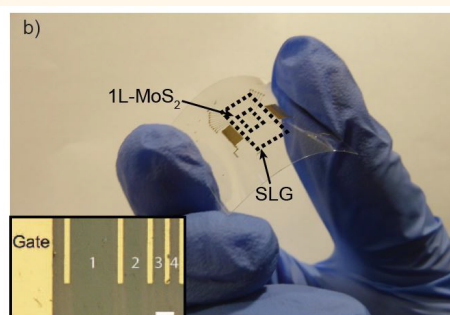
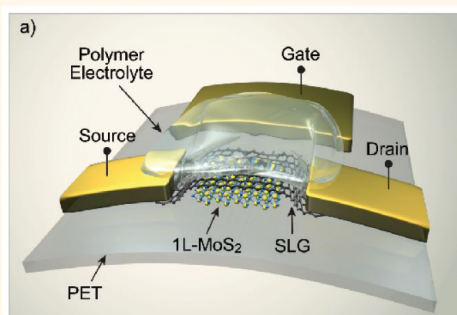


High Responsivity, Large-Area Graphene/MoS₂ Flexible Photodetectors

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ABSTRACT: We present flexible photodetectors (PDs) for visible wavelengths fabricated by stacking centimeter-scale chemical vapor deposited (CVD) single layer graphene (SLG) and single layer CVD MoS₂, both wet transferred onto a flexible polyethylene terephthalate substrate. The operation mechanism relies on injection of photoexcited electrons from MoS₂ to the SLG channel. The external responsivity is 45.5A/W and the internal 570A/W at 642 nm. This is at least 2 orders of magnitude higher than bulk-semiconductor flexible membranes. The photoconductive gain is up to 4×10^5 . The photocurrent is in the 0.1–100 μ A range. The devices are semitransparent, with 8% absorptance at 642 nm, and are stable upon bending to a curvature of 1.4 cm. These capabilities and the low-voltage operation (<1 V) make them attractive for wearable applications.

KEYWORDS: photodetectors, graphene, MoS₂, heterostructures, flexible optoelectronics

Modern electronic and optoelectronic systems, such as smart phones, smart glasses, smart watches, wearable devices, and electronic tattoos, increasingly require ultrathin, transparent, low-cost, and energy efficient devices on flexible substrates.¹ The rising demand for flexible electronics and optoelectronics requires materials that can provide a variety of electrical and optical functionalities, with constant performance upon application of strain.² A wide range of optoelectronic devices on flexible substrates have been reported to date, such as photodetectors (PDs),^{3,4} light emitting diodes (LEDs),⁵ optical filters,⁶ optical interconnects,^{7,8} photovoltaic devices,^{9,10} and biomedical sensors.^{11,12}

Major challenges in the development of flexible optoelectronic devices stem from the limitations associated with the high stiffness of bulk semiconductors.^{13,14} In the case of flexible PDs, the current approaches primarily rely on thin (μ m-thick) semiconductor membranes^{4,15} and compound semiconductor nanowires (NWs),^{3,16–18} mainly because of their ability to absorb light throughout the whole visible range (0.4–0.7 μ m) and the possibility to adapt their fabrication techniques from rigid to plastic, or deformable substrates.¹

One of the key parameters for PDs characterization is the responsivity. This is defined as the ratio between the collected photocurrent (I_{ph}) and the optical power. The responsivity is named external ($R_{ext} = I_{ph}/P_o$)¹⁹ or internal ($R_{int} = I_{ph}/P_{abs}$),¹⁹ whenever the incident (P_o) or absorbed (P_{abs}) optical power is used in the denominator. Since not all incident photons are absorbed by a PD, *i.e.*, $P_{abs} < P_o$, then R_{int} is typically larger than R_{ext} .¹⁹

In flexible PDs, R_{ext} up to ~ 0.3 A/W was reported for crystalline semiconductor membranes (InP, Ge)^{4,15} with integrated p–i–n junctions, showing photocurrent up to ~ 100 μ A, with $\sim 30\%$ degradation upon bending at a radius $r_b \sim 3$ cm.¹⁵ PDs made of a single semiconductor NW on flexible substrates^{3,16–18} demonstrated R_{ext} up to $\sim 10^5$ A/W, for r_b down to 0.3 cm.³ Yet, these provide limited I_{ph} in the order of nA,^{3,16,18} up to <1 μ A.¹⁷ For flexible devices exploiting NW arrays by drop-casting,^{3,16,18} rather than based on single NWs,

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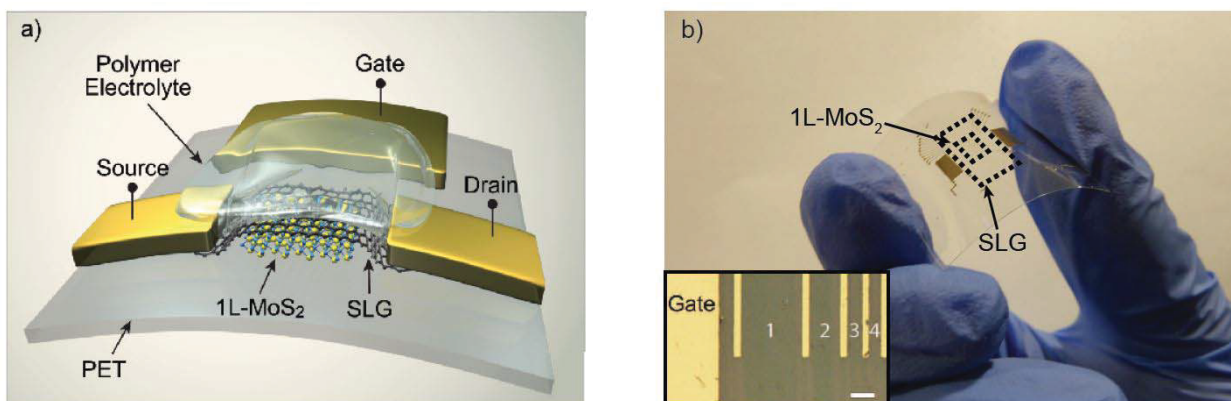


Figure 1. (a) Schematic SLG/MoS₂ flexible PD, side-gated with a PE. (b) Picture of a typical PD, showing transparency and flexibility. (Inset) Optical image of 4 PDs with different channel lengths and common side gate electrode. Scale bar is 200 μm .

R_{ext} degrades significantly from $\sim 10^5$ A/W to the mA/W range,^{3,16,18} due to photocurrent loss at multiple junctions in the NW network.^{3,16,18}

Graphene and related materials (GRMs) have great potential in photonics and optoelectronics.^{20–23} A variety of GRM-based devices have been reported, such as flexible displays,²⁴ photovoltaic modules,^{25,26} photodetectors,^{22,27–29} optical modulators,³⁰ plasmonic devices,^{31–35} and ultrafast lasers.²³ Heterostructures, obtained by stacking layers of different materials, were also explored,^{21,22} e.g., in photovoltaic³⁶ and light emitting devices.³⁷

Flexible PDs based on GRMs were studied for ultraviolet,^{38,39} visible,^{40–45} and near-infrared bands.^{46,47} In these devices, different materials and heterostructures produced by mechanical exfoliation,^{40,41} Chemical Vapor Deposition (CVD),^{42,43,46} and liquid-phase exfoliation (LPE)^{44,45,47} were employed. The flexible PDs produced by mechanical exfoliation^{40,41} have a small ($\sim \mu\text{m}^2$) photoactive area, and they cannot be scaled up to mass production. LPE-based PDs have low (< mA/W)^{44,45} responsivity. Ref 47 showed that thick ($\sim \mu\text{m}$) films of chemically modified, charge-transfer optimized, LPE-produced, MoSe₂ and MoS₂ polymer composites can provide \sim A/W responsivity⁴⁷ at near-infrared bands. Nevertheless, these PDs require high (~ 10 V) operation voltage and are nontransparent. Flexible PDs at 450 nm using CVD MoS₂ transistors⁴² and MoS₂/WS₂ heterostructures⁴³ were previously reported, and PDs at 780 nm were prepared from doped SLG p–n junctions.³⁸ However, these devices have responsivity in the mA/W range. CVD-based SLG/MoS₂ heterostructures⁴⁸ showed good photodetection on rigid Si/SiO₂ substrates, with back-gate-dependent $R_{\text{int}} \sim 10^8$ A/W for optical intensities < 0.1 pW/ μm^2 .

Here we demonstrate a polymer electrolyte (PE) gated, CVD-based, flexible PD, for visible wavelengths, with large ($\sim \text{mm}^2$) photoactive area combined with high responsivity (\sim hundreds A/W), high (> 80%) transparency, gate tunability, low (< 1 V) operation voltage, and stable ($\pm 12\%$) I_{ph} upon multiple (> 30) bending cycles. The device is assembled by stacking on a PET substrate a centimeter-scale CVD single layer graphene (SLG) on top of a CVD-grown single layer MoS₂ (1L-MoS₂). In this configuration, 1L-MoS₂ acts as visible light absorber, while SLG is the conductive channel for photocurrent flow.⁴⁸ We show that R_{ext} can be increased by promoting carrier injection from 1L-MoS₂ to SLG using PE gating, or by increasing the source-drain voltage. This is

achieved in devices with $\sim 82\%$ transparency, twice that reported for semiconductor membrane devices.¹⁵ We get $R_{\text{int}} \sim 570$ A/W for ~ 0.1 nW/ μm^2 at 642 nm, similar to SLG/MoS₂ PDs⁴⁸ on rigid substrates operating at the same optical power. This shows that SLG/MoS₂ heterostructures on PET retain their photodetection capabilities. We note that the devices from ref 48 have at least 2 orders of magnitude smaller photoactive area with respect to ours, and they are not flexible, not transparent, and require tens of volts operation, unlike the < 1 V of ours. Upon bending, our PDs have stable performance for r_b down to ~ 1.4 cm. This is comparable to r_b measured in semiconductor membranes PDs,^{4,15} which show lower (< 0.3 A/W) responsivities.^{4,15} Although our r_b is 1 order of magnitude larger than for flexible single NWs,^{3,16–18} the latter had at least 3 orders of magnitude smaller device areas (< 5 μm^2)^{3,16–18} compared to our PDs (> 0.2 mm²). Given the responsivity, flexibility, transparency, and low operation voltage, our PDs may be integrated in wearable, biomedical, and low-power optoelectronic applications.^{11,12,17}

RESULTS

Figure 1a plots a schematic drawing of our PDs. We fabricated 4 PD arrays with 10 devices each, with channel lengths of 100 μm , 200 μm , 500 μm , and 1 mm. Each device consists of a 1L-MoS₂ absorber covered by a SLG channel, clamped between source and drain electrodes. We chose PET as a flexible substrate due to its $\sim 90\%$ transparency in the visible range⁴⁹ and ability to withstand solvents (e.g., acetone and isopropyl alcohol)⁵⁰ commonly used in the transfer processes of layered materials grown by CVD (e.g., transfer of SLG grown on Cu).⁵¹ A 1L-MoS₂ is used as absorber in order to preserve > 80% transparency, considered suitable by industry for wearable applications,⁵² Figure 1b. The SLG/1L-MoS₂ heterostructure is gated using a PE.^{53,54}

The operation principle of our devices is depicted in Figure 2. For energy bands alignment, the electron affinity of 1L-MoS₂ and the Dirac point of SLG are assumed to be ~ 4 – 4.2 eV^{55,56} and ~ 4.6 eV,^{57,58} respectively. We also assume SLG to be initially p-doped (Figure 2a), as reported in previous works involving SLG transferred on PET.^{59,60} At zero voltage the device is in thermodynamic equilibrium with a constant Fermi level (E_{F}) and zero current flow between the layers. During illumination and photon absorption in MoS₂, part of the photogenerated electrons would be injected from the 1L-MoS₂ conduction band into the lower energy states in p-doped

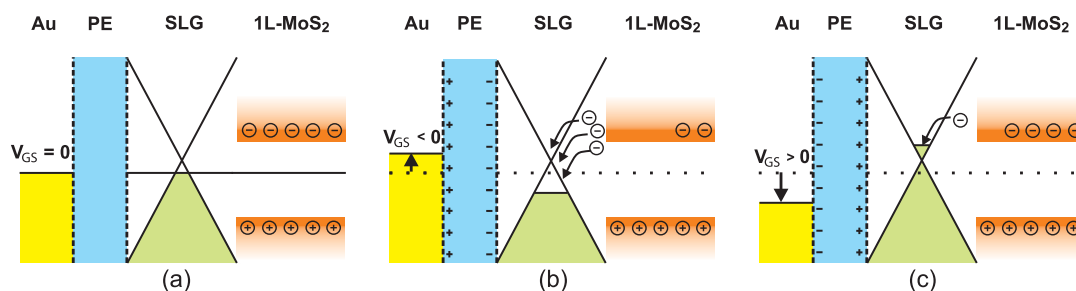


Figure 2. Schematic band diagram of PE gated SLG/1L-MoS₂ PD at (a) zero, (b) negative, and (c) positive V_{GS} .

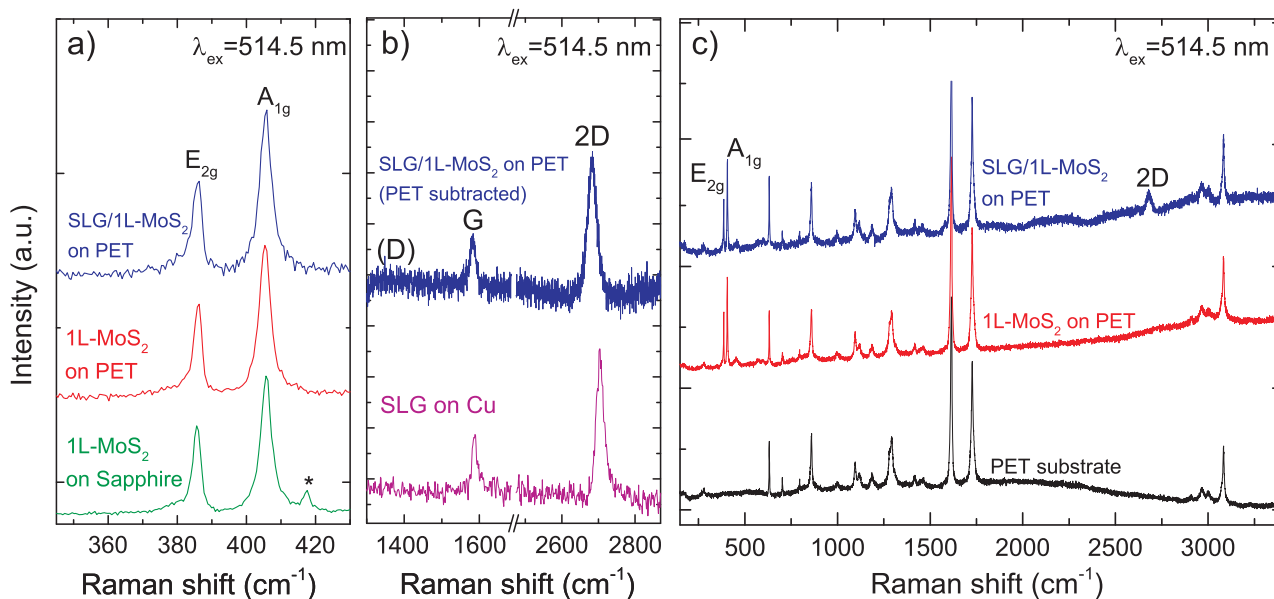


Figure 3. (a) Raman spectra at 514 nm for 1L-MoS₂ on sapphire, 1L-MoS₂ on PET, and SLG/1L-MoS₂. (b) Comparison at 514 nm of the Raman spectra of as-grown SLG on Cu (magenta curve) and SLG/1L-MoS₂ after transfer on PET and normalized subtraction of the PET substrate signal (blue curve). (c) Raman spectra at 514 nm of PET substrate (black curve), 1L-MoS₂ on PET (red curve) and SLG/1L-MoS₂ on PET (blue curve).

SLG,⁴⁸ leaving behind the uncompensated charge of photo-generated holes. The latter would be trapped in 1L-MoS₂ and act as an additional positive gate voltage, V_{GS} , applied to the SLG channel, resulting in a shift of the charge neutrality point (V_{CNP}) to more negative voltages. The injected electrons from 1L-MoS₂ would occupy energy states above E_F (Figure 2b), thus reducing the hole concentration and decreasing the hole current in the SLG channel. Electron injection can be further promoted by gating. When a negative V_{GS} is applied, higher p-doping of the SLG channel would induce a stronger electric field at the SLG/1L-MoS₂ interface,⁴⁸ thus favoring electron transfer from 1L-MoS₂ (Figure 2b). Hence, for negative V_{GS} , R_{ext} is expected to increase, due to injection of more photoelectrons to SLG and consequent more pronounced current reduction. The opposite should happen for positive V_{GS} , where the gate-induced negative charge in SLG would reduce the p-doping and shift E_F toward the Dirac point. In this case, the photogenerated electrons in 1L-MoS₂ would experience weaker electric fields at the SLG/1L-MoS₂ interface⁴⁸ and would become less attracted by the SLG channel. Thus, we expect R_{ext} to decrease. For high enough positive V_{GS} , E_F would cross the Dirac point, and SLG becomes n-doped (Figure 2c). As a result, only a weak electron injection from 1L-MoS₂ would be possible, if E_F in SLG remains below the 1L-MoS₂ conduction band, retaining a weak electric field at the

interface. In this regime, the transferred electrons would increase the free carrier concentration in the n-doped channel, hence only minor increments of R_{ext} and I_{ph} are expected.

Our devices are built as follows: 1L-MoS₂ is epitaxially grown by CVD on c-plane sapphire substrates,⁶¹ while SLG is grown on a 35 μm Cu foil, following the process described in refs 51 and 62 (see Methods for details). Prior to assembling the SLG/MoS₂ stack, the quality and uniformity of MoS₂ on sapphire and SLG on Cu are inspected by Raman spectroscopy and photoluminescence (PL), using a Horiba Jobin Yvon HR800 spectrometer equipped with a 100 \times objective. The laser power is kept below 100 μW (spot size <1 μm) to avoid possible heating effects or damage. Figure 3a (green curve) plots the Raman spectrum of CVD MoS₂ on sapphire for 514 nm excitation. The peak at $\sim 385\text{ cm}^{-1}$ corresponds to the in-plane (E_{2g}^1) mode,^{63,64} while that at $\sim 404\text{ cm}^{-1}$ is the out of plane (A_{1g}) mode,^{63,64} with full width at half-maximum FWHM (E_{2g}^1) = 2.5 and FWHM(A_{1g}) = 3.6 cm^{-1} , respectively. The E_{2g}^1 mode softens, whereas the A_{1g} stiffens with increasing layer thickness,^{65,66} so that their frequency difference can be used to monitor the number of layers.⁶⁵ The peak position difference $\sim 20\text{ cm}^{-1}$ is an indicator of 1L-MoS₂.⁶⁵ The peak at $\sim 417\text{ cm}^{-1}$ (marked by an asterisk in Figure 3a) corresponds to the A_{1g} mode of sapphire.⁶⁷

The Raman spectrum measured at 514 nm of SLG on Cu is shown in Figure 3b (magenta curve). This is obtained after the removal of the background PL of Cu.⁶⁸ The two most intense features are the G and the 2D peak, with no significant D peak. The G peak corresponds to the E_{2g} phonon at the Brillouin zone center.⁶⁹ The D peak is due to the breathing modes of sp^2 rings and requires a defect for its activation by double resonance.^{69–72} The 2D peak is the second order of the D peak.⁶⁹ This is always seen, even when no D peak is present, since no defects are required for the activation of two phonons with the same momentum, one backscattering from the other.⁶⁹ In our sample, the 2D peak is a single sharp Lorentzian with $\text{FWHM}(2D) \sim 26 \text{ cm}^{-1}$, a signature of SLG.⁷⁰ Different (~ 20) measurements show similar spectra, indicating uniform quality throughout the sample. The position of the G peak, $\text{Pos}(G)$, is $\sim 1588 \text{ cm}^{-1}$, with $\text{FWHM}(G) \sim 6 \text{ cm}^{-1}$. The 2D peak position, $\text{Pos}(2D)$ is $\sim 2705 \text{ cm}^{-1}$, while the 2D to G peak intensity and area ratios, $I(2D)/I(G)$ and $A(2D)/A(G)$, are ~ 2.6 and ~ 5.8 , respectively, indicating a p-doping $\sim 300 \text{ meV}$,^{53,73,74} which corresponds to a carrier concentration $\sim 6 \times 10^{12} \text{ cm}^{-2}$.

Another evidence for 1L-MoS₂ comes from the PL spectrum [Figure 4a (green curve)], showing a peak at $\sim 658 \text{ nm}$ ($\sim 1.88 \text{ eV}$), due to band-to-band radiative recombination in 1L-MoS₂.⁷⁵

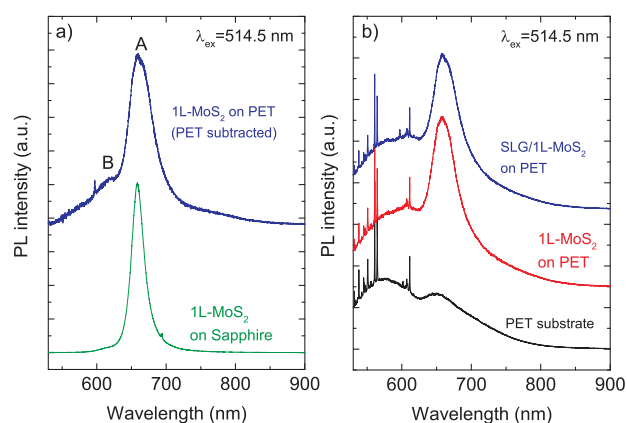


Figure 4. (a) PL spectrum at 514 nm (2.41 eV) of 1L-MoS₂ on sapphire and 1L-MoS₂ after transfer on PET. (b) PL spectra of PET substrate (black curve), 1L-MoS₂ on PET (red curve), and SLG/1L-MoS₂ on PET (blue curve).

Then, the MoS₂ film is transferred onto a PET substrate from sapphire using a KOH-based approach.⁶¹ The samples are first spin coated with $\sim 100 \text{ nm}$ poly(methyl methacrylate) (PMMA). This is detached in a 30% KOH solution, washed in deionized (DI) water, and transferred onto PET. The PMMA is then dissolved in acetone. Subsequently, SLG is transferred on the 1L-MoS₂ on PET. PMMA is spin coated on the SLG/Cu substrate and then placed in a solution of ammonium persulfate (APS) in DI water until Cu is etched.^{51,76} The PMMA membrane with attached SLG is then transferred to a beaker filled with DI water for cleaning APS residuals. The membrane is subsequently lifted with the target PET substrate, having 1L-MoS₂ on top. After drying, PMMA is removed in acetone leaving SLG on 1L-MoS₂.

Raman and PL characterizations are performed at each step of the SLG/1L-MoS₂ assembly on PET, *i.e.*, on 1L-MoS₂ transferred on PET and on SLG on 1L-MoS₂. This is to confirm no degradation during the fabrication process. Figure 3a (red curve) plots the Raman spectrum of 1L-MoS₂ on PET. The frequency difference between E_{2g}^1 and A_{1g} and the FWHMs are preserved on PET, suggesting no degradation. The PL spectrum of 1L-MoS₂ on PET is shown in Figure 4b (red curve). The signal from 1L-MoS₂ is convolved within the background due to the PET substrate [Figure 4b (black curve)]. In order to reveal the underlying PL signature of 1L-MoS₂, we use a point-to-point subtraction between the spectrum of 1L-MoS₂ on PET [Figure 4b (red curve)] and the reference PET spectrum [Figure 4b (black curve)]. Prior to subtraction, the spectra are normalized to the intensity of the Raman peak at $\sim 1615 \text{ cm}^{-1}$ (corresponding to the peak at $\sim 560 \text{ nm}$ in Figure 4b), due to the stretching vibrations of benzene rings in PET.⁷⁷ As a result, the PL signal of 1L-MoS₂ can be seen in Figure 4a (blue curve) revealing no significant changes after transfer. The subsequent transfer of SLG on 1L-MoS₂ does not alter the 1L-MoS₂ PL position and line shape [Figure 4b (blue curve)].

We then characterize the SLG transferred on 1L-MoS₂/PET. The intense Raman features of the underlying PET substrate⁷⁷ [Figure 3c (black curve)] mask the SLG peaks. In order to reveal the Raman signatures of SLG, we first measure the reference spectrum, shown in Figure 3c (black curve), of a PET substrate, using identical conditions as those for SLG/1L-MoS₂/PET. We then implement a point-to-point subtraction, normalized to the intensity of the PET peak at $\sim 1615 \text{ cm}^{-1}$, of the PET reference spectrum, Figure 3c (black curve), from the total spectrum, Figure 3c (blue curve). The result is in Figure

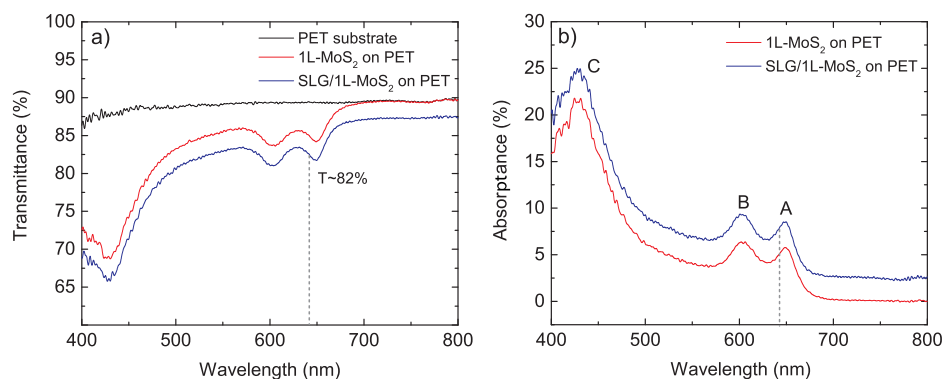


Figure 5. (a) Transmittance of PET (black curve), 1L-MoS₂ on PET (red curve), and SLG/1L-MoS₂ on PET (blue curve). (b) Absorbance of 1L-MoS₂ and SLG/1L-MoS₂ as derived from the transmittance measurements. Dashed lines indicate our test wavelength.

3b (blue curve). The 2D peak retains its single-Lorentzian line-shape with $\text{FWHM}(2\text{D}) \sim 28 \text{ cm}^{-1}$, validating the transfer of SLG. The negligible D peak indicates that no significant defects are induced during transfer. $\text{Pos}(G) \sim 1583 \text{ cm}^{-1}$, $\text{FWHM}(G) \sim 17 \text{ cm}^{-1}$, $\text{Pos}(2\text{D}) \sim 2683 \text{ cm}^{-1}$, and $A(2\text{D})/A(G) \sim 4.8$, indicating a p-doping $\sim 4 \times 10^{12} \text{ cm}^{-2}$ ($\sim 250 \text{ meV}$).^{53,73}

We then measure the absorptance and transmittance of SLG/1L-MoS₂ using a broadband (400–1300 nm) white light from a tungsten halogen lamp. The transmitted light is collected by a 10× objective lens (NA = 0.25) with a Horiba Jobin Yvon HR800 spectrometer equipped with a 300 grooves/mm grating, charged coupled device (CCD) detector and a 50 μm pinhole. Figure 5a plots the optical transmittance of bare PET (T_{PET} , black line), 1L-MoS₂ on PET (T_{MoS_2} , red line), and the final SLG/1L-MoS₂ stack on PET (T_{Hetero} , blue line) measured in the 400–800 nm wavelength range. Figure 5b plots the absorptance of 1L-MoS₂ on PET ($\text{Abs}_{\text{MoS}_2}$, red line) and of SLG/1L-MoS₂ on PET ($\text{Abs}_{\text{Hetero}}$, blue line), calculated as $\text{Abs}_{\text{MoS}_2} = (T_{\text{PET}} - T_{\text{MoS}_2})/T_{\text{PET}}$ and $\text{Abs}_{\text{Hetero}} = (T_{\text{PET}} - T_{\text{Hetero}})/T_{\text{PET}}$. The three peaks in Figure 5b at $\sim 650 \text{ nm}$ (1.91 eV), $\sim 603 \text{ nm}$ (2.06 eV), and $\sim 428 \text{ nm}$ (2.90 eV) correspond to the A, B, C excitons of 1L-MoS₂.^{75,78} Their positions remain unchanged after SLG transfer. The absorptance difference between the two curves (red and blue) is $\sim 2.6\%$, consistent with the additional SLG absorptance.⁷⁹

The PD area is shaped by etching, whereby SLG extending beyond the 1L-MoS₂ flake is removed in an oxygen plasma. The source, drain and gate electrodes are then defined by patterning the contacts area, followed by Cr/Au (6 nm/60 nm) evaporation and lift-off. PDs with different channel lengths (100 μm to 1 mm), 2 mm channel width, and common side-gate electrodes (1 × 0.5 cm) are built (Figure 1b).

Ref 48 showed that the responsivity of SLG/MoS₂ PDs can be enhanced by gating. This induces a stronger electric field at the SLG/MoS₂ interface and promotes charge transfer. Various gating techniques have been exploited for GRM-based devices, including conventional Si/SiO₂ back-gates,⁸⁰ high-k dielectrics (Al₂O₃, HfO₂),⁸¹ chemical dopants,⁸² ionic liquids,⁸³ and PEs.^{53,74} In order to gate our SLG/1L-MoS₂ on PET, we employ the latter due to its compatibility with flexible substrates⁸⁴ and the ability to substantially dope SLG ($\pm 0.8 \text{ eV}$)^{53,74} using small gate voltages (up to 4 V), unlike other gating techniques, which would require considerably higher biases to reach the same doping.^{80,82} We use a PE consisting of LiClO₄ and poly(ethylene oxide) (POE).^{53,74} We place the PE over both the SLG channel and the side-gate electrode. To evaluate the effect of PE deposition on the SLG channel doping, we use Raman analysis. We get $\text{Pos}(G) \sim 1583 \text{ cm}^{-1}$, $\text{FWHM}(G) \sim 19 \text{ cm}^{-1}$, $\text{Pos}(2\text{D}) \sim 2686 \text{ cm}^{-1}$, and $A(2\text{D})/A(G) \sim 5.3$, consistent with a small reduction of p-doping to $\sim 230 \text{ meV}$.^{53,73} For electrical measurements we apply $-1 \text{ V} < V_{\text{GS}} < 1 \text{ V}$ in order to avoid electrochemical reactions, such as hydrolysis of residual water in the electrolyte.^{85,86} These may permanently modify the graphene electrode^{85,86} and compromise the stability and performance of the device. To control the stability of the PE gating, we continuously monitor the gate leakage current (I_{gate}), keeping $I_{\text{gate}} < 1 \text{ nA}$ throughout the experiments. The devices are tested ~ 30 times, showing no degradation in the leakage current over at least six months.

We then characterize the responsivity at 642 nm ($\sim 1.93 \text{ eV}$), slightly above the A exciton peak, where absorption of 1L-MoS₂ is maximized (Figure 5b). At 642 nm the SLG/1L-MoS₂

heterostructure shows $\sim 8\%$ absorptance (Figure 5b), and the device retains $\sim 82\%$ transparency (Figure 5a).

The $I_{\text{DS}}-V_{\text{GS}}$ measurements in Figure 6a are done at room temperature using a probe station and a parameter analyzer

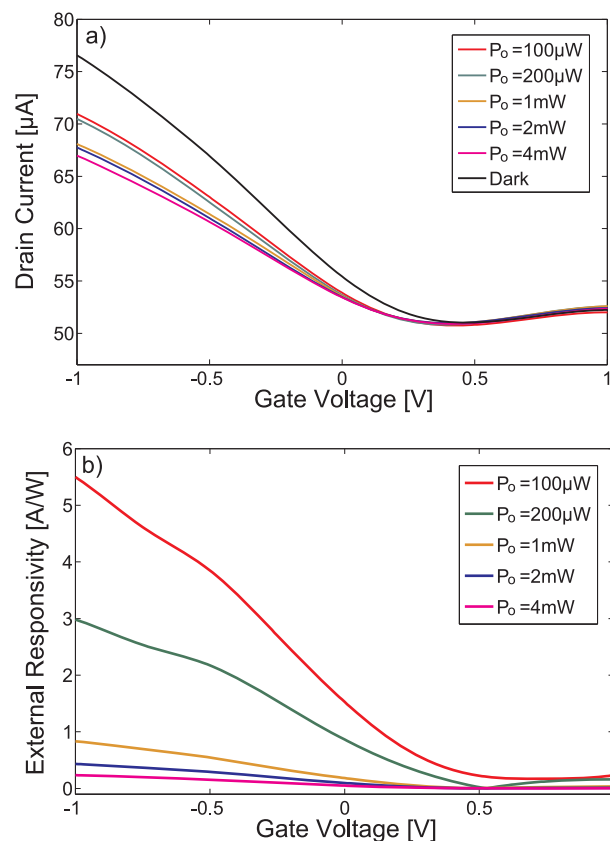


Figure 6. (a) Transfer characteristics as a function of P_o . (b) R_{ext} as a function of V_{GS} and P_o . Channel length and width are 100 μm and 2 mm, respectively.

(Keithley 4200). The PD is illuminated at normal incidence by a collimated laser with P_o ranging from 100 μW to 4 mW. At these P_o and with $V_{\text{DS}} = 0.1 \text{ V}$ we measure a positive V_{CNP} ranging from ~ 0.39 to 0.47 V , indicating an initial SLG p-doping $\sim 220 \text{ meV}$, consistent with the Raman estimate.

Figure 6a shows that for $-1 \text{ V} < V_{\text{GS}} < 0.5 \text{ V}$, where SLG transport is hole dominated, the current decreases under illumination ($\sim 10 \mu\text{A}$ at $V_{\text{GS}} = -1 \text{ V}$), as anticipated from the band-diagram of Figure 2. For $V_{\text{GS}} > 0.5 \text{ V}$, where SLG is electron-doped, the PD shows a small (up to $\sim 0.2 \mu\text{A}$) current increase under illumination. Figure 6b plots R_{ext} as a function of V_{GS} , as derived from transconductance measurements using:¹⁹

$$R_{\text{ext}} = \frac{|I_{\text{light}} - I_{\text{dark}}|}{P_o \cdot A_{\text{PD}}/A_o} \quad (1)$$

where I_{light} and I_{dark} are the PD current under illumination and in dark, $|I_{\text{light}} - I_{\text{dark}}| = I_{\text{ph}}$ is the photocurrent defined as the absolute change in the device current upon illumination, A_o is the laser spot area, A_{PD} is the PD area, and A_{PD}/A_o is a scaling factor that takes into account the fact that only a fraction of optical power impinges on the PD. As expected from the band-diagram in Figure 2, R_{ext} tends to increase for more negative V_{GS} , up to $\sim 5.5 \text{ A/W}$ at $V_{\text{GS}} = -1 \text{ V}$, $V_{\text{DS}} = 0.1 \text{ V}$ for $P_o = 100 \mu\text{W}$. By taking into account that only 8% of light is absorbed

($P_{\text{abs}} = 0.08 \times P_o$), we derive $R_{\text{int}} = R_{\text{ext}}/0.08 = 69\text{A/W}$. Figure 6b implies that the higher P_o , the lower R_{ext} . This can be explained considering that the more photogenerated electrons are injected into the p-doped channel, the lower the electric field at the SLG/1L-MoS₂ interface, therefore a reduced injection of electrons causes R_{ext} to decrease.

Given that $R_{\text{ext}}R_{\text{int}} > 1\text{A/W}$, we expect a photoconductive gain (G_{PD}),^{19,87} whereby absorption of one photon results in multiple charge carriers contributing to I_{ph} . Our PDs act as optically gated photoconductors, where the SLG conductance is modulated by optical absorption in the 1L-MoS₂. In this configuration, the presence of G_{PD} implies that the injected electrons in SLG can recirculate multiple times between source and drain, before recombining with trapped holes in 1L-MoS₂. Consequently, G_{PD} can be estimated as the ratio of electrons recombination (τ_{rec}) and transit (t_{tr}) times in the SLG channel: $G_{\text{PD}} = \tau_{\text{rec}}/t_{\text{tr}}$.^{19,21,22,87} For higher V_{DS} , the free carriers drift velocity v_d in the SLG channel increases linearly with bias (ohmic region) until it saturates, because of carriers scattering with optical phonons.⁸⁸ The linear increase in v_d results in shorter t_{tr} , with $t_{\text{tr}} = L/v_d$, where L is the channel length.^{19,21,22,87} Therefore, G_{PD} is also expected to grow linearly with V_{DS} , providing higher R_{ext} . To confirm the photoconductive nature of G_{PD} in our devices and test the dependence of R_{ext} on V_{DS} , we measure $I_{\text{DS}}-V_{\text{DS}}$ under illumination at $P_o = 100 \mu\text{W}$ for $V_{\text{GS}} = -1\text{V}$ and calculate R_{ext} using eq 1. The $I_{\text{DS}}-V_{\text{DS}}$ characteristics of the PD show linear dependence, confirming ohmic behavior of the metal-SLG-metal channel.⁸⁹ We use $V_{\text{DS}} < 1\text{V}$ to keep the device operation in the linear (ohmic) regime and minimize the effects of the nonlinear dependence of v_d on V_{DS} (such as velocity saturation) that might appear for $V_{\text{DS}} > 1\text{V}$.⁸⁸ As shown in Figure 7, R_{ext} scales with V_{DS} and reaches $\sim 45.5\text{A/W}$ ($R_{\text{int}} \sim 570\text{A/W}$) at $V_{\text{DS}} = 1\text{V}$. This is almost 1 order of magnitude higher than at $V_{\text{DS}} = 0.1\text{V}$, consistent with the similar increase in V_{DS} . These results are at least 2 orders of magnitude higher than semiconductor flexible membranes.^{4,15} Furthermore, such a combination of high responsivity with μA range photocurrent surpasses that found in other GRM-based PDs in the visible range.^{40-45,47} We also fabricate a control device with a 1L-MoS₂ channel only, without SLG. This has $R_{\text{ext}} \sim 2\text{mA/W}$, which is 4 orders of magnitude smaller than that of our SLG/1L-MoS₂ heterostructure. We thus conclude that SLG/1L-MoS₂ heterostructures are necessary to achieve high (hundreds A/W) responsivity, due to the presence of photoconductive gain.

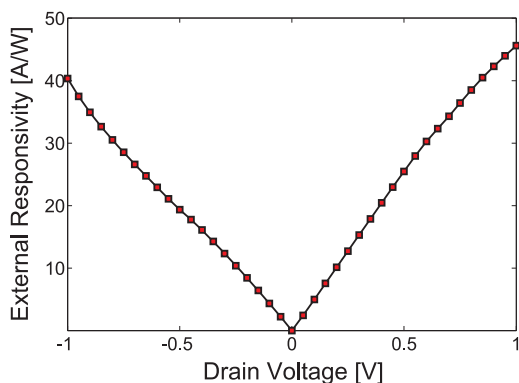


Figure 7. R_{ext} as a function of V_{DS} for $P_o = 100 \mu\text{W}$ at $V_{\text{GS}} = -1\text{V}$.

To assess the photoresponse uniformity in our SLG/1L-MoS₂ heterostructures, we perform photocurrent mapping using the same laser source (642 nm) as for optoelectronic characterizations. We scan areas of $80 \times 140 \mu\text{m}$ (pixel size $3 \times 3 \mu\text{m}$) at different locations. At each position (pixel) the device photocurrent is measured for $V_{\text{DS}} = 0.3\text{V}$ (Figure 8a). We also collect the backscattered light to give a reflection map (Figure 8b). Figure 8a indicates that the entire channel area confined between the source-drain electrodes is photoactive, and shows uniform photocurrent photoresponse with standard deviation $\pm 15\%$. We thus conclude that interface imperfections (e.g., bubbles, polymer residuals, etc.) have minor effect on the charge transfer process from MoS₂ to graphene.

We define G_{PD} as the ratio between electrons recirculating in the SLG channel, thus sustaining I_{ph} , and the initial electron concentration injected into SLG from 1L-MoS₂:⁴⁸

$$G_{\text{PD}} = \frac{|I_{\text{light}} - I_{\text{dark}}|}{q \cdot A_{\text{PD}} \cdot \Delta n_{\text{ch}}} \quad (2)$$

where q is the electron charge and Δn_{ch} is the concentration per unit area and per unit time of the injected electrons. Δn_{ch} is equal to the trapped-hole concentration per unit area and per unit time in 1L-MoS₂, which is related to a charge neutrality point shift $\Delta V_{\text{GS}} = \Delta V_{\text{CNP}}$ in the transfer characteristics. To calculate Δn_{ch} , we first write the potential balance in the metal-dielectric-SLG structure. When V_{GS} is applied, it creates a gate-to-channel potential drop (V_{diel}), and it induces a local electrostatic potential in the graphene channel ($V_{\text{ch}} = E_{\text{F}}/q$):^{19,53}

$$V_{\text{GS}} = V_{\text{diel}} + V_{\text{ch}} = \frac{Q_{\text{G}}}{C_{\text{G}}} + V_{\text{ch}} \quad (3)$$

where Q_{G} and C_{G} are the charge concentration and the geometrical capacitance per unit area associated with the gate electrode, respectively. $|Q_{\text{G}}| = |q \cdot n_{\text{ch}}|$, reflecting the charge neutrality of the gate capacitor, with n_{ch} the charge carrier concentration per unit area in the channel induced by V_{GS} . Any variations of n_{ch} change Q_{G} and V_{GS} . From eq 3 we get:

$$\frac{dV_{\text{GS}}}{dQ_{\text{G}}} = \frac{1}{C_{\text{G}}} + \frac{dV_{\text{ch}}}{dQ_{\text{G}}} \quad (4)$$

which leads to

$$\Delta Q_{\text{G}} = (1/C_{\text{G}} + 1/C_{\text{Q}})^{-1} \cdot \Delta V_{\text{GS}} \quad (5)$$

where $C_{\text{Q}} = dQ_{\text{G}}/dV_{\text{ch}}$ is the SLG quantum capacitance^{90,91} that characterizes the changes of the channel potential ΔV_{ch} as a result of additional gating ΔQ_{G} , and $(1/C_{\text{G}} + 1/C_{\text{Q}})^{-1}$ is the total capacitance C_{tot} .

To calculate Q_{G} and Δn_{ch} , we first need to find C_{G} and C_{Q} . In PE gating, C_{G} is associated with the electric double layer (EDL) at the SLG/electrolyte interface.^{53,90,92,93} The EDL acts like a parallel-plate capacitor with a dielectric layer thickness of the order of the Debye length λ_{D} , so that $C_{\text{G}} = C_{\text{EDL}} = \epsilon \epsilon_0 / \lambda_{\text{D}}$, where ϵ is the PE dielectric constant, and ϵ_0 is the vacuum permittivity. In principle, for a monovalent electrolyte, λ_{D} can be explicitly calculated⁹⁴ if the electrolyte concentration is known. However, in the presence of a polymer matrix, the electrolyte ions can form complexes with polymer chains,⁹⁵ therefore the precise ion concentration is difficult to measure. For PE gating, different EDL thicknesses in the range $\sim 1-5\text{nm}$ have been reported.^{53,54,92,93} To estimate C_{EDL} in our devices,

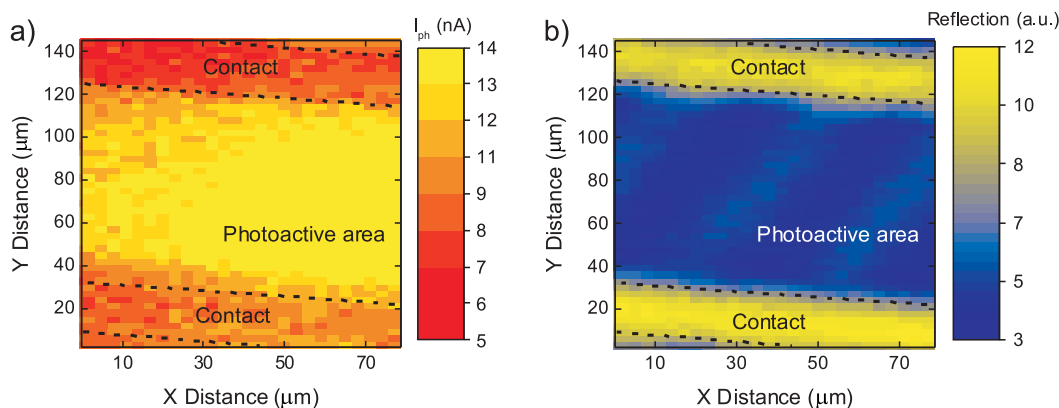


Figure 8. (a) Photocurrent map of channel area, simultaneously measured with backscattered light map. A uniform signal is observed in the channel area (between the electrodes). (b) Reflection map of backscattered light from the device channel. The yellow areas, corresponding to the contact areas, show higher reflectance than the substrate (in blue).

we take $\lambda_D \sim 2 \text{ nm}$ ⁵³ and the dielectric constant of the poly(ethylene oxide) matrix to be $\epsilon \sim 5$,⁹⁶ as done in ref 53. As a result, we obtain $C_{EDL} = 2.2 \times 10^{-6} \text{ F/cm}^2$. This is the same order of magnitude as the SLG C_Q .⁹⁰ Therefore, the latter cannot be neglected in eq 5. C_Q is given by⁹⁰

$$C_Q \approx \frac{2q^2}{\hbar v_F \sqrt{\pi}} \cdot \sqrt{n_{ch} + n_i} \quad (6)$$

where \hbar is the reduced Planck constant, $v_F = 1.1 \times 10^6 \text{ m/s}$ is the SLG Fermi velocity,^{80,97} and n_i is the intrinsic carrier concentration in SLG near the Dirac point induced by charge impurities, defects and local potential fluctuations in the SLG channel.^{90,98–100} From our Raman and transconductance measurements we estimate $n_i \sim 3 \times 10^{12} \text{ cm}^{-2}$. From eq 6 we then get $C_Q = 4 \times 10^{-6} \text{ F/cm}^2$ at V_{CNP} . From Figure 6a, and extracting ΔV_{CNP} between the dark current and the transfer curves measured under illumination, and with eq 5, we get Δn_{ch} ranging from 4 to $8 \times 10^{11} \text{ cm}^{-2}$ for P_o going from 100 μW to 4 mW. As a result, we obtain $G_{PD} \sim 5 \times 10^4$ at $V_{DS} = 0.1 \text{ V}$ for different P_o as shown in Figure 9. As discussed previously, G_{PD} becomes larger for higher V_{DS} . Thus, we measure an increase of almost 1 order of magnitude ($G_{PD} \sim 4 \times 10^5$ at $P_o = 100 \mu\text{W}$) for V_{DS} going from 0.1 to 1 V.

Finally, we test I_{ph} as a function of bending using a Deben Microtest three-point bending setup (Figure 10a). In this case, $r_b = [h^2 + (L/2)^2]/2h$, where L is the chord of circumference connecting the two ends of the arc, and h is the height at the

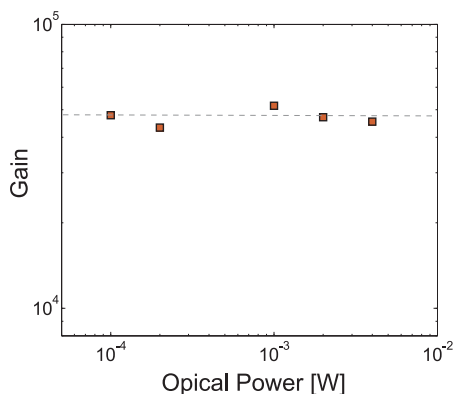


Figure 9. G_{PD} as a function of P_o at $V_{GS} = -1 \text{ V}$ and $V_{DS} = 0.1 \text{ V}$.

chord midpoint. The plotted values of I_{ph} in the bent state at each r_b ($I_{ph,bend}$) are normalized to the values of I_{ph} measured at rest with the sample in the flat position ($I_{ph,rest}$). Figure 10b plots the normalized $I_{ph,bend}/I_{ph,rest}$ for different r_b , showing

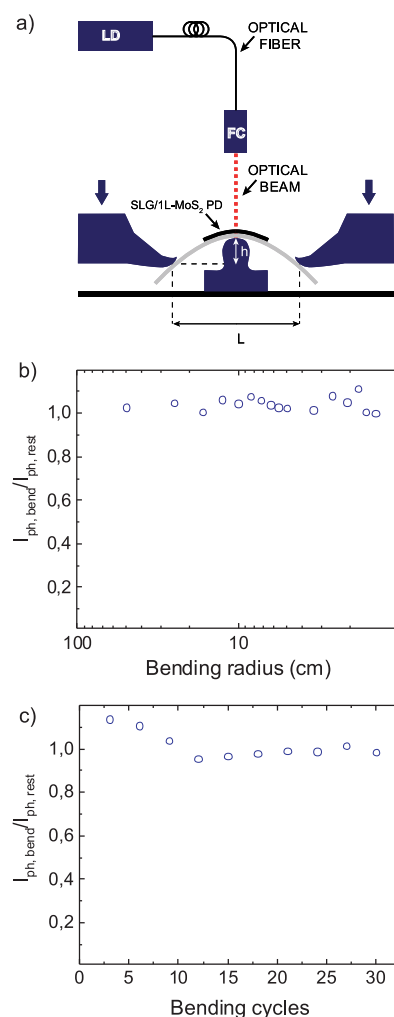


Figure 10. (a) Schematic three-point bending setup. LD = laser diode; FC = fiber collimator; (b) $I_{ph,bend}$ normalized to the value at rest $I_{ph,rest}$ as a function of r_b ; (c) $I_{ph,bend}$ normalized to the value at rest $I_{ph,rest}$ as a function of the number of bending cycles.

deviations within 15% for r_b down to 1.4 cm. This r_b is comparable to that reported for semiconductor membrane PDs,^{4,15} yet the latter have 2 orders of magnitude lower (<0.3 A/W) responsivities.^{4,15} Although our r_b is five times larger than that reported for flexible single NW devices,^{3,16–18} the area of our PDs (>40 mm²) is at least 6 orders of magnitude larger than that of the NW devices (<5 μm²). To test the device performance upon bending cycles, we first measure the photocurrent at rest ($I_{ph,rest}$ flat position) and then at the smallest r_b ($I_{ph,bend}$, $r_b \sim 1.4$ cm), repeating these measurements for 30 bending cycles. Figure 10c plots $I_{ph,bend}/I_{ph,rest}$ as a function of bending cycles. This shows that our PDs retain stable photocurrent after multiple bending tests with a $I_{ph,bend}/I_{ph,rest}$ standard deviation $\pm 12\%$.

CONCLUSIONS

We reported polymer electrolyte gated, flexible photodetectors, for visible wavelengths with external responsivity up to ~ 45.5 A/W, photoconductive gain $\sim 4 \times 10^5$, operation voltage <1 V, and optical transparency >82%. The responsivity is at least 2 orders of magnitude higher than in semiconductor flexible membranes. The devices show stable performance upon bending for radii of curvature larger than ~ 1.4 cm. Owing to their responsivity, flexibility, transparency, and low operation voltage, our photodetectors can be attractive for wearable, biomedical, and low-power optoelectronic applications.^{11,12,17}

METHODS

1L-MoS₂ is epitaxially grown by CVD on c-plane sapphire substrates.⁶¹ These are annealed at 1000 °C in air for 1 h after consecutive cleaning by acetone/isopropyl alcohol/DI water. They are then placed face-down above a crucible containing ~ 5 mg MoO₃ ($\geq 99.998\%$ Alfa Aesar). This is loaded into a 32 mm outer diameter quartz tube placed in a split-tube three-zone furnace. A second crucible containing 350 mg sulfur ($\geq 99.99\%$ purity, Sigma-Aldrich) is located upstream from the growth substrates. Ultrahigh-purity Ar is used as carrier gas at atmospheric pressure. The procedure is to ramp the temperature to 300 °C with 200 sccm Ar flow, set to 300 °C for 10 min, ramp to 700 °C (50 °C/min increase temperature rate) with 10 sccm Ar flow, set at 700 °C for 10 min, cool to 570 °C with 10 sccm of Ar, increase the gas flow to 200 sccm and open the furnace for rapid cooling.⁶¹ SLG is grown on a 35 μm Cu foil, following the process described in ref 51. The substrate is annealed in hydrogen atmosphere (H₂, 20 sccm) up to 1000 °C for 30 min. Then, 5 sccm CH₄ is added to initiate growth.^{51,62} The sample is then cooled in vacuum (1 mTorr) to room temperature and removed from the chamber.

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Notes

The authors declare no competing financial interest.

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