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## Cyclic Heptapeptides from the Jamaican Sponge *Stylissa caribica*

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### Abstract

Two new cyclic heptapeptides, stylisin 1 (**1**) and stylisin 2 (**2**), were isolated and characterized from the Jamaican sponge *Stylissa caribica*, in addition to the cyclic heptapeptide phakellistatin 13 (**3**) and the known bromopyrrole alkaloids sceptrin (**4**), stevensine (**5**), and oroidin (**6**). The new structures were assigned on the basis of 1D and 2D NMR spectroscopic data, as well as chemical methods for the elucidation of the absolute configuration of amino acids. The peptides and alkaloids have been evaluated for their antimicrobial, antimalarial, anticancer, anti-HIV-1, anti-Mtb, and antiinflammatory activities.

Peptides are well-established bioactive metabolites from marine invertebrates and include linear peptides, depsipeptides, and cyclic and bicyclic peptides. Species of the sponge genera *Phakellia* and *Axinella* have yielded several bioactive cyclic peptides reported to have cytotoxic activity against a number of cancer cell lines. It has been noted that most of these cyclic peptides are rich in proline amino acid moieties, which may be in either *cis* or *trans* conformation. There is evidence that some of the sponge cyclic peptides are able to bind undetectable quantities of a very potent antineoplastic substance that can only be revealed through biological assays.

In this report as part of an investigation of active sponge extracts against infectious diseases two new cyclic heptapeptides, stylisin 1 (**1**) and stylisin 2 (**2**), and the known peptide phakellistatin 13 (**3**) were isolated as well as the known alkaloids sceptrin (**4**), stevensine (**5**), and oroidin (**6**). Here we describe the isolation and structure elucidation of the two new cyclic peptides. Several reports describing the chemistry of different marine sponges belonging to *Stylissa* spp. have been published including aldisines from *Stylissa massa*, latonduines from *Stylissa carteri*, and massadine from *Stylissa aff. massa*. In a study regarding the chemistry of *Stylissa caribica* Lehnert & van Soest, 1998 (Halichondrida: Dictyonellidae), a series of bromopyrrole alkaloids have been identified. No cyclic peptides have been previously reported from *Stylissa* species, and these results have suggested a possible microbial or symbiotic origin for these metabolites.

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## Results and Discussion

The sponge extract was dissolved in ethanol and subjected to silica gel vacuum-liquid chromatography followed by column chromatography, Sephadex LH-20, RP HPLC, and PTLC chromatography to yield two new cyclic heptapeptides, stylisin 1 (**1**) and stylisin 2 (**2**), the known cyclic peptide phakellistatin 13 (**3**), and the known alkaloids sceptrin, stevensine, and oroidin (**4–6**).

Stylisin 1 (**1**) (Table 1) was obtained as a white solid with the molecular formula  $C_{45}H_{61}N_7O_8$  as determined by HRESIMS 850.4428  $[M + Na^+]$  (calcd 850.4479). The  $^{13}C$  NMR spectrum contained seven carbonyls, and the  $^1H$  NMR showed four secondary amide protons ( $\delta$  6.77–8.95) and 19 degrees of unsaturation. All the amino acids were elucidated on the basis of COSY, HMQC, HMBC, and NOESY experiments (Figures 2 and 3). Each amino acid was connected to the adjacent amino acid in a cyclic form based on the HMBC correlations between the  $\alpha H$  and NH with the adjacent CO. An HMBC correlation was assigned between the isoleucine NH and CO of Pro1 at  $\delta$  171.48 and generated a partial structure for a Pro1-Ile. The amide proton at  $\delta$  7.44 of Phe gave an HMBC correlation to CO of Ile at  $\delta$  170.5 while the CO of Phe at  $\delta$  172.1 showed an HMBC to the  $\alpha H$  of Pro3, thus generating a partial structure of four amino acids. The remaining amino acids were assigned through the following HMBC correlations of the residual COs and vicinal NH protons, as well as COSY correlations between NH and  $\alpha H$  signals. An HMBC correlation between the amide proton of Leu at  $\delta$  8.15 and CO of Pro3 at  $\delta$  172.7 provided the partial structure Leu-Pro3-Phe-Ile-Pro1. Pro2 and Tyr yielded the second partial structure through an HMBC correlation between the Tyr amide proton at  $\delta$  6.77 and CO of Pro2 at  $\delta$  170.2. A NOESY correlation between the  $\alpha H$  of Leu at  $\delta$  4.71 and  $\alpha H$  of Pro2 at  $\delta$  4.42 allows the connection of Pro2 to Leu. The last connection between Tyr and Pro1 revealed the cyclic structure with a NOESY correlation between the  $\alpha H$  of Tyr at  $\delta$  4.55 and  $\alpha H$  of Pro1 at  $\delta$  4.17. The *cis/trans* conformation of Pro1, 2, and 3 was determined to be *cis* on the basis of the  $\delta_{\beta\gamma}$  (differential value of  $^{13}C$  chemical shifts of  $C_\beta$  and  $C_\gamma$  in proline); the value of the three proline moieties, which was  $>9$  ppm, and the NOE correlations between  $\alpha H$  of Pro 1, 2, and 3 and the  $\alpha H$  of the adjacent amino acid. Marfey's analysis was used to determine the absolute configuration as L for all the constituent amino acids of stylisin 1.

Stylisin 2 (**2**) (Table 2) was obtained as a white solid with the molecular formula  $C_{44}H_{57}N_7O_8$ , determined by HRESIMS to be 834.4088  $[M + Na^+]$  (calcd 834.4166). The heptapeptide skeleton was suggested from the presence of seven carbonyl carbons in the  $^{13}C$  NMR spectrum at  $\delta$  168.5–172.0. The large number of methylene carbons (14  $CH_2$ ) and an unsaturation number of 20 suggested a greater number of proline moieties constituting the structural skeleton. Close inspection of the 1D and 2D NMR data [ $^1H$ ,  $^{13}C$ , COSY, HMQC, HMBC and ROESY (Figures 4 and 5)] resulted in three main partial structures, Phe-Pro1, Pro1-Ile, Pro4-Tyr, and a remaining Pro3 moiety. The two aromatic systems of Tyr and Phe were elucidated on the basis of both COSY and HMBC correlations. The Phe amide proton at  $\delta$  6.79 showed COSY correlations to the  $\alpha H$  of Phe at  $\delta$  4.34 and showed an HMBC correlation of the Pro2 carbonyl at  $\delta$  171.3, which resulted in the partial structure Phe-Pro2. The HMBC and COSY correlations of the Ile amide proton at  $\delta$  8.98 to the Pro1 carbonyl at  $\delta$  170.8 and to the  $\alpha H$  of Ile at  $\delta$  4.18, respectively, allowed the formation of the second

partial structure Ile-Pro1. The  $\delta\text{H}$  of Pro1 at  $\delta$  3.25 showed an HMBC correlation to the Phe carbonyl at  $\delta$  168.5 and allowed the connection between the first and second partial structure (Ile-Pro1-Phe-Pro2). An HMBC correlation between the Tyr amide proton at  $\delta$  7.62 and the carbonyl carbon of Pro4 at  $\delta$  170.7 connected Pro4 and Tyr to form the third partial structure. An HMBC correlation between the carbonyl carbon of the Ile moiety at  $\delta$  171.6 and the  $\alpha\text{H}$  of Pro4 at  $\delta$  3.57 allowed the formation of the larger partial structure Tyr-Pro4-Ile-Pro1-Phe-Pro2. Three COSY correlations between the protons at  $\delta$  4.22 and 1.90,  $\delta$  1.90 and 1.85, and  $\delta$  1.85 and 3.50 and an HMBC correlation between the carbonyl at  $\delta$  169.4 and the proton at  $\delta$  4.22 allowed the construction of the Pro3 moiety, which lacks any HMBC correlations to its carbonyl carbon,  $\alpha$  or  $\delta\text{H}$ . Seven amide carbonyls along with the presence of 20 degrees of unsaturation including eight for two aromatic systems, four for the proline moieties, and seven for the seven carbonyl carbons gave a total of 19 degrees of unsaturation, leaving one degree of unsaturation for the cyclic structure of stylisin 1. Two key ROESY correlations between  $\gamma\text{H}$  of Pro3 at  $\delta$  1.85 and the  $\beta\text{H}$  of Pro2 at  $\delta$  1.47 and between the  $\alpha\text{H}$  of Pro3 at  $\delta$  4.22 and the  $\beta\text{H}$  of Tyr at  $\delta$  2.70 allowed the connection of the Pro3 to Pro2 and Tyr to produce the complete structure of stylisin 2. The amino acid sequence of stylisin 2 was Tyr-Pro4-Ile-Pro1-Phe-Pro2-Pro3. The differential value of  $\delta_{\beta\gamma}$  was calculated for Pro1 and 4 (6 and 6.2 ppm, respectively), which indicated that Pro1 and Pro4 were in *trans* conformation. The conformation was *cis* for Pro2 and Pro3 ( $\delta_{\beta\gamma}$  are 9.1 and 8.2 ppm, respectively). The ROESY correlations between the  $\alpha\text{H}$  of Pro2 and 3 and the  $\alpha\text{H}$  of the adjacent amino acid further support the *cis* conformation of these two prolines. Using Marfey's analysis the absolute configuration of the amino acids of stylisin 2 was determined to be L.

The conformations of stylisin 1 (**1**) and stylisin 2 (**2**) were analyzed using molecular modeling combined with NOE NMR experiments (Figure 1a,b). The lowest energy conformers for stylisins 1 and 2 (163.4, 185.0 kcal/mol, respectively) were obtained by simulated annealing and were highly consistent with the experimentally determined NOESY and ROESY data (Figures 3, 5). The distances between the correlated protons, some of which are shown in Figure 1, are less than 3 Å, indicating that the conformation of stylisins 1 and 2 in acetone is highly consistent with the modeled conformers shown in Figure 1.

An investigation of active sponge extracts yielded two new cyclic heptapeptides, stylisins 1 and 2, and the known peptide phakellistatin 13 from the sponge *S. caribica*. Phakellistatin 13 was previously isolated from the sponge *Phakellia fusca* Thiele and was reported to have potent cytotoxic activity against the human hepatoma BEL-7404 cell line with an  $\text{ED}_{50} < 0.01 \mu\text{g}/\text{mL}$ . Our evaluation of phakellistatin 13 did not show any cytotoxic activity against the NCI tumor cell lines. These results again highlight the problem of lost activity associated with these metabolites, which may be due to changes in the conformation of the cyclic peptide or due to its binding with highly toxic natural products, which can be detected only in the biological assays. The isolated peptides have shown no cytotoxicity against a number of cancer cell lines. They have been tested for their antimicrobial (*C. albicans*, *C. neoformans*, *S. aureus*, MRS, *P. aeruginosa*, *M. intracellulare*, and *A. fumigatus*), antimalarial (*P. falciparum*, D6 and W2 clone), anti-Mtb, and anti-inflammatory activity in rat neonatal microglia and anti-HIV-1 in PBM cells and were inactive in all of the assays. The

three alkaloids have been evaluated for their antimalarial activity (*P. falciparum*, D6 and W2 clone), and only oroidin and stevensine showed marginal activity, with IC<sub>50</sub> values of 1200 and 1800 ng/mL against the D6 clone. Oroidin and stevensine showed 71% and 95% inhibition at 128 µg/mL, respectively, against *M. tuberculosis*, while scep trin showed no activity at a concentration of >128 µg/mL. The three alkaloids have been tested for their activity against several pathogenic fungi and bacteria, including *C. albicans*, *C. neoformans*, *S. aureus*, MRS, *P. aeruginosa*, *M. intracellulare*, and *A. fumigatus*. Only scep trin showed activity against *C. neoformans*, with an IC<sub>50</sub> of 3.5 µg/mL. The three alkaloids were not active against *Leishmania donovani*, nor were they found to be active when assayed against HIV-1 in PBM cells.

In summary the peptides and alkaloids have been evaluated for their antimicrobial, antimalarial, anticancer, anti-HIV-1, anti-Mtb, and antiinflammatory activities. None of the peptides have shown any activity in the stated biological assays. Scep trin was moderately active against *C. neoformans*, with an IC<sub>50</sub> of 3.5 µg/mL, while oroidin and stevensine were slightly active at 128 µg/mL in the assay against *M. tuberculosis*.

## Experimental Section

### General Experimental Procedures

Optical rotations were measured with a JASCO DIP-310 digital polarimeter. UV and IR spectra were obtained using a Perkin-Elmer Lambda 3B UV/vis and AATI Mattson, Genesis Series FTIR spectrophotometer, respectively. The <sup>1</sup>H and <sup>13</sup>C NMR spectra were recorded in acetone-*d*<sub>6</sub> using NMR spectrometers operating at 500 or 600 MHz for <sup>1</sup>H and 125 or 150 MHz for <sup>13</sup>C NMR. Chemical shifts, with  $\delta$  values expressed in parts per million (ppm), are referenced to the residual solvent signals with resonances at  $\delta_{\text{H}}/\delta_{\text{C}}$  2.09/29.91, 206.71 (acetone-*d*<sub>6</sub>). The HRMS spectra were measured using a Bioapex FTESI-MS with electrospray ionization. Chromatographic procedures were carried out using normal and reversed-phase (RP) Si gel chromatography, Sephadex LH-20, and C8 and NH<sub>2</sub> HPLC (Waters 510 model system, Luna 5 µm, C<sub>8</sub>(2) 100 Å, 60 × 21.20 mm and Phenomenex Luna 5 µm NH<sub>2</sub> 100 Å, 250 × 21.29 mm, 5 µm). TLC analysis was carried out on precoated Si gel G<sub>254</sub>.

### Taxonomy, Extraction, and Isolation

This common sponge was collected from a silted reef, between 22 and 30 m deep, at Columbus Park, Jamaica, on November 16, 2002. In life the sponge forms thin fleshy conulose corrugated lamellae emanating from a central thickened region, arising from a rounded stalk. The sponge is deep orange, the texture fleshy and smooth, the sponge flexible and elastic. The skeleton consists of a loose large-meshed reticulation of predominantly angular styles embedded in spongin or free in the choanosome, in a loosely compressed axial region. The external parts of the sponge are fleshy with relatively few spicules. The sponge is closest to *Stylissa caribica* Lehnert & van Soest 1998 (order Halichondrida, family Dictyonellidae) and can be easily mistaken for *Axinella corrugata* (George & Wilson, 1919) *sensu* Alvarez et al. (1998), formerly known as *Teichaxinella morchella* Wiedenmayer, 1977.

A voucher specimen has been deposited in the Natural History Museum, London (BMNH 2003.9.19.1).

Four kilograms of the fresh sponge was freeze-dried and blended with EtOH. The blended material was extracted with EtOH and EtOH/H<sub>2</sub>O and dried under vacuum. The dried residue was subjected to vacuum-liquid chromatography using Si gel and a gradient of hexanes, EtOAc, EtOH, and H<sub>2</sub>O to give 11 fractions. Fraction 7, eluted with EtOAc/EtOH (1:1), was dissolved in MeOH. The MeOH-soluble portion was chromatographed using a C<sub>18</sub> flash Si column and a Sephadex LH-20 column, followed by RP C<sub>8</sub> HPLC, and further purified on a HPLC amino column. Two new peptides, stylisin 1 and stylisin 2, were obtained in addition to phakellistatin 13, which was further purified using PTLC. The <sup>1</sup>H NMR and <sup>13</sup>C NMR of the insoluble MeOH portion of fraction 7 showed a mixture of small metabolites related to the bromopyrrole alkaloid class, which on further purification by gel chromatography and HPLC produced the three known alkaloids stevensine (**5**) (1.5 g), oroidin (**6**) (0.5 g), and sceptrin (**4**) (100 mg).

The absolute configuration of the stylisins was determined by Marfey's analysis method. One milligram of stylisin 1 (**1**) and 0.5 mg of stylisin 2 (**2**) were subjected to hydrolysis by 6 N HCl in a 1 mL ReactiVial for 24 h at 100 °C. The hydrolyzate was then dried under vacuum and derivatized by Marfey's reagent (40 μL of FDAA 10% in acetone). Eight microliters of 1 M NaHCO<sub>3</sub> was added and heated for 1 h at 40 °C. Then 4 μL of 2 N HCl was added, and the mixture was dried and dissolved in 50 μL of DMSO. The products were analyzed by a HPLC Waters Nova Pack 3.9 × 150 mm column at λ 340 nm, using a gradient of triethylamine phosphate buffer (pH 3.00 ± 0.02) and acetonitrile (90% to 60% TEAP in 1 h). The retention times in minutes of the L and D amino acids were as follows (Thr 22.5, 36.0; Pro 30.7, 36.6; Leu 49.2, 59.3; Ile 49.1, 59.1; Phe 51.1, 58.6; Tyr 63.7, 70.1; alloIle 51.6, 60.3). The retention times of stylisin 1 hydrolyzate were as follows (Leu 49.3, Pro 30.5, Phe 51.0, Ileu 49.1, Tyr 63.9), and those for stylisin 2 hydrolyzate were as follows (Ileu 49.0, Pro 30.6, Phe 51.0, Tyr 64.3).

Molecular modeling was performed using SYBYL 6.8 (Tripos Associates, St. Louis, MO). Initial conformations of the molecule were obtained by 10 rounds of dynamics simulations, in which the molecule was heated to 500 K within 500 fs and then allowed to cool to 200 K within 2000 fs. Twenty lowest energy conformations were selected and refined by molecular mechanics minimization using Powell's gradient algorithm method with the MMFF94 force field, a constant-dependent dielectric of 4.00, and partial atomic charges, until a root-mean-square deviation of 0.001 kcal/mol·Å was achieved. Finally, from these refined conformations, the conformer with the lowest energy was selected as the global minimum, representing the most likely molecular configuration.

**Stylisin 1 (1):** white solid (MeOH);  $[\alpha]_D^{25} -44.9$  (*c* 0.2, MeOH); UV (MeOH)  $\lambda_{\max}$  (log  $\epsilon$ ) 212 (3.12), 278 (1.84) nm; IR (KBr film)  $\nu_{\max}$  3279, 2960, 1635, 1516, 1447, 1240, 703 cm<sup>-1</sup>; NMR data, see Table 1; HRESIMS *m/z* 850.4428 [M + Na<sup>+</sup>], calcd for C<sub>45</sub>H<sub>61</sub>N<sub>7</sub>O<sub>8</sub>, 850.4479

**Stylisin 2 (2):** white solid (MeOH);  $[\alpha]_D^{25} -39.9$  ( $c$  0.2, MeOH); UV (MeOH)  $\lambda_{\max}$  (log  $\epsilon$ ) 214 (3.17), 278 (2.07) nm; IR (KBr film)  $\nu_{\max}$  3403, 2963, 1642, 1516, 1449, 1348, 703  $\text{cm}^{-1}$ ; NMR data, see Table 2; HRESMS 834.4088  $[M + \text{Na}^+]$ , calcd for  $\text{C}_{44}\text{H}_{57}\text{N}_7\text{O}_8$ , 834.4166.

**Phakellistatin 13 (3):** pale yellow solid (MeOH);  $[\alpha]_D^{25} -127$  ( $c$  0.09, MeOH), the reported  $[\alpha]_D^{25}$  was  $-136$  ( $c$  0.09, MeOH); UV (MeOH)  $\lambda_{\max}$  (log  $\epsilon$ ) 224 (3.22), 280 nm (2.63); IR (KBr film)  $\nu_{\max}$  3315, 2955, 1641, 1530, 1454, 1082, 745  $\text{cm}^{-1}$ ; NMR data; HRESIMS  $m/z$  821.4514  $[M + \text{Na}^+]$ , calcd for  $\text{C}_{42}\text{H}_{53}\text{N}_8\text{O}_8$ .

**Scepttrin (4):** brownish solid with the molecular formula  $\text{C}_{22}\text{H}_{24}\text{N}_{10}\text{O}_2\text{Br}_2$ , which was determined by HRESIMS 619.329, 621.425, and 623.526, indicating dibromo substitution. By comparison of ESIMS,  $^1\text{H}$  NMR, DEPT, and  $^{13}\text{C}$  NMR data with the literature<sup>25</sup> the compound was identified as scepttrin.

**Stevensine (5):** yellowish powder with the molecular formula  $\text{C}_{11}\text{H}_9\text{N}_5\text{OBr}_2$ , which was determined by HRESIMS 385.929, 387.925, and 389.926, indicating dibromo substitution. By comparison of ESIMS,  $^1\text{H}$  NMR, DEPT, and  $^{13}\text{C}$  NMR data with the literature the compound was identified as stevensine.

**Oroidin (6):** yellowish powder with the molecular formula  $\text{C}_{11}\text{H}_{11}\text{N}_5\text{OBr}_2$ , which was determined by HRESIMS 387.94, 389.931, and 391.931, indicating dibromo derivatives. By comparison of ESIMS,  $^1\text{H}$  NMR, DEPT, and  $^{13}\text{C}$  NMR data with the literature<sup>25</sup> the compound was identified as oroidin.

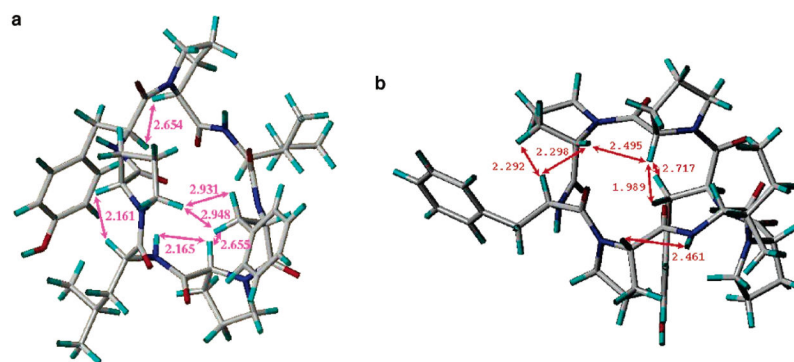
## Acknowledgments

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## References and Notes

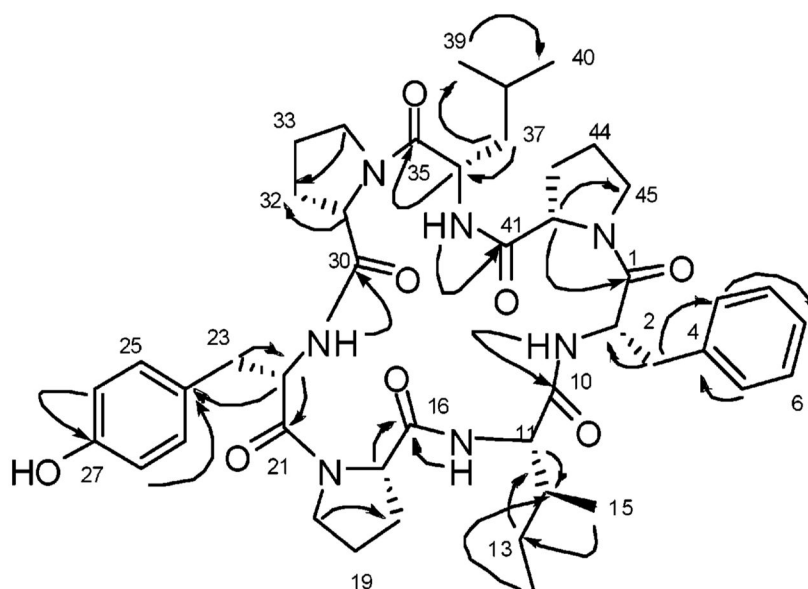
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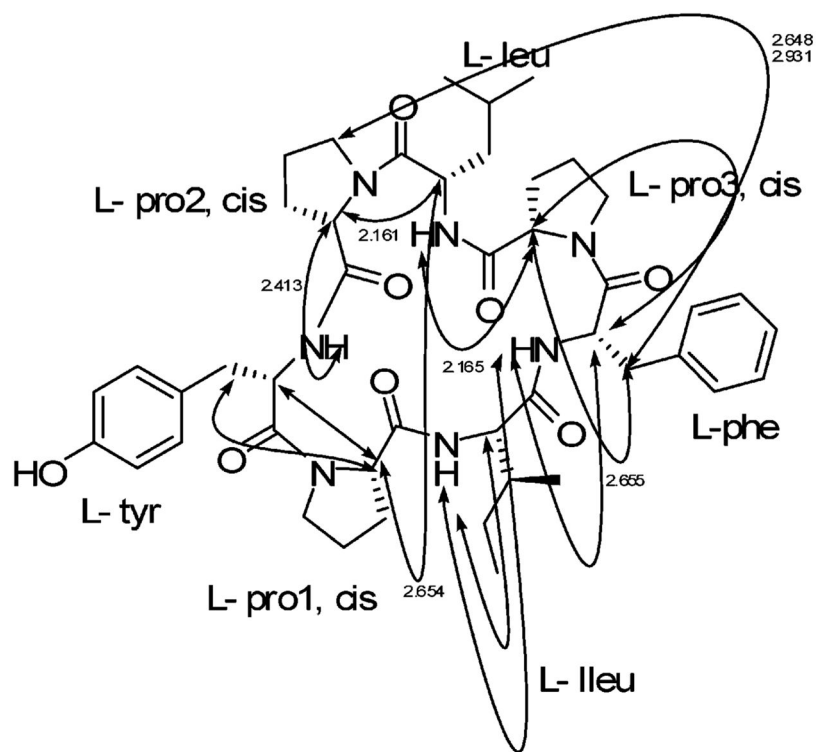


**Figure 1.** (a) Conformation of **1** with minimized energy. (b) Conformation of **2** with minimized energy. The arrows show the NOE correlations, and the data are the distances in Å between correlated protons.

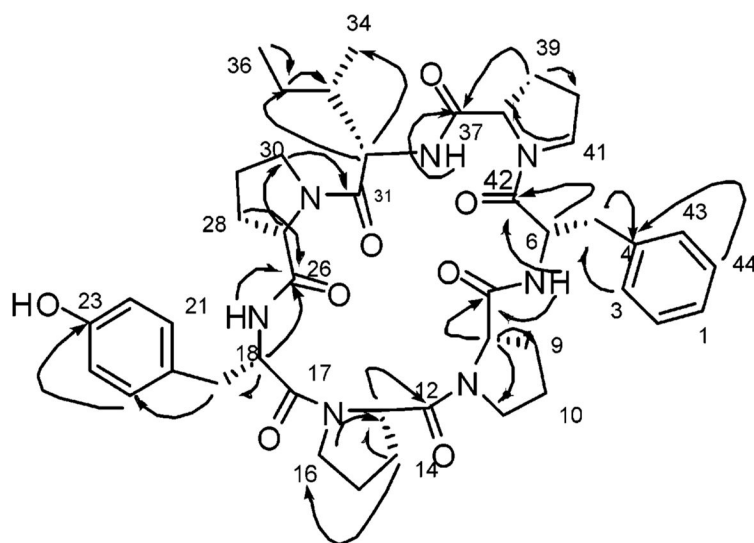




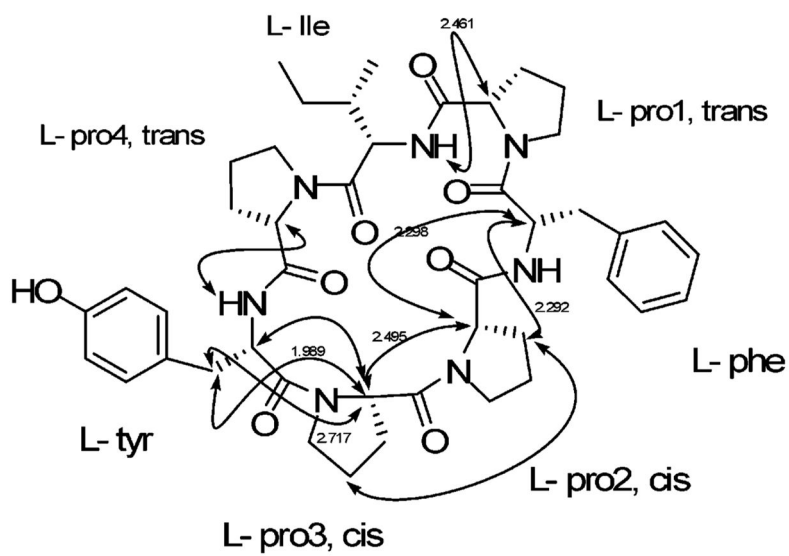
**Figure 2.**  
Key HMBC correlations of stylisin 1 (1).



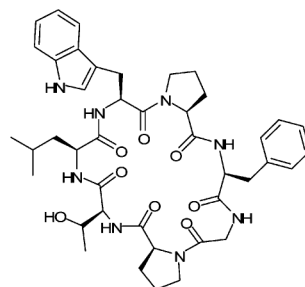
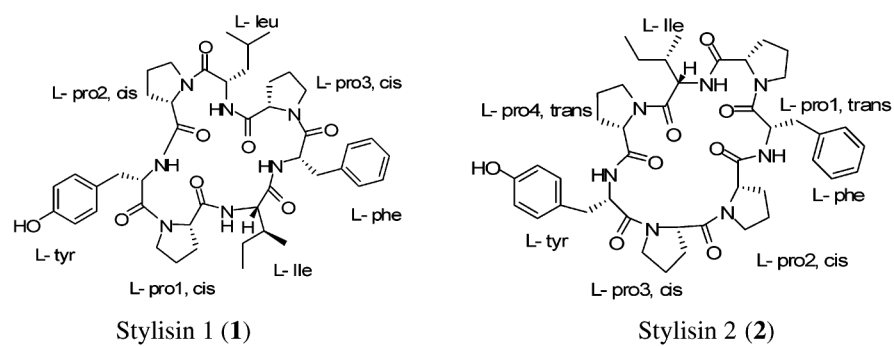
**Figure 3.**  
Key NOESY correlations of stylisin 1 (1).



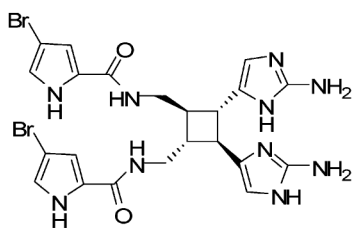
**Figure 4.**  
Key HMBC correlations of stylisin 2 (2).



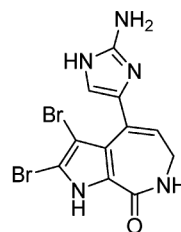
**Figure 5.**  
Key ROESY correlations of stylisin 2 (2).



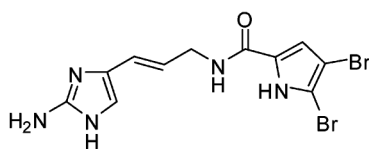
Phakellistatin 13 (3)



Scepterin (4)



Stevensine (5)



Oroidin (6)

Chart 1.

Table 1

<sup>13</sup>C and <sup>1</sup>H NMR Data of Stylistin 1 (1)<sup>a</sup>

amino acid	position	$\delta_C$	$\delta_H$	HMBC ( <sup>1</sup> H– <sup>13</sup> C)
Phe	C <sub>1</sub>	172.1, qC		
	C <sub>2</sub>	55.8, CH	4.55, m	C1, C3
	C <sub>3</sub>	36.2, CH <sub>2</sub>	3.00, m; 3.05, m	C2, C9
	C <sub>4</sub>	139.7, qC		
	C <sub>5,5'</sub>	129.5, CH	7.37, t; 16.0, 9.0 Hz	C9, C7
	C <sub>6,6'</sub>	128.1, CH	7.28, t; 15.0, 8.0 Hz	C8, C7
	C <sub>7</sub>	126.1, CH	7.18, t; 15.0, 7.5 Hz	C6, C5
	NH		7.44, d; 9.2 Hz	C2, C10
Ile	C <sub>1</sub>	170.5, qC		
	C <sub>2</sub>	53.9, CH	4.53, m	C10, C12, C13, C14
	C <sub>3</sub>	38.2, CH	1.70, m	C13, C14, C15
	C <sub>4</sub>	14.8, CH <sub>3</sub>	0.89, d; 7.0 Hz	C11, C12, C14
	C <sub>5</sub>	24.1, CH <sub>2</sub>	1.20, m; 1.51, m	C12, C13, C15
	C <sub>6</sub>	10.2, CH <sub>3</sub>	0.86, m	C12, C14
	NH		8.95, d; 8.4 Hz	C16
Pro1 ( <i>cis</i> )	C <sub>1</sub>	171.5, qC		
	C <sub>2</sub>	60.9, CH	4.17, d; 8.0 Hz	C16, C18, C19
	C <sub>3</sub>	31.5, CH <sub>2</sub>	2.11, m; 2.30, m	C17, C20
	C <sub>4</sub>	21.9, CH <sub>2</sub>	1.61, m; 1.72, m	C17
	C <sub>5</sub>	46.8, CH <sub>2</sub>	3.25, m; 3.41, m	C17, C18
Tyr	C <sub>1</sub>	169.3, qC		
	C <sub>2</sub>	53.0, CH	4.55, m	C21, C23
	C <sub>3</sub>	36.3, CH <sub>2</sub>	3.27, m; 3.25, m	C21, C22, C29
	C <sub>4</sub>	127.1, qC		
	C <sub>5,5'</sub>	131.1, CH	6.98, d; 8.5 Hz	C26, C27, C29
	C <sub>6,6'</sub>	115.1, CH	6.78, d; 8.5 Hz	C27, C28, C24
	C <sub>7</sub>	156.5, qC		
NH			6.77, d; 4.8 Hz	C30
Pro2 ( <i>cis</i> )	C <sub>1</sub>	170.2, qC		
	C <sub>2</sub>	61.9, CH	4.42, m	C30, C32, C34
	C <sub>3</sub>	31.8, CH <sub>2</sub>	2.11, m	C31
	C <sub>4</sub>	22.2, CH <sub>2</sub>	1.97, m	C31
	C <sub>5</sub>	46.3, CH <sub>2</sub>	3.48, m; 3.62, m	C31
Leu	C <sub>1</sub>	170.9, qC		
	C <sub>2</sub>	52.2, CH	4.71, m	C37, C35, C38
	C <sub>3</sub>	38.8, CH <sub>2</sub>	1.35, m; 1.62, m	C39, C40

amino acid	position	$\delta_C$	$\delta_H$	HMBC ( $^1H$ - $^{13}C$ )
	C <sub>4</sub>	24.9, CH	1.17, m	C39, C40
	C <sub>5</sub>	20.7, CH <sub>3</sub>	1.02, d; 6.5 Hz	C38, C40
	C <sub>6</sub>	23.4, CH <sub>3</sub>	0.88, d; 6.5 Hz	C38, C40
	NH		8.15, d; 3.6 Hz	C41
Pro3 ( <i>cis</i> )	C <sub>1</sub>	172.7, qC		
	C <sub>2</sub>	58.6, CH	4.58, m	C44, C45, C1
	C <sub>3</sub>	30.7, CH <sub>2</sub>	1.85, m; 2.01, m	
	C <sub>4</sub>	21.7, CH <sub>2</sub>	1.54, m; 1.51, m	C42
	C <sub>5</sub>	47.12, CH <sub>2</sub>	3.72, m	C42

<sup>a</sup>In acetone-*d*<sub>6</sub>, 600 MHz for  $^1H$  and 150 MHz for  $^{13}C$  NMR. Carbon multiplicities were determined by DEPT experiments. Coupling constants (*J*) are in Hz.

Table 2

<sup>13</sup>C and <sup>1</sup>H NMR Data of Stylistin 2 (2)<sup>a</sup>

amino acid	position	$\delta_C$	$\delta_H$	HMBC ( <sup>1</sup> H- <sup>13</sup> C)
Phe	C <sub>1</sub>	168.5, qC		
	C <sub>2</sub>	53.3, CH	4.34, m	C42, C5
	C <sub>3</sub>	38.1, CH <sub>2</sub>	3.06, m; 3.21, m	C4, C6, C3
	C <sub>4</sub>	136.9, qC		
	C <sub>5,5'</sub>	130.0, CH	7.11, t; 13.8, 7.2 Hz	C1, C2, C5, C43
	C <sub>6,6'</sub>	128.4, CH	7.27; 14.0, 7.8 Hz	C44, C3, C1
	C <sub>7</sub>	127.1, CH	7.26, t; 15.0, 7.8 Hz	C2, C3
Pro2 ( <i>cis</i> )	NH		6.79, d; 9.6 Hz	C7
	C <sub>1</sub>	171.3, qC		
	C <sub>2</sub>	61.3, CH	3.62, m	C7, C10
	C <sub>3</sub>	31.4, CH <sub>2</sub>	1.47, m	C11, C10
	C <sub>4</sub>	22.3, CH <sub>2</sub>	1.88, m	C8
Pro3 ( <i>cis</i> )	C <sub>5</sub>	47.5, CH <sub>2</sub>	3.64, m	C8
	C <sub>1</sub>	169.4, qC		
	C <sub>2</sub>	57.2, CH	4.22, d, 7.8 Hz	C12
	C <sub>3</sub>	29.7, CH <sub>2</sub>	1.90, m	C16
	C <sub>4</sub>	21.5, CH <sub>2</sub>	1.85, m	C13
Tyr	C <sub>5</sub>	47.0, CH <sub>2</sub>	3.50, m, 3.52, m	C14
	C <sub>1</sub>	172.0, qC		
	C <sub>2</sub>	52.0, CH	4.62, q, 10.0, 6.0 Hz	C17, C19
	C <sub>3</sub>	38.2, CH <sub>2</sub>	2.70, q, 12.0, 5.5 Hz; 2.83, m	C20, C21
	C <sub>4</sub>	136.9, qC		
	C <sub>5,5'</sub>	130.6, CH	6.95, d; 8.4 Hz	C21, C22
	C <sub>6,6'</sub>	115.5, CH	6.80, d; 9.0 Hz	C22, C23
Pro4 ( <i>trans</i> )	C <sub>7</sub>	156.7, qC		C23, C24
	NH		7.62, d; 4.8 Hz	C26
	C <sub>1</sub>	170.7, qC		
	C <sub>2</sub>	61.3, CH	3.57, m	C26, C29
	C <sub>3</sub>	28.1, CH <sub>2</sub>	2.19, m	C30
Ile	C <sub>4</sub>	21.9, CH <sub>2</sub>	1.61, m	C27
	C <sub>5</sub>	46.5, CH <sub>2</sub>	3.57, m	C27, C31
	C <sub>1</sub>	171.6, qC		
	C <sub>2</sub>	58.4, CH	4.18, d, 8.4 Hz	C31, C34
	C <sub>3</sub>	35.5, CH	2.08, m	C36, C34
	C <sub>4</sub>	15.8, CH <sub>3</sub>	0.80, d; 7.8 Hz	C35, C32
	C <sub>5</sub>	25.8, CH <sub>2</sub>	1.27, m; 1.4, m	C36, C33



amino acid	position	$\delta_C$	$\delta_H$	HMBC ( $^1H$ - $^{13}C$ )
	C <sub>6</sub>	10.6, CH <sub>3</sub>	0.82, t; 6.6, 3.0 Hz	C35, C33
	NH		8.98, d; 3.6 Hz	C37
Pro1 ( <i>trans</i> )	C <sub>1</sub>	170.8, qC		
	C <sub>2</sub>	59.3, CH	4.44, m	C37, C41
	C <sub>3</sub>	31.3, CH <sub>2</sub>	2.69, q; 5.4, 12.0 Hz	C37, C41
	C <sub>4</sub>	25.6, CH <sub>2</sub>	2.08, m; 2.21, m	C38
	C <sub>5</sub>	46.2, CH <sub>2</sub>	3.25, m	C42

<sup>a</sup>In acetone-*d*<sub>6</sub>, 600 MHz for  $^1H$  and 150 MHz for  $^{13}C$  NMR. Carbon multiplicities were determined by DEPT experiments. Coupling constants (*J*) are in Hz.