Versatile synthesis and biological evaluation of novel 3'-fluorinated purine nucleosides

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Full Research Paper

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Keywords:

anticancer; 3'-fluororibonucleoside; purine nucleoside; 6-substituted purine; synthesis

Beilstein J. Org. Chem. **2015**, *11*, 2509–2520. doi:10.3762/bjoc.11.272

Received: 30 July 2015 Accepted: 26 November 2015 Published: 09 December 2015

Associate Editor: V. Gouverneur

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Abstract

A unified synthetic strategy accessing novel 3'-fluorinated purine nucleoside derivatives and their biological evaluation were achieved. Novel 3'-fluorinated analogues were constructed from a common 3'-deoxy-3'-fluororibofuranose intermediate. Employing Suzuki and Stille cross-coupling reactions, fifteen 3'-fluororibose purine nucleosides 1–15 and eight 3'-fluororibose 2-chloro/2-aminopurine nucleosides 16–23 with various substituents at position 6 of the purine ring were efficiently synthesized. Furthermore, 3'-fluorine analogs of natural products nebularine and 6-methylpurine riboside were constructed via our convergent synthetic strategy. Synthesized nucleosides were tested against HT116 (colon cancer) and 143B (osteosarcoma cancer) tumor cell lines. We have demonstrated 3'-fluorine purine nucleoside analogues display potent tumor cell growth inhibition activity at sub- or low micromolar concentration.

Introduction

Antimetabolites are extremely useful for the treatment of cancers and viral infections and are one of the largest classes of drugs. Most antimetabolite drugs are nucleoside derivatives that substitute for endogenous nucleosides and prevent DNA and protein replication [1]. Many of the drugs described by the World Health Organization as "essential medicines" are nucleo-

side derivatives [2] and nearly 20% of all drugs for the treatment of cancers are nucleoside derivatives [3]. The design of new antimetabolites is an active field of research and several nucleoside derivatives have recently come to market such as gemcitabine, capecitabine, and decitabine [4-6]. Purine nucleoside analogues such as fludarabine, nelarabine, cladribine, and

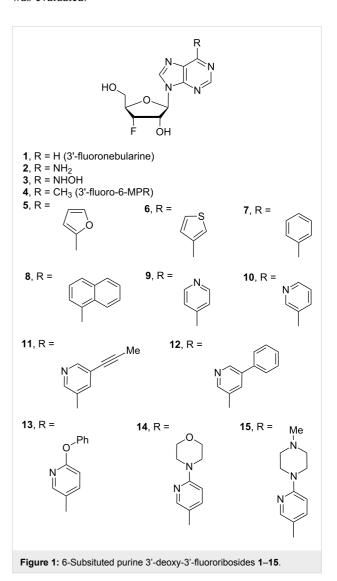
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clofarabine are an important subset of nucleoside drugs [7-10]. Purine ribonucleosides substituted at position 6 have exhibited potent antimetabolite activity [11-13] and aryl or heterocyclic substituents have imparted cytostatic activities against various tumor cell-lines [14-16]. Moreover, some 6-heterocyclic substituted purine ribonucleosides also demonstrate strong antiviral activities [17]. Purine derivatives such as, 2'-β-C-methyl-6-substituted purine nucleosides exhibit promising anti-HCV activity by blocking RNA dependent RNA polymerase [18-20]. Design and synthesis of purine-based nucleosides are still needed to enable new therapies for the treatment of drug-resistant tumors and viruses.

One of the first antimetabolite drugs rationally designed from biochemical data was 5-fluorouracil. A hydrogen atom on uracil was replaced with a fluorine atom for specific reasons. The fluorine atom possesses unique characteristics; it enhances the lipophilicity of organic compounds and C-F bonds have low chemical reactivity imparting high enzymatic stability and resistance to metabolic processes. The high electronegativity of fluorine and the lipophilicity it imparts improve the bioavailability of fluorine-containing drugs. Relative to the unfluorinated derivative, fluorinated drugs have demonstrated favorable pharmacological, physicochemical, pharmacokinetic, pharmacodynamic and safety profiles for a number of compounds [21-23]. Several blockbuster drugs such as Lipitor[®], Seretide[®], Crestor®, Takepron®, Sustiva®, Celebrex®, and the recently described fluorapacin and azvudine [24-29], all contain fluorine atom(s). Not surprisingly, 20% of marketed drugs contain fluorine atom(s). Gemcitabine, the 2'-deoxy-2',2'-difluorocytidine has been routinely utilized to treat solid tumors [30]. However, 3'-fluorine-modified nucleosides have not been wellstudied because of the challenges associated with the synthesis of modified carbohydrate moieties [31-35]. As a result, 3'-fluoro-6-heterocyclic-substituted purine nucleosides are not well represented in the literature. We chose to explore the biological potential of synthetic purine analogues combining substitutions at position 6 and a 3'-fluorine. Novel 6-heterocyclic substituted purine 3'-deoxy-3'-fluororibonucleosides were designed to discover more selective and potent novel antiviral and anticancer therapeutics. Various fluorine-modified ribonucleoside derivatives were designed, synthesized, and tested. The preliminary results are presented herein.

We utilized the structural characteristics of 3'-fluorine and 6-substituted purine nucleosides to expand their biological application. Herein, we report the synthesis of twenty three 3'-deoxy-3'-fluoro-6-modified purine nucleoside derivatives, of which twenty one are novel compounds, including 6-substituted purine 3'-deoxy-3'-fluororibosides 1–15 (Figure 1), 2-chloro-6-substituted purine 3'-deoxy-3'-fluororibosides

16–20 and 2-amino-6-substituted purine 3'-deoxy-3'-fluororiboses **21–23** (Figure 2). In addition, their anticancer activity was evaluated.



Results and Discussion

Current strategies to the synthesis of 3'-fluorine and 6-substituted purine require harsh conditions and laborious protecting group manipulation that results in low product yields. It was reported that 3'-deoxy-3'-fluoroadenosine (2) (Figure 1) was synthesized in 3.5% yield starting from adenosine [31,32]. This eight-step synthesis required harsh reaction conditions and HPLC purification of the final product. In addition, the product cannot be utilized for further derivatization to reach our objectives. This 3'-fluorine-modified adenosine 2 has also been synthesized starting from a well-protected xylofuranosyladenine derivative using a complicated strategy [33]. Similarly, 3'-deoxy-3'-fluoroguanosine (23, Figure 2) was isolated in 2% yield after protecting group manipulation from arabinoguano-

Figure 2: 2-Chloro- and 2-aminopurine 3'-deoxy-3'-fluororibosides 16–23.

sine [34]. Complicated orthogonally protected adenosine and guanosine derivatives with three or four different protecting groups have also been used for the synthesis of compounds 2 and 23, and the protocols required extensive manipulation of the protecting groups [35]. Compound 2 has also been synthesized starting from xylofuranoside by manipulating the protecting groups on the carbohydrate moiety [36]. De Clercq and co-workers [37] developed a protocol for the synthesis of 3'-fluororibofuranose in 10 steps, and it requires epoxide formation and ring opening as well as reversion of the hydroxy group on the sugar ring. Jeong and co-workers [38-40] synthesized fluorine-substituted ribofuranose but isolation required a challenging separation. Therefore, the synthetic challenges for preparing 3'-fluorine modified sugars and nucleosides have limited the synthesis and biological testing of these promising fluorine modified nucleoside derivatives. To open up research and therapeutic exploration of this class of compounds, we developed efficient synthetic routes for the construction of 3'-deoxy-3'-fluoroadenosine and guanosine derivatives with a number of different modifications on position 6.

Substituted purine 3'-deoxy-3'-fluoro-β-D-ribofuranosyl nucleosides were successfully constructed from the universal intermediate 1',2'-di-O-acetyl-5'-O-p-toluyl-3'-fluoro-3'-deoxy-β-D-ribofuranose (25, Figure 3). The final 3'-deoxy-3'-fluoro-β-D-ribofuranose derivatives 1–23 (Figure 1 and Figure 2) were derivatized from the corresponding intermediates 26, 29, 30–40, 42, 48 and 51. All of these compounds are novel derivatives except compounds 2 and 23. The syntheses of these compounds are outlined in Schemes 1–5. 1,2-O-Iso-

propylidene-5-O-(4-methylbenzoyl)- α -D-xylofuranose (24) was synthesized from D-xylose according to literature procedures [41]. Compound 24 was treated with iodine in methanol, and fluorinated with diethylaminosulfur trifluoride (DAST). The universal intermediate 1',2'-di-O-acetyl-5'-O-p-toluyl-3'fluoro-3'-deoxy-β-D-ribofuranose (25) was then obtained in 33.13% overall yield after further treatment with acetic anhydride-acetic acid-sulfuric acid system. More than 200 g scale was achieved for the synthesis, and high purity (98%) product was obtained. This key intermediate 25 can be used to synthesize a variety of 3'-fluoro-modified nucleoside derivatives, and it was utilized for the synthesis of all analogues reported herein. This strategy avoids tedious orthogonal protecting group manipulations previously reported in literature [31-35]. Our strategy provides the desired nucleoside intermediates and also opens up the opportunity for modification on any class of nucleosides with a 3'-fluorine atom to explore their biological and therapeutic potential. While this work is related with purine nucleosides, the strategy can be used for the synthesis of a variety of nucleosides with a wide range of heterocyclic moieties to investigate the impact of a 3'-fluorine atom on the biological activity of nucleosides.

The 6-chloropurine was silylated with trimethylsilyl triflate and then glycosylated with the universal intermediate 3'-fluoro-3'deoxy-β-D-ribofuranose 25 using 1,8-diazabicyclo[5.4.0]undec-7-ene (DBU) to provide the desired protected key intermediate 26 in 90% yield (Scheme 1). To construct the first series of fluorinated purine analogues, compound 26 was treated with a saturated solution of ammonia in methanol, which resulted in the amination at the 6-position and deprotection of the protecting groups to furnish 3'-deoxy-3'-fluoroadenosine (2) in 85% yield. Our synthetic strategy provided compound 2 in excellent yield (76%, 2 steps) compared to previously reported literature protocols (3.5%, 8 steps) [31,32]. The 6-chlorine of compound 26 was replaced by hydroxylamine, with concomitant removal of the protecting groups to yield N^6 -hydroxy-3'fluoro-3'-deoxyadenosine (3). Hydrogenation of compound 26 under hydrogen pressure (50 psi) over 10% Pd/C resulted in the de-chlorinated compound 27, which was further deprotected in a saturated solution of ammonia in methanol providing the desired 6-deaminated 3'-fluoro-adenosine 1 in 93% yield. We targeted 9-(3-deoxy-3-fluoro-β-D-ribofuranosyl)purine (1) in particular because it is the 3'-fluorine analogue of nebularine, a naturally occurring antibacterial and antineoplastic agent [42,43].

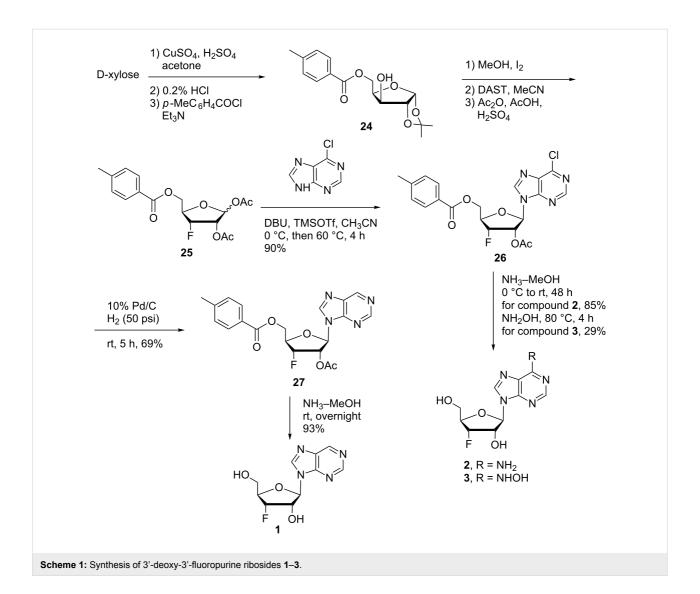
3'-Fluoro-6-methylpurine riboside $\bf 4$, a 6- β -D-MPR mimic

6-Methylpurine- β -D-riboside (6- β -D-MPR) is an isolated antibiotic agent that possesses potent antifungal, antiviral, and

antitumor activities [44]. In order to explore the effect of fluorine on the biological activity of this pharmacophore, we synthesized 6-methylpurine-3'-deoxy-3'-fluoro-β-D-riboside (4) (Scheme 2). 6-Methylpurine (28) was synthesized from 6-chloropurine according to the reported protocol [44]. Compound 28 was silylated with BSA and glycosylated with the 3'-fluoro-sugar 25 to provide the desired compound 29 in 78% yield. Subsequent deprotection furnished the targeted novel fluorine modified 6-methylpurine riboside 4 in 80% yield.

3'-Fluoro-6-C-substituted purine nucleosides

The 6-C-substituted purine nucleosides have demonstrated anticancer, antiviral and other biological activities [11-20]. In order to explore the effect of fluorine on the biological activity of 6-C-substituted purine nucleosides, we designed and synthesized 3'-deoxy-3'-fluoro- β -D-ribofuranosyl purine derivatives 5–15 with various aromatic and heterocyclic moieties at position 6 of the purine base from the chlorine intermediate 26 (Scheme 3).



The synthesis of 3'-deoxy-3'-fluoro-β-D-ribofuranosyl purine derivatives began with a palladium-catalyzed cross coupling [45,46] to install the aromatic moieties to the 6-position of the purine ring. In order to accomplish efficient cross-couplings of a wide range of aromatic rings, we utilized three different protocols that were employed for Stille and Suzuki reaction conditions. To this end, 2-(tributylstannyl)furan was coupled with 6-chloropurine nucleoside 26 by Stille cross coupling [47,48] catalyzed by bis(triphenylphosphine)palladium(II) chloride in DMF (Method I) (Scheme 3). The resulting 6-aryl compound 30 was obtained in 91% yield and deprotected with a saturated solution of ammonia in methanol to give the desired analogue 5 with furan-2-yl substituent at position 6 of the purine base.

Considering the possible toxicity of organostannyl reagents, we then utilized various organoboronic acids for Suzuki cross couplings [49-51] to synthesize compounds 6–15. The 6-chloropurine intermediate 26 was coupled with 3-thienyl-boronic acid and phenylboronic acid catalyzed by tetrakis(triphenylphosphine)palladium (Pd(PPh₃)₄) in toluene (Method II), resulting in the desired protected intermediates 31 and 32 in 56% and 52% yields, respectively. Suzuki coupling efficiency in DMF (entries 2 and 3, Table 1) was not as high as Stille coupling (entry 1, Table 1). Suzuki coupling of intermediate 26 with 1-naphthylboronic acid, pyridine-4-boronic acid, pyridine-3-boronic acid and 5-propynylpyridine-3-boronic acid (Method II) (entries 4, 6, 8, and 10, Table 1) provided no detectable

product. After screening various solvent systems, we discovered that dimethoxyethane (DME)—water was the optimal solvent system (Method III), and achieved construction of 6-aryl compounds **33–36** (entries 5, 7, 9 and 11, Table 1), albeit

in low to moderate yields. Next, substituted pyridine boronic acids [52-54] were coupled with intermediate 26 using Method III in DME-water to provide the desired 6-aryl products 37-40 (entries 12-15, Table 1). These results indicated that the

ntry	Method ^a	Reagents	Coupling product (yield)	Final product (yield)
	I	\bigcirc SnBu ₃	30 (91%)	5 (70%)
	II	S B(OH) ₂	31 (56%)	6 (92%)
	II	$B(OH)_2$	32 (52%)	7 (84%)
	II		33 (no product)	-
	III	B(OH) ₂	33 (57%)	8 (92%)
	II	.//	34 (no product)	_
	III	N'B(OH) ₂	34 (32%)	9 (89%)
	II	7	35 (no product)	_
	III	$N \longrightarrow B(OH)_2$	35 (29%)	10 (84%)
	II	B(OH) ₂	36 (no product)	_
	III	Me————————————————————————————————————	36 (17%)	11 (70%)
	Ш	B(OH) ₂	37 (21%)	12 (77%)
	III	$ \begin{array}{c} Ph \\ O \longrightarrow \\ N = \\ \end{array} $ $ B(OH)_2$	38 (46%)	13 (87%)
	III	$O N - N - B(OH)_2$	39 (49%)	14 (90%)
	III	$Me-N$ $N N=$ $B(OH)_2$	40 (16%)	15 (72%)
	1	\bigcirc SnBu $_3$	43 (24%)	17 (85%)
	II	S B(OH) ₂	44 (16%)	18 (70%)
	II	B(OH) ₂	45 (21%)	19 (72%)
	III	B(OH) ₂	46 (22%)	20 (71%)

DME—water solvent system was more favourable for more challenging Suzuki C–C coupling reactions on the 6-chloropurine nucleosides. Intermediate compounds **31–40** were deprotected with a saturated solution of ammonia in methanol to furnish the desired products **6–15** in 70–92% yields (Table 1).

3'-Fluoro-2-chloropurine nucleosides – cladribine and clofarabine mimics

Anticancer drugs cladribine and clofarabine [55-57] are purine nucleoside derivatives having a chlorine atom at position 2 of the purine moiety. With this in mind, we designed and synthesized 2-chloropurine nucleosides 16–20 (Figure 2 and Scheme 4) with 3'-fluorine and 6-C-aromatic or heterocyclic modifications to explore a broad biological space for this class of nucleoside derivatives. The 2,6-dichloropurine (41) was glycosylated with the 3'-fluororibose intermediate 25 to furnish the 2,6-dichloropurine intermediate 42 in 89% yield (Scheme 4). Chemoselective amination of the 6-position over the 2-position of the purine and deprotection of 2,6-dichloropurine 42 was achieved with a saturated solution of ammonia in methanol to furnish 2-chloro-3'-deoxy-3'-fluoroadenosine (16). The 2,6-dichloro-intermediate 42 was coupled with 2-(tributyl-

stannyl)furan catalyzed by bis(triphenylphosphine)palladium(II) chloride in DMF to provide monochloro-intermediate 43 (Method I, entry 16, Table 1). Monochloro-intermediates 44 and 45 were constructed by Suzuki coupling of dichloro-intermediate 42 with 3-thienylboronic acid and phenylboronic acid in toluene (Method II; entries 17 and 18, Table 1), and monochloro-intermediate 46 was synthesized by coupling of 42 with 1-naphthylboronic acid in DME-water (Method III; entry 19, Table 1). The resulted coupling products 43–46 were deprotected with a saturated solution of ammonia in methanol to provide corresponding final products 17–20 in 70–90% yields.

Of particular importance, the 2-chlorine atom on the intermediate 42 was stable under the Stille and Suzuki reaction conditions and under ammonia deprotection conditions. However, the palladium-catalyzed cross coupling of dichloro-intermediate 42 with organostannane and organoboronic acid reagents resulted in lower yields when compared to the cross coupling of monochloro-intermediate 26. From the amination studies of 2,6-dichloropurines, the 6-position of the purine possesses higher reactivity towards nucleophiles than the 2-position. In addition, the selectivity for the 6-position is also higher for the Stille than

for the Suzuki cross coupling. The lower yields obtained from the cross-coupling of 2,6-dichloropurines is most likely contributed to the 2-chlorine reducing the reactivity of the 6-chlorine, or undesired cross-coupled products at the 2-position of the purine that may have resulted, but were not isolated.

3'-Fluoro-2-aminopurine nucleosides

As mentioned above, chloropurine [55-57] and deaminopurine [42,43] nucleosides have exhibited potent biological activities. We synthesized 2-amino-6-chloropurine 3'-deoxy-3'-fluororiboside 22 and 2-aminopurine-3'-deoxy-3'-fluororiboside 21 to explore the SAR of 3'-fluorine-substituted purine nucleoside derivatives (Scheme 5). The 2-amino-6-chloropurine (47) was glycosylated with the 3'-fluorine riboside 25 under DBU and trimethylsilyl triflate conditions. The corresponding protected intermediate nucleoside 48 was obtained in 78% yield and utilized for further derivatization. Deprotection of compound 48 with a saturated solution of ammonia in methanol at 0 °C resulted in the desired product 22 in 85% yield. In addition, the 6-chlorine atom was left untouched during the deprotection. Compound 48 was hydrogenated over 10% Pd/C under

hydrogen pressure (50 psi) giving the de-chlorinated compound 49, which was deprotected under ammonia treatment to complete the desired final product 21. In order to further explore the SAR of this pharmacophore, we constructed 3'-deoxy-3'-fluoroguanosine (23, Scheme 5). N²-Acetyl-6-O-(diphenylcarbamoyl)guanine (50) was prepared according to reported protocol [58,59]. Compound 50 was silylated and glycosylated with 3'-fluorine riboside 25 providing the fully protected intermediate 51 in 72% yield. The final product 23 was obtained in 96% yield after global deprotection of 51 using a saturated solution of ammonia in methanol. This protocol of direct glycosylation for the synthesis of 3'-fluorine modified guanosine derivatives is highly advantageous compared to the previously reported methods utilizing orthogonal protecting groups, selective deprotection, and fluorination of the starting guanosine [34].

Biological evaluation

Newly synthesized compounds were tested against the human colon cancer cell line HCT-116 and the human osteosarcoma cancer cell line 143B (Table 2). The fluorinated nebularine

Compound	Cytotoxicity	Cytotoxicity IC ₅₀ (µM)		Cytotoxicity IC ₅₀ (µM)	
	HCT116	143B	Compound	HCT116	143B
1	0.5–1.0	0.5–1.0	22	0.5–1.0	0.5–1.0
2	1.0-5.0	1.0-5.0	27	0.5-1.0	0.5-1.0
3	5.0-10	5.0-10	29	0.5-1.0	0.5-1.0
4	5.0-10	5.0-10	30	1.0-5.0	1.0-5.0
5	>10	>10	31	>10	>10
6	>10	>10	32	0.5-1.0	0.5-1.0
7	>20	>20	33	>20	>30
8	>20	>20	34	>10	>10
9	>10	>10	35	>10	>10
10	>10	>10	36	>20	>20
11	>20	>20	37	>20	>20
12	1.0-5.0	1.0-5.0	38	>10	>10
13	>10	>10	39	0.5-1.0	0.5-1.0
14	>10	>10	40	0.5-5.0	0.5-5.0
15	>20	>20	43	>10	>10
16	>10	>10	44	>10	>10
17	>10	>10	45	>20	>20
18	>10	>10	46	>20	>20
19	>20	>20	49	1.0-5.0	1.0-5.0
20	>20	>20	Camptothecin	0.30	1.20

analog 1 and its protected analog 27 demonstrated potent tumor cell growth inhibition at sub-micromolar concentration. The protected 6-methylpurine riboside 29 exhibited better antitumor activity than its deprotected analog 4, the 3'-fluoro-6-MPR analog. The 6-chlorine compound 22, as well as protected purine nucleosides 32, 39 and 40, also showed potent inhibitory activity. 3'-Fluorine purine nucleosides 2–4, 12, 30, and 49 showed moderate levels of inhibitory activity, but other derivatives did not show detectable activity against the tested tumor cell lines. Antiviral and other biological evaluation of these 3'-fluorine modified nucleosides is in progress and will be reported in due course.

Conclusion

The 6-chloropurine, 2,6-dichloropurine, and 2-amino-6-chloropurine were glycosylated with the protected 3'-deoxy-3'-fluororibose **25** to provide the corresponding key intermediates **26**, **42**, and **48**. These intermediates were then further derivatized to furnish final products **1–3** and **5–23**. The glycosylation of 6-methylpurine **28** with **25** furnished 3'-deoxy-3'-fluoro-6-methylpurine riboside **4**, the analogue of biologically active natural product, 6-methylpurine-β-D-riboside (6-β-D-MPR). We constructed 3'-deoxy-3'-fluororibofuranosylpurine nucleosides **5–11** with various aromatic and heterocyclic moieties at position 6 from the key intermediate **26** by Stille and Suzuki cross-coupling reactions. 3'-Deoxy-3'-fluororibofuranosyl 2-chloropurine nucleosides **17–20** were synthesized by the

similar strategy from the key intermediate 42. DME—water was found to be an efficient solvent system for Suzuki reaction on challenging substrates. De-chlorination of the 6-chloro-intermediates 26 and 48, followed by deprotection, resulted in 6-deaminopurine nucleosides 1 and 21, new analogues of the naturally isolated antibacterial and antineoplastic agent 6-deaminoadenosine, nebularine. Newly synthesized compounds were evaluated for their antitumor activity. Eleven compounds exhibited potent tumour cell growth inhibition activity against the human colon tumor cell line HCT116 and the human osteosarcoma cell line 143B at sub- or low micromolar concentration.

Supporting Information

Supporting Information File 1

Experimental procedures, characterization data, and ¹H NMR and mass spectral data for new compounds. [http://www.beilstein-journals.org/bjoc/content/supplementary/1860-5397-11-272-S1.pdf]

References

 Henry, J. R.; Mader, M. M. Recent Advances in Antimetabolite Cancer Chemotherapies. In *Annual Reports in Medicinal Chemistry*; Doherty, A. M., Ed.; Elsevier: San Diego, CA, U.S.A., 2004; Vol. 39, pp 159–172. doi:10.1016/s0065-7743(04)39013-5

- 19th WHO Model List of Essential Medicines. WHO, 2015;
 http://www.who.int/medicines/publications/essentialmedicines/EML201
 5_8-May-15.pdf (accessed July 8, 2015).
- Parker, W. B. Chem. Rev. 2009, 109, 2880–2893. doi:10.1021/cr900028p
- Johnson, S. A.; Thomas, W. Hematol. Oncol. 2000, 18, 141–153. doi:10.1002/1099-1069(200012)18:4<141::AID-HON666>3.0.CO;2-#
- Johnson, S. A. Expert Opin. Pharmacother. 2001, 2, 929–943. doi:10.1517/14656566.2.6.929
- Parker, W. B.; Secrist, J. A., III; Waud, W. R. Curr. Opin. Invest. Drugs 2004, 5, 592–596
- Keating, M. J.; Kantarjian, H.; Talpaz, M.; Redman, J.; Koller, C.; Barlogie, B.; Velasquez, W.; Plunkett, W.; Freireich, E. J.; McCredie, K. B. *Blood* 1989, 74, 19–25.
- Rai, K. R.; Peterson, B. L.; Appelbaum, F. R.; Kolitz, J.; Elias, L.; Shepherd, L.; Hines, J.; Threatte, G. A.; Larson, R. A.; Cheson, B. D.; Schiffer, C. A. N. Engl. J. Med. 2000, 343, 1750–1757. doi:10.1056/NEJM200012143432402
- Bonate, P. L.; Arthaud, L.; Cantrell, W. R., Jr.; Stephenson, K.; Secrist, J. A., III; Weitman, S. Nat. Rev. Drug Discovery 2006, 5, 855–863. doi:10.1038/nrd2055
- Jeha, S.; Kantarjian, H. Expert Rev. Anticancer Ther. 2007, 7, 113–118. doi:10.1586/14737140.7.2.113
- Gundersen, L.-L.; Nissen-Meyer, J.; Spilsberg, B. J. Med. Chem. 2002, 45, 1383–1386. doi:10.1021/jm0110284
- Nauš, P.; Perliková, P.; Pohl, R.; Hocek, M. Collect. Czech. Chem. Commun. 2011, 76, 957–988. doi:10.1135/cccc2011082
- Montgomery, J. A.; Hewson, K. J. Med. Chem. 1968, 11, 48–52. doi:10.1021/jm00307a010
- Hocek, M.; Holý, A.; Votruba, I.; Dvořáková, H. J. Med. Chem. 2000, 43, 1817–1825. doi:10.1021/jm991167+
- Hocek, M.; Holý, A.; Votruba, I.; Dvořáková, H.
 Collect. Czech. Chem. Commun. 2001, 66, 483–499.
 doi:10.1135/cccc20010483
- Nauš, P.; Pohl, R.; Votruba, I.; Džubák, P.; Hajdúch, M.; Ameral, R.;
 Birkuš, G.; Wang, T.; Ray, A. S.; Mackman, R.; Cihlar, T.; Hocek, M.
 J. Med. Chem. 2010, 53, 460–470. doi:10.1021/jm901428k
- Hocek, M.; Nauš, P.; Pohl, R.; Votruba, I.; Furman, P. A.;
 Tharnish, P. M.; Otto, M. J. J. Med. Chem. 2005, 48, 5869–5873.
 doi:10.1021/jm050335x
- 18. Smith, K. L.; Lai, V. C. H.; Prigaro, B. J.; Ding, Y.; Gunic, E.; Girardet, J.-L.; Zhong, W.; Hong, Z.; Lang, S.; An, H. *Bioorg. Med. Chem. Lett.* **2004**, *14*, 3517–3520. doi:10.1016/j.bmcl.2004.04.067
- Ding, Y.; Girardet, J.-L.; Hong, Z.; Lai, V. C. H.; An, H.; Koh, Y.-h.; Shaw, S. Z.; Zhong, W. Bioorg. Med. Chem. Lett. 2005, 15, 709–713. doi:10.1016/j.bmcl.2004.11.020
- Olsen, D. B.; Eldrup, A. B.; Bartholomew, L.; Bhat, B.;
 Bosserman, M. R.; Ceccacci, A.; Colwell, L. F.; Fay, J. F.; Flores, O. A.;
 Getty, K. L.; Grobler, J. A.; LaFemina, R. L.; Markel, E. J.;
 Migliaccio, G.; Prhavc, M.; Stahlhut, M. W.; Tomassini, J. E.;
 MacCoss, M.; Hazuda, D. J.; Carroll, S. S.
 Antimicrob. Agents Chemother. 2004, 48, 3944–3953.
 doi:10.1128/AAC.48.10.3944-3953.2004
- Fustero, S. In Fluorine in Medicinal Chemistry and Chemical Biology;
 Ojima, I., Ed.; Wiley-Blackwell: Chichester, U.K., 2009; pp 527–583.
- 22. Filler, R.; Saha, R. Future Med. Chem. 2009, 1, 777–791. doi:10.4155/fmc.09.65

- 23. Guo, X.; Li, Y.; Tao, L.; Wang, Q.; Wang, S.; Hu, W.; Pan, Z.; Yang, Q.; Cui, Y.; Ge, Z.; Dong, L.; Yu, X.; An, H.; Song, C.; Chang, J. Bioorg. Med. Chem. Lett. 2011, 21, 6770–6772. doi:10.1016/j.bmcl.2011.09.040
- An, H.; Zhu, J.; Wang, X.; Xu, X. Bioorg. Med. Chem. Lett. 2006, 16, 4826–4829. doi:10.1016/j.bmcl.2006.06.070
- 25. Xu, W.; Xi, B.; Wu, J.; An, H.; Zhu, J.; Abassi, Y.; Feinstein, S. C.; Gaylord, M.; Geng, B.; Yan, H.; Fan, W.; Sui, M.; Wang, X.; Xu, X. Mol. Cancer Ther. 2009, 8, 3318–3330.
 doi:10.1158/1535-7163.MCT-09-0548
- 26. Bao, Y.; Mo, X.; Xu, X.; He, Y.; Xu, X.; An, H. *J. Pharm. Biomed. Anal.* **2008**, *48*, 664–671. doi:10.1016/j.jpba.2008.06.013
- 27. An, H.; Hu, X.; Gu, J.; Chen, L.; Xu, W.; Mo, X.; Xu, W.; Wang, X.; Xu, X. AAPS PharmSciTech 2008, 9, 551–556. doi:10.1208/s12249-008-9081-5
- 28. Zheng, L.; Wang, Q.; Yang, X.; Guo, X.; Chen, L.; Tao, L.; Dong, L.; Li, Y.; An, H.; Yu, X.; Wang, Q.; Chang, J. Antiviral Ther. 2012, 17, 679–687. doi:10.3851/IMP2094
- Wang, Q.; Hu, W.; Wang, S.; Pan, Z.; Tao, L.; Guo, X.; Qian, K.;
 Chen, C.-H.; Lee, K.-H.; Chang, J. Eur. J. Med. Chem. 2011, 46, 4178–4183. doi:10.1016/j.ejmech.2011.06.020
- Yokoyama, A.; Nakai, Y.; Yoneda, S.; Kurita, Y.; Niitani, H. *Anti-Cancer Drugs* 1997, 8, 574–580. doi:10.1097/00001813-199707000-00004
- 31. Pankiewicz, K. W.; Krzeminski, J.; Watanabe, K. A. *J. Org. Chem.* **1992**, *57*, 7315–7321. doi:10.1021/jo00052a055
- Giordani, A.; Ermoli, A.; Battisthi, C.
 Nucleosides, Nucleotides Nucleic Acids 1991, 10, 719–721.
 doi:10.1080/07328319108046581
- van Aerschot, A.; Herdewijn, P.; Janssen, G.; Cools, M.; De Clercq, E.
 Antiviral Res. 1989, 12, 133–150. doi:10.1016/0166-3542(89)90047-8
- 34. Puech, F.; Gosselin, G.; Imbach, J.-L. *J. Chem. Soc., Chem. Commun.* **1989**, 955–957. doi:10.1039/c39890000955
- Battistini, C.; Giordani, A.; Ermoli, A.; Franceschi, G. Synthesis 1990, 900–905. doi:10.1055/s-1990-27048
- Morizawa, Y.; Nakayama, T.; Yasuda, A.; Uchida, K.
 Bull. Chem. Soc. Jpn. 1989, 62, 2119–2120. doi:10.1246/bcsj.62.2119
- Mikhailopulo, I. A.; Poopeiko, N. E.; Pricota, T. I.; Sivets, G. G.;
 Kvasyuk, E. I.; Balzarini, J.; De Clercq, E. J. Med. Chem. 1991, 34, 2195–2202. doi:10.1021/jm00111a040
- Kim, H. O.; Lim, M. H.; Park, J. G.; Moon, H. R.; Jacobson, K. A.; Kim, H.-D.; Chun, M. W.; Jeong, L. S. Nucleosides, Nucleotides Nucleic Acids 2003, 22, 923–925. doi:10.1081/NCN-120022701
- 39. Lim, M. H.; Kim, H. O.; Moon, H. R.; Lee, S. J.; Chun, M. W.; Gao, Z.-G.; Melman, N.; Jacobson, K. A.; Kim, J. H.; Jeong, L. S. Bioorg. Med. Chem. Lett. 2003, 13, 817–820. doi:10.1016/S0960-894X(03)00027-1
- Mikhailopulo, I. A.; Sivets, G. G. Helv. Chim. Acta 1999, 82, 2052–2065.
 doi:10.1002/(SICI)1522-2675(19991110)82:11<2052::AID-HLCA2052> 3.0.CO;2-7
- Ozols, A. M.; Azhayev, A. V.; Dyatkina, N. B.; Krayevsky, A. A. Synthesis 1980, 557–559. doi:10.1055/s-1980-29120
- Bennett, L. L., Jr.; Schnebli, H. P.; Vail, M. H.; Allan, P. W.; Montgomery, J. A. Mol. Pharmacol. 1966, 2, 432–443.
- 43. Lynch, T. P.; Paran, J. H.; Paterson, A. R. P. Cancer Res. 1981, 41,
- Marasco, C., Jr.; Pera, P.; Spiess, A. J.; Bernacki, R.; Sufrin, J. Molecules 2005, 10, 1015–1020. doi:10.3390/10081015

- Söderberg, B. C. G. Coord. Chem. Rev. 2006, 250, 300–387. doi:10.1016/j.ccr.2005.03.033
- Agrofoglio, L. A.; Gillaizeau, I.; Saito, Y. Chem. Rev. 2003, 103, 1875–1916. doi:10.1021/cr010374q
- 47. Greco, N. J.; Tor, Y. Tetrahedron 2007, 63, 3515–3527. doi:10.1016/j.tet.2007.01.073
- Srivatsan, S. G.; Tor, Y. J. Am. Chem. Soc. 2007, 129, 2044–2053. doi:10.1021/ja066455r
- 49. Miyaura, N.; Suzuki, A. Chem. Rev. 1995, 95, 2457–2483. doi:10.1021/cr00039a007
- Strouse, J. J.; Jeselnik, M.; Arterburn, J. B. Acta Chim. Slov. 2005, 52, 187–199.
- Bellina, F.; Carpita, A.; Rossi, R. Synthesis 2004, 2419–2440. doi:10.1055/s-2004-831223
- Parry, P. R.; Bryce, M. R.; Tarbit, B. Synthesis 2003, 1035–1038. doi:10.1055/s-2003-39158
- 53. Li, W.; Nelson, D. P.; Jensen, M. S.; Hoerrner, R. S.; Cai, D.; Larsen, R. D.; Reider, P. J. J. Org. Chem. 2002, 67, 5394–5397. doi:10.1021/jo025792p
- Cioffi, C. L.; Spencer, W. T.; Richards, J. J.; Herr, R. J. J. Org. Chem. 2004, 69, 2210–2212. doi:10.1021/jo034664d
- 55. Bauta, W. E.; Schulmeier, B. E.; Burke, B.; Puente, J. F.; Cantrell, W. R., Jr.; Lovett, D.; Goebel, J.; Anderson, B.; Ionescu, D.; Guo, R. Org. Process Res. Dev. 2004, 8, 889–896. doi:10.1021/op049884n
- 56. Kluin-Nelemans, H. C.; Oldhoff, J. M.; van Doormaal, J. J.; van't Wout, J. W.; Verhoef, G.; Gerrits, W. B. J.; van Dobbenburgh, O. A.; Pasmans, S. G.; Fijnheer, R. *Blood* 2003, 102, 4270–4276. doi:10.1182/blood-2003-05-1699
- Aleem, A.; Anjum, F.; Algahtani, F.; Iqbal, Z.; Alsaleh, K.; AlMomen, A. Asian Pac. J. Cancer Prev. 2013, 14, 1089–1092.
 doi:10.7314/APJCP.2013.14.2.1089
- Dalpozzo, R.; De Nino, A.; Maiuolo, L.; Procopio, A.; De Munno, G.; Sindona, G. *Tetrahedron* 2001, *57*, 4035–4038. doi:10.1016/S0040-4020(01)00284-8
- Robins, M. J.; Zou, R.; Guo, Z.; Wnuk, S. F. J. Org. Chem. 1996, 61, 9207–9212. doi:10.1021/jo9617023

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doi:10.3762/bjoc.11.272