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Synthesis of *Cis,syndiotactic* A-*alt*-B Copolymers from Two Enantiomerically Pure *Trans*-2,3-Disubstituted-5,6-Norbornenes

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Supporting Information

ABSTRACT: *Cis,syndiotactic* **A***-alt-***B** copolymers, where **A** and **B** are two enantiomerically pure *trans-*2,3-disubstituted-5,6-norbornenes with "opposite" chiralities, can be prepared with stereogenic-at-metal initiators of the type M(NR)-(CHR')(OR")(pyrrolide). Formation of a high percentage of alternating **AB** copolymer linkages relies on an inversion of chirality at the metal with each propagating step and a relatively fast formation of an **AB** sequence as a consequence of a preferred diastereomeric relationship between the chirality at the metal and the chirality of the monomer. This approach to formation of an alternating **AB** copolymer sequence and the antically with the principle of forming **AB** copolymers from achiral monomers and catalysts.

$A^{R} + A^{R} + B^{R'} + B^{R'} + B^{R'} + B^{R'} + Cis, syndiotactic-poly(A-alt-B)$ M = Mo or W; R" = 1-Adamantyl or 2,6-Me₂C₆H₃

INTRODUCTION

Copolymers in which monomers A and B are incorporated in an alternating manner, poly(A-alt-B), are rare.¹⁻⁵ Examples are alternating AB copolymers formed from CO and olefins or CO₂ and epoxides. In these cases alternation is greatly assisted by the fact that one partner (CO or CO_2) does not itself polymerize. A few alternating AB copolymers have been formed through ringopening metathesis polymerization (ROMP) of cyclic olefin monomers,⁶⁻²⁸ but in these circumstances both A and B usually can be homopolymerized and the stereochemistry of the C=Cbond in the polymer is not fixed. One exception is the alternating AB copolymer which has all trans C=C bonds and <5% AA errors formed from a norbornene-like monomer (\mathbf{B}) that is slow to homopolymerize and cyclooctene or cycloheptene (A). The most successful initiators are of the type Mo(NR)(CHCMe₂Ph)- $[OCMe(CF_3)_2]_2$ (R = 2,6-Me₂C₆H₃ or 2,6-*i*-Pr₂C₆H₃), two welldefined alkylidene initiators that contain Mo or W out of many that have proven useful for preparing stereoregular polymers from norbornenes and norbornadienes.²

Among the well-defined Mo or W initiators are those that contain a stereogenic metal, e.g., 1a-1c. These initiators can produce a special category of stereoregular A-*alt*-B copolymers made from a racemic chiral monomer where A and B are enantiomers. These "A-*alt*-A*" copolymers have a basic *cis, syndiotactic* structure (eq 1), which is readily proven through ¹H



and ¹³C NMR studies.^{36,37} Only one example of a stereoregular A-alt-A* copolymer has been reported in the older literature.^{38,39} The cis structure is formed when 1a or 1b reacts with monomer to yield all *cis* metallacycles in trigonal bipyramidal (TBP) intermediates in which the terphenoxide and the imido ligands are in apical positions, while syndiotacticity results from an inversion of chirality at the metal center with each step in the polymerization. Inversion of chirality at the metal forces the olefin to approach first one side of the M=C bond and then the other. Incorporation of enantiomers in an alternating fashion is a consequence of one enantiomer of the racemic monomer reacting more rapidly with one enantiomer (at the metal center) of each propagating species. We have called this mode of control of polymer structure "stereogenic metal control"; although the chirality of the chain end nearest the metal that results from last inserted monomer is not necessarily irrelevant, the determining feature is the lowest energy diastereomeric combination of chirality at the metal and chirality of the monomer. The "errors" in the *cis,syndiotactic*-poly(A-alt-A*) structure arise through formation of AA and A*A* cis,syndiotactic and trans,isotactic dyads. Trans, isotactic dyads arise through formation of a trans metallacyclobutane intermediate (instead of a cis metallacyclobutane), which "flips over" before opening, a rearrangement that preserves the configuration at the metal and leads to a *trans* C= C linkage.³⁷ This mechanistic proposal is based on the fact that polymerization of (+)-DCMNBE (DCMNBE = 2,3-dicarbomethoxynorbornene) by 1a yields a polymer that contains \sim 75% trans, isotactic dyads and 25% cis, syndiotactic dyads, while 1c yields a polymer that contains $\sim 92\%$ trans, isotactic dyads and $\sim 8\%$

Received: July 19, 2016 Published: September 6, 2016 cis,syndiotactic dyads (Figure 1).³⁷ The olefinic protons in *trans,isotactic* polymer are inequivalent and on the same C=C



Figure 1. Olefinic region of the ¹H NMR spectrum of the poly[(R,R)-DCMNBE] prepared from initiator **1c**. Reprinted with permission from ref 37. Copyright 2012 American Chemical Society.

bond (and therefore coupled to each other with $J_{\rm HH} \sim 16 \, {\rm Hz}^{37}$) while the olefinic protons in *cis,syndiotactic* polymer are inequivalent and on different C=C bonds (and therefore not coupled to each other). It was also shown that W(O)(CH-*t*-Bu)(OHMT)(Pyr)(PMe₂Ph) polymerizes (+)-DCMNBE to give only *cis,syndiotactic*-poly[(+)-DCMNBE].



An interesting question is whether *stereogenic metal control* will direct formation of a copolymer where **A** and **B** are not strictly enantiomers, but have similar structures *and* reactivities toward homopolymerization, are enantiomerically pure, and have "opposite" chirality. If **A** and **B** are significantly different chemically, the resulting polymer could be further manipulated through selective reactions that involve one of the two components within the polymer. We show here that several such *cis,syndiotactic* **A**-*alt*-**B** copolymers can be prepared with Mo (primarily) and W alkylidene initiators.

RESULTS AND DISCUSSION

Initial screening experiments employed the four monomers shown in Figure 2, where the A monomers have the (2R,3R)



Figure 2. First four monomers employed in this study.

configuration and the **B** monomers have the (2S,3S) configuration. The ¹³C NMR spectra of *cis,syndiotactic*-poly(**A**-*alt*-**B**) should reveal four different olefinic carbon resonances, and ¹H NMR spectra *could* reveal up to four first order resonances for four different olefinic protons that are coupled pairwise (eq 2). (Overlap of proton resonances could result in non first order ¹H NMR spectra.) Racemic **A**₁ is known to be polymerized by **1a** to give *cis,syndiotactic*-poly[**A**₁(*R*,*R*)-*alt*-



 $A_1(S,S)$];³⁷ we find that *cis,syndiotactic,alt* polymers are also formed from racemic A_2 , B_1 , and B_2 (see Supporting Information).

Copolymerization of a mixture of 25 equiv of A_1 and 25 equiv of B_2 with 1a (0.1 M in toluene- d_8) as the initiator was complete within seconds to give *cis,syndiotactic*-poly(A_1 -*alt*- B_2). Its partial ¹³C NMR spectrum in CDCl₃ showed primarily four different olefinic resonances (Figure 3, right), while its ¹H NMR spectrum



Figure 3. ¹H NMR (500 MHz, CDCl₃, left) and ¹³C NMR (125 MHz, CDCl₃, right) spectra of *cis,syndiotactic*-poly(A_1 -*alt*- B_2) (olefinic resonances only).

showed four overlapping first order (pseudo triplet) olefinic proton resonances (Figure 3, left). (See Supporting Information for details.) The broad resonance shown between 5.45 and 5.50 ppm in Figure 3 we propose is half of the pattern that arises from *trans,isotactic* A_1A_1 and B_2B_2 "errors" (see Figure 1). The other half of the pattern, along with any (minor) pattern that is characteristic of *cis,syndiotactic* A_1A_1 and B_2B_2 errors (see Figure 1), is buried under the main pattern of four triplets for *cis,syndiotactic* $-poly(A_1-alt-B_2)$ around 5.30 ppm. If we *assume* that only *trans,isotactic* A_1A_1 and B_2B_2 dyad resonances are present under the main four triplet resonance, we can estimate that ~94% of the polymer contains *cis,syndiotactic*-poly($A_1-alt-B_2$) dyads. The olefinic carbon resonances for any errors cannot be identified reliably in the partial carbon NMR spectrum shown in Figure 3.

Copolymerization of 25 equiv of (R,R)-2,3-(CO₂Me)₂norbornene (A_1) and 25 equiv (S,S)-2,3- $(CO_2CH_2CF_3)_2$ norbornene (\mathbf{B}_1) with $\mathbf{1a}$ (0.1 M in toluene- d_8) as the initiator was also complete within seconds. The ¹³C NMR spectrum of the resulting polymer again showed primarily four different olefinic carbon resonances (Figure 4, right), while its ¹H NMR spectrum showed four overlapping first order (pseudo triplet) olefinic proton resonances (Figure 4, left), consistent with the formation of *cis,syndiotactic*-poly(A₁-*alt*-B₁). A virtually identical *cis,syndiotactic*-poly(A₁-*alt*-B₁) polymer was prepared employing 1b as the initiator (see Supporting Information). It is clear from the spectra in Figure 5 that this cis, syndiotactic A_1 -alt- B_1 copolymer contains more trans, isotactic errors than the *cis,syndiotactic* A₁-*alt*-B₂ copolymer described above, most likely as a consequence of the more significant differences in reactivity between A_1 and B_1 than between A_1 and B_2 . The 5.50 ppm resonance was integrated, and the % cis, syndiotactic-poly(A_1 -alt- B_1) dyads were calculated to be ~90%. Two olefinic carbon resonances for the A1A1 and B1B1 "errors" in this case can be seen



Figure 4. ¹H NMR (500 MHz, CDCl₃, left) and ¹³C NMR (125 MHz, CDCl₃, right) spectra of *cis,syndiotactic*-poly(A_1 -*alt*- B_1) (olefinic resonances only).



Figure 5. ¹H NMR (500 MHz, CDCl₃, left) and ¹³C NMR (125 MHz, CDCl₃, right) spectra of *cis,syndiotactic*-poly(A_2 -*alt*- B_2) (olefinic resonances only).

at ~133.0 and 128.3 ppm (Figure 4). The ¹⁹F NMR spectrum of *cis,syndiotactic*-poly(A_1 -*alt*- B_1) also reveals two types of overlapping fluorine resonances for A_1B_1 and B_1B_1 errors (see Supporting Information), integration of which suggests that the % *cis,syndiotactic*-poly(A_1 -*alt*- B_1) dyads is ~80%.

A third *cis,syndiotactic* polymer was prepared through copolymerization of a mixture of 25 equiv of (R,R)-2,3- $(CH_2OAc)_2$ -norbornene (A_2) and 25 equiv of (S,S)-2,3- $(CO_2Et)_2$ -norbornene (B_2) with **1b** as the initiator (Figure 5). The ¹³C NMR spectrum showed primarily four different olefinic carbon resonances, while the ¹H NMR spectrum showed two essentially first order triplet resonances for protons coupled to one another, along with a second order resonance at ~5.23 ppm for two coupled olefinic protons. On the basis of the carbon NMR spectrum we can estimate the number of A_2A_2 and B_2B_2 errors to be on the order of 5%.

A fourth example is *cis,syndiotactic*-poly(A_2 -*alt*- B_1). The ¹H NMR spectrum of *cis,syndiotactic*-poly(A_2 -*alt*- B_1) (Figure 6) provides little evidence that the polymer is relatively regular.



Figure 6. ¹H NMR (500 MHz, CDCl₃, left) and ¹³C NMR (125 MHz, CDCl₃, right) spectra of *cis,syndiotactic*-poly(A_2 -*alt*- B_1) (olefinic resonances only).

However, inspection of the ^{13}C NMR spectrum shows primarily four olefinic resonances, consistent with a relatively high percentage (estimated ~90%) of the proposed structure. The complexity seen in the ^{1}H NMR spectrum can be traced to the overlap and second order nature of the proton resonances. At least four carbon resonances for A_2A_2 and B_1B_1 errors can be seen in the ^{13}C NMR spectrum.

Twelve other Mo and W initiators were explored for making *cis,syndiotactic*-poly(A_1 -*alt*- B_1), but none was as efficient as 1a or 1b, at least according to proton NMR spectroscopy (see Supporting Information for details). The failure of more than two initiators (so far) to produce *cis,syndiotactic* A-*alt*-B copolymers of the type described here is not surprising if one considers the complexity of the stereoregular ROMP reaction²⁹ and the need to control formation of A_1A_1 and B_1B_1 errors. The requirements that the metal has a stereogenic center, that its configuration must switch with each insertion of A or B, and that the polymerization be controlled primarily by the chirality of the stereogenic metal are demanding.

Four additional enantiomerically pure monomers (Figure 7) were prepared, and five $A_x B_y$ combinations were found to give



Figure 7. Four additional monomers.

copolymers with >90% alternating **AB** dyads using **1b** as the initiator, according to their ¹H and ¹³C NMR spectra (see Supporting Information for a complete list of reactions employing A_x and B_y). The percentage of *trans,isotactic* and/or *cis,syndiotactic* errors was estimated to be in the range of 5–10%.

Copolymerization of A_2 and B_3 ((*S*,*S*)-(CO₂-*t*-Bu)₂-norbornene) using 1b yielded a CDCl₃-soluble polymer whose ¹H NMR spectrum showed primarily two resonances, a triplet at 5.34 ppm and a second order resonance at 5.23 ppm that integrated to three times its relative intensity (Figure 8). Weak



Figure 8. ¹H NMR (500 MHz, CDCl₃, left) and ¹³C NMR (125 MHz, CDCl₃, right) spectra of *cis,syndiotactic*-poly(A_2 -*alt*- B_3) (olefinic resonances only).

and broad resonances near 5.50 and 5.45 can be attributed to *trans,isotactic* errors. The presence of primarily four olefinic resonances in the ¹³C NMR spectrum at 134.0, 133.3, 130.8, and 130.6 ppm suggests that the *cis,syndiotactic*-poly(A_2 -*alt*- B_3) structure is of the order of 90%.

Copolymerization of A_1 and (S,S)- $(CH_2OAc)_2$ -norbornene (B_4) proceeded smoothly to give another CDCl₃ soluble polymer. The ¹H NMR spectrum of the isolated polymer showed two pairs of overlapping olefinic proton resonances and

weak resonances at 5.45-5.50 for *trans,isotactic* errors (Figure 9). However, the ¹³C NMR spectrum showed primarily four olefinic



Figure 9. ¹H NMR (500 MHz, CDCl₃, left) and ¹³C NMR (125 MHz, CDCl₃, right) spectra of *cis,syndiotactic*-poly(A_1 -*alt*- B_4) (olefinic resonances only).

resonances, which confirm that the polymer has largely the *cis,syndiotactic,alt* structure. Given the successful copolymerization of (R,R)- $(CH_2OAc)_2$ -norbornene (A_2) with (S,S)- $(CO_2Et)_2$ -norbornene $(B_2;$ Figure 5), the formation of *cis,syndiotactic*-poly $(A_1$ -*alt*- B_4) is not surprising. The resonance attributed to *trans,isotactic* errors is much more pronounced when **1a** is used as the initiator (see Supporting Information).

The copolymer derived from A_3 ((*R*,*R*)-(CO₂-*t*-Bu)₂norbornene) and B_5 ((*S*,*S*)-(CH₂OMe)₂-norbornene) showed two higher ordered olefinic proton resonances of equal intensity in CDCl₃, (Figure 10) along with resonances for A_3A_3 and B_5B_5



Figure 10. ¹H NMR (500 MHz, CDCl₃, left) and ¹³C NMR (125 MHz, CDCl₃, right) spectra of *cis,syndiotactic*-poly(A_3 -*alt*- B_5) (olefinic resonances only).

errors. The two overlapping proton resonances were not wellresolved, making assessment of the errors in the polymer structure difficult. However, the ¹³C NMR spectrum showed primarily four olefinic resonances, consistent with formation of largely *cis,syndiotactic*-poly(A_3 -*alt*- B_5). Copolymerization of A_1 and B_5 gives *cis,syndiotactic*-poly(A_1 -*alt*- B_5) (Figure 11), the proton NMR spectrum of which resembles that of *cis,syndiotactic*poly(A_2 -*alt*- B_2) (Figure 5). The second order olefinic proton resonance at 5.33 ppm was shifted to higher frequency with respect to two coupled triplet proton resonances, revealing the resonances for *trans,isotactic* errors. Copolymerization of A_3 with B_2 gave *cis,syndiotactic*-poly(A_3 -*alt*- B_2) (Figure 12), which contains <10% errors.

It is important to establish whether the tungsten analogue of **1b** (**1b**_w) is an equally efficient catalyst. Addition of 50 equiv of *rac*-DCBNBE (DCBNBE = 2,3-dicarbo-*t*-butoxynorbornene) to a toluene solution of **1b**_w led to full consumption of the monomer within 10 min. Only two pseudo triplet olefinic proton resonances (${}^{3}J_{\text{HH}} = 10 \text{ Hz}$) are present in the ¹H NMR spectrum



Figure 11. ¹H NMR (500 MHz, CDCl₃, left) and ¹³C NMR (125 MHz, CDCl₃, right) spectra of *cis,syndiotactic*-poly(A_1 -*alt*- B_5) (olefinic resonances only).



Figure 12. ¹H NMR (500 MHz, CDCl₃, left) and ¹³C NMR (125 MHz, CDCl₃, right) spectra of *cis,syndiotactic*-poly(A_3 -*alt*- B_2) (olefinic resonances only).

of the resulting polymer (Figure 13a). The ¹³C NMR spectrum is also sharp and free of any significant fine structure associated with structural irregularities (see Supporting Information). These results are consistent with a *cis,syndiotactic,alt* structure for the polymer. The two small broad resonances assigned to



Figure 13. ¹H NMR (500 MHz, CDCl₃) spectra for *cis,syndiotactic* polymers synthesized using **1b**_w (olefinic resonances only).

trans, isotactic dyads, in the olefin region of poly(rac-DCBNBE) prepared from 1a, are absent from the spectrum of poly(rac-DCBNBE) prepared from $1b_{w}$. The reason is that polymerization of (S,S)-DCBNBE with $1b_w$ gives cis,syndiotacticpoly[(S,S)-DCBNBE] with the ¹H NMR spectrum shown in Figure 13b; there is no evidence for a *trans,isotactic* structure. The olefinic proton resonances of *cis,syndiotactic*-poly[(S,S)-DCBNBE] are located in the middle of the olefinic proton resonances for poly(rac-DCBNBE) (Figure 13a), which makes it difficult to assess the percentage of microstructural errors formed in this copolymer using ¹H NMR spectroscopy. However, when racemic DCENBE (DCENBE = 2,3-dicarboethoxynorbornene) and rac-DCMNBE are polymerized by 1bw under similar conditions, the olefinic proton triplet resonances are broadened, Figures 13c and 13e. The positions of the minor component within the olefinic proton resonances of poly(rac-DCENBE) and poly(rac-DCMNBE) are visible and can be unambiguously ascribed to cis,syndiotactic dyads in the largely cis,syndiotactic,alt structure (compare Figures 13c-13e).

As reported previously,³⁷ (+)-DCMNBE and *rac*-DCMNBE are polymerized at approximately the same rate using W(O)-(CHCMe₃)(OHMT)(Pyr)(PMe₂Ph). In contrast, (-)-DCBNBE and *rac*-DCBNBE are polymerized at different rates. The sharp olefinic resonances in the ¹H and ¹³C spectra of poly(*rac*-DCBNBE) prepared from **1a** as the initiator are consistent with a lower percentage of *cis,syndiotactic* errors. A similar trend is seen when one inspects the ¹H and ¹³C NMR spectra of poly(*rac*-DCBNBE) using **1b**_w (Figure 13a).

The proton and carbon NMR spectra of the A_3 -alt- B_2 copolymers derived from 1a, 1b, and 1b_w are compared in Figures 14 and 15. Initiators 1b and 1b_w appear to yield the



Figure 14. ¹³C NMR spectra (125 MHz, CDCl₃) of *cis,syndiotactic*-poly(A_3 -*alt*- B_2) (olefinic resonances only) formed with initiators 1a, 1b, and $1b_w$.

highest percentages of *cis,syndiotactic* A_3 -*alt*- B_2 structures with *trans,isotactic* errors being formed when 1b is employed and *cis,syndiotactic* errors being formed when 1b_w is employed.

CONCLUSIONS

Cis,syndiotactic **A**-*alt*-**B** copolymers, where **A** and **B** are two enantiomerically pure *trans*-2,3-disubstituted-5,6-norbornenes with "opposite" chiralities, can be prepared with stereogenic-atmetal initiators of the type M(NR)(CHR')(OHMT)(pyrrolide)(R = 1-adamantyl or 2,6-Me₂C₆H₃; R' = CMe₂Ph; M = Mo or W). The errors when Mo initiators are employed are primarily *trans,isotactic* **AA** and **BB** dyads, while the errors when a W initiator is employed are *cis,syndiotactic* **AA** and **BB** dyads. Formation of a high percentage of alternating **AB** copolymer



Figure 15. ¹H NMR spectra (500 MHz, CDCl₃) of *cis,syndiotactic*-poly(A_3 -*alt*- B_2) (olefinic resonances only) formed with initiators 1a, 1b, and $1b_w$.

linkages relies on an inversion of chirality at the metal center with each propagating step and faster formation of an **AB** sequence than an **AA** or **BB** sequence as a consequence of a preferred diastereomeric relationship between the chirality at the metal and the chirality of the monomer.

EXPERIMENTAL SECTION

Representative Polymerization. A mixture of 21.0 mg (0.1 mmol, 25 equiv) of A_1 and 34.6 mg (0.1 mmol, 25 equiv) of B_1 in 0.5 mL of toluene- d_8 was added to a solution of 3.0 mg (0.004 mmol) of **1b** in 0.5 mL of toluene- d_8 . The reaction mixture thickened within seconds. ¹H NMR spectroscopy was used to monitor the course of the reaction. Once complete, the reaction mixture was exposed to air and poured into 35 mL of MeOH. The precipitated *cis,syndiotactic*-poly(A_1B_1) was allowed to settle. The solvent was decanted, and the polymer was dried *in vacuo*.

ASSOCIATED CONTENT

Supporting Information

The Supporting Information is available free of charge on the ACS Publications website at DOI: 10.1021/acscentsci.6b00200.

Experimental details for all reactions and all supporting NMR characterization of polymers (PDF)

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Notes

The authors declare no competing financial interest.

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REFERENCES

(1) Odian, G. *Principles of Polymerization*, 4th ed.; John Wiley & Sons, Inc.: Hoboken, NJ, 2004.

(2) Coates, G. W. Precise Control of Polyolefin Stereochemistry Using Single-Site Metal Catalysts. *Chem. Rev.* **2000**, *100*, 1223–1252.

(3) Cheng, M.; Lobkovsky, E. B.; Coates, G. W. Catalytic Reactions Involving C1 Feedstocks: New High-Activity Zn(II)-Based Catalysts for the Alternating Copolymerization of Carbon Dioxide and Epoxides. J. Am. Chem. Soc. **1998**, 120, 11018–11019.

(4) Super, M.; Berluche, E.; Costello, C.; Beckman, E. Copolymerization of 1,2-Epoxycyclohexane and Carbon Dioxide Using Carbon Diioxide as Both Reactant and Solvent. *Macromolecules* **1997**, *30*, 368– 372.

(5) Darensbourg, D. J.; Holtcamp, M. W. Catalytic Activity of Zinc(II) Phenoxides Which Possess Readily Accessible Coordination Sites. Copolymerization and Terpolymerization of Epoxides and Carbon Dioxide. *Macromolecules* **1995**, *28*, 7577–7579.

(6) Vehlow, K.; Wang, D.; Buchmeiser, M. R.; Blechert, S. Alternating Copolymerizations Using a Grubbs-Type Initiator with an Unsymmetrical, Chiral N-Heterocyclic Carbene Ligand. *Angew. Chem., Int. Ed.* **2008**, 47, 2615–2618.

(7) Song, A.; Parker, K. A.; Sampson, N. S. Synthesis of Copolymers by Alternating ROMP (AROMP). J. Am. Chem. Soc. 2009, 131, 3444–3445.

(8) Song, A.; Parker, K. A.; Sampson, N. S. Cyclic Alternating Ring-Opening Metathesis Polymerization (CAROMP). Rapid Access to Functionalized Cyclic Polymers. *Org. Lett.* **2010**, *12*, 3729–3731.

(9) Sutthasupa, S.; Shiotsuki, M.; Masuda, T.; Sanda, F. Alternating Ring-Opening Metathesis Copolymerization of Amino Acid Derived Norbornene Monomers Carrying Nonprotected Carboxy and Amino Groups Based on Acid-Base Interaction. J. Am. Chem. Soc. 2009, 131, 10546–10551.

(10) Nakade, H.; Ilker, M. F.; Jordan, B. J.; Uzun, O.; LaPointe, N. L.; Coughlin, E. B.; Rotello, V. M. Duplex strand formation using alternating copolymers. *Chem. Commun.* **2005**, 3271–3273.

(11) Lichtenheldt, M.; Wang, D.; Vehlow, K.; Reinhardt, I.; Kühnel, C.; Decker, U.; Blechert, S.; Buchmeiser, M. R. Alternating Ring-Opening Metathesis Copolymerization by Grubbs-Type Initiators with Unsymmetrical N-Heterocyclic Carbenes. *Chem. - Eur. J.* **2009**, *15*, 9451–9457.

(12) Vehlow, K.; Lichtenheldt, M.; Wang, D.; Blechert, S.; Buchmeiser, M. R. Alternating Ring-Opening Metathesis Copolymerization of Norborn-2-ene with cis-Cyclooctene and Cyclopentene. *Macromol. Symp.* **2010**, *296*, 44–48.

(13) Ilker, M. F.; Coughlin, E. B. Alternating Copolymerizations of Polar and Nonpolar Cyclic Olefins by Ring-Opening Metathesis Polymerization. *Macromolecules* **2002**, *35*, 54–58.

(14) Bornand, M.; Torker, S.; Chen, P. Mechanistically Designed Dual-Site Catalysts for the Alternating ROMP of Norbornene and Cyclooctene. *Organometallics* **2007**, *26*, 3585–3596.

(15) Romulus, J.; Tan, L.; Weck, M.; Sampson, N. S. Alternating Ring-Opening Metathesis Polymerization Copolymers Containing Charge-Transfer Units. *ACS Macro Lett.* **2013**, *2*, 749–752.

(16) Daeffler, C. S.; Grubbs, R. H. Catalyst-Dependent Routes to Ring-Opening Metathesis Alternating Copolymers of Substituted Oxanorbornenes and Cyclooctene. *Macromolecules* **2013**, *46*, 3288–3292.

(17) Abbas, M.; Wappel, J.; Slugovc, C. Alternating Diene Metathesis Polycondensation (ALTMET) – Opitimizing Catalyst Loading. *Macromol. Symp.* **2012**, *311*, 122–125.

(18) Buchmeiser, M. R.; Ahmad, I.; Gurram, V.; Kumar, P. S. Pseudo-Halide and Nitrate Derivatives of Grubbs andGrubbs-Hoveyda Initiators: Some Structural Features Related to the Alternating Ring-Opening Metathesis Copolymerization of Norborn-2-ene with Cyclic Olefins. *Macromolecules* **2011**, *44*, 4098–4106.

(19) Demel, S.; Slugovc, C.; Stelzer, F.; Fodor-Csorba, K.; Galli, G. Alternating Diene Metathesis Polycondensation (ALTMET) – AVersatile Tool for the Preparation of Perfectly Alternating AB Copolymers. *Macromol. Rapid Commun.* **2003**, *24*, 636–641.

(20) Choi, T.-L.; Rutenberg, I. M.; Grubbs, R. H. Synthesis of A,B-Alternating Copolymers by Ring-Opening-Insertion-Metathesis Polymerization. *Angew. Chem., Int. Ed.* **2002**, *41*, 3839–3841.

(21) Al Samak, B.; Amir-Ebrahimi, V.; Corry, D. G.; Hamilton, J. G.; Rigby, S.; Rooney, J. J.; Thompson, J. M. Dramatic solvent effects on ring-opening metathesis polymerization of cycloalkenes. *J. Mol. Catal. A: Chem.* **2000**, *160*, 13–21. (22) Konzelman, J.; Wagener, K. B. Acyclic Diene Metathesis (ADMET) Polymerization. Synthesis of Perfectly Alternating Copolymers from a Single Monomer. *Macromolecules* **1996**, *29*, 7657–7660.

(23) Wu, Z.; Grubbs, R. H. Preparation of Alternating Copolymers from the Ring-Opening Metathesis Polymerization of 3-Methylcyclobutene and 3,3-Dimethylcyclobutene. *Macromolecules* **1995**, *28*, 3502– 3508.

(24) Ding, L.; Zheng, X.-Q.; Lu, R.; An, J.; Qiu, J. Perfectly ABalternating copolymers via alternating diene metathesis polymerization: one-step synthesis, characterization and properties. *Polym. Int.* **2014**, *63*, 997–1002.

(25) Tan, L.; Parker, K. A.; Sampson, N. S. A Bicyclo[4.2.0]octene-Derived Monomer Provides Completely Linear Alternating Copolymers via Alternating Ring-Opening Metathesis Polymerization (AROMP). *Macromolecules* **2014**, *47*, 6572–6579.

(26) Lee, H.-K.; Bang, K.-T.; Hess, A.; Grubbs, R. H.; Choi, T.-L. Multiple Olefin Metathesis Polymerization That Combines All Three Olefin Metathesis Transformations: Ring-Opening, Ring-Closing, and Cross Metathesis. J. Am. Chem. Soc. **2015**, *137*, 9262–9265.

(27) Elling, B. R.; Xia, Y. Living Alternating Ring-Opening Metathesis Polymerization Based on Single Monomer Additions. *J. Am. Chem. Soc.* **2015**, *137*, 9922–9926.

(28) Tan, L.; Li, G.; Parker, K. A.; Sampson, N. S. Ru-Catalyzed Isomerization Provides Access to Alternating Copolymers via Ring-Opening Metathesis Polymerization. *Macromolecules* **2015**, *48*, 4793– 4800.

(29) Schrock, R. R. Synthesis of Stereoregular Polymers through Ring-Opening Metathesis Polymerization. *Acc. Chem. Res.* **2014**, *47*, 2457–2466.

(30) Forrest, W. P.; Weis, J. G.; John, J. M.; Axtell, J. C.; Simpson, J. H.; Swager, T. M.; Schrock, R. R. Stereospecific Ring-Opening Metathesis Polymerization of Norbornadienes Employing Tungsten Oxo Alkylidene Initiators. *J. Am. Chem. Soc.* **2014**, *136*, 10910–10913.

(31) Jeong, H.; John, J. M.; Schrock, R. R.; Hoveyda, A. H. Synthesis of Alternating *trans*-AB Copolymers Employing Ring-Opening Metathesis Polymerization Initiated by Molybdenum Alkylidenes. *J. Am. Chem. Soc.* **2015**, *137*, 2239–2242.

(32) Jeong, H.; Ng, V. W. L.; Börner, J.; Schrock, R. R. Stereoselective Ring-Opening Metathesis Polymerization (ROMP) of Methyl-N-(1phenylethyl)-2-azabicyclo[2.2.1]hept-5-ene-3-carboxylate by Molybdenum and Tungsten Initiators. *Macromolecules* **2015**, *48*, 2006–2012.

(33) Autenrieth, B.; Jeong, H.; Forrest, W. P.; Axtell, J. C.; Ota, A.; Lehr, T.; Buchmeiser, M. R.; Schrock, R. R. Stereospecific Ring-Opening Metathesis Polymerization (ROMP) of *Endo*-Dicyclopentadiene by Molybdenum and Tungsten Catalysts. *Macromolecules* **2015**, *48*, 2480– 2492.

(34) Autenrieth, B.; Schrock, R. R. Stereospecific Ring-Opening Metathesis Polymerization (ROMP) of Norbornene and Tetracyclododecene by Mo and W Initiators. *Macromolecules* **2015**, *48*, 2493–2503.

(35) Hyvl, J.; Autenrieth, B.; Schrock, R. R. Proof of Tacticity of ROMP Polymers Through Post Polymerization Modification. *Macromolecules* **2015**, *48*, 3148–3152.

(36) Flook, M. M.; Ng, V. W. L.; Schrock, R. R. Synthesis of *cis,syndiotactic* ROMP Polymers Containing Alternating Enantiomers. *J. Am. Chem. Soc.* **2011**, *133*, 1784–1786.

(37) Flook, M. M.; Börner, J.; Kilyanek, S.; Gerber, L. C. H.; Schrock, R. R. Five-Coordinate Rearrangements of Metallacyclobutane Intermediates During Ring-Opening Metathesis Polymerization (ROMP) of 2,3-Dicarboalkoxynorbornenes by Molybdenum and Tungsten Monoalkoxide Pyrrolide (MAP) Initiators. *Organometallics* **2012**, *31*, 6231–6243.

(38) Hamilton, J. G.; Ivin, K. J.; Rooney, J. J. ¹³C N.m.r. Spectra of Ring-opened Polymers of 1-Methylbicyclo[2.2.1]hept-2-ene and their Hydrogenated Products *Brit. Br. Polym. J.* **1984**, *16*, 21–33.

(39) Hamilton, J. G.; Ivin, K. J.; Rooney, J. J.; Waring, L. C. Alternating Copolymerization of Enantiomers of 1-Methylbicyclo[2.2.1]hept-2-ene by a Metathesis Catalyst. *J. Chem. Soc., Chem. Commun.* **1983**, 159–161.