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Chlorine-induced cardiopulmonary injury

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Abstract

Chlorine (Cl₂) is utilized worldwide for a diverse range of industrial applications, including pulp bleaching, sanitation, and pharmaceutical development. Though Cl₂ has widespread use, little is known regarding the mechanisms of toxicity associated with Cl₂ exposure, which occurs during industrial accidents or acts of terrorism. Previous instances of Cl₂ exposure have led to reported episodes of respiratory distress that result in high morbidity and mortality. Furthermore, studies suggest that acute Cl₂ exposure also results in systemic vascular injury and subsequent myocardial contractile dysfunction. Here we review both lung and cardiac pathology associated with acute Cl₂ inhalation and discuss recently published data that suggests that mitochondrial dysfunction underlies the pathogenesis of Cl₂-induced toxicity. Lastly, we discuss our findings that suggest that upregulation of autophagy protects against Cl₂-induced lung inflammation and can be a potential therapeutic target for ameliorating the toxic effects of Cl₂ exposure.

Keywords

chlorine; mitochondrial dysfunction; autophagy; lung injury; cardiac injury

Chlorine as an occupational and public health hazard

Chlorine (Cl₂), a common toxic inhalant, is an essential chemical widely used in numerous industrial applications, including plastics manufacturing, waste sanitation, water treatment, and pharmaceutical development. Cl₂ is a water-soluble yellow-green gas that commonly irritates the eyes, integumentary, and respiratory systems following exposure.¹ Like any

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toxic inhalant, extent of injury is limited primarily by duration of exposure and intensity of dose. Most injuries are localized to sites in direct contact with the gas.² Skin exposure to Cl₂ can result in chemical burns and cell death resulting in the generation of dermal lesions.³ The Cl₂ Institute in Arlington, Virginia, estimates that approximately 65 million tons of Cl₂ are manufactured yearly worldwide, with the United States alone producing 13.8 million tons. With such widespread production and use, Cl₂ exposure has the potential to be a significant global public health threat. An estimated 9000 calls occur annually to U.S. poison control centers for Cl₂-related exposures.⁴

Wide-scale Cl₂ exposures can occur in a number of scenarios, from accidental spills and occupational exposures to chemical warfare and acts of terrorism. Cl₂ accidents have resulted in thousands of individuals being exposed to the toxic gas. Over the last 20 years, there have been 30 reported cases of large-scale accidental Cl₂ release into urban centers. On January 6, 2005, a train derailment and subsequent 54,422-kg (120,000-lb) chlorine gas spill occurred in the cotton mill town of Graniteville, South Carolina.⁵⁻⁹ Several hundred people became immediately sick and thousands more were exposed.¹⁰⁻¹² This is the largest civilian U.S. population ever exposed to chlorine gas.¹³ Data has been aggregated from 1979 to 2006 for over 8000 Graniteville millworkers who have at least 3 years of pre-event spirometry and related health and covariate assessment.¹⁴ Individual chlorine exposures from the event have been estimated using validated plume models.^{13,15,16} Concentrations of Cl₂ upon the initial release of the gas reached up to 500 ppm, a dose above the lethal concentration of Cl₂ in many animal studies.^{1,17} In addition, adverse environmental damages from the accident were extensive, as animal deaths and bleaching of vegetation occurred up to 1 km from the site of the accident.¹⁷⁻¹⁹

Other notable Cl₂ exposure incidents include an industrial accident at a hazardous waste facility in Apex, North Carolina in 2006 and a Cl₂ delivery-system malfunction in a Sacramento, California waterpark in 2011, both of which led to numerous hospitalizations for severe respiratory complaints.²⁰ Even more recently, a British military nurse developed respiratory distress and reactive airway disease following Cl₂ gas inhalation during routine procedures for disinfecting an Ebola treatment facility in Sierra Leone.⁶ These events highlight the hazards that Cl₂ poses as both an accidental and occupational exposure risk.

More troublingly, Cl₂ facilities are also implicated as potential terrorist targets. The U.S. Department of Homeland Security considers Cl₂ production and storage facilities as potential high-risk targets that, according to the projections, could result in thousands of fatalities and up to 100,000 hospitalizations (The Homeland Security Council. Planning Scenarios: Executive Summaries. 004; 8-1). The use of Cl₂ as a weapon is not a novel concept. During World War I, German forces would wait for favorable meteorological conditions before opening cylinders of Cl₂ to surprise unsuspecting forces caught downwind of the suffocating greenish-yellow cloud.¹ In 2007, insurgents in the Iraqi Conflict adopted this chemical warfare agent to terrorize regions of conflict by detonating explosives containing Cl₂ gas in large urban centers.²¹

There is no specific antidote for individuals exposed to Cl₂. Standards for treatment following Cl₂ exposure are principally symptomatic, focusing on ameliorating conditions

that potentially develop, such as pulmonary edema and upper and lower airway obstruction.^{1,2} Postexposure therapeutic management includes oxygen administration and treatment of bronchospasms with beta agonists.²² In cases of severe respiratory distress, intubation may be required. Corticosteroids have been shown to ameliorate acute respiratory distress following Cl₂ exposure in animal studies, but their efficacy in improving respiratory function following human exposures has yet to be established.²³

Lung injury after Cl₂ exposure

In the year of the Graniteville accident, a significant reduction in mean forced expiratory volume in 1 second (FEV1) (−4.2% predicted, $P=0.019$) was recorded when compared to the year before the incident. In the second year, partial recovery in the mean forced vital capacity (FVC) percent predicted level was seen, but the cohorts' average FEV1/FVC percent predicted level continued to decrease over time. Severe annual FEV1 decline was most prevalent in the year of the accident and independent of millworker smoking status.¹¹

The toxic effects of Cl₂ are most related to its oxidant potential. Upon inhalation, Cl₂ first reacts with antioxidants in lung epithelial lining fluid (ELF).²⁴ Once antioxidant stores are depleted, soluble Cl₂ rapidly undergoes hydrolysis to generate hypochlorous (HOCl) and hydrochloric acids (HCl).¹ These Cl₂ products then react with protein side chains, DNA, and lipids of the cells that line the airway epithelium to generate toxic reactants.^{24–26} Some of these reactants formed are long lasting, contributing to injury long after the duration of the initial exposure event. Chloramines generated by Cl₂ reactions with proteins activate inflammatory cascades through the mitogen-activated protein kinase (MAP kinase)²⁷ and the nuclear factor κ light chain enhancer of activated B cells (NK- κ B) pathways,²⁸ resulting in inflammatory cell infiltration into the alveolar space. Furthermore, chloramines may predispose individuals to pulmonary edema by damaging epithelial sodium channel activity and disrupting fluid clearance in the lung.²⁹ In addition, chlorinated lipids (sterols, fatty acids, and phospholipid chlorohydrines generated from reactions with Cl₂ or HOCl) are considered proinflammatory.³⁰

Along the respiratory tract, Cl₂ distribution is primarily influenced by the concentration of the gas. It has been noted that, at low concentrations, Cl₂ injury is predominantly limited to nasal and tracheal injury.^{31, 32} On the basis of diffusion modeling, at low Cl₂ concentrations, up to 90% of Cl₂ is absorbed within the nasal cavity and nasopharynx, with less than 10% reaching the hypopharynx or beyond.³¹ However, at higher concentrations, Cl₂ not only damages the upper airways, but also penetrates deeper and damages the lower respiratory system.^{31,33}

The dose of Cl₂ also determines physiologic symptoms, responses, and consequences. Mild nasal irritation and throat irritation can occur at levels as low as 0.014–0.04 ppm, and ocular irritation, dyspnea, and headaches can occur at levels of 1 ppm.² Acute exposure to doses as low as 1ppm causes sensory irritation,³⁴ while chronic exposure to even lower doses (0.1–0.4ppm) can evoke ocular irritation and airway epithelial degeneration.^{32,35} These irritating effects were shown to be primarily mediated by the activation of transient receptor potential ankyrin 1 (TRPA1) ion channels, which interact with airway neurons to mediate

neurological and airway responses to inhaled irritants, such as the cough reflex.^{36,37} Other clinical effects at low concentrations include lung inflammation, reversible bronchospasm, bradypnea, and dyspnea.^{38–41}

At concentrations greater than 50 ppm, chemical pneumonitis develops.¹ Animal studies have shed light into the pathologic sequence of injury following Cl₂ exposure at higher concentrations: acute exposure to high doses of Cl₂ induces bronchial epithelial sloughing, followed by a period of interstitial edema and infiltration of immunocompetent cells into the airways.⁴² These pathological events result in the development of pneumonitis, necrosis, and pulmonary edema.^{24,32} Uncorrected injury can potentially progress to respiratory failure and death.⁴³ Mortality can occur at exposure to concentrations of 430 ppm for 30 min or more or concentrations greater than 1000 ppm for a few minutes.^{1,17}

Another common sequel of Cl₂ exposures in humans is reactive airway disease syndrome (RADS), a disorder characterized by wheezing, airflow obstruction, air trapping, and other asthma-like symptoms that persist beyond the typical recovery period for an inhaled toxic irritant. Not all individuals that are exposed to Cl₂ develop RADS, and the phenotypic and genotypic risk factors that predispose individuals to its pathogenesis are still poorly understood.⁴⁴ Although initially postulated to be secondary to only high-level inhalation exposures,⁴⁵ later case reports demonstrated that low levels of Cl₂ can also induce long-term reactive changes in the airway.⁴⁴ These findings of reactive airway disease syndrome were recapitulated in animal studies: animals that survive the initial Cl₂ insult often develop severe airway hyperreactivity, elevated airway resistance to methacholine challenge, and mucous metaplasia following exposure.^{38, 46}

Evidence also suggests that Cl₂ exposure increases susceptibilities to pulmonary infections. Following the Graniteville derailment, the South Carolina Department of Health and Environmental Control sent out a health alert message to all healthcare providers within the South Carolina health alert network within 24 h of the Graniteville disaster. This message alerted medical providers to the risks of secondary pneumonia infection within patients treated for chlorine injury and recommended that all hospitalized patients receive a pneumococcal vaccine before hospital discharge.

Following Cl₂ exposure, mucous metaplasia results in an increase in mucous-producing goblet cells and a subsequent decrease in secretory serous cells in the conducting airways. Overproduction of mucous impairs respiratory clearance of pathogens and leads to airway obstruction and atelectasis. In addition, lower numbers of serous cells results in decreased production of their antimicrobial granules, lactoferrin and lysozyme, which predisposes to infection.^{42,47} Furthermore, HOCl reacts with amino acids in carbohydrate-recognition domains of surfactant protein (SP)-A, thereby decreasing its ability to bind and kill pathogens.⁴⁸ HOCl also disrupts disulfide crosslinking of SP-D, inhibiting its pathogen-aggregating activity following infectious challenge.⁴⁹ Finally, studies have demonstrated that Cl₂ impairs antimicrobial superoxide generation and inflammatory production of IL-17A and IL-22 (both critical cytokines in inducing the epithelial antimicrobial response) of myeloid cells in the lung, thereby increasing susceptibility to certain pulmonary fungal infections by 500 fold.⁵⁰

Cardiovascular effects of Cl₂ toxicity

There have been some isolated reports in humans of systemic injury following gas exposure targeting the brain, liver, and heart. Specifically, white matter hemorrhages and liver transaminitis and cardiomegaly have been observed in cases of high-dose Cl₂ exposure.^{17,53,54} In animal models, Cl₂ exposure has also been shown to injure systemic vasculature and impair vascular function. Rats exposed to Cl₂ (250–400 ppm) for 30 min had significantly decreased endothelial nitric oxide synthase (eNOS) protein expression. These animals also displayed marked attenuation in eNOS-dependent vasodilation (stimulated by acetylcholine) at 24–48 h after exposure, suggesting that NO dysregulation may underlie the pathophysiology of Cl₂ inhalation-induced systemic endothelial dysfunction.⁵⁵ However, results also demonstrated that during the initial stages of Cl₂ induced injury, inflammatory, cell-derived, inducible NOS (iNOS) may compensate for the loss of eNOS and prevent spike in the mean blood pressure. However, when iNOS was inhibited, significantly higher systemic blood pressures were recorded.⁵⁵

Cardiotoxicity is another important risk factor for more severe immediate outcomes from Cl₂ injury. In rats, inhalation of 500 ppm Cl₂ for 30 min increases lactate in the coronary sinus, suggesting an increase in anaerobic metabolism by the heart. Cl₂ inhalation also attenuates myocardial contractile force, reduces systolic and diastolic blood pressures, and causes biventricular failure and death.⁵⁶ These results can be reproduced by sole exposure to chloramine (a potential circulating Cl₂ reactant product), suggesting an independent and distinctive role of Cl₂ (and its reactants) in inducing cardiac toxicity.²⁶

Interestingly, cardiomegaly was observed on autopsy in eight of nine immediate victims of the acute chlorine release from the Graniteville, South Carolina train derailment in January 2005.¹⁹ Upon further examination, the one victim who died in the hospital was found to have cardiomegaly by chest X-ray in the hospital, later confirmed by autopsy. Similarly, four of the 71 victims hospitalized were found to have had cardiomegaly by chest X-ray. Furthermore, of those hospitalized, 64% presented with tachycardia and 52% with hypertension in the emergency department. Before the Graniteville chlorine disaster, there was little recognition of any potential cardiovascular effects from chlorine exposure in humans. There was some evidence that exposure to high levels of chlorine gas may result in vascular injury and the depression of cardiac function in humans and rats,^{56,57} which could exacerbate existing conditions leading to stroke.^{56–58} But there had not been any evidence of such long-term cardiovascular effects in a human cohort study.

The Graniteville cohort,⁵⁹ which now has 305 patients enrolled, is the only systematic long-term longitudinal cohort study of a population exposed to a single high-concentration chlorine gas event with pre–post individual exposure and health comparisons and prospective longitudinal lung function assessment. In the cohort, it was observed that the immediate lung function loss between 2004 and 2005¹⁴ had not fully recovered at 7–9 years postevent for some individuals. Furthermore, surprisingly, 12% of the participants were medically ineligible for spirometry testing owing to their high blood pressure even after repeated attempts to medically control it, and the modeled chlorine dose was significantly associated with their increased blood pressure ($P < 0.05$) 7–9 years later. When studying the

rates of hospitalizations for cardiovascular diseases within the larger Graniteville population, hospitalizations for high blood pressure doubled from 110 per 10,000 residents in 2005 to 220 hospitalizations per 10,000 residents in 2012. This data indicates that the cardiovascular effects of chlorine exposure may last for many years.

Mitochondrial bioenergetic dysfunction in Cl₂-induced injury

Although Cl₂ exposure results in local and systemic injury with both acute and chronic consequences, effective therapeutics to ameliorate Cl₂ toxicity are lacking. Further dissection of the pathophysiologic mechanisms of Cl₂-induced injury can direct the generation of novel therapeutics. Mitochondria play a significant role in determining cell survival and apoptosis following a stress response.⁶⁰ Mitochondria are also an important source of reactive oxygen species that may be elevated after cell injury. Interestingly, these mitochondrial oxidants can further propagate cell injury and disrupt other mitochondria that were not affected by the initial injury, resulting in an indirect perpetuation of ongoing inflammation and cell death.^{61–63} This results in both an acute and delayed deleterious effect from Cl₂, even with sublethal Cl₂ exposures.

In our recent study, we found that mitochondrial function was impaired after Cl₂ exposure.⁶⁴ We exposed H441 (human airway Clara cell-like cells) cells to 100 ppm Cl₂ for 15 min, a dose and duration similar to those found in documented Cl₂ disasters. Cl₂ exposure decreased the maximal mitochondrial oxygen consumption rate (OCR) and the bioenergetic reserve capacity, an index of cellular energy for repair and resistance to oxidative damage.⁶⁵ Cl₂ also increased non-mitochondrial OCR, which represents an increased capacity to generate reactive oxygen species.

Interestingly, Cl₂ exposure increased superoxide production in the mitochondria of Clara cells⁶⁴ and primary cultures of alveolar type II cells.⁶⁶ Further, we also demonstrated that mitochondrial oxidative stress played a significant role in the inhibition of cellular bioenergetics.⁶⁴ Treatment of Cl₂-exposed H441 cells with MitoQ, a mitochondrial-targeted antioxidant, partially prevented the Cl₂-dependent decrease in maximal OCR. Exposure to Cl₂ also decreased the mitochondrial membrane potential, which was prevented by MitoQ pretreatment.⁶⁴ Surprisingly, exposure to Cl₂ also resulted in a significant decrease in the maximal extracellular acidification rate (ECAR), an indicator of glycolysis,⁶⁷ suggesting that glycolysis may also be impaired in these cells. Again, treatment with MitoQ attenuated this decrease in maximal ECAR to near normal levels. Our study also found that Cl₂ specifically inhibits complex I and II of the mitochondrial electron transport chain.⁶⁴

The mitochondrial injury also seems to be an important mediator of Cl₂-induced cardiac toxicity. In rats, (500 ppm for 30 min) inhalation decreased ATP content in heart tissue and primary cardiomyocytes, resulting in increased anaerobic metabolism, as evidenced by the accumulation of lactate in the coronary sinus.⁶⁸ Ahmad *et al.* attributed this impairment of the mitochondrial function to the formation of circulating Cl₂ reactants and chlorination and inactivation of cardiac sarcoendoplasmic Ca²⁺ ATPase (SERCA) and subsequent increase in cytosolic Ca²⁺ overload. Excessive cytosolic Ca²⁺ causes increased production of reactive oxygen species by the mitochondria.^{69,70} In addition, the mitochondrial reactive species can

itself perturb the cytosolic Ca^{2+} homeostasis and cause cardiac dysfunction.^{71,72} SERCA regulates cardiac intracellular Ca^{2+} homeostasis by transporting cytosolic Ca^{2+} into the sarco/endoplasmic reticulum at diastole and thus lowering intracellular Ca^{2+} levels.^{68,73} SERCA can be readily oxidized and its activity impaired by reactive species like hypochlorous acid (HOCl), a product of chlorine and water.^{68, 73} Interestingly, the administration of the SERCA stabilizer ranolazine or the SERCA activator istaroxime prevented chlorine-induced cardiomyocyte death and preserved mitochondrial membrane potential and ATP after chlorine exposure.⁶⁸

The role of autophagy in Cl_2 -induced injury

Autophagy is generally an inducible adaptive response to lung injury. During autophagy, damaged organelles or denatured proteins are disposed through the lysosomal degradation pathway.⁷⁴ Autophagic degradation of dysfunctional or damaged mitochondria is termed mitophagy.⁷⁴ Mitochondria that are damaged by reactive oxygen species are typically removed by the lysosomal–mitophagy systems in order to maintain adequate mitochondrial function and control.^{60,75,76} During a minimal stress response, the lysosomal–autophagy system is capable of allowing mitophagy of dysfunctional mitochondria; however, if a more significant insult occurs, this system cannot appropriately maintain the large amount of mitochondria damaged by the reactive species.

The ubiquitin-like conjugation protein microtubule-associated protein 1, light chain 3B (LC3B) is a principal mediator of autophagy. In mammals, the conversion of LC3BI (unconjugated cytosolic form) to LC3BII (autophagosomal membrane–associated phosphatidylethanolamine-conjugated form) is a hallmark of autophagosome formation.^{50, 51} In addition, the nucleoporin p62 protein is involved in recognizing toxic cellular waste and targeting it for autophagy.⁷⁷ Decreased levels of p62 are associated with induction of autophagy.⁷⁸ Interestingly, in our studies we found that LC3BII levels were increased, while the levels of p62 protein were decreased, 6 h after Cl_2 exposure in H441 cells. These results suggested that Cl_2 exposure leads to an increase in autophagy, presumably to clear damaged mitochondria.

In addition, we explored whether the modulation of autophagy would affect mitochondrial bioenergetics following Cl_2 inhalation. For this, we pretreated human club cells with either trehalose, an activator of autophagy, or 3-MA (3-methyladenine), an inhibitor of autophagy, before exposure to Cl_2 . Preincubation of cells with trehalose resulted in an increase in LC3BII in both air- and Cl_2 -exposed cells, associated with elevated levels of autophagy. More interestingly, trehalose attenuated the decrease in maximal OCR of cells following Cl_2 exposure, suggesting an improvement of mitochondrial respiration function. Conversely, 3-MA pretreatment resulted in further bioenergetic dysfunction of Cl_2 -exposed cells.⁶⁴

We also studied the effects of autophagy upregulation on lung inflammatory responses *in vivo* after Cl_2 -inhalation injury. Acute pretreatment of mice with trehalose via aerosolization increased lung LC3BII levels after Cl_2 (400 ppm, 30 min) exposure, which correlated with decreased lung permeability and airway protein leakage into bronchoalveolar lavage fluid.

Chronic treatment of mice with trehalose 6 weeks before Cl₂ exposure resulted in a decrease in lung inflammatory cells and improved tissue healing.⁶⁴

Conclusions

In conclusion, given the widespread use and easy attainability of Cl₂, exposure to the toxic gas poses a major health threat that currently has few available therapeutic treatments. Exposure to Cl₂ gas results in both local and systemic responses, resulting in high morbidity and mortality secondary to both cardiac and pulmonary dysfunction. Our studies show that an increase in mitochondrial ROS and subsequent dysfunction in mitochondrial bioenergetics are hallmarks of Cl₂-inhalation injury. Furthermore, preventing oxidative damage of SERCA or stabilizing the SERCA activity can prevent Ca²⁺ overload, mitochondrial dysfunction, and cardiac damage. In addition, upregulating mitochondrial autophagy pathways in damaged cells can likely ameliorate mitochondrial bioenergetics dysfunction and subsequent pulmonary damage resulting from acute Cl₂ exposure (Fig. 1). Trehalose administration represents a novel treatment method that improves autophagy of dysfunctional mitochondria and can possibly decrease the deleterious effects of Cl₂-inhalational injury. It is important to note that immediate treatment after Cl₂ exposure may play significant role in mitigating long-term cardiopulmonary pathologies associated with chronic Cl₂ toxicity. However, further studies are required to assess the long-term beneficial effects of mitigating mitochondrial dysfunction after Cl₂ exposure.

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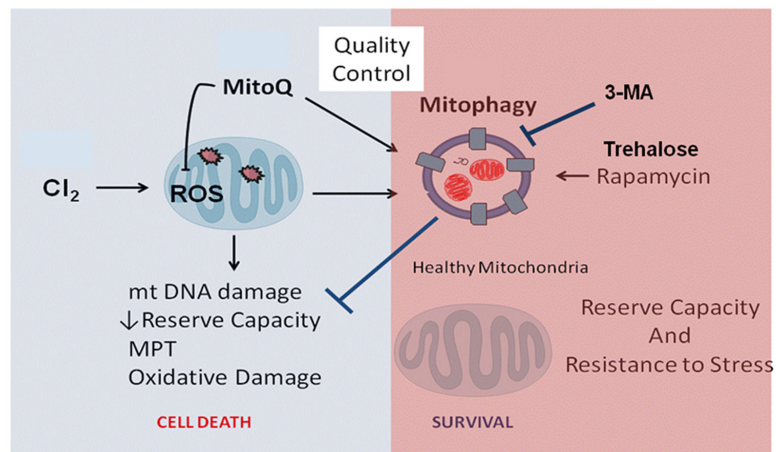


Figure 1.

Role of autophagy on chlorine (Cl_2)-induced mitochondrial dysfunction. Mitochondria are the key regulators of cell survival in response to Cl_2 -induced cellular stress. Healthy mitochondria produce reactive oxygen species (ROS) that serve a cell signaling function; however, damaged mitochondria produce excessive amounts of ROS and become a major contributor of cellular and organ injury leading to mitochondrial DNA damage, formation of mitochondrial permeability transition (MPT) pores, and cell death. Mitochondria modified by ROS are targeted for removal by mitophagy, a critical step in maintaining mitochondrial quality control and limit cellular and organ injury. Upregulation of mitophagy by rapamycin or trehalose prevents mitochondrial oxidative damage and is beneficial in ameliorating Cl_2 toxicity. Alternatively, decreasing mitophagy with 3-methyladenine (3-MA) enhances Cl_2 -induced injury.