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## Estrogenicity and androgenicity screening of PCB sulfate monoesters in human breast cancer MCF-7 cells

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### Abstract

Recent studies identified PCB sulfate esters as a major product of PCB metabolism. Since hydroxy-PCBs (HO-PCBs), the immediate precursors of PCB sulfates and important contributors to PCB toxicity, were shown to have estrogenic activity, we investigated the estrogenicity/androgenicity of a series of PCB sulfate metabolites. We synthesized the five possible structural sulfate monoester metabolites of PCB 3, a congener shown to be biotransformed to sulfates, a sulfate ester of the paint-specific congener PCB 11, and sulfate monoesters of two HO-PCBs reported to interact with sulfotransferases (PCB 39, no *ortho* chlorines, and PCB 53, 3 *ortho* chlorines). We tested these PCB sulfates and 4'-HO-PCB 3 as positive control for estrogenic, androgenic, anti-estrogenic and anti-androgenic activity in the E- and A-screen with human breast cancer MCF7 derived cells at 100  $\mu\text{M}$  – 1 pM concentrations. Only 4'-HO-PCB 3 was highly cytotoxic at 100  $\mu\text{M}$ . We observed structure-activity relationships: compounds with a sulfate group in the chlorine-containing ring of PCB 3 (2PCB 3 and 3PCB 3 sulfate) showed no interaction with the estrogen (ER) and androgen (AR) receptor. The 4'-HO-PCB 3 and its sulfate ester had the highest estrogenic effect, but at 100 fold different concentrations, i.e. 1  $\mu\text{M}$  and 100  $\mu\text{M}$ , respectively. Four of the PCB sulfates were estrogenic (2'PCB 3, 4'PCB 3, 4PCB 39, 4PCB 53 sulfates; at 100  $\mu\text{M}$ ). These sulfates and 3'PCB 3 sulfate also exhibited anti-estrogenic activity, but at nM and pM concentrations. The 4'PCB 3 sulfate (*para-para*' substituted) had the strongest androgenic activity, followed by 3'PCB 3, 4PCB 53, 4PCB 11, and 4PCB 39 sulfates and the 4'HO-PCB 3. In contrast, anti-androgenicity was only observed with the two compounds that have the sulfate group in *ortho*- or *meta*- position in the second ring (2'PCB 3 and 3'PCB 3 sulfate). No dose-response was observed in any screen, but, with exception of estrogenic activity (only seen at 100  $\mu\text{M}$ ), endocrine activity was often displayed at several concentrations and even at 1 pM concentration. These data suggest that sulfation of HO-PCBs is indeed reducing their cytotoxicity and estrogenicity, but may produce other endocrine disruptive activities at very low concentrations.

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**Conflict of Interest**

The authors declare no conflict of interest.

## Keywords

Polychlorinated biphenyl; PCB-sulfate; MCF-7; estrogenicity; androgenicity; E-screen; A-screen

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## Introduction

Polychlorinated Biphenyls (PCBs), a group of 209 individual congeners with different chlorination pattern on a biphenyl core, are a major class of persistent organic pollutants. They bioaccumulate, biomagnify in the food chain, and are ubiquitously detectable in the environment (water, soil, air), wildlife, food, and human tissues (Safe 1993). PCBs were commercially produced from 1928 to the 1980s as mixtures, sold in the US under the trade name Aroclor, and used in large quantities for a multitude of industrial applications (Robertson & Hansen 2001). An estimated 0.2 million tons of PCBs are distributed worldwide in the environment and PCBs are still in use in older devices that could leak PCBs into the environment through failure and/or inappropriate disposal (Erickson & Kaley 2011). In addition, new research found that PCBs are currently produced as inadvertent by-products in the manufacture of commercial paints and pigments (Hu & Hornbuckle 2010).

The International Agency for Research on Cancer (IARC) has recently upgraded PCBs to Group 1, human carcinogens (Lauby-Secretan et al. 2013). Moreover aside from cancer, exposure to PCBs is associated with a broad range of adverse health effects (Silberhorn et al. 1990), including reproductive abnormalities in animal models, developmental toxicity, and endocrine disruptive effects, to mention a few (DeCastro et al. 2006, Ropstad et al. 2006, Safe 1994). Impaired reproduction, changes in behaviour, altered fetal and child development, and even promotion of carcinogenesis are often associated with hormonally active agents (Longnecker et al. 1997, National Research Council U.S. 1999, Negri et al. 2003), raising the question whether PCBs or their derivatives are hormonally active compounds.

PCB congeners especially those possessing few chlorine atoms, are subject to metabolic attack and give rise to a large number of hydroxyl and other metabolites, each with its own biologic activity (Grimm et al. 2015a). Nevertheless, most research to date has focused on commercial PCB mixtures, individual congeners, and to a lesser extend their hydroxylated metabolites. Particularly the semi-volatile lower chlorinated PCBs that are commonly found in indoor and outdoor air resulting in considerable inhalation exposure (Ampleman et al. 2015, Herrick et al. 2007, Lehmann et al. 2014, Tue et al. 2013) undergo cytochrome P-450 catalysed oxidative metabolism to mono- and dihydroxy-biphenyls (McLean et al. 1996). Those hydroxylated PCBs are substrates of cytosolic phase II enzymes resulting in glucuronidation and sulfation (Ekuase et al. 2014a, Liu et al. 2009, Sacco et al. 2008, Tampal et al. 2002). Dhakal and coworkers discovered that formation of sulfate conjugates is the major metabolic pathway for PCB 3 in rats *in vivo* (Dhakal et al. 2013, Dhakal et al. 2012, Dhakal et al. 2014). While quite a bit has been learned recently about HO-PCB as substrates and inhibitors of sulfotransferases (Ekuase et al. 2014b, Ekuase et al. 2011, Liu et al. 2011) very little is known about the potential toxicity and endocrine disruption of PCB sulfate metabolites. However, some HO-PCBs and PCB sulfate metabolites are ligands for

human transthyretin which could play a potential role in disruption of thyroid hormone homeostasis (Grimm et al. 2013a).

HO-PCBs were reported to elicit estrogenic and anti-estrogenic effects in different test systems (Connor et al. 1997, DeCastro et al. 2006, Gregoraszcuk et al. 2008, Kramer et al. 1997, Krishnan & Safe 1993, Machala et al. 2004, Soto et al. 1995). We hypothesized that sulfation of HO-PCBs would decrease their potential cytotoxic, estrogenic and androgenic activity. Our intention in this study was to screen a variety of different PCB sulfates over a wide range of concentrations in four different types of endocrine disruption assays to gain a better understanding of the potential toxicity of this kind of metabolite. We used the E- and A-screen in which estrogenic or androgenic activity of a test compound is determined by receptor-mediated induction or inhibition of proliferation of MCF7-BOS or MCF7-AR1 cells, respectively (Soto et al. 1998, Soto et al. 1995, Szelei et al. 1997). We examined the endocrine effects of eight synthetic PCB sulfate monoesters in comparison to a hydroxylated PCB. We report here that some PCB sulfate esters show estrogenic activity, but only at very high concentrations (100  $\mu$ M), while anti-estrogenic, androgenic, or anti-androgenic activity was more frequently observed and visible at orders of magnitude lower concentrations (pM).

## Materials and Methods

### Chemical Substances

Reagents, media and media supplements were from Fischer Scientific (Pittsburgh, PA) if not stated otherwise. Beta-Estradiol (> 98% pure) was purchased from MP Biomedicals; Fulvestrant, ICI 182,780 (> 98% pure) and R1881 (> 98% pure) were purchased from Sigma-Aldrich. The synthesis and characterization of the positive control, 4'-chloro-biphenyl-4-ol (4'-HO-PCB 3), and the ammonium salts of 2'-sulfooxy-4-chloro-biphenyl (2'PCB 3 sulfate), 3'-sulfooxy-4-chloro-biphenyl (3'PCB 3 sulfate), 4'-sulfooxy-4-chloro-biphenyl (4'PCB 3 sulfate), and 4'-sulfooxy-3,3'-dichloro-biphenyl (4PCB 11 sulfate) has been described previously (Grimm et al. 2013b, Li et al. 2010). Four additional PCB sulfates, including 2-sulfooxy-4-chlorobiphenyl (2PCB 3 sulfate), 3-sulfooxy-4-chloro-biphenyl (3PCB3 sulfate), 4-sulfooxy-3,4',5-trichloro-biphenyl (4PCB 39 sulfate) and 4-sulfooxy-2,2',5,6'-tetrachloro-biphenyl (4PCB 53 sulfate) were synthesized and characterized as described below. The chemical structures and the respective abbreviations of the PCB sulfates and the positive control, 4'-HO-PCB 3, are shown in Figure 1. All chemicals were dissolved in dimethyl sulfoxid (DMSO) and added to the medium at the indicated concentration. The final concentration of solvent in the culture medium did not exceed 0.5% or 0.6% v/v of DMSO. Control samples obtained medium containing 0.5% or 0.6% DMSO without test compound. This concentration was found to have no effect on cell yields.

### Synthesis and Characterization of PCB sulfates

**General methods**—Melting points were measured on a Mel-Temp melting point apparatus and are uncorrected. If not stated otherwise, all PCB sulfates and the corresponding intermediates, were characterized by  $^1\text{H}$  and  $^{13}\text{C}$  NMR spectroscopy. The NMR spectra were recorded on a Bruker Avance DRX-400 spectrometer in the University

of Iowa Central NMR Research Facility (Iowa City, IA, USA). The NMR samples were prepared in CDCl<sub>3</sub> or CD<sub>3</sub>OD (Cambridge Isotope Laboratories, Andover, MA), and tetramethylsilane (TMS) was used as internal standard. GC-MS analysis of TCE-PCB sulfates diester intermediates was performed in the electron impact (EI) mode on an Agilent 6890N gas chromatograph coupled with an Agilent 5975 Mass Selective Detector (Agilent Technologies, CA, USA) as reported previously (Telu et al. 2010). Only the isotopic ion with the lowest mass is reported for major fragments observed in the MS spectra. Accurate mass determinations of the PCB sulfates were performed by the High Resolution Mass Spectrometry Facility of the University of California Riverside (Riverside, CA, USA). The purity of all PCB sulfates was verified by thin layer chromatography as described previously (Li et al. 2010) prior to preparing DMSO stock solutions for cell culture experiments to ensure that no hydrolysis to the corresponding OH-PCB had occurred.

### Characterization of TCE-protected PCB sulfate diesters

**Sulfuric acid 4-chlorobiphenyl-2-yl 2,2,2-trichloroethyl sulfate:** Reaction of 4-chlorobiphenyl-2-ol (300 mg) (Zhai et al. 2011) and 2,2,2-trichloroethyl chlorosulfate (441 mg) in the presence of Et<sub>3</sub>N/DMAP in THF yielded 300 mg (49 %) of the product as a colorless oil (Li et al. 2010). <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>): δ/ppm 3.92 (s, 2H), 7.37- 7.41 (m, 2H), 7.42-7.45 (m, 2H), 7.46-7.47 (m, 2H), 7.48-7.51 (m, 1H), 7.57 (d, 1H, *J* = 8.0 Hz); <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>): δ/ppm 79.9, 92.3, 123.1, 128.6, 128.7, 129.0, 129.6, 132.3, 133.6, 134.4, 135.4, 146.9. GC-MS *m/z* (relative abundance, %): 416(18), 284(10), 204(100), 175(36), 139(32).

**Sulfuric acid 4-chlorobiphenyl-3-yl 2,2,2-trichloroethyl sulfate:** Reaction of 4-chlorobiphenyl-3-ol (400 mg) (Zhai et al. 2011) and 2,2,2-trichloroethyl chlorosulfate (497 mg) in the presence of Et<sub>3</sub>N/DMAP in THF yielded 550 mg (80 %) of the product as a colorless oil (Li et al. 2010). <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>): δ/ppm 4.97 (s, 2H), 7.39-7.41 (m, 1H), 7.43-7.46 (m, 2H), 7.49-7.55 (m, 4H), 7.74 (d, 1H, *J* = 8.0 Hz); <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>): δ/ppm 80.8, 92.4, 121.6, 125.5, 127.1, 127.4, 128.6, 129.3, 131.4, 138.4, 142.2, 146.2. GC-MS *m/z* (relative abundance, %): 416(16), 284(13), 204(51), 168(100), 139(38).

**Sulfuric acid 3,4',5-trichlorobiphenyl-4-yl 2,2,2-trichloroethyl sulfate:** Reaction of 3,4',5-trichloro-biphenyl-4-ol (160 mg) (Joshi et al. 2011) and 2,2,2-trichloroethyl chlorosulfate (175 mg) in the presence of Et<sub>3</sub>N/DMAP in THF yielded 220 mg (78 %) of the product as a white solid (Li et al. 2010), mp: 117-118 °C. <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>): δ/ppm 5.04 (s, 2H), 7.43 (s, 4H), 7.55(s, 2H). <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>): δ/ppm 80.4, 92.4, 127.9, 128.5, 129.6, 129.9, 135.4, 135.8, 141.4. GC-MS *m/z* (relative abundance, %): 484(11), 273(100), 243(21), 207(11), 138(32), 61(18).

### Characterization of PCB sulfate monesters, ammonium salts

**2-Sulfooxy-4-chloro-biphenyl, ammonium salt (2PCB 3 sulfate):** Deprotection of sulfuric acid 4-chlorobiphenyl-2-yl 2,2,2-trichloroethyl sulfate (250 mg) with zinc powder-ammonium formate in ethanol yielded the desired ammonium salt (165 mg; 92 %) as a white solid (Li et al. 2010), mp: 99 °C (dec.); <sup>1</sup>H NMR (400 MHz, CD<sub>3</sub>OD): δ/ppm 7.22 (dd, 1H, *J* = 4.0 Hz, *J* = 8.0 Hz), 7.31-7.41 (m, 4H), 7.57 (AA'XX'' system, 2H), 7.78 (d, 1H, *J* = 2.0

Hz);  $^{13}\text{C}$  NMR (100 MHz,  $\text{CD}_3\text{OD}$ ):  $\delta/\text{ppm}$  122.4, 125.7, 128.3, 129.1, 130.6, 132.8, 134.1, 138.5, 151.6; HRMS (ESI, negative):  $[\text{M}-\text{NH}_4]^-$  found  $m/z$  282.9832, calculated for  $\text{C}_{12}\text{H}_8^{35}\text{ClO}_4\text{S}$  282.9837.

**3-Sulfoxy-4-chloro-biphenyl, ammonium salt (3PCB 3 sulfate):** Deprotection of sulfuric acid 4-chlorobiphenyl-3-yl 2,2,2-trichloroethyl sulfate (474 mg) with zinc powder-ammonium formate in ethanol yielded the desired ammonium salt (320 mg; 93 %) as a white solid (Li et al. 2010), mp: 85 °C (dec.);  $^1\text{H}$  NMR (400 MHz,  $\text{CD}_3\text{OD}$ ):  $\delta/\text{ppm}$  7.33-7.45 (m, 4H), 7.47 (d, 1H,  $J = 8.4$  Hz), 7.62 (AA'XX' system, 2H), 7.89 (d, 1H,  $J = 2.4$  Hz);  $^{13}\text{C}$  NMR (100 MHz,  $\text{CD}_3\text{OD}$ ):  $\delta/\text{ppm}$  122.2, 125.0, 126.7, 127.9, 128.8, 130.0, 131.4, 140.8, 142.2, 150.2; HRMS (ESI, negative):  $[\text{M}-\text{NH}_4]^-$  found  $m/z$  282.9840, calculated for  $\text{C}_{12}\text{H}_8^{35}\text{ClO}_4\text{S}$  282.9837.

**4-Sulfoxy-3,4',5-trichloro-biphenyl, ammonium salt (4PCB 39 sulfate):** Deprotection of sulfuric acid 3,4',5-trichlorobiphenyl-4-yl 2,2,2-trichloroethyl sulfate (170 mg) with zinc powder-ammonium formate in ethanol yielded the desired ammonium salt (70 mg; 54 %) as a white solid (Li et al. 2010), mp: 150 °C (dec.);  $^1\text{H}$  NMR (400 MHz,  $\text{CD}_3\text{OD}$ ):  $\delta/\text{ppm}$  7.45 (AA'XX' system, 2H), 7.58 (AA'XX' system, 2H), 7.62 (s, 2H);  $^{13}\text{C}$  NMR (100 MHz,  $\text{CD}_3\text{OD}$ ):  $\delta/\text{ppm}$  128.2, 129.5, 130.2, 132.1, 135.4, 138.1, 139.5, 146.7; HRMS (ESI, negative):  $[\text{M}-\text{NH}_4]^-$  found  $m/z$  350.9049, calculated for  $\text{C}_{12}\text{H}_6^{35}\text{Cl}_3\text{O}_4\text{S}$  350.9058.

**4-Sulfoxy-2,2',5',6-tetrachloro-biphenyl, ammonium salt (4PCB 53 sulfate):** Reaction of 2,2',5',6-tetrachloro-biphenyl-4-ol (210 mg) (Joshi et al. 2011) and 2,2,2-trichloroethyl chlorosulfate (203 mg) in the presence of  $\text{Et}_3\text{N}/\text{DMAP}$  in THF yielded 320 mg (90 %) of crude 2,2',5',6-tetrachloro-biphenyl-4-yl 2,2,2-trichloroethyl sulfate as a colorless oil. Subsequent deprotection of the crude TCE-protected PCB sulfate (300 mg) with zinc powder-ammonium formate in ethanol yielded the desired ammonium salt (190 mg; 81 %) as a white solid (Li et al. 2010), mp: 140 °C (dec.);  $^1\text{H}$  NMR (400 MHz,  $\text{CD}_3\text{OD}$ ):  $\delta/\text{ppm}$  7.29 (d, 1H,  $J = 2.4$  Hz), 7.45 (dd, 1H,  $J = 2.4$  Hz,  $J = 8.4$  Hz), 7.66 (s, 1H), 7.53 (d, 1H,  $J = 8.4$  Hz);  $^{13}\text{C}$  NMR (100 MHz,  $\text{CD}_3\text{OD}$ ):  $\delta/\text{ppm}$  121.9, 131.2, 131.9, 132.3, 133.2, 133.7, 133.8, 135.9, 138.6, 154.8; HRMS (ESI, negative):  $[\text{M}-\text{NH}_4]^-$  found  $m/z$  386.8637, calculated for  $\text{C}_{12}\text{H}_5^{35}\text{Cl}_4\text{O}_4\text{S}$  384.8668.

**MCF7 cell culture**—The estrogen-sensitive human breast cancer cell lines MCF-7 BOS and MCF7-AR1 were kindly provided by Drs. Ana M. Soto and Carlos Sonnenschein (Tufts University School of Medicine, Department of Anatomy and Cellular Biology, Boston, Massachusetts). For maintenance cells were grown in Dulbecco's modified Eagle's medium (DMEM) supplemented with 10% fetal bovine serum (FBS), 100 U/ml penicillin and 100  $\mu\text{g}/\text{ml}$  streptomycin (DMEM/10%FBS), at 37°C with 6%  $\text{CO}_2$  in a humidified incubator as described by DeCastro and coworkers (DeCastro et al. 2006).

**Cytotoxicity measurement with Resazurin**—To examine cell viability  $5 \times 10^3$  MCF-7 BOS cells were seeded in 100  $\mu\text{l}$  DMEM/10%FBS per well in 96-well plates. After 24 h cells received fresh medium containing test compounds at a range from 1pM or 5 nM to 100  $\mu\text{M}$  or solvent alone (DMSO, final concentration 0.5% in medium). Normal medium and FCS was used in this assay to enable the assessment of cytotoxic and proliferation

inhibiting activity of the test compounds, since these cells do hardly proliferate in medium with charcoal-treated FBS as used below in the endocrine disruption assay. After 120 h exposure the medium was removed, cells were washed once with phosphate buffered saline (PBS) and incubated in phenol red-free DMEM/10%FBS containing resazurin (50  $\mu$ M) for 1 h. The assay measures the reduction of non-fluorescent resazurin to fluorescent resorufin by viable, metabolically active cells. The rate of fluorescence is directly proportional to the number of viable cells per well. The fluorescence was measured with a Tecan Genios Pro plate reader at 535 nm excitation and 590 nm emission wavelengths.

#### **E-screen with MCF7-BOS cells to test estrogenicity and anti-estrogenicity—**

Estrogenicity of a test compound is assessed in the E-screen assay by measuring compound-induced proliferation of MCF-7 BOS human breast cancer cells in estrogen-free medium. These cells respond to estrogenic compounds that bind to the cytoplasmic estrogen receptor (ER) with stimulated cell proliferation. According to a technique slightly modified from Soto et al. (1995),  $3 \times 10^4$  MCF7-BOS cells per well in 1 ml DMEM/10%FBS were seeded into 24-well plates. A 48 h attachment time was used to obtain good adherence of the cells to the well surface. The cells were then carefully washed with phenol red-free DMEM supplemented with 5% Charcoal-Dextran-treated FBS (DMEM/5%CD-FBS) and then incubated in duplicates for each data point in this DMEM/5%CD-FBS with test compounds (concentration range of 1 pM – 100  $\mu$ M) or 0.5% DMSO (hormone-free negative control) or estradiol (E2, 100 pM, positive control). After 120 h incubation the wells were checked visually to assure that the samples were subconfluent, the medium was removed, and the Sulforhodamine B (SRB) protein assay was used to evaluate cell growth. Briefly, cells were fixed onto the plates with cold 10% trichloroacetic acid for 30 min at 6°C, then washed 3 times with water and afterwards air dried. Cells were stained with 0.4% SRB in a 1% acetic acid solution for 20 min at room temperature and afterwards washed with 1% acetic acid until the wash solution in the plate was clear. After air drying of the fixed and stained cells, exactly 600  $\mu$ L Tris Base Buffer (10mM) was added to each well, the plates were shaken rapidly for at least 10 min until the color was extracted from the cells and a homogeneous solution was visible. Absorption of the solution was measured at 540 nm using a microtiter plate reader (Tecan Genios Pro) which is proportional to the cell number/protein content per well. To confirm that the proliferation induction was ER mediated, parallel cultures were prepared and co-exposed to the ER antagonist ICI 182,780 (10 nM in 0.1% DMSO) and the test compound which should give negative results.

The estrogenic activity of xenobiotics was assessed by calculating the Relative Estrogenic proliferative Effect (REE), which is 100 times the ratio of the highest proliferative effect (PE) obtained with the chemical (Compound PE-1) compared to 100 pM E2 (E2 PE-1), i.e. the formula used is:  $100 \times \text{compound PE-1}/\text{E2 PE-1}$ . The REE compares quantitatively the magnitude of the proliferative response of the test compound, the efficacy of the compound compared to the one of E2. This response is then categorized as follows: REE = 80-100% as full agonist, 25-80% as partial agonist, 10-25% as weak agonist, and 10% a negative result. Also calculated were the Relative Estrogenic Potency (RPP) which is the ratio between the minimal concentration of E2 needed for maximal cell yield and the minimal concentration of the test compound needed to achieve a similar effect times 100.



Anti-estrogenicity: The E-screen can also assess anti-estrogenicity by inducing the maximal proliferation of MCF7-BOS cells with E2 and measuring the reduction in proliferation that occurs as a result of co-incubation with the test compound. Cells were seeded into 24-well plates and 48 h later washed with phenol red free DMEM/5% CD-FBS as described above and then treated in duplicates with E2 (100 pM in 0.1% DMSO) in the presence of different concentrations (1 pM - 100 µM) of the test compound or 0.5% DMSO. After 5 days exposure the SRB measurement was carried out as described above. A reduction of the stimulated cell proliferation effect from E2 indicates an anti-estrogenic activity.

#### **A-screen with MCF7-AR1 cells to test androgenicity and anti-androgenicity—**

Androgenicity was examined with MCF7-AR1 cells which are MCF7-BOS cells stably transfected with the human androgen receptor (AR). If a test compound acts as androgen it induces an AR-mediated proliferative shutoff and reduces the proliferation induced through E2. According to a technique slightly modified from Soto et al. (1998),  $5 \times 10^4$  MCF7-AR1 cells were seeded into 24-well plate in 1 ml DMEM/10%FBS. This little higher seeding cell number compared to the MCF7-BOS cells was used to compensate for the slightly slower proliferation and attachment rate of the MCF7-AR1 cells. After 48h attachment time cells were carefully washed with phenol red-free DMEM/5%CD-FBS and then co-treated in duplicates with different concentrations (1 pM – 100 µM) of the test compound or 0.5% DMSO and E2 (100 pM in 0.1% DMSO). After 120 h the assay was stopped by removing the medium from the wells and SRB measurement was carried out.

The Relative androgenic effect (RAE) was assessed by calculating a ratio from the reduced cell yield of R1881 (500 pM), the androgen 'gold standard', and the most significant reduced cell yield from the test compound times 100.

Anti-androgenicity:  $5 \times 10^4$  MCF7-BOS cells were seeded as described above and 48 h later washed with phenol red-free DMEM/5%CD-FBS and then co-treated with different concentrations (100 µM – 1 pM) of the test compound and E2 (100 pM in 0.05% DMSO) and R1881 (10 µM in 0.05% DMSO), an androgen receptor agonist. After 120 h the assay was stopped and SRB measurement was carried out. If test compound blocks the androgen-induced inhibition of cell proliferation by R1881 and stimulates cell proliferation it acts as an anti-androgen.

### **Statistics**

Experiments were performed in duplicates and every experiment was repeated at least once. A test compounds evaluated with the E-screen assay was considered estrogenic if it induced a statistically significant increase in cell yield over the hormone-free control. Results were normalized to steroid-free control 1% (for Estrogenicity), E2 (Anti-Estrogenicity and Androgenicity) or E2+R1881 (Anti-Androgenicity) and expressed as the mean  $\pm$  standard error (SE). Differences between treatments were analyzed statistically by using one-way ANOVA with Dunnett's test at the 5% level. Levels of significance are indicated as: \*  $p < 0.05$ ; \*\*  $p < 0.01$  and \*\*\*  $p < 0.001$ .

## Results

### Synthesis and characterization of PCB sulfates

Eight PCB sulfates, including five sulfate esters of HO-PCB 3s, were selected to investigate their estrogenicity and androgenicity *in vitro* (Fig. 1). Four new PCB sulfates, including 2PCB 3 sulfate, 3PCB 3 sulfate, 4PCB 39 sulfate and 4PCB 53 sulfate, were synthesized as the respective ammonium salts using a published approach (Grimm et al. 2013b, Li et al. 2010) and fully characterized (see Materials and Methods). This approach allows the preparation of milligram quantities of PCB sulfates in high purity (> 99%) as determined using thin layer chromatography (Li et al. 2010) and HPLC (Grimm et al. 2013b). Briefly, the corresponding HO-PCB was reacted with 2,2,2-trichloroethyl chlorosulfate (Hedayatullah et al. 1972) to yield a 2,2,2-trichloroethyl (TCE-)PCB sulfates diester. Deprotection with zinc powder-ammonium formate yielded the ammonium salt of the corresponding PCB sulfate. The synthesis and characterization of the other four PCB sulfates has been reported earlier (Grimm et al. 2013b, Li et al. 2010). The ammonium salts of all eight PCB sulfates are colorless solids and well soluble in DMSO. PCB sulfates can be stored both as neat compound and solution in DMSO for several months without detectable degradation to the corresponding HO-PCBs.

### Cytotoxicity/cell viability

To establish the experimental concentration range for the cell proliferation assays, the cytotoxicity of PCB sulfates was determined at concentrations from 5 nM or 1pM to 100  $\mu$ M after 120 h (5 days) of exposure in DMEM/10%FBS medium. The percent of life cells compared to solvent treated controls was quantified using the Resazurin assay, an assay based on metabolic activity of living cells. Cells were also visually inspected using an inverted microscope to evaluate viability, cell number and morphology of the cells. One compound, 4'-HO-PCB 3, was 100% toxic at 100  $\mu$ M, and this concentration was therefore not used in the E- and A-screen assay. Also, 4PCB 53 sulfate was weakly cytotoxic after 120 h. Since the reduction of viable cells was only 16% compared to the control this concentration was included in the E- and A-screen (Supplementary Material, Fig. S1). No significant cytotoxicity was observed with any of the other PCB sulfates up to concentrations of 100  $\mu$ M.

### E-screen for estrogenic and anti-estrogenic activity

The MCF7-BOS cell line was used to measure estrogenicity/induction of cell proliferation in estrogen-free medium. 4'-HO-PCB 3, used as a positive PCB control, was the most potent of all tested PCB metabolites, inducing significant cell proliferation at 1  $\mu$ M concentration (RPP of 0.1%; Fig 2-1 and Table 1). Four of the eight tested PCB sulfate monoesters also showed estrogenic activity but only at the highest concentration tested, 100  $\mu$ M, equivalent to an RPP of 0.001% (Fig. 2-1 and Table 1). Of these the 4'PCB 3 sulfate had the highest estrogenic efficacy, surpassing the effect seen with its parent compound 4'-HO-PCB 3 at 1  $\mu$ M, and reaching that of E2, although only at a 6 orders of magnitude higher concentration (100  $\mu$ M vs 100 pM, respectively, RPP of 0.001% and highest REE of 78%). The other sulfate esters followed in the order 2'PCB 3 sulfate > 4PCB 39 sulfate > 4PCB 53 sulfate at 100  $\mu$ M. The addition of an ER antagonist ICI 182,780 in parallel cultures confirmed that



these inductions of proliferation were mediated through ER activation since ICI 182,780 completely blocked the effect (Fig. S4). The other tested sulfate esters, 2PCB 3 and 3PCB 3 sulfate, 3'PCB 3 sulfate and 4PCB 11 sulfate did not show any estrogenicity at any concentration investigated, i.e. no higher cell proliferation compared to DMSO controls (Supplementary Material, Fig. S2).

Five of the eight PCB sulfate monoesters showed anti-estrogenic activity by reducing the E2 stimulated cell proliferation (Fig. 2-2). 4'PCB 3 sulfate was anti-estrogenic at almost all, i.e. 4 out of 5 concentrations tested, ranging from 1 pM up to 100 μM. Similarly 2'PCB 3 sulfate and 3'PCB 3 sulfate were anti-estrogenic at 3 of 5 concentrations tested, particularly at the lower ones. 4PCB 39 sulfate was only active at 1 pM and 4PCB 53 sulfate only at 10 nM and 100 μM concentrations. 2PCB 3 sulfate, 3PCB 3 sulfate, and 4PCB 11 sulfate did not cause any significant reduction of cell proliferation compared to E2 controls at the tested concentrations, neither did 4'-HO-PCB 3 (Supplemental Material, Fig S2).

### A-screen for androgenicity and anti-androgenicity

Five out of eight tested PCB sulfate monoesters had androgenic effects over a wide range of concentrations (some from 1 pM to 100 μM). 4PCB 53 sulfate at 100 μM had the highest androgenic efficacy with a RAE of 70%, followed by 4'PCB 3 sulfate (100 μM; RAE of 60%, Fig 3-1 and Table 1). 4'PCB 3 sulfate and 3'PCB 3 sulfate had the highest potency, exhibiting significant androgenic activity at a concentration of 1 pM. Both compounds were androgenic at 4 of the 5 concentrations tested over a range of 1pM to 100 μM. 4PCB 11 and 4PCB 39 were less androgenic, each at 2 of the concentrations in the middle range of concentrations tested with a RAE of 51% and 48% respectively. This qualifies these 4 PCB sulfates as partial agonists. In contrast, 2'PCB 3 sulfate, 2PCB 3 sulfate and 3PCB 3 sulfate did not show any androgenicity, i.e. they did not lower E2 induced cell proliferation at any tested concentration (Supplemental Material, Fig. S3).

Only two (2'PCB 3 and 3'PCB 3 sulfate) of the eight PCB sulfate monoesters showed weak anti-androgenic activity, i.e. higher cell proliferation compared to the E2 + R1881, and both had this effect at very low concentrations (100 pM and 1 pM, Fig 3-2). All other seven compounds did not increase cell proliferation at the concentrations tested (Supplementary Material, Fig. S3).

### Comparison of ED effects of all tested compounds

These newly synthesized PCB sulfate monoesters are not cytotoxic up to 100 μM concentration in MCF-7 cells. Table 1 provides a summary of calculations for relative estrogenic effect/efficacy (REE), relative estrogenic potency (RPP) and relative androgenic effect/efficacy at the concentrations with the most significant effect. Two compounds (2PCB 3 sulfate and 3PCB 3 sulfate) did not act as disruptor in any of the screens. While 4'-HO-PCB 3 acted as potent estrogen with a RPP of 0.1% and REE of 60% at 1 μM concentration, some sulfates like the 4'PCB 3 sulfate acted like estrogens only at very high concentrations (100 μM) with an RPP of 0.001% and REE of 78%. Anti-androgenicity was restricted to 2 sulfates while 5 sulfates had anti-estrogenic and/or androgenic activity. We saw the most significant effect of androgenic activity with the 4PCB 53 sulfate at 100 μM with a RAE of

70%. 4'-HO-PCB 3 also displayed androgenic activity at two concentrations and the most significant one at 100  $\mu$ M resulted in an RAE of 55%. However, with most compounds no classical dose-response curves (increasing effect with increasing concentration) were seen in our experiments. These preliminary data suggest that these PCB sulfate monoesters have low cytotoxic potential in human breast cancer cells as well as low potential to act as endocrine disrupting agent.

## Discussion

PCBs are ubiquitous human and environmental contaminants and cause a multitude of adverse health effects, but the mechanisms of action are often not known. Lower chlorinated PCBs undergo oxidative metabolism to hydroxylated PCBs which are important contributors to PCB toxicity (Grimm et al. 2015a, Ludewig et al. 2008). These metabolites are substrates of cytosolic phase II enzymes. Using a rat model, Dhakal and co-workers recently discovered that sulfation is a major phase II pathway, that sulfate metabolites of PCB 3 were detected in blood and urine even after low dose inhalation exposure, and that the concentration of PCB 3 sulfates exceeds those of their respective HO-PCBs in serum *in vivo* (Dhakal et al. 2013, Dhakal et al. 2012, Dhakal et al. 2014). Also a new study showed that the sulfated metabolites of PCB 3 and PCB 11 are high affinity ligands of transthyretin *in vitro* and could play a role in Alzheimer disease as inhibitors of transthyretin fibrillogenesis (Grimm et al. 2013a, 2015b). Very little is known about this dominant class of PCB metabolites and their toxicity. To gain better knowledge about their cytotoxicity and potential biological activity as hormone mimics we synthesized and evaluated all possible mono-sulfates of PCB 3, a sulfate metabolite of PCB 11, a non Aroclor congener discovered in high levels in paint, indoor and outdoor air, and human blood (Hu et al. 2008, Marek et al. 2013), and one congener each with 3 and 4 chlorines in specific positions, PCB 39 with no *ortho* chlorines and PCB 53 with 3 *ortho* chlorines, respectively (Hu & Hornbuckle 2010). To our knowledge this is the first study examining the potential of PCB sulfates for (anti-)estrogenic and (anti-)androgenic activities and cytotoxicity in cells in culture.

In recent years global concerns have been raised that about 10 percent of man-made chemicals could affect the endocrine system even though only a handful of endocrine disruptors have been confirmed so far (Bergman et al. 2013). Human and wildlife health depends on the ability to reproduce and develop normally and endocrine disruptors were specifically considered the new theme of the United Nations Integrated Management of Chemicals 2012 initiative and are the emphasis of the US EPA Endocrine Disruptor Screening Program. Among the ED test systems the E-screen and A-screen are simple and stable assays to determine the activity of compounds to interact with the ER and AR steroid hormone receptors (Soto et al. 1995, Szelei et al. 1997). We used these assays to gain first insight into the endocrine disrupting activity of PCB sulfate metabolites.

The cytotoxicity prescreening showed that only 4'-HO-PCB 3, not its 4'PCB 3 sulfate metabolite or any of the other 6 PCB sulfate metabolites tested, had significant cytotoxic effects in MCF7-BOS cells after 120 h of exposure. This indicates that the introduction of a sulfate group strongly reduces the cytotoxicity of the hydroxylated parent compound. This conclusion is further supported by the results of Machala et al. who reported a cytotoxicity

LOEC of 50  $\mu\text{M}$  for 4'-HO-PCB 3 and 20  $\mu\text{M}$  for 2'- and 3'-HO-PCB 3 in T47D.Luc cells (Machala et al. 2004), while we did not see any toxicity up to 100  $\mu\text{M}$  with the sulfate derivatives of these compounds in our assay. Thus phase II metabolism is a true detoxification mechanism for these HO-PCBs with respect to cytotoxicity.

A recent study of 20 PCB congeners employing an ER dual-luciferase reporter gene assay reported that PCB 18, 28, 49, 52, 99, 101, 103, 110, and 128 exhibited estrogenic effects, whereas PCB 118, 138, 163, 170, 180, 187, 194, 199 and 203 behaved as anti-estrogens and PCB 30 and 44 had both effects (Zhang et al. 2014). This infers that less chlorinated PCBs behave estrogenic while higher chlorinated ones behave as anti-estrogens. Machala and coworkers reported significant ER-mediated activity with hydroxylated low molecular weight PCBs like 3'-HO-PCB3 and 4'-HO-PCB 3 in human mammary cell lines MVLN and T47D.Luc transfected with a luciferase reporter gene. They screened 16 mono-, 10 di-hydroxy and 6 quinone metabolites of PCBs with 1 to 5 chlorine substituents. Only mono-hydroxylated PCBs with 1 or 2 chlorines, one 3,4-diOH congener, and 4'-HO-PCB 34 had strong estrogenic activity, displayed at low  $\mu\text{M}$  concentrations in both cell lines (Machala et al. 2004). This is in agreement with our observation of an estrogenic effect with 4'-HO-PCB3 at 1  $\mu\text{M}$  concentration and a REE of 60% and RPP of 0.1%. We now could show that sulfation strongly reduced the efficacy of these metabolites as estrogens. Machala and coworkers also reported that HO-PCBs had significantly higher estrogenic potencies in MVLN and T47D.Luc cells than E2 and they acted in an additive way with E2 (Machala et al. 2004). Of our sulfate derivatives, only 4'PCB 3 sulfate produced an estrogenic effect with a REE of 78% and RPP of 0.001% that came close to the activity of E2 (a REE of 80% is considered a full agonist). This sulfate also had an anti-estrogenic effect in the presence of E2. Machala et al. reported anti-estrogenic activity with most mono- and di-hydroxylated metabolites, but only at concentrations in the cytotoxic range (Machala et al. 2004). Connors and coworkers used a battery of assays to examine estrogenic/anti-estrogenic activity of OH-PCBs and found that most compounds bound to the ER with high affinity, but only few activated the ER while many inhibited ER activation (Connor et al. 1997). We observed anti-estrogenic activity with 5 of 7 sulfate metabolites of PCBs. These findings suggest that while hydroxylation increases estrogenicity of PCBs, sulfation of these hydroxyl-metabolites decreases the estrogenic activity but increases/creates an anti-estrogenic activity. It should also be emphasizes that 4 of the sulfates are anti-estrogenic at very low, nano- and pico-molar concentrations and estrogenic only at the highest concentration tested (100  $\mu\text{M}$ ), suggesting that the anti-estrogenicity should be the major concern *in vivo*.

Very few studies examined the effects of PCBs and their metabolites on the androgen receptor. Endo and coworkers reported concentration-dependent anti-androgenic activity of two dioxin-like congeners, PCB77 and PCB126, in LNCap prostate cancer cells (Endo et al. 2003). The *ortho*-substituted PCB118 and PCB153 had a biphasic effect, stimulated cell proliferation at low and reduced proliferation at high concentrations. The authors suggest both AR and AhR mediated pathways as well as modulation of post-transcriptional targets as mechanisms. The PCB mixtures Delor 103 and Aroclor 1254 activated the human AR in a reporter assay *in vitro* (Portugal et al. 2002, Svobodova et al. 2009), while Aroclor mixtures (1248 > 1254 > 1242 > 1260) and the congeners PCB 42, 128, and 138, but not

PCB 31, 99, 118, 153, 168, 180, or 198 acted as anti-androgens mostly at  $\mu\text{M}$  concentrations (Bonfeld-Jorgensen et al. 2001, Portugal et al. 2002, Sperry & Thomas 1999). Fish have two androgen receptors and PCB mixtures, Aroclor 1254 > 1260, and hydroxylated PCB congeners displaced DHT from AR2 at medium to high  $\mu\text{M}$  concentrations (Sperry & Thomas 1999). We observed androgenic activity with the 4'-HO-PCB 3 with a RAE of 55% and five sulfate metabolites and anti-androgenicity with two PCB sulfates, with most of these effects visible in the pico- or nanomolar range. Unlike the interaction with the ER, only one compound, 3'PCB 3 sulfate, both activated and inhibited the AR and both reactions occurred at the same picomolar concentration. Overall it appears that many PCBs have anti-androgenic activity whereas HO-PCBs and their sulfate metabolites are more prone to be activators of the androgen receptor.

Six of eight compounds tested showed significant androgenic and/or anti-estrogenic/androgenic activity at more than one concentration tested, but the effect was not dose-dependent, often spanning several orders of magnitude and starting at a picomolar concentration, suggesting that these compounds could interfere with the endocrine system at very low concentrations. Endocrine-disrupting chemicals may display U-shaped or inverted U-shaped non-monotonic dose-response curves and it was reported that they rarely exhibit a clear dose-response (Vandenberg et al. 2012, Zoeller & Vandenberg 2015). This weak, constant effect instead of the more common dose-response may be due to specificities of the assay that limit the anti-estrogenic and androgenic effect to a range between no-effect to estrogen-induced level.

We were interested in structure-activity relationships (SAR). Since only a limited number of compounds could be synthesized and screened we decided to examine 1) the effect of the 5 possible different positions of the sulfate group on the PCB 3 molecule and 2) the effect of non-*ortho* (PCB 39) vs *ortho*-chlorinated (PCB 53) and 3) *para*- (PCB 3, 39) vs *meta*-chlorinated (PCB 11, 53) metabolites with the sulfate in *para* position. Overall some interesting SARs emerged from these studies that need further examination. The position of the sulfate group on the PCB 3 molecule had a strong influence. If it was in the same ring with the chlorine group it abolished any interaction with the ER and AR (2PCB 3 and 3PCB 3 sulfate). Compounds with the sulfate group in the other ring were active in 3 of the 4 endpoints, with 3'PCB 3 sulfate being not estrogenic, 2'PCB 3 sulfate not androgenic, and 4'PCB 3 sulfate not anti-estrogenic. This latter compound was also the most potent estrogenic compound of all sulfates tested, suggesting that the long *para-to-para* alignment of the substitutions favors interactions with the ER. Unexpectedly and opposite to findings with pure PCB congeners, non-*ortho* vs *ortho*-chlorination did not have an influence in these assays, while the one compound with only *meta*-chlorination (4PCB 11) only showed some androgenicity. It is of interest that the only anti-androgenic compounds (2'PCB 3, 3'PCB 3) are the PCB 3's with sulfates in the opposite ring in *ortho*- and *meta*- position.

The ER and AR are very similar in structure. The C-terminal region contains the ligand binding domain. This region consists of 11  $\alpha$ -helices that form the barrel-like hydrophobic pocket of the binding cavity. A 12<sup>th</sup>  $\alpha$ -helix at the extreme end of the C-terminus blocks the coactivator binding site, but changes the position to 'cap' the lid over the barrel after binding of a ligand, thereby exposing the coactivator binding site (Pike et al. 2000, Pike et al. 1999,

Portugal et al. 2002). The binding pocket can accept a large number of compounds with those that fit only loosely acting as weak activators while those that have portions of their structure protruding over the edge acting as antagonists (Eick et al. 2012, White et al. 1994). It can be speculated that having a hydrophilic sulfate in the chlorinated ring (2PCB 3 and 3PCB 3 sulfates) may prevent entry into the barrel-like structure of the ER and AR thereby preventing any interaction with these receptors. *Ortho*- and *meta*- sulfate in the opposite ring may be hanging over the lid of the barrel, thereby hindering the closure and activation of the ER (3'PCB 3 sulfate) or AR (2'PCB 3 sulfate) receptor.

A negative correlation was observed between PCBs and sperm motility in fish (Jenkins et al. 2014), a positive correlation between PCBs and abnormal phenotypic sex differentiation in rodents (Gray & Kelce 1996) and cryptorchidism in boys (Brucker-Davis et al. 2008), an inverse correlation between PCBs and blood testosterone levels in humans and polar bears (Goncharov et al. 2009, Oskam et al. 2003), and a positive correlation between PCBs and feminization of behavior in male mice (Fanini et al. 1990). This very partial list of PCB effects on development and reproduction indicate that PCBs as a group have endocrine disrupting activity. However, our results suggest that this activity may, at least in part, be mediated by their metabolites, including the sulfate esters and that the type of endocrine activity, estrogenic, androgenic, anti-estrogenic, anti-androgenic, or no activity, depends on the metabolites formed. This supports the hypothesis that the specificity of the steroid hormone receptors action is a consequence of endogenous metabolic activity (Lathe & Kotelevtsev 2014).

## Conclusions

Our results show that the sulfate metabolites of PCBs are less toxic and have less estrogenic activity than their hydroxylated progenitors, but that they may have anti-estrogenic and androgenic activity at very low concentrations. Thus it is not clear whether sulfation is a true detoxification or possibly an activation pathway, at least with respect to endocrine activity. Sulfate metabolites appear to be the major product of phase II metabolism of HO-PCBs and even though their endocrine activity was mostly relatively small, it occurred at very low concentrations. PCB 3 sulfates were readily detectable in blood and urine of rats even after low-dose inhalation exposure (Dhakal et al. 2014) and were also identified in poplar trees (Zhai et al. 2013) however, it is currently unclear if PCB sulfates are also major PCB metabolites in humans. This suggests that more investigations into their occurrence and effects are warranted.

## Supplementary Material

Refer to Web version on PubMed Central for supplementary material.

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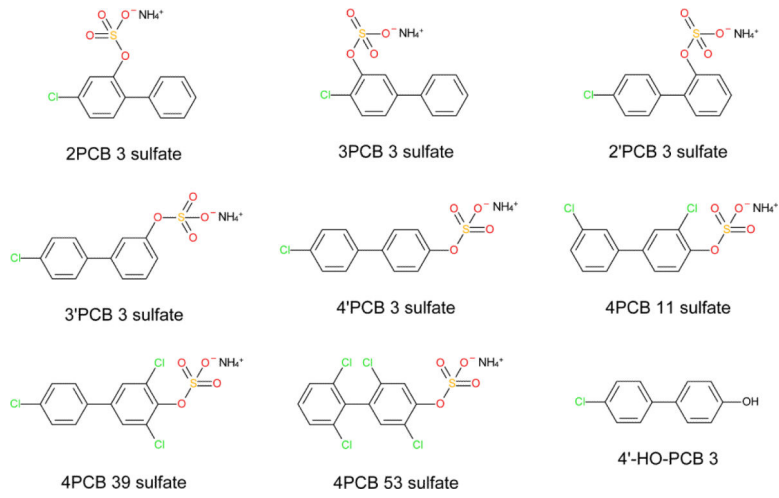
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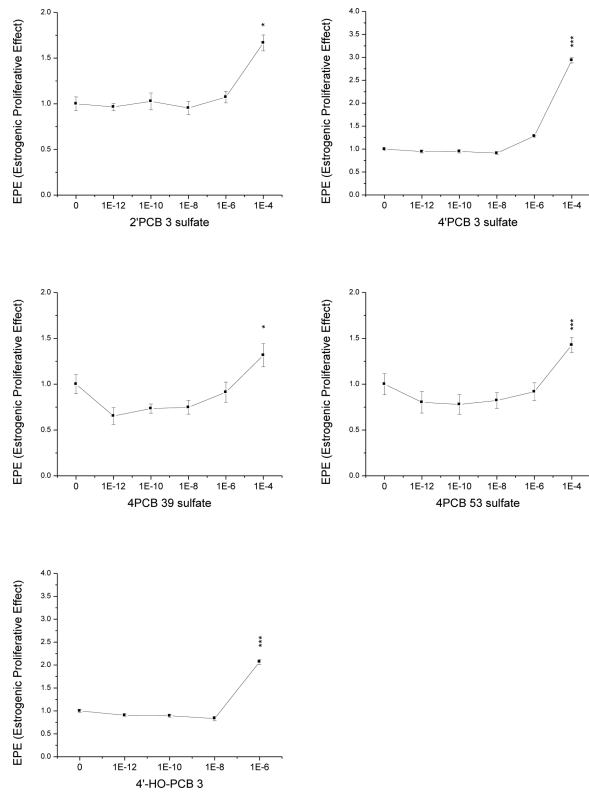
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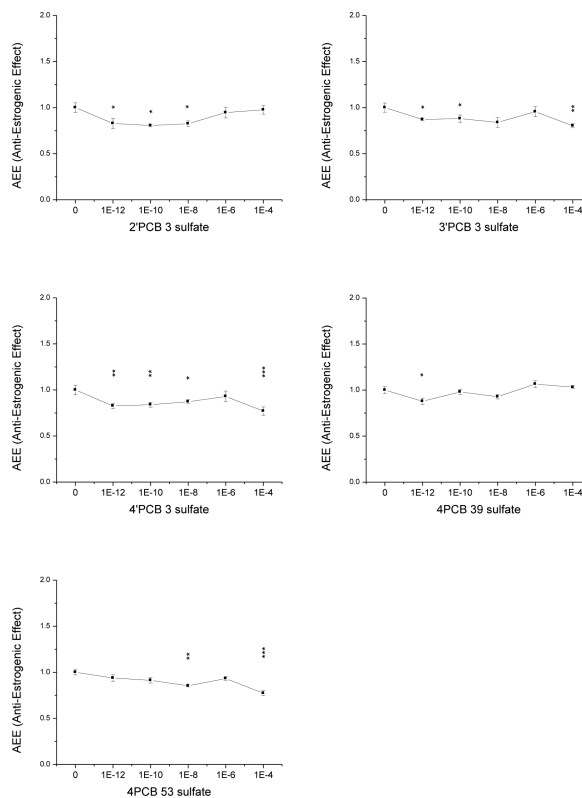


**Fig. 1. Chemical structures of PCB sulfates and HO-PCB3**

The PCB sulfates used in this study were synthesized as ammonium salts.

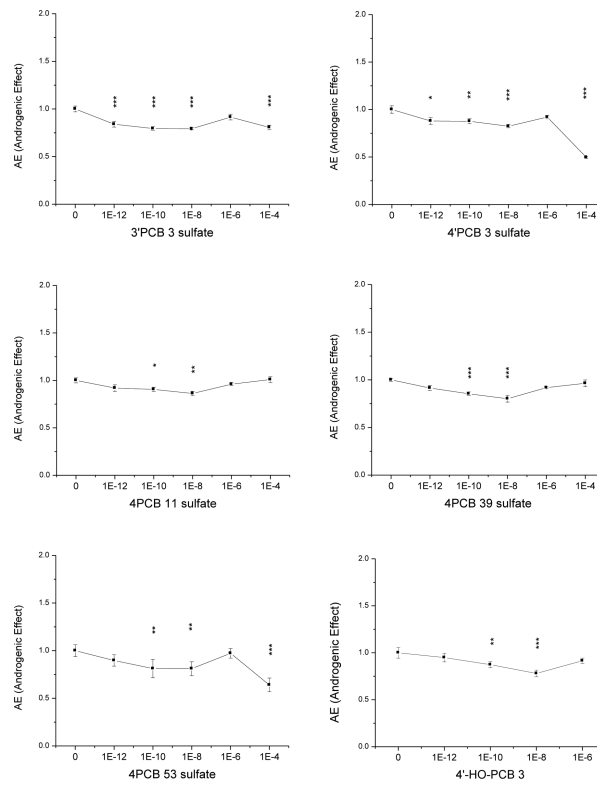


**Fig. 2-1. Estrogenic Proliferative Effect (EPE) of five PCB sulfates and 4'-HO-PCB 3**  
 Normalized proliferative data compared to control (0.5% DMSO). Data show the mean of 2 individual experiments with duplicate samples  $\pm$  SE. Levels of significance are: \*  $p < 0.05$ ; \*\*  $p < 0.01$  and \*\*\*  $p < 0.001$ .

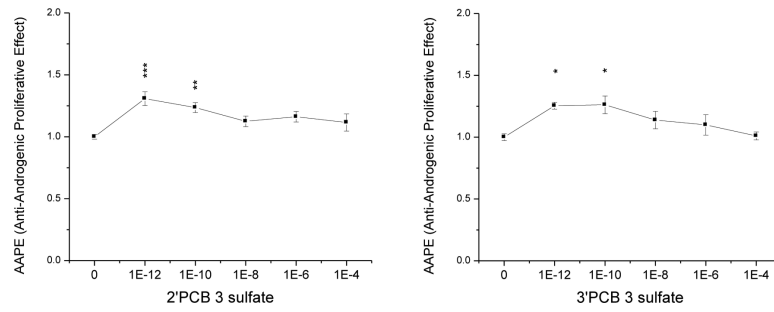


**Fig. 2-2. Anti-Estrogenic Effect (AEE) of five PCB sulfates and 4'-HO-PCB 3**  
 Normalized proliferative data compared to 100 pM E2. Data depicted are the mean of 2 individual experiments with duplicate samples  $\pm$  SE. Levels of significance are: \* p < 0.05; \*\* p < 0.01 and \*\*\* p < 0.001.





**Fig. 3-1. Androgenic Effect (AE) of six PCB sulfates and 4'-HO-PCB 3**  
 Normalized proliferative data (control is 100 pM E2). Data depicted the mean of 2 individual experiments, each with duplicate samples,  $\pm$  SE. Levels of significance: \*  $p < 0.05$ ; \*\*  $p < 0.01$  and \*\*\*  $p < 0.001$ .



**Fig. 3-2. Anti-Androgenic Proliferative Effect (AAPE) of two PCB sulfates**

Data were normalized against the control (100 pM E2 + 500 pM R1881). Data depicted are the mean of 2 individual experiments, each with duplicate samples,  $\pm$  SE. Levels of significance: \*  $p < 0.05$ ; \*\*  $p < 0.01$  and \*\*\*  $p < 0.001$ .

**Table 1**

Summary of the E- and A-screen results for PCB sulfates. The values are calculated for the most significant effect if there were more than one significant concentration

compound	REE <sup>1</sup> , %	RPP <sup>2</sup> , %	RAE <sup>3</sup> , %
17P-E2	100	100	--
R1881	--	--	100
2'PCB 3	25	0.001	--
3'PCB 3	--	--	38
4'PCB 3	78	0.001	60
2PCB 3	--	--	--
3PCB 3	--	--	--
4PCB 11	--	--	51
4PCB 39	20	0.001	48
4PCB 53	26	0.001	70
4'-HO-PCB 3	60	0.1	55