

# Light-driven carbon dioxide reduction to methane by nitrogenase in a photosynthetic bacterium

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**Nitrogenase is an ATP-requiring enzyme capable of carrying out multielectron reductions of inert molecules. A purified remodeled nitrogenase containing two amino acid substitutions near the site of its FeMo cofactor was recently described as having the capacity to reduce carbon dioxide (CO<sub>2</sub>) to methane (CH<sub>4</sub>). Here, we developed the anoxygenic phototroph, *Rhodospseudomonas palustris*, as a biocatalyst capable of light-driven CO<sub>2</sub> reduction to CH<sub>4</sub> by *R. palustris* required constitutive expression of nitrogenase, which was achieved by using a variant of the transcription factor NifA that is able to activate expression of nitrogenase under all growth conditions. Also, light was required for generation of ATP by cyclic photophosphorylation. CH<sub>4</sub> production by *R. palustris* could be controlled by manipulating the distribution of electrons and energy available to nitrogenase. This work shows the feasibility of using microbes to generate hydrocarbons from CO<sub>2</sub> in one enzymatic step using light energy.**

nitrogenase | *Rhodospseudomonas* | bioenergy | methane | engineered bacterium

An essential process for life and an important step in the biogeochemical nitrogen cycle is nitrogen fixation by nitrogenase, in which nitrogen gas (N<sub>2</sub>) is converted to ammonia (NH<sub>3</sub>) (1). The difficult reduction of N<sub>2</sub> to two NH<sub>3</sub> occurs at an FeMoS cluster called FeMo cofactor in Mo-dependent nitrogenase in a reaction that requires ATP hydrolysis and dihydrogen production as shown in 1



Nitrogenase deprived of access to N<sub>2</sub> but provided with a source of electrons produces H<sub>2</sub> exclusively. Also, the ability of nitrogenase to carry out the multielectron reduction of an inert molecule is not limited to reduction of N<sub>2</sub>. This enzyme can also reduce other molecules with double and triple bonds, including carbon-containing compounds (reviewed in ref. 2). Recently, we found that a remodeled nitrogenase with substitutions in two key amino acids near the FeMo cofactor is capable of reducing carbon dioxide (CO<sub>2</sub>) to methane (CH<sub>4</sub>) in vitro. This enzyme did not retain its ability to reduce N<sub>2</sub> but was active in H<sub>2</sub> production (3). It was unclear if the remodeled nitrogenase gene could confer to bacteria the ability to reduce CO<sub>2</sub> to CH<sub>4</sub>.

Here, we describe a biocatalyst capable of generating the energy-rich hydrocarbon CH<sub>4</sub> by reduction of CO<sub>2</sub> using a remodeled nitrogenase. Development of this biocatalyst required selection of an appropriate microbial host, because large amounts of cellular reductant and ATP are used by nitrogenase, and as a consequence, this energetically expensive enzyme is repressed by both transcriptional and posttranslational regulatory mechanisms when an alternative nitrogen source, like ammonium, is available (4). Use of nitrogenase to generate a product not used by the organism would require overcoming these regulatory constraints to achieve expression of active enzyme, while at the same time providing cells with ammonium for growth. We reasoned that the anoxygenic

photosynthetic bacterium *Rhodospseudomonas palustris* would be a good chassis for studying the remodeled nitrogenase in the context of a biological system, because it can fix nitrogen, and an activating mutation in *nifA*, encoding the transcription activator of nitrogenase genes, has been identified that can bypass the regulatory networks that repress nitrogenase (5–7). Also, *R. palustris* can generate the considerable amount of ATP needed for the activity of a remodeled nitrogenase from light by cyclic photophosphorylation.

We found that *R. palustris* reduced CO<sub>2</sub> to CH<sub>4</sub> using a remodeled nitrogenase with NifD<sup>V75AH201Q</sup> amino acid substitutions when the enzyme was expressed in a genetic background that allows for its constitutive production and cells were incubated in light. We also found that *R. palustris* is a tractable system that allows control of CH<sub>4</sub> production by manipulating the distribution of electrons and energy available to nitrogenase. In nature, the eight-electron reduction of CO<sub>2</sub> to CH<sub>4</sub> requires a multistep metabolic pathway found only in methanogenic archaea (8). The results presented here describe a one-step route to biological hydrocarbon production.

## Results

A variant of the FeMo-containing nitrogenase protein, NifD, from *Azotobacter vinelandii* containing amino acid substitutions V70A and H195Q reduces CO<sub>2</sub> to CH<sub>4</sub> and also produces H<sub>2</sub> in vitro, but it is unable to reduce N<sub>2</sub> to NH<sub>3</sub> (3). Nucleotide substitutions to generate homologous substitutions of residues in *R. palustris* NifD (V75A and H201Q) were introduced into the *nifD* coding sequence by homologous recombination. The *R. palustris*

## Significance

One of the most important life-sustaining metabolisms that results from a net reduction reaction is the conversion of nitrogen gas to ammonia by nitrogenase in a process known as nitrogen fixation. A nitrogenase variant has been described that converts carbon dioxide to methane in vitro. Here, we expressed this variant in an engineered strain of the photosynthetic bacterium *Rhodospseudomonas palustris* and showed biological light-driven methane production. Its ability to reduce carbon dioxide at ambient temperature and pressure with a single enzyme and energy provided by light makes the methane-producing strain of *R. palustris* an excellent starting point to understand how a biological system can marshal resources to produce an energy-rich hydrocarbon in one enzymatic step.

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**Table 1. CH<sub>4</sub> and H<sub>2</sub> production by *R. palustris* NifA\* strains expressing either a WT nitrogenase or a remodeled nitrogenase**

Genotype	Growth rate (h)	H <sub>2</sub> production	CH <sub>4</sub> production
		( $\mu\text{mol/mg}$ total protein)	(nmol/mg total protein)
<i>nifA* nifD<sup>WT</sup></i>	11.6 (0.1)	36.2 (6.2)	ND
<i>nifA* nifD<sup>V75AH201Q</sup></i>	12.7 (0.9)	55.1 (3.2)	30.1 (2.0)

Data are the average of three or more experiments, with the SDs shown in parentheses. Cells were grown with 20 mM acetate and 10 mM NaHCO<sub>3</sub> and incubated in light. ND, not detected.

*nifD<sup>V75AH201Q</sup>* allele was integrated into the chromosome of an *R. palustris* strain CGA009 *nifA\** mutant that expresses Mo nitrogenase constitutively, even in the presence of ammonium (5, 7). *R. palustris* can also synthesize alternative V or Fe nitrogenase isozymes, but these isozymes are not expressed in the NifA\* strain grown with ammonium (5, 9). The NifA\* strain expressing the remodeled nitrogenase produced CH<sub>4</sub> when grown with acetate as a carbon source and HCO<sub>3</sub><sup>-</sup> as a source of CO<sub>2</sub> (Table 1). The growth rate of this strain was similar to that of its parent strain expressing WT nitrogenase. Both strains also produced comparable levels of H<sub>2</sub> (Table 1). *R. palustris* CGA009 is able to produce H<sub>2</sub> only through the activity of its nitrogenase, and thus, H<sub>2</sub> production is a measure of nitrogenase activity (10). The *nifA\* nifD<sup>V75AH201Q</sup>* strain produced almost 2,000 times more H<sub>2</sub> than CH<sub>4</sub> (Fig. 1 and Table 1). A similarly high ratio of H<sub>2</sub> to CH<sub>4</sub> produced was observed in vitro with purified remodeled nitrogenase (3).

To verify that the remodeled nitrogenase was directly responsible for CH<sub>4</sub> production by *R. palustris*, the product profile of the enzyme was examined. Remodeled enzyme purified from *R. palustris* converted CO<sub>2</sub> to CH<sub>4</sub> (1.95 nmol CH<sub>4</sub>/nmol MoFe protein). No CH<sub>4</sub> production was detected for purified WT nitrogenase. In addition, nongrowing cells of the *R. palustris nifA\* nifD<sup>V75AH201Q</sup>* strain supplied with thiosulfate as an electron donor and H<sup>13</sup>CO<sub>3</sub><sup>-</sup> as a source of CO<sub>2</sub> produced a product that, when analyzed by gas chromatography - mass spectrometry (GC-MS), had an *m/z* of 17 and a retention time that correlates with <sup>13</sup>CH<sub>4</sub>. A peak with an *m/z* of 16 and a retention time corresponding to CH<sub>4</sub> occurred when H<sup>12</sup>CO<sub>3</sub><sup>-</sup> was provided. Almost 90% of the CH<sub>4</sub> produced was from CO<sub>2</sub> reduction (Fig. 2). These data confirm that the CH<sub>4</sub> produced by *R. palustris* cells comes directly from CO<sub>2</sub>. As expected, the remodeled nitrogenase purified from *R. palustris* was unable to reduce N<sub>2</sub> to NH<sub>3</sub>, but it reduced acetylene (C<sub>2</sub>H<sub>2</sub>) to ethylene (C<sub>2</sub>H<sub>4</sub>) (Fig. S1). These results indicate that *R. palustris* functions to catalyze the production of CH<sub>4</sub> and H<sub>2</sub> as shown in Fig. 3. Electrons

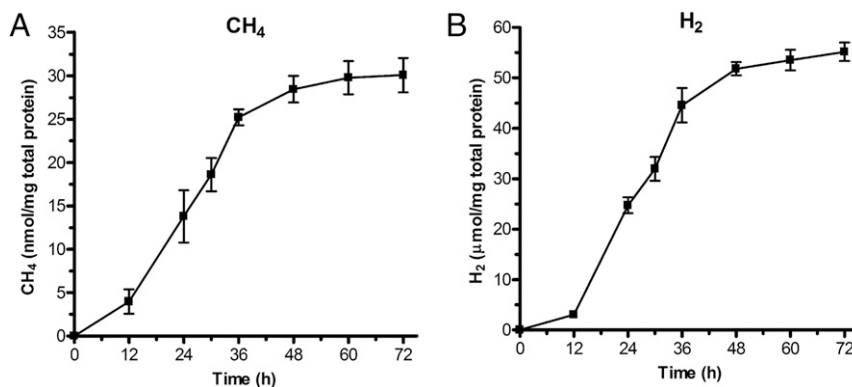
derived from the oxidation of either an organic carbon source (acetate) or an inorganic compound (thiosulfate) are used to reduce CO<sub>2</sub> to CH<sub>4</sub>. HCO<sub>3</sub><sup>-</sup> acquired from outside the cell is converted to intracellular CO<sub>2</sub> by carbonic anhydrase (11). Some CO<sub>2</sub> is also generated from acetate metabolism.

A common method used to divert cellular resources to microbial biocatalysts is to use nongrowing cells. Such cells are spared the imperative of carrying out biosynthetic reactions required for growth, which frees up electrons derived from the oxidation of external electron donors for use in reduction reactions (12). To test if diversion of electrons from biosynthesis leads to more CH<sub>4</sub> production by the remodeled nitrogenase, the cells of the *R. palustris nifA\* nifD<sup>V75AH201Q</sup>* strain grown to midlogarithmic phase were harvested, resuspended in nitrogen-free medium, and tested for their ability to produce CH<sub>4</sub>. As shown in Fig. 4A, nongrowing cells produced almost 20 times more CH<sub>4</sub> than growing cells.

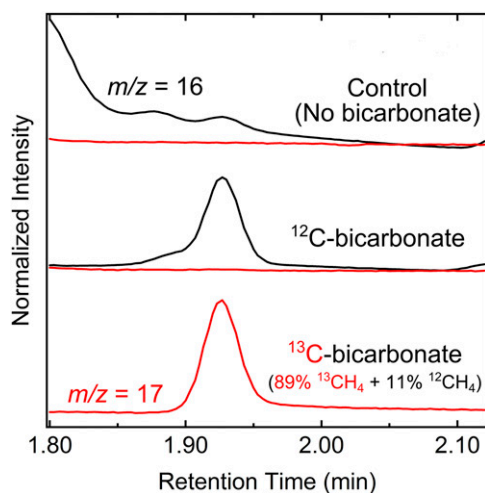
*R. palustris* has the Calvin Bassham Benson (CBB) cycle for CO<sub>2</sub> fixation, and it is known that nitrogenase competes with the CBB cycle for electrons (7, 13). Cells in which CBB cycle activity has been genetically disrupted produce more hydrogen, because more reducing equivalents are available to nitrogenase (7, 13). We found that *R. palustris nifA\* nifD<sup>V75AH201Q</sup> ΔcbbMLS* cells, which had both sets of Rubisco genes deleted, produced twice as much CH<sub>4</sub> as cells with an intact CBB cycle (Fig. 4B).

When nongrowing cells of the *nifA\* nifD<sup>V75AH201Q</sup>* strain were incubated in the dark, no detectable CH<sub>4</sub> was produced (Fig. 5). This result was expected, because light is required for ATP production, and ATP is required for nitrogenase activity. As shown in Fig. 5, the amount and rate of CH<sub>4</sub> production correlated with light intensity, and more CH<sub>4</sub> was produced under high light intensities than under low light intensities. The exception was at the two highest light intensities: 30 and 60  $\mu\text{mol photons/m}^2$  per second, where the amount and rate of CH<sub>4</sub> production were not significantly different. This result likely reflects that ATP is not the limiting parameter for CH<sub>4</sub> production under higher light intensities.

The results shown in Fig. 4 indicate that CH<sub>4</sub> production can be modulated by controlling the electron flux to nitrogenase. In *R. palustris*, electrons are generated during oxidation of organic carbon sources, and we have found that the organic electron donor provided influences the amount of electrons available to nitrogenase (13). The substrate oxidation state, the route that the substrate takes to generate biosynthetic precursors, and the amount of CBB cycle flux that occurs on a given substrate all influence the amount of electrons available to nitrogenase in *R. palustris* (13). When we grew *R. palustris nifA\* nifD<sup>V75AH201Q</sup>* cells on different carbon sources to determine how different electron donors may influence CH<sub>4</sub> production, we found that, with the



**Fig. 1.** An engineered strain of *R. palustris* expressing a remodeled nitrogenase produces CH<sub>4</sub>. *R. palustris* NifA\* cells expressing a remodeled nitrogenase produce (A) CH<sub>4</sub> and (B) H<sub>2</sub> during growth. These data are the average of three independent experiments, and error bars represent SDs. Cells were grown with 20 mM acetate and 10 mM NaHCO<sub>3</sub> and incubated in light.



**Fig. 2.** CH<sub>4</sub> produced by *R. palustris* NifA\* cells expressing a remodeled nitrogenase is caused by CO<sub>2</sub> reduction by the remodeled nitrogenase. GC-MS confirmation of CH<sub>4</sub> production from CO<sub>2</sub> reduction by *R. palustris* NifA\* cells expressing remodeled nitrogenase. The black traces are the GC-MS data for monitoring <sup>12</sup>CH<sub>4</sub> ( $m/z = 16$ ), and the red traces are the data for monitoring <sup>13</sup>CH<sub>4</sub> ( $m/z = 17$ ). Thiosulfate was supplied as an electron donor for the reaction.

exception of fumarate and succinate, the amount of CH<sub>4</sub> produced was similar, regardless of the type of organic carbon source used (Fig. 6). *R. palustris* expressing remodeled nitrogenase produced H<sub>2</sub> as well as CH<sub>4</sub> under all conditions that we tested. The amount of H<sub>2</sub> that cells produced was always on the order of 1,000–7,000 times greater than that of CH<sub>4</sub>.

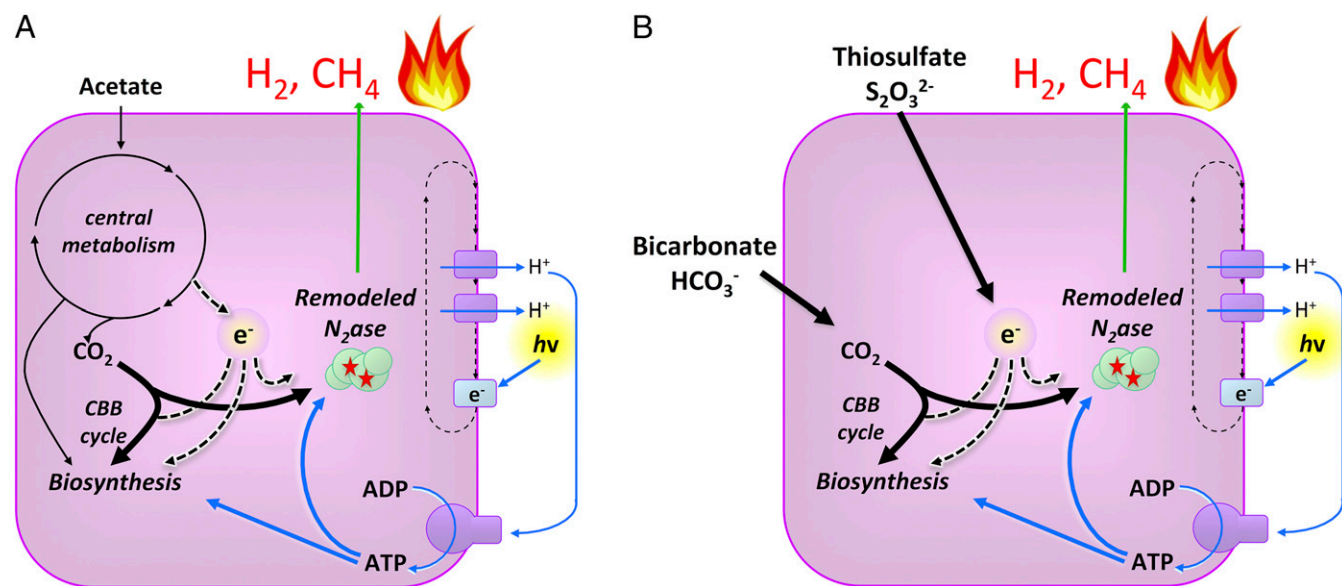
### Discussion

*R. palustris* has clear advantages as a system to develop the use of a remodeled nitrogenase in synthetic biology. One is that active remodeled nitrogenase can be produced constitutively. This feature is important, because it allows *R. palustris* to grow with ammonium as a nitrogen source and use an energetically expensive

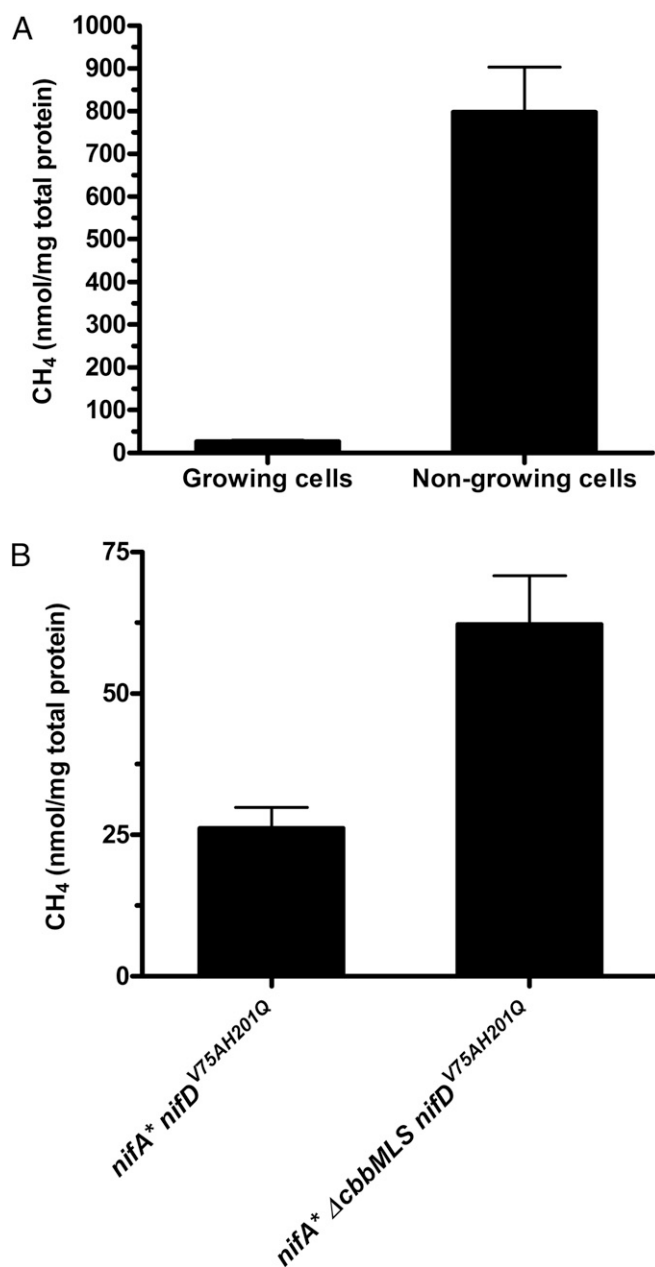
enzyme to produce a product that it cannot use. The other major advantage to using *R. palustris* is that it can meet the high ATP requirement for CH<sub>4</sub> production by nitrogenase by using cyclic photophosphorylation (Fig. 3). We have shown that the amount and rate of CH<sub>4</sub> production are dependent on the light intensity provided (Fig. 5). The simplest explanation for this correlation is that decreased light intensity results in a change in the energy status of the cell, so that less ATP is available for nitrogenase. However, this effect on CH<sub>4</sub> production could be more complex, because cellular energy levels may affect posttranslational modification of nitrogenase (14).

CH<sub>4</sub> production can be improved by diverting electrons away from biosynthesis or the CBB cycle (Fig. 4), and maximal CH<sub>4</sub> production will likely involve manipulating both of these parameters. *R. palustris* can maintain a nongrowing but metabolically active state for months as long as it has access to light and electrons (15). It can obtain these electrons from both organic and inorganic donors and even use an electric current from an electrode as an electron donor (13, 16, 17). Substrate conversion efficiencies of 86% have been shown using an inorganic electron donor, much higher than the 60% conversion efficiency on an organic electron donor (16). This ability sets up the possibility of exploiting nongrowing *R. palustris* nifA\* nifD<sup>V75AH201Q</sup> in microbial electrolysis cells (MECs), where electrons are provided by a cathode and used to generate CH<sub>4</sub>. Generation of a chemical product at the cathode has both environmental and monetary benefits that increase the potential of MECs in biofuel generation (18).

Despite an ability to increase CH<sub>4</sub> production by diverting electrons to nitrogenase, maximizing CH<sub>4</sub> production using a biocatalyst like *R. palustris* will ultimately require altering the ratio of CH<sub>4</sub> to H<sub>2</sub> produced by the remodeled nitrogenase. The nifA\* strain used in this study has a defect in its uptake hydrogenase (10). One way to recycle the reductant used to make hydrogen would be to restore the uptake hydrogenase. Additionally, increasing the ratio of the Fe protein to the MoFe protein has been shown to favor CH<sub>4</sub> production, resulting in a reduction in the ratio of H<sub>2</sub> to CH<sub>4</sub> produced (3). A similar approach could be taken by overexpressing the Fe protein in *R. palustris* nifA\* nifD<sup>V75AH201Q</sup>.



**Fig. 3.** Metabolic route of CH<sub>4</sub> production by an engineered strain of *R. palustris* expressing a remodeled nitrogenase. ATP is produced by cyclic photophosphorylation, in which electrons energized by light are cycled through a proton-pumping electron transport chain rather than transferred to a terminal electron acceptor. Electrons are generated by oxidation of (A) organic compounds or (B) inorganic compounds, such as thiosulfate. CO<sub>2</sub> from bicarbonate or generated from acetate oxidation is converted by remodeled nitrogenase to CH<sub>4</sub> and the CBB cycle to cell material.



**Fig. 4.** Diverting more electrons to the remodeled nitrogenase results in more CH<sub>4</sub> production by *R. palustris* whole cells. Diversion of electrons away from (A) biosynthesis using nongrowing cells or (B) the CBB cycle through genetic mutation ( $\Delta cbbMLS$ ) results in more CH<sub>4</sub> production by *R. palustris* NifA\* cells expressing a remodeled nitrogenase. These data are the average of three or more independent experiments, and the error bars represent SDs.

Future studies will focus on manipulating light, electron flux, and nitrogenase itself to improve CO<sub>2</sub> reduction to CH<sub>4</sub>. *R. palustris* can also synthesize alternative V or Fe nitrogenase isozymes (9). These isozymes could allow for even greater expansion of the catalytic repertoire of nitrogenase and further our understanding of how to produce energy-rich hydrocarbons in one step.

### Materials and Methods

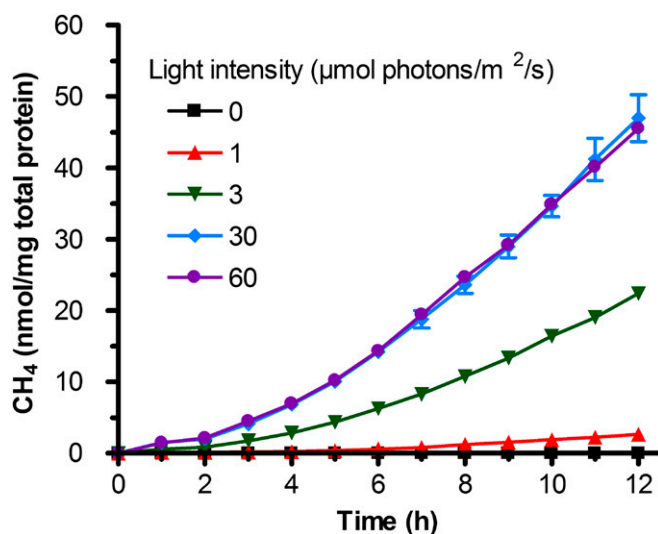
**Reagents, Bacteria, and Culture Methods.** All reagents were obtained from Sigma-Aldrich or Fisher Scientific and used without additional purification. Gases (argon, dinitrogen, hydrogen, CH<sub>4</sub>, ethane, C<sub>2</sub>H<sub>2</sub>, and C<sub>2</sub>H<sub>4</sub>) were purchased from Air Liquide or Gasco. Manipulation of proteins and buffers was done in septum-sealed serum vials under an argon atmosphere or on a

Schlenk line. All gas transfers were made using gas-tight syringes. All *R. palustris* strains were grown on photosynthetic medium (PM) as previously described (5–7, 19). The parent *R. palustris* strain, CGA009, used in all experiments is defective in expression of its uptake hydrogenase enzyme, and thus, hydrogen production by this strain can be used as a direct measure of nitrogenase activity. All cultures were initially grown anaerobically with 30  $\mu\text{mol photons/m}^2$  per s from a 60-W incandescent light bulb (General Electric) and then, diluted into fresh PM medium. This light intensity supports the maximum growth rate of *R. palustris*. Where appropriate, organic acids and alcohols were added at a final concentration of 40 mM carbon, and NaHCO<sub>3</sub> was added at a final concentration of 10 mM. Light intensity was altered using a 15-W incandescent light bulb controlled by a dimmer switch to provide light intensities of 3 and 1  $\mu\text{mol photons/m}^2$  per second, and light intensity of 60  $\mu\text{mol photons/m}^2$  per second was obtained by using a 200-W incandescent light bulb. We generated suspensions of nongrowing cells by washing midexponential-phase cultures with nitrogen-free PM (NFM) and flushing the headspace with argon gas after degassing the medium. Additional culture methods and growth conditions can be found in *SI Materials and Methods*.

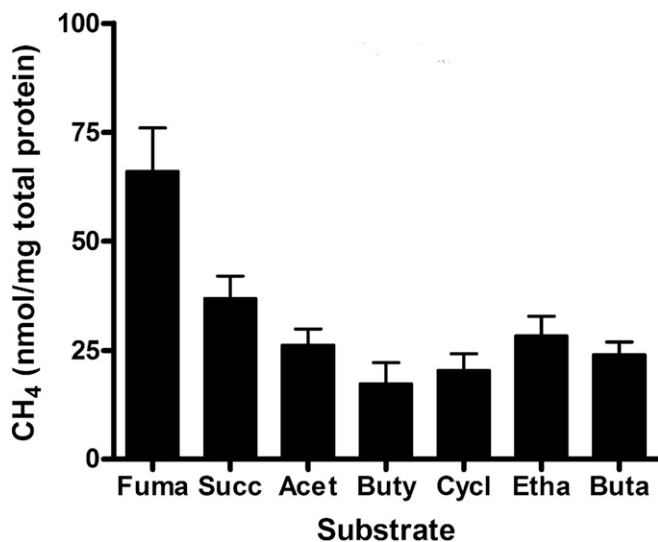
**Genetic Manipulation of *R. palustris*.** All strains and plasmids used are listed in Table S1. Plasmids were mobilized into *R. palustris* CGA676 or CGA679 by conjugation with *Escherichia coli* S17-1, and double-cross-over events for allelic exchange were achieved using a selection and screening strategy as described previously (5). Details of strain and plasmid construction for this study can be found in *SI Materials and Methods*.

**Protein Purification.** Cell extracts from *R. palustris* cells were prepared by using a French pressure cell operated at 1,500 lb/in<sup>2</sup> in a degassed 50 mM Tris-HCl buffer (pH 8.0) with 2 mM sodium dithionite under Ar. His-tagged MoFe proteins were purified by an immobilized metal affinity chelation chromatography protocol (20), with minor modifications. Protein concentrations were determined by the Biuret assay using BSA as the standard. The purities of these proteins were determined based on SDS/PAGE analysis with Coomassie staining.

**CO<sub>2</sub> Reduction Assays.** CO<sub>2</sub> reduction assays were conducted in 9.4-mL serum vials containing an assay buffer consisting of an MgATP regeneration system (15 mM MgCl<sub>2</sub>, 90 mM phosphocreatine, 15 mM ATP, 0.6 mg/mL creatine phosphokinase, 1.2 mg/mL BSA) and 12 mM sodium dithionite in 100 mM 4-morpholinepropanesulfonic acid sodium salt (Mops) buffer at pH 7.0. After solutions were made anaerobic, 0.45 atm CO<sub>2</sub> was added, and the gas and liquid phases were allowed to equilibrate for ~20 min. MoFe protein was then added, the vials were ventilated to atmospheric pressure, and the



**Fig. 5.** CH<sub>4</sub> production by *R. palustris* NifA\* cells expressing a remodeled nitrogenase is a light-driven process that can be controlled by light intensity. Altering the light intensity provided to cells alters the amount of CH<sub>4</sub> produced. These data are the average of three or more independent experiments, and the error bars represent SDs.



**Fig. 6.** CH<sub>4</sub> produced by *R. palustris* NifA\* cells expressing remodeled nitrogenase is affected by the organic compound supplied as a growth substrate. CH<sub>4</sub> production by cells grown on different substrates. Organic acids and alcohols were added at a final concentration of 40 mM carbon. The data shown are the average of five or more independent experiments, and the error bars represent SDs. Acet, acetate; Buta, butanol; Buty, butyrate; Cycl, cyclohexane carboxylate; Etha, ethanol; Fuma, fumarate; Succ, succinate.

reaction was initiated by the addition of Fe protein. Reactions were conducted at 30 °C for 60 min and then quenched by the addition of 700  $\mu$ L 400 mM EDTA (pH 8.0). Quantification of CH<sub>4</sub> was done according to a published protocol (3).

**Dinitrogen, C<sub>2</sub>H<sub>2</sub>, and Proton Reduction Assays.** Reduction assays were conducted in 9.4-mL serum vials containing an assay buffer consisting of an MgATP regeneration system (6.7 mM MgCl<sub>2</sub>, 30 mM phosphocreatine, 5 mM ATP, 0.2 mg/mL creatine phosphokinase, 1.2 mg/mL BSA) and 12 mM sodium

dithionite in 100 mM Mops buffer at pH 7.0. After solutions were made anaerobic, the headspace gases in the reaction vials were adjusted to proper partial pressures of different gases for different substrates (1 atm N<sub>2</sub> for N<sub>2</sub>, 0.1 atm C<sub>2</sub>H<sub>2</sub> and 0.9 atm Ar for C<sub>2</sub>H<sub>2</sub> reduction, and 1 atm Ar for proton reduction). This step was followed by the addition of the MoFe protein. After the vials were ventilated to atmospheric pressure, the reactions were initiated by the addition of the Fe protein. Reactions were conducted at 30 °C for 8 min and then, quenched by the addition of 300  $\mu$ L 400 mM EDTA (pH 8.0). The products (NH<sub>3</sub>, H<sub>2</sub>, and C<sub>2</sub>H<sub>4</sub>) from different substrate reduction assays were quantified according to published methods with minor modifications (3, 21, 22).

**H<sub>2</sub> and CH<sub>4</sub> Measurements from *R. palustris* Cells.** Gas-phase samples (50  $\mu$ L) were withdrawn with a Hamilton sample lock syringe from the culture vial headspace at intervals, and hydrogen was measured with a Shimadzu GC-2014 Gas Chromatograph as previously described (5, 7, 12, 13). Total protein concentrations were determined using the Bio-Rad Protein Assay Kit. Ethane was used as an internal standard to quantify CH<sub>4</sub>; 200  $\mu$ L ethane (0.08% or 0.8%; 1 atm) was placed into a 27-mL anaerobic tube with 10 mL culture and then, shaken gently to distribute the gases uniformly. A 100- $\mu$ L gas sample was taken from the headspace of the tube for gas chromatography - flame ionization (GC-FID) analysis using a Hamilton Sample Lock Syringe. The separation of CH<sub>4</sub> was performed using a stainless steel 80/100 Porapak N Column (1.8 m  $\times$  1/8 in  $\times$  2.1 mm i.d.) purchased from Sigma-Aldrich. The column temperature was held at 60 °C, and the injector and detector temperatures were set to 85 °C and 150 °C, respectively. Argon was used as the carrier gas at a rate of 35 mL/min.

**GC-MS Analysis for in Vivo CO<sub>2</sub> Reduction to CH<sub>4</sub> by *R. palustris*.** GC-MS analysis was conducted to confirm the production of CH<sub>4</sub> from CO<sub>2</sub> reduction using a Shimadzu GC-2010 Gas Chromatograph equipped with a programmed temperature vaporizing injector and a Shimadzu GCMS-QP2010S Mass Spectrometer by using <sup>12/13</sup>C-enriched NaHCO<sub>3</sub> as the CO<sub>2</sub> source for *R. palustris* V75A/H201Q mutant. Additional details can be found in *SI Materials and Methods*.

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