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Data Article

Nano-cellulose derived bioplastic biomaterial data for vehicle bio-bumper from banana peel waste biomass



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ARTICLE INFO

Article history: Received 26 September 2015 Received in revised form 27 April 2016 Accepted 16 May 2016 Available online 25 May 2016

Keywords: Nano-celluloses Biopolymer Banana peel waste Biobumper

ABSTRACT

The innovative study was carried out to produce nano-cellulose based bioplastic biomaterials for vehicle use coming after bioprocess technology. The data show that nano-cellulose particle size was 20 nm and negligible water absorption was 0.03% in the bioplastic. Moreover, burning test, size and shape characterizations, spray coating dye, energy test and firmness of bioplastic have been explored and compared with the standardization of synthetic vehicle plastic bumper following the American Society for Testing and Materials (ASTM). Tensile test was observed 120 MPa/kg m³. In addition to that pH and cellulose content were found positive in the bioplastic compared to the synthetic plastic. Chemical tests like K, CO₃, Cl₂, Na were determined and shown positive results compared to the synthetic plastic using the EN-14214 (European Norm) standardization.

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Specialization Table

Subject area Biological Chemistry More specific subject area Biological Chemistry

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http://dx.doi.org/10.1016/j.dib.2016.05.029

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Type of data	Table and figure
How data was	SEM, pH meter, Tensile test was performed by Universal Test Machine,
acquired	absorption test, burning test, crack test, energy test, chemical test by ASTM
	and EN standard
Data format	Raw data collection and analysis
Experimental	Samples pyrolysis, acid hydrolysis
factors	
Experimental	3 Replicates were used in the experiment as Complete Randomized Design
features	(CRD). The sample was selected randomly from the different lots
Data source	Hail city, Saudi Arabia
location	
Data accessibility	Data are presented in this article

Value of the data

- The data are an innovative and important in the research area of nano-cellulose based bio-plastic biomaterials utilizing banana peel waste biomass in the pharmaceuticals, medical, biomedical and bioengineering aspect.
- Innovative information regarding biopolymer production from biomass has been explored. The data can be most useful for the researcher, research student and academician to acquire innovative knowledge.
- Identification of the suitability of produced nano-cellulosed based bio-plastic materials plays a significant role for the researcher in further studies for automotive biobumper production.

1. Data

The data have been shown the sample collection, preparation and biochemical process in Figs. 1.1– 1.5. Fig. 1.1a shows the procedure of nano-bioplastic and biobumper preparation. Moreover, Fig. 1.2 demonstrates the absorption test of nano-cellulose based bioplastic. Fig. 1.3 indicates the burning test image. The Universal Test Machine procedure is denoted in Fig. 1.4. Besides, nanobioplastic having spray coating test has been exhibited (Fig. 1.5). From the data it had been seen that nano-cellulose particle size was 20 nm found in the bioplastic. It was considered for plastic bumper as mentioning the result of ASTM (0–0.16) (Table 1.1). In addition to that burning test (Table 1.2), spray coating dye (Table 1.3), size and shape characterizations (Table 1.4), energy test (Table 1.5), firmness test (Table 1.6) were found positive and standard in bioplastic compared to the synthetic plastic following ASTM standardization. Tensile test was observed 120 MPa/kg m³ (Table 1.7). In addition to that chemical analysis like K⁺, CO₃⁻, Cl₂, Na were determined (Table 1.8) and shown positive results compared to the synthetic plastic in the laboratory using the EN (166) standardization (Table 1.9).

2. Experimental design, materials and methods

2.1. Sample collection and preparation

A total of 10 kg waste banana samples (*Musa acuminata*) were collected from the local market, Hail city, KSA. Afterwards, banana peel was removed and washed to ensure proper cleaning. Washed peel was sliced with scissors, then the sliced peel was blended using conventional blender. After blending, the sample was ground again with mortar and pestle, to get a fine mixture and put it into the beaker.

2.2. pH determination

The pH was determined using Horiba Scientific pH meter.

2.3. Cellulose determination

2.3.1. Dinitrosalicylic acid (DNS) method for cellulose determination

Cellulose content was determined by utilizing 3,5-dinitrosalicylic acid (DNS). A standard curve was drawn by measuring the absorbance of known concentration of cellulose solution at 575 nm. DNS reagent consisted of 1% dinitrosalicylic acid, 0.2% phenol, 0.05% sodium sulfite and 1% sodium hydroxide. To measure cellulose content, 3 ml of unknown cellulose solution was poured into a test tube followed by addition of 3 ml of DNS reagent. The test tubes were then heated in boiling water bath for 15 min. 1 ml of 40% potassium sodium tartrate solution was then added prior to cooling the sample. All test tubes were then cooled under running tap water and its absorbance was measured at 575 nm.

2.4. Samples pyrolysis

Blended and ground sample was heated at 150 °C in pressure cooker for 2 h at 30 psi until the sample became liquid paste. After heating, the liquid paste sample was cooled down.

2.5. Acid hydrolysis

Paste sample was hydrolyzed (100 ml/50 g sample) by hydrochloric acid (HCl, 99%) to make it micro to nano size particle for 8 h. The water bath was used during the hydrolysis came after in bioprocess technique. After 8 h, the samples were separated using separation funnel and washed by distilled water.

2.6. Nanoparticle measurement

Nano particle size of the sample was measured by employing Scanning Electron Microscopy (SEM).

2.7. Plastilizer and mixture of chemicals

Chlorinated paraffin liquid plasticizers (1:8 of sample), 5% acetic acid (5 ml/100 g sample), 5 ml/ 100 g (polyvinyl chloride), cellulose (25%) and 25% starch powder, 5% film forming agent (toluene, Phthalates ester) and 10% water were added with the 1000 g of samples. Later 10 ml/100 g polyvinyl chloride (PVC) and glycerin were added with the mixture of samples and kept for 10 min to mix up well. The dye (benozophenone and stearalkonium) was mixed and kept for 5 min to mix up well and then the mixture was heated implementing pyrolysis method following ASTM standard [1] (at 150 °C in the oven for 30 min; at 30 psi pressure) until visual plasticity in the oven for bioplastic material. The samples were taken out from the oven and stored in room temperature (28 °C) to cool down for 10 min. Afterwards, it was put in the exposed aluminum foils for few hours to make it dry bioplastic. Then it was set for molding process. Finally it was oven dried at 80 °C. The bioplastic was used for different test for fitness.

2.8. Molding in the laboratory

The bio-plastic sample was molded using manually prepared mold former and burner for 24–30 h drying and shaping of vehicle bumper. Dryer was used every 5 h to form a curve shape of the biobumper. It was kept at -5 °C for one day and taken out from the deep freezer, dried by dryer and kept for hardening at normal temperature.



a. Waste banana peel

b. Blended peel paste



c. Heating of paste d. Liquid paste after heating e. Sample paste after cooling



f. nanoparticle sized after acid hydrolysis, g. pyrolysis and plastilizer mixture



h. Raw bioplastic before drying i. Bioplastic after oven drying j. Spray dye coating and drying





- k. Molding process 18 hrs and heating (3rd step)
- l. Final drying 24 hrs after molding

Fig. 1.1. (a) Procedure of nano-bioplastic and biobumper preparation. (b) Nanocellulose.



Fig. 1.2. Absorption test.



Fig. 1.3. Burning test image.



Fig. 1.4. (a) Universal Test Machine procedure, (b) Universal Test Machine, (c) plastic bumper, and (d) bioplastic.



Fig. 1.5. Nano-bioplastic with spray coating test.

Table 1.1

Determination of water absorption by ASTM D570.

Materials	Water absorption	ASTM D570
Bioplastic bumper Plastic bumper	0.03% 0-0.16%	Water absorption Water absorption by ASTM is 0-0.16%.

Table 1.2

Burning test according to the ASTM D3801.

Materials	Odor	Color of flame	Speed of burning	Spark or not
Bioplastic bumper	Low odor	Yellow-orange	Slow	Spark
Plastic bumper (by ASTM D3801)	Low odor	Yellow-orange	Slow	Spark

Table 1.3

Spray coating dye was used as the mode of application by ASTM B 117

Materials	Spray coating test	ASTM B117
Bioplastic bumper Plastic bumper By ASTM B 117	(Drying time) 1.5 h 2 h (max).	Maimum 2 hours

Table 1.4

Determination of size and shape characteristics

Materials	Size and shape	ASTM A500
Bioplastic bumper Plastic bumper	No swell or shrink	Resistant characters
By ASTM A 500	No swell or shrink	

2.9. Fitness testing of bioplastic

2.9.1. Absorption test as ASTM D570 [2]

For the water absorption test, the specimens were dried in an oven for a specified time and temperature and then placed into the desiccator to cool. Immediately upon cooling the specimens were weighed. The material was then merged in water upon conditions, often 23 °C for 24 h.

Table 1.5

Energy to	est of	bioplastic	that can	be	used	for	bumper
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Materials	Energy	ASTM E1886
Bioplastic bumper	E = 1/2x MxV2 2.5 Joule	
Plastic bumper By ASTM E 1886	1.75-25 Joule	

Table 1.6

Firmness test represented by bore and crack test.

	Bore and crack test	
Materials	Firmness test	Firmness test
	Bore test ASTM D2925	Crack test by ASTM D5419
Bioplastic bumper	No bore symptom	No crack symptom
Plastic bumper	No bore symptom	No Crack symptom

Table 1.7

Determination of tensile test by using ASTM by ASTM D5083

Materials	Tensile strength (MPa/kg.m3)	Tensile Modulus (GPa)
Bioplastic bumper	120.0	1.1
By ASTM D5083	70-230 (ASTM)	1.0- 3.0 (ASTM)

Table 1.8

pH and Cellulose determination.

		pH and Cellulose determ	nination
Test Banana samples Plastic	sample	pH 7.7±0.01 Alkaline ≥ 7	Cellulose 39.4 $\%$ ±0.1 (It is zero if from gas or oil sample, if from cellulose sample it is 20- 40%)

Mean \pm SE (standard Error, n=3)

Table 1.9

Chemical element test.

Chemical element	Banana sample based bio- plastic (PPM)	Plastic by EN (European Standard EN166)
K	10 ± 0.5	10
Na	5.1 ± 0.2	5
Cl ₂	0.67 ± 0.01	2
CO ₃	132 ± 2.0	5-440

Mean \pm standard error (SE, n=3).

Specimens were removed, patted dry with a lint free cloth, and weighed (Fig. 1.2). Diameter of disk was 5 cm and 2 mm thick. Water absorption was calculated (Table 1.1).

2.9.2. Burning test

Bioplastic was burnt by using gas burner. Odor, color of flame, speed of burning and spark were observed by visual observation and compared with the synthetic bumper by ASTM D3801 [3] (Fig. 1.3).

2.9.3. Tensile/tension test

Tensile test was performed by Universal Test Machine for bioplastics as ASTM D5083 [4].

2.9.3.1. Test procedure. Specimens were placed in the grips of a Universal Test Machine at a specified grip separation and pulled until failure. For ASTM D5083 [4], the test speed was determined by the material specification. The default test speed was 5 mm/min (0.2 in./min), but modulus determinations were made at 2 mm/min (0.079 in./min). An extensometer or strain gauge was used to determine elongation and tensile modulus. Depending upon the reinforcement and type, testing in more than one orientation might be necessary. Max load capacity was 50 kN/m².

2.9.3.2. Sample size. The standard specimen for ASTM D5083 had a constant rectangular cross section, 25 mm (1 in.) wide and at least 250 mm (10 in.) long. Thickness was between 2 mm (0.079 in.) and 14 mm (0.55 in.). Optional tabs could be bonded to the ends of the specimen to prevent gripping damage. Inter-tek PTL could machine the specimens from larger samples and bond tabs if required (Fig. 1.4). Tensile strength (MPa or psi) was displayed from tensile test.

2.9.4. Shape and size test

The specimen was continuously beaten by hammer for 2 min and pulled on for 5 min. There was no change of its shape and size (Fig. 1.5).

2.9.5. Energy test

Energy was tested for the bioplastic based bumper using the following equation: The energy= $1/2mv^2$; m=mass of hammer, V (velocity)=m/s, energy=? J (Table 1.5).

2.9.6. Firmness test (bore test)

Bioplastic was hit with a 2 kg hammer on the screw set on the bioplastic. Hammering was completed after 5 min.

2.9.7. Crack test

A 10 kg weight (5 J of energy) was dropped on to the bioplastic from 1 m height. There was no crack on it. (5 J of energy was calculated by the following equation: $energy=1/2mv^2$; m=mass of stone=10 kg, V (velocity)=m/s=1/1, m=distance meter, s=second. Therefore, $energy=1/2 \times 10 \times 1=5$ J.)

2.9.8. Spray coating test

Spray coating dye was used as the mode of application. It was attached properly with plastic and dried after 1 h.

2.9.9. Chemical test

Chemical test like K, CO_3 , Cl_2 . Na were determined by using different meters and exhibited positive results (Table 1.8) compared to the synthetic plastic in the laboratory using the EN (14214) [5] standardization.

2.9.10. Statistical analysis

Standard deviation was calculated and standard error was analyzed using 3 replicates of the samples where necessary (n=3).

Acknowledgment

Authors are thankful to the Department of Biology, Faculty of Science, University of Hail for financial support of this project (UOH/S-11). Also thankful to their M.S. and Ph.D. student, who assisted and analyzed the data.

Transparency document. Supplementary material

Transparency data associated with this article can be found in the online version at http://dx.doi. org/10.1016/j.dib.2016.05.029.

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