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Prompt repair of hydrogen peroxide-induced DNA lesions prevents catastrophic chromosomal fragmentation

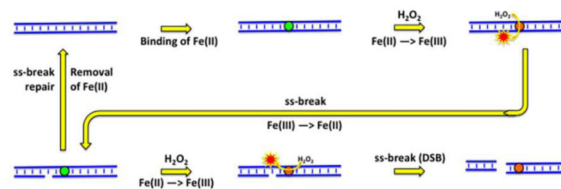
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Abstract

Iron-dependent oxidative DNA damage in vivo by hydrogen peroxide (H_2O_2 , HP) induces copious single-strand(ss)-breaks and base modifications. HP also causes infrequent double-strand DNA breaks, whose relationship to the cell killing is unclear. Since hydrogen peroxide only fragments chromosomes in growing cells, these double-strand breaks were thought to represent replication forks collapsed at direct or excision ss-breaks and to be fully repairable. We have recently reported that hydrogen peroxide kills *Escherichia coli* by inducing catastrophic chromosome fragmentation, while cyanide (CN) potentiates both the killing and fragmentation. Remarkably, the extreme density of CN+HP-induced chromosomal double-strand breaks makes involvement of replication forks unlikely. Here we show that this massive fragmentation is further amplified by inactivation of ss-break repair or base-excision repair, suggesting that unrepaired primary DNA lesions are directly converted into double-strand breaks. Indeed, blocking DNA replication lowers CN+HP-induced fragmentation only ~2-fold, without affecting the survival. Once cyanide is removed, recombinational repair in *E. coli* can mend several double-strand breaks, but cannot mend ~100 breaks spread over the entire chromosome. Therefore, double-strand breaks induced by oxidative damage happen at the sites of unrepaired primary one-strand DNA lesions, are independent of replication and are highly lethal, supporting the model of clustered ss-breaks at the sites of stable DNA-iron complexes.

Graphical Abstract



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Keywords

cyanide; hydrogen peroxide; catastrophic chromosome fragmentation; base excision repair; ss-break repair; recombinational repair

1. Introduction

Hydrogen peroxide (H_2O_2 , sometimes abbreviated as HP) is a potent life-specific oxidation agent, used by our immune cells to kill invading microbes [1, 2]. The general lack of reactivity of H_2O_2 outside the cell at physiological pH and temperatures [3-5] contrasts with its potent toxicity inside the cell, explained by the distribution of soluble ferrous iron (Fe(II)) between “life” and the surrounding “non-life”. Indeed, in oxic conditions, Fe(II) is generally absent outside the cell [6, 7], while plentiful (~1 mM total iron) inside the cell [8, 9]. According to the Fenton's reaction [10], ferrous iron donates electron to hydrogen peroxide, splitting the molecule in half and thus generating hydroxyl radical, which reacts with organic molecules at diffusion rates [5, 11].

However, because of the powerful cellular defense systems [12], H_2O_2 alone kills fast only at concentrations of 20 mM and higher [13-15], while typical in vivo concentrations of it even in lysosomes are orders of magnitude lower [16-18]. To make the physiological concentrations kill, cells of our immune system use various potentiator molecules that are only bacteriostatic by themselves, yet make low concentrations of H_2O_2 bactericidal. Perhaps the most widely recognized of such potentiators is nitric oxide [19-21], but several other potentiators are known, including aminoacids histidine and cysteine [22-24], and cyanide (CN) [15, 25]. In spite of years of research, the mechanisms behind potentiation of H_2O_2 toxicity remain elusive. The initial objective of our project was to test the idea that CN potentiates H_2O_2 toxicity by inactivating DNA repair mechanisms that mend various H_2O_2 -induced DNA lesions. Cyanide could do it, for example, by extracting metal cofactors from certain enzymes or by blocking ATP production and thus stalling repair pathways requiring substantial energy consumption.

Hydrogen peroxide (via formation of hydroxyl radicals) induces a wide variety of DNA lesions. H_2O_2 treatment causes direct one-strand DNA breaks (ss-breaks), repaired in *E. coli* by DNA pol I plus DNA ligase (ss-break repair), a variety of modified sugars, as well as modified DNA bases, like hypoxanthine, 5,6-dihydrothymine, fapy-G and 8-oxo-G, that are removed by DNA glycosylases, with the resulting abasic sites incised by abasic site endonucleases (ABS-endo) and subsequent repair completed by the same DNA pol I and DNA ligase (base-excision repair) [26-28]. This diverse chemistry of DNA damage notwithstanding, at the end H_2O_2 either breaks individual DNA strands directly, or the modified sugars or bases are repaired via strand incision intermediates, both leading to accumulation of ss-breaks during cell treatment with H_2O_2 [29]. With all this H_2O_2 -induced ss-breaks, it is not surprising that some of them represent double-strand DNA breaks [23, 29, 30], the chromosome lesions of the highest killing potential [31-33]. In fact, there are proposals that the killing lesions after hydrogen peroxide treatment are these infrequent double-strand breaks [15, 23, 31, 34], rather than the copious oxidative one-strand DNA

damage. The acute sensitivity to hydrogen peroxide of mutants in double-strand break repair [13, 14, 35] supports this suspicion.

The traditional explanation for these infrequent double-strand breaks would be stochastic coincidence of two direct/repair ss-breaks in the opposite DNA strands, according to the scenario called “clustered excision” [36-38] (Fig. 1A). However, aggravation of the repairable H₂O₂-induced DNA damage to the irreparable status is only observed in growing cells [15, 39], suggesting involvement of DNA replication or segregation in formation of double-strand DNA breaks. There are several models of replication-dependent chromosomal fragmentation [40, 41], but only one of them, the replication fork collapse model, features preexisting DNA ss-breaks. According to the replication fork collapse scenario [42-44], replication fork runs into a ss-break in template DNA and comes apart, generating a one-ended double-strand break and the full-length molecule (a hybrid between the template and one of the replicated daughter arms) (Fig. 1A). Finally, segregation in bacteria is suspected to break duplex DNA at unrepaired ss-breaks, in the double-strand break-behind the replication fork scenario [45] (Fig. 1A). The growth requirement for H₂O₂-induced double-strand breaks strongly favors either replication- or segregation-dependent scenarios for the chromosome fragmentation in H₂O₂-treated cells.

While investigating mechanisms of cyanide-potentiated H₂O₂ toxicity in *E. coli*, we have recently reported a novel phenomenon, that we have called “catastrophic chromosome fragmentation” [15]. We have found that chromosomes in the treated cells were literally pulverized, leaving no chance of survival, but in wild type (WT) *E. coli* fragmentation only affected growing cells [15]. Then we also found that in the *dps* mutants, H₂O₂-induced fragmentation is still observed in stationary cultures [15]. This surprising result meant that replication forks *per se* are not required for H₂O₂-induced chromosome fragmentation and opened a possibility that even in growing cells, H₂O₂-induced double-strand breaks have a non-replicative nature. Another aspect of H₂O₂-induced chromosomal fragmentation that begged further investigation was its massive nature (thus, the term “catastrophic chromosome fragmentation”) [15], which was intuitively inconsistent with the replication- or segregation-dependence of the breakage. To explore mechanisms behind this catastrophic chromosomal fragmentation, and their potentiation by cyanide, we sought answers to three major questions: 1) what is the role of various DNA repair mechanisms in preventing or mending this fragmentation? 2) what is the effect of cyanide on the fragmentation or its repair? 3) are double-strand breaks dependent on replication or protein synthesis?

2. Materials and Methods

2.1. Strains and Plasmids

Escherichia coli strains used are listed in Table 1 and are all K-12 BW25113 derivatives [46], except for *dnaA46*, *dnaC2* and *dut recBC*(Ts), which are in the AB1157 background. Alleles were moved between strains by P1 transduction [47]. Unless indicated otherwise, the mutants were all deletions from the Keio collection [46], purchased from the *E. coli* Genetic Stock Center, and were verified by PCR or by their characteristic UV-sensitivities. For double mutant construction, the resident kanamycin-resistance cassette was first removed by transforming the strain with pCP20 plasmid [48].

2.2. Reagents

Hydrogen peroxide was purchased from Sigma; potassium cyanide (KCN) was from Mallinckrodt.

2.3. Growth Conditions and Viability Assay

To quantify survival kinetics, fresh overnight cultures were diluted 500-fold into LB medium (10 g of tryptone, 5 g of yeast extract, 5 g of NaCl, 250 μ L of 4 M NaOH per liter [47]) and were shaken at 37°C for about two and a half hours until they reached $OD_{600} \sim 0.3$. At this point, the cultures were made 3 mM for CN and/or 2 mM for H₂O₂ (or the indicated treatment) and the shaking at 37°C was continued. In order to measure survival/revival in cells treated with CN + H₂O₂, the cells were spun down, resuspended in fresh LB and shaken at 37°C for various amount of time. Viability of cultures was measured at the indicated time points by spotting 10 μ L of serial dilutions in 1% NaCl on LB plates (LB medium supplemented with 15 g of agar per liter). The plates were incubated overnight at 28°C, the next morning colonies in each spot were counted under the stereomicroscope. All titers have been normalized to the titer at time 0 (just before the treatment).

2.4. Measuring Chromosomal Fragmentation via Pulsed-Field Gel Electrophoresis

This was done exactly as before [15] and follows our general protocol [49, 50].

3. Results

3.1. Conditions with reduced chromosome fragmentation for sensitive mutants

Concentrations of KCN up to 300 mM and H₂O₂ up to 10 mM are bacteriostatic for WT *E. coli* grown in a rich medium, while their 3 mM KCN + 2 mM H₂O₂ combination is strongly bactericidal [15, 25]. To gain insights into both the primary DNA lesions and the ultimate chromosomal consequences after KCN + H₂O₂ treatment, we used the survival and chromosomal fragmentation as the two readouts with select mutants in DNA repair. However, some of these mutants proved to be so sensitive to our standard 3 mM KCN + 2 mM H₂O₂ treatment, that we had to develop milder treatment regimens to observe any survivors at the earliest time points. We found that reducing CN concentration 10 times does not affect the early rate of killing of WT cells, while reducing H₂O₂ concentration 10 times decreases the early rate of killing somewhat (Fig. 1B and Fig. S1AB). At the same time, if the concentration of both CN and H₂O₂ is reduced 10-fold, the treatment becomes bacteriostatic for WT cells (Fig. 1B), as we have reported before [15]. The rate of early (5 minutes) chromosome fragmentation during CN + H₂O₂ treatment is reduced ~2-fold when 10 times lower CN concentration is used, whereas it is reduced ~8-fold when both CN and H₂O₂ concentrations are decreased 10-fold (Fig. 1CD). We used 0.3 mM CN + 2 mM H₂O₂ conditions for the sensitive mutants, while reserving the 0.3 mM CN + 0.2 mM H₂O₂ conditions (that do not kill WT cells) for the hyper-sensitive mutants.

3.2. ABS-endonucleases prevent double-strand breaks

The two major pathways for repair of one-strand DNA lesions in *E. coli* are nucleotide excision repair (NER) and base excision repair (BER) [51]. In addition, there is also the

most basic pathway to close all kinds of single-strand interruptions, catalyzed by DNA pol I and DNA ligase that serves both the excision repair pathways, as well as DNA replication [52]. We found that the NER-deficient *uvrA* and *uvrB* mutants have the wild type sensitivity to CN + H₂O₂ (Fig. 2A), indicating that NER has no role in mending the CN + H₂O₂ induced DNA lesions and suggesting that oxidative DNA lesions are unlike the bulky and/or DNA helix-distorting lesions, or interstrand crosslinks, all repaired by NER [28].

We blocked base-excision repair at the critical stage of abasic site nicking, inactivating both the major abasic site endonuclease (XthA) and the minor one (Nfo) [51]. Both *xthA* single and *xthA nfo* double mutants are hypersensitive to the CN + H₂O₂ treatment (Fig. 2B and Fig. S2), indicating that this treatment causes massive base damage. Since any mechanism of base-damage-induced chromosome fragmentation assumes conversion of abasic sites into ss-breaks, and the H₂O₂-induced double-strand breaks were supposed to be due to ss-break clustering (Fig. 1A), we expected much reduced chromosome fragmentation in the *xthA nfo* double mutant. To our surprise, we found that the ABS-endo mutants exhibit massive chromosomal fragmentation far greater than the WT cells under similar conditions (see Fig. 2CD for *xthA nfo* and Fig. 5BC for *xthA* alone), suggesting conversion of unrepaired abasic sites into double-strand breaks. At the same time, the ABS-endo mutants are neither sensitive to the corresponding H₂O₂-alone treatment (Fig. 2B and S2), nor show any chromosomal fragmentation after it (Fig. 2CD). Therefore, abasic sites are *not* produced by H₂O₂ acting alone, yet they are massively induced when the same H₂O₂ treatment is potentiated by cyanide.

3.3. The two types of CN + H₂O₂-induced one-strand DNA lesions and their repair

The mutants most sensitive to CN + H₂O₂ killing turned out to be the ss-break repair mutants, *polA* and *ligA* (Fig. 3AB) (the difference in the treatment doses is because the *polA12*(Ts) mutant we use has the mildest DNA pol I defect of all *polA* mutants [53], in contrast to our *ligA251*(Ts) mutant, which is a complete inactivation at the non-permissive temperature [45, 54, 55]). Remarkably, unlike the abasic site endonuclease mutants, both ss-break repair mutants show a distinct pattern of similar sensitivities to both CN + H₂O₂ and H₂O₂ alone treatments (Fig. 3AB) and exhibit similarly enhanced catastrophic chromosomal fragmentation in response to both treatments (Fig. 3CDE). (The level of fragmentation goes down at later times due to over-fragmentation and short linear DNA migrating out of the gel [15].) This demonstrates that, far from being innocuous, H₂O₂ alone treatment induces copious ss-breaks in DNA (confirming prior observations [29, 35]), but these ss-breaks are promptly repaired by DNA pol I and DNA ligase. Interestingly, CN-potentiation only moderately increases the number of these “direct” ss-breaks.

Taken together, the results from the most sensitive mutants suggest that ss-breaks dominate among H₂O₂-induced DNA lesions, while CN potentiation of H₂O₂ treatment leads to the additional induction of base modifications, removed by base-excision repair via the abasic site intermediate (Fig. 3F). Most surprisingly, without immediate and complete repair, a significant fraction of these one-strand lesions and repair intermediates is converted into double-strand breaks, which we detect as catastrophic, “pulverizing” chromosome fragmentation in these mutants (Figs. 2C and 3C). The mode of the size distribution of the

chromosomal fragments in ligase mutants after CN(0.3) + H₂O₂(0.2) treatment is ~50 kbp, translating into at least 100 double-strand breaks per genome-equivalent.

3.4. Recombinational repair mends H₂O₂-induced, but not CN + H₂O₂-induced double-strand breaks

The major pathway to mend double-strand DNA breaks in *E. coli* is recombinational repair [56-58] (Fig. 4A), and since CN + H₂O₂ treatment induces double-strand breaks, recombinational repair mutants were expected to show extreme sensitivity to CN + H₂O₂ treatment. Because of the known sensitivity of *recA* and *recBC* mutants to H₂O₂-alone treatment [13, 14, 35], we expected recombinational repair to mend double-strand breaks induced by hydrogen peroxide alone and we have confirmed these expectations, both by survival (Fig. 4B) and by physical analysis of chromosomal fragmentation in WT cells versus *rec* mutants (Fig. 4CD). It should be noted that the *recBCD* mutants, deficient in both linear DNA repair and degradation, show the level of total fragmentation, whereas the WT cells show the level of irreparable fragmentation (which is very low, in this case, indicating complete repair).

Remarkably, recombinational repair mutants are much less sensitive to H₂O₂-alone treatment than the most sensitive mutants in ss-break repair, which is reflected in their 2 mM H₂O₂ killing concentration (Fig. 4B), compared to the 10-times lower 0.2 mM killing H₂O₂ concentrations for the *ligA* mutants (Fig. 3B). The apparent reason for this difference is the much lower number of double-strand breaks in recombinational repair mutants after the same H₂O₂ treatment, most likely because of the functional ss-break repair. This is a strong, though indirect, evidence that timely ss-break repair prevents massive chromosomal fragmentation during oxidative damage.

Addition of cyanide to the H₂O₂-alone treatment decreases survival of the WT cells, as well as the *recA*, *recBCD* and the double *recG ruvABC* mutants, two-three orders of magnitude from the corresponding levels after H₂O₂-alone treatment, generally preserving the relationship between the four strains (Fig. 5A). The level of CN + H₂O₂-induced chromosomal fragmentation, even though increased, also appears to be similar in the WT cells and in the *rec* mutants (and generally less than in the BER mutants) (Fig. 5BC). Thus, instead of the expected devastating effect of the CN + H₂O₂ treatment on recombinational repair mutants compared to the WT cells, we observed a comparable deterioration for all of them, independently of their *rec* status (Fig. 5). This means that, although recombinational repair does efficiently mend a limited number of double-strand breaks due to H₂O₂-alone treatment, additional double-strand breaks induced by CN + H₂O₂ treatment are not repaired, at least not during the treatment itself.

3.5. Recombinational repair efficiently mends several breaks per chromosome, but is paralyzed by CN

In fact, this conclusion was expected, as all recombinational repair enzymes hydrolyze ATP to fuel their activities [56], while even 2 mM CN is known to lower ATP production to 5-10% of the WT levels [59, 60], which should inhibit recombinational repair. For example, RecBCD exonuclease is expected to degrade fragmented chromosome during the CN+HP

treatment, but no such chromosomal DNA degradation is observed until cyanide is removed from the medium (T.M. and A.K., unpublished). Because of this dependence of recombinational repair on ATP-hydrolysis, we originally considered recombinational repair a likely target of CN-potential, similar to our earlier suspicion about catalases [15]. However, a much higher sensitivity of recombinational repair mutants to CN + H₂O₂ treatment compared to H₂O₂ -alone treatment (Fig. 5A) argues against this possibility, as mutants inactivating the target of CN potentiation of H₂O₂ toxicity are expected to be equally sensitive to both H₂O₂ -alone and CN + H₂O₂ treatments [15].

To reveal the role of recombinational repair in mending CN + H₂O₂-induced chromosomal fragmentation, we removed the treatment by pelleting cells and resuspending them in fresh medium to allow resumption of ATP-production, and then followed both the culture titer and the level of fragmentation with time. We detected no repair after 45 minute CN + H₂O₂ treatment (not shown), apparently due to the massive chromosomal fragmentation overwhelming the WT double-strand break repair capacity. In contrast, after a short 5 minute treatment followed by removal of CN + H₂O₂, we observed an almost three orders of magnitude recovery in the culture titer (to 30% of the original titer) (Fig. 6A), accompanied by a complete disappearance of chromosomal fragmentation (Fig. 6BC). The median size of sub-chromosomal fragments is ~500 kbp after 5 minute treatment (Fig. 6B), translating into ~10 double-strand breaks per genome-equivalent. There was neither recovery of the culture titer, nor significant repair of chromosomal fragmentation in the *recA* single mutant or in the *ruvABC recG* double mutant (Fig. 6), indicating that both phenomena are due to recombinational repair. There was also no recovery or repair (but evident DNA degradation) in the *xthA nfo* double mutant, in which the density of double-strand breaks after even 5 minute treatment is much higher and is similar to the density of double-strand breaks in WT cells after 45 minute treatment (the median size of sub-chromosomal fragments is ~50 kbp (Fig. 6B), translating into ~100 double-strand breaks per genome-equivalent). We conclude that when ATP-production resumes, recombinational repair is still capable of reassembling fragmented chromosomes if there are 10 or fewer double-strand breaks per genome equivalent, but its capacity is saturated when the density of double-strand breaks is increased 10-fold.

3.6. Blocking replication and protein synthesis does not save cells from CN + H₂O₂ induced double-strand breaks and killing

The extremely high density of double-strand breaks (~100 per genome equivalent) is unique to (CN +) H₂O₂-induced chromosome fragmentation [15] (phleomycin-induced fragmentation may be another example [49]), but the nature of these breaks is perplexing. Indeed, as our genetic analysis suggests, the primary DNA lesions caused by H₂O₂ are one-strand interruptions (ss-breaks), while CN potentiation causes additional base modifications, which eventually translate into more ss-breaks (Fig. 3F). Typically, ss-breaks by themselves cannot fragment DNA; they cause chromosomal fragmentation only during the replication-segregation transition (Fig. 1A), as either replication fork collapse events (Fig. 7A) [56] or segregation fork collapse events [41, 45]. Therefore, the subchromosomal fragments released as a result of fork collapse events in asynchronous cultures have the length distribution from close to zero to the full chromosome size [50, 61], which is *not* what we

observe after 45 minutes of CN + H₂O₂ treatment, that produces chromosomal DNA broken into uniformly short fragments (Fig. 1B) [15].

To test the replication-dependence of CN + H₂O₂-induced fragmentation, we blocked initiation of new replication rounds for two hours, while allowing ongoing rounds to finish, which aligns the chromosomes in the fully-replicated state, devoid of any replication forks. A classic and reliable way to align the chromosomes in bacteria is to block protein synthesis with chloramphenicol (because replication initiation requires new protein synthesis) [62]. The chloramphenicol block does reduce the killing of WT cells slightly (Fig. 7B) and reduces the overall fragmentation about two-fold (Fig. 7CD). However, chloramphenicol pretreatment fails to prevent both fragmentation and cell killing, demonstrating their significant independence of replication or segregation events and even of protein synthesis.

Since blocking protein synthesis would preclude any kind of inducible repair in WT cells (that may negate the positive consequences of replication removal), we thought that DNA repair mutants that are hyper-sensitive to CN + H₂O₂ treatment could be a more sensitive system to detect the effect of replication block. In other words, if chloramphenicol-treated DNA repair mutants were to show the sensitivity of chloramphenicol-treated WT cells, this would mean a huge boost to their CN + H₂O₂ resistance. However, the chloramphenicol-treated DNA repair mutants were still 100-1,000 times more sensitive to CN + H₂O₂ compared to WT cells (Fig. 7E). The levels of chromosome fragmentation in them were also significant (Fig. 7FG), leaving no chances for increased survival. We conclude that blocking replication via protein synthesis inhibition does not save from CN + H₂O₂ poisoning via catastrophic chromosomal fragmentation, indicating that a good half of induced double-strand breaks require no replication forks for their formation.

3.7. Chromosome alignment in the initiation mutants

The most non-invasive way to align the chromosome is to utilize *dnaA*(Ts) and *dnaC*(Ts) mutants, that have heat-sensitive replication initiation and in 2 hours at 42°C have complete chromosomes without replication forks, but with all other cellular processes unaffected [63, 64]. We found that after 2 hours at 42°C the *dnaA*(Ts) and *dnaC*(Ts) mutants are killed with exactly the same kinetics as the corresponding WT strain (Fig. 8A), even though they reach early stationary phase at this point and barely fragment their chromosome (Fig. S3). Since we have already observed a similar effect (loss of viability without chromosome fragmentation) in the stationary cultures of the *dps* mutants before [15], we kept the *dna*(Ts) cultures from saturating by deep dilution before switching to 42°C. We found that CN + H₂O₂ induced significant chromosome fragmentation in non-replicating diluted cultures of *dna*(Ts) mutants at 42°C, even though this level was still ~1.5 times lower than in WT cells (Fig. 8BC) (which matches the chloramphenicol result above (Fig. 7CD)). Overall, we conclude that at least half of the CN + H₂O₂-induced chromosomal fragmentation is independent of DNA replication. Together with the finding that this fragmentation is at least partially irreparable by recombinational repair (Fig. 6A), the replication-independence means that these double-strand breaks form by mechanisms other than replication fork disintegration (Fig. 7A and 8D). For example, they could be direct double-strand breaks,

reparable in the replicated part of the chromosome, but irreparable in the unreplicated part around the terminus (Fig. 8E).

4. Discussion

Catastrophic chromosomal fragmentation is a novel phenomenon that we have discovered in *E. coli* cells killed by hydrogen peroxide (either by high, killing concentrations of H₂O₂-alone or by combined treatment of hydrogen peroxide and cyanide at concentrations of the two chemicals that individually are only bacteriostatic). In our previous work [15], we have shown that cyanide potentiation of oxidative damage by H₂O₂ works at two cellular levels: 1) via iron recruitment from the intracellular depots to fuel Fenton's reaction; 2) by depositing the recruited iron directly onto DNA and thus promoting the so-called DNA self-targeting Fenton's reaction, which, by catalyzing hydroxyl radical formation right on DNA, generates double-strand DNA breaks.

In this work, we have characterized the DNA damage aspects of the CN-potentiated H₂O₂ toxicity by measuring survival and chromosomal fragmentation in various DNA repair mutants. As explained previously [15, 25], finding a mutant equally sensitive to H₂O₂-alone and CN + H₂O₂ treatments could mean that the corresponding repair enzyme is a target of cyanide potentiation. However, the only mutant in DNA repair showing this behavior is deficient in DNA ligase, an enzyme that requires Mg²⁺ as the only metal cofactor and that cannot be, therefore, inactivated by CN via demetallation (cyanide forms tight complexes with transition metals, like iron or copper, but not with Mg²⁺). Interestingly, unlike eukaryotic ATP-dependent ligases, bacterial NAD⁺-dependent ligase [65] is not inactivated by CN via ATP depletion either.

Even though our search for cyanide-inhibition targets among the DNA repair enzymes proved futile, we found evidence consistent with formation of double-strand DNA breaks due to *stable iron-DNA complexes*: 1) the ABS-endo deficiency, instead of suppressing chromosome fragmentation, increases it; 2) the inability to repair ss-breaks, instead of modestly increasing chromosomal fragmentation, literally pulverizes the chromosome; 3) H₂O₂-induced double-strand breaks are partially independent of replication or segregation; 4) even in replicating cells, at least some of these breaks cannot be mended by recombinational repair. Our specific findings include:

— The ss-break repair mends the bulk of both the H₂O₂-alone-induced or CN + H₂O₂-induced primary DNA lesions, suggesting that they are ss-breaks.

— Compared to ss-break repair, base-excision repair plays a lesser role in mending H₂O₂-alone-induced primary DNA damage, but is critical for repair a significant part of the CN + H₂O₂-induced DNA damage, indicating that CN-potentiation redirects this part of oxidative damage to DNA bases.

— CN + H₂O₂-induced double-strand breaks form at ~50% efficiency in non-replicating chromosome and (judging by the unchanged size distribution of resulting chromosomal fragments) appear to be uniformly distributed over the chromosome.

— Recombinational repair efficiently mends a few double-strand breaks generated by H₂O₂-alone treatment.

— Recombinational repair does not function during the CN + H₂O₂ treatment, because cyanide inhibits production of ATP, while all recombinational repair functions depend on ATP hydrolysis. Once the treated cells are transferred to a CN-free medium, recombinational repair can mend ~10 double-strand breaks per genome equivalent. Recombinational repair in *E. coli* cannot mend 100 double-strand breaks per genome equivalent, at least not under our growth conditions.

Below we discuss these specific finding in detail.

4.1. The primary H₂O₂-induced DNA lesions and the possible nature of CN-potentiated base modifications

By itself, hydrogen peroxide has low reactivity with organic matter [3-5]; the DNA damage comes from hydroxyl radicals, that are products of H₂O₂ splitting by electrons derived from the free intracellular Fe(II) atoms [16, 66]. Since hydroxyl radicals interact with various organic substances at diffusion rates [5, 11], the expectation is that all three chemical constituents of DNA: sugars, phosphates and nitrogen bases, will be equally susceptible to hydroxyl radicals coming from cytosol. However, the observed sensitivity of the DNA repair mutants to H₂O₂-alone treatments indicates that the sugar-phosphate backbone is hit preferentially over DNA bases: the ABS-endo-deficient mutant shows no sensitivity to 0.2 mM H₂O₂, while the ligase mutant is extremely sensitive to this treatment. If we assume that hydroxyl radicals are indeed generated in the cytosol around DNA, this difference in sensitivities between the two mutants may reflect the fact that the duplex DNA structure hides the genetic information-carrying bases in the protective double spiral of the sugar-phosphate backbone, which absorbs most of the chemical reactivity coming to DNA [67].

Perhaps more likely, Fe(II) atoms form complexes with DNA phosphates, essentially targeting sugar-phosphate backbone for attacks by nascent hydroxyl radicals. If so, then cyanide potentiation not only dramatically induces production of hydroxyl radicals, but also expands their targets, as both the ligase and DNA pol I mutants, on the one hand, and the ABS-endo-deficient mutants on the other become sensitive to CN + H₂O₂ treatment. The sensitivity of ABS-endo-deficient mutants indicates that cyanide potentiation redirects part of the H₂O₂ damage to DNA bases which, according to the previous results, suggests that Fe(II) atoms directly bind to the DNA bases (in addition to sugar-phosphate backbone). In other words, CN not only releases Fe(II) from the intracellular depots and delivers it to DNA [15], but it also helps depositing Fe(II) onto both sugar-phosphate backbone and DNA bases. Two binding sites of iron on DNA, one at the backbone, while the other at the bases, have been proposed [68-70].

4.2. DNA-iron removal as a result of one-strand repair

One of the most surprising findings of this work is the critical role of base excision repair in *prevention* of double-strand breaks. The general expectation is that, in the absence of ss-break processing by DNA pol I or ligase, ss-breaks will accumulate, leading to a modest increase in double-strand breaks due to occasional coincidence of ss-breaks in opposite

DNA strands. The genome-pulverizing density of double-strand breaks in ligase mutants apparently exceeds this expectation, but in the absence of actual measurements of the density of ss-breaks during these treatments one can only speculate about the mismatch between the two numbers. Others have observed in vitro that the density of double-strand breaks from $H_2O_2 + Fe(II)$ treatment significantly exceeds the one expected from the density of ss-breaks in the same reaction (assuming that double-strand breaks result from coincidence of random ss-breaks) [71].

In the ABS-endo-deficient mutant, much fewer ss-breaks are expected due to the blocked base-excision repair. Yet, chromosomal fragmentation, instead of going down, is extreme in this mutant (Fig. 2C), indicating that completion of this repair somehow prevents subsequent formation of double-strand breaks. We would like to speculate that, since the original cause of the primary one-strand lesions appears to be the DNA-bound Fe(II), this prevention could work by removing the DNA-bound iron (Fig. 9). This removal could be potentially done by either any of the DNA repair enzymes (ABS-endonuclease, DNA pol I, DNA ligase) or, more likely, by the specialized iron depot protein Dps, that could be targeted to the culprit iron by the DNA repair activity in the region. The concept of proteins mopping up transition metal ions to take subsequent oxidative damage on themselves, has experimental support [72].

If not removed from DNA, this iron could continuously cycle between Fe(II) and Fe(III), catalyzing formation of hydroxyl radicals that would eventually break both DNA strands in the same location, inducing a double-strand break (Fig. 9). The idea that oxidatively-induced double-strand DNA breaks are due to stable binding of a catalytic transition metal to DNA has a long history [71, 73-75] and has been formally presented [76].

4.3. The replication-independent HP-induced double-strand breaks

Another unexpected finding, that we had to document carefully, was the partial replication independence of CN + H_2O_2 -induced double-strand breaks. Since oxidative damage is supposed to break (directly or via excision repair) only one DNA strand at a time [29, 35], we expected complete dependence of fragmentation on replication or segregation, like in the cases of UV irradiation or ligase-deficiency [45, 77]. However, aligning the chromosome by blocking new initiations and allowing the existing replication forks to finish did not save the cells from killing and only reduced the observed fragmentation ~2-fold, suggesting that at least half of the double-strand breaks are replication/segregation-independent, in other words — direct. This dovetails with the proposed model of Fenton's reaction-promoted double-strand DNA breakage as a result of stable DNA-iron complex catalyzing formation of multiple hydroxyl radicals at the same DNA location and thus increasing the likelihood of breaks in both DNA strands (Fig. 9). Such double-strand breaks are expected to be completely independent of replication or segregation. Recently, the replication fork-independence of H_2O_2 -induced double-strand breaks (with H2AX foci as a readout) was reported in human cells [78, 79], even triggering a suspicion that H2AX foci do not necessarily identify double-strand breaks [79].

4.4. Recombinational repair of H₂O₂-induced DNA lesions

In contrast to H₂O₂-induced one-strand DNA lesions, whose nature differs depending on whether cyanide is present or absent, there appears no indication of heterogeneity among double-strand breaks induced by H₂O₂-alone or by CN + H₂O₂. Neither there is any difference in how they are recombinationally repaired, as long as their density appears similar. We did not quantify it systematically, but it is clear that *E. coli* is still in the position to reassemble one functional chromosome after experiencing ~10 double-strand breaks per genome equivalent, even though the survival is only 30%. At the other extreme, when the density of double-strand breaks reaches ~100 per genome equivalent, no repair or survival becomes possible. These numbers are most likely similar to other bacterial and eukaryotic cells alike [33, 80], as the massive experience in radiation sterilization testifies [81, 82]. The only exception is *Deinococcus radiodurans* and its relatives, capable of assembling chromosome with no loss of viability after 100 double-strand breaks per genome equivalent [83-85]. One reason for such a high resistance of *Deinococcus* to double-strand breaks is the presence of at least four genome-equivalents in resting cells, putting the minimal number of any chromosomal part per *Deinococcus* cells at four [86]. In contrast, the stationary *E. coli* cells, like most other bacteria, have a single chromosome, and even in the rapidly-growing *E. coli* cells, the copy number of chromosomal segments around the terminus is close to one, making any double-strand break there an irreparable lesion (Fig. 8E).

5. Conclusion

In conclusion, our results clearly support the idea that oxidative damage by hydrogen peroxide kills by inducing double-strand breaks in the chromosomal DNA. In addition, our investigation into the nature of DNA damage induced by hydrogen peroxide alone or by cyanide-potentiated hydrogen peroxide highlighted the important differences between the two in the position of the primary DNA damage and revealed the critical role of timely excision repair in prevention of the subsequent double-strand breaks. In combination with the unexpected lack of the critical role of DNA replication in formation of these double-strand breaks, this indirectly but strongly suggests that a significant fraction of lethal oxidative DNA lesions comes from stable DNA-iron complexes, as has been suggested before [71, 73-76]. In the presence of hydrogen peroxide, such stable complexes eventually “burn through” DNA, breaking both strands in the same location. In the future it would be important to detect such complexes inside the cell, as well as their promotion by cyanide and removal after excision repair. It would be also important to test whether H₂O₂ induces the same catastrophic chromosome fragmentation in *Deinococcus* and how efficiently the pulverized genome will be repaired in that unusual bacterium.

Supplementary Material

Refer to Web version on PubMed Central for supplementary material.

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Abbreviations

CN	cyanide
HP	hydrogen peroxide
ss-break	single-strand DNA break

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Highlights

- Blocking BER of oxidative DNA damage pulverizes the chromosome in CN+HP-treated cells
- Thus, timely repair of one-strand lesions prevents the bulk of double-strand breaks
- Recombination repairs HP-induced double-strand breaks, but is poisoned by CN
- Blocking DNA replication halves CN+HP-induced fragmentation, does not affect survival
- We propose that HP-induced double-strand breaks happen at stable DNA-iron complexes

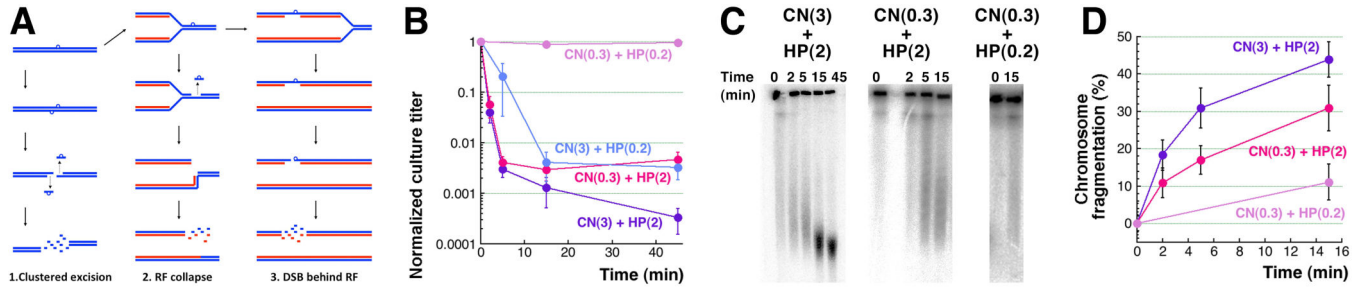


Fig. 1. Three levels of CN + H₂O₂-induced chromosomal fragmentation and their consequences for survival

A. The three models of ss-break-mediated double-strand DNA breaks: 1) the clustered excision model of direct breaks; 2) the replication fork collapse model of replication-dependent breaks; 3) the “DSB-behind the fork” model of segregation-dependent breaks. Parental DNA strands are in blue, while the newly-synthesized DNA strands are in red.

B. Kinetics of death of WT cells treated with varying concentrations of either CN (3 or 0.3 mM) or H₂O₂ (2 or 0.2 mM). Here and in the rest of the paper, all values are means of 3 or more independent measurements ± SEM.

C. A representative pulsed-field gel demonstrating chromosomal fragmentation induced in WT cells treated with 3 mM CN + 2 mM H₂O₂, or 0.3 mM CN + 2 mM H₂O₂, or 0.3 mM CN + 0.2 mM H₂O₂.

D. Kinetics of chromosomal fragmentation upon treatment with 3 mM CN + 2 mM H₂O₂, or 0.3 mM CN + 2 mM H₂O₂, or 0.3 mM CN + 0.2 mM H₂O₂ (from several gels like in “C”). Here and for the rest of the paper, fragmentation level at any time point is shown over the fragmentation level at time = 0.

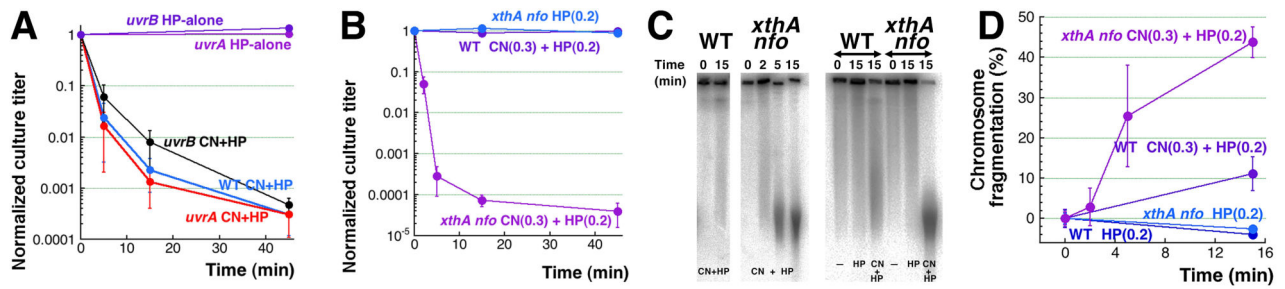


Fig. 2. Mutants in base excision repair are extremely sensitive to CN + H₂O₂ treatment, but not to H₂O₂-alone treatment

A. Kinetics of death of the *uvrA* and *uvrB* mutants, compared to WT cells, treated with 3 mM CN + 2 mM H₂O₂ or with 2 mM H₂O₂ alone.

B. Kinetics of death of the *xthA nfo* double mutant treated with 0.3 mM CN + 0.2 mM H₂O₂ or 0.2 mM H₂O₂ alone.

C. Representative pulsed-field gels demonstrating chromosomal fragmentation induced in the *xthA nfo* double mutant treated with 0.3 mM CN + 0.2 mM H₂O₂ compared to WT cells (left) or comparison of H₂O₂-alone treatment with CN + H₂O₂ treatment (right).

D. Quantification of the kinetics of chromosomal fragmentation in *xthA nfo* and WT cells upon treatment with 0.2 mM H₂O₂ alone or with 0.3 mM CN + 0.2 mM H₂O₂ (from several gels like in "C").

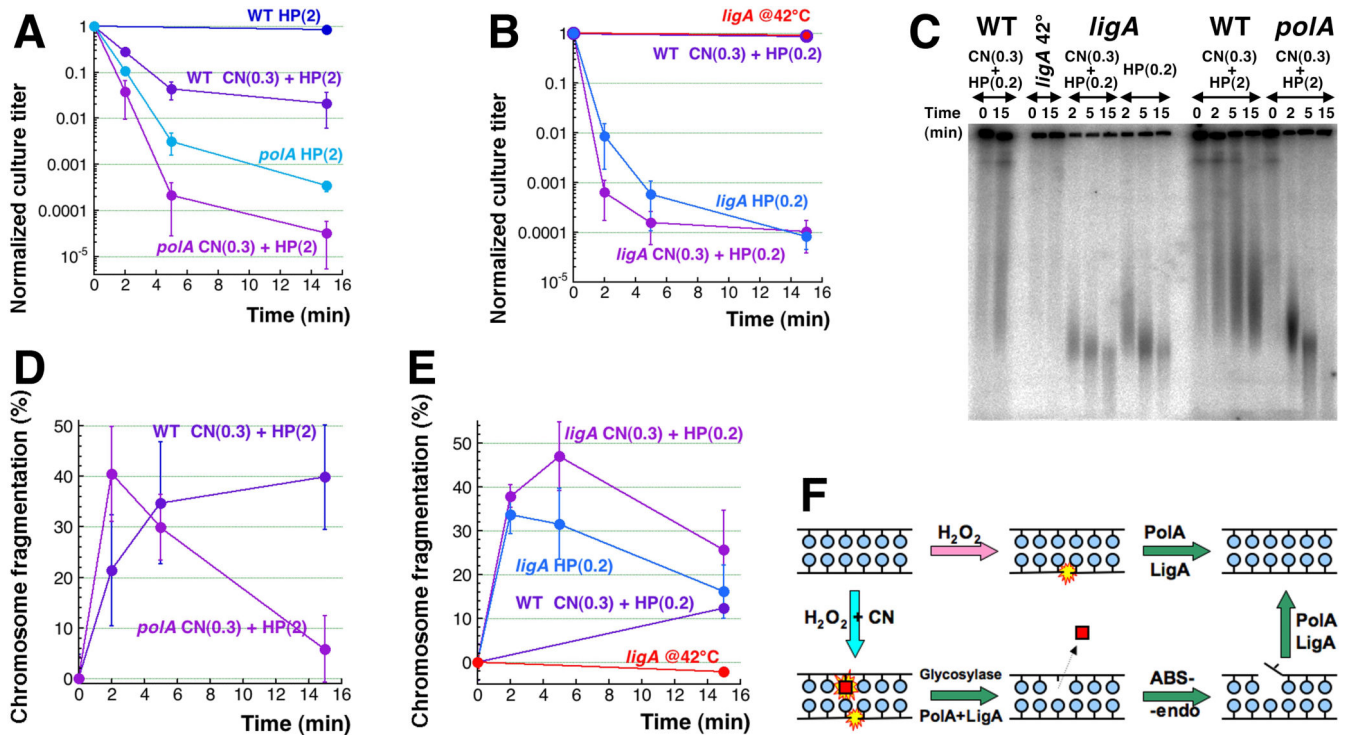


Fig. 3. Mutants in single-strand break repair are similarly sensitive to both treatments

A. Kinetics of death of the *polA12* (Ts) mutant and WT cells treated with 0.3 mM CN + 2 mM H₂O₂ or 2 mM H₂O₂ alone at 42°C.

B. Kinetics of death of the *ligA251* (Ts) mutant and WT cells treated with 0.3 mM CN + 0.2 mM H₂O₂ or 0.2 mM H₂O₂ alone at 42°C. The *ligA251* (Ts) mutant and WT cells are pre-grown at 28°C for two and a half hours prior to addition of 0.3 mM CN + 0.2 mM H₂O₂ or 0.2 mM H₂O₂, after which they are shifted to 42°C for the duration of the treatment. An untreated *ligA* control is also included to demonstrate that the mutant does not start dying due to the ligase defect for up to 15 minutes after being shifted to 42°C [45].

C. A representative pulsed-field gel demonstrating chromosomal fragmentation induced in the *ligA* mutant treated with 0.3 mM CN + 0.2 mM H₂O₂ or 0.2 mM H₂O₂ alone compared to WT at 42°C (along with untreated *ligA* control), and in the WT and *polA* mutant treated with 0.3 mM CN + 2 mM H₂O₂ at 42°C.

D. Quantification of the kinetics of chromosomal fragmentation upon treatment with 0.3 mM CN + 2 mM H₂O₂ in *polA* and WT cells (from several gels like in “C”).

E. Quantification of the kinetics of chromosomal fragmentation upon treatment with 0.3 mM CN + 0.2 mM H₂O₂ or 0.2 mM H₂O₂ alone in *ligA* and WT cells (from several gels like in “C”).

F. A scheme of base-excision repair and ss-break repair of H₂O₂-alone or CN + H₂O₂-induced DNA lesions.

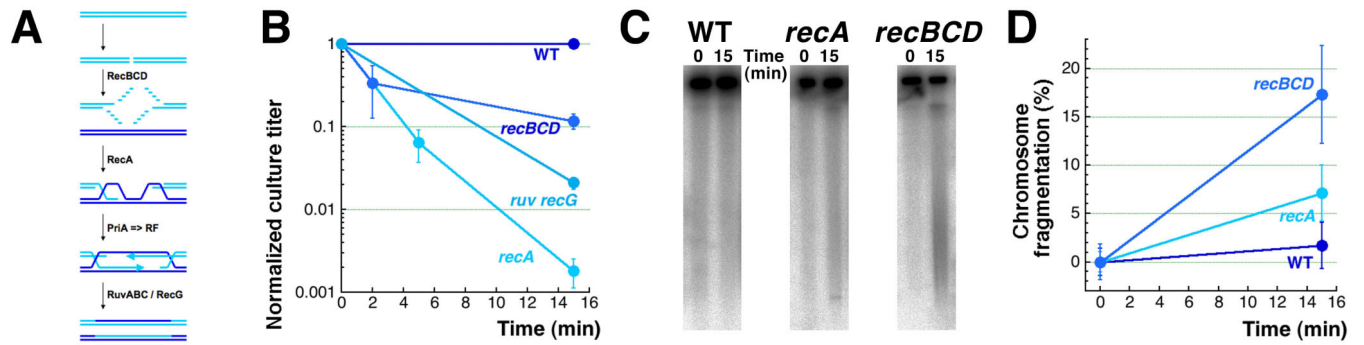


Fig. 4. H_2O_2 -alone-induced double-strand breaks and their recombinational repair

A. A scheme of double-strand break repair in *E. coli*, with the critical enzymes marking the corresponding stages.

B. Survival of recombinational repair mutants after 2 mM H_2O_2 -alone treatment.

C. A representative gel showing chromosome fragmentation in WT versus *recA* and *recBCD* mutants treated with 2 mM H_2O_2 .

D. Quantification of the chromosomal fragmentation upon treatment with 2 mM H_2O_2 .

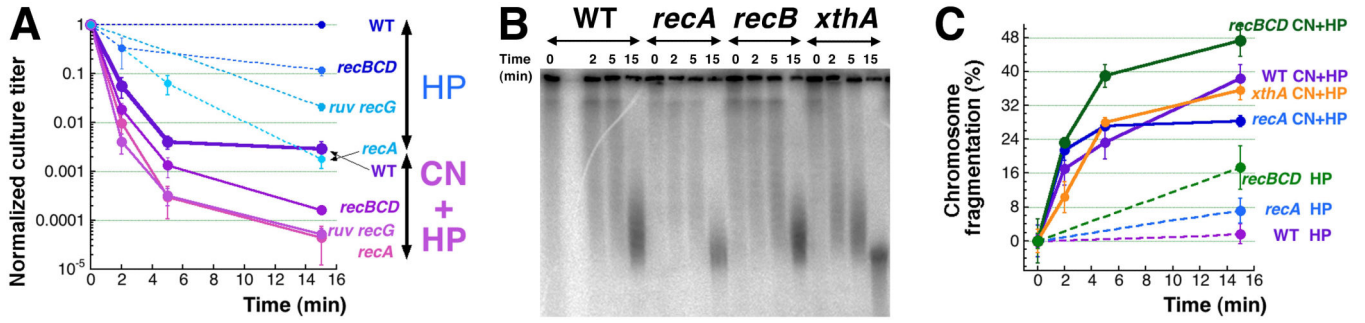


Fig. 5. Recombinational repair mutants are more sensitive than the WT cells to CN + H₂O₂ treatment

A. Kinetics of survival of the recombinational repair mutants treated with 0.3 mM CN + 2 mM H₂O₂. The survival of 2 mM H₂O₂-alone treatment is shown for comparison from Fig. 4B.

B. A representative pulsed-field gel of kinetics of chromosomal fragmentation in the *recA* and *recBC* mutants treated with 0.3 mM CN + 2 mM H₂O₂ compared with WT and *xthA* mutant cells.

C. Quantification of the kinetics of chromosomal fragmentation upon treatment with 0.3 mM CN + 2 mM H₂O₂ in *recA* and *recBC* mutants compared to WT and *xthA* mutant cells (from several gels like in “B”). For comparison, fragmentation levels of H₂O₂-alone treatment from Fig. 4D are also shown.

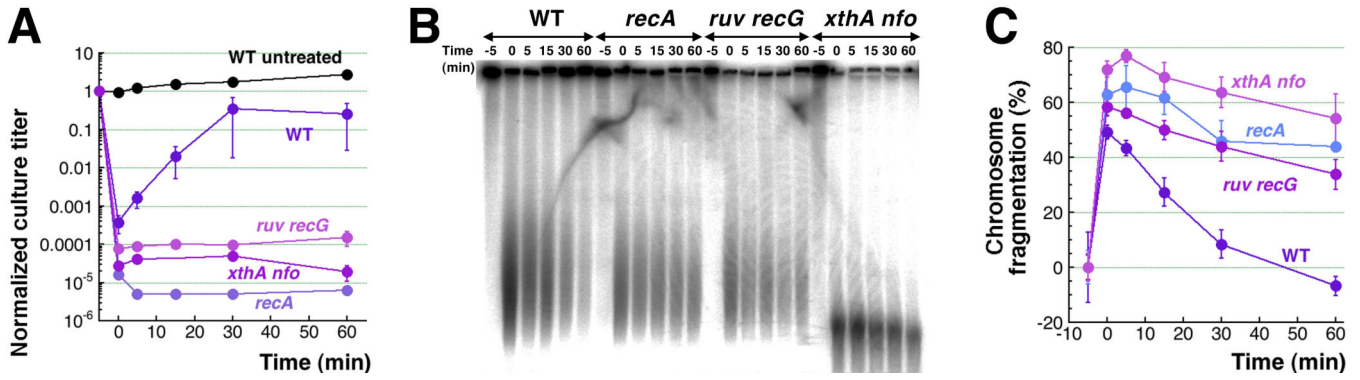


Fig. 6. Recombinational repair of CN + H₂O₂-induced double-strand breaks

A. Kinetics of survival/‘revival’ after 3 mM CN + 2 mM H₂O₂ treatment. After 5 minutes of treatment, CN and H₂O₂ were removed by pelleting cells by centrifugation, resuspending in fresh LB and allowing to recover at 37°C. Repair-deficient mutants, such as *recA*, *recG* *ruvABC* and *xthA nfo*, were included as negative controls. Growth of the untreated WT culture was also monitored in parallel.

B. A representative pulsed-field gel showing the disappearance of catastrophic chromosomal fragmentation in WT cells induced by 5-minute CN + H₂O₂ treatment, upon removal of the treatment at time = 0. In contrast, *recA*, *recG* *ruvABC* and *xthA nfo* mutants after the same treatment show only decrease in the levels of chromosomal fragmentation consistent with some linear DNA degradation.

C. Quantification of the disappearance of catastrophic chromosomal fragmentation in WT cells induced by 5-minute CN + H₂O₂ treatment upon their removal, compared to the lack of it in *recA*, *recG* *ruvABC* and *xthA nfo* mutants (from several gels like in ‘B’).

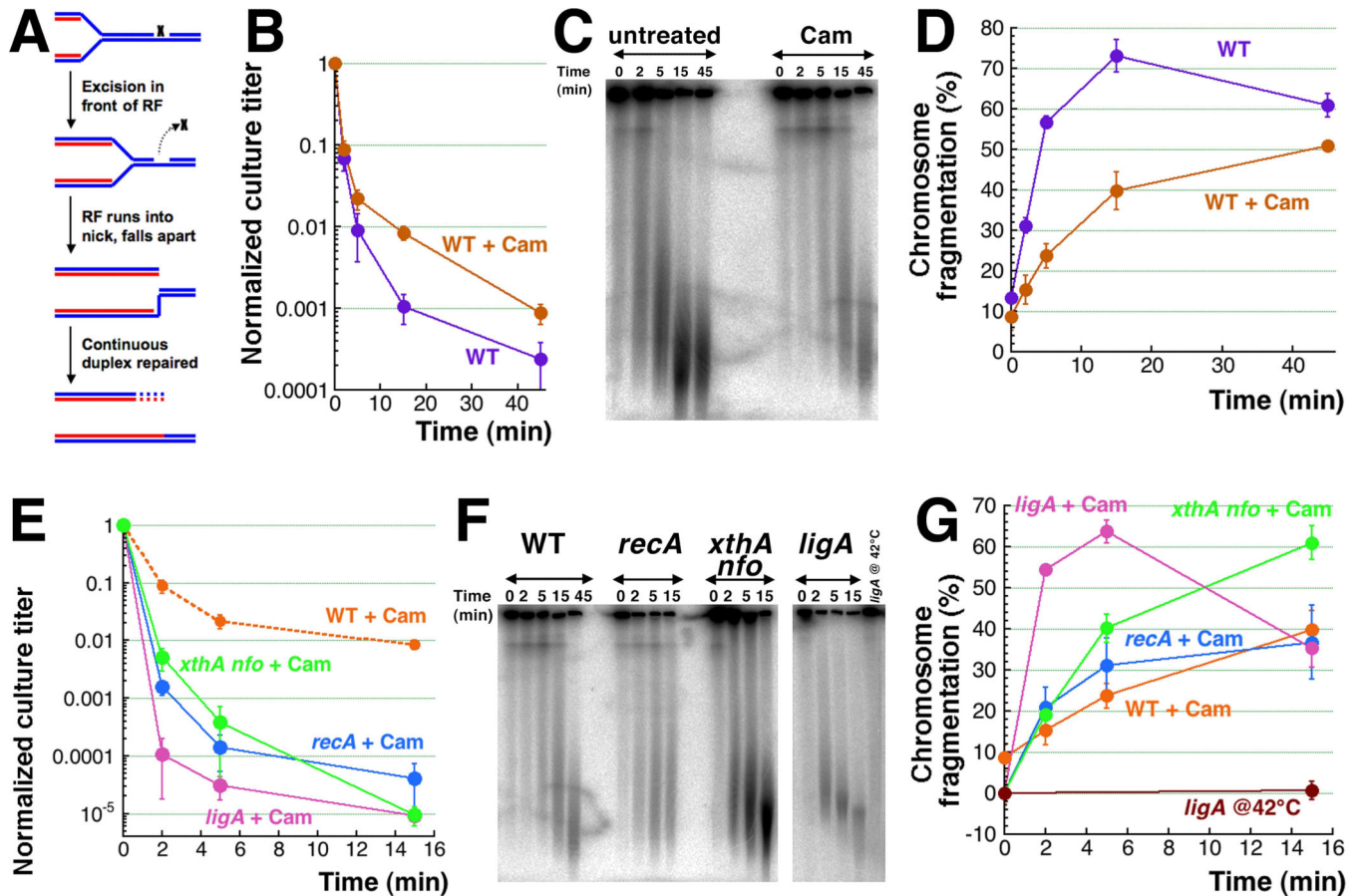


Fig. 7. CN+H₂O₂-induced killing and chromosome fragmentation in chloramphenicol-treated cells

A. A scheme of replication fork (RF) collapse at a ss-break in template DNA.

B. Kinetics of death in WT cells treated with 40 µg/ml chloramphenicol for two hours, to stop any replication in the chromosome, before treatment with 3 mM CN + 2 mM H₂O₂.

C. A representative gel showing the level of CN + H₂O₂-induced chromosomal fragmentation in chloramphenicol-treated versus untreated WT cells.

D. Quantification of the kinetics of chromosomal fragmentation from several gels like in "C".

E. Kinetics of death of non-replicating WT, *recA*, *ligA*(Ts) and *xthA nfo* cultures pre-treated with 40 µg/ml chloramphenicol (Cam) for 2 hours before treatment with 3 mM CN + 2 mM H₂O₂. The *ligA*(Ts) mutant was pre-grown and treated with chloramphenicol at 28°C and was shifted to 42°C upon adding CN + H₂O₂.

F. A representative pulsed-field gel demonstrating the catastrophic chromosomal fragmentation induced in chloramphenicol pre-treated non-replicating WT, *recA*, *ligA*(Ts) and *xthA nfo* cultures by the 3 mM CN + 2 mM H₂O₂ treatment.

G. Quantification of the kinetics of chromosomal fragmentation upon treatment with 3 mM CN + 2 mM H₂O₂ (from several gels like in 'F') in chloramphenicol pre-treated non-replicating WT, *recA*, *ligA*(Ts) and *xthA nfo* cultures. Lack of fragmentation in the untreated *ligA*(Ts) mutant at 42°C within the time frame of experiment is also shown.

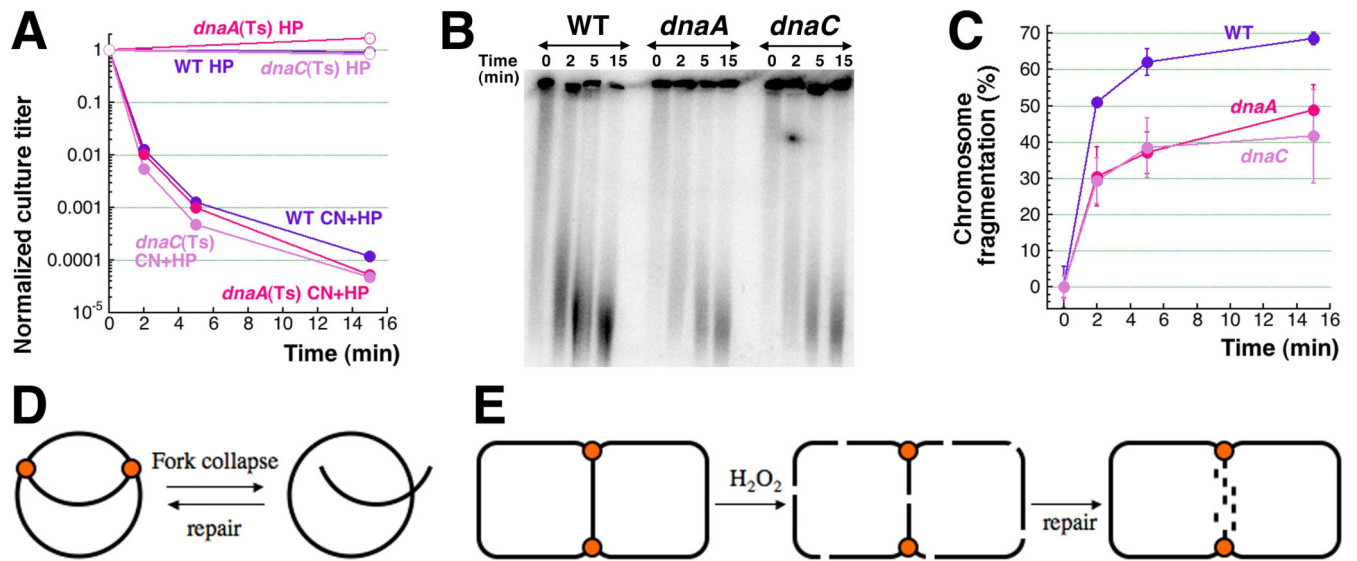


Fig. 8. CN + H₂O₂-induced killing and chromosome fragmentation in *dnaA(Ts)* and *dnaC(Ts)* mutant cells at 42°C

A. Kinetics of death of non-replicating *dnaA46* and *dnaC2* cultures in the AB1157 background (compared to exponential AB1157 cultures) upon treatment with 3 mM CN + 2 mM H₂O₂. The AB1157, *dnaA46(Ts)* and *dnaC2(Ts)* mutants were pre-grown at 28°C for two hours and then shifted to 42°C for two hours, following which the CN + H₂O₂ treatment was carried out at 42°C.

B. A representative pulsed-field gel demonstrating the catastrophic chromosomal fragmentation induced in non-replicating *dnaA46* and *dnaC2* cultures by the 3 mM CN + 2 mM H₂O₂ treatment (compared to the fragmentation observed in exponentially growing CN + H₂O₂ treated AB1157 cells). The AB1157, *dnaA46(Ts)* and *dnaC2(Ts)* mutants were pre-grown at 28°C for two hours and then diluted 20 times and shifted to 42°C for two hours, followed by the CN + H₂O₂ treatment at 42°C.

C. Quantification of the kinetics of chromosomal fragmentation upon treatment with 3 mM CN + 2 mM H₂O₂ (from several gels like in “B”) in non-replicating diluted *dnaA46* and *dnaC2* cultures, compared with WT cells.

D. Replication fork collapse and recombinational repair at the chromosome level.

E. Replication-independent double-strand breaks all over the replicating chromosome and their subsequent repair in the duplicated parts. Chromosome is presented as a theta-replicating structure with replication forks identified by the orange dots. Note the impossibility of recombinational repair in the unreplicated part of the chromosome, which is instead degraded, killing the cell.

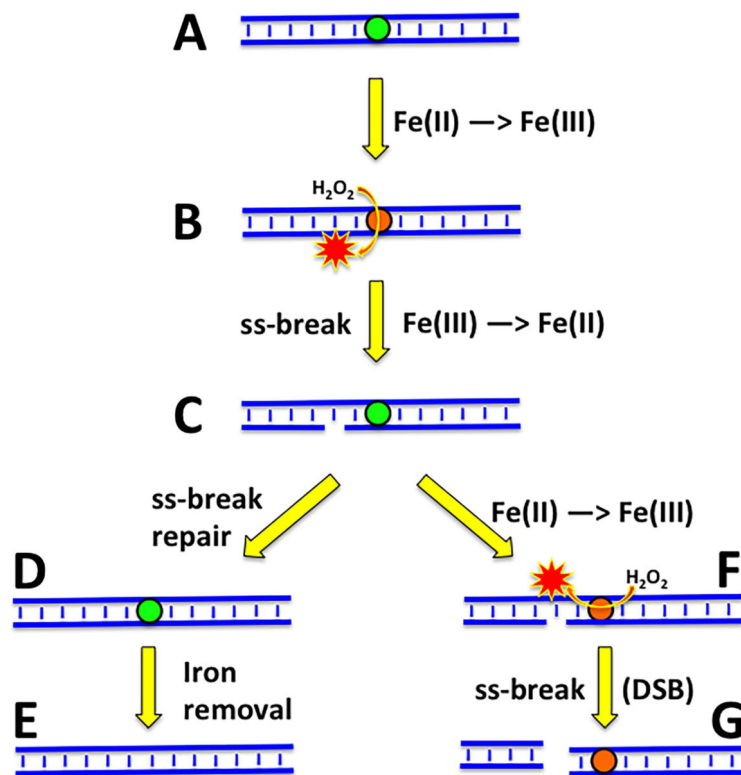


Fig. 9. A scheme of how hydrogen peroxide could induce direct double-strand DNA breaks, and how timely repair of one-strand DNA lesions could prevent them
 DNA duplex is shown in blue, the iron atoms are indicated by colored circles; green circles, Fe(II); orange circles, Fe(III). Red stars, hydroxyl radicals. **A**, formation of Fe(II)-DNA complex. **B**, hydrogen peroxide undergoes Fenton's reaction on DNA to generate hydroxyl radical. **C**, hydroxyl radical breaks one DNA strand. **D**, repair of the ss-break restores DNA integrity. **E**, the iron atom is removed from DNA (by Dps). **F**, in the absence of ss-break repair and subsequent iron removal, another Fenton's reaction with the same iron atom generates another hydroxyl radical nearby. **G**, the second DNA strand is disrupted opposite the first ss-break, breaking the DNA duplex.

Table 1

E. coli strains and plasmids (all strains are in the BW25113 background unless indicated otherwise).

Strain name	Relevant Genotype	Source/ CGSC# or Construction
AB1157	F ⁻ λ^- <i>rac- thi-1 hisG4 (gpt-proA)62 argE3 thr-1 leuB6 kdgK51 rfbD1 araC14 lacY1 galK2 xylA5 mtl-1 tsx-33 glnV44 rpsL31</i> .	[87]
L-216	AB1157 <i>dnaA46</i> (Ts) [88] <i>lac/CE-ori::bla</i>	Elena Kouzminova
L-393	AB1157 <i>dnaC2</i> (Ts) [64]	10827
AK4	AB1157 (<i>srlR-recA306</i>):Tn10	[89]
JB1	AB1157 <i>recBCD3::kan</i>	[90]
N2731	AB1157 <i>recG258</i> :Tn10	[91]
JJC754	AB1157 <i>ruvABC232::cat</i>	[92]
AK25	AB1157 <i>polA12</i> (Ts)(Tn10)	Lab Collection
GR501	<i>Hfr</i> (PO45), λ , <i>ligA251(ts)</i> , <i>relA1</i> , <i>spoT1</i> , <i>thiE1</i>	6087
LA20	GR501 <i>ligA251 ypeB::kan</i>	[93]
SX1253	F ⁻ , (<i>argF-lac</i>)169, <i>gal-490</i> , (<i>modF-ybhJ</i>) 803, λ [<i>c1857 (cro-bioA)</i>], <i>xthA791-YFP::cat</i> , <i>IN(rrnD-rrnE)1</i> , <i>rph-1</i>	12808
BW535	F ⁻ , <i>thr-1</i> , <i>araC14</i> , <i>leuB6</i> (Am), (<i>gpt-proA</i>) 62, <i>lacY1</i> , <i>tsx-33</i> , <i>glnX44</i> (AS), <i>galK2</i> (Oc), λ^- , <i>Rac-0</i> , <i>nth-1::kan, ble</i> , (<i>xthA-pncA</i>) 90, <i>hisG4</i> (Oc), <i>rfbC1</i> , <i>mgl-51</i> , <i>nfo-1::kan</i> , <i>rpsL31</i> (<i>strR</i>), <i>kdgK51</i> , <i>xylA5</i> , <i>mtl-1</i> , <i>argE3</i> (Oc), <i>thiE1</i>	7047
N3055	F ⁻ , λ^- , <i>IN(rrnD-rrnE)1</i> , <i>rph-1</i> , <i>uvrA277</i> :Tn10	6661
JW0762-2	<i>uvrB751::kan</i>	8819
BW25113	F ⁻ , (<i>araD-araB</i>)567, <i>lacZ4787::rrnB-3</i> , λ^- , <i>rph-1</i> , (<i>rhaD-rhaB</i>)568, <i>hsdR514</i>	[46]
TM21	(<i>srlR-recA306</i>):Tn10	BW25113 \times P1 AK4
TM22	<i>recBCD3::kan</i>	BW25113 \times P1 JB1
TM23	<i>ruvABC232::cat</i>	BW25113 \times P1 JJC754
TM24	<i>recG258</i> :Tn10 <i>ruvABC232::cat</i>	TM23 \times P1 N2731
TM25	<i>xthA791-YFP::cat</i>	BW25113 \times P1 SX1253
TM26	<i>xthA791-YFP::cat nfo-1::kan</i>	TM25 \times P1 BW535
TM 27	<i>ligA251 ypeB::kan</i>	BW25113 \times P1 LA20
TM 28	<i>polA12</i> (Ts) (Tn10)	BW25113 \times P1 AK25
TM 29	<i>uvrA277</i> :Tn10	BW25113 \times P1 N3055
Plasmids		
pCP20	Flp recombinase gene on a temperature-sensitive replicon	[48]