# *Bacillus anthracis* Prolyl 4-Hydroxylase Modifies Collagen-like Substrates in Asymmetric Patterns\*

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Proline hydroxylation is the most prevalent post-translational modification in collagen. The resulting product trans-4hydroxyproline (Hyp) is of critical importance for the stability and thus function of collagen, with defects leading to several diseases. Prolyl 4-hydroxylases (P4Hs) are mononuclear nonheme iron  $\alpha$ -ketoglutarate ( $\alpha$ KG)-dependent dioxygenases that catalyze Hyp formation. Although animal and plant P4Hs target peptidyl proline, prokaryotes have been known to use free L-proline as a precursor to form Hyp. The P4H from Bacillus anthracis (BaP4H) has been postulated to act on peptidyl proline in collagen peptides, making it unusual within the bacterial clade, but its true physiological substrate remains enigmatic. Here we use mass spectrometry, fluorescence binding, x-ray crystallography, and docking experiments to confirm that BaP4H recognizes and acts on peptidyl substrates but not free L-proline, using elements characteristic of an Fe(II)/aKG-dependent dioxygenases. We further show that BaP4H can hydroxylate unique peptidyl proline sites in collagen-derived peptides with asymmetric hydroxylation patterns. The cofactor-bound crystal structures of BaP4H reveal active site conformational changes that define open and closed forms and mimic "ready" and "product-released" states of the enzyme in the catalytic cycle. These results help to clarify the role of BaP4H as well as provide broader insights into human collagen P4H and proteins with poly-L-proline type II helices.

Prolyl 4-hydroxylases (P4Hs)<sup>2</sup> are ubiquitous enzymes that catalyze the post-translational hydroxylation of proline residues forming 4-hydroxyproline (Hyp) (1). The P4Hs belong to the mononuclear non-heme iron  $\alpha$ -ketoglutarate ( $\alpha$ -KG) family of oxygenases requiring Fe(II),  $\alpha$ KG, and O<sub>2</sub> for catalysis (2, 3). These enzymes are characterized by 2His-1-carboxylate (Asp/Glu) facial triad residues that coordinate to the active site iron center (4–7). The common co-substrate  $\alpha$ KG binds to iron via its oxo and carboxylate groups and provides reducing equivalents for these enzymes to reduce  $O_2$  to a hydroxyl group (2). The oxidation of various substrates containing prolyl residues is accomplished via the activation of a C-H bond of an unactivated alkyl group (8–10). Although one atom from molecular oxygen is incorporated into the substrate, the second atom is utilized in decarboxylation of  $\alpha KG$  to form the by-product succinate.

The P4Hs are remarkable in their ability to recognize varied substrates with diverse sequences motifs and are central to the growth, assembly, and function of the host organism (1, 11–13). Although these enzymes exhibit substrate diversity, they share a general reaction mechanism and a conserved fold at their core. A double-stranded  $\beta$ -helix (DSBH) comprising of eight anti-parallel  $\beta$ -strands contains the conserved active site residues required for iron and  $\alpha$ KG binding (Fig. 1) (14, 15). Although the DSBH motif is highly conserved in all P4Hs and oxygenases in general, the overall sequence identity is low (11) and the area surrounding the DSBH is the region that gives specificity to each enzyme and substrate (14).

In humans, there are two types of P4Hs, a collagen P4H (C-P4H) and hypoxia inducible factor-1 $\alpha$  (HIF1 $\alpha$ )-prolyl hydroxylase domain proteins (PHDs) (13, 16–18). The HIF1 $\alpha$ -PHDs are involved in the post-translational hydroxylation of a LXXLAP motif in HIF1 $\alpha$  and play a key role in cellular response to hypoxia (19-21). Whereas, C-P4H catalyzes the hydroxylation of proline residues at the Y position of (X-Y-G), repeats (where, *n* denotes the number of repeated sequences and X is frequently proline) in collagen and other collagenous domain containing proteins (17, 22). The  $(P-P-G)_n$  repeats found in collagen adopt a poly-L-proline type II (PPII) conformation in each strand of the collagen triple helix. The PPII helical residues are known to interact with proline recognition domains (PRDs) and are involved in a variety of processes including bacterial and viral pathogenesis, elasticity, transcription, cell motility, and self-assembly (23, 24). The hydroxylation of Y proline residues in collagen is of utmost importance in providing stability to the collagen helical structure. Defects in collagen stability due to insufficient hydroxylation or excessive C-P4H activity can cause diseases like scurvy and fibrosis, respectively (1). Because C-P4H is the rate-limiting enzyme in collagen biosynthesis, therefore, it is an attractive target for developing corresponding mechanism-based activators and inhibitors for scurvy and fibrotic disorders.

Three C-P4H isoenzymes exist in vertebrates and these are all  $\alpha_2\beta_2$  tetramers. The  $\beta$ -subunit is identical to protein-disulfide isomerase and is necessary for preventing the aggregation of  $\alpha$ -subunit and to retain it in the endoplasmic reticulum (22, 25). The catalytic  $\alpha$ -subunit consists of 3 domains: an N-do-



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The atomic coordinates and structure factors (codes 5IAT, 5IAV, and 5IAX) have been deposited in the Protein Data Bank (http://wwpdb.org/).

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<sup>&</sup>lt;sup>2</sup> The abbreviations used are: P4H, prolyl 4-hydroxylase; Hyp, hydroxyproline; αKG, α-ketoglutarate; MLI, malonate; PPII, poly-L-proline type II; DSBH, double-stranded β-helix; PRD, proline recognition domain; PSB, peptidesubstrate binding; SBL, substrate binding loop; vCPH, viral collagen P4H; XIC, extracted ion chromatogram; SBL, substrate binding loop; HIF-1α, hypoxia inducible factor-1α; PDB, Protein Data Bank; PHD, prolyl hydroxylase domain protein.



FIGURE 1. Sequence alignment of BaP4H with vCPH, human C-P4H, and CrP4H and secondary structures defined above the sequence.  $\alpha$  Helices are indicated by helical *lines* with  $\alpha$  labels,  $\beta$  strands are indicated by *arrows* with  $\beta$  labels. Active site residues are highlighted in *magenta*. The missing loop in BaP4H comprises of residues 70–83. The 3<sub>10</sub> helices have not been labeled. The sequences were aligned by their secondary structures using ESPript 3.

main responsible for tetramer assembly, a peptide-substrate binding (PSB) domain, and a C-terminal catalytic domain containing the active site (22, 26). The crystal structures of the N and PSB domains have been determined in presence of a C-P4H inhibitor (poly-L-proline) and (P-P-G)<sub>3</sub> substrate (26, 27). The two peptides adopt a PPII helical conformation and bind to the PSB domain in a groove lined by conserved tyrosines. The PSB domain of C-P4H consists of  $\alpha$ -helices that forms tetratricopeptide repeat motifs. The tetratricopeptide repeat motifs are promiscuous in mediating protein-protein interactions and recognize many ligands including proline-rich peptides (28). Because C-P4H acts on X-P-G repeats in collagen and other proteins with collagen-like sequences, the substrate specificity of this enzyme must arise from residues beyond the conserved region of tetratricopeptide repeat. An open question, which is mainly due to the lack of a structure of C-P4H with an intact catalytic domain, is how the PSB and catalytic domains communicate to position a substrate around the active site for sitespecific prolyl hydroxylation. The poor solubility of the  $\alpha$ -subunit containing the catalytic site has proved challenging for structural studies and there has been an ongoing effort in pursuit of an improved model for C-P4H.

Previous structures of P4Hs have provided initial insights into enzyme structure and function. A viral collagen P4H (vCPH) has been shown to hydroxylate poly-L-proline and prolines in proteins containing (P-A-P-K)<sub>n</sub> proline-rich repeats (29, 30). The crystal structures of vCPH with manganese-bound and a zinc- $\alpha$ KG complex show residues involved in metal and  $\alpha$ KG binding and related conformational changes when  $\alpha$ KG binds (30). Docking simulations using a PKPAPK peptide showed possible interactions with vCPH and the peptide model forms a PPII helix in the PSB groove of this enzyme (30). Plants and green algae have collagenous domain containing glycoproteins with proline-rich motifs where Hyp is important for cellwall assembly (31, 32). A structure of an algal P4H from *Chlamydomonas reinhardtii* (CrP4H) complexed with a  $(S-P)_5$  peptide is one of the few P4H crystal structures solved to date with a substrate bound in the active site (33). The structure illustrates various enzyme-substrate interactions in the binding groove and in the substrate binding loop (SBL) region adjacent to the active site, with the peptide adopting the PPII helix conformation (34, 35). These highly mobile SBL regions are conserved through P4Hs and in the absence of a substrate one of the loops is usually disordered (30, 36, 37).

Similar to other organisms bacteria contain Hyp, but a distinctive characteristic of bacterial P4Hs is to produce Hyp from free L-proline instead of peptidyl proline (38). On the other hand, Bacillus anthracis P4H (BaP4H) is unique among bacterial P4Hs in that it is reported to bind the human collagen peptide (P-P-G)<sub>10</sub> with an affinity similar to that of human C-P4H (1, 36, 39). A sequence alignment of BaP4H with human C-P4H shows  $\sim$  30% sequence identity. Structural alignment of BaP4H with CrP4H and vCPH structures shows conserved secondary structure similarities between various P4Hs (Fig. 1). Examining the proteome of B. anthracis reveals collagen-like proteins that contain stretches of (G-X-Y) repeats. A genome-based study of collagen structural motifs in bacterial and viral proteins showed that bacterial collagen structural motifs favor a (G-P-T) triplet unit rather than the second most prevalent (G-P-Q) repeat unit (40). Because the proteome of *B. anthracis* contains (G-P-T) repeats in some proteins, it has been proposed that a possible physiological substrate for BaP4H could be a protein consisting of  $(G-P-T)_n$  repeats (36, 41). However, the substrate for BaP4H is not yet fully established.

The previously reported crystal structure of BaP4H is an apoform of the enzyme lacking necessary cofactors (36, 41). To gain insight into the site-specific prolyl hydroxylation activity of BaP4H, it is important to understand how the active site resi-



dues are organized to assemble the cofactors, iron and  $\alpha {\rm KG},$  for substrate recognition.

In the current study we examine cofactor and substrate binding, including site-specific hydroxylation activity of BaP4H by a combination of biochemical, mass spectrometry, and x-ray crystallographic methods. Attempts toward determining the crystal structure of a substrate-bound form of BaP4H has not been fully successful, therefore molecular docking was used to simulate the binding mode of a collagen-like proline-rich (P-P-G)<sub>5</sub> peptide. Here we describe the first biochemical and structural analysis of the cofactor-bound forms of BaP4H, and identified collagen-derived (P-P-G)<sub>5,10</sub> peptides as substrates with hydroxylation of proline residues occurring in both X and Y positions. Furthermore, the crystal structures reveal conformational changes upon cofactor binding, and represent "ready" and "product-released" states of the enzyme in the catalytic mechanism (42).

### **Experimental Procedures**

*Materials*—All buffers, media ingredients, and other reagents were acquired from Sigma and, unless otherwise stated, were of the highest purity available. Solutions were prepared using Nanopure deionized water. N<sub>2</sub> (99.98%), NO (>99%), and liquid helium (>99%) were obtained from Praxair (Cedar Rapids, IA).

*Protein Expression and Purification*—The *bap4h* gene was synthesized by Genscript (Piscataway, NJ), and then subcloned into the pET28a expression vector using restriction sites NcoI and NdeI. The BaP4H plasmid was transformed into *Escherichia coli* BL21 star DE3 pLysS cells (Life Technologies). Apo-BaP4H was expressed and purified with a slight modification as reported previously (41).

The BaP4H-(P-P-G)<sub>3</sub> construct was made by PCR using appropriate primers to extend the C terminus end to contain a  $(P-P-G)_3$  peptide. The PCR product was cut with NcoI and BamHI and ligated back into pET28a. This new plasmid was transformed, expressed, and purified as above.

UV-visible Spectroscopy Measurements—To prevent any turnover, all UV-visible measurements were performed in an anaerobic chamber (Vacuum Atmospheres, Hawthorne, CA) using an Ocean Optics DH-2000-BAL light source (Ocean Optics, Dunedin, FL). Anaerobic solutions were made from dry stocks in a glovebox. Spectra were measured between 250 and 800 nm in a buffer containing 50 mM Tris, pH 7.4, 30 mM NaCl in a 100- $\mu$ l quartz cuvette. The spectrum of Fe(II)-BaP4H was measured after adding equimolar concentrations of Fe(NH<sub>4</sub>)<sub>2</sub>- $(SO_4)_2$  and 776  $\mu$ M apo-BaP4H. To detect any spectral changes associated with  $\alpha$ KG binding, stoichiometric aliquots of Fe(II) and  $\alpha$ KG were added to a 776  $\mu$ M solution of apo-BaP4H and spectra were recorded after each addition. The difference spectrum of Fe(II)-BaP4H- $\alpha$ KG ternary complex (solid line, Fig. 2A), was obtained by subtracting the Fe(II)-BaP4H spectrum (dashed line, Fig. 2A) from the original ternary complex spectrum to highlight the absorption maximum at 520 nm characteristic of *a*KG-dependent non-heme iron dioxygenases (43, 44).

To determine the metal stoichiometry, a 100- $\mu$ l solution containing equimolar concentration of BaP4H and  $\alpha$ KG (800

 $\mu\rm M)$  was titrated with a 2 mM stock solution of Fe(NH\_4)\_2(SO\_4)\_2 anaerobically in a glovebox. Aliquots of Fe(II) were added to this solution of BaP4H/ $\alpha\rm KG$  until the absorption at 520 nm had saturated.

Electron Paramagnetic Resonance Spectroscopy (EPR)-Xband EPR spectra were recorded on a Bruker EMX spectrometer (Bruker Biospin Corp., Billerica, MA), equipped with an Oxford ITC4 temperature controller, a Hewlett-Packard model 5340 automatic frequency counter, and Bruker gauss meter. The EPR samples were prepared anaerobically with an enzyme concentration of 100 μM in 50 mM Tris, pH 7.4, 150 mM KCl, 5 mm  $\beta$ -mercaptoethanol. Reduced Fe(II)-BaP4H samples were prepared by adding sodium dithionite to a final concentration of 1 mm. NO gas was bubbled into the protein solutions with cofactors in an anaerobic sealed vial and the solution transferred to an EPR tube via a Hamilton syringe. All spectra were collected at microwave frequency, 9.43 GHz; receiver gain, 2  $\times$ 10<sup>4</sup>; modulation frequency, 100 kHz; temperature, 4 K; microwave power, 10 milliwatts; modulation amplitude, 10 G; sweep time, 83.89 s; and 16 scans. The Fe(II)-BaP4H spectrum (data not shown) was recorded using 1 scan keeping all other parameters the same.

Fluorescence Binding Titrations-Fluorescence spectra were recorded at room temperature using a Cary Eclipse Fluorescence Spectrophotometer (Agilent Technologies, Santa Clara, CA). Fluorescence titration was used to determine the binding affinity of substrates and cofactors to BaP4H. The decrease in intensity of intrinsic tryptophan fluorescence at 295 nm served as an indicator of cofactor binding. Titration experiments were performed using an enzyme concentration of 1  $\mu$ M in 50 mM Tris, pH 7.4, 30 mM NaCl. All titrations were performed anaerobically using solutions that were prepared in a glove box. In a typical anaerobic titration, a degassed protein solution (3 ml) was sealed in a screw top quartz cuvette with a septum to prevent diffusion of O<sub>2</sub> into the anaerobic solution. Anaerobic cofactor solutions were added in aliquots via syringe to the quartz cuvette. The wavelength used for excitation was 295 nm and the emission was monitored from 310-380 nm using excitation and emission slit widths of 5 nm. The signal intensity was recorded at the maxima, 329 nm.

Hydroxylation Product Identification by Liquid Chromatography Tandem Mass Spectrometry-All mass spectra LC-MS/ MS data were collected on an Orbitrap Fusion mass spectrometer (Thermo Fisher Scientific) equipped with an Easy nLC 1000 for sample handling and peptide separations. Approximately 200 fmol of peptide resuspended in 5% formic acid + 5% acetonitrile was loaded onto a 125-µm inner diameter fusedsilica microcapillary with a needle tip pulled to an internal diameter less than 5  $\mu$ m. The column was packed in-house to a length of 15 cm with a C<sub>18</sub> reverse phase resin (120 Å pore size, 5.0 µm particle size, GP-C18, SePax Technologies). The peptides were separated using a 20-min linear gradient from 3 to 35% buffer B (100% ACN + 0.125% formic acid) equilibrated with buffer A (3% ACN + 0.125% formic acid) at a flow rate of 600 nl/min across the column. The Orbitrap Fusion mass spectrometer was operated in the data-dependent positive ion mode using the top speed strategy. In brief, the scan sequence for the Fusion Orbitrap began with an MS1 spectrum (Orbitrap



analysis, resolution 120,000, 800-1000 m/z scan range for (P-P-G)<sub>10,</sub> 600 – 800 m/z scan range for (P-P-G)<sub>5</sub>), AGC target 5 × 10<sup>5</sup>, maximum injection time 100 ms, dynamic exclusion of 5 s). "Top speed" (2 s) was selected for MS2 analysis, which consisted of HCD (quadrupole isolation set at 2.0 Da and ion trap analysis, AGC  $1 \times 10^5$ , collision energy 25%, maximum injection time 250 ms). MS2 fragment ions were analyzed in the Orbitrap, resolution 15,000. A suite of in-house software tools were used for .RAW file processing and controlling peptide false discovery rates (45). MS/MS spectra were searched against a custom database containing the  $(P-P-G)_{10}$  sequence and 100 common contaminants with both the forward and reverse sequences. Database search criteria are as follows: tryptic with two missed cleavages, a precursor mass tolerance of 1 Da, fragment ion mass tolerance of 0.1 Da, and variable hydroxylation of proline (15.99491 Da). Peptides were filtered to a 1% false discovery rate using linear discriminate analysis (45). We used a modified version of the Ascore algorithm to quantify the confidence with which each hydroxylation site could be assigned to a particular residue. Hydroxylation sites with Ascore values >13 ( $p \le 0.05$ ) were considered confidently localized to a particular residue (45). The enzymatic assays to verify peptide hydroxylation were performed aerobically in 100 mM Tris, pH 7.4, using (10 or 50  $\mu$ M) apo-BaP4H preincubated anaerobically for 30 min with (10 or 50  $\mu$ M) Fe(II). A reaction mixture containing 500  $\mu$ M  $\alpha$ KG, ascorbate (10 or 50  $\mu$ M), and 100  $\mu$ M of either  $(P-P-G)_5$  or  $(P-P-G)_{10}$  were prepared at 20 °C. The reactions were initiated by the addition of Fe(II)-BaP4H solution and incubated for 45 min at 20 °C. The reactions were quenched in 0.5% TFA and vacuum concentrated (SpeedVac concentrator, Savant, SPD131DDA). The dried powder containing the product was collected and stored at 4 °C until it was analyzed by mass spectrometry.

*Data Analysis and Equations*—Data in Fig. 2 were fit with Sigma Plot 12.0 (Systat Software Inc., Point Richmond, CA) using a ligand-binding model (Equation 1) as described previously (46, 47),

$$f = f_o + (f_m - f_o) \frac{(nP + x + K_d) - \sqrt[2]{(nP + x + K_d)^2 - 4nPx}}{2nP}$$
(Eq. 1)

where *f* is the fluorescence signal resulting from binding of the metal cofactor,  $f_o$  is the signal from a protein solution in the absence of any cofactors,  $f_m$  corresponds to the maximal quenched fluorescence intensity when a cofactor was bound,  $K_d$  is the dissociation constant, *P* and *x* are total protein and added metal ion/ligand concentrations, respectively, *n* is the number of binding sites. A better approximation of the binding affinity can be obtained by using Equation 1 in cases where the protein concentration is higher than the estimated  $K_d$  (46, 47).

*Crystallization of Apo-BaP4H*—Initial crystals of apo-BaP4H were obtained using the hanging-drop vapor-diffusion method with optimization around a previously reported condition (41) by varying pH and PEG concentrations. Crystal drops were set up at 20 °C with 1  $\mu$ l of 24 mg/ml of apo-BaP4H mixed with 1  $\mu$ l of precipitant solution containing 0.04 M potassium phosphate monobasic, pH 4.0–7.0, 13–18% PEG 8000, and 20% glycerol.

Crystals would appear in 1-2 days and the best crystal used for data collection was obtained from 0.04 M potassium phosphate monobasic, pH 6.0, 14% PEG 8000, and 20% glycerol. Crystals were taken from the precipitant solution and directly frozen in liquid nitrogen.

Crystallization of Co(II)-BaP4H-MLI—A solution of 9 mg/ml of apo-BaP4H, 1 mM CoCl<sub>2</sub>, and 1 mM L-Pro was incubated on ice for 30 min before setting crystal drops. Crystals were obtained using hanging-drop vapor-diffusion at 20 °C from the PACT screen (Molecular Dimensions) by mixing 1  $\mu$ l of the above protein solution with 1  $\mu$ l of reservoir solution. Initial crystals appeared in 0.1 M malonate/imidazole/boric acid, pH 6.0, 25% PEG 1500 and were optimized by varying pH and precipitant concentration. Diffraction quality crystals were obtained from 0.1 M malonate/imidazole/boric acid, pH 6.5, 18% PEG 1500. The crystals of Co(II)-BaP4H-MLI were plate shaped and grew in about 2 days. The crystals were soaked in a cryoprotectant solution of 0.1 M malonate/imidazole/boric acid, pH 6.5, 18% PEG 1500, and 20% glycerol for 5 min and submerged in liquid nitrogen.

Crystallization of Co(II)-BaP4H-PPG—Crystals of BaP4H fused with (P-P-G)<sub>3</sub> peptide to the C terminus were obtained from the JCSG screen (Qiagen) using sitting-drop vapor-diffusion at 20 °C. A solution of 12 mg/ml of BaP4H-(PPG)<sub>3</sub>, 1 mM CoCl<sub>2</sub>, and 1 mM  $\alpha$ KG was incubated on ice for 30 min. Crystal drops were set up by mixing equal volumes of above protein solution with reservoir solution. Crystals were obtained in 0.15 M KBr and 30% PEG 2000 MME. Plate-shaped crystals would appear in about 2 days with dimensions of  $\sim$ 70  $\times$  70  $\times$  20  $\mu$ m. The crystals were soaked in cryoprotectant solution containing 0.15 M KBr, 30% PEG 2000 MME, and 20% glycerol for 5 min, prior to freezing in liquid nitrogen.

Data Collection and Analysis—Data sets for apo-BaP4H and Co(II)-BaP4H-PPG were collected at 100 K at the Advanced Light Source, beamline 4.2.2. Data collection parameters were: apo-BaP4H, 0.5° oscillation, 4-s exposure time, 120-mm detector distance; and Co(II)-BaP4H-PPG, 0.1° oscillation, 0.4-s exposure time, 250 mm detector distance. The data set for Co(II)-BaP4H-MLI was collected at 100 K at the Advanced Photon Source, beamline 19-ID-D using 0.5° oscillation, 0.6-s exposure time, and 182-mm detector distance. The data were indexed and integrated using the XDS package (48) and merged and scaled using SCALA (49) (CCP4 suite) (50). Fluorescence scans were taken for cobalt containing crystals and a peak for cobalt was observed at 6.92 KeV. Data collection and refinement statistics are summarized in Table 2.

Structure Determination—Molecular replacement for apo-BaP4H was done with PHASER (51) using PDB code 3ITQ as a starting model. Phases for Co(II)-BaP4H-MLI and Co(II)-BaP4H-PPG were obtained by molecular replacement with our apo-BaP4H structure. Matthews coefficients and estimated solvent contents for apo-BaP4H, Co(II)-BaP4H-MLI, and Co(II)-BaP4H-PPG were 2.15 Å<sup>3</sup>/Da, 43%; 1.96 Å<sup>3</sup>/Da, 37%; and 2.03 Å<sup>3</sup>/Da, 39%, respectively. Refinement was carried out using phenix.refine from the PHENIX software package (52). Structures containing metals had bond length and angle parameters generated by phenix.metal\_coordination. Following iterative rounds of refinement using simulated annealing, energy mini-





FIGURE 2. **Spectral properties and substrate/cofactor binding affinities** ( $K_d$ ) of Fe(II)-BaP4H. A, UV-visible spectra of the reaction of 776  $\mu$ M apo-BaP4H in the presence of Fe(II) and  $\alpha$ KG. Apo-BaP4H, *dotted line*; Fe(II)-BaP4H, *dashed line*; Fe(II)-BaP4H- $\alpha$ KG ternary complex, *solid line*. B, EPR spectra of BaP4H in the presence of NO: top, 100  $\mu$ M apo-BaP4H with equimolar concentration of Fe(NH<sub>4</sub>)<sub>2</sub>(SO<sub>4</sub>)<sub>2</sub>; *bottom*, NO-Fe(II)-BaP4H with 100  $\mu$ M  $\alpha$ KG. C, the intrinsic tryptophan fluorescence was monitored as corresponding peptides were titrated into an anaerobic solution containing 1  $\mu$ M apo-BaP4H, 3  $\mu$ M Fe(NH<sub>4</sub>)<sub>2</sub>(SO<sub>4</sub>)<sub>2</sub>; *and* 100  $\mu$ M  $\alpha$ KG. The binding affinities for the (P-P-G)<sub>10</sub> and (P-P-G)<sub>5</sub> peptides were determined to be 0.74 ± 0.11  $\mu$ M (*circles* and *solid line*) and 0.3 ± 0.02  $\mu$ M (*triangles* and *dashed line*), respectively. D, the  $K_d$  for L-proline was determined to be 7.9 ± 2.5  $\mu$ M and was titrated into an anaerobic solution containing 1  $\mu$ M BaP4H, 3  $\mu$ M Fe(NH<sub>4</sub>)<sub>2</sub>(SO<sub>4</sub>)<sub>2</sub>, *and* 100  $\mu$ M  $\alpha$ KG. E, the  $K_d$  (4.2 ± 0.5  $\mu$ M) for malonate was obtained by titrating it into an anaerobic solution containing 1  $\mu$ M BaP4H and 3  $\mu$ M Fe(NH<sub>4</sub>)<sub>2</sub>(SO<sub>4</sub>)<sub>2</sub>. F, the  $K_d$  for Co(II) was determined to be 0.81 ± 0.09  $\mu$ M. CoCl<sub>2</sub> was titrated into an anaerobic solution containing 1  $\mu$ M BaP4H.

mization, real space refinement, and B-factor refinement, model building was done in COOT (53). Addition of ligands into distinct positive electron density was based on a simulated annealing omit map generated by omitting the ligand. Water molecules were added into clear densities in later rounds of refinement and composite omit maps were generated to verify the structures. All structures contain a disordered N terminus and flexible loop region that varies slightly for each model. The disordered residues that were not modeled for each structure and chain are as follows: apo-BaP4H chain A 1-10, chain B 1-11 and 73-82; Co(II)-BaP4H-MLI chain A: 1-11 and 72-80, chain B: 1-11 and 71-81; Co(II)-BaP4H-PPG chain A: 1-11, 70-73, and 79-83, chain B: 1-11 and 70-83. The 2nd and 3rd repeat of (P-P-G)<sub>3</sub> in the Co(II)-BaP4H-PPG structure were also disordered and not included in the final model. In Co(II)-BaP4H-MLI, no density for proline appeared in the structure and instead density for malonate (from crystallization solution) was observed in the final model. The Ramachandran geometries of each structure were analyzed by PROCHECK (54, 55) (Table 2). All structural figures were made using PyMOL.

Computational Ligand Docking—The Co(II)-BaP4H-PPG structure (excluding the modeled P-P-G repeat from the C terminus end) was used for docking calculations with a peptide from PDB entry 1NAY truncated to  $(P-P-G)_5$ . The Co(II)-BaP4H-PPG structure and the  $(P-P-G)_5$  peptide were submitted to the ZDOCK server (56). ZDOCK was used for initial rigid body and low resolution docking. Most of the outputs from ZDOCK docked the peptide in the peptide binding groove

region. The most reasonable conformation (based on CrP4H peptide bound structure, PDB code 3GZE) of peptide docking from ZDOCK was used for flexible docking with the FlexPepDock server (57, 58). The FlexPepDock server performs a rotamer search on protein and treats the peptide as a completely flexible entity. It performs a rigid body optimization and a peptide backbone optimization, thereby optimizing the peptide conformation.

#### Results

Analysis of Cofactor and Substrate Binding to BaP4H-To probe the nature of spectral changes associated with cofactor and substrate binding to BaP4H, we employed UV-visible and EPR in conjunction with fluorescence saturation studies. The UV-visible spectrum of as-isolated BaP4H exhibited no absorption features above 280 nm. Anaerobic addition of Fe(II) had no effect on the absorption spectrum. However, when  $\alpha$ KG was added to complex of Fe(II)-BaP4H the absorbance in the 300-500 nm region developed a broad shoulder with the absorption maximum at 520 nm (Fig. 2A). This feature is due to a charge transfer transition of the  $\alpha$ KG-Fe(II)-Ba4H ternary complex and is characteristic of *a*KG-dependent non-heme iron dioxygenases (43, 44, 59). The absorbance at 520 nm was monitored to determine the stoichiometry of Fe(II) binding to the enzyme. Titrating Fe(II) into an anaerobic solution of BaP4H in the presence of  $\alpha$ KG resulted in an increase in absorption at 520 nm and showed a clear inflection point at 1:1 ratio of Fe(II) to BaP4H monomer (data not shown).





FIGURE 3. **Hydroxylated peptide product identification by LC-MS.** *A*, XIC peak profiles of  $[(P-P-G)_5 + OH]^{+2}$  (green) and  $(P-P-G)_5$  (black) as separated on a C<sub>18</sub> column. *B*, XIC peak profiles for  $[(P-P-G)_{10} + OH]^{+3}$  (green) and  $(P-P-G)_{10}$  (black) separated on a C<sub>18</sub> column. *C*, MS analysis of  $(P-P-G)_5$  before and after hydroxylation. The *top panel* shows  $[(P-P-G)_5 + OH]^{+2}$ , the isotopic mass envelope starts at *m/z* 645.820. The *bottom panel* shows the substrate peptide  $((P-P-G)_5)$ , with an isotopic mass envelope starting at *m/z* 637.8345. Hydroxylation of  $(P-P-G)_5$  results in a *m/z* change of 7.9945 for the double charged, this is equivalent to a 15.989 Da increase to the peptide mass, which corresponds to incorporation of a single oxygen atom. *D*, MS analysis of  $(P-P-G)_{10}$  before and after hydroxylation. The *top panel* shows  $[(P-P-G)_{10} + OH]^{+3}$ , the isotopic mass envelope starts at *m/z* 849.4371. The *bottom panel* shows the substrate peptide mass (P-P-G)\_{10} + OH]^{+3}, the isotopic mass envelope starts at *m/z* 849.4371. The *bottom panel* shows the substrate peptide  $[(P-P-G)_{10}]$ , with an isotopic mass envelope starting at *m/z* 844.1041. Hydroxylation of  $(P-P-G)_{10}$  results in a *m/z* change of 5.3323 for the triply charged ion, this is equivalent to a 15.989 Da increase to the peptide mass, which corresponds to incorporation of a single oxygen atom.

To investigate the interaction of ferrous iron with dioxygen in BaP4H we used nitric oxide (NO), which serves as a surrogate of dioxygen and assessed the NO-Fe(II)-BaP4H complex by EPR. BaP4H with reduced Fe(II) exists in an EPR-silent (S = 2) spin state, similar to other non-heme iron proteins (data not shown). When NO interacts with the Fe(II) center, the spin state is changed from S = 2 into an EPR-active S = 3/2 species. EPR signals associated with this species are g = 3.94, 1.97 (Fig. 2B, top) (60-63). Anaerobic addition of NO to the reduced  $\alpha$ KG-Fe(II)-BaP4H complex diminished the intensity of the signal at g = 3.91 and the nitrosyl signal at g = 1.98 predominates (Fig. 2B, bottom). In addition, a signal due to oxidized ferric iron is detectable in both NO spectra at g = 4.31. These results suggest that  $\alpha KG$  binds to the Fe(II) center and alters the electronic environment without displacing bound NO, as indicated by the small changes in the observed g values.

To determine the binding affinities of various cofactor and potential substrates with BaP4H, intrinsic tryptophan fluores-

cence served as a probe for ligand binding. Because (P-P-G)<sub>10</sub> was suggested to bind to BaP4H with an affinity similar to that of C-P4H (36) we sought to determine the binding affinities of (P-P-G)<sub>10</sub> and (P-P-G)<sub>5</sub> peptides in addition to L-proline. By measuring the decreasing tryptophan fluorescence intensity at 329 nm, whereas titrating in the appropriate ligand to  $\alpha$ KG-Fe(II)-BaP4H ternary complex, the binding affinities were determined. The  $K_d$  of (P-P-G)<sub>10</sub> and (P-P-G)<sub>5</sub> peptides were calculated to be 0.74  $\pm$  0.11 and 0.3  $\pm$  0.02  $\mu$ M, respectively (Fig. 2*C*). Although free L-proline binds to Fe(II)-BaP4H ( $K_d$  of 7.9  $\pm$  2.5  $\mu$ M), the affinity is an order of magnitude lower than the peptides (Fig. 2, *C* and *D*). The binding affinity of a known P4H inhibitor malonate with the Fe(II)-BaP4H complex was determined to be 4.2  $\pm$  0.5  $\mu$ M and Co(II) binds to apo-BaP4H with a  $K_d$  of 0.81  $\pm$  0.09  $\mu$ M (Fig. 2, *E* and *F*).

Hydroxylation of Human Collagen-like  $(P-P-G)_{5,10}$  Peptides by BaP4H—It has been proposed that BaP4H hydroxylates peptidyl proline but not free L-proline, a common substrate for



#### TABLE 1

#### Amino acid sequence of (P-P-G)<sub>5</sub> and (P-P-G)<sub>10</sub> with the unique hydroxylation sites shown

Xcorr values were calculated by the SEQUEST algorithm. The hydroxylation sites and localization scores were determined by a modified version of the Ascore algorithm. The table shows only peptides that have one or more hydroxylation site(s) localized to >99% confidence. The first three hydroxylated peptides for  $(P-P-G)_5$  were observed under both steady state and single-turn over conditions and the last two peptides were observed only in the single turnover condition. All peptides listed for  $(P-P-G)_{10}$  were discovered under single turnover conditions. RT, retention time.

							Site 1		Site 2
Peptide	Xcorr	Sequence	Observed <i>m/z</i>	Precursor intensity	RT (min)	Position	Localization score	Position	Localization score
	4.3	Nter-PPGPPGPPGP <b>P(OH)</b> GPPG-Cter	645.82804	5.99E+05	11.62	11	40	-	-
	4.8	Nter-PPGPPGPPGP(OH)PGPPG-Cter	645.83138	2.64E+06	12.09	10	40	-	-
(P-P-G) <sub>5</sub>	4.2	Nter-PPGPPGPP(OH)GPPGPPG-Cter	645.83033	1.17E+06	12.39	8	40	-	-
	4.2	Nter-PPGP <b>P(OH)</b> GPPGPPGPC-Cter	645.82847	7.49E+05	12.31	5	32.3	-	-
	1.8	Nter-PPGPPGPPGPPGP(OH)P(OH)G-Cter	653.82512	3.07E+05	12.98	13	25.7	14	25.7
	7.4	Nter-PPGPPGPPGPPGPPGPPGP <b>(OH)</b> PGPPGPPGPPG-Cter	849.43506	7.78E+05	15.72	19	36.9	-	-
	7.1	Nter-PPGPPGPPGPPGPPGPPGPPGPPGP <b>P(OH)</b> GPPG-Cter	849.4339	1.08E+05	15.62	26	32.3	-	-
(P-P-G) <sub>10</sub>	7.8	Nter-PPGPPGPPGPPGPPGPPGPPGPPGPPG-Cter	849.43425	3.83E+05	15.8	13	20.8	-	-
	5.6	Nter-PPGPPGPPGPPGPPGPPGPPGP(OH)P(OH)GPPGPPGPPG-Cter	854.76981	9.60E+04	16.05	19	19.6	20	19.6

bacterial P4Hs. To establish the specific targets recognized by BaP4H, we conducted enzymatic assays using  $(P-P-G)_5$ ,  $(P-P-G)_{10}$ , and free L-proline. Although BaP4H catalyzes the hydroxylation of both  $(P-P-G)_5$  and  $(P-P-G)_{10}$ , it is unreactive toward free L-proline.

To identify the hydroxylated peptides from BaP4H-catalyzed reactions, LC-MS/MS was performed on products isolated from reactions carried out under steady state and single turnover conditions using either  $(P-P-G)_{10}$  or  $(P-P-G)_5$  peptides. The extracted ion chromatogram (XIC) of the (P-P-G)<sub>5</sub> peptide eluted as a sharp peak around 13.3 min (Fig. 3A, black trace) and the hydroxylated-(P-P-G)<sub>5</sub> ((P-P-G)<sub>5</sub>-OH) eluted as a major peak at 12.3 min and a shoulder at 11.8 min (Fig. 3A, green trace). The major peak consists primarily of (P-P-G)<sub>5</sub>-OH, with prolines hydroxylated in the Y position of the second and third repeats (positions 8 and 5, Table 1). The shoulder consists of (P-P-G)<sub>5</sub>-OH, with hydroxylated prolines in the fourth repeat (positions 11 and 10, Table 1). The XIC of the (P-P-G)<sub>10</sub> peptide eluted as a sharp peak around 16.3 min and the hydroxylated- $(P-P-G)_{10}$  ((P-P-G)\_{10}-OH) eluted as a broad peak from 15.5 to 17.3 min with multiple shoulders (Fig. 3B). Unlike (P-P- $G)_5$ -OH we are uncertain of the exact hydroxylated species of (P-P-G)<sub>10</sub>-OH that corresponds to each of the shoulders in Fig. 3B. This is due to the fact that a small modification such as hydroxylation relative to a large molecule like  $(P-P-G)_{10}$  molecule has little effect on the retention time of the modified peptide. Therefore, it made it difficult to resolve the different hydroxylated forms under our experimental conditions. Although hydroxylated (P-P-G)<sub>5</sub> appears at m/z 645.8298 that corresponds to a doubly charged species, the spectrum of  $(P-P-G)_{10}$  contained a peak with m/z 849.4352, corresponding to a triply charged single hydroxylation of  $(P-P-G)_{10}$  (Fig. 3, C and D).

Under steady state conditions hydroxylation sites were localized to both X and Y prolines on the third and fourth repeats of the  $(P-P-G)_5$  peptide (Table 1). Whereas, with  $(P-P-G)_{10}$ , although hydroxylated peptides were identified, due to low localization scores we were unable to assign specific hydroxylation sites.

The single turnover conditions uncovered additional hydroxylation sites on both the  $(P-P-G)_5$  and  $(P-P-G)_{10}$  peptides. The

MS2 data of  $(P-P-G)_5$  identified two additional peptides: one with a hydroxyl contained on the Y proline of the second repeat; and a second peptide that is doubly hydroxylated with the hydroxyl groups located on adjacent prolines in the fifth repeat (Table 1). The MS2 fragment ion pattern for the Y proline on the fourth repeat is shown in Fig. 4A. Interestingly, hydroxylation localized to the X proline of the fourth repeat was also observed (Table 1). With (P-P-G)<sub>10</sub>, hydroxylated peptides were indeed detected with acceptable localization site scores (>15), which irrevocably allowed assignment of hydroxylation sites. Preferential hydroxylation of proline residues toward the C terminus end of the (P-P-G)<sub>10</sub> peptide were observed, with peptides containing a hydroxyl group on the X proline of the fifth and seventh repeats, and the Y proline of the ninth repeat (Table 1). The MS2 fragment ion pattern for the X proline on the fifth repeat is shown in Fig. 4B. In addition, a doubly hydroxylated peptide containing hydroxyl groups on adjacent prolines of the seventh repeat was observed (Table 1). Although small hydroxylation of the proline residues in the X positions of (P-P-G)10 were observed previously with Arabidopsis thaliana P4H (At-P4H-1), there was greater uncertainty associated with the true degrees of hydroxylation (31). The MS results presented here irrefutably revealed the peptides recognized by BaP4H, including specific prolyl hydroxylation sites.

BaP4H Structural Overview—To establish how cofactors and substrate peptides are assembled, two structures of BaP4H in the presence of Co(II) were determined: one in complex with  $\alpha$ KG and a (P-P-G) peptide; and the other structure in complex with a P4H inhibitor, malonate (Figs. 5 and 6, and Table 2). In addition, a structure of apo-BaP4H was obtained (Fig. 7, Table 2) near physiological pH and in a space group different from the previously published apo-BaP4H structure (36). To obtain a peptide-bound structure of BaP4H, a shorter (P-P-G)<sub>3</sub> peptide was incorporated into the C terminus end of the protein close to the predicted substrate binding groove. The resulting structure had clear density only for one repeat that is located close to the active site of the symmetry related molecule (Fig. 5). Attempts to determine the Fe(II)-BaP4H structure have failed because of the poor diffraction quality of the resultant crystals.

All three structures solved contain two molecules in the asymmetric unit, therefore, for simplicity all discussion hereaf-





FIGURE 4. **MS2 fragment ion pattern of hydroxylated (P-P-G)**<sub>5</sub> and (P-P-G)<sub>10</sub> peptides. Assignment of *b* (*blue*) and *y* (*red*) fragment ion series in the MS2 scans from [(P-P-G)<sub>5</sub> + OH]<sup>+2</sup> (A) and [(P-P-G)<sub>10</sub> + OH]<sup>+3</sup> (B) precursor ions. Using a variation of the Ascore algorithm, the hydroxylation was localized with >99% confidence to the Y position in the 4th repeat for (P-P-G)<sub>5</sub> and to the X position in the 5th repeat for (P-P-G)<sub>10</sub>. The precursor peptides were selected from the single-turnover experiments.

ter will be based on chain A from each structure. The overall structures are similar, with root mean square deviations (r.m.s. deviations) of 0.21-0.25 Å over almost  $155 C\alpha$  atoms. With the exception of chain A in apo-BaP4H, all other chains in the BaP4H structures have about 10-15 disordered residues encompassing the region between residues 70-83, which we refer to as the substrate binding loop (SBL). Although there is electron density around the SBL region in both Co(II)-bound BaP4H structures indicating the presence of flexible amino acids, the density quality is too poor to model and the structures re-converge at residue 84.

Structural alignment between BaP4H and an algal P4H (CrP4H), with an r.m.s. deviation of 0.60 Å over 117 C $\alpha$  atoms, shows that two short  $\beta$ -strands ( $\beta$ 4 and  $\beta$ 5) should be located in the disordered SBL region and are flanked between two  $\alpha$ -helices ( $\alpha$ 1 and  $\alpha$ 2) and the  $\beta$ 3 and  $\beta$ 6 strands (Fig. 1). Although the apo-BaP4H structure is the only one with a complete loop, subtle differences were seen in the two cobalt-bound structures with partially complete SBLs. The SBL in the apo-form is positioned in an open conformation perpendicular and away from the active site. Although Co(II)-BaP4H-MLI and Co(II)-BaP4H-PPG structures do not have a completely ordered SBL, however, it appears to be adopting a slightly more closed conformation with the SBL oriented toward the protein surface, which

# Collagen Peptide Hydroxylation by BaP4H

could be due to crystallographic contacts with residues that are ordered in the loop region. A truly closed conformation of the SBL would not be expected until a peptide that is not crystallographically constrained is bound.

Active Site Features of Co(II)-BaP4H-PPG and Co(II)-BaP4H-MLI Complexes—Structural comparison of Co(II)bound forms of BaP4H with our apo-BaP4H shows that regions and residues around the active site within the core  $\beta$ -barrel motif become more organized upon binding of cofactors and/or peptide (Figs. 1 and 5–8). Strands  $\beta$ II and  $\beta$ VII bring triad residues His<sup>127</sup> (2.1 Å), Asp<sup>129</sup> (2.1 Å), and His<sup>193</sup> (2.0 Å) into proximity of coordination to the cobalt atom. The remaining cobalt coordination sites are occupied by either  $\alpha$ KG or malonate along with water molecules, resulting in a six-coordinate complex (Fig. 6).

In Co(II)-BaP4H-PPG,  $\alpha$ KG coordinates cobalt in a planar fashion via its C1 carboxylate (2.0 Å) and the C2 ketone (2.1 Å) opposite of Asp<sup>129</sup> (Fig. 6, A and B). The C5 carboxylate of  $\alpha$ KG is anchored in the active site by hydrogen bonding interactions with Tyr<sup>118</sup> (2.6 Å), Lys<sup>203</sup> (2.9 Å), and Thr<sup>159</sup> (2.6 Å). One of the C1 carboxylates is hydrogen bonded to Thr<sup>207</sup> (2.6 Å) (Fig. 6*B*). The bidentate binding of  $\alpha KG$  to the metal center and stabilization by lysine, arginine, and threonine interactions is conserved in the Fe(II)/ $\alpha$ KG-dependent enzymes, although the positioning of  $\alpha$ KG differs among various enzymes (2). Surprisingly, in the Co(II)-BaP4H-PPG structure, the peptide was located in a symmetry related molecule adjacent to the active site (Fig. 5). Although the (P-P-G)<sub>3</sub> peptide was disordered with density existing for only one repeat, however, the structure displays that the central Y proline of the peptide points to the cobalt center and is at a distance of 5.2 Å (Fig. 6C), which is slightly longer than that observed for CrP4H (30, 34). In Co(II)-BaP4H-MLI, malonate occupies the  $\alpha$ KG binding space in the active site, with its C1 carboxylate located in the same position as the C2 ketone of  $\alpha$ KG (Figs. 6 and 8). Malonate coordinates cobalt in a monodentate mode with the C1 carboxylate trans to Asp<sup>129</sup> at a distance of 2.6 Å from the cobalt center. The C3 carboxylate forms a salt bridge with Lys<sup>203</sup> and is hydrogen bonded to Thr<sup>159</sup> (Fig. 6, *D* and *E*).

In both Co(II)-bound structures, in addition to the catalytic triad residues and  $\alpha$ KG or, malonate that coordinate cobalt, the remaining coordinate sites are occupied by water molecules thus forming a six-coordinate cobalt complex. In the  $\alpha$ KGbound Co(II)-BaP4H-PPG, a water molecule is located 4 Å away from cobalt nearly trans to His<sup>193</sup> in both chains in the structure, which is likely the site where dioxygen would bind with iron present (Fig. 6B). In Co(II)-BaP4H-MLI, two water molecules are coordinating cobalt at distances of  $\sim 2.5$  Å (Fig. 6, D and E). The apo-BaP4H structure has multiple water molecules spread through the active site. One of the waters (W1) occupies the space where metal would bind and is hydrogen bonded to Asp<sup>129</sup> and His<sup>193</sup> (Fig. 7*A*). Two of the other water molecules (W2 and W3) are hydrogen bonding with Thr<sup>159</sup> and Lys<sup>203</sup>, whereas a fourth water (W4) is hydrogen bonding with W2 and W3 (Fig. 7B).

Structures of BaP4H Reveal Open and Closed Active Site Conformations—In the Co(II)-BaP4H-PPG structure, binding of  $\alpha$ KG and peptide induces a conformational change in the





FIGURE 5. Structure of the Co(II)-BaP4H-PPG monomer (green) showing the C terminus of a symmetry related molecule (cyan) interacting with the active site of chain A (green). The  $\beta$ -strands of the DSBH-fold (*light blue*) are labeled  $\beta$ I- $\beta$ VIII. Other  $\beta$  strands (*light orange*) are labeled  $\beta$ 0,  $\beta$ 1,  $\beta$ 2,  $\beta$ 6,  $\beta$ 7, and  $\beta$ 8. Missing residues in the substrate binding loop region are indicated by a *dashed line*. The metal binding residues, His<sup>127</sup>, Asp<sup>129</sup>, and His<sup>193</sup>,  $\alpha$ KG (*yellow*), and the fused (P-P-G) peptide are shown as sticks (oxygen, *red*; nitrogen, *blue*). Cobalt is shown as a *magenta sphere*.



FIGURE 6. Active site of BaP4H with bound cofactors. *A*, bidentate binding mode for  $\alpha$ KG in Co(II)-BaP4H-PPG structure. *B*, active site residues of Co(II)-BaP4H-PPG involved in hydrogen bonding interactions between protein,  $\alpha$ KG, and water are indicated by *black dashes* with corresponding distances. The distance between Tyr<sup>124</sup> and W1 is 2.8 Å and was excluded from the picture for clarity. *C*, the PPG peptide (*green*) at the C terminus interacts with the active site of a symmetry related molecule (*cyan*). The Y proline (P2) is oriented with the C4 carbon 5.2 Å away from the cobalt metal center. The 2nd and 3rd repeats of (P-P-G)<sub>3</sub> were disordered. *D*, monodentate binding mode for malonate in Co(II)-BaP4H-MLI structure. *E*, active site residues of Co(II)-BaP4H-MLI showing hydrogen bonding interactions between protein, malonate, and water with labeled distances.  $\alpha$ KG, malonate, and protein residues are shown as *sticks*: carbon (protein backbone color), oxygen (*red*), nitrogen (*blue*), water molecules as spheres (*cyan*), cobalt as a sphere (*magenta*). The 2*F*<sub>o</sub> – *F*<sub>c</sub> composite omit maps (also shown as *blue mesh*) for  $\alpha$ KG and MLI are contoured at 3.0  $\sigma$  in *A* and *D*.



TABLE 2	1
Data col	lection and refinement statistics

	Apo-BaP4H	Co(II)-BaP4H-MLI	Co(II)-BaP4H-PPG/αKG
Data Collection			
Space group	P 2, 2, 2,	P 1 2, 1	P 2, 2, 2, 2,
Unit cell dimensions	1 1 1	1	1 1 1
a b c (Å)	51.6 80.1 103.8	42.8 41.5 111	50.8 75.6 105.8
$\alpha \beta \gamma (deg)$	90 90 90	90 97 90	90 90 90
Beam line	ALS 4.2.2	APS 19-ID-D	ALS 4.2.2
Wavelength (Å)	1.00	0.979	1.00
Resolution range $(Å)^a$	33.3-1.67 (1.76-1.67)	55.2-1.70	61.5-2.10
		(1.79 - 1.70)	(2.21 - 2.10)
Observations <sup>a</sup>	347.255 (32.419)	155.721 (23.012)	111.271 (15.669)
Unique reflections <sup><math>a</math></sup>	50.418 (6.990)	42.773 (6.257)	24.447 (3.527)
Redundancy <sup>a</sup>	69(46)	36(37)	46(44)
Completeness $(\%)^a$	99.5 (96.5)	99.6 (100)	99.8 (100)
$I/\sigma(I)^a$	27 1 (3 2)	25.5 (6.6)	13.1 (3.0)
$R \qquad (0)^{a,b}$	42(460)	26.0(0.0)	68 (49 6)
$R_{\text{merge}} (\%) = R (\%)^{a,c}$	5.0 (65.6)	37(222)	88 (60 7)
CC CC	1,000 (0,661)	0.999(0.967)	0.998(0.911)
200 <sub>1/2</sub>	1.000 (0.001)	0.555 (0.567)	0.000 (0.011)
Refinement			
$R_{\text{work}}, R_{\text{free}}$ (%)	17.5 (21.1)	21.2 (22.9)	19.7 (23.4)
Resolution range (A)	33.3-1.67	29.7-1.70	61.5-2.10
No. of reflections	50,232	42,742	24,291
Number of atoms			
Protein	3,165	3,004	3,063
Co(II)		2	2
αKG			20
PPG			36
Malonate		14	
Water	396	254	154
Average B-factor (Ų)			
Protein	16.8	20.0	29.0
Co(II)		25.4	19.4
αKG			24.0
PPG			46.2
Malonate		26.3	
Water	33.3	33.0	34.8
R.m.s. deviation bond lengths (Å)	0.010	0.020	0.011
R.m.s. deviation bond angles (degree)	1.2	1.1	1.1
Ramachandran favored (%)	97.5	97.7	97.6
Ramachandran allowed (%)	2.5	2.3	2.4
Clash score	4.5	3.4	3.2

Values in parentheses are the highest-resolution shell.

 $R_{\text{merge}} = \sum_{hkl} |I_i(hkl) - \langle I(hkl) \rangle |\sum_{hkl} Z_i I_i(hkl), \text{ where } I_i(hkl) \text{ is the } i\text{ the measured diffraction intensity and } \langle I(hkl) \rangle \text{ is the mean intensity for the Miller index (hkl).} \\ R_{\text{meas}} = \sum_{hkl} \langle I_i(hkl) - \langle I(hkl) \rangle |Z_{hkl} Z_i I_i(hkl) - \langle I(hkl) \rangle |Z_{hkl} Z_i I_i(hkl).$ 



FIGURE 7. *A*, active site of apo-BaP4H structure with no cofactors bound. *B*, hydrogen bonding interactions of active site residues of apo-BaP4H between protein and water are indicated by *black dashes* with corresponding distances. The metal binding residues (His<sup>127</sup>, Asp<sup>129</sup>, and His<sup>193</sup>) and residues involved in hydrogen bonds are shown as sticks: carbon (protein backbone color), oxygen (*red*), nitrogen (*blue*). The  $2F_o - F_c$  composite omit maps (blue mesh) are contoured at 1.0  $\sigma$  around the active site residues.

 $\beta$ -domain involving strands  $\beta$ I,  $\beta$ II, and  $\beta$ IV. These strands are relatively disordered in the apo- and malonate-bound structures, but become more ordered in the Co(II)-BaP4H-PPG structure. Furthermore, the backbone between residues 122125 in the apo-BaP4H structure is shifted out away from the active site by about 2 Å compared with both cobalt-bound structures (Fig. 8), which indicates that regions of the active site must have greater conformational flexibility.



In the BaP4H structures, we speculate that the residues important for tuning the active site include Tyr<sup>118</sup>, Tyr<sup>124</sup>, and Phe<sup>160</sup>. These residues are located on the "tunable" face of the main  $\beta$ -barrel motif and are around 5–10 Å away from the metal center (Figs. 6, B and E, and 8).  $Tyr^{124}$  exhibits different conformations in the active site. In the apo-form, Tyr<sup>124</sup> is 10 Å (measured from the OH to water located at the metal site) away from the metal site and flipped upwards about 90° from the carbon backbone (Figs. 7 and 8). In Co(II)-BaP4H-PPG, this tyrosine is closer to the active site at  $\sim$  5.3 Å from cobalt and is oriented such that the plane of aromatic ring is parallel to the metal-binding site (Figs. 6B and 8). In Co(II)-BaP4H-MLI, malonate binding induces a conformational change causing movement within key active site residues (Tyr<sup>118</sup>, Tyr<sup>124</sup>, and Phe<sup>160</sup>). The aromatic ring plane of Tyr<sup>124</sup> is oriented perpendicular to the metal binding site and the hydroxyl group is about



FIGURE 8. BaP4H structures highlighting conformational changes involving residues Tyr<sup>118</sup>, Tyr<sup>124</sup>, and Phe<sup>160</sup> upon metal and  $\alpha$ KG or malonate binding. Active site comparison between the Co(II)-BaP4H-PPG "closed" active site (green), Co(II)-BaP4H-MLI "partly open" active site (pink), and apo-BaP4H "fully open" active site (light blue). Cobalt is shown as a magenta sphere.

7 Å away from cobalt (Figs. 6E and 8). The different open and closed conformations of Tyr<sup>124</sup> suggest that it is presumably a catalytically important residue.

Similarly, Tyr<sup>118</sup> has multiple conformations and is important in  $\alpha$ KG and malonate binding. The apo-form has Tyr<sup>118</sup> oriented perpendicular to the  $\alpha$ KG site. In Co(II)-BaP4H-PPG, it has a similar orientation but is shifted down by 1.6 Å to position itself about 2.6 Å from the C5 carboxylate of  $\alpha$ KG. However, in Co(II)-BaP4H-MLI, Tyr<sup>118</sup> is swung outwards away from the active site, at distances of 4.9 and 6.5 Å from malonate C3 and C2 carboxylate ends, respectively. Another residue that is more open in the apo structure is Phe<sup>160</sup>, located in strand  $\beta$ IV. It protrudes out in apo-BaP4H but is oriented more toward the active site when cobalt is bound (Fig. 8).

Docking of (P-P-G)<sub>5</sub> Peptide with Co(II)-BaP4H-PPG Struc*ture*—In the absence of a true peptide bound crystal structure of BaP4H, (P-P-G)<sub>5</sub> was modeled into the active site of the  $\alpha$ KGbound Co(II)-BaP4H complex. The model shows that the (P-P-G)<sub>5</sub> peptide forms an elongated PPII helical conformation similar to that observed in CrP4H (34, 35) (Fig. 9A). The peptide binds BaP4H between \u00b34, \u00b35 (missing loops), and \u00b36 loops and the region between  $\alpha 1$ ,  $\beta I$ , and  $\beta II$ . It interacts with the protein through hydrophobic residues Phe<sup>85</sup> from  $\beta$ 6 and Phe<sup>131</sup> (Fig. 9B). It forms hydrogen bonds with protein residues Glu<sup>111</sup>, Arg<sup>142</sup>, Lys<sup>163</sup>, and Tyr<sup>215</sup> through backbone carbonyl atoms and side chains. A proline from the second repeat (P3) is involved in hydrogen bonding with Arg<sup>142</sup>. The peptide is positioned within the active site of BaP4H with the C4 carbon atom of the Y proline of the second repeat (P4) directed toward the metal center at a distance of 4.6 Å (Fig. 9B), thus indicating that it is a probable hydroxylation site. This model is supported by our mass spectrometry data showing that the Y proline residue in the second repeat of (P-P-G)<sub>5</sub> peptide (P4) is hydroxylated by BaP4H (Table 1).



FIGURE 9. **Model of (P-P-G)**<sub>5</sub> **peptide (***cyan***) into the**  $\alpha$ **KG-bound structure of Co(II)-BaP4H-PPG (***green***).** *A*, surface representation of BaP4H monomer showing the peptide binding groove residues colored in gray and the overall position of (P-P-G)<sub>5</sub>. The peptide is shown as *cyan sticks* with oxygen (*red*) and nitrogen (*blue*). The active site region is colored *orange*. The residues representing the ends of the missing loop are colored *violet*. Cobalt and  $\alpha$ KG are shown as a *magenta sphere* and *yellow sticks*, respectively. *B*, location of the (P-P-G)<sub>5</sub> peptide with potential interacting BaP4H residues (*green sticks*). The C4 position of the Y proline in the second repeat of (P-P-G)<sub>5</sub> is located 4.6 Å away from the metal center.



#### Discussion

The high-resolution MS results demonstrate for the first time that BaP4H can hydroxylate proline residues in both X and Y positions of collagen-like (P-P-G)<sub>5</sub> and (P-P-G)<sub>10</sub> peptides. Animal C-P4H is known to solely hydroxylate proline in the Y position (64-67). Earlier studies have demonstrated that Hyp in the Y position of collagen repeats contributes greater stability to the collagen triple helix through stereoelectronic effects of the inserted oxygen atom (1). Although preferential hydroxylation of a proline in the X position of  $(P-P-G)_{10}$  has been observed with CrP4H-1A (32), At-P4H-1 acts preferentially but not exclusively on proline in the Y position (31). Interestingly, the MS data revealed asymmetric hydroxylation of (P-P-G)<sub>5.10</sub> peptides by BaP4H (Table 1). Both C-P4H (65, 67) and At-P4H-1 (31) displayed an asymmetrical hydroxylation preference in (P-P-G)<sub>5,10</sub> peptides, whereas, CrP4H-1A (32) evenly hydroxylated the Pro residues in (P-P-G)<sub>10</sub>. Furthermore, BaP4H hydroxylates the Y proline in the ninth triplet of (P-P-G)<sub>10</sub> in addition to other sites (Table 1), which is consistent with the hydroxylation pattern observed previously using <sup>14</sup>Clabeled (P-P-G)<sub>10</sub> peptide and either C-P4H (67) or At-P4H-1 (31). With (P-P-G)<sub>5</sub>, BaP4H catalyzes the hydroxylation of at least one proline residue in the triplets, with the exception of the first repeat (Table 1). Hydroxylation of both X and Y proline residues were identified, which is quite distinct from the hydroxylation pattern of (P-P-G)<sub>5</sub> by C-P4H (65). Using <sup>14</sup>C-labeled (P-P-G)<sub>5</sub>, previous studies have shown that C-P4H preferentially hydroxylates the Y proline in the fourth triplet of  $(P-P-G)_5$  (65).

Docking studies with (P-P-G)<sub>5</sub> illustrates that the peptide is located over the active site with the Y proline residue of the second repeat (P4) in proximity of the metal (Fig. 9). The C-4 carbon atom of this proline is positioned optimally to the metal center for C-H bond activation and subsequent hydroxylation, via a reactive Fe(IV)-oxo intermediate. Although modeling displayed P4 in the proper orientation for hydroxylation, alternatives with prolines from other repeat units of the peptide in correct conformation for hydroxylation are feasible. The MS data revealed that proline residues of the second, fourth, and fifth repeats of the (P-P-G)<sub>5</sub> peptide are indeed hydroxylated by BaP4H (Table 1).

Based on the crystal structures of BaP4H, the surface contains a groove across the active site comprised of aromatic and other conserved residues found in Src homology 3 domain proteins known to recognize proline-rich motifs (68). The Src homology 3 domain-like proteins have been reported to occur in prokaryotes (23). Based on structural studies with many protein families, the mechanism of proline-rich peptide recognition is converging, which suggests that a groove composed of aromatic residues is responsible for proline recognition (68). However, the specificity toward a target is mostly dependent on variable loops and neighboring domains (68). In BaP4H, the peptide binding groove along with the SBL most likely contribute to the observed tighter binding affinities for both  $(P-P-G)_{10}$  (K<sub>d</sub>, 0.7  $\mu$ M) and (P-P-G)<sub>5</sub> ( $K_d$ , 0.3  $\mu$ M) peptides. These  $K_d$  values are comparable with the reported  $K_m$  for human C-P4H (20 and 170  $\mu$ M for (P-P-G)<sub>10</sub> and (P-P-G)<sub>5</sub>, respectively (22)), but sig-

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nificantly lower than the  $K_m$  determined for algal CrP4H (>1.5 mM for (P-P-G)<sub>10</sub> (32)) and vCPH (2.9 mM for (P-P-G)<sub>10</sub> (29, 30)). Although the binding affinities of BaP4H for both (P-P-G)<sub>5,10</sub> peptides are similar, it can catalyze the hydroxylation of (P-P-G)<sub>5</sub> with relatively greater ease based on hydroxylated peptides observed in the MS results. It is possible that the longer (P-P-G)<sub>10</sub> is more flexible and may not be positioned in the active site in a favorable conformation for hydroxylation. We have also established that BaP4H binds free L-proline but does not catalyze the formation of Hyp. Although there are distinct bacterial P4Hs known to hydroxylate free L-proline (1, 69), it is unclear why BaP4H does not recognize it for hydroxylation. In the absence of a proline- or, a true peptide-bound crystal structure, no additional predictions can be made.

Currently, the structures of only five P4H enzymes have been solved. These are HIF $\alpha$ -PHD2 (70), *Pseudomonas aeruginosa* PHD (PPHD) (37), CrP4H (34, 35), vCPH (30), and apo-BaP4H (36). The crystal structures of BaP4H with bound cofactors display structural features characteristic of Fe(II)/ $\alpha$ KG-dependent enzymes consisting of the 2His-1carboxylate iron-binding motif. Based on sequence and secondary structural similarities with CrP4H, we expect that three  $\beta$ -strands ( $\beta$ 3,  $\beta$ 4, and  $\beta$ 5), including the disordered residues in BaP4H (Fig. 1), are likely involved in substrate recognition and we hypothesize that these residues will become ordered upon peptide binding as observed in CrP4H (35).

The structures of BaP4H provide insight into active site reorganization upon binding of cofactors and the inhibitor, malonate. The binding of cobalt and  $\alpha$ KG induces a conformational change bringing the active site from an open state (apo-BaP4H) to a more closed form required for substrate and dioxygen binding prior to catalysis (Fig. 8). It is evident that due to the movement of various important residues, BaP4H can adopt alternate conformations when bound to different ligands. The residue that undergoes the largest movement is Tyr<sup>124</sup>. Although Tyr<sup>124</sup> does not directly interact with  $\alpha$ KG, it serves as a lid to keep the active site closed when  $\alpha$ KG is present. This tyrosine is conserved across P4Hs and similar movement has been observed in both vCPH and CrP4H (30, 34, 35). It was proposed that this tyrosine could serve as a "conformational switch" to modulate loop movement and substrate binding or release, which is further supported in the BaP4H structures (30, 34, 35). Thus the Co(II)-BaP4H-PPG structure with  $\alpha$ KG bound represents the ready or pre-catalytic state of the enzyme.

In BaP4H  $\alpha$ KG coordinates cobalt in the "in line" binding mode observed in many Fe(II)/ $\alpha$ KG-dependent dioxygenases (2), with the C1 carboxylate opposite of His<sup>127</sup> and carbonyl oxygen opposite of Asp<sup>129</sup>, indicating that the substrate will be positioned near the open site trans to His<sup>193</sup> for productive catalysis (Figs. 6, 8, and 9). The  $\alpha$ KG binding site in BaP4H uses similar residues that are conserved in the Fe(II)/ $\alpha$ KG-dependent enzymes, including other P4Hs. Although BaP4H has two less residues involved in  $\alpha$ KG binding compared with vCPH, it is unclear as to why it has a much higher binding affinity for  $\alpha$ KG ( $K_d \sim 1 \mu$ M) (71) compared with vCPH ( $K_d \sim 700 \mu$ M) (30). PHD2 on the other hand has an affinity for  $\alpha$ KG ( $K_d < 2 \mu$ M) similar to that determined for BaP4H (72). It was proposed that the weaker binding of  $\alpha$ KG in vCPH compared with PHD2



could be due to a difference in the basic residue utilized in binding to the C5 carboxylate of  $\alpha$ KG (Lys<sup>231</sup> in vCPH *versus* Arg<sup>383</sup> in PHD2 (30). However, BaP4H has Lys<sup>203</sup> that interacts with the C5 carboxylate, yet has a high binding affinity toward  $\alpha$ KG.

Although malonate is a known P4H inhibitor (73), to the best of our knowledge, this is the first structure of a  $\alpha$ KG-dependent dioxygenase with malonate bound. The open and solvent accessible active site of Co(II)-bound BaP4H can accommodate malonate from the crystallization conditions. Malonate occupies the  $\alpha$ KG binding space in the active site and mimics binding of the  $\alpha$ KG decarboxylation product succinate. The malonate bound form of BaP4H appears to be more open than the  $\alpha$ KG-BaP4H-PPG bound form because it simulates a product state of the enzyme post-hydroxylation, where an open form is necessary for product release. The C1 carboxylate of malonate is coordinated to the metal trans to Asp<sup>129</sup> (Figs. 6 and 8). The binding mode is similar for this family of enzymes in that the keto group of  $\alpha$ KG always binds trans to the carboxylate residue in the catalytic triad (2). Even with two fewer residues responsible for malonate binding to BaP4H compared with  $\alpha$ KG, malonate still binds BaP4H with high affinity ( $K_d \sim 4 \mu M$ ) (Fig. 2E). The slightly lower binding affinity of malonate with BaP4H compared with  $\alpha$ KG ( $K_d \sim 1 \mu$ M) is somewhat surprising. However, this could lend some evidence that similar metabolites, such as succinate, could provide a feedback type of inhibition to BaP4H. It has been suggested previously that concentrations of  $\alpha$ KG could serve as an indicator of the metabolic status of the cell and that the Fe(II)/ $\alpha$ KG dependent family of enzymes could possess a wider role in regulation of cellular processes (74, 75). This leads to the conclusion that there does not appear to be an obvious trend in analyzing binding affinities of aKG for different enzymes. A more complex combination of local cellular environment and structural dynamics is most likely involved.

In conclusion, the results described here uncovered that BaP4H can target proline-rich human collagen-like peptides for site-specific hydroxylation. The BaP4H structure-function studies have broader implications in deciphering molecular mechanisms that regulate the PSB domains in P4Hs and PPII substrate binding enzymes, including substrate specificity and the overall reaction mechanisms. Further studies to determine the x-ray crystal structure of active Fe(II)-BaP4H with a peptide substrate bound are underway, in addition to site-directed mutagenesis to confirm the catalytic residues, which altogether would allow the identification of the molecular determinants of substrate binding and site-specific catalysis.

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