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Research Article

Phytochemical Profiling of Leaf, Stem, and Tuber Parts of Solena amplexicaulis (Lam.) Gandhi Using GC-MS

Karthika Krishnamoorthy and Paulsamy Subramaniam

PG and Research Department of Botany, Kongunadu Arts and Science College, Coimbatore, Tamil Nadu 641029, India

Correspondence should be addressed to Paulsamy Subramaniam; paulsami@yahoo.com

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Objective. To explore the possible bioactive compounds in the methanolic extracts of leaf, stem, and tuber parts of the medicinal climber, *Solena amplexicaulis*, using GC-MS. *Methods*. GC-MS analysis of the plant extracts were performed by using GC-MS-5975C [Agilent] and mass spectra of the compounds found in the extract was matched with the data in the library of National Institute of Standards and Technology (NIST). *Results*. Thirty-five compounds were determined to be present in the parts studied. The active principles with their retention time, molecular formula, molecular weight, peak area, structure, category of the compounds, and activities were predicted. The most prevailing compounds were phytol (38.24%) in leaf, 4-(4-ethoxyphenyl) but-3-en-2-one (56.90%) in stem, and 9,17-octadecadienal, (Z)- (21.77%) in tuber. *Conclusion*. This study revealed that the species *S. amplexicaulis* is a good source of many bioactive compounds like terpenes, triazines, esters, alkanes, alcohols, hydrocarbons, aldehydes, amides, and so forth. That justifies the traditional usage of this species.

1. Introduction

Herbal plants are valuable gift of nature for mankind and they are the source of a variety of phytochemicals which are utilized for human and animal diets also. It is capable of synthesizing an overwhelming variety of low molecular weight organic compounds called secondary metabolites, usually with unique and complex structures. The medicinal actions of plants unique to particular plant species or groups are consistent with the concept that the combination of secondary products in a particular plant is taxonomically distinct [1]. It states that around 85-90% of the world's population consumes traditional herbal medicines [2]. In recent decades, studies on phytochemical constituents of medicinal plant and its pharmacological activities have received wide attention [3-6]. WHO has emphasized the need to ensure the quality of medicinal plant products using modern techniques with the application of suitable standards. Many modern methods are adapted for identification and quantification of active principle compounds in plant materials. Of them, gas chromatography-mass spectrometry (GC-MS) has become firmly established as a key technological platform for secondary metabolite profiling in both plant and nonplant species [7, 8].

The plant species Solena amplexicaulis is commonly called creeping cucumber and belongs to the family Cucurbitaceae distributed very seldom in the dry deciduous forest and scrub jungles of Tamil Nadu [9]. The medicinal uses of this species are multifaceted. The local healers of Tamil Nadu and Andhra Pradesh are prescribing this species for many ailments owing to its effective healing property [10]. The traditional healers are prescribing the tubers, leaves, and seeds of this species for various ailments like spermatorrhoea, thermogenics, diuretics, haemorrhoids, and invigorating and it is a very good appetizer and cardiotonic [11]. The whole plant is a potential source of natural antioxidant [12, 13], antidiabetic [10], and antibacterial agent [14] also. As the leaves have good anti-inflammatory activity, it is recommended for inflammation, skin lesions, and skin diseases [15]. Crude leaf juice is used to cure jaundice [16]. Unripe fruits are eaten raw to strengthen the body [17]. The decoction of the root is taken orally to cure stomachache [18]. As the reproductive parts like seeds and tubers are exploited severely for medicinal uses, this species becomes rare sighted in its habitats of Tamil Nadu.

Despite these wide medicinal uses, no information on qualitative account of phytochemicals is available for this species. To address this lacuna, GC-MS studies were undertaken to explore the phytochemical constituents present in the leaf, stem, and tuber parts of *S. amplexicaulis*.

2. Materials and Methods

2.1. Collection, Identification and Preparation of Plant Materials. The leaf, stem, and tuber parts of *S. amplexicaulis* were collected separately from the thorny scrub jungles of Madukkarai, Coimbatore District, Tamil Nadu, India. The authenticity of the plant was confirmed in Botanical Survey of India, Southern Regional Centre, Coimbatore, by referring to the deposited specimen (Voucher specimen number: CPS 313). They were washed thoroughly in tap water, shade-dried, and then homogenized to fine powder and stored in air tight bottles.

2.2. Preparation of Extract. 50 g of powdered leaf, stem, and tuber parts of *S. amplexicaulis* was separately extracted with 250 mL methanol at the temperature between 60 and 65°C for 24 h by using soxhlet extractor. The solvent was evaporated by rotary vacuum evaporator to obtain viscous semisolid masses. This semidry methanolic crude extract was subjected to GC-MS analysis.

2.3. GC-MS Analysis. GC-MS analysis was carried out on a 5975C Agilent equipped with a DB-5ms Agilent fused silica capillary column (30 \times 0.25 mm ID; film thickness: 0.25 μ m), operating in electron impact mode at 70 eV. Pure helium (99.999%) was used as carrier gas at a constant flow of 1 mL/min and an injection volume of 1 μ L was employed (split ratio is 10:1). Mass transfer line and injector temperature were set at 230 and 250°C, respectively. The oven temperature was programmed from 70 (isothermal for 3 min) to 300°C (isothermal for 9 min) at the rate of 10°C/min. Total GC running time was 34 min and the MS detection was completed within 35 min.

By GC-MS, the compounds were separated and then they were eluted from the column and made enter into the detector which was capable of creating an electronic signal. Then they were processed by the computer for generating chromatogram. Then the compound entered into the electron ionization (mass spectroscopy) detector, where they were bombarded with a stream of electrons causing them to break apart into fragments. These fragments were actually charged ions with certain mass. The m/z (mass/charge) ratio obtained was calibrated from the graph, called the mass spectrum, and is the fingerprint of the molecule.

2.4. Identification of the Compounds. To identify the compounds, the extract was assigned for comparison of their retention indices and mass spectra fragmentation patterns with those stored on the computer library and also with the published literature. National Institute of Standards and

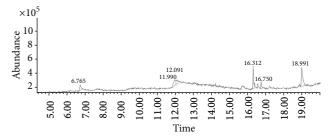


FIGURE 1: GC-MS chromatogram of methanolic leaf extract of *Solena amplexicaulis*.

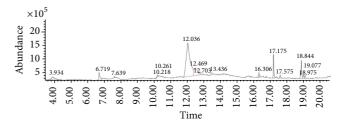


FIGURE 2: GC-MS chromatogram of methanolic stem extract of Solena amplexicaulis.

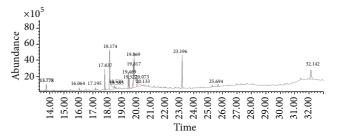


FIGURE 3: GC-MS chromatogram of methanolic tuber extract of Solena amplexicaulis.

Technology library sources were also used for matching the identified compounds from the plant materials [19, 20].

3. Results

The gas chromatograms of leaf, stem, and tuber parts of *S. amplexicaulis* confirmed the presence of various interesting compounds with different retention times as illustrated in Figures 1, 2, and 3. These compounds were identified through mass spectrometry attached with GC. The identified compounds and their retention time, molecular formula, molecular weight, peak area (%), structure, category of the compound, and activities related with medicinal uses are given in Tables 1, 2, and 3 for leaf, stem, and tuber, respectively. The compound prediction is based on Dr. Duke's Phytochemical and Ethnobotanical Databases. Six compounds were detected in the methanolic leaf extract of *S. amplexicaulis*. Among them, the most prevailing major compounds were phytol, a diterpene (peak area: 38.24%) (Figure 4(a)), carane, a terpene (peak area: 18.76%)

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TABLE 1: Compounds identified in the methanolic leaf extract of Solena amplexicaulis by GC-MS.

	•		TABLE I: CC	ompounds id	entihed in the	TABLE I: Compounds identified in the methanolic leaf extract of Solena amplexicaulis by GC-MS.	by GC-MS.	
S. number	Name of the compound	RT	Molecular formula	Molecular weight	Peak area %	Struct	Category of the compound	Activity*
1	Hexahydropyridine, 1-methyl-4-[4,5- dihydroxyphenyl]-	6.761	$\mathrm{C}_{12}\mathrm{H}_{17}\mathrm{NO}_2$	207.12	10.75	CH ₃ N HO OH	Aromatic piperidine	No activity reported
2	1-Octanamine	11.990	$\mathrm{C_8H_{19}N}$	129.24	16.16	H ₃ C NH ₂	Aliphatic amine	No activity reported
3	1-Tetradecanamine	12.091	$C_{14}H_{31}N$	213.40	10.24	NH ₂	Aliphatic amine	No activity reported
4	Carane	16.317	$ m C_{10}H_{18}$	138.24	18.76	H ₃ C CH ₃	Terpene	Antifeedant, antioxidant
rV	Pentane-2,4-dione, 3-(1-adamantyl)	16.753	$C_{lS}H_{22}O_2$	234.33	5.85	CH ₃ CH ₃	Aliphatic diketone	No activity reported
9	Phytol	18.990	$\mathrm{C}_{20}\mathrm{H}_{40}\mathrm{O}$	296.53	38.24	H ₃ C OH	Diterpene	Anticancer, antioxidant, anti-inflammatory, diuretic, antitumor, chemopreventive, antimicrobial, use in vaccine

^{*}Source: Dr. Duke's Phytochemical and Ethnobotanical Databases (online database).

TABLE 2: Compounds identified in the methanolic stem extract of Solena amplexicaulis by GC-MS.

Activity*	No activity reported	Antimicrobial agents, transducer for immunosensor and its method of production. carcinogens, enzyme inhibitors, solvents	No activity reported	No activity reported	Antimicrobial, anti-lipopolysaccharide, anti-tumor properties, anti-infective agents, antineoplastic agents
Category of the compound	Cyclic diketone	Alkane	Triazine	Aromatic nitro compound	Taurine amino acid derivative
Structure	0=	H ₃ C CH ₃	HN N NH N N NH N N N NH H N N N		
Peak area %	4.47	3.92	0.36	0.52	0.17
Molecular weight	98.09	156.30	172.14	259.21	284.35
Molecular M formula	$C_5H_6O_2$	$C_{11}H_{24}$	$\mathrm{C_4H_8N_6O_2}$	$C_{13}H_9NO_5$	10.261 C ₇ H ₁₆ N ₄ O ₄ S ₂
RT	3.929	6.718	7.633	10.218	10.261
S. number Name of the compound	1,3-Cyclopentanedione	Undecane	1,2,4-Triazino [5,6-E] [1,2,4]-triazine-3,6-dione, hexahydro-	4-Hydroxyphenyl 3-nitrobenzoate	Taurolidine
S. nuı	-	7	60	4	rυ

Continued.	
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TABLE	

	e Activity*	No activity reported	Treat amyloidosis (prevent the deposition of amyloid protein in the body)	No activity reported	No activity reported	r No activity reported
	Category of the compound	Aliphatic acid	Sucrose	Aldo heptose	Aromatic benzaldehyde	Aliphatic ester
IABLE 2: Continued.	Structure	CH_3	CH ₂ OH CH ₂ OH OH OH OH	О НО НО ОНО	H ₃ C OH	H ₃ C CH ₃
IABI	Peak area %	56.90	11.49	1.68	1.71	1.40
	Molecular weight	190.24	342.29	210.18	180.20	254.40
	Molecular formula	$G_{12}H_{14}O_{2}$	$C_{12}H_{22}O_{11}$	$C_7H_{14}O_7$	$C_{10}H_{12}O_3$	C ₁₆ H ₃₀ O ₂
	RT	12.033	12.469	12.701	13.442	16.303
	S. number Name of the compound	4-(4-Ethoxyphenyl) but-3-en-2-one	Trehalose	d-Glycero-d-tallo-heptose 12.701	Benzal dehyde, 6-hydroxy- 4-methoxy-2,3-dimethyl-	9-Tetradecen-1-ol, acetate, 16.303 (Z)-
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Hexadecanoic acid, 17174 C ₀ H ₂₄ O ₂ 270.45 6.52 H ₃ C O CH ₃ Linoleic acid mentaloguic mentaloguic mentaloguic mentaloguic antimotroguic antimotroguic antimotroguic mentaloguic antimotroguic antimotroguic methyl ester 17581 C ₀ H ₂₅ O ₂ 296.48 6.76 H ₃ C O C ₉ H ₃₅ O ₂ 207.18 1.42 O H ₃ C O H ₃	numbe	S. number Name of the compound	RT	Molecular formula	Molecular weight	TAB Peak area %	TABLE 2: Continued. rea Structure	Category of the compound	Activity* Anti-inflammatory,
C ₁₉ H ₃₆ O ₂ 296.48 6.76 CH ₃ Bicyclic alkane C ₁₉ H ₃₆ O ₂ 296.48 6.76 C ₁₉ H ₃₆ O ₂ 207.18 1.42 O H N H C ₁₉ H ₃₈ O ₂ 298.50 1.29 H ₃ C C ₁₉ H ₃₈ O ₂ 298.50 1.29 H ₃ C C ₁₉ H ₃₈ O ₂ 298.50 1.29 H ₃ C C ₁₉ H ₃₈ O ₂ 298.50 Fatty ester		Hexadecanoic acid, methyl ester	17.174	$C_D H_{34} O_2$	270.45	6.52		Linoleic acid ester	preventive, hepatoprotective, nematicide, insectifuge, antihistaminic, antieczemic, antiacne, alpha reductase inhibitor, antiandrogenic, antiarthritic, anticoronary
$C_{19}H_{36}O_{2} \qquad 296.48 \qquad 6.76 \qquad \qquad \\ H_{3}C \qquad \qquad \\ \\ C_{19}H_{38}O_{2} \qquad 298.50 \qquad 1.29 \qquad \\ H_{3}C \qquad \qquad \\ \\ C_{19}H_{3}C \qquad \qquad \\ \\ C_{19}H_{3}C \qquad \qquad \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ $		I-Methyl-3- ethyladamantane	17.581	$C_{13}H_{22}$	178.31	1.37		Bicyclic alkane	No activity reported
18.975 $C_8H_9N_5O_2$ 207.18 1.42 $O_1H_1 H_1 H_2 H_3 H_3 H_3 H_3 H_3 H_3 H_3 H_3 H_3 H_3$		9-Octadecenoic acid (Z)-, methyl ester	18.844	$C_{l9}H_{36}O_{2}$	296.48	92.9		Linoleic acid ester	Anti-inflammatory, antiandrogenic cancer preventive, dermatitigenic hypocholesterolemic, 5-alpha reductase inhibitor, anemiagenic, insectifuge
19.077 $C_{19}H_{38}O_{2}$ 298.50 1.29 $H_{3}C$ Eatty ester CH_{3}		Benzaldehyde, 2-nitro-, diaminomet hylidenhydrazone	18.975	C ₈ H ₉ N ₅ O ₂	207.18	1.42	H Z-Z	Nitrogen compound	Antimicrobial
	i l	Heptadecanoic acid, 10-methyl-, methyl ester	19.077		298.50	1.29	CH ₃	Fatty ester	No activity reported

*Source: Dr. Duke's Phytochemical and Ethnobotanical Databases (online database).

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		TABI	E 3: Compoun	ıds identifieα	l in the me	Table 3: Compounds identified in the methanolic tuber extract of Solena amplexicaulis by GC-MS.	GC-MS.	
S. number	Name of the compound	RT	Molecular formula	Molecular weight	Peak area %	Structure	Category of the compound	Activity*
1	Dodecanoic acid	13.776	$C_{12}H_{24}O_{2}$	200.31	2.40	HO CH ₃	Fatty acids	No activity reported
7	Tetradecanoic acid	16.071	$\mathrm{C_{14}H_{28}O_2}$	228.37	0.95	HO	Myristic acid	Antioxidant, cancer preventive, nematicide, hypocholesterolemic, lubricant
ro	1,2-Benzenedicarboxylic acid, bis(2-methylpropyl) ester	17.189	$\mathrm{C_{l6}H_{22}O_4}$	278.34	0.74	$H_3C \xrightarrow{CH_3} CH_3$	Phthalic ester	Used in preparation of perfumes and cosmetics, plasticized vinyl seats on furniture, cars, and clothing including jackets, raincoats, and boots and used in textiles, as dyestuffs, cosmetics, and glass making
4	Pentadecanoic acid, 14-methyl-, methyl ester	17.842	$C_{17}H_{34}O_2$	270.45	4.61	H ₃ C CH ₃	Fatty ester	No activity reported
rV.	n-Hexadecanoic acid	18.176	$C_{16}H_{32}O_2$	256.42	21.75	H ₃ C OH	Palmitic acid	Antioxidant, hypocholesterolemic, nematicide, pesticide, lubricant, hemolytic inhibitor, antiandrogenic
9	Cystodytin	18.510	$18.510 ext{ C}_{22} ext{H}_{19} ext{O}_3 ext{N}_3$	373.78	1.58	H ₃ C CH ₃ CH ₃ O O	Aromatic alkaloid	Antiproliferative activity in human tumor cell lines
7	1-Decanol, 2-hexyl-	18.583	$\mathrm{C_{16}H_{34}O}$	242.44	1.21	H ₃ C CH ₃	Aliphatic alcohols	Antimicrobial
∞	10,13-Octadecadienoic acid, methyl ester	19.469	$C_{19}H_{34}O_2$	294.47	4.72	H ₃ C OCH ₃	Linoleic acid esters	Anti-inflammatory, hypocholesterolemic, cancer preventive, hepatoprotective, nematicide, insectifuge, antieczemic, anticancer, antiarthritic, insectifuge, antihistaminic, anticonnary

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			Molecular	Molecular	Deak		Category of	
S. number	Name of the compound	RT	formula	weight	area %	Structure	the compound	Activity*
	trans-13-Octadecenoic acid, methyl ester	19.527	$C_{l9}H_{36}O_{2}$	296.48	3.55	H ₃ C CH ₃	Linoleic acid esters	Anti-inflammatory, antiandrogenic, cancer preventive, dermatitigenic, irritant, antileukotriene—D4, hypocholesterolemic, 5-alpha reductase inhibitor, anemiagenic, insectifuge, flavor
	9,12-Octadecadienoic acid (Z,Z)-	19.817	$C_{l8}H_{32}O_2$	280.44	9.35	CH ₃ OH	Linolenic acid	Anti-inflammatory, hypocholesterolemic, cancer preventive, insectifuge, antiarthritic, hepatoprotective, antiandrogenic, nematicide, antihistaminic, antieczemic
	9,17-Octadecadienal, (Z)-	19.876	$C_{\mathrm{lg}}H_{32}O$	264.44	21.77	CH ₃	Unsaturated aldehyde	Antimicrobial
	Phthalic acid, di(2-propylpentyl) ester	23.201	$C_{24}H_{38}O_4$	390.55	9.48	$\begin{array}{c} CH_3 \\ CH_3 \\ CH_3 \end{array}$	Dicarboxylic acid ester	Oral toxicity during pregnancy and sucking in the Long-Evans Rat
	Anthracene, 9-ethyl-9,10- dihydro-10-t-butyl-	25.699	$\mathrm{C}_{20}\mathrm{H}_{24}$	264.40	1.26	H_3C CH_3 H_3C	Hydrocarbons	No activity reported

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Activity*	No activity reported
Category of the compound	Tyramine derivative
Structure	
Peak area %	6.72
Molecular weight	298.29
Molecular Molecular Peak formula weight area %	32.148 C ₁₆ H ₁₄ N ₂ O ₄
RT	32.148
S. number Name of the compound RT	4-Dehydroxy-N-(4,5- methylenedioxy-2- nitrobenzylidene) tyramine
S. number	14

*Source: Dr. Duke's Phytochemical and Ethnobotanical Databases (online database).

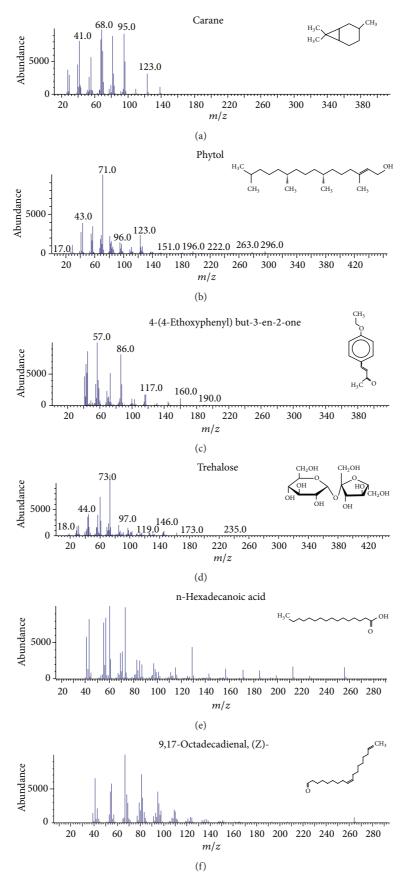


FIGURE 4: (a) Mass spectrum of carane. (b) Mass spectrum of phytol. (c) Mass spectrum of 4-(4-ethoxyphenyl) but-3-en-2-one. (d) Mass spectrum of trehalose. (e) Mass spectrum of n-hexadecanoic acid. (f) Mass spectrum of 9,17-octadecadienal, (Z)-.

(Figure 4(b)), and 1-octanamine, an aliphatic amine (peak area: 16.16%). The methanolic stem extract of S. amplexicaulis showed the presence of fifteen different organic compounds. The major phytochemical compounds among them were 4-(4-ethoxyphenyl) but-3-en-2-one, an aliphatic acid (peak area: 56.90%) (Figure 4(c)), trehalose, sucrose (peak area: 11.49%) (Figure 4(d)), hexadecanoic acid, methyl ester, a linoleic acid ester (peak area: 6.52%), and 9octadecenoic acid (Z)-, methyl ester, another linoleic acid ester (peak area: 6.76%). Fourteen compounds were identified in the methanolic tuber extract. In this account, 9,17-octadecadienal (Z)-, an unsaturated aldehyde (peak area: 21.77%) (Figure 4(e)), n-hexadecanoic acid, a palmitic acid (peak area: 21.75%) (Figure 4(f)), phthalic acid, di(2propylpentyl) ester, a dicarboxylic acid ester (peak area: 9.48%), and 9,12-octadecadienoic acid (Z,Z)-, a linolenic acid (peak area: 9.35%) were the major phytochemicals on the basis of quantity.

4. Discussion

The gas chromatogram shows that the relative concentrations of various compounds are getting eluted as a function of retention time. The height of the peaks indicates the relative concentrations of the compounds present in the plant. The mass spectrometer analyzes of the compounds eluted at different times to identify the nature and structure of the compounds. The large compound fragments into small compounds give rise to appearance of peaks at different m/z ratios. These mass spectra are fingerprint of that compound which can be identified from the data library.

Generally, the reliability of medicinal plant for its usage is evaluated by correlating the phytochemical compounds with their biological activities [21]. In the present study, the GC-MS analysis of the methanolic extracts of leaf, stem, and tuber parts of S. amplexicaulis altogether showed the presence of 35 compounds. In this account, the leaf extract contained six compounds among them, phytol (38.24%) is having anticancer, antioxidant, anti-inflammatory, antitumor, antimicrobial, diuretic, and chemopreventive properties and used in vaccine formulations [22, 23]. The other compound, carane (18.76%) is having antifeedant and antioxidant properties [24, 25]. The methanolic stem and tuber extracts showed the presence of greater number of 14 and 15 compounds, respectively. The six phytoconstituents, namely, undecane, taurolidine, trehalose, hexadecanoic acid methyl ester, 9octadecenoic acid (Z)-, methyl ester, and benzaldehyde, 2nitro-, diaminomet hylidenhydrazone in stem extracts have possessed medicinal properties [26]. Undecane, an alkane, is an antimicrobial agent, used as carcinogen [27, 28]. Similarly, the other compound, taurolidine, a taurine amino acid derivative, has antimicrobial, antilipopolysaccharidal, and antitumor properties [29, 30]. The sucrose compound, trehalose, is used for the treatment of amyloidosis [31]. The linoleic acid esters present in the stem, hexadecanoic acid methyl ester, are reported to have anti-inflammatory, cancer preventive, hepatoprotective, antiarthritic, and anticoronary properties. The other linoleic acid ester, 9-octadecenoic

acid (Z)-, methyl ester, is also having anti-inflammatory, antiandrogenic, and anemiagenic properties [32]. The nitrogen compound, benzaldehyde, 2-nitro-, diaminomet hylidenhydrazone, is known to have the property of curing infectious diseases by its antimicrobial activity. In the tuber extracts, the compounds identified, namely, 10,13-octadecadienoic acid methyl ester, trans-13-octadecenoic acid, methyl ester, and 9,12-octadecadienoic acid (Z,Z)-, are possessed with anti-inflammatory and cancer preventive characters. The two compounds, namely, tetradecanoic acid and n-hexadecanoic acid, are antioxidants. The phthalic acid, 1,2-benzenedicarboxylic acid, bis(2-methylpropyl) ester, is used in the preparation of perfumes and cosmetics. The unsaturated alcoholic compound, 9,17-octadecadienal, (Z)-, is reported to have antimicrobial property [33]. The study species S. amplexicaulis is endowed with various medicinal properties maybe due to the presence of all these compounds described. In a similar fashion, certain traditional medicinal plant species of Cucurbitaceae have been analyzed phytochemically by using GC-MS and suggested for drug preparation after succeeding in clinical trials [34, 35]. The therapeutic properties of the other compounds in all the three parts of *S. amplexicaulis* were not yet reported.

Our investigation through the present study revealed that the species *S. amplexicaulis* is a reliable source of bioactive compounds like fatty acid esters, alcohols, hydrocarbons, alkanes, amines, terpenes, and sugars that justify the traditional usage of this species [16–18] by the local healers in Coimbatore and Tirupur districts of Tamil Nadu, India, for various ailments. As GC-MS is the first step towards understanding the nature of active principles [36, 37], further investigation in this species is suggested for the development of novel drugs.

Conflict of Interests

The authors declare that they have no conflict of interests regarding the publication of this paper.

Acknowledgment

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