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Antiepileptic Ceramides from the Red Sea Sponge Negombata corticata

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Abstract

A new antiepileptic ceramide mixture **1** was isolated from the Red Sea sponge Negombata corticata. The structures of the metabolites were determined by extensive spectroscopic analysis. The anticonvulsant activity of **1** was measured in vivo using the pentylenetetrazole-induced seizure model. This finding has important implications for biological studies with this class of compounds.

> Sterols and fatty acid are potentially excellent biomarkers in marine samples due to their stability and diversity of structures. They are present in all eukaryotes and share with phospholipids a structural function in membranes due to their role in chemotaxonomic purposes and for food web tracing.¹ Sphingolipids have emerged as a new class of modulators of various cell functions. Ceramides, which are the central moiety in the biosynthesis of sphingolipids and glycosphingolipids, are involved in the regulation of different cellular events, including cell senescence, differentiation, and programmed cell death (apoptosis).² Ceramides also act as regulators of many biochemical and cellular responses to stress, such as exposure to heat, radiation, oxidative conditions, and chemotherapeutic agents.³ Ceramides as well as more complex sphingolipids are required for activation of membrane fusion of Semliki Forest virus (SFV) and other alphaviruses.⁴ An interest in ceramides as regulators of growth, differentiation, and cellular apoptosis has recently increased. Many marine invertebrates are rich sources of ceramides that differ in structure and biological properties from those of terrestrial organisms. Unusual ceramides have been isolated from sponges, 5^{-8} Coelenterata, 9^{-11} crabs, 1^2 sea stars, 1^3 and ascidia. 1^4 Marine ceramides manifest cytotoxic, anti-tumor, antimicrobial, 9 and antifungal activities.⁷ Others are sex heromones¹² and enzyme inhibitors.¹⁴

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In this work, we report on the isolation and identification of a new ceramide mixture **1** (**1^a** and $\mathbf{1}_b$) from the Red Sea sponge *Negombata corticata*. The genus *Negombata* is represented in the Red Sea by two species, namely, Negombata magnifica (Keller) (formerly Latrunculia magnifica) and Negombata corticata (Carter), family Podospongiidae.¹⁵ The genus Negombata was shown to be a source of biologically active macrolides¹⁶⁻¹⁹ and lipids.²⁰

The structure elucidation of **1** began with an analysis of HRMS data. The high-resolution ESI-TOF mass spectrum of 1 displayed a pseudomolecular ion peak at m/z 676.6230 [M + Na^+ , which when combined with the detailed analysis of the ¹³C spectrum and DEPT indicated a molecular formula of $C_{41}H_{83}O_4N$, representing one unit of unsaturation. The ¹H NMR spectrum in C₅D₅N showed resonances of an amide proton doublet at δ 8.44 (1H, d, J $= 8.4$ Hz) and protons of a long methylene chain at δ 1.28, indicating a sphingolipid skeleton. The characteristic resonances of 2-amino-1,3,4-triol of the hydrocarbon chain were observed at δ 5.10 (1H, m), 4.50 (2H, dd, J = 8.0, 4.8), 4.38 (1H, m), and 4.29 (1H, m) in the ¹H NMR spectrum and at δ 54.1 (CHN), 62.5 (CH₂O), 77.0 (CHOH), and 73.3 (CHOH) in the ${}^{13}C$ NMR spectrum. In addition, the ${}^{1}H$ NMR spectrum showed resonances corresponding to aliphatic hydrocarbons at $\delta 0.89$ (6H, d, $J = 7.2$, H-4' and H-5'), 0.88 (3H, t, $J = 6.8$, H-3'''), 1.28 (overlapped H, m), 1.84 (1H, sep, H-3'), 1.95 (2H, m, H-3''), and 2.47 (2H, t, $J = 7.6$, H-2"). The ¹³C NMR spectrum showed resonances due to one terminal methyl group at δ 14.6 and two branched methyl groups in aliphatic hydrocarbon chains at δ 23.1 and an amide carbonyl at δ 173.7. Analysis of the ¹H–¹H COSY, HMOC, and HMBC spectra led to the assignment of proton and carbon signals for **1**. The positions of the hydroxyl groups were confirmed by a ${}^{1}H-{}^{1}H$ COSY spectrum between H₂-1/H-2, H-2/H-3, H-3/H-4, and H-4/H₂-5 and also from HMBC of H₂-1/C-2 (²J_{CH}), H₂-1/C-3 (³J_{CH}), H-2/C-3 $(^{2}J_{\text{CH}})$, H-3/C-2 $(^{2}J_{\text{CH}})$, H-3/C-4 $(^{2}J_{\text{CH}})$, H-4/C-2 $(^{3}J_{\text{CH}})$, and H-4/C-6 $(^{3}J_{\text{CH}})$, leading to the assignment of C-1/C-2/C-3/C-4. GC-MS analysis of the fatty acid methyl ester of **1** was carried out after hydrolysis and yielded two peaks with molecular ions of m/z 270 and 298 on the chromatogram, corresponding to C_{16} and C_{18} fatty acid methyl esters with a ratio of 62.5:37.5, respectively. The characteristic ¹H NMR resonance in CDCl₃ of methyl esters at 0.87 (3H, t, $J = 6.8$) indicates the presence of only terminal fatty acids palmitic (C_{16:0}) and stearic ($C_{18:0}$) acids. The ¹H NMR spectrum in CDCl₃ of the liberated sphingosine bases after hydrolysis showed a characteristic resonance at 0.81 (6H, d, $J = 7.2$), indicating the presence of only isopropyl terminal sphingosine bases. LRESIMS analysis of the sphingosine bases showed molecular ions at m/z 388.3 [M + H]⁺ and m/z 416.4 [M + H]⁺, corresponding to C_{23} and C_{25} , respectively.

The configurations of the ceramide moieties were assigned by comparison of the physical data and ¹H NMR and ¹³C NMR in two different solvents (C_5D_5N and CDCl₃) with the analogues as reported in the literature, in which the optical rotation $[+17.4 \,(\text{c } 0.08, \text{MeOH})$ and +11.1 (c 1.00, MeOH)] and the chemical shifts of H-2 (δ 5.10), H-3 (δ 4.38), and H-4 (δ

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4.29) in C_5D_5N were in good agreement with those of known synthetic ceramide (2S,3S, $4R$)-2-[(2R)-2-hydroxytetracosanoylamino]-1,3,4-hexadecanetriol²¹ and natural ceramide gracilamide B.²² Also the chemical shifts of H-2 (δ 4.15), H-3 (δ 3.59), and H-4 (δ 3.62) in CDCl₃ were similar to those of $(2S, 3S,$ and $4R)$ ceramides from the sea sponge *Grayella* cyatophora²³ and Oceanapia sp.²⁴ This evidence indicates the absolute configurations of C-2, C-3, and C-4 to be $2S$, 3S, and 4R, respectively.

A detailed analysis of COSY and HMBC correlations was found to be in complete agreement with the proposed structure for **1** (Table 1).

Experimental Section

General Experimental Procedures

Optical rotations were measured at ambient temperature using a Rudolph Research Analytical Autopol IV automatic polarimeter. The IR spectrum was recorded on a Bruker Tensor 27 spectrophotometer. 1D and 2D NMR spectra were recorded in CDCl₃ and C₅D₅N on a Bruker Avance DPX-400 spectrometer and on a Varian AS 400 spectrometer. The lowresolution ESIMS were obtained using a Finnigan Mat LCQ. The high-resolution ESIMS was measured using a Bruker Daltonic (GmbH, Germany) micro-TOF series with electrospray ionization.

Biological Material, Collection, and Identification

Negombata corticata, Carter (coll. no. SAA-8) was collected by hand using scuba at depths of 15–20 m from Safaga in the Egyptian Red Sea. The sponge materials were frozen immediately and kept frozen at −20 °C until processed. A voucher specimen is deposited at the Zoological Museum of the University of Amsterdam, under registration No. ZMAPOR. 18569.

Extraction and Isolation

The sponge was freeze-dried (400 g dry weight), ground, and extracted with a mixture of MeOH/CH₂Cl₂ (1:1) (3 \times 2 L) at room temperature. The extract was evaporated under vacuum to afford 100 g of red oil. This extract was subjected to vacuum liquid chromatography on a flash silica gel column using a hexanes, EtOAc, and MeOH gradient.

Fractions eluted from 70% EtOAc in hexanes to EtOAc were concentrated to afford 7 g of reddish residue. Purification of this fraction was carried out by column chromatography using flash silica gel. The mobile phase was made from $CHCl₃$ with an increasing gradient of MeOH. Fractions eluted with 5% MeOH in CHCl3 were collected and evaporated to dryness under reduced pressure to afford 250 mg of a green residue. Further purification was carried out on column chromatography over a silica gel 60/230–400 mesh (flash) column $(50 \times 0.5 \text{ cm})$ using 2% MeOH in CHCl₃. Final purification was carried out on Sephadex LH-20 using MeOH/CHCl₃ (1:1) to afford 50 mg of 1.

Hydrolysis of Ceramide Mixture 1

The mixture (3 mg) was heated with 5 mL of 1 N HCL in 15 mL of MeOH for 4 h at 90 °C. The reaction mixture was extracted with hexanes, and the hexanes layer was concentrated under vacuum to give a mixture of fatty acid methyl esters. NaOH solution (10%) was added to the methanolic layer containing a mixture of sphingosine bases up to pH 12.0, the sphingosine bases were extracted with CHCl₃, and the extract was evaporated in a vacuum to dryness.

Compound 1: white solid; $R_f = 0.65$, 10% MeOH in CHCl₃; [α]²⁵_D +17.4 (*c* 0.08, MeOH) and $[\alpha]^{25}$ _D +11.1 (*c* 1.00, MeOH); IR (KBr) (thin film) v_{max} 3309, 2920, 2850, 2371, 1642, 1545, 1467, 1048.2 cm⁻¹; ¹H NMR (C₅D₅N, 400 MHz and CDCl₃, 400 MHz) and ¹³C NMR (C_5D_5N , 100 MHz and CDCl₃, 100 MHz), see Table 1; LRESIMS, found m/z 652.6 [M – H]⁻; HRTOFMS, found m/z 676.6230 [M + Na]⁺ (calcd for C₄₁H₈₃O₄NNa, 676.6220).

Anticonvulsant Bioassay. Animals

A total of 24 male albino rats (125–155 g) were housed separately, two to a cage, under standard laboratory conditions. They were kept at constant room temperature (27 \pm 2 °C) and relative humidity of 55–65% under a 12/12 h light/dark cycle at least 10 days prior to testing. Commercial food pellets and tap H₂O were freely available. The experiments were performed during the light portion of the cycle, between 8:00 and 12:00 a.m., to avoid circadian influences. Rats were divided into 4 groups, 6 rats each: control group, reference group, drug group (dose, 0.5 mg/kg, ip), and drug group (dose, 1 mg/kg, ip).

Drugs and Dosage

The effects of **1** on seizure susceptibility were evaluated using the pentylenetetrazole (PTZ) test.25 Thirty minutes after administration of various doses of **1** (0.5 mg/kg, ip and 1 mg/kg, ip), animals received an intraperitoneal (ip) injection of PTZ (45 mg/kg, ip) and were then observed for a 30 min period for clonic seizures. The effect of **1** was compared against a reference group, where the rats were pretreated with diazepam (DZP) (1 mg/kg, ip) (Valpam amp., Amoun Co., Egypt), 30 min before each PTZ injection, and a control group, where the rats were injected with subconvulsive doses of PTZ dissolved in saline (45 mg/kg, ip) $(Sigma)$.²⁵

Evaluation of Seizures

After each injection, the convulsive behavior was observed for 30 min. The resultant seizures were classified according to the Racine rating scale as shown in Table 2.²⁶ Values are expressed as mean \pm SEM. For statistical analysis, one-way analysis of variances was applied. For all comparisons, differences were considered significant at $p < 0.05$.

In PTZ-induced seizure, the administration of **1** at doses of 0.5 mg/kg, ip and 1 mg/kg, ip 30 min before the injection of PTZ prolonged the latency and reduced the duration of tonicclonic seizures and showed antiepileptic effect comparable to that of DZP in a PTZ acute model, Table 2. There was no significant difference between seizure parameters for rats in the reference group (DZP) and those treated with **1**. Furthermore, there was a significant

difference between seizure parameters of rats in the control group (PTZ) and those treated with **1**, Table 2. In addition **1** was shown to prevent death significantly.

Supplementary Material

Refer to Web version on PubMed Central for supplementary material.

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Table 1

¹³C (100 MHz), 1H (400 MHz), COSY, and HMBC NMR Spectroscopic Data of Ceramide Mixture **1**

 $a_{n_1} = 13, 11, n_2 = 9, 11.$

Table 2

Anticonvulsant Activity of Ceramide Mixture **1**

 a_0 , No seizure response; 1, Immobility, eye closure, ear twitching, facial clonus; 2, Head nodding associated with more severe facial clonus; 3, Clonus of one forelimb; 3.5, Bilateral forelimb clonus without rearing; 4, Bilateral forelimb clonus with rearing; 4.5, Falling on a side (without rearing), loss of righting reflex accompanied by generalized; clonic seizures; 5, Rearing and falling on back accompanied by generalized clonic seizures.