

Reference Standardization for Mass Spectrometry and High-resolution Metabolomics Applications to Exposome Research

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ABSTRACT

The exposome is the cumulative measure of environmental influences and associated biological responses throughout the lifespan, including exposures from the environment, diet, behavior, and endogenous processes. A major challenge for exposome research lies in the development of robust and affordable analytic procedures to measure the broad range of exposures and associated biologic impacts occurring over a lifetime. Biomonitoring is an established approach to evaluate internal body burden of environmental exposures, but use of biomonitoring for exposome research is often limited by the high costs associated with quantification of individual chemicals. High-resolution metabolomics (HRM) uses ultra-high resolution mass spectrometry with minimal sample preparation to support high-throughput relative quantification of thousands of environmental, dietary, and microbial chemicals. HRM also measures metabolites in most endogenous metabolic pathways, thereby providing simultaneous measurement of biologic responses to environmental exposures. The present research examined quantification strategies to enhance the usefulness of HRM data for cumulative exposome research. The results provide a simple reference standardization protocol in which individual chemical concentrations in unknown samples are estimated by comparison to a concurrently analyzed, pooled reference sample with known chemical concentrations. The approach was tested using blinded analyses of amino acids in human samples and was found to be comparable to independent laboratory results based on surrogate standardization or internal standardization. Quantification was reproducible over a 13-month period and extrapolated to thousands of chemicals. The results show that reference standardization protocol provides an effective strategy that will enhance data collection for cumulative exposome research. In principle, the approach can be extended to other types of mass spectrometry and other analytical methods.

Key words: amino acids; metabolomics; analytical chemistry; environment; mass spectrometry

Previous research shows that Fourier-transform mass spectrometry (MS) coupled to liquid chromatography (LC), termed “high-resolution metabolomics (HRM),” provides a practical approach

to detail physiological chemistry of an individual (Johnson *et al.*, 2010; Jones *et al.*, 2012; Soltow *et al.*, 2013). With this approach, LC is coupled with ultra-high resolution MS to detect >100 000 ions,

characterized by accurate mass, retention time (RT) and ion intensity. Although multiple ions are generated from individual chemicals, single ions are detected for many low-abundance chemicals so that the number of chemicals detected exceeds 20 000. Resolution of low-abundance chemicals with mass accuracy within a few parts-per-million is a key characteristic enabling this unsurpassed coverage of endogenous metabolites and exogenous chemicals in biologic samples.

Advances in LC, MS and data extraction (Chen *et al.*, 2012; Ivanisevic *et al.*, 2013; Uppal *et al.*, 2013) have further expanded the dynamic range for detection of chemicals in biological samples by at least an order of magnitude in the last decade. The use of triplicate analytical replicates, in particular, has transformed reliable measurement of low-abundance ions. Results show that 20 000 ions can be measured in human serum with median coefficient of variation (CV) $\leq 15\%$ (Uppal and Jones, unpublished), sufficient to measure thousands of chemicals in a single analysis.

Importantly, HRM detects relevant environmental chemicals that are needed to establish the exposome (Jones *et al.*, 2012; Miller and Jones, 2013). For instance, plasticizers, insecticides, fungicides, herbicides, drugs, and bacterial products are detected simultaneously with endogenous metabolites (Cribbs *et al.*, 2014; Frediani *et al.*, 2014; Go *et al.*, 2014a; Jones *et al.*, 2012; Neujahr *et al.*, 2014; Osborn *et al.*, 2013; Park *et al.*, 2012; Roede *et al.*, 2013). Environmental chemicals correlate with plasma phenylalanine (Go *et al.*, 2015) and with health behaviors, eg, cotinine with smoking (Go *et al.* 2014a; Park *et al.*, 2012). Exposome research is facilitated by use of HRM to study aging (Hoffman *et al.*, 2014; Zhao *et al.*, 2014), Parkinson's disease (Roede *et al.*, 2013), asthma (Fitzpatrick *et al.*, 2014), human immunodeficiency virus infection (Cribbs *et al.*, 2014), lung transplantation (Neujahr *et al.*, 2014), red blood cells stored for transfusion (Roback *et al.*, 2014), and macular degeneration (Osborn *et al.*, 2013). Human studies are also available on intraindividual variations due to diet (Stamler *et al.*, 2013) and nutritional deficiency (Gregory *et al.*, 2013), and model studies are available for environmental exposures (Go *et al.* 2014a; Park *et al.*, 2012), molecular mechanisms in immunity (Li *et al.*, 2013; Ravindran *et al.*, 2014; Xu *et al.*, 2014) and complex environmental disease mechanisms (Go *et al.*, 2014a; Roede *et al.*, 2014). This widespread use emphasizes the need for methods to facilitate the integration of data from different studies into common analytic structures to characterize the exposome.

Measurement of 20 000 chemicals in human plasma (Ivanisevic *et al.*, 2013; Uppal *et al.*, 2013) by HRM provides a useful capability for exposome research. However, HRM is limited by reliance upon relative quantification, which allows for the study of chemicals among a set of samples but limits comparisons between platforms or within the same facility at different times of analysis. In a targeted study of aminothiols in human plasma (Johnson *et al.*, 2008), we found that chemicals differing by nearly 4 orders of magnitude were quantifiable by LC-MS with isotopic internal standards, but this approach is impractical for large numbers of analytes. Complex mixtures of external standards provide a possible alternative, but external calibration is usually avoided for LC-MS due to nonlinearity and ion suppression effects (Muller *et al.*, 2002).

The ability to measure chemicals by LC-MS differing in concentration by nearly 4 orders of magnitude suggests that simultaneous quantification of thousands of chemicals with substantial differences in concentration may be possible. This is supported by method-of-addition studies in biological matrices, which show linear increases in intensity with analyte addition

(Chelius and Bondarenko, 2002; Wang *et al.*, 2003), suggesting linearity is sufficient to allow single point calibration. Van der Greef *et al.* (2007) discussed the use of pooled reference standards for nontargeted analysis of large sample batches with significant time between each batch and established feasibility for single point calibration using glucose as an example.

This study builds upon this analytic structure for quantification in exposome research, with the specific goal to develop analytical procedures for HRM over time and between research facilities (Table 1). We analyzed human samples with external calibration, internal standardization, surrogate standardization, and reference standardization. External calibration involved calibration of amino acid (AA) content in 117 human samples against measures obtained by conventional AA analysis. Internal standardization involved measurement of methionine and tyrosine in these samples by stable isotope dilution. Surrogate standardization involved quantification of other AAs against a stable isotopic internal standard using relative response factors (RRFs) (Greizerstein *et al.*, 1997). Reference standardization involved quantification of individual chemicals relative to pooled reference plasma that was calibrated to National Institute of Standards and Technology (NIST) Standard Reference Material for human plasma (SRM 1950) (McGaw *et al.*, 2010). Results of these analyses show that the latter approach, which represents a simple extension of the practice advocated by van der Greef *et al.* (2007), provides an effective framework to quantitatively compare endogenous metabolites and environmental chemicals for use in cumulative exposome research.

MATERIALS AND METHODS

Chemicals and HPLC columns. Acetonitrile (HPLC grade), formic acid (HPLC grade), and water (HPLC grade) were obtained from Sigma-Aldrich (St. Louis). A mixture of internal standard stable isotopic chemicals (Soltow *et al.*, 2013) included [¹⁵N]-L-tyrosine, [trimethyl-¹³C₃]-caffeine and [¹⁵N,¹³C₅]-L-methionine from Cambridge Isotope Laboratories, Inc (Andover, Pennsylvania).

Human plasma samples. Subsets of samples (n = 117) and (n = 157) from 2 studies (ClinicalTrials.gov Identifier: NCT00248638 and NCT00336570, respectively) were used for this methods development research. Details of the complete cohort studies are provided elsewhere (GLND, TR Ziegler, MD, Principal Investigator; PREMED, AA Quyyumi, MD, Principal Investigator). For the present purposes, the demographic and clinical details were not considered relevant to the tests of analytic procedures, and therefore, were not included in the analyses. The use of samples for additional analysis was allowed in the informed consent; both studies were reviewed and approved by the Emory University Investigational Review Board. GLND subjects were participants in a trial (NCT00248638) to determine whether the glutamine (GLN)-containing dipeptide alanyl-GLN protects against morbidity or mortality in surgical intensive care unit patients. The samples analyzed included individuals from treatment and control groups. PREMED subjects were healthy participants between 30 and 90 years who were studied to define a "normal" value or range of values in healthy people and to evaluate tests specifically designed to look for evidence of early multiorgan disease (NCT00336570). Both studies included both sexes and individuals of different races and ethnicities. Plasma was stored at -80°C prior to LC-MS analysis.

NIST SRM 1950 and Q-Standard. Our standard operating procedures involves use of 2 pooled human reference plasma

TABLE 1. Calibration Methods for Exposome Research

Method	Comments
External calibration with method of additions	<ul style="list-style-type: none"> • Tested for AAs in 117 samples • Provides accurate quantification • Impractical for use to quantify large numbers of metabolites
Internal standardization with stable isotopic standards	<ul style="list-style-type: none"> • Tested for methionine and tyrosine in 117 samples • Provides accurate quantification • Impractical for use to quantify large numbers of metabolites
Surrogate standardization with stable isotopic standards	<ul style="list-style-type: none"> • Tested for AAs in reference standard and 117 samples • Accuracy dependent upon consistency of RRF • Useful secondary method for chemicals unstable in reference
Reference standardization using calibrated reference	<ul style="list-style-type: none"> • Tested for AAs in 117 samples • Tested for selected metabolites in 2 sample sets • Tested for unidentified chemicals in 2 reference standards • Best primary method for quantification

HRM provides relative quantification of thousands of metabolites. While useful for many metabolomics studies, relative quantification is difficult to combine to create cumulative databases for exposome research. The present research examined 4 methods for quantification of plasma metabolite concentrations for data obtained using HRM.

samples for each sample set, with quality control and assurance defined by the consistency of data within concurrently analyzed batches of 20 samples (see Go et al., 2014b). The rationale for 2 pooled references was that the SRM1950, although preferable because it has certified concentrations, is costly and of limited supply. Thus, we adopted a procedure to analyze SRM1950 at the beginning and end of each study (or every 25 days) and analyze a second pooled reference sample that was prepared within our laboratory, at the beginning and end of each batch of 20 samples (twice daily). SRM1950 consisted of pooled plasma from an equal number of healthy men and women aged 40–50 years. Our pooled reference Q-Std (Q-Standard, independently prepared pooled reference plasma) consisted of plasma pooled from 2 separate lots, each from an unknown number of males and females without demographic information, from Equitech-Bio, Inc, (Kerrville, Texas); direct comparison of QStd and SRM1950 showed that they were sufficiently similar to allow calibration of the QStd relative to SRM1950 for specific metabolite quantification (see Results).

High-resolution metabolomics. Human plasma samples (see above) were extracted and analyzed as previously described (Soltow et al., 2013; Uppal et al., 2013). Briefly, extractions were performed with acetonitrile containing a mixture of internal standards and maintained in an autosampler at 4°C until injection. LC was performed using a C₁₈ column (Higgins Analytical, Targa, 2.1 × 10 cm) with a short, end-capped C₁₈ precolumn (Higgins Analytical, Targa guard) and an acetonitrile gradient. A flow rate of 0.35 ml/min was used for the first 6 min and then increased to 0.5 ml/min for the remaining 4 min. The first 2-min period consisted of 5% solution A (2% formic acid), 60% water, 35% acetonitrile, followed by a 4-min linear gradient to 5% A, 0%

water, 95% acetonitrile. The final 4-min period was maintained at 5% A, 95% acetonitrile. Mass spectrometry was performed using an (Linear Trap Quadrupole) LTQ-Velos-Orbitrap mass spectrometer (Thermo Fisher, San Diego, California): HESI probe with S-lens combination for Electrospray Ionization (ESI); MS1 mode scanning *m/z* (mass to charge, ion) range of 85–2000; resolution, 60 000; maximum number of ions collected, 5.00 × 10⁵. The maximum injection time, 5 ms; capillary temperature, 275°C; source heater; 45°C; voltage, 4.6 kV; sheath gas, 45 (arbitrary units); auxiliary gas flow, 5 (arbitrary units); sweep gas flow, 0 (arbitrary units). Each sample was run in triplicate using a 10 μl injection volume. Data were collected continuously over the 10-min chromatographic separation and stored as .raw files.

The MS files were converted using XCalibur file converter software (Thermo Fisher, San Diego, California) to .cdf files for data processing. The .raw files are time and date stamped, and electronically archived as the original data for use in subsequent reanalysis. The data were processed for peak extraction and quantification of ion intensities using xMSanalyzer (MS¹ mode) (Uppal et al., 2013) with aPLCMS (Yu et al., 2009). The parameter settings used for data extraction determine the number of features retained, where an *m/z* feature is defined as an accurate mass *m/z* (defined as ≤10 parts per million mass resolution) with associated RT and intensity. For this study, feature and sample filtering were set to retain features that had a median CV less than 50%, a Pearson correlation greater than 0.7 among 3 replicate injections for each sample, and fewer than 30% missing values. The identities of selected features were validated using MS² and compared with either authentic reference standards (when available) or online databases (Smith et al., 2005; Wishart et al., 2013). An online supplement (Supplementary Data S1) is provided with details on chemical identification. Statistical analysis included Pearson correlation (*r*) analyses between analytic platforms. Significance levels were calculated using a standard *t* test ($t = r/\sqrt{1 - r^2/n - 2}$). Raw *P* values are provided, where *P* < .0025 is significant for 20 comparisons using Bonferroni correction.

Amino acid analysis. AA analysis was performed using either a Beckman System 6300 High Performance Amino Acid Analyzer (Beckman Instruments, Inc. Palo Alto, California) or a Biochrom 30 Amino Acid Analyzer (Biochrom US, Holliston, Massachusetts) at the Emory Genetics Laboratory.

RESULTS

Consistency of Analysis

For 117 human plasma samples, the median CV for intensity among triplicate technical replicates for 22 625 ions (*m/z*) across all samples was 9.1%. Results for total ion intensity for the entire sample set showed a CV of 39.6%, with only a relatively small decrease to 33.4% obtained by averaging triplicates prior to calculating the standard deviation. Thus, the major variation in total signal intensity was due to differences in characteristics of individual samples resulting from either total chemical content, matrix effects that impact total signal detection, or both. The smaller median CV for individual metabolites suggests that the variation in total ion intensity was due to differences in total metabolite content among samples rather than analytical variations. Differences in total metabolite content could contribute to a matrix effect that impacts quantification, and normalization procedures for—omics data sometime assume that variations in total signal are due to analytic, rather than

biologic, differences. Consequently, to determine whether variation in total ion intensity impacted signal detection for individual chemicals, we analyzed intensities for internal standards with and without normalization to total signal intensity for each respective sample. For $^{15}\text{N}^{13}\text{C}$ -Met, the mean CV was 6.0% and the median CV was 13.3%; for ^{15}N -Tyr, the mean CV was 8.9% and the median CV was 13.4%. Normalization of signal intensities for $^{15}\text{N}^{13}\text{C}$ -Met to the total ion intensity for each respective sample did not improve CV (mean, 14.0; median 17.9%). Similarly, normalization of signal intensities for ^{15}N -Tyr to the respective total ion intensities did not improve coefficients of variation (mean, 28.0%; median 16.2%). Thus, for subsequent analyses, intensities were used without normalization to total ion intensity.

To test whether normalization relative to a surrogate standard decreased variability for replicate analyses, $^{15}\text{N}^{13}\text{C}$ -Met was normalized relative to ^{15}N -Tyr. Because these standards were added together during extraction with acetonitrile, they are present in each sample at an exact ratio relative to each other. The results showed that normalization of $^{15}\text{N}^{13}\text{C}$ -Met signal relative to ^{15}N -Tyr increased the CV (median, from 8.0 to 16.2%; mean from 13.3% to 28.0%). That is, directly referencing 1 chemical against another chemical amplified the variation of the target chemical, apparently due to the variability in detection of the reference. Consequently, these data indicate that for quantification in HRM, signal intensities for individual ions without normalization to an internal standard, provides smaller CV.

Comparison of HRM Signals for AAs to Results From Conventional AA Analysis

To directly test HRM analyses of individual chemicals, signal intensities for AAs in the 117 samples were compared with measurements obtained by conventional AA analysis performed on matched samples by an independent laboratory. These analyses were performed in random order and operators were blinded to the nature of the samples. Correlations for 17 AA, ranging from relatively poor correlation for cystine (CySS, $r=0.38$, $P=2.2 \times 10^{-5}$; not shown) to high correlations for Met ($r=0.97$, $P=1.34 \times 10^{-74}$) and Phe ($r=0.97$, $P=8.28 \times 10^{-73}$) are shown in Figure 1 and Table 2. To determine whether the correlations for Met were driven by the most extreme high values, analyses were repeated after removal of the top 2 values and top 6 values; respective r and P values were $r=0.90$, $P=3.44 \times 10^{-43}$ and $r=0.90$, $P=8.01 \times 10^{-40}$. For Phe, results for analyses after removal of the top 1 and top 5 values were $r=0.95$, $P=1.59 \times 10^{-43}$ and $r=0.94$, $P=4.81 \times 10^{-53}$.

Poor correlation for CySS was found to be due to partial oxidation to sulfoxide and sulfone forms, and efforts to improve accuracy by summation of the respective forms was not useful, apparently due to the additive error of the individual measurements. Leucine (Leu) and Isoleucine (Ile) were not resolved by LC-MS, but AA analyses for each were correlated with the combined LC-MS signal. Based upon " r " values, the reliability of the LC-MS quantification from best to worst relative to conventional AA analysis followed the order: Met=Phe>Tyr>

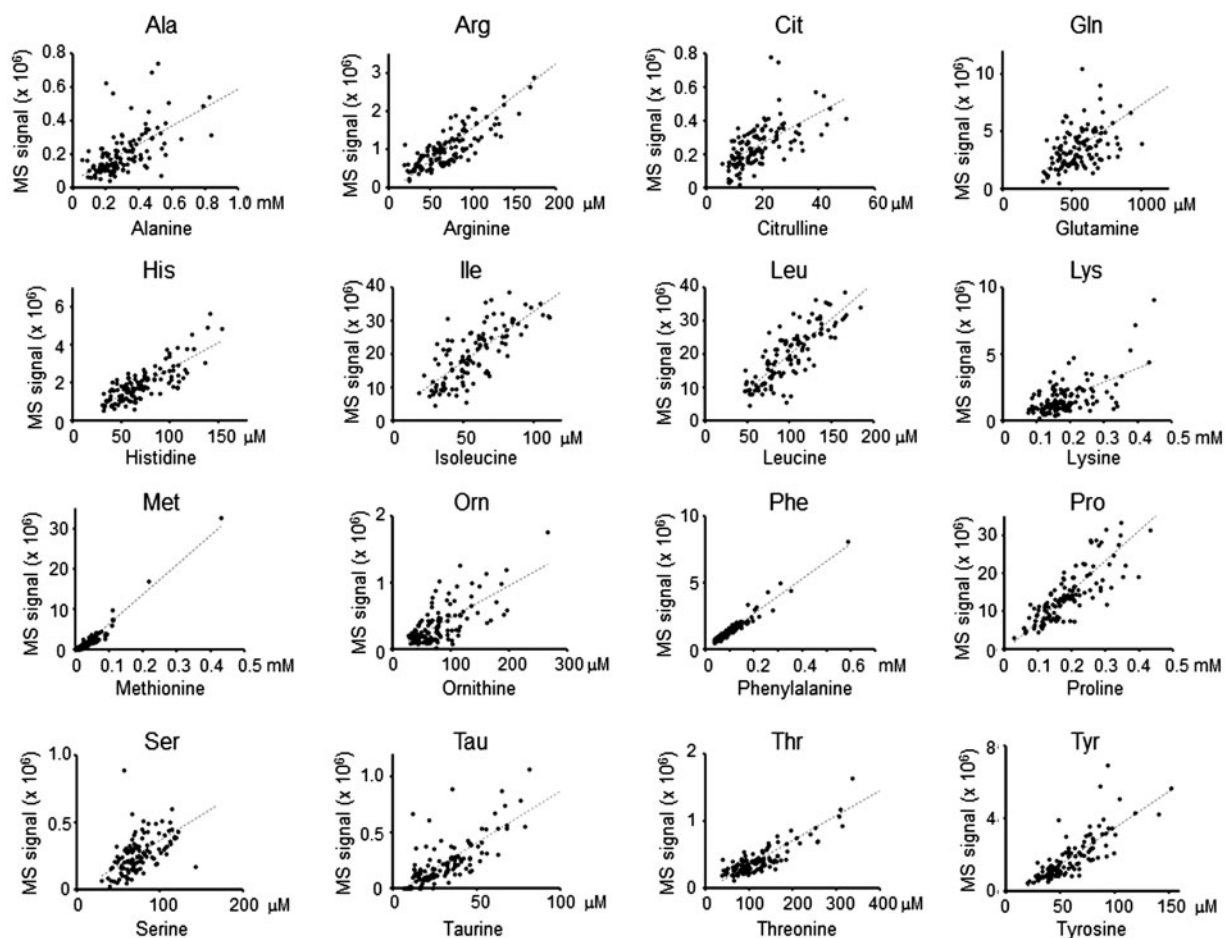


FIG. 1. Cross-laboratory comparison of amino acid (AA) quantification. Randomized and blinded analyses of 117 human plasma samples were performed in independent laboratories by HRM and automated AA analysis. Pearson correlation coefficients (r) are provided in Table 2; all correlations were significant at $P < .05$.

TABLE 2. Comparison of AA Detection by HRM to Conventional AA Analysis

AA	AA Analysis Mean (μM)	LC-FTMS C ₁₈ Mean Intensity	R	P	m/z	Adduct	RT (s)	Median CV (%)
Alanine	321	216 532	0.70	2.97E-18	90.0543	H ⁺	61	15.3
Arginine	73	1 095 024	0.84	4.76E-32	219.0818	2Na ⁺ -H ⁺	47	5.4
		405 164	0.79	6.34E-26	197.1000	Na ⁺	46	10.4
		1 825 927	0.79	1.10E-25	175.1180	H ⁺	50	6.0
Aspartic acid	7	No highly associated identifiable m/z						
Citrulline	19	252 184	0.61	3.16E-13	176.1024	H ⁺	71	17.1
Cystine	46	277 006	0.38	2.23E-05	284.9939	2Na ⁺ -H ⁺	52	8.6
Glutamate acid	58	No highly associated identifiable m/z						
Glutamine	554	6 305 193	0.68	4.54E-17	191.0393	2Na ⁺ -H ⁺	49	2.5
		295 356	0.66	1.35E-15	192.0427	2Na ⁺ -H ⁺	51	8.7
		7 382 949	0.59	4.67E-12	147.0756	H ⁺	60	5.4
Glycine	242	Not within m/z range measured						
Histidine	71	1 914 683	0.81	5.97E-28	200.0398	2Na ⁺ -H ⁺	52	5.4
		546 666	0.76	2.72E-23	178.0578	Na ⁺	48	5.3
Isoleucine	59	20 296 291	0.79	5.17E-26	132.1012	H ⁺	57	6.6
		1 158 099	0.61	5.45E-13	133.1046	H ⁺ (¹³ C)	56	8.6
Leucine	104	20 296 291	0.84	1.36E-32	132.1012	H ⁺	57	6.6
		1 158 099	0.68	1.02E-16	133.1046	H ⁺ (¹³ C)	56	8.6
		301 942	0.57	3.37E-11	176.0649	2Na ⁺ -H ⁺	53	10.1
Lysine	187	1 505 166	0.73	1.58E-20	191.0759	2Na ⁺ -H ⁺	49	7.7
		1 715 298	0.61	6.81E-13	147.1120	H ⁺	59	13.0
Methionine	41	2 047 964	0.97	1.34E-74	150.0575	H ⁺	55	6.2
		247 136	0.82	1.12E-29	194.0213	2Na ⁺ -H ⁺	52	16.7
Ornithine	82	385 853	0.69	1.29E-17	177.0599	2Na ⁺ -H ⁺	47	14.3
		646 111	0.62	1.88E-13	133.0967	H ⁺	63	9.3
Phenylalanine	108	1 455 188	0.97	8.28E-73	210.0490	2Na ⁺ -H ⁺	53	3.4
		17 603 867	0.86	1.91E-35	166.0854	H ⁺	61	5.5
		158 595	0.77	2.91E-24	211.0524	H ⁺ (¹³ C)	54	17.6
Proline	192	14 676 899	0.84	8.79E-32	116.0700	H ⁺	57	4.9
		401 834	0.65	2.32E-15	160.0337	2Na ⁺ -H ⁺	49	8.6
Serine	74	273 417	0.58	1.26E-11	106.0494	H ⁺	61	17.8
		235 221	0.58	1.29E-11	150.0132	2Na ⁺ -H ⁺	42	17.3
Taurine	34	277 462	0.83	2.91E-31	148.0033	Na ⁺	57	10.5
		175 084	0.79	2.14E-26	126.0213	H ⁺	70	9.1
Threonine	124	427 113	0.88	9.01E-39	164.0286	2Na ⁺ -H ⁺	49	6.0
		908 010	0.81	3.03E-28	120.0648	H ⁺	55	7.6
Tyrosine	59	653 421	0.92	3.45E-49	226.0441	2Na ⁺ -H ⁺	52	3.3
		1 817 830	0.81	4.03E-28	182.0802	H ⁺	56	11.1
Tryptophan		Not detected by AA analysis						
Valine	226	No highly associated identifiable m/z						

Data are summarized for 117 GLND serum samples. Pearson correlation (r) with respective P is given for accurate mass ions (m/z) detected at indicated RT. The median CV is given for the triplicate analyses by HRM.

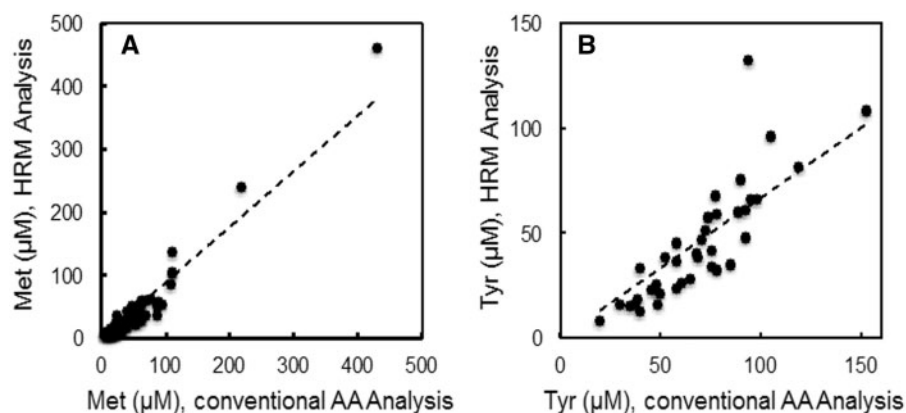


FIG. 2. Comparison of absolute concentration measurements for methionine (Met) and tyrosine (Tyr) by HRM and amino acid (AA) analysis. Met (A) and Tyr (B) concentrations were obtained by HRM using stable isotope dilution and directly compared to corresponding measurements by conventional AA analysis.

TABLE 3. AA Quantification in NIST1950 by HRM Using Surrogate Internal Standards ($^{15}\text{N}^{13}\text{C}$ -Met and ^{15}N -Tyr)

AA	Adduct	McGaw et al. (μM)	RRF	vs $^{15}\text{N}^{13}\text{C}$ C-Met (μM)	RRF	vs ^{15}N -Tyr (μM)	Detected N	Median CV
Alanine	H^+	300	74	—	16	—	0	—
Arginine	$2\text{Na}^+-\text{H}^+$	81.4	3.3	—	0.74	—	0	—
	Na^+		9.0	90 ± 24	2.0	62 ± 16	6	24
Citrulline	H^+		2.0	71 ± 12	0.44	48 ± 8	9	13
	H^+		3.8	33 ± 7	0.83	22 ± 5	5	23
Cystine	$2\text{Na}^+-\text{H}^+$	7.78	8.3	52 ± 8	1.8	36 ± 5	9	33
Glutamine	$2\text{Na}^+-\text{H}^+$		4.4	503 ± 44	0.97	344 ± 30	12	4
	$2\text{Na}^+-\text{H}^+$ (^{13}C)		94	294 ± 132	21	201 ± 90	4	22
Histidine	H^+		3.8	422 ± 36	0.83	289 ± 24	12	10
	$2\text{Na}^+-\text{H}^+$	72.6	1.8	68 ± 11	0.41	47 ± 8	12	8
Isoleucine	Na^+		6.5	58 ± 9	1.4	40 ± 6	10	21
	H^+	55.5	0.15	65 ± 6	0.03	45 ± 4	8	6
Leucine	H^+ (^{13}C)		2.5	100 ± 10	0.56	69 ± 7	12	24
	H^+	100	0.26	115 ± 11	0.06	79 ± 7	8	6
Lysine	H^+ (^{13}C)		4.5	177 ± 17	0.99	121 ± 12	12	24
	$2\text{Na}^+-\text{H}^+$		17	252 ± 26	3.8	172 ± 18	11	11
Methionine	$2\text{Na}^+-\text{H}^+$	140	6.2	135 ± 18	1.38	93 ± 12	11	15
	H^+		5.4	139 ± 20	1.21	95 ± 14	12	26
Ornithine	H^+	22.5	1.0	29 ± 2	0.22	20 ± 2	12	10
	$2\text{Na}^+-\text{H}^+$		8.3	—	1.84	—	0	—
Phenylalanine	$2\text{Na}^+-\text{H}^+$	52.1	10.6	53 ± 6	2.4	37 ± 4	7	19
	H^+		6.3	82 ± 23	1.4	56 ± 16	4	35
Proline	$2\text{Na}^+-\text{H}^+$	50.8	3.7	93 ± 4	0.82	64 ± 3	11	9
	H^+		0.31	101 ± 8	0.07	69 ± 5	12	13
Serine	H^+ (^{13}C)		34	62 ± 18	7.5	43 ± 12	4	47
	H^+	177	0.65	221 ± 18	0.14	151 ± 13	12	11
Taurine	$2\text{Na}^+-\text{H}^+$		24	190 ± 23	5.3	130 ± 15	12	20
	H^+	95.9	14	—	3.0	—	0	—
Threonine	$2\text{Na}^+-\text{H}^+$		16	55 ± 19	3.5	38 ± 13	8	49
	Na^+		6.1	37 ± 10	1.4	25 ± 7	12	24
Tyrosine	H^+		9.7	24 ± 3	2.2	17 ± 2	9	28
	$2\text{Na}^+-\text{H}^+$	119	14	129 ± 10	3.2	89 ± 7	12	15
Tyrosine	H^+		6.8	131 ± 13	1.5	90 ± 9	12	11
	$2\text{Na}^+-\text{H}^+$	57.3	4.5	91 ± 8	1.0	62 ± 6	12	6
	H^+		1.6	—	0.36	—	0	—

Intensity data for individual adducts were used with RRFs calculated from data in Table 2. Summary data for quantification of AAs by 5 different platforms from McGaw et al (McGaw et al., 2010) are provided for comparison. Four aliquots of NIST were analyzed in triplicate; the number of times that adduct was detected is expressed as "Detected N." The median CV is the variation for triplicate analyses.

Thr>Pro=Arg=Leu/Ile>Tau>His>Lys>Ala>Orn>Gln>Citrulline
>Ser>CySS (Table 2). Importantly, the plots show little evidence for nonlinearity of response between analytical platforms and reasonable extrapolation to the origin (Fig. 1).

Poor correlation ($r < 0.7$) is likely due to combined analytic variation of the 2 methods. The conventional AA analyses were performed only once so reproducibility of analyses for these individual samples is unknown. The LC-MS analyses, which were performed in triplicate, allowed determination of CV for AA detection in individual samples. With the exception of alanine and serine, all had at least 1 detected m/z ion, ie, H^+ adduct, Na^+ adduct, with $\text{CV} < 10\%$ (Table 2). Signals with median $\text{CV} > 10\%$ had lower r , indicating that lower analytic variability of detection by LC-MS was associated with better correlation. For glutamine, the median CV for LC-MS data was small, but correlation was relatively poor, perhaps reflecting a poor reproducibility of conventional AA analysis of glutamine. Correlations of lysine (Lys) and arginine (Arg) within and between platforms further indicate contributions of both platforms to analytic variation. Lys and Arg are transported into and out of plasma by common transport systems, and as

expected, are correlated within the AA analysis ($r = 0.84$) and within the HRM analysis ($r = 0.70$). The similar correlations of Lys ($r = 0.73$) and Arg ($r = 0.84$) between methods (Table 2) indicate that similar extents of analytical variation exist for both platforms. Based upon the median CV values, the LC-MS reliability from best to worst follows the order: Gln>Phe>Tyr>Pro>His>Arg>Thr>Met>Leu/Ile>Lys>CySS>Tau>Orn>Ala>Citrulline>Ser. Importantly, the results show that external standardization against data obtained from an independent, authenticated method can be used for absolute HRM quantification.

Absolute Quantification Based Upon Stable Isotope Internal Standardization

^{15}N , ^{13}C -Met and ^{15}N -Tyr were included in each sample, so that these AAs could be directly determined in the 117 samples relative to their respective internal standard. Results showed correlations between the measurements using internal standardization and the independent measurements by conventional AA analysis (Figs. 2A and 2B), although the absolute value was lower for Met. As for Figure 1, the correlations remained significant after removal of the most extreme values.

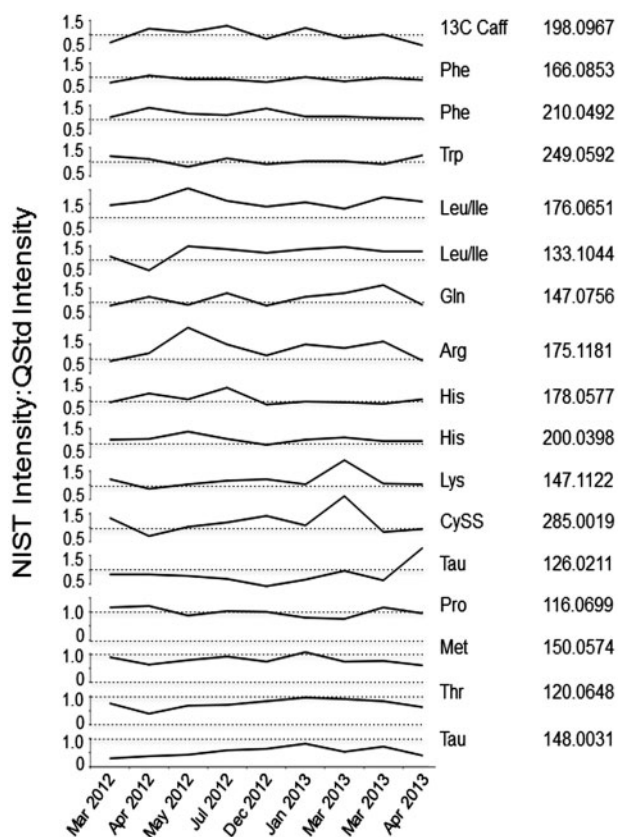


FIG. 3. Comparison of signal intensities for amino acids (AAs) in NIST SRM1950 and Qstd with repeated analyses over 13 months. To evaluate reproducibility of reference standardization over time, AA signals were compared for analyses performed during different time periods. The results confirm that long-term referencing is robust. The results also showed that variability for different AAs occurred with triplicate analyses, providing a basis to change standard operating procedures to perform 6 technical replicates of the reference standards after each batch of 20 samples.

Specifically, the mean value for Met was 29.0 μM (Table 3), compared with 41 μM by conventional AA analysis (Table 2). The respective Tyr values were 62 μM (Table 3) and 59 μM (Table 2). For comparison, values for Met and Tyr in NIST1950 were 22.5 and 57.3 μM , respectively (Table 3) (McGaw et al., 2010).

Absolute Quantification Based Upon Stable Isotope Surrogate Internal Standardization

Surrogate standardization was based upon the methods of Greizerstein et al. (1997). RRFs for ionization and detection of specific AA ions in Table 2 were calculated relative to Met $[+\text{H}]^+$ and Tyr $[+2\text{Na}^+-\text{H}]^+$ (Table 3). Using these values, concentrations of AA in NIST1950 were determined on 4 different weeks with $^{15}\text{N}^{13}\text{C}$ -Met and ^{15}N -Tyr as surrogate standards. Results show quantification for all detected AA except cystine and serine, which had poor CV (Table 3). The m/z for Gly $[+\text{H}]^+$ is less the operational range for m/z (85–850) used for the measurements, and the Na^+ and $2\text{Na}^+-\text{H}^+$ adducts were not detected. H^+ adduct was not detected in the NIST samples for either Ala or Ser. Thus, for the detected AA, these results support previous findings (Greizerstein et al., 1997) indicating that surrogate standardization can be used for absolute quantification.

Use of NIST SRM1950 (NIST1950) as an External Reference for Absolute Quantification

The data in Table 3 provide independent confirmation of the AA measurements of McGaw (McGaw et al., 2010), and additionally suggest that NIST1950 can be used as a reference standard for quantification of AA. The stability of NIST1950 for quantification was verified by comparing the signals of AAs in NIST1950 and ^{13}C -caffeine added on the day of analysis to signals of AAs in pooled reference plasma (QStd) prepared independently and analyzed together over a period of 13 months (Fig. 3). Results showed that signals for AAs in NIST1950 and Qstd over this time period were stable relative to each other and also relative to the added ^{13}C -caffeine internal standard. These results

TABLE 4. Use of NIST-Calibrated QStd for Calculation of AA Concentrations in 117 Human Samples

AA	Adduct	Reference Concentration QStd	Reference Standardization 117 samples	Conventional AA Analysis 117 Samples
Alanine	H^+	295	322 ± 228	321 ± 196
Arginine	H^+	60 ± 8	59 ± 41	73 ± 37
Aspartic Acid	—	—	—	7 ± 5
Citrulline	H^+	33	19 ± 9	19 ± 9
Cystine	H^+	15	52 ± 63	46 ± 27
Glutamic Acid	—	—	—	58 ± 34
Glutamine	$2\text{Na}^+-\text{H}^+$	367	559 ± 225	554 ± 163
Glycine	—	—	—	242 ± 90
Histidine	Na^+	61 ± 9	85 ± 39	71 ± 28
Leucine/Isoleucine	H^+	127 ± 17	148 ± 82	175 ± 60
Lysine	H^+	133 ± 38	123 ± 94	187 ± 75
Methionine	H^+	30 ± 5	45 ± 78	41 ± 47
Ornithine	$2\text{Na}^+-\text{H}^+$	28 ± 4	36 ± 28	82 ± 42
Phenylalanine	H^+	51 ± 15	98 ± 58	108 ± 72
Proline	H^+	191 ± 28	231 ± 134	192 ± 88
Serine	H^+	108	75 ± 42	74 ± 24
Taurine	H^+	37	24 ± 18	34 ± 25
Threonine	$2\text{Na}^+-\text{H}^+$	112 ± 37	141 ± 47	124 ± 67
Tyrosine	$2\text{Na}^+-\text{H}^+$	52 ± 6	49 ± 20	59 ± 24
Valine	—	—	—	226 ± 73

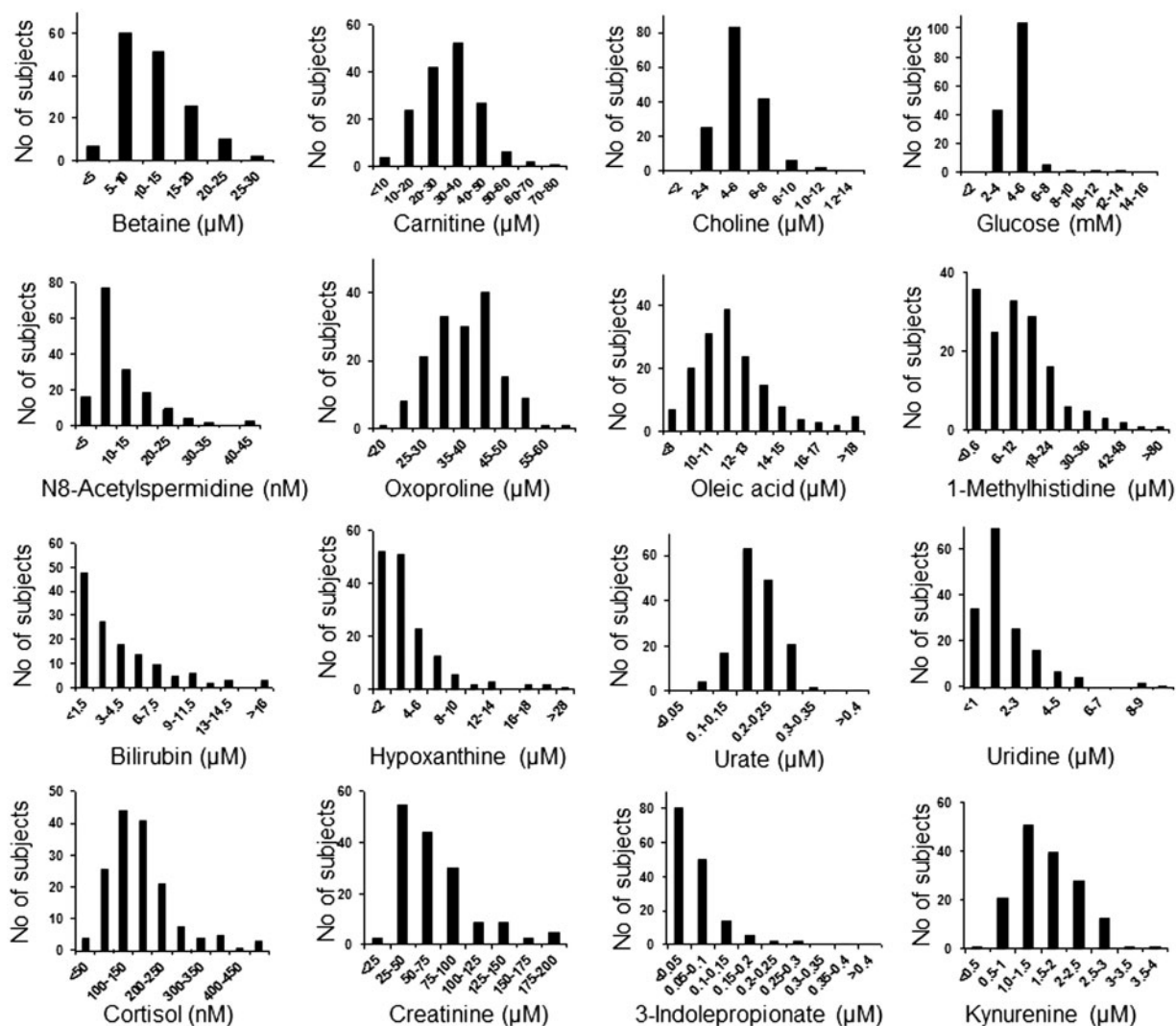


FIG. 4. Histograms of selected metabolite concentrations in 157 healthy adults. Samples were analyzed in triplicate by HRM and quantification was performed using reference standardization.

establish that pooled reference plasma can be calibrated relative to NIST1950 (Table 4), and that such preparations are stable at -80°C for at least 13 months. To test use of Qstd for absolute quantification, AA concentrations in the 117 human serum samples were calculated based upon corresponding AA signals in Qstd, analyzed concurrently. The results were comparable to the direct measurements by AA analysis (Table 4). Consequently, the results demonstrate that "Reference Standardization" can be used as an alternative to external standardization, stable isotope dilution, or surrogate standardization for quantification of HRM data.

General Applicability of Reference Standardization

To test the general applicability of the method, we selected metabolites with confirmed identifications for which concentrations in NIST1950 are available (http://srm1950.nist.gov/srm_search.php?cert=on) and/or for which we had performed independent external method-of-addition calibration of values in the Qstd. Analysis of metabolite concentrations in plasma from 157 healthy adults showed distributions comparable to data in Human Metabolomics Database (HMDB), Version 3.0 (Fig. 4 and Table 5). Thus, the evaluation shows that reference

standardization can be used to convert HRM data to absolute concentrations for metabolites that have been calibrated in NIST1950 or in a comparable pooled reference material.

Application of Reference Standardization to Measurement of Environmental Chemicals

In contrast to most of the endogenous metabolites measured above, environmental chemicals were detected in the nanomolar and subnanomolar concentration ranges (Fig. 5 and Table 6). Of those quantified, octylphenol (57 ± 61 nM) and dipropylphthalate (60 ± 24 nM) were present at highest concentrations. Dibutylphthalate was present in all samples, but calibration of standards was inconsistent, and 6 samples had levels calculated to be in the micromolar range and were excluded because of possible contamination. Styrene, flame retardants (triethylphosphate, triphenylphosphate), and some insecticides and related metabolites were also detected in the nanomolar range (Fig. 5 and Table 6). Pirimicarb was present in most samples at subnanomolar concentration. The 157 healthy individuals were all nonsmokers and extensively screened to assure absence of disease; accordingly, measured values of cotinine (4.6 ± 4.2 nM; median 4.0 nM; Fig. 5) were lower than pooled reference NIST

TABLE 5. Comparison of Metabolite Concentrations in 157 Healthy Adults Obtained Using Reference Standardization to Metabolite Data Summarized in HMDB

Metabolite	Reference Standardization			HMDB	
	Median	Mean	SD	Mean	SD
N8-Acetylspermidine (nM)	9.1	11.0	6.7	50	14
Betaine (μ M)	11	11.7	5.1	33.6	—
Bilirubin (μ M)	2.5	4.1	4.4	8.0	0.9
Carnitine (μ M)	32	31.6	12.1	30	7.6
Choline (μ M)	2.5	5.4	1.6	6.0	0.3
Cortisol (nM)	151	169	89	320	190
Cotinine (nM)	3.9	5.5	4.2	1400	900
Creatinine (μ M)	59	71	36	70	10
Glucose (mM)	4.4	4.6	1.3	5.3	1.2
Hypoxanthine (μ M)	2.9	4.4	5.4	4.87	0.36
3-Indolepropionate (μ M)	0.48	0.067	0.070	0.48	—
Kynurenine (μ M)	1.55	1.63	0.6	1.6	0.1
Oleic acid (μ M)	47.4	49.7	12.9	49	19
Oxoproline (μ M)	38	38.4	13.7	19.5	3.7
1-Methylhistidine (μ M)	12.2	14.9	12.2	12.7	2.9
Urate (μ M)	190	200	51	272	43
Uridine (μ M)	1.64	2.00	1.37	3.12	1.31

The subject characteristics were not well matched between the present study and those for the HMDB data, likely accounting for many of the differences observed. MS/MS data and notes concerning identification of metabolites listed in this table are provided in [Supplementary Data S1](#).

(10 nM) and Qstd (15 nM), but similar to HMDB values for non-smokers. Caffeine, which is directly derived from diet, and hippurate, which is formed by glycine conjugation of dietary benzoic acid, were present with micromolar concentrations consistent with HMDB data.

Application of Reference Standardization to Unidentified Metabolites
HRM detects large numbers of ions that have not been characterized. To determine the suitability of reference standardization to support quantification of currently unidentified metabolites in plasma, including those derived from microbiome, diet, or environmental exposures, we compared high-resolution m/z features detected in NIST1950 with those detected in QStd. Results showed that >8000 ions were detected in at least 50% of replicates for both NIST and QStd across 13 months of analyses. A similar examination of data for the 117 plasma samples showed that >14 000 of the 22629 metabolites detected were also detected at least 50% of the time in QStd; respective values for the 157 healthy subjects were >10 000 out of 16665 metabolites. These data suggest that as chemicals are identified and quantified in NIST SRM1590, reference standardization will support quantification of thousands of chemicals using HRM methods.

DISCUSSION

Exposome research to study links between disease and exposure history across a lifespan requires new analytic methods to

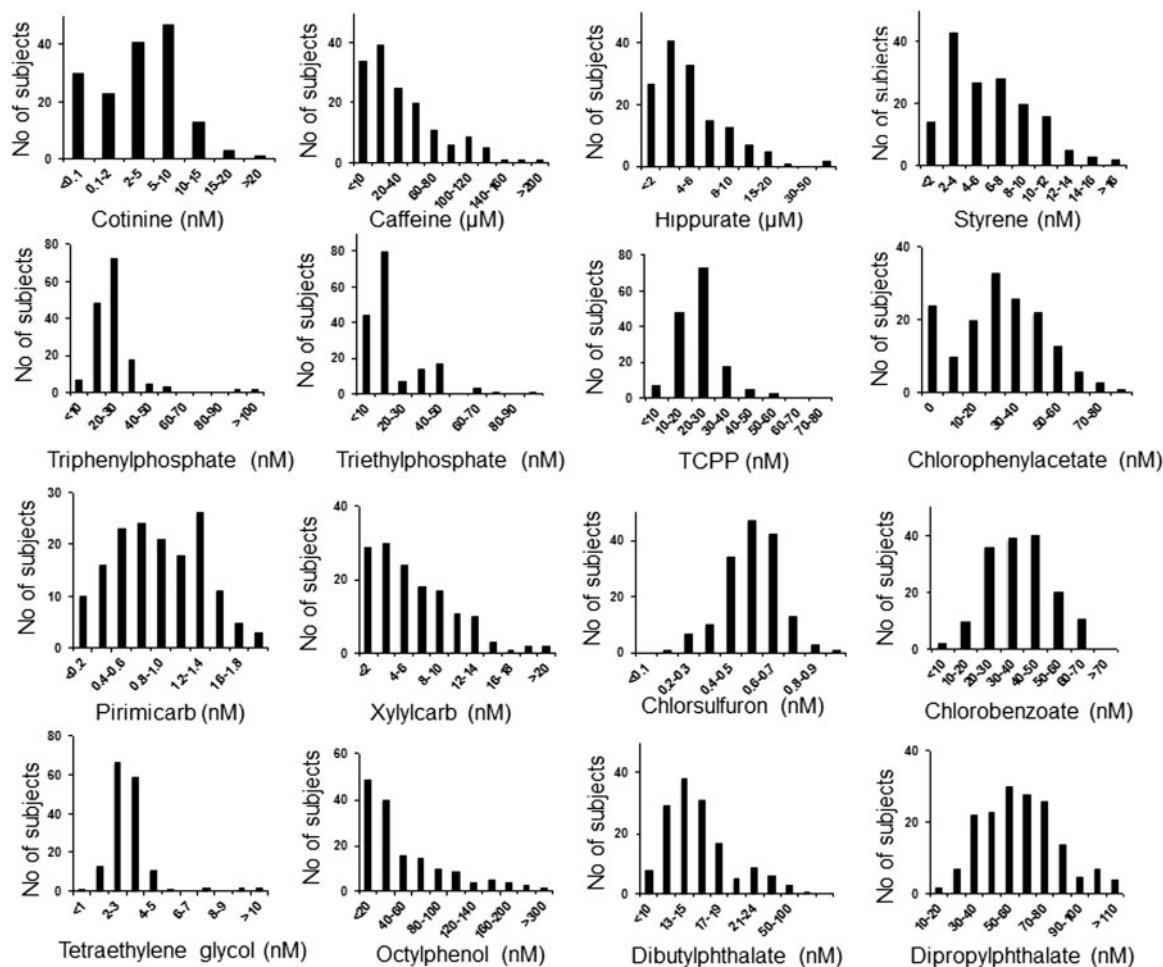


FIG. 5. Histograms of selected environmental and dietary chemicals in 157 healthy adults. Samples were analyzed in triplicate by HRM and quantification was performed using reference standardization.

TABLE 6. Environmental Chemicals Concentrations in Plasma of 157 Healthy Adults Obtained Using Reference Standardization

Metabolite	Reference Standardization			Literature Values	
	Median	Mean	SD	Mean or Range	References
Caffeine (μM)	22	36	46	26–129	HMDB
Chlorobenzoic acid (nM)	38	38	13	—	—
Chlorophenylacetic acid (nM)	28	34	18	—	—
Chlorsulfuron (nM)	0.56	0.55	0.13	—	—
Cotinine (nM)	3.9	5.5	4.2	6.7–13.5	HMDB
Dibutylphthalate (nM)	20.5	25.4	16.4	15–989	ToxNet
Dipropylphthalate (nM)	57	60	24	—	—
Hippuric acid (μM)	4.7	7.4	11.7	17 \pm 11	HMDB
Octylphenol (nM)	34	57	61	0.14–2.2 ^a	(Qin et al., 2013)
Pirimicarb (nM)	0.82	0.79	0.36	—	—
Styrene (nM)	5.3	6.0	3.6	0.4–5.3	HMDB
Tetraethylene glycol (nM)	2.9	3.2	1.6	—	—
Triethylphosphate (nM)	6.9	10.0	8.1	—	—
Triphenylphosphate (nM)	22	25	12	0.6 nM	(Shah et al., 2006)
Tris(2-chloropropyl)phosphate (nM)	45	51	23	—	—
Xylylcarb (nM)	4.6	6.4	8.2	—	—

Previous literature values are provided for serum or plasma where available. For dibutylphthalate, 6 samples were in the micromolar range and were excluded because of possible contamination. Additionally, standardization was inconsistent for dibutylphthalate. ToxNet, <http://toxnet.nlm.nih.gov> (Shah et al., 2006). MS/MS data and notes concerning identification of metabolites listed in this table are provided in [Supplementary Data S1](#).

^a Value for urinary 4-n-octylphenol, expressed as nmole/l urine

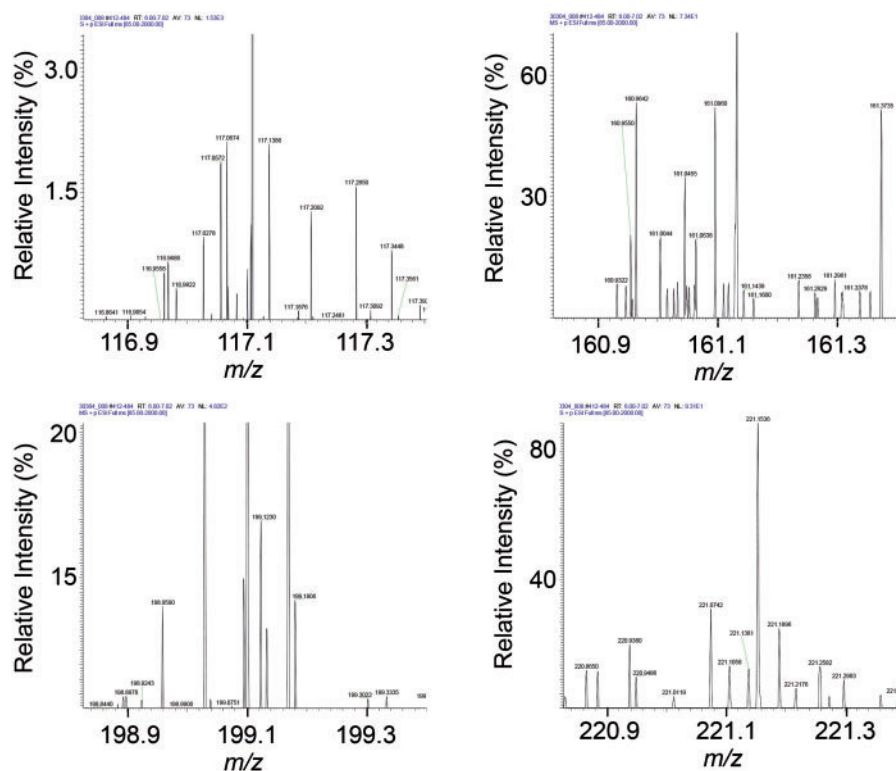


FIG. 6. Representative ultra-high resolution MS¹ scans of human plasma show that many low abundance ions are present within a 0.5 atomic mass unit (AMU) window. Four examples of mass spectra are given and are representative of nominal mass <1000.

address both the range of exposures and the time frame between exposure and outcome. Human biomonitoring with HRM provides means to capture chemical exposure information in terms of measured body burden and simultaneously measure metabolic responses to current or prior exposures (Park et al., 2012; Soltow et al., 2013). This study shows that multiple

standardization methods can be applied to HRM data to create cumulative exposome research databases (Table 1). The most salient finding is that reference standardization, a simple modification of analytic workflows commonly used for metabolomics, provides a framework for creation of cumulative exposome reference databases. The results demonstrate that by

calibrating the concentrations of chemicals in a pooled reference sample, data can be converted to absolute concentrations for direct comparisons to data collected with different platforms and/or at different times. In practical use, the method is critically dependent upon detection of low abundance chemicals in the reference standard. We have adopted standard operating procedures to analyze the reference standard 12 times daily, 6 preceding and 6 following batches of 20 samples, to facilitate quantification of metabolites that are inconsistently detected upon repeat analysis. This strategy overcomes limitations of external calibration and internal standardization to measure thousands of chemicals in a routine and affordable manner. At the same time, environmental chemicals that are present in only a small fraction of people are diluted in pooled reference samples. Thus, for optimization of the approach, pooled reference samples with known, relevant concentrations of environmental chemicals will be needed.

External calibration with a method of additions is a labor-intensive approach that is useful for determination of environmental chemical concentrations in samples but requires a suitable quality standard and care in determination. In the current evaluation, we used additions of relevant small amounts of standards into a constant small volume of diluent added to the Qstd. At least 3 different amounts were used to increase target signal from 1.3- to 20-fold. This approach allows for extrapolation back to signal obtained with only the diluent to obtain an estimate of the concentration in Qstd. For exposome research, this methodology is useful for calibrating the Reference Standard.

Internal standardization is the "gold standard" for targeted analyses by LC-MS and provides quantitatively accurate data. However, this approach is impractical for use to measure thousands of chemicals in a high-throughput format because of the cost of large numbers of stable isotopic standards (Table 1). Nonetheless, inclusion of a small number of stable isotopic standards in every sample provides means for quality control and for surrogate standardization of chemicals that are unstable in the reference standard. Additionally, internal standardization provides a reliable means to calibrate the Reference Standard.

Surrogate standardization provides a useful alternative to reference standardization for absolute quantification. Surrogate standardization requires concentrations of known metabolites to obtain RRFs, but this can be done retrospectively in the same manner as suggested above for reference standardization. Both reference standardization and surrogate standardization are expected to lose reliability for absolute quantification for values that deviate substantially from the mean. Additionally, surrogate standardization is limited when the RRF is calculated from a regression curve with a nonzero intercept (data not shown).

Whether reference standardization is better than surrogate standardization is not clearly evident for the metabolites analyzed.

The present analysis of quantification strategies shows that reference standardization, with only a modest change in analytic workflow for nontargeted metabolomics, can enable use of HRM for cumulative exposome research. Nontargeted metabolomic work flows typically include a pooled reference sample to provide a means to ensure analytical quality control and batch correction (Bijlsma *et al.*, 2006; Dunn *et al.*, 2011; Masson *et al.*, 2010; O'Kane *et al.*, 2013; Want *et al.*, 2010). The resulting data demonstrate that by calibrating this pooled reference sample to a calibration reference, such as NIST SRM1950, one creates an analytic structure that is simple and quantitatively reliable. Although validated for only a small number of chemicals, these

data suggest suitability for use with thousands of chemicals. In principle, reference standardization also provides a means for batch correction by calibrating each chemical within each sample to the same chemical analyzed concurrently in the reference material. At present, the number of quantifiable analytes based upon NIST is approximately 100 chemicals, ie, too small to make this highly useful for exposome research. However, quantification using traditional methods of addition with authentic standards (Bueschl *et al.*, 2013; Hewavitharana, 2011; Niessen *et al.*, 2006) can readily expand this capability. Additionally, availability of certified reference materials with relevant low environmental concentrations of chemicals of interest would facilitate interlaboratory comparisons of exposome research for various populations, analytical platforms and phenotypes.

These analytic procedures do not address sample quality, which is largely determined by patient characteristics, sample collection procedures, and sample storage prior to delivery to an analytical facility. However, the information-rich nature of HRM data allows retrospective evaluation of sample quality in databases. For instance, signals for EDTA, citrate, and lithium adducts directly show whether a plasma sample was collected as described in the metadata. Loss of cystine and presence of oxidation products (cystine sulfoxide and cystine sulfone) may indicate adverse or prolonged storage of samples.

The results presented here establish feasibility for use of reference standardization to build cumulative databases. This approach needs to be validated by application to multiple datasets and datasets from multiple analytic facilities. At least 3 conditions are needed to assure reliability: authenticated chemical identification and/or accurate *m/z* and calibrated LC-RT; quantitative accuracy of reference material calibration; and similarity of instrument response characteristics over a similar dynamic range. In this regard, multicenter cooperation and blinded cross-validation will be needed to assure integrity of data.

This study was performed using ultra-high resolution MS and comparable studies will be needed to test suitability of the approach for other mass spectrometers and other analytic approaches. Analyses were performed using an analytic structure optimized for 10-min analyses of plasma (Johnson *et al.*, 2010; Soltow *et al.*, 2013); the LTQ-Velos-Orbitrap was operated at 60 000 resolution, which is not the maximal resolution of this or other Fourier transform instruments (Marshall and Hendrickson, 2008). Commonly used Q-TOF MS operate at 35 000 resolution, and similar quantitative reliability using a reference standardization protocol can be expected, at least for relatively high abundance chemicals. For lower abundance chemicals, which include many environmental chemicals, quantification may be more problematic due to the presence of multiple low-intensity ions within the ion selection window for ion dissociation (Fig. 6).

Cross-laboratory comparisons of AA measurements in NIST1950 show that differences occur among values obtained from competent analytic laboratories (McGaw *et al.*, 2010). This implies that regardless of interlaboratory comparisons, computational methods will be required to normalize metabolomics data entered into cumulative databases. Such integration was extensively addressed for gene expression microarray data, and this experience suggests that despite analytic hurdles, computational methods can be developed to manage such differences in metabolomics research. Consequently, one can anticipate that reference standardization, perhaps with a concurrent surrogate standardization structure, will facilitate development of cumulative exposome databases. The resulting databases will

serve to foster exposome-related metabolomics research for detailed examination of cumulative environmental as well as dietary, microbiome, and therapeutic influences on health and disease.

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SUPPLEMENTARY DATA

Supplementary data are available online at <http://toxsci.oxfordjournals.org/>.

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