

Inhibition of Alzheimer amyloid β aggregation by polyvalent trehalose

Yoshiko Miura^{1,2}, Chouga You¹ and Reiko Ohnishi²

¹ School of Materials Science, Japan Advanced Institute of Science and Technology, 1-1 Asahidai, Nomi, Ishikawa 923-1292, Japan

² Department of Molecular Design and Engineering, Graduate School of Engineering, Nagoya University, Furo-cho, Chikusa-ku, Nagoya 464-8603, Japan

E-mail: miuray@jaist.ac.jp

Received 31 January 2008

Accepted for publication 30 June 2008

Published 24 July 2008

Online at stacks.iop.org/STAM/9/024407

Abstract

A glycopolymer carrying trehalose was found to suppress the formation of amyloid fibrils from the amyloid β peptide (1–42) ($A\beta$), as evaluated by thioflavin T assay and atomic force microscopy. Glycopolymers carrying sugar alcohols also changed the aggregation properties of $A\beta$, and the inhibitory effect depended on the type of sugar and alkyl side chain.

Neutralization activity was confirmed by *in vitro* assay using HeLa cells. The glycopolymer carrying trehalose strongly inhibited amyloid formation and neutralized cytotoxicity.

Keywords: glycopolymer, amyloid, multivalent effect, amphiphilicity

(Some figures in this article are in colour only in the electronic version)

1. Introduction

Alzheimer's disease (AD) is a progressive dementia, and the number of patients increases exponentially with life expectancy [1]. The development of an effective inhibitor of AD is urgently required. AD is characterized neuropathologically by extracellular deposition of amyloid senile plaques and neurofibrillary tangles in vulnerable AD brain regions. These plaques are primarily composed of fibrils of the amyloid β ($A\beta$) peptide, a small peptide composed of 39–43 amino acids [2]. The most abundant forms are 40 and 42 amino acids in length [3, 4]. These types of peptide are cleaved from a large protein called the amyloid precursor protein (APP) by a secretase enzyme [5]. Production of $A\beta$ s is a normal occurrence, although its function remains unclarified. However, in AD patients the peptide forms ordered fibrillar aggregates. This process involves a secondary structure transition from a disordered random structure to an ordered β -sheet conformation [6].

Therefore, it is required to develop medicinal compounds that inhibit $A\beta$ aggregation. Preventing $A\beta$ aggregation can be accomplished using various compounds such as dopamine [7] and heparin [8]. All of these agents work

through similar mechanisms, reducing cytotoxicity by inhibiting and delaying aggregation of $A\beta$ [9].

A nonreducing disaccharide with an $\alpha(1-1)$ linkage, trehalose, is the focus of much attention in view of its inhibitory effect on protein aggregation [10]. It has been reported that trehalose inhibits Huntington's disease *in vivo* [11] and amyloid formation of proteins (insulin [12] and $A\beta$ s [9]) *in vitro* owing to the hydration of proteins by the water-like hydrogen bond. Because the inhibitory effect of trehalose is not sufficient for medicinal application, an ingenious modification is required to amplify its activity.

The biological abilities of saccharides can be amplified by multivalency [13]. Many groups have reported the multivalent effect and the generation of the multivalent compounds such as glycopeptides [14], glycolixarenes [15], glycodendrimers [16], and glycopolymers [17]. In particular, glycopolymers carrying a saccharide at the side chain exhibit a large multivalent effect and applicability as biomaterials.

We have studied various glycopolymers with biorecognition abilities, such as lectin recognition [18]; hepatocyte culture; and amyloid inhibitor [19]. In this report, we investigated glycopolymers carrying trehalose to generate an efficient inhibitor of $A\beta$ aggregation in terms of the

effect of amplification of trehalose hydration properties. The glycopolymers carrying trehalose and sugar alcohols (lactitol and maltitol) were synthesized, and the inhibitory effect on amyloid aggregation was examined in terms of the saccharide structure and amphiphilicity of the glycopolymers.

2. Experimental procedure

2.1. Materials

The following reagents were used as received: amyloid β -protein ($A\beta(1-42)$) (Bachem AG, Switzerland), divinyl sebacate (Polyscience Inc., US), trehalose (Kanto Chemical, Japan), lactitol (Sigma-Aldrich, US), Dulbecco's modified Eagle's medium (DMEM), fetal bovine serum (FBS) (Invitrogen, US), thioflavin T (ThT), Omnipore membrane filter (0.45 μm , Millipore, USA), 3-[4,5-dimethylthiazol-2-yl]-2,5-diphenyltetrazolium bromide (MTT) (Wako Chemical, Japan), 8-anilino-1-naphthalenesulfonic acid ammonium salt (ANS), maltitol, and divinyl adipate (Tokyo Kasei, Japan). Lipase from *Candida antarctica* (CA) was kindly donated by Novo Nordisk Bioindustry Ltd.

2.2. Characterization

$^1\text{H-NMR}$ (300 MHz) and $^{13}\text{C-NMR}$ (75 MHz) spectra were recorded on Varian Gemini-300, and the spectra were measured in D_2O at room temperature. Gel permeation chromatography (GPC) was conducted using a JASCO 800 high-performance liquid chromatography instrument, on a Shodex SB-804HQ column with PBS(-) as an eluent, and molecular weight was determined using a pullulan standard. Fluorescence spectra were measured on FP-6500 (JASCO, Japan) at room temperature. The diameters of the glycopolymer were measured by dynamic light scattering (DLS) with Nano-ZS (Malvern, UK). Atomic force microscopy (AFM) was performed using SPA400 (Seiko Instrument Inc., Japan).

2.3. Syntheses of glycopolymers

The glycopolymers were synthesized by enzymatic esterification of lipase CA and radical polymerization according to the literature [20]. Syntheses of saccharide vinyl esters were confirmed from NMR spectra. The molecular structures and molecular weights of the polymers are summarized in figure 1 and table 1, respectively.

2.4. In vitro amyloid formation of $A\beta(1-42)$ [19]

$A\beta(1-42)$ was dissolved in 0.02% ammonia solution at a concentration of 200 μM , and any aggregates formed were removed by centrifugation using a CS 120 FX (Hitachi, Tokyo, Japan) at 16000 $\times g$ for 30 min at 4 °C. Next, the supernatant was mixed with phosphate buffer (20 mM phosphate buffer, pH 7.4, and 100 mM NaCl) to a final peptide concentration of 20 μM . The peptide solution was incubated with each glycopolymer at 37 °C.

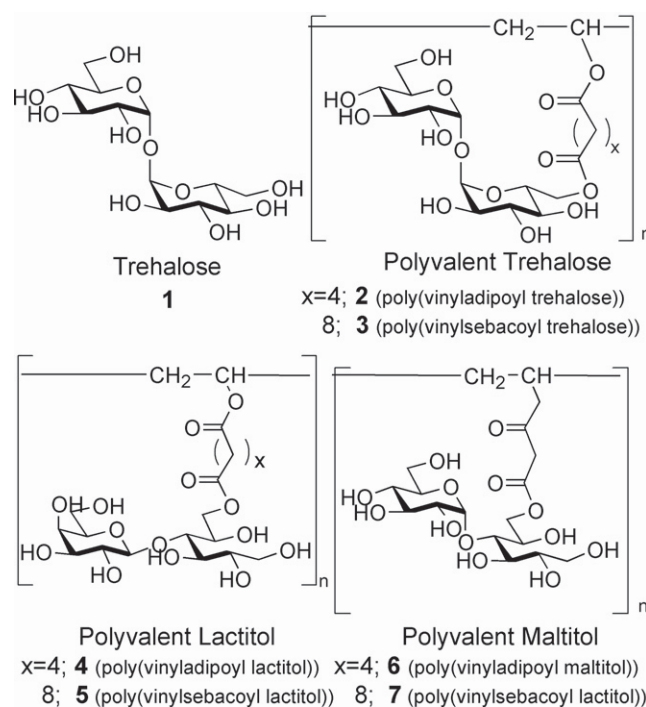


Figure 1. Molecular structures of the glycopolymers with trehalose and sugar alcohols.

Table 1. Molecular weights of the glycopolymers^a

Sample	M_w	M_w/M_n
2	4.2×10^4	1.3
3	2.1×10^4	1.2
4	3.7×10^4	1.7
5	3.5×10^4	1.8
6	7.1×10^3	1.3
7	5.0×10^4	1.2

^ausing pullulan standard.

2.5. ThT fluorescence assay

Amyloid fibril formation was evaluated on the basis of fluorescence emission of ThT using a 3-mm light-path quartz cuvette. $A\beta(1-42)$ (20 μM) was incubated at 37 °C in 50 mM phosphate buffer, 100 mM NaCl, at pH 7.4, 100 μM glycopolymer, and 50 μL of $A\beta$ solution was periodically added to 300 μL of aqueous solution of 5 μL of ThT (50 μM) in 50 mM glycine-NaOH. ThT fluorescence intensity was measured at an excitation wavelength of 440 nm and an emission wavelength of 482 nm. The fluorescence intensity of the control sample after 12 h incubation without a glycopolymer was used as the standard (1.0). Fluorescence intensity was taken as the average of at least three samples.

2.6. AFM measurements

$A\beta(1-42)$ solution (20 μl) was incubated with 100 μM glycopolymer in phosphate buffer (50 mM phosphate buffer, 100 mM NaCl, at pH 7.4). An aliquot of a sample solution (5 μl) was placed on freshly cleaved mica, rinsed with deionized water, and dried. The resultant sample was scanned using SPA400.

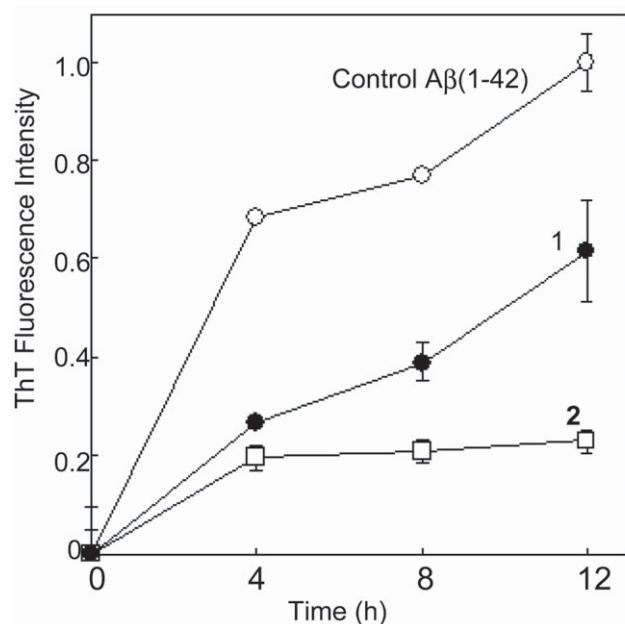


Figure 2. Effect of trehalose (1) and polyvalent trehalose (2) on $A\beta(1-42)$ aggregation. Control (○), 1 (●) and 2 (□).

2.7. Neutralization of $A\beta(1-42)$ with a glycopolymer

Neutralization activity of a glycopolymer (2) was examined *in vitro* by HeLa cell cytotoxicity assay. HeLa cells were cultured in DMEM containing 10% FBS at 37 °C incubation in 5% CO₂. An aliquot containing 1×10^3 cells in 100 μ l of DMEM was transferred to each well of a 96-well plate coated with collagen, and DMEM was removed. $A\beta(1-42)$ solution (30 μ l, 20 μ M) was preincubated with 100 μ M 2 at 37 °C for 8 h in 20 mM HEPES, 100 mM NaCl, at pH 7.4, and 30 μ l of the medium was added to each well. The mixed sample solution was incubated for 4 h at 37 °C. The plate was centrifuged (DU-600), and the medium was removed using an aspirator. A 100 μ l aliquot of MTT solution was added to each well. The absorbance at 570 and 650 nm was measured using a plate reader. The difference in absorbance between 570 and 650 nm was plotted to measure cell viability. The absorbance of HeLa cells alone was used as the standard (100%). The experiments were performed using at least three samples.

3. Results

3.1. Inhibitory effect of glycopolymers on $A\beta(1-42)$ aggregation

The aggregation behavior of $A\beta(1-42)$ was investigated by adding trehalose additives (1, 2, and 3) (figure 2). The aggregation behavior of $A\beta$ was examined on the basis of fluorescence intensity of ThT. The fluorescence intensity of $A\beta$ without additives (control) gradually increased, suggesting amyloid formation. The fluorescence intensity after 12 h incubation was set as 1.0, and fluorescence intensities in the presence of trehalose additives were compared with that of the standard. The addition of trehalose

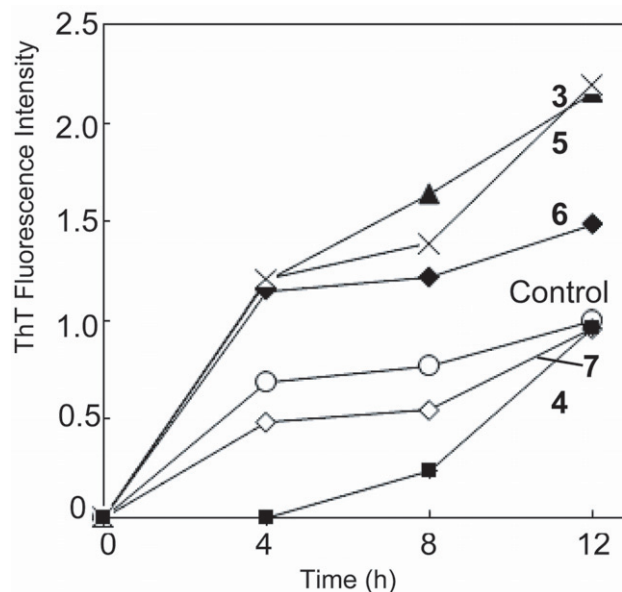


Figure 3. Effect of glycopolymer (3, 4, 5 and 6) on $A\beta(1-42)$ aggregation. Control (○), 3 (▲), 4 (■), 5 (X), 6 (◆), and 7 (◇).

attenuated the aggregation properties of the peptide, in which trehalose decreased fluorescence intensity up to 60 %.

The addition of polyvalent trehalose with a short alkyl chain (2, poly(vinyladipoyl trehalose)) markedly decreased the fluorescent intensity of ThT. The fluorescence intensity of 2 was only 20% of that of control. The efficacy lasted for a long time over 15 h. These results indicate that the inhibitory effect of trehalose was amplified by multivalency. On the other hand, the polyvalent trehalose with a longer alkyl side chain (3, poly(vinylsebacoyl trehalose)) did not inhibit amyloid formation but rather induced it.

Polyvalent sugar alcohols (4–7) were also investigated for their inhibitory effect on amyloid formation (figure 3). Poly(vinyladipoyl lactitol) (4) strongly inhibited amyloid formation from 0 to 8 h, but amyloid formation rapidly proceeded after 8 h incubation. Poly(vinylsebacoyl lactitol) (5) induced amyloid formation. Polyvalent maltitol (6 (poly(vinyladipoyl maltitol) and 7 (poly(vinylsebacoyl maltitol)) also changed the time course curve of ThT fluorescence intensity, and the effects on amyloid formation were much smaller than those of polyvalent trehalose (2 and 3) and polyvalent maltitol (6 and 7).

Polyvalent trehalose with a short alkyl chain (2) showed the strongest inhibition effect on $A\beta(1-42)$ aggregation.

3.2. Morphology of $A\beta(1-42)$ observed by AFM

The inhibitory effect on amyloid formation was also investigated by AFM (figure 4). $A\beta(1-42)$ formed amyloid fibrils of 30–120 nm diameter and 300 nm–2 μ m length owing to aggregation properties. The addition of 2 totally changed the morphology, and amyloid fibrils were not observed. Nonuniform round aggregates of 40–120 nm diameter were observed. The addition of 3 also changed the morphology into round aggregates, but fibril formation with 30–150 nm in width was still observed, which corresponded to the ThT

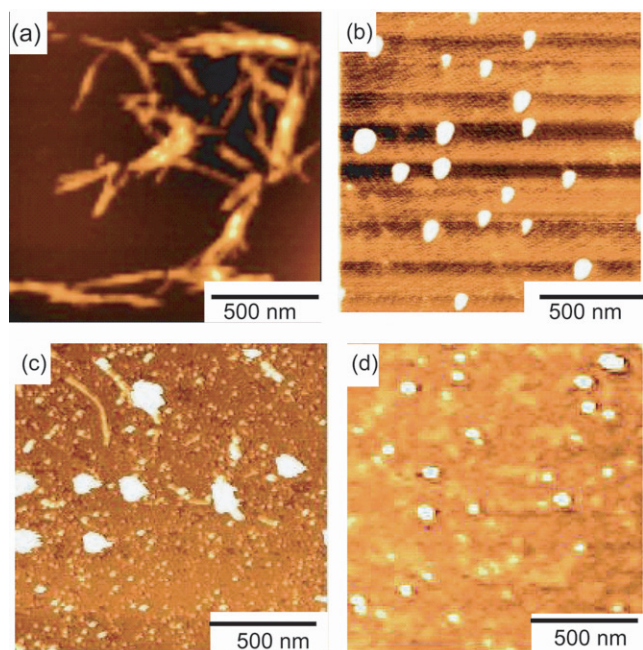


Figure 4. AFM of $A\beta(1-42)$ (a) without glycopolymer, (b) in the presence of **2** and (c) in the presence of **3**. Control sample of glycopolymer (**2**) without $A\beta(1-42)$ (d).

fluorescence intensity. The round aggregates with **2** and **3** resembled the morphology of the glycopolymer itself. Therefore, the round aggregates in figures 4(b) and (c) were a mixture of $A\beta$ and glycopolymers. The glycopolymers had affinities with $A\beta(1-42)$ owing to trehalose and the amphiphilic properties of the polymer, which induced the formation of round aggregates.

3.3. Neutralization effect of glycopolymer with $A\beta(1-42)$

Polyvalent trehalose of **2** was evaluated for its activity of neutralizing $A\beta(1-42)$ in HeLa cells (figure 5). $A\beta$ was incubated in advance and added to HeLa cells. When the cells were incubated with $A\beta$ alone, $A\beta$ exhibited cytotoxicity reducing MTT activity by about 50%. On the other hand, coincubation of $A\beta$ with a polyvalent trehalose (**2**) decreased cytotoxicity, indicating protection against $A\beta(1-42)$. The cell survival rate was nearly 100% compared with that without additions. Moreover, the incubation with **2** alone did not show cytotoxicity. These results showed that polyvalent trehalose can be an effective inhibitor of $A\beta$ aggregation.

3.4. Amphiphilic properties of glycopolymers

The glycopolymers formed a self-assembling structure in aqueous solution owing to the amphiphilic structure of the polymer, and the size of the glycopolymer in aqueous solution was measured by DLS (table 2). The diameters of the polyvalent saccharides (**2-7**) depended on alkyl chain length and the type of saccharides. The diameters of **2** and **4** were small, approximately 30–40 nm, suggesting the compaction of the polymer. Other glycopolymers (**3, 5, 6** and **7**) formed larger aggregates with diameters over 100 nm. The order of the diameters is $4 < 2 \ll 5 \ll 3, 6 \ll 7$.

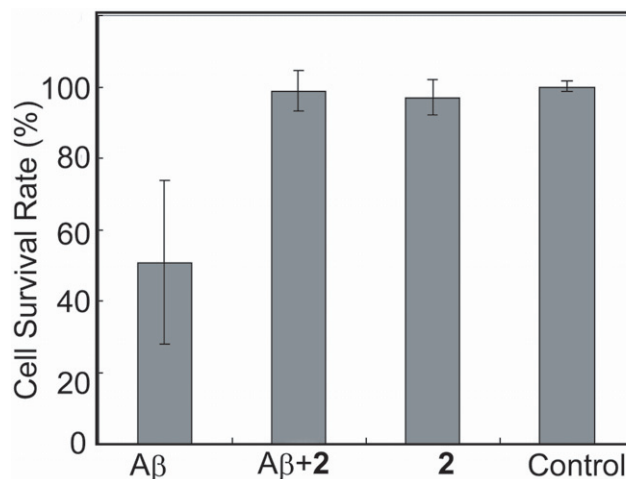


Figure 5. Neutralization of $A\beta(1-42)$ with a glycopolymer (**2**). The survival rate of HeLa cells against $A\beta(1-42)$ were monitored in the presence of the glycopolymer. Results were the average of at least three samples.

Table 2. The diameters of glycopolymers in aqueous solution.

Sample	Diameter (nm)
2	45.1
3	145
4	39.4
5	64.1
6	133
7	454

The inhibitory effect on amyloid formation determined from ThT fluorescence intensity was plotted against the diameter of the glycopolymers (figure 6). The glycopolymers with small diameters (0–50 nm, **2** and **4**) inhibited amyloid formation, and ThT fluorescence intensities in the presence of the glycopolymers were below 0.4. The glycopolymer with medium diameters (approximately 100 nm) of **3, 5** and **6** rather enhanced amyloid fibril formation with ThT fluorescence intensities above 1.0. The glycopolymer with a larger diameter (over 400 nm) (**7**) did not show a marked difference from the control without sugar additives.

The amphiphilic property of the glycopolymer (**2**) was also investigated using a hydrophobic indicator of 1-anilinonaphthalene-8-sulfonic acid (ANS). The fluorescence intensity of ANS showed rather hydrophobic properties with a peak of 470 nm (figure 7).

4. Discussion

Glycopolymer additives changed the amyloid formation of $A\beta(1-42)$. The extents of inhibitory effect were in the order of $2 > 4 \gg 7 \gg 6 > 5 > 3$. Glycopolymers **2, 4** and **7** inhibited amyloidosis, but glycopolymers **3, 5** and **6** rather induced it. The best inhibitor was polyvalent trehalose of **2**. These results indicate that the inhibitory effect of the saccharides was amplified by a multivalency-like protein-saccharide interaction. Trehalose is a unique disaccharide with a clam

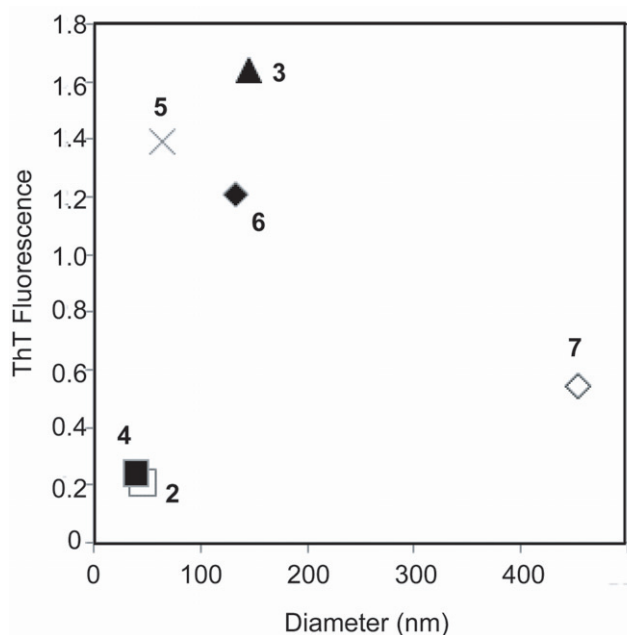


Figure 6. Correlation between diameter of glycopolymer and aggregation properties of $A\beta(1-42)$. The diameter of the glycopolymer was plotted against ThT fluorescence intensity after 8 h incubation.

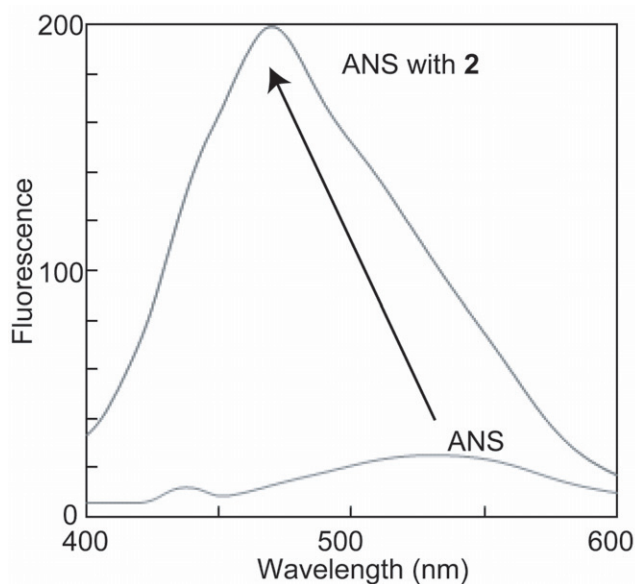


Figure 7. Fluorescence spectra of ANS in **2** ($100 \mu\text{M}$) with excitation wavelength at 345 nm.

shell structure [21], which induces the water-like hydrogen bonding with lipids [22] and proteins [23] to preserve the biological activities. The hydrogen bonding formation of trehalose with water clusters was also reported [24]. The biological activities of trehalose are different from those of other saccharides with open structures such as lactose and maltose. Although the modification of trehalose might partially change the structure, the multivalent trehalose of **2** did not cause the loss of the ability of hydrogen-bonding formation with ambient biomolecules and water. The glycopolymer **2** rather induced the alignment of trehalose to

facilitate the hydrogen bonding with $A\beta(1-42)$, resulting in an efficient inhibitory effect.

At the same time the inhibitory effect was greatly affected by the alkyl chain length of the glycopolymers. For example, polyvalent trehalose with a short alkyl chain (**2**) showed a strong inhibitory effect, but polyvalent trehalose with a longer alkyl chain (**3**) induced amyloid formation. It has been reported that amyloid formation is induced by surfactants of amphiphilic molecules [25–27]. Not only the saccharide structure of the polymer but also amphiphilicity plays important roles in the amyloid formation and inhibition. Interestingly, the inhibitory effect on amyloid formation correlated to the diameter of the glycopolymers owing to amphiphilicity. The glycopolymers with small diameters (**2** and **4**) reflected the hydration properties of saccharides and showed the effective inhibition activity. The glycopolymer **4** did not inhibit the aggregation of $A\beta$ for a long time (over 8 h), but the polymer inhibited the aggregation until 8 h of incubation. The glycopolymers with diameters of about 100 nm (**3**, **5** and **6**) showed a hydrophobic cavity inside the polymer micelle, and the hydrophobic interaction with $A\beta$ was stimulated, resulting in the aggregation of the peptide. The glycopolymer with a larger diameter (**7**) did not exhibit the markedly strong inhibitory effect on $A\beta$. The amphiphilic properties of **2** were also investigated using a hydrophobic indicator of ANS. Although glycopolymer **2** showed the strongest inhibitory effect on $A\beta$ aggregation owing to the hydration properties of trehalose, the fluorescence intensity of ANS showed rather hydrophobic properties with a peak of 470 nm. These data indicate that the glycopolymer has a densely packed trehalose layer at the periphery for the hydration of $A\beta$. The amphiphilic structures of the glycopolymers are still under investigation to develop drugs for the inhibition of $A\beta$ aggregation.

Trehalose preserves biological activities owing to its special physical chemical properties of hydration. Our results indicate that the polyvalent trehalose can amplify the hydration properties, which is utilized for not only the inhibition of $A\beta$ aggregation, but also for treatment of other conformation diseases. In addition, the glycopolymers with trehalose show amphiphilic properties, which can be utilized as a novel surfactant or an interface-reforming reagent.

5. Conclusions

The glycopolymers were analyzed as an amyloid inhibitor. The glycopolymer with trehalose and sugar alcohols changed the aggregation properties of $A\beta(1-42)$. The inhibitory effect on $A\beta$ aggregation depended on the type of saccharide and amphiphilicity. Poly(vinyladipoyl trehalose) showed the strongest inhibitory effect on amyloid formation, and the morphology of $A\beta$ changed from fibrils to round aggregates. The cytotoxicity of $A\beta$ was completely neutralized by the polymer added. The glycopolymer with trehalose was a novel inhibitor of $A\beta$ aggregation.

Acknowledgments

This work was supported by a Grant-in-Aid for Young Scientists (B), JST seeds, and Shibuya Foundation.

References

- [1] Selkoe D J 2001 *Physiol. Rev.* **81** 741
- [2] Selkoe D J 1994 *Annu. Rev. Neurosci.* **17** 489
- [3] Glenner G G and Wong C W 1984 *Biochem. Biophys. Res. Commun.* **122** 1131
- [4] Glenner G G and Wong C W 1984 *Biochem. Biophys. Res. Commun.* **120** 885
- [5] Sinha S and Lieberbur I 1999 *Proc. Natl Acad. Sci. USA* **96** 11049
- [6] Thompson L K 2003 *Proc. Natl Acad. Sci. USA* **100** 383
- [7] Opazo C et al 2002 *J. Biol. Chem.* **277** 40302
- [8] Yoshiike Y, Tanemura K, Murayama O, Akagi T, Murayama M, Sato S, Sun X, Tanaka N and Takashima A 2001 *J. Biol. Chem.* **276** 32293
- [9] Janus C 2003 *CNS Drugs* **17** 457
- [10] Liu R, Barkhordarian H, Emadi S, Park C B and Sierks M R 2005 *Neurobiol. Disease* **20** 74
- [11] Tanaka M, Machida Y, Niu S, Ikeda T, Jane N R, Doi H, Kurosawa M, Nekooki M and Nukina N 2004 *Nat. Med.* **10** 148
- [12] Arora A, Ha C and Park C-B 2004 *FEBS Lett.* **564** 121
- [13] Mammen M, Choi S-K and Whitesides G M 1998 *Angew. Chem. Int. Ed.* **37** 2754
- [14] Lundquist J J, Debenham S D and Toone E J 2000 *J. Org. Chem.* **65** 8245
- [15] Dondroni A, Kleban M, Hu X, Marra A and Banks H D 2002 *J. Org. Chem.* **67** 4722
- [16] Roy R and Kim J M 1999 *Angew. Chem. Int. Ed.* **38** 369
- [17] Kobayashi A, Akaïke T, Kobayashi K and Sumitomo H 1986 *Makromol. Chem. Rapid. Commun.* **7** 645
- [18] Miura Y, Ikeda T and Kobayashi K 2003 *Biomacromolecules* **4** 410
- [19] Miura Y, Yasuda Y, Yamamoto K, Koike M, Nishida Y and Kobayashi K 2007 *Biomacromolecules* **8** 2129
- [20] Miura Y, Wada N, Nishida Y, Mori H and Kobayashi K 2004 *J. Polym. Sci. A Polym. Chem.* **42** 4598
- [21] Brown G M, Rohre D C, Berking B, Beevers C A, Gould R O and Simpson R 1972 *Acta Crystallogr. B* **28** 3145
- [22] Lambruschini C, Relini A, Ridi A, Cordone L and Gliozzi A 2000 *Langmuir* **16** 5467
- [23] Sola-Penna M and Meyer-Fernandes J R 1998 *Arc. Biochem. Biophys.* **360** 10
- [24] Pagnotta S E, Ricci M A, Bruni F, McLain S and Magazu S 2008 *Chem. Phys.* **345** 159
- [25] Li Y, Gao M and Wang Y 2006 *J. Phys. Chem. B* **110** 18040
- [26] Cao M, Han Y, Wang J and Wang Y 2007 *J. Phys. Chem. B* **111** 13436
- [27] Hamill A C, Wang S C and Lee C T Jr 2007 *Biochemistry* **46** 7694