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Heterogeneous H-Bonding in a Foldamer Helix

Brian F. Fisher, Li Guo[†], Brian S. Dolinar, Ilia A. Guzei, and Samuel H. Gellman

Department of Chemistry, University of Wisconsin-Madison, Madison, Wisconsin 53706, United States

Abstract

Structural characterization of new α/γ -peptide foldamers containing the cyclically-constrained γ -amino acid **I** is described. Crystallographic and 2D NMR analysis shows that γ residue **I** promotes formation of a 12/10-helical secondary structure in α/γ -peptides. This helix contains two different types of internal H-bond, and the data show that the 12-atom C=O(i)→H-N(i+3) H-bond is more favorable than the 10-atom C=O(i)→H-N(i-1) H-bond. Several foldamer helices featuring topologically distinct H-bonds have been discovered, but our findings are the first to show that such H-bonds may differ in their favorability.

Biology relies on proteins and nucleic acids to carry out diverse functions, most of which require that the biopolymer adopts a complex and specific folding pattern. The relationship between sophisticated function and higher-order structure has inspired many groups to seek non-biological oligomers that favor specific conformations (“foldamers”).¹ Extrapolation from the poly- α -amino acid backbone of proteins has led to the study of β -peptides, γ -peptides and higher homologues.² Secondary structural motifs reminiscent of (but geometrically distinct from) those well-known in proteins, including helices, sheets and reverse turns, have been characterized for these new backbones, and approaches to tertiary structure have been reported.³ The folding rules established for β - and γ -peptides have enabled the development of specific examples that display biomimetic function.⁴

Expansion beyond the biopolymer prototypes allows deviation from particular structural parameters associated with proteins and nucleic acids. The backbones of these biopolymers, for example, are homogeneous in that each subunit is drawn from a single chemical class (e.g., α -amino acids), but heterogeneous backbones are readily accessed among synthetic oligomers.⁵ A variety of discrete secondary structures have been identified among peptidic foldamers with mixed backbones, including combinations of $\alpha + \beta$ residues or $\alpha + \gamma$ residues. For some applications, such as functional mimicry of a natural α -helix,

Corresponding Author, gellman@chem.wisc.edu.

[†]Present Address

Halliburton, 3000 N Sam Houston Parkway E, Houston, Texas 77032, United States.

Supporting Information

Experimental details including synthetic routes, NMR data, details and PDB files of NMR structure calculations of **4** and other foldamers, DMSO titration data, HDX data, helix and H-bond parameters for crystal structures, and crystallographic data, including CIF files. This material is available free of charge via the Internet at <http://pubs.acs.org>.

Notes

The authors declare no competing financial interest.

heterogeneous backbones containing both α and β residues have proven to be superior to the homogeneous β residue backbone.⁶

Within the regular helices found in proteins, each type of internal non-covalent contact is topologically equivalent across all subunits involved. In an α -helix, for example, all $C=O(i) \rightarrow H-N(i+4)$ H-bonds should be comparable to one another, excluding terminal effects, since the $C=O$ and $H-N$ groups are all similar. In contrast, different types of internal H-bonds occur within many of the helices that have been documented among peptidic foldamers with unnatural backbones. Such differences are inherent for helices formed by heterogeneous backbones because of subunit diversity. Thus, for example, an α/β -peptide contains H-bond accepting groups ($C=O$) and donating groups ($N-H$) from both α and β residues. Different types of H-bonds are found also in foldameric helices in which H-bond directionality alternates along the backbone, whether the backbone is homogeneous (as in the β -peptide 10/12-helix⁷ and 18/20-helix⁸) or heterogeneous (as in the α/β -peptide 11/9-helix⁹ and 18/16-helix¹⁰). These systems raise a fundamental question: are the different types of intrahelical H-bonds comparably favorable? Here we describe a new type of α/γ -peptide foldamer and provide the first evidence that distinct types of H-bonds formed within a regular secondary structure can differ in terms of favorability.

The new foldamers contain ring-constrained γ residues of type **I** (Figure 1). Either enantiomer of the necessary γ -amino acid building block can be efficiently prepared.¹¹ Previous characterization of oligomers containing the stereoisomeric γ residue of type **II** established that this subunit, alone¹² or in alternation with α or with β residues,^{11, 13} favors helices defined by $C=O(i) \rightarrow H-N(i+3)$ H-bonds. The present studies of α/γ -peptides containing **I** were intended to test the generality of the “stereochemical patterning” hypothesis of Martinek, Fülöp *et al.*¹⁴ These workers proposed a correlation between H-bond directionality within foldamer helices and the signs of the torsion angles about the bonds that flank amide linkages (ψ, ϕ pairs from adjacent residues). If these torsion angles are all of the same sign, then all intrahelical H-bonds should be oriented in the same direction (as observed, for example in the α -helix). If the torsion angles alternate between neighboring amides (i.e., if amides that have positive flanking torsion angles are adjacent to amides that have negative flanking torsion angles and vice versa), then intrahelical H-bonds should alternate in directionality relative to the helix axis.

The Martinek-Fülöp hypothesis led us to predict that altering only the stereocenter adjacent to the ϕ torsion angle in γ residues (i.e., **II** \rightarrow **I**) would cause a wholesale change in helix geometry in α/γ -peptides, from a conformation with unidirectional H-bonds to a conformation with bidirectional H-bonds. This prediction relies on the fact that α residues can readily adopt conformations in which the backbone torsion angles ϕ and ψ have the same sign (α -helix) or different signs (β -sheet). γ Residues of type **I** with absolute configuration of (*S,S,R*) at the (α, β, γ) carbons must be paired with L- α -amino acid residues, and (*R,R,S*)-**I** with D- α residues, to test this prediction.

Our initial efforts were aimed at crystallographic characterization of the new α/γ (**I**)-peptides. Seven structures were obtained¹⁵ for oligomers containing four to six residues, three by racemic crystallization.¹⁶ The structures of tetramers **1a-c**, collectively, were not

wholly consistent with our expectations. All three formed a 12-atom H-bond between the N-terminal Boc C=O and N-H of the α residue closest to the C-terminus (Figure 2). This C=O(i) \rightarrow H-N($i+3$) interaction is characteristic of both the 12/10-helix we expected and the 12-helix previously documented for α/γ -peptides containing γ residue **II**. Only *rac*-**1a** (α residues = Phe), however, displayed the 10-atom C=O(i) \rightarrow H-N($i-1$) H-bond characteristic of the 12/10-helix. In this structure, the 10-atom H-bond was significantly longer than the 12-atom H-bond (N \cdots O = 3.00 vs. 2.83 Å). For **1b** (α residues = Ala), a second 12-atom H-bond formed, i.e. this α/γ -peptide adopted a 12-helix rather than a 12/10-helix. α/γ -Peptide **1c** (α residues = Ala) displayed an interesting variation on the 10-atom H-bond: the carbonyl of Ala-3 was linked to N-H($i-1$) via an intermolecular H-bond chain that included a water molecule along with the N-terminal Ala carbonyl of its lattice neighbor.¹⁷

Longer α/γ -peptides containing γ residue **I** manifested a stronger preference for the 12/10-helix in the solid state, but the crystal structures offered hints that the 12-atom H-bonds are more favorable than the 10-atom H-bonds. Pentamer **2a** (α residues = Val) displayed a 12/10-helical conformation in which the maximum number of amide-to-amide H-bonds was formed, two in 12-atom rings and one in a 10-atom ring (Figure 3). All of the H-bonded N \cdots O distances in this structure were relatively long (> 3.00 Å). The two structurally very similar symmetry-independent conformers of pentamer **2b**¹⁷ (α residues = Ala) adopted a 12/10-helix-like conformation in the crystal, but an ethanol molecule was found to be interpolated between the C=O and H-N groups that would have formed the 10-atom intrahelical H-bond (Figure 3). The two 12-atom H-bonds in this structure featured significantly shorter N \cdots O distances than were found in the crystal structure of **2a**.¹⁷

Hexamer *rac*-**3** (α residues = Phe) provided two crystal forms under slightly different solvent conditions (Figure 4). One crystal contained a 12/10-helix that seemed to unravel toward the C-terminus; this structure contained two short 12-atom H-bonds (2.85/2.86 Å), and a longer 10-atom H-bond (3.02 Å). A second amide-amide 10-atom H-bond would be possible toward the C-terminus, but in this case the poor geometric parameters (N \cdots O distance of 4.09 Å and N-H \cdots O angle of 135°) suggest at least that this interaction is substantially weaker than all other H-bonds described here.¹⁸ The other crystalline form of *rac*-**3** revealed a 12/10-helix-like conformation in which a water molecule was found to be interpolated between the C=O and H-N groups that would have formed the C-terminal 10-atom H-bond. The set of seven crystal structures, collectively, suggests that the new α/γ -peptide backbone has a propensity for 12/10-helix-like conformations, but that H-bonding may be more favorable in the 12-atom rings relative to the 10-atom rings.

To gain insight on the folding propensity of α/γ (**I**)-peptides in solution, we employed 2D NMR (COSY, TOCSY, and ROESY) to evaluate the conformations in CDCl₃ of five oligomers containing from five to nine residues. Evidence for 12/10-helical folding was obtained in each case in the form of inter-residue NOEs between protons separated by 7 to 12 covalent bonds. Here we focus on nonamer **4**. Three non-sequential backbone proton NOE patterns previously reported to be consistent with 12/10-helical secondary structure¹⁹ were observed along most or all of **4** (Figure 5). In addition, we observed a set of strong $i, i+2$ NOEs from C α H of each Val residue to C β H of the next Val residue, which is consistent with the 12/10-helical conformation. NMR-restrained simulated annealing calculations were

carried out for each of the five oligomers with the Crystallography and NMR System (CNS) software suite (Figure 6). A total of 24 NOE-derived distance restraints from **4** was used to generate an ensemble of 10 lowest-energy structures. The N-terminal four-residue segment in the average generated from these 10 structures overlays well with the crystal structure of pentamer **2a**.

We used H/D exchange (HDX) measurements to test the hypothesis that 12/10-helices formed by the new $\alpha/\gamma(\mathbf{I})$ -peptide backbone contain H-bond types with different intrinsic favorabilities. These studies focused on octamer **5**, Boc-([Phe][$\gamma(\mathbf{I})$])₄-OBn, for which 2D NMR data provided evidence of 12/10-helical folding in CDCl₃.¹⁷ Amino acid residue identity exerts a large influence on HDX;²⁰ therefore, results for octamer **5**, expressed as half-life ($t_{1/2}$) for disappearance of amide ¹H-NMR resonances after dissolution in 95:5 CDCl₃/CD₃OD, were normalized to results for model dipeptide **6**. Specifically, $t_{1/2}$ values for each α residue NH in **5** were divided by $t_{1/2}$ for the α residue NH in **6**, and $t_{1/2}$ values for each γ residue in **5** were divided by $t_{1/2}$ for the γ residue in **6**. This normalization should highlight the impact of folding on HDX at each site within the octamer.

The normalized $t_{1/2}$ values (Table 1) are generally largest toward the center of **5**, which is consistent with the hypothesis that this α/γ -peptide adopts a helical conformation featuring internal H-bonds of alternating directionality. The α residue NH resonances show larger normalized $t_{1/2}$ values than do the γ residue NH resonances. This trend is consistent with the hypothesis that the H-bonds formed by the α residue NH groups (12-atom rings) are more favorable than the H-bonds formed by the γ residue NH groups (10-atom rings). Further support for the proposed distinction between 12- and 10-atom internal H-bonds was provided by experiments in which small aliquots of DMSO were added to a CDCl₃ solution of octamer **5** (“DMSO titrations”).¹⁷

The crystallographic and NMR data reported here show that α/γ -peptide secondary structure can be engineered based on the stereochemical patterning hypothesis of Martinek, Fülöp *et al.* Altering just one configuration in the dipeptide repeating unit of the backbone, by replacing $\gamma(\mathbf{II})$ residues with $\gamma(\mathbf{I})$ residues, leads to a change in conformational preference, from a helix with unidirectional H-bonds to a helix in which H-bond directions alternate. The ability to alter helix type by switching a single configuration within the γ residues represents a significant accomplishment in foldamer design.

Exploration of the new $\alpha/\gamma(\mathbf{I})$ backbone led to the unexpected discovery that the 12/10-helix formed by these foldamers contains H-bonds that differ in terms of favorability, with the γ residue C=O(i) \rightarrow α residue H-N($i+3$) H-bonds intrinsically superior to the α residue C=O(i) \rightarrow γ residue H-N($i-1$) H-bonds. This possibility was initially suggested by the seven crystal structures described here, in which the α residue C=O(i) \rightarrow γ residue H-N($i-1$) H-bond distances were long, or other molecules were interpolated into these interactions. Such interpolations have occasionally been observed in proteins²¹ or short conventional peptides,²² and more recently in α/γ -peptides,²³ but the frequency of this unusual phenomenon in our structural data set stands out. HDX and DMSO titration results support the conclusion that there is an energetic differentiation between the two types of H-bond in the 12/10-helical conformation formed by the $\alpha/\gamma(\mathbf{I})$ backbone. We speculate that the $\gamma(\mathbf{I})$

residue is not ideally preorganized for the 10-atom H-bonding mode.¹⁷ Several foldamer secondary structures containing different H-bond types have been discovered, and our results raise the possibility that topologically distinct H-bonds will in general be energetically differentiated.

Supplementary Material

Refer to Web version on PubMed Central for supplementary material.

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REFERENCES

- 1 (a). Gellman SH. *Acc. Chem. Res.* 1998; 31:173.(b) Hill DJ, Mio MJ, Prince RB, Hughes TS, Moore JS. *Chem. Rev.* 2001; 101:3893. [PubMed: 11740924] (c) Guichard G, Huc I. *Chem. Commun.* 2011; 47:5933.
- 2 (a). Cheng RP, Gellman SH, DeGrado WF. *Chem. Rev.* 2001; 101:3219. [PubMed: 11710070] (b) Seebach D, Hook DF, Glättli A. *Biopolymers.* 2006; 84:23. [PubMed: 16235225]
3. Reinert ZE, Horne WS. *Org. Biomol. Chem.* 2014; 12:8796. [PubMed: 25285575]
- 4 (a). Hamuro Y, Schneider JP, DeGrado WF. *J. Am. Chem. Soc.* 1999; 121:12200.(b) Porter EA, Wang X, Lee HS, Weisblum B, Gellman SH. *Nature.* 2000; 404:565. [PubMed: 10766230] (c) Stephens OM, Kim S, Welch BD, Hodsdon ME, Kay MS, Schepartz A. *J. Am. Chem. Soc.* 2005; 127:13126. [PubMed: 16173723] (d) English EP, Chumanov RS, Gellman SH, Compton T. *J. Biol. Chem.* 2006; 281:2661. [PubMed: 16275647] (e) Gademann K, Kimmerlin T, Hoyer D, Seebach D. *J. Med. Chem.* 2001; 44:2460. [PubMed: 11448228] (f) Seebach D, Schaeffer L, Brenner M, Hoyer D. *Angew. Chem., Int. Ed.* 2003; 42:776.
- 5 (a). Horne WS, Gellman SH. *Acc. Chem. Res.* 2008; 41:1399. [PubMed: 18590282] (b) Pendem N, Nelli YR, Douat C, Fischer L, Laguerre M, Ennifir E, Kauffmann B, Guichard G. *Angew. Chem., Int. Ed.* 2013; 52:4147.(c) Wu H, Qiao Q, Hu Y, Teng P, Gao W, Zuo X, Wojtas L, Larsen RW, Ma S, Cai J. *Chem. Eur. J.* 2015; 21:2501. [PubMed: 25504756]
- 6 (a). Sadowsky JD, Schmitt MA, Lee HS, Umezawa N, Wang S, Tomita Y, Gellman SH. *J. Am. Chem. Soc.* 2005; 127:11966. [PubMed: 16117535] (b) Sadowsky JD, Fairlie WD, Hadley EB, Lee HS, Umezawa N, Nikolovska-Coleska Z, Wang S, Huang DC, Tomita Y, Gellman SH. *J. Am. Chem. Soc.* 2007; 129:139. [PubMed: 17199293] (c) Johnson LM, Gellman SH. *Methods Enzymol.* 2013; 523:407. [PubMed: 23422441]
7. Seebach D, Gademann K, Schreiber JV, Matthews JL, Hintermann T, Jaun B, Oberer L, Hommel U, Widmer H. *Helv. Chim. Acta.* 1997; 80:2033.
8. Szolnoki É, Hetényi A, Mándity IM, Fülöp F, Martinek TA. *Eur. J. Org. Chem.* 2013:3555.
9. Sharma GV, Nagendar P, Jayaprakash P, Radha Krishna P, Ramakrishna KV, Kunwar AC. *Angew. Chem., Int. Ed.* 2005; 44:5878.
- 10 (a). Berlicki L, Pils I, Wéber E, Mándity IM, Cabrele C, Martinek TA, Fülöp F, Reiser O. *Angew. Chem., Int. Ed.* 2012; 51:2208.(b) Legrand B, André C, Moulat L, Wenger E, Didierjean C, Aubert E, Averlant-Petit MC, Martinez J, Calmes M, Amblard M. *Angew. Chem., Int. Ed.* 2014; 53:13131.
11. Guo L, Chi Y, Almeida AM, Guzei IA, Parker BK, Gellman SH. *J. Am. Chem. Soc.* 2009; 131:16018. [PubMed: 19886693]
12. Guo L, Zhang W, Reidenbach AG, Giuliano MW, Guzei IA, Spencer LC, Gellman SH. *Angew. Chem., Int. Ed.* 2011; 50:5843.

13. Guo L, Almeida AM, Zhang W, Reidenbach AG, Choi SH, Guzei IA, Gellman SH. *J. Am. Chem. Soc.* 2010; 132:7868. [PubMed: 20491510]
14. Mándity IM, Weber E, Martinek TA, Olajos G, Tóth GK, Vass E, Fülöp F. *Angew. Chem., Int. Ed.* 2009; 48:2171.
15. Crystallographic data has been deposited at the Cambridge Structural Data Centre with accession codes 1056028-1056034 and can be accessed free of charge at www.ccdc.cam.ac.uk/data_request/cif.
- 16 (a). Yeates TO, Kent SB. *Annu. Rev. Biophys.* 2012; 41:41. [PubMed: 22443988] (b) Lee M, Shim J, Kang P, Guzei IA, Choi SH. *Angew. Chem., Int. Ed.* 2013; 52:12564.
17. See Supporting Information.
18. Jeffrey, GA. *An Introduction to Hydrogen Bonding*. Oxford University Press; New York: 1997.
- 19 (a). Sharma GV, Jadhav VB, Ramakrishna KV, Jayaprakash P, Narsimulu K, Subash V, Kunwar AC. *J. Am. Chem. Soc.* 2006; 128:14657. [PubMed: 17090052] (b) Giuliano MW, Maynard SJ, Almeida AM, Guo L, Guzei IA, Spencer LC, Gellman SH. *J. Am. Chem. Soc.* 2014; 136:15046. [PubMed: 25301259]
20. Bai Y, Milne JS, Mayne L, Englander SW. *Proteins.* 1993; 17:75. [PubMed: 8234246]
21. Sundaralingam M, Sekharudu YC. *Science.* 1989; 244:1333. [PubMed: 2734612]
22. Karle IL, Flippen-Anderson J, Uma K, Balaram P. *Proc. Natl. Acad. Sci. USA.* 1987; 85:299. [PubMed: 2448772]
- 23 (a). Basuroy K, Dinesh B, Shamala N, Balaram P. *Angew. Chem.* 2013; 125:3218.(b) Basuroy K, Dinesh B, Shamala N, Balaram P. *Angew. Chem., Int. Ed.* 2013; 52:3136.(c) Karle IL, Flippen-Anderson JL, Uma K, Balaram P. *Int. J. Pept. Protein Res.* 1994; 44:491. [PubMed: 7896509]

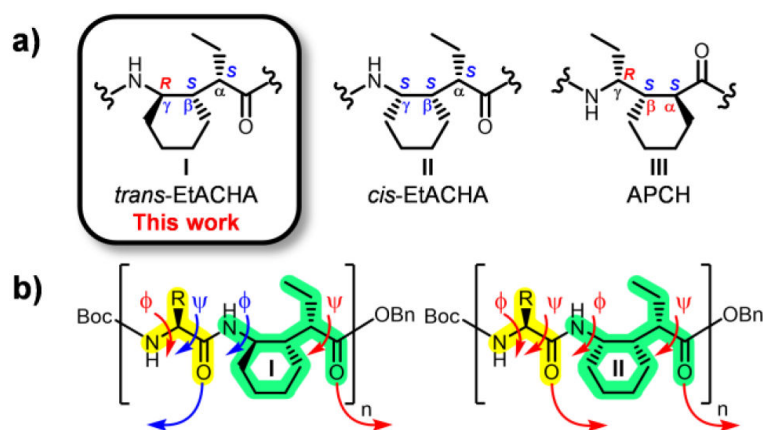
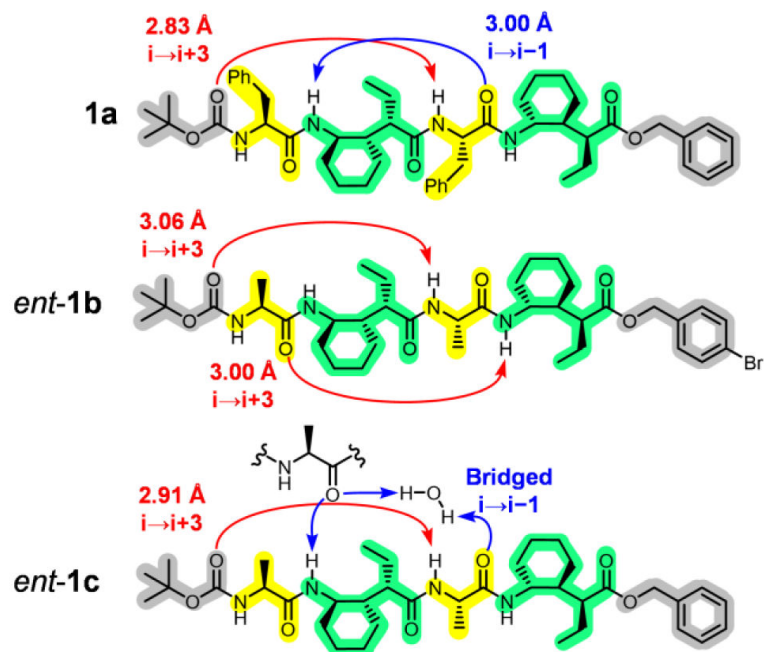


Figure 1.
 a) Comparison of cyclic γ -amino acids studied by our group. **b)** Inversion of the C_γ stereocenter of **II**, forming **I**, is predicted to invert the adjacent H-bond directionality in the full-length H-bonded foldamer.

**Figure 2.**

Structures of tetramers **1a-c** depicting H-bonds observed in the crystal structures.

Enantiomers of **1b** and **1c** are shown. α Residues are shaded in yellow, γ residues in green, and protecting groups in gray. 12-Atom H-bonds are shown in red, and 10-atom H-bonds in blue. Geometric criteria used for all H-bond assignment are distance $N\cdots O < 4.0 \text{ \AA}$ and angle $N-H\cdots O > 130^\circ$.

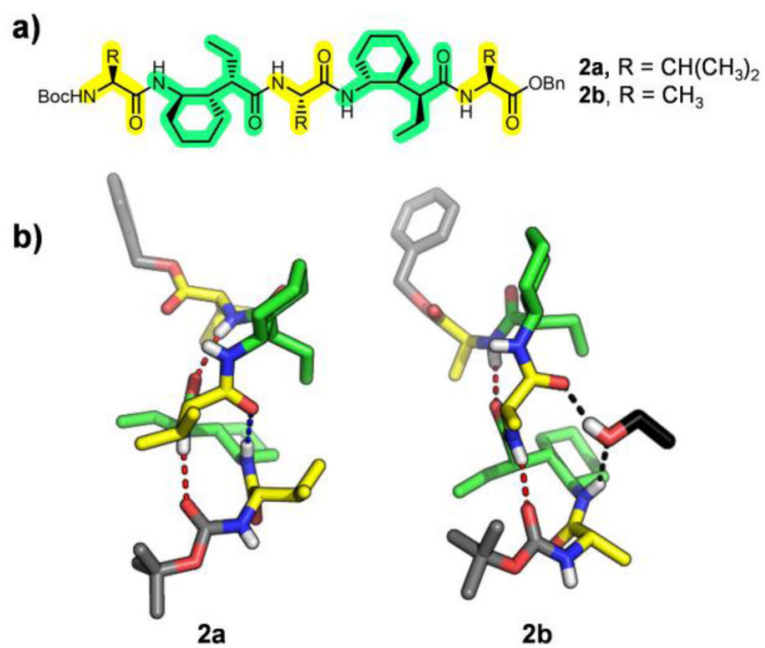


Figure 3.
a) Structures of pentamers **2a-b**. **b)** Crystal structures of pentamers **2a** and **2b**, depicting the inserted EtOH molecule of **2b** in black. 12-Atom H-bonds are shown as red dashes, and 10-atom H-bonds as blue dashes.

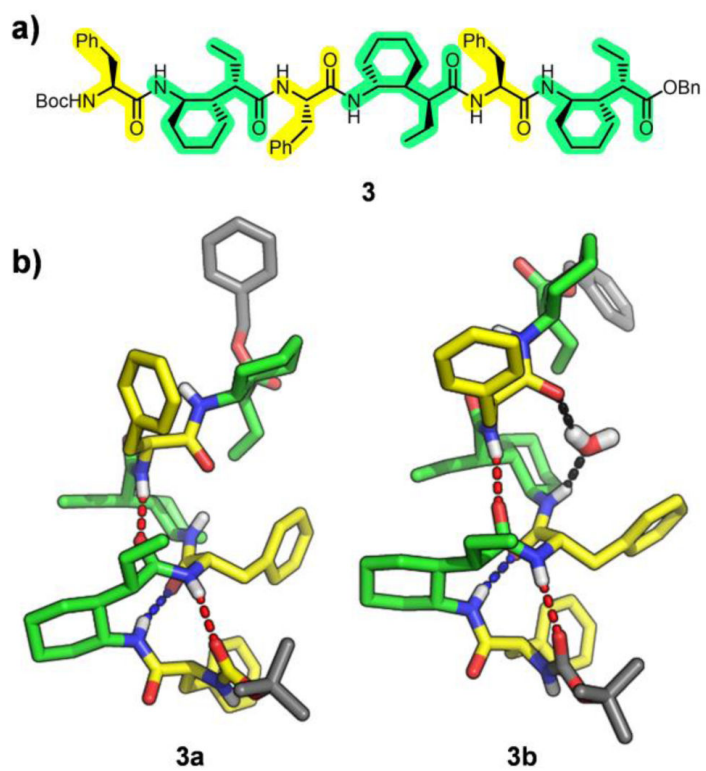
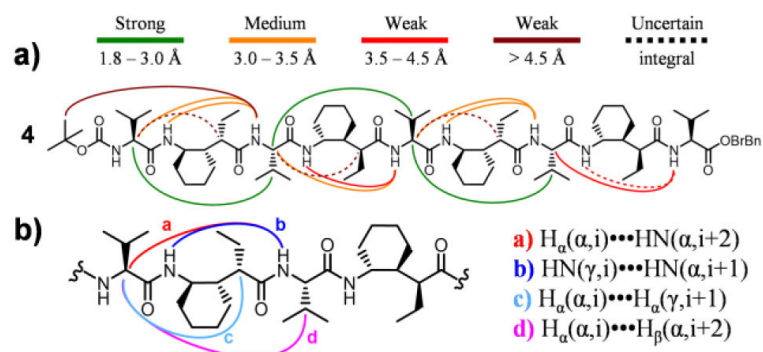


Figure 4.
a) Structure of hexamer **3**. b) Crystal structures **3a** and **3b**, depicting the inserted H₂O molecule of **3b**.

**Figure 5.**

a) Non-sequential ROESY crosspeaks observed in nonamer **4**. Uncertain integrals were unambiguously assigned but could not be reliably software-integrated. **b)** 12/10-Helical crosspeak patterns observed in **4**.

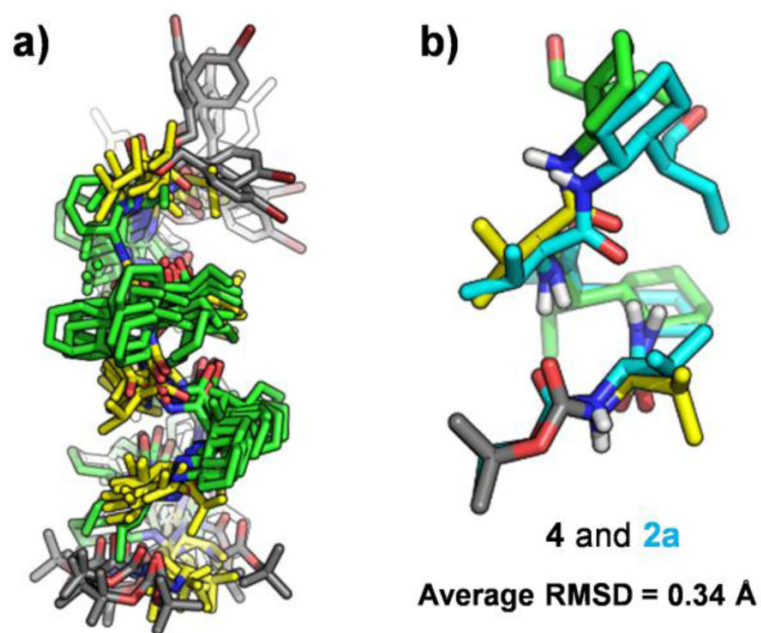


Figure 6.
a) NMR ensemble generated by CNS using distance restraints from ROESY data of nonamer **4** in CDCl₃. The 10 lowest-energy structures were collected from 1000 trial structures. b) Backbone alignment of the first four residues of the minimized average of the 10 lowest-energy NMR structures with pentamer crystal structure **2a**.

Table 1Relative HDX Half-Life Change of Octamer **5**.

	Phe-1	γ (I)-2	Phe-3	γ (I)-4	Phe-5	γ (I)-6	Phe-7	γ (I)-8
$t_{1/2}$ (hr)	2.2	102	95	199	121	– ^a	46	37
Normalized $t_{1/2}$	1.7	13	73	25	93	– ^a	35	4.6
H-bonded ring size	–	10	12	10	12	10	12	–

^aResonance overlapped with phenyl H.