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# Phenolic-rich extract from the Costa Rican guava (*Psidium friedrichsthalianum*) pulp with antioxidant and anti-inflammatory activity. Potential for COPD therapy

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#### Abstract

The potential therapeutic effects of Costa Rican guava (*Psidium friedrichsthalianum*) extracts for chronic obstructive pulmonary disease were examined. The ethyl acetate fraction displayed the highest antioxidant activity, as compared to the hexane, chloroform, and *n*-butanol fractions, as well as the crude extract. This fraction was evaluated for its anti-inflammatory activity response relationship against interleukin-8 (IL-8) and inhibition of matrix metalloproteinase-1 (MMP-1) expression before and after treatment with cigarette smoke. The ethyl acetate fraction exhibited inhibitory activity against IL-8 production and MMP-1 expression, showing the most potent inhibitory activities in both assays at 100 µg/mL, and nine compounds (**1**–**9**) were found. Phenolic compounds 1-*O*-*trans*-cinnamoyl- $\beta$ -D-glucopyranose (**2**), ellagic acid (**3**), myricetin (**4**), quercitrin (**7**), and quercetin (**9**) were identified using standard compounds or literature reports from related species. Compounds **1**, **5**, **6**, and **8** were tentatively identified as 1,5-dimethyl citrate (**1**), sinapic aldehyde 4-*O*- $\beta$ -D-glucopyranose (**5**), 3,3',4-tri-*O*-methylellagic acid-4'-*O*-D-glucopyranoside (**6**), and 1,3-*O*-diferuloylglycerol (**8**), All nine compounds are reported for the first time in Costa Rican guava.

#### Keywords

Costa Rican guava; *Psidium friedrichsthalianum*; Antioxidants; Interleukin-8 (IL-8); Matrix metalloproteinase-1 (MMP-1); Chronic obstructive pulmonary disease (COPD)

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#### 1. Introduction

The genus *Psidium* consists of approximately 150 species, but only about twenty produce commonly eaten fruits. The most widely cultivated species is the common guava (*Psidium guajava* L.), and other cultivated species include strawberry guava (*Psidium cattleianum* Sabine), the Brazilian guava (*Psidium guineense* Sw.), and Costa Rican guava (*Psidium friedrichsthalianum* Ndz.) (Mani, Mishra, & Thomas, 2011).

The Costa Rican guava is a small tree with small and sour fruits native to the seasonally flooded forest of Central America, from south Mexico to northern South America (Bailey, 1941; Dinesh & Iyer, 2005). The fruits are consumed as juices and also made into sweets and jellies; they are often described as being more aromatic than the common guava (Pino, Marbot, & Vazquez, 2002). A large number of terpenes and terpenic derivatives have been identified in Costa Rican guava (Pino et al., 2002), as well as in a variety of other guava species (Pino, Marbot, & Vazquez, 2001; Pino, Ortega, & Rosado, 1999).

Common guavas (*P. guajava*) are often included among super-fruits (Sanda, Grema, Geidman, & Bukar-Kolo, 2011). A review focused on guavas reported that the leaves and fruits of this plant showed anti-oxidant, anti-inflammatory, antimicrobial, antispasmodic, hepatoprotective, anti-allergy, antigenotoxic, antiplasmodial, antidiabetic, cardioactive, anti-cough, and anticancer effects (Gutierrez, Mitchell, & Solis, 2008). We hypothesise that as a closely related edible *Psidium*, Costa Rican guava shares some of these useful pharmacological properties. However, no reports have examined the biological properties and the composition of this guava species.

Chronic obstructive pulmonary disease (COPD), a major disease which causes death and disability, is expected to be the third leading cause of death worldwide by 2020 (Murray & Lopez, 1997). Cigarette smoke is the main aetiological factor associated with the development of COPD, for which there is no cure. Currently, little progress has been made toward developing effective therapies for COPD, and although treatments can improve symptoms, their effects are limited and they have not reduced disease progression (Calverley et al., 2007; Vestbo et al., 1999).

The inflammatory reaction to cigarette smoke is crucial in the pathogenic mechanisms of COPD (Repine, Bast, & Lankhorst, 1997). The high concentration of oxidant molecules in cigarette smoke, in addition to the oxidants endogenously formed by inflammatory cells (macrophages and neutrophils), overcome the capacity of the antioxidant protective physiological mechanisms and induce oxidative stress (Cross, Van der Vliet, O'Neill, Louie, & Halliwell, 1994; Frei, Forte, Ames, & Cross, 1991). This oxidative stress may result in direct damage to structural cells, amplification of inflammation, and promotion of proteolytic degradation of tissues by inhibiting antiprotease systems. *In vitro* studies of human alveolar macrophages have shown that oxidative stress caused by acute cigarette smoke extract (CSE) exposure increases the release of interleukin-8 (IL-8) (Walters et al., 2005). Matrix metalloproteinases (MMPs), a family of zinc endo-peptidases, may also play an important role in COPD pathology (Babusyte et al., 2007; Haq et al., 2010). Particularly it has been described that MMP-1 is expressed in the lung of human patients with

emphysema but not in normal control subjects (Imai et al., 2001). Kim et al. (2004) demonstrated that cigarette smoke stimulates MMP-1 production by human lung fibroblasts through the extracellular-signal-regulated kinases (ERK)1/2 pathway.

Therefore, it is anticipated that drugs that inhibit MMP-1 expression, reduce pulmonary inflammation and decrease oxidative stress in the lungs of patients with COPD will provide effective disease therapies.

Several epidemiological studies have established a beneficial link between phenolic compounds intake and reduced risk of disease, which were attributed to both their antioxidant and anti-inflammatory properties (Arts & Hollman, 2005). A significant inverse correlation between phenolic compounds intake and the incidence of COPD has been reported in a study with over 13,000 subjects (Tabak, Arts, Smit, Heederik, & Kromhout, 2001). It was reported that increased phenolic compounds, such as catechin, flavonol and flavone, intake can improve symptoms such as phlegm production, cough, and breathlessness; lung function also can improved by measured by forced expiratory volume in 1 s (Tabak et al., 2001).

Other studies have demonstrated a direct impact of specific phenolic compounds on inflammation *in vitro* and *in vivo*. For example, the flavanoid resveratrol inhibits inflammatory cytokine release from macrophages isolated from COPD patients (Culpitt et al., 2003). Meja et al. (2008) have observed that curcumin can inhibit inflammation and restore glucocorticoid efficacy in response to oxidative stress.

In our ongoing study of phenolic compounds with therapeutic effects for COPD from tropical fruits (Dastmalchi, Flores, Petrova, Pedraza-Penalosa, & Kennelly, 2011; Flores et al., 2012a; Flores et al., 2012b; Reynertson et al., 2006), the Costa Rican guava was investigated. The focus of this study is to evaluate the phenolic constituents of Costa Rican guava for their antioxidant and anti-inflammatory activities. Our ultimate goal is to identify natural products as new therapeutics for the treatment of COPD.

#### 2. Materials and methods

#### 2.1. General experimental procedures

Solvents for chromatography, HPLC-grade MeOH, formic acid and acetonitrile were obtained from J.T. Baker (Phillipsburg, NJ). GR-grade MeOH, hexane, chloroform, ethyl acetate, and *n*-butanol were supplied by VWR Inc. (Bridgeport, PA). Ultrapure water was prepared using a Millipore Milli-RO 12 plus system (Millipore Corp., Bedford, MA). 1,1-Diphenyl-2-picrylhydrazyl (DPPH), Trolox, and potassium peroxosulfate were purchased from Sigma–Aldrich (St. Louis, MO). 2,2'-Azinobis(3-ethylbenzothiazoline-6-sulfonate) diammonium salt (ABTS) was obtained from TCI-Ace (Tokyo, Japan). Ellagic acid, myricetin, quercitrin, and quercetin were supplied by Extrasynthèse (Genay, France).

#### 2.2. Plant material

Fruits of Costa Rican guava were collected at the Fruit and Spice Park (Homestead, FL). Fruits were frozen and shipped by overnight courier on dry ice to the laboratory, where they were kept in cold  $(-20 \text{ }^{\circ}\text{C})$  dark storage until processed.

#### 2.3. Extraction

Fruits of Costa Rican guava (900 g) were freeze–dried to obtain 90 g of fruit. The freeze–dried edible pulp (10 g) was extracted three times with 200 mL of MeOH/H<sub>2</sub>O (70:30) at room temperature with a blender for 5 min per extraction. The combined extract was dried *in vacuo* at temperatures not exceeding 40 °C.

The extract was suspended in water and sequentially partitioned three times with hexane (60 mL  $\times$  3), chloroform (60 mL  $\times$  3), ethyl acetate (60 mL  $\times$  3), and *n*-butanol (60 mL  $\times$  3). The four fractions were concentrated *in vacuo*.

#### 2.4. ABTS assay

Determination of the ABTS<sup>++</sup> scavenging assay effect of the Costa Rican guava crude extract and of the hexane, chloroform, ethyl acetate, and *n*-butanol fractions was performed according to the method of Re et al. (1999). Initially, an ABTS<sup>++</sup> stock solution was prepared by reacting ABTS aqueous solution (7 mM) with  $K_2S_2O_8$  (2.45 mM, final concentration) at ambient temperature in the dark for 12–16 h. Absorbance of the reactant was later adjusted to 0.700 ± 0.020 at ambient temperature at a wavelength of 734 nm. A Molecular Devices Versa<sub>max</sub> microplate reader (Sunnyvale, CA) was used. In a final volume of 200 µL, the reaction mixture was compromised of 198 µL of ABTS<sup>++</sup> solution and 2 µL of the sample at different concentrations (50–750 µg/mL). Absorbance of the ABTS<sup>++</sup> solution in the presence of different concentrations of Trolox (0.5–10 µg/mL) was also determined. The ABTS<sup>++</sup> percentage inhibition of the extracts, the fractions and Trolox was calculated from Eq. (1). Results were expressed as µmol Trolox equivalent antioxidant capacity (TEAC) per gram of sample:

$$Percentage Inhibition = \left[\frac{(Absorbance control) - (Absorbance sample)}{(Absorbance control)}\right] \times 100$$
(1)

#### 2.5. DPPH assay

The DPPH assay was performed according to the method developed by Smith, Reeves, Dage, and Schnettler (1987) slightly modified. To a 50  $\mu$ L aliquot of the sample 150  $\mu$ L of DPPH (400  $\mu$ M) were added. Decrease of absorbance was monitored at 517 nm after 30 min of incubation at 37 °C on a Molecular Devices Versa<sub>max</sub> microplate reader (Sunnyvale, CA). The percentage inhibition of the DPPH by each dilution of samples was calculated considering the percentage of the steady DPPH in solution after reaction (Eq. (1)). A plot of percentage inhibition *versus* concentration was made for the reference standard Trolox. On the basis of this plot, the TEAC values for different samples were calculated.

#### 2.6. IL-8 immunoassay

Human small airway epithelial (SAE) cells were cultured according to supplier instructions (Lonza, Walkersville, MD) and maintained in a controlled atmosphere of air/5% CO<sub>2</sub> at 37 °C. 80% confluent SAE cells at passages 2–5 were used for experiments. CSE was prepared using a modified protocol (Laurent, Janoff, & Kagan, 1983). Briefly, a Barnet vacuum pump operating at constant flow was used to draw the smoke of one 3R4F research grade cigarette (University of Kentucky) through 25 mL of Dulbecco's phosphate-buffered saline. This solution (100% CSE) was adjusted to pH 7.4, filtered, diluted with small airway growth medium to a final concentration of 5%, and added to the cells immediately.

Cells were treated with 5% CSE or pure compounds or pre-treated with pure compounds 1 h prior to 5% CSE exposure. After 24 h, measurement of human IL-8 in cell culture supernatants was performed by ELISA (R&D Systems Inc., Minneapolis, MN).

#### 2.7. MMP-1 mRNA expression

The cells were cultured as described above. After 24 h of treatment, total RNA from the human SAE cells was isolated (RNeasy kit, Qiagen, Valencia, CA) and converted into cDNA (high capacity cDNA kit, Applied Biosystems, Carlsbad, CA). Relative expression of MMP-1 was measured using real-time quantitative PCR and Taqman probes with GAPDH as an endogenous control (Applied Biosystems).

#### 2.8. HPLC-PDA

The analytical HPLC system (Waters Corp., Milford, MA) consisted of a Waters 2695 Separation Module equipped with a 2996 photodiode-array detector (PDA) and coupled to the Waters Empower (version 5.0) for data acquisition and processing. Separation was carried out using a 250 × 4.6 mm, 4 µm Synergi Hydro-RP 80A column (Phenomenex, Torrance, CA). The mobile phase consisted of solvents A (1% aqueous formic acid solution) and B (acetonitrile) as follows: 80-70% A over 5 min; 70-60% A over 5-10 min; 60-50% from 10 to 35 min at 1 mL/min, and these conditions were kept isocratic for 10 min. The composition was then changed to initial conditions in 5 min, and maintained for 10 min prior to the next injection. Stock solution of ellagic acid, myricetin, quercitrin, and quercetin were prepared in 70% (v/v) methanol to final concentration of 1 mg/mL. Each stock solution was further diluted to obtain six concentrations of the standard (ranging from 10 to 750 g/mL) for HPLC–PDA quantification and they were injected in triplicate. To determine the concentration of 3,3',4-tri-O-methylellagic acid-4'-O-D-glucopyranoside, ellagic acid was used as standard. The fruit extracts were also reconstituted in 70% (v/v) methanol and injected in triplicate at a concentration of 10 mg/mL. Peak areas for the extracts and standards were integrated from HPLC-PDA chromatograms by use of Waters Empower2 software at 360 nm.

#### 2.9. LC–MS analyses

An LCT Premier XE TOF mass spectrometer (Waters Corp.) equipped with an ESI interface and controlled by MassLynx V4.1 software was used to perform the LC–MS analysis. Mass spectra were acquired in both positive and negative modes over the mass range m/z 100– 1000. MS parameters were: capillary voltage, 3000 V (positive mode) and 2800 V (negative

mode); cone voltage, 20 V; desolvation and cone gas flow rates, 600 and 20 L/h; desolvation temperature, 400 °C; and the source temperature, 120 °C. The analytical column used was a  $250 \times 4.6$  mm, 4 µm Synergi Hydro-RP 80A column (Phenomenex), and the same elution solvent and method as the one described above for HPLC–PDA were applied.

#### 2.10. Statistical analysis

Data are expressed as means values  $\pm 95\%$  confidence interval. One-way analysis of variance (ANOVA) was performed with significant differences between means determined by the Student's *t*-test. JMP Statistics software package version 8 was used for statistical analyses (SAS Institute Inc., Cary, NC).

#### 3. Results and discussion

The potential effects of Costa Rican guava extract and fractions were examined for the treatment of COPD. Oxidative stress and inflammation are two causes for COPD; therefore polyphenols may be a useful treatment for COPD (Rahman, 2012). The ABTS<sup>++</sup> and DPPH<sup>+</sup> scavenging activities of the crude extract and the hexane, chloroform, ethyl acetate, and *n*-butanol fractions of Costa Rican guava were evaluated. Ethyl acetate fraction demonstrated higher antioxidant activity in the ABTS and DPPH assay, and therefore the inhibitory effect of IL-8 and MMP-1 expression in cells treated with CSE was investigated. Further, a compositional analysis of this fraction was performed using LC–PDA and TOF LC–MS and nine compounds (1–9) were found.

#### 3.1. Extraction

The extraction yield for the crude extract was 476.2 (mg/g of dry fruit). The highest extraction yield of the four fractions was obtained for *n*-butanol (30.23% w/w) followed by ethyl acetate (4.12% w/w) then chloroform (0.75% w/w). The lowest yield was obtained for hexane (0.25% w/w).

#### 3.2. Antioxidant activity

In order to measure the antioxidant activities of the Costa Rican guava extracts, ABTS<sup>++</sup> and DPPH<sup>-</sup> scavenging assays were used. The ABTS<sup>++</sup> scavenging activity was monitored over time, allowing the slow-acting antioxidants to have enough time to exert their effects. Due to their contribution to the scavenging activity, the order of activity among the samples changed during the assay, reaching their highest activity at 40 min (Fig. 1). All the extracts demonstrated a wide range of ABTS<sup>++</sup> scavenging activities. They exerted an increase in their activity over time; and the order of activity remained the same from 0 to 40 min. The ethyl acetate fraction was the most potent, followed by the crude extract, and the *n*-butanol, chloroform, and hexane fractions. The order of DPPH<sup>-</sup> scavenging activity of the Costa Rican guava extracts was ethyl acetate fraction > *n*-BuOH fraction and crude extract (not significantly different, p > 0.05) > chloroform fraction > hexane fraction (Fig. 2).

#### 3.3. IL-8 inhibition

The ethyl acetate fraction, due to its high antioxidant activity in the ABTS and DPPH assays, was evaluated for its efficacy in the inhibition of IL-8 in small airway epithelial

(SAE) cells untreated and treated with CSE. Concentrations of 5, 25, 50, and 100  $\mu$ g/mL were used. When control cells were exposed to CSE the amount of IL-8 increased threefold (Fig. 3a). A dose of 5  $\mu$ g/mL of the ethyl acetate fraction was enough to decrease the concentration of IL-8. SAE cells were more susceptible to IL-8 inhibition when 50 and 100  $\mu$ g/mL were added to the medium and 100  $\mu$ g/mL showed the highest decrease on the IL-8 production. In untreated cells the ethyl acetate fraction did not decrease the basal production of IL-8 at all of the concentrations tested.

#### 3.4. MMP-1

Before treatment with the guava ethyl acetate fraction, SAE cells demonstrated MMP-1 expression, which increased threefold after 24 h of CSE exposure (Fig. 3b). The addition of 5  $\mu$ g/mL of the ethyl acetate fraction reduced the production of MMP-1. This reduction was higher when 25 and 50  $\mu$ g/mL were added to the cells. The highest inhibitory activity was observed when 100  $\mu$ g/mL of the extract were added to the cells. In untreated cells the expression of MMP-1 decreased with all the concentrations evaluated in this study to the same level.

#### 3.5. Characterisation and quantification of the ethyl acetate fraction components by LC– PDA and LC–TOF

The ethyl acetate fraction of Costa Rican guava was analysed by LC–PDA and LC–TOF, to identify the components with significant biological activity. The nine major peaks detected in the ethyl acetate fraction by HPLC–PDA at 254 nm (Fig. 4), were identified by their elution order, UV/Vis and MS characteristics were compared with reported data in the literature, and, when possible, by co-injection of standards. In this study negative and positive modes of ESI mass detection were employed. TOF LC–MS (negative and positive modes) with ESI mass detection was conducted, and fragmentation data, retention time, spectrum information, and quantitative data are displayed in Table 1.

Compounds **3**, **4**, **7**, and **9** were identified as ellagic acid, myricetin, quercitrin, and quercetin, respectively. Their identification was confirmed by co-injection of standards. Mahattanatawee et al. (2006) identified ellagic acid in the leaf and roots of *P. guajava*. Quercitin, quercitrin, and myrcetin have been previously reported in the leaf and fruits of *P. guajava* (Kubola, Siriamornpun, & Meeso, 2011; Lozoya et al., 1994).

Compound **1** showed in the positive mode m/z 243.0492 corresponding to  $[M+Na]^+$  (C<sub>8</sub>H<sub>12</sub>O<sub>7</sub>Na), and the molecular ion  $[M+H]^+$  at 221.0649 (C<sub>8</sub>H<sub>13</sub>O<sub>7</sub>) (Table 1). In the negative mode a fragmental ion  $[M-H+HCOOH]^-$  at 265.0540 (C<sub>9</sub>H<sub>13</sub>O<sub>9</sub>), and the molecular ion  $[M-H]^-$  at 219.0480 (C<sub>8</sub>H<sub>11</sub>O<sub>7</sub>) were found (Table 1). The maximum UV absorbance was registered at 205 nm. This compound was tentatively associated with 1,5-dimethyl citrate.

The parent ion of compound **2** was obtained at m/z 333.0946 [M+Na]<sup>+</sup> (C<sub>15</sub>H<sub>18</sub>O<sub>7</sub>Na). It showed a deprotonated molecular ion in the negative mode at m/z 355.1011 (Table 1) corresponding, as already observed (Lunkenbein et al., 2006) to the formate adduct [M–H +COOH]<sup>-</sup> and thus, a molecular weight of 310.1048. The UV spectra showed absorption

maxima at 300, 224 (sh). This compound was determined to be 1-*trans*-cinnamoyl- $\beta$ -D-glucopyranoside, which was previously identified in *P. guajava* by Latza et al. (1996).

Based on their fragmentation pattern, molecular formula and UV profile (Table 1), compounds **5** and **8** were tentatively identified as sinapic aldehyde-4-O- $\beta$ -D-glucopyranoside and 1,3-O-diferuloylglycerol, respectively.

Compound **6** had a similar UV absorbance profile to **3**. It showed a molecular ion  $[M+H]^+$  at 507.1158 in the positive mode. In the negative mode a fragment at 551.1033 corresponding to the formate adduct  $[M-H+HCOOH]^-$  was found. The mass spectrum showed a fragment corresponding to the presence of glucose in the structure at m/z 345.0554  $[M+H-glucosyl group]^+$  (C<sub>17</sub>H<sub>13</sub>O<sub>8</sub>) in the positive mode and at m/z 343.0470  $[M-H-glucosyl group]^-$  (C<sub>17</sub>H<sub>11</sub>O<sub>8</sub>) in the negative mode (Table 1). This compound was tentatively identified as 3,3',4-tri-*O*-methylellagic acid-4'-*O*-D-glucopyranoside. Fig. 5 shows the structures of compounds **1–9**.

Although the therapeutical effect of Costa Rican guava ethyl acetate fraction on COPD remains to be investigated further, the antioxidant, IL-8, and MMP-1 inhibitory activities of this fraction are likely to be due to the phytochemicals mentioned above.

The antioxidant activity of compounds **3**, **4**, **7**, and **9** has been well documented (Boots, Haenen, & Bast, 2008; Tabart, Kevers, Pincemail, Defraigne, & Dommes, 2009). Luo, Li, and Kong (2011) studied the antioxidant activity of compound **8** using ABTS and DPPH assays. Several researchers have reported the anti-inflammatory activity of these polyphenols. Mueller, Hobiger, and Jungbauer (2010) reported that compounds **4** and **9** were notably more effective on the reduction of IL-6 and TNF- $\alpha$  secretion compared to cortisol. Previous studies carried out in our laboratory revealed that compound **3**, the major phenolic compound, demonstrated IL-8 and MMP-1 inhibitory activity in cells treated and untreated with CSE (Dastmalchi et al., 2012). Reactive oxygen species can initiate transcription factors as well as signal-transduction pathways (Rahman & Adcock, 2006). The activation of these pro-inflammatory mediators enhances transcription of downstream inflammatory chemokines (Calixto, Campos, Otuki, & Santos, 2004). By reducing the oxidative attack, these compounds may reduce the amount of pro-inflammatory mediators, which are directly tied to IL-8 production.

Despite the many promising *in vitro* effects of phenolic compounds for COPD and other diseases, it is important to note that many phenolic compounds either have limited bioavailability *in vivo* and/or are biotransformed in the gastrointestinal tract into compounds that may have less therapeutic efficacy (Spencer, Schroeter, Rechner, & Rice-Evans, 2001). Future *in vivo* studies using these phenolic compounds should be designed keeping in view the preceding observations.

For the treatment of COPD, it has been proposed that the development of improved inhaled delivery techniques would allow clinically relevant concentrations of antioxidants to be deposited in the lung while avoiding the first pass metabolism that occurs during systemic absorption (Zhu, Chen, & Li, 2000).

#### 4. Conclusion

The antioxidant activity of the Costa Rican guava ethyl acetate fraction coupled with its ability to reduce inflammation caused by secondary smoke exposure shows the potential that the constituents within this fraction may have for further drug development. Based on these results, further studies of these chemical constituents are currently being carried out in our laboratory.

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#### Fig. 1.

ABTS<sup>++</sup> scavenging activity of Costa Rican guava crude extract, and hexane, chloroform, ethyl acetate, and *n*-butanol fractions. Values are expressed as means  $\pm 95\%$  confidence intervals (*n* = 8) of Trolox equivalent antioxidant capacity (TEAC) (µmol of Trolox per gram of dry extract).



#### Fig. 2.

DPPH scavenging activity of Costa Rican guava crude extract, and hexane, chloroform, ethyl acetate, and *n*-butanol fractions. Values are expressed as means  $\pm 95\%$  confidence intervals (*n* = 8) of Trolox equivalent antioxidant capacity (TEAC) (µmol of Trolox per gram of dry extract). Bars with different letters (a–e) are significantly (*p* < 0.05) different.



#### Fig. 3.

Dose–response relationship of Costa Rican guava ethyl acetate fraction at 5, 25, 50, and 100 mg/mL on the expression of (a) IL-8 (b) and MMP-1 mRNA in SAE cells untreated (open bars) and treated (bold bars) with cigarette smoke extract (CSE). Data are presented as mean values  $\pm$ 95% confidence limits (*n* = 3). Open bars with the same lower case letters (a) and bold bars with the same upper case letters (A–C) are not significantly (*p* > 0.05) different.









#### Fig. 5.

Chemical structures of compounds identified in Costa Rican guava ethyl acetate fraction. 1,5-Dimethyl citrate (1), 1-*O*-*trans*-cinnamoyl- $\beta$ -D-glucopyranoside (2), ellagic acid (3), myricetin (4), sinapic aldehyde 4-*O*- $\beta$ -D-glucopyranoside (5), 3,3<sup>'</sup>,4-tri-*O*-methylellagic acid-4<sup>'</sup>-*O*-D-glucopyranoside (6), quercitrin (7), 1,3-*O*-diferuloylglycerol (8), and quercetin (9).

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## Table 1

Chemical profile of the identified compounds in the Costa Rican guava ethyl acetate fraction.

Identification	1,5-Dimethyl citrate	1-O-trans-Cinnamoyl-B-D-glucopyranose	Ellagic acid	Myricetin	Sinapic aldehyde 4- $\mathcal{O}oldsymbol{eta}$ D-glucopyranoside	3,3',4-Tri- <i>O</i> -methylellagic acid-4' - O-D-glucopyranoside	Quercitrin
Selected fragmental ion exact masses [adduct molecular ions-neutral molecules or radicals] <sup>+,-</sup> (molecular formula, ppm)	$\begin{array}{l} 243.0492 \ [M+Na]^+ \\ (C_8H_{1,2}O_7Na, 4.5); \\ 463.1057 \ [2M+Na]^+ \\ (C_{16}H_{24}O_{14}Na, -1.5) \\ 219.0480 \ [M-H]^- \\ (C_8H_{11}O_{1}, -11.4); \\ 265.0540 \ [M-H] \\ +HCOOH]^- \ (C_9H_{13}O_{9}, -7.5); 439.1057 \ [2 M-H] \\ -7.5; 439.1057 \ [2 M-H] \\ (C_{16}H_{23}O_{14}, -7.1) \end{array}$	$\begin{array}{l} 643.2001 \ [2M+Na]^+\\ (C_{30}H_{36}O_{14}Na_{1}-0.3)\\ 355.1011 \ [M-H\\ +HCOOH]^- \ (C_{16}H_{19}O_{9},\\ -5.1) \end{array}$	603.0048 [M-H +HCOOH] <sup>-</sup> (C <sub>28</sub> H <sub>11</sub> O <sub>16</sub> , 0.2)	319.0465 [M+H] <sup>+</sup> (C <sub>15</sub> H <sub>11</sub> O <sub>8</sub> , 3.4); 635.0687 [2 M-H] <sup>-</sup> (C <sub>30</sub> H <sub>19</sub> O <sub>16</sub> , 0.5)	393.1161 [M+Na] <sup>+</sup> (C <sub>17</sub> H <sub>21</sub> O <sub>9</sub> Na, -0.3)	507.1158 $[M+H]^+$ (C <sub>23</sub> H <sub>23</sub> O <sub>13</sub> , 3.7); 529.0938 $[M+Na]^+$ (C <sub>23</sub> H <sub>22</sub> O <sub>13</sub> Na, -3.8); 345.0554 $[M+H-$ glucosyl group] <sup>+</sup> (C <sub>17</sub> H <sub>13</sub> O <sub>8</sub> , -16.2) 343.0470 $[M-H-glucosylgroup]^- (C17H11O8, 4.7)$	493.0970 [M-H +HCOOH] <sup>-</sup> (C <sub>22</sub> H <sub>21</sub> O <sub>13</sub> , -2.4)
Marker ion exact mass (ppm)	221.0649 [M+H] <sup>+</sup> (C <sub>8</sub> H <sub>13</sub> O <sub>7</sub> , -5.4)	333.0946 [M+Na] <sup>+</sup> (C <sub>15</sub> H <sub>18</sub> O <sub>7</sub> Na, -1.2)	300.0073 [M-H] <sup>-</sup> (C <sub>14</sub> H <sub>5</sub> O <sub>8</sub> , -3.7)	317.0280 [M-H] <sup>-</sup> (C <sub>15</sub> H <sub>9</sub> O <sub>8</sub> , -5.4)	415.1219 [M-H +HCOOH] <sup>-</sup> (C <sub>18</sub> H <sub>23</sub> O <sub>11</sub> , -5.1)	551.1033 [M–H +HCOOH] - (C <sub>24</sub> H <sub>23</sub> O <sub>15</sub> , –0.7)	449.1099 [M+H] <sup>+</sup> (C <sub>21</sub> H <sub>21</sub> O <sub>11</sub> , 3.3)
UV data (A <sub>max</sub> )	205	224, 300	254, 364	254, 370	265, 365	250, 365	257, 353
Concentration (mg/g of dry extract)			10.5	1.1		5.8	6.0
Retention time (min)	12.7	15.0	17.1	17.4	18.2	18.9	20.9
Compound number <sup>a</sup>	_	6	£	4	Ŋ	v	۲

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Identification	1,3- <i>O</i> Diferuloylglycerol	Quercetin
Selected fragmental ion exact masses [adduct molecular ions-neutral molecules or radicals] <sup>+,-</sup> (molecular formula, ppm)	443.1326 [M-H] <sup>-</sup> (C <sub>23</sub> H <sub>23</sub> O <sub>9</sub> , -3.6)	303.0534 [M-H] <sup>+</sup> (C <sub>15</sub> H <sub>11</sub> O <sub>7</sub> , 9.6)
Marker ion exact mass (ppm)	445.1501 [M+H] <sup>+</sup> (C <sub>23</sub> H <sub>25</sub> O <sub>9</sub> , 0.4)	301.0366 [M-H] <sup>-</sup> (C <sub>15</sub> H <sub>9</sub> O <sub>7</sub> , 6.0)
UV data ( $\mathcal{X}_{max}$ )	250, 365	254, 367
Concentration (mg/g of dry extract)		9.2
Retention time (min)	22.0	22.9
Compound number <sup>a</sup>	8	6

 $^{a}\!\mathrm{Compounds}$  numbers and retention times refers to the numbers given in Fig. 4.