

THE BACTERIOSTATIC ACTIVITY OF CERIUM, LANTHANUM, AND THALLIUM

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Received for publication June 23, 1947

The salts of cerium, lanthanum, and thallium have long been known to possess bacteriostatic properties in low concentrations, but there is little information in the literature concerning the variability of their toxicity toward different species of microorganisms. A report of the concentrations of the salts limiting development of various types of bacteria may have importance in relation to their possible use as bacteriostatic or bactericidal agents.

Bokorny (1894) found cerium compounds relatively much more toxic for bacteria than for algae. Hebert (1907) reported that cerium and lanthanum sulfates in concentrations of 5 to 10 grams per liter showed little toxicity toward *Aspergillus niger* and yeast. Sartory and Bailly (1922) reported that 0.2 per cent lanthanum sulfate depressed the growth of *Aspergillus fumigatus* in Raulin's solution and practically inhibited spore formation. Frouin (1912) observed that 0.005 grams of lanthanum sulfate per 100 ml of medium stimulated the growth of the tubercle bacillus but that higher concentrations were toxic. Frouin and Roudski (1914) studied the toxicity of lanthanum and thorium for the cholera and dysentery organisms.

Other investigators who reported bacteriostatic or lethal effects of salts of the rare earth group include Drossback (1897), Brooks (1921), Grenet and Drouin (1927), Zirpolo (1924), Frouin (1920), Simonini (1914), Doerr (1920), Eisenberg (1918), and Hotchkiss (1923). A general review of the earlier literature on this subject is found in Buchanan and Fulmer (1930). McKenzie (1941) employed thallium acetate in a medium recommended for the enrichment of the streptococci causing mastitis. The effect of cerium on enzyme activity was reported by Gould (1936). Olszewski (1932) observed no significant reduction in the bacteria of river water when 1 ppm of cerous or ceric chloride or ceric sulfate was employed. Richards (1932) reported thallium to be a growth stimulant for yeast.

The present paper reports a further investigation of the bacteriostatic activity of the salts of cerium, lanthanum, and thallium.

METHODS

Thirty-nine species of bacteria, representing 16 different genera, were employed in this study. Also, 35 species of fungi, comprising 18 genera, were used in a limited comparison of the mycostatic and bacteriostatic effects of the compounds. The salts used were cerium chloride (CeCl_3 , cp, E. H. Sargent); cerium nitrate ($\text{Ce}(\text{NO}_3)_3 \cdot 6\text{H}_2\text{O}$, cp, General Chemical Co.); anhydrous ceric sulfate ($\text{Ce}(\text{SO}_4)_2$, G. F. Smith Chemical Co.); lanthanum chloride (cp, E. H. Sargent), and thallium nitrate (E. H. Sargent).

Other chemicals were used in some of the tests in order to determine their effect on the toxicity of the test substances. These chemicals included the sulfates and chlorides of sodium, magnesium, barium, and ammonium, and the chlorides of calcium and lithium. Stock solutions of the salts were made and the amount required for each test medium was removed by pipette. The tests were made on petri plates of solidified agar containing the specified amounts of the salts. The basal medium consisted of 1 per cent Difco peptone and 1.5 per cent Difco agar in distilled water. Inoculations of the agar plates were made by means of a 1½-mm nichrome wire loop using a 24-hour broth culture. Radial streak inoculations were made, using 8 cultures to each plate. Incubation was at 37 C except for those soil and water forms which grow better at a lower temperature and were incubated at room temperature (22 to 27 C). Observations and records were made after incubation for 1, 2, 3, and 5 days.

RESULTS

The toxicity of cerium salts for bacteria. The bacteriostatic action of three cerium salts, the trivalent cerium chloride and cerium nitrate and the tetravalent ceric sulfate, was determined against 40 different species. The results are presented in table 1. The chloride was found to be definitely less toxic than either the sulfate or nitrate of cerium. No significant difference in toxicity between the sulfate and nitrate of cerium was noted.

The reaction of the media was not adjusted after the addition of the cerium salts, and the pH values were found to vary as follows: for cerium chloride agar, 6.3 at 0.0002 M concentration to 5.95 at 0.0014 M concentration; for cerium nitrate, pH 5.8 at 0.0001 M to 5.65 at 0.0009 M; and for cerium sulfate, pH 6.7 at 0.0002 M to 5.4 at 0.0008 M.

The toxicity of cerium nitrate and cerium chloride varied relatively little among the different species of bacteria tested, and even fewer variations were observed with cerium sulfate. Of the bacterial species tried, *Aerobacter aerogenes*, *Aerobacter cloacae*, *Salmonella aertrycke*, and *Achromobacter lipolyticum* were most tolerant of cerium. The *Torula rosea* culture proved to be far more resistant to the cerium compounds than the most resistant bacteria.

The effect of pH and the presence of other salts on the toxicity of cerium compounds. In order to determine the effect of pH on cerium toxicity, the media were prepared using cerium nitrate in concentrations varying from 0.0003 M to 0.0007 M, then adjusted to pH 6.0 and 8.0. The results obtained after 2 days' incubation are presented in table 2. It was observed that at pH 8.0 all cultures except *Staphylococcus albus* developed without hindrance in 0.0007 M cerium nitrate, whereas at pH 6.0 a considerable number of cultures failed to grow in the 0.0003 M concentration of cerium.

The effect of other salts on the bacteriostatic activity of cerium compounds. The effect of various salts which are sometimes used in culture media on the bacteriostatic activity of cerium compounds was determined by adding the salts separately to the cerium-containing media and observing for bacterial growth after inoculations. It was found that sodium chloride in concentrations up to

TABLE 1
The toxicity of various cerium compounds for certain bacteria
 Incubation for 2 days

CULTURE	MOLECULAR CONCENTRATION					
	CeCl ₃		Ce(NO ₃) ₃		Ce(SO ₄) ₂	
	A	B	A	B	A	B
<i>Salmonella paratyphi</i>	0.0006	0.0008	0.0004	0.0005	0.0004	0.0006
<i>Salmonella pullorum</i>	0.0006	0.0008	0.0004	0.0005	0.0004	0.0006
<i>Salmonella schottmuelleri</i>	0.0008	0.0010	0.0004	0.0005	0.0004	0.0006
<i>Salmonella enteritidis</i>	0.0008	0.0010	0.0004	0.0005	0.0004	0.0006
<i>Salmonella aertrycke</i>	0.0008	0.0010	0.0004	0.0005	0.0006	0.0008
<i>Salmonella gallinarum</i>	0.0006	0.0008	0.0004	0.0005	0.0004	0.0006
<i>Salmonella suispestifer</i>	0.0008	0.0010	0.0004	0.0005	0.0004	0.0006
<i>Eberthella typhosa</i>	0.0008	0.0010	0.0004	0.0005	0.0004	0.0006
<i>Shigella sonnei</i>	0.0008	0.0010	0.0005	0.0006	0.0004	0.0006
<i>Shigella dysenteriae</i>	0.0008	0.0010	0.0003	0.0004	0.0004	0.0006
<i>Aerobacter aerogenes</i>	0.0008	0.0010	0.0005	0.0006	0.0006	0.0008
<i>Aerobacter cloacae</i>	†	†	0.0005	0.0006	0.0006	0.0008
<i>Escherichia coli</i>	0.0008	0.0010	0.0004	0.0005	0.0004	0.0006
<i>Escherichia communior</i>	0.0008	0.0010	0.0004	0.0005	0.0004	0.0006
<i>Escherichia acidilactici</i>	0.0008	0.0010	0.0004	0.0005	0.0004	0.0006
<i>Citrobacter intermedium</i>	0.0008	0.0010	0.0004	0.0005	0.0004	0.0006
<i>Alcaligenes faecalis</i>	0.0010	0.0012	0.0004	0.0005	0.0004	0.0006
<i>Proteus vulgaris</i>	0.0008	0.0010	0.0004	0.0005	0.0004	0.0006
<i>Pseudomonas aeruginosa</i>	0.0004	0.0006	0.0003	0.0004	0.0004	0.0006
<i>Pseudomonas ovalis</i>	0.0006	0.0008	0.0003	0.0004	0.0004	0.0006
<i>Pseudomonas graveolens</i>	0.0004	0.0006	0.0001	0.0002	0.0004	0.0006
<i>Pseudomonas syncyanea</i>	0.0006	0.0008	0.0003	0.0004	0.0004	0.0006
<i>Pseudomonas mucedolens</i>	0.0004	0.0006	0.0001	0.0002	0.0004	0.0006
<i>Flavobacterium sauevolens</i>	0.0010	0.0012	0.0005	0.0006	0.0004	0.0006
<i>Achromobacter lipolyticum</i>	0.0010	0.0012	0.0006	0.0007	0.0006	0.0008
<i>Serratia marcescens</i>	0.0010	0.0012	0.0006	0.0007	0.0004	0.0006
<i>Bacillus subtilis</i>	0.0010	0.0012	0.0006	0.0007	0.0004	0.0006
<i>Bacillus mesentericus</i>	0.0008	0.0010	0.0004	0.0005	0.0004	0.0006
<i>Bacillus mycoides</i>	0.0008	0.0010	0.0005	0.0006	0.0004	0.0006
<i>Bacillus fusiformis</i>	0.0008	0.0010	0.0004	0.0005	0.0004	0.0006
<i>Bacillus metiens</i>	0.0008	0.0010	0.0005	0.0006	0.0004	0.0006
<i>Staphylococcus candidus</i>	0.0008	0.0010	0.0005	0.0006	0.0004	0.0006
<i>Staphylococcus flavus</i>	0.0008	0.0010	0.0003	0.0004	0.0004	0.0006
<i>Staphylococcus aureus</i>	0.0008	0.0010	0.0007	0.0008	0.0004	0.0006
<i>Staphylococcus albus</i>	0.0010	0.0012	0.0004	0.0005	0.0004	0.0006
<i>Sarcina lutea</i>	0.0006	0.0008	0.0004	0.0005	0.0004	0.0006
<i>Sarcina conjunctivae</i>	0.0006	0.0008	0.0007	0.0008	0.0004	0.0006
<i>Rhodococcus agilis</i>	0.0006	0.0008	0.0003	0.0004	0.0004	0.0006
<i>Rhodococcus rosaceus</i>	0.0006	0.0008	0.0003	0.0004	0.0004	0.0006
<i>Torula rosea</i>	0.0012	*	0.0009	*	0.0008	*

A = concentration permitting growth in 2 days.

B = concentration inhibiting growth in 2 days.

* Inhibiting concentration was not determined.

† Concentration less than .0003 M was not employed.

TABLE 2

The effect of pH on the toxicity of Ce(NO₃)₂ for certain bacteria

CULTURE	MOLECULAR CONCENTRATION			
	pH 6		pH 8	
	A	B	A	B
<i>S. paratyphi</i>	0.0003	0.0004	0.0007	*
<i>S. pullorum</i>	0.0003	0.0004	0.0007	*
<i>S. schottmuelleri</i>	0.0005	0.0007	0.0007	*
<i>S. enteritidis</i>	0.0004	0.0005	0.0007	*
<i>S. aertrycke</i>	0.0004	0.0005	0.0007	*
<i>S. gallinarum</i>	0.0004	0.0005	0.0007	*
<i>S. suispestifer</i>	0.0005	0.0006	0.0007	*
<i>E. typhosa</i>	0.0004	0.0005	0.0007	*
<i>S. sonnei</i>	†	0.0003	0.0007	*
<i>S. dysenteriae</i>	0.0003	0.0004	0.0007	*
<i>A. aerogenes</i>	0.0007	*	0.0007	*
<i>A. cloacae</i>	0.0006	0.0007	0.0007	*
<i>E. coli</i>	0.0004	0.0005	0.0007	*
<i>E. communior</i>	0.0004	0.0005	0.0007	*
<i>E. acidilactici</i>	0.0003	0.0004	0.0007	*
<i>C. intermedium</i>	0.0006	0.0007	0.0007	*
<i>A. faecalis</i>	0.0003	0.0004	0.0007	*
<i>P. vulgaris</i>	0.0003	0.0004	0.0007	*
<i>P. aeruginosa</i>	†	0.0003	0.0007	*
<i>P. ovalis</i>	†	0.0003	0.0007	*
<i>P. graveolens</i>	†	0.0003	0.0007	*
<i>P. syncyanea</i>	†	0.0003	0.0007	*
<i>P. mucedolens</i>	†	0.0003	0.0007	*
<i>F. suaveolens</i>	0.0007	*	0.0007	*
<i>A. lipolyticum</i>	0.0007	*	0.0007	*
<i>S. marcescens</i>	0.0007	*	0.0007	*
<i>B. subtilis</i>	†	0.0003	0.0007	*
<i>B. mesentericus</i>	0.0003	0.0004	0.0007	*
<i>B. mycooides</i>	†	0.0003	0.0007	*
<i>B. fusiformis</i>	†	0.0003	0.0007	*
<i>B. meliens</i>	†	0.0003	0.0007	*
<i>S. aureus</i>	0.0006	0.0007	0.0007	*
<i>S. candidus</i>	0.0004	0.0005	0.0007	*
<i>S. albus</i>	0.0003	0.0004	0.0004	0.0005
<i>S. flava</i>	0.0003	0.0004	0.0007	*
<i>S. lutea</i>	†	0.0003	0.0007	*
<i>S. conjunctivae</i>	†	0.0003	0.0007	*
<i>R. agilis</i>	0.0003	0.0004	0.0007	*
<i>R. rosaceous</i>	†	0.0003	0.0007	*
<i>Torula rosea</i>	0.0007	*	0.0007	*

A = concentration permitting growth in 2 days.

B = concentration inhibiting growth in 2 days.

* Concentration greater than 0.0007 M was not employed.

† Concentration less than 0.0003 M was not employed.

TABLE 3

The effect of 0.4 M of NaCl and 0.1 M Na₂SO₄ on toxicity of Ce(SO₄)₂

CULTURE	Ce(SO ₄) ₂ CONC. (M)	CONTROLS FOR			Ce(SO ₄) ₂ PLUS	
		Ce(SO ₄) ₂	NaCl	Na ₂ SO ₄	NaCl	Na ₂ SO ₄
Vigor of growth: incubation for 2 days						
<i>S. paratyphi</i>	0.0006	-	++	++	-	+
<i>S. pullorum</i>	0.0006	-	++	++	-	++
<i>S. schottmuelleri</i>	0.0008	-	+++	+++	-	++
<i>S. enteritidis</i>	0.0008	-	++	++	-	+
<i>S. aertrycke</i>	0.0008	+	++	++	-	+
<i>S. gallinarum</i>	0.0004	+	++	++	++	+
<i>S. suispestifer</i>	0.0008	-	++	++	-	+
<i>E. typhosa</i>	0.0006	-	+++	+++	-	++
<i>S. sonnei</i>	0.0006	-	++	++	-	+
<i>S. dysenteriae</i>	0.0004	++	++	++	+	+
<i>A. aerogenes</i>	0.0008	+	++	++	-	++
<i>A. cloacae</i>	0.0008	++	+++	++	-	++
<i>E. coli</i>	0.0006	-	+++	++	-	+
<i>E. communior</i>	0.0008	-	+++	+++	-	+
<i>C. intermedium</i>	0.0006	-	++	++	-	+
<i>A. faecalis</i>	0.0006	-	+++	++	-	+
<i>P. vulgaris</i>	0.0006	-	++	++	-	+
<i>P. aeruginosa</i>	0.0004	++	++	+++	-	++
<i>P. ovalis</i>	0.0004	++	++	++	+	++
<i>P. graveolens</i>	0.0004	++	++	++	-	+
<i>P. syncyanea</i>	0.0004	+	+++	+++	-	++
<i>P. mucedolens</i>	0.0004	+	++	++	-	+
<i>F. suaveolens</i>	0.0006	-	++	++	-	++
<i>A. lipolyticum</i>	0.0008	+	++	++	+	+
<i>S. marcescens</i>	0.0006	+	++	++	+	++
<i>B. subtilis</i>	0.0006	+	++	+++	-	++
<i>B. mesentericus</i>	0.0008	-	++	+++	-	+
<i>B. mycoides</i>	0.0004	++	++	+++	+	++
<i>B. fusiformis</i>	0.0004	++	++	+++	++	++
<i>B. metiens</i>	0.0004	++	++	++	++	++
<i>S. aureus</i>	0.0004	++	++	++	++	++
<i>S. candidus</i>	0.0006	-	++	++	-	+
<i>S. albus</i>	0.0004	++	++	++	+	++
<i>S. flavus</i>	0.0004	++	++	++	+	+
<i>S. lutea</i>	0.0004	+	++	++	-	+
<i>S. conjunctivae</i>	0.0004	++	++	++	+	++
<i>R. agilis</i>	0.0004	+	++	++	+	+
<i>R. rosaceus</i>	0.0004	+	++	++	+	+
<i>Torula rosea</i>	0.0004	++	++	++	++	++

(-) = complete inhibition of growth.

(+) = moderate growth.

(++) = good growth.

(+++)= growth better than on nutrient agar control.

TABLE 4
The bacteriostatic activity of lanthanum chloride and thallium nitrate

ORGANISM	MOLECULAR CONCENTRATION			
	Permitting growth		Preventing growth	
	LaCl ₃	TlNO ₃	LaCl ₃	TlNO ₃
<i>S. paratyphi</i>	0.0002	0.0006	0.0004	0.0007
<i>S. pullorum</i>	0.0002	0.0006	0.0004	0.0007
<i>S. schottmuelleri</i>	0.0004	0.0007	0.0006	0.0008
<i>S. enteritidis</i>	0.0004	0.0007	0.0006	0.0008
<i>S. aertrycke</i>	0.0004	0.0007	0.0006	0.0008
<i>S. gallinarum</i>	0.0004	0.0006	0.0006	0.0007
<i>S. suispestifer</i>	0.0004	0.0006	0.0006	0.0007
<i>E. typhosa</i>	0.0004	0.0006	0.0006	0.0007
<i>S. conjunctivae</i>	0.0006	0.0008	0.0008	0.0010
<i>S. sonnei</i>	0.0004	0.0007	0.0006	0.0008
<i>S. dysenteriae</i>	0.0004	0.0005	0.0006	0.0007
<i>A. aerogenes</i>	0.0004	0.0007	0.0008	0.0008
<i>A. cloacae</i>	0.0004	0.0007	0.0008	0.0008
<i>E. coli</i>	0.0002	0.0005	0.0004	0.0007
<i>E. communior</i>	0.0004	0.0007	0.0006	0.0008
<i>E. acidilactia</i>	0.0004	0.0006	0.0006	0.0008
<i>P. aeruginosa</i>	0.0002	*	0.0004	0.0005
<i>P. ovalis</i>	0.0002	0.0005	0.0004	0.0006
<i>P. graveolens</i>	0.0001	*	0.0002	0.0005
<i>P. syncyanea</i>	0.0002	*	0.0004	0.0005
<i>P. mucedolens</i>	0.0002	*	0.0004	0.0005
<i>S. marcescens</i>	0.0006	0.0007	0.0008	0.0010
<i>R. agilis</i>	0.0001	0.0005	0.0004	0.0006
<i>R. rosaceous</i>	0.0001	0.0005	0.0004	0.0006
<i>Torula rosea</i>	0.0020	0.0011	†	0.0080
<i>F. saueolens</i>	0.0006	0.0007	0.0008	0.0008
<i>A. lipolyticum</i>	0.0006	0.0007	0.0008	0.0010
<i>B. subtilis</i>	0.0006	0.0008	0.0008	0.0010
<i>B. mesentericus</i>	0.0004	0.0007	0.0006	0.0008
<i>B. mycoides</i>	0.0004	0.0007	0.0006	0.0008
<i>B. fusiformis</i>	0.0002	0.0005	0.0004	0.0007
<i>B. metiens</i>	0.0004	0.0008	0.0006	0.0010
<i>C. intermedium</i>	0.0004	0.0007	0.0006	0.0008
<i>A. faecalis</i>	0.0004	0.0007	0.0006	0.0008
<i>P. vulgaris</i> x 19.....	0.0004	0.0007	0.0006	0.0008
<i>S. aureus</i>	0.0004	0.0007	0.0008	0.0008
<i>S. candidus</i>	0.0002	0.0007	0.0006	0.0008
<i>S. albus</i>	0.0002	0.0005	0.0006	0.0007
<i>S. flava</i>	0.0002	0.0005	0.0004	0.0006
<i>Sarcina lutea</i>	0.0002	0.0005	0.0004	0.0007

* No growth in lowest concentration employed.

† Growth in highest concentration employed.

0.4 M depressed slightly or had no effect on the toxicity of cerium chloride and increased slightly or had no effect on the toxicity of cerium sulfate. Sodium sulfate in 0.1 M concentration markedly reduced the toxicity of cerium sulfate (table 3), but was without significant effect when used with cerium chloride.

Magnesium chloride (0.5 M) generally diminished the toxicity of cerium chloride, but magnesium sulfate (0.5 M) was without effect. The chlorides (0.5 M) of calcium and barium slightly increased the toxicity of cerium chloride, whereas barium sulfate, lithium chloride, ammonium sulfate, and ammonium chloride were without significant effect.

The toxicity of cerium chloride for fungi. In all the preceding experiments it was observed that the *Torula* culture was far more tolerant of the cerium compounds than were the bacteria. In order to determine whether other common fungi are equally tolerant of cerium, a yeast extract glucose peptone medium was prepared with concentrations of cerium chloride sufficient to inhibit all the bacteria employed in this study. Thirty-five strains of fungi were inoculated on the media by streaking, and the results were read after incubation for 2 days at room temperature.

The following organisms were employed: *Debaromyces tyrocola*, *Endomyces hordei*, *Monilia krusei*, *Mycoderma valida*, *Pichia farinosus*, *Saccharomyces cerevisiae* Froberg, *Saccharomyces* of Curtis, *Saccharomyces cerevisiae* Saaz, *Schizosaccharomyces mellacei*, *Torula* "Hansen" sp., *Torula humicola*, *Torula mucilaginoso*, *Torula spherica*, *Torula datilla*, *Torula colliculosa*, *Torula sanguinea*, *Torula* "pink" sp., *Torula fructicola*, *Torula liconde*, *Torula fermentati*, *Torula kefyri*, *Torula lactosa*, *Torula candida*, *Zygosaccharomyces priorianus*, *Zygosaccharomyces chevalieri*, *Vermicularia* sp., *Fusarium* sp., *Phytophthora* sp., *Neocosmospora* sp., *Dothiorella* sp., *Cunninghamella* sp., *Trichoderma* sp., *Pythium* sp., *Rhizopus* sp., and *Aspergillus* sp.

All the fungus cultures grew as well in the presence of 0.0014 M cerium chloride, the highest concentration employed, as in the control medium.

The bacteriostatic activity of lanthanum and thallium. The toxicity of lanthanum chloride and thallium nitrate for the selected bacteria was determined by the same methods employed in the preceding experiments. The results are presented in table 4. The order of toxicity of lanthanum and thallium was found to be approximately the same as that of cerium. Again, some species were observed to be relatively more resistant than others. The organisms most tolerant of the salts were found to be certain species of *Bacillus*, *Serratia marcescens*, *Sarcina conjunctivae*, *Achromobacter lipolyticum*, and *Torula rosea*. The most susceptible organisms were species of the genus *Pseudomonas*.

CONCLUSIONS

The salts of cerium, lanthanum, and thallium were found to be definitely more toxic for the bacteria than for the fungi included in this study.

The 39 species of bacteria were prevented from growth in concentrations of cerium chloride varying from 0.0006 to 0.0012 M; in cerium nitrate from 0.0004 to 0.0008 M; in cerium sulfate from 0.0006 to 0.0008 M; in lanthanum chloride from 0.0002 to 0.0008 M; and in thallium nitrate from 0.0005 to 0.0010 M.

The toxicity of cerium sulfate for most bacteria was reduced by the addition of sodium sulfate (0.05 M) to the medium, and the toxicity of cerium chloride was generally decreased by the addition of magnesium chloride (0.05 M). The addition of other salts had little effect on the toxicity of cerium.

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