

# Different models provide equivalent predictive power for cross-biome response of leaf respiration to temperature

Mark A. Adams<sup>a,1</sup>, Heinz Rennenberg<sup>b</sup>, and Jorg Kruse<sup>b</sup>

In their recent paper, Heskell et al. (1) claim for leaf respiration that “Neither current physiological nor terrestrial biosphere models adequately describe its short-term temperature response. . .” and present an empirical, polynomial model of leaf respiration suitable for all biomes and plant functional types. Our polynomial model (2), an extension of the Arrhenius function, provides the same predictive power as described by Heskell et al. (1). In fact, different approaches and methodologies have yielded models of the same form, and generate close to identical results (Figs. 1 and 2).

The two models have very different origins. Heskell et al. (1) used well-established gas exchange methods and equipment to measure leaf respiration of a wide range of species and life forms, using almost identical protocols at many sites across the globe. Their dataset is a significant development. Our approach has been based on calorimetry and was summarized in a Tansley review in 2011 (2). Recently, Noguchi et al. (3) provided an independent test of whether the polynomial model developed using calorimetric techniques could be confirmed using gas exchange techniques [as used by Heskell et al. (1)]. In addition to providing a clear positive answer to this question, they highlighted that activation energies ( $E_o$ ) for the entire respiratory pathway could vary little between growth temperatures. In our studies (2, 4) and the study of Noguchi et al. (3), dynamic change of  $E_o$  is hard to explain. However, the strong correlation between  $E_o$  (at 0 °C reference temperature) and  $\delta$  (Fig. 2), and between the  $b$ - and  $c$ -parameters of the model of Heskell et al. (1) (Fig. 2),

suggests a common mechanistic basis of respiratory regulation for all plant species. Likewise, prompt change in respiratory capacity with growth temperature (3, 4), which is proportional to change in the logarithm of respiration rate at a low reference temperature in our model (2) or the  $a$ -parameter in the model of Heskell et al. (1) (Fig. 1), is seemingly an explanation of plant adaptation (1, 4), albeit we lack a full physiological/anatomical explanation.

Congruence of papers by Heskell et al. (1), Noguchi et al. (3) and Kruse et al. (2, 4) is support for how science moves forward. Congruence, not just of data from across the globe but of knowledge derived using different methods and techniques, remains a central test of the substance of any new idea. It is not sufficient that the same method used repeatedly by the same group of researchers with different study subjects yields the same result.

As noted by Heskell et al. (1), empirical models are useful additions to the armory of those individuals charged with the responsibility of predicting global carbon budgets. Researchers and modelers now have a choice. They may use a fully abstract polynomial function to describe respiration (1), and set aside concepts such as the activation energy of enzyme reactions, or they may use a polynomial function derived from classic Arrhenius kinetics (2).

More mechanistic, and preferably process-based, models (e.g., ref. 5) are still needed to synthesize the complexity of respiration, and to help frame questions about its regulation.

<sup>a</sup>Centre for Carbon, Water and Food, University of Sydney, Camden 2570, NSW, Australia; and <sup>b</sup>Institut für Forstwissenschaften, University of Freiburg, 79085 Freiburg, Germany

Author contributions: M.A.A., H.R., and J.K. designed research; J.K. performed research; M.A.A. and J.K. analyzed data; and M.A.A., H.R., and J.K. wrote the paper.

The authors declare no conflict of interest.

<sup>1</sup>To whom correspondence should be addressed. Email: mark.adams@sydney.edu.au.

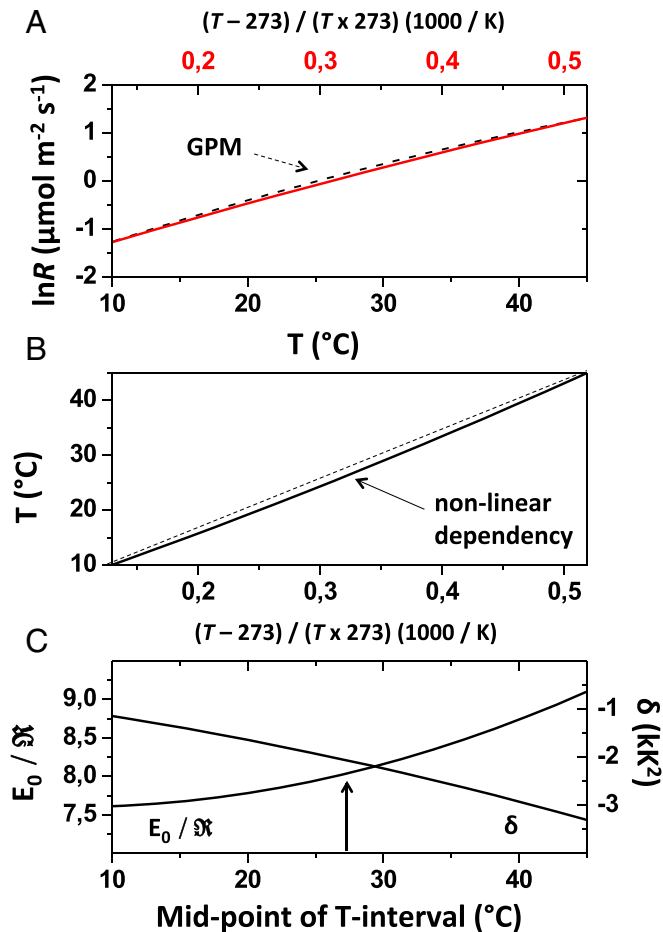


Fig. 1. Equivalence between polynomial and extended Arrhenius models. (A) Extended Arrhenius model, where  $E_0$  is the activation energy and  $R$  is the ideal gas constant [ $\ln R = \ln R_{ref} + \frac{E_0(Ref)}{R} \times \frac{T-273}{T \times 273} + \delta \times (\frac{T-273}{T \times 273})^2 = -2.27 + 8.0 \times \frac{T-273}{T \times 273} - 2.1 \times (\frac{T-273}{T \times 273})^2$ ; red line] can be fitted with  $R^2 > 0.99$  to the global polynomial model ( $\ln R = a + b \times T + c \times T^2 = -2.23 + 0.1012 \times T - 0.0005 \times T^2$ ; black dotted line;  $T$  range of 10–45  $^{\circ}\text{C}$ ). The fitted value of  $\ln R_{ref}$  differs by less than 2% from  $a = -2.23$ , so that  $E_0(Ref)$  and the  $\delta$ -parameter can be predicted with similar precision from  $b$ - and  $c$ -parameters. GPM, global polynomial R–T model. (B) Nonlinear dependency between  $T$  ( $^{\circ}\text{C}$ ) and reciprocal temperature (1,000/K):  $T$  ( $^{\circ}\text{C}$ )  $\approx 73.5 \times x + 25.5 \times x^2$ ; where  $x = \frac{T-273}{T \times 273}$ . (C) For accurate prediction ( $\pm 2\%$ ) of  $E_0(Ref)$  and the  $\delta$ -parameter from  $b$ - and  $c$ -parameters, the temperature dependency between  $T$  ( $^{\circ}\text{C}$ ) and  $x$  must be accounted for (*sensu B*). For any given  $b$ - and  $c$ -parameters, Arrhenius-type parameters can thus be predicted from the midpoint of the  $T$  interval under consideration (i.e., 27.5  $^{\circ}\text{C}$  for the global temperature range).

