsive increment in the  $T_m$  of  $TS_{VZV}$  (Fig 3b). When 100  $\mu$ M of raltitrexed was added to  $TS_{VZV}$  and 100  $\mu$ M of BVDU<sub>P</sub>, the  $T_m$  of  $TS_{VZV}$  increased from 57°C to 64°C (Fig 3c).





**Fig 2. DSF melting curves of TS**<sub>VZV</sub> with dUMP. (a) The DSF melting curve of apo-TS<sub>VZV</sub> has two melting transitions and two T<sub>m</sub> can be estimated from the mid-slope of each sigmoidal transition. Dotted lines were added to the mid-slopes of each melting curve as references to the TS<sub>VZV</sub> T<sub>m</sub> for each melting transition. (b) Increasing concentrations of dUMP only increased the T<sub>m</sub> of the first melting transition. 100 μM of dUMP was sufficient to stabilize the first melting transition of TS<sub>VZV</sub> to produce a single melting transition. Further increase of the dUMP concentration could not increase the T<sub>m</sub> beyond the second melting transition. (c) Raltitrexed further stabilized TS<sub>VZV</sub> in the presence of 100 μM of dUMP in the DSF. In the absence of dUMP, raltitrexed could not stabilize TS<sub>VZV</sub> in DSF.

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**Fig 3. DSF melting curves of TS**<sub>VZV</sub> with **BVDU**<sub>P</sub>. (a) BVDU could not stabilize TS<sub>VZV</sub> before phosphorylation but after an *in vitro* phosphorylation with TK<sub>HS</sub>, BVDU<sub>P</sub> increased the T<sub>m</sub> of TS<sub>VZV</sub> by a larger extent than dUMP. (b) Varying concentrations of BVDU<sub>P</sub> increased the T<sub>m</sub> of TS<sub>VZV</sub> in a dose responsive manner. (c) Further stabilization of TS<sub>VZV</sub> was observed in the presence of 100 μM raltitrexed with 100 μM of BVDU<sub>P</sub>.

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## **Discussion**

TS is a highly conserved enzyme in many organisms including human, mouse, bacteria, protozoa and some viruses. TS is essential for the *de novo* synthesis of dTMP and its expression is up-regulated in actively replicating cells. It is an attractive drug target and several TS inhibitors including 5-fluorouracil, raltitrexed and pemetrexed are extensively used for treatments of



solid tumors [39, 40]. Structural information of TS from several different species in complex with its substrates and inhibitors is also available [17–30]. TS is seemingly only present in two human herpesviruses, VZV and KSHV [32, 33]. Both TS<sub>VZV</sub> and TS<sub>KSHV</sub> share high sequence similarities with TS<sub>HS</sub>, with at least 80% sequence identity at the TS active site. Enzymatic studies have showed that TS<sub>VZV</sub> and TS<sub>KSHV</sub> display similar catalytic kinetics as human TS in the presence of dUMP and mTHF [32]. However, both herpesviral TS remain poorly characterized till date.

In an attempt to further characterize  $TS_{VZV}$ , the crystal structures of the apo-, dUMP and BVDU<sub>P</sub> bound  $TS_{VZV}$  were determined. These constitute the first crystal structure of an eukaryotic viral TS. The overall structure is, as expected, well conserved with the human TS (S1 Fig) [35]. However, the  $TS_{VZV}$ +dUMP and  $TS_{VZV}$ +BVDU<sub>P</sub> structures have disordered regions that cannot be modeled in two of the four subunits of the crystallographic asymmetric unit. It is tempting to speculate that the disorder is correlated to the binding of the dUMP/BVDU<sub>P</sub>. However, these disordered regions are observed only in one of the monomer of the  $TS_{VZV}$  dimer. Although the binding of dUMP and BVDU<sub>P</sub> appears to induce minor conformational changes to the amino acids lining the active site, the positioning of both dUMP/BVDU<sub>P</sub> and the amino acids lining the active site are largely conserved in all the monomers (Fig 1). An inspection of the crystal packing in the unit cell of the  $TS_{VZV}$ +dUMP and  $TS_{VZV}$ +BVDU<sub>P</sub> structures reveals that these disordered regions are located on the protein surfaces that lack crystal contact with a neighboring monomer. Thus, the observed structural disorder is most likely due to differences in crystal packing and crystal contacts of the two molecules and not due to the binding of dUMP/BVDU<sub>P</sub> to  $TS_{VZV}$ .

The  $TS_{VZV}$ +dUMP structure adopts an open conformation, as concluded from the lack of covalent bonds between the thiol group of the catalytic cysteine and C6 of the dUMP pyrimidine ring (Fig 1b). The presence of the cofactor and the C-terminal loop have been proposed to be required for the closed conformation [41]. However, the  $TS_{VZV}$  protein construct that produced well diffracting crystals for structural determination of  $TS_{VZV}$  lacks the last six amino acids at the C-terminal. These six amino acids correspond to the C-terminal loop that has been reported to act as a lid over the TS active site and assist in folate binding [22]. Thus, the reason for an open conformation in  $TS_{VZV}$ +dUMP is likely partly due to the absence of the folate cofactor in the active site and the C-terminal loop in the  $TS_{VZV}$  structures.

The interactions between  $dUMP/BVDU_P$  and an anti-folate drug raltitrexed with  $TS_{VZV}$  were characterized with DSF. While the DSF demonstrated that  $dUMP/BVDU_P$  and raltitrexed are able to bind to  $TS_{VZV}$  in vitro, it also showed that raltitrexed is not able to bind  $TS_{VZV}$  in the absence of the nucleotides (Figs  $\underline{2c}$  and  $\underline{3c}$ ). This is consistent with the well-characterized sequential binding mode in other TSs  $[\underline{18}]$ . Thus, the lack of binding of raltitrexed to  $TS_{VZV}$  in the DSF is likely due to the requirement for bound dUMP in the active site before raltitrexed can bind.

Nucleoside analogs have routinely been used for treatments of viral infections [7, 42]. Many of these drugs are administered as a pro-drug and requires an activation step through phosphorylation by the viral or human TK [43]. To study the possibility of  $TS_{VZV}$  being a target protein of nucleoside analogs, which are often not commercially available in their phosphorylated form, *in vitro* phosphorylation by  $TK_{HS}$  was performed prior to the binding studies. The assay was tested with deoxythymidine, which should be phosphorylated by  $TK_{HS}$  to TMP in the presence of ATP. The *in vitro* phosphorylation was validated with DSF, as deoxythymidine is unable to bind to  $TS_{VZV}$  while dTMP gave significant shifts (§2 Fig).

Several clinically approved anti-herpesviral nucleoside analogs are 5-substituted pyrimidine nucleosides, which are modifications of the TS binding dUMP. BVDU is a clinically approved 5-substituted pyrimidine nucleoside analog that displays the highest potency against HSV-1



and VZV [42, 44]. It has also been shown to be an inhibitor of  $TS_{HS}$  [45]. Our DSF experiments demonstrated that BVDU is able to bind  $TS_{VZV}$  after *in vitro* phosphorylation by  $TK_{HS}$  (Fig 3a). This supports that the nucleoside analog has been successfully phosphorylated by  $TK_{HS}$  *in vitro*. The successful phosphorylation of BVDU<sub>P</sub> was eventually confirmed by the presence of the phosphate group on BVDU<sub>P</sub> in the complex structure of  $TS_{VZV}$  with BVDU<sub>P</sub> (Fig 1d). This may seem contradicting to previous studies that have reported a selective phosphorylation of BVDU by the herpesviral TK but  $TK_{HS}$  has been shown to be catalytically active towards BVDU, albeit with low efficiency [10, 46]. The effectiveness of  $TK_{HS}$  on BVDU may be attributed to the higher concentration of the kinase used *in vitro* and the efficiency of  $TK_{HS}$  activity on BVDU is likely to diminish *in vivo*.

The binding of BVDU<sub>P</sub> to  $TS_{VZV}$  was further confirmed with the complex structure,  $TS_{vzv}$ +BVDU<sub>P</sub>. The coordination of BVDU<sub>P</sub> is highly similar to the coordination of dUMP in  $TS_{VZV}$  (Fig 1b and 1d). Though BVDU<sub>P</sub> is not covalently bound to the catalytic thiol, a comparison of the structure of  $TS_{vzv}$ +BVDU<sub>P</sub> with the closed  $TS_{HS}$  shows that the coordination of BVDU<sub>P</sub> by  $TS_{VZV}$  is more conserved with the coordination of dUMP in the closed- $TS_{HS}$  (Fig 1c and 1d). Even in the absence of the folate cofactor and the C-terminal loop, BVDU<sub>P</sub> appears to be held in a closed conformation by the additional hydrophobic interactions contributed by its 5-bromovinyl substituent (Fig 1e). The presence of a bound PEG 400 in the  $TS_{VZV}$  active site, in the region where folate binds, may have also helped in stabilizing the closed conformation. However, rather than a catalytically active closed conformation, the BVDU<sub>P</sub> bound  $TS_{VZV}$  structure is likely to represent a catalytically inhibited  $TS_{VZV}$ . Together, these data suggest that phosphorylated BVDU inhibits  $TS_{VZV}$  by binding competitively to the active site and yield further support for that  $TS_{VZV}$  could be a target protein for BVDU in the cell after the activation by the herpesvirus TK.

BVDU<sub>P</sub> has been previously demonstrated to inhibit the catalytic activities of both  $TS_{VZV}$  and  $TS_{HS}$  in vitro [32, 47]. The high degree of structural conservation and the lack of selectivity of BVDU<sub>P</sub> between  $TS_{VZV}$  and  $TS_{HS}$  could potentially lead to  $TS_{HS}$  mediated adverse effects. However, structural differences observed between  $TS_{VZV}$  and  $TS_{HS}$  may confer additional selectivity of BVDU<sub>P</sub> towards  $TS_{VZV}$ . His 184 do not share the same conformation as the corresponding His 196 in  $TS_{HS}$  (Fig 1c and 1d). As a result, the imidazole ring of His 184 does not interact with O4 of the dUMP/BVDU<sub>P</sub> pyrimidine ring. Even though the amino acids lining the active sites of  $TS_{VZV}$  and  $TS_{HS}$  are highly conserved, there is a single amino acid substitution at Tyr 100 in  $TS_{VZV}$  that corresponds to Asn 112 in  $TS_{HS}$ . This could potentially aid in the design of more specific inhibitors towards  $TS_{VZV}$ . Regardless of the structural differences between  $TS_{VZV}$  and  $TS_{HS}$ , the potency of BVDU towards VZV infected cells is likely due to the preferential phosphorylation by herpesvirus TK [10, 46]. Nevertheless, further studies are required to evaluate whether  $TS_{VZV}$  is indeed a physiological relevant target protein for the mono-phosphorylated BVDU.

In conclusion, substrate binding in  $TS_{VZV}$  has been characterized with a thermal shift assay and the apo and complex structures with nucleotides have been determined. The different  $TS_{VZV}$  structures share features with the evolutionary conserved  $TS_{HS}$  but also show some small but significant differences. Binding between  $TS_{VZV}$  and an activated herpesviral drug BVDU was demonstrated and a complex structure with an *in vitro* phosphorylated BVDU was determined. Together this work adds further support BVDU as a good inhibitor for  $TS_{VZV}$  and provides the structural basis for structure driven design of more specific  $TS_{VZV}$  inhibitors.

#### **Materials and Methods**

#### Cell culture and harvest

The starter culture supplemented with 50  $\mu$ g/mL kanamycin (Sigma Aldrich) and 34  $\mu$ g/mL chloroamphenicol (Sigma Aldrich) was inoculated with a streak from a TS<sub>VZV</sub> glycerol stock.



The culture was incubated overnight at 37°C, 220 rpm. 2 mL of the starter culture was added into each flask containing 750 mL of fresh TB, 50  $\mu$ g/mL kanamycin and 34  $\mu$ g/mL chloroamphenicol. The cultures were incubated at 37°C, 180 rpm until OD<sub>600</sub> was 0.8. Expression was induced overnight with 0.5 mM IPTG at 18°C and harvested by centrifugation (4500 g for 10 minutes at 15°C) the next day. The cell pellet was flash frozen with liquid nitrogen, pounded and stored at -20°C.

## Protein purification

TS<sub>VZV</sub> was purified with a two-step standard purification protocol, consisting of an immobilized metal affinity chromatography step and a size exclusion chromatography step using the ÄKTAxpress protein purification systems at 4°C. Ice-cold lysis buffer (100 mM Na-HEPES pH 8.0 (Sigma Aldrich), 500 mM NaCl (Sigma Aldrich), 10 mM imidazole pH 8.0 (Sigma Aldrich), 10% glycerol (Sigma Aldrich) and 0.5 mM TCEP (Sigma Aldrich)) supplemented with 0.1 mg/ mL lysozyme (Sigma Aldrich), 1 μL/mL protease inhibitor cocktail (Nacalai) and 125 U/mL benzonase (Merck Millipore) was added to the frozen TS<sub>VZV</sub> cell pellet. The cell lysate was sonicated before clarifying by centrifugation at 47000 g for 25 minutes at 4°C and filtration of the supernatant with an 1.2 µm syringe filter. The filtrate was loaded onto a pre-equilibrated (10 mM Na-HEPES pH 7.5, 500 mM NaCl, 10 mM imidazole pH 7.5, 10% glycerol and 0.5 mM TCEP) 5 mL HiTrap IMAC HP column (GE Healthcare) and the column was washed with 25 mL of purification wash buffer (10 mM Na-HEPES pH 7.5, 500 mM NaCl, 10 mM imidazole pH 7.5, 10% glycerol). The protein was eluted with a step elution of 156 mM and 335 mM imidazole. The samples eluted with 335 mM imidazole were pooled and concentrated to 5 mL before loading onto a pre-equilibrated (20 mM Na-HEPES pH 7.5, 300 mM NaCl, 10% glycerol and 0.5 mM TCEP) HiLoad 16/600 Superdex 200 gel filtration column (GE Healthcare). The fractions were analyzed on a 4-12% NuPAGE Bis-Tris gel (Life Technologies) and the purest fractions were pooled and concentrated to 12 mg/mL with a 10 kD MWCO protein concentrator (GE Healthcare). The concentrated TS<sub>VZV</sub> samples were flash frozen in liquid nitrogen and stored at -80°C.

# In vitro phosphorylation of nucleosides using TK<sub>HS</sub>

Deoxythymidine was phosphorylated by incubating 1 mM deoxythymidine (Sigma-Aldrich) with 1 mM ATP (Sigma-Aldrich) and 40  $\mu g$  of recombinant purified TK $_{\rm HS}$  (gift from Chen Dan, Nanyang Technological University, Singapore) overnight at 37°C in a reaction buffer (20 mM HEPES pH 7.5, 300 mM NaCl, 25 mM MgCl $_2$ ). The reaction was stopped by heating the samples at 95°C for 15 minutes. The denatured TK $_{\rm HS}$  was removed by centrifugation at 13000 rpm for 5 minutes. The supernatant was removed and used for the crystallization and differential scanning fluorimetry (DSF) experiments with TS $_{\rm VZV}$ . The phosphorylated BVDU (BVDU $_{\rm P}$ ) was prepared by incubating BVDU (Santa Cruz Biotechnology) with TK $_{\rm HS}$  as with deoxythymidine.

## Crystallization and data collection

The apo-TS $_{\rm VZV}$  was crystallized in two crystal forms. The first crystal form was obtained by mixing 200 nL of the crystallization solution (12% Ethylene glycol, 0.1 M HEPES pH 7.5) with 100 nL of the protein solution (12 mg/mL) in a 96-wells sitting drop Intelli-plate (Art Robbins) at 20°C. The second crystal form was obtained by adding 200 nL of the crystallization solution (0.1 M Tris-HCl pH 8.0 and 40% PEG 300) to 100 nL of the protein solution (12 mg/mL) in a 96-wells sitting drop Intelli-plate at 20°C. Crystals of the second crystal form were soaked with 1 mM dUMP for three minutes and 1 mM BVDU $_{\rm P}$  for six minutes for the structures of the



complex of  $TS_{VZV}$  with dUMP ( $TS_{VZV}$ +dUMP) and  $TS_{VZV}$  with BVDU<sub>P</sub> ( $TS_{VZV}$ +BVDU<sub>P</sub>) respectively. The crystals were then flash frozen with liquid nitrogen. Diffraction data were collected at the MX2 Micro Crystallography beam line at the Australian Synchrotron, Australia using the Blu-Ice data collection software [48]. All diffraction datasets were integrated and scaled with HKL2000 [49].

## Structure determination

The apo structure of  $TS_{VZV}$  was determined by molecular replacement with Phaser, using a molecule of  $TS_{HS}$  (PDB ID: 1HZW) as the search model [35]. Inspection of the electron density maps revealed spherical positive electron densities at one end of the  $TS_{VZV}$  active site and a phosphate ion was manually modeled into the positive electron densities using coot [50]. The structure of  $TS_{VZV}$  with phosphate was further refined with phenix refine from the Phenix suite [51, 52].

The structures of  $TS_{VZV}$  in complex with dUMP ( $TS_{VZV}$ +dUMP) and  $BVDU_P$  ( $TS_{VZV}$ +- $BVDU_P$ ) were phased with molecular replacement using Phaser from the CCP4 suite [53, 54]. A  $TS_{VZV}$  monomer from the apo- $TS_{VZV}$  structure was used as the molecular replacement search model. Inspection of the electron density maps of  $TS_{VZV}$ +dUMP and  $TS_{VZV}$ +BVDU<sub>P</sub> revealed clear electron densities for dUMP and  $BVDU_P$  at their respective  $TS_{VZV}$  active sites. dUMP and  $BVDU_P$  were manually modeled into the respective positive electron densities using coot [50]. Iterations of automated and manual refinements of the structure were performed with phenix.refine and coot [50, 52]. The final structure refinement of the  $TS_{VZV}$  structures were validated with MolProbity [55]. All figures of the  $TS_{VZV}$  structures were displayed with pymol [56]. The apo  $TS_{VZV}$ ,  $TS_{VZV}$ +dUMP and  $TS_{VZV}$ +BVDU<sub>P</sub> structures were deposited on the PDB as 4XSE, 4XSD and 4XSC.

# Differential scanning fluorimetry

 $TS_{VZV}$  was diluted to 0.2 mg/mL with the DSF buffer (20 mM HEPES pH 7.5 and 300 mM NaCl) supplemented with 5x SYPROrange (Life Technologies) and was added to the 96-wells PCR plate. Triplicates were performed. Samples were heated from 25°C to 80°C, in increments of 1°C. The average absolute fluorescence intensity of the triplicates was plotted against the temperature (°C) on Prism 6 (GraphPad, Prism). The curves were then normalized with Prism 6 to obtain the DSF melting curves of  $TS_{VZV}$ . The melting temperatures ( $T_{m}$ ) of  $TS_{VZV}$  were estimated from the mid-point of each sigmoidal curve.

The effect of different ligands on  $TS_{VZV}$  was also performed with DSF as above, with slight modifications. In addition to diluting  $TS_{VZV}$  in the DSF buffer,  $TS_{VZV}$  was mixed with 1 mM dUMP (Sigma-Aldrich), 1 mM dTMP (Sigma-Aldrich), 1 mM deoxythymidine (Sigma-Aldrich) before and after phosphorylation or 1 mM BVDU (Santa Cruz) before and after phosphorylation. In the binding studies of raltitrexed with  $TS_{VZV}$ , the DSF experiment was repeated with 100  $\mu$ M of raltitrexed (Selleck Chemicals) in the presence and absence of 100  $\mu$ M dUMP or BVDU<sub>P</sub>. The dose responsive DSF of  $TS_{VZV}$  with dUMP was obtained by mixing  $TS_{VZV}$  with dUMP at varying concentrations of 1 mM, 200  $\mu$ M, 100  $\mu$ M, 50  $\mu$ M, 25  $\mu$ M, 12.5  $\mu$ M, 6.25  $\mu$ M, 3.125  $\mu$ M, 1.56  $\mu$ M, 0.78  $\mu$ M and 0  $\mu$ M. Likewise, the dose responsive DSF of  $TS_{VZV}$  with the phosphorylated BVDU was also performed by mixing  $TS_{VZV}$  with 625  $\mu$ M, 313  $\mu$ M, 156  $\mu$ M, 78  $\mu$ M and 39  $\mu$ M of BVDU<sub>P</sub>. The normalized melting curves and the  $T_m$  of  $TS_{VZV}$  in the presence of different ligands were obtained as described above.

## **Supporting Information**

**S1 Fig. Crystal structure of apo-TS**<sub>VZV</sub>. Two orientations of the apo-TS<sub>VZV</sub> dimer are illustrated in cartoon. A phosphate ion has been found in each TS<sub>VZV</sub> active site and both



phosphate ions are displayed as sticks. (TIF)

S2 Fig. DSF melting curves of  $TS_{VZV}$  with deoxythymidine before and after phosphorylation. Deoxythymidine did not stabilize  $TS_{VZV}$  before phosphorylation but after an *in vitro* phosphorylation with  $TK_{HS}$ , it stabilizes  $TS_{VZV}$  to a similar extent as dTMP. (TIF)

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### **Author Contributions**

Conceived and designed the experiments: KH SLD SV LXP TC PN. Performed the experiments: KH SV LXP. Analyzed the data: KH SLD SV LXP TC PN. Contributed reagents/materials/analysis tools: KH SLD SV LXP TC PN. Wrote the paper: KH SLD PN.

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