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## DELIVERY OF ZICONOTIDE TO CEREBROSPINAL FLUID VIA INTRANASAL PATHWAY FOR THE TREATMENT OF CHRONIC PAIN

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### Abstract

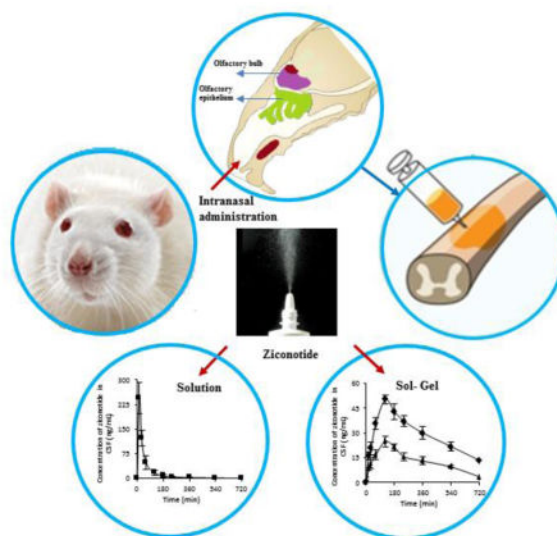
The purpose of the current study was to investigate the plausibility of delivery of ziconotide to the cerebrospinal fluid (CSF) via intranasal administration. Ziconotide was administered either in the form of solution or Kolliphor P 407 gels (KP 407) intranasally in Sprague-Dawley rats. The effect of incorporation of chitosan in the formulation was also investigated. Time course of drug in the CSF was investigated by collecting CSF from cisterna magna. Pharmacokinetics of ziconotide in CSF following intrathecal and intravenous (i.v) administration of ziconotide was investigated. Upon intrathecal administration the elimination rate constant of ziconotide in CSF was found to be  $1.01 \pm 0.34 \text{ h}^{-1}$ . The  $C_{\text{max}}$  and  $T_{\text{max}}$  of ziconotide in CSF following intravenous administration were found to be  $37.78 \pm 6.8 \text{ ng/mL}$  and  $\sim 2 \text{ h}$  respectively. The time required to attain maximum concentration ( $T_{\text{max}}$ ) in CSF was less upon intranasal administration (15 min) compared to i.v administration (120 min). Presence of chitosan enhanced the overall bioavailability of ziconotide from intranasal solution and gel formulations. The elimination rate constant of ziconotide in CSF following intranasal and intravenous administration of ziconotide solution was found to be  $0.54 \pm 0.08 \text{ h}^{-1}$  and  $0.42 \pm 0.10 \text{ h}^{-1}$  respectively. Whereas, intranasal administration of ziconotide in the form of *in situ* forming gel lowered the elimination rate significantly. These results suggest that intranasal administration could be a potential noninvasive and patient compliant method of delivering ziconotide to CSF to treat chronic pain.

### Graphical Abstract

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## Introduction

Treatment of chronic or recurring pain is a major therapeutic challenge. Chronic pain is a widely prevalent universal problem and has been reported that more than 15% of the world's population experience chronic pain. Reports support the fact that chronic pain significantly affects the quality of life of patients and obviously interferes with their participation in daily activities leading to reduced productivity [1, 2].

A variety of medications are generally administered via oral, transdermal and parenteral routes for the treatment of chronic neuropathic pain. Despite the availability of very potent drugs, the treatment of chronic pain remains challenging [3]. It has been found that systemic administration of drugs fail to achieve adequate pain relief in up to ~30% of patients [2]. The poor clinical responses to drugs are attributed mainly to poor bioavailability of drugs to the brain and cerebrospinal fluid (CSF) (due to the physiological barriers protecting the CNS) and short duration of activity of drugs due to rapid clearance of drug from these regions [4–6]. In addition, serious adverse effects restrict the use of analgesics systemically [7].

Ziconotide ( $\omega$ -conotoxin) is a synthetic peptide (25 amino acid sequence) isolated from the magician's cone snail, *Conus magus*. It is used for the treatment of chronic pain and has a novel mechanism of action that involves potent and selective blockade of presynaptic neuronal N-type calcium channels in the spinal cord [8, 9]. Ziconotide when bound to the N-type calcium channel blocks the release of neurotransmitters from the primary afferent nerve terminals to the synaptic cleft [10, 11].

Ziconotide is found to be effective in patients who respond poorly to opioids and other analgesic drugs [12]. However, due to the peptide nature of drug and poor ability to cross the blood-CSF barrier, bioavailability to CSF following systemic administration is poor. Moreover, systemic administration of ziconotide is also known to cause profound side

effects [13]. Therefore, recently this non-opioid drug was approved only for intrathecal infusion by FDA and European Medicines Agency.

Although, intrathecal administration of ziconotide is considered to be an effective route to target the spinal cord region, the procedural and device complications may be a major concern when delivering drugs via intrathecal route. Meningitis is a potential complication caused because of the contamination of the micro infusion device, pump pocket, or catheter tract. Meningitis occurred in 3% (40 cases) and 1% (1 case) of the ziconotide and placebo groups respectively during the ziconotide clinical trial. Of the 41 meningitis cases 38 were because of the use of external infusion systems [14, 15]. Complications like pump failure, program error, and incorrect refill may lead to overdose of ziconotide which might lead to exaggerated pharmacological effects like ataxia, spinal myoclonus or nystagmus. Failure in adopting proper titrations of ziconotide may also lead to cognitive and neuropsychiatric side effects [16].

Neurological deficits can be developed because of the procedure failure and inflammatory mass development catheter tip. Therefore, considering all the complications associated with existing therapy for intrathecal administration of ziconotide there is an urgent need to develop safer methods for delivery of ziconotide to the CSF.

Intranasal delivery is a noninvasive route that offers a direct pathway from nose to CSF via the olfactory apparatus [17–21]. Nose to CSF pathway could deliver drugs directly to the CSF bypassing the blood-CSF barrier. In addition, intranasal delivery would be patient compliant and allows frequent administration. Nasal formulations can be self-administered and do not require physician supervision during administration unlike in case of intrathecal or parenteral formulations [22]. Therefore, in this project the plausibility of delivering ziconotide to the CSF via intranasal route was investigated.

Intranasal administration of drugs to the CNS is limited because of the poor permeability of the mucosal epithelium to drugs. Tight junctions present in the epithelia are known to limit the permeability and bioavailability of the drug molecules. It was reported by Vaka and coworkers that chitosan leads to reversible permeabilization of olfactory mucosa leading to enhanced bioavailability of intranasally administered drugs [23]. The bioavailability of drugs is also limited by the rapid mucociliary clearance which could be resolved by administration of formulations that would increase the drug residence duration on the olfactory mucosa [24]. Therefore, in the current study Kolliphor P 407 (KP 407) was used in formulation of *in situ* gel to minimize the mucociliary clearance of drug

## 2. MATERIALS AND METHODS

### 2.1. Chemicals

Ziconotide acetate, chitosan (MW ~250 kDa, 75–80% deacetylation), Krebs-ringer bicarbonate (KRB) (premixed powder), acetone, hydrochloric acid were procured from Sigma chemicals (St. Louis, MO). Ziconotide acetate was radio labeled by PerkinElmer (Waltham, MA) using  $^{125}\text{I}$ . Sodium perchlorate, methanol, trifluoro acetic acid and

acetonitrile were obtained from Fischer Scientific (Atlanta, GA). Kolliphor P 407 (KP 407) was obtained from BASF (Florham Park, NJ).

## 2.2. Preparation of Olfactory mucosa

*In vitro* permeation studies were carried out across bovine olfactory mucosa (PelFreez Biologicals, Rogers, AR). Freshly excised frozen tissue was obtained from the supplier and used immediately. The tissue was thawed in KRB for a period of 30 min before carrying out the permeation studies.

## 2.3. Analytical Method

**A). Quantification of unlabeled neat ziconotide**—The HPLC system (Waters, 1525) consisting of a Phenomenex C-18 analytical column (4.6 mm × 150 mm, Luna 5.0 μ) and a variable wavelength detector (Waters, 2487). The mobile phase was made up of a gradient elution starting with 100% mobile phase A for 2 minutes, shifting to 70% mobile phase A: 30% mobile phase B over 12 minutes, then to 30% mobile phase A:70% mobile phase B for next 10 min, and to 100% mobile phase A for approximately 6 minutes to re-equilibrate the column. The flow rate was 1.0 mL/min and the column effluent was monitored at 212 nm. Mobile phase A consisted of 98% 25 mmol/L sodium perchlorate, 2% methanol, and 0.05% trifluoroacetic acid; mobile phase B consisted of 49.95% acetonitrile, 49.95% water, and 0.1% trifluoroacetic acid [25]. The limit of detection for neat ziconotide was found to be 10 μg/mL.

**b). Estimation of <sup>125</sup>I ziconotide**—The ziconotide samples labelled with <sup>125</sup>I were injected into a HPLC column using the same conditions mentioned above. The fractions eluted from the column were collected at regular intervals to separate free iodine from the intact <sup>125</sup>I ziconotide. The fractions collected every minute were subjected to radioactivity measurement using a Beckmann Gamma counter (Pasadena, CA) [26].

## 2.4. Stability of ziconotide

**A). Stability of ziconotide in olfactory tissue homogenate**—Bovine olfactory mucosa was homogenized in a glass vial using a high shear tissue homogenizer (Fisher Tissuemizer®) in KRB. To this, a known concentration of ziconotide was incorporated and incubated at 34°C for 6 h. Samples were collected at regular intervals, filtered and analyzed using HPLC.

**b). Stability of ziconotide in CSF**—CSF was collected from the cisternae magna of the Sprague-Dawley rats. The CSF was incorporated with a known concentration of ziconotide. The drug with CSF was incubated at 37°C for a period of 6 h and samples were collected intermittently at regular intervals. The collected samples were analyzed using HPLC.

## 2.5. Preparation of chitosan and <sup>125</sup>I ziconotide solution

Chitosan solution was prepared by dissolving the required quantity of chitosan in 1% glacial acetic acid solution prepared in KRB (pH 5.5). The cold ziconotide and <sup>125</sup>I ziconotide were dissolved separately in KRB. The ziconotide and chitosan solutions were mixed by vortexing. The solutions prepared without incorporation of chitosan served as control.

## 2.6. Experimental setup for *In vitro* permeation

*In vitro* permeation studies were carried out using vertical Franz diffusion apparatus (Logan Instruments, Somerset, NJ). The olfactory mucosa was sandwiched between the donor and receiver compartments such that the dorsal side of the tissue is facing the donor compartment and the ventral side facing the receiver compartment. Ag/AgCl electrodes (procured from Alfa Aesar, Ward Hill, MA) in the form of circular ring with a diameter of 0.5 mm were placed 2 mm away from the tissue in both donor and receiver compartments [22]. A load resistor  $R_L$  (100 k $\Omega$ ) was placed in series with olfactory mucosa after filling the donor and receiver compartments with 500  $\mu$ L and 5 mL of KRB respectively. The voltage drop across the whole circuit ( $V_o$ ) and across the mucosa ( $V_E$ ) were measured using a waveform generator and multimeter (Agilent Technologies, Santa Clara, CA) [27]. The resistance in k $\Omega$  cm<sup>2</sup> was measured by applying a small voltage of about 100 mv at 10 Hz [23, 28].

$$R_E = \frac{V_E R_L}{V_o - V_E}$$

Where  $R_E$  is the olfactory mucosa resistance and  $R_L$  is the load resistor in k $\Omega$ .

## 2.7. Effect of chitosan on permeation of ziconotide, *in vitro*

The effect of different concentrations of chitosan (0.1%, 0.25% and 0.5% w/v) on the permeation of ziconotide was investigated. Drug-chitosan solution was prepared as mentioned in section 2.5. Drug solution with chitosan (100  $\mu$ L) was placed in the donor compartment and 5 ml KRB was filled in the receiver compartment. Control set of experiments were run without incorporation of chitosan in donor solution. The receiver compartment buffer was sampled at different time points, injected in to the HPLC to separate the free Iodine and the fractions of <sup>125</sup>I-ziconotide were quantitated using a Beckmann gamma counter.

## 2.8. Preparation of *in situ* gel forming formulations of ziconotide

*In situ* gels were prepared by cold process. KRB was cooled to 4°C, appropriate amount of KP 407 was added and stirred overnight in the refrigerator. Drug-chitosan solution was prepared by dissolving the required quantity of chitosan in 1% glacial acetic acid solution in polymeric solution followed by addition of unlabeled neat ziconotide and <sup>125</sup>I ziconotide. In case of control, unlabeled neat ziconotide and <sup>125</sup>I ziconotide were dissolved in KP 407 solution without chitosan.

## 2.9. Rheological characterization of *in-situ* gel forming formulations

Rheological studies were carried out using a HR2 rheometer (TA Instruments, New Castle, DE) equipped with 25 mm parallel plate fixture. Temperature was controlled by using a Peltier stage connected to the rheometer. The effect of chitosan addition on gel strength, gelation temperature, and complex viscosity was determined using oscillatory shear

rheology. Samples were equilibrated in the rheometer for 5 minutes before starting any experiments. Each sample was examined in triplicates.

**a) Gelation temperature**—Oscillatory rheology was used to determine the gelation temperature ( $T_{\text{sol/gel}}$ ) of the *in situ* forming gels. The samples were subjected to a constant strain (0.5%) and frequency (0.5 Hz). The sample was heated from 10 °C to 37 °C at a rate of 3 °C and the elastic modulus ( $G'$ ) as a function of temperature was measured. The phase transition from liquid to gel ( $T_{\text{sol/gel}}$ ) was obtained by determining the intersection point of  $G'$  and  $G''$ .

**b) Complex viscosity**—The complex viscosity as a function of temperature (10 °C to 37 °C) during the *in situ* gelation process was determined at a constant frequency of 4.75 rad/sec.

**c) Gel strength**—Elastic modulus ( $G'$ ) is an indirect measure of the gel strength. The elastic modulus ( $G'$ ) recorded during *in situ* gelation process, as described above, was considered.

#### 2.10. *In vitro* release of ziconotide from the gels

*In vitro* release studies were carried out using vertical Franz diffusion cells (Logan Instruments Ltd, NJ). A spectra pore membrane with a cut of molecular weight 5kD, was used to carry out the release studies. The donor and receiver compartments were filled with 0.1 ml of the drug polymeric solution and 5 mL of KRB respectively. The temperature of the chamber was regulated at  $34 \pm 1^\circ\text{C}$  by water circulation. The receiver compartment buffer was sampled at different time points and the amount of ziconotide was measured by HPLC and gamma counter.

#### 2.11. *In vitro* permeation of ziconotide from gels

*In vitro* permeation studies were carried out as discussed in section 2.6. The donor compartments were replaced with 0.1 mL of polymeric solutions, allowed to equilibrate at  $34 \pm 1^\circ\text{C}$ . Samples were collected from the receiver compartment at regular intervals of time and were estimated for ziconotide using HPLC and Gamma counter.

#### 2.12. Extraction of ziconotide from plasma

Blank plasma obtained from untreated rats was spiked with known concentrations of ziconotide and the calibration curve was plotted using the analytical method as mentioned in section 2.3. The rat plasma samples containing ziconotide were extracted initially by vortexing 100  $\mu\text{L}$  of plasma with 0.9 ml of acid acetone (Acetone:Water:HCl; 40:6:1). Proteins were then removed by centrifugation at  $4^\circ\text{C}$  for 15 min. The supernatant was injected into the HPLC for separation of free label  $^{125}\text{I}$  and intact  $^{125}\text{I}$ -ziconotide. Gamma counter was used to estimate the intact  $^{125}\text{I}$ -ziconotide [29].

#### 2.13. Pharmacokinetic studies

Pharmacokinetic studies were performed in Sprague-Dawley rats (male, 250–300 g; Harlan Company, Indianapolis, IN, USA). The rats were categorized into six main groups ( $n=6$ ).

Each main group (I–III) was subdivided into seven sub-groups for seven time points with each sub-group having 6 rats. Whereas, the groups IV–VI, were subdivided into nine sub-groups corresponding to nine time points. The rats were anesthetized using ketamine (80 mg/kg) and xylazine (10 mg/kg) (i.p injection).

Ziconotide (100 µg/kg) solution without chitosan, with chitosan, Ziconotide KP 407 without chitosan (ZKP) and Ziconotide KP 407 with Chitosan (ZKP-0.25 Ch) was administered via intranasal route (i.e. administration of drug directly into the posterior segment of the nose using a microsyringe connected with a soft polymer capillary) to rats in groups III–VI respectively. Whereas, rats were administered with ziconotide (100 µg/kg) by i.v route in group II.

Intrathetically catheterized rats (Group I) were obtained from Charles River (Charles River, Wilmington MA). Ziconotide solution (100 µg/kg) was administered to carry out the pharmacokinetic studies of ziconotide in CSF and plasma following intrathecal administration. In groups I–III the time course of drug in the CSF and plasma was investigated by collecting CSF from cisterna magna and blood from retro orbital sinus at 15, 30, 60, 120, 180, 240 and 360 min. Whereas in groups IV–VI the time course of drug in CSF was obtained at 15, 30, 60, 120, 180, 240, 360, 540 and 720 min. Blood samples in groups IV–VI were collected at time points similar to groups I–III. Plasma was collected from blood samples by centrifuging at 1500 rpm for 15 min and ziconotide was extracted as mentioned in section 2.12. Both blood and CSF collected from each group at respective time points were analyzed using HPLC and gamma counter.

#### 2.14. Aspiration of CSF

The rats were anesthetized using ketamine/xylazine and secured on a stereotaxic frame (Harvard Instruments, Holliston, MA, USA). An incision was made on the skin over the occipital bone and the first layer of the muscle was cut using a scalpel blade. A capillary tube was molded to a pointed edge on one end, whereas the other end with circular tip was connected to a 1 mL syringe with the aid of polyethylene tubing and was used for a cisternal puncture. Following i.v, intranasal or intrathecal administration of ziconotide formulations (25 µg in 100 µL) the CSF samples were collected at specified time intervals as mentioned in section 2.13 and quantified using HPLC and gamma counter.

#### 2.15. Data analysis

The statistical analysis was carried out using GraphPad Prism 5 software. The *t*-test was selected as the test of significance, and  $p < 0.05$  was considered statistically significant.

#### 2.16. Pharmacokinetic data analysis

The area under the CSF concentration-time curve (AUC) from the start of drug administration to the time of the last quantifiable concentration in CSF ( $AUC_{0-t}$ ) was calculated by using the trapezoidal rule. The slope and elimination rate constant ( $K_e$ ) were calculated from the concentration-time curve on a semi logarithmic scale. The pharmacokinetic parameters between the groups were compared by *t*-test at a level of significance of  $p$  value  $< 0.05$ .



### 3. Results and Discussion

#### 3.1. Quantitation of ziconotide

The lowest concentration of neat unlabeled ziconotide that was detected by HPLC with the aid of UV detector at 212 nm was 10 µg/ml at a retention time of ~24.79 min. Ziconotide when radiolabeled with  $^{125}\text{I}$  and quantitated using gamma counter the limit of detection was  $1.24 \times 10^{-3}$  ng/mL. To quantitate the amount of ziconotide, the free  $^{125}\text{I}$  was separated from each biological sample using HPLC. The chromatogram representing the separation of free  $^{125}\text{I}$  and  $^{125}\text{I}$  bound to ziconotide is represented in figure 1. Mass spectral analysis was carried out for the fractions collected from 2–4 min using a Matrix-assisted laser desorption/ionization technique (MALDISYNAPT MS/HDMS). According to the mass spectra, peak of free Iodine was found at 126.87 m/z, which resembles the theoretical m/z value of Iodine value obtained from the Mass Lynx software and confirms the separation of the free Iodine (Figure 2).

#### 3.2. Stability of ziconotide in olfactory tissue homogenates and CSF

A large pool of enzymes is known to be present in the olfactory mucosa. These enzymes act as a protective agents and prevent accidental entry of toxins via nose to brain pathway [30]. The nasal mucosal surface has a variety of proteolytic enzymes which could result in degradation of peptide and protein drugs. Therefore, the stability of ziconotide in the homogenate of the mucosal tissue was investigated for a period of 6 h. Ziconotide was found to be stable for a period of 4 h and there was no significant loss of drug observed.

CSF is a product obtained from the filtrate of plasma and membrane secretions. It is a clear, colorless fluid with a composition of various ions, proteins, enzymes and other substances [31]. However, the ziconotide was found to be stable without any significant degradation in freshly spiked rat CSF which is attributable to unique structure of this peptide.

#### 3.3. Kinetics of ziconotide in CSF - Intrathecal and Intravenous route of administration

Ziconotide was administered into the lumbar region of the spinal cord via intrathecal route. A concentration of  $992.2 \pm 76.43$  ng/ml was attained within a period of 15 min following intrathecal administration. The elimination rate constant of ziconotide in CSF was found to be  $1.01 \pm 0.34$  h<sup>-1</sup>. As mentioned earlier, the drug remained stable without significant degradation for 6 h in freshly aspirated rat CSF at 37°C. CSF is a dynamic system that generally has a high turnover rate. Generally, the clearance of drugs from the CSF is predominantly due to its high turnover rate. Some of the other drugs used in the treatment of pain management were also reported to have a short elimination half-life predominantly due to the rapid turnover of CSF. Vigdis and coworkers [32] reported that sufentanil when administered intrathecally in humans has a short half-life of 0.6 h and a low mean residence time of 0.9 h in CSF. Similarly, morphine was also found to have shorter half-life of ~1.2 h in CSF when administered intrathecally [33, 34]. Therefore, it is reasonable to speculate that the high fluid turnover rate of CSF (125 ml/h in rats) is the likely reason for rapid disappearance of ziconotide as well from the CSF.



When administered via intravenous route, the  $C_{max}$  and  $T_{max}$  of ziconotide in CSF were found to be  $37.78 \pm 6.8$  ng/mL and  $\sim 2$  h respectively. The  $AUC_{0-6}$  of ziconotide in CSF following i.v administration was  $4.98 \pm 0.97$  min. $\mu$ g/mL (100  $\mu$ g/kg dose). The poor bioavailability of drug in CSF following intravenous administration is likely because of the poor ability of ziconotide to cross the blood-CSF barrier which is an interface formed by the epithelial cells of the choroid plexus [35, 36]. From the CSF concentration-time profile data following intravenous administration of drug shown in figure 3, it appears that blood-CSF barrier strongly limits the bioavailability of ziconotide into the CSF.

### 3.4. *In vitro* permeation across the olfactory mucosa

*In vitro* permeation studies were carried out across the olfactory mucosa using Franz diffusion cells. The amount of ziconotide permeated across the olfactory mucosa from neat ziconotide solution (Control) following 4 h was found to be  $0.65 \pm 0.05$   $\mu$ g/cm<sup>2</sup>. Earlier studies have shown that chitosan acts as a permeation enhancer for both small (Cefotaxime) and macromolecules (NGF and BDNF) across the olfactory mucosa. Chitosan used in concentrations of 0.1, 0.25 and 0.5 % w/v enhanced the permeation of ziconotide across bovine olfactory mucosa by  $\sim 5$ , 12 and 14 fold at the end of 4 h respectively, over control (The permeability coefficients for control, 0.1, 0.25 and 0.5 % chitosan were  $1.2 \pm 0.06 \times 10^{-4}$  cm/sec,  $4.6 \pm 0.3 \times 10^{-4}$  cm/sec,  $11.2 \pm 0.5 \times 10^{-4}$  cm/sec,  $11.8 \pm 0.3 \times 10^{-4}$  cm/sec respectively) (Figure 4).

Generally, with increase in concentration of the polymer, the viscosity increases reducing the diffusion coefficient and thus the permeation across mucosal membrane. In this case, it appears that the concentration dependent effect of chitosan on the membrane is predominant over the influence of viscosity. However, 0.25% w/v solution was used *in vivo* to investigate the pharmacokinetics of ziconotide following intranasal administration to minimize the potential discomfort to the animal due to interruption of respiratory pathway at higher polymer concentration.

### 3.5. Pharmacokinetics of ziconotide in CSF following administration of intranasal solution

The  $C_{max}$  of ziconotide in the CSF following intranasal administration of ziconotide with chitosan was 7 fold higher than control (neat ziconotide) ( $33.55 \pm 7.39$  ng/mL) (Figure 5). But the amount of ziconotide in CSF following i.v administration ( $37.78 \pm 6.8$  ng/mL) was comparable to that of intranasal control group.

The difference in  $C_{max}$  attained following intranasal administration of neat ziconotide and ziconotide with chitosan is likely due to the ability of chitosan to open the tight junctions present in the olfactory epithelia. In addition, chitosan also acts as a mucoadhesive agent and would retain the drug for a longer period of time compared to the control.

In case of the group administered by ziconotide with chitosan, the bioavailability ( $AUC_{0-6}$ ) of ziconotide in CSF following intranasal administration was  $\sim 2$  fold greater than the bioavailability ( $AUC_{0-6}$ ) following i.v administration ( $4.98 \pm 0.98$  min. $\mu$ g/mL). Whereas, the bioavailability of neat ziconotide following intranasal administration ( $1.57 \pm 0.44$  min.

$\mu\text{g/mL}$ ) was  $\sim 3$  fold less compared to bioavailability ( $\text{AUC}_{0-6}$ ) following i.v administration (Table 1).

The time required to attain maximum concentration ( $T_{\text{max}}$ ) in CSF was less upon intranasal administration (15 min) compared to i.v administration (120 min). Similar observations have been reported in case of cephalexin. Sakane and co-workers [19] suggested that following intranasal administration of cephalexin, the drug CSF levels were higher in 15 min than at 30 min. It was reported that the rapid absorption of drug was attributed to the rapid distribution of the drug into the cervical lymph node via the perineural spaces of the olfactory neurons [19]. From figure 5 it is evident that the drug delivered via intranasal route in presence of chitosan attained a concentration of  $244.34 \pm 48.6 \text{ ng/mL}$  (92 nM) in 15 min. The objective is to achieve this order of drug levels in the CSF in humans. Although we cannot extrapolate the rodent data to humans as there exists significant anatomical differences in the nose to brain pathway between the human and rats, the pharmacokinetic data in rats serve as proof of concept to escalate to the next step in humans.

The production or turnover rate of CSF is about  $125 \mu\text{L/h}$  in rats [37]. This might contribute to considerable level of elimination of drug from the CSF. The elimination rate of ziconotide in CSF following intranasal and intravenous administration is  $0.6 \pm 0.17 \text{ h}^{-1}$  and  $0.42 \pm 0.10 \text{ h}^{-1}$  respectively. Although, the onset of action via intranasal route is rapid, the elimination of ziconotide from CSF is also considerably high because of its high turnover rate as discussed earlier in section 3.3 [8].

### 3.6. Rheological evaluation of ziconotide gels

The gelation of KP 407 occurs as a result of change in polymer desolvation and subsequent micellization. As the temperature was increased, the negative coefficient of the solubility effects micellization, expels the water from the core of micelles, leads to conformational changes in methyl group (PPO) orientation and results in gel formation [38–41]. Chen et al reported that KP 407 in concentrations that forms gel at physiological temperature, possess higher viscosity than normal nasal fluid, fast phase switching capability and withstand the antidilution effect [24]. Preliminary studies revealed that a minimum of 18% w/w KP 407 is required to obtain a gel at a temperature lower than  $37 \text{ }^\circ\text{C}$ . As the concentration of the KP 407 increases, the gelation temperature decreases. The temperature of the nasal cavity is  $\sim 34 \text{ }^\circ\text{C}$  and a polymeric solution that forms gel close to the nasal temperature would be ideal for controlled release drug delivery applications. Further studies were carried out to determine a poloxamer concentration that gels in the range of  $25 \text{ }^\circ\text{C}$  to  $37 \text{ }^\circ\text{C}$ . The gelation temperatures of 18, 20 and 22% KP 407 were found to be  $\sim 31$ , 28 and  $22 \text{ }^\circ\text{C}$ , respectively (Table 2 and Figure 6).

The complex viscosity of 18, 20 and 22% KP 407 at  $34 \text{ }^\circ\text{C}$  was found to be  $1153 \pm 41$ ,  $1976 \pm 74$  and  $3085 \pm 94 \text{ Pa.s}$ , respectively.

Gel strength is an indirect measurement of storage modulus ( $G'$ ). The gel strength of 18, 20 and 22% KP 407 at an angular frequency of 0.5 Hz and  $34 \text{ }^\circ\text{C}$  were found to be  $5311 \pm 25$ ,  $9366 \pm 61$  and  $14654 \pm 110 \text{ Pa}$  respectively (Figure 6). The gel strength increased with increase in the concentration of KP 407.

KP 407 with 20% w/w was considered for further studies as it has the gelation temperature close to the physiological temperature, considerable gel strength and a viscosity higher than nasal mucosa.

### 3.7 Effect of chitosan on gelation temperature, gel strength and viscosity

The effect of chitosan on gelation temperature, gel strength and viscosity were also examined. Chitosan when incorporated in a concentration of 0.25 % w/v in the KP407 polymeric solution, shifted the gelation temperature of 20% KP 407 from 30 °C to 34 °C. Addition of chitosan hinders gel formation and as a result gelation temperature increases to a higher temperature [42]. A similar change in temperature on addition of chitosan was also reported by Zaki et al [43]. In addition to increase in gelation temperature, the gel strength and the complex viscosity of the gel decreased due to incorporation of chitosan (Table 2). Although the viscosity of the formulation was lower than that observed for formulation without chitosan, it was still higher than the nasal mucosal viscosity which ranges between 0.02–500 Pa.s at a frequency of 0.01–100 1/sec [44, 45].

### 3.8. *In vitro* release and permeation studies of ziconotide gels

The percentage of ziconotide released from ZKP and ZKP-0.25Ch in 6 h was found to be  $85.01 \pm 0.86$  and  $69.22 \pm 1.04$ , respectively.

Chitosan used in concentration of 0.25 % w/v enhanced the permeation of ziconotide across the bovine olfactory mucosa by ~5 fold at the end of 6 h compared to ziconotide alone ( $1.06 \pm 0.09 \mu\text{g}/\text{cm}^2$ ). Therefore, ZKP and ZKP-0.25Ch were used to carry out the *in vivo* studies following intranasal administration.

### 3.9. Intranasal ziconotide gels

The  $C_{\text{max}}$  of ziconotide in the CSF following intranasal administration of ZKP-0.25Ch was ~ 2 fold higher than ZKP ( $24.92 \pm 2.91 \text{ ng/mL}$ ). Whereas, the amount of ziconotide in CSF following intranasal administration of ziconotide solution with chitosan was ~5 fold higher than ZKP-0.25Ch ( $50.21 \pm 3.48 \text{ ng/mL}$ ) (Figure 7). The difference in  $C_{\text{max}}$  attained following administration of solution and gel formulation could be attributed to enhanced viscosity of gel formulation compared to solution.

The elimination rate of ziconotide in CSF following intranasal gel was reduced by ~7.5 fold compared to the elimination rate attained by intranasal solution ( $0.6 \pm 0.17 \text{ h}^{-1}$ ) as observed in sustained release dosage forms (Figure 7). It appears that the mucoadhesive formulation approach using *in situ* gel forming polymer is an excellent approach to retain drug in CSF longer by replenishing the CSF.

### 3.10 Pharmacokinetics of ziconotide in plasma

The  $\text{AUC}_{0-6}$  levels in the plasma following intrathecal administration of ziconotide solution, intranasal administration of ziconotide solution with and without chitosan were  $53.15 \pm 8.25 \mu\text{g}\cdot\text{min}/\text{mL}$ ,  $54.84 \pm 5.11 \mu\text{g}\cdot\text{min}/\text{mL}$  and  $27.88 \pm 9.18 \mu\text{g}\cdot\text{min}/\text{mL}$ , respectively. These levels were almost ~5–10 fold less compared to that of i.v administration ( $275.07 \pm 23.44 \mu\text{g}\cdot\text{min}/\text{mL}$ ) (Figure 8). This data suggests that low systemic bioavailability following

intranasal administration is likely to result in less systemic side effects compared to parenteral administration.

The AUC<sub>0-6</sub> levels of ziconotide in plasma following intranasal administration of ZKP-0.25C and ZKP was found to be  $37.89 \pm 4.86 \mu\text{g}\cdot\text{min}/\text{mL}$  and  $31.77 \pm 3.22 \mu\text{g}\cdot\text{min}/\text{mL}$  (Figure 8). These levels are almost 7 and 8 fold less compared to that of i.v administration ( $275.07 \pm 23.44 \mu\text{g}\cdot\text{min}/\text{mL}$ ).

This data suggests that low systemic bioavailability following intranasal administration is likely to result in less systemic side effects compared to parenteral administration.

#### 4. Conclusion

These results indicate that intranasal route could be utilized as one of the potential routes for delivery of ziconotide to the CSF compared to intrathecal and intravenous route. Ziconotide solution when administered via intranasal route attained a faster T<sub>max</sub> compared to intravenous administration. Co-administration of ziconotide solution with chitosan via intranasal route enhanced the C<sub>max</sub> in CSF by ~5 folds when compared to its control, but had a faster elimination rate from CSF irrespective of the formulation.

Ziconotide when delivered using poloxamer as a vehicle not only attains therapeutic concentration rapidly but also prolonged the release of ziconotide and maintained therapeutic levels in CSF till 9 h compared to solution. Prolonged maintenance of ziconotide would potentially reduce the frequency of administration. The elimination rate of ziconotide from CSF was reduced by ~7.5 fold using gel formulations compared to intranasal solutions. Despite reducing the rate of elimination, intranasal route is also preferred over i.v and intrathecal route because of its noninvasiveness, opportunity for frequent administration, and potentially minimal side effects. The intranasal administration of ziconotide was also found to result in relatively low systemic bioavailability. These results suggest that intranasal delivery of ziconotide could be developed as a potential treatment approach for chronic pain.

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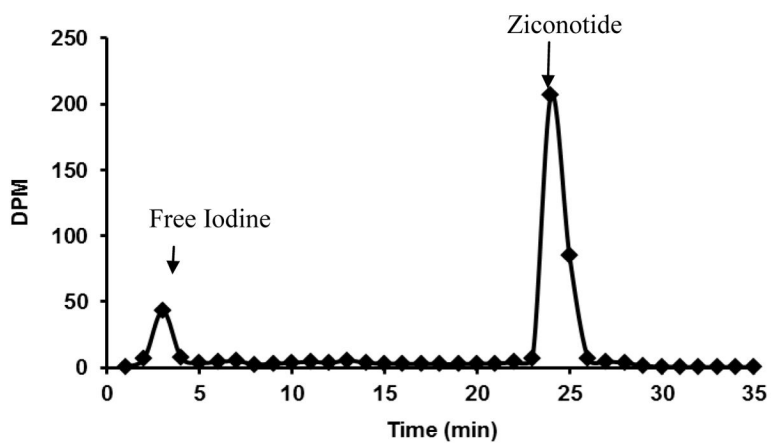
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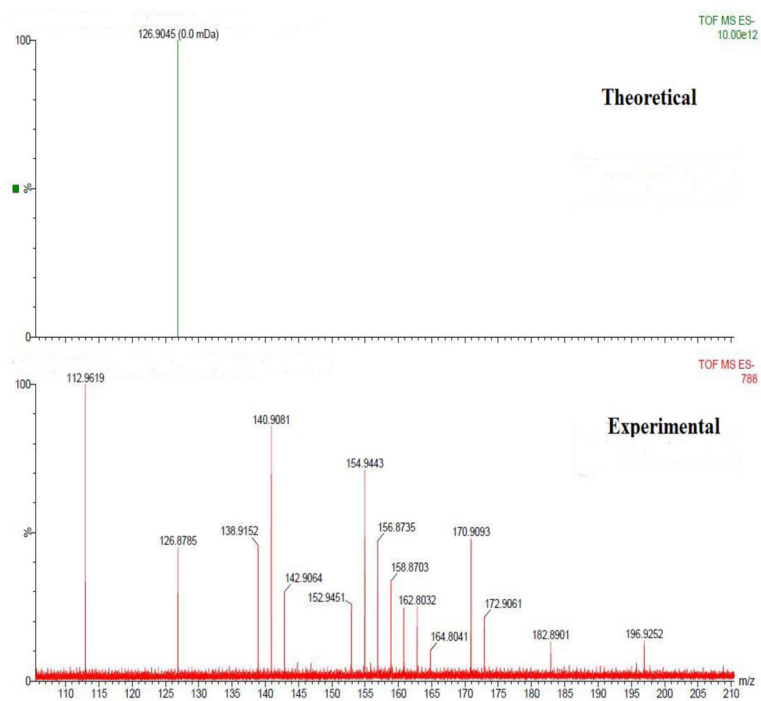
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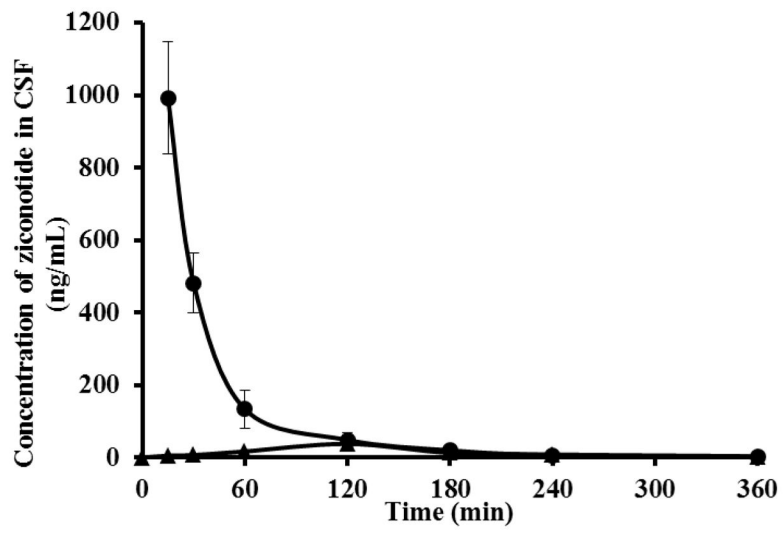




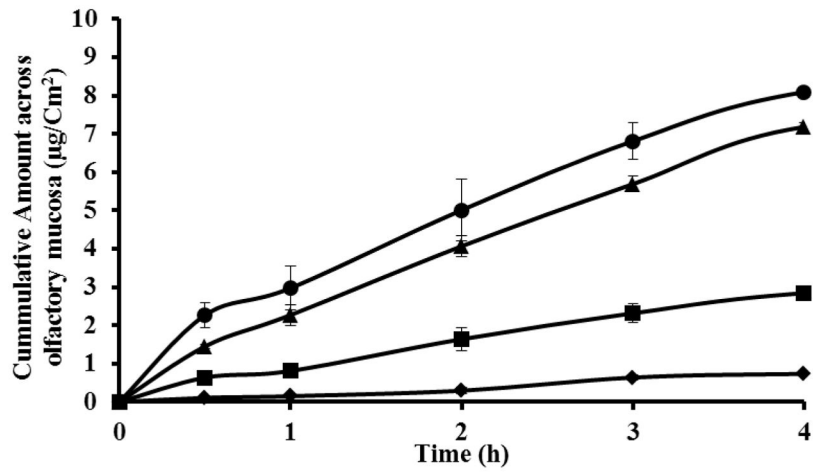
**Figure 1.** Chromatogram representing  $^{125}\text{I}$  and  $^{125}\text{I}$  ziconotide.



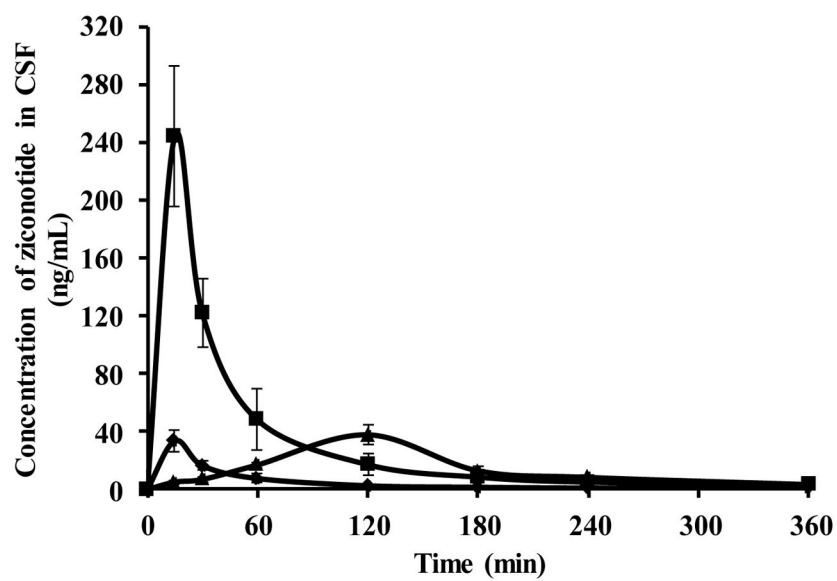
**Figure 2.** Negative mode ESI-MS spectra of free Iodine acquired on a Waters synapt HDMS.



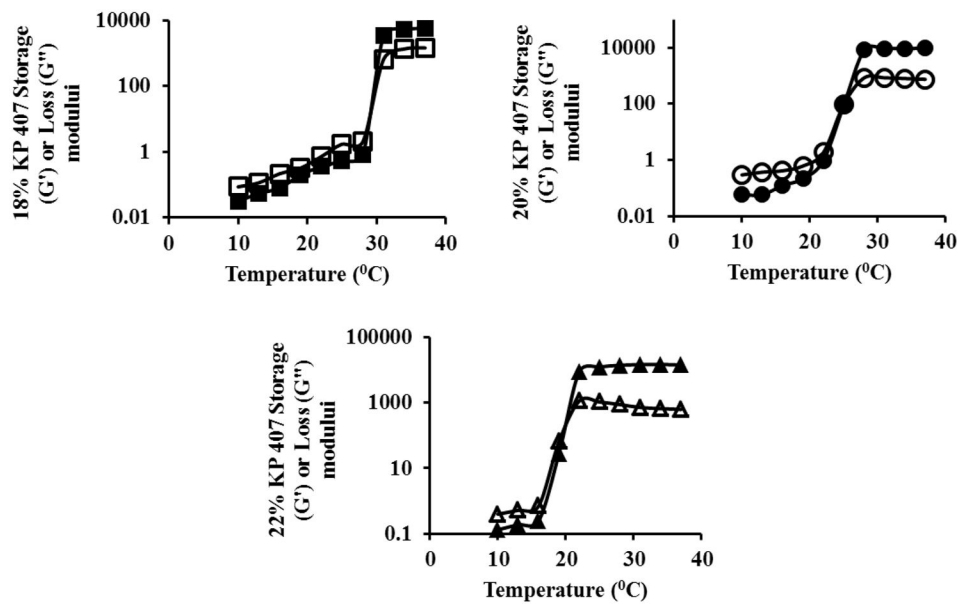
**Figure 3.** Pharmacokinetics of ziconotide in the CSF following intrathecal (●) and intravenous (▲) administration.



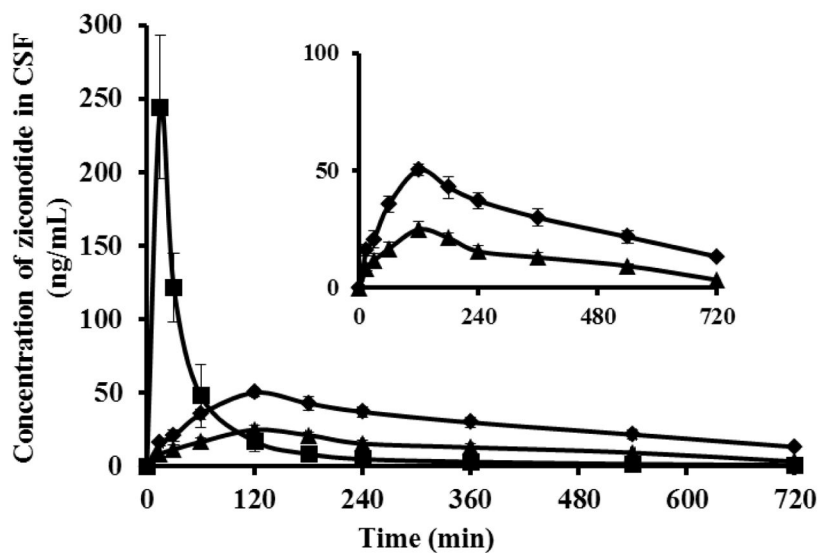
**Figure 4.** In vitro permeation of ziconotide across the bovine olfactory mucosa in presence of control (◆), 0.1% chitosan (■), 0.25% chitosan (▲) and 0.5% chitosan (●).



**Figure 5.** Pharmacokinetic profile of ziconotide in CSF following intranasal administration of ziconotide with chitosan (■), without chitosan (◆) and intravenous (▲) administration of ziconotide.

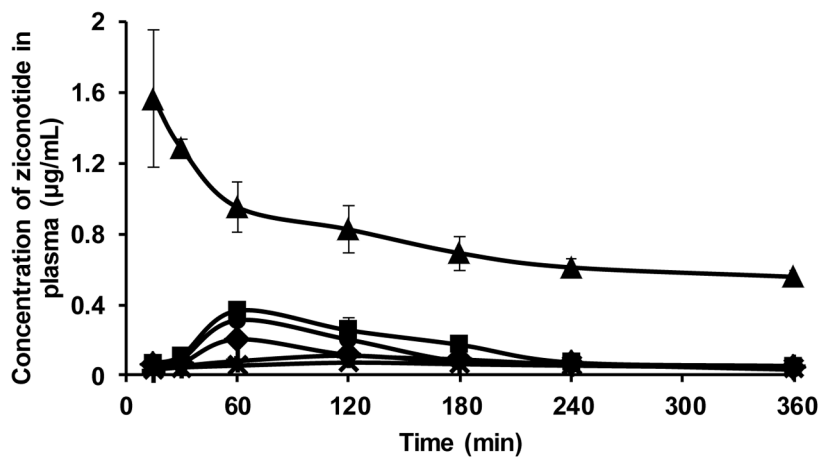


**Figure 6.** Determination of gelation temperature, storage and loss modulus of 18% (■,□), 20% (●,○) and 22% (▲,△) KP 407. Filled symbols represent storage modulus and empty symbols represent loss modulus.



**Figure 7.** Pharmacokinetic profile of ziconotide in CSF following intranasal administration of ziconotide KP 407 without chitosan (▲), with chitosan (◆) and ziconotide solution with chitosan (■).





**Figure 8.** Pharmacokinetic profile of ziconotide in plasma following intranasal administration of ziconotide KP 407 without chitosan (X), with chitosan (+), ziconotide solution with chitosan (■) without chitosan (◆), i.v (▲) and intrathecal (●) administration of ziconotide solution.

**Table 1**

PK parameters of ziconotide in CSF following intranasal and intravenous administration.

CSF	Intranasal ziconotide	Intranasal ziconotide with Chitosan	Intravenous ziconotide
AUC <sub>0-6</sub> (min.µg/mL)	1.57 ± 0.44	10.73 ± 2.92	4.98 ± 0.98
C <sub>max</sub> (ng/mL)	33.55 ± 7.39	244.34 ± 48.6	37.78 ± 6.8
T <sub>max</sub> (min)	15	15	120
K <sub>E</sub> (h <sup>-1</sup> )	0.54 ± 0.08	0.69 ± 0.17	0.43 ± 0.10

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**Table 2**

Gelation temperature, storage modulus and complex viscosity of 18, 20 and 22% KP 407.

<b>Poloxamer Concentration (%w/w)</b>	<b>Gelation Temperature (°C)</b>	<b>Storage Modulus (Pa) @ 34 °C</b>	<b>Complex Viscosity (Pa.s) @ 34 °C</b>
<b>18% KP 407</b>	31	5311 ± 25	1153 ± 41
<b>20% KP 407</b>	28	9366 ± 61	1976 ± 74
<b>22% KP 407</b>	22	14654 ± 110	3085 ± 94
<b>20% KP 407-0.25% Chitosan</b>	34	7578 ± 94	1605.46 ± 38

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