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Biomimetic synthesis of hemiketal eicosanoids for biological testing

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Abstract

The hemiketal (HK) eicosanoids HKE_2 and HKD_2 are the major products resulting from the biosynthetic cross-over of the 5-lipoxygenase and cyclooxygenase-2 pathways. They are formed by activated human leukocytes ex vivo, and, therefore, may be involved in regulation of the inflammatory response as autocrine or paracrine mediators. HKE_2 and HKD_2 are not commercially available and, so far, no method for their total chemical synthesis has been reported. The limited availability has impeded the characterization of their biological effects. Here, we describe a method for biomimetic preparation of HKE_2 and HKD_2 by reaction of recombinant human cyclooxygenase-2 with chemically synthesized 5*S*-HETE. We found that HKE_2 did not induce or inhibit the release of TNFa and IL-1 β by human THP-1 monocytes and phorbol ester treatment-derived macrophages.

Keywords

lipoxygenase; cyclooxygenase; hemiketal; di-endoperoxide; macrophage; cytokine

Introduction

The 5-LOX product, 5*S*-HETE, is an efficient substrate for COX-2 thereby linking the two enzymes in a converging biosynthetic pathway (1). A di-endoperoxide is the major product resulting from this cross-over pathway (Fig. 1) (1). The reaction also yields two by-products, 5S,11*R*-diHETE and 5S,15*R*,*S*-diHETE that are the 5-hydroxy analogs to the 11*R*- and 15R,*S*-HETE by-products in the reaction of COX-2 with arachidonic acid (2-4). The diendoperoxide is functionally equivalent to the prostaglandin endoperoxide PGH₂ formed in the COX reaction with arachidonic acid (2). Like PGH₂ the di-endoperoxide is an unstable intermediate that can undergo further enzymatic and non-enzymatic transformations (5).

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Treatment with redox active metal ions or hematin results in cleavage of the peroxide moieties, yielding the aldehydes 8-oxo-5-hydroxy-6*E*-octenoic acid, malondialdehyde, and 4-hydroxy-2*E*-nonenal (6). In the absence of a redox catalyst the di-endoperoxide rearranges by opening of the peroxide moieties into keto/hydroxy groups that engage in intramolecular acetal formation yielding the hemiketals HKE₂ and HKD₂ (7).

The di-endoperoxide is a substrate for the hematopoietic type of PGD synthase forming HKD_2 . Reaction with the lipocalin-type PGDS may result in HKE_2 (7). Activated human leukocytes stimulated with LPS and calcium ionophore A23187 ex vivo release HKE_2 and HKD_2 but it is not known whether these are enzymatic or non-enzymatic rearrangement products of the COX-2 derived di-endoperoxide (7). Stimulation of human leukocytes also gives 5,11- and 5,15-diHETE as products of the 5-LOX/COX-2 biosynthetic cross-over pathway (8).

Little is known about the biological role of the 5-LOX/COX-2 cross-over pathway and its products. One of the reasons for this lack in knowledge is that the HKs are not readily available. They are not sold commercially, and their chemical synthesis has not been reported. Due to the lack of methods for the synthesis of HKs, biochemical synthesis is an adequate method to obtain HKs for biological testing. In this report we provide a detailed description of the procedure we have developed for their preparation.

The HKs have been shown to stimulate tubulogenesis in microvascular endothelial cells isolated from mouse lungs (7). The endothelial cells also migrated in response to HK treatment. Because HKs are formed by mixtures of human leukocytes in response to stimulation by LPS and calcium ionophore, we decided to test whether they can regulate cytokine release by human THP-1 monocytes and THP-1 derived macrophages.

Materials and Methods

Materials

Arachidonic acid (90%+) was obtained from NuChek Prep. (Elysian, MN). Recombinant human COX-2 was expressed and purified as described (9). THP-1 cells were from ATCC (Manassas, VA).

Synthesis of 5S-HETE

A modified procedure for synthesis and chiral resolution of 5S-HETE has been described in detail previously (6,10).

Enzymatic synthesis of HKE₂ and HKD₂

Recombinant human COX-2 (0.2 μ M) was added to 2 ml of 100 mM Tris-HCl pH 8 containing 1 μ M hematin and 0.5 mM phenol. The solution was warmed to 37°C, and 30 μ g 5*S*-HETE were added. The reaction was kept at 37°C for 5 min followed by incubation at room temperature for 45 min. To the reaction were added 16 μ l glacial acetic acid and 50 μ l methanol before extraction using a preconditioned 30 mg Waters HLB cartridge. The cartridge was preconditioned by washing with 2 ml of methanol followed by 2 ml of water. Samples were loaded, washed with water, and eluted with 0.5 ml of ethyl acetate followed

by 0.5 ml of methanol into a conical vial. The water bubble in the bottom of the vial after ethyl acetate elution was removed with a pointy glass syringe prior to elution of the cartridge with methanol. The solvents were evaporated under a stream of N_2 , and the sample was reconstituted in 100 µl of methanol for HPLC analysis. For preparative isolation of HKs 10 or 20 2 ml-reactions were conducted in parallel, and two reactions each were extracted using one HLB cartridge.

HPLC isolation of HKE₂ and HKD₂

RP-HPLC was performed using a Waters Symmetry C18 5 μ m-column (4.6 × 250 mm) eluted with a gradient starting with acetonitrile/water/acetic acid 20/80/0.01 (v/v) to 70/30/0.01 (v/v) in 10 min at a flow rate of 0.4 ml/min. From 10 to 20 min the flow rate was held at 0.4 ml/min and then switched to 1.0 ml/min for the next 10 min before returning to the starting solvent. Elution was monitored using an Agilent 1200 diode array system programmed to record 4 wavelengths simultaneously: 205, 220, 235, and 270 nm. Products collected from RP-HPLC were evaporated under a stream of N₂ to remove organic solvent, diluted to 1 ml using 0.1% acetic acid and extracted using a 30 mg Waters HLB cartridge as described above.

SP-HPLC was performed using an Agilent Zorbax RX-SIL 5 μ m-column (4.6 × 250 mm) eluted with hexane/isopropanol/acetic acid 80/20/0.1 (v/v) at 1.0 ml/min flow rate and UV detection at 4 wavelengths as described above. HKE₂ used for biological testing was purified by SP-HPLC prior to use in the assays.

Quantification of HKE₂

HKE₂ was dissolved in 150 μ l acetonitrile- d_3 and transferred to a 3 mm NMR tube. The sample was spiked with 0.25 μ moles CH₂Cl₂ dissolved in 10 μ l CCl₄. The sample was analyzed using standard pulse frequencies with the delay time (D1) set to 30 s to ensure complete relaxation and correct integration of the signals. NMR spectra were recorded using a Bruker DRX 600-MHz spectrometer equipped with a cryoprobe at 298 K.

Stability of HKE₂

HKE2 (1 μ M) was dissolved in 2.7 ml of RPMI and placed in a cell culture incubator at 37°C. At 0, 1, 2, 4, and 8 an aliquot of 250 μ l was removed and extracted using a preconditioned 30 mg Waters HLB cartridge. The cartridge was eluted with methanol and evaporated. The sample was dissolved in 40 μ l of acetonitrile/water (1:1, by vol.) and analyzed by LC-ESI-MS in the negative ion mode.

THP-1 monocyte treatment and analysis

THP-1 cells were maintained in RPMI 1640 medium supplemented with 10% FBS, 2 mM glutamine, 100 U/ml penicillin, and 100 mg/ml streptomycin, and grown at 37°C in 5% CO₂ at constant humidity. The cells were maintained between 300,000 - 800,000 cells/ml, and subcultured every 2-3 days. Passages 5-10 and population doubling level from 10-30 were used for all the experiments. Cells were seeded in 96-well plates at a concentration of 500,000 cells/ml in complete medium (10% FBS) or deprived-medium (0.1% FBS), incubated for 24 h, and treated with 100 ng/ml LPS in the presence or absence of 1 μ M of

 HKE_2 for 5 h. Cells treated with vehicle (ethanol) were used as control. After treatment, media from 3 different wells corresponding to each treatment were pooled and kept at $-80^{\circ}C$ until analysis. TNF-a and IL-1 β were analyzed in culture media using an ELISA kit from Peprotech (Rocky Hill, NJ).

THP-1 derived macrophage treatment and analysis

Differentiation of THP-1 monocytes into macrophages was performed as previously described (11). 500,000 monocytes/ml were cultured in 96-well plates and treated with 100 ng/ml of phorbol-12-myristate-13-acetate (PMA) for 72 h. The medium was removed, and the cells were washed 3 times with PBS. The macrophages were incubated with complete or FBS-deprived medium for 24 h. The cells were treated with 1 μ M of HKE₂ in the presence or absence of LPS (100 ng/ml) for 5 h, and TNF-a and IL-1 β were analyzed as described for monocytes.

Results and Discussion

Synthesis of 5S-HETE

5S-HETE is readily synthesized following a procedure originally developed by Corey and Hashimoto (10). The procedure uses iodolactonization of arachidonic acid to a δ -lactone for the introduction of oxygen at the 5-carbon. Base-catalyzed removal of the iodine as HI yields the 6,7-trans double bond. The lactone is opened using triethylamine in methanol to form racemic 5-HETE methyl ester. The synthesis can be performed starting from arachidonic acid of low purity (>90%). The reaction steps do not require specialized equipment and can be carried out in a standard laboratory.

We routinely use a semi-preparative Chiralpak AD column (1 cm inner diameter) operated in straight phase mode for the resolution of milligram amounts of the enantiomers of 5-HETE methyl ester per chromatographic run (12). The *R*- elutes before the *S*-enantiomer (13).

Hydrolysis of the collected 5*S*-HETE methyl ester is achieved by dissolving the collected enantiomer in ethanol and adding an equal volume of 15% KOH. If large enough volumes of solvents are used (ca. 1 ml each for 1 mg of 5-HETE methyl ester) the solvent mixture results in a single phase with 5-HETE methyl ester dissolved without micelle formation. This is required for efficient hydrolysis. If no micelles are formed, the hydrolysis can be complete in as fast as 1 min at room temperature. If micelles form, these may be dissolved by adding 10 or 20 vol.-% dichloromethane. Success of hydrolysis can be assessed by direct injection of a small aliquot of the reaction mixture on RP-HPLC before termination of the reaction. If hydrolysis is complete, the solution is acidified using 1 N HCl to pH = 3, evaporated from ethanol, and extracted twice using dichloromethane. It is advisable to wash the combined dichloromethane phase 2-3 times with water to remove acid in order to reduce the chance for lactonization of 5-HETE during evaporation and storage. On the other hand, the more complete the acid is removed from the dichloromethane solution the greater is the tendency of 5-HETE to act as a detergent and form an emulsion during washing. Persistent emulsions are best dealt with by centrifugation.

5-HETE is quantified by UV spectroscopy using an ϵ value of 25,000 M⁻¹cm⁻¹ at 238 nm (14). An alcoholic solution of 10 µg/ml of 5-HETE gives an absorbance of 0.75 AU at 238 nm. Acetonitrile is a good solvent for preparation and storage of 5-HETE stock solutions. It reduces the chance of esterification to the methyl or ethyl ester derivatives that can occur when stock solutions are prepared in the corresponding alcohols.

Preparation of HKs

It is advisable to use enantiomerically pure 5*S*-HETE because the *R*-enantiomer will react to some degree with COX-2 (1) and form a diastereomeric product that will be difficult to resolve from the 5*S*-derived products. The recombinant COX-2 enzyme can be obtained commercially. We use recombinant human COX-2 expressed in *St9* insect cells that is purified by ease of a His₆-tag that has been encoded into the cDNA following the N-terminal signal peptide cleavage site (9,15).

Small amounts of standards of HKE₂ and HKD₂ for MS or HPLC analyses can be prepared using a 100 µl reaction scale and containing 2 µg 5*S*-HETE and 200 nM COX-2. All reactions require phenol (0.5 mM) as a co-substrate in the peroxidase active site and hematin (1 µM, freshly prepared) to replenish heme in the POX active site that may have been lost during purification of the enzyme. The hemiketals can be analyzed by their molecular ion using LC-MS in the negative ion mode (m/z 399) or by characteristic fragment ions in SRM analyses (HKE₂: m/z 399 -> 151; HKD₂: m/z 399 -> 183) (7).

For isolation of HKs we scale up by using 2 ml reactions with 30 µg 5*S*-HETE substrate. We test the enzyme activity by performing pilot reactions with different amounts of enzyme to optimize conversion of 5*S*-HETE. We usually perform 10 or 20 parallel 2 ml reactions since increasing the volume beyond 2 ml and scaling reagents accordingly has not resulted in an increase in the absolute yield of HKs. We warm the buffer with enzyme, hematin, and phenol to 37°C and add 5*S*-HETE to start the reaction which is held at 37°C for 5 min (during the enzymatic reaction) and then incubated at room temperature for 45 min to allow for the rearrangement of the di-endoperoxide into the HKs (7).

HPLC isolation

The extracted reaction mixture is first analyzed using RP-HPLC. A typical chromatogram shows elution of HKE₂ and HKD₂ as well as 5,11- and 5,15-diHETE ahead of the unreacted 5*S*-HETE substrate (Fig. 2A). We modified the flow rate for the gradient elution in order to achieve resolution of the HKs from phenol which tended to elute between HKE₂ and HKD₂ when using more generic conditions. The starting material and products can be differentiated by their UV spectra (Fig. 2B). HKE₂ eluted as a sharp peak whereas HKD₂ consistently eluted as a broad peak, and at times appeared to have several maxima. Inspection of the UV spectra showed that the HK chromophore (λ max 236 nm) was present throughout the entire peak of HKD₂. The unfavorable peak shape may be related to the opening and closing of the hemiketal ring during the chromatographic run that appears to be more prevalent with HKD₂ than HKE₂.

The collected products are extracted from RP-HPLC solvent and further purified to remove the co-eluting isomers using SP-HPLC (Fig. 3). The peaks collected from SP-HPLC are evaporated directly and prepared for quantification.

The di-endoperoxide precursor of the HKs can readily be isolated when the reaction times with COX-2 are kept short and the ensuing extraction and HPLC analyses are performed at cold temperature (1,7). We have noticed, however, that the di-endoperoxide can break down during RP-HPLC if a "wrong" column was used. Breakdown of the diendoperoxide is apparent from an early eluting peak of 8-oxo,5*S*-hydroxy-6*E*-octenoic acid and a later peak of 4-hydroxy-2*E*-nonenal which elutes closely to the diendoperoxide; both aldehydes can be recognized by their typical enone chromophore with λ max around 220 nm (6,16). We are uncertain as to what factors render a column unfit for preparation of di-endoperoxide but they may include old columns that have accumulated redox-active metal ions or other reactive material.

Quantification of HKE₂

HKE₂ collected from SP-HPLC was quantified by ¹H NMR by adding a defined amount of CH₂Cl₂ dissolved in CCl₄ as internal standard. We chose CH₂Cl₂ and CCl₄ because both can be readily removed from the sample by evaporation. In the NMR solvent used (acetonitrile-*d*₃), HKE₂ exists as two isomers of about similar abundance. For the signal chosen for quantification (H-7) the integrated areas of the *dd* signals at 6.69 and 6.82 ppm were added together. Comparison of the areas of the methylene signal of CH₂Cl₂ (5.42 ppm) to the signals for H-7 allowed the amount of HKE₂ to be calculated. The quantified sample was then evaporated and analyzed by UV spectroscopy to determine the molar extinction coefficient ϵ . The procedure was performed twice using two different samples resulting in ϵ_{236} (MeOH) = 11,000 ± 1,000 M⁻¹cm⁻¹ for HKE₂. This value should be considered preliminary since only microgram amounts of HKE₂ were available for quantification. The ϵ of HKD₂ is assumed to be similar although this was not confirmed experimentally.

Despite many trials to improve the yield of the procedure we were unable to prepare more than ca. 20 μ g each of isolated and highly purified HKE₂ and HKD₂ starting from 500 μ g 5*S*-HETE. This is less than 5% overall yield, and a major factor is likely the need to perform both RP- and SP-HPLC purification steps. Attempts to use only SP-HPLC as a single purification step were unsuccessful.

Stability of HKE₂

We analyzed the stability of HKE₂ in RPMI cell culture medium at 37°C (Fig. 4). Aliquots were removed from the incubation at 0, 1, 2, 4, and 8 h and extracted. Remaining HKE₂ was quantified using LC-MS. After 4 h there was 38% of the starting amount remaining and about 20% after 8 h. Although HKE₂ undergoes transformation in buffer it is sufficiently stable for use in bioassays that require less than 4 h incubation time. For long term storage, stock solutions of HKE₂ in ethanol or methanol are best kept at -80° C.

Effect of HKE₂ on cytokine release by THP-1 monocytes and macrophages

Assuming that HKs, like many other eicosanoids, could function as autocrine or paracrine mediators in inflammation (17-19), we tested whether HKE₂ can regulate the release of TNF- α and IL-1 β from activated monocytes and macrophages. THP-1 cells were used as monocytes or differentiated into macrophages by culturing in the presence of PMA for 72 h. We also tested the effect of FBS supplementation of the culture medium, using either 10% or 0.1% FBS. LPS stimulation resulted in the release of TNF- α into the culture medium by the macrophages whereas it had much less, if any, effect on IL-1 β (Figure 5A, B). Treatment of the macrophages with 1 μ M HKE₂ in the absence or presence of LPS did not have an effect on TNF- α or IL-1 β release (Figure 5A, B). In the monocytes HKE₂ was likewise without an effect on LPS-induced release of TNF- α or IL-1 β (Figure 5C, D). The effect of HKE₂ in the absence of LPS could not be determined as the levels of TNF- α and IL-1 β were below the limit of detection of the analysis. These data indicate that HKE₂ may not function as a simple regulator of cytokine release from monocyte/macrophages and may affect other target cells or be acting in a more subtle or more complex manner than what was tested here.

Conclusion

Preparation of HKE₂ and HKD₂ on a small scale as reference compounds in MS analyses can be achieved using all commercial reagents. Both 5*S*-HETE and COX-2 are available from various suppliers, and the enzymatic reaction is a simple incubation in buffer. Together with co-workers at the Chemistry Department at Vanderbilt University we have begun to develop a process for the total chemical synthesis of HKE₂ and HKD₂ (Boer, R.E., Schneider, C., Sulikowski, G.A. et al., unpublished). The synthesis is a multi-step process and entails reactions that could be difficult to perform in a molecular biology lab. In the absence of a straightforward chemical synthesis for preparation of HKE₂ and HKD₂ enzymatic synthesis with recombinant COX-2 starting from 5*S*-HETE is a feasible approach for preparation of small amounts of HKs for biological testing or as a standard in LC-MS analyses.

Acknowledgement

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Abbreviations

COX	cyclooxygenase
(di)HETE	(di)hydroxyeicosatetraenoic acid
НК	hemiketal
LOX	lipoxygenase
PG	prostaglandin
PMA	phorbol-12-myristate-13-acetate

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Highlights

- HKE₂ and HKD₂ are formed by cross-over of the 5-LOX and COX-2 pathways.
- A simple method for the preparation of HKE₂ and HKD₂ is described.
- HKE_2 and HKD_2 do not regulate release of TNFa and IL-1 β from THP-1 cells.

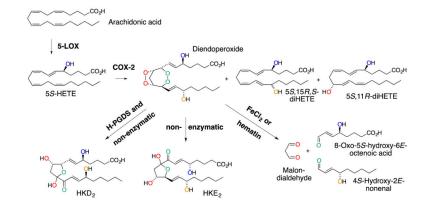


Fig. 1.

Consecutive oxygenation of arachidonic acid by 5-LOX and COX-2 gives a diendoperoxide as the major and 5,11- and 5,15-diHETE as by-products. Further enzymatic or non-enzymatic transformation of the di-endoperoxide gives HKD₂ and HKE₂. The diendoperoxide can be cleaved by FeCl₂ or hematin into three aldehydes. COX-2 reacts 5*S*-HETE with 3 molecules of oxygen, and their incorporation into the di-endoperoxide and subsequent fate in the HKs and other products is marked by the different colors.

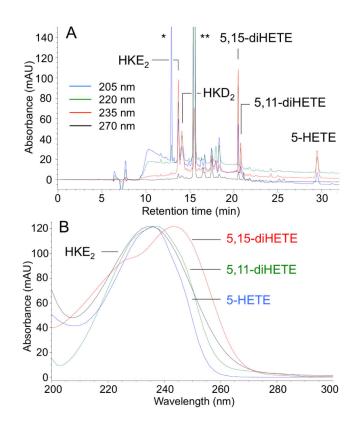
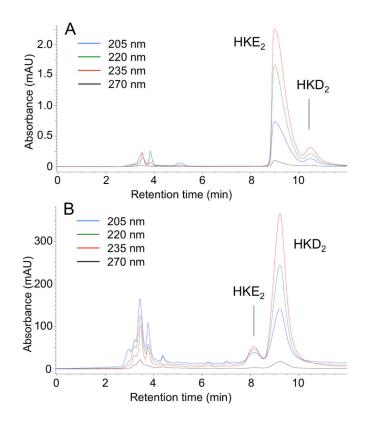


Fig. 2.

RP-HPLC/diode array analysis of the reaction of 5S-HETE with COX-2. (A) RP-HPLC analysis showing the elution of HKE₂, HKD₂, 5,15- and 5,11-diHETE, as well as unreacted substrate, 5-HETE. The single (*) and double (**) asterisk indicate the elution of detergent from the protein solution and of phenol, respectively. (B) UV spectra for products and substrate obtained in the HPLC analysis.





SP-HPLC analysis of HKE_2 (A) and HKD_2 (B) collected off RP-HPLC. The products were analyzed at different days which resulted in a slight discrepancy of the retention times between the chromatograms in (A) and (B).



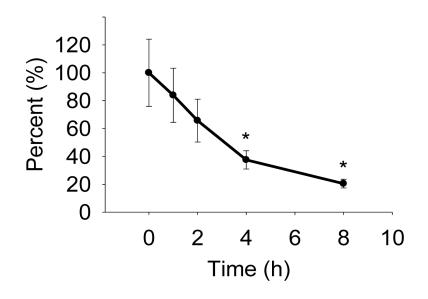


Fig. 4.

Stability of HKE₂ in cell culture medium. HKE₂ (1 μ M) was dissolved in RPMI and incubated at 37°C. At the times indicated an aliquot was removed, extracted, and quantified by LC-MS analysis. The asterisk indicates significant difference in ANOVA analysis (p<0.05) from the starting conditions at t=0 h (n = 3 independent experiments).

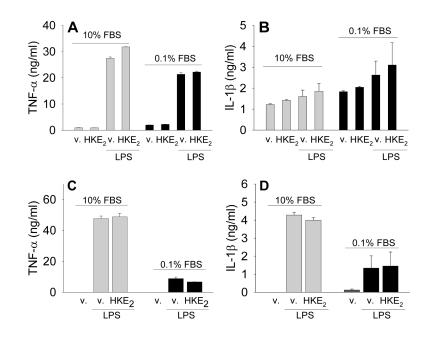


Fig. 5.

Effect of HKE₂ on the release of TNF- α and IL-1 β by THP-1 monocytes and phorbol estertreatment derived macrophages. Macrophages (A, B) and monocytes (C, D) were treated with or without LPS (100 ng/ml) in the absence or presence of HKE₂ (1 μ M) for 5 h before collection of the cell culture media for ELISA. Cells were supplemented with either 10% or 0.1% FBS during treatment. Data are expressed as mean \pm SD of two (A,C) or three (B,D) independent experiments.