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DNA Polymerases η and ζ Combine to Bypass O^2 -[4-(3-Pyridyl)-4-oxobutyl]thymine, a DNA Adduct Formed from Tobacco Carcinogens

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Abstract

4-(Methylnitrosamino)-1-(3-pyridyl)-1-butanone (NNK) and *N*'-nitrosonornicotine (NNN) are important human carcinogens in tobacco products. They are metabolized to produce a variety 4-(3-pyridyl)-4-oxobutyl (POB) DNA adducts including O^2 -[4-(3-pyridyl)-4-oxobut-1-yl]thymidine (O^2 -POB-dT), the most abundant POB adduct in NNK- and NNN-treated rodents. To evaluate the mutagenic properties of O^2 -POB-dT, we measured the rate of insertion of dNTPs opposite and extension past O^2 -POB-dT and O^2 -Me-dT by purified human DNA polymerases η , κ , ι , and yeast polymerase ζ in vitro. Under conditions of polymerase in excess, polymerase η was most effective at the insertion of dNTPs opposite O^2 -alkyl-dTs. The time courses were biphasic suggesting the formation of inactive DNA-polymerase complexes. The k_{pol} parameter was reduced approximately 100-fold in the presence of the adduct for pol η , κ , and ι . Pol η was the most reactive polymerase for the adducts due to a higher burst amplitude. For all three polymerases, the nucleotide preference was dATP > dTTP \gg dGTP and dCTP. Yeast pol ζ was most effective in bypassing the adducts; the k_{cat}/K_m values were reduced only 3-fold in the presence of the adducts. The identity of the nucleotide opposite the O^2 -alkyl-dT did not significantly affect the ability of pol ζ to bypass the adducts. The data support a model in which pol η inserts ATP or dTTP opposite O^2 -POB-dT, and then, pol ζ extends past the adduct.

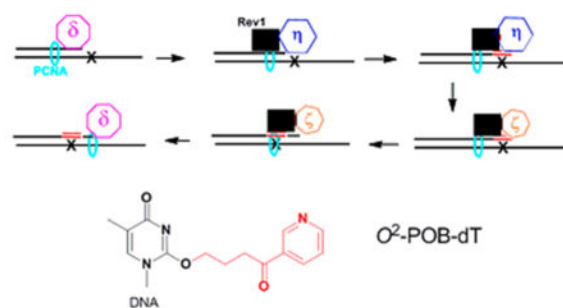
Graphical Abstract

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Supporting Information

The Supporting Information is available free of charge on the ACS Publications website at DOI: 10.1021/acs.chemrestox.5b00468. Kinetic plots for the incorporation opposite and extension past dT, O^2 -Me-dT, and O^2 -POB-dT by pol η , ζ , ι , and κ are presented (PDF)



INTRODUCTION

Lung cancer is the most common cause of cancer death in the US. Cigarette smoking is the main risk factor for lung cancer.¹ Tobacco specific nitrosamines (TSNAs) are a significant class of tobacco carcinogens.²⁻⁴ *N*'-Nitrosornicotine (NNN) and 4-(methylnitrosamino)-1-(3-pyridyl)-1-butanone (NNK) are the two most potent TSNAs in causing lung tumors in animals.²⁻⁴ NNK and NNN are considered to be among the causative agents of lung and oral cavity cancers in individuals who use tobacco products.

Both NNN and NNK are metabolically activated to alkylating agents. NNK produces both a methyl and a 4-(3-pyridyl)-4-oxobutyl (POB) diazonium ion, while NNN produces the POB-diazonium ion.⁵ The methylation pathway, with the formation of the mutagenic O^6 -methyl-2'-deoxyguanosine, is an important contributor to mutagenicity and carcinogenicity of NNK.^{6,7} The pyridyloxobutylation pathway, although less well studied, also plays a role in carcinogenesis in NNK and NNN. Four POB adducts have been characterized in rodents, in which the POB-group is bound to the *N*7- and O^6 -positions of dG and the O^2 -positions of dC and dT.^{13,34-36} The chemical toxicology of O^6 -POB-dG has been the most studied. It is repaired by O^6 -alkylguanine-DNA alkyltransferase⁸⁻¹⁰ and is mutagenic in bacteria and human cells causing mostly G to A transitions with some A to T transversions.^{11,12} DNA polymerase η is the most reactive Y-family polymerase toward O^6 -POB-dG.¹³

O^2 -[4-(3-Pyridyl)-4-oxobut-1-yl]thymidine (O^2 -POB-dT) may also play a role in carcinogenesis. Its formation is shown in Scheme 1. In long-term studies, in which rats are continuously administered NNK or NNN, O^2 -POB-dT is the most abundant POB-adduct.¹⁴⁻¹⁷ NNK-treated transgenic mice produce approximately equal amounts of mutations at G/C and A/T base pairs.^{18,19} If the mutagenesis were solely due to the methyl diazonium ion pathway, then the mutagenic spectrum would be dominated by G/C to A/T mutations caused by O^6 -Me-dG, as observed in mice treated with *N,N*-dimethyl-*N*-nitrosamine.²⁰ Therefore, O^2 -POB-dT is a potential cause of the A/T mutations in NNK-treated rodents.

DNA polymerases are crucial in maintaining genome integrity. Seventeen human DNA polymerases are involved in high fidelity DNA replication, DNA repair, and replication of DNA damage.²¹⁻²³ Polymerases α , γ , δ , ϵ , and telomerase are involved in the high fidelity replication of the genome. The Y-family polymerases η , κ , ι , and Rev1 as well as the B-family pol ζ are involved in translesion DNA synthesis (TLS).^{21,24} Several other

EXPERIMENTAL PROCEDURES

General

[³²P]ATP (6000 Ci/mmol) was purchased from PerkinElmer and T4 polynucleotide kinase from USB/Affymetrix. The dNTPs (ultrapure grade) were purchased from GE Healthcare, and the concentrations were determined by UV absorbance.⁴⁵ yPol ζ was purchased from Enzymax (Lexington, KT).

Oligodeoxynucleotides

Oligodeoxynucleotides containing *O*²-Me-dT and *O*²-POB-dT were synthesized, purified, and characterized as described.^{42,46} The oligodeoxynucleotide containing 6-carboxyfluorescein was purchased from Integrated DNA Technologies (Coralville, Iowa). The sequences of the oligodeoxynucleotides are shown in Chart 1. The concentrations of oligodeoxynucleotides were determined from the absorbance at 260 nm, using the method of Borer⁴⁷ in which it was assumed that the spectroscopic properties of *O*²-Me-dT and *O*²-POB-dT were identical to those of the dT. The primer was ³²P-labeled with γ-[³²P]ATP and annealed with a 20% excess of the template.⁴⁸

Polymerase Purification and Concentration

Polymerases κ, η, and ι were purified from Sf9 insect cells as described.^{49–51} The polymerases are full length proteins with an N-terminal His-tag. The active concentration of the enzymes were determined by evaluating the magnitude of the burst using an undamaged template (data not shown). The polymerases were reacted with 300 nM DNA (P15/T24G) and 100 μM dCTP and fit to eq 2. The amplitude was set as the polymerase concentration.

Polymerase Kinetics

Enzyme reactions were initiated by mixing equal volumes of the DNA (P15/T24)/polymerase solution in buffer with dNTP and Mg²⁺ in H₂O at 37 °C. The final buffer concentration was 40 mM Tris-HCl (pH 8.0), 3 mM DTT, 10 μg/mL BSA, and 2.5% glycerol. The reactions were quenched with equal volumes of STOP solution containing 10% 0.5MNa₂EDTA, 90% formamide, 0.025% (w/v) xylene cyanol, and 0.025% (w/v) bromophenol blue. Rapid reactions were performed on a RQF-3 (Kin-Tek Corporation) and quenched with 300 mM EDTA (pH 8).

The time course of the reaction was analyzed by denaturing PAGE, and the radioactivity on the gel was visualized with a Typhoon 9200. The progress of the reaction was quantitated by dividing the total radioactivity in the product band(s) by the radioactivity in the product and reactant bands. Multiple product bands appeared when the incorrect dNTP was added to the reaction.

Polymerase-DNA Binding Experiments

The binding affinity of the DNA to the polymerases was evaluated with fluorescence anisotropy. The DNA consisted of the primer strand (P15) modified on the 5'-end with 6-carboxyfluorescein and the template strand containing dT, *O*²-Me-dT, or *O*²-POB-dT. The fluorescence anisotropy was measured with 1 nM DNA and 0–100 nM polymerase at 37 °C.

Anisotropy was measured with a Tecan Safire 2 plate reader with half-volume 96 well plates. The excitation and emission wavelengths were 471 and 525 nm, respectively, with an emission bandwidth of 20 nm.

Data Analysis

Data were fitted by nonlinear regression using the program Prism, version 5, for Windows (GraphPad Software, San Diego, California, USA, www.graphpad.com). The V_{max} and K_m values were determined by fitting the data to eq 1; v_0 is the initial rate, E_0 the enzyme, and N_0 the dNTP concentrations.

$$v_0 = \frac{V_{max}N_0}{N_0 + K_m} \quad (1)$$

The time courses for the insertion enzyme in excess reactions with O^2 -Me-dT and O^2 -POB-dT were fitted to eq 2 in which P is the product concentration, A the burst amplitude, k the burst rate constant, k_{ss} the steady-state rate constant, and t the time. The dNTP concentration dependence was quantified by fitting the burst equation parameters to the hyperbolic eq 1 in which v_0 is the value of the parameter at a specific [dNTP], and V_{max} is the maximum value of that parameter. The time courses for the enzyme in excess extension reactions with O^2 -Me-dT and O^2 -POB-dT were fitted to eq 3 in which P is the product concentration, A the amplitude, and k the first-order rate constant. The dNTP concentration dependence on the kinetic parameters in eqs 2 and 3 was quantified by fitting the parameters to the hyperbolic eq 1.

$$P = A(1 - e^{-kt}) + k_{ss}t \quad (2)$$

$$P = A(1 - e^{-kt}) \quad (3)$$

The K_d^{DNA} was determined by fitting the measured anisotropy (A) to eq 4, in which K is K_d^{DNA} , E_0 and D_0 (set to 1 nM) are the total concentrations of the polymerase and DNA, respectively, and A_{max} is the increase in anisotropy caused by the polymerase binding to the DNA.

$$A = \frac{A_{max}}{[E]_0} \frac{([E]_0 + [D]_0 + K) - \sqrt{([E]_0 + [D]_0 + K)^2 - 4[E]_0[D]_0}}{2} \quad (4)$$

RESULTS

Running Start Incorporation of dNTPs Opposite O^2 -alkyl-dT

The incorporation of dNTPs into a DNA substrate containing O^2 -alkyl-dT was examined with the running start analyses as illustrated in Figure 1. The lowest bands represent the 12-mer primer, while the top band is the 24-mer product. The left-most lanes show that templates containing dT are rapidly replicated by all three polymerases. With O^2 -alkyl-dT in the template, the gels display a band corresponding to a 15-mer, indicating that the polymerases stalled prior to the adduct. A more faint band just above the 15-mer indicates that extension past the adduct is also inefficient.

DNA-Polymerase Binding to Oligodeoxynucleotides Containing O^2 -alkyl-dT

The binding affinities of the DNA substrates to the polymerases were measured by fluorescence anisotropy using a fluorescein-labeled primer strand. The increase in anisotropy was plotted against polymerase concentration and fitted to the quadratic equation as shown in Figure 2. The results, presented in Table 1, indicate that the modification, either methyl or POB, does not affect the affinity of the DNA to the polymerase.

Steady-State Incorporation of dNTPs Opposite O^2 -alkyl-dT

The Michaelis–Menten kinetic parameters were determined with polymerase concentration from 0.01 to 0.1 nM, 10 nM DNA, and 50 μ M to 1 mM dNTP. The kinetic parameters are presented in Tables 2–4 for the different polymerases. For each enzyme, we also evaluated the correct incorporation of dATP opposite dT and the misincorporation of dTTP opposite dT. The relative k_{cat}/K_m values are visualized in Figure 2. Fluorescence anisotropy was determined with 1 nM DNA in which the 5'-terminus of the primer strand was modified with fluorescein with variable concentrations of (A) pol η , (B) pol ι , and (C) pol κ . The DNA contained dT (open circle), O^2 -Me-dT (black circle), and O^2 -POB-dT (square). The data points are the mean \pm standard deviation of three determinations. The lines are the best fit to eq 3.

As shown in Figure 3, the incorporation of dATP opposite dT was >1000-fold faster than that of dTTP opposite dT for all three polymerases. The increase in k_{cat}/K_m was due to both k_{cat} and K_m effects. The k_{cat}/K_m values for the incorporation of the dNTPs opposite the O^2 -alkyl-dTs are similar to the misincorporation k_{cat}/K_m values. The selectivity for a particular dNTP is low. The greatest selectivity is for the incorporation of dATP opposite O^2 -Me-dT by pol η in which the k_{cat}/K_m for dATP is 7-, 3.5-, and 5-fold greater than that for dCTP, dGTP, and dTTP, respectively. This selectivity is due to K_m differences. A similar selectivity is found with O^2 -POB-dT in which the k_{cat}/K_m for the pol η catalyzed insertion of dATP is about 4-fold greater than that for the other dNTPs. Pol ι is the least selective polymerase with dATP, dGTP, and dTTP having similar k_{cat}/K_m values with dCTP having k_{cat}/K_m values 5-fold less. The k_{cat}/K_m values for each enzyme are also very similar. Upon the basis of the steady-state kinetic analysis, we would predict that O^2 -alkyl-dTs would be bypassed with low fidelity with pol η , κ , and ι .

Incorporation of dNTPs Opposite O^2 -alkyl-dT with Polymerase in Excess

The TLS bypass of DNA adducts does not occur under steady-state kinetic conditions in the cell. The TLS polymerase would be brought to the DNA, perform an insertion reaction, and then hand the DNA off to a subsequent polymerase. This process can be kinetically driven by the relative reactivity of the polymerases or actively driven by protein–protein interactions.³¹ While k_{cat}/K_m values and consequently the selectivity of the polymerase under steady-state conditions is not affected by nonproductive binding complexes, the relative reactivity during single-turnover reaction will be. Consequently, we analyzed the single-nucleotide incorporation reactions with enzyme in excess conditions with polymerase concentrations of 250 nM and a DNA concentration of 25 nM. The DNA was P15/T24 in which the template base was dT, O^2 -Me-dT, or O^2 -POB-dT. The time courses for pol η catalyzed incorporation of each dATP opposite O^2 -POB-dT are shown in Figure 4A, with the early time points in Figure 4B. The data were fit to the burst equation, and the burst amplitude (A), burst rate constant (k), and steady-state rate constant (k_{ss}) were determined. Depending on the polymerase and the dNTP, the amplitude, burst rate constant, and/or steady-state rate constants were dependent on the dNTP concentration. If the parameters were dependent on the dNTP concentration, the parameters were fit to the hyperbolic eq 1. Figure 4C and D show the graphs for the amplitude and burst rate constant for pol η with each dNTP. The complete set of time course plots and the dNTP concentration dependence of the kinetic parameters for pol η , ι , and κ are presented in Figures S1–S12. The fitted parameters are presented in Tables 5–7. The parameters in the tables are the A^{max} , the maximum amplitude, K_A , and the dNTP concentration at half maximal amplitude, k_{pol} , the maximum burst rate constant, and $K_d^{dNTP} (app)$ the apparent dNTP dissociation constant, k_{ss}^{max} , the maximum k_{ss} , and K_{ss} , the dNTP concentration at half maximal k_{ss}^{max} . In some cases, the parameter did not exhibit a [dNTP] dependence, in which case the K value was not listed.

The relative reactivity of the polymerase/DNA/dNTP pairings are shown in the time course plots in Figure 5 in which the dNTP concentration was 50 μ M. This dNTP concentration is shown because cellular dNTP concentrations typically range from 5 to 50 μ M.⁵² The left panels have O^2 -Me-dT, and the right panels have O^2 -POB-dT as the template. Each panel also shows the correct incorporation of dATP opposite dT (black cricle). Several general observations can be made. First, the rate of incorporation of dATP opposite dT is 100–1000-fold faster than incorporation opposite the O^2 -alkyl-dTs. Second, pol η is the most reactive polymerase toward the alkylated substrates. Third, the incorporation of dATP and dTTP are faster than the incorporation of dCTP and dGTP opposite the O^2 -alkyl-dTs. These observations are discussed further below.

O^2 -alkyl-dT Are Poorer Substrates than dT—The decreased reactivity of the incorporation of dATP opposite the alkylated substrates is due to decreased k_{pol} and increased K_d^{dNTP} and K_A values. This is evident in Tables 5–7, in which the k_{pol} values for dATP decreases from 160, 72, and 62 s^{-1} to less than 10 s^{-1} with O^2 -alkyl-dT for all three enzymes. For pol η , the $K_d^{dNTP} (app)$ increased \sim 10-fold, but for ι and κ , the $K_d^{dNTP} (app)$ did not appreciably change. The A^{max} for pol η remained at \sim 0.7 for both dT and the O^2 -alkyl-dTs, while the K_A and K_d^{dNTP} increased 5–10-fold for the O^2 -alkyl-dTs.

Pol η Is More Reactive than Pol ι or κ with Both O²-Me-dT and O²-POB-dT—In Figure 5, the increased reactivity of pol η is evident by comparing the half-lives of the reactions. The half-lives of the pol η -catalyzed incorporation of dATP is ~1 s, while that for pol ι and κ are ~30 s. The increased reactivity of pol η is due to an increased burst amplitude for pol η relative to pols ι and κ . These conclusions are evident in Tables 5–7 in which the A^{max} is the highest and the K_A is the lowest for pol η when compared with those of pols ι and κ . For example, for dATP, the A^{max} values are 0.72 and 0.68, and K_A values are 25 and 14 μ M for O²-Me-dT and O²-POB-dT, respectively. The A^{max} values drop to between 0.24 and 0.49, and the K_A values rise to 90–340 μ M for pols ι and κ . In contrast, the k_{pol} and $K_d^{dNTP}(app)$ parameters are very similar for each polymerase. For example, for pol η and O²-alkyl-dT, the k_{pol} ranges from 5.8 to 7.3 s⁻¹, and very similar values are observed for pol ι (3–3.7 s⁻¹) and pol κ (3 s⁻¹). The $K_d^{dNTP}(app)$ values are also very similar: η (67–162 μ M), ι (8–43 μ M), and κ (8–11 μ M).

dATP and dTTP Are the Most Reactive Nucleotides—The increased reactivity of dATP and dTTP over dCTP and dGTP is due primarily to the burst amplitudes. Examination of Table 5 for pol η shows that the $A^{max}/[pol]$ for dATP and dTTP is ~0.7, while that for dCTP and dGTP is ~0.2. In contrast, the K_A , k_{pol} , and $K_d^{dNTP}(app)$ parameters are quite similar for each dNTP.

Extension Past O²-alkyl-dT

We examined the ability of human DNA pols η , κ , and ι , and yeast pol ζ to extend past X/Y base pairs using the P16/T24 DNA substrates. Figure 6 shows polyacrylamide gels for the extension past dA/dT (left), dA/O²-Me-dT (middle), and dA/O²-POB-dT (right) by pols η , κ , ι , and ζ . In each panel, the lower band is the 16-mer starting material, and the upper bands are product bands. The polymerase concentrations were set so that the reactivity with the dA/dT DNA substrate was similar for each polymerase. As can be seen in the left panels, the polymerases react with the DNA almost to completion within 1 min. Also evident is the robust activity of yeast pol ζ (lower panels), while the human Y-family polymerases show little if any activity. These experiments were duplicated for pol ζ with dC, dG, and dT as the base pair partner for the O²-alkyl-dTs. Plots showing the quantification of these data are shown in the Supporting Information. These experiments indicate that pol ζ catalyzes the extension past the dN/O²-alkyl-dT more efficiently than the Y-family polymerases and that the extension is not dependent on the base pair partner or the alkyl chain.

To quantitate the relative reactivity of pol ζ in extending undamaged versus NNK-damaged base pairs, we performed steady-state kinetic analysis. The initial rate kinetics of the bypass of dN/O²-alkyl-dT were performed at various concentrations of the next correct dNTP. The data were fit to eq 1, and the resulting Michaelis–Menten parameters are presented in Table 8. The V_{max}/K_m parameters are summarized in Figure 7. Several observations are made. First, the presence of mispairs in the terminal base pair (X/Y) does not inhibit the insertion of the correct dNTP. This result is consistent with previous reports that pol ζ is effective at extending mispairs.⁵³ Second, the presence of the O²-alkyl-dT only inhibits replication by a factor of 3. Third, the identity of the alkyl chain, being methyl or POB, does not affect the

V_{max}/K_m parameter. Finally, the base pair opposite the O^2 -alkyl-dT does not affect the rate of extension.

Since pol ζ is effective in extending dN/ O^2 -POB-dT base pairs, we briefly examined if the enzyme was capable of inserting a nucleotide opposite the adducts. As is shown in Figure S18, the enzyme is very inefficient at inserting dATP opposite O^2 -alkyl-dT.

Extension Past O^2 -alkyl-dT with Polymerase in Excess

As discussed above, the steady-state kinetic parameters may not accurately reflect the activity of the Y-family polymerases during the incorporation of a single nucleotide at the replication fork. We therefore examined the extension past the adducts with enzyme in excess. We found that extension past the adducts was slower than the insertion opposite the adducts for the three Y-family polymerases studied. The reactions with 3 nM DNA, 30 nM pol η , and 50 μ M dNTP are shown in Figure 8. The black triangles show the rapid insertion of dATP opposite dT, and the open triangles show the slower insertion of dATP opposite O^2 -alkyl-dT. The extension past the dN/ O^2 -POB-dT base pairs are represented by the solid and open circles and squares. As is evident, the pol η catalyzed extension is much slower than the insertion reaction. The extension past the adducts by pols ι and κ are shown in Figure S19.

DISCUSSION

NNK is a potent human lung carcinogen.⁴ It is unique among lung carcinogens in that irrespective of the route of administration, rodents given NNK develop lung tumors.⁵ Understanding the molecular mechanisms underlying its mutagenicity may shed some light on its remarkable organ-specificity. NNK is a bifunctional alkylating agent, producing both methyl and POB-DNA adducts. While similar sites on the DNA are alkylated, the relative proportions are different. In particular, O^2 -POB-dT is the most abundant POB-DNA adduct in rats chronically treated with NNK or NNN.^{14,54-56} In contrast, the corresponding O^2 -Me-dT is a minor adduct from the corresponding methylating agent. Rodents given NNK have increased levels of mutations at AT base pairs when compared with rodents given a methylating agent,^{19,20} supporting the role of NNK in mutagenesis at AT base pairs.

Recently, we found that siRNA knockdown of pol η , ζ , and Rev1 impacted the bypass of O^2 -Me-dT and O^2 -POB-dT in human cells.⁴⁴ These results, along with known reactivities of the enzymes, led to the hypothesis illustrated in Scheme 2 in which pol η is involved in the insertion of dNTPs opposite the adducts, pol ζ is involved in the extension past the adduct, while Rev1 is a structural protein. To test the hypotheses that catalytic activities of pols η and ζ are critical to bypass and mutagenesis of O^2 -POB-dT, we evaluated the *in vitro* kinetics of insertion of dNTPs opposite and bypass of O^2 -POB-dT by pols η , ι , κ , and ζ .

We initially examined the steady-state kinetics for the incorporation of each dNTP opposite both O^2 -Me-dT and O^2 -POB-dT. The kinetic parameters in Tables 2-4 show that pols η , κ , and ι are equally poor at insertion opposite the adducts. This result is inconsistent with a role for pol η in bypass. However, TLS polymerases do not operate under steady-state kinetics *in vivo*. A current model of TLS bypass is the handoff model in which polymerases

are recruited for insertion opposite the adduct, while another polymerase can perform the extension past the adduct.^{21,34} In this model, a polymerase acts once, and consequently, steady-state kinetics do not accurately describe the reactivity.

Under conditions of polymerase in excess, the time courses for the insertion of dNTPs opposite the adducts are biphasic with a rapid burst followed by a slower reaction. As illustrated in Figure 5, we show that pol η is more effective than pols ι and κ in the insertion of dNTPs opposite O^2 -POB-dT. All three enzymes are more effective than Kf(exo-) and Dpo4.³⁹ The biphasic kinetic behavior under conditions with polymerase in excess can be explained by two mechanisms: (1) formation of nonproductive enzyme–substrate complexes^{39,57–61} and (2) a rate limiting step after rapid phosphodiester bond formation.^{62,63} The formation of nonproductive complexes has been observed in DNA damage bypass.^{39,57–61} In addition, pol κ exhibits this behavior during the formation of a correct base pair.⁶⁴

The substrate specificity of polymerases have been reported by their relative k_{cat}/K_m and presteady-state k_{pol}/K_d^{dNTP} values.^{65,66} However, this relationship does not hold in this system. For example, as shown in Table 5, the k_{pol}/K_d value for pol η -catalyzed incorporation of dGTP ($555 \text{ mM}^{-1} \text{ s}^{-1}$) is greater than that for dATP ($74 \text{ mM}^{-1} \text{ s}^{-1}$). However, it is clearly evident in Figure 5B that the incorporation of dATP (solid square) is faster than the incorporation of dGTP (solid diamond). Therefore, the relative reactivity of the substrates is not simply the k_{pol}/K_d ratios but must take into account the amplitude and the steady-state parameters.

Pol κ and its *E. coli* analogue pol IV accurately bypass N^2 -dG adducts of various sizes with good efficiency.^{51,67–70} For example, human pol κ bypasses the very bulky N^2 -benzopyrene adduct, while the prokaryotic analogue (pol IV/dinB) bypasses the less bulky N^2 -furfuryl-dG and N^2 -(1-carboxyethyl)-2'-dG adducts. The crystal structure of pol κ displays an open area on the minor groove of DNA.⁷¹ Thus, one can envision that pol κ could bypass O^2 -POB-dT by binding the POB-group in the minor groove pocket of pol κ , while dGTP would bind to dT in a Watson–Crick-like conformation in Figure 9d. However, we found that neither O^2 -POB-dT nor the smaller O^2 -Me-dT is a good substrate for pol κ . This result agrees with Andersen et al., who reported similar findings for O^2 -Me-dT and O^2 -Et-dT.^{40,41}

The ability of pol κ to accommodate N^2 -alkyl-dG DNA damage is not a strictly steric issue. For example, the replacement of Phe13 by the smaller valine eliminates the ability of pol IV to bypass N^2 -furfuryl-dG.⁶⁹ This result appears to suggest that van der Waals interactions play a role in pol κ 's substrate specificity. In addition, functional/kinetic studies show that pol κ requires Watson–Crick hydrogen bonds for rapid catalysis.⁷² A difference between N^2 -dG adducts and O^2 -POB-dT is that O^2 -POB-dT cannot form Watson–Crick hydrogen bonds with the incoming dNTP. While O^2 -POB-dT has a bulky adduct in the minor groove, perhaps the inability of O^2 -POB-dT to form Watson–Crick base pairs makes this a poor substrate for pol κ . This rationale was proposed to explain why pol IV is not involved in the bypass of O^2 -alkyl-dT adducts in *E. coli*.⁷³ In this article, we extend this rationale to human pol κ .

Pol ι flips template purines from the *anti*- to the *syn*-configuration thereby displaying the Hoogsteen hydrogen bonding face to the incoming dNTP.^{74–77} This property allows for the replication of adducts in which the Watson–Crick hydrogen bonding face is blocked.⁷⁸ In addition, O^6 -Et-dG and N^2 -Et-dG adducts are bypassed in this manner.^{79,80} However, pol ι is inefficient at bypassing more bulky N^2 -dG adducts.⁵⁰ Template pyrimidines react with pol ι in the *anti*-conformation. Kinetic evidence indicates that dGTP/dC is replicated with Watson–Crick base pairs, while dGTP/dT occurs via wobble base pairing.⁸¹ Upon the basis of these properties, there is no reason to expect that pol ι would be efficient at bypassing O^2 -POB-dT.

Pol η evolved to accurately bypass *cis-syn* cyclobutane pyrimidine dimers. Inactivation of the POLH gene leads to the XPV form of xeroderma pigmentosum in which individuals are predisposed to skin cancer.^{82–84} Pol η is also proficient at bypassing other adducts such as *N*-(deoxyguanosin-8-yl)-1-aminopyrene,⁶¹ O^6 -alkyl-dG,¹³ and 8-oxo-dG.^{85–88} Synthesis opposite the bulky N^2 -dG adducts is slow,⁶⁷ although pol η may play a role in the mutagenic bypass of N^2 -BP-dG.^{68,89,90} These studies do not provide any mechanistic rationale for the ability of pol η to insert dTTP and dATP opposite O^2 -alkyl-dT. Our *in vitro* kinetic data do agree with the mammalian cell culture studies in which pol η is crucial to the bypass.⁴⁴

The mechanism underlying the pol η catalyzed insertion of dTTP and dATP opposite O^2 -alkyl-dT is not clear. The ability to form Watson–Crick-like structures has been proposed to explain the mutagenic potential of DNA adducts. This type of simplistic structural analysis worked well with O^6 -Me-dG in which the Watson–Crick-like structure (Figure 9a) initially proposed in 1976⁹¹ explains the mutagenic incorporation of dTTP.^{92–94} However, an attempt to predict the base pairing of O^2 -alkyl-dT is not as satisfying. Figure 9b and c illustrates the potential hydrogen-bonding interactions between dA and dT with O^2 -alkyl-dT. While the dA/ O^2 -alkyl-dT structure requires protonation, the alkyl group can also make a hydrophobic interaction with the 2-position of adenine stabilizing the structure.⁹⁵ Figure 9c displays a base-pair structure between dT and O^2 -alkyl-dT that is similar to the wobble structure that was observed in an oligodeoxynucleotide duplex containing a dT/dT mispair.⁹⁶ Figure 9b and c does not exhibit any chemical interactions that would explain the preference for pol η to preferentially insert dATP and dTTP opposite O^2 -alkyl-dT more often than dGTP. Figure 9d shows a potential Watson–Crick-like structure between dG and O^2 -alkyl-dT. This structure is very similar to that in Figure 9a, except that the alkyl group is in the minor groove. Perhaps this structure explains the mutagenic incorporation of dGTP opposite O^2 -alkyl-dT by pol V in *E. coli*.^{42,73} However, the preference for the incorporation of dT and dA opposite O^2 -alkyl-dG in human cells is not evident. Structural and functional studies must be undertaken to elucidate the interactions that control the identity of the nucleotide inserted opposite O^2 -alkyl-dTs.

While pol η is the most efficient Y-family polymerase to incorporate dNTPs opposite O^2 -alkyl-dG, neither it nor pol ι or κ were effective at further extending the primer. We did find that yeast pol ζ is able to extend past the O^2 -alkyl-dT adducts. Pol ζ is a B-family polymerase that does not insert dNTPs opposite DNA damage but readily extends past mismatches,^{97,98} abasic sites,⁹⁹ γ -hydroxy-1, N^2 -propano-dG,¹⁰⁰ 8-oxo-dG,¹⁰¹ O^6 -Me-dG,¹⁰¹ thymine glycol,¹⁰² and mismatched N^2 -BP-dG/T.^{103,104} The polymerase we

employed was the heterodimer between yeast Rev3 and Rev7. Human Rev3 is twice as large as the yeast protein.^{105,106} Human and yeast Rev7 are homologous, and the activity of the catalytic subunit, Rev3, is increased 10-fold by Rev 7.¹⁰⁶ *In vivo*, the likely active species is a tetramer among Rev3, Rev7, and the pol δ subunits, pol31 and pol32.^{107–110} We found that yeast pol ζ readily extends dN/*O*²-alkyl-dT base pairs, with k_{cat}/K_m values one-third that of undamaged DNA. There is no sequence specificity with respect to the base pair partner.

Our model for the bypass of *O*²-POB-dT is illustrated in Scheme 1. A replicative polymerase, such as pol δ , synthesizes up to the adduct. We found that the high fidelity polymerase Kf(exo-) can synthesize up to the adduct, but have very little activity at the insertion of a dNTP opposite the adduct.³⁹ The replicative polymerase stalls at the adduct and is replaced by a Y-family polymerase. Pol η has higher activity toward *O*²-POB-dT than κ or ι , and based upon relative kinetics, it will be the polymerase that inserts the correct dA or an incorrect dT opposite the adduct.^{40,41} Pol η , as well as ι and κ , has low activity extending past *O*²-POB-dT. Following insertion, pol η will be replaced by pol ζ , which can extend the primer past the adduct. Subsequently, a high fidelity polymerase can continue DNA synthesis. The activity of pol ζ does not depend on the identity of the nucleotide that is inserted opposite the adduct. In our model, we have Rev1 acting as a scaffold protein. The involvement of Rev1 is supported from cell studies using siRNA to deplete polymerases.⁴⁴ Rev1 is highly selective for inserting C opposite normal and adducted template G because it uses arginine as a template for dCTP binding.^{111,112}

Supplementary Material

Refer to Web version on PubMed Central for supplementary material.

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ABBREVIATIONS

7-POB-dG	7-[4-(3-pyridyl)-4-oxobut-1-yl]-2'-deoxyguanosine
Dpo4	<i>Sulfolobus solfataricus</i> DNA polymerase IV
Kf(exo-)	Klenow fragment of proofreading deficient <i>E. coli</i> DNA polymerase I
NNK	4-(methylnitrosamino)-1-(3-pyridyl)-1-butanone
NNN	<i>N</i> '-nitrosornicotine
<i>O</i>²-Me-dT	<i>O</i> ² -methylthymidine
<i>O</i>²-POB-dC	<i>O</i> ² -[4-(3-pyridyl)-4-oxobut-1-yl]-2'-deoxycytidine

***O*²-POB-dT** *O*²-[4-(3-pyridyl)-4-oxobut-1-yl]-thymidine

***O*⁶-POB-dG** *O*⁶-[4-(3-pyridyl)-4-oxobut-1-yl]-2'-deoxyguanosine

POB 4-(3-pyridyl)-4-oxobut-1-yl

pol polymerase

yPol *Saccharomyces cerevisiae* DNA polymerase

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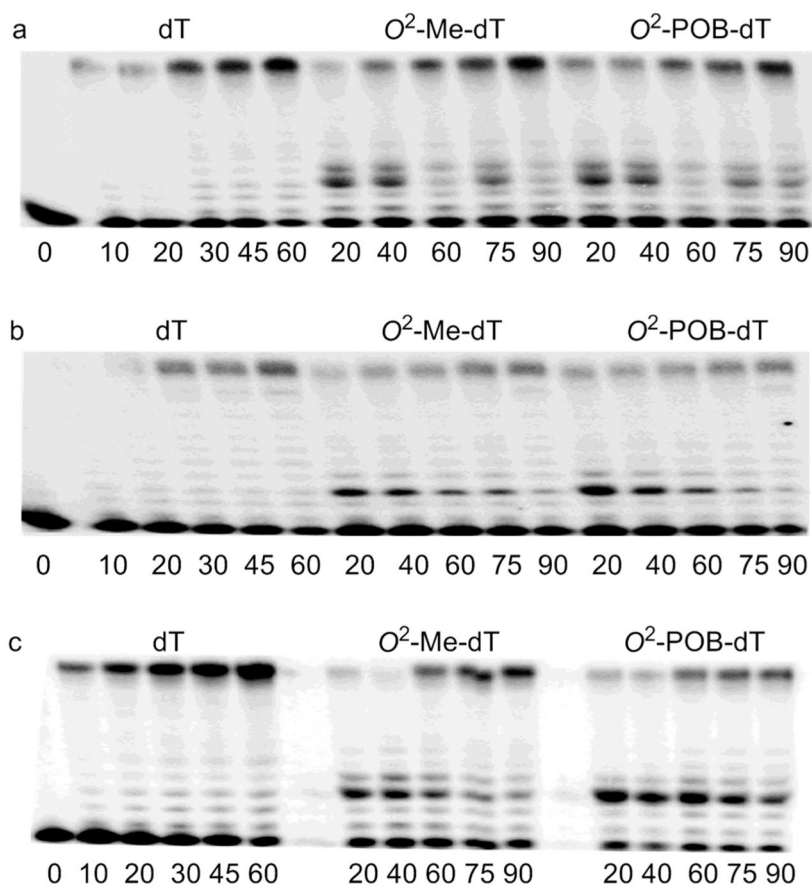


Figure 1. PAGE analysis of running start polymerase reactions. DNA substrates (10 nM, P12/T24 with dT, O²-Me-dT, and O²-POB-dT) were reacted with pol κ (a), pol η (b), and pol ι (c) and all four dNTPs for the indicated time (min). The DNA polymerase concentrations were 0.1 nM for dT and 0.5 nM for O²-alkyl-dT substrates. The dNTP concentrations were 5 μ M for dT and 50 μ M for O²-alkyl-dT substrates.

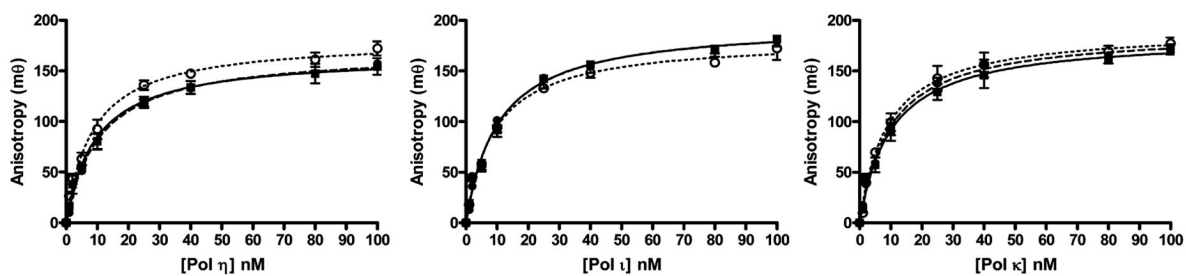


Figure 2.

Polymerase-DNA affinity. Fluorescence anisotropy was determined with 1 nM DNA in which the 5'-terminus of the primer strand was modified with fluorescein with variable concentrations of (A) pol η , (B) pol ι , and (C) pol κ . The DNA contained dT (open circle), O²-Me-dT (black circle), O²-POB-dT (square). The data points are the mean \pm standard deviation of three determinations. The lines are the best fit to eq 3.

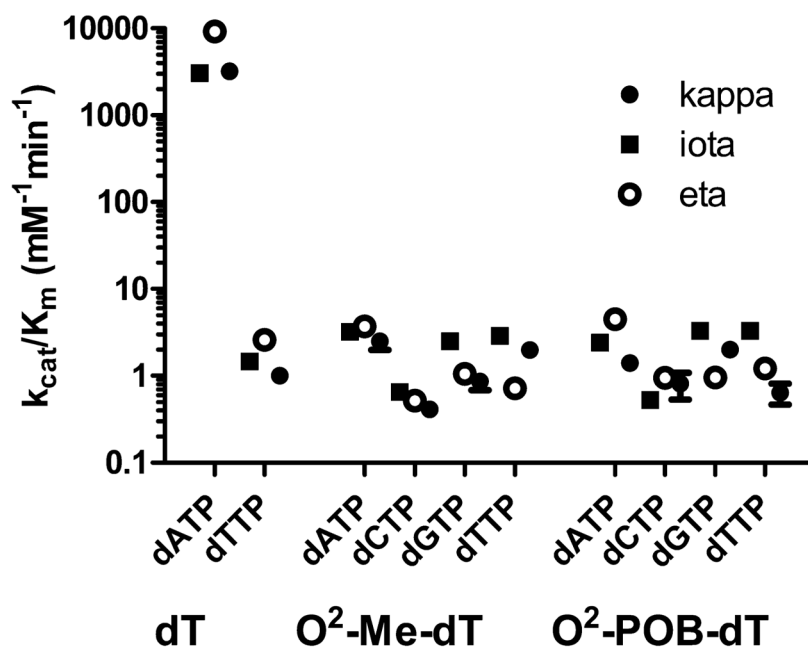


Figure 3. k_{cat}/K_m values \pm standard errors for the incorporation of dNTPs opposite dT, O²-Me-dT, and O²-POB-dT catalyzed by pol κ (black circle), ι (square), and η (open circle).

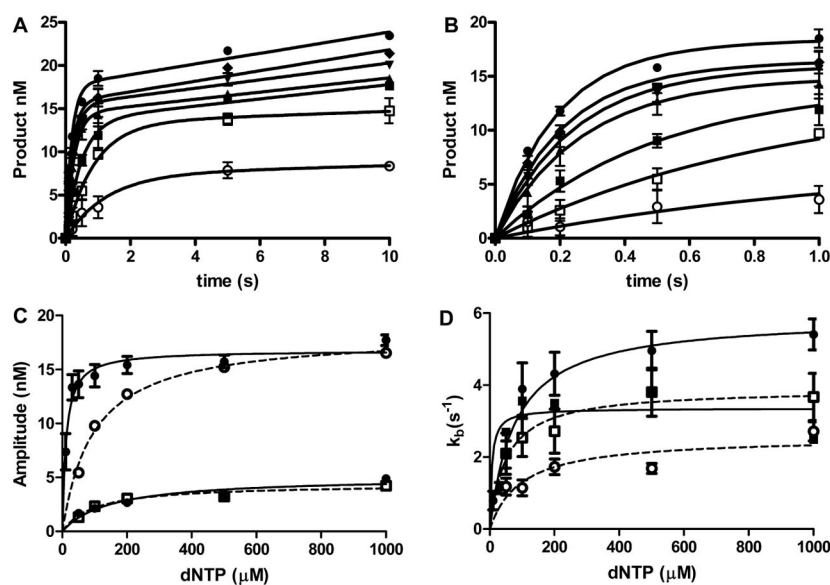


Figure 4.

Pol η catalyzed insertion of dNTPs opposite O²-POB-dT. (A) Pol η (250 nM) and O²-POB-dT DNA (25 nM) were reacted with 10 μ M (open circle), 30 μ M (open square), 50 μ M (square), 100 μ M (up triangle), 200 μ M (down triangle), 500 μ M (diamond), and 1000 μ M (circle) dATP. (B) Early time points from panel A. The lines are the best fit to the burst equation. The data points are the mean of three experiments \pm standard deviation. dNTP concentration dependence on the (C) amplitude and burst rate constant (D) of the pol η catalyzed insertion of dNTPs opposite O²-POB-dT. The error bars are the standard errors. The lines (solid for dATP and dGTP, and dotted for dCTP and dTTP) are the best fit to eq 1 for dATP (black circle), dCTP (open square), dGTP (black square), and dTTP (open square).

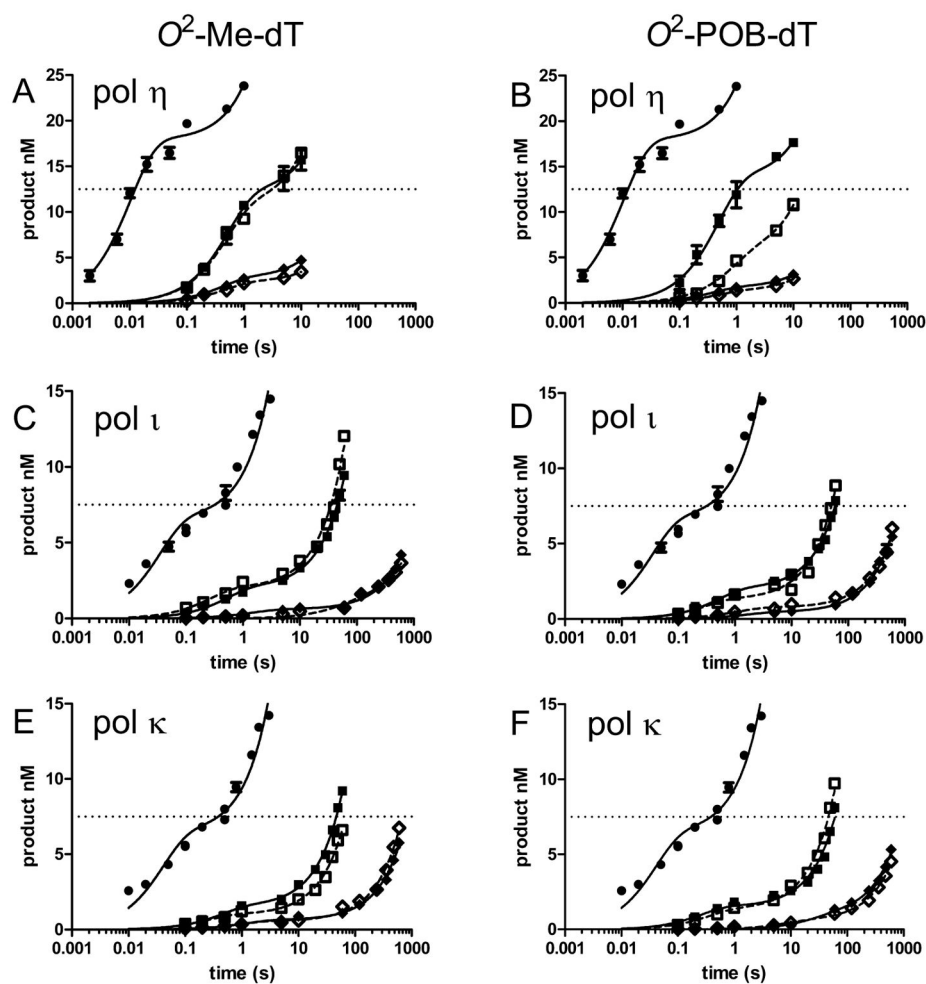


Figure 5.

Comparisons of the relative reactivity of polymerases with DNA and dNTP. The pol and O^2 -alkyl-dT DNA pairing in each panel is as follows: (A) pol η and O^2 -Me-dT; (B) pol η and O^2 -POB-dT; (C) pol ι and O^2 -Me-dT; (D) pol ι and O^2 -POB-dT; (E) pol ι and O^2 -Me-dT; (F) pol ι and O^2 -POB-dT. Each panel shows the insertion of 25 μ M dATP opposite dT (black circle). The polymerase (250 nM) and DNA (25 nM) were reacted with 50 μ M dATP (black square), dCTP (open diamond), dGTP (black diamond), dTTP (open square); dT, O^2 -Me-dT, or O^2 -POB-dT. The solid lines are the best fit to the burst equation with the purine dNTP represented by the solid line and the pyrimidine dNTP represented by the dashed line.

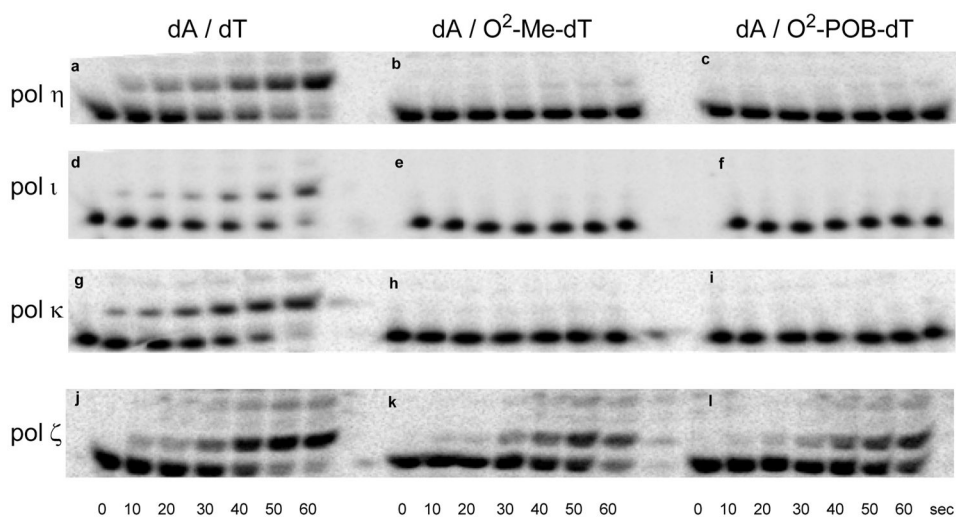


Figure 6. Relative bypass of O^2 -alkyl-dT by human (a–c) pol η , (d–f) pol κ , (g–i) pol ι , and (j–l) yeast pol ζ . The concentration of the next incoming dNTP was $50 \mu\text{M}$, and the concentration of the DNA containing a dA/dT (left), dA/ O^2 -Me-dT (middle), and dA/ O^2 -POB-dT (left) base pair was 3 nM . The concentration of pol η was 0.05 nM , pol κ and ι were 0.1 nM , and pol ζ was $6 \mu\text{g}/\mu\text{L}$.

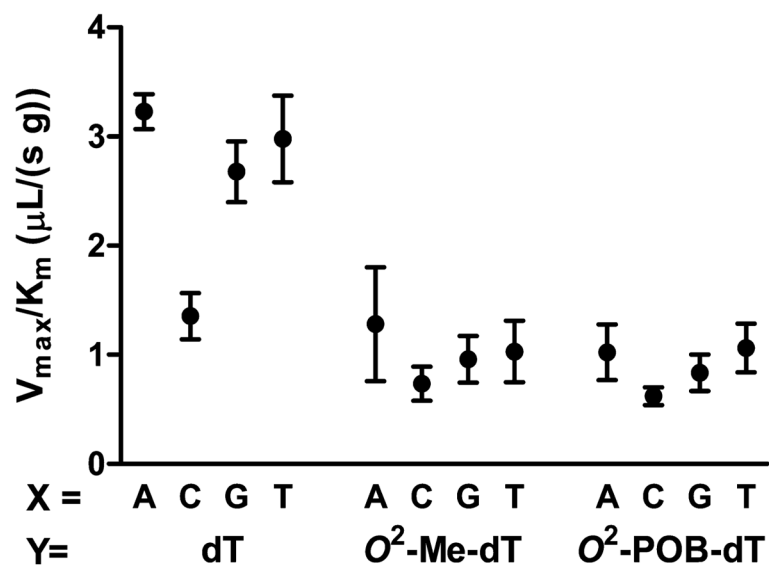


Figure 7. Pol ζ catalyzed extension of O^2 -alkyl-dT base pairs. The relative V_{\max}/K_m values for dA, dC, dG, and dT as at the primer terminus (X) with dT, O^2 -Me-dT, and O^2 -POB-dT as base pair partner (Y). The steady-state kinetics were performed with 30 nM DNA (P16/T24) and 1 $\mu\text{g}/\mu\text{L}$ pol ζ . The data points are the mean \pm SD of three independent determinations.

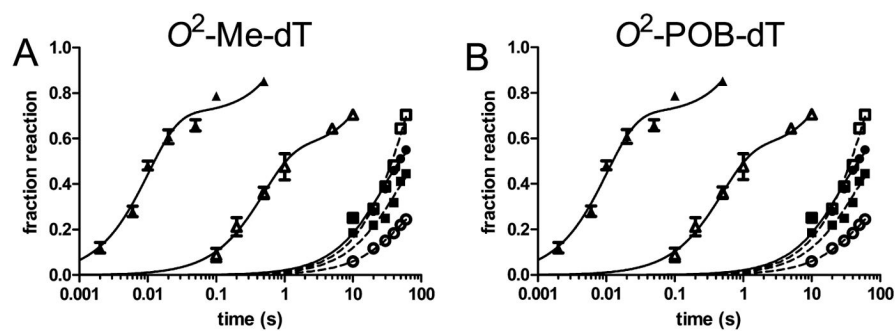


Figure 8. Insertion and extension reactivity of pol η past O^2 -Me-dT (A) and O^2 -POB-dT (B). Pol η (30 nM) and DNA (3 nM) were reacted with 50 μ M dNTP. Each panel shows the insertion of dATP opposite dT (triangle) and O^2 -alkyl-dT (open triangle) and the extension past dA (square), dC (open circle), dG (black circle), and dT (open square) opposite O^2 -alkyl-dT. The solid lines are the best fit to the burst equation, and the dashed lines are fit to a first-order equation.

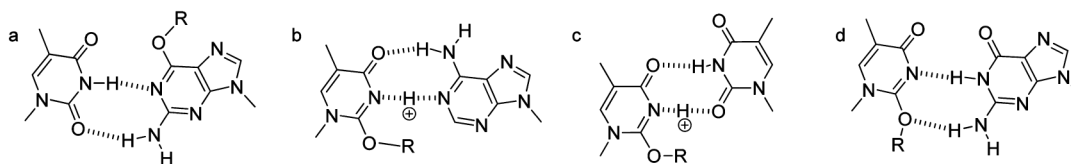
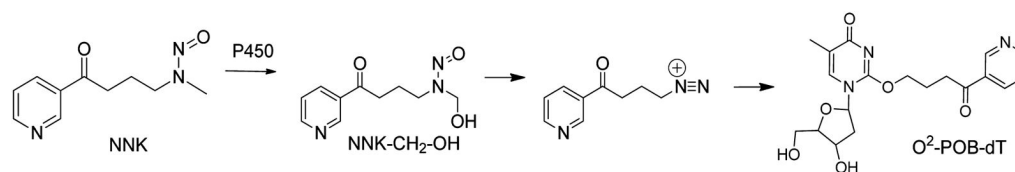
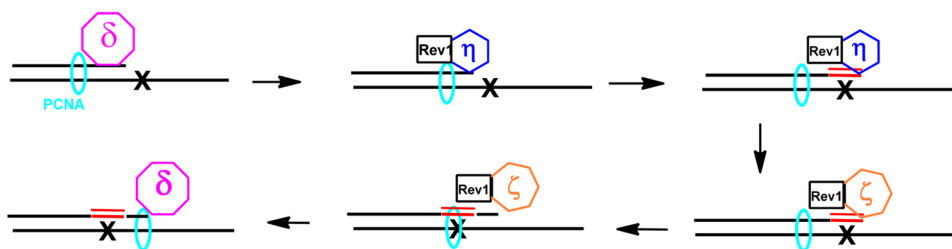


Figure 9.
Potential base pair structures.



Scheme 1.
Formation of O²-POB-dT from the Bioactivation of NNK



Scheme 2.
Model for the Bypass of O^2 -alkyl-dT

P12-	5'- G C A C C G C A G A C G	-3'
P15-	5'- G C A C C G C A G A C G C A G	-3'
P16(Y)	5'- G C A C C G C A G A C G C A G Y	-3'
T24(X)	3'- C G T G G C G T C T G C G T C X G C A G C G T C	-5'

Chart 1.
Oligodeoxynucleotide Sequences

Table 1Dissociation Constants for Normal and Damaged DNA to Polymerases^a

X	pol η (nM)	pol κ (nM)	pol ι (nM)
dT	7.9 ± 0.7	8.3 ± 0.7	8.4 ± 0.7
O ² -Me-dT	10.1 ± 0.7	9.1 ± 0.5	9.7 ± 0.5
O ² -POB-dT	8.9 ± 0.8	9.4 ± 0.9	9.7 ± 0.8

^aCalculated value with standard error. The fluorescence anisotropy was determined with 1 nM DNA with 0–100 nM polymerase in the appropriate buffer. The experiment was performed three times, and the data were fitted to the quadratic equation.

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Table 2Steady-State Parameters for the Pol η Catalyzed Insertion Opposite O^2 -alkyl-dT^a

dNTP	template	k_{cat} (min ⁻¹)	K_m (μ M)	k_{cat}/K_m (mM ⁻¹ min ⁻¹)
dATP	dT	49 \pm 2	5.3 \pm 0.3	9230 \pm 230
dTTP	dT	0.140 \pm 0.004	77.8 \pm 5.5	1.8 \pm 0.1
dATP	O^2 -Me-dT	0.068 \pm 0.003	18.5 \pm 2.4	3.7 \pm 0.4
dCTP	O^2 -Me-dT	0.068 \pm 0.003	130.2 \pm 11.4	0.52 \pm 0.02
dGTP	O^2 -Me-dT	0.039 \pm 0.001	38.4 \pm 1.8	1.05 \pm 0.04
dTTP	O^2 -Me-dT	0.119 \pm 0.004	166 \pm 9	0.72 \pm 0.02
dATP	O^2 -POB-dT	0.098 \pm 0.004	21.7 \pm 2.5	4.50 \pm 0.37
dCTP	O^2 -POB-dT	0.141 \pm 0.008	148.6 \pm 16.2	0.95 \pm 0.05
dGTP	O^2 -POB-dT	0.121 \pm 0.004	126.2 \pm 8.0	0.96 \pm 0.03
dTTP	O^2 -POB-dT	0.126 \pm 0.005	103.3 \pm 8.6	1.22 \pm 0.06

^a Initial rates were conducted with 0.25 to 0.5 nM pol, 5 nM DNA, and 0 to 200 μ M dNTP. The values are the mean \pm standard deviation of three determinations.

Table 3Steady-State Parameters for the Pol α Catalyzed Insertion Opposite O^2 -alkyl-dT^a

dNTP	template	k_{cat} (min^{-1})	K_m (μM)	k_{cat}/K_m ($\text{mM}^{-1} \text{min}^{-1}$)
dATP	dT	8.0 ± 0.6	2.6 ± 0.4	3070 ± 270
dTTP	dT	0.103 ± 0.003	70 ± 4	1.46 ± 0.06
dATP	O^2 -Me-dT	0.076 ± 0.002	24 ± 1	3.2 ± 0.1
dCTP	O^2 -Me-dT	0.061 ± 0.004	93 ± 14	0.65 ± 0.06
dGTP	O^2 -Me-dT	0.114 ± 0.002	46 ± 2	2.5 ± 0.1
dTTP	O^2 -Me-dT	0.121 ± 0.005	42 ± 5	2.9 ± 0.2
dATP	O^2 -POB-dT	0.086 ± 0.003	35 ± 3	2.4 ± 0.1
dCTP	O^2 -POB-dT	0.059 ± 0.003	112 ± 13	0.53 ± 0.03
dGTP	O^2 -POB-dT	0.094 ± 0.003	28 ± 3	3.3 ± 0.2
dTTP	O^2 -POB-dT	0.121 ± 0.001	37 ± 1	3.27 ± 0.08

^aInitial rates were conducted with 0.25 to 0.5 nM pol, 5 nM DNA, and 0 to 200 μM dNTP. The values are the mean \pm standard deviation of three determinations.

Table 4Steady-State Parameters for the Pol κ Catalyzed Insertion Opposite O^2 -alkyl-dT^a

dNTP	template	k_{cat} (min ⁻¹)	K_m (μ M)	k_{cat}/K_m
dATP	T	4.3 \pm 0.2	1.4 \pm 0.1	3200 \pm 200
dTTP	T	0.11 \pm 0.01	120 \pm 20	1.0 \pm 0.1
dATP	O^2 -Me-dT	0.087 \pm 0.008	35 \pm 8	2.5 \pm 0.3
dCTP	O^2 -Me-dT	0.067 \pm 0.006	160 \pm 30	0.41 \pm 0.03
dGTP	O^2 -Me-dT	0.055 \pm 0.004	64 \pm 12	0.86 \pm 0.10
dTTP	O^2 -Me-dT	0.067 \pm 0.001	34 \pm 2	1.99 \pm 0.07
dATP	O^2 -POB-dT	0.047 \pm 0.001	33 \pm 2	1.41 \pm 0.05
dCTP	O^2 -POB-dT	0.022 \pm 0.002	29 \pm 7	0.81 \pm 0.16
dGTP	O^2 -POB-dT	0.13 \pm 0.01	66 \pm 12	2.0 \pm 0.2
dTTP	O^2 -POB-dT	0.14 \pm 0.03	215 \pm 79	0.64 \pm 0.10

^a Initial rates were conducted with 0.25 to 0.5 nM pol, 5 nM DNA, and 0 to 200 μ M dNTP. The values are the mean \pm standard deviation of three determinations.

Table 5

Kinetic Parameters for the dNTP Dependence of the Pol κ Catalyzed Insertion Opposite O^2 -alkyl-dIT^a

template	dNTP	amplitude			burst rate constant				steady-state rate constant	
		$A^{\max}/[\text{pol}]$	K_A (μM)	$A^{\max}/[\text{pol}]/K_A$ (mM^{-1})	k_{pol} (s^{-1})	K_d dNTP (μM)	k_{pol}/K_d ($\text{s}^{-1} \text{mM}^{-1}$)	k_{ss}^{\max} (s^{-1})	K_{ss} (μM)	
dT	dATP	0.69 ± 0.03	1.9 ± 0.4	363	155 ± 7	7.9 ± 0.8	19600	0.31 ± 0.03	8.0 ± 2.0	
O^2 -Me-dT	dATP	0.72 ± 0.04	22 ± 3	33	7.4 ± 0.4	168 ± 20	44	0.023 ± 0.003	73 ± 41	
O^2 -Me-dT	dCTP	0.22 ± 0.02	100 ± 29	2.2	3.5 ± 0.28	32 ± 15	109	0.014 ± 0.002	130 ± 70	
O^2 -Me-dT	dGTP	0.25 ± 0.02	62 ± 17	4	3.9 ± 0.23	41 ± 12	95	0.013 ± 0.001	41 ± 13	
O^2 -Me-dT	dTTP	0.72 ± 0.04	97 ± 12	7.4	2.6 ± 0.4	96 ± 60	27	0.020 ± 0.002		
O^2 -POB-dT	dATP	0.68 ± 0.04	11 ± 2	62	5.8 ± 0.4	78 ± 16	74	0.026 ± 0.002	55 ± 17	
O^2 -POB-dT	dCTP	0.18 ± 0.02	101 ± 29	1.8	3.89 ± 0.26	51 ± 16	76	0.010 ± 0.002	78 ± 45	
O^2 -POB-dT	dGTP	0.20 ± 0.03	149 ± 67	1.3	3.35 ± 0.45	6 ± 15	555	0.011 ± 0.001	37 ± 14	
O^2 -POB-dT	dTTP	0.72 ± 0.04	97 ± 12	7.4	2.6 ± 0.4	96 ± 60	27	0.020 ± 0.002		

^aReactions were conducted with 250 nM pol, 25 nM DNA, and 0 to 1000 μM dNTP. The individual time courses were fit to eq 2 to obtain amplitude (A), burst rate constants (k_b), and steady-state rate constants (k_{ss}) as a function of dNTP concentration. These values were fit to the hyperbolic equation to obtain the parameters in the table. The values are the mean \pm standard error of three determinations. If the K values are missing, then the parameter was not dependent on the [dNTP].

Table 6

Kinetic Parameters for the dNTP Dependence of the Pol α Catalyzed Insertion Opposite O^2 -alkyl-dT^a

template	dNTP	amplitude			burst rate constant			steady-state rate constant	
		$A^{\max}/[\text{pol}]$	K_A (μM)	$A^{\max}/[\text{pol}]/K_A$ (mM^{-1})	k_{pol} (s^{-1})	K_d^{dNTP} (μM)	k_{pol}/K_d ($\text{s}^{-1} \text{mM}^{-1}$)	k_{ss}^{\max} (s^{-1})	K_{ss} (μM)
dT	dATP			3.0 ± 0.4	22 ± 6			0.25 ± 0.03	10 ± 5
O^2 -Me-dT	dATP	0.420 ± 0.047	190 ± 60	8.9	3.3 ± 0.3			0.027 ± 0.267	78 ± 37
O^2 -Me-dT	dCTP	0.180 ± 0.020	601 ± 130	0.3	2.3 ± 0.6			0.013 ± 0.003	250 ± 160
O^2 -Me-dT	dGTP	0.093 ± 0.007	213 ± 56	0.4	5.5 ± 1.1	25		0.017 ± 0.006	592 ± 426
O^2 -Me-dT	dTTP	0.380 ± 0.033	104 ± 34	3.6	3.6 ± 0.2			0.023 ± 0.003	110 ± 40
O^2 -POB-pT	dATP	0.313 ± 0.020	90 ± 20	3.5	3.7 ± 0.2	86		0.017 ± 0.001	80 ± 20
O^2 -POB-pT	dCTP	0.187 ± 0.013	300 ± 53	0.6	2.53 ± 0.05	94		0.004 ± 0.001	8 ± 15
O^2 -POB-pT	dGTP	0.147 ± 0.013	421 ± 105	0.3	2.42 ± 0.24	54		0.007 ± 0.001	186 ± 63
O^2 -POB-pT	dTTP	0.307 ± 0.033	82 ± 31	3.7	3.7 ± 0.3	88		0.024 ± 0.004	170 ± 90

^aReactions were conducted with 150 nM pol, 15 nM DNA, and 0 to 1000 μM dNTP. The individual time courses were fit to eq 2 to obtain amplitude (A), burst rate constants (k_b), and steady-state rate constants (k_{ss}) as a function of dNTP concentration. These values were fit to the hyperbolic equation to obtain the parameters in the table. The values are the mean \pm standard error of three determinations. If the K values are missing, then the parameter was not dependent on the [dNTP].

Table 7

Kinetic Parameters for the dNTP Dependence of the Pol ζ Catalyzed Insertion Opposite O^2 -alkyl-dT^a

template	amplitude				burst rate constant			steady-state rate constant	
	dNTP	$A^{\max}/[\text{pol}]$	K_A (μM)	$A^{\max}/[\text{pol}]/K_A$ (mM^{-1})	k_{pol} (s^{-1})	K_d dNTP (μM)	k_{pol}/K_d ($\text{s}^{-1} \text{mM}^{-1}$)	k_{ss}^{\max} (s^{-1})	K_{ss} (μM)
dT	dATP	0.3 ± 0.01			62 ± 2	8 ± 1	7750	0.86 ± 0.15	14 ± 5
O^2 -Me-dT	dATP	0.49 ± 0.11	340 ± 190	1.4	3.3 ± 0.5			0.033 ± 0.005	202 ± 84
O^2 -Me-dT	dCTP	0.18 ± 0.01	320 ± 27	0.6	3.2 ± 0.7			0.013 ± 0.003	438 ± 215
O^2 -Me-dT	dGTP	0.14 ± 0.015	195 ± 21	0.7	4.8 ± 0.6	210 ± 80	22	0.010 ± 0.001	152 ± 30
O^2 -Me-dT	dTTP	0.87 ± 0.47	1120 ± 960	0.7	3.2 ± 0.5			0.039 ± 0.006	364 ± 139
O^2 -POB-dT	dATP	0.29 ± 0.03	85 ± 37	3.4	3.3 ± 0.3	20 ± 7	165	0.029 ± 0.006	260 ± 150
O^2 -POB-dT	dCTP	0.25 ± 0.07	860 ± 410	0.3	2.9 ± 0.4	60 ± 40	48	0.014 ± 0.002	490 ± 170
O^2 -POB-dT	dGTP	0.80 ± 0.67	5200 ± 4800	0.2	2.3 ± 0.1	10 ± 4	230	0.011 ± 0.002	380 ± 200
O^2 -POB-dT	dTTP	0.51 ± 0.03	350 ± 60	1.5	2.8 ± 0.3			0.027 ± 0.004	110 ± 50

^aReactions were conducted with 150 nM pol, 15 nM DNA, and 0 to 1000 μM dNTP. The individual time courses were fit to eq 2 to obtain amplitude (A), burst rate constants (k_b), and steady-state rate constants (k_{ss}) as a function of dNTP concentration. These values were fit to the hyperbolic equation to obtain the parameters in the table. The values are the mean \pm standard error of three determinations. If the K values are missing, then the parameter was not dependent on the [dNTP].

Table 8

Steady-State Kinetic Parameters for Pol ζ Catalyzed the Extension Past O^2 -alkyl-dT^a

Y	Z	dNTP	X	V_{\max} nmol s ⁻¹ g ⁻¹	K_m μ M	V_{\max}/K_m μ L/(s g)
A	T	A	dT	0.0674 \pm 0.0067	20.9 \pm 2.4	3.22 \pm 0.16
A	T	C	dT	0.0472 \pm 0.0047	35.7 \pm 8.2	1.35 \pm 0.21
A	T	G	dT	0.0674 \pm 0.0040	25.2 \pm 1.3	2.68 \pm 0.28
A	T	T	dT	0.0678 \pm 0.0090	23.4 \pm 6.5	2.98 \pm 0.40
T	A	A	O^2 -Me-dT	0.0421 \pm 0.0078	38 \pm 18	1.28 \pm 0.52
T	A	C	O^2 -Me-dT	0.0411 \pm 0.011	60 \pm 31	0.73 \pm 0.16
T	A	G	O^2 -Me-dT	0.0424 \pm 0.00035	46 \pm 14	0.96 \pm 0.21
T	A	T	O^2 -Me-dT	0.0444 \pm 0.0103	48 \pm 26	1.03 \pm 0.28
T	A	A	O^2 -POB-dT	0.0431 \pm 0.0075	44 \pm 16	1.02 \pm 0.25
T	A	C	O^2 -POB-dT	0.0401 \pm 0.0048	66 \pm 16	0.62 \pm 0.08
T	A	G	O^2 -POB-dT	0.0416 \pm 0.0013	51 \pm 11	0.83 \pm 0.17
T	A	T	O^2 -POB-dT	0.0371 \pm 0.0038	36 \pm 11	1.06 \pm 0.24

^aData are expressed as the mean \pm SD of three independent experiments. The DNA substrate was P16/T24 as in Chart 1.