

MINIREVIEW



Are Antimalarial Hybrid Molecules a Close Reality or a Distant Dream?

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ABSTRACT Emergence of drug-resistant Plasmodium falciparum strains has led to a situation of haste in the scientific and pharmaceutical communities. Hence, all their efforts are redirected toward finding alternative chemotherapeutic agents that are capable of combating multidrug-resistant parasite strains. In light of this situation, scientists have come up with the concept of hybridization of two or more active pharmacophores into a single chemical entity, resulting in "antimalarial hybrids." The approach has been applied widely for generation of lead compounds against deadly diseases such as cancer and AIDS, with a proven potential for use as novel drugs, but is comparatively new in the sphere of antimalarial drug discovery. A sudden surge has been evidenced in the number of studies on the design and synthesis of hybrids for treating malaria and may be regarded as proof of their potential advantages over artemisinin-based combination therapy (ACT). However, it is evident from recent studies that most of the potential advantages of antimalarial hybrids, such as lower toxicity, better pharmacokinetics, and easier formulation, have yet to be realized. A number of questions left unaddressed at present need to be answered before this approach can progress to the late stages of clinical development and prove their worth in the clinic. To the best of our knowledge, this compilation is the first attempt to shed light on the shortcomings that are surfacing as more and more studies on molecular hybridization of the active pharmacophores of known antimalarials are being published.

KEYWORDS antimalarials, combination therapy, hybrids, covalent bitherapy

alaria continues to be a prominent killer of populations in the tropics. In the recent past, we had already been hit with a major setback in the effective control of malaria, when chloroquine (CQ) lost its position as a first-line antimalarial drug (1, 2). Consequently, most 4-aminoquinoline (AQ) drugs show cross-resistance due to their structural relationship to chloroquine. More recently, the endoperoxide sesquiterpene lactone-artemisinin (lactone-ART) (and its derivatives) became the mainstay for treating malaria. They were considered to represent the only class of potential drugs available to vitiate the impact of multidrug-resistant strains of Plasmodium. The World Health Organization (WHO), over the last decade, has been advocating the deployment of artemisinin-based combination therapy (ACT) as the "gold standard" for treatment of all malaria infections in areas afflicted by Plasmodium falciparum. It involves the simultaneous use of two or more blood schizonticidal drugs with independent modes of action and different biochemical targets in the parasite (3). However, recent reports on the emergence of artemisinin resistance have increased awareness of the risk of returning our efforts at reducing the worldwide malaria burden to ground zero (4, 5). As ACT is one of the last viable treatment options that we presently have, the current state of affairs is extremely worrisome.

Although true clinical resistance to artemisinins has not been confirmed in parasites collected from patients, there have been reports of clinical failures of artemisinin



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FIG 1 (A) Hybrid compounds classified according to (a) mode of interaction of the individual pharmacophores with target, (b) nature/form of presentation, and (c) nature of the linker unit employed. (B) Chemical structures of compounds representing antimalarial hybrids in each category. The chemical structure representing a merged hybrid (F) is that of a potential anticancer agent.

treatment (6). A small number of cases with poor responses to artesunate or artemether have also been reported in western Thailand, India, and Sierra Leone (7, 8). Some clinical parasite isolates from Nigeria and Madagascar appear to exhibit reduced sensitivity (9, 10).

Factors such as failure of mosquito control methods, development of parasite resistance to the existing antimalarials, lack of an effective antimalarial vaccine reaching clinical application, shortfall in drug supply versus demand, inadequate drug deployment, and poor patient compliance contribute to the severity of the malaria menace (11).

The present scenario has directed the attention of researchers toward exploring new, multifaceted avenues of drug discovery, the outcome of which has been the design and synthesis of antimalarial "hybrids."

Hybrid compounds can be defined as chemical entities with two or more structural moieties with different biological functions, thus combining two or more pharmacophores in a single molecule (12, 13). In simple terms, it is a rational chemistry-based approach which involves the covalent linking of two molecules, each with its own antimalarial activity, to produce a single hybrid molecule with dual activity (14). Therefore, the concept, also known as "covalent biotherapy" or "double drugs," can be regarded as an extension of the concept of a fixed-dose combination of two or more drugs in a single tablet. Hybridization of molecules is a powerful tool that has been utilized by several research groups to develop compounds with the potential to treat a number of diseases such as cancer, AIDS, and tuberculosis and is now gaining momentum in the field of antimalarial drug discovery.

Hybrid molecules can be classified on three different bases:

- Mode of interaction of the individual pharmacophores with target
- Nature/form of presentation
- Nature of the linker unit employed

Figure 1 illustrates a detailed classification of hybrids based on the categories listed above, along with an example of a hybrid compound in each category (12–24).

ADVANTAGES OF HYBRIDS

The suddenly accelerated pace of studies on the design and synthesis of hybrid antimalarial drugs stands strongly in support of the potential advantages of hybrids over ACTs, which are listed below.

Drug resistance. The use of double-drug or dual molecules can be regarded as an approach that would reduce the risk of drug resistance development by the mutual protection of each pharmacophoric moiety. This is especially useful in designing drugs such as the aminoquinolines (for example, chloroquine), where resistance does not surface because of an altered target but because of an inability to latch onto the target. It will be interesting to evaluate whether the concept of "covalent bitherapy" can be exploited to develop hybrid molecules with the ability to restore the activity of members of other drug classes, such as antifolates (e.g., sulfadoxine/pyrimeth-amine), which have been rendered ineffective by emergence of resistance (25).

Solubility. In case of fixed-dose combinations, different levels of bloodstream uptake occur due to differences in the solubility of the partner drugs. However, with a hybrid, one can surpass the fine-tuning required to ensure similar blood levels of drugs administered in the same tablet. If one moiety of the hybrid molecule is more soluble than the other, its uptake capacity can be used to contribute to the bioavailability of the other (26). Additionally, the nature of the linker employed can also contribute to the solubility of the entire unit. For example, the linker employed in chloroquine-pyrimethamine hybrids has two ethylene oxide units, which take part in hydrogen bonding with the water molecules and most likely contribute to its good solubility in both acidic and neutral media (27).

Synergism. If the active moieties of the two partner drugs are linked and if the spacing is appropriate, they may interact synergistically and display higher activity than as free agents (13). Walsh et al. gave a proof of concept for the hybridization of artemisinin and quinine in a single molecule, which showed an enhanced antimalarial effect in comparison to that of each of the parent compounds as well as in comparison to that of a 1:1 mixture of artemisinin and quinine (28). Another example illustrating the synergistic effect of covalent linkage was furnished by Benoit-Vical and coworkers (29). They combined trioxane and chloroquine into a single molecule, the hybrid was also able to inhibit the properties of trioxane-containing molecules, the hybrid was also able to inhibit the polymerization of β -hematin, a property of chloroquine. Similar results have been obtained in the case of artemisinin-primaquine phosphate (artemisinin-Primaquine [PQ]) and stilbene-chalcone (30, 31) hybrids.

Pharmacokinetics. The pharmacokinetic properties of a hybrid are easier to predict and hence to manipulate than those of the two individual drugs. Therefore, problems pertaining to pharmacokinetics, metabolic stability, or side effects of individual molecules are rectified in the form of a hybrid, as the entire drug molecule might be toxic in a few cases, but its pharmacophore might not necessarily be as toxic. If a toxicophore fragment of a given drug molecule does not overlap the pharmacophore, then it may be possible to redesign the molecule (32). For example, thalidomide derivatives devoid of teratogenic effects have been developed for treating chronic inflammation employing hybridization techniques, based on studies demonstrating that the teratogenicity is due to its distinct toxic subunit, glutarimide (33). PQ, an 8-aminoquinoline, has serious side effects and is known to induce hemolysis, especially in glucose-6-phosphate dehydrogenase (G6PD)-deficient individuals. However, Vangapandu and coworkers (34) demonstrated that 8-quinoline amine conjugates as well as the corresponding "double prodrugs" had promising in vivo activity in mice. If these compounds, in which the basic pharmacophore is primaguine phosphate, were modified to improve their blood schizonticide activity, they would have the capability to be employed as broad-spectrum (tissue and blood schizonticides) antimalarial agents.

Stability. Joubert et al. (35) employed the differential scanning calorimetry (DSC) and thermogravimetric analysis (TGA) techniques to ascertain whether the addition of



FIG 2 Chemical structures of (A) reversed chloroquine (RCQ), as proposed by Burgess et al. (36), and (B) chloroquine and (C) imipramine (RA), employed for its synthesis.

9-aminoacridine moiety would impart stability to the otherwise unstable artemisinin pharmacophore. Affirming the stability of potential antimalarial compounds during the initial stages of development is of utmost importance, so as to ensure that the compound will be able to tolerate the extreme conditions present in regions of malaria endemicity. The hybrids that emerged in their study were extremely stable, with minimum weight losses. It is clearly evident from that study that the bulkiness and rigidity of the acridine ring imparted stability to the complete hybrid structure.

Incorporation of adducts. Forming hybrids allows grafting of suitable inhibitors of parasite resistance or water-soluble molecules to overcome poor solubility. For example, the first attempt to counteract chloroquine resistance in *P. falciparum*, by blocking its export from the parasite digestive vacuole (DV), involved linking a CQ-like moiety to a reversal agent (RA) via an alkyl linker (36). A number of calcium channel blockers, such as nifedipine and verapamil, and their derivatives, such as imipramine, dibenzylmethylamines, primaquine phosphate, and dihydroanthracenes, are known to restore sensitivity to chloroquine in resistant strains (37-40). Certain RAs are known to inhibit P. falciparum digestive vacuole membrane protein PfCRT (P. falciparum chloroguine resistance transporter)-associated export of chloroquine from its site of action in the DV. This occurs as a consequence of mutations in PfCRT. It was hypothesized that such a hybrid molecule would deliver the quinoline moiety and the RA in a 1:1 ratio, lowering the dose of reversal agent required in comparison to the dose that would be needed in cases in which the two components would be administered individually, making the effective dose much lower. The hybrid was termed a "reversed chloroquine" (Fig. 2), and although it was effective against chloroquine-resistant parasites both in vitro and in vivo, it did not move into further development on account of being highly lipophilic.

Multistage antimalarial strategy. Most antimalarial drugs in the current treatment strategies primarily target the erythrocytic stages of the malaria parasite in the human blood system. But to ensure malaria eradication, new drugs are urgently needed that restrict transmission of the parasite between the human host and the mosquito vector and that eliminate the parasite in its various stages during its cycle in the human body. The first study based on the covalent combination of molecules acting on different stages of the parasite life cycle was conducted by Capela et al. (30). The work describes the synthesis of hybrid molecules containing PQ and artemisinin (ART) pharmacophoric units, and their efficacies against Plasmodium hepatic and erythrocytic stages, both in vitro and in animal models of malaria. PQ is the only drug approved against liver stages of Plasmodium, including parasites acutely infecting the liver and hypnozoites. None of the other approved drugs reliably clear hypnozoites. PQ is also effective against sexual stages, i.e., the gametocytes, thereby disrupting the transmission of infection to mosquitoes. Therefore, PQ is administered in combination with a therapeutic agent that acts against blood-stage parasites. This strategy is aimed at reliably curbing infections with P. vivax or P. ovale and thereby preventing relapses due to the development of subsequent blood-stage infections from hypnozoites.

Miranda et al. (18) reported a series of hybrid compounds combining either a 1,2,4-trioxane or 1,2,4,5-tetraoxane and 8-aminoquinoline moieties. The hybrids were synthesized and screened for their antimalarial activity.

Both those studies indicated that peroxide-8-aminoquinoline hybrids can serve as promising lead compounds to develop potent agents that possess all the desired antimalarial multistage activities in a single chemical entity that may emerge as drugs of choice in malaria elimination campaigns.

Commercial aspects. Goals such as cost-effective production (leading to greater chances of reaching the masses), better patient compliance, and freedom from patent restrictions, if achieved, would contribute to the path of hybrid compounds in emerging as the next-generation antimalarials.

WHICH ARE MORE EFFICACIOUS: HYBRIDS OR COMBINATIONS?

Although interesting from a chemical biology perspective, the same antimalarial potency of hybrid molecules might be achieved with combination therapy. The prime force driving the concept of hybrids is their potentially higher efficacy than that of either of the parent compounds administered as monotherapy or as a fixed-dose combination. This would account for all the efforts put into designing of the hybrid molecule rather than simply formulating a mixture of the drugs. However, it is a matter of concern that the majority of the published studies have neglected the fundamental control experiment, i.e., comparison of the activity of the hybrid with that of a 1:1 ratio combination of the individual drugs. Most of the published studies on design and synthesis of antimalarial hybrids involving different pharmacophore components, starting from 2001 and continuing until the present, have been listed in Table 1. As seen from the table, among the various classes of hybrids synthesized so far, viz., endoperoxide-quinoline-based hybrids, endoperoxide-chalcone-based hybrids, etc., only a few studies have reported such a comparison to date. The advantage of the covalent linkage of individual functional moieties for antimalarial activity over their combination has been proven experimentally and documented in the form of published literature in only a minute number of studies for cultured parasites and in no cases for animal models. The hybrid may actually be less efficacious than the combination of the individual constituents. Here, one should consider the fact that the strategy employed in forming a hybrid would add bulk to the molecule, which might hinder its passage into the parasite cell, which in turn would lead to reduced penetration and suppressed activity. Pretorius et al. (27) investigated this issue to ascertain the advantages of hybrids over equimolar combinations, in terms of antimalarial activity. They concluded that there were none. The most potent hybrid in their study was as effective as its two components, chloroquine and pyrimethamine, against strain D10 and only slightly superior to chloroquine alone against strain Dd2. Those results prove that all the possible partners should be thoroughly investigated and chosen for any hybrid antimalarial project and that the advantage(s) of the hybrid over the combination needs to be proven expeditiously. Taking into consideration the current preference for combination therapy in the treatment of malaria, the hybrid drug would have to compare favorably not only to the known single agents but also to the combinations in use.

NEW METABOLIC LIABILITIES THAT ARISE BY MOLECULAR HYBRIDIZATION

Hybridization can lead to loss or gain of favorable absorption, distribution, metabolism, excretion, and pharmacokinetics (ADME/PK) properties of the individual pharmacophoric moieties. Thelingwani et al. (83) addressed the various metabolic challenges that arise in covalently linking two active pharmacophores by characterizing artemisinin-chloroquinoline hybrids (47) and concluded that though the technique combines the desirable properties, certain unfavorable properties are also carried along in the process and need to be accounted for. The experiment investigating metabolic stability and metabolite identification showed that the hybrids were not extensively metabolized, with the major amount of the parent compound remaining unchanged after 1 h of incubation in hepatocytes. As a result, the proportions of the detected metabolites were very low compared to those of the parent hybrid molecule. Contrastingly, in another study by the same group, metabolism of 4-aminoquinoline-3-

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TABLE 1

				Activity with respect to		
Component 1	Component 2	Activity in culture (IC ₅₀)	Activity in animal model	components 1 and 2 in culture (component no.)	Activity with respect to mixture (1:1) in culture	Reference
Aminoquinoline	Trioxane	\sim 40 to 48 nmol/liter	QN	DN	ND	41
Aminoquinoline	Trioxane or trioxolane	22 to 536 nM	ED ₅₀ , CD ₅₀ , 0.7 to $>$ 25 mg/kg/day	ND	ND	42
Aminoquinoline	Aminopeptidase inhibitor	150 nM to 1.4 μ M	ND	ND	ND	43
Aminoquioline	Hexanoic acid derivative	107 to 259 nM	ND	ND	ND	44
Primaguine	Statine	0.4 to 6.2 mM	QN	DN	ND	45
Aminoquinoline	Isatin derivative	33 nM to 1.5 µM	ND	ND	ND	46
Trifluoromethyl-artemisinin	Mefloquine	2.4 to 17.2 nM	Effective at 35 μ mol/kg/day	Higher (2)	ND	47
Primaquine	Statine (PS777621, plasmepsin inhibitor)	0.1 to 4.5 μ M	ND	Higher (2)	ND	48
Chloroguine	Imipramine	2.9 to 5.3 nM	Effective at 64 ma/ka/dav	Hiaher (1)	ND	36
Ferroquine	Thiosemi-carbazone	0.1 to 96 μ M	ND	Lower (1)	ND	49
Artemisinin	Quinine	9.0 to 10 nM	ND	Higher (1 & 2)	Higher (3-fold)	28
Aminoquinoline	Trioxolane	4.0 to 32.0 nM	$ED_{50'}$ $CD_{50'}$ 0.7 to >25 mg/kg/day	Higher	Higher	29
Aminoquinoline	Tetraoxane	2.0 to 33 nM	Most effective at 320 mg/kg/day, partly effective at 80 mg/kg/day	Higher (1)	ND	50
Aminoguinoline	Triazine	4.4 to 256 ng/ml	Partly effective at 50 mg/kg/day	Lower (1)	ND	51
Atovaquone and derivatives	Statine	0.61 to 5.3 nM	ND	ND	ND	52
Aminoquinoline/aminoacridine	Clotrimazole derivatives	3.0 nM to 1.4 μ M	Some compounds effective at 50 mg/kg/day; ED ₅₀ of one	Variable with strains (1), higher (2)	ND	53
			compound, 6.3 mg/kg/day			
Chloroquine	Astemizole	0.02 to 0.61 μ M	Effective at 80 to 200 mg/kg	Higher (1), higher (2)	ND	54
Artemisinin	Dipeptidyl vinyl sulfone	2.08 to 4.81 nM	ND	Higher (1), higher (2)	ND	55
Chalcone	Thiolactone/Isatin	0.68 to 14.9 μ M	ND	Higher (2)	ND	56
Artemisinin (1,2,4-trioxane pharmacophore)	Ferroquine (ferrocenel-quinoline)	16 to 43 nM	10 mg/kg/day	Comparable to artemisinin, ferroquine, & trioxaquine; hicher than chlorocuine	DZ	57
Artemisinin	Primaguine	9.1 to 12.5 nM	9 ma/ka/dav	Lower (1). higher (2)	DN	30
Chloroquine	Thiazolidinone	0.02 to $>$ 22 μ M	Two compounds effective at 10 mg/kg/day	Variable with strains (1)	ND	58
Artemisinin/naphthoquinone	Aminoquinoline	0.02 to 0.64 μ M	ND	Comparable (1), comparable (2)	ND	59
Stilbene	Chalcone	1.4 to 6.4 µM	ND	Higher (1), higher (1)	Higher	31
Chloroquine	Chalcone	29 to 314 nM	ND	Variable with strains (1)	ND	60
Coumarin	Trioxane	39 to >500 ng/ml	One compound partly effective at 96 mg/kg/dav	DN	ND	61
Chloroquine	Chalcone	3.6 to 380 ng/ml	Two compounds effective at	Lower (1)	ND	62
Amodiaguine	Phenyl furoxan	0.630 mM	nou mig/kg/day ND	Hiaher	ND	63
Chloroquine	Primaquine	0.08 to 0.64 μ M	Effective at 60 mg/kg/day	Variable with strains (1 & 2)	Higher (strain K1), lower	64
					(strains 3D7 & Dd2)	
Artemisinin	Quinoline	5.0 to 21.5 nM	ED ₅₀ , <0.8 to 1.4 mg/kg/day	Lower (1), higher (2)	DN ND	65 27
Aminoquinoline	I riazine	0.1/ to 1./ hM	UN	Variable with strains (1)	UN	00
Tetraoxane	Dipeptidyl vinyl sulfone	10.7 to 175 nM	QN	Variable with strains (1)	QN	67
Chioroquine Ouinoline	Hydroxy-pyndone Triazole	0.004 to /1.6 µJM 14 to 109 JJM		Higher (1) Lower (1)		69 69
Artemisinin	Triazine	5.5 to 85 nM	QN	Comparable (1), higher (2)	Lower (strain NF54),	70
					comparable (strain Dd2)	
Artemisinin	Acridine	2.6 to 430 nM	ND	Lower (1)	ND	35
					(Continued on fo	llowing page)

TABLE 1 (Continued)						
Component 1	Component 2	Activity in culture (IC ₅₀)	Activity in animal model	Activity with respect to components 1 and 2 in culture (component no.)	Activity with respect to mixture (1:1) in culture	Reference
Quinoline	Diarylether	20.3 to >1,200 nM	ND	Lower (1)	ND	71
Tetraoxane	Pyrimidine	9.8 to 81.2 nM	ND	ND	ND	72
Trioxane	Egonol and/or ferrocene	0.3 to 88 nM	ND	Higher (1), lower (2)	ND	73
Quinoline	Pyrimidine	22 to 4,310 nM	ND	Lower (1), lower (2)	ND	74
Artesunate	Indoloquinoline	0.4 to 3.2 nM	One compound effective at 40 mg/kg/day	Higher (1), higher (1)	ND	75
Chloroquine	Hydroxi-pyridone	0.04 to 0.46 μ M	Partly effective at 8 mg/kg/day	Variable with strains (1)	ND	76
Aminoquinoline	Pyrazolopyrimidine	0.005 to 1.6 μ M	•	Variable with strains (1)	ND	77
Primaquine	Pyrimidine	1.93 to 6.46 µM	ND	Lower (1)	ND	78
Quinoline	Sulfonamide	0.05 to 1.63 µM	Partly effective at 10 mg/kg/day	Higher (1), higher (1)	ND	79
Trioxane	Ferrocene	7.2 to 30.2 nM	DN	Lower (1), higher (2)	ND	80
Aminoquinoline	s-Triazine	0.22 to 8.3 μ M	ND	Higher (1)	ND	81
Aminoquinoline	Atorvastatin	0.4 to 6.39 $\mu { m M}$	ND	Higher (1), higher (2)	ND	82
^a CD ₅₀ , 50% cytotoxic dose; ED	50, 50% effective dose; GR, glucocortic	oid receptor; kg, kilograms of bo	dy weight; ND, not determined.			

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hydroxypyridin-4-one hybrids was predicted to be extensive in cryopreserved human hepatocytes (mainly via the linker chain), leading to formation of various primary and secondary metabolites (84).

The long half-life property of chloroquine was lost in both artemisinin-chloroquinoline and 4-aminoquinoline-3-hydroxypyridin-4-one hybrids, which in turn behaved more like the other pharmacophore present (artemisinin in the first case) by displaying a short half-life. Intermediate to fast clearance in hepatocytes signifies that their effective regimen should be dosing more than once daily in the treatment or prophylaxis of malaria. The most potent compounds were the fastest cleared. Therefore, the strategy for molecular hybridization should focus on designing more metabolically stable hybrids, thereby achieving the desired regimen of dosing once a day.

Additionally, the artemisinin-chloroquinoline hybrid compounds were observed to inhibit the enzymes involved in their own metabolism, the cytochrome P450 (CYP) enzymes. Hence, they inhibited their own metabolism and that of other compounds as well that share a common clearance mechanism. This gives rise to a liability because of the risk of drug-drug interactions. Although the parent compounds have been reported to inhibit a few CYP enzymes, a striking difference here is the inhibition of all the five isoforms by the hybrids (CYP1A2, CYP2C9, CYP2C19, CYP2D6, and CYP3A4). The 50% inhibitory concentration (IC₅₀) for CYP3A4 inhibition was noted to be less than 1 μ M, further substantiating the risk for drug-drug interactions. Intermediate risk of interactions with the substrates of CYP3A4 and CYP2D6 was observed for 4-aminoquinoline-3-hydroxypyridin-4-one hybrids, with IC₅₀s ranging from 1 to 10 μ M. Hence, hybridization, instead of masking an undesirable property of the parent compounds, led to broadening it.

Joubert et al. (35) observed unfavorable drug-likeness properties in artemisininacridine hybrids. The hybrids showed extremely low solubility and absorption levels under physiological conditions and undefined blood-brain barrier (BBB) penetration levels, which they proposed to result from the blocking of the polar primary/secondary amine groups in the intermediates, thereby leading to a reduction in the formation of hydrogen bonds. Another possible reason could be the presence the lipophilic artemisinin moiety. Also, the hybrids were unable to serve efficaciously against the chloroquineresistant *P. falciparum* Dd2 strain, although dihydroartemisinin was active against the same strain. Hence, one can infer that hybridization of the artemisinin and acridine moieties resulted in an antagonistic interaction and that the presence of acridine rendered artemisinin ineffective against the Dd2 strain.

Ferrocene-pyrimidine conjugates were also reported by Chopra et al. (85) to exhibit moderate to poor aqueous solubility, and the lack of hydrophilicity was distinctly reflected in their average levels of *in vitro* antiplasmodial activity against CQ⁵ NF54 strain. The hybrid with the lowest IC_{50} was the one with the highest hydrophilicity. That study proved that water solubility should be regarded as an essential property of newly synthesized anyimalarial hybids to ensure adequate absorption and plasma concentrations.

SOME METABOLIC LIABILITIES PERSIST AFTER MOLECULAR HYBRIDIZATION

It has been proposed that the side effects manifested by individual drugs may be masked when they are linked covalently in a hybrid. Contrastingly, some hybrids display the same metabolic liabilities as are exhibited by their constituents. For example, the 4-aminoquinolines, which have been drugs of choice for various antimalarial hybrid development programs, have been implicated in the occurrence of clinically significant cardiovascular effects (86). They cause significant prolongation of the electrocardiograph QT interval, raising the risk for fatal ventricular arrhythmias such as "torsades de pointes" and sudden cardiac death (87). When injected rapidly, chloroquine is potentially hypotensive (88).

Very low solubility and absorption levels of artemisinin-acridine hybrids have been attributed to inherent pharmacokinetic limitations of artemisinin, *viz.*, poor water solubility, absorption, and plasma bioavailability (35). Hepatotoxicity of acridine and neurotoxicity of dihydroartemisinin are also well reported (89, 90). The process could neither reduce the toxicities associated with the acridine or artemisinin moieties nor overcome the resistance of parasites to chloroquine.

LINKER SELECTION

In the design of hybrid drugs for malaria, very little attention has been given to the relative proximity of the reputed cellular targets and the distance between the two components of the hybrid. Although the data may be difficult to obtain in practice, they nevertheless should be considered in the design of hybrid drugs, especially in situations where a metabolically resistant linker unit is employed.

Lombard et al. (65) have devised a number of strategies to covalently link quinolines and artemisinins, making use of different linkers. Their hybrids displayed activity either similar to or higher than that of chloroquine against CQ-sensitive *P. falciparum* strains and activity greater than that of chloroquine against CQ-resistant *P. falciparum* strains. The results of all their studies indicated that cyclic linkers should be avoided as they contribute to decreased antiplasmodial properties and that the length of the linker should be curtailed to two or three carbon atoms. Hybrids with a linker chain length of greater than three were found to be less potent than chloroquine. The members of the O'Neill group, which worked on synthetic 1,2,4-trioxolaquines (91), and several others (62, 92), have reported similar results. Therefore, the length and nature of the linker exert a strong influence on the antimalarial efficacy of the conjugates.

It is noteworthy that, in contrast with the studies described above, another study to do with linking astemizole derivatives with an aminoquinoline via a piperazine/aminopiperidine linker concluded that hybrids with conformationally constrained cyclic linkers also exhibited potent activity against a CQ-resistant K1 strain (54). Similarly, the two most potent hybrids from a series of quinoline-pyrimidine hybrids evaluated by Pretorius et al. (27) contained rigid aromatic and piperazine linkers. This indicates that the flexibility of the linker between the two pharmacophores does not govern the activity of these compounds.

CONCLUDING REMARKS

The potential advantages of antimalarial hybrids, such as lower toxicity, better pharmacokinetics, and easier formulation, have yet to be realized. The paucity of information about pharmacokinetics, pharmacodynamics, and rational dosing of drugs represents a critical knowledge gap that needs to be addressed in order to use current drugs in conjunction with other tools to reduce malaria transmission, as well as to provide rationally designed treatment strategies. The hybrid approach is interesting in itself but likely no more so than the others. However, it can take a substantial position in the strategy for searching for new antimalarials. It is important to indicate, that instead of exhibiting potential promising pharmacological interest, no antimalarial hybrid drug is currently either in development and or at a preclinical step in this portfolio. From over 100 trioxaquines tested so far, the trioxaquine PA1103/SAR116242 was selected by Palumed in collaboration with Sanofi-Aventis as a drug development candidate for treatment of uncomplicated malaria as the first "fusion" antimalarial (93). However, it no longer appears in the global malaria portfolio as it was abandoned in preclinical development. Our opinion is that for any hybrid antimalarial project, the choice of partners needs careful justification, and the advantage(s) of the hybrid over the combination needs to be proven at the earliest opportunity. Before going ahead with designing a hybrid, both the benefits and demerits of the product, and the strategy employed, should be thoroughly investigated and the hybrid candidate should then be taken a level up only when the advantages overshadow the drawbacks.

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