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Catalytic Promiscuity and Heme-dependent Redox Regulation of H₂S Synthesis

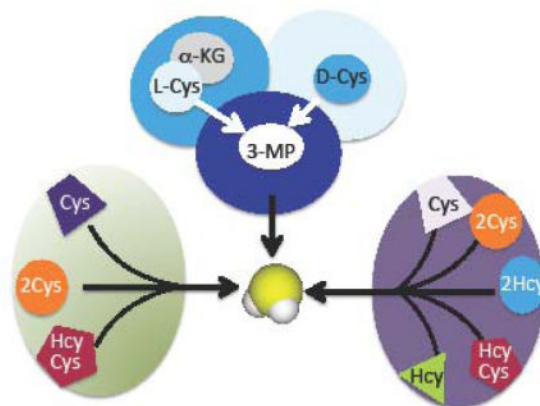
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Abstract

The view of enzymes as punctilious catalysts has been shifting as examples of their promiscuous behavior increase. However, unlike a number of cases where the physiological relevance of breached substrate specificity is questionable, the very synthesis of H₂S relies on substrate and reaction promiscuity, which presents the enzymes with a multitude of substrate and reaction choices. The transsulfuration pathway, a major source of H₂S, is inherently substrate-ambiguous. A heme-regulated switch embedded in the first enzyme in the pathway can help avert the stochastic production of cysteine versus H₂S and control switching between metabolic tracks to meet cellular needs. This review discusses the dominant role of enzyme promiscuity in pathways that double as sulfur catabolic and H₂S synthetic tracks.

Graphical Abstract



Enzymes can exhibit considerable laxity in both substrate and reaction specificity, contributing to the growing view that promiscuity and fidelity coexist in biocatalysts [1]. An evolutionary advantage of a specificity cushion is that it affords a latent functional repertoire

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that is broader than the genome encoding it, and provides an adaptive advantage under pressure for the emergence of new catalytic functions as seen in both natural and laboratory settings [2–4]. Enzymes can exhibit promiscuous behavior toward xenobiotic substrates or towards naturally occurring metabolites. The term “underground metabolism” was coined to refer to the stream of secondary metabolic activity with endogenous substrates that is generally invisible due to low flux but might be phenotypic under certain conditions [5]. At one extreme of the specificity spectrum are enzymes involved in DNA replication, which operate with low albeit nonzero error rates, their ability to slip up and introduce mutations being advantageous from an evolutionary perspective. At the other end of the spectrum, are enzymes involved in some amino acid metabolism pathways such as the ones shared for cysteine and H₂S synthesis as discussed in this review.

It is posited that ancient enzymes were generalists with broad specificity and that metabolic pathways were inherently leaky [6]. Enzyme promiscuity and underground metabolism play a surprisingly prominent role in multiple facets of H₂S synthesis [7–9]. The enzymes involved in H₂S biogenesis are distinct from the highly specific nitric oxide synthases and heme oxygenases, dedicated to synthesizing the other two gaseous signaling molecules, NO and CO, respectively. Also in striking contrast to NO and CO synthesis, three unrelated enzymes support H₂S synthesis of which two, serve alternative metabolic functions (Fig. 1A). Cystathionine β -synthase (CBS) and γ -cystathionase (CSE) comprise the cytoplasmic transsulfuration pathway that functions to direct homocysteine derived from methionine to cysteine synthesis, particularly under conditions of sulfur excess [10]. The third enzyme, β -mercaptopyruvate sulfurtransferase (MST) resides in the cysteine catabolic branch of the sulfur network and is both cytoplasmic and mitochondrial [11]. The reactions catalyzed by these H₂S synthesizing enzymes and their regulation, are discussed in this review.

H₂S Synthesis via the Transsulfuration Pathway

Parallel tracks within the transsulfuration pathway lead to cysteine synthesis from serine and homocysteine and to H₂S synthesis from cysteine and homocysteine (Fig. 1A). The first enzyme in the pathway, cystathionine β -synthase (CBS) catalyzes the β -replacement of serine and homocysteine eliminating water and forming cystathionine. The latter is a substrate for γ -cystathionase (CSE), which catalyzes its α - γ elimination to cysteine, α -ketobutyrate and ammonia. In this configuration of the transsulfuration pathway, sulfur is fated for transfer from homocysteine to cysteine. However, CBS [12] and CSE [13] exhibit both substrate and reaction ambiguity (Fig. 2A, B). Thus, CBS can swap cysteine for serine eliminating H₂S while still forming cystathionine in the presence of homocysteine (Fig. 1A). It can also generate H₂S from one or two moles of cysteine (Fig. 2A). Of the three routes for CBS-catalyzed H₂S-production, the dominant one is β -replacement of cysteine by homocysteine [12].

CSE, the second enzyme in the transsulfuration pathway, exhibits even greater promiscuity than CBS. In addition to the three H₂S generating reactions that it catalyzes in common with CBS, it also produces H₂S from one or two moles of homocysteine (Fig. 2B). The major routes for CSE-catalyzed H₂S generation are via α - β elimination of cysteine to form pyruvate and ammonia and by α - γ elimination of homocysteine forming α -ketobutyrate and

ammonia. The former reaction is favored at physiologically relevant substrate concentrations [12]. In addition to H₂S, the transsulfuration enzymes catalyze the synthesis of persulfides from homocystine (CSE only) and cystine (CBS and CSE), the oxidized forms of the respective amino acids (Fig. 2A, B) [14–16]. In the reducing intracellular milieu, cystine and homocystine concentrations are low and the persulfide-generating reactions are predicted to be quantitatively insignificant [15]. However, under oxidizing conditions, these reactions might become significant.

The first step in the CBS and CSE catalyzed reactions involves formation of an external aldimine with the incoming amino acid. While CBS forms a Schiff base with serine or cysteine, CSE can also accommodate homocysteine, with an extra methylene group at this position (Fig. 1B), explaining the wider range of reactions that it catalyzes [12]. H₂S synthesis by CSE is responsive to the grade of homocystinuria [13], a metabolic disorder characterized by elevated homocysteine [17]. The physiological relevance of homocysteine-derived H₂S is supported by elevated homolanthionine in homocystinuric patients [18,19]. Homolanthionine is a side product of H₂S-generation via condensation of two moles of homocysteine (Fig. 2B).

Human CBS does not discriminate between serine and cysteine at the level of the respective specificity constants, which are virtually identical ($k_{\text{cat}}/K_{\text{m}(\text{Cys})} = 2.9 \times 10^3 \text{ M}^{-1}\text{s}^{-1}$ versus $k_{\text{cat}}/K_{\text{m}(\text{Ser})} = 2.7 \times 10^3 \text{ M}^{-1}\text{s}^{-1}$ at pH 7.4 and 37 °C) [12]. However, the K_{d} for serine is ~7-fold lower than for cysteine [12]. Human CSE exhibits a preference for cystathionine ($k_{\text{cat}}/K_{\text{m}(\text{Cyst})} = 8 \times 10^3 \text{ M}^{-1}\text{s}^{-1}$) over cysteine ($k_{\text{cat}}/K_{\text{m}(\text{Cys})} = 0.3 \times 10^3 \text{ M}^{-1}\text{s}^{-1}$) or homocysteine ($k_{\text{cat}}/K_{\text{m}(\text{Hcy})} = 0.4 \times 10^3 \text{ M}^{-1}\text{s}^{-1}$) [13]. Both CBS and CSE exhibit high K_{m} values for cysteine and homocysteine (2–7 mM) that are 10–100 fold higher than the intracellular concentrations of these substrates in most tissues [12,13]. The corresponding enzymes in other organisms also exhibit high K_{m} values for their substrates [20]. It is not known if small molecule modulators or supramolecular organization of pathway enzymes as seen in purinosomes [21] influence the affinity of the transsulfuration enzymes for their substrates or their kinetic efficiencies in vivo.

H₂S Synthesis via Cysteine Catabolism

The conversion of cysteine to H₂S via the cysteine catabolic pathway occurs in two steps catalyzed by a transaminase and by MST (Fig. 1A) [22,23]. Aspartate aminotransferase (AAT) is a notoriously promiscuous pyridoxal phosphate-dependent enzyme that catalyzes the transamination reaction between pairs of amino and keto acids (Fig. 2C). It catalyzes a cysteine aminotransferase (CAT) reaction in which aspartate is substituted with cysteine forming 3-mercaptopyruvate. The promiscuity of CAT/AAT is further demonstrated by its ~10-fold higher activity under V_{max} conditions with cysteine sulfinic acid (Fig. 2C) than with aspartate [24]. Cysteine sulfinic acid is the product of cysteine dioxygenase, which is also involved in cysteine catabolism [10].

In the absence of a known mechanism for regulating substrate selectivity, the reaction choice for CAT/AAT is presumably determined by a combination of substrate concentrations and the relevant specificity constants ($k_{\text{cat}}/K_{\text{m}}$). Mitochondrial and cytoplasmic isoenzymes of

AAT/CAT exist and despite the wealth of structural and mechanistic information on them, direct comparison of the kinetic parameters for the competing reactions at physiologically relevant pH, are not readily available. The K_m values for the mitochondrial rat liver CAT/AAT are 22 mM for cysteine and 0.5–1.6 mM for aspartate at pH 9.7 [23] and the specificity constants are estimated to be $1.4 \times 10^4 \text{ M}^{-1} \text{ s}^{-1}$ (aspartate) and $1.4 \times 10^2 \text{ M}^{-1} \text{ s}^{-1}$ (cysteine) at pH 9.7 and 37 °C [25]. At pH 7, the CAT activity is ~10-fold lower than at its optimal pH of 9.7 [25]. The CAT activity is potently inhibited by aspartate [23,25], which is more abundant than cysteine in most tissues. Despite the kinetic parameters favoring AAT over CAT activity, the physiological relevance of the CAT reaction is borne out by the accumulation of mercaptolactate disulfide in individuals with a genetic deficiency of MST [26]. Mercaptolactate is the product of lactate dehydrogenase-catalyzed reduction of mercaptopyruvate.

An alternative route to 3-mercaptopyruvate is via the oxidative deamination of D-cysteine catalyzed by the flavoprotein, D-amino acid oxidase, which is yet another promiscuous enzyme [27,28]. While its “physiological” substrate is presumed to be D-serine, it exhibits substantial or even higher activity with other D-amino acids [29]. For the human enzyme, the k_{cat}/K_m for D-serine is $0.4 \times 10^3 \text{ M}^{-1} \text{ s}^{-1}$ at pH 8.5 and 25 °C [30] while the corresponding value for D-cysteine is not known. D-amino acid oxidase is a peroxisomal enzyme while MST is predominantly mitochondrial. Hence, the contribution of this pair of enzymes to H_2S generation in intact cells and the source of D-cysteine are not known.

MST also exhibits promiscuity utilizing either 3-mercaptopyruvate or thiosulfate as substrate (equations 1,2). MST catalyzes a sulfurtransferase reaction forming an enzyme-bound persulfide intermediate, which subsequently donates the sulfane sulfur atom to an acceptor e.g. cyanide (equation 3). The physiological sulfur acceptor is predicted to be thioredoxin [31,32].



The K_m values of rat liver MST for mercaptopyruvate (1.2 mM) and thiosulfate (62 mM) are vastly different. Furthermore, their specificity constants ($5.6 \times 10^3 \text{ M}^{-1} \text{ s}^{-1}$ for mercaptopyruvate (pH 9.55 and 25 °C) and $1.5 \times 10^2 \text{ M}^{-1} \text{ s}^{-1}$ for thiosulfate (pH 5.0 and 25 °C)) are difficult to compare given the difference in the assay conditions [33]. The $k_{\text{cat}}/K_m(3\text{-MP})$ for human MST with mercaptopyruvate as donor and thioredoxin as acceptor is $3.7 \times 10^3 \text{ M}^{-1} \text{ s}^{-1}$ at pH 7.4 and 37 °C [31]. In human MST, Arg188, Arg197 and Ser250

make contacts with the carbonyl and carboxyl oxygens of mercaptopyruvate and are important determinants of selectivity against thiosulfate [33].

Heme-dependent Metabolic Switching

Some cellular strategies for averting the potentially adverse effects of inherently lax substrate specificity are regulation of protein expression levels, limiting active site access via substrate or product inhibition and metabolite repair [34,35]. In the transsulfuration pathway, the single or combinatorial use of even a limited number of amino acids creates a multitude of reaction choices for CBS and CSE (Fig. 2A, B), which must be regulated to service cellular needs for cysteine versus H₂S synthesis. CBS is poised at a key metabolic decision point where the choice between recycling or transmuting homocysteine is made. Hence, CBS is a major hub of regulation; it is allosterically stabilized [36] and activated by S-adenosylmethionine (AdoMet), and by glutathionylation [37], but inhibited by CO [38–41], NO [42,43], nitrite [44] and by SUMOylation [45]. Human CBS comprises an N-terminal regulatory domain that houses a heme [46,47] and a C-terminal domain that has a tandem repeat of CBS domains, a secondary structure motif that is often utilized in energy sensing modules [48].

In human CBS, Cys52 and His65 serve as ligands to the heme in the ferric and ferrous states (Fig. 1C). Although located ~20 Å from the catalytic site where the PLP cofactor is housed, the heme exerts long-range allosteric effects [49]. Binding of NO or CO to the ferrous heme results in the formation of 5- and 6-coordinate species, respectively in which Cys52 or both endogenous ligands are displaced (Fig. 1C). The heme also exhibits catalytic activity reducing nitrite to NO and forming the inhibitory ferrous-nitrosyl complex [44]. Changes in the native heme environment are communicated to the PLP pocket [49–51] and results in a shift in the tautomeric equilibrium from the active ketoenamine to the inactive enolimine form [52]. Inhibition of CBS by CO and NO is readily reversed in the presence of oxygen, which rapidly oxidizes ferrous CBS [53].

The heme in CBS is a key operator that can switch the transsulfuration pathway between the cysteine and H₂S production tracks [54]. When the heme is coordinated by endogenous ligands, synthesis of cystathionine via the canonical reaction is favored due to the higher intracellular concentration of serine and its higher affinity for CBS versus cysteine. Cystathionine ($K_m = 0.28$ mM) in turn, competes with cysteine ($K_m = 1.7$ mM) and homocysteine ($K_m = 2.7$ mM) for CSE resulting in the transsulfuration pathway operating in the canonical cysteine-producing track (Fig. 3). Under conditions that induce nitric oxide synthase or heme oxygenase, e.g. ER stress [55] or inflammation [56], enhanced production of NO or CO could lead to ferrous nitrosyl or ferrous carbonyl CBS, which are inactive. Consequently, homocysteine levels rise and cystathionine levels fall, promoting H₂S synthesis by CSE (Fig. 3). As NO or CO levels drop, or the ferrous heme in CBS is oxidized, the transsulfuration pathway switches back to the cysteine track. Both the transsulfuration pathway and transporters feed the cysteine pool and conditions such as ER stress enhance cysteine import [57]. In liver, where the transsulfuration pathway is best characterized, CSE is estimated to account for ~97% of H₂S produced at physiologically

relevant substrate concentrations and taking into account differences in the protein levels of CBS and CSE [58].

Other strategies for controlling H₂S production also exist in cells including substrate level activation, posttranslational modification of CBS [37,59] and CSE [60] and regulation of protein levels of the transsulfuration pathway enzymes. While CSE is more abundant than CBS in liver and kidney [58], CBS predominates in brain [61].

Conclusions

The rampant promiscuity of enzymes involved in mammalian H₂S synthesis is not surprising from an evolutionary perspective. CSE and CBS orthologs in lower organisms condense cysteine and H₂S (or thiosulfate) with O-phosphohomoserine and O-acetylserine forming cystathionine and cysteine (or sulfocysteine), respectively [20,62,63]. While metabolic regulation during evolution has resulted in reversal of the transsulfuration pathway from sulfur assimilation in lower organisms to dissimilation in higher organisms, lax substrate specificity in the pathway enzymes has endured. Other enzymes such as AAT/CAT and D-amino acid oxidase, which feed MST-dependent H₂S synthesis, are inherently broad specificity enzymes. The heme-regulated metabolic track switching discussed here is one strategy for regulating the multipurpose enzymes involved in H₂S biogenesis; other strategies must exist and remain to be identified. Whether the use of small molecule regulators for switching enzyme specificity and redirecting flux might be a strategy deployed by other metabolic pathways remains to be elucidated.

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Highlights

- This review highlights the prevalence of promiscuity not only in the enzymes that synthesize hydrogen sulfide, a signaling molecule, but also in the metabolic pathways in which they reside.
- The role of heme-dependent metabolic track switching as a mechanism of regulating flux between competing pathways is discussed.

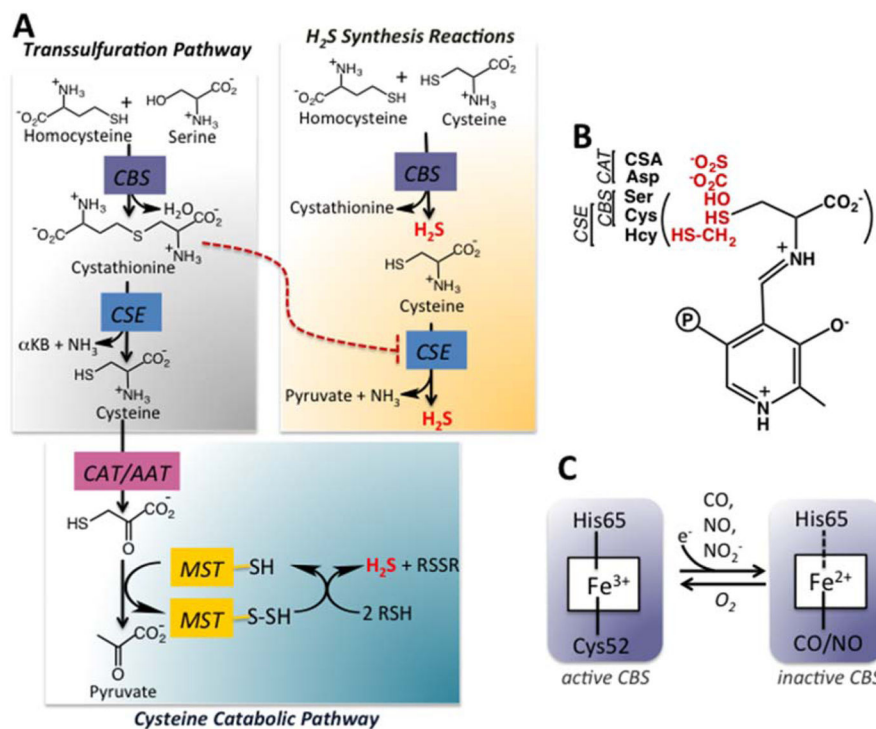


Figure 1. Overview of H₂S synthesizing reactions. **A.** H₂S can be synthesized by the transsulfuration pathway enzymes, CBS and CSE or by the cysteine catabolism pathway enzymes, CAT/AAT and MST. The canonical transsulfuration reactions catalyzed by CBS and CSE results in the conversion of serine and homocysteine to cysteine. However, these enzymes can also utilize cysteine and homocysteine to generate H₂S. Cystathionine, an intermediate in the canonical transsulfuration pathway competes with cysteine for binding to CSE, thus inhibiting H₂S synthesis (red dotted line). MST is a sulfurtransferase, which catalyzes the transfer of the sulfur atom from mercaptopyruvate to an active site cysteine thiol to form a cysteine persulfide. The latter, in the presence of reductants can release H₂S. αKB denotes α-ketobutyrate. **B.** The first step in the reactions catalyzed by CBS, CSE and CAT/AAT is the formation of an external aldimine via a Schiff base linkage between PLP and the amino acid. CBS can bind either serine or cysteine, CSE can bind cysteine or homocysteine, while CAT/AAT can bind aspartate or cysteine sulfinic acid (CSA) in addition to cysteine at this position. **C.** CBS has a regulatory heme cofactor that is ligated by His65 and Cys52 (human protein numbering). One electron reduction to the ferrous state promotes binding of exogenous ligands such as CO or NO leading to inactive enzyme. The heme harbors nitrite reductase activity and forms nitrosyl heme, which is 5-coordinate. The broken line to His65 indicates that this residue serves as a ligand when CO but not when NO is bound. The ferrous nitrosyl and ferrous carbonyl forms of CBS are readily converted to the ferric state in the presence of O₂.

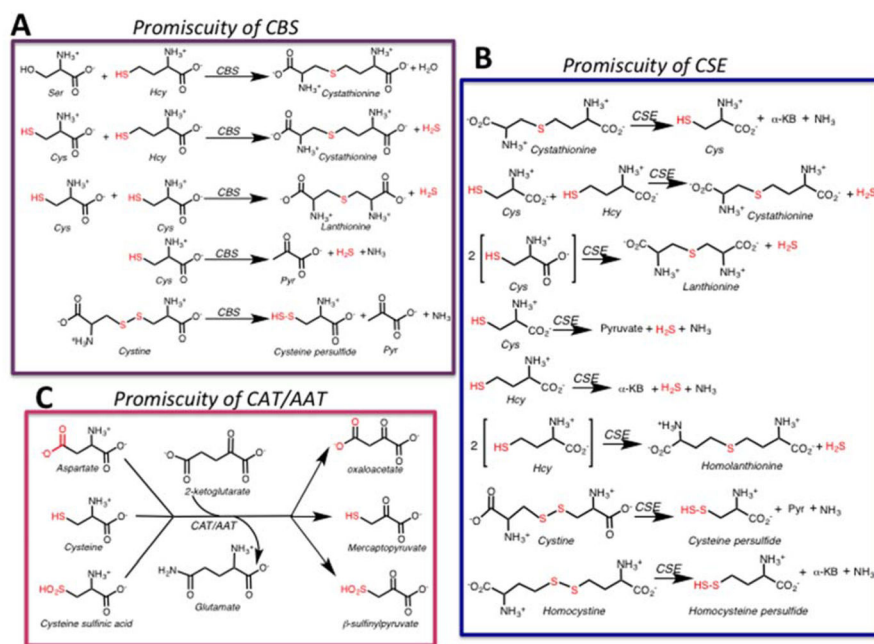


Figure 2. Promiscuity of PLP enzymes involved in H₂S synthesis. H₂S and persulfide-generating reactions catalyzed by the transsulfuration pathway enzymes CBS (**A**) and CSE (**B**). Reactions catalyzed by CAT/AAT (**C**). Pyr and α -KB denote pyruvate and α -ketobutyrate respectively.

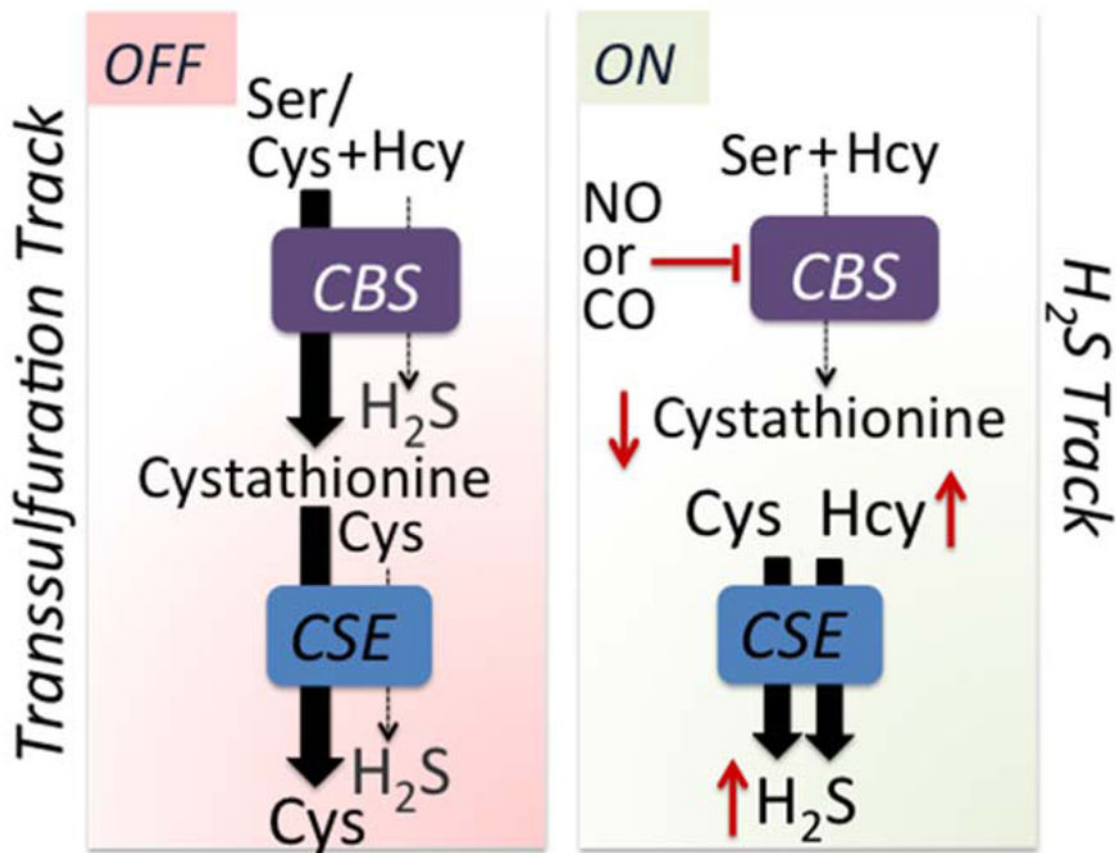


Figure 3.

Heme-dependent metabolic track switching. The canonical transsulfuration track operates when the heme in CBS is coordinated by its endogenous ligands and serine, which is more abundant than cysteine and binds with higher affinity, competes effectively for the active site. The product, cystathionine, is then converted by CSE to cysteine. The enzymes switch metabolic tracks when ferrous CBS binds either NO or CO, inhibiting activity, which leads to an increase in homocysteine and a decrease in cystathionine. Under these conditions, H₂S synthesis from cysteine, which is catalyzed by CSE, is promoted. The red up and down arrows denote changes in metabolite levels.