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All-Atom Molecular Dynamics Simulations of Entire Virus Capsid Reveal the Role of Ion Distribution in Capsid's Stability

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Abstract

Present experimental methods do not have sufficient resolution to investigate all processes in virus particles at atomistic details. We report the results of molecular dynamics simulations and analyze the connection between the number of ions inside an empty capsid of PCV2 virus and its stability. We compare the crystallographic structures of the capsids with unresolved N-termini and without them in realistic conditions (room temperature and aqueous solution) and show that the structure is preserved. We find that the chloride ions play a key role in the stability of the capsid. A low number of chloride ions results in loss of the native icosahedral symmetry, while an optimal number of chloride ions create a neutralizing layer next to the positively charged inner surface of the capsid. Understanding the dependence of the capsid stability on the distribution of the ions will help clarify the details of the viral life cycle that is ultimately connected to the role of packaged viral genome inside the capsid.

Graphical Abstract

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Author Contributions

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Supporting Information

The Supporting Information is available free of charge on the ACS Publications website at DOI: 10.1021/acs.jpclett.6b02759. Detailed description of the preparation of the structures for the simulations, the simulation process, RMSD(*t*) and $R_g(t)$ plots for GROMOS force field, and the charge distribution in final configurations (PDF)



Detailed knowledge of the structure and function of viruses at the molecular level is necessary for developing vaccines and drug treatments for diseases caused by these viruses. Recently such information became possible to obtain because experiments allow the measurement of structures of viral particles at atomistic resolution by X-ray crystallography and cryo-electron microscopy.^{1–5} However, since the measurements are performed at nonphysiological conditions (e.g., low temperature, crystal packing of the sample), the obtained structure may differ from the cellular environment. In addition, experiments do not resolve some parts of the viral particles (typically flexible or disordered) that can be important for the structure and the dynamics of the virus.^{1,2}

Simulations of whole virus particles using molecular dynamics (MD) is an actively developing area of research.^{6–15} They (i) can provide detailed information about the structure and dynamics of the system in solution at atomistic and molecular levels and (ii) become feasible for relatively long simulated times by using high performance computation, including special-purpose computer systems, such as ANTON¹⁶ and MDGRAPE-4,¹⁷ and new algorithms.^{18–21}

Here we report on the simulation study of Porcine Circovirus type 2 (PCV2). It refers to the family *Circoviridae* and has circular single-stranded DNA placed inside a nonenveloped icosahedral capsid. PCV2 is pathogenic and causes porcine circovirus-associated diseases.^{1,22–24} This virus represents a convenient object for MD simulations because it is one of the smallest viruses, with a diameter of about 20 nm, and its atomistic structure has been determined experimentally at 2.3 Å resolution by X-ray crystallography.¹ The capsid consists of 60 copies of single gene product (elementary subunits), the structure of which is available in the Protein Data Bank (PDB ID 3R0R). Importantly, the flexible N-terminal ends of the proteins (33 residues) could not be resolved in the structure and, therefore, are absent in the PDB. Also, the crystallographic structure does not reflect the complete picture of interactions involving proteins constituting the capsid and the solvent at room temperature,^{12,25} where molecules of water and ions play a key role in the dynamical

The proteins of the PCV2 capsid are capable of spontaneously assembling into an empty capsid (also termed virus-like particle), which is stable in the absence of the DNA inside, as shown in experiments at physiological conditions (refs 27 and 28, and unpublished results) albeit at rather low resolution (only the overall shape of the empty capsid was visible). However, the reason for this stability of the empty capsid without the DNA inside is still unclear.

By performing the simulations under physiological conditions (e.g., pH, salinity, and temperature) using the crystallographic coordinates, we hypothesize that one of the possible reasons for the stability of the empty capsid is linked to the charges carried by the ions in solution, particularly chloride ions, that could mimic the genome in its absence and substitute its functions related to the maintenance of the capsid structure. The imitation of genome by negative charges has been already shown for satellite tobacco mosaic virus and satellite tobacco necrosis virus, where chloride ions are located at the same regions as the genome.⁹ Therefore, we consider cases with varying number of ions to check this hypothesis.

In this work, we report that while the crystallographic structure agrees reasonably well with the MD structures, a noticeable difference arises between the simulations of the capsids with low and high number of ions that confirms our hypothesis.

The PCV2 capsid consists of 60 copies of the structural subunits arranged in an icosahedral symmetric structure. The subunits are assembled without gaps with the exception of small pores at 5-fold axes, large enough for water and ion transport.

We perform MD simulations of the whole virus capsid at all-atom resolution in explicit water. The initial structure of the elementary subunit was taken from the X-ray measurements excluding the first 33 amino acids, the N-terminus (which we call "tail"), located inside the capsid.¹ The sequence for the missing N-terminus is known from the sequence of the gene used to express the elementary subunit. In this work we used the structure of PCV2 consensus sequence virus-like particle, which was chosen for X-ray crystallography as the most convenient for expression. The structure of the N-termini was impossible to resolve most probably because of their disordered conformations.² Note that the N-termini that we used are slightly negatively charged, which is in contrast to the real virus, where they possess significant positive charge. This facilitates their interaction with negatively charged DNA. In our case, however, the DNA is absent and, thus, the charge of the N-termini is not critical for the self-assembly of the capsid.

During the preparation of the initial capsids without the tails, the elementary subunit without the N-terminus was copied 60 times, taking into account the symmetry of the capsid, and assembled into a complete icosahedral capsid using VIPERdb Utilities.²⁹ The same procedure was applied for the assembly of the capsid including tails, with the structure of the tails reconstructed based on the known sequence using the Homology Modeling method

by Modeller³⁰ and adding it to the elementary subunit from the PDB. Next, the structural refinements of the tails inside the whole capsid were performed using Molecular Operating Environment (MOE)³¹ program (Chemical Computing Group, Inc.)

The capsid carries a large total positive charge of 360 |e| (without tails) or 180 |e| (with tails, which are negatively charged). The charge is distributed unevenly between the inner and the outer surfaces. In order to neutralize this excess charge correctly, we plotted the dependences of the integral charge of the capsid atoms on the distance from the capsid center of mass (COM), Figure 1. We used adjacent averaging to remove noise from the plots.

For comparison we also plotted the density of capsid atoms versus the distance from the capsid COM. The distinct well on the plot for the capsid with the tails corresponds to tails that are stretched toward the capsid center. The maxima on the density plots are located at ~8.5 nm and correspond to the middle of the capsid wall. The maxima of the charge plots, however, are located at a shorter distance of ~7.8 nm. This allowed us to conclude that the positive charge is concentrated at the inner surface of the capsid and has the magnitude equal to the value at the maximum of the charge plot (606 for the capsid without tails and 407 for the capsid with them). The outer surface carries the negative charge of smaller magnitude (246 for the capsid without the tails and 227 for the capsid with them). In total the capsid has positive charge. Because the capsid appeared to be highly polar, we decided to separately neutralize the charge of each surface.

The positive charge of the inner surface was neutralized with Cl⁻ ions placed inside the capsid (606 ions for the capsid without the tails and 407 ions for the capsid with them). The negative charge of the outer surface was neutralized with Na⁺ ions placed in the bulk solution (246 for the capsid without the tails and 227 for the capsid with them). We will call the systems with these numbers of ions "neutralizing" further in the text. Finally, to simulate physiological conditions, we added 1720 Na⁺ and Cl⁻ ions uniformly distributed in the system, which corresponds to 0.9 mass percent of NaCl buffer.

Also for both capsids we performed simulations with a reduced number of ions, where we placed half of the needed number of Cl^- inside the capsids (300 in the capsid without tails and 200 in another one). To maintain the electroneutrality of the whole system, we also added ions outside the capsid (60 Cl⁻ for the capsid without tails and 20 Na⁺ for the capsid with tails). The systems with these numbers of ions will be called "low" to distinguish them from the "neutralizing" systems. NaCl in physiological concentration (1720 Na⁺ and Cl⁻) was added as well. A detailed algorithm of adding ions is presented in the Supporting Information. We would like to stress that all simulated systems were electroneutral as a whole.

Two different force fields were used for each simulation: GROMOS53a6³² and AMBER03,³³ to investigate the dependence of the results on the force field details of interatomic interactions.

Different models of water were also chosen for the different force fields: SPC for GROMOS and TIP3P for AMBER. In total, eight systems were studied: four capsids with the neutralizing number of ions (two capsid without the tails and two capsids with the tails

included), and four of the same systems with the deficit of ions inside. Table 1 summarizes the initial conditions of the systems in the AMBER force field.

The initial structures were equilibrated before the start of MD simulations via energy minimization and subsequent MD run with position restraints of the capsid, increasing the temperature from 200 to 300 K (see Supporting material for details). We used PME treatment of Coulomb interactions for the MD simulations. After the initial equilibration, we simulated each system for 10 ns and used these intervals for the analysis.

At the end of the 10 ns simulation, all capsids with the neutralizing number of ions preserved symmetry, whereas simulations with the low number of the ions lost the capsid symmetry during the simulation and showed clear signs of disintegration. The simulation of the system with the low number of ions and no tails in AMBER force field was then extended for additional 10 ns.

To quantitatively demonstrate the stability of the structure, we analyzed the change of rootmean-square deviation (RMSD) and the radius of gyration (R_g) for the backbone atoms of the whole capsids with time. Plots for AMBER force field are shown on Figure 2, plots for GROMOS force field are presented in the Supporting Information. The calculations of RMSD and R_g were performed for the same set of amino acids for all systems and did not include the tails.

First consider the results obtained for the capsids with the neutralizing number of ions.

Figure 2a,c show RMSD and R_g values for the capsids with the neutralizing number of ions without tails (black curve) and with tails (red curve). A stable value of RMSD for the capsid without tails at 0.2 nm indicates a stable state of the viral particle and small deviation from the initial structure, which is a usual consensus in literature,^{8,34} together with constant values of R_g allow us to conclude that the capsid without tails is stable during 10 ns of simulation. Despite the presence of slow growth of RMSD for the capsid with the tails we can conclude that the capsid is in a stable state, because R_g remains constant. The growth of RMSD in the case with the tails could be explained by local rearrangements in the proteins or some movements of the parts of protein, which are connected with flexible tails that do not affect the overall shape of the capsid. This behavior coincides with experimental data.

Distinct results are obtained for the case where the low number of ions are placed inside the capsid (Figure 2b,d). Both structures demonstrate the continuous growth of RMSD. The value for the capsid without tails reaches almost ~0.4 nm, and the value for the structure with the tails is almost ~0.3 nm. Also, the continuous increase of R_g for both capsids suggests instability of the initial conformation of each capsid with the low number of ions.

It is noted that for all cases with low number of ions (Figure 2b,d), the capsids demonstrate drastical change and they lose their symmetry, which suggests that the initial structure is not stable for the conditions. The loss of symmetry for the capsid without tails and with the tails happens at different rates. We conclude that the capsid with tails has additional interactions between the tails and the inner surface of the capsid that play the role of anchors, which

prevent rapid destruction of the structure; however, it is not enough for maintaining the symmetrical structure.

Plots for the same systems in GROMOS force field show similar behavior (Supporting Information).

Figure 3 shows the results of the simulations for the capsids without tails in the GROMOS force field with the neutralizing number of ions (a,c) and with the low number of ions (b,d). The systems with low number of ions develop cracks in the capsid shell in the region of 5-fold axis (Figure 3b,d), which is the reason for the loss of the symmetry and growing values of RMSD and R_{g} .

Coinciding values of R_g with respect to x-, y-, and z-axes unchanged with time (Figure 3e,f) demonstrate the preservation of the symmetry for the system with the neutralizing number of ions (Figure 3e), while increasing in different way values of R_g indicate the loss of the symmetry for the system with the low number of ions (Figure 3f) that shows that the capsid expands unevenly, developing large cracks in the shell. This behavior does not have a tendency to change with time, even after additional 10 ns of simulation, Figure 3f.

Summarizing, the results obtained for eight systems after 10 ns of MD simulation show that the number of ions inside plays the key role in the stability of the capsid and in maintaining its symmetry.

We additionally plotted the dependences of the integral charge of the capsid atoms on the distance from the capsid's center of mass for the final structures in order to assess the change in comparison to the initial structure. The plots are shown in the Supporting Information. The plots are very similar to the initial ones, and the heights of the maxima remained unchanged.

The loss of the capsid symmetry and the formation of the cracks at certain regions, that we observed, suggests the hypothesis that there is a qualitative difference in the distribution of the ions inside the capsids. Figure 4d shows that next to the positive inner wall of the capsid, a tight layer of chloride ions is formed when the number of ions is correct. However, in the case of the low number of ions (Figure 4b), the chloride ions are accumulated unevenly on the inner capsid surface, forming distinct voids near the pores. We illustrate the case of ion distribution in the capsid without tails because the presence of the tails complicates the picture.

We suggest two possible mechanisms that explain the loss of the capsid symmetry.

1. When the capsid has insufficient number of chloride ions inside, its positive inner wall is covered by negative ions unevenly, and some areas of the wall have a deficit of the negative ion charges (these areas are mostly in the vicinity of the capsid's pores; Figure 4b). Because of this unbalanced charge distribution, the unneutralized positive parts of the inner wall are attracted toward the negative cloud of the chloride ions inside the capsid, causing this part of the wall to collapse inside and flattening the roughly spherical shape of the capsid at this

location. This correlates with the conclusions of ref 9, where the authors observed accumulation of chloride ions inside the 5-fold symmetry pores.

2. As we observed during the simulation, the lack of chloride ions in and around the pores allows sodium ions to pass through the pore, increasing the already excessive unneutralized positive charge of the inner wall in the vicinity of the pore, which leads to weakening of the interactions between the subunits and eventual appearance of wide cracks in the capsid at these locations. In the case of the neutralizing number of chloride ions inside the capsid, the chloride ions form a tight layer of negative charge evenly distributed on the inside surface. This layer is matched by a similar positive layer of sodium ions on the outside surface of the capside (Figure 4c,d). These two layers of large opposite charges are strongly attracted to each other, stabilizing the whole structure of the capsid.

In conclusion, our simulations demonstrate that the neutralizing number of ions (especially chloride ions) is necessary for the stability of the capsid, and the chloride ions cover the positive inner wall of the capsid, creating the tight layer of negative charge.

Lastly, comparing the two used force fields, we conclude that AMBER produces the structure of the stable capsid that more closely matches the crystallographic structure. Whether this signifies the better suitability of AMBER force field for PCV2 simulations or GROMOS will also present the stable result is unclear; more simulation time is needed for conclusive results.

Also, our results show that the RMSD between the structure with the neutralizing number of ions with and without the N-termini in solution at 300 K and X-ray structure at 100 K crystal packing environment is not high. This demonstrates that the crystallographic structure reflects the structure in vivo.

Supplementary Material

Refer to Web version on PubMed Central for supplementary material.

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Figure 1.

Dependences of integral charge (top) and number density (bottom) of capsid atoms on the distance from the capsid center of mass. Red curves correspond to the capsid without tails, blue curves correspond to the capsid with tails. The vertical line corresponds to the middle of the capsid wall.



Figure 2.

Time change of the root-mean-square deviation, RMSD, (top plots) and radius of gyration, R_g , (bottom plots) of backbone atoms from their initial structure as a reference structure in AMBER force field. The calculations of RMSD and R_g did not include the tail sequence for all structures and consisted of the same number of amino acids. Red is for the capsid with the tails, black is for the capsid without the tails. (a,c) Capsids with the neutralizing number of ions; (b,d) capsids with the low number of ions.



Figure 3.

Atomistic structures of PCV2 capsid without tails after simulations: (a,c) the atomistic structures of PCV2 capsid with the neutralizing number of ions, (e) radius of gyration for different axis (green is *X*-axis, red is *Y*-axis, blue is *Z*-axis); (b,d) the capsid structures with the low number of ions and (f) radius of gyration for different axis.



Figure 4.

Distribution of ions after the first nanosecond of MD simulation for the low number of ions (top) and for the neutralizing number of ions (bottom). Red density (a,c) represents the density of sodium ions, and blue density (b,d) is the density of chloride ions averaged over one nanosecond of simulation. The pictures of the capsids with ions are sliced along the middle of the capsid.

Table 1

Initial Composition of Modelled System^a

| | | capsid wit | hout tails | | | capsid w | ith tails | |
|-----------------------|---------------------|--------------------|------------|---------|---------------------|-------------------------|---------------------|----------------------|
| system | 2 | 11 | Γ | ľ | 2 | I | Г | ľ |
| box size (nm) | 26 | .68 | 26 | .68 | 26 | .65 | 26. | .65 |
| | inside | outside | inside | outside | inside | outside | inside | outside |
| $N_{ m compens}$ | 606 Cl ⁻ | 246 Na^+ | 300 CI- | 60 CI- | 407 CI ⁻ | $227 \ \mathrm{Na^{+}}$ | 200 CI ⁻ | $20 \ \mathrm{Na^+}$ |
| $N_{\mathrm{Na}}{}^+$ | 15 | 966 | 17 | 20 | 19 | 47 | 17 | 40 |
| $N_{ m Cl}{}^-$ | 25 | 326 | 20 | 80 | 21 | 27 | 19 | 20 |
| $N_{ m water}$ | 568 | : 707 | 569 | 199 | 559 | 368 | 559 | 782 |
| $N_{ m atoms}$ | 1 89 | 8 573 | 1 899 |) 557 | 1 89 | 2 998 | 1 898 | 3 826 |

number of ions added to capsids for charge compensation. MN_{a+} – total number of sodium ions; NC1- - total number of chloride ions; Nwater - total number of water molecules; Natoms -total number of atoms in system. - subdens a