Fine-tuning of Substrate Affinity Leads to Alternative Roles of *Mycobacterium tuberculosis* Fe²⁺-ATPases*

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Sarju J. Patel[‡], Brianne E. Lewis[§], Jarukit E. Long[¶], Subhalaxmi Nambi[¶], Christopher M. Sassetti^{¶|}, Timothy L. Stemmler[§], and José M. Argüello^{‡1}

From the [‡]Department of Chemistry and Biochemistry, Worcester Polytechnic Institute, Worcester, Massachusetts 01609, [§]Department of Pharmaceutical Sciences, Wayne State University, Detroit, Michigan 48201, [¶]Department of Microbiology and Physiological Systems, University of Massachusetts Medical School, Worcester, Massachusetts 01655, and [∥]Howard Hughes Medical Institute, Chevy Chase, Maryland 20815

Little is known about iron efflux transporters within bacterial systems. Recently, the participation of Bacillus subtilis PfeT, a P_{1B4} -ATPase, in cytoplasmic Fe²⁺ efflux has been proposed. We report here the distinct roles of mycobacterial P_{1B4} -ATPases in the homeostasis of Co²⁺ and Fe²⁺. Mutation of Mycobacterium smegmatis ctpJ affects the homeostasis of both ions. Alternatively, an *M. tuberculosis ctpJ* mutant is more sensitive to Co²⁺ than Fe^{2+} , whereas mutation of the homologous *M. tuberculosis ctpD* leads to Fe²⁺ sensitivity but no alterations in Co²⁺ homeostasis. In vitro, the three enzymes are activated by both Fe²⁺ and Co^{2+} and bind 1 eq of either ion at their transport site. However, equilibrium binding affinities and activity kinetics show that *M. tuberculosis* CtpD has higher affinity for Fe²⁺ and twice the Fe²⁺-stimulated activity than the CtpJs. These parameters are paralleled by a lower activation and affinity for Co^{2+} . Analysis of Fe²⁺ and Co²⁺ binding to CtpD by x-ray absorption spectroscopy shows that both ions are five- to six-coordinate, constrained within oxygen/nitrogen environments with similar geometries. Mutagenesis studies suggest the involvement of invariant Ser, His, and Glu residues in metal coordination. Interestingly, replacement of the conserved Cys at the metal binding pocket leads to a large reduction in Fe²⁺ but not Co²⁺ binding affinity. We propose that CtpJ ATPases participate in the control of steady state Fe²⁺ levels. CtpD, required for *M. tuberculosis* virulence, is a high affinity Fe^{2+} transporter involved in the rapid response to iron dyshomeostasis generated upon redox stress.

Iron is an essential micronutrient required for numerous biological processes as it is used as a prosthetic group by several different enzymes (1, 2). However, in excess, it can be toxic due to its participation in Fenton chemistry and potential mismetallation in non-iron-containing metalloproteins. In this context, damage of iron-sulfur centers and mononuclear iron enzymes produced by various redox stresses are particular con-

tributors to iron dyshomeostasis and consequent toxicity (3-6). Characterization of bacterial Fe²⁺ homeostasis has mainly been focused in mechanisms of uptake (by divalent metal, siderophore, and heme transporters), transcriptional regulation (by Fur and IdeR systems), and Fe²⁺ sequestration (by bacterioferritin and Dps proteins) (2, 7-9). Nevertheless, studies have suggested that cation diffusion facilitators and iron-citrate transporters participate in Fe^{2+} efflux (10–12). We recently observed that Bacillus subtilis PfeT, a P1B4-ATPase, confers Fe²⁺ tolerance (13). PfeT is expressed under the control of PerR in response to peroxide exposure (14). Initial biochemical characterization showed that Fe²⁺ activates isolated PfeT ATPase, leading to a higher V_{max} than generated by Co^{2+} , which is the proposed substrate of P_{1B4} -ATPases (13, 15–17). Interestingly, phenotypic analysis of Listeria monocytogenes lacking the P_{1B4}-ATPase FrvA showed a role of this ATPase in resistance to heme toxicity (18). These observations suggest a significant role of this subfamily of P-type ATPases in Fe²⁺ homeostasis (13, 14).

P1B4-ATPases present in prokaryotes and plant chloroplasts are part of the large family of P-type ATPases (15, 19, 20). P-type ATPases are polytopic membrane proteins that transport a variety of ions using the energy provided by ATP hydrolysis (21–23). The P_{1B} subgroup includes proteins responsible for the efflux of cytoplasmic transition metals including Cu⁺, Zn^{2+} , Co^{2+} , and Ni^{2+} (19, 22, 23). The specificity of their transmembrane metal binding sites (TM-MBSs)² is determined by invariant amino acid sequences in their last three transmembrane segments (TMs) (17, 19, 24-26). However, activation by non-cognate substrates has been reported for most P1B-ATPase subgroups (22, 27). In particular, activation of P_{1B4}-ATPases by Co²⁺, Ni²⁺, Ca²⁺, Cu⁺, Zn²⁺, and Cd²⁺ has been proposed (15–17, 28–30). We previously reported in vivo and in vitro functional studies directed at understanding the metal selectivity and consequent physiological roles of mycobacterial P_{1B4}-ATPases (15, 16). The presence of one or two P_{1B4}-ATPasecoding genes in mycobacterial species enabled comparative studies of Mycobacterium smegmatis CtpJ and Mycobacterium



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¹ To whom correspondence should be addressed: Dept. of Chemistry and Biochemistry, Worcester Polytechnic Institute, 100 Institute Rd., Worcester, MA 01609. Tel.: 508-831-5326; Fax: 508-831-4116; E-mail: arguello@ wpi.edu.

² The abbreviations used are: TM-MBS, transmembrane metal binding site; AAS, atomic absorbance spectroscopy; EXAFS, extended x-ray absorption fine structure; LIMM, low iron defined medium; STN, streptonigrin; TM, transmembrane segment; XANES, x-ray absorption near edge spectroscopy; XAS, x-ray absorption spectroscopy; Ms, *M. smegmatis*; Mt, *M. tuberculosis*; TCEP, tris(2-carboxyethyl)phosphine.

tuberculosis CtpJ and CtpD. *In vitro*, MsCtpJ and MtCtpJ display a higher activation by Co²⁺ and Ni²⁺ compared with Zn²⁺, although equilibrium binding affinities show K_D values for $Zn^{2+} < Co^{2+} = Ni^{2+}$ (15, 16). *In vivo*, *ctpJ* expression is induced by Co²⁺, whereas mutant strains show accumulation and sensitivity to the metal. On the contrary, the expression of the homologous *MtctpD* is not induced by Co²⁺ but rather by redox stress. Mutation of *MtctpD* does not lead to Co²⁺ sensitivity or higher intracellular levels of this metal. Nevertheless, MtCtpD ATPase activity is partially activated by Co²⁺. Surprisingly, MtCtpD but not MtCtpJ is required for *M. tuberculosis* virulence.

Previous studies have not explored the activation of mycobacterial P_{1B4} -ATPases by Fe²⁺. Could a differential activation by Co²⁺/Fe²⁺ explain the presence of paralogous genes in *M. tuberculosis*? Why is MtCtpD but not MtCtpJ required for virulence? To address these questions, we examined the activation of *M. smegmatis* and *M. tuberculosis* P_{1B4} -ATPases by Fe²⁺ and their participation in Fe²⁺ homeostasis and stress response. In addition, we explored the molecular basis of the different Fe²⁺ and Co²⁺-ATPase activities by determining the coordination of these metals during transport by MtCtpD.

Experimental Procedures

Mycobacterium Strains and Culture Conditions-M. smegmatis mc2155, M. tuberculosis H37Rv, and derived strains were grown in 7H9 liquid medium (BD Biosciences, Difco) supplemented with 0.2% glycerol, 0.05% Tween 80, and 10% ADN supplement (0.5% bovine serum albumin, 0.2% dextrose, and 0.085% NaCl) or in low iron defined medium (LIMM) containing 0.5% L-asparagine, 0.5% KH₂PO₄, 2% glycerol, 0.05% Tween 80, and 10% ADN, pH 6.8 (31). LIMM was treated with Chelex-100 (Sigma) and before use supplemented with 3.7 µM ZnCl₂, 0.8 µM MnCl₂, and 0.4 mM MgCl₂. This medium contained less than 1 μ M residual iron as determined by atomic absorption spectroscopy (AAS) (PerkinElmer Life Sciences PinAAcle 900z). Construction of $Ms\Delta ctpJ$ (MSMEG 5403), $Mt\Delta ctpD$ (Rv1469), $Mt\Delta ctpJ$ (Rv3743), and $Mt\Delta ctpD:\Delta ctpJ$ mutant and complemented strains was described previously (15, 16).

Iron and Hemin Sensitivity Tests—Liquid LIMM cultures of *M. smegmatis* mc2155, *M. tuberculosis* H37Rv, mutant, and complemented strains were inoculated at 0.05 A_{600} from late exponential phase cultures and supplemented with the desired concentration of FeCl₃ or hemin (Sigma). A hemin stock solution was prepared at 25 mg/ml in 1.4 M NaOH. Cells were incubated for 16 h (*M. smegmatis*) or 5 days (*M. tuberculosis*), and A_{600} was measured. To avoid hemin interference in A_{600} readings, cells grown in hemin-containing medium were collected, washed twice with LIMM, and suspended in the original LIMM volume, and A_{600} was measured.

Streptonigrin Sensitivity Tests—M. smegmatis mc2155, M. tuberculosis H37Rv, mutant, and complemented strains grown in LIMM to midlog phase were diluted to 0.05 A_{600} in LIMM. The cultures were supplemented with 1 µg ml⁻¹ streptonigrin (STN) and 10 µM FeCl₃ as indicated in the figures. Cells were incubated for 16 h (*M. smegmatis*) or 5 days (*M. tuberculosis*), and A_{600} was measured.

Metal Accumulation Assays—Liquid LIMM cultures in midexponential phase ($A_{600} \sim 1.0$) were supplemented with increasing concentrations of FeCl₃ and incubated for 4 (*M. smegmatis*) or 8 h (*M. tuberculosis*). After this incubation, cells were harvested and washed with 5 mM EDTA and 0.9% NaCl. Aliquots were taken for protein determinations (32). Pellets were acid-digested with 0.5 ml of NO₃H for 1 h at 80 °C and then overnight at 20 °C. Digestions were concluded by adding ½ volume of 30% (v/v) H₂O₂ followed by a 1:5 dilution with water. Metal contents in digested samples were measured by AAS.

Protein Expression and Purification-Preparation of E. coli LMG194 $\Delta copA$ strains carrying *M. smegmatis ctpJ* (MSMEG_ 5403) or M. tuberculosis ctpD (Rv1469) or ctpJ (Rv3743) in pBAD-TOPO/His vectors was described previously (15, 16). Cells were grown at 37 °C in ZYP-505 autoinduction medium supplemented with 0.05% arabinose, 100 mg ml⁻¹ ampicillin, and 50 mg ml⁻¹ kanamycin (33). Cells were harvested at 16 h postinoculation; washed with 25 mM Tris, pH 7.0, 100 mM KCl, and 20% glycerol; and stored at -70 °C. Expressed mycobacterial proteins contained a C-terminal His₆ tag sequence preceded by a tobacco etch virus protease recognition sequence. Protein purification was carried out as described previously (15, 24, 34). Briefly, cells were disrupted in a French press, and membranes were isolated by centrifugation. Membranes were treated with 0.75% dodecyl β-D-maltoside (Calbiochem), 25 mm Tris, pH 8.0, 100 mm sucrose, 500 mm NaCl, and 1 mm phenylmethylsulfonyl fluoride. The solubilized membrane protein suspension was cleared by centrifugation at 163,000 \times *g* for 1 h, and proteins were affinity-purified using Ni²⁺-nitrilotriacetic acid resin. The His₆ tag was removed from the C terminus by treatment with His₆-tagged tobacco etch virus protease (35). Tobacco etch virus-His₆ was removed by affinity purification with Ni²⁺-nitrilotriacetic acid resin. Protein purity was analyzed by 10% SDS-PAGE followed by Coomassie Brilliant Blue staining or Western blotting using an anti-His₆ tag antibody (GenScript, Piscataway, NJ). Isolated proteins (3 mg/ml) were stored at -20 °C in 25 mM Tris, pH 8.0, 100 mM sucrose, 50 mM NaCl, 0.01% dodecyl β-D-maltoside, 0.01% asolectin, 1 mM phenylmethylsulfonyl fluoride, and 20% glycerol. Prior to ATPase activity determinations, proteins (1 mg/ml) were treated with 0.5 mM EDTA and 0.5 mM tetrathiomolybdate for 45 min at room temperature. Chelators were removed using Ultra-30 Centricon (Millipore, Darmstadt, Germany) filtration devices.

Mutagenesis of MtCtpD Metal Binding Site—MtctpD cloned into pBAD-TOPO/His vector was used as a template to introduce the mutations coding for the single substitutions Ser-316 (S316A), Cys-318 (C318A), His-642 (H642A), Glu-643 (E643A and E643D), Gly-644 (G644A), Ser-645 (S645A), and Thr-646 (T646A and T646S) and the multiple replacements S316C/ C318S. All mutations were introduced using a Q5[®] site-directed mutagenesis kit (New England Biolabs, Ipswich, MA). The sequences of primers used in this study are available upon request to the corresponding author. DNA sequences were confirmed by automated sequencing.



Fe²⁺ Transport ATPases

 Fe^{2+} Binding to Proteins—Metal binding to isolated enzymes was measured as described previously (15, 36). Five micromolar His-less enzyme was incubated for 1 min at 4 °C in 25 mM HEPES-NaOH, pH 7.5, 150 mM NaCl, 1 mM tris(2-carboxyethyl)phosphine (TCEP), and 20 μ M either FeSO₄ or CoCl₂. Excess metal was removed by washing in a 30-kDa-cutoff Centricon filtration device. Protein samples were acid-digested as described above, and metal concentrations were measured using AAS.

Metal binding affinities were determined using the divalent metal-binding chromophore mag-fura-2 (Invitrogen) (15, 36). Five micromolar His-less protein and 10 µM mag-fura-2 were titrated with 1 mM Fe^{2+} or Co^{2+} in Chelex-treated buffer (25 тм HEPES-NaOH, pH 7.5, 150 mм NaCl, and 1 mм TCEP). Free mag-fura-2 was determined by monitoring A_{366} (ϵ_{366} 29,900 1/M·cm). Free metal concentrations were calculated from $K_I = [I \cdot Me^{2+}]/[I_{\text{free}}][Me^{2+}]_{\text{free}}$ where *I* is mag-fura-2, Me is the metal ion, and K_I is the association constant of mag-fura-2 with each metal. K_D values of 1.5 μ M for Fe²⁺ and 2.8 μ M for Co²⁺ were experimentally determined for the metal/magfura-2 interactions. The metal-protein K_D values were calculated from $\nu = n[\text{Me}^{2+}_{\text{free}}]/K_D(1 + ([\text{Me}^{2+}_{\text{free}}]/K_D))$ where ν is the molar ratio of metal bound to protein and *n* is the apparent stoichiometry (37). Reported errors for K_D and n are asymptotic standard errors provided by the fitting software KaleidaGraph (Synergy, Reading, PA).

ATPase Assays—ATPase assays were performed as described (15, 24, 34). The assay mixture contained 50 mM Tris, pH 7.4, 50 тм NaCl, 3 тм MgCl₂, 3 тм ATP, 0.01% asolectin, 0.01% dodecyl β -D-maltoside, 2.5 mM TCEP, 20 μ g/ml purified protein, and freshly prepared transition metal ions at the desired concentrations. Fe³⁺ was added as FeCl₃, Cu²⁺ was added as CuSO₄, and in both cases TCEP was not included in the assay medium. Cu⁺ was obtained by including TCEP with CuSO₄ salt. Fe²⁺ and Zn²⁺ were included in the assay medium as the sulfate salts, whereas Co²⁺, Ni²⁺, and Mn²⁺ were included as their chloride salts. ATPase activity was stopped after a 20-min incubation at 37 °C, and released P_i was determined (38). ATPase activity measured in the absence of transition metals was subtracted from plotted values. Curves of ATPase activity metal dependence were fit to $\nu = V_{\text{max}}[\text{Me}^{+/2+}]/([\text{Me}^{+/2+}] +$ $K_{1/2}$). The reported standard errors for V_{max} and $K_{1/2}$ are asymptotic standard errors reported by the fitting software KaleidaGraph.

X-ray Absorption Spectroscopy (XAS)—XAS samples were loaded in a Coy anaerobic chamber at a 1:1 metal:protein molar ratio (1.2 mM Fe²⁺ or 0.43 mM Co²⁺). Sample was injected into a Kapton-wrapped Lucite cell, flash frozen, and stored in liquid nitrogen. XAS data were collected at the Stanford Synchrotron Radiation Lightsource on beamline 7-3 equipped with a Si(220) double crystal monochromator with a harmonic rejection mirror. Fluorescence spectra were collected using a 30-element germanium solid-state detector (Canberra, Meriden, CT). During data collection, the continuous flow liquid helium cryostat (Oxford Instruments, Concord, MA) was stabilized at 10 K. Iron and cobalt data were collected using a 6- μ m magnesium or a 3- μ m iron filter, respectively, placed between the cryostat and the detector to reduce unassociated scattering. Iron and cobalt foil spectra were collected simultaneously with protein data for direct energy calibration of the data. The first inflection points for iron and cobalt were set at 7111.3 and 7709.5 eV, respectively. Iron XAS spectra were recorded using 5-eV steps in the pre-edge regions (6900–7094 eV), 0.25-eV steps in the edge regions (7095–7135 eV), and 0.05-Å⁻¹ increments in the extended x-ray absorption fine structure (EXAFS) region (to k = 13.5Å⁻¹), integrating from 1 to 20 s in a k^3 -weighted manner. Cobalt XAS spectra were recorded using 5-eV steps in the edge regions (7702–7780 eV), and 0.05-Å⁻¹ increments in the edge regions 1 to 25 s in a k^3 -weighted manner. A total of eight scans were taken for each sample, and these were then averaged.

XAS spectra were processed and analyzed using the EXAF-SPAK program suite for Macintosh OSX (39). A Gaussian function was used in the pre-edge region, and a three-region cubic spline was used in the EXAFS region. EXAFS data were converted to k space using E_0 values of 7130 and 7745 eV for iron and cobalt, respectively. Spectra were simulated using single and multiple scattering amplitude and phase functions generated using the Feff v8 software integrated within EXAFSPAK. Single scattering models were calculated for oxygen, nitrogen, and carbon to simulate possible iron- or cobalt-ligand environment. Calibrated scale factors and model E₀ values were not allowed to vary during fitting; the scale factor for iron was 0.95, and that for cobalt was 0.98. Iron data were fit out to a k value of 13.5 Å⁻¹. Calibration from Fe²⁺ and Fe³⁺ model compounds was used for determination of E_0 and scale factor parameters for iron. E_0 values for Fe–O and Fe–C were set at -10 eV. Cobalt data were fit out to a *k* value of 13.5 Å⁻¹. Calibration from Co^{2+} model compounds was used for determination of E_0 and scale factor parameters. E₀ values for Co-O and Co-C were set at -11.3 eV. EXAFS spectra were simulated using both filtered and unfiltered data; however, simulation results are presented only for fits to unfiltered (raw) data. Simulation protocols and criteria for determining the best fit have been described elsewhere (25).

Results

Mycobacterial P_{1B4} -ATPases Confer Fe^{2+} Tolerance—We previously reported the activation of mycobacterial P_{1B4}-ATPases by Co²⁺ (15, 16). However, *M. tuberculosis* CtpJ and CtpD appear to have distinct roles in Co²⁺ tolerance and cellular response to redox stress (16). We recently observed that B. subtilis PfeT, a P_{1B4}-ATPase in the PerR regulon, transports and confers tolerance to Fe^{2+} in addition to Co^{2+} (13). Similarly, the L. monocytogenes P1B4-ATPase, FrvA, confers resistance to heme toxicity (18). Thus, we hypothesized that selective activation by Fe²⁺ might explain the presence of the *ctpJ* and ctpD paralogs in the M. tuberculosis genome. To test this idea, the capability of mycobacterial P_{1B4}-ATPases to confer tolerance to Fe²⁺ was assessed. Fig. 1 shows the growth of *Ms* Δ *ctpJ*, *Mt* Δ *ctpD*, *Mt* Δ *ctpJ*, and *Mt* Δ *ctpD*: Δ *ctpJ* double mutant strains in LIMM supplemented with different concentrations of FeCl_3 or hemin. The Ms $\Delta ctpJ$ strain showed a growth defect at high FeCl₃ or hemin as compared with



Fe²⁺ Transport ATPases



FIGURE 1. Growth response of mycobacterial P₁₈₄-ATPase mutant strains to iron, hemin, and streptonigrin/iron. *A* and *B*, *M*. smegmatis WT ($\textcircled{\bullet}$; black) and $\Delta ctpJ$ ($\textcircled{\bullet}$; blue) and complemented (\bigtriangledown ; blue) strains were grown in the presence of increasing concentrations of FeCl₃ (*A*) or hemin (*B*) in the LIMM for 16 h, and A_{600} was measured. *C*, *M*. smegmatis WT (gray bars), $\Delta ctpJ$ (blue bars), and complemented (white bars) strains were grown in LIMM supplemented with 1 μ g ml⁻¹ STN and 10 μ M FeCl₃, and growth was measured at 16 h. *D* and *E*, *M*. tuberculosis WT ($\textcircled{\bullet}$; black), $\Delta ctpD$ ($\textcircled{\blacksquare}$; blue) and complemented (\Box ; blue), $\Delta ctpJ$ (\bigstar ; green) and complemented (Δ ; green), and $\Delta ctpD:\Delta ctpJ$ (\bigstar ; red) strains were grown in the presence of increasing concentrations of FeCl₃ (*D*) or hemin (*E*) in LIMM for 5 days, and A_{600} was measured. *F*, *M*. tuberculosis WT (gray bars), $\Delta ctpD$ (green bars), $\Delta ctpD$ complemented (white bars), $\Delta ctpJ$ (blue bars), $\Delta ctpJ$ complemented (white bars), bctpJ (blue bars), $\Delta ctpJ$ complemented (white bars), and $\Delta ctpD:\Delta ctpJ$ (red bars) strains were grown in LIMM supplemented with 1 μ g ml⁻¹ STN and 10 μ M FeCl₃, growth was measured at 5 days, and A_{600} was measured. Data are the mean \pm S.E. (error bars) of three independent experiments. Significant differences from the WT as determined by Student's t test are indicated (*, p < 0.05).

M. smegmatis WT (Fig. 1, A and B). The complemented $Ms\Delta ctpJ$ strain, carrying the plasmid pMV306 harboring the full-length MsctpJ gene under the regulation of its native promoter, showed similar growth as the M. smegmatis WT. A comparable deficiency was observed in the $Mt\Delta ctpJ$ strain grown in the presence of 1 mM FeCl_3 (Fig. 1D). However, this mutant strain was not affected by the presence of hemin in the medium. On the contrary, mutation of the MtctpD gene significantly affected the growth in both FeCl₃- and hemin-supplemented medium. Interestingly, the $Mt\Delta ctpD:\Delta ctpJ$ double mutant strain showed a behavior identical to that of the $Mt\Delta ctpD$ strain. In all cases, complemented strains showed the growth phenotype of *M. tuberculosis* WT (Fig. 1, *D* and *E*). The data indicate that MsCtpJ and MtCtpD confer iron tolerance when cells are exposed to relatively high metal levels. Exploring their role at lower iron levels, the sensitivity to STN was tested. STN is a guinone antibiotic whose activity is correlated with intracellular iron availability (40). The $Ms\Delta ctpJ$, $Mt\Delta ctpD$, $Mt\Delta ctpJ$, and $Mt\Delta ctpD:\Delta ctpJ$ mutant strains displayed a significantly increased STN sensitivity in LIMM supplemented with 1 μ g ml⁻¹ STN and only 10 μ M FeCl₃ (Fig. 1, C and *F*). In contrast to their distinct tolerance to high Fe^{3+} in the medium, there was no significant difference in the sensitivity of $Mt\Delta ctpD$ and $Mt\Delta ctpJ$ strains to STN-Fe²⁺.

These results suggest that to different extents mycobacterial P_{1B4} -ATPases contribute to Fe²⁺ homeostasis by driving this metal efflux. To further explore this hypothesis, $Ms\Delta ctpJ$, $Mt\Delta ctpD$, $Mt\Delta ctpJ$, and $Mt\Delta ctpD:\Delta ctpJ$ mutant strains were challenged with sublethal concentrations of FeCl₃, and the resulting cellular Fe²⁺ levels were determined. Consistent with the iron sensitivity phenotypes (Fig. 1), Fe^{2+} accumulation was observed in mutant strains (Fig. 2, A and B). Iron levels in the $Ms\Delta ctpJ$ strain were approximately 5 times higher than those in *M. smegmatis* WT (Fig. 2A). The partial recovery observed in the $Ms\Delta ctpJ$ mutant strain complemented with MsctpJ appears to be associated with lower levels of transcript (35% of WT; not shown). Reinforcing the predominant role of MtCtpD in Fe²⁺ homeostasis, a significant increase of Fe²⁺ content was observed in the $Mt\Delta ctpD$ strain, whereas 50% smaller changes were observed in the $Mt\Delta ctpJ$ strain. Similar Fe²⁺ accumulation was observed in the $Mt\Delta ctpD$ and $Mt\Delta ctpD$: $\Delta ctpJ$ double mutant strains (Fig. 2B). These results suggest that although mycobacterial CtpJs are involved in controlling Co^{2+} levels they also participate



FIGURE 2. Iron levels in mycobacterial P_{1B4}-ATPase mutant strains. *A*, *M*. smegmatis WT (\bigcirc ; black), $\Delta ctpJ$ (\bigtriangledown ; blue), and complemented (\bigtriangledown ; blue) strains grown in LIMM supplemented with increasing concentrations of FeCl₃ for 4 h. *B*, *M*. tuberculosis WT (\bigcirc ; black), $\Delta ctpJ$ (\blacksquare ; green), $\Delta ctpD$ complemented (\square ; green), $\Delta ctpJ$ (\blacktriangle ; blue), and $\Delta ctpD$: $\Delta ctpJ$ (\blacklozenge ; green), $\Delta ctpJ$ (\blacksquare ; green), $\Delta ctpJ$ (\blacksquare ; green), $\Delta ctpJ$ (\blacksquare ; green), $\Delta ctpJ$ complemented (\square ; green), $\Delta ctpJ$ (\blacklozenge ; blue), and $\Delta ctpD$: $\Delta ctpJ$ (\blacklozenge ; red) strains grown in LIMM supplemented with increasing concentrations of FeCl₃ for 8 h. Data are the mean \pm S.E. (error bars) of three independent experiments.

in Fe²⁺ efflux, particularly when they are the only P_{1B4} -ATPase in the organism as in *M. smegmatis*. In contrast, MtCtpD appears to play a dominant role in maintaining the cytoplasmic Fe²⁺ level in this organism.

Distinct Biochemical Properties of Mycobacterial P_{1B4}-ATPase-P-ATPases couple the transmembrane transport of their substrate to ATP hydrolysis following the Albers-Post E1/E2-like mechanism (23). Consequently, the metal dependence of ATPase activity provides a starting point to analyze substrate selectivity. Previous reports showed that MsCtpJ, MtCtpD, and MtCtpJ are differently activated by Co²⁺, Ni²⁺, and to a lesser extent Zn^{2+} (15, 16). The activation of mycobacterial P_{1B4} -ATPases by Fe^{2+/3+} was tested using purified proteins stabilized in lipid/detergent micelles. All three proteins were strongly activated by Fe²⁺ and only minimally by Fe³⁺ (Fig. 3). For comparison, activation by Co^{2+} , Ni^{2+} , and Zn^{2+} at 0.1 and 1 mM concentrations is shown. MtCtpD Fe²⁺-dependent activity was ~2-fold higher than those observed in MtCtpJ and MsCtpJ (Table 1 and Fig. 3D) and quite similar to that of B. subtilis PfeT (3.25 \pm 0.21 μ mol/mg/h) (13). MtCtpD also showed significant activation at 1 mM Zn^{2+} (Fig. 3B). Zn^{2+} binding to P_{1B4} -ATPases as well as Zn^{2+} transport has been reported (15, 30). The $K_{1/2}$ for Fe²⁺ activation of the mycobacterial enzymes confirmed a tendency observed in *B. subtilis* PfeT: the larger activation by Fe^{2+} is associated with a $K_{1/2}$ much larger than that of Co^{2+} (Table 1). However, the observed $K_{1/2}$ values do not describe the selectivity to the enzymes. These parameters result from the $k_{\rm on}/k_{\rm off}$ of the metals binding the cytoplasmic facing transmembrane sites and the k_{on}/k_{off} for the release/backward binding of the metal to the periplasmic facing sites (41). As shown below, equilibrium binding determinations of K_D better report the relative selectivity for the activating metals.

The described Fe²⁺-ATPase activities require the binding of the transported substrate to the TM-MBS. The stoichiometry of this interaction was verified by measuring Fe²⁺ binding to MsCtpJ, MtCtpD, and MtCtpJ in non-turnover conditions lacking ATP (Table 1). The His₆-less enzymes were incubated with excess Fe²⁺, unbound metal was removed by filtration, and bound metal was quantified by AAS. As expected, the proteins bind Fe²⁺ in a 1:1 molar ratio. Discarding the possibility of nonspecific interactions, metal binding was largely abolished in the presence of 1.5 mM vanadate (not shown). This binding stoichiometry is similar to that previously observed for Zn²⁺, Ni²⁺, and Co²⁺ binding to P_{1B4}-ATPases (15, 17). Notably, although the TM-MBSs of these enzymes appear to accommodate divalent cations, no significant binding of Fe³⁺ was observed (not shown).

Mycobacterial P_{1B4} -ATPase affinities for Fe^{2+} and Co^{2+} under equilibrium conditions were determined by titration of isolated enzymes in the presence of the fluorescence indicator mag-fura-2 (15, 42). In these experiments, mag-fura-2 forms 1:1 indicator-metal complexes of known K_D . The concentration of free indicator can be spectrophotometrically monitored, and the free metal and metal-protein complex levels can be calculated. The enzyme-metal K_D and the apparent stoichiometry of the interactions were obtained by fitting mag-fura- $2A_{366}$ versus free metal concentration curves (Table 1). MsCtpJ and MtCtpJ showed a similar K_D for Fe²⁺. These were also comparable with those previously reported for Co²⁺ (included in Table 1 for comparison). Notably, MtCtpD has \sim 3-fold higher affinity for Fe^{2+} compared with Co^{2+} . Moreover, the affinity of MtCtpD is 20 times higher for Fe²⁺ (lower K_D) and 5 times higher for Co²⁺ when compared with those observed in the CtpJ enzymes. The relative preference of MtCtpD for Fe²⁺ when compared with MtCtpJ further supports a dominant role of MtCtpD in Fe²⁺ tolerance (Fig. 1).

Distinct Co^{2+} and Fe^{2+} Coordination by MtCtpD—Full appreciation of the different enzymatic activities and metal selectivity observed in P_{1B4}-ATPases requires understanding of the structural basis of these phenomena. P_{1B4}-ATPases share a number of invariant residues in the transmembrane region proposed to participate in metal coordination (19, 24–26). A sixcoordinate Co^{2+} species by the *Sulfitobacter* sp. P_{1B4}-ATPase has been postulated with participation of a Ser in the conserved SPC in the fourth TM and invariant His, Glu, and Thr in the



Fe²⁺ Transport ATPases



FIGURE 3. Activation of mycobacterial P_{1B4} -ATPases by Fe^{2+} . ATPase activity of purified MsCtpJ (A), MtCtpD (B), and MtCtpJ (C) in the presence of a 0.1 (green bars) or 1.0 mm (blue bars) concentration of the indicated metal ions was determined. D, Fe^{2+} dependence of MsCtpJ (\blacksquare ; blue), MtCtpD (\blacksquare ; black), and MtCtpJ (\blacklozenge ; green) ATPase activities. Data are the mean \pm S.E. (error bars) of three independent experiments performed in duplicate.

TABLE 1

Summary	of kinetic	parameters,	metal binding	g stoichiometry	, and affinity	of m	vcobacterial P	ATPases
/					,			104

	M. smegmatis CtpJ		M. tuberculosis CtpD		M. tuberculosis CtpJ	
	Fe ²⁺	Co ²⁺	Fe ²⁺	Co ²⁺	Fe ²⁺	Co ²⁺
ATPase activity ^{<i>a</i>}						
$V_{\rm max}$ (μ mol/mg/h)	2.73 ± 0.12	0.62 ± 0.03^{b}	4.25 ± 0.19	0.35 ± 0.03^{b}	2.16 ± 0.07	1.38 ± 0.12^{b}
$K_{1/2}(\mu M)$	350 ± 45	4.1 ± 0.9^b	443 ± 54	4.8 ± 1.4^b	143 ± 18	6.3 ± 2.2^{b}
Metal stoichiometry ^c	1.0 ± 0.1	0.9 ± 0.1^b	1.1 ± 0.1	1.1 ± 0.1^b	1.1 ± 0.1	1.1 ± 0.3^b
Metal binding affinity ^d						
$K_{D}(\mu M)$	2.24 ± 0.43	2.59 ± 0.32^{b}	0.12 ± 0.04	0.33 ± 0.04	2.41 ± 0.27	1.60 ± 0.16
N	1.44 ± 0.17	1.39 ± 0.22^{b}	1.32 ± 0.11	1.26 ± 0.04	1.35 ± 0.06	1.38 ± 0.05

^b Values were previously reported (15, 16).

^c Values are the mean \pm S.E. (n = 3).

^{*d*} Values obtained by competitive metal binding with mag-fura-2. Values are the mean \pm S.D. (n = 3).

sixth TM (17). Surprisingly, this coordination did not include the invariant Cys located in the fourth transmembrane segment of all P_{1B} -ATPases.

Considering the results shown above and the possible distinct Fe^{2+}/Co^{2+} coordination, the binding environment of Fe^{2+} and Co^{2+} in MtCtpD was analyzed by XAS. The x-ray absorption near edge spectroscopy (XANES) portion of the XAS spectrum is element-specific and local bonding-sensitive; therefore it is useful for reporting the oxidation and coordination states of metals bound to the enzyme. The spectra of Fe^{2+} loaded protein were compared with Fe^{2+} and Fe^{3+} model systems (Fig. 4A). The first inflection point energy for proteinbound iron occurs at 7127.6 eV, consistent with a 50%/50% Fe^{2+}/Fe^{3+} redox state mixture. Although all spectra were closely screened for photoreduction, iron oxidation during protein concentration postmetal loading may have led to this observation. Pre-edge features observed in the XANES of iron-MtCtpD are characteristic of $1s \rightarrow 3d$ electronic transitions. These pre-edge features are consistent with pseudosymmetric six-coordinate iron-ligand systems (43). Cobalt-MtCtpD XANES pre-edge features and the general edge structure are consistent with Co²⁺ bound to protein systems in a six-coordinate ligand environment as observed previously (Fig. 4*B*) (17, 44). The EXAFS region of an XAS spectrum provides high res-





FIGURE 4. Normalized iron and cobalt XANES spectra of MtCtpD. *A*, XANES plot for iron-MtCtpD (*green*), FeSO₄ (*red*), and Fe₂(SO₄)₃ (*blue*). Peak maximum, 7113.5 eV; total peak area, 7.91. *B*, XANES plot for cobalt-MtCtpD (*blue*). Peak maximum, 7709.6 eV; total peak area, 1.92. *Insets* show expansion of spline-subtracted pre-edge feature for the $1s \rightarrow 3d$ transition for each element.

olution distances for ligand coordination environments of metals bound to metalloproteins. Fourier transform of the EXAFS provides a pseudoradial distribution function of ligand environments surrounding the metal. EXAFS analysis was used to compare differences in coordination of iron and cobalt to MtCtpD (Fig. 5). The resulting coordination number, ligand identity, bond lengths, and statistical fitting parameters are described in Table 2. The iron-MtCtpD spectrum was best simulated by a ligand environment containing five oxygen/nitrogen ligands at two sets of coordinating distances. Long range iron-ligand scattering was best fit with four Fe–C ligands. The general features in the iron EXAFS and bond lengths obtained from the simulations suggest that iron is most likely coordinated by six oxygen/nitrogen-based side chain ligands from amino acids and water molecules. The cobalt-MtCtpD EXAFS served as a comparison for the iron bound to MtCtpD. The best fit simulation contained two unique Co–O/N environments (Table 2). The Co-O/N bond lengths, compared with crystallographically characterized model compounds in the Cambridge Structural Database, are again consistent with five- to six-coordinate Co– $(O/N)_6$ compounds (45). This is similar to what has been observed in the Sulfitobacter sp. P_{1B4}-ATPase (17).

The spectroscopy analysis points to a distinct coordination for Co^{2+} and Fe^{2+} . However, the spectroscopy is not able to reveal alternative ligands. Conserved residues of MtCtpD possibly involved in metal coordination are Ser-316 and Cys-318 in TM4 and His-642, Glu-643, Gly-644, Ser-645, and Thr-646 in TM6. Seeking a more detailed understanding of the Fe^{2+} and Co²⁺ coordination at the TM-MBS, residues likely involved in the metal coordination were exchanged by site-directed mutagenesis, and the resulting proteins were functionally characterized. The single mutants E643D and T646S (included as conservative control modifications) and G644A and S645A did not alter the Co²⁺- or Fe²⁺-ATPase activities of MtCtpD (Fig. 6). In agreement with previous reports, mutation of S316A, H642A, E643A, or T646A led to significant loss of Co²⁺ ATPase activity (17). The single substitutions S316A, E643A, and T646A equally affected the Co²⁺- and Fe²⁺-ATPase activities. However, the H642A mutation largely abolished Co^{2+} stimulation (6% of WT activity) while preserving significant Fe^{2+} sensitivity (33%) at saturating metal concentrations. In contrast, the C318A mutation had diminished Fe^{2+} activation (18%) while retaining 39% of the Co²⁺-ATPase activity. The differential effects of H642A and C318A mutations on the ATPase activity point toward a plausible mechanism. It appears that CtpD differentiates Co^{2+} and Fe^{2+} as substrates perhaps via alternative coordination despite binding these ions with quite similar affinities.

Considering that the ATPase activities might be affected by the removal of a ligating group or by the inability to undergo structural changes required for transport, the capability of the C318A and H642A mutant proteins to bind Co²⁺ and Fe²⁺ was tested. In comparison, the C318A mutant showed significantly lowered MtCtpD affinity for Fe²⁺, but it did not change the binding affinity for Co²⁺, suggesting an important role of this conserved Cys in Fe²⁺ binding (Fig. 7 and Table 3). The critical role of Cys-318 was further confirmed by the determination of the metal binding by AAS after incubation of the C318A mutant protein with metals at concentrations 10 times over the observed K_D . The C318A mutant protein was able to bind $1.05\pm0.14\,Co^{2+}$ but only 0.32 \pm 0.06 $Fe^{2+}.$ Finally, the H642A mutant had no detectable effect on Fe²⁺ or Co²⁺ binding affinities compared with WT (Table 3), raising the possibility that His-642 has a role other than the direct participation in the TM-MBS.

Discussion

The substrates and consequent functional roles of bacterial and eukaryotic P_{1B4} -ATPases have remained elusive as their capabilities to transport different transition metals have been reported (15–17, 28–30). This is relevant as some of these transporters are required for bacterial virulence (16, 18) and critical for metal homeostasis in chloroplasts (29, 46). An extra layer of complexity is present in bacterial systems containing homologous non-redundant P_{1B4} -ATPases (16). From a different perspective, defining P_{1B4} -ATPases substrates is significant to understanding the coordination of transition metals by





FIGURE 5. Iron and cobalt EXAFS and Fourier transform with best fit simulations. Iron-MtCtpD raw k^3 -weighted EXAFS (A) and phase-shifted Fourier transform (B) are shown. Cobalt-MtCtpD raw k^3 -weighted EXAFS (C) and phase-shifted Fourier transform (D) are shown. Raw unfiltered data are shown in *black*, and best fit simulations are shown in *green*.

TABLE 2

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Summary of the iron and cobalt EXAFS simulations for metals bound to MtCtpD
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Data utilized a *k* range of 1–13.5 Å⁻¹.

	Nearest neighbor ligand environment ^a					Long range ligand environment ^a				
Metal	Atom ^b	R^{c}	$C.N.^d$	σ^{2e}	Atom ^b	R^{c}	$C.N.^d$	σ^{2e}	F'^{f}	
		Å				Å				
Iron	Oxygen/nitrogen Oxygen/nitrogen	1.93 2.07	2.5 2.5	1.93 2.41	Carbon	3.09	4.0	2.46	0.32	
Cobalt	Oxygen/nitrogen Oxygen/nitrogen	2.09 1.97	4.0 1.0	3.73 3.91	Carbon	2.91	4.0	4.43	0.33	

^a Independent metal-ligand scattering environment.

^b Scattering atoms.

^c Average metal-ligand bond length.

^d Average metal-ligand coordination number.

 e Average Debye-Waller factor (Å $imes 10^3$).

^fNumber of degrees of freedom weighted mean square deviation between data and fit.

transport proteins as well as likely additional selectivity mechanisms acting *in vivo*. Here we describe the roles of mycobacterial P_{14B}-ATPases in Fe²⁺ homeostasis, the kinetics of transport, and the structural elements that in part determine the selectivity of these enzymes. These results indicate that mycobacterial P_{14B}-ATPases are Fe²⁺/Co²⁺-ATPases; however, the various isoforms show differential participation in the homeostasis of these ions. *Mycobacterial* P_{1B4} -ATPases Participates in Both Fe^{2+} and Co^{2+} Homeostasis—We observed that mycobacterial CtpJ proteins contribute to the homeostasis of Fe^{2+} and Co^{2+} to different extents. *M. smegmatis* has a single P_{1B4} -ATPase, MsCtpJ. Expression of the coding gene is induced by Co^{2+} and partially by the superoxide generator paraquat but not by H_2O_2 (15). Deletion of *MsctpJ* leads to lower tolerance to Co^{2+} , Fe^{2+} , and hemin as well as increments in intracellular Co^{2+} and Fe^{2+}



FIGURE 6. ATPase activity of MtCtpD proteins carrying substitutions of residues likely participating in the metal binding site in the presence of 0.1 mm Co²⁺ (green bars) or 1 mm Fe²⁺ (blue bars). Replacement SPC16CPS indicates the double mutant S316C/C318S. Activities were normalized to those of the wild type MtCtpD. Data are the mean \pm S.E. (error bars) of three independent experiments performed in duplicate. Significant differences from the WT as determined by Student's t test are indicated (*, p < 0.01).

levels (Fig. 1) (15). The *M. tuberculosis* genome encodes two P_{1B4} -ATPases. MtCtpJ expression, like MsCtpJ, is induced by Co^{2+} and to a lesser extent by redox stressors and Fe^{2+} (16). The *Mt* Δ *ctpJ* strain accumulates higher levels of both Co²⁺ and Fe^{2+} (Fig. 1) (16), again in a fashion similar to that of the $Ms\Delta ctpJ$ strain. These characteristics appear similar to those observed for PfteT, the single P_{1B4}-ATPase present in *B. subtilis* (13). The comparable functions suggested by the observed phenotypes correlate with the analogous biochemistry of MsCtpJ, MtCtpJ, and B. subtilis PfteT. These three ATPases transport Fe^{2+} and Co^{2+} with surprisingly similar V_{max} and $K_{1/2}$ for activation. Moreover, both CtpJs bind Fe²⁺ and Co²⁺ with micromolar affinities (equilibrium binding determinations have not been performed for PfeT). These affinities explain the observed capability of these enzymes to influence the cellular response to STN when extracellular iron is maintained at just 10 μ M. More importantly, 2–3 μ M Fe²⁺ affinities appear consistent with reported free Fe $^{2+}$ levels in the 1–10 μ M region (4). In fact, the iron-sensing transcriptional regulators Fur and IdeR have 9 μ M K_D for Fe²⁺ (47, 48), indicating that these regulators are sensitive to the same concentration of Fe^{2+} as the P_{1B4}-ATPases. Consequently, efflux CtpJ ATPases and influx transporter regulators are likely to coordinately respond to changes in metal levels not only under Fe²⁺ stress conditions but also under normal conditions.

Homologous P_{1B4} -ATPases Present in Mycobacterial Genomes Have Distinct Roles—M. tuberculosis, as in other mycobacteria, has an additional P_{1B4} -ATPase, CtpD. Notably, MtCtpD, but

Fe²⁺ Transport ATPases

not MtCtpJ, is required for bacterial virulence. What unique function does CtpD provide? The $Mt\Delta ctpD$ strain is more sensitive to iron stress and accumulates higher levels of this metal than $Mt\Delta ctpJ$. The phenotypic differences between the $Mt\Delta ctpD$ and $Mt\Delta ctpJ$ strains should have a molecular basis either in the biochemistry of these enzymes or the iron pool that they transport. Notably, the phenotype of the $Mt\Delta ctpD:\Delta ctpJ$ double mutant strain is similar to that of the $Mt\Delta ctpD$ cells, suggesting that CtpD and CtpJ use the same iron pool as a substrate, and this can be controlled by MtCtpD alone. Alternatively, the molecular activities of MtCtpD and MtCtpJ appear distinct. MtCtpD has significantly higher Fe²⁺-ATPase activity. Moreover, if the relative activation induced by Fe^{2+}/Co^{2+} is considered, MtCtpD shows Fe²⁺ activity 12 times larger than that generated by Co²⁺. On the contrary, MtCtpJ shows higher activation by Co^{2+} than MtCtpD and only a 1.5 Fe²⁺-ATPase: Co²⁺-ATPase ratio. Although these relative activities approximately correlate with the observed phenotypes, the higher affinity of MtCtpD for Fe²⁺ (K_D of 0.1 μ M) appears to confer its dominant role in Fe²⁺ homeostasis. This K_D is 1 order of magnitude smaller than that reported for Fe²⁺sensing transcriptional regulators of influx systems (47, 48). Distinct from CtpJ, CtpD is not induced by divalent metals but by redox stressors, such as the nitric oxide generator nitroprusside and the respiratory poison cyanide (16). In fact, the region upstream of *ctpD* contains the TTG XXXXTTCXXG operator sequence for the redox-sensing MtFurA regulator (49). Considering the release of Fe²⁺ from iron-sulfur and mononuclear iron-containing proteins upon redox stress, it can be hypothesized that CtpD constitutes an early response to Fe²⁺ dyshomeostasis that is independent of efflux (CtpJ), storage (bacterioferritin), and regulators (IdeR) that respond to higher free Fe^{2+} levels.

The Coordination of Fe^{2+} by P_{1B4} -ATPases Likely Requires the Invariant Cys in the Fourth TM-Metal selectivity is central to the physiological roles of P1B-ATPases. In early studies, invariant Cys in the sixth TM (fourth TM in P_{1B4}-ATPases) were instrumental in defining P1B-ATPases. Detection of other conserve residues in the transmembrane region led to the identification of P_{1B} -ATPases subgroups (19). The participation of these signature residues in the binding sites of P_{1B1} Cu⁺-ATPases and P_{1B2} Zn²⁺-ATPases was later established (25, 26). Then it was relevant to establish the metal coordination in P_{1B4} -ATPases. Previous studies proposed that P1B4-ATPases coordinate Co²⁺ with a Ser in the conserved SPC in the fourth TM and invariant His, Glu, and Thr in the sixth TM of these proteins (17). Surprisingly, no participation of the archetypical Cys in the fourth TM in Co²⁺ coordination was observed. However, a different coordination of Fe²⁺ by MtCtpD might explain its distinct biochemistry, *i.e.* higher affinity for Fe^{2+} and Co^{2+} and higher activity in the presence of Fe²⁺. We studied the coordination of Fe²⁺ and Co²⁺ while bound to MtCtpD TM-MBS by XAS and functionally analyzed variants carrying mutations in putative coordinating groups. XAS data indicate that both Fe²⁺ and Co²⁺ are coordinated by five to six oxygen/nitrogen ligands in a manner similar to that described previously for the Sulfitobacter sp. P_{1B4}-ATPase. That is, the spectroscopy does not





FIGURE 7. Fe²⁺ and Co²⁺ binding affinities to MtCtpD WT, C318A, and H642A protein variants. Dissociation constants of MtCtpD WT (\oplus ; black), C318A (\blacksquare ; blue), and H642A (\oplus ; green) for Fe²⁺ (A) and Co²⁺ (B) were determined. The data were fit to $\nu = n[Me^{2+}_{free}]/K_D(1 + ([Me^{2+}_{free}]/K_D))$ using values shown in Table 3. Data are the mean \pm S.D. (error bars) of three independent experiments.

TABLE 3 Comparison of Fe²⁺ and Co²⁺ dissociation constants of MtCtpD and C318A and H642A variants

	Fe	2+	Co ²⁺		
MtCtpD	$K_D^{\ a}$	n ^a	K _D	п	
	μм		μM		
WT	0.12 ± 0.04	1.32 ± 0.11	0.33 ± 0.04	1.26 ± 0.04	
C318A	4.01 ± 0.89	0.74 ± 0.07	0.28 ± 0.08	1.20 ± 0.05	
H642A	0.28 ± 0.08	1.14 ± 0.10	0.34 ± 0.05	1.33 ± 0.14	

^{*a*} Values obtained by fitting curves resulting from competitive metal binding with mag-fura-2 (Fig. 7). Values are the mean \pm S.D. (n = 3).

show the participation of sulfur atoms from the invariant Cys in the fourth TM as a metal ligand.

Notably, mutagenesis studies showed an alternative portrait of MtCtpD TM-MBS. As shown in the case the Sulfitobacter sp. P_{1B4} -ATPase, we observed that mutation of S316A, H642A, E643A, and T646A led to an almost complete inhibition of Co²⁺ activation, whereas replacement C318A retains significant (39%) Co2+-ATPase activity. A different pattern is observed, however, for the effects of these mutations on the Fe²⁺-ATPase. In this case, H642A retains some activity, whereas C318A causes a larger decrease in the activation by Fe²⁺. Although these differences are not dramatic, they suggest a putative differential involvement of these residues. Determination of the equilibrium binding affinities provided a more detailed view. Surprisingly, mutation H642A did not affect the metal binding to MtCtpD, suggesting that the reduced V_{max} of this mutant is associated with alterations in rate-limiting conformational steps rather than ion coordination. Keep in mind that metal release is the rate-limiting step in P-type ATPases (21, 50). More remarkably, replacement of C318A leads to a large reduction in the affinity for Fe^{2+} without affecting Co^{2+} binding. This datum in itself does not show a role of the conserved Cys in coordinating metals but suggests a direct effect, perhaps steric or through the second coordination sphere, in determining the affinity for Fe^{2+} . In this case, the conservation of this Cys in the CPS signature sequence appears to be a logical consequence of the need to maintain a high binding affinity for Fe^{2+} .

In summary, our observations suggest that mycobacterial P_{1B4} -ATPases play a central role in Fe²⁺ homeostasis. CtpD in

particular, likely regulated by FurA, constitutes part of the cellular response to redox-induced damage of iron centers.

Author Contributions—S. J. P. conducted most of the experiments, analyzed the results, and wrote the initial draft of the manuscript. B. E. L. performed X-ray spectroscopy analysis. J. E. L. and S. N. conducted growth and metal tolerance experiments using *M. tuberculosis* sis strains. C. M. S. oversaw *in vivo* experiments with *M. tuberculosis* and participated in manuscript revision. T. L. S. supervised x-ray spectroscopic analysis and participated in manuscript revision. J. M. A. conceived the idea for the project, directed the project, and wrote the paper with S. J. P.

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